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**Maeda et al.**

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(54) **ACTIVE ENERGY BEAM-CURABLE COMPOSITION, INK CONTAINING THE SAME, AND PRINTER ACCOMMODATING THE SAME INK**

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(\*) Notice: Subject to any disclaimer, the term of this patent is extended or adjusted under 35 U.S.C. 154(b) by 0 days.

4,256,828 A	3/1981	Smith	
4,387,216 A	6/1983	Irving	
4,394,403 A	7/1983	Smith	
5,059,266 A	10/1991	Yamane et al.	
5,178,996 A	1/1993	Kobayashi et al.	
5,302,439 A	4/1994	Malhotra et al.	
5,463,084 A	10/1995	Crivello et al.	
5,476,752 A	12/1995	Noguchi et al.	
5,674,922 A	10/1997	Igarashi et al.	
5,681,643 A	10/1997	Noguchi et al.	
5,721,020 A	2/1998	Takami et al.	
5,750,590 A	5/1998	Schaefer et al.	
5,889,084 A	3/1999	Roth	
6,092,890 A *	7/2000	Wen et al.	347/101
6,197,409 B1	3/2001	Bodager et al.	
6,232,361 B1 *	5/2001	Laksin et al.	106/31.28

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(52) **U.S. Cl.** ..... **347/96; 347/100; 347/101**

(58) **Field of Search** ..... 347/96, 100, 95, 347/101, 102; 523/160; 106/31.6, 31.27, 31.13, 31.28, 31.85

(56) **References Cited**

**U.S. PATENT DOCUMENTS**

3,835,003 A	9/1974	Schlesinger
4,069,055 A	1/1978	Crivello
4,139,655 A	2/1979	Tsao
4,197,174 A	4/1980	Chang

**FOREIGN PATENT DOCUMENTS**

EP	1 138 739 A1	10/1991
EP	1 243 629 A1	9/2002
EP	1 357 159 A2	10/2003

(List continued on next page.)

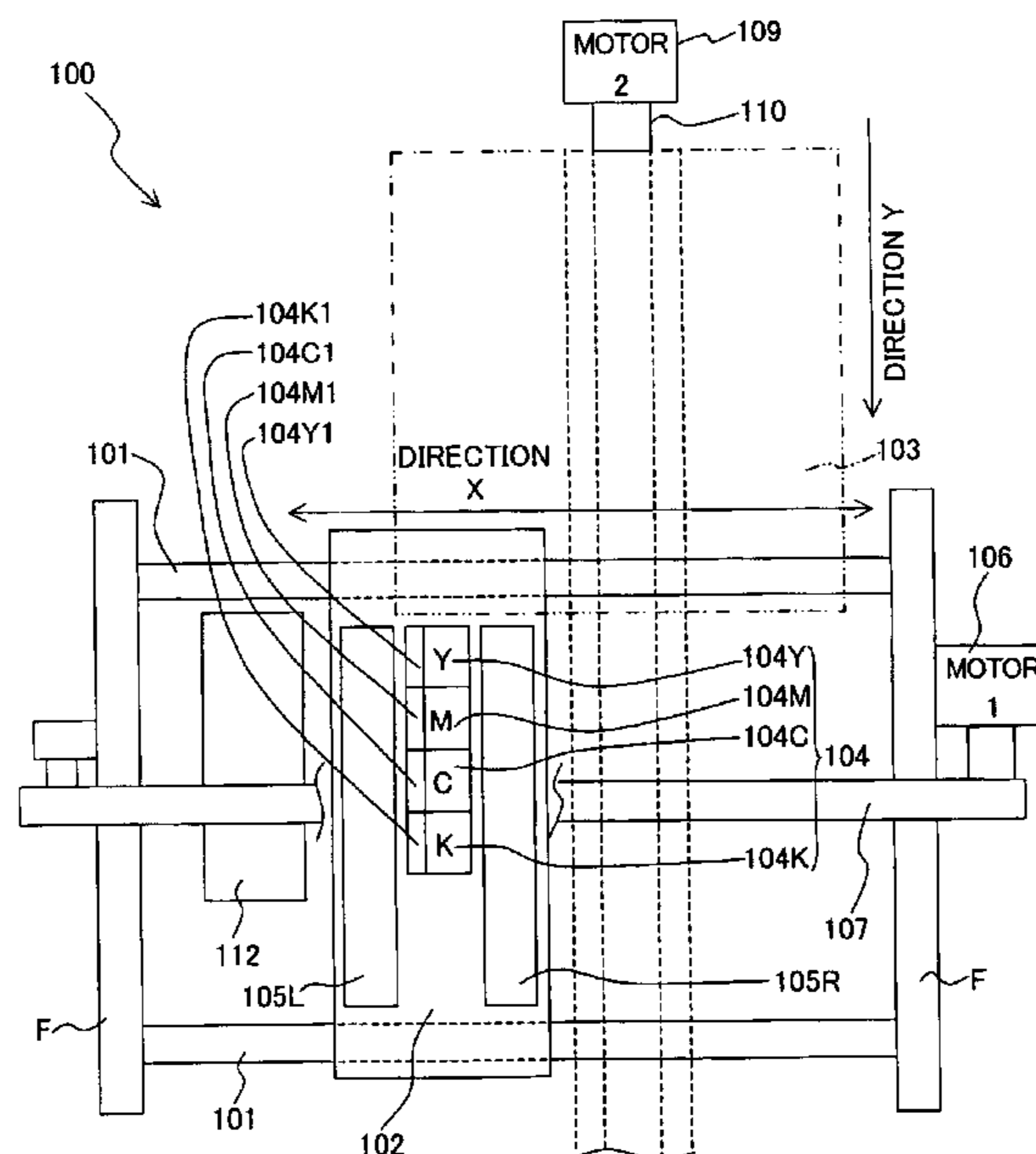
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(57) **ABSTRACT**

An active energy beam-curable composition includes a photo-cationic polymerizable substance; a photo-cationic polymerization initiator; and a polymerizable compound having an ethylenic unsaturated double bond such as monomethacrylate, dimethacrylate, monoacrylate, and diacrylate. Owing to the compound having the ethylenic unsaturated double bond contained in the composition, it is possible to suppress the occurrence of odor and cloudiness even when the composition is heated after being cured.

**15 Claims, 4 Drawing Sheets**



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FOREIGN PATENT DOCUMENTS						
			JP	A 63-51472		3/1988
			JP	A 4-11626		1/1992
			JP	A 8-143806		6/1996
			JP	08143806	*	6/1996 .....
			JP	A 8-218296		8/1996 C09D/11/00
			JP	0 844 255	A1	5/1998
			JP	A 10-195117		7/1998
			JP	A 11-140279		5/1999
			JP	11140279	*	5/1999 .....
			JP	A 11-246647		9/1999 C08L/63/00
			JP	11246647	*	9/1999 .....
			JP	A 2000-144043		5/2000 C08G/59/40
			JP	WO 00/50245		8/2000
			JP	A 2002-317139		10/2002

