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(54) **MCRALY BOND COAT WITH ENHANCED YTTRIUM LAYER**

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See application file for complete search history.

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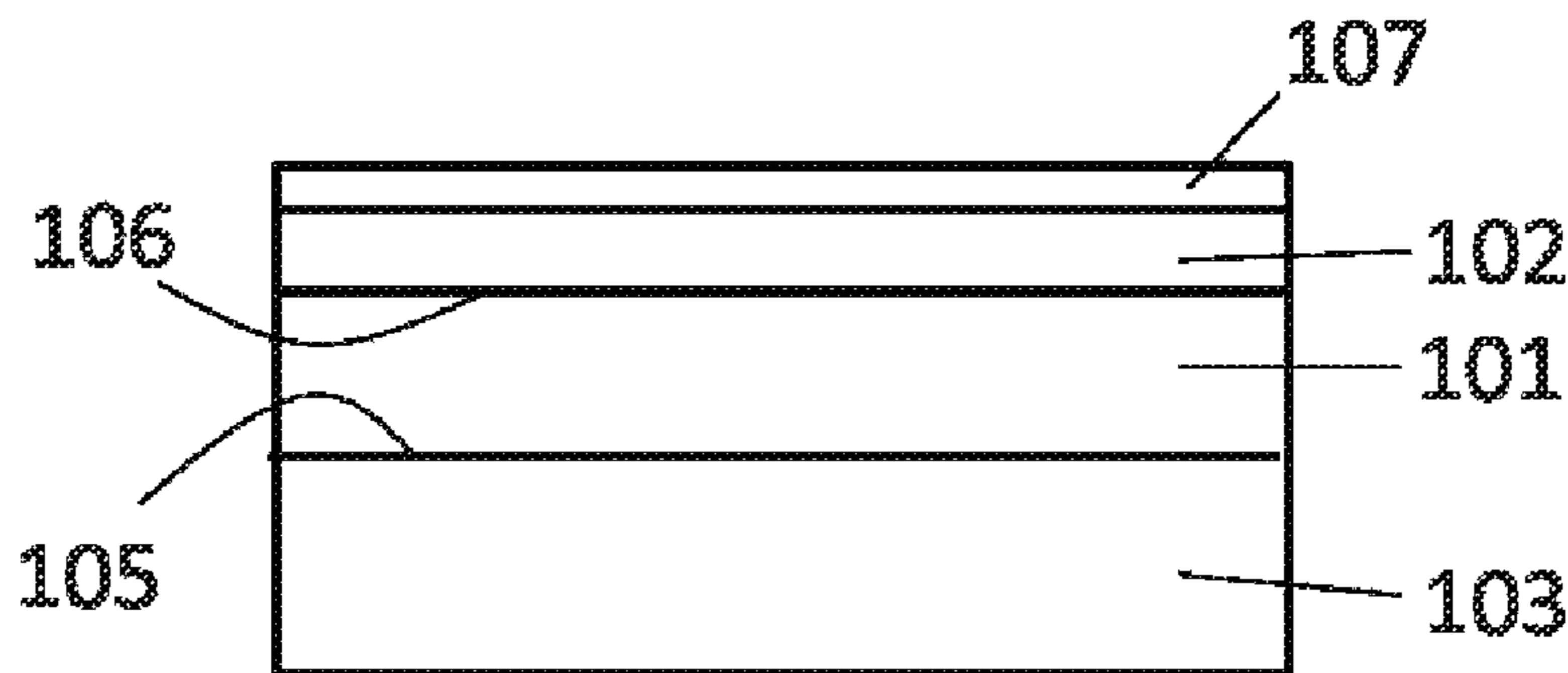
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(57) **ABSTRACT**

One or more embodiments relates to an MCrAlY bond coat comprising an MCrAlY layer in contact with a Y—Al₂O₃ layer. The MCrAlY layer is comprised of a γ-M solid solution, a β-MAl intermetallic phase, and Y-type intermetallics. The Y—Al₂O₃ layer is comprised of Yttrium atoms coordinated with oxygen atoms comprising the Al₂O₃ lattice. Both the MCrAlY layer and the Y—Al₂O₃ layer have a substantial absence of Y—Al oxides, providing advantage in the maintainability of the Yttrium reservoir within the MCrAlY bulk. The MCrAlY bond coat may be fabricated through application of a Y₂O₃ paste to an MCrAlY material, followed by heating in a non-oxidizing environment.

18 Claims, 2 Drawing Sheets



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4/085 (2013.01); *B32B 15/18* (2013.01); *Y10T*
428/12611 (2015.01); *F01D 5/288* (2013.01);
F01D 5/28 (2013.01); *B32B 15/04* (2013.01);
B32B 15/20 (2013.01); *Y10T 428/12618*
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428/12937 (2015.01); *Y10T 428/12944*
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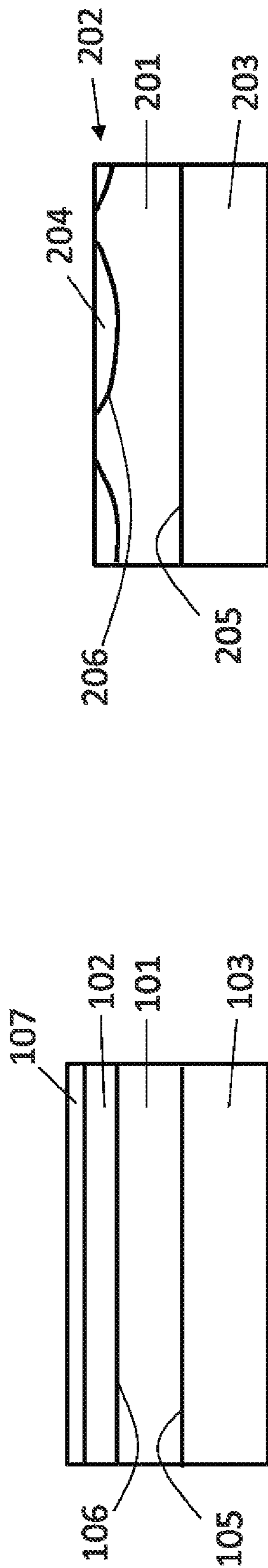


