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Baeck et al.

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[54] **COMPACT DETERGENT COMPOSITIONS WITH HIGH ACTIVITY CELLULASE**

4,738,682	4/1988	Boegh et al.	8/401
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FOREIGN PATENT DOCUMENTS

0220016	4/1987	European Pat. Off. .
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8909259	10/1989	WIPO .
WO9105841	5/1991	WIPO .
9117243	11/1991	WIPO .

[73] Assignee: **The Procter & Gamble Company**, Cincinnati, Ohio

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[30] Foreign Application Priority Data

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[51] **Int. Cl.⁶** **C11D 3/386**

[52] **U.S. Cl.** **252/174.12; 252/DIG. 12; 252/173; 252/90; 435/209; 435/183; 435/264**

[58] **Field of Search** **252/174.12, DIG. 12, 252/173, 90; 435/209, 183, 264**

[56] References Cited

U.S. PATENT DOCUMENTS

4,435,307 3/1984 Barbesgaard 252/174.12

[57] ABSTRACT

The present invention concerns cellulase-containing granular detergent compositions which are in a "compact" form, i.e. they are of a relatively high density and contain a relatively low amount of inorganic filler salt compared to conventional detergent compositions. In the detergent compositions herein the cellulase is defined by the C14CMC method described herein and preferably comprises a specific single-component endoglucanase.

21 Claims, No Drawings

COMPACT DETERGENT COMPOSITIONS WITH HIGH ACTIVITY CELLULASE

TECHNICAL FIELD

The present invention concerns cellulase-containing granular detergent compositions which are in a "compact" form, i.e. they are of a relatively high density and contain a relatively low amount of inorganic filler salt, compared to conventional detergent compositions. In the detergent compositions herein the cellulase comprises a cellulase of high activity defined by the C14CMC method described herein. Preferably the cellulase is a specific single-component endoglucanase.

BACKGROUND OF THE INVENTION

The need for detergent compositions which exhibit not only good cleaning properties, but also good fabric-softening performance, and other fabric care benefits, is well-established in the art.

The efficiency of cellulolytic enzymes, i.e. cellulases, in terms of textile cleaning and harshness-reducing agent for fabrics, has been recognized for some time; GB-A-2,075,028, GB-A-2,095,275 and GB-A-2,094,826, disclose detergent compositions with cellulase for improved cleaning performance; GB-A-1,368,599 discloses the use of cellulase for reducing the harshness of cotton-containing fabrics; U.S. Pat. No. 4,435,307 teaches the use of a cellulolytic enzyme derived from *Humicola insolens* as well as a fraction thereof, designated ACXI, as a harshness-reducing detergent additive.

EP-A-0 269 168 discloses optimized detergent compositions containing cellulase, which are formulated at a mild alkaline pH range and provide combined fabric cleaning, fabric softening, and fabric care performance.

In WO 89109259 have been disclosed cellulase preparations useful for reducing the harshness of cotton-containing fabrics, comprising an endoglucanase component with a high endoase activity and affinity towards cellulose.

The practical exploitation of cellulases has however, been set back by the fact that cellulase preparations such as those disclosed in the above-mentioned prior art documents, are complex mixtures, of which only a certain fraction is effective in the fabric-care context; it was thus difficult to implement cost effective industrial production of cellulase for the detergent industry; and large quantities of such cellulase preparations would need to be applied, in order to obtain the desired effect on fabrics.

Improvements in cellulase production also often have not proven to be sufficiently identifiable in terms of applicability in detergents. Defining a cellulase selection criterium relevant for detergent application of cellulase was made possible by the C14CMC-method disclosed in EP-A-350 098. A minimum of 10% removal of immobilized radioactive labelled carboxymethylcellulose has been found to provide high activity cellulase. A preferred group of cellulase falling under the high activity definition according to the present invention has been disclosed in copending Danish Patent Application No.: 1159/90 filed May 5, 1990. There is disclosed a cellulase preparation consisting essentially of a homogeneous endoglucanase component which is immunoreactive with a monoclonal antibody raised against a partially purified 43kD cellulase derived from *Humicola insolens* DM1800.

The finding that this particular endoglucanase component of cellulase is advantageous for the treatment of cellulose-containing materials now permits to produce the cellulase cost-effectively, e.g. by employing recombinant DNA techniques, and allows to apply only a small quantity of the cellulase preparation, and obtain the desired effect on fabrics.

On the other hand, a new generation of detergent compositions is now being marketed, which can be best pictured as "compact detergents" although they have been given a variety of trade names such as "Ultra", "Supra", "Micro". . . The particularity of such detergent compositions is their relatively high density compared to conventional detergent compositions, and their ability to achieve the same efficiency than conventional detergent compositions by using a considerably lesser amount of "compact" detergent composition. This particularity is best reflected, in terms of composition, by a relatively low amount of inorganic filler salt. The efficiency of such "compact" detergent compositions is best achieved by eliminating the pre-wash cycle and by using dispersing and diffusing devices, which are put directly in the drum of the washing machine at the start of the main washing cycle.

It is an object of the present invention to provide detergent compositions in a compact form, having a relatively high density and containing a low amount of inorganic filler salt, which exhibit optimum cellulase efficiency.

In EP-A-381 397 has been disclosed the effect of low ionic-strength on enzyme performance, in particular lipase.

It has been surprisingly found however, that the effect of the compact matrix on the selected enzymes of the present invention is much higher than what could be expected from state of the art cellulases such as disclosed in EP-A-381 397.

It is another object of the present invention to provide a method for treating fabrics in a washing machine, comprising the utilization of the present detergent compositions at low levels, for the main wash cycle.

SUMMARY OF THE INVENTION

The present invention relates to granular detergent compositions containing a surface-active agent, a builder, an enzyme, and if desired conventional additives, characterized in that the enzyme comprises a cellulase preparation providing at least 10% removal of immobilized radioactive labelled carboxymethylcellulose according to the C14CMC-method, at 25×10^{-6} % by weight of cellulase protein in the laundry test solution.

Preferably, the cellulase compound consists essentially of a homogeneous endoglucanase component which is immunoreactive with a monoclonal antibody raised against a partially purified about ≈ 43 kD cellulase derived from *Humicola insolens*, DSM 1800, or which is homologous to said ≈ 43 kD endoglucanase.

DETAILED DESCRIPTION OF THE INVENTION

The present detergent compositions are in granular form and are characterized by their density, which is higher than the density of conventional detergent compositions. The density of the compositions herein ranges from 550 to 950 g/liter, preferably 650 to 850 g/liter of composition, measured at 20° C.

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The "compact" form of the compositions herein is best reflected, in terms of composition, by the amount of inorganic filler salt; inorganic filler salts are conventional ingredients of detergent compositions in powder form; In conventional detergent compositions, the filler salts are present in substantial amounts, typically 17-35% by weight of the total composition.

In the present compositions, the filler salt is present in amounts not exceeding 15% of the total composition, preferably no exceeding 10%, most preferably not exceeding 5% by weight of the composition.

Inorganic filler salts, such as meant in the present compositions are selected from the alkali and alkaline-earth-metal salts of sulphates and chlorides.

A preferred filler salt is sodium sulphate.

SURFACTANT

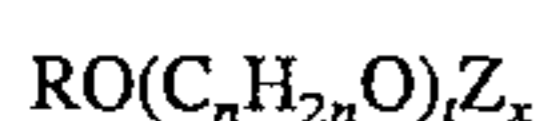
A wide range of surfactants can be used in the detergent compositions. A typical listing of anionic, nonionic, ampholytic and zwitterionic classes, and species of these surfactants, is given in U.S. Pat. No. 3,664,961 issued to Norris on May 23, 1972.

Mixtures of anionic surfactants are particularly suitable herein, especially mixtures of sulphonate and sulphate surfactants in a weight ratio of from 5:1 to 1:2, preferably from 3:1 to 2:3, more preferably from 3:1 to 1:1. Preferred sulphonates include alkyl benzene sulphonates having from 9 to 15, especially 11 to 13 carbon atoms in the alkyl radical, and alphasulphonated methyl fatty acid esters in which the fatty acid is derived from a C₁₂-C₁₈ fatty source preferably from a C₁₆-C₁₈ fatty source. In each instance the cation is an alkali metal, preferably sodium. Preferred sulphate surfactants are alkyl sulphates having from 12 to 18 carbon atoms in the alkyl radical, optionally in admixture with ethoxy sulphates having from 10 to 20, preferably 10 to 16 carbon atoms in the alkyl radical and an average degree of ethoxylation of 1 to 6. Examples of preferred alkyl sulphates herein are tallow alkyl sulphate, coconut alkyl sulphate, and C₁₄₋₁₅ alkyl sulphates. The cation in each instance is again an alkali metal cation, preferably sodium.

