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(54) **CALIBRATION GAS DELIVERY APPARATUS**

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(75) Inventors: **Frank R. Schaedlich**, Toronto (CA);  
**Daniel R. Schneeberger**, Scarborough  
(CA)

(73) Assignee: **Tekran Instruments Corporation**,  
Halifax (CA)

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*Primary Examiner*—Lyle A Alexander

*Assistant Examiner*—Dean Kwak

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(74) *Attorney, Agent, or Firm*—Bereskin & Parr LLP; H.  
Samuel Frost

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**G01J 3/30** (2006.01)

(57) **ABSTRACT**

(52) **U.S. Cl.** ..... **436/9**; 422/100; 422/68.1;  
422/83; 436/18; 73/1.03

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436/18; 422/100, 68.1, 83; 356/316; 73/1.03  
See application file for complete search history.

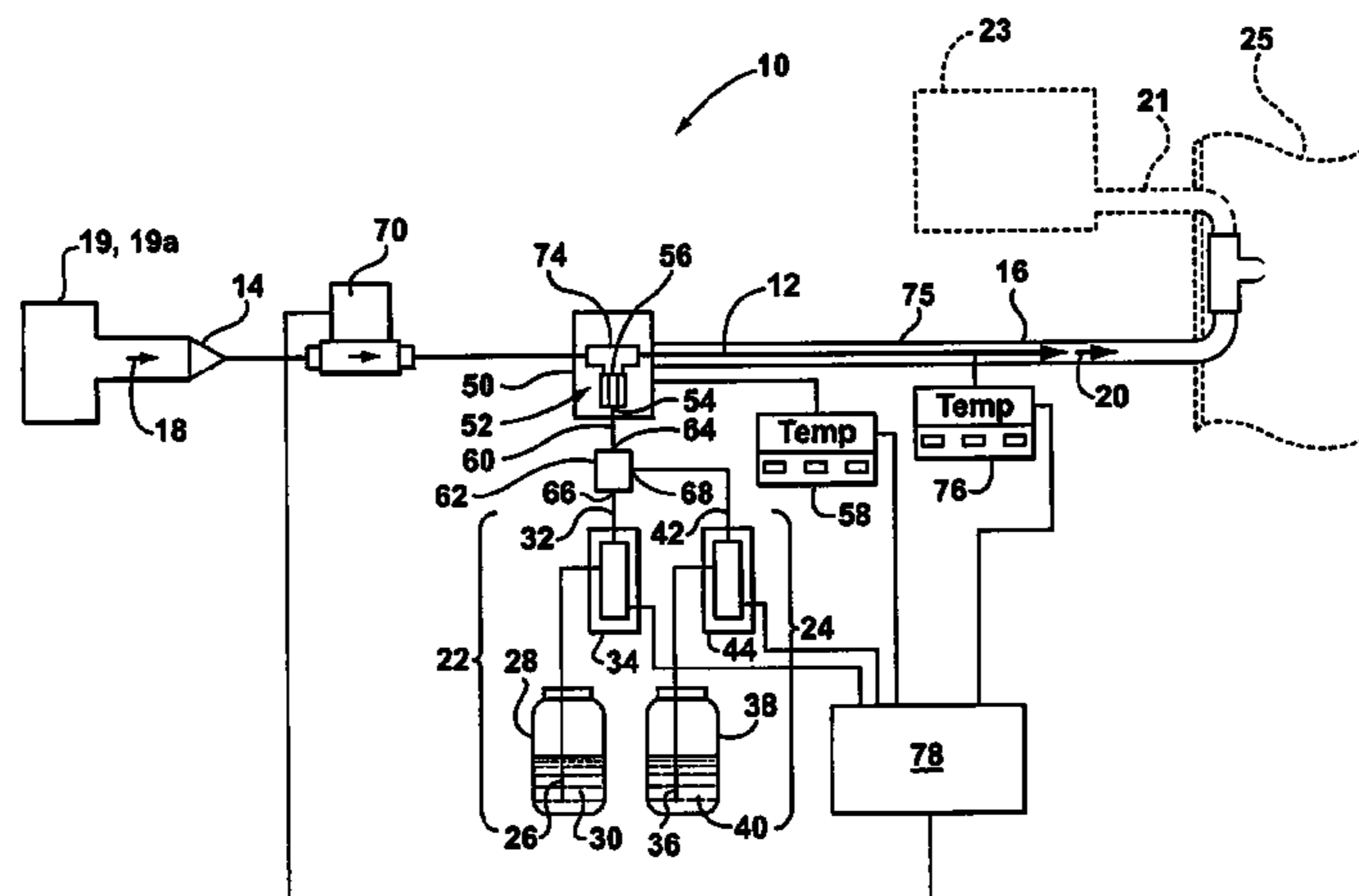
A calibration gas generation method and apparatus for gener-  
ating a selectively humidified calibration gas to a measur-  
ment probe includes a delivery conduit having a conduit inlet  
adapted to receive a carrier gas stream and a conduit outlet for  
delivering a calibration gas stream. The apparatus is provided  
with a first injection unit having a first intake in fluid com-  
munication with a first reservoir and a first outlet in fluid  
communication with the delivery conduit, the first reservoir  
being adapted to hold a first analyte in liquid form, and a  
second injection unit having a second intake in fluid commu-  
nication with a second reservoir and a second outlet in fluid  
communication with the delivery conduit, the second reser-  
voir being adapted to hold a humidificant in liquid form. The  
apparatus further includes at least one vaporizer downstream  
of the first and second outlets and upstream of the conduit  
outlet for converting the analyte and humidificant liquids to  
vapor form and delivering a calibration gas including the  
carrier gas, analyte vapor, and humidificant vapor to the con-  
duit outlet.

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**12 Claims, 3 Drawing Sheets**