FIG. 1

FIG. 2

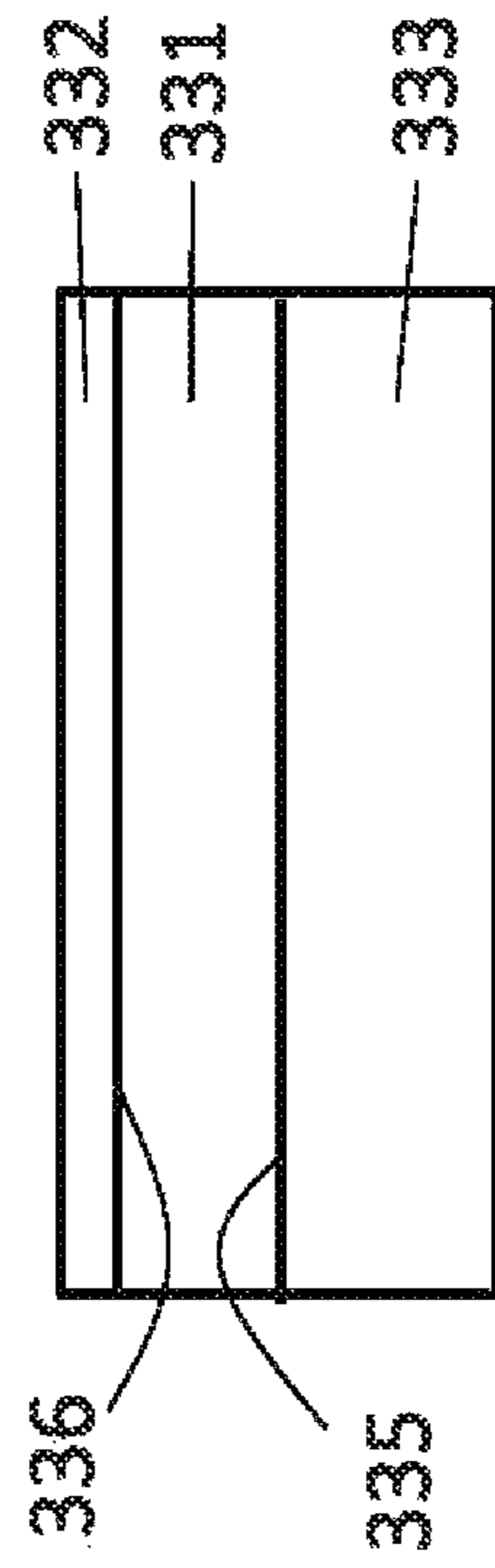


FIG. 3

MCRALY BOND COAT WITH ENHANCED YTTRIUM LAYER

RELATION TO OTHER APPLICATIONS

This, patent application is a divisional of and claims priority from non-provisional patent application 13/363,486 filed Feb. 1, 2012, which is hereby incorporated by reference.

GOVERNMENT INTERESTS

The United States Government has rights in this invention pursuant to the employer-employee relationship of the Government to the inventors as U.S. Department of Energy employees and site-support contractors at the National Energy Technology Laboratory.

FIELD OF THE INVENTION

One or more embodiments relates to an MCrAlY bond coat comprised an MCrAlY layer in contact with a Y—Al₂O₃ layer. The MCrAlY composition is comprised of Yttrium and Aluminum, and further comprised of greater than 50 weight percent (wt. %) of a metal M, where the metal M is Ni, Co, Fe, or combinations thereof, and the MCrAlY layer is comprised of a γ -M solid solution, a β -MAl intermetallic phase, and Y-type intermetallics. The Y—Al₂O₃ layer is comprised of Yttrium atoms coordinated with oxygen atoms comprising the Al₂O₃ lattice. Both the MCrAlY layer and the Y—Al₂O₃ layer have a substantial absence of Y—Al oxides, providing advantage in the maintainability of the Yttrium reservoir within the MCrAlY bulk. A method of fabrication is additionally disclosed.

BACKGROUND

MCrAlY type coatings are commonly used as overlay coatings and bondcoats for the protection of components against high-temperature oxidation and corrosion. During exposure to high temperatures, an oxide scale forms on the MCrAlY surface, becoming part of the original system. This thermally grown oxide is largely comprised of alumina and often represents the starting point for failure. Correspondingly, the growth rate and adhesion of the oxide scale and the aluminum depletion in the bondcoat are among the determining factors for the useful lifetime of an MCrAlY bond coated article.

At long term exposures, the alumina scales are prone to spallation. The scale spallation is a common event under temperature cycling conditions due to the thermal expansion mismatch between the oxide and the metallic substrate, and failure is generally related to scale thickness, growth rate and adhesion. The two latter parameters are crucially affected by the addition of Yttrium, as well as other reactive elements such as Lanthanum, Zirconium, and Hafnium. These reactive elements will commonly constitute some percentage, typically less than 0.5 wt. % to the base MCrAl-composition. These reactive elements have been found to greatly improve alumina scale adherence and significantly increase the critical scale thickness for spallation. However, this beneficial effect is highly sensitive to the Y-concentration. In some cases, optimum oxidation and improved adherence is obtained by low concentrations of Y in the coating (around 0.1 wt. %), whereas a few tenths of a weight percent more (over 0.5 wt. %) can lead to accelerated oxidation and detrimental oxide morphology. This implies that relatively low amounts of Yttrium are desired in the coating. Issues have arisen as a result of this Yttrium content sensitivity however, because

Yttrium can be tied up by Y—Al oxides resulting from typical fabrication processes, and the effective Yttrium concentration of the MCrAlY coating may significantly deviate from that of the original MCrAlY source material.

Depletion of the Yttrium reservoir through the formation of excessive Y—Al oxides will therefore promote TGO spallation in service. Such depletion results in deviation from the optimum Yttrium content designed to obtain a compromise between the beneficial effect on oxide adherence without extensively enhancing the growth rate of the oxide. Defining and maintaining this optimum yttrium content it is absolutely necessary to take into account the actual Yttrium reservoir in the MCrAlY coating. See e.g., Toscano et al., "Parameters affecting TGO growth and adherence on MCrAlY-bond coats for TBC's," *Surface & Coatings Technology* 201 (2006), among others. It would be advantageous to provide an MCrAlY bond coat wherein the formation of Y—Al oxides is suppressed, and the Yttrium reservoir of the final heat treated composition could be largely defined by the Yttrium concentration of the original MCrAlY source material. Such an MCrAlY bond coat would mitigate variations in the Y-distribution and the reservoir of metallic Y in MCrAlY coatings and avoid the significant variations in the alumina scale growth rate and adherence which accrue from nominally equivalent MCrAlY bond coat compositions.

