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Hari et al.

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[54] AMINE SALTS OF AZO PIGMENTS BASED ON PYRAZOLONE DERIVATIVES

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27, 1987, abandoned.

[30] Foreign Application Priority Data

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[56] References Cited

U.S. PATENT DOCUMENTS

839,605	12/1906	Kuchenbecker	534/784
901,675	10/1908	Boniger	534/784
2,490,703	12/1949		534/728
2,842,537	7/1958	Strobel et al	534/784
3,671,254	6/1972	Dostes	534/728 X
3,770,638	11/1973	Checkak	534/728 X
3,802,836	4/1974	Speck	534/784 X
4,388,115	6/1983	Sugiuyama et al	534/793 X
4,594,411	6/1986	Henning	534/784
4,992,495	2/1991	Hari et al	524/106

FOREIGN PATENT DOCUMENTS

0225553 6/1987 European Pat. Off. . 2616981 10/1977 Germany .

763555 12/1956 United Kingdom . 1286770 8/1972 United Kingdom . 1410862 10/1975 United Kingdom . 1490139 10/1977 United Kingdom .

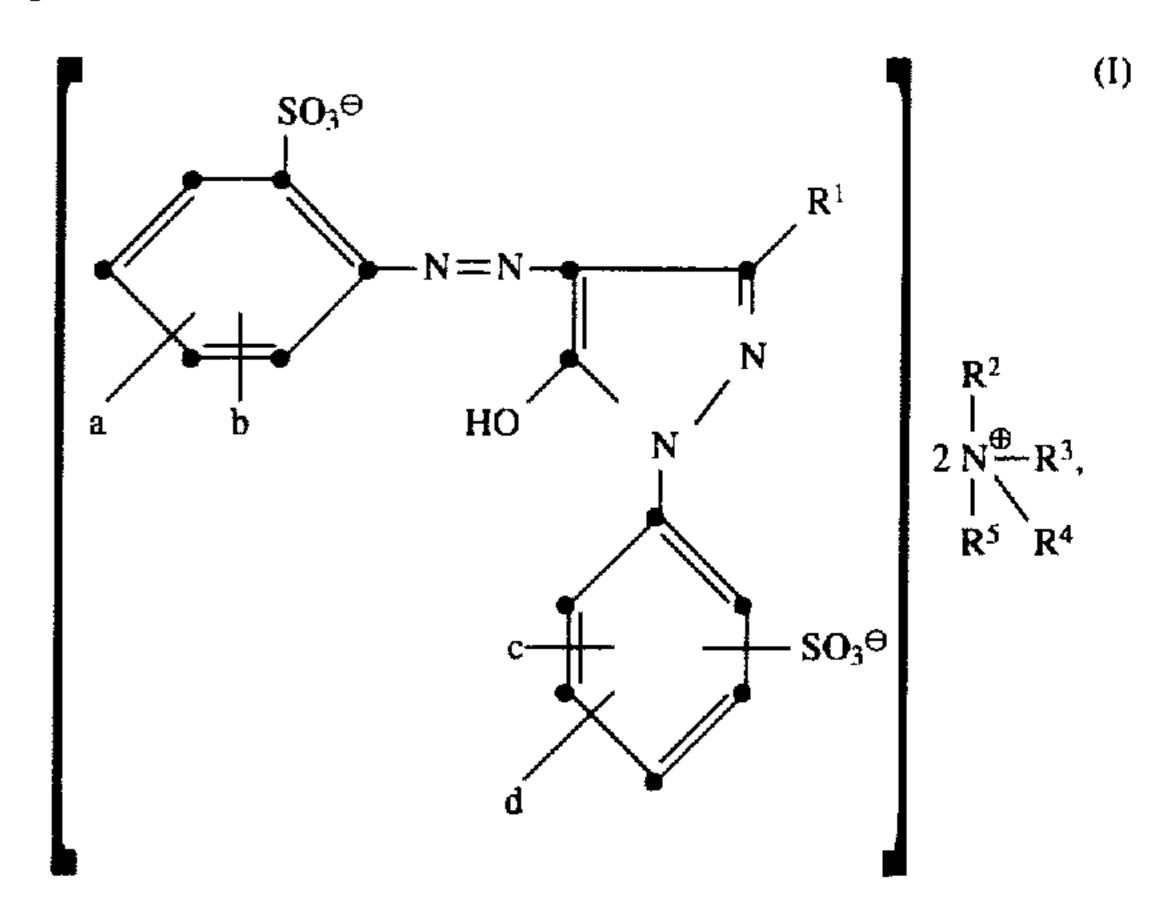
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[57]