\* cited by examiner

Fig. 1A

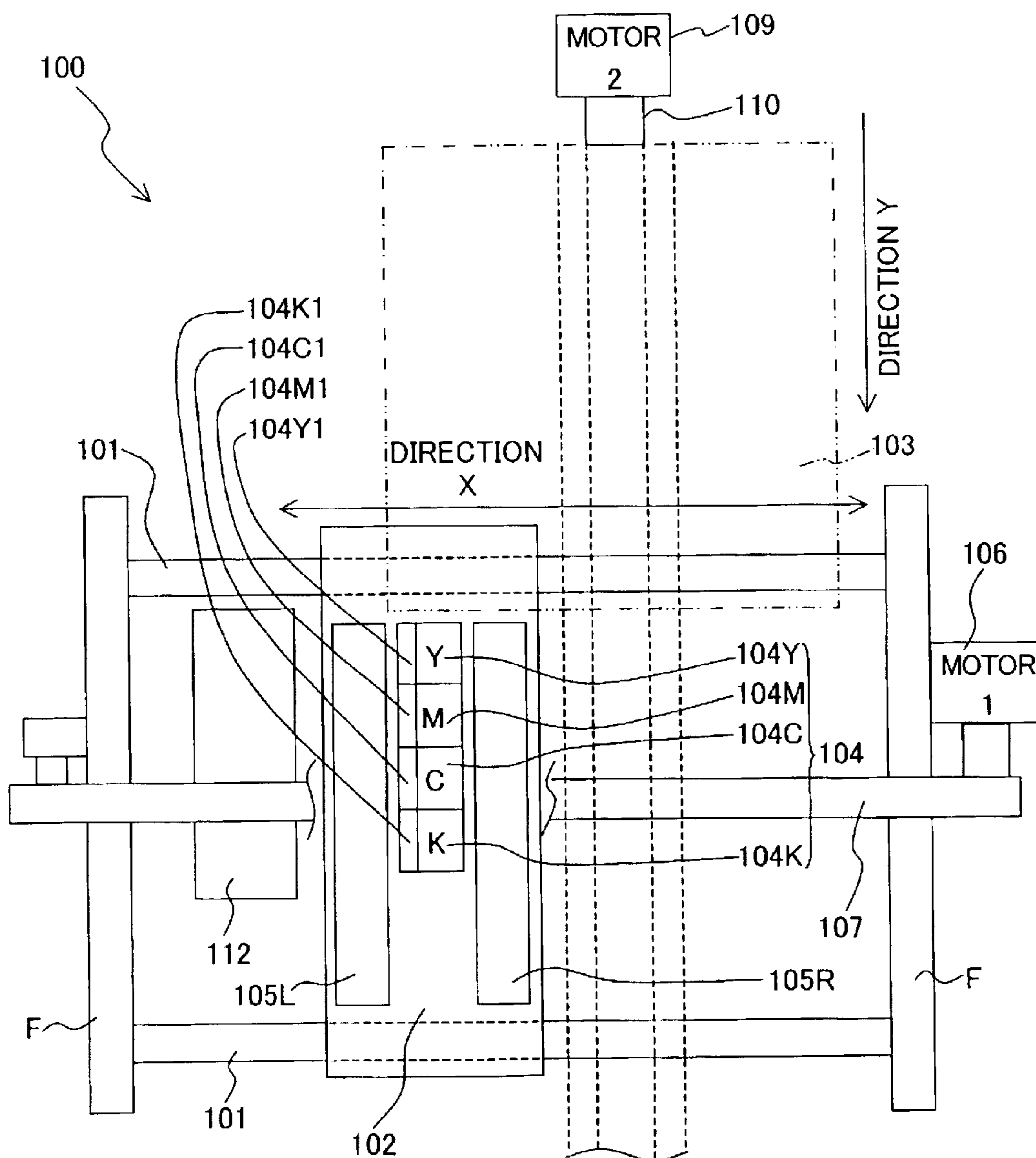


Fig. 1B

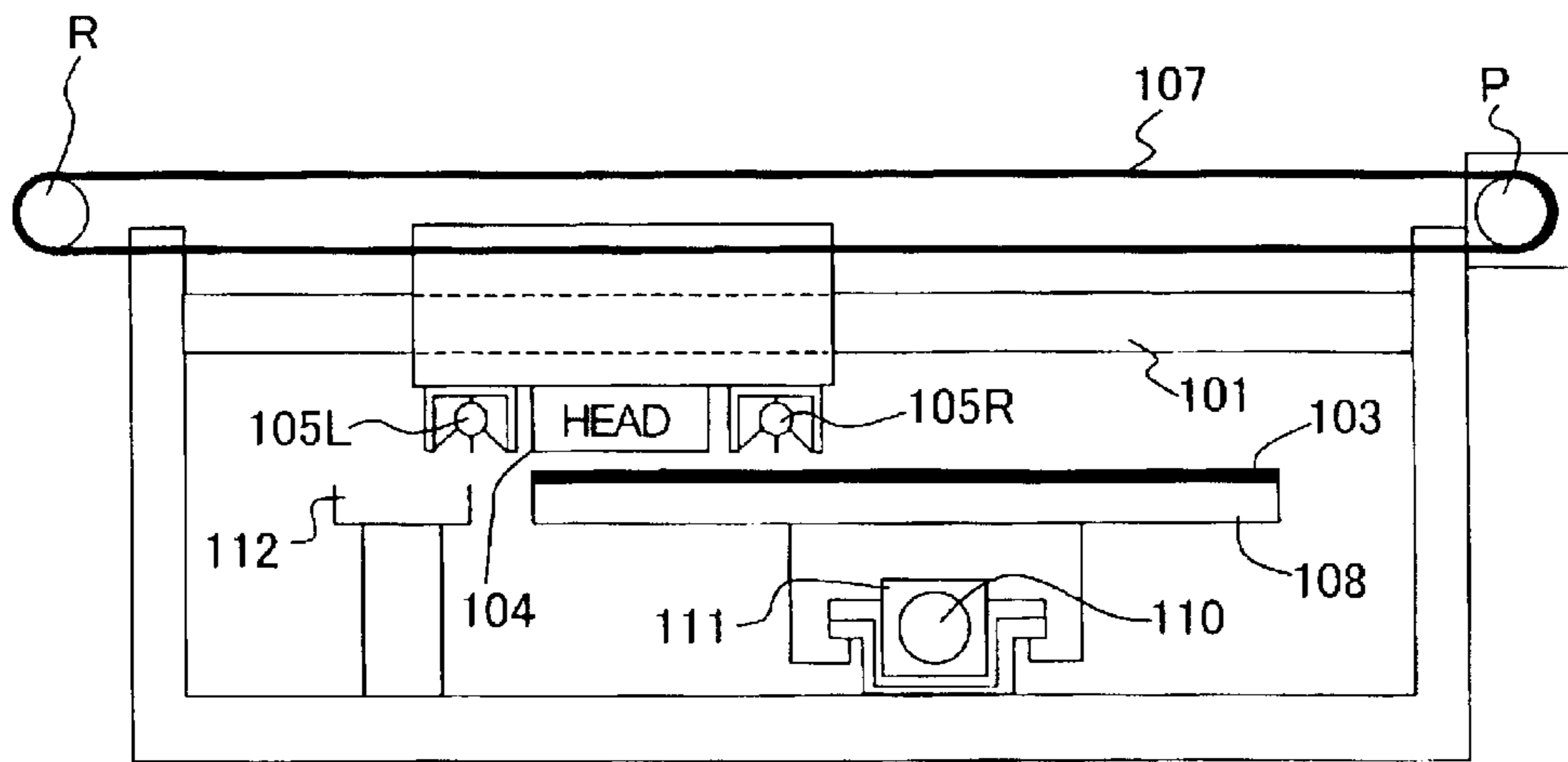


Fig. 2

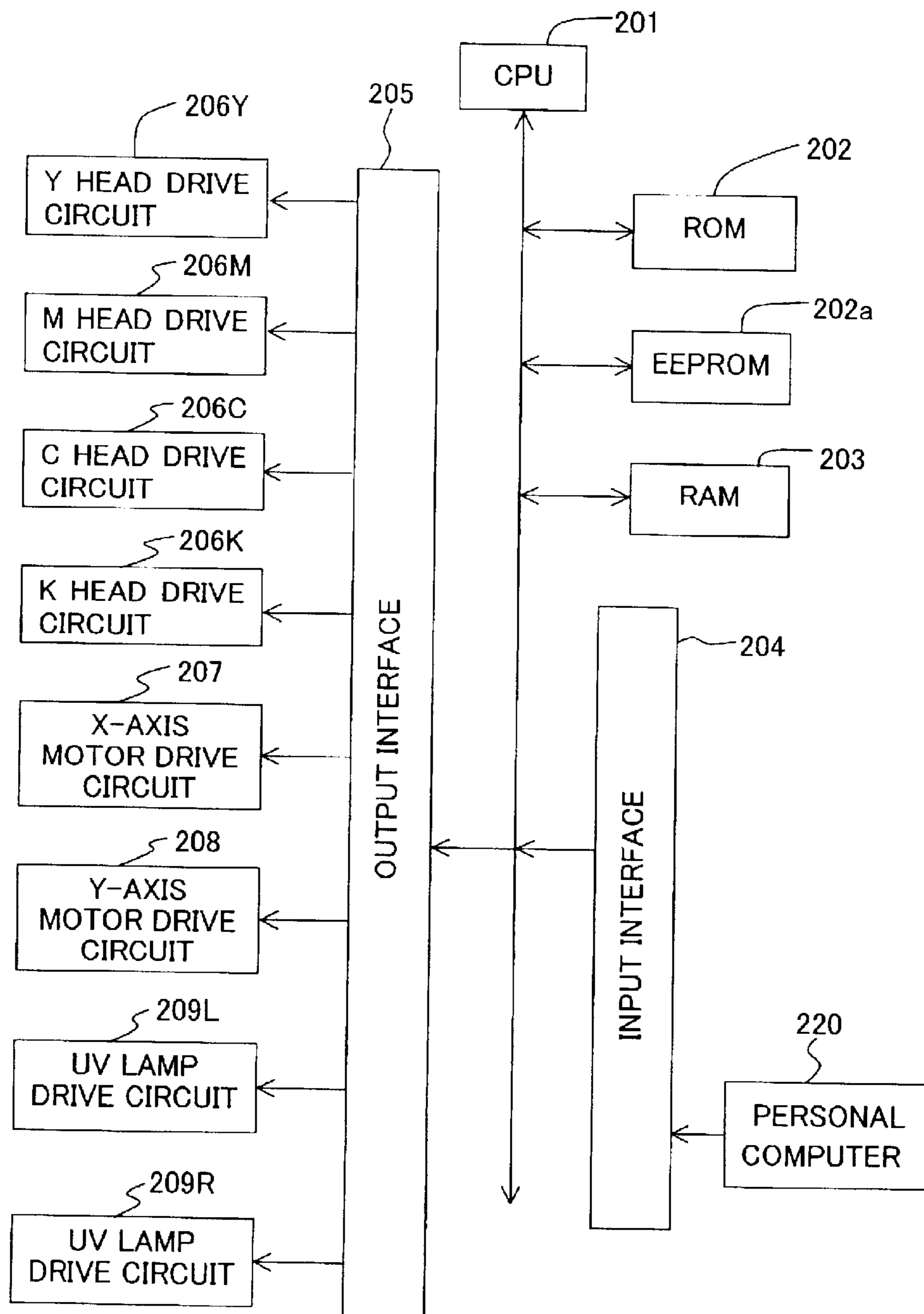
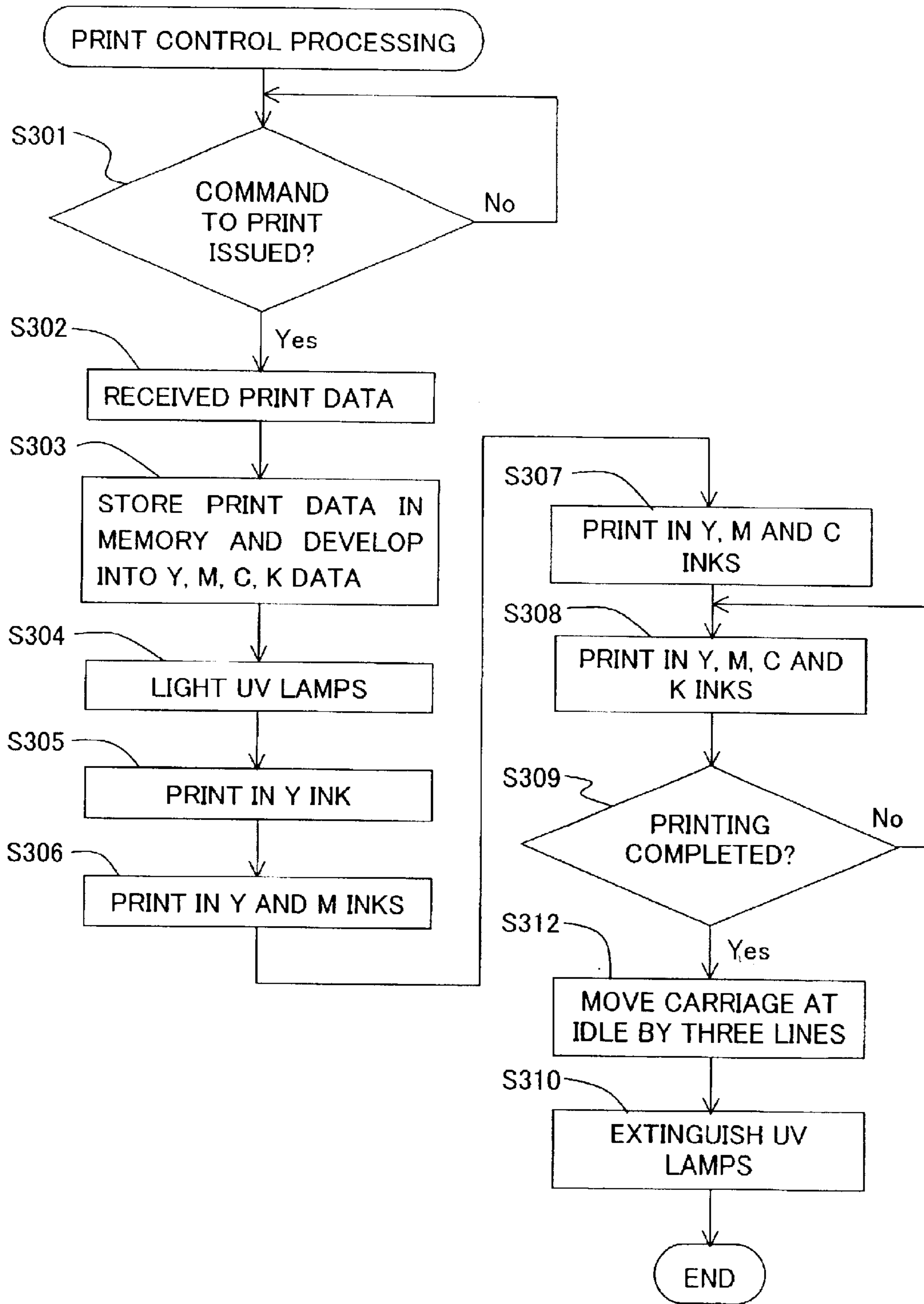


Fig. 3



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**ACTIVE ENERGY BEAM-CURABLE  
COMPOSITION, INK CONTAINING THE  
SAME, AND PRINTER ACCOMMODATING  
THE SAME INK**

BACKGROUND OF THE INVENTION

1. Field of the Invention

The present invention relates to an active energy beam-curable composition which is quickly cured by irradiation with an active energy beam such as ultraviolet radiation and electron beam, which causes neither odor nor cloudiness even when the cured composition is heated, and which has a low viscosity with good operation performance. The present invention also relates to an ink which contains the energy beam-curable composition, and a printer which accommodates the ink.

2. Description of the Related Art

The active energy beam curing technique is extremely important in a variety of industries including, for example, wood coating, metal painting, and printing, because of various characteristics including, for example, the quick curing speed, the good operation performance generally brought about by the absence of solvent, and the extremely low amount of required energy. Especially, the photo-cationic polymerization is attractive, because of the sufficient possibility to realize a variety of chemical and physical characteristics by means of polymerization of various monomers. In the photo-cationic polymerization, the polymerization is not inhibited by oxygen. Therefore, it is unnecessary to carry out the polymerization in an inert atmosphere. The photo-cationic polymerization is advantageous in that the polymerization can be performed quickly and completely in the air.

However, the conventional active energy beam-curable composition involves the following problem. That is, when the active energy beam-curable composition is further heated after being subjected to coating and curing on a base material such as wood, metal, and paper, any odor is caused by the volatile component thereof, for example, any uncured remaining monomer.

When the active energy beam-curable composition is used at the inside of an instrument or the like, the following problem also arises. That is, when the active energy beam-curable composition itself undergoes a high temperature, any cloudiness appears in the instrument due to the volatile component of the active energy beam-curable composition.

Taking the handling of the active energy beam-curable composition into consideration, it is desirable that the active energy beam-curable composition itself has a low viscosity before being cured. Therefore, when the active energy beam-curable composition contains a photo-cationic polymerizable substance and a polymerization initiator, a low viscosity polymerizable substance such as vinyl ether is used as a diluent in order to lower the viscosity. However, when the low viscosity polymerizable substance such as vinyl ether is used, the following problem arises. That is, when the cured composition is heated as described above, then the volatilization occurs, and the odor and the cloudiness are caused.

SUMMARY OF THE INVENTION

The present invention has been made in order to solve the problems as described above, an object of which is to provide an active energy beam-curable composition which

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has a low viscosity with good operation performance, which avoids the appearance of the volatile component even when the active energy beam-curable composition is cured and heated thereafter, and which makes it possible to avoid the occurrence of the odor and the cloudiness, an ink containing the composition, and an ink discharging apparatus with an ink container which accommodates the ink.

According to a first aspect of the present invention, there is provided an active energy beam-curable composition comprising:

- a photo-cationic polymerizable substance;
- a photo-cationic polymerization initiator; and
- a polymerizable compound having an ethylenic unsaturated double bond.

In the present invention, the active energy beam-curable composition, which comprises the photo-cationic polymerizable substance and the photo-cationic polymerization initiator, contains the polymerizable compound having the ethylenic unsaturated double bond. The polymerizable compound having the ethylenic unsaturated double bond may be, for example, monomethacrylate, dimethacrylate, monoacrylate, or diacrylate. It is considered that the polymerizable compound having the ethylenic unsaturated double bond as described above is polymerized by the irradiation with the active energy beam (radical polymerization), and the polymerizable compound having the ethylenic unsaturated double bond forms a bonding state by means of any chemical reaction together with the photo-cationic polymerizable substance. Therefore, even when the composition, which has been cured by the irradiation with the active energy beam, is heated again, then the volatile component, which is any remaining monomer, exists in an extremely small amount, and thus the odor and the cloudiness are scarcely caused. Further, the compound as described above has a relatively low viscosity. Therefore, it is easy to handle the active energy beam-curable composition.

In view of the prevention of the cloudiness and the odor which would be otherwise caused by the cured composition, the photo-cationic polymerizable substance preferably contains an epoxy resin compound. Especially, an alicyclic epoxy resin compound is preferred, because it enhances the crosslinking degree of the active energy beam-curable composition. Alternatively, the photo-cationic polymerizable substance may be an oxetane compound. When the oxetane compound is used, it is possible to more effectively avoid the cloudiness and the odor when the cured composition is heated.

The photo-cationic polymerization initiator may contain an onium salt. When the onium salt is used, then the crosslinking degree of the active energy beam-curable composition is enhanced, and it is possible to more effectively avoid the cloudiness and the odor when the cured composition is heated.

The active energy beam-curable composition may be a color or colorless (transparent) ink, for example, an ink to be preferably used for the ink-jet (or ink discharging) application. In this case, the composition may contain, for example, a coloring agent of a synthetic dye or a pigment, an inorganic filler, a surface-treating agent (a surfactant and/or a dispersing agent), a viscosity-adjusting agent, a treating agent, and an ultraviolet radiation-blocking agent, in addition to the photo-cationic polymerizable substance and the polymerization initiator. When the composition is used as the ink for the ink-jet application, the composition is prepared so that the viscosity of the composition upon discharging the ink from an ink jet head (ink discharging head) is 2 cps to 30 cps.

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The content of the polymerizable compound having the ethylenic unsaturated double bond is preferably about 5% by weight to about 58% by weight, in view of the fact that the occurrence of the odor and the cloudiness is avoided. Further, when the composition is used for an ink head of an ink jet apparatus (or ink discharging apparatus) which performs photo-molding or photo-curing printing, the content of the polymerizable compound having the ethylenic unsaturated double bond is preferably about 10% by weight to about 48% by weight. The content of the polymerizable compound having the ethylenic unsaturated double bond affects the viscosity of the composition. Therefore, when the composition is used as the ink for the ink-jet application, the content is preferably adjusted within a range in which the viscosity of the composition upon discharging the ink from an ink jet head is 2 cps to 30 cps.

According to a second aspect of the present invention, there is provided an ink discharging apparatus comprising:

- an ink discharging head which discharges an ink;
- an ink tank which stores the ink to be supplied to the ink discharging head; and
- a light source which is used to cure the discharged ink, wherein the ink comprises:
  - a photo-cationic polymerizable substance;
  - a photo-cationic polymerization initiator; and
  - a polymerizable compound having an ethylenic unsaturated double bond.

In the ink discharging apparatus of the present invention, the ink, which has been discharged by the ink discharging head, for example, onto the impermeable surface of those of wood, metal or the like, is cured with the light source such as the ultraviolet radiation. The ink contains the photo-cationic polymerizable substance, the photo-cationic polymerization initiator, and the polymerizable compound having the ethylenic unsaturated double bond. Therefore, the ink has a preferred viscosity to be discharged from the ink discharging head. Even when the cured ink is heated, neither odor nor cloudiness is caused. The ink discharging apparatus may be provided, for example, as a dispenser, a coater, and a printer. The ink tank may be an ink container fixedly provided in the apparatus or an ink cartridge which is replaceable.

## BRIEF DESCRIPTION OF THE DRAWINGS

An embodiment of the invention will be described in detail with reference to the following figures wherein:

FIG. 1A is a plan view of a color ink jet printer of a first embodiment of the invention;

FIG. 1B is a front view of the color ink jet printer of the first embodiment;

FIG. 2 is a schematic block diagram showing a control system of the ink jet printer of the first embodiment; and

FIG. 3 is a flowchart of control processing to be executed in the ink jet printer of an embodiment.

## DESCRIPTION OF THE PREFERRED EMBODIMENTS

The present invention will be explained in detail below. At first, explanation will be made for the substances to be used for the present invention.

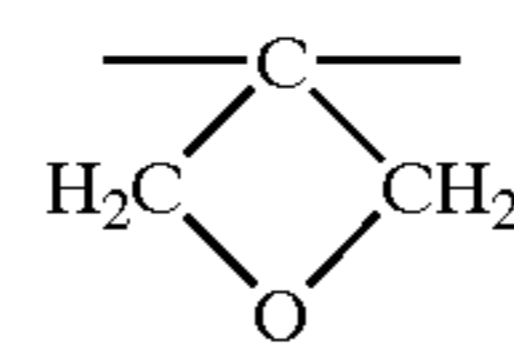
## (A) Photo-Cationic Polymerizable Substance

The photo-cationic polymerizable substance (A) (hereinafter referred to as "component (A)" as well) is an

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organic compound which causes the polymerization reaction and/or the crosslinking reaction by being irradiated with light in the presence of the cationic photopolymerization initiator. The photo-cationic polymerizable substance (A) includes, for example, epoxy compound, oxetane compound, oxolane compound, cyclic acetal compound, cyclic lactone compound, thiirane compound, thietane compound, Spiro ortho ester compound as reaction product of epoxy compound and lactone, cyclic ether compound, and cyclic thioether compound.

The photo-cationic polymerizable substance based on the oxetane compound is a compound having one or more oxetane ring or oxetane rings represented by the following formula (1).

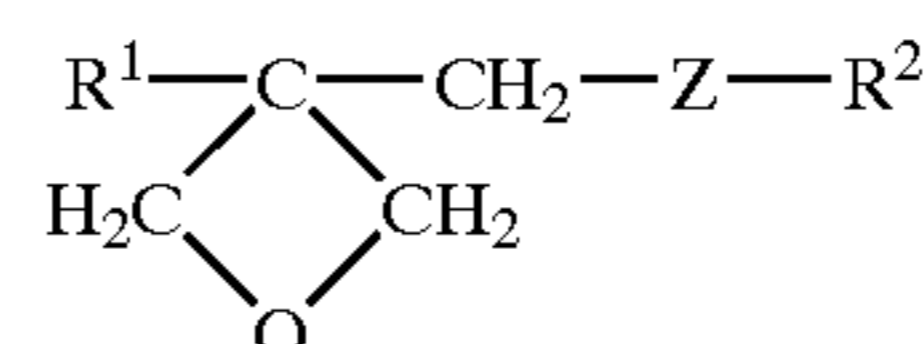


(1)

The compound causes the polymerization reaction and/or the crosslinking reaction by being irradiated with light in the presence of the cationic polymerizable photopolymerization initiator.

Various compounds may be used as the compound having the oxetane ring as described above, provided that the compound has one or more oxetane ring or oxetane rings. The compound will be exemplified below.

The compound, which has one oxetane ring, includes, for example, compounds represented by the following general formula (2).



(2)

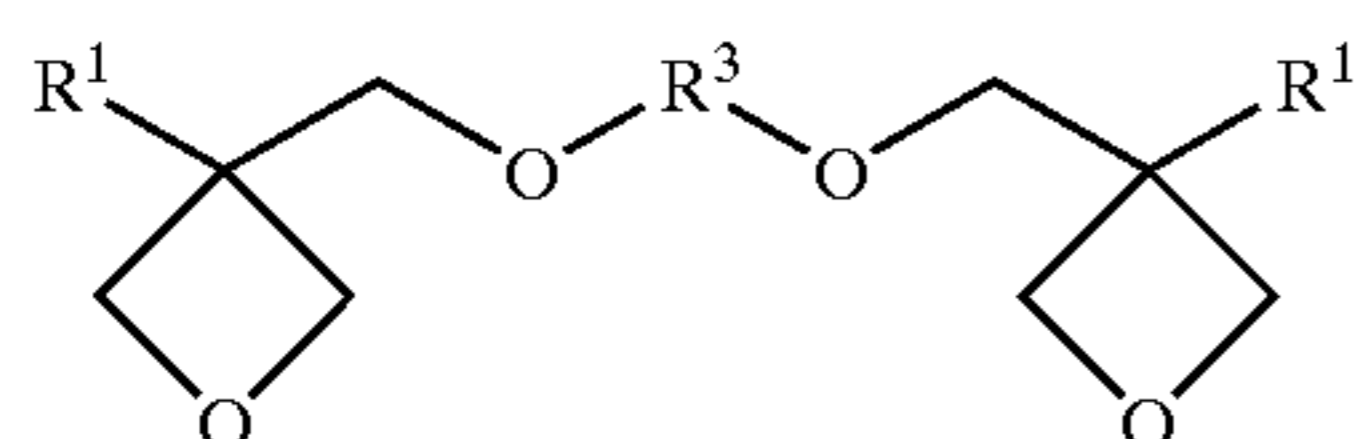
In the general formula (2), Z represents oxygen atom or sulfur atom. R<sup>1</sup> represents hydrogen atom; fluorine atom; alkyl group having a number of carbon atom or carbon atoms of 1 to 6 such as methyl group, ethyl group, propyl group, and butyl group; fluoroalkyl group having a number of carbon atom or carbon atoms of 1 to 6 such as trifluoromethyl group, perfluoroethyl group, and perfluoropropyl group; aryl group having a number of carbon atoms of 6 to 18 such as phenyl group and naphthyl group; furyl group; and thienyl group. R<sup>2</sup> represents, for example, hydrogen atom; alkyl group having a number of carbon atom or carbon atoms of 1 to 6 such as methyl group, ethyl group, propyl group, and butyl group; alkenyl group having a number of carbon atoms of 2 to 6 such as 1-propenyl group, 2-propenyl group, 2-methyl-1-propenyl group, 2-methyl-2-propenyl group, 1-butenyl group, 2-butenyl group, and 3-butenyl group; aryl group having a number of carbon atoms of 6 to 18 such as phenyl group, naphthyl group, anthranyl group, and phenanthryl group; substituted or unsubstituted aralkyl group having a number of carbon atoms of 7 to 18 such as benzyl group, fluorobenzyl group, methoxybenzyl group, fenetyl group, styryl group, cinnamyl group, and ethoxybenzyl group; group having another aromatic ring such as aryloxyalkyl including, for example, phenoxyethyl group and phenoxyethyl group; alkylcarbonyl group having a number of carbon atoms of 2 to 6 such as ethylcarbonyl group, propylcarbonyl group, and butylcarbonyl group; alkoxy carbonyl group having a number of carbon atoms of 2 to 6 such as ethoxycarbonyl group, propoxycarbonyl