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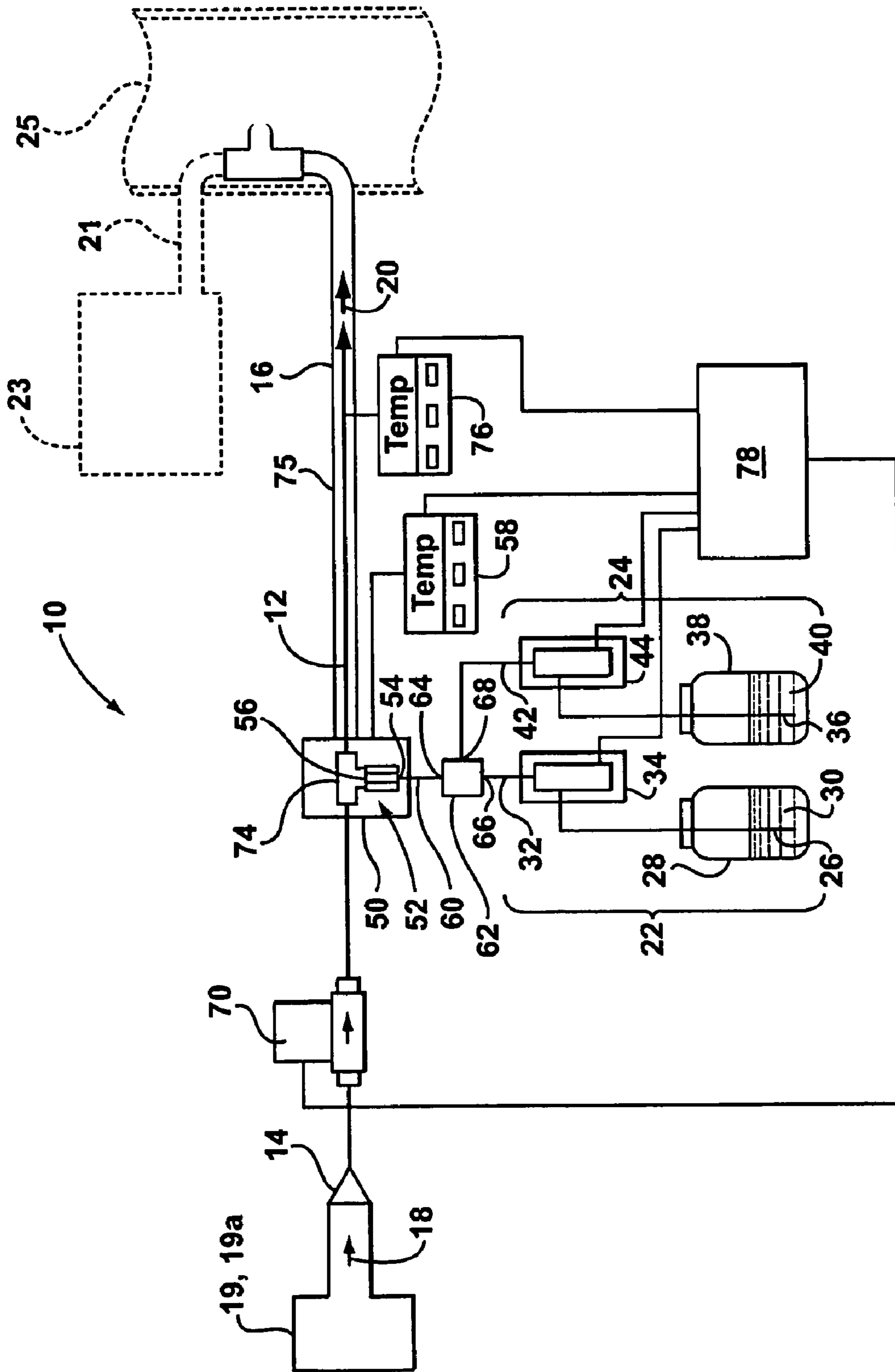