Additionally, a common feature of the oxide morphology in MCrAlY bond coats is the formation of Y₂O₃ compounds near or at the metal/oxide interface, known as pegs. The coefficient of diffusion of oxygen in Y₂O₃ is about 2.2×10^{-11} cm²/s, and is considerably higher than the surrounding Al₂O₃, where the value is generally around 1×10^{-17} cm²/s (at 700° C.). As a result, the presence of Y₂O₃ in the alumina scale is associated with locally accelerated oxidation along these Y-compounds, which leads to internal oxidations intruding in the coating alloy. In real oxide scales, this short-circuit path diffusion significantly impacts the kinetics of internal MCrAlY bulk oxidation. It would be additionally advantageous to provide an MCrAl bond coat wherein an upper surface layer of Y—Al₂O₃ with an absence of Y₂O₃ could be formed, in order to mitigate internal MCrAlY bulk oxidations enabled by short-circuit path diffusion through Y₂O₃ in the upper layer. Such an absence of Y₂O₃ and other Y—Al oxides in the MCrAlY upper layer would further preserve the reservoir of metallic Y in MCrAlY coatings and promote MCrAlY effective lifetimes.

Provided here is an MCrAlY bond coat comprised of an MCrAlY layer in contact with a Y—Al₂O₃ layer. The MCrAlY layer is comprised of a γ -M solid solution, a β -MAl intermetallic phase, and Y-type intermetallics. The Y—Al₂O₃ layer is comprised of Yttrium atoms coordinated with oxygen atoms comprising the Al₂O₃ lattice. The MCrAlY layer and the Y—Al₂O₃ layer have a substantial absence of Y—Al oxides. The absence of selected Y—Al oxides within the MCrAlY layer provides advantage in the maintainability of the Yttrium reservoir within the MCrAlY bulk, and the absence of selected Y—Al oxides within the Y—Al₂O₃ layer mitigates oxygen diffusion and undesired modification of the originally intended Yttrium distribution in the underlying MCrAlY bulk.

These and other objects, aspects, and advantages of the present disclosure will become better understood with reference to the accompanying description and claims.

SUMMARY

The disclosure provides a bond coat comprised of an MCrAlY layer in contact with a Y—Al₂O₃ layer. The

MCrAlY layer may be further in contact with a substrate surface. The MCrAlY layer is comprised of Yttrium and Aluminum, and further comprised of greater than 50 weight percent (wt. %) of a metal M, where the metal M is Ni, Co, Fe, or combinations thereof. In an embodiment, the MCrAlY layer is in contact with a nickel-based substrate, and the metal M is Ni, Co, or Ni and Co. The MCrAlY layer is comprised of a γ -M solid solution, a β -MAl intermetallic phase, and Y-type intermetallics, and has a substantial absence of Yttrium-Aluminum Garnet (YAG) and Yttrium-Aluminum Perovskite (YAP) phases. The Y—Al₂O₃ layer is comprised of Yttrium atoms and Al₂O₃, where the Yttrium atoms are coordinated with oxygen atoms comprising the Al₂O₃ lattice. Generally, the Y—Al₂O₃ layer is comprised of equiaxed grains of Al₂O₃ where the Al₂O₃ grain boundaries are enriched with Yttrium at grain-boundary cation sites.

Additionally, the MCrAlY layer and Y—Al₂O₃ layer have a substantial absence of various Y—Al oxide phases, as opposed to typical MCrAlY bond coats generated through the vacuum heat treatment of MCrAlY materials. The absence of these Y—Al oxides maintains the Y-oxide distribution within the MCrAlY bulk and has a significant effect on the growth rate and adherence of the alumina scale and lifetime of the bond coat. The absence of these Y—Al oxides within the MCrAlY layer and the Y—Al₂O₃ layer is both advantageous and difficult to achieve.

In a particular embodiment, MCrAlY layer **101** and Y—Al₂O₃ layer **102** are fabricated by a process of depositing an MCrAlY material on a substrate, preparing a Y₂O₃ paste comprised of a halide activator, and heat treating the article in a non-oxidizing atmosphere to generate the MCrAlY layer and the Y—Al₂O₃ layer. The Y₂O₃ paste is comprised of Y₂O₃ and a halide activator, and may be further comprised of inert filler and inert transport fluid. The halide activator may be selected from any of the activators known for use in the pack cementation technique and suitable for the production of yttrium containing vapors when placed in the presence of Y₂O₃ at sufficient temperature. The Y₂O₃, halide activator and optionally, inert filler, are combined and thoroughly mixed as required to make the Y₂O₃ paste composition. The Y₂O₃ paste is applied to the MCrAlY upper surface by any convenient, cost effective manner, and the paste covered article is heated in a non-oxidizing atmosphere at a temperature of from about 400° C. to about 1300° C. for a sufficient period of time sufficient to generate the MCrAlY layer and the Y—Al₂O₃ layer. Following the heating period, any excess remaining Y₂O₃ paste may be removed from the heat treated article, generating a bond coated article.

The novel process and principles of operation are further discussed in the following description.

BRIEF DESCRIPTION OF THE DRAWINGS

FIG. **1** illustrates an embodiment of a bond coated article comprised of the MCrAlY layer and the Y—Al₂O₃ layer.

FIG. **2** illustrates a second embodiment of a bond coated article comprised of the MCrAlY layer and the Y—Al₂O₃ layer.

FIG. **3** illustrates an embodiment of a paste covered article comprised of an MCrAlY material and a Y₂O₃ paste.

DETAILED DESCRIPTION

The following description is provided to enable any person skilled in the art to use the invention and sets forth the best mode contemplated by the inventor for carrying out the invention. Various modifications, however, will remain readily

apparent to those skilled in the art, since the principles of the present invention are defined herein specifically to provide a bond coated article comprised of a Y—Al₂O₃ layer atop an MCrAlY layer, where the Y—Al₂O₃ layer and the MCrAlY layer have a substantial absence of Yttrium-Aluminum Garnet and Yttrium-Aluminum Perovskite phases.

The disclosure provides a bond coat comprised of an MCrAlY layer in contact with a Y—Al₂O₃ layer. The MCrAlY layer may be further in contact with a substrate surface. The MCrAlY layer is comprised of Yttrium and Aluminum, and further comprised of greater than 50 weight percent (wt. %) of a metal M, where the metal M is Ni, Co, Fe, or combinations thereof. In an embodiment, the MCrAlY layer is in contact with a nickel-based substrate, and the metal M is Ni, Co, or Ni and Co. The MCrAlY layer is comprised of a γ -M solid solution, a β -MAl intermetallic phase, and Y-type intermetallics, and has a substantial absence of Yttrium-Aluminum Garnet (YAG) and Yttrium-Aluminum Perovskite (YAP) phases. The absence of YAG and YAP phases provides advantages in Y-distribution and maintainability of the Yttrium reservoir within the MCrAlY bulk over typical MCrAlY bond coats generated through the vacuum heat treatments.