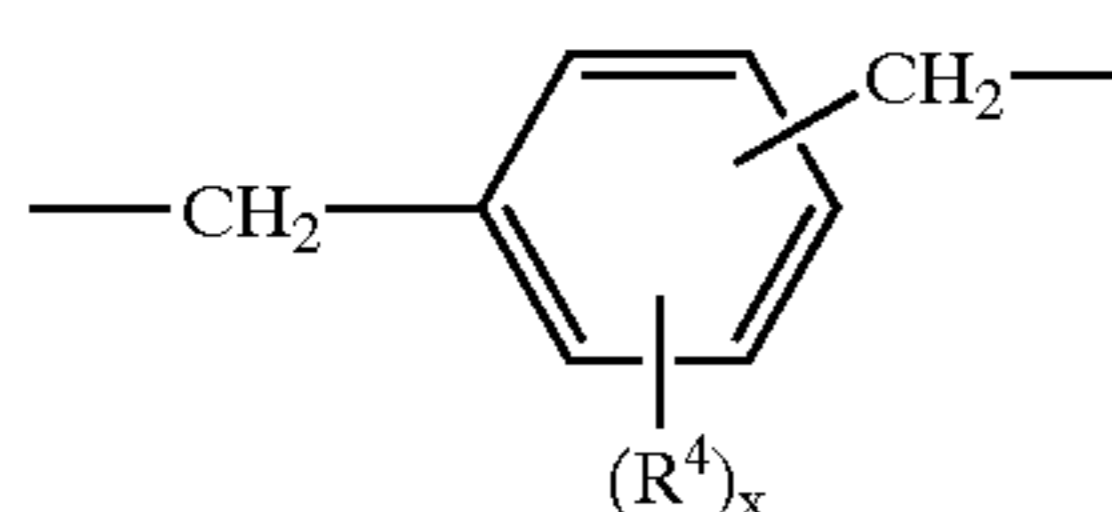
## 5

group, and butoxycarbonyl group; and N-alkylcarbamoyl group having a number of carbon atoms of 2 to 6 such as ethylcarbamoyl group, propylcarbamoyl group, butylcarbamoyl group, and pentylcarbamoyl group.

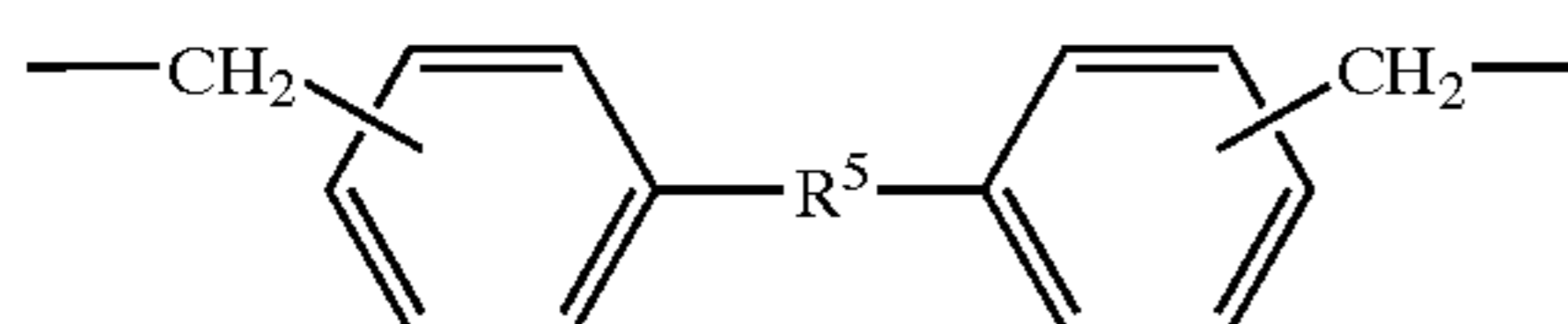
The compound, which has two oxetane rings, includes, for example, compounds represented by the following general formula (3).



In the general formula (3),  $R^1$  represents those as defined in the general formula (2) described above.  $R^3$  represents, for example, linear or branched alkylene group usually having a number of carbon atom or carbon atoms of 1 to 20 such as ethylene group, propylene group, and butylene group; linear or branched poly(alkyleneoxy) group usually having a number of carbon atom or carbon atoms of 1 to 120 such as poly(ethyleneoxy) group and poly(propyleneoxy) group; linear or branched unsaturated hydrocarbon group such as propenylene group, methylpropenylene group, and butenylene group; carbonyl group; alkylene group containing carbonyl group; alkylene group containing carboxyl group at intermediate position of molecular chain; and alkylene group containing carbamoyl group at intermediate position of molecular chain. Alternatively,  $R^3$  may be polyvalent group selected from groups represented by the following general formulas (4), (5), and (6).

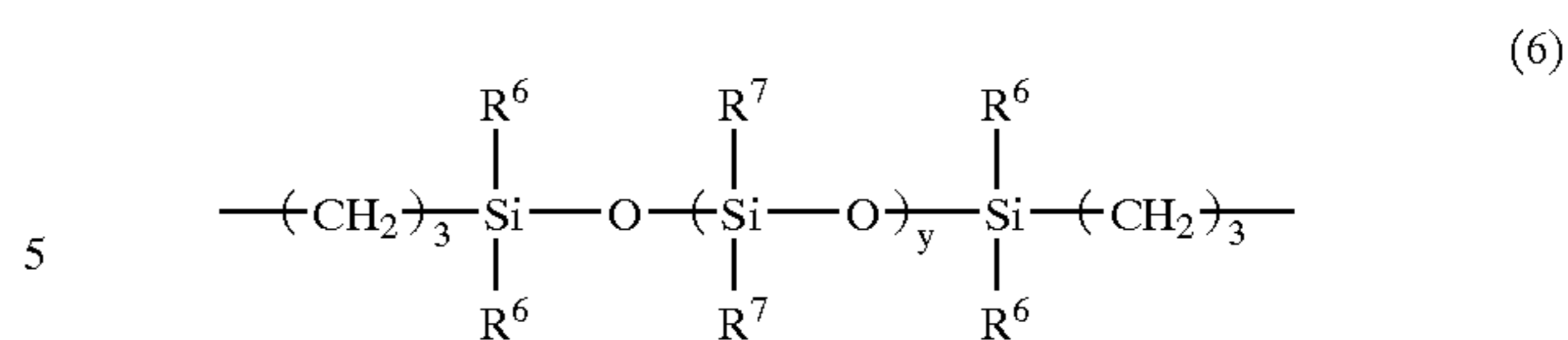


In the general formula (4),  $R^4$  represents hydrogen atom; alkyl group having a number of carbon atom or carbon atoms of 1 to 4 such as methyl group, ethyl group, propyl group, and butyl group; alkoxy group having a number of carbon atom or carbon atoms of 1 to 4 such as methoxy group, ethoxy group, propoxy group, and butoxy group; halogen atom such as chlorine atom and boron atom; nitro group; cyano group; mercapto group; lower alkylcarboxyl group; carboxyl group; or carbamoyl group.  $x$  represents an integer of 1 to 4.

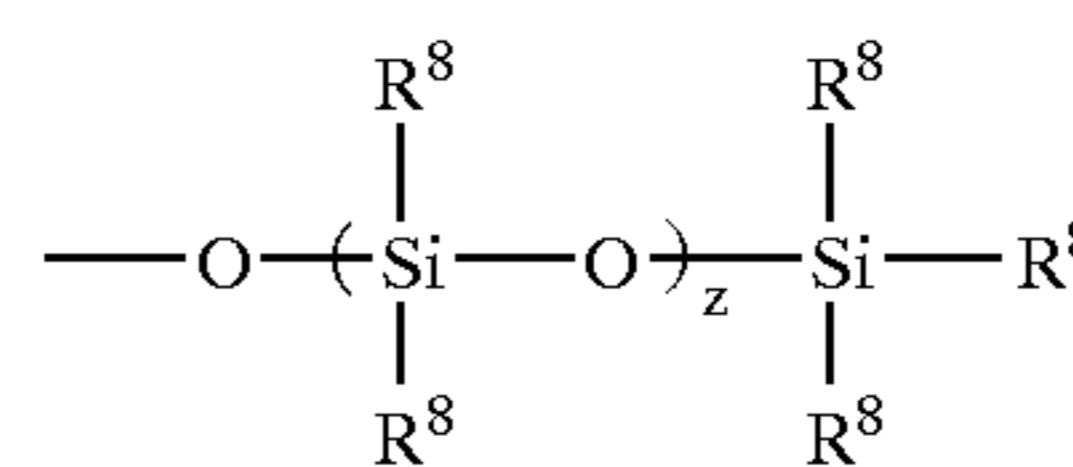


In the general formula (5),  $R^5$  represents oxygen atom, sulfur atom, methylene group,  $-\text{NH}-$ ,  $-\text{SO}-$ ,  $-\text{SO}_2-$ ,  $-\text{C}(\text{CF}_3)_2-$ , or  $-\text{C}(\text{CH}_3)_2-$ .

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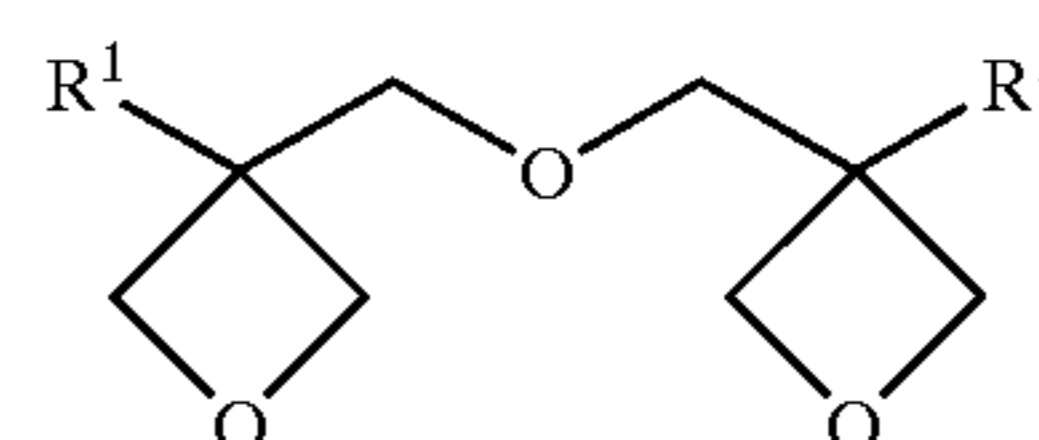
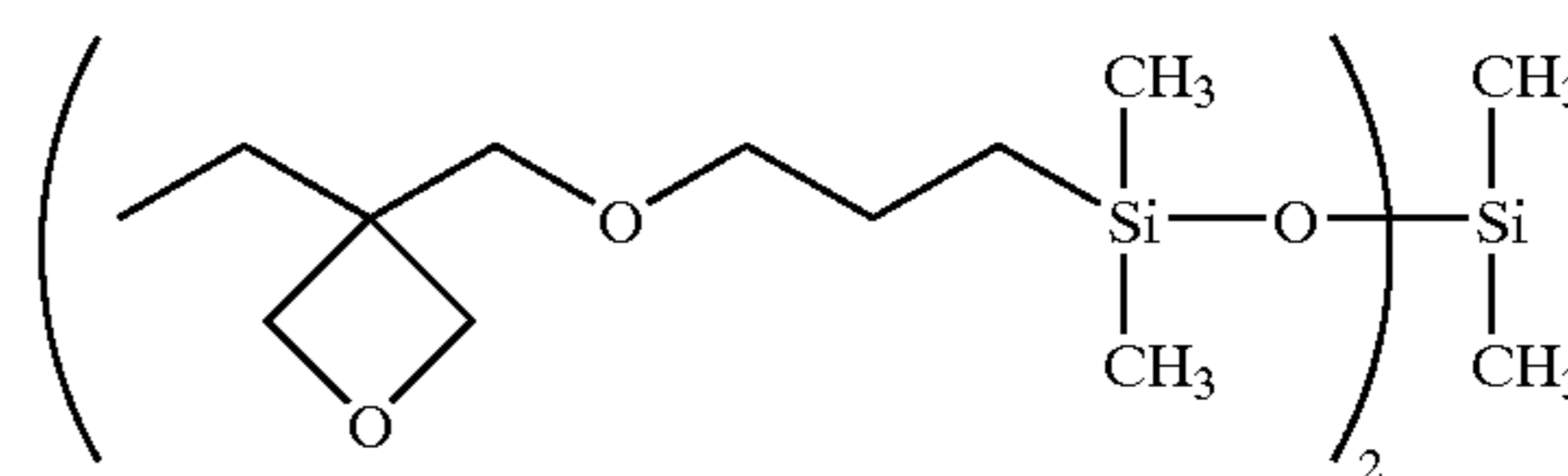
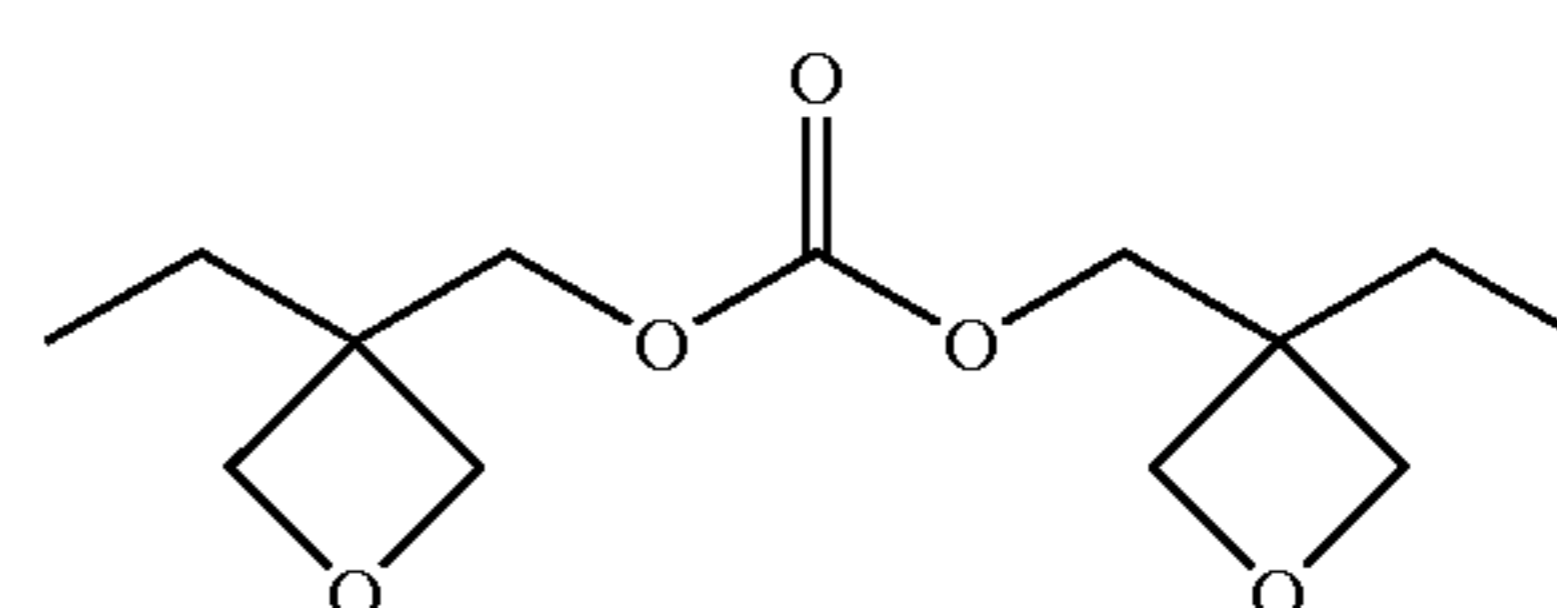


In the general formula (6),  $R^6$  represents alkyl group having a number of carbon atom or carbon atoms of 1 to 4 such as methyl group, ethyl group, propyl group, and butyl group; or aryl group having a number of carbon atoms of 6 to 18 such as phenyl group and naphthyl group.  $y$  is an integer of 0 to 200.  $R^7$  represents alkyl group having a number of carbon atom or carbon atoms of 1 to 4 such as methyl group, ethyl group, propyl group, and butyl group; or aryl group having a number of carbon atoms of 6 to 18 such as phenyl group and naphthyl group.  $R^7$  may be a group represented by the following general formula (7).



In the general formula (7),  $R^8$  represents alkyl group having a number of carbon atom or carbon atoms of 1 to 4 such as methyl group, ethyl group, propyl group, and butyl group; or aryl group having a number of carbon atoms of 6 to 18 such as phenyl group and naphthyl group.  $z$  is an integer of 0 to 100.