FIG. 1

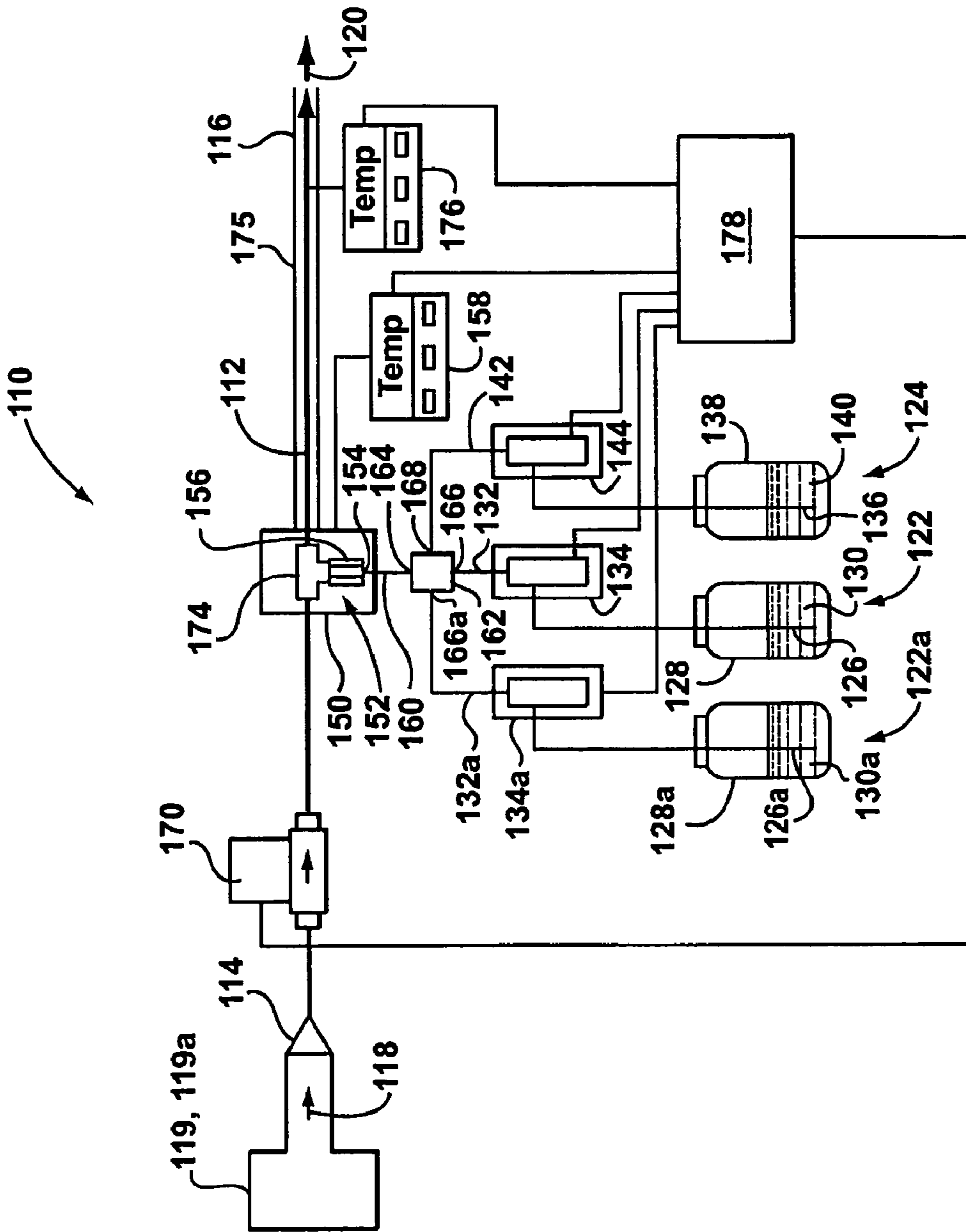


FIG. 2

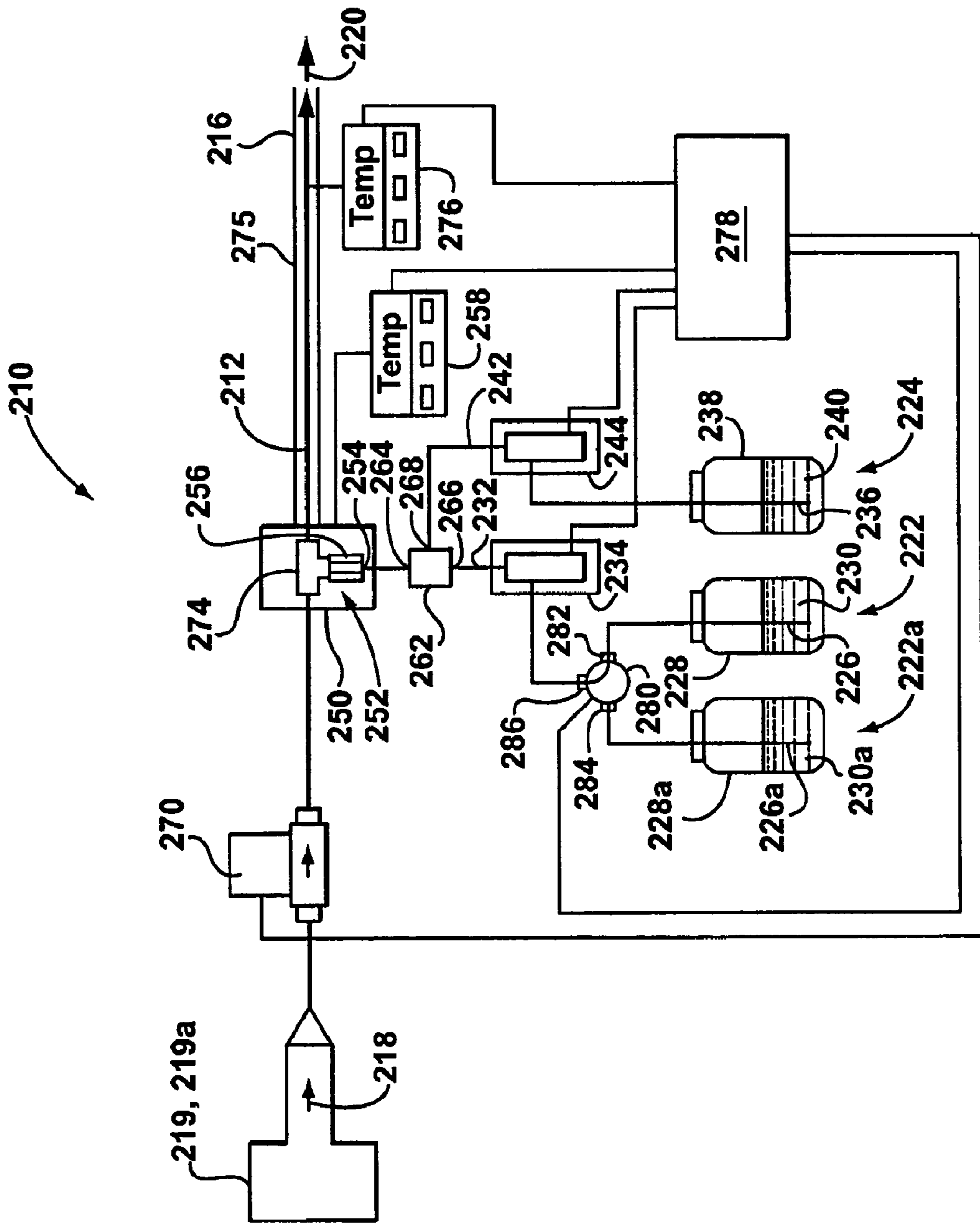


FIG. 3

**CALIBRATION GAS DELIVERY APPARATUS**

## FIELD

This invention relates to an apparatus and method for providing a calibration gas to, for example, a continuous emissions monitoring device.

## INTRODUCTION

Calibration of gas monitoring systems is generally a mandatory procedure in the maintenance of gas monitoring systems to ensure that accurate readings from the systems are being obtained. It is common practice to introduce a known concentration of the analyte in an inert gas or in a zero air matrix to calibrate a gas monitoring system.

An example of a gas monitoring system is a continuous emissions monitor (CEM) used to monitor the amount of mercury discharged in smoke stacks of power generation installations or incinerators. To ensure that the CEM system is operating satisfactorily, a calibration routine is periodically performed in which the discharged gas normally being monitored is replaced with a calibration gas containing a known amount of mercury. The mercury level indicated by the CEM is then compared to the known amount contained in the calibration gas, and corrective action can be taken where any unacceptable deviation is found. The calibration procedure must typically conform to government standards (e.g. US EPA (Environmental Protection Agency)), and in the case of mercury CEM systems, both elemental and halogenated mercury standard gasses may need to be accommodated for calibration.

Various apparatus and methods for calibration of mercury gas monitoring systems are known. Elemental mercury at low volumetric concentrations may be provided in cylinders (Spectra Gases). Each of such prepared cylinders must be compared against a certified standard in order to have a concentration value assigned to it. These cylinders are expensive, and thus are not economical to generate the high volumes of gas required for the calibration of most analytical systems. Generally all of the mercury CEM systems that are currently available employ an inertial filter arrangement. To calibrate through this type of filter, a very large sample flow of calibration gas, typically in excess of 20 l/m, is required.

Calibrators which generate elemental mercury with a saturated mercury vapor chamber, such as Tekran® Model 3310 Elemental Mercury Calibrator, are known. Calibrators of this type rely on first principles. The vapor pressure of elemental mercury is a well-characterized function of temperature. A known gas flow is first passed through a chamber containing liquid mercury. By saturating the gas with mercury vapor, the gas exits the chamber in equilibrium at the prescribed chamber temperature. A second known gas flow is then used to dilute the saturated gas stream. This method will produce known concentrations if the temperature of the source, chamber flow rate and dilution gas flow rate are known. These variables are easily measured in a manner that is traceable to standards set by NIST (National Institute of Science and Technology). Saturated sources can generate large volumes of elemental calibration gases over a wide range of concentrations at little cost.

Permeating devices may also be used for the generation of elemental mercury. At high emission rates, permeation devices may be certified gravimetrically. At low levels, gravimetric certification is not practical so the sources must be calibrated against some other primary standard.

