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[45] Reissued Nov. 14, 1978

[54]	4-HYDROXY-3-(3-ISOXAZOLYLCAR-			
	BAMOYL)-2H-1,2-BENZOTHIAZINE			
	1,1-DIOXIDES AND PROCESS FOR THEIR			
	PRODUCTION			

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[21] Appl. No.: 840,929

[22] Filed: Oct. 11, 1977

Related U.S. Patent Documents

Reissue of:

[64] Patent No.: 3,822,258

Issued: Jul. 2, 1974

Appl. No.: 356,026 Filed: Apr. 30, 1973

U.S. Applications:

[60] Continuation-in-part of Ser. No. 333,821, Feb. 20, 1973, Pat. No. 3,816,828, which is a division of Ser. No. 119,967, Mar. 1, 1971, Pat. No. 3,787,324.

[56] References Cited

U.S. PATENT DOCUMENTS

3,646,021	2/1972	Zinnes et al	14****	544/49
3,704,298	11/1972	Zinnes et al	******	544/49
3.714.155	1/1973	Zinnes et al	******************	544/49

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[57] ABSTRACT

Compounds having the following structural formula are disclosed:

OH O R2
CNH N-R1 N O R3

wherein R_1 is hydrogen or methyl and R_2 , and R_3 are hydrogen or alkyl. These compounds are useful as anti-inflammatory agents, antipyretics, analgesics.

2 Claims, No Drawings

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4-HYDROXY-3-(3-ISOXAZOLYLCARBAMOYL)-2H-1,2-BENZOTHIAZINE 1,1-DIOXIDES AND PROCESS FOR THEIR PRODUCTION

Matter enclosed in heavy brackets [] appears in the original patent but forms no part of this reissue specification; matter printed in italics indicates the additions made by reissue.

This application is a continuation-in-part of our copending application U.S. Ser. No. 33,821, filed Feb. 20, 1973, which in turn is a divisional of U.S. Ser. No. 119,967 filed Mar. 1, 1971, now abandoned.

This application is a reissue of U.S. Pat. No. 3,822,858, filed as U.S. Ser. No. 356,026 on Apr. 30, 1973, which is a continuation-in-part of U.S. Ser. No. 333,821, filed Feb. 20, 1973, now U.S. Pat. No. 3,816,828, which in turn is a divisional of U.S. Ser. No. 119,967, filed Mar. 1, 1971, now U.S. Pat. No. 3,787,324.

The present invention relates to 4-hydroxy-3-(3-isox-azolylcarbamoyl)-2H-1,2-benzothiazine, 1,1-Dioxides having the following structural formula:

wherein R₁ is hydrogen or methyl and R₂, and R₃ are hydrogen or alkyl having from 1 to 7 carbon atoms, for example, methyl, ethyl, propyl, isopropyl, butyl, isobutyl and the like.

The present invention also includes within its scope the corresponding alkali metal, alkaline earth and amine salts.

The compounds of this invention are useful as anti-inflammatory agents, antipyretics, and analgesics in several mammalian species. When administered orally to rats at a dose of 10-200 mg./kg., they are able to cause reduction in swelling of the paw induced by injection into the foot pads of an irritant such as carrageenin. Therapeutically or prophylactically administered orally at a dose of 15-200 mg./kg., the compounds inhibit adjuvant induced polyarthritis in the rat. Oral doses of 25-100 mg./kg. are sufficient to inhibit yeast induced hyperthermia in the rat. At oral doses of 25-200 mg./kg. they exhibit a significant analgesic effect as 50 determined by the phenylquinone writhing procedure in mice.

Generally speaking, these compounds are indicated in conditions such as pain resulting from arthritis, bursitis and the like. A daily dosage regimen of about 0.5 grams 55 to about 2 grams in several divided doses is recommended for a mammal weighing about 70 kg. body weight to relieve the pain and swelling associated with these conditions. These compounds are administered either orally or by injection.

In order to use these compounds, they are formulated into dosage forms such as tablets or syrups by blending with an inert pharmaceutical carrier such as lactose or simple syrup by methods well known to the pharmacists art. For injectionable dosage forms, they are formulated 65 with vehicles such as water, peanut oil, sesame oil, and the like. In these dosage forms, the active ingredient is from about 0.5 grams to 1 gram per dosage unit.