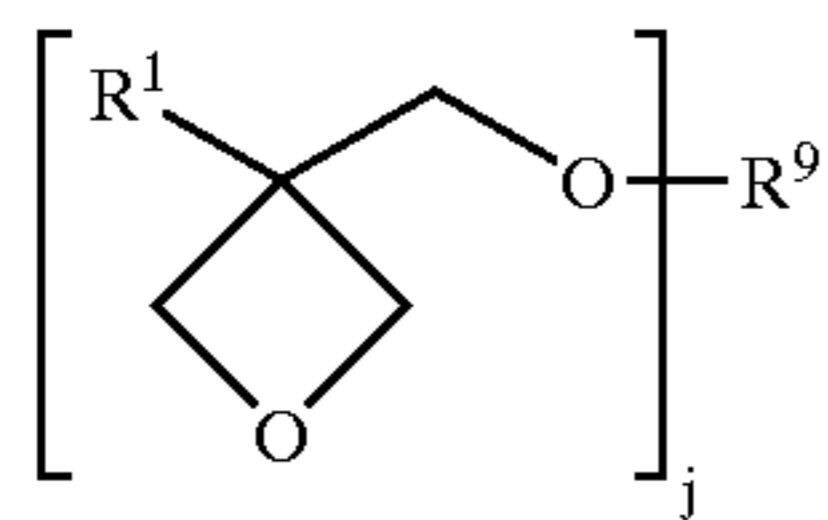
Specifically, the compound, which has two oxetane rings, includes, for example, compounds represented by the following general formulas (8), (9), and (10).



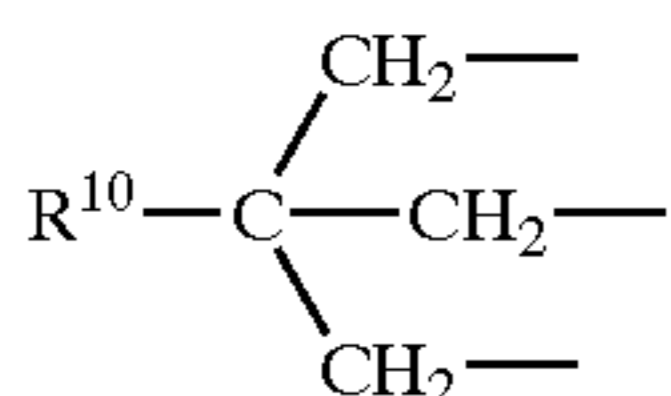
In the general formula (10),  $R^1$  represents those as defined in the general formula (2) described above.

The compound, which has three or more oxetane rings, includes, for example, compounds represented by the following general formula (11).

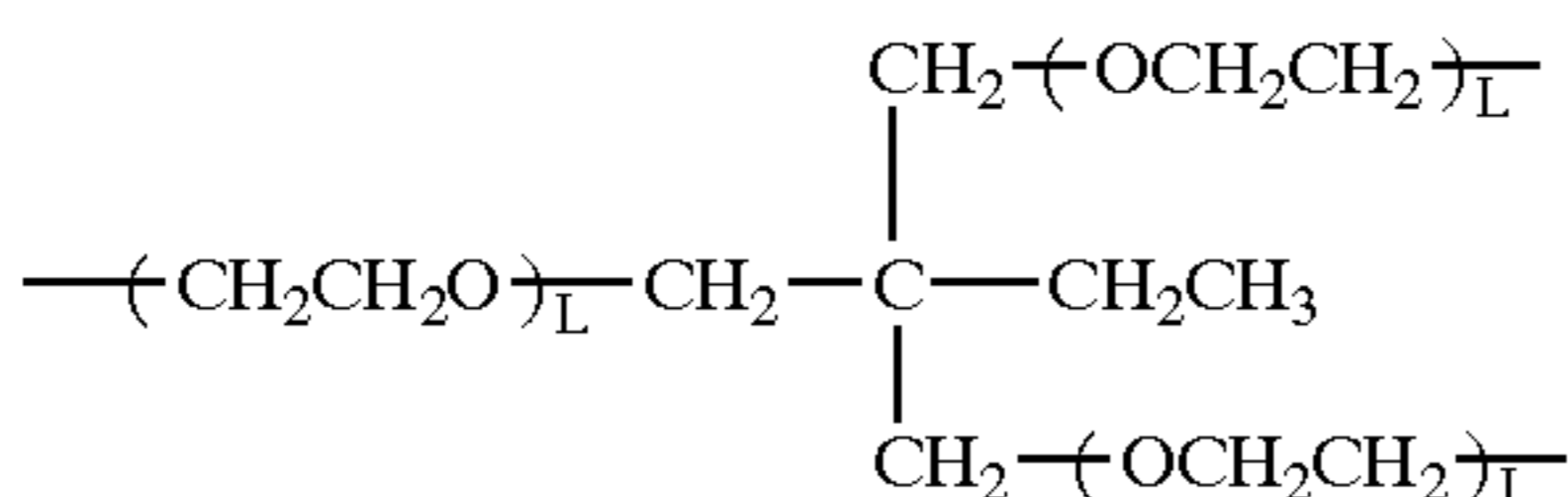
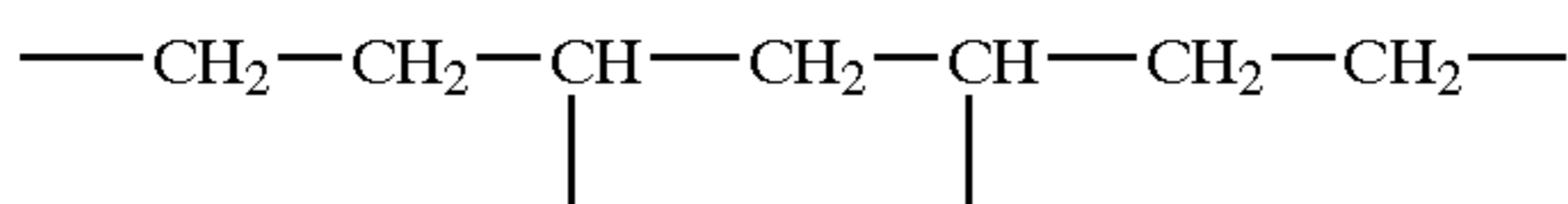
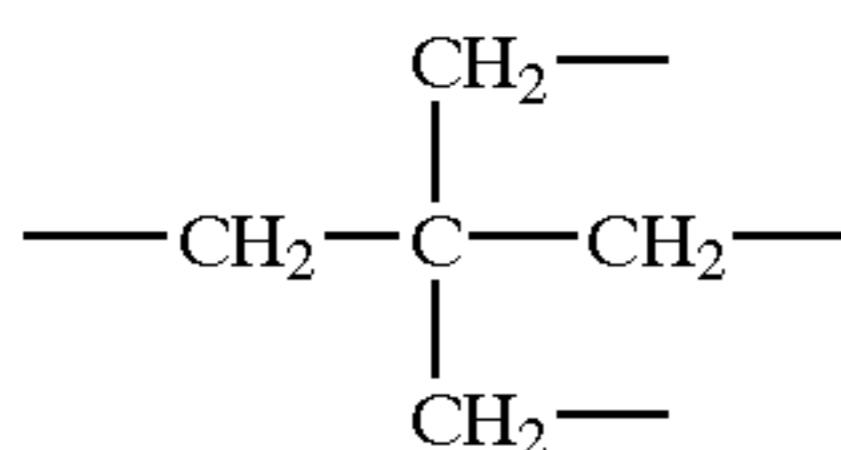
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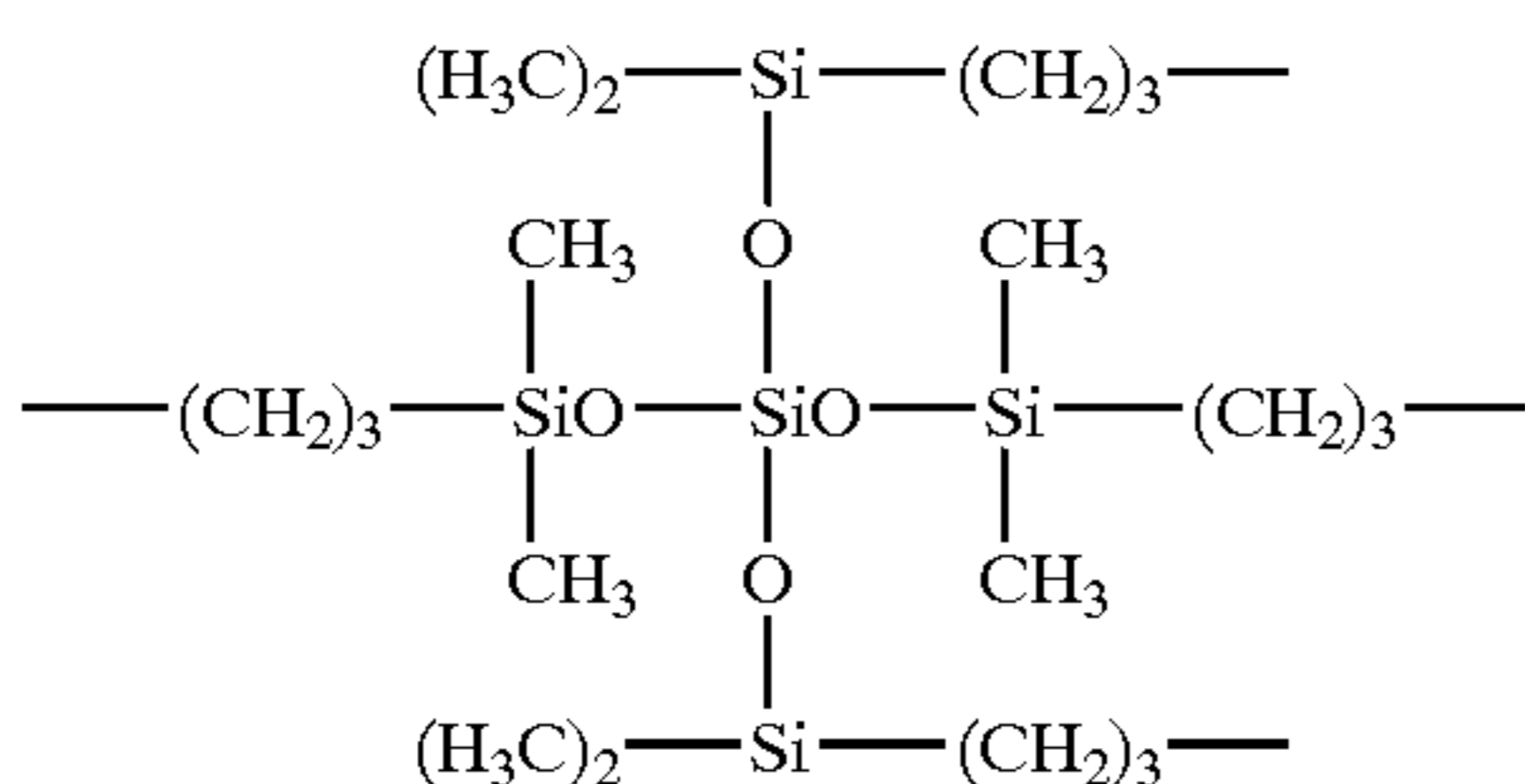
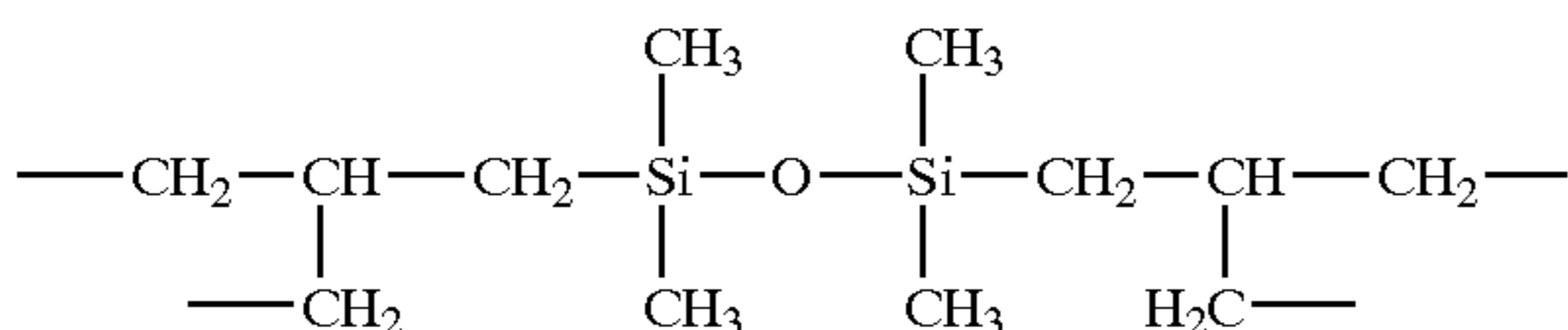
In the general formula (11),  $R^1$  represents those as defined in the general formula (2) described above.  $R^9$  represents trivalent to decavalent organic groups including, for example, branched or linear alkylene group having a number of carbon atom or carbon atoms of 1 to 30 such as groups represented by the following formulas (12) to (14), branched poly(alkyleneoxy) group such as groups represented by the following formula (15), and linear or branched polysiloxane-containing groups represented by the following formula (16) or (17).  $j$  represents an integer of 3 to 10 equal to the valency number of  $R^9$ .



In the formula (12),  $R^{10}$  represents alkyl group having a number of carbon atom or carbon atoms of 1 to 6 such as methyl group, ethyl group, and propyl group.

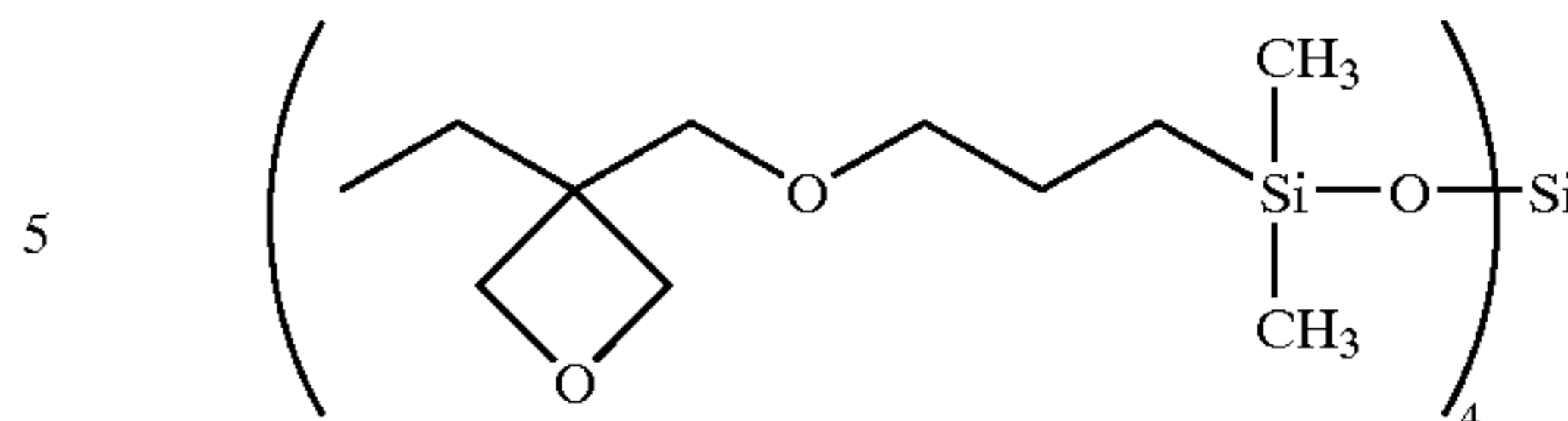


In the formula (15),  $L$ 's are the same or different from each other, each of which is an integer of 1 to 10.