In U.S. Pat. No. 6,852,542 (Mandel et al.), a method and system for creating a mercury halide standard for use in testing a mercury analyzer system is disclosed. This system uses a known reaction for producing mercury chloride. A known amount of elemental mercury and a gaseous stream containing chlorine are fed to a reaction chamber to form mercury halide. The mercury halide is then fed to a mercury analyzer system where it is converted to form gaseous elemental mercury which is then measured by a mercury analyzer. Comparing either the amount of elemental mercury supplied to the reaction chamber or the amount of mercury halide formed in the reaction chamber with the amount of elemental mercury converted from the mercury halide, the conversion of mercury halide to gaseous elemental mercury by the mercury analyzer system can be evaluated. In practice, the rate of conversion from elemental mercury to mercury chloride is problematical.

In U.S. Pat. No. 6,475,802 (Schaedlich et al.), a method and apparatus for collecting a sample of gaseous mercury and to differentiate between the different gaseous mercury components is disclosed. A quartz denuder module is provided having a coated extended surface for adsorbing reactive gaseous mercury. After collection of a sample, the coating is heated to desorb the mercury as elemental gaseous mercury, which can then be detected and measured in a conventional analyzer. This device may be calibrated using any type of device capable of producing sufficiently low levels. Other patents and applications to the same inventors and relating to mercury detection are U.S. Pat. Nos. 5,597,535, 5,660,795 and 6,475,802 and U.S. patent application Ser. Nos. 10/931,987 and 11/086,480, all of which are hereby incorporated by reference.

In another calibration gas generator marketed under the brand name Hovacal™, a dilute mercury solution is pumped to a vaporizer by a peristaltic pump. The solution is weighed over time using a precision balance to determine the rate of use. A known gas flow is used to dilute the mercury vapor to a known concentration.

There are problems with these known approaches. Most of the systems described above produce a dry calibration gas. Mercury chloride does not travel well in a dry gas, resulting in sample transport problems and lengthy equilibration times. A more severe problem is the poor transport of all mercury compounds, including elemental mercury, through the front end CEM components, including the probe and filter components. In practice, QA/QC techniques such as standard additions may show that a continuous mercury analyzer (CEM) is operating properly when confronted with the normal stack gas matrix being monitored. However, when confronted with the radically different matrix of a typical calibration gas, these components will often fail to transport the calibration gas properly until they reach equilibrium with the new gas. This can result in very lengthy calibration times which do not meet regulatory requirements. In cases where the coal produces a particularly reactive fly ash, the losses when using a dry calibration gas are consistent until the probe is mechanically cleaned. This can occur even when using inertial separator filters rather than conventional filters.

The Hovacal is a system that generates humidified gas, however, it is manual in operation. It has a single, manually controlled injection port and cannot automatically generate various concentrations of analyte while retaining, for example, a constant water concentration in the gas without continually changing solutions, i.e. the water or vapor concentration in the gas inherently depends on the concentration of the analyte in the solution and the rate at which the solution

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is added to the gas flow. To change the vapor concentration different supplies of the analyte solution with different concentrations are needed.

## SUMMARY

The present invention provides an apparatus and method for generating a calibration gas for a gas monitoring system, that is simple and economical and may be fully automated. It provides accurate, reproducible, and stable calibration gas with the capability to be transported through analytical instrumentation systems. The present invention can also provide an apparatus and method that is effective in generating a wide variety of different calibration gas compositions such that multi-point calibrations may be provided and that the gas can continue to emulate the composition of the actual flue gas being monitored as the analyte concentration is varied.

According to one aspect of the invention, a calibration gas delivery apparatus for delivering a selectively humidified calibration gas to a measurement probe is provided. The apparatus includes a delivery conduit having a conduit inlet adapted to receive a carrier gas stream and a conduit outlet for delivering a calibration gas stream. The apparatus is provided with a first injection unit having a first intake in fluid communication with a first reservoir and a first outlet in fluid communication with the delivery conduit, the first reservoir being adapted to hold a first analyte in liquid form, and a second injection unit having a second intake in fluid communication with a second reservoir and a second outlet in fluid communication with the delivery conduit, the second reservoir being adapted to hold a humidificant in liquid form. The apparatus further includes at least one vaporizer downstream of the first and second outlets and upstream of the conduit outlet for converting the analyte and humidificant liquids to vapor form and delivering a calibration gas including the carrier gas, analyte vapor, and humidificant vapor to the conduit outlet.

In accordance with a first aspect of the present invention, there is provided a calibration gas delivery apparatus for delivering a selectively humidified calibration gas to a measurement probe, the apparatus comprising:

- a) a delivery conduit having a conduit inlet adapted to receive a carrier gas stream and a conduit outlet for delivering a calibration gas stream;
- b) a first injection unit having a first intake in fluid communication with a first reservoir and a first outlet in fluid communication with the delivery conduit, the first reservoir adapted to hold a first analyte in liquid form;
- c) a second injection unit having a second intake in fluid communication with a second reservoir and a second outlet in fluid communication with the delivery conduit, the second reservoir adapted to hold a humidificant in liquid form; and
- d) at least one vaporizer downstream of the first and second outlets and upstream of the conduit outlet for converting the analyte and humidificant liquids to vapor form and delivering a calibration gas including the carrier gas, analyte vapor, and humidificant vapor to the conduit outlet.

In accordance with another aspect of the present invention, there is provided a calibration gas delivery apparatus for delivering a selectively humidified calibration gas to a measurement probe, the apparatus comprising:

- a) a delivery conduit having a conduit inlet adapted to receive at least one carrier gas stream including an analyte and a conduit outlet for delivering a calibration gas stream;
- b) an injection unit having an intake in fluid communication with a reservoir and an outlet in fluid communication

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with the delivery conduit, the reservoir adapted to hold a humidificant in liquid form; and

- c) at least one vaporizer downstream of the outlet and upstream of the conduit outlet for converting the humidificant to vapor form and delivering a calibration gas including all of said at least one carrier gas, and humidificant vapor to the conduit outlet.

In accordance with a further aspect of the present invention, there is provided a method of generating a calibration gas including at least one analyte and at least one humidificant, the method comprising:

- a) supplying at least one carrier gas;
- b) providing the carrier gas with at least one analyte, each analyte being provided at known concentration in the carrier gas;
- c) adding humidity to the carrier gas flow, to give a desired humidity level; and
- d) supplying the carrier gas including the analyte and added humidity to measuring equipment for detecting the analyte to enable calibration of the measuring equipment.

## BRIEF DESCRIPTION OF THE DRAWINGS

For a better understanding of the present invention and to show more clearly how it would be carried into effect, reference will now be made by way of example, to the accompanying drawings that show embodiments of the present invention, and in which:

FIG. 1 is a schematic diagram of an embodiment of a calibration gas delivery apparatus in accordance with the present invention;

FIG. 2 is a schematic diagram of another embodiment of a calibration gas delivery apparatus in accordance with the present invention; and

FIG. 3 is a schematic diagram of a further embodiment of a calibration gas delivery apparatus in accordance with the present invention.