[11] E

ABSTRACT

Compounds of formula



wherein a and b are each independently of the other halogen, $-CH_3$, $-C_2H_5$, $-C_2H_5$ or the group of formula -NHCO-C₁-C₄alkyl, c and d are each independently of the other hydrogen, halogen, —CH₃, —C₂H₅, —OCH₃, $--OC_2H_5$ or the group of formula $--NHCO--C_1-C_4$ alkyl, and R^1 is $-CH_3$, phenyl, $-COOCH_3$, $-COOC_2H_5$, $-NHCO-C_1-C_4$ alkyl or $-CONH-C_1-C_4$ alkyl, and R², R³, R⁴ and R⁵ are each independently hydrogen, C_1 - C_4 alkyl, unsubstituted or substituted by a chlorine atom or a —OH group, or are C_7 - C_9 aralkyl, C_6 - C_8 cycloalkyl, phenyl or phenyl which is substituted by 1 to 3 halogen atoms or methyl or methoxy groups, or is a heterocyclic radical of the benzimidazole, benzoxazole, benzthiazole or pyridine series, or R³ and R⁴, together with the linking nitrogen atom, are a pyrrolidine, imidazolidine, piperidine, piperazine or morpholine radical, or R³, R⁴ and R⁵, together with the linking nitrogen atom, are a pyrrole, pyridine, picoline, pyrazine, quinoline or isoquinoline radical, are most suitable for coloring high molecular organic material.

5 Claims, No Drawings

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This is a divisional of application Ser. No. 207,414, filed on Jun. 16, 1988, now U.S. Pat. No. 4,992,495, issued on 10 Feb. 12, 1991, which is in turn a continuation of application Ser. No. 020,059, filed on Feb. 27, 1987, now abandoned.

The present invention relates to novel amine salts of azo compounds, to the preparation thereof, and to the use thereof for colouring high molecular organic material.

Amine salts of azo compounds are known in the art. Thus, for example, ammonium salts of sulfonated azo dyes of the pyrazolone series are disclosed in European patent application 0 073 972. Such compounds are used as pigments for colouring plastics materials.

Specifically, the present invention relates to compounds of formula

wherein a and b are each independently of the other halogen, $-CH_3$, $-C_2H_5$, $-OCH_3$, $-OC_2H_5$ or the group of formula —NHCO—C₁-C₄alkyl, c and d are each independently of the other hydrogen, halogen, --CH₃, --C₂H₅, 40 -OCH₃, -OC₂H₅ or the group of formula -NHCO- C_1-C_4 alkyl, and R^1 is $-CH_3$, phenyl, $-COOCH_3$, $-COOC_{25}$, $-NHCO-C_1-C_4$ alkyl or -CONH-C₁-C₄alkyl, and R², R³, R⁴ and R⁵ are each independently hydrogen, C₁-C₄alkyl, unsubstituted or substituted by a 45 chlorine atom or a —OH group, or are C_7 – C_9 aralkyl, C₅-C₈cycloalkyl, phenyl or phenyl which is substituted by 1 to 3 halogen atoms or methyl or methoxy groups, or is a heterocyclic radical of the benzimidazole, benzoxazole, benzthiazole or pyridine series, or R³ and R⁴, together with the 50 linking nitrogen atom, are a pyrrolidine, imidazolidine, piperidine, piperazine or morpholine radical, or R³, R⁴ and R⁵, together with the linking nitrogen atom, are a pyrrole, pyridine, picoline, pyrazine, quinoline or isoquinoline radical.

The C₁-C₄alkyl moiety of -NHCO-C₁-C₄alkyl and -CONH-C₁-C₄alkyl radicals may be methyl, ethyl, n-propyl, isopropyl, n-butyl and tert-butyl, with methyl being preferred.

Halogen by itself as a, b, c and d and as substituent R², R³, 60 R⁴ and R⁵ of phenyl is fluorine, chlorine, bromine or iodine, with chlorine or bromine being preferred and chlorine most preferred.

R², R³, R⁴ and R⁵ as C₁-C₄alkyl may be methyl, ethyl, n-propyl, isopropyl, n-butyl and tert-butyl.