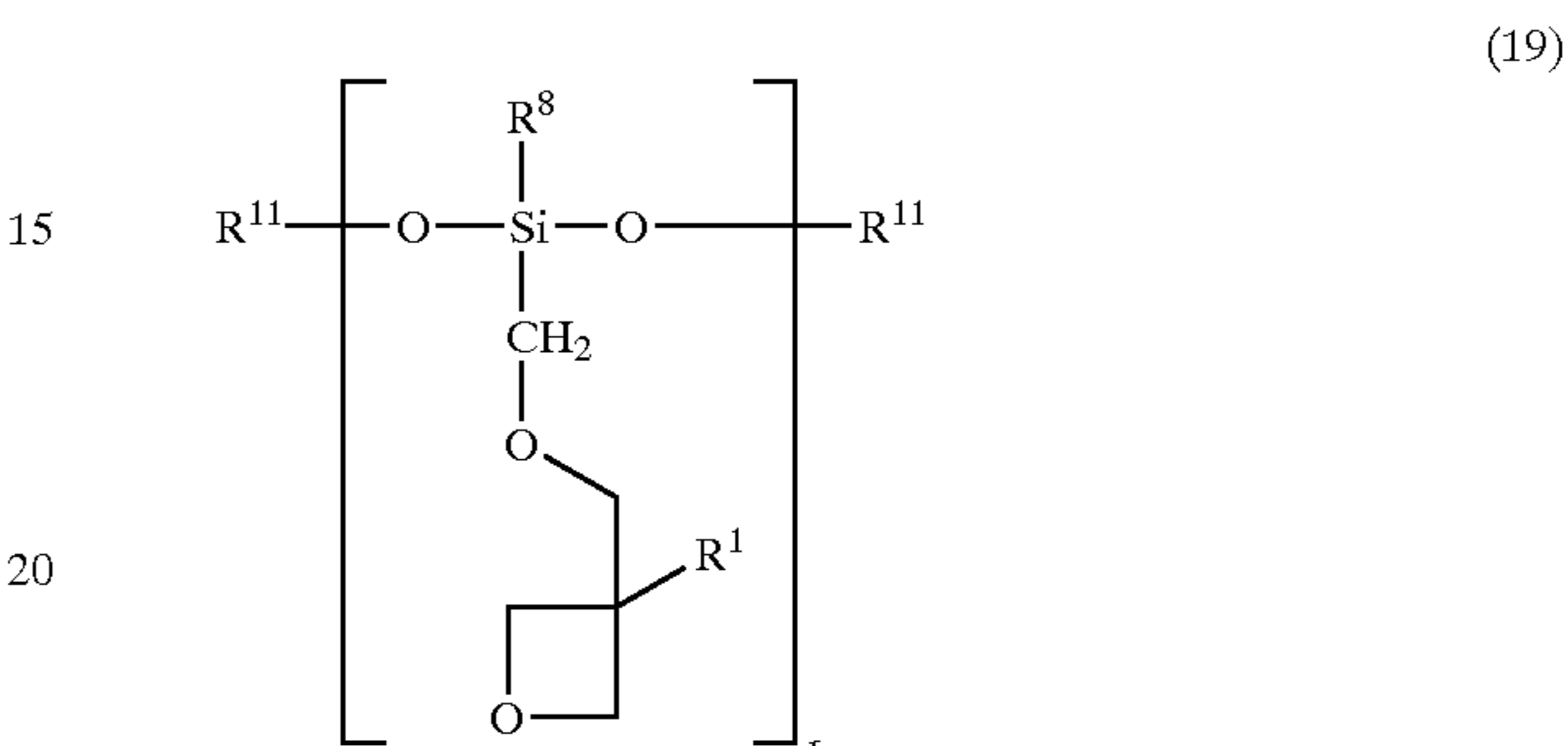
Specifically, the compound, which has four or more oxetane rings, includes, for example, a compound represented by the following formula (18).

(11)



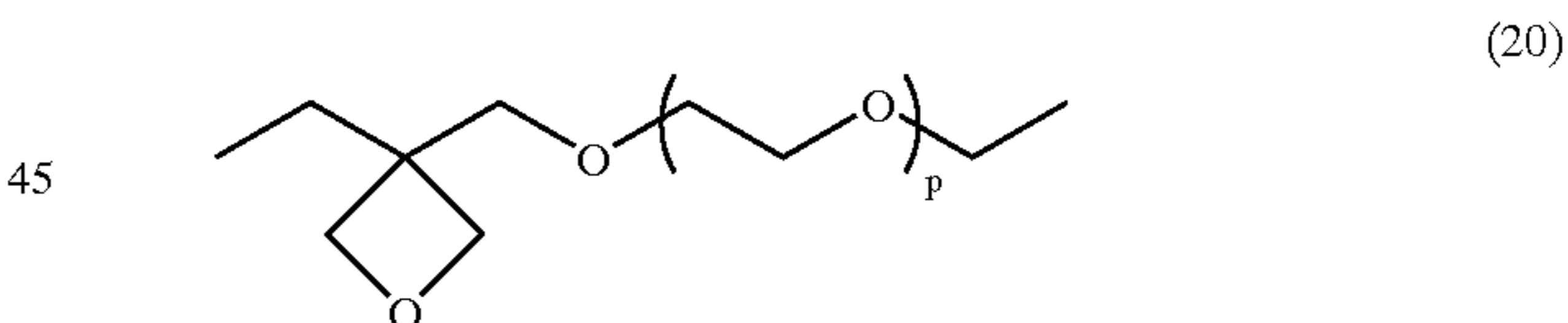
(18)

The compound represented by the following general formula (19) may have 1 to 10 oxetane ring or oxetane rings.

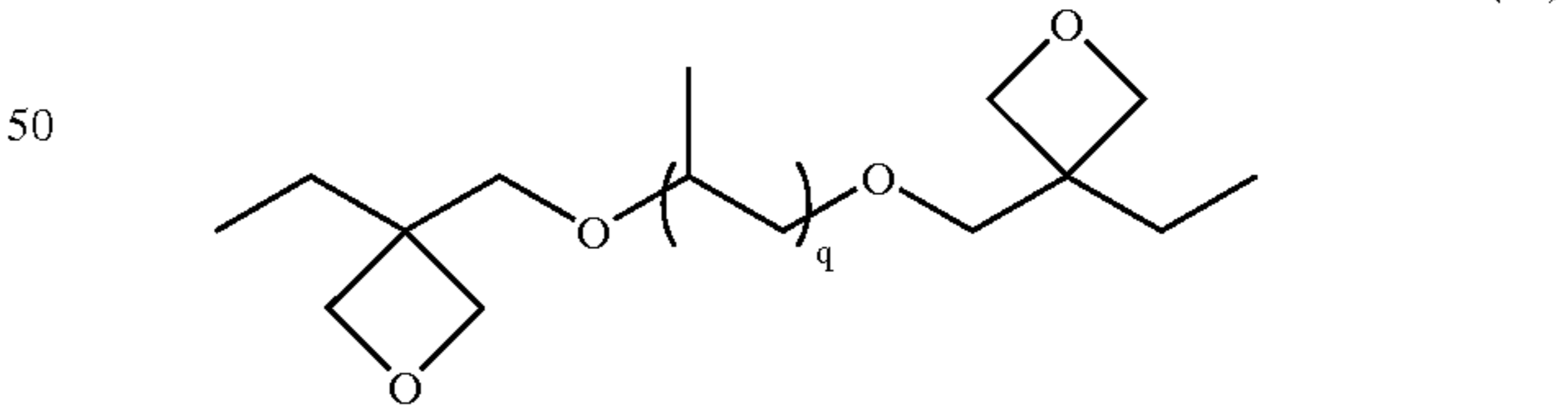


In the formula (19),  $R^1$  represents those as defined in the general formula (2).  $R^8$  represents those as defined in the general formula (7).  $R^{11}$  represents alkyl group having a number of carbon atom or carbon atoms of 1 to 4 such as methyl group, ethyl group, propyl group, and butyl group; or trialkylsilyl group (in this case, alkyl groups are the same or different from each other, including those having a number of carbon atoms of 3 to 12 such as trimethylsilyl group, triethylsilyl group, tripropylsilyl group, and tributylsilyl group), and  $r$  represents an integer of 1 to 10.

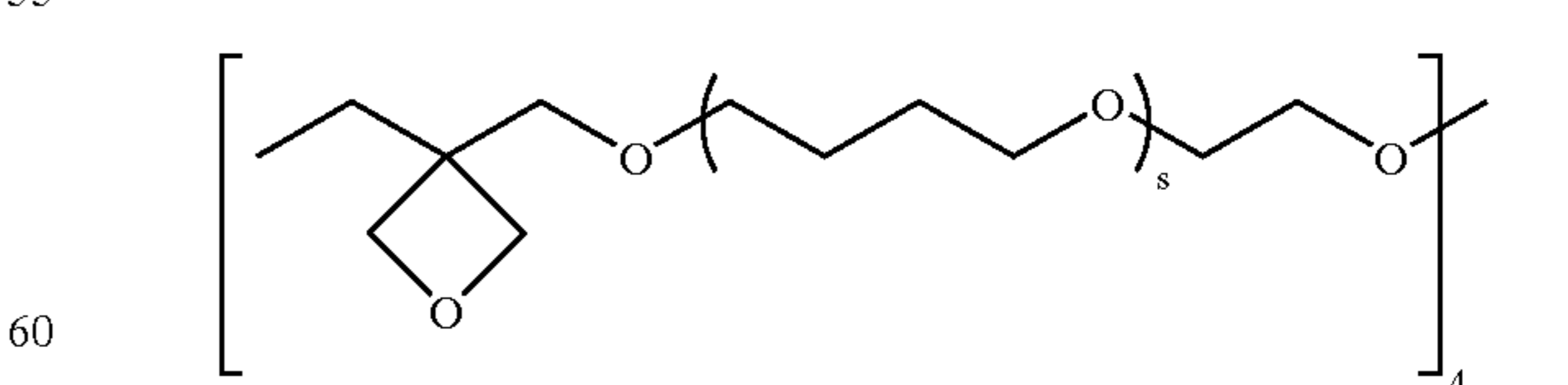
Other than the examples described above, the compound (A) having the oxetane ring further includes compounds having a high molecular weight, i.e., a number average molecular weight of about 1000 to 5000 as converted into a value of polystyrene measured by the gel permeation chromatography. Such a compound includes compounds represented by the following formulas (20), (21), and (22) by way of example.



(20)



(21)



In the formula (20),  $p$  represents an arbitrary integer. In the formula (21),  $q$  represents an arbitrary integer. In the formula (22),  $s$  represents an integer of 20 to 200. Specified examples of the compound (A) having the oxetane ring or oxetane rings explained above are as follows.

## Compound Having One Oxetane Ring

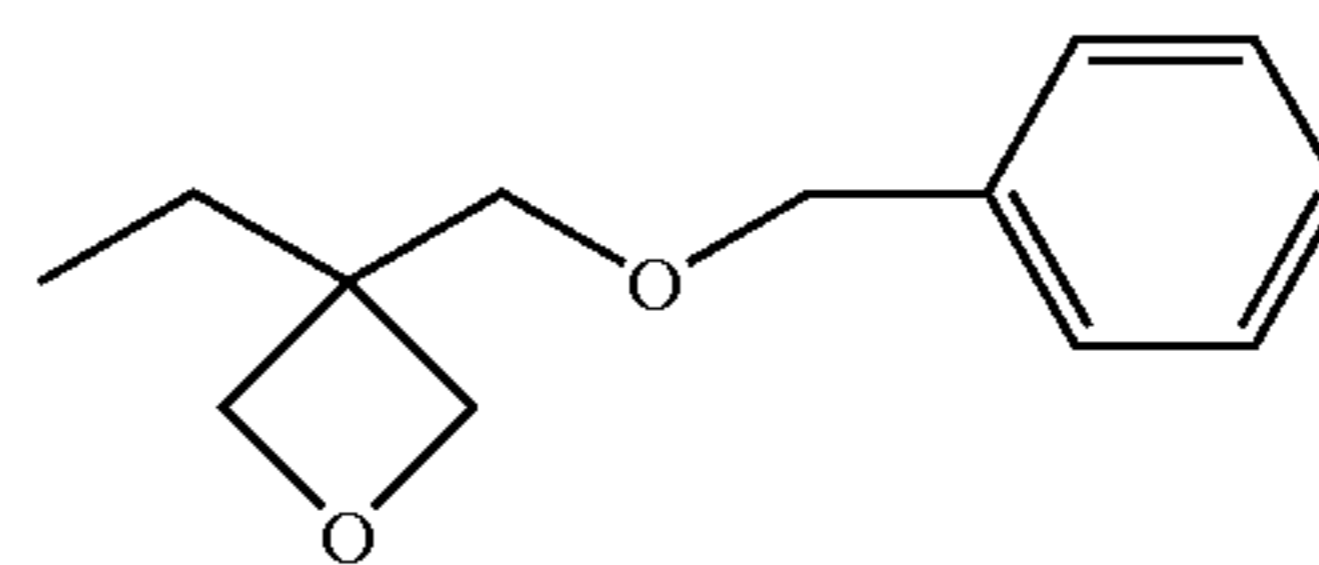
For example, there may be exemplified 3-ethyl-3-hydroxymethyloxetane, 3-(meta)allyloxymethyl-3-ethyloxetane, (3-ethyl-3-oxetanylmethoxy)methylbenzene, 4-fluoro-[1-(3-ethyl-3-oxetanylmethoxy)methyl]benzene, 4-methoxy-[1-(3-ethyl-3-oxetanylmethoxy)methyl]benzene, [1-(3-ethyl-3-oxetanylmethoxy)ethyl]phenyl ether, isobutoxymethyl(3-ethyl-3-oxetanylmethyl)ether, isobornyloxyethyl(3-ethyl-3-oxetanylmethyl)ether, isobornyl(3-ethyl-3-oxetanylmethyl) ether, 2-ethylhexyl(3-ethyl-3-oxetanylmethyl) ether, ethyldiethylene glycol(3-ethyl-3-oxetanylmethyl) ether, dicyclopentadiene(3-ethyl-3-oxetanylmethyl) ether, dicyclopentenyl(3-ethyl-3-oxetanylmethyl)ether, tetrahydrofurfuryl(3-ethyl-3-oxetanylmethyl)ether, tetrabromophenyl(3-ethyl-3-oxetanylmethyl)ether, 2-tetrabromophenoxyethyl(3-ethyl-3-oxetanylmethyl)ether, tribromophenyl(3-ethyl-3-oxetanylmethyl)ether, 2-tribromophenoxyethyl(3-ethyl-3-oxetanylmethyl)ether, 2-hydroxyethyl(3-ethyl-3-oxetanylmethyl)ether, 2-hydroxypropyl(3-ethyl-3-oxetanylmethyl)ether, butoxyethyl(3-ethyl-3-oxetanylmethyl)ether, pentachlorophenyl(3-ethyl-3-oxetanylmethyl)ether, pentabromophenyl(3-ethyl-3-oxetanylmethyl)ether, and bornyl(3-ethyl-3-oxetanylmethyl)ether.

## Compound Having Two or More Oxetane Rings

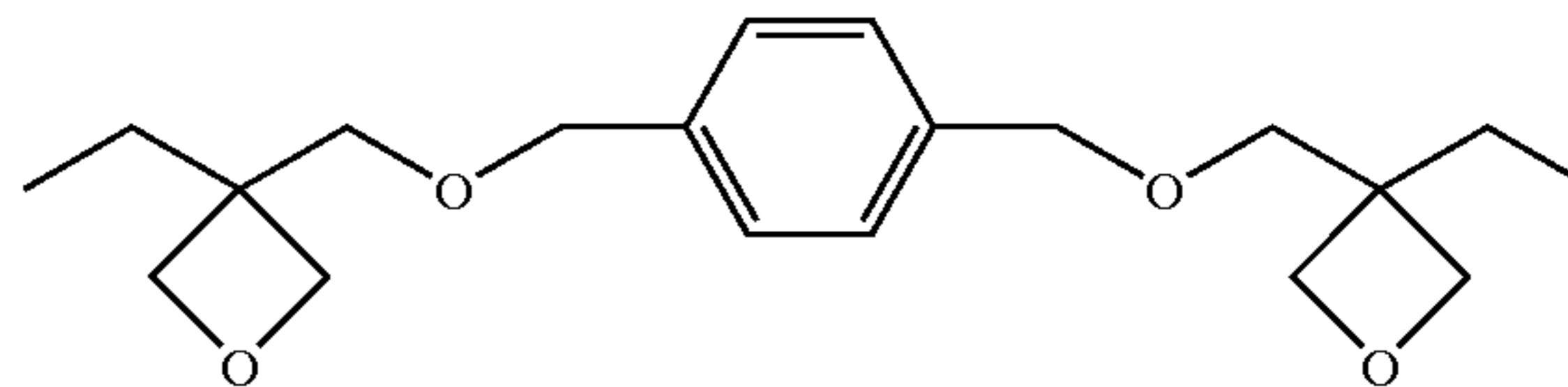
For example, there may be exemplified 3,7-bis(3-oxetanyl)-5-oxa-nonane, 3,3'-(1,3-(2-methylene)propanediylbis(oxymethylene))bis(3-ethyloxetane), 1,4-bis[(3-ethyl-3-oxetanylmethoxy)methyl]benzene, 1,2-bis[(3-ethyl-3-oxetanylmethoxy)methyl]ethane, 1,3-bis[(3-ethyl-3-oxetanylmethoxy)methyl]propane, ethylene glycol bis(3-ethyl-3-oxetanylmethyl)ether, dicyclopentenylbis(3-ethyl-3-oxetanylmethyl)ether, triethylene glycol bis(3-ethyl-3-oxetanylmethyl)ether, tetraethylene glycol bis(3-ethyl-3-oxetanylmethyl)ether, tricyclodecanediyl dimethylene(3-ethyl-3-oxetanylmethyl)ether, trimethylolpropane tris(3-ethyl-3-oxetanylmethyl)ether, 1,4-bis(3-ethyl-3-oxetanylmethoxy)butane, 1,6-bis(3-ethyl-3-oxetanylmethoxy)hexane, pentaerythritol tris(3-ethyl-3-oxetanylmethyl)ether, pentaerythritol tetrakis(3-ethyl-3-oxetanylmethyl)ether, polyethylene glycol bis(3-ethyl-3-oxetanylmethyl)ether, dipentaerythritol hexakis(3-ethyl-3-oxetanylmethyl)ether, dipentaerythritol pentakis(3-ethyl-3-oxetanylmethyl)ether, dipentaerythritol tetrakis(3-ethyl-3-oxetanylmethyl)ether, caprolactone modified dipentaerythritol hexakis(3-ethyl-3-oxetanylmethyl)ether, caprolactone modified dipentaerythritol pentakis(3-ethyl-3-oxetanylmethyl)ether, ditrimethylolpropane tetrakis(3-ethyl-3-oxetanylmethyl)ether, EO modified bisphenol A bis(3-ethyl-3-oxetanylmethyl)ether, PO modified bisphenol A bis(3-ethyl-3-oxetanylmethyl)ether, EO modified hydrogenated bisphenol A bis(3-ethyl-3-oxetanylmethyl)ether, PO modified hydrogenated bisphenol A bis(3-ethyl-3-oxetanylmethyl)ether, and EO modified bisphenol F (3-ethyl-3-oxetanylmethyl)ether. One species of the compound as described above may be used singly. Alternatively, two or more species of the compounds as described above may be used in combination.