The Y—Al₂O₃ layer is comprised of Yttrium atoms and Al₂O₃, where the Yttrium atoms are coordinated with oxygen atoms comprising the Al₂O₃ lattice. The Yttrium enrichment along the grain boundaries provides significant benefit in terms of oxygen diffusivity. Additionally, the Y—Al₂O₃ layer has a substantial absence of Yttria, YAG, and YAP phases, as opposed to the typical upper layers generated through the vacuum heat treatment of MCrAlY materials. The absence of Yttria and mixed Y—Al oxide compounds in the Y—Al₂O₃ layer mitigates diffusion and undesired modification of the originally intended Yttrium distribution in the underlying MCrAlY bulk. The absence of Yttria and mixed Y—Al oxide compounds additionally reduces scale spallation due to stress concentration and thermal expansion mismatch.

Within this disclosure, the descriptive “MCrAlY” when applied to a layer or material denotes a material composition comprised of a base metal M, chromium (Cr), aluminum (Al), and yttrium (Y), where the base metal M is typically Iron (Fe), Nickel (Ni), Cobalt (Co), or a combination of Ni and Co, and where the material composition is at least 50 wt. % of the base metal M, at least 10 wt. % Cr, and at least 7 wt. % Al, where the wt. % values are based on the total weight of the material composition. In an embodiment, the material composition is about 0.05 to about 1.5 wt. % yttrium. In a further embodiment, an MCrAlY material has a composition of 22 wt. % Cr, 10 wt. % Al, 1.0 wt. % Y, balance Ni. In another embodiment, an MCrAlY material has a composition of 32 wt. % Ni, 21 wt. % Cr, 8 wt. % Al, 0.5 wt. % Y, balance Co.

Within this disclosure, an “MCrAlY layer” means a layer having an MCrAlY composition and comprised of an γ -M solid solution, a β -MAl intermetallic phase, and Y-type intermetallics. A “ γ -M solid solution” means a face-centered cubic (FCC) solid solution of the base metal M in the MCrAlY composition, where the base metal M is the solvent and Al, Cr, and possibly other elements, are the solute. “ β -MAl intermetallics” means a body-centered cubic (BCC) intermetallic of the base metal M and Al. “Y-type intermetallics” means an intermetallic of Yttrium and the base metal M or other metallic elements. As is understood, typically the γ -M solid solution, a β -MAl intermetallic phase, and Y-type intermetallics will result from the heat treatment of an MCrAlY material under specific temperature, time, and atmospheric requirements.

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Within this disclosure, “YAG” means a Yttrium-Aluminum Garnet having the chemical structure YAl_5O_{12} , “YAP” means a Yttrium-Aluminum Perovskite having the chemical structure YAl_3 , and “Ytria” means a Yttrium-oxide having the chemical structure Y_2O_3 .

Within this disclosure, a “substantial absence” of Ytria, YAG, and YAP phases within a $Y-Al_2O_3$ layer means that cathodoluminescence of the $Y-Al_2O_3$ layer generates a CL emission spectra where absorption peaks corresponding to Ytria, YAG, and YAP wavelengths are undetectable above background. Similarly, a “substantial absence” of YAG and YAP phases within a MCrAlY layer means that cathodoluminescence of the MCrAlY layer generates a CL emission spectra where absorption peaks corresponding to YAG and YAP wavelengths are undetectable above background.

The disclosure further provides a process for producing the bond coated article disclosed comprised briefly of depositing an MCrAlY material on a substrate, preparing a Y_2O_3 paste comprised of a halide activator, and heat treating the article in a non-oxidizing atmosphere. The process generates the MCrAlY layer and the $Y-Al_2O_3$ layer disclosed by blocking oxygen diffusion into the MCrAlY bulk during heat treatment and mitigating Yttrium diffusion toward the surface, thereby avoiding significant variation in the intended Y-distribution and the Yttrium reservoir.

FIG. 1 illustrates an embodiment of a bond coat having MCrAlY layer **101** in contact with substrate **103** at substrate surface **105**, and further in contact with $Y-Al_2O_3$ layer **102** at MCrAlY upper surface **106**, where substrate surface **105** is a surface of substrate **103** and MCrAlY upper surface is a surface of MCrAlY layer **101**.

At FIG. 1, $Y-Al_2O_3$ layer **102** is comprised of Yttrium atoms and Al_2O_3 , where the Yttrium atoms are coordinated with oxygen atoms comprising the Al_2O_3 lattice. Generally and without being bound by theory, $Y-Al_2O_3$ layer **102** is comprised of equiaxed grains of Al_2O_3 where the Al_2O_3 grain boundaries are enriched with Yttrium at grain-boundary cation sites. Typically the grain-boundary enrichment is reflected by an enrichment factor of at least 100, where the enrichment factor is the ratio of Yttrium concentration at grain boundaries to that within some statistically significant population of grains, with an average grain-boundary cation coverage of typically between 6 and 10%. See e.g., Thompson et al., “Dopant Distributions in Rare-Earth-Doped Alumina,” *J. Am. Ceram. Soc.*, 80 [2] (1997). The Yttrium enrichment along the grain boundary provides significant benefit in terms of oxygen diffusivity within $Y-Al_2O_3$ layer **102**, as is understood. In some cases, Yttrium enrichment along the grain boundaries of Al_2O_3 has generated diffusivities up to 10 times lower than that of pristine grain boundaries. See e.g., Nakagawa et al., “Yttrium doping effect on oxygen grain boundary diffusion in $\alpha-Al_2O_3$,” *Acta Materialia* 55 (2007).

Additionally, $Y-Al_2O_3$ layer **102** has a substantial absence of Ytria, YAG, and YAP phases, as opposed to the typical upper layers generated through the vacuum heat treatment of MCrAlY materials. In typical vacuum heat treatments, where residual oxygen partial pressures on the order of 10^{-5} mbar are experienced, the formation of Ytria and mixed $Y-Al$ oxide compounds is expected. These compounds result from the high oxygen affinity of Yttrium driving a Yttrium diffusion from within the bulk of the MCrAlY material, which modifies the originally intended Y-distribution in the heat-treated MCrAlY bondcoat. Additionally, Ytria and mixed $Y-Al$ oxide compounds, particularly large YAG particles, are linked to increased scale spallation due to stress concentration and thermal expansion mismatch. See e.g., Pint, “Optimization of Reactive-Element Additions to

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Improve Oxidation Performance of Alumina-Forming Alloys,” *J. Am. Ceram. Soc.*, 86 [4] (2003). The absence of Ytria, YAG, and YAP phases within $Y-Al_2O_3$ layer **102** is thus a configuration which is both difficult to achieve and advantageous in operational service.