One class of nonionic surfactants useful in the present invention are condensates of ethylene oxide with a hydrophobic moiety to provide a surfactant having an average hydrophilic-lipophilic balance (HLB) in the range from 8 to 17, preferably from 9.5 to 13.5, more preferably from 10 to 12.5. The hydrophobic (lipophilic) moiety may be aliphatic or aromatic in nature and the length of the polyoxyethylene group which is condensed with any particular hydrophobic group can be readily adjusted to yield a water-soluble compound having the desired degree of balance between hydrophilic and hydrophobic elements.

Especially preferred nonionic surfactants of this type are the C₉-C₁₅ primary alcohol ethoxylates containing 3-8 moles of ethylene oxide per mole of alcohol, particularly the C₁₄-C₁₅ primary alcohols containing 6-8 moles of ethylene oxide per mole of alcohol and the C₁₂-C₁₄ primary alcohols containing 3-5 moles of ethylene oxide per mole of alcohol.

Another class of nonionic surfactants comprises alkyl polyglucoside compounds of general formula

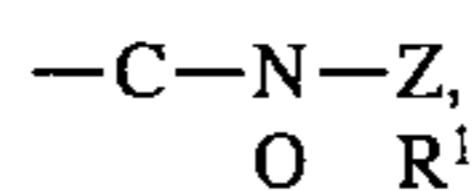


wherein Z is a moiety derived from glucose; R is saturated hydrophobic alkyl group that contains from 12 to 18 carbon atoms; t is from 0 to 10 and n is 2 or 3; x is from 1.3 to 4, the compounds including less than 10% unreacted fatty alcohol and less than 50% short chain alkyl polyglucosides. Compounds of this type and their use in detergent are

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disclosed in EP-B 0 070 077, 0 075 996 and 0 094 118.

Also suitable as nonionic surfactants are poly hydroxy fatty acid amide surfactants of the formula R²



wherein R¹ is H, C₁₋₄ hydrocarbyl, 2-hydroxy ethyl, 2-hydroxy propyl or a mixture thereof, R₂ is C₅₋₃₁ hydrocarbyl, an Z is a polyhydroxyhydrocarbyl having a linear hydrocarbyl chain with at least 3 hydroxyls directly connected to the chain, or an alkoxyated derivative thereof. Preferably, R₁ is methyl, R₂ is a straight, C₁₁₋₁₅ alkyl or alkenyl chain such as coconut alkyl or mixtures thereof, and Z is derived from a reducing sugar such as glucose, fructose, maltose, lactose, in a reductive amination reaction.

A further class of surfactants are the semi-polar surfactants such as amine oxides. Suitable amine oxides are selected from mono C₈-C₂₀, preferably C₁₀-C₁₄ N-alkyl or alkenyl amine oxides and propylene-1,3-diamine dioxides wherein the remaining N positions are substituted by methyl, hydroxyethyl or hydroxypropyl groups.

Another class of surfactants are amphoteric surfactants, such as polyamine-based species.

Cationic surfactants can also be used in the detergent compositions herein and suitable quaternary ammonium surfactants are selected from mono C₈-C₁₆, preferably C₁₀-C₁₄ N-alkyl or alkenyl ammonium surfactants wherein remaining N positions are substituted by methyl, hydroxyethyl or hydroxypropyl groups.

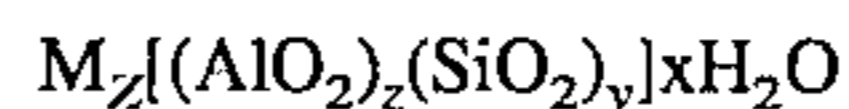
Mixtures of surfactant types are preferred, more especially anionic-nonionic and also anionic-nonionic-cationic mixtures. Particularly preferred mixtures are described in British Patent No. 2040987 and European Published Application No. 0 087 914. The detergent compositions can comprise from 1%-70% by weight of surfactant, but usually the surfactant is present in the compositions herein an amount of from 1% to 30%, more preferably from 10-25% by weight.

BUILDER

Builder materials will typically be present at from 10% to 60% of the detergent compositions herein. The compositions herein are free or substantially free of phosphate-containing builders (substantially free being herein defined to constitute less than 1% of the total detergent builder system), and the builder system herein consists of water-soluble builders, water-insoluble builders, or mixtures thereof.

Water insoluble builders can be an inorganic ion exchange material, commonly an inorganic hydrated aluminosilicate material, more particularly a hydrated synthetic zeolite such as hydrated Zeolite A, X, B or HS.

Preferred aluminosilicate ion-exchange materials have the unit cell formula

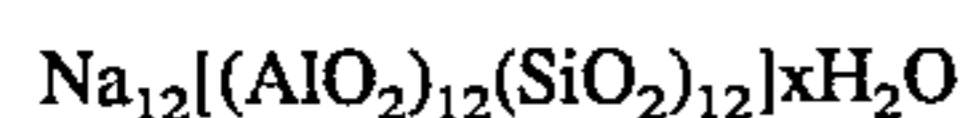


wherein M is a calcium-exchange cation, z and y are at least 6; the molar ratio of z to y is from 1.0 to 0.5 and x is at least 5, preferably from 7.5 to 276, more preferably from 10 to 264. The aluminosilicate materials are in hydrated form and are preferably crystalline containing from 10% to 28%, more preferably from 18% to 22% water.

The above aluminosilicate ion exchange materials are further characterized by a particle size diameter of from 0.1 to 10 micrometers, preferably from 0.2 to 4 micrometers. The term "particle size diameter" herein represents the average particle size diameter of a given ion exchange material as determined by conventional analytical tech-

niques such as, for example, microscopic determination utilizing a scanning electron microscope. The aluminosilicate ion exchange materials are further characterized by their calcium ion exchange capacity, which is at least 200 mg equivalent of CaCO_3 water hardness/g of aluminosilicate, calculated on an anhydrous basis, and which generally is in the range of from 300 mg eq./g to 352 mg eq./g. The aluminosilicate ion exchange materials herein are still further characterized by their calcium ion exchange rate which is described in detail in GB-1,429,143.

Aluminosilicate ion exchange materials useful in the practice of this invention are commercially available and can be naturally occurring materials, but are preferably synthetically derived. A method for producing aluminosilicate ion exchange materials is discussed in U.S. Pat. No. 3,985,669. Preferred synthetic crystalline aluminosilicate ion exchange materials useful herein are available under the designation Zeolite A, Zeolite B, Zeolite X, Zeolite HS and mixtures thereof. In an especially preferred embodiment, the crystalline aluminosilicate ion exchange material is Zeolite A and has the formula



wherein x is from 20 to 30, especially 27. Zeolite X of formula $\text{Na}_{86}[(\text{AlO}_2)_{86}(\text{SiO}_2)_{106}]-10.276\text{H}_2\text{O}$ is also suitable, as well as Zeolite HS of formula $\text{Na}_6[(\text{AlO}_2)_6(\text{SiO}_2)_6]7.5\text{H}_2\text{O}$.

Another suitable water-insoluble, inorganic builder material is layered silicate, e.g. SKS-6 (Hoechst). SKS-6 is a crystalline layered silicate consisting of sodium silicate ($\text{Na}_2\text{Si}_2\text{O}_5$). The high $\text{Ca}^{++}/\text{Mg}^{++}$ binding capacity is mainly a cation exchange mechanism. In hot water, the material becomes more soluble.

The water-soluble builder can be a monomeric or oligomeric carboxylate chelating agent.

Suitable carboxylates containing one carboxy group include lactic acid, glycolic acid and ether derivatives thereof as disclosed in Belgian Patent Nos. 831,368, 821,369 and 821,370. Polycarboxylates containing two carboxy groups include the water-soluble salts of succinic acid, malonic acid, (ethylenedioxy) diacetic acid, maleic acid, diglycolic acid, tartaric acid, tartronic acid and fumaric acid, as well as the ether carboxylates described in German Offenlegenschrift 2,446,686, and 2,446,687 and U.S. Pat. No. 3,935,257 and the sulfinyl carboxylates described in Belgian Patent No. 840,623. Polycarboxylates containing three carboxy groups include, in particular, water-soluble citrates, aconitrates and citraconates as well as succinate derivatives such as the carboxymethyloxysuccinates described in British Patent No. 1,379,241, lactoxysuccinates described in Netherlands Application 7205873, and the oxypolycarboxylate materials such as 2-oxa-1,1,3-propane tricarboxylates described in British Patent No. 1,387,447.