Among them, the compound having the oxetane ring, which may be used especially preferably as the component (A) of the resin composition of the present invention, includes, for example, (3-ethyl-3-oxetanylmethoxy)methylbenzene represented by the following formula (23), 1,4-bis[(3-ethyl-3-oxetanylmethoxy)methyl]benzene represented by the formula (24), 1,2-bis(3-ethyl-3-

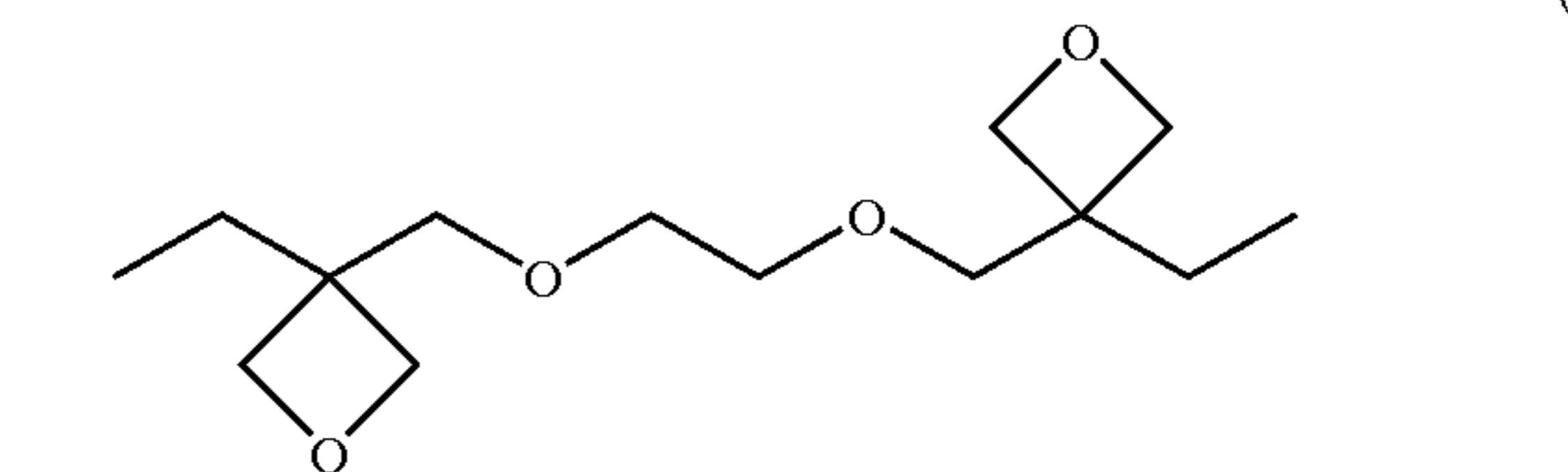
oxetanylmethoxy)ethane represented by the formula (25), trimethylolpropane tris(3-ethyl-3-oxetanylmethyl)ether represented by the formula (26), and the compounds represented by the general formula (19) described above.



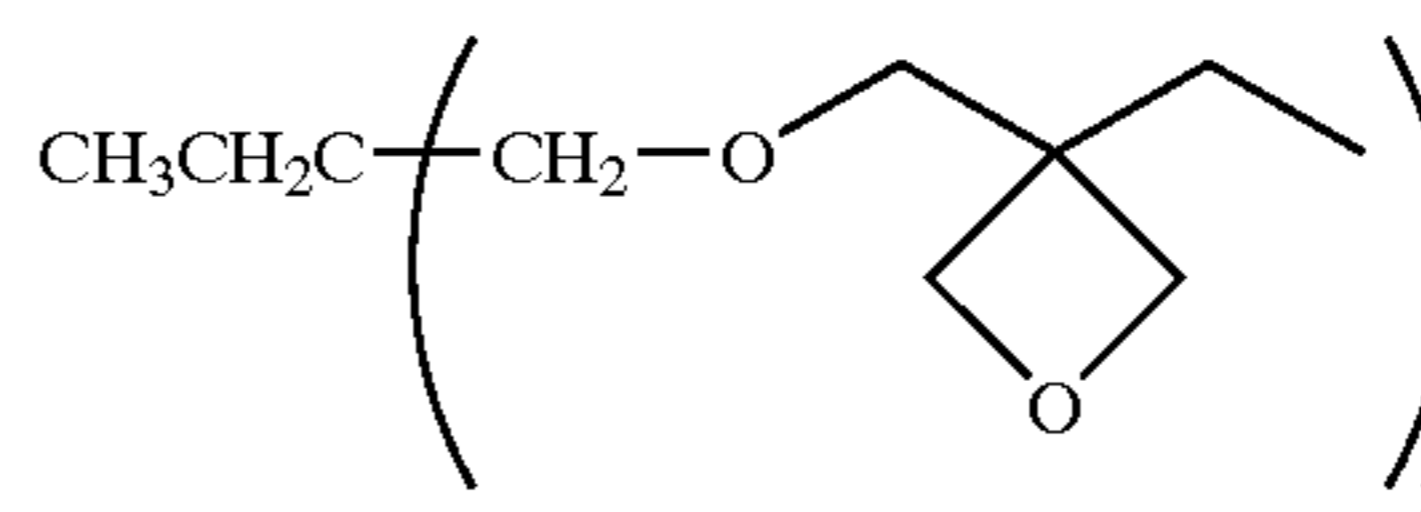
(23)



(24)



(25)



(26)

One species of the compound having the oxetane ring as described above may be used singly. Alternatively, two or more species of the compounds having the oxetane ring as described above may be used in combination.

Especially, it is desirable to use 3-ethyl-3-hydroxymethyloxetane, di[1-ethyl(3-oxetanyl)]methyl ether, or 3-ethyl-3-(2-ethylhexyloxymethyl)oxetane. When 3-ethyl-3-hydroxymethyloxetane, di[1-ethyl(3-oxetanyl)]methyl ether, or 3-ethyl-3-(2-ethylhexyloxymethyl)oxetane is used, then the reaction velocity of the photo-cationic polymerization is quickened in the present invention, and it is possible to quickly obtain the active energy beam-curable composition of the present invention. The oxetane compound is disclosed in detail in U.S. Pat. No. 5,463,084, the content of which has been incorporated herein by reference.

The alicyclic epoxy resin compound may include, for example, 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate, 2-(3,4-epoxycyclohexyl-5,5-spiro-3,4-epoxy)cyclohexane-meta-dioxane, bis(3,4-epoxycyclohexylmethyl)adipate, vinylcyclohexene oxide, 4-vinylepoxycyclohexane, bis(3,4-epoxy-6-methylcyclohexylmethyl)adipate, 3,4-epoxy-6-methylcyclohexyl-3',4'-epoxy-6'-methylcyclohexane carboxylate, methylenebis(3,4-epoxycyclohexane), dicyclopentadiene diepoxide, di(3,4-epoxycyclohexylmethyl)ether of ethylene glycol, ethylenebis(3,4-epoxycyclohexane carboxylate), epoxydized tetrabenzyl alcohol, and lactone modified 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate.

The cationic polymerizable substance, which is especially preferred as the alicyclic epoxy resin compound, is the epoxy compound having two or more alicyclic epoxy groups in one molecule, including, for example, 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate and bis(3,4-epoxycyclohexylmethyl)adipate.

Commercially available products of the cationic polymerizable substance, which are preferably usable as the alicyclic

epoxy resin compound, may include, for example, UVR-6100, UVR-6105, UVR-6110, and UVR-6128 (all produced by Union Carbide Corporation), Celloxide 2021, Celloxide 2021P, and Celloxide 2021A (all produced by Daicel Chemical Industries, Ltd.), and KRM-2100 and KRM-2199 (both

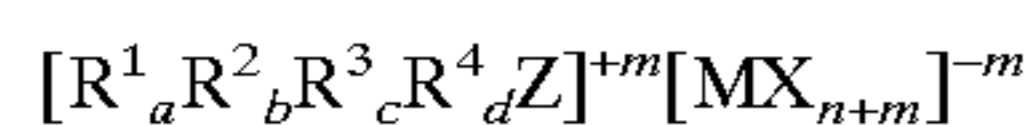
produced by Asahi Denka Co., Ltd.). It has been confirmed that the use of the epoxy resin compound makes it possible to further avoid the appearance of the volatile component and further avoid the occurrence of the odor and the cloudiness, even when the active energy beam-curable composition of the present invention is heated after being cured.

Especially, when the alicyclic epoxy resin compound is used, it is possible to enhance the crosslinking degree of the active energy beam-curable composition of the present invention. The occurrence of the odor and the cloudiness has been successfully avoided, when the active energy beam-curable composition of the present invention is heated after being cured.

It is preferable that the photo-cationic polymerizable substance other than the vinyl ether compound is used as the photo-cationic polymerizable substance to be used for the present invention, for the following reason. That is, if the vinyl ether compound is contained in the composition, when the cured composition is heated, then acetaldehyde and crotonaldehyde are volatilized, and they cause the odor and the cloudiness. However, it is allowable that the photo-cationic polymerizable substance other than the vinyl ether compound is used as a major component of the photo-cationic polymerizable substance, and a minute amount of the vinyl ether compound is contained within a range in which no harmful influence is exerted on the odor and the cloudiness.

#### (B) Photo-Cationic Polymerization Initiator

The photo-cationic polymerization initiator (B) (hereinafter referred to as "component (B)" as well), which constitutes the active energy beam-curable composition of the present invention, is the compound capable of releasing the substance to initiate the cationic polymerization by receiving the energy beam such as light. Especially preferred compounds may include the onium salt having the structure represented by the general formula (1). The onium salt is the compound which releases Lewis acid when it receives light.



In the formula, the cation is onium. Z is S, Se, Te, P, As, Sb, Bi, O, I, Br, Cl, or N≡N. R<sup>1</sup>, R<sup>2</sup>, R<sup>3</sup>, and R<sup>4</sup> are identical or different organic groups. a, b, c, and d are integers of 0 to 3 respectively. (a+b+c+d) is equal to the valency number of Z. M represents metalloid or metal to constitute the central atom of the halide complex, including, for example, B, P, As, Sb, Fe, Sn, Bi, Al, Ca, In, Ti, Zn, Sc, V, Cr, Mn, and Co. X represents halogen atom. m represents the net charge of the halide complex ion. n represents the number of atoms in the halide complex ion. Specifically, the anion (MX<sub>n+m</sub>) in the general formula (1) described above includes, for example, tetrafluoroborate (BF<sub>4</sub><sup>-</sup>), hexafluorophosphate (PF<sub>6</sub><sup>-</sup>), hexafluoroantimonate (SbF<sub>6</sub><sup>-</sup>), hexafluoroarsenate (AsF<sub>6</sub><sup>-</sup>), and hexachloroantimonate (SbCl<sub>6</sub><sup>-</sup>).

Further, it is possible to use the onium salt having the anion represented by the general formula [MX<sub>n</sub>(OH)]<sup>-</sup>. Furthermore, it is also possible to use the onium salt having another anion such as perchlorate ion (ClO<sub>4</sub><sup>-</sup>), trifluoromethanesulfonate ion (CF<sub>3</sub>SO<sub>3</sub><sup>-</sup>), fluorosulfonate ion (FSO<sub>3</sub><sup>-</sup>), toluenesulfonate ion, trinitrobenzenesulfonate anion, and trinitrotoluenesulfonate anion.

Among the onium salts as described above, the onium salt, which is especially effective as the component (B), is the aromatic onium salt. Especially, it is preferable to use, for example, aromatic halonium salt described, for example, in Japanese Patent Application Laid-open No. 50-151996 corresponding to England Patent No. 1516351 and Japanese Patent Application Laid-open No. 50-158680 corresponding to U.S. Pat. No. 4,394,403; VIA group aromatic onium salt described, for example, in Japanese Patent Application Laid-open No. 50-151997 corresponding to England Patent No. 1516511, Japanese Patent Application Laid-open No. 52-30899 corresponding to U.S. Pat. No. 4,256,828, Japanese Patent Application Laid-open No. 56-55420 corresponding to England Patent No. 2061280, and Japanese Patent Application Laid-Open No. 55-125105 corresponding to U.S. Pat. No. 4,197,174; VA group aromatic onium salt described, for example, in Japanese Patent Application Laid-open No. 50-158698 corresponding to U.S. Pat. No. 4,069,055; oxosulfoxonium salt described, for example, in Japanese Patent Application Laid-open No. 56-8428, Japanese Patent Application Laid-open No. 56-149402, and Japanese Patent Application Laid-open No. 57-192429 corresponding to U.S. Pat. No. 4,387,216; aromatic diazonium salt described, for example, in Japanese Patent Application Laid-open No. 49-17040; and thiopyrylium salt described in U.S. Pat. No. 4,139,655. It is also possible to refer, for example, to initiators based on aluminum complex/photolysis silicon compound and iron/allene complex.

Commercially available products of the cationic photopolymerization initiator, which are preferably usable as the component (B), may include, for example, UVI-6950, UVI-6970, UVI-6974, and UVI-6990 (all produced by Union Carbide Corporation); ADEKA Optomer SP-150, SP-151, SP-170, and SP-171 (all produced by Asahi Denka Co., Ltd.); Irgacure 261 (produced by Ciba-Geigy Corporation); CI-2481, CI-2624, CI-2639, and CI-2064 (all produced by NIPPON SODA CO., LTD.); CD-1010, CD-1011, and CD-1012 (all produced by Sartomer Company); and DTS-102, DTS-103, NAT-103, NDS-103, TPS-103, MDS-103, MPI-103, and BBI-103 (all produced by Midori Kagaku Co., Ltd.). Among the commercially available products described above, UVI-6970, UVI-6974, ADEKA Optomer SP-170, SP-171, CD-1012, and MPI-103 are especially preferred, because it is possible to allow the resin composition containing each of them to express the high curing sensitivity.

When the onium salt is contained, then it is possible to enhance the crosslinking degree of the active energy beam-curable composition of the present invention, and the occurrence of the odor and the cloudiness has been successfully avoided when the composition is heated after being cured.

#### (C) Polymerizable Compound Having Ethylenic Unsaturated Double Bond

The polymerizable compound having the ethylenic unsaturated double bond (C) (hereinafter referred to as "component (C)" as well), which constitutes the active energy beam-curable composition of the present invention, may include, for example, monomethacrylate, dimethacrylate, monoacrylate, diacrylate monomer, multifunctional monomer, and oligoacrylate.

Components of monomethacrylate, dimethacrylate, monoacrylate, and diacrylate monomers are summarized in Table 1 below.

TABLE 1

Classification	Component (C)	Trade Name	Maker	
monomethacrylate	methoxydiethylene glycol methacrylate (n = 2)	M20G	Shin-Nakamura Chemical Co., Ltd.	5
	methoxydiethylene glycol methacrylate (n = 2)	M40G	Shin-Nakamura Chemical Co., Ltd.	10
	laurylmethacrylate (n = 12)	LMA	Shin-Nakamura Chemical Co., Ltd.	10
	stearylmethacrylate (n = 18)	S	Shin-Nakamura Chemical Co., Ltd.	15
dimethacrylate	isostearyl-methacrylate	S-1800M	Shin-Nakamura Chemical Co., Ltd.	15
	2-hydroxy-3-acryloyloxypropyl-methacrylate	G-201P	Kyoeisha Chemical Co., Ltd.	20
	ethylene glycol dimethacrylate (n = 1)	1G	Shin-Nakamura Chemical Co., Ltd.	20
	diethylene glycol dimethacrylate (n = 2)	2G	Shin-Nakamura Chemical Co., Ltd.	25
	triethylene glycol dimethacrylate (n = 3)	3G	Shin-Nakamura Chemical Co., Ltd.	25
	polyethylene glycol 200 dimethacrylate (n = 4)	4G	Shin-Nakamura Chemical Co., Ltd.	30
	neopentyl glycol dimethacrylate	NPG	Shin-Nakamura Chemical Co., Ltd.	30
	1,3-butylene glycol dimethacrylate	BG	Shin-Nakamura Chemical Co., Ltd.	35
	1,4-butanediol dimethacrylate	BD	Shin-Nakamura Chemical Co., Ltd.	35
	3-methyl-1,5-pentanediol dimethacrylate	MPD	Shin-Nakamura Chemical Co., Ltd.	40
	1,6-hexanediol methacrylate	HD	Shin-Nakamura Chemical Co., Ltd.	40
	1,9-nonanediol methacrylate	NOD	Shin-Nakamura Chemical Co., Ltd.	45
2-methyl-1,8-octanediol dimethacrylate (85%) + 1,9-nonanediol dimethacrylate (15%)	IND	Shin-Nakamura Chemical Co., Ltd.	45	
1,10-decanediol dimethacrylate	DOD-N	Shin-Nakamura Chemical Co., Ltd.	50	
monoacrylate	phenoxyethylene glycol acrylate (n = 1)	AMP-10G	Shin-Nakamura Chemical Co., Ltd.	55
	phenoxyethylene glycol acrylate (n = 2)	AMP-20GY	Shin-Nakamura Chemical Co., Ltd.	55
	methoxyethylene glycol acrylate (n = 3)	AMP-30G	Shin-Nakamura Chemical Co., Ltd.	60
	lauryl acrylate	LA	Shin-Nakamura Chemical Co., Ltd.	60
	isostearyl acrylate	S-1800S	Shin-Nakamura Chemical Co., Ltd.	65
	morpholine acrylate	ACMO	Kohjin Co., Ltd.	65