The $Y-Al_2O_3$ layer may be a continuous or discontinuous layer. For example, FIG. 2 illustrates MCrAlY layer **201** in contact with substrate **203** at substrate surface **205**, and further in contact with $Y-Al_2O_3$ layer **202** at MCrAlY upper surface **206**, where $Y-Al_2O_3$ layer **202** is a discontinuous layer comprised of $Y-Al_2O_3$ regions such as $Y-Al_2O_3$ region **204**. When the $Y-Al_2O_3$ layer is comprised of regions such as $Y-Al_2O_3$ region **204**, the individual $Y-Al_2O_3$ regions have the same properties as those described for the $Y-Al_2O_3$ layer, such as $Y-Al_2O_3$ layer **102**. As such, whether the $Y-Al_2O_3$ layer is a continuous or discontinuous layer is not a limitation of this disclosure. Within this disclosure, it is only necessary that the $Y-Al_2O_3$ layer whether continuous or discontinuous is in contact with the MCrAlY upper surface, and that when the MCrAlY layer is in contact with a substrate surface, that the $Y-Al_2O_3$ layer is separated from the substrate surface by at least some portion of the MCrAlY layer.

At FIG. 1, MCrAlY layer **101** is comprised of Yttrium and Aluminum, and further comprised of at least 50 wt. % of a metal M, at least 10 wt. % Cr, and at least 7 wt. % Al, where the metal M is Ni, Co, Fe, or combinations thereof, and comprised of phases including a γ -M solid solution, a β -MAl intermetallic phase, and Y-type intermetallics. Depending on the exact chemical composition of MCrAlY layer **101**, additional phases may also be present. Generally, the structure of MCrAlY layer **101** is comprised of distinct γ -M solid solution and β -MAl intermetallic phases, with Y-type intermetallics between the phases. In an embodiment, an MCrAlY material utilized to form MCrAlY layer **101** has an MCrAlY composition comprised of from about 15 to about 25 wt. % chromium, from about 7 to about 15 wt. % aluminum, from about 0.05 to about 0.5 wt. % yttrium, and at least 50 wt. % of the base metal M, where the wt. % values are based on the total weight of the composition.

Additionally, MCrAlY layer **101** has a substantial absence of YAG and YAP phases, as opposed to typical MCrAlY bond coats generated through the vacuum heat treatment of MCrAlY materials. Generally, residual oxygen partial pressure in typical vacuum chambers is on the order of 10^{-5} mbar, resulting in the formation of mixed $Y-Al$ oxide compounds. Depending on the Yttrium content of the MCrAlY material and the oxygen partial pressure employed, $Y-Al$ oxide compounds can additionally be expected within the bulk of the MCrAlY bond coat layer, along with, in some cases, a generally continuous layer of yttria and significant Yttrium diffusion toward the surface. See e.g., Gil et al., “Y-rich oxide distribution in plasma sprayed MCrAlY-coatings studied by SEM with a cathodoluminescence detector and Raman spectroscopy,” *Surface & Coatings Technology* 204 (2009); and see Subanovic et al., “Effect of manufacturing related parameters on oxidation properties of MCrAlY-bondcoats,” *Materials and Corrosion* 59 (2008). This generated Y-oxide distribution modifies the Y-distribution within the MCrAlY bulk and has a significant effect on the growth rate and adherence of the alumina scale and lifetime of the bond coat. Additionally, because significant variations in the Y-distribution and the reservoir of metallic Y in MCrAlY coatings can be caused by only minor variations in the processing parameters, the alumina scale growth rate and adherence may be different for nominally the same MCrAlY bondcoats, negatively impacting the reproducibility of bond coat lifetimes. See e.g.,

Toscano et al., “Parameters affecting TGO growth rate and the lifetime of TBC systems with MCrAlY-bondcoats,” *Materials and Corrosion* 59 (2008), among others. Thus, the substantial absence of YAG and YAP phases within MCrAlY layer **101** provides a more uniform and predictable Y-distribution and significantly mitigates issues associated with Y-oxide formation and altered Y distribution profiles.

When MCrAlY layer **101** is a cast layer, it is expected that MCrAlY layer **101** will additionally have a substantial absence of Ytria in addition to the absences of YAG and YAP phases. However, depending on the method of fabrication, some Ytria formed during application may be unavoidably present. See e.g., Gudmundsson et al., “Yttrium oxides in Vacuum-plasma Sprayed CoNiCrAlY Coatings,” *Thin Solid Films*, 173 (1989), among others.

It is known that Ytria, YAG, and YAP exhibit significant cathodoluminescence and may be utilized to determine the morphology and distribution of Y_2O_3 and Y—Al mixed oxide phases within MCrAlY coatings. See Gil et al. and see Subanovic et al., among others. Correspondingly, as used herein and above, a “substantial absence” of Ytria, YAG, and YAP phases within Y— Al_2O_3 layer **102** means that cathodoluminescence of some portion of Y— Al_2O_3 layer **102** generates a CL emission spectra where absorption peaks corresponding to Ytria, YAG, and YAP wavelengths are undetectable above background. Similarly, a “substantial absence” of YAG and YAP phases within MCrAlY layer **101** means that cathodoluminescence of some portion of MCrAlY layer **101** generates a CL emission spectra where absorption peaks corresponding to YAG and YAP wavelengths are undetectable above background.