According to the present invention, the above compounds are prepared by a reaction scheme as follows:

II $R_4 = alkyl$

$$\begin{array}{c|c} H_2N & \longrightarrow & I + R_4OH \\ \hline & N & O & R_3 & \longrightarrow & I \end{array}$$

Generally speaking, starting compound II is refluxed with 3-aminoisoxazole III in an inert solvent such as xylene. In a preferred embodiment of the present invention, the reactants are refluxed in the presence of a molecular sieve which promote the desired reaction by removing the alcohol which is formed as a by-product. The use of molecular sieves results in a more convenient and practical process in that lengthy distillation to remove the alcohol is no longer required. Typically the reaction is carried out in a Soxhlet apparatus with the molecular sieves contained in the thimble.

Examples of the molecular sieves, which can be used in this process, are commercially available molecular sieves under the trade name Linde type 4A molecular sieve from Matheson Coleman & Bell Company. See also Fieser & Fieser, "Reagents for Organic Synthesis," Vol. 1.

The starting material II are known compounds and they are prepared in accordance with the description in U.S. Pat. No. 3,501,466.

Starting compound 3-aminoisoxazole is prepared in accordance with the description in Chem. Pharm. Bull. 14, 1277 (1969). The starting compound 3-amino-5-methylisoxazole is commercially available from Hoffmann-La Roche. It is prepared in accordance with the description set forth in the Netherlands Pat. No. 6,511,924, issued Mar. 15, 1966.

The corresponding salts with metal or with amine are prepared by treating the above compounds with the desired base e.g. sodium alkoxide, potassium alkoxide, sodium hydroxide, potassium hydroxide, calcium hydroxide, pyrrolidine and the like by conventional procedures.

According to a further feature of this invention, a 2-unsubstituted derivative of structure IV is reacted with sodium hydride in dimethylformamide and the resulting anion is treated with methyl iodide. Under the conditions described, methylation takes place solely at the 2-position. Dilution of the reaction mixture with water results in the precipitation of the crystalline product I in high yield and a high state of purity. This reaction is as follows:

$$\begin{array}{c|c}
OH \\
\hline
CONH \\
NH \\
N \\
O
\end{array}$$

$$\begin{array}{c}
1. \text{ NaH} \\
\hline
2. \text{ CH}_3\text{ IV}
\end{array}$$

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In order to further illustrate the practice of this invention, the following examples are included. All temperatures are given in degrees centigrade.

EXAMPLE I

4-Hydroxy-3-(5-methyl-3-isoxazolylcarbamoyl)-2-methyl-2H-1,2-benzothiazine 1,1-Dioxide

A mixture of 40.5 g. (0.15 mole) of 3-carbethoxy-4-hydroxy-2-methyl-2H-1,2-benzothiazine 1,1-dioxide, 20.6 g. (0.21 mole) of 3-amino-5-methylisoxazole, and 2500 ml. of xylene was refluxed for 24 hr. in a Soxhlet apparatus, the thimble of which contained 60 g. of 35 Linde type 4A molecular sieve. The mixture was cooled to 25° and the resulting crystalline precipitate was collected and washed with ether to give 44 g. of crude product. Recrystallization from 1600 ml. of 1,4-dioxan gave 34.7 of material, m.p. 265-271° dec.

Anal.—Calcd. for C₁₄H₁₃N₃O₅S: C, 50.14; H, 3.91; N, 12.53; S, 9.56. Found: C, 50.33; H, 3.88; N, 12.30; S, 9.49.

The corresponding sodium salt was prepared by treating the above compound with sodium hydroxide 45 which has the following structural formula:

The sodium salt after recrystallization from ethanol has 55 m.p. 270-272° dec.

Anal.—Calcd. for $C_{14}H_{12}N_3O_5SNa$: C, 47.06; H, 3.39; N, 11.76; S, 8.97. Found: C, 47.09; H, 3.69; N, 11.50; S, 8.81.

EXAMPLE II

4-Hydroxy-3-(3-isoxazolylcarbamoyl)-2-methyl-2H-1,2-benzothiazine 1,1-Dioxide

A mixture of 4.15 g. (0.0155 mole) of 3-carbethoxy-4-hydroxy-2-methyl-2H-1,2-benzothiazine 1,1-dioxide, 1.3 g. (0.0155 mole) of 3-aminoisoxazole, and 500 ml. of xylene was refluxed for 24 hr. in a Soxhlet apparatus, the thimble of which contained 10 g. of Linde 4A molecular sieve. The mixture was cooled to 25° and the resulting crystalline precipitate was collected and washed with ether to give 3.2 g. of product, m.p. 235-240° dec. Recrystallization from 175 ml. of ethyl acetate gave 1.6 g. of material, m.p. 237-240° dec.

Anal.—Calcd. for $C_{13}H_{11}N_3O_5\bar{S}$: C, 48.60; H, 3.45; N, 13.08; S, 9.98. Found: C, 48.77; H, 3.44; N, 12.86; S, 9.85.