## DESCRIPTION OF EMBODIMENTS OF THE INVENTION

Aspects of the present invention and applicants' teachings may be further understood in light of the following examples, which should not be construed as limiting the scope of the present teachings in any way.

A calibration gas delivery apparatus **10** in accordance with the present invention is shown in FIG. 1. The apparatus **10** includes a delivery conduit **12** having a conduit inlet **14** and a conduit outlet **16** between which the conduit **12** is adapted to convey fluids.

The inlet **14** is, in the embodiment illustrated, adapted to receive a carrier gas stream **18**, which can be provided from a carrier gas source **19**. The outlet **16** provides delivery of a calibration gas **20**. The calibration gas **20** can provide a known level of a component being measured to a measurement probe **21** for comparison against the level indicated by an associated measurement device **23**. In FIG. 1, the probe **21** is schematically shown within an emissions stack **25**.

Usually, the outlet **16** is connected to the sample probe **21** by a T connector as shown. The stem of the T extends into the emissions stack for collection of a sample of the emission or flue gas in normal operation. The stem is usually short.

The apparatus **10** includes a first injection unit **22** and a second injection unit **24** for introducing respective components into the delivery conduit **12**.

In the embodiment illustrated, the first injection unit **22** has a first intake **26** located within a first reservoir **28**. The first

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reservoir **28** contains a supply of an analyte **30**. The analyte **30** is generally defined by the component to be measured by the measurement device **23** to which the calibration gas **20** is being supplied. In the embodiment illustrated, the analyte **30** can be in the form of a liquid mercury chloride, or other ionic mercury solution, and the calibration gas **20** can be supplied to a mercury continuous emission monitoring system.

The first injection unit **22** has a first outlet **32** in fluid communication with the delivery conduit **12**. To provide flow of the analyte **30** from the first intake **26** to the first outlet **32**, the first injection unit **22** can be provided with a first flow inducer **34**. The first flow inducer **34** can include a precise automated liquid delivery mechanism, such as, for example, but not limited to, a calibrated piston displacement pump. The flow inducer can thus deliver a known volume of the analyte **30** to the delivery conduit **12**, to provide a known flow rate of ionic mercury, to provide a known flow rate of ionic mercury.

The second injection unit **24** has a second intake **36** in fluid communication with a second reservoir **38**. The second reservoir **38** contains a supply of humidificant **40**, and is also referred to herein as “humidifier reservoir” **38**. In the embodiment illustrated, the humidificant **40** may be in the form of deionized liquid water, but for some applications, the humidificant can comprise water containing a mixture of one or more dilute acids.

The second injection unit **24** has a second outlet **42** in fluid communication with the delivery conduit **12**. To facilitate flow of the water **40** from the second intake **36** to the second outlet **42**, the second injection unit **24** can be provided with a second flow inducer **44**. In the embodiment illustrated, the flow inducer **44** is in the form of a calibrated piston displacement pump, similar to, or the same as, the first flow inducer **34**. The second injection unit **24**, in the embodiment illustrated, can selectively deliver known amounts of the humidificant **40** to the delivery conduit **12** over a wide range of volumes and known flow rates. Levels or flow rates may, by way of example, be in the range 0.01 to 5 ml/min. This allows selective humidification of the calibration gas **20** to generally any desired level of water concentration.

The calibration gas **20**, in the embodiment illustrated, is delivered to the measurement probe in vapor phase. To convert the liquid phase analytes **30** and **40** of the illustrated embodiment into vapor phase, the apparatus **10** is provided with a vaporizer **50**. The vaporizer **50** generally provides a heated flow path **52** that has an upstream end **54** in fluid communication with the first and second outlets **32** and **42** of the first and second injection units **22**, **24**, and a downstream end **56** in fluid communication with the delivery conduit outlet **16**. The liquid analyte **30** and water **40** delivered by the first and second injection units **22** and **24** is directed through the heated flow path **52** of the vaporizer **50**, where it is converted to vapor phase. The temperature of the heated flow path **52** can be controlled by a vaporizer temperature controller **58**.

In the illustrated embodiment of the apparatus **10**, both the analyte **30** and water **40** are consumed from respective sources in the same phase (i.e. the liquid phase). Therefore, the delivery of the analyte **30** and water **40** can be controlled (metered) independently using the same principles (e.g. displacement of the generally non-compressible liquids in a positive displacement pump). In the liquid phase and at constant temperatures, they can be treated as incompressible (and in any event need not be subject to any excess pressures) so the mass flow rates are readily determined from volume flow rates.

Furthermore, supplying the components (e.g. the analyte **30**) for introduction into the carrier gas **18** in liquid form

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allows introduction of some analytes that cannot readily be supplied in gaseous form, for example mercury chloride. As well, at least for some analytes, the analyte stored in liquid form can be more stable over time than the corresponding vapor form. Stabilizing agents, such as, for example, but not limited to, weak acids can be added to the liquid analyte **30** in the analyte reservoir **28** to further enhance the stability of the analyte. As detailed below, the presence of stabilizing acids is an additional humidificant that can help to simulate the gas conditions present in normal use.

In the embodiment illustrated, the vaporizer **50** receives a combined liquid flow **60** at its upstream end **54** that includes a mixture of the analyte **30** and the humidificant **40**. The apparatus **10**, in the embodiment illustrated, is provided with a manifold **62** having an exhaust port **64** for discharging the combined liquid flow **60** to the upstream end **54** of the vaporizer **50**. The manifold **62** has first and second inlet ports **66** and **68** that are connected for fluid communication with the first and second outlets **32** and **42**, respectively, of the first and second injection units **22** and **24**, respectively.

In use, the carrier gas **18** is directed into the inlet **14** of the delivery conduit **12**. The carrier gas **18** can be a zero gas, such as air or nitrogen, for generating a “zero” reading on the measurement device **23**. The carrier gas **18** is generally dry (free of any water vapor). The supply rate of the carrier gas **18** to the delivery conduit **12** can be controlled and measured by a mass flow controller **70** located adjacent the inlet **14** of the delivery conduit **12**.

The liquid analyte **30** and humidificant **40** can be pumped from the respective reservoirs **28** and **38**, and fed to the vaporizer **50** for conversion from the liquid phase to the vapor phase, forming a mixed component gas **72** at the downstream end **56** of the heated flow path **52**. The mixed component gas **72** (i.e. water vapor and analyte vapor mixture) can then be fed into the delivery conduit **12** via a T-fitting **74**. The T-fitting **74** connects the downstream end **56** of the flow path **52** to the delivery conduit **12**.