As stated, MCrAlY layer **101** is comprised of a γ -M solid solution, a β -MAl intermetallic phase, and Y-type intermetallics. As defined, a “Y-type intermetallic” as used here means an intermetallic compound consisting of Yttrium and at least one other metal. For example, when MCrAlY layer **101** derives from a source material composition where the base metal M is Nickel and Cobalt, MCrAlY layer **101** may be comprised of Ni_5Y , Co_5Y , and Cr_5Y , among others, as Y-type intermetallics. These intermetallics define a Yttrium atomic concentration within MCrAlY layer **101** which is relatively stable around the original Yttrium atomic concentration of the MCrAlY source material utilized to generate the MCrAlY layer. For example, when the MCrAlY source material has an MCrAlY composition comprised of 0.1 wt. % Yttrium, and MCrAlY layer **101** comprised of γ -M solid solution, a β -MAl intermetallic phase, and Y-type intermetallics results from heat treatment of the MCrAlY source material, the resulting Y-type intermetallics define a Yttrium atomic concentration within MCrAlY layer **101** approximately equivalent 0.1 wt. %. This is distinguished from typical Yttrium containing MCrAlY layers, where even low oxygen partial pressures during heat treatment are sufficient to tie up a significant portion of the MCrAlY source material’s Yttrium in Y—Al oxides within the MCrAlY bulk. Additionally, as will be discussed infra, the diffusion of Yttrium toward the surface is mitigated during vacuum heat treatment, resulting in a relatively homogenous Yttrium distribution within MCrAlY layer **101**.

Maintaining a significant majority of the Yttrium as a Y-type intermetallic has distinct advantage over MCrAlY layers where a significant portion of the Yttrium component may be tied up in Y—Al oxide inclusions, effectively removing that Yttrium from the available Yttrium reservoir of the MCrAlY layer. There is strong indication that an important life time governing factor of MCrAlY coatings is the Yttrium incorporation into the alumina based Thermal Grown Oxide

(TGO) formed during service, which correspondingly generates a decrease of the yttrium concentration in the coating over a component lifetime. If the yttrium concentration is decreased beneath a critical level, its positive effect on TGO adherence is lost, resulting in TGO spallation. See e.g., Toscano et al., “Parameters affecting TGO growth rate and the lifetime of TBC systems with MCrAlY-bondcoats,” *Materials and Corrosion* 59 (2008). Thus, within the MCrAlY layer of this disclosure, maintaining a large majority of the Yttrium component in the form of Y-type intermetallics with a substantial absence of Y—Al oxides acts to maximize the Yttrium reservoir of the material, delaying spallation of thermally grown oxides during subsequent service.

In an embodiment, the Yttrium within an MCrAlY layer such as MCrAlY layer **101** defines a Yttrium component, and the Y-type intermetallics within the MCrAlY layer are comprised of at least 90 wt. % of the Yttrium within the Yttrium component. Stated differently, 90 wt. % of the Yttrium in the MCrAlY layer is in the form of a Y-type intermetallic.

As indicated at FIG. 1, MCrAlY layer **101** may be further in contact with substrate surface **105** of substrate **103**. In an embodiment, substrate **103** is nickel-base alloy, and the base metal M comprising MCrAlY layer **101** is Ni, Co, or Ni and Co. Additionally, Y— Al_2O_3 layer **102** may be further in contact with topcoat **107**. In an embodiment, topcoat **107** is comprised of yttria stabilized zirconia.

In a particular embodiment, MCrAlY layer **101** and Y— Al_2O_3 layer **102** are fabricated by a process of depositing an MCrAlY material on a substrate, preparing a Y_2O_3 paste comprised of a halide activator, and heat treating the article in a non-oxidizing atmosphere to generate the MCrAlY layer and the Y— Al_2O_3 layer, as follows:

Depositing the MCrAlY Material on the Substrate:

As earlier defined, an MCrAlY material is a material having a material composition comprised of a base metal M, chromium, aluminum, and yttrium. The base metal M is typically Iron, Nickel, Cobalt, or a combination of Nickel and Cobalt. Generally, the MCrAlY material has an MCrAlY composition comprised of from about 15 to about 25 weight percent (wt. %) chromium, from about 7 to about 15 wt. % aluminum, from about 0.05 to about 0.5 wt. % yttrium, and greater than about 50 wt. % of the base metal M, where the wt. % values are based on the total weight of the composition. In an embodiment, the MCrAlY material has an MCrAlY composition comprised of 22 wt. % Cr, 10 wt. % Al, 1.0 wt. % Y, balance Ni. In another embodiment, an MCrAlY material has an MCrAlY composition of 32 wt. % Ni, 21 wt. % Cr, 8 wt. % Al, 0.5 wt. % Y, balance Co.

The MCrAlY material composition described can be manufactured as a powder for use in depositions using, for example, a spraying technique. The material composition including all elemental constituents may be prepared with powders generated by conventional powder processing methods, such as inert gas atomization from ingots. Alternatively, powder blends may be prepared by mixing separate powders of individual elements or alloys. In a final powder composition prepared in this manner the weight percentage of each elemental constituent corresponds to the ranges earlier provided. A preferred diameter for the metallic powder particles, regardless how formed, is between about 5 to about 50 microns.

The MCrAlY material is deposited as a coating layer on a substrate surface, such as substrate surface **335** of substrate **333** at FIG. 3. The MCrAlY material **331** is deposited using conventional methods, such as low pressure plasma spray (LPPS), electron beam physical vapor deposition (EBPVD), High Velocity Oxygen Fuel (HVOF) spray, or other methods

known in the art for deposition of an MCrAlY material onto a substrate surface. See e.g., Pawlowski, *The Science and Engineering of Thermal Spray Coatings* (2nd Ed. 2008). In an embodiment, the MCrAlY material **331** is deposited on the substrate surface to a thickness of about 20 microns. In another embodiment where the base metal M of the MCrAlY material **331** is Nickel or Cobalt, the substrate is comprised of a nickel-base alloy. Deposition of the MCrAlY material **331** onto substrate surface **335** of substrate **333** generates an MCrAlY material surface **336**, where the MCrAlY material surface **336** is a surface of the MCrAlY material **331** separated from substrate surface **335** by some portion of MCrAlY material **331**.

In an embodiment, the MCrAlY material has an MCrAlY composition comprised of 22 wt. % Cr, 10 wt. % Al, 1.0 wt. % Y, balance Ni, or comprised of 32 wt. % Ni, 21 wt. % Cr, 8 wt. % Al, 0.5 wt. % Y, balance Co, and the MCrAlY material is deposited by atmospheric plasma spray (APS) or high velocity oxygen fuel (HVOF) on a substrate comprised of an advanced Ni-based single crystal alloy, such as RENE N5 developed by General Electric Company, PW1484 developed by Pratt & Whitney, RR2000 developed by Rolls-Royce, or CMSX-4 developed by Cannon Muskegon Corporation, and the equivalents.