TABLE 1-continued

Classification	Component (C)	Trade Name	Maker
monoacrylate	lauryl acrylate	Light-Acrylate L-A	Kyoeisha Chemical Co., Ltd.
	stearyl acrylate	Light-Acrylate S-A	Kyoeisha Chemical Co., Ltd.
	butoxyethyl acrylate	Light-Acrylate BO-A	Kyoeisha Chemical Co., Ltd.
	methoxytriethylene glycol acrylate	Light-Acrylate MTG-A	Kyoeisha Chemical Co., Ltd.
	isobornyl acrylate	Light-Acrylate IB-XA	Kyoeisha Chemical Co., Ltd.
	phenoxypolyethylene acrylate	Light-Acrylate P-200A	Kyoeisha Chemical Co., Ltd.
	2-ethylhexylcarbitol acrylate	Aronix M-120	Toagosei Co., Ltd.
	phenol EO modified acrylate (n = 2)	Aronix M-101	Toagosei Co., Ltd.
	4-hydroxybutyl acrylate	4-HBA	Osaka Organic Chemical Industry Ltd.
	3-methoxybutyl acrylate	V#158	Osaka Organic Chemical Industry Ltd.
	benzyl acrylate	V#160	Osaka Organic Chemical Industry Ltd.
	phenoxyethyl acrylate	V#192	Osaka Organic Chemical Industry Ltd.
diacrylate	dipropylene glycol diacrylate	APG-100	Shin-Nakamura Chemical Co., Ltd.
	tripropylene glycol diacrylate	APG-200	Shin-Nakamura Chemical Co., Ltd.
	neopentyl glycol diacrylate	A-NPG-S	Shin-Nakamura Chemical Co., Ltd.
	1,8-hexanediol acrylate	A-HD	Shin-Nakamura Chemical Co., Ltd.
	1,9-nonanediol acrylate	A-NOD	Shin-Nakamura Chemical Co., Ltd.
	2-methyl-1,8-octanediol acrylate (85%) + 1,9-nonanediol acrylate (15%)	A-IND	Shin-Nakamura Chemical Co., Ltd.
	triethylene glycol diacrylate	Light-Acrylate 3EG-A	Kyoeisha Chemical Co., Ltd.
	PEG400# diacrylate	Light-Acrylate 9EG-A	Kyoeisha Chemical Co., Ltd.
	1,9-nonanediol diacrylate	Light-Acrylate 1,9ND-A	Kyoeisha Chemical Co., Ltd.
	tripropylene glycol diacrylate	Aronix M-220 (TPGDA)	Toagosei Co., Ltd.
	polyethylene glycol diacrylate	Aronix M-240 (TEGDA)	Toagosei Co., Ltd.
	1,9-nonanediol diacrylate	V#260	Osaka Organic Chemical Industry Ltd.
tripropylene glycol diacrylate	V#310HP	Osaka Organic Chemical Industry Ltd.	

TABLE 1-continued

Classification	Component (C)	Trade Name	Maker
	tripropylene glycol diacrylate	V#310HG	Osaka Organic Chemical Industry Ltd.

The multifunctional monomer may include, for example, trimethylolpropane triacrylate, pentaerythritol triacrylate, trimethylolpropane EO-added triacrylate, glycerol PO-added triacrylate, trisacryloyloxyethyl phosphate, and pentaerythritol tetraacrylate.

Further, the oligoacrylate may include, for example, tetra furfuryl alcohol oligoacrylate, ethylcarbitol oligoacrylate, 1,4-butanediol oligoacrylate, neopentyl glycol oligoacrylate, 1,6-hexanediol oligoacrylate, trimethylolpropane oligoacrylate, and pentaerythritol oligoacrylate.

Commercially available products (other than those shown in Table 1) of the polymerizable compound having the ethylenic unsaturated double bond, which are preferably usable as the component (C), may include, for example, V#295, V#300, V#360, V#GPT, V3PA, V#400, V#150D, V#190d, V#195D, V#215D, V#230D, V#295D, and V#400D (all produced by Osaka Organic Chemical Industry Ltd.), and TMPT, TMM-360, A-TMPT, A-TMPT-3EO, AT-30E, A-TMPT-3PO, A-9300, A-TMM-3, A-TMM-3L, A-TMM-T, ATM-2.4E, ATM-4E, ATM-35E, ATM-4P, AD-TMP, AD-TMP-L, and A-9530 (all produced by Shin-Nakamura Chemical Co., Ltd.). The content of the polymerizable compound having the ethylenic unsaturated double bond is preferably about 5% by weight to about 58% by weight, in view of the fact that the occurrence of the odor and the cloudiness is avoided. Further, when the composition is used for an ink head of an ink jet apparatus (or ink discharging apparatus) which performs photo-molding or photo-curing printing, the content of the polymerizable compound having the ethylenic unsaturated double bond is preferably about 10% by weight to about 48% by weight. The content of the polymerizable compound having the ethylenic unsaturated double bond affects the viscosity of the composition. Therefore, when the composition is used as the ink for the ink-jet application, the content is preferably adjusted within a range in which the viscosity of the composition upon discharging the ink from an ink jet head is 2 cps to 30 cps.

#### (D) Other Components

In addition to the essential components described above, the active energy beam-curable composition of the present invention may be optionally blended with inactive components such as an inorganic filler, a dye, a pigment, a viscosity-adjusting agent, a treating agent, and an ultraviolet radiation-blocking agent. Further, if necessary, it is possible to blend a component capable of radical polymerization with the active energy beam, such as a radical photopolymerization initiator and a compound having (meth)acryloyl group.

When the active energy beam-curable composition of the present invention is cured with ultraviolet radiation, a photosensitizing agent may be blended in order to further improve the curing performance. Typical sensitizing agents, which are usable in the present invention, include compounds disclosed by "Crivello" in Adv. in Polymer Sci., 62, 1 (1984). Specifically, the sensitizing agent includes, for example, pyrene, perylene, acridine orange, thioxanthone,

2-chlorothioxanthone, and benzoflavin. When the active energy beam-curable composition of the present invention is used as an ink for ink-jet printing, the active energy beam-curable composition may contain, for example, a coloring agent such as a synthetic dye or a pigment, an inorganic filler, a surface-treating agent (a surfactant and/or a dispersing agent), a viscosity-adjusting agent, a treating agent, and an ultraviolet radiation-blocking agent, in addition to the photo-cationic polymerizable substance, the photo-cationic polymerization initiator, and the polymerizable compound having the unsaturated double bond. When the composition is used as an ink for ink jet application, the composition is prepared so that the viscosity of the composition upon ink jetting from an ink jet head is 2 cps to 30 cps. U.S. Pat. No. 5,059,266 discloses, for example, additives, contents thereof, and characteristics of the ink when the active energy beam-curable composition is used as an ink for ink-jet printing. The content of this United States Patent has been incorporated herein by reference. The active energy beam, which is used to cure the composition of the present invention, includes, ultraviolet radiation, infrared radiation, and electron beam (EB).

Next, explanation will be made for inks (active energy beam-curable compositions) obtained in Examples and Comparative Examples.

#### COMPARATIVE EXAMPLE 1

A transparent liquid composition was obtained by mixing and agitating, at room temperature in a dark room, a photo-cationic polymerizable substance, i.e., 30 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 30 parts by weight of limonene dioxide (produced by elf atochem), 40 parts by weight of triethylene glycol divinyl ether (trade name: DVE-3, produced by ISP), and a photo-cationic polymerization initiator, i.e., 4 parts by weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%).

#### COMPARATIVE EXAMPLE 2

A transparent liquid composition was obtained in the same manner as in Comparative Example 1 except that a photo-cationic polymerizable substance, i.e., 20 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 80 parts by weight of ethylene glycol dimethacrylate (trade name: 1G, produced by Shin-Nakamura Chemical Co., Ltd.), and a photo-cationic polymerization initiator, i.e., 4 parts by weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%) were used.

#### COMPARATIVE EXAMPLE 3

A transparent liquid composition was obtained in the same manner as in Comparative Example 1 except that a photo-cationic polymerizable substance, i.e., 30 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 70 parts by weight of ethylene glycol dimethacrylate (trade name: 1G, produced by Shin-Nakamura Chemical Co., Ltd.), and a photo-cationic polymerization initiator, i.e., 4 parts by

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weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%) were used.

## COMPARATIVE EXAMPLE 4

A transparent liquid composition was obtained in the same manner as in Comparative Example 1 except that a photo-cationic polymerizable substance, i.e., 20 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 80 parts by weight of 1,3-butylene glycol dimethacrylate (trade name: BG, produced by Shin-Nakamura Chemical Co., Ltd.), and a photo-cationic polymerization initiator, i.e., 4 parts by weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%) were used.

## COMPARATIVE EXAMPLE 5

A transparent liquid composition was obtained in the same manner as in Comparative Example 1 except that a photo-cationic polymerizable substance, i.e., 30 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 70 parts by weight of 1,3-butylene glycol dimethacrylate (trade name: BG, produced by Shin-Nakamura Chemical Co., Ltd.), and a photo-cationic polymerization initiator, i.e., 4 parts by weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%) were used.

## EXAMPLE 1

A transparent liquid composition was obtained in the same manner as in Comparative Example 1 except that a photo-cationic polymerizable substance, i.e., 30 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 30 parts by weight of limonene dioxide (produced by elf atochem), 40 parts by weight of isobornyl acrylate (trade name: IBXA, produced by Kyoisha Chemical Co., Ltd.), and a photo-cationic polymerization initiator, i.e., 4 parts by weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%) were used.

## EXAMPLE 2

A transparent liquid composition was obtained in the same manner as in Comparative Example 1 except that a photo-cationic polymerizable substance, i.e., 30 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 30 parts by weight of limonene dioxide (produced by elf atochem), 40 parts by weight of methoxydiethylene glycol methacrylate (trade name: M20G, produced by Shin-Nakamura Chemical Co., Ltd.), and a photo-cationic polymerization initiator, i.e., 4 parts by weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%) were used.

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## EXAMPLE 3

A transparent liquid composition was obtained in the same manner as in Comparative Example 1 except that a photo-cationic polymerizable substance, i.e., 30 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 30 parts by weight of limonene dioxide (produced by elf atochem), 40 parts by weight of ethylene glycol dimethacrylate (trade name: 1G, produced by Shin-Nakamura Chemical Co., Ltd.), and a photo-cationic polymerization initiator, i.e., 4 parts by weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%) were used.

## EXAMPLE 4

A transparent liquid composition was obtained in the same manner as in Comparative Example 1 except that a photo-cationic polymerizable substance, i.e., 30 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 30 parts by weight of limonene dioxide (produced by elf atochem), 40 parts by weight of polyethylene glycol 200 dimethacrylate (trade name: 4G, produced by Shin-Nakamura Chemical Co., Ltd.), and a photo-cationic polymerization initiator, i.e., 4 parts by weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%) were used.

## EXAMPLE 5

A transparent liquid composition was obtained in the same manner as in Comparative Example 1 except that a photo-cationic polymerizable substance, i.e., 30 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 30 parts by weight of limonene dioxide (produced by elf atochem), 40 parts by weight of neopentyl glycol dimethacrylate (trade name: NPG, produced by Shin-Nakamura Chemical Co., Ltd.), and a photo-cationic polymerization initiator, i.e., 4 parts by weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%) were used.

## EXAMPLE 6

A transparent liquid composition was obtained in the same manner as in Comparative Example 1 except that a photo-cationic polymerizable substance, i.e., 30 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 30 parts by weight of limonene dioxide (produced by elf atochem), 40 parts by weight of 1,3-butylene glycol dimethacrylate (trade name: BG, produced by Shin-Nakamura Chemical Co., Ltd.), and a photo-cationic polymerization initiator, i.e., 4 parts by weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%) were used.

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## EXAMPLE 7

A transparent liquid composition was obtained in the same manner as in Comparative Example 1 except that a photo-cationic polymerizable substance, i.e., 30 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 30 parts by weight of limonene dioxide (produced by elf atochem), 40 parts by weight of 1,6-hexanediol dimethacrylate (trade name: HD, produced by Shin-Nakamura Chemical Co., Ltd.), and a photo-cationic polymerization initiator, i.e., 4 parts by weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%) were used.

## EXAMPLE 8

A transparent liquid composition was obtained in the same manner as in Comparative Example 1 except that a photo-cationic polymerizable substance, i.e., 30 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 30 parts by weight of limonene dioxide (produced by elf atochem), 40 parts by weight of tripropylene glycol diacrylate (trade name: APG-200, produced by Shin-Nakamura Chemical Co., Ltd.), and a photo-cationic polymerization initiator, i.e., 4 parts by weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%) were used.

## EXAMPLE 9

A transparent liquid composition was obtained in the same manner as in Comparative Example 1 except that a photo-cationic polymerizable substance, i.e., 30 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 10 parts by weight of limonene dioxide (produced by elf atochem), 60 parts by weight of ethylene glycol dimethacrylate (trade name: 1G, produced by Shin-Nakamura Chemical Co., Ltd.), and a photo-cationic polymerization initiator, i.e., 4 parts by weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%) were used.

## EXAMPLE 10

A transparent liquid composition was obtained in the same manner as in Comparative Example 1 except that a photo-cationic polymerizable substance, i.e., 30 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 20 parts by weight of limonene dioxide (produced by elf atochem), 50 parts by weight of ethylene glycol dimethacrylate (trade name: 1G, produced by Shin-Nakamura Chemical Co., Ltd.), and a photo-cationic polymerization initiator, i.e., 4 parts by weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%) were used.

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## EXAMPLE 11

A transparent liquid composition was obtained in the same manner as in Comparative Example 1 except that a photo-cationic polymerizable substance, i.e., 20 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 75 parts by weight of limonene dioxide (produced by elf atochem), 5 parts by weight of ethylene glycol dimethacrylate (trade name: 1G, produced by Shin-Nakamura Chemical Co., Ltd.), and a photo-cationic polymerization initiator, i.e., 4 parts by weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%) were used.

## EXAMPLE 12

A transparent liquid composition was obtained in the same manner as in Comparative Example 1 except that a photo-cationic polymerizable substance, i.e., 30 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 10 parts by weight of limonene dioxide (produced by elf atochem), 60 parts by weight of 1,3-butylene glycol dimethacrylate (trade name: BG, produced by Shin-Nakamura Chemical Co., Ltd.), and a photo-cationic polymerization initiator, i.e., 4 parts by weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%) were used.

## EXAMPLE 13

A transparent liquid composition was obtained in the same manner as in Comparative Example 1 except that a photo-cationic polymerizable substance, i.e., 30 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 20 parts by weight of limonene dioxide (produced by elf atochem), 50 parts by weight of 1,3-butylene glycol dimethacrylate (trade name: BG, produced by Shin-Nakamura Chemical Co., Ltd.), and a photo-cationic polymerization initiator, i.e., 4 parts by weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%) were used.

## EXAMPLE 14

A transparent liquid composition was obtained in the same manner as in Comparative Example 1 except that a photo-cationic polymerizable substance, i.e., 20 parts by weight of 3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane carboxylate (trade name: UVR-6110, produced by Union Carbide Corporation), 75 parts by weight of limonene dioxide (produced by elf atochem), 5 parts by weight of 1,3-butylene glycol dimethacrylate (trade name: BG, produced by Shin-Nakamura Chemical Co., Ltd.), and a photo-cationic polymerization initiator, i.e., 4 parts by weight of a mixture of triarylsulfonium hexafluorophosphate salts and propylene carbonate (trade name: UVI-6990, produced by Union Carbide Corporation, active ingredient: 50%) were used.

## I. Curing Method

The obtained composition was applied to a polyester film so that the thickness was about 15  $\mu\text{m}$ , and then the



composition was cured by allowing the composition to pass through twice an ultraviolet radiation-irradiating apparatus of the conveyer type, the apparatus being installed with a super high voltage mercury lamp of 80 W/cm (lamp height: 155 mm, conveyer speed: 7.5 m/min., radiation intensity: 80 mW/cm<sup>2</sup>, 160 mJ/cm<sup>2</sup>). Test machine was Eye Mini Grandage ECS-151U (lamp output: 1.0 kW, belt conveyer type, produced by Eye Graphics Co., Ltd.).

## II. Cloudiness Test

A test piece (cured composition) was placed on a hot plate heated to 110° C., and the test piece was covered with a glass vessel arranged thereon. The cloudiness in the glass vessel, which possibly appeared after being left to stand for 24 hours, was visually confirmed. The case with no cloudiness is indicated by "+", and the case with any cloudiness was indicated by "-".

The transparent liquid compositions (active energy beam-curable compositions) obtained in Examples 1 to 14 and Comparative Examples 1 to 5 described above were cured to perform the cloudiness test. Obtained results are summarized in Table 2 below. The polymerizable compounds each having the unsaturated double bond (components (C)) used for the respective compositions are entered in Table 2.

TABLE 2

	Component (C)	Cloudiness test (110° C. × 24 H)
Comparative Example 1	DVE-3Tech	-
Comparative Example 2	1G	-
Comparative Example 3	1G	-
Comparative Example 4	BG	-
Comparative Example 5	BG	-
Example 1	IBXA	+
Example 2	M20G	+
Example 3	1G	+
Example 4	4G	+
Example 5	NPG	+
Example 6	BG	+
Example 7	HD	+
Example 8	APG-200	+
Example 9	1G	+
Example 10	1G	+
Example 11	1G	+
Example 12	BG	+
Example 13	BG	+
Example 14	BG	+

According to Table 2, it has been confirmed that no cloudiness appears in the cloudiness test by containing monomethacrylate, dimethacrylate, monoacrylate, or diacrylate.