C₁-C₄Alkyl substituted by a chlorine atom or a —OH group may be β -chloroethyl or 2-hydroxyethyl.

R², R³, R⁴ and R⁵ as C₇-C₉aralkyl may be benzyl or ethylphenyl.

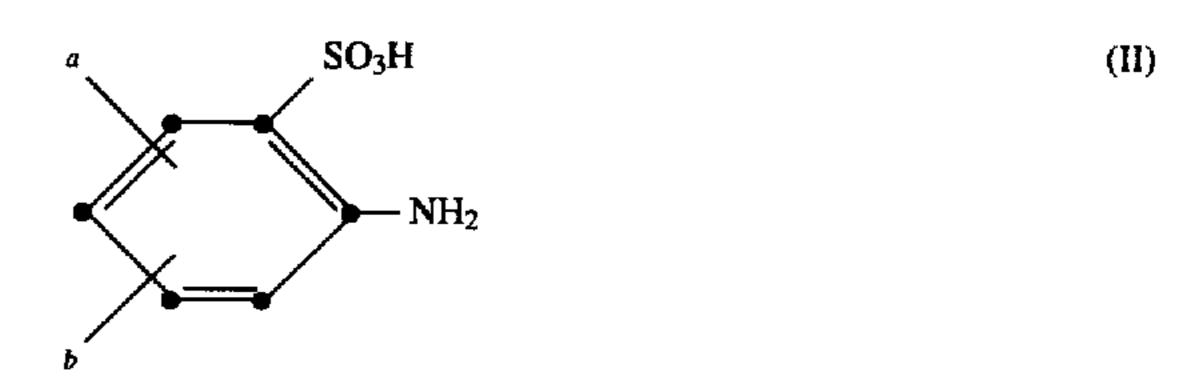
C₅-C₈Cycloalkyl may be cyclopentyl, cyclohexyl, cycloheptyl or cyclooctyl.

Heterocyclic radicals of the benzimidazole, benzoxazole, benzthiazole and pyridine series are e.g. radicals derived from 5-amino-benzimidazole, 5-amino-2-methylbenzimidazole, 5-aminobenzoxazole, 5-aminobenzthiazole, 2-amino-6-ethoxybenzthiazole, 2-aminobenzimidazole, and 2-, 3- or 4-aminopyridine.

Particularly preferred compounds of formula I are those wherein a and b are each independently of the other —Cl, —Br, —CH₃ or —OCH₃, c and d are each independently of the other hydrogen, —Cl, —Br, —CH₃ or —OCH₃, R¹ is -CH₃, -COOC₂H₅, -CONHCH₃ or -NHCOCH₃, and R², R³, R⁴ and R⁵ are each independently hydrogen or $-CH_3$.

The most preferred compounds of formula I are those wherein a and b are each independently of the other —Cl or -CH₃, c and d are each independently of the other hydrogen, —Cl or —CH₃, R^1 is —CH₃ and R^2 , R^3 , R^4 and R^5 are hydrogen.

The compounds of formula I are prepared, for example, by reacting a diazo compound of an amine of formula II



with a coupling component of formula III

$$R^1$$
 N
 N
 SO_3H
 $(IIII)$

under reactions which are known per se, and subsequently treating the reaction product, which is preferably in the form of the free disulfonic acid, with a compound of formula IV

$$N^{\oplus}(R^2)(R^3)(R^4)(R^5)$$
 (IV)

in the form of its chloride or sulfate, in which formulae II, III and IV above a, b, c, d, R¹ and R² to R⁵ are as defined above.

Instead of the chloride or sulfate of this compound, it is also possible to use the free compound.

A variant of the process comprises carrying out the diazotisation and coupling steps without using an inorganic acid by mixing the amine of formula II and the coupling component of formula III, in water or in an organic solvent, with the stoichiometric amount of a nitrite, e.g. NaNO₂, and converting the resultant azo compound into the compound of formula I by conventional methods. Examples of amines of formula II are: 2-amino-3-methyl-5-chlorobenzenesulfonic acid, 2-amino-3,4-dichlorobenzenesulfonic acid, 2-amino-3, 5-dichlorobenzenesulfonic acid, 2-amino-3,5-dimethylbenzenesulfonic acid, 2-amino-4,5-dimethylbenzenesulfonic acid, 2-amino-4-chloro-5-acetylaminobenzenesulfonic acid, 2-amino-4-methoxy-5-acetylaminobenzenesulfonic 2-amino-4-chloro-5-methylbenzenesulfonic acid, 2-amino-4-methyl-5-chlorobenzenesulfonic acid, 2-amino-4,5dichlorobenzenesulfonic acid, 2-amino-4-chloro-5-methoxybenzenesulfonic acid.