Polycarboxylates containing four carboxy groups include oxydisuccinates disclosed in British Patent No. 1,261,829, 1,1,2,2-ethane tetracarboxylates, 1,1,3,3-propane tetracarboxylates and 1,1,2,3-propane tetracarboxylates. Polycarboxylates containing sulfo substituents include the sulfosuccinate derivatives disclosed in British Patent Nos. 1,398,421 and 1,398,422 and in U.S. Pat. No. 3,936,448, and the sulfonated pyrolysed citrates described in British Patent No. 1,082,179, while polycarboxylates containing phosphone substituents are disclosed in British Patent No. 1,439,000.

Alicyclic and heterocyclic polycarboxylates include cyclopentane-cis,cis,cis-tetracarboxylates, cyclopentadienide pentacarboxylates, 2,3,4,5-tetrahydrofuran-cis, cis, cis-tetra-

carboxylates, 2,5-tetrahydrofuran-cis-dicarboxylates, 2,2,5,5-tetrahydrofuran-tetracarboxylates, 1,2,3,4,5,6-hexanehexacarboxylates and and carboxymethyl derivatives of polyhydric alcohols such as sorbitol, mannitol and xylitol. Aromatic polycarboxylates include mellitic acid, pyromellitic acid and the phthalic acid derivatives disclosed in British Patent No. 1,425,343.

Of the above, the preferred polycarboxylates are hydroxycarboxylates containing up to three carboxy groups per molecule, more particularly citrates.

Preferred builder systems for use in the present compositions include a mixture of a water-insoluble aluminosilicate builder such as zeolite A, and a water-soluble carboxylate chelating agent such as citric acid.

Other builder materials that can form part of the builder system for the purposes of the invention include inorganic materials such as alkali metal carbonates, bicarbonates, silicates, and organic materials such as the organic phosphonates, amino polyalkylene phosphonates and amino polycarboxylates.

Other suitable water-soluble organic salts are the homo- or co-polymeric acids or their salts, in which the polycarboxylic acid comprises at least two carboxyl radicals separated from each other by not more than two carbon atoms.

Polymers of this type are disclosed in GB-A-1,596,756. Examples of such salts are polyacrylates of MW 2000-5000 and their copolymers with maleic anhydride, such copolymers having a molecular weight of from 20,000 to 70,000, especially about 40,000.

CELLULASE

The activity of enzymes and particularly the activity of cellulase enzyme has been defined for various applications by different analytical methods. These methods all attempt to provide a realistic assessment of the expected use performance or at least a measurement correlating with the in use performance. As has been detailed in European Patent Application EP-A-350098, many of the methods, particularly these frequently used by cellulase manufacturers, are not sufficiently correlated with the in use performance of cellulase in laundry detergent compositions. This is due to the various other usage conditions for which these activity measurement methods have been developed.

The method described in EP-A-350098, has been developed to be and to have a predictive correlation for the ranking of cellulase activity in laundry detergent compositions.

The present invention therefore uses the method disclosed in EP-A-350098 to screen cellulases in order to distinguish cellulases which are useful in the present invention and those which would not provide the objectives of the present invention. The screening method, hereinafter referred to as C14CMC-Method, which has been adopted from the method disclosed in EP-A-350098, can be described as follows:

Principle:

The principle of the C14CMC-Method for screening is to measure at a defined cellulase concentration in a wash solution the removal of immobilized carboxy methyl cellulose (CMC) from a cloth substrate. The removal of CMC is measured by radio-active labelling of some of the CMC by using C14 radio-active carbon. Simple counting of the amount of radio-active C14 on the cloth substrate before and after the cellulase treatment allows the evaluation of the cellulase activity.

Sample preparation:

CMC preparation:

The radio-active CMC stock solution is prepared according to Table I. The radio-active CMC can be obtained by methods referred to in EP-A-350098.

Fabric substrates:

The fabric substrates are muslin cotton swatches having a size of 5 cm×5 cm. They are inoculated with 0.35 ml of the radio-active labelled CMC stock solution in their center. The muslin cotton swatches are then airdried.

Immobilization of CMC:

To immobilize the radio-active labelled CMC on the muslin cotton swatches, laundero-meter equipment "Linitest Original Haunau" made by Original Haunau, Germany, is used. A metal jar of the laundero-meter is filled with 400 ml of hard water (4 mmol/liter of Ca⁺⁺ ions). A maximum number of 13 swatches can be used per jar. The jar is then incubated in a heat-up cycle from 20° C. to 60° C. over 40 minutes in the laundero-meter equipment. After incubation the swatches are rinsed under running city water for 1 minute. They are squeezed and allowed to airdry for at least 30 minutes.

According to EP-A-350098 samples of the swatches with immobilized radio-active CMC can also be measured as "blank samples" without washing.

Sample treatment:

Laundry test solution:

The laundry test solution is prepared according to the composition of Table II. It is balanced to pH 7.5. The laundry test solution is the basis to which a cellulase test sample is added. Care should be taken to not dilute the laundry test solution by adding water to a 100% balance prior to having determined the amount of cellulase to be added. The amount of cellulase which is used in this screening test should be added to provide 25×10⁻⁶ weight percent of cellulase protein in the laundry test solution (equivalent to 0.25 milligram/liter at 14.5° C.).

Wash procedure:

The swatches thus inoculated with radio-active labelled CMC are then treated in a laundry simulation process. The laundry process is simulated in the laundero-meter type equipment, "Linitest, Original Haunau", by Original Haunau, Haunau Germany. An individual swatch is put into a 20 cm³ glass vial. The vial is filled with 10 ml of the laundry test solution and then sealed liquid tight. Up to 5 vials are put into each laundero-meter jar. The jar is filled with water as a heat transfer medium for the laundering simulation. The laundering simulation is conducted as a heat-up cycle from 20° C. to 60° C. over 40 minutes.

After the processing of the samples the vials are submerged in cold water and subsequently each swatch is taken out of its vial, rinsed in a beaker under running soft water, squeezed and allowed to airdry for at least 30 minutes.

Measurement:

In order to measure radio-active labelled CMC removal, a scintillation counter, for example, a LKB 1210 Ultrabeta Scintillation Counter, is used. In order to obtain most accurate results, the instruction manual for optimum operation of the particular scintillation counter should be followed. For example, for the LKB 1210 Ultrabeta Scintillation Counter, the following procedure should be followed. The swatch to be measured is put into a plastic vial filled with 12 ml of scintillator liquid (e.g. scintillator 299 from Packard). The swatch is then allowed to stabilize for at least 30 minutes. The vial is then put into the LKB 1210 Ultrabeta Scintillation Counter and the respective radio-activity counts for the swatch is obtained.

In order to measure the amount of CMC removal due only to the cellulase, a measurement of a swatch which has been inoculated at the same time but has been treated in the laundry test solution without cellulase, is necessary. The activity of the cellulase is then expressed as percent of

radio-active labelled CMC removal. This percentage is calculated by the following formula:

$$\% \text{ of radio-active CMC removal} = \frac{XO - XC}{XO} \times 100$$

Wherein

XO is the radioactivity scintillation count of a swatch treated with the laundry test solution without cellulase

XC is the radioactivity scintillation count of a swatch treated with the laundry test solution containing the cellulase to be evaluated

Statistical considerations, procedure confirmation:

In order to provide statistically sound results, standard statistical analysis should be employed. For the given example, using the LKB 1210 Ultrabeta Scintillation Counter, it has been found that a sample size of 3 swatches for each radioactivity scintillation count can be used.

In order to confirm the procedure by internal crosschecking, measurement and calculation of the "blank sample" according to EP-A-350098 are recommended. This will allow to detect and eliminate errors.

Interpretation of results:

The described screening test does provide a fast, unique and reliable method to identify cellulases which satisfy the activity criteria of the present invention versus cellulases which are not part of the present invention.

It has been found that a removal of 10% or more of the immobilized radioactive labelled CMC according to the above C14CMC-method, indicates that the respective cellulase satisfies the requirements of the invention.

It will be obvious to those skilled in the art that removal percentages above 10% indicate a higher activity for the respective cellulase. It therefore is contemplated that cellulase providing above 25% or preferably above 50% removal of radioactive labelled CMC, at the protein concentration in the laundry test solution according to the C14CMC-method, would provide indication of an even better performance of the cellulase for use in laundry detergents.

It also has been contemplated that usage of higher concentrations of cellulase for C14CMC-method, would provide higher removal percentages. However, there exists no linear proven correlation between cellulase concentration and removal percentage obtained by it.