The liquid compositions obtained in Examples 1 to 14 were used as transparent inks to perform the printing by using a printer which will be explained later. As a result, good output results were successfully obtained, in which neither odor nor cloudiness appeared. On the other hand, the liquid compositions obtained in Comparative Examples 1 to 5 were used as inks to perform the printing by using a printer. As a result, both of odor and cloudiness were confirmed.

An embodiment of an ink jet printer as an ink discharging apparatus in accordance with the invention will be described as below with reference to the accompanying drawings.

Embodiments of the invention will be described with reference to the accompanying drawings. As shown in FIG. 1, a color ink jet printer 100 of a first embodiment includes a carriage 102 that can move in a main scanning direction

(an X-axis direction) while guided by X-axis guide bars 101, 101 fixed to a frame F. A print head 104 as an ink discharging head and ultraviolet (UV) lamps 105L, 105R are provided on the carriage 102. The print head 104 includes a Y head 104Y for yellow (Y) ink, an M head 104M for magenta (M) ink, a C head 104C for cyan (C) ink and a K head 104K for black (K) ink. The ink to be ejected from the heads 104Y, 104M, 104C, 104K is an ultraviolet curing type ink (UV ink). The ink can be prepared by adding an coloring agent to the transparent liquid compositions prepared in the above examples. The heads 104Y, 104M, 104C and 104K are aligned in a sub scanning direction (a Y-axis direction), which is perpendicular to the main scanning direction, on the carriage 102. The UV lamps 105L, 105R are provided on right and left sides of the print head 104 so as to sandwich the print head 104 therebetween.

The UV lamps 105L, 105R irradiate the UV ink, ejected from the respective ink heads onto a recording medium 103, with ultraviolet light to increase its viscosity to cure the ink.

Each of the heads 104Y, 104M, 104C, 104K is formed with a plurality of nozzles, which face a platen 108 (described later) and are basically aligned in parallel to the sub scanning direction. The Y head 104Y, the M head 104M, the C head 104C and the K head 104K are connected with an yellow ink tank 104Y1 containing UV yellow ink, a magenta ink tank 104M1 containing UV magenta ink, a cyan ink tank 104C1 containing UV cyan ink, a black ink tank 104K1 containing UV black ink, respectively. Each UV ink is supplied from the respective ink tanks 104Y1, 104M1, 104C1, 104K1 to the respective nozzles through each ink flow path in the respective heads 104Y, 104M, 104C, 104K. Therefore, when one-line printing is performed by ejecting ink droplets of the UV ink from each head 104Y, 104M, 104C, 104K while the carriage 102 is moved in the main scanning direction, a printed portion is formed on the recording medium 103. A width of the printed portion is equal to a carriage traveling distance in the main scanning direction. A length of the printed portion is equal to a width of ink ejection from each head (a width of ink ejection from the nozzles in each head at a time).

An X-axis motor 106 that moves the carriage 102 in the sub scanning direction is attached to one of side walls the frame F. A driven pulley R is rotatably provided to the other side wall of the frame F. A timing belt 107 is stretched between the driven pulley R and a drive pulley P fixed to an output shaft of the X-axis motor 106. The timing belt 107 is engaged with the carriage 102. Thus, as the X-axis motor 106 is driven, the carriage 102 travels from side to side along the X-axis guide bars 101, 101, that is, the X-axis guide bars 101, 101 and the X-axis motor 106 moves the carriage 102 in the main scanning direction with respect to the recording medium 103. The platen 108 is supported by the frame F, under the carriage 102 and the X-axis guide bars 101, 101 so as to be movable in the sub scanning direction perpendicular to the main scanning direction. The platen 108 is formed with a plurality of air-intakes at its top, and a recording medium 103, such as a plastic plate, is placed on the top. Air is taken in from the air-intakes with the recording medium 103 placed on the top (a surface for placing the recording medium 103), so that a negative pressure is generated and thus the recording medium 103 is statically held on the platen 108. As described above, the platen 108 is provided with an air taking mechanism for holding the recording medium 103. The air taking mechanism also functions as a holding device for holding the recording medium 103. The holding device is not limited to the air taking mechanism. However, various types of holding

devices can be adopted. For example, a clamp mechanism that mechanically presses and holds the recording medium **103** on the platen **108** can be adopted.

A screw shaft **110**, extending in a direction perpendicular to the main scanning direction, is rotatably supported by the frame **F**. One end of the screw shaft **110** is secured to an output shaft of a Y-axis motor **109**. A bearing **111**, into which the screw shaft **110** is threaded, is fixed under the platen **108**. As the screw shaft **110** is rotated by the Y-axis motor **109**, the platen **108** is moved in the sub scanning direction. A screw mechanism having the screw shaft **110**, the bearing **111** and the Y-axis motor **109** moves the platen **108** in the sub scanning direction with respect to the carriage **102**.

The amount and direction of travel of the carriage **102** is controlled by controlling the amount and direction of rotation of the X-axis motor **106**. For example, a pulse motor is preferably used as the X-axis motor **106**. The amount and direction of travel of the platen **108** is controlled by controlling the amount and direction of rotation of the Y-axis motor **109**. For example, a pulse motor is preferably used as the Y-axis motor **109**.

An area **112** is a maintenance area in which purging, wiping, flushing and the like are performed. As shown in FIG. 1A, the UV lamps **105L**, **105R** are longer than the print head **104** in a direction of feeding the recording medium **103** by a predetermined length.

As shown in FIG. 2, a control system of the color ink jet printer **100** includes a CPU **201**, a ROM **202**, an EEPROM **202a**, a RAM **203**, an input interface **204** and an output interface **205**. The output interface **205** is connected with a Y head drive circuit **206Y**, an M head drive circuit **206M**, a C head drive circuit **206C**, a K head drive circuit **206K**, an X-axis motor drive circuit **207**, a Y-axis motor drive circuit **208** and UV lamp drive circuits **209L**, **209R**. A control signal and data from a personal computer **220** are inputted to the input interface **204**.

FIG. 3 is a flowchart for performing print control processing. At first, the CPU **201** determines whether a command to start printing has been issued from the personal computer **220** (S301). When the CPU **201** determines that the command to start printing has been issued (S301:YES), the CPU **201** receives print data (S302). After the CPU **201** receives a page of print data, the CPU **201** stores the print data in the RAM **203** and develops the print data into one page of data of each color of Y, M, C, K (S303). Then, the CPU **201** lights the UV lamps **105L**, **105R** (S304). After that, based on the Y data, while outputting one line of dot data to the Y head drive circuit **206Y**, the CPU **201** performs one-line printing in the UV yellow ink by moving the carriage **102** in the main scanning direction (S305).

One-line printing is performed as described below. A print data developing area for one page is provided according to color in the RAM **203**. One page of dot data is developed into each print data developing area in the bitmap format in which dots are aligned in the main scanning direction and in the sub scanning direction. Each dot in the dot data represents ejection (nonejection) of an ink droplet from one nozzle. One-line printing of dot data is that dots equal to the number of nozzles aligned in the sub scanning direction are selected in the sub scanning direction and rows of the selected dots are aligned in the main scanning direction from a print start side to a print end side. Dots equal to the number of nozzles aligned in the sub scanning direction are selected and referred to as dot rows. While the dot rows are sequentially read from the print data developing area from the print start side to the print end side and supplied to each heads

**104Y**, **104M**, **104C**, **104K**, the print head **104** is moved in the main scanning direction, thereby one line of a printed portion is formed on a recording medium **103**. Upon performing the printing, a printed portion is irradiated with ultraviolet light by the UV lamps **105L**, **105R** disposed adjacent to the print head **104** to increase its viscosity.

Next, the CPU **201** moves the platen **108** in the sub scanning direction by one line (by the width of the ink ejection from the print head **104**). Then, while outputting each one line of the dot data to the respective Y and M head drive circuits **206Y** and **206M** based on each Y and M data, respectively, the CPU **201** performs one-line printing in each of the UV yellow ink and the UV magenta ink by moving the carriage **102** in the main scanning direction (S306).

That is, at S305, one-line printing of the first line of the Y data has been performed. Therefore, next, dot data of the Y data for a second line is selected and provided to the Y head drive circuit **206Y** and dot data of the M data for a first line is selected and provided to the M head drive circuit **206M**. In other words, ink droplets of the UV magenta ink to be ejected according to the M data are ejected on the first line of the UV yellow ink ejected at S305. However, the viscosity of the UV yellow ink ejected prior to the ejection of the UV magenta ink has already been increased by the irradiation by the UV lamps **105L**, **105R**, so that the UV yellow ink and the UV magenta ink do not intermix and clarity of the ink colors is not degraded. Upon performing the printing in each of the UV yellow ink and the UV magenta ink, the UV yellow ink and the UV magenta ink ejected at S306 are irradiated with ultraviolet light by the UV lamps **105L**, **105R** to increase their viscosity.

Next, the CPU **201** moves the platen **108** in the sub scanning direction by one line. Then, while outputting each one line of the dot data to the respective Y, M and C head drive circuits **206Y**, **206M** and **206C** based on each Y, M and C data, respectively, the CPU **201** performs one-line printing in each of the UV yellow ink, the UV magenta ink and the UV cyan ink by moving the carriage **102** in the main scanning direction (S307).

That is, dot data of the Y data for a third line, dot data of the M data for a second line and dot data of the C data for a first line are selected and supplied to the Y head drive circuit **206Y**, the M head drive circuit **206M** and the C head drive circuit **206C**, respectively. In other words, ink droplets of the UV cyan ink to be ejected according to the C data are ejected on the first line of the UV magenta ink ejected at S306. However, the viscosity of the UV magenta ink has already been increased by the irradiation by the UV lamps **105L**, **105R**, so that the UV magenta ink and the UV cyan ink do not intermix and clarity of the ink colors is not degraded. In short, the first line of the UV magenta ink is ejected over the first line of the UV yellow ink, and the first line of the UV cyan ink is ejected over the first line of the UV magenta ink. One-line printing according to the M and Y data is also performed in the same manner described above.

Next, the CPU **201** moves the platen **108** in the sub scanning direction by one line. Then, while outputting each one line of the dot data to the respective Y, M, C and K head drive circuits **206Y**, **206M**, **206C** and **206K** based on each Y, M, C and K data, respectively, the CPU **201** performs one-line printing in each of the UV yellow ink, the UV magenta ink, the UV cyan ink and the UV black ink by moving the carriage **102** in the main scanning direction. (S308).

That is, dot data of the Y data for a fourth line, dot data of the M data for a third line, dot data of the C data for a

second line and dot data of the K data for a first line are selected and supplied to the Y head drive circuit **206Y**, the M head drive circuit **206M**, the C head drive circuit **206C** and the K head drive circuit **206K**, respectively. At **S308**, if the K data to be read is dot data for a Nth line (N is a whole number), the Y dot data is dot data for N+3th line, the M data is dot data for N+2th line and the C data is dot data for N+1th line. If the CPU **201** makes a determination in the negative at **S309** described later, the CPU **201** sequentially increments N and reads dot data. The ejected ink droplets are irradiated with the ultraviolet light to increase their viscosity immediately after the ink droplets are ejected.

Then, the CPU **201** determines whether all printing has completed (**S309**). If the printing has not completed yet (**S309:NO**), dot data of each of the Y, M, C and K data for a next line is read and printing is performed according to the data at **S308** until the printing is completed.

The repetition of the one-line printing and the transport of the platen **108** will make the printing operation reach a last line in order of the Y head **104Y**, the M head **104M**, the C head **104C** and the K head **104K**. When dot data that is to be used for printing is not left as to all the heads **104Y**, **104M**, **104C**, **104K**, the CPU **201** determines that the printing has completed. At **S308**, dot data of each color is read. At that time, if dot data of each color has already been used for printing in a last line, the CPU **201** outputs dot data that means nonejection of ink as to the dot data of each color.

When the CPU **201** determines that the printing has completed (**S309:YES**), the CPU **201** moves the carriage **102** at idle by three lines without ejecting ink droplets from the heads **104Y**, **104M**, **104C**, **104K** (**S312**). The UV black ink ejected last is irradiated with the ultraviolet light for the time of the one-line printing at the moment. That is, the irradiation time of the ultraviolet light to the last printed portion in the UV black ink is shorter than that to the other printed portion, so that the carriage **102** is moved at idle to irradiate the ultraviolet light to cover a shortfall. That is, the UV yellow ink is irradiated with ultraviolet light three times before the UV black ink is ejected, the UV magenta ink is irradiated twice before the UV black ink is ejected, and the UV cyan ink is irradiated once before the UV black ink is ejected. Therefore, if the irradiation of the ultraviolet light is finished when the ejection of the UV black ink in the last line is completed, the irradiation amount of the ultraviolet light to the area of three lines from the last printed line is less than that to the other portions. Due to lack of the irradiation, the viscosity of the UV black ink ejected in the last line may not be sufficiently increased or the ink may not be sufficiently cured. To avoid this, the carriage **102** is moved at idle the number of times that is one less than the number of heads to average the time of the irradiation of the ultraviolet light. By doing so, the inks ejected onto the recording medium **103** are equally cured. After that, the CPU **201** extinguishes the UV lamps **105L**, **105R** (**S310**), and thus this processing has completed.

According to the color ink jet printer **100** of the first embodiment, the UV lamps **105L**, **105R** are mounted on the carriage **102** to be placed front and/or behind of the heads **104Y**, **104M**, **104C**, **104K** in the main scanning direction. Thus, cure of ink can be performed in parallel with ink ejection from each heads **104Y**, **104M**, **104C**, **104K**. The color ink jet printer **100** does not become large in size, and the time required for printing can be shortened. Further, the colors of the printing can be clearly represented.

The UV lamps **105L**, **105R** are longer than the heads **104Y**, **104M**, **104C**, **104K** in the sub scanning direction, so

that cure of ink can be performed with respect to all heads **104Y**, **104M**, **104C**, **104K** at a time in parallel with ink ejection from the heads **104Y**, **104M**, **104C**, **104K**. Further, the substantially same amount of energy can be irradiated to each color of ink on a recording medium **103**, so that the ink can be evenly cured.

The active energy beam-curable composition of the present invention has been explained with reference to Examples. However, the present invention is not limited thereto. The active energy beam-curable composition can be constituted by using other components and compositions within appended claims. 3,4-Epoxy cyclohexylmethyl-3',4'-epoxycyclohexane carboxylate has been used as the photo-cationic polymerizable substance. In place of this compound, it is possible to use a variety of photo-cationic polymerizable substances as described in the specification. Especially, it is preferable to use the oxetane compound as the photo-cationic polymerizable substance. The active energy beam-curable composition of the present invention can be applied to a variety of objectives. Even when the composition is heated after being cured, the occurrence of the odor and the cloudiness is suppressed.

What is claimed is:

1. An active energy beam-curable composition comprising:
  - a photo-cationic polymerizable substance;
  - a photo-cationic polymerization initiator;
  - limonene dioxide; and
  - a polymerizable compound having an ethylenic unsaturated double bond,
 wherein a weight ratio of a total weight of the limonene dioxide and the polymerizable compound having an ethylenic unsaturated double bond to a weight of the photo-cationic polymerizable substance is 7:3 to 8:2.
2. The composition according to claim 1, wherein the polymerizable compound having the ethylenic unsaturated double bond is a monomer selected from the group consisting of monomethacrylate, dimethacrylate, monoacrylate, and diacrylate.
3. The composition according to claim 1, wherein the photo-cationic polymerizable substance includes an epoxy resin compound.
4. The composition according to claim 3, wherein the epoxy resin compound is an alicyclic epoxy resin compound.
5. The composition according to claim 1, wherein the photo-cationic polymerizable substance is an oxetane compound.
6. The composition according to claim 1, wherein the photo-cationic polymerization initiator contains an onium salt.
7. The composition according to claim 1, wherein the composition is an ink.
8. The composition according to claim 7, further comprising a coloring agent.
9. An ink discharging apparatus comprising:
  - an ink discharging head which discharges an ink;
  - an ink tank which stores the ink to be supplied to the ink discharging head; and
  - a light source which is used to cure the discharged ink, wherein the ink comprises:
    - a photo-cationic polymerizable substance;
    - a photo-cationic polymerization initiator;
    - limonene dioxide; and
    - a polymerizable compound having an ethylenic unsaturated double bond,

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wherein a weight ratio of a total weight of the limonene dioxide and the polymerizable compound having an ethylenic unsaturated double bond to a weight of the photo-cationic polymerizable substance is 7:3 to 8:2.

**10.** The ink discharging apparatus according to claim **9**,<sup>5</sup> wherein the polymerizable compound having the ethylenic unsaturated double bond is a monomer selected from the group consisting of monomethacrylate, dimethacrylate, monoacrylate, and diacrylate.

**11.** The ink discharging apparatus according to claim **9**,<sup>10</sup> wherein the photo-cationic polymerizable substance includes an epoxy resin compound.

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**12.** The ink discharging apparatus according to claim **11**, wherein the epoxy resin compound is an alicyclic epoxy resin compound.

**13.** The ink discharging apparatus according to claim **9**, wherein the photo-cationic polymerizable substance is an oxetane compound.

**14.** The ink discharging apparatus according to claim **9**, wherein the photo-cationic polymerization initiator contains an onium salt.

**15.** The ink discharging apparatus according to claim **9**, wherein the ink further comprises a coloring agent.

\* \* \* \* \*