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Examples of coupling components of formula III are:

- 1-(2'-sulfophenyl)-3-methyl-5-pyrazolone;
- 1-(4',5'-dichloro-2'-sulfophenyl)-3-methyl-5-pyrazolone,
- 1-(3'-sulfophenyl)-3-methyl-5-pyrazolone,
- 1-(6'-chloro-3'-sulfophenyl)-3-methyl-5-pyrazolone,
- 1-(4'-sulfophenyl-3-methyl-5-pyrazolone,
- 1-(2'-chloro-6'-methyl-4'-sulfophenyl)-3-methyl-5-pyrazolone,
- 1-(2'-methyl-4'-sulfophenyl)-3-methylpyrazolone,
- 1-(2',5'-dichloro-4'-sulfophenyl)-3-methylpyrazolone,
- 1-(3'-sulfophenyl)-3-carbomethoxy-5-pyrazolone,
- 1-(2',5'-dichloro-4'-sulfophenyl)-3-carbomethoxy-5-pyra-zolone,
- 1-(3'-sulfophenyl)-3-carboethoxy-5-pyrazolone,
- 1-(2',5'-dichloro-4'-sulfophenyl)-3-carboethoxy-5-pyra-zolone,
- 1-(3'-sulfophenyl)-3-(N-methylcarbamyl)-5-pyrazolone,
- 1-(3'-sulfophenyl)-3-acetylamino-5-pyrazolone,
- 1-(2',5'-dichloro-4'-sulfophenyl)-3-acetylamino-5-pyra-zolone.

Examples of free compounds of formula IV are: ammo- 20 nia, methylamine, ethylamine, n-propylamine, n-butylamine, cyclohexylamine, benzylamine, aniline, 2-chloroaniline, 2-, 3- or 4-methylamiline, 2,4-dimethylamiline, 2,4,6-trimethylamine, dimethylamine, trimethylamine, tetramethylamine, 2-, 3- or 4-aminopyridine, morpholine, 25 piperidine, 2-, 3- or 4-picoline as well as pyridine.

The compounds of formula I can be used as pigments for colouring high molecular organic material. They can normally be used direct in the form in which they are obtained in their synthesis. Depending on the envisaged utility, the 30 compounds of formula I can be converted into a more opaque or more transparent form.

If it is desired to obtain a more opaque pigment form, then it is usually expedient to effect a thermal aftertreatment in water or an organic solvent, under normal or elevated 35 pressure. It is preferred to use organic solvents such as benzenes which are substituted by halogen atoms, alkyl or nitro groups, for example xylenes, chlorobenzene, o-dichlorobenzene or nitrobenzene, as well as pyridine bases such as pyridine, picoline or quinoline, and also ketones such as cyclohexanone, alcohols such as isopropanol, butanols or pentanols, ethers such as ethylene glycol monomethyl or monoethyl ether, amides such as dimethylformamide or N-methylpyrrolidone, as well as dimethylsulfoxide or sulfolane. The aftertreatment can also be carried out in water, 45 under normal or elevated pressure, in the presence of organic solvents and/or with the addition of surfactants.

Examples of high molecular organic materials which may be coloured with the compounds of formula I are cellulose ethers and esters such as ethyl cellulose, nitrocellulose, 50 cellulose acetate, cellulose butyrate, natural resins or synthetic resins such as polymerisation resins or condensation resins, for example aminoplasts, in particular urea/formal-dehyde and melamine/formaldehyde resins, alkyd resins, phenolic plastics, polycarbonates, polyolefins, polystyrene, 55 polyvinyl chloride, polyamides, polyurethanes, polyesters, rubber, casein, silicone and silicone resins, singly or in mixtures.

The compounds of formula I are particularly suitable for colouring polyvinyl chloride and polyolefins such as polyethylene and polypropylene.