Downstream from the T-fitting, the conduit comprises a chemically inert delivery line, optionally provided with a heater **75**, to ensure that analytes and humidificants do not condense onto the walls of the line. Whether a heater is required will depend on a number of factors such as: nature and vapor pressures of the analytes and humidificants at the prevailing temperatures; their concentrations in the gas flow

Downstream of the T-fitting **74**, the mixed component gas **72** mixes with the carrier gas **18** to form the calibration gas **20**. The calibration gas **20** can include an amount of the analyte **30** in vapor form, the concentration of which can be controlled by adjusting the rate at which the first flow inducer **34** provides the liquid analyte **30** to the first injection outlet **32**. For example, where the first flow inducer **34** includes a positive displacement pump, the number of rotations of the pump per unit time can be increased or decreased to adjust the relative concentration of the vapor analyte **30** per unit volume of the calibration gas **20**.

Similarly, the calibration gas **20** can include an amount of a humidificant, such as water **40**, in vapor form, the concentration of which can be adjusted by adjusting the rate at which the second flow inducer **44** provides the liquid humidificant **40** to the second injection outlet **42**.

The apparatus **10** can be used to supply a calibration gas **20** that has a precisely known amount of components or chemicals to be measured. The calibration gas **20** can contain a precisely known amount of water vapor to humidify the calibration gas **20**. Humidification of the calibration gas **20** can facilitate the transport of the analyte through the probe **21** and other CEM components **23** even after the surfaces have been



coated by reactive fly ash deposits. Humidification should be such as to simulate the conditions present during normal operation and sample testing. The present inventors have realized that, if this is done, then any built up fly ash or other contaminants will, it is believed, behave in essentially the same way as they do when normal stack gases etc. are passing through. In particular it is believed that with the calibration gas humidified to simulate the usual sample gas, mercury or other components of interest will not tend to be adsorbed by the fly ash. Humidification of the calibration gas **20** can also facilitate prevention of species conversion of the mercury (for example, oxidation of the mercury) by enabling members of the measurement/calibration system, such as, for example, the probe **21**, to operate at cooler temperatures.

The apparatus **10** can be provided with an electronic controller **78** for controlling and/or monitoring one or more of the first flow inducer **34**, the second flow inducer **44**, the vaporizer temperature controller **58**, the mass flow controller **70** and the temperature controller for the downstream portion of the delivery conduit **12**. The electronic controller **78** can be programmed to automatically adjust the amounts of analyte **30** and water **40** being supplied to the delivery conduit **12** for changing the respective concentrations in the calibration gas **20**, thereby facilitating multi-point automatic calibration of the measurement device **23**. The electronic controller **78** (or a separate, second controller in communication with the controller **78**) can calculate the concentration and total delivery rate of the analyte **30** in the calibration gas **20** for comparison against the corresponding values measured by the measurement device **23**. The calculations can be made on both a wet and dry gas basis. The “dry” concentration calculations, as referred to herein, exclude (or factor out) the contribution of water from the humidificant **40** and analyte **30** in increasing the calibration gas **20** volume, which reduces the concentration of the analyte **30** per unit volume of the calibration gas **20**.

In some cases, it may be desirable that the apparatus **10** provide a calibration gas **20** that includes a second analyte (not shown in FIG. 1). The second analyte can be provided as an alternative to, or in addition to, the first analyte **30**. To provide the second analyte, the carrier gas source **19** can be changed from providing a “zero gas” carrier gas **18** to a carrier gas source **19a** that includes an amount of the second analyte. For example, the carrier gas source **19a** could be in the form of a standard gas cylinder providing a carrier gas **18** having a known composition of the second analyte. If the carrier gas source **19a** can provide the carrier gas stream **18** at a known flow rate, the mass flow controller **70** can be omitted. The carrier gas source **19** can be, for example, but not limited to, a saturated vapor source containing elemental mercury at a known concentration for delivery of a known flow rate, or a calibration gas cylinder.

An alternative calibration gas delivery apparatus **110** in accordance with the present invention is shown in FIG. 2. The apparatus **110** has many similarities to the apparatus **10**, and like features are identified by like reference characters, incremented by 100. The apparatus **110** includes a first injection unit **122**, a second injection unit **124**, and a third injection unit **122a**. The third injection unit **122a** is adapted to introduce a third component **130a** into the calibration gas **120**, and has a third intake **126a** in fluid communication with a third reservoir **128a**.

In the embodiment illustrated, the third component **130a** is in the form of a second analyte. The third reservoir **128a** is adapted to contain an amount of the second analyte **130a** in liquid form. The third injection unit **122a** has a third outlet **132a** in fluid communication with the delivery conduit **112**,

and a third flow inducer **134a** to facilitate flow of the second analyte **130a** from the third intake **126a** to the third outlet **132a**. The apparatus **110** has a manifold **162** having a third inlet port **166a**, in addition to the first and second inlet ports **166**, **168**, that is connected in fluid communication with the third outlet **132a**.

The third flow inducer is, in the embodiment illustrated, similar to, or the same as, the first flow inducer **134**. The third flow inducer **134a** can be adjusted independently of the first and second flow inducers **134**, **144**, by, for example, the controller **178**. This can allow selective addition of the first and second analytes **130**, **130a** to the calibration gas **120**. Any of the three flow inducers **134**, **144**, or **134a** can also be adjusted to an off position, in which the respective component **130**, **140**, or **140a** is not supplied to the delivery conduit **112** and is absent from the calibration gas **120**. By providing for automatic adjustment of each of the three flow inducers **134**, **144**, and **134a** via the controller **178**, distinct calibration cycles can be performed automatically, providing for convenient calibration of the measurement device **123** over a wide range of concentrations of the respective components **130**, **140** and **130a**.

It is to be appreciated that, in accordance with the present invention, further additional injection units **122b**, **122c** . . . (etc.) can be provided with the apparatus **110**. The additional injection units can be controlled by the controller **178**, and can introduce other respective analytes into the calibration gas **120**. The apparatus **110** can thus conveniently provide a calibration gas **120** that with a variety of components and/or concentrations and with selectively varied levels of humidification. The humidificant **140** can be included or omitted as required; for some applications, adequate humidification may be provided by the liquid solvents of the analyte(s) and/or be included in the carrier gas **118**. The calibration gas **120** can also be configured to closely match the nature of the actual emission gas being monitored by the device **123**, by selecting components for the respective injection units that match components known to exist in the emission gas.