Preparing a Y₂O₃ Paste:

The Y₂O₃ paste is comprised of Y₂O₃ and a halide activator, and may be further comprised of inert filler and inert transport fluid. In an embodiment, the Y₂O₃ comprising the Y₂O₃ paste is at least 10 wt. % of the combined weight of the Y₂O₃, halide activator, and inert filler, and the halide activator is at least about 0.1 wt. % of the combined weight of the Y₂O₃ and the halide activator.

Y₂O₃ is commercially available as a powder with typical purities of 99.5 wt. % to 99.99 wt. %. Although any particle size may be employed, it has been found advantageous to use particles of relatively fine size. Preferably, the Y₂O₃ is relatively fine for ease in suspending in a paste or slurry. Generally, Y₂O₃ particle size is less than 200 microns, preferably less than 45 microns, more preferably less than 10 microns.

The halide activator serves to react with the Y₂O₃ powder during a high temperature coating process to produce yttrium containing vapors which are carried to the MCrAlY upper surface. The halide activator may be selected from any of the activators known for use in the pack cementation technique and suitable for the production of yttrium containing vapors when placed in the presence of Y₂O₃ at sufficient temperature. Suitable halide activators may be selected from any of the activators known for this purpose and combinations thereof and equivalents. Typically the halide activator includes one or more halide salts. Suitable halide activators include AlF₃, AlCl₃, NH₄F, NH₄Cl, NaF, NaCl, KF, and KCl, among others. In an embodiment, the halide activator is present in an amount of at least about 0.1 wt. % of the combined weight of the Y₂O₃ and the halide activator. In an embodiment, the Y₂O₃ paste is comprised of 0.5 g NaCl per 150 g Y₂O₃.

The inert filler is a material which is not reduced by yttrium-containing vapors and which prevents the sintering of Y₂O₃. Generally, the inert filler may comprise 10 wt. % to 80 wt. % of the Y₂O₃ paste. Fillers are known in the art that inhibit sintering of Y₂O₃ and include alumina, thoria, calcia, zirconia and other stable and inert refractory oxides and mixtures thereof. The inert filler may be added in the same particle size range as the Y₂O₃, with typical particle size less than 200 microns, preferably less than 45 microns, more preferably less than 10 microns.

The Y₂O₃, halide activator and optionally, inert filler, are combined and thoroughly mixed as required to make the

Y₂O₃ paste composition. An inert transport fluid such as water may be added, generally in an amount of about 30 wt % to 40 wt % of the Y₂O₃ paste. In an embodiment, the Y₂O₃ paste is comprised of 0.5 g NaCl and 150 g Y₂O₃ per 100 cc of water. This is stirred until it has a consistence of about milk or cream. Frequent stirring or continuous stirring during application provides the best coating.

Applying the Y₂O₃ Paste and Heating the Paste Covered Article:

The Y₂O₃ paste is applied to the MCrAlY material surface by any convenient, cost effective manner. Dipping, brushing and spraying are all effective. In an embodiment, the Y₂O₃ paste is applied to the MCrAlY material surface in a layer less than about 1 micron thick. FIG. 3 illustrates Y₂O₃ paste **332** applied to MCrAlY material surface **336**. Application of Y₂O₃ paste **332** to MCrAlY material surface **336** generates a paste covered article, where the paste covered article is comprised of substrate **333**, MCrAlY material **331**, and Y₂O₃ paste **332**, where MCrAlY material **331** is in contact with substrate **333** at substrate surface **335** and further in contact with Y₂O₃ paste **332** at MCrAlY material surface **336**, and where Y₂O₃ paste **332** is separated from substrate **333** by at least some portion of MCrAlY material **331**, as illustrated at FIG. 3. In an embodiment, a topcoat material such as yttria stabilized zirconia is then applied to the surface of the Y₂O₃ layer.

The paste covered article is then heated in a non-oxidizing atmosphere at a temperature of from about 400° C. to about 1300° C. for a sufficient period of time sufficient to generate the MCrAlY layer and the Y—Al₂O₃ layer, where the MCrAlY layer and the Y—Al₂O₃ layer have properties as described. In an embodiment, the paste covered article is heated at a temperature of about 900° C. for a period of about 2 hours. A “non-oxidizing atmosphere” as used in this disclosure means an atmosphere having either a total pressure or partial pressure of oxygen less than about 10⁻³ mbar.

Following the heating period, any excess remaining Y₂O₃ paste may be removed from the heat treated article, generating a bond coated article.

The method as described above may be utilized for the production of the bond coated article comprised of the Y—Al₂O₃ layer and the MCrAlY layer as described. Without being bound by theory, it is believed that utilization of the Y₂O₃ paste in the non-oxidizing atmosphere as described acts to severely mitigate the diffusion of any residual oxygen existing during the heat treatment prescribed. This residual oxygen is typically present in vacuum heat treatment furnaces, which generally exhibit oxygen pressures on the order of 10⁻⁵ mbar during operation. As is understood, Yttrium has the highest oxygen affinity in typical MCrAlY materials, and Yttria is generally the first oxide to form during early oxidation stages. The low oxygen pressure exhibited by standard vacuum heat furnaces is sufficient to generate a strong diffusion of Yttrium within the MCrAlY material toward the surface and subsequent oxidation, as well as the formation of YAG and YAP phases within the MCrAlY bulk. The method as described above allows the formation of a bond coated article comprised of the Y—Al₂O₃ layer and the MCrAlY layer as described by blocking oxygen diffusion into the MCrAlY bulk during heat treatment, resulting in a Y—Al₂O₃ layer comprised of equiaxed grains of Al₂O₃ enriched with Yttrium at grain-boundary cation sites and having a substantial absence of Yttria, YAG, and YAP phases, and an MCrAlY layer in contact with the Y—Al₂O₃ layer, where the MCrAlY layer is comprised of γ -M solid solution, β -MAl intermetallics, and Y-type intermetallics with a substantial absence of YAG and YAP phases. The produced material thus avoids

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significant variation in the intended Y-distribution and the reservoir of metallic Y in MCrAlY coatings, and mitigates the negative impacts to bond coat lifetimes.

It is to be understood that the above-described arrangements are only illustrative of the application of the principles of the present invention and it is not intended to be exhaustive or limit the invention to the precise form disclosed. Numerous modifications and alternative arrangements may be devised by those skilled in the art in light of the above teachings without departing from the spirit and scope of the present invention. It is intended that the scope of the invention be defined by the claims appended hereto.

In addition, the previously described versions of the present invention have many advantages, including but not limited to those described above. However, the invention does not require that all advantages and aspects be incorporated into every embodiment of the present invention.

All publications and patent documents cited in this application are incorporated by reference in their entirety for all purposes to the same extent as if each individual publication or patent document were so individually denoted.