The above high molecular organic compounds may be singly or as mixtures in the form of plastics, metals or of spinning solutions, varnishes, paints or printing inks. Depending on the utility, it is advantageous to use the 65 compounds of formula I as toners or in the form of preparations.

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The compounds of formula I are preferably employed in an amount of 0.01 to 30% by weight, preferably 0.1 to 10% by weight, based on the high molecular organic material to be pigmented.

The pigmenting of the high molecular organic compounds with the compounds of formula I is carried out for example by incorporating such a compound, optionally in the form of a masterbatch, into the substrates using roll mills, mixing or grinding machines. The pigmented material is then brought into the desired final form by methods which are known per se, for example calendering, moulding, extruding, coating, spinning, casting or by injection moulding. It is often desirable to incorporate plasticisers into the high molecular compounds before processing in order to produce nonbrittle mouldings or to diminish their brittleness. Suitable plasticisers are for example esters of phosphoric acid, phthalic acid or sebacic acid. The plasticisers may be incorporated before or after working the compound of formula I into the polymers. To obtain different shades, it is also possible to add fillers or other chromophoric components such as white, coloured or black pigments, in any amount, to the high molecular organic compounds, in addition to the compounds of formula I.

For pigmenting varnishes and printing ink, the high molecular organic materials and the compounds of formula I, together with optional additives such as fillers, other pigments, siccatives or plasticisers, are finely dispersed or dissolved in a common organic solvent or mixture of solvents. The procedure may be such that the individual components by themselves, or also several jointly, are dispersed or dissolved in the solvent and subsequently all the components are mixed.

The colourations obtained, for example, in polyvinyl chloride or polyolefins with the compounds of formula I, have a good allround fastness properties such as good dispersibility, high tinting strength and purity, good fastness to migration, heat, light, and weathering, as well as good hiding power.

In the following Examples, parts and percentages are by weight.

EXAMPLE 1

5.2 g of the azo compound of formula

(obtainable by known methods) are dissolved at room temperature in a mixture of 50 ml of acetic acid and 50 ml of deionised water. To the clear, dark red solution is added, over 2 minutes, a solution of 2.6 g of aniline hydrochloride in 10 ml of distilled water. The resultant orange yellow suspension is stirred for 3 hours at room temperature and filtered. The filter residue is washed with 100 ml of deionised water and vacuum dried at 60° C., affording 5.1 g of a deep orange

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powder of formula

SO₃
$$\Theta$$

N=N
CH₃
 Θ
2NH₃C₆H₅
10
SO₃ Θ

Analysis: theory (in %): C 48.49, H 3.78; Cl 10.22; N 12.11; S 9.24; found (in %): C 48.32; H 3.87; Cl 10.46; N 12.20; S 8.90.

This compound colours polyvinyl chloride and polyeth- 20 ylene in strong, fast yellow shades.

Instead of using aniline hydrochloride, it is possible to use other amine hydrochlorides listed in the following table to give the compounds of formula

$$\begin{bmatrix} SO_3^{\ominus} & CH_3 \\ N & N \end{bmatrix} = \begin{bmatrix} SO_3^{\ominus} & CH_3 \\ N & N \end{bmatrix}$$

in accordance with the process conditions described in this ³⁵ Example.

Example	Amine hydrochloride	A	Shade in soft PVC (0.2% of pigment)
2	ammonium chloride	⊕ NH₄	yellow
3	methylamine hydrochloride	⊕ H ₃ NCH ₃	orange yellow
4	benzylamine hydrochloride	$_{3}^{\oplus}$ CH ₂ C ₆ H ₅	yellow
5	trimethylamine hydrochloride	⊕ HN(CH ₃) ₃	yellow
6	tetramethylammonium chloride	⊕ N(CH ₃) ₄	greenish yellow
7	2,5-dimethylaniline hydrochloride	H ₃ N CH ₃	yellow
8	pyridine hydrochloride	HN \oplus	yellow

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Example	Amine hydrochloride	A	Shade in soft PVC (0.2% of pigment)
9	2-hydroxyethyl hydrochloride	⊕ H ₃ NCH ₂ CH ₂ — OH	yellow
10	cyclohexylamine hydrochloride	H_3N H	yellow
11	morpholine hydrochloride	H ₂ N H O	yellow
12	2-aminopyridine hydrochloride	N \bigoplus NH ₃	yellow
13	6-amino-benzthiazole hydrochloride	H_3N H_3N S H	yellow
14	5-amino-2-methylbenzimidazole hydrochloride	H ₃ N CH ₃	yellow