It also has been contemplated that usage of higher concentrations of cellulase for C14CMC-method, would provide higher removal percentages.

TABLE I

Radioactive C ₁₄ labelled CMC stock solution (all percentages by weight of total solution)	
Total CMC* (CMC should be detergent grade CMC with a degree of substitution from about 0.47 to about 0.7)	99.2 × 10 ⁻³ %
Ethanol	14985.12 × 10 ⁻³ %
Deionized Water	84915.68 × 10 ⁻³ %
Total:	100%

*Total CMC contains non-radio-active and radio-active CMC to provide a radio-activity which allows sufficiently clear readings on the scintillation counter used. For example, the radio-active CMC can have an activity of 0.7 millicurie/g and be mixed

TABLE II

Laundry test solution (all percentages by weight of total solution)	
Linear C ₁₂ alkyl benzene sulphonic acid	0.110%
Coconut alkyl sulphate (TEA salt)	0.040%
C ₁₂₋₁₅ alcohol ethoxylate (E07)	0.100%
Coconut fatty acid	0.100%
Oleic acid	0.050%
Citric acid	0.010%
Triethanolamine	0.040%
Ethanol	0.060%
Propanediol	0.015%
Sodium hydroxide	0.030%
Sodium formate	0.010%
Protease	0.006%
Water (2.5 mmol/liter Ca ⁺⁺), pH adjustment agent (HCL or NaOH solutions) and cellulase	balance to 100%

According to the present invention, preferred cellulase are those as described in Danish Patent Application 1159/90. For example, a cellulase preparation useful in the compositions of the invention can consist essentially of a homogeneous endoglucanase component, which is immunoreactive with an antibody raised against a highly purified 43kD cellulase derived from *Humicola insolens*, DSM 1800, or which is homologous to said 43kD endoglucanase.

It should be stressed that all cellulase enzymes according to the present invention have to meet the criteria of the above mentioned screening test. However, in the Danish Patent Application 1159/90 additional criteria are established allowing to identify preferred cellulase enzymes in combination with the present screening test.

Cellulase preparations particularly useful in the compositions of the invention are those in which in addition to the screening test, the endoglucanase component exhibits a CMC-endoase activity of at least about 50, preferably at least about 60, in particular at least about 90 CMC-endoase units per mg of total protein. In particular, a preferred endoglucanase component exhibits a CMC-endoase activity of at least 100 CMC-endoase units per mg of total protein.

In the present context, the term "CMC-endoase activity" refers to the endoglucanase activity of the endoglucanase component in terms of its ability to degrade cellulose to glucose, cellobiose and triose, as determined by a viscosity decrease of a solution of carboxymethyl cellulose (CMC) after incubation with the cellulase preparation of the invention, as described in detail below.

The CMC-endoase (endoglucanase) activity can be determined from the viscosity decrease of CMC, as follows: A substrate solution is prepared, containing 35 g/l CMC (Hercules 7 LFD) in 0.1M tris buffer at pH 9.0. The enzyme sample to be analyzed is dissolved in the same buffer. 10 ml substrate solution and 0.5 ml enzyme solution are mixed and transferred to a viscosimeter (e.g. Haake VT 181, NV sensor, 181 rpm), thermostated at 40° C. Viscosity readings are taken as soon as possible after mixing and again 30 minutes later. The amount of enzyme that reduces the viscosity to one half under these conditions is defined as 1 unit of CMC-endoase activity.

SDS polyacrylamide gel electrophoresis (SDS-PAGE) and isoelectric focusing with marker proteins in a manner known to persons skilled in the art were used to determine the molecular weight and isoelectric point (pI), respectively, of the endoglucanase component in the cellulase preparation

useful in the present context. In this way, the molecular weight of a specific endoglucanase component was determined to be 43kD. The isoelectric point of this endoglucanase was determined to be about 5.1.

The cellobiohydrolase activity may be defined as the activity towards cellobiose p-nitrophenyl. The activity is determined as umole nitrophenyl released per minute at 37° C. and pH 7.0. The present endoglucanase component was found to have essentially no cellobiohydrolase activity.

The endoglucanase component in the cellulase preparation herein has initially been isolated by extensive purification procedures, i.a. involving reverse phase HPLC purification of a crude *H. insolens* cellulase mixture according to U.S. Pat. No. 4,435,307. This procedure has surprisingly resulted in the isolation of a 43kD endoglucanase as a single component with unexpectedly favourable properties due to a surprisingly high endoglucanase activity.

Also, in addition to the screening test, the cellulase enzymes useful in the present compositions can further be defined as enzymes exhibiting endoglucanase activity (in the following referred to as an "endoglucanase enzyme"), which enzymes have the amino acid sequence shown in the appended Sequence Listing ID#2, or a homologue thereof exhibiting endoglucanase activity.

In the present context, the term "homologue" is intended to indicate a polypeptide encoded by DNA which hybridizes to the same probe as the DNA coding for the endoglucanase enzyme with this amino acid sequence under certain specified conditions (such as presoaking in 5×SSC and prehybridizing for 1 h at 40° C. in a solution of 20% formamide, 5×Denhardt's solution, 50 mM sodium phosphate, pH 6.8, and 50 ug of denatured sonicated calf thymus DNA, followed by hybridization in the same solution supplemented with 100 uM ATP for 18 h at 40° C.). The term is intended to include derivatives of the aforementioned sequence obtained by addition of one or more amino acid residues to either or both the C- and N-terminal of the native sequence, substitution of one or more amino acid residues at one or more sites in the native sequence, deletion of one or more amino acid residues at either or both ends of the native amino acid sequence or at one or more sites within the native sequence, or insertion of one or more amino acid residues at one or more sites in the native sequence.

The endoglucanase enzyme herein may be one producible by species of *Humicola* such as *Humicola insolens* e.g. strain DSM 1800, deposited on Oct. 1, 1981 at the Deutsche Sammlung von Mikroorganismen, Mascheroder Weg 1B, D-3300 Braunschweig, FRG, in accordance with the provisions of the Budapest Treaty on the International Recognition of the Deposit of Microorganisms for the Purposes of Patent Procedure (the Budapest Treaty).

In still a further aspect, the cellulase enzymes useful herein can be defined, in addition to the screening test, as endoglucanase enzymes which have the amino acid sequence shown in the appended Sequence Listing ID#4, or a homologue thereof (as defined above) exhibiting endoglucanase activity. Said endoglucanase enzyme may be one producible by a species of *Fusarium*, such as *Fusarium oxysporum*, e.g. strain DSM 2672, deposited on Jun. 6, 1983 at the Deutsche Sammlung von Mikroorganismen, Mascheroder Weg 1B, D-3300 Braunschweig, FRG, in accordance with the provisions of the Budapest Treaty.

Furthermore, it is contemplated that homologous endoglucanases may be derived from other microorganisms producing cellulolytic enzymes, e.g. species of *Trichoderma*, *Myceliophthora*, *Phanerochaete*, *Schizophyllum*, *Penicillium*, *Aspergillus*, and *Geotricum*.

For industrial production of the cellulase preparation herein, however, it is preferred to employ recombinant DNA techniques or other techniques involving adjustments of fermentations or mutation of the microorganisms involved to ensure overproduction of the desired enzymatic activities. Such methods and techniques are known in the art and may readily be carried out by persons skilled in the art.

The endoglucanase component may thus be one which is producible by a method comprising cultivating a host cell transformed with a recombinant DNA vector which carries a DNA sequence encoding said endoglucanase component or a precursor of said endoglucanase component as well as DNA sequences encoding functions permitting the expression of the DNA sequence encoding the endoglucanase component or precursor thereof, in a culture medium under conditions permitting the expression of the endoglucanase component or precursor thereof and recovering the endoglucanase component from the culture.

DNA constructs comprising a DNA sequence encoding an endoglucanase enzyme as described above, or a precursor form of the enzyme, include the DNA constructs having a DNA sequence as shown in the appended Sequence Listings ID#1 or ID#3, or a modification thereof. Examples of suitable modifications of the DNA sequence are nucleotide substitutions which do not give rise to another amino acid sequence of the endoglucanase, but which correspond to the codon usage of the host organism into which the DNA construct is introduced or nucleotide substitutions which do give rise to a different amino acid sequence and therefore, possibly, a different protein structure which might give rise to an endoglucanase mutant with different properties than the native enzyme. Other examples of possible modifications are insertion of one or more nucleotides at either end of the sequence, or deletion of one or more nucleotides at either end or within the sequence.