What is claimed is:

1. A bond coat comprised of:

an MCrAlY layer, where the MCrAlY layer has an MCrAlY composition comprised of yttrium, aluminum, chromium, and at least 50 wt. % of a base metal M, where the base metal M is Ni, Co, or Fe, and where the MCrAlY layer is comprised of γ -M solid solution, β -MAl intermetallic, and Y-type intermetallics, and where the MCrAlY layer has a substantial absence of Yttrium-Aluminum Garnet and Yttrium-Aluminum Perovskite, such that a cathodoluminescence of the MCrAlY layer generates a MCrAlY CL emission spectra where absorption peaks corresponding to Yttrium-Aluminum Garnet and Yttrium-Aluminum Perovskite are undetectable above background, and where the MCrAlY layer has an MCrAlY upper surface, where the MCrAlY upper surface is a surface of the MCrAlY layer; and

a Y—Al₂O₃ layer, where the Y—Al₂O₃ layer is in contact with the MCrAlY upper surface, where the Y—Al₂O₃ layer is comprised of yttrium atoms and Al₂O₃, where the yttrium atoms are coordinated with oxygen atoms comprising the Al₂O₃, and where the Y—Al₂O₃ layer has a substantial absence of Yttrium-Aluminum Garnet, Yttrium-Aluminum Perovskite, and yttria, such that a cathodoluminescence of the Y—Al₂O₃ layer generates a Y—Al₂O₃ CL emission spectra where absorption peaks corresponding to Yttrium-Aluminum Garnet, Yttrium-Aluminum Perovskite, and yttria wavelengths are undetectable above background.

2. The bond coat of claim 1 where the MCrAlY composition is comprised of from about 15 to about 25 wt. % chromium, from about 10 to about 15 wt. % aluminum, and from about 0.05 to about 1.5 wt. % yttrium.

3. The bond coat of claim 2 where the MCrAlY layer has a weight percent of yttrium within the MCrAlY layer, and where the Y-type intermetallics of the MCrAlY layer comprise at least 90% of the weight percent of yttrium within the MCrAlY layer.

4. The bond coat of claim 3 further comprised of a substrate having a substrate surface, where the MCrAlY layer is in contact with the substrate surface, such that the MCrAlY layer is between the substrate and the Y—Al₂O₃ layer.

5. The bond coat of claim 4 further comprised of a topcoat in contact with the Y—Al₂O₃ layer, such that the Y—Al₂O₃ layer is between the topcoat and the MCrAlY layer.

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6. The bond coat of claim 4 where the MCrAlY composition is comprised of less than about 0.15 wt. % yttrium.

7. The bond coat of claim 6 where the substrate is comprised of a nickel-base alloy, and where the metal base M is Ni or Co.

8. The bond coat of claim 7 where the MCrAlY layer is less than about 20 microns thick and the Y—Al₂O₃ layer is less than about 1 micron thick.

9. The bond coat of claim 8 further comprised of a topcoat in contact with the Y—Al₂O₃ layer, such that the Y—Al₂O₃ layer is between the topcoat and the MCrAlY layer.

10. The bond coat of claim 9 where the topcoat is comprised of yttria stabilized zirconia.

11. The bond coat of claim 2 where the substrate is comprised of a nickel-base alloy, and where the metal base M is Ni, Co, or Ni and Co, and where the MCrAlY composition is comprised of less than about 0.15 wt. % yttrium, and where the MCrAlY layer is less than about 20 microns thick and the Y—Al₂O₃ layer is less than about 1 micron thick.

12. The bond coat of claim 11 further comprised of a topcoat in contact with the Y—Al₂O₃ layer, such that the Y—Al₂O₃ layer is between the topcoat and the MCrAlY layer.

13. A bond coat comprised of:

an MCrAlY layer, where the MCrAlY layer has an MCrAlY composition comprised of from about 15 to about 25 wt. % chromium, from about 10 to about 15 wt. % aluminum, from about 0.05 to about 0.5 wt. % yttrium, and at least 50 wt. % of a base metal M, where the base metal M is Ni, Co, or Fe, and where the MCrAlY layer is comprised of γ -M solid solution, β -MAl intermetallic, and Y-type intermetallics, and where the MCrAlY layer has a substantial absence of Yttrium-Aluminum Garnet and Yttrium-Aluminum Perovskite, such that a cathodoluminescence of the MCrAlY layer generates a MCrAlY CL emission spectra where absorption peaks corresponding to Yttrium-Aluminum Garnet and Yttrium-Aluminum Perovskite are undetectable above background, and where the MCrAl layer has a weight percent of Yttrium within the MCrAlY layer, where the Y-type intermetallics of the MCrAlY layer comprise at least 90% of the weight percent of yttrium within the MCrAlY layer, and where the MCrAlY layer has an MCrAlY upper surface, where the MCrAlY upper surface is a surface of the MCrAlY layer; and

a Y—Al₂O₃ layer, where the Y—Al₂O₃ layer is in contact with the MCrAlY upper surface, where the Y—Al₂O₃ layer is comprised of Yttrium atoms and Al₂O₃, where the Yttrium atoms are coordinated with oxygen atoms comprising the Al₂O₃, and where the Y—Al₂O₃ layer has a substantial absence of Yttrium-Aluminum Garnet, Yttrium-Aluminum Perovskite, and Yttria, such that a cathodoluminescence of the Y—Al₂O₃ layer generates a Y—Al₂O₃ CL emission spectra where absorption peaks corresponding to Yttrium-Aluminum Garnet, Yttrium-Aluminum Perovskite, and yttria wavelengths are undetectable above background.

14. The bond coat of claim 13 where the metal base M is Ni or Co.

15. The bond coat of claim 14 where the MCrAlY composition is comprised of less than about 0.15 wt. % Yttrium.

16. The bond coat of claim 15 further comprised of a substrate having a substrate surface, where the MCrAlY layer is in contact with the substrate surface, such that the MCrAlY layer is between the substrate and the Y—Al₂O₃ layer, and where the substrate is comprised of a nickel-base alloy.

17. The bond coat of claim 16 where the MCrAlY layer is less than about 20 microns thick and the Y—Al₂O₃ layer is less than about 1 micron thick.

18. The bond coat of claim 17 further comprised of a topcoat in contact with the Y—Al₂O₃ layer, such that the Y—Al₂O₃ layer is between the topcoat and the MCrAlY layer, and where the topcoat is comprised of yttria stabilized zirconia. 5

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