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EXAMPLE 15

Example 1 and 4.4 g of ammonium chloride are heated to 80° C. in 100 ml of deionised water. The deep yellow suspension is stirred for 4 hours at this temperature, cooled to room temperature and filtered. The filter residue is washed with a small amount of deionised water and vacuum dried at 60° C., affording 5.1 g (94.2% of theory) of a product that is identical structurally and in all properties with the pigment obtained in Example 2. In analogues manner, further compounds of formula I can be prepared starting from an amine of formula II, a coupling component of formula III and an amine of formula IV, as listed in the following table.

Example	Amine of formula II	Coupling component of formula III	Amine of formula IV	Shade in soft PVC (0.2% of pigment)
16	H ₃ C Cl	HO N SO ₃ H	NH ₃	yellowish orange
17	NH ₂ SO ₃ H Cl	HO N SO ₃ H	CH ₃ NH ₂	yellowish orange
18	NH ₂ SO ₃ H CH ₃	HO N SO ₃ H	NH ₃	yellow
19	SO ₃ H CH ₃	HO N SO ₃ H	CH ₃ NH ₂	yellow
20 H	SO ₃ H CH ₃	HO N SO ₃ H	NH ₃	orange yellow

Example	Amine of formula II	Coupling component of formula III	Amine of formula IV	Shade in soft PVC (0.2% of pigment)
21	NH ₂ SO ₃ H OCH ₃	CH ₃	NH ₃	orange yellow
22	CH ₃ O NHCOCH ₃	CH ₃ N SO ₃ H	NH ₃	orange
23	NH_2 H_5C_2 C_1 SO_3H	HO N SO ₃ H	NH ₃	yellow
24	NH ₂ SO ₃ H	HO ₃ S CH ₃	NH ₃	greenish yellow
25	NH ₂ SO ₃ H Cl Cl	HO N CH ₃	NH ₃	greenish yellow

Example Amine of formula II	Coupling component of formula III	Amine of formula IV	Shade in soft PVC (0.2% of pigment)
26 NH ₂ SO ₃ H	HO N SO ₃ H	NH ₃	greenish yellow
27 NH ₂ SO ₃ H	CH ₃ O OCH ₃	NH ₃	greenish yellow
28 NH ₂ SO ₃ H	HO N CI	NH ₃	greenish yellow
NH ₂ SO ₃ H	HO N SO ₃ H	NH ₃	yellow
30 NH ₂ SO ₃ H	HO N SO ₃ H	CH ₃ NH ₂	orange

Example	Amine of formula II	Coupling component of formula III	Amine of formula IV	Shade in soft PVC (0.2% of pigment)
31	NH ₂ SO ₃ H	HO NHCOCH ₃	NH ₃	yellow