Another alternative calibration gas delivery apparatus **210** in accordance with the present invention is shown in FIGS. 1 and 3. The apparatus **210** has many similarities to the apparatus **10**, and like features are identified by like reference characters, incremented by 200 from FIG. 1. The apparatus **210** includes a first injection unit **222**, a second injection unit **224** and a third injection unit **222a**. The first injection unit **222** includes a first reservoir **228**, and the third injection unit **222a** includes a third reservoir **228a**. The third reservoir **228a** is adapted to supply a third component **230a** for injection into the calibration gas **230**. The first injection unit **222** includes a third intake **226a** in fluid communication with the third reservoir **228a**.

The third component **230a** can be, but need not be, a third analyte for measurement by, and calibration of, the measurement device **223**. The third component **230a** can be an agent for otherwise treating or maintaining the measurement apparatus. In the embodiment illustrated, the third component **230a** is in the form of an acidic liquid solution that, when combined in vapor form with the carrier gas **218**, can serve to clean and condition members of the measurement system including, for example, the probe **221**.

The first injection unit **222** is provided with first and third intakes **226** and **226a** that are in selective, adjustable fluid communication with the first outlet **232** via an intake valve **280**. The intake valve **280** can provide adjustment of the relative amount of the first and third components **230**, **230a** to be introduced into the calibration gas **220**.

In the embodiment illustrated, the intake valve **280** is a three-port, two-position valve. The valve **280** has a first valve inlet **282** in fluid communication with the first intake **226** of the first injection unit **222**. The valve **280** has a second intake **284** in fluid communication with the third intake **226a** of the first injection unit **222**. The valve **280** has a valve outlet **286** in fluid communication with the flow path **252** of the vaporizer **250**.

When the intake valve **280** is in the first valve position, the valve outlet **286** is in fluid communication with the first valve inlet **282** and isolated from the second valve inlet **284**, for drawing an amount of the first analyte **230**. When the intake valve **280** is in the second valve position, the valve outlet **286** is in fluid communication with the second valve inlet **284**, and is isolated from the first valve inlet **282**, for drawing an amount of the third component **230a**.

By changing the position of the valve (from the first to the second valve positions and back again), the first and third components **230**, **230a**, respectively, can alternatively be introduced into the calibration gas **220**. Where the third component **230a** is, as in the illustrated embodiment, formed of a cleaning solution, the valve **280** can be moved from the first position to the second position to perform a periodic cleaning cycle. The valve **280** can be electronically controlled, and can be moved between the first and second positions by the controller **278**. Periodic cleaning can thus be conducted automatically at regularly scheduled (i.e. programmed) intervals. The cleaning step can be conducted shortly prior to the actual measurement of the calibration gas by the device **123** to enhance the accuracy and repeatability of the overall calibration procedure.

While techniques are known for generating and supplying calibration gases for calibration of various types of measuring equipment, a common characteristic of known systems is simply to provide an analyte of interest in a desired concentration in a carrier gas; no attempt is made to replicate the general characteristics of the sample gas usually sampled and measured by the measuring equipment. Correspondingly, it has not been realized that, in the detection of some analytes, the characteristics of the calibration gas stream can strongly influence the ability of the measuring equipment to transport said calibration gases.

This characteristic is evident when: the analyte to be detected may only be present at extremely low levels; the analyte has a strong tendency to react with materials present or be absorbed on available surfaces; the usual gas environment is complex and includes various materials, e.g. fly ash and other particulates, that can promote such reaction of absorption. All these features are usually found in the detection of mercury in stack gases.

Accordingly what the present inventors have realized is that the calibration gas, with an entrained analyte should simulate the gas usually sampled, to the extent necessary to prevent the occurrence of the effects listed in the preceding paragraph. Thus, for mercury detection, and in particular for the detection of ionic mercury, one needs to be aware that the environment in a flue stack includes deposited fly ash conducive to causing mercury to absorb or deposit on it; and the actual flue gas is usually a complex and acidic matrix that strongly effects the behaviour of the mercury. Thus any significant change in the composition of the gas flowing through the flue stack can significantly alter the behaviour of the mercury. This is commonly handled by running existing calibration equipment for a long enough time for a steady state to be reached; for example if the use of a dry gas promoted deposition of mercury, then the calibration is run for a long enough time for sufficient mercury to be deposited that a

steady state is reached where there is no net additional deposition and all the mercury as the analyte in the calibration gas passes through to the measuring equipment. However, in many cases the time required is too long and is not acceptable.

Accordingly the present invention provides a method and apparatus intended to overcome, or at least mitigate these problems. To achieve this, a number of variations in the described embodiments are within the scope of the present invention.

For the carrier gas supplied, this could comprise a combination of two or more gases, either supplied separately and mixed by the apparatus of the present invention, or supplied in an already mixed form. Additionally, one or more the supplied gases can include one or more analytes of interest at a desired concentration. For example one of the gas sources, or the only gas source, can be an elemental mercury source, such as the Tekran® Model 3310 CEM Mercury Calibrator, that serve to supply simultaneously a desired gas and an analyte. Any other conventional calibration gas generator can be used, with the basic requirement being that it produces a known concentration of analyte; it can also be such as to measure or regulate the flow rate of the gas/analyte mixture, or if this feature is not present, mass flow regulation can be provided by the apparatus of the present invention.

The number of analyte sources provided can vary. At a minimum, if the gas source also provides the analyte of interest, e.g. an elemental mercury source, then it may only be necessary to supply humidificant to the gas flow, i.e. there would be no separate supply of an analyte solution. Otherwise, there can be one, two or more separate supplies of analyte in solution. Each of these can include one or more analytes, and in each case, the solvent can be simple single solvent or it can be a mixture.

The humidificant supply may not always be needed. The concept is that any analyte in solution will provide a base level of humidification, the characteristics of which will depend on the solvent and the concentration of the analyte in the solution. (Commonly gases provided in compressed state in a cylinder are essentially dry). Then, as required, additional humidification is added to give the calibration gas a desired level of humidification, both in terms of relative humidity level(s) and also in terms of the humidificants used. These can include, in addition to water, various acids that simulate the properties of usual sample gases.

Where the apparatus and method are used to supply mercury for calibration, typically levels are 0.05-200  $\mu\text{g}/\text{m}^3$  both elemental and ionic mercury, but note that measuring equipment being calibrated may be capable of measuring to much higher levels, e.g. 5000  $\mu\text{g}/\text{m}^3$  and hence may need to be calibrated at that level. For the humidity level, this should usually correspond to that found in a stack, e.g. 6-10%, and more generally a humidity level in the range 1-30% could be used. Also, it is envisaged that the present invention can be used to clean the probe **21** by providing a high level of humidity, e.g. 100%, without any analyte.

It is to be understood that what has been described are preferred embodiments of the invention. The invention nonetheless is amenable to certain changes and alternative embodiments without departing from the subject invention, the scope of which is defined in the following claims.

The invention claimed is:

1. A calibration gas delivery apparatus for delivering a selectively humidified calibration gas to a measurement probe, the apparatus comprising:

a) a delivery conduit having a conduit inlet adapted to receive a carrier gas stream and a conduit outlet for delivering a calibration gas stream;

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- b) a mass flow controller in the delivery conduit for controlling the flow of the carrier gas;
- c) a first injection unit having a first intake in fluid communication with a first reservoir and a first outlet in fluid communication with the delivery conduit, the first reservoir adapted to hold an ionic mercury solution, the first injection unit delivering the ionic mercury solution at a known flow rate in liquid form;
- d) a second injection unit having a second intake in fluid communication with a second reservoir and a second outlet in fluid communication with the delivery conduit, the second reservoir adapted to hold water as a humidificant, the second injection unit delivering the water at a known flow rate; and
- e) at least one vaporizer downstream of the first and second outlets and upstream of the conduit outlet for converting the ionic mercury solution and the water into vapor form and delivering a calibration gas including the carrier gas, ionic mercury, and water vapor to the conduit outlet, whereby the concentrations of the ionic mercury and water in the carrier gas are known.
2. The apparatus of claim 1 further including respective first and second flow inducers for drawing fluid from the first and second reservoirs and providing precise amounts of the first analyte and humidificant liquids to the first and second outlets, respectively.
3. The apparatus of claim 2 wherein each of the first and second injection units is selectively adjustable to adjust the amounts of the first analyte and humidificant liquids being delivered to the respective first and second outlets.
4. The apparatus of claim 2 wherein each of the first and second flow inducers comprises a precision displacement pump.
5. A calibration gas delivery apparatus for delivering a selectively humidified calibration gas to a measurement probe, the apparatus comprising:
- a delivery conduit having a conduit inlet adapted to receive a carrier gas stream and a conduit outlet for delivering a calibration gas stream;
  - a mass flow controller in the delivery conduit for controlling the flow of the carrier gas;
  - a first injection unit having a first intake in fluid communication with a first reservoir and a first outlet in fluid communication with the delivery conduit, the first reservoir adapted to hold an ionic mercury solution, the first injection unit delivering the ionic mercury solution at a known flow rate in liquid form;
  - a second injection unit having a second intake in fluid communication with a second reservoir and a second outlet in fluid communication with the delivery conduit, the second reservoir adapted to hold water as a humidificant, the second injection unit delivering the water at a known flow rate;
  - a manifold having first and second inlet ports connected to the first and second outlets, respectively, of the first and second injection units, and an exhaust port;
  - a vaporizer in the delivery conduit and connected to the manifold exhaust port and including a heater for converting the ionic mercury solution and the water into vapor form and for delivering a calibration gas including the carrier gas, analyte vapor, and humidificant vapor to the conduit outlet, whereby the concentrations of the ionic mercury and the water in the carrier gas are known.
6. The apparatus of claim 1, including at least one additional injection unit and a corresponding additional reservoir for at least one of an additional analyte and an additional

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- humidificant, for delivering said at least one of an additional analyte and an additional humidificant at a known flow rate.
7. The apparatus of claim 6, further including a valve connected to at least two of the injection units for switching therebetween.
8. The apparatus of claim 1, wherein the conduit inlet is adapted to receive at least two carrier gas streams.
9. A calibration gas delivery apparatus for delivering a selectively humidified calibration gas to a measurement probe, the apparatus comprising:
- a delivery conduit having a conduit inlet adapted to receive at least one carrier gas stream including an analyte and a conduit outlet for delivering a calibration gas stream at a known flow rate;
  - an injection unit having an intake in fluid communication with a reservoir and an outlet in fluid communication with the delivery conduit, the reservoir adapted to hold a water, the injection unit delivering the water at a known flow rate;
  - at least one vaporizer downstream of the outlet and upstream of the conduit outlet for converting the water to vapor form and delivering a calibration gas including all of said at least one carrier gas and water vapor to the conduit outlet, whereby the concentration of the water vapour in the carrier gas stream is known;
  - an elemental mercury source that provides the carrier gas stream with elemental mercury entrained therein, wherein the conduit inlet is in fluid communication with the elemental mercury source, and the elemental mercury is provided at a known flow rate, whereby the concentration of the elemental mercury in the carrier gas stream is known.
10. A calibration gas delivery apparatus for delivering a selectively humidified calibration gas including mercury chloride to a measurement probe, the apparatus comprising,
- a delivery conduit having a conduit inlet adapted to receive a carrier gas stream and a conduit outlet for delivering a calibration gas stream;
  - a mass flow controller in the delivery conduit for controlling the flow of the carrier gas;
  - a first injection unit having a first intake in fluid communication with a first reservoir and a first outlet in fluid communication with the delivery conduit, the first reservoir holding a liquid mercury chloride solution and the first injection unit delivering the mercury chloride solution at a known flow rate;
  - a second injection unit having a second intake in fluid communication with a second reservoir and a second outlet in fluid communication with a delivery conduit, the second reservoir adapted to hold (a water, the second injection unit delivering the water at a known flow rate; and
  - at least one vaporizer downstream of the first and the second outlets and upstream of the conduit for converting the mercury chloride solution and the water to vapour form and delivering a calibration gas including the carrier gas, mercury chloride vapour and water vapour to the conduit outlet, with known concentrations of water vapor and mercury chloride.
11. A calibration gas delivery apparatus for delivering a selectively humidified calibration gas to a measurement probe, the apparatus comprising:
- a delivery conduit having a conduit inlet adapted to receive a carrier gas stream and a conduit outlet for delivering a calibration gas stream;
  - a mass flow controller in the delivery conduit for controlling the flow of the carrier gas;

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- c) a first injection unit having a first intake in fluid communication with a first reservoir and a first outlet in fluid communication with the delivery conduit, the first reservoir adapted to hold a first analyte in liquid form and the first injection unit delivering the analyte at a known flow rate; 5
- d) a second injection unit having a second intake in fluid communication with a second reservoir and a second outlet in fluid communication with the delivery conduit, the second reservoir adapted to hold a water and to deliver the water at a know flow rate; 10
- e) a manifold having first and second inlet ports connected to the first and second outlets, respectively, of the first and second injection units, and an exhaust port;

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- f) a vaporizer in the delivery conduit and connected to the manifold exhaust port and including a heater for converting the analyte and the water to vapor form and for delivering a calibration gas including the carrier gas, analyte vapor, and water vapor to the conduit outlet, wherein the vaporizer is connected to a T fitting in the delivery conduit, whereby the analyte and the humidificand liquids are heated prior to being supplied to the delivery conduit, and whereby the concentrations of the analyte and water vapor in the carrier gas stream are known.
- 12.** An apparatus is claimed in claim **11**, including a heater for the delivery conduit downstream from the T fitting.

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