EXAMPLE III

4-Hydroxy-3-(5-methyl-3-isoxazolylcarbamoyl)-2H-1,2-benzothiazine 1,1-Dioxide

A mixture of 15.3 g. (0.06 mole) of 3-carbethoxy-4-30 hydroxy-2-methyl-2H-1,2-benzothiazine 1,1-dioxide, 5.9 g. (0.06 mole) of 3-amino-5-methylisoxazole, and 800 ml. of xylene was refluxed for 24 hr. in a Soxhlet apparatus, the thimble of which contained 20 g. of Linde type 4A molecular sieve. The mixture was cooled to 25° and the resulting precipitate was collected and washed with ether to give 11.5 g. of crude product, m.p. 242-248° dec. Recrystallization from 300 ml. of 1,4-dioxan gave 11.2 g. of crystalline product, m.p. 254-257° dec.

Anal.—Calcd. for $C_{13}H_{11}N_3O_5S$: C, 48.60; H, 3.45; N, 13.08; S, 9.98. Found: C, 48.67; H, 3.44; N, 12.91; S, 10.23.

EXAMPLE IV

To a mixture of 0.01 mole (as a 57% dispersion in mineral oil) and 3.21 g. (0.01 mol) of Compound IV was added 20 ml. of dry dimethylformamide. The mixture was stirred for one hour during which hydrogen evolved and the dark brown anion formed. A solution of 1.42 g. (0.01 mol) of methyl iodide in dimethylformamide was added, the reaction mixture was stirred at 60° for 4 hours, and then at 25° for 16 hours. It was diluted with 20 ml. of water and stirred for one hour. The resulting precipitate was collected, washed with two 10

ml. portions of dimethylformamide and one 20 ml. portion of methanol and air dried. Yield: 2.7 g. of white solid, m.p. 259-262° dec.

We claim:

[1. A compound of the formula:

wherein R₁ is hydrogen or methyl and R₂, and R₃ are hydrogen or alkyl having 1 to 7 carbon atoms, or its corresponding alkali metal, akaline earth and amine salts.

[2. A compound according to claim 1 which is 4-Hydroxy-3-(5-methyl-3-isoxazolylcarbamoyl)-2-methyl-2H-1,2-benzothiazine 1,1-Dioxide.]

[3. A compound according to claim 1 which is 4-25 Hydroxy-3-(3-isoxazolylcarbamoyl)-2-methyl-2H-1,2-benzothiazine 1,1-Dioxide.]

[4. A compound according to claim 1 which is 4-30 Hydroxy-3-(5-methyl-3-isoxazolylcarbamoyl)-2H-1,2-benzothiazine 1,1-Dioxide.]

[5. A process for production of the compound of claim 1 which comprises refluxing together a com- 35 III: pound of Structure II:

$$\begin{array}{c} OH \\ CO_2R_4 \\ SO_2 \end{array}$$

in which R₄ is lower alkyl with a compound of Structure III:

$$R_2$$
 R_2
 R_3
 R_3

in an inert solvent in the presence of a molecular sieve adapted to remove the by-product alcohol formed.]

[6. A process for the production of the compound of claim 2 which comprises treating a compound of the formula:

with sodium hydride in dimethylformamide, treating the resulting anion with methyl iodide; adding water to the reaction mixture; and recovering compound I as a precipitate.

7. A process for the production of a compound of the formula I:

$$CNH \longrightarrow R_{I}$$

$$SO_{I}$$

$$R_{I}$$

$$R_{I}$$

$$R_{I}$$

wherein R_1 is hydrogen or methyl and R_2 and R_3 are hydrogen or alkyl having 1 to 7 carbon atoms or its corresponding alkali metal, alkaline earth and amine salts; which comprises refluxing together a compound of the formula II:

in which R₄ is lower alkyl with a compound of the formula III:

$$H_{\bullet}N$$
 R_{\bullet} R_{\bullet} R_{\bullet}

in an inert solvent in the presence of a molecular sieve adapted to remove the by-product alcohol formed.

8. A process for the production of 4-hydroxy-3-(5-meth-yl-3-isoxazolylcarbamoyl)-2-methyl-2H-1,2-benzothiazine 1,1-dioxide which comprises treating a compound of the formula IV:

with sodium hydride in dimethylformamide, treating the resulting anion with methyl iodide; adding water to the reaction mixture; and recovering 4-hydroxy-3-(5-methyl-3-isoxazolylcarbamoyl)-2-methyl-2H-1,2-benzothiazine 1,1-dioxide as a precipitate.

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