EXAMPLE 32

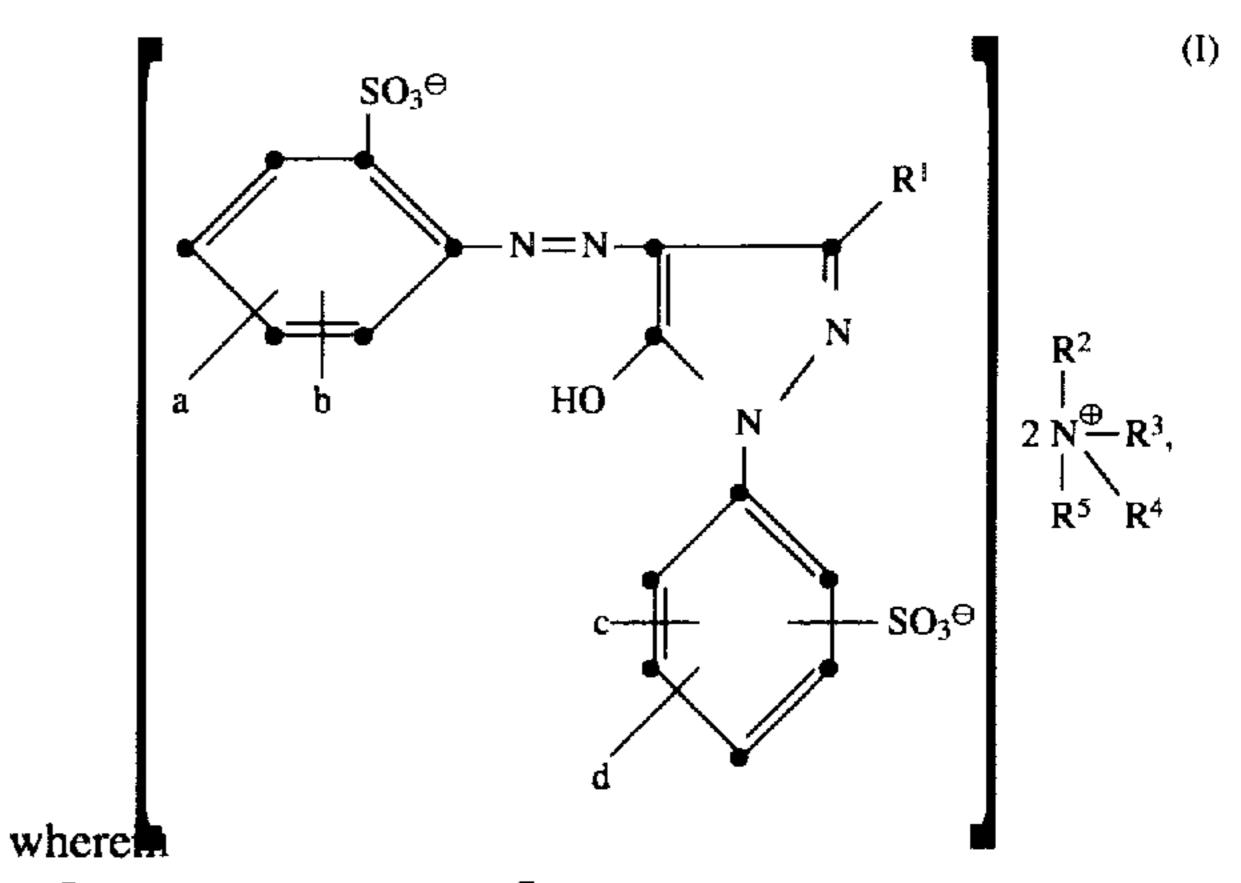
13 g of 93% 2-amino-4,5-dichlorobenzenesulfonic acid are dissolved in 90 ml of deionised water and 5.7 ml of 25% ammonia solution. Then 27.5 ml of 23.5% hydrochloric acid are added to the clear, reddish violet solution. The resultant thick, white suspension is cooled to 5° C. and 12.5 ml of 4N 25 sodium nitrite solutions are then added over 5 minutes. Then 15.5 g of 90.1% 1-(3'-sulfophenyl)-3-methyl-5-pyrazolone in the form of a fine suspension in 100 ml of deionised water are added dropwise over 5 minutes to the slightly beige- 30 coloured suspension. The pH of the resultant pale yellow suspension is adjusted to 6.5 with 15.5 ml of 25% ammonia solution. The now orange-coloured suspension is stirred for 3 hours, during which time the temperature of the suspen- 35 sion rises from 5° to 22° C. When coupling is complete, this suspension is heated to 80° C., stirred for 6 hours at this temperature, then cooled to room temperature and filtered. The filter residue is washed with a small amount of deionised water and vacuum dried at 60° C., affording 26.9 g (99.5% of theory) of a product which is identical with that of Example 2.

EXAMPLE 33

5.2 g of the azo compound used in Example 1 (disulfonic acid) are dissolved at 70° C. in 150 ml of deionised water. Then 1.85 g of aniline are added to the clear, red solution. The resultant orange suspension is stirred for 2 hours 55 without heating and then filtered. The residue is washed with 100 ml of deionised water and vacuum dried at 60° C., affording 5.6 g (80.8% of theory) of an orange product that is identical in structure and all properties with the pigment 60 obtained in Example 1.

What is claimed is:

1. A compound of formula I



[a and b are each —Cl,] a is —Cl and b is —Cl or —CH₃, c and d are each independently of the hydrogen, —Cl or —CH₃,

R¹ is —CH₃, and

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R², R³, R⁴ and R⁵ are each hydrogen.

2. A compound according to claim 1 wherein c and d are each hydrogen.

3. A compound according to claim 2 wherein a is —Cl and b is —CH₃.

4. A compound according to claim 3 of the formula

5. A compound according to claim 1, wherein a and b are each —Cl.