DNA constructs encoding endoglucanase enzymes useful herein may be prepared synthetically by established standard methods, e.g. the phosphoamidite method described by S. L. Beaucage and M. H. Caruthers, *Tetrahedron Letters* 22, 1981, pp. 1859-1869, or the method described by Matthes et al., *EMBO Journal* 3, 1984, pp. 801-805. According to the phosphoamidite method, oligonucleotides are synthesized, e.g. in an automatic DNA synthesizer, purified, annealed, ligated and cloned in suitable vectors.

A DNA construct encoding the endoglucanase enzyme or a precursor thereof may, for instance, be isolated by establishing a cDNA or genomic library of a cellulase-producing microorganism, such as *Humicola insolens*, DSM 1800, and screening for positive clones by conventional procedures such as by hybridization using oligonucleotide probes synthesized on the basis of the full or partial amino acid sequence of the endoglucanase in accordance with standard techniques (cf. Sambrook et al., *Molecular Cloning: A Laboratory Manual*, 2nd. Ed. Cold Spring Harbor, 1989), or by selecting for clones expressing the appropriate enzyme activity (i.e. CMC-endoase activity as defined above), or by selecting for clones producing a protein which is reactive with an antibody against a native cellulase (endoglucanase).

Finally, the DNA construct may be of mixed synthetic and genomic, mixed synthetic and cDNA or mixed genomic and cDNA origin prepared by ligating fragments of synthetic, genomic or cDNA origin (as appropriate), the fragments corresponding to various parts of the entire DNA construct, in accordance with standard techniques. The DNA construct may also be prepared by polymerase chain reaction using specific primers, for instance as described in U.S. Pat. No. 4,683,202 or R. K. Saiki et al., *Science* 239, 1988, pp. 487-491.

Recombinant expression vectors into which the above DNA constructs are inserted include any vector which may conveniently be subjected to recombinant DNA procedures, and the choice of vector will often depend on the host cell into which it is to be introduced. Thus, the vector may be an autonomously replicating vector, i.e. a vector which exists as an extrachromosomal entity, the replication of which is independent of chromosomal replication, e.g. a plasmid. Alternatively, the vector may be one which, when introduced into a host cell, is integrated into the host cell genome and replicated together with the chromosome(s) into which it has been integrated.

In the vector, the DNA sequence encoding the endoglucanase should be operably connected to a suitable promoter and terminator sequence. The promoter may be any DNA sequence which shows transcriptional activity in the host cell of choice and may be derived from genes encoding proteins either homologous or heterologous to the host cell. The procedures used to ligate the DNA sequences coding for the endoglucanase, the promoter and the terminator, respectively, and to insert them into suitable vectors are well known to persons skilled in the art (cf., for instance, Sambrook et al., op.cit.).

Host cells which are transformed with the above DNA constructs or the above expression vectors may be for instance belong to a species of *Aspergillus*, most preferably *Aspergillus oryzae* or *Aspergillus niger*. Fungal cells may be transformed by a process involving protoplast formation and transformation of the protoplasts followed by regeneration of the cell wall in a manner known per se. The use of *Aspergillus* as a host microorganism is described in EP 238 023 (of Novo Industri A/S), the contents of which are hereby incorporated by reference. The host cell may also be a yeast cell, e.g. a strain of *Saccharomyces cerevisiae*.

Alternatively, the host organism may be a bacterium, in particular strains of *Streptomyces* and *Bacillus*, and *E. coli*. The transformation of bacterial cells may be performed according to conventional methods, e.g. as described in Sambrook et al., *Molecular Cloning: A Laboratory Manual*, Cold Spring Harbor, 1989.

The screening of appropriate DNA sequences and construction of vectors may also be carried out by standard procedures, cf. Sambrook et al., op.cit.

The medium used to cultivate the transformed host cells may be any conventional medium suitable for growing the host cells in question. The expressed endoglucanase may conveniently be secreted into the culture medium and may be recovered therefrom by well-known procedures including separating the cells from the medium by centrifugation or filtration, precipitating proteinaceous components of the medium by means of a salt such as ammonium sulphate, followed by chromatographic procedures such as ion exchange chromatography, affinity chromatography, or the like.

By employing recombinant DNA techniques as indicated above, techniques of protein purification, techniques of fermentation and mutation or other techniques which are well known in the art, it is possible to provide endoglucanases of a high purity.

The level in the present composition of cellulase described above should be such that the amount of enzyme protein to be delivered in the wash solution is from 0.005 to 40 mg/liter of wash solution, preferably 0.01 to 10 mg/liter of wash solution.

OPTIONAL INGREDIENTS

The present compositions will typically include optional ingredients that normally form part of detergent composi-

tions Antiredeposition and soil suspension agents, optical brighteners, bleaches, bleach activators, suds suppressors, anticaking agents, dyes and pigments are examples of such optional ingredients and can be added in varying amounts as desired.

Antiredeposition and soil suspension agents suitable herein include cellulose derivatives such as methylcellulose, carboxymethylcellulose and hydroxyethylcellulose, and homo- or co-polymeric polycarboxylic acids or their salts. Polymers of this type include the polyacrylates and maleic anhydride-acrylic acid copolymers previously mentioned as builders, as well as copolymers of maleic anhydride with ethylene, methylvinyl ether or methacrylic acid, the maleic anhydride constituting at least 20 mole percent of the copolymer. These materials are normally used at levels of from 0.5% to 10% by weight, more preferably from 0.75% to 8%, most preferably from 1% to 6% by weight of the composition.

Preferred optical brighteners are anionic in character, examples of which are disodium 4,4'-bis-(2-diethanolamino-4-anilino-s-triazin-6-ylamino)stilbene-2,2'-disulphonate, disodium 4,4'-bis-(2-morpholino-4-anilino-s-triazin-6-ylamino)stilbene-2,2'-disulphonate, disodium 4,4'-bis-(2,4-dianilino-s-triazin-6-ylamino)stilbene-2,2'-disulphonate, monosodium 4,4'-bis-(2,4-dianilino-s-triazin-6-ylamino)stilbene-2-sulphonate, disodium 4,4'-bis-(2-anilino-4-(N-methyl-N-2-hydroxyethylamino)-s-triazin-6-ylamino)stilbene-2,2'-disulphonate, disodium 4,4'-bis-(4-phenyl-2,1,3-triazol-2-yl)stilbene-2,2'-disulphonate, disodium 4,4'-bis-(2-anilino-4-(1-methyl-2-hydroxyethylamino)-s-triazin-6-ylamino)stilbene-2,2'-disulphonate and sodium 2(stilbyl-4'-naphtho-1',2':4,5)-1,2,3-triazole-2'-sulphonate.

Any particulate inorganic perhydrate bleach can be used, in an amount of from 3% to 40% by weight, more preferably from 8% to 25% by weight and most preferably from 12% to 20% by weight of the compositions. Preferred examples of such bleaches are sodium perborate monohydrate and tetrahydrate, percarbonate, and mixtures thereof.

Another preferred separately mixed ingredient is a peroxy carboxylic acid bleach precursor, commonly referred to as a bleach activator, which is preferably added in a prilled or agglomerated form. Examples of suitable compounds of this type are disclosed in British Patent Nos. 1586769 and 2143231 and a method for their formation into a prilled form is described in European Published Patent Application No. 0 062 523. Preferred examples of such compounds are tetracetyl ethylene diamine and sodium 3, 5, 5 trimethyl hexanoyloxybenzene sulphonate.

Bleach activators are normally employed at levels of from 0.5% to 10% by weight, more frequently from 1% to 8% and preferably from 2% to 6% by weight of the composition.

Another optional ingredient is a suds suppressor, exemplified by silicones, and silica-silicone mixtures. Silicones can be generally represented by alkylated polysiloxane materials while silica is normally used in finely divided forms exemplified by silica aerogels and xerogels and hydrophobic silicas of various types. These materials can be incorporated as particulates in which the suds suppressor is advantageously releasably incorporated in a water-soluble or water-dispersible, substantially non-surface-active detergent impermeable carrier. Alternatively the suds suppressor can be dissolved or dispersed in a liquid carrier and applied by spraying on to one or more of the other components.

As mentioned above, useful silicone suds controlling agents can comprise a mixture of an alkylated siloxane, of the type referred to hereinbefore, and solid silica. Such

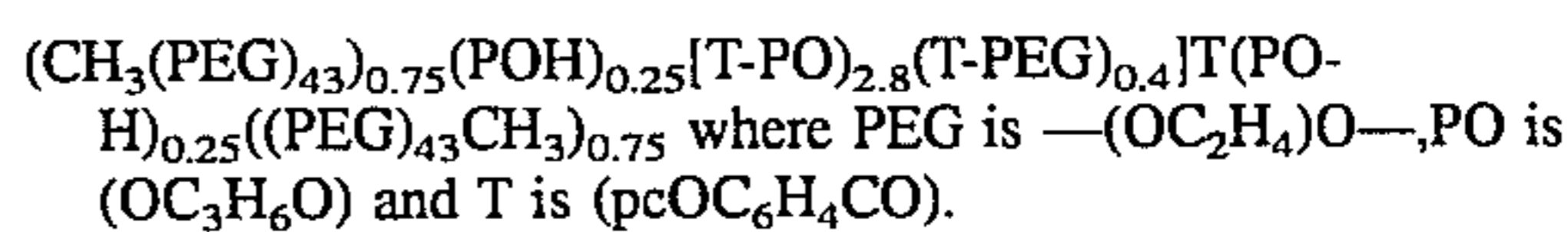
mixtures are prepared by affixing the silicone to the surface of the solid silica. A preferred silicone suds controlling agent is represented by a hydrophobic silanated (most preferably trimethyl-silanated) silica having a particle size in the range from 10 millimicrons to 20 millimicrons and a specific surface area above 50 m²/g intimately admixed with dimethyl silicone fluid having a molecular weight in the range from about 500 to about 200,000 at a weight ratio of silicone to silanated silica of from about 1:1 to about 1:2.

A preferred silicone suds controlling agent is disclosed in Bartollota et al. U.S. Pat. No. 3,933,672. Other particularly useful suds suppressors are the self-emulsifying silicone suds suppressors, described in German Patent Application DTOS 2,646,126 published Apr. 28, 1977. An example of such a compound is DC-544, commercially available from Dow Corning, which is a siloxane/glycol copolymer.

The suds suppressors described above are normally employed at levels of from 0.001% to 2% by weight of the composition, preferably from 0.01% to 1% by weight. The incorporation of the suds modifiers is preferably made as separate particulates, and this permits the inclusion therein of other suds controlling materials such as C20-C24 fatty acids, microcrystalline waxes and high MW copolymers of ethylene oxide and propylene oxide which would otherwise adversely affect the dispersibility of the matrix. Techniques for forming such suds modifying particulates are disclosed in the previously mentioned Bartollota et al U.S. Pat. No. 3,933,672.

Other useful polymeric materials are the polyethylene glycols, particularly those of molecular weight 1000-10000, more particularly 2000 to 8000 and most preferably about 4000. These are used at levels of from 0.20% to 5% more preferably from 0.25% to 2.5% by weight. These polymers and the previously mentioned homo- or co-polymeric polycarboxylate salts are valuable for improving whiteness maintenance, fabric ash deposition, and cleaning performance on clay, proteinaceous and oxidizable soils in the presence of transition metal impurities.

Soil release agents useful in compositions of the present invention are conventionally copolymers or terpolymers of terephthalic acid with ethylene glycol and/or propylene glycol units in various arrangements. Examples of such polymers are disclosed in the commonly assigned U.S. Pat. Nos. 4116885 and 4711730 and European Published Patent Application No. 0 272 033. A particular preferred polymer in accordance with EP-A-0 272 033 has the formula



Certain polymeric materials such as polyvinyl pyrrolidones typically of MW 5000-20000, preferably 10000-15000, also form useful agents in preventing the transfer of labile dyestuffs between fabrics during the washing process.

Fabric softening agents can also be incorporated into detergent compositions in accordance with the present invention. These agents may be inorganic or organic in type. Inorganic softening agents are exemplified by the smectite clays disclosed in GB-A-1,400,898. Organic fabric softening agents include the water-insoluble tertiary amines as disclosed in GB-A-1514276 and EP-B-0 011 340 and their combination with mono C₁₂-C₁₄ quaternary ammonium salts are disclosed in EP-B-0 026 527 and EP-B-0 026 528 and di-long-chain amides as disclosed in EP-B-0 242 919. Other useful organic ingredients of fabric softening systems

include high molecular weight polyethylene oxide materials as disclosed in EP-A-0 299 575 and 0 313 146.

Levels of smectite clay are normally in the range from 5% to 20%, more preferably from 8% to 15% by weight with the material being added as a dry mixed component to the remainder of the formulation. Organic fabric softening agents such as the water-insoluble tertiary amines or di-long-chain amide materials are incorporated at levels of from 0.5% to 5% by weight, normally from 1% to 3% by weight whilst the high molecular weight polyethylene oxide materials and the water-soluble cationic materials are added at levels of from 0.1% to 2%, normally from 0.15% to 1.5% by weight. These materials are normally added to the spray dried portion of the composition, although in some instances it may be more convenient to add them as a dry mixed particulate, or spray them as a molten liquid on to other solid components of the composition.

Enzymes other than the specific cellulase preparation herein can be present in the composition herein, such as proteases, lipases and amylases.

MAKING PROCESS

Compositions according to the present invention can be made via a variety of methods including dry mixing, spray drying, agglomeration and granulation and combinations of any of these techniques.

PREFERRED MAKING PROCESS

A preferred method of making the compositions herein involves a combination of spray drying, agglomeration in a high speed mixer and dry mixing.

A first granular component containing a relatively insoluble anionic surfactant is spray dried and part of the spray dried product is diverted and subjected to a low level of nonionic surfactant spray on before being reblended with the remainder. A second granular component is made by dry neutralisation of an anionic surfactant acid using sodium carbonate as the neutralising agent in a continuous high speed blender such as a Lodige KM mixer. The first and second components together with other dry mix ingredients such as the carboxylate chelating agent, inorganic peroxygen bleach, bleach activator, soil suspension agent, silicate and enzyme are then fed to a conveyor belt from which they are transferred to a horizontally rotating drum in which perfume and silicone suds suppressor are sprayed on to the product. In highly preferred compositions, a further drum mixing step is employed in which a low (approx. 2%) level of finely divided crystalline aluminosilicate is introduced to increase density and improve granular flow characteristics.

PROCESS OF WASHING

The compact detergent compositions herein have the ability to achieve the same efficiency than conventional detergent compositions, when a considerably lesser amount of composition herein, is used in the main wash cycle of a washing machine.

Accordingly, in an other embodiment of the invention, it is herewith provided for a process for washing fabrics in a washing machine wherein an amount of from 15 to 170 g of a detergent composition according to the present invention is used for the main wash cycle.

Typically, under European conditions, the recommended usage is from 80 to 140 g of detergent composition for the main wash cycle, without the need of a pre-wash.

The detergent compositions herein are preferably delivered directly to the drum and not indirectly via the outer casing of the machine. This can most easily be achieved by incorporation of the composition in a bag or container from which it can be released at the start of the wash cycle in response to agitation, a rise in temperature or immersion in

the wash water in the drum. Such a container will be placed in the drum, together with the fabrics to be washed. Alternatively the washing machine itself may be adapted to permit direct addition of the composition to the drum e.g. by a dispensing arrangement in the access door.

Products comprising a detergent composition enclosed in a bag or container are usually designed in such a way that container integrity is maintained in the dry state to prevent egress of the contents when dry, but are adapted for release of the container contents on exposure to a washing environment, normally on immersion in an aqueous solution.

Usually the container will be flexible, such as a bag or pouch. The bag may be of fibrous construction coated with a water impermeable protective material so as to retain the contents, such as is disclosed in European published Patent Application No. 0 018 678. Alternatively it may be formed of a water insoluble synthetic polymeric material provided with an edge seal or closure designed to rupture in aqueous media as disclosed in European published Patent Application Nos. 0 011 500, 0 011 501, 0 011 502, and 0 011 968. A convenient form of water frangible closure comprises a water soluble adhesive disposed along and sealing one edge of a pouch formed of a water impermeable polymeric film such as polyethylene or polypropylene.

In a variant of the bag or container product form, laminated sheet products can be employed in which a central flexible layer is impregnated and/or coated with a composition and then one or more outer layers are applied to produce a fabric-like aesthetic effect. The layers may be sealed together so as to remain attached during use or may separate on contact with water to facilitate the release of the coated or impregnated material.

An alternative laminate form comprises one layer embossed or deformed to provide a series of pouch-like containers into each of which the detergent components are deposited in measured amounts, with a second layer overlying the first layer and sealed thereto in those areas between the pouch-like containers where the two layers are in contact. The components may be deposited in particulate, paste or molten form and the laminate layers should prevent egress of the contents of the pouch-like containers prior to their addition to water. The layers may separate or may remain attached together on contact with water, the only requirement being that the structure should permit rapid release of the contents of the pouch-like containers into solution. The number of pouch-like containers per unit area of substrate is a matter of choice but will normally vary between 500 and 25,000 per square meter.

Suitable materials which can be used for the flexible laminate layers in this aspect of the invention include, among others, sponges, paper and woven and non-woven fabrics.

However the preferred means of carrying out the washing process according to the present invention includes the use of a reusable dispensing device having walls that are permeable to liquid but impermeable to the solid composition.

Devices of this kind are disclosed in European Patent Application Publication Nos. 0 343 069 and 0 344 070. The latter Application discloses a device comprising a flexible sheet in the form of a bag extending from a support ring defining an orifice, the orifice being adapted to admit to the bag sufficient products for one washing cycle in a washing cycle. A portion of the washing medium flows through the orifice into the bag, dissolves the product, and the solution then passes outwardly through the orifice into the washing medium. The support ring is provided with a masking arrangement to prevent egress of wetted, undissolved, prod-

uct, this arrangement typically comprising radially extending walls extending from a central boss in a spoked wheel configuration, or a similar structure in which the walls have a helical form.

EXAMPLES

The following examples illustrate the invention and facilitate its understanding.

The abbreviations for the individual ingredients have the following meaning:

LAS: sodium salt of linear dodecyl benzene sulfonate

TAS: sodium salt of tallow alcohol sulfate

AS: sodium salt of alkyl (C14-C15) sulfate

AO: C12-C14 alkyl dimethylamine oxide

FA45E7: fatty alcohol (C14-C15) ethoxylated with about 7 moles of ethylene oxide

CAT: C12 alkyl trimethyl ammonium chloride

Clay: smectite clay

Zeolite 4A: sodium salt of zeolite 4A with average particle size between 1-10 micrometer

SKS-6: crystalline layered silicate (Hoechst)

Copolymer AA/MA: copolymer of acrylic acid and maleic acid

PAA: polyacrylic acid MW 100→10000

CMC: carboxymethylcellulose Phosphonate: sodium salt of ethylenediamine tetramethylene phosphonic acid

EDTA: sodium salt of ethylenediamine tetra acetate

PB1: NaBO₂.H₂O₂

PB4: NaBO₂.H₂O₂.3H₂O

TAED: tetra acetyl ethylene diamine

NOBS: - nonanoyl oxybenzene sodium sulfonate

P.A.: sulphonated zinc phthalocyanine Silicate (R=n): SiO₂/Na₂O=n

Amylase: Termamyl 60 T (Novo-Nordisk)

Lipase: Lipolase 100 T (Novo-Nordisk)

Protease: Savinase 4 T (Novo-Nordisk)

SSS: Suds Suppressing System (silica/silicone mixture)

EXAMPLE I

Criticality of the cellulase performance parameter of claim 1

The following test was conducted:

Test conditions:

Washing temperature: 60° C. (heat up cycle)

-continued

Washing time: 40 min.

pH = 7.5

Water hardness: 4 mmol/L

Detergent concentration: 1%

Detergent composition: cfr. EPA 350 098 ex. 1

Cellulases:

1) Celluzyme^R supplied by Novo Nordisk = reference

2) 43kD endoglucanase = cellulase according to the invention

Test Results:	% C14-CMC Removal by Cellulase
Detergent without cellulase (=reference)	0
Detergent + Celluzyme ^R	
0.25 mg protein/L	below 3
0.9 mg protein/L	10
1.5 mg protein/L	12.7
3.0 mg protein/L	17.7
4.5 mg protein/L	21.5
Detergent + 43kD endoglucanase	
0.3 mg protein/L	20.3
0.25 mg protein/L	18.5

Discussion of the results:

The above data clearly demonstrate the criticality of the claimed parameter for the cellulases of the invention over the commercially available Celluzyme.

EXAMPLE II

The following base compositions were prepared:

	COMPOSITIONS: (all levels in % by weight)	
	Compact Detergent	Non-compact Detergent
LAS	9.40	6.27
TAS	3.00	2.00
FA45E7	2.65	1.77
Na citrate/citric acid	18.50	12.33
Zeolite 4A	32.65	21.77
Copolymer AA/MA	4.90	3.27
Phosphonate	0.19	0.13
Na carbonate	3.00	2.00

SEQUENCE LISTING

(1) GENERAL INFORMATION:

(i i i) NUMBER OF SEQUENCES: 4

(2) INFORMATION FOR SEQ ID NO:1:

(i) SEQUENCE CHARACTERISTICS:

- (A) LENGTH: 1060 base pairs
- (B) TYPE: nucleic acid
- (C) STRANDEDNESS: unknown
- (D) TOPOLOGY: unknown

(i x) FEATURE:

- (A) NAME/KEY: CDS
- (B) LOCATION: 10..924

-continued

His Gln Cys Leu

285

CGAAATGACA CTCCAATCA CTGTATTAGT TCTTGACAT AATTTCGTCA TCCCTCCAGG 1024
 GATTGTCACA TAAATGCAAT GAGGAACAAT GAGTAC 1060

(2) INFORMATION FOR SEQ ID NO:2:

(i) SEQUENCE CHARACTERISTICS:

(A) LENGTH: 305 amino acids

(B) TYPE: amino acid

(D) TOPOLOGY: linear

(i i) MOLECULE TYPE: protein

(x i) SEQUENCE DESCRIPTION: SEQ ID NO:2:

Met	Arg	Ser	Ser	Pro	Leu	Leu	Pro	Ser	Ala	Val	Val	Ala	Ala	Leu	Pro
-21	-20					-15					-10				
Val	Leu	Ala	Leu	Ala	Ala	Asp	Gly	Arg	Ser	Thr	Arg	Tyr	Trp	Asp	Cys
-5					1				5					10	
Cys	Lys	Pro	Ser	Cys	Gly	Trp	Ala	Lys	Lys	Ala	Pro	Val	Asn	Gln	Pro
			15					20					25		
Val	Phe	Ser	Cys	Asn	Ala	Asn	Phe	Gln	Arg	Ile	Thr	Asp	Phe	Asp	Ala
		30					35					40			
Lys	Ser	Gly	Cys	Glu	Pro	Gly	Gly	Val	Ala	Tyr	Ser	Cys	Ala	Asp	Gln
	45					50					55				
Thr	Pro	Trp	Ala	Val	Asn	Asp	Asp	Phe	Ala	Leu	Gly	Phe	Ala	Ala	Thr
60					65					70					75
Ser	Ile	Ala	Gly	Ser	Asn	Glu	Ala	Gly	Trp	Cys	Cys	Ala	Cys	Tyr	Glu
				80					85					90	
Leu	Thr	Phe	Thr	Ser	Gly	Pro	Val	Ala	Gly	Lys	Lys	Met	Val	Val	Gln
			95					100					105		
Ser	Thr	Ser	Thr	Gly	Gly	Asp	Leu	Gly	Ser	Asn	His	Phe	Asp	Leu	Asn
		110					115					120			
Ile	Pro	Gly	Gly	Gly	Val	Gly	Ile	Phe	Asp	Gly	Cys	Thr	Pro	Gln	Phe
	125					130					135				
Gly	Gly	Leu	Pro	Gly	Gln	Arg	Tyr	Gly	Gly	Ile	Ser	Ser	Arg	Asn	Glu
140					145					150					155
Cys	Asp	Arg	Phe	Pro	Asp	Ala	Leu	Lys	Pro	Gly	Cys	Tyr	Trp	Arg	Phe
				160					165					170	
Asp	Trp	Phe	Lys	Asn	Ala	Asp	Asn	Pro	Ser	Phe	Ser	Phe	Arg	Gln	Val
			175					180					185		
Gln	Cys	Pro	Ala	Glu	Leu	Val	Ala	Arg	Thr	Gly	Cys	Arg	Arg	Asn	Asp
		190					195					200			
Asp	Gly	Asn	Phe	Pro	Ala	Val	Gln	Ile	Pro	Ser	Ser	Ser	Thr	Ser	Ser
	205					210					215				
Pro	Val	Asn	Gln	Pro	Thr	Ser	Thr	Ser	Thr	Thr	Ser	Thr	Ser	Thr	Thr
220					225					230					235
Ser	Ser	Pro	Pro	Val	Gln	Pro	Thr	Thr	Pro	Ser	Gly	Cys	Thr	Ala	Glu
				240					245					250	
Arg	Trp	Ala	Gln	Cys	Gly	Gly	Asn	Gly	Trp	Ser	Gly	Cys	Thr	Thr	Cys
			255					260					265		
Val	Ala	Gly	Ser	Thr	Cys	Thr	Lys	Ile	Asn	Asp	Trp	Tyr	His	Gln	Cys
		270					275					280			

Leu

(2) INFORMATION FOR SEQ ID NO:3:

-continued

(i) SEQUENCE CHARACTERISTICS:

- (A) LENGTH: 1473 base pairs
 (B) TYPE: nucleic acid
 (C) STRANDEDNESS: unknown
 (D) TOPOLOGY: unknown

(i x) FEATURE:

- (A) NAME/KEY: CDS
 (B) LOCATION: 97..1224

(x i) SEQUENCE DESCRIPTION: SEQ ID NO:3:

GAATT	CGCGG	CCGCT	CATTC	ACTTC	ATTCA	TTCTT	TAGAA	TTACA	TACAC	TCTCT	TTCAA						60
AACAG	TCACT	CTTAA	AACAA	AACA	ACTTTT	GCAACA	ATG	CGA	TCT	TAC	ACT	CTT					114
								Met	Arg	Ser	Tyr	Thr	Leu				
								1				5					
CTC	GCC	CTG	GCC	GGC	CCT	CTC	GCC	GTG	AGT	GCT	GCT	TCT	GGA	AGC	GGT		162
Leu	Ala	Leu	Ala	Gly	Pro	Leu	Ala	Val	Ser	Ala	Ala	Ser	Gly	Ser	Gly		
			10					15					20				
CAC	TCT	ACT	CGA	TAC	TGG	GAT	TGC	TGC	AAG	CCT	TCT	TGC	TCT	TGG	AGC		210
His	Ser	Thr	Arg	Tyr	Trp	Asp	Cys	Cys	Lys	Pro	Ser	Cys	Ser	Trp	Ser		
		25					30					35					
GGA	AAG	GCT	GCT	GTC	AAC	GCC	CCT	GCT	TTA	ACT	TGT	GAT	AAG	AAC	GAC		258
Gly	Lys	Ala	Ala	Val	Asn	Ala	Pro	Ala	Leu	Thr	Cys	Asp	Lys	Asn	Asp		
	40					45					50						
AAC	CCC	ATT	TCC	AAC	ACC	AAT	GCT	GTC	AAC	GGT	TGT	GAG	GGT	GGT	GGT		306
Asn	Pro	Ile	Ser	Asn	Thr	Asn	Ala	Val	Asn	Gly	Cys	Glu	Gly	Gly	Gly		
	55				60					65					70		
TCT	GCT	TAT	GCT	TGC	ACC	AAC	TAC	TCT	CCC	TGG	GCT	GTC	AAC	GAT	GAG		354
Ser	Ala	Tyr	Ala	Cys	Thr	Asn	Tyr	Ser	Pro	Trp	Ala	Val	Asn	Asp	Glu		
				75					80						85		
CTT	GCC	TAC	GGT	TTC	GCT	GCT	ACC	AAG	ATC	TCC	GGT	GGC	TCC	GAG	GCC		402
Leu	Ala	Tyr	Gly	Phe	Ala	Ala	Thr	Lys	Ile	Ser	Gly	Gly	Ser	Glu	Ala		
			90					95					100				
AGC	TGG	TGC	TGT	GCT	TGC	TAT	GCT	TTG	ACC	TTC	ACC	ACT	GGC	CCC	GTC		450
Ser	Trp	Cys	Cys	Ala	Cys	Tyr	Ala	Leu	Thr	Phe	Thr	Thr	Gly	Pro	Val		
		105					110					115					
AAG	GGC	AAG	AAG	ATG	ATC	GTC	CAG	TCC	ACC	AAC	ACT	GGA	GGT	GAT	CTC		498
Lys	Gly	Lys	Lys	Met	Ile	Val	Gln	Ser	Thr	Asn	Thr	Gly	Gly	Asp	Leu		
	120			125							130						
GGC	GAC	AAC	CAC	TTC	GAT	CTC	ATG	ATG	CCC	GGC	GGT	GGT	GTC	GGT	ATC		546
Gly	Asp	Asn	His	Phe	Asp	Leu	Met	Met	Pro	Gly	Gly	Gly	Val	Gly	Ile		
	135			140					145						150		
TTC	GAC	GGC	TGC	ACC	TCT	GAG	TTC	GGC	AAG	GCT	CTC	GGC	GGT	GCC	CAG		594
Phe	Asp	Gly	Cys	Thr	Ser	Glu	Phe	Gly	Lys	Ala	Leu	Gly	Gly	Ala	Gln		
				155				160						165			
TAC	GGC	GGT	ATC	TCC	TCC	CGA	AGC	GAA	TGT	GAT	AGC	TAC	CCC	GAG	CTT		642
Tyr	Gly	Gly	Ile	Ser	Ser	Arg	Ser	Glu	Cys	Asp	Ser	Tyr	Pro	Glu	Leu		
			170					175					180				
CTC	AAG	GAC	GGT	TGC	CAC	TGG	CGA	TTC	GAC	TGG	TTC	GAG	AAC	GCC	GAC		690
Leu	Lys	Asp	Gly	Cys	His	Trp	Arg	Phe	Asp	Trp	Phe	Glu	Asn	Ala	Asp		
		185					190					195					
AAC	CCT	GAC	TTC	ACC	TTT	GAG	CAG	GTT	CAG	TGC	CCC	AAG	GCT	CTC	CTC		738
Asn	Pro	Asp	Phe	Thr	Phe	Glu	Gln	Val	Gln	Cys	Pro	Lys	Ala	Leu	Leu		
	200					205					210						
GAC	ATC	AGT	GGA	TGC	AAG	CGT	GAT	GAC	GAC	TCC	AGC	TTC	CCT	GCC	TTC		786
Asp	Ile	Ser	Gly	Cys	Lys	Arg	Asp	Asp	Asp	Ser	Ser	Phe	Pro	Ala	Phe		
	215			220						225					230		
AAG	GTT	GAT	ACC	TCG	GCC	AGC	AAG	CCC	CAG	CCC	TCC	AGC	TCC	GCT	AAG		834
Lys	Val	Asp	Thr	Ser	Ala	Ser	Lys	Pro	Gln	Pro	Ser	Ser	Ser	Ala	Lys		
				235					240					245			

-continued

AAG Lys	ACC Thr	ACC Thr	TCC Ser 250	GCT Ala	GCT Ala	GCT Ala	GCC Ala	GCT Ala	CAG Gln	CCC Pro	CAG Gln	AAG Lys	ACC Thr 260	AAG Lys	GAT Asp	882
TCC Ser	GCT Ala	CCT Pro 265	GTT Val	GTC Val	CAG Gln	AAG Lys	TCC Ser 270	TCC Ser	ACC Thr	AAG Lys	CCT Pro	GCC Ala 275	GCT Ala	CAG Gln	CCC Pro	930
GAG Glu 280	CCT Pro	ACT Thr	AAG Lys	CCC Pro	GCC Ala	GAC Asp 285	AAG Lys	CCC Pro	CAG Gln	ACC Thr	GAC Asp 290	AAG Lys	CCT Pro	GTC Val	GCC Ala	978
ACC Thr 295	AAG Lys	CCT Pro	GCT Ala	GCT Ala	ACC Thr 300	AAG Lys	CCC Pro	GTC Val	CAA Gln	CCT Pro 305	GTC Val	AAC Asn	AAG Lys	CCC Pro	AAG Lys 310	1026
ACA Thr	ACC Thr	CAG Gln	AAG Lys	GTC Val 315	CGT Arg	GGA Gly	ACC Thr	AAA Lys	ACC Thr 320	CGA Arg	GGA Gly	AGC Ser	TGC Cys	CCG Pro	GCC Ala 325	1074
AAG Lys	ACT Thr	GAC Asp	GCT Ala 330	ACC Thr	GCC Ala	AAG Lys	GCC Ala	TCC Ser 335	GTT Val	GTC Val	CCT Pro	GCT Ala	TAT Tyr 340	TAC Tyr	CAG Gln	1122
TGT Cys	GGT Gly	GGT Gly	TCC Ser	AAG Lys	TCC Ser	GCT Ala	TAT Tyr 350	CCC Pro	AAC Asn	GGC Gly	AAC Asn	CTC Leu 355	GCT Ala	TGC Cys	GCT Ala	1170
ACT Thr 360	GGA Gly	AGC Ser	AAG Lys	TGT Cys	GTC Val	AAG Lys 365	CAG Gln	AAC Asn	GAG Glu	TAC Tyr	TAC Tyr	TCC Ser 370	CAG Gln	TGT Cys	GTC Val	1218
CCC Pro 375	AAC Asn	TAAATGGTAG ATCCATCGGT TGTGGAAGAG ACTATGCGTC TCAGