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(54) **CARBIDE NANOSTRUCTURES AND METHODS FOR MAKING SAME**

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H01J 9/12 (2006.01)

(52) **U.S. Cl.** **445/49; 445/51**

(58) **Field of Classification Search** **445/49-51**
See application file for complete search history.

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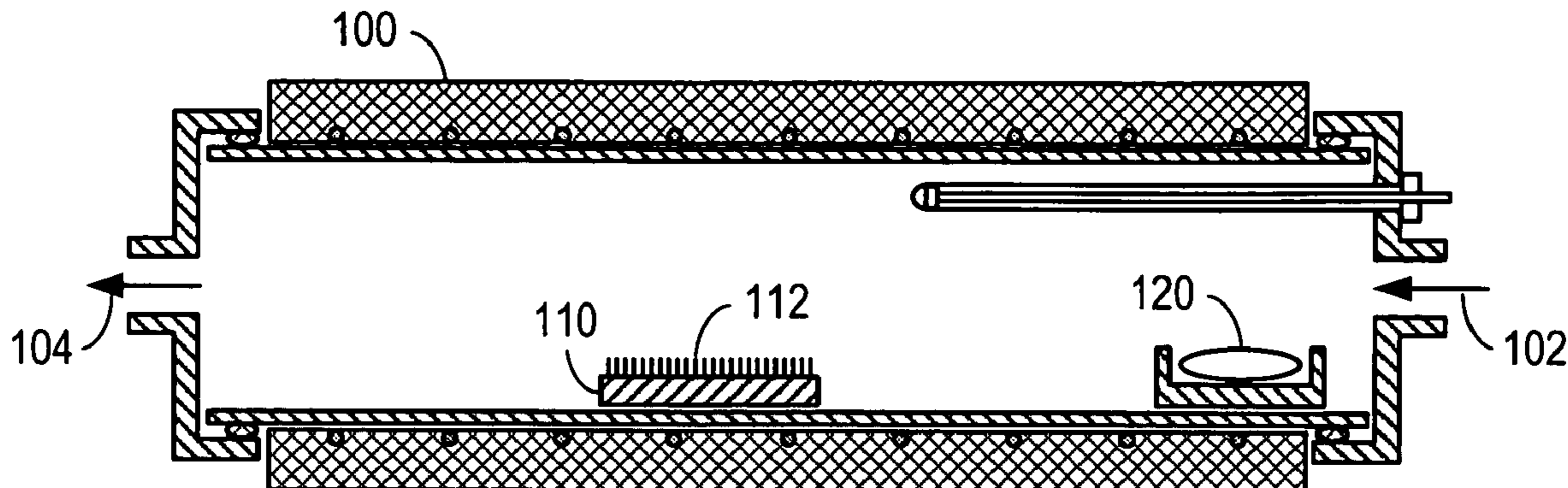
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(57) **ABSTRACT**

A structure includes a substrate and a metallized carbon nano-structure extending from a portion of the substrate. In a method of making a metallized carbon nanostructure, at least one carbon structure formed on a substrate is placed in a furnace. A metallic vapor is applied to the carbon nano-structure at a preselected temperature for a preselected period of time so that a metallized nanostructure.

16 Claims, 3 Drawing Sheets



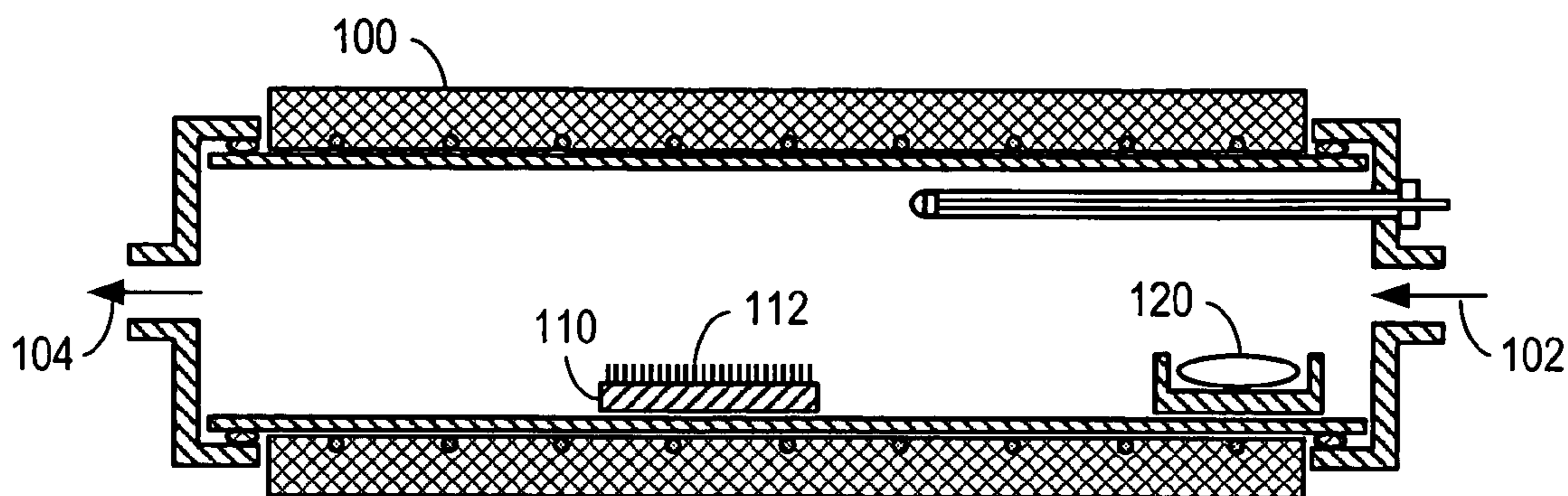


FIG. 1

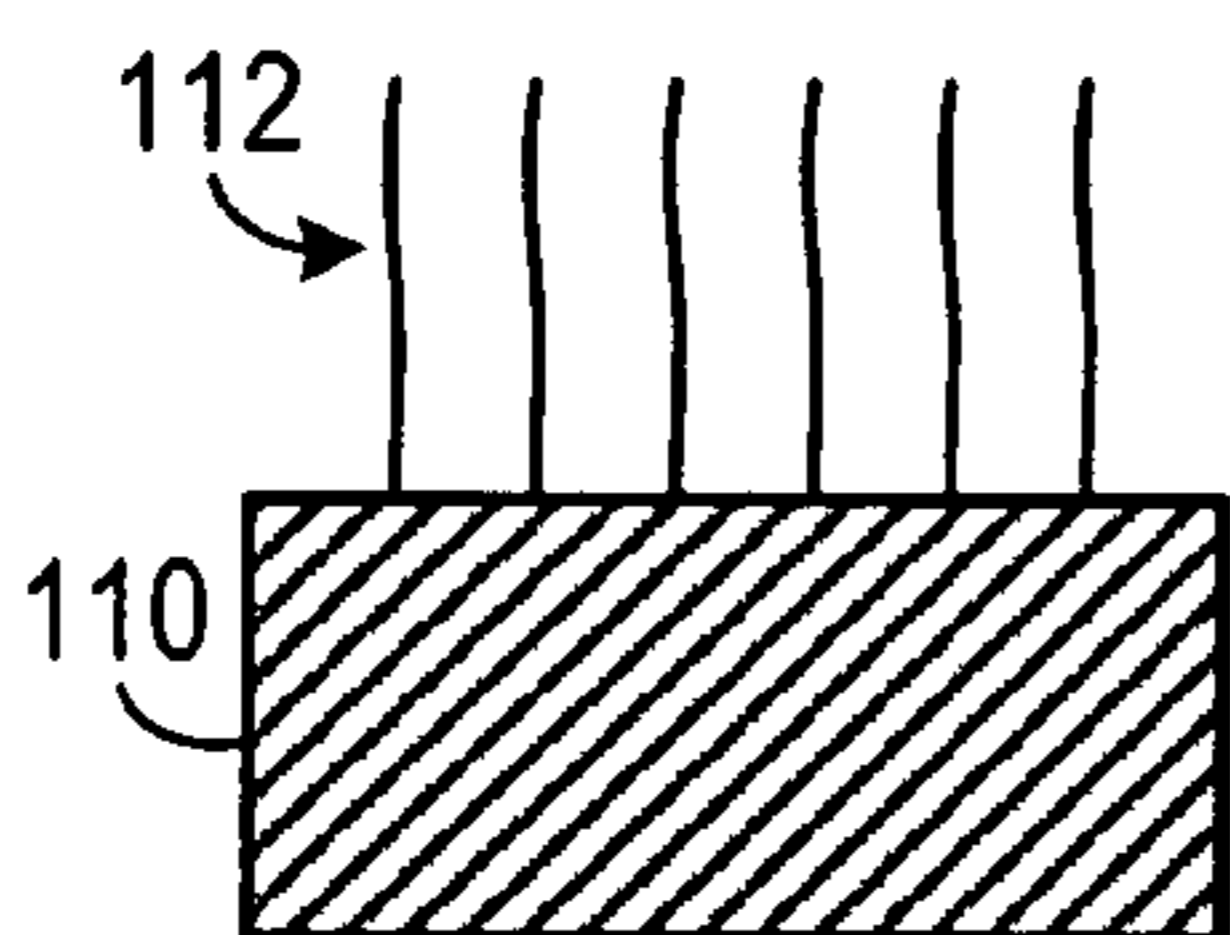


FIG. 2A

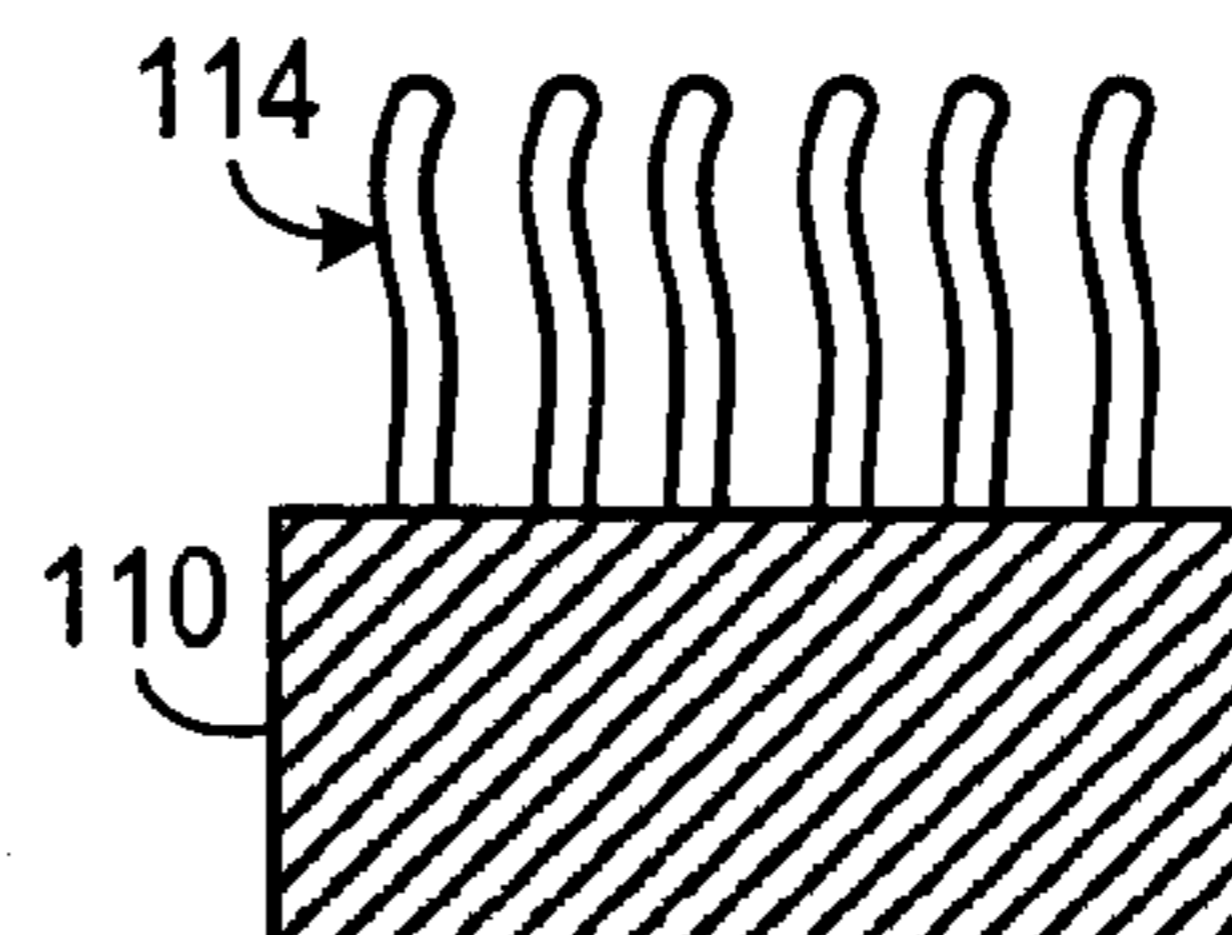


FIG. 2B

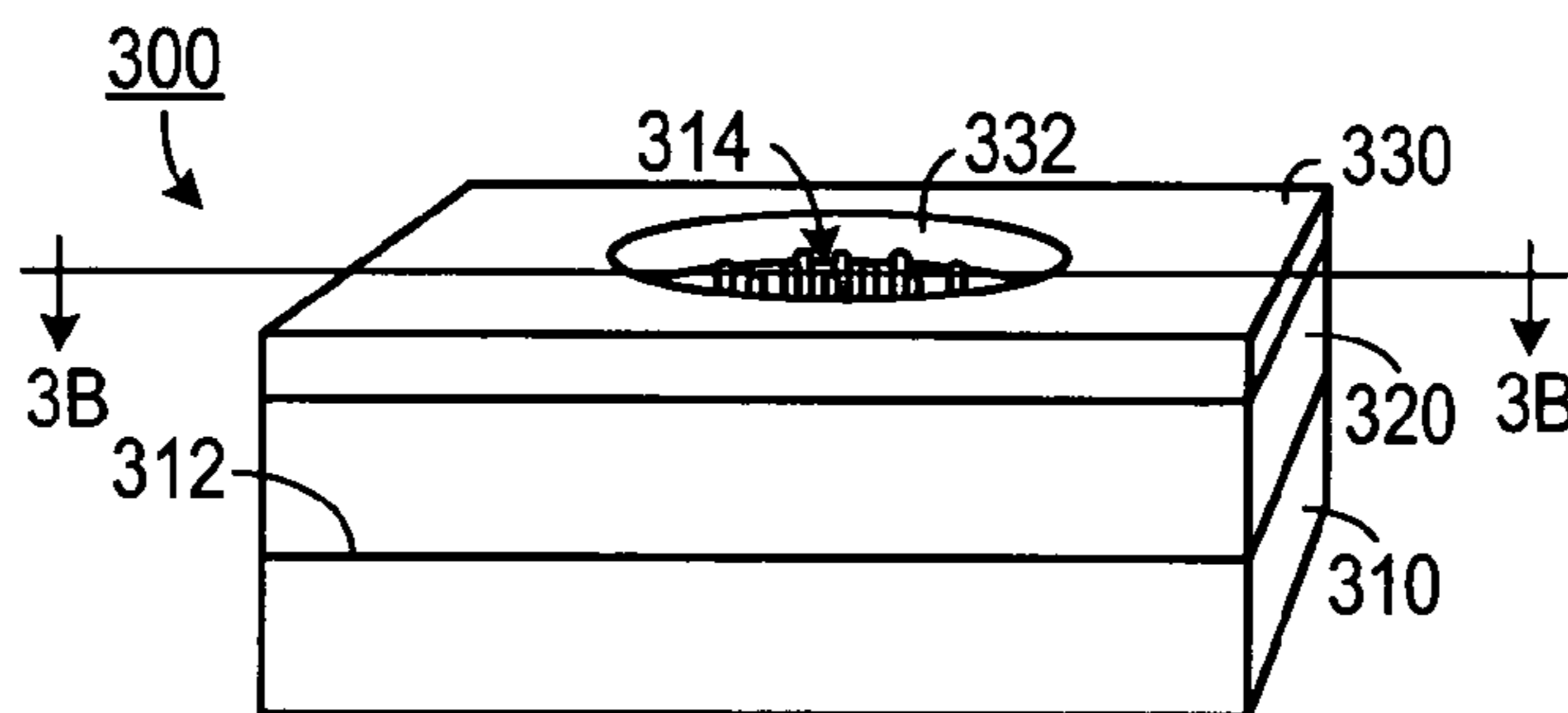


FIG. 3A

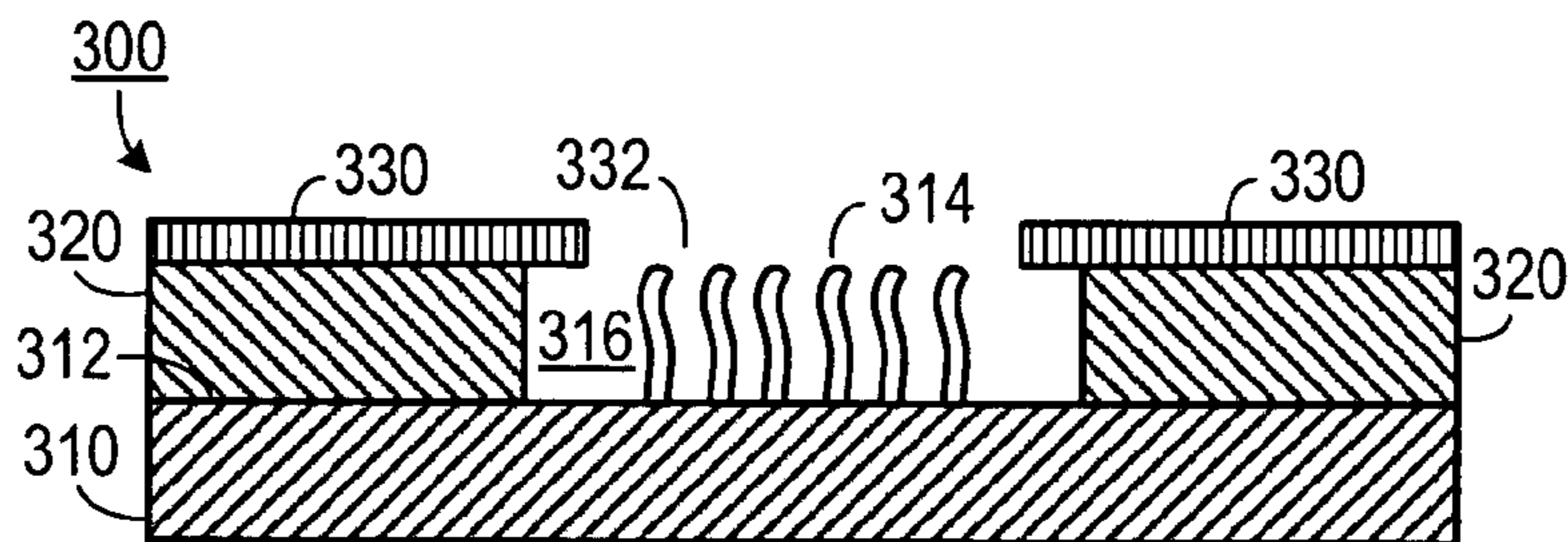


FIG. 3B

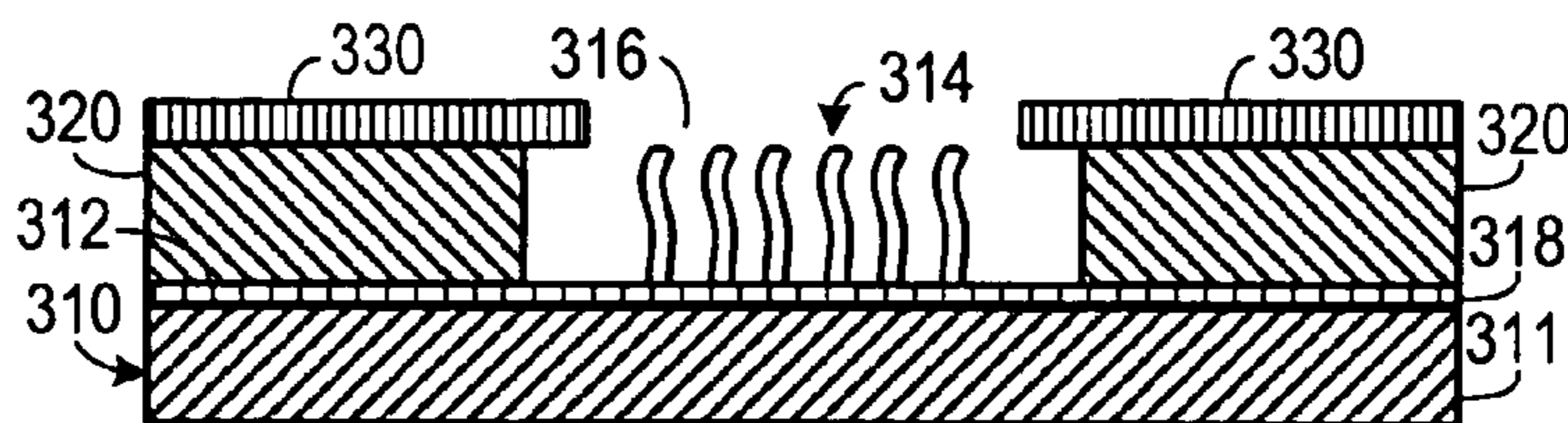


FIG. 3C

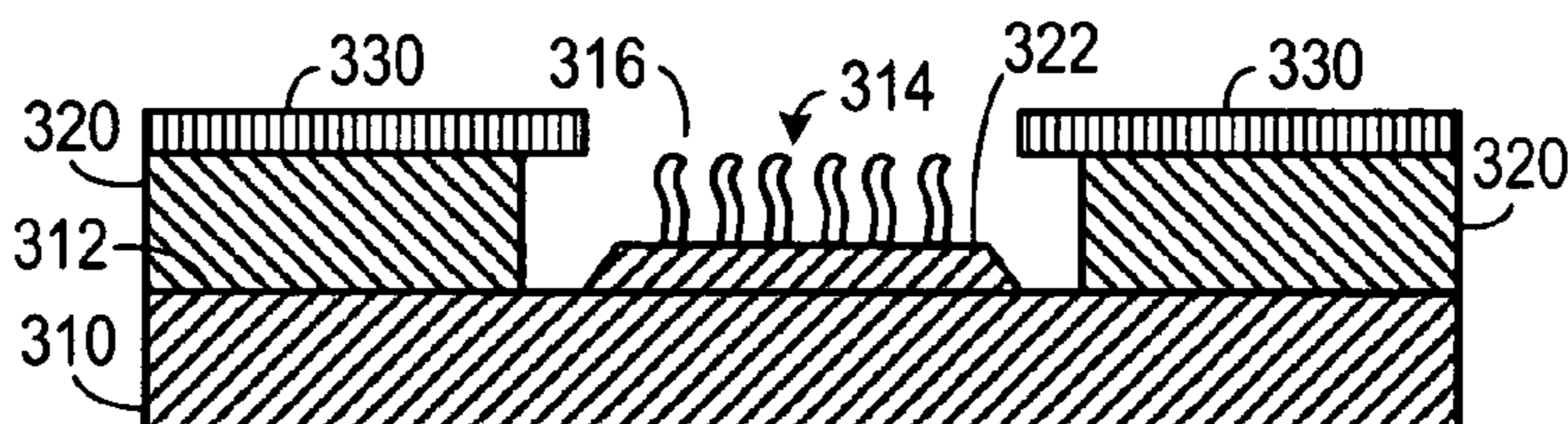


FIG. 3D

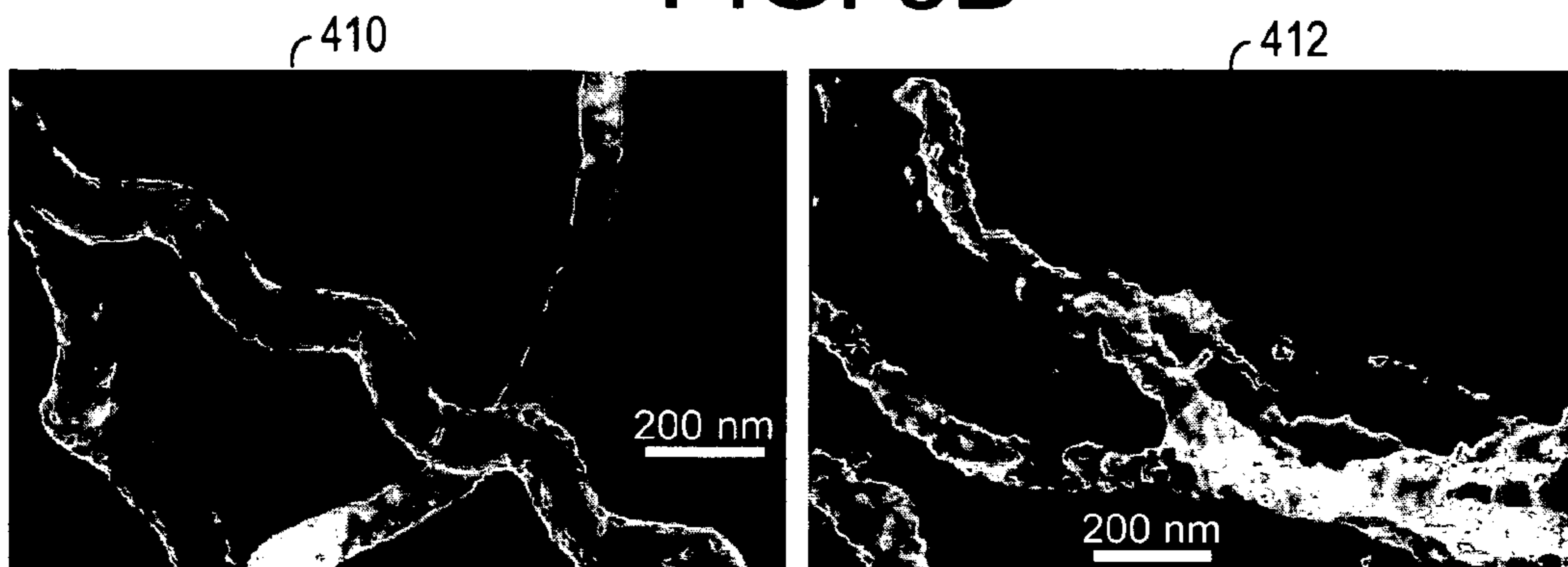


FIG. 4A

FIG. 4B

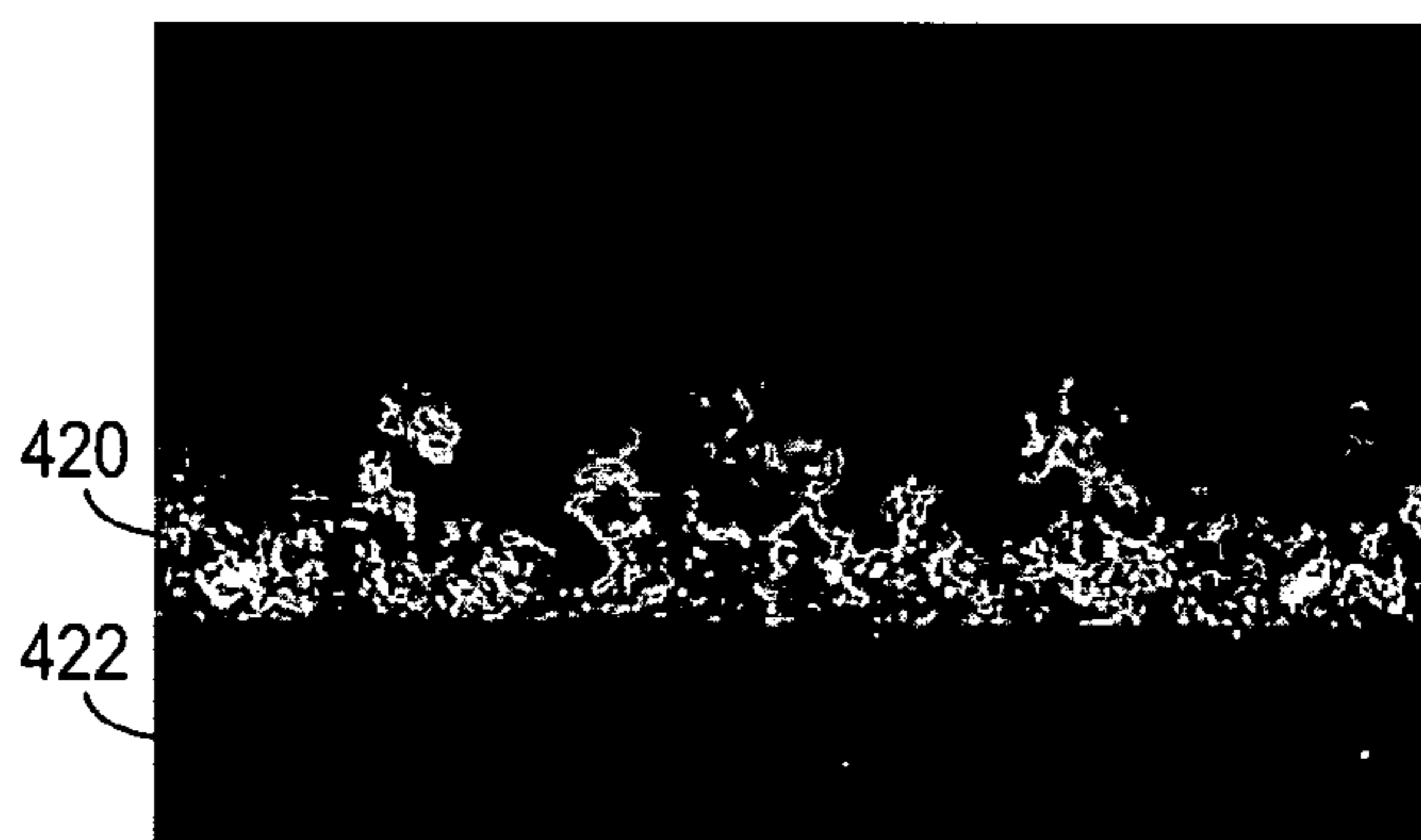


FIG. 4C

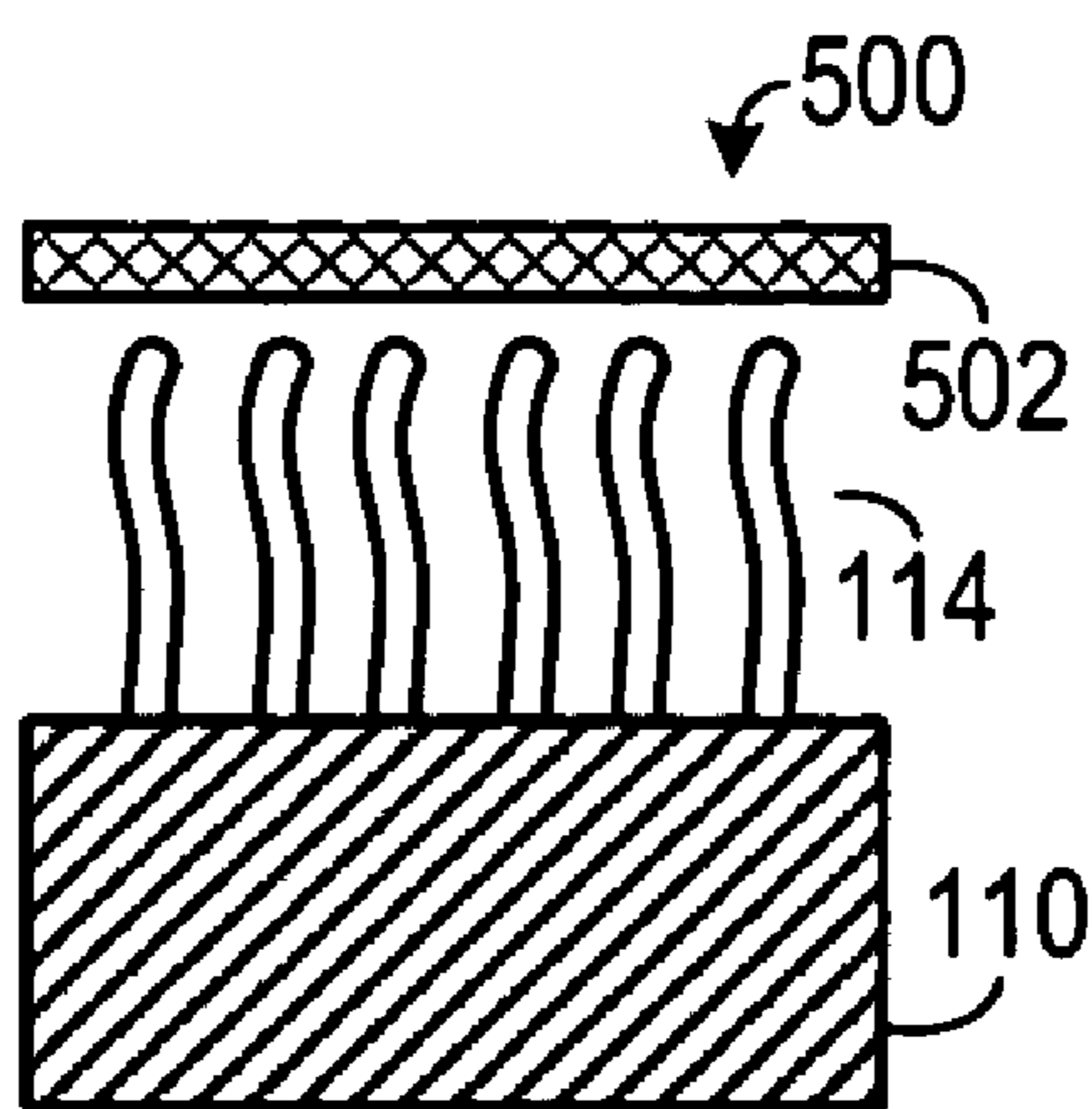


FIG. 5A

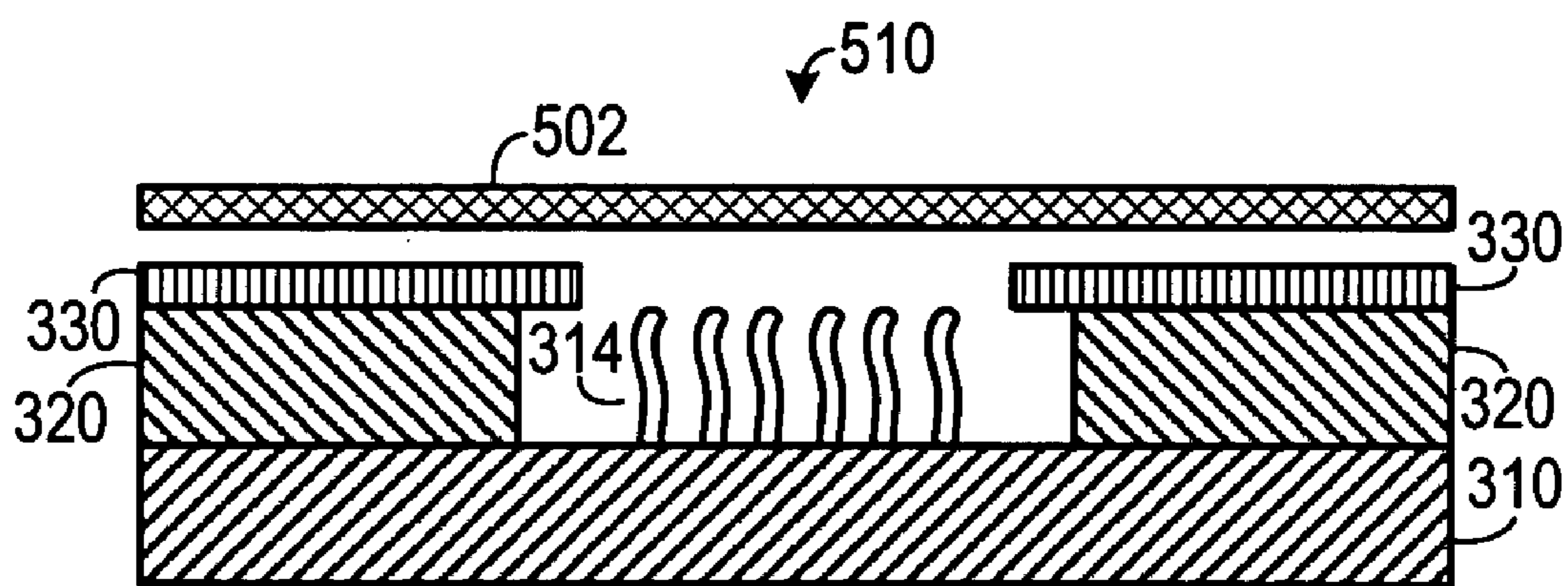


FIG. 5B

CARBIDE NANOSTRUCTURES AND METHODS FOR MAKING SAME

FEDERALLY SPONSORED RESEARCH

The United States Government may have certain rights in this invention pursuant to Cooperative Agreement No. 70NANB2H3030, awarded by the National Institute of Standards and Technology, United States Department of Commerce.

BACKGROUND

1. Field of the Invention

The invention relates to nano-scale structures and, more specifically, to a method of making nano-structures by carbon nanotube confined reactions.

2. Description of the Prior Art

Field emission devices (gated or ungated) have applications in X-ray imaging systems, displays, electronics, microwave amplifiers, fluorescent lamp cathodes, gas discharge tubes, and many other electrical systems. Other applications for field emission devices include sensors, photonic bandgap devices, and wide bandgap semiconductor devices.

Carbon nanotubes are currently being researched as electron emission sources in, for example, flat panel field emission display applications, microwave power amplifier applications, transistor applications and electron-beam lithography applications. The carbon nanotubes are typically synthesized through an arc discharge method, a chemical vapor deposition (CVD) method or a laser ablation method. Carbon nanotubes offer the advantage of having high aspect ratios which increases the field enhancement factor and therefore the extraction of electrons at relatively low electric fields. Carbon nanotubes, however, exhibit a fairly high work function, and are prone to damage under typical operating conditions, limiting the life and effectiveness of the devices. What is needed, therefore, is a material more robust and with a lower work function than carbon, but with a cylindrical geometry and diameters in the 10-800 nm range.

Metal carbide nanorods are candidate materials for use in field emission gated devices, which have applications in stationary computed tomography systems, displays, etc. A fabrication procedure is required that allows for seamless integration with gated device structures as well as control of the lateral density of nanorods so that electric field shielding does not occur. Carbide materials may be preferred due to their chemical stability, mechanical hardness and strength, high electrical conductivity, and relatively low work function. These characteristics make them particularly suited to the environment that may be found in a computed tomography system and the like. Such materials may also be important in superconducting nanodevices, optoelectronic nanodevices, display systems, lighting systems and other similar systems.

The main approach to synthesizing carbide nanorods to date has been to use a carbon nanotube (CNT) as a template on which a reaction is carried out between the CNT and a metal, metal oxide, or metal iodide in vapor form to produce metal carbide nanorods. This process is typically done in a vacuum-sealed tube with reaction times being more than 24 hours. However, demonstration of CNT conversion on a substrate such as silicon or in a device structure is not known.

Therefore, there is a need for a system that generates metallized carbon nanostructures from a substrate in a relatively short period of time.

There is also a need for a system that grows elongated nanostructures in situ directly in device structures.

SUMMARY OF THE INVENTION

The disadvantages of the prior art are overcome by the present invention, which, in one aspect, is a structure that includes a substrate and a metallized carbon nano-structure extending from a portion of the substrate.

In another aspect, the invention is a field emitter that includes a conductive substrate, an insulating layer, a conductive gate layer, and at least one metallized carbon nano-structure. The conductive substrate has a top surface. The insulating layer is disposed adjacent to the top surface of the conductive layer and defines a passage that exposes a portion of the top surface of the substrate. The conductive gate layer is disposed adjacent to the insulating layer and defines a hole that is in substantial alignment with the passage defined by the insulating layer. The metallized carbon nano-structure extends from the portion of the top layer of the substrate through at least a portion of the passage defined by the insulating layer.

In another aspect, the invention is a method of making a metallized carbon nanostructure, in which at least one carbon structure formed on a substrate is placed in an furnace. A metallic vapor is applied to the carbon nanostructure at a preselected temperature for a preselected period of time so that a metallized nanostructure forms about the carbon nanostructure.

In yet another aspect, the invention is a method of making a field emitter, in which an insulating layer is placed on a conductive substrate. A conductive gate layer is placed on the insulating layer, opposite the conductive substrate. An opening is defined in the conductive gate layer and the insulating layer, so as to expose a portion of the conductive substrate. At least one carbon nanostructure is grown on the exposed portion of the substrate. A metal vapor is applied to the carbon nanotube so as to form a metallized carbon nanostructure in the opening.

These and other aspects of the invention will become apparent from the following description of the preferred embodiments taken in conjunction with the following drawings. As would be obvious to one skilled in the art, many variations and modifications of the invention may be effected without departing from the spirit and scope of the novel concepts of the disclosure.

BRIEF DESCRIPTION OF THE FIGURES OF THE DRAWINGS

FIG. 1 is a schematic view of a tube furnace used in a method according to one illustrative embodiment of the invention.

FIG. 2A is a cross-sectional view showing a plurality of carbon nanotubes extending from a substrate.

FIG. 2B is a cross-sectional view showing a plurality of metallized carbon nanotubes extending from a substrate.

FIG. 3A is a perspective view of a field emitter.

FIG. 3B is a cross-sectional view of the field emitter shown in FIG. 3A, taken along line 3B-3B.

FIG. 3C is a cross-sectional view of a field emitter in which a conductive layer is deposited on the substrate.

FIG. 3D is a cross-sectional view of a field emitter in which the substrate includes a raised portion.

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FIG. 4A is a micrograph of a carbon nanotube metallized with an oxide.

FIG. 4B is a micrograph of a carbon nanotube metallized with a metal.

FIG. 4C is a micrograph of a cross section of NbC nano-structures on a substrate.

FIG. 5A is a cross-sectional view of a field emitter diode configuration.

FIG. 5B is a cross-sectional view of a field emitter triode configuration.

DETAILED DESCRIPTION OF THE INVENTION

A preferred embodiment of the invention is now described in detail. Referring to the drawings, like numbers indicate like parts throughout the views. As used in the description herein and throughout the claims, the following terms take the meanings explicitly associated herein, unless the context clearly dictates otherwise: the meaning of “a,” “an,” and “the” includes plural reference, the meaning of “in” includes “in” and “on.” Unless otherwise specified herein, the drawings are not necessarily drawn to scale.

Also, as used herein, nanostructure means a structure having a narrowest dimension diameter of less than 800 nanometers (nm). as used herein, “metallized carbon nano-structures” include metal carbide nanostructures and metal carbide structures with carbon nanotube cores.

Metal carbide nanostructures are made as disclosed below. Such nanostructures may be used to form field emitting devices. Initially carbon nanotubes, either grown from a substrate through methods known to the art (such as by an in situ catalyst-based carbon nanotube procedure) or deposited onto a substrate in the form of a paste, are exposed to a metal vapor and form metallized nanostructures. Different types of carbon nanotubes (e.g., single-walled nanotubes, double-walled nanotubes and multi-walled nanotubes) may be employed. The type of nanotube employed will affect the properties of the resulting nanostructures.

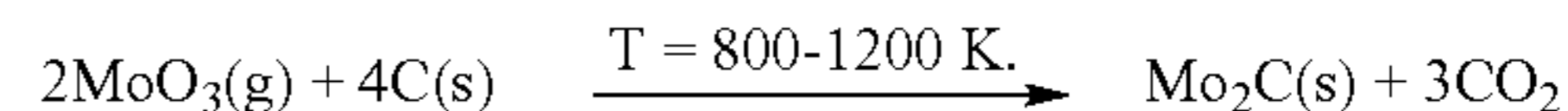
In one example of a method of making metallized carbon nanostructures, as shown in FIG. 1, carbon nanotubes **112**, or other carbon nanostructures, disposed on a substrate **110** are placed in an furnace **100** (such as a quartz tube furnace), along with a metal precursor **120**. The metal precursor **120** could be a metal, such as a transition metal, or an oxide of a metal. Examples of suitable metals include: molybdenum, tungsten, zirconium, hafnium, silicon, niobium, titanium, and tantalum. Combinations of these metals, or oxides of these metals, or iodides, bromides, organometallics, etc. employing these metals may also give desirable results. The use of a metal oxide as the metal precursor **120** may be desirable in some embodiments because the resulting vapor pressure may be higher than with a non-oxidized metal.

The furnace **100** is heated to a preselected temperature for a preselected period of time so that a least a portion of the metal precursor **120** vaporizes and metallized nanostructures form about the carbon nanostructures **112** as part of a chemical vapor deposition process. A gas is used to carry the metal vapor to the carbon nanotubes **112**.

In one illustrative example, an inert gas (e.g., argon) at a gas flow rate of between 300 sccm and 1000 sccm, is forced to flow through the furnace at a predetermined gas flow rate, entering at a gas intake **102** and flowing out through an exit **104**. Alternatively, a combination of an inert gas and a reducing gas may be used. For example, the inert gas could include argon applied at a gas flow rate of between 300 sccm and 1000 sccm, and the reducing gas could include hydrogen

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at a gas flow rate that comprises between 1 sccm and 300 sccm. Adding a reducing gas reduces the effects of residual oxygen in the furnace **100** associated with some precursors. In one exemplary embodiment of the method, the pressure inside the furnace **100** is approximately one atmosphere, the temperature is maintained in a range from about 800 K to about 1200 K and the metallized carbon nanostructures are formed in about one to four hours. In this embodiment, using molybdenum oxide as the metal precursor, the following reaction takes place:



As shown in FIG. 2A, in executing the above-described method, a plurality of carbon nanotubes **112**, which extend from the substrate **110** (which may include a material such as, for example, silicon or silicon carbide), are placed in the furnace. As shown in FIG. 2B, the chemical vapor deposition process transforms the carbon nanotubes into metallized carbon nanostructures **114**. Depending on the time and temperature parameters employed, and the type of carbon nanotube used, the metallized carbon nanostructures **114** could be metal carbide throughout, they could have a carbon core with a carbide exterior, or they could have a carbon or carbide core surrounded by a metal sheath.

Selection of the metal influences the work function of the resulting device. The following table lists work functions associated with various metal carbides, and gallium nitride (GaN), which is included for comparison:

Material	Work Function (eV)
Mo ₂ C	3.8-4.74
WC	3.6
ZrC	2.1-4.39
HfC	2.04-4.15
SiC (0001)	4.58
NbC	2.24-4.1
TiC	2.35-4.12
GaN	2.9-3.9

As shown in FIGS. 3A and 3B, a field emitter **300** may be constructed according to the above-disclosed method. The field emitter **300** includes a conductive substrate **310** having a top surface **312**. An insulating layer **320** is disposed adjacent to the top surface **312** of the conductive layer **310**. The insulating layer **320** defines a passage **316** that exposes a portion of the top surface **312** of the substrate **310**. A conductive gate layer **330** is disposed adjacent to the insulating layer **320** and defines a hole **332** that is in substantial alignment with the passage **316** defined by the insulating layer. A plurality of metallized carbon nano-structures **314** extends from the portion of the top layer **312** of the substrate **310** through at least a portion of the passage **316** defined by the insulating layer **320**.

As shown in FIG. 3C, the conductive substrate **310** could include a conductive layer **311** and a conductive film **318** (which could include titanium nitride) that acts as a diffusion barrier disposed on the conductive layer **311**.

As shown in FIG. 3D, the substrate **310** may include a raised portion **322** extending from a top surface **312** of the substrate **310**. The raised portion **322** could be conical in shape. The metallized carbon nano-structures **314** extend upwardly from the raised portion **322**.

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Two micrographs of structures made according to the disclosed method are shown in FIGS. 4A and 4B. These figures show carbide structures in which the following reaction occurred:

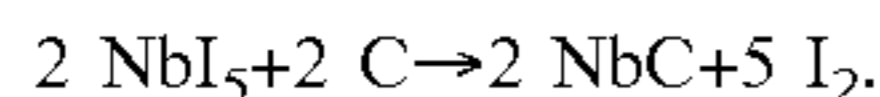


FIG. 4A shows several metallized carbon nano-structures formed from metallization of carbon nanotubes **412** in which the metal precursor used was a metal iodide. A cross-sectional micrograph of NbC nano-structures **420** on a substrate **422**, made according to the method disclosed herein, is shown in FIG. 4C.

As shown in FIGS. 5A and 5B, the addition of an anode **502** may be used in constructing an active electronic device. For example, FIG. 5A shows a diode **500** configuration of a field emitter and FIG. 5B shows a triode **510** configuration of a field emitter. If the anode **502** is coated with a phosphor, then the active device may generate light when a suitable potential is applied. Such devices could be used in imaging systems, such as, for example, x-ray imaging systems, displays, and in lighting systems. Devices made as disclosed herein would result in a lower work function than conventional field emitters and make systems employing field emitters more robust. Also, as is understood, there are many field emitter configurations that could be made according with the present disclosure (e.g., different spatial dispositions of the gate, different materials added to the anode, etc.).

Field emitter **300** may also be serve as a cathode to generate an electron beam in an x-ray generating device, or x-ray source. In this application, a plurality of field emitters **300** are arranged in an array. Electrons emitted by field emitters **300** impinge upon an anode, which in turn emits x-rays. Such an x-ray source may be used in imaging systems, such as computer tomography systems and the like, in medical, inspection, and other radiographic applications. The use of field emitters as x-ray sources are described in U.S. Pat. No. 6,385,292 by Bruce M. Dunham et al., entitled "Solid-State CT System and Method," issued on May 7, 2002; United States Patent Application Publication US 2002/0085674 by Bruce M. Dunham et al., entitled "Radiography Device with Flat Panel X-Ray Device," filed on Dec. 29, 2000; and United States Patent Application Publication US 2003/0002628 by Colin R. Wilson et al., entitled "Method and System for Generating an Electron Beam in X-Ray Generating Devices," filed on Jun. 27, 2001, the contents of which are incorporated by reference herein in their entirety.

Additional coatings may be applied to the resulting devices to achieve desired properties. For example, certain coatings to the structures (which could be applied in a chemical vapor process or through some other chemical process) could lower the work function of the devices and/or could provide additional protection from ion bombardment. An example of this is coating of the carbide nanostructures by a boron nitride or barium oxide coating.

One typical method of growing carbon nanotubes includes placing a film of a catalyst on a substrate, heating the catalyst so that thin film forms islands, and applying a carbon compound to the islands to grow carbon nanotubes. The catalyst islands may remain at substrate or they may stay at the growing end of the carbon nanotube (or they may even be incorporated within the nanotube).

The above described embodiments are given as illustrative examples only. It will be readily appreciated that many

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deviations may be made from the specific embodiments disclosed in this specification without departing from the invention. Accordingly, the scope of the invention is to be determined by the claims below rather than being limited to the specifically described embodiments above.

What is claimed is:

1. A method of making a metallized carbon nanostructure, comprising the steps of:

- a. placing at least one carbon nanostructure formed on a substrate in a furnace; and
- b. applying a metallic vapor to the carbon nanostructure at a preselected temperature for a preselected period of time so that the metallic vapor reacts with carbon nanostructure to form a metallized nanostructure including a solid core about the carbon nanostructure.

2. The method of claim 1, wherein the carbon nanostructure comprises a carbon nanotube.

3. The method of claim 2, wherein the step of applying a metallic vapor to the carbon nanotube employs a chemical vapor deposition process.

4. The method of claim 3, wherein the chemical vapor deposition process comprises the steps of:

- a. placing a metal precursor in the furnace;
- b. flowing a gas through the furnace at a predetermined gas flow rate; and
- c. heating the furnace to the preselected temperature.

5. The method of claim 4, wherein the metal precursor comprises a metal.

6. The method of claim 5, wherein the metal comprises a transition metal.

7. The method of claim 4, wherein the metal precursor comprises a metal oxide.

8. The method of claim 4, wherein the furnace comprises a quartz tube furnace.

9. The method of claim 8, wherein the gas comprises an inert gas.

10. The method of claim 9, wherein the inert gas comprises argon applied at a gas flow rate of between 300 sccm and 1000 sccm.

11. The method of claim 8, wherein the gas comprises:

- a. an inert gas; and
- b. a reducing gas.

12. The method of claim 11, wherein the inert gas comprises argon.

13. The method of claim 12, wherein the predetermined gas flow rate comprises between 100 sccm and 1,000 sccm.

14. The method of claim 11, wherein the reducing gas comprises hydrogen.

15. The method of claim 14, wherein the predetermined gas flow rate comprises between 1 sccm and 300 sccm.

16. A method of making a field emitter, comprising:

- a. placing an insulating layer on a conductive substrate;
- b. placing a conductive gate layer on the insulating layer, opposite the conductive substrate;
- c. defining an opening in the conductive gate layer and the insulating layer, so as to expose a portion of the conductive substrate;
- d. grow at least one carbon nanostructure on the exposed portion of the substrate; and
- e. applying a metal vapor to the carbon nanostructure so as to form a metallized carbon nanostructure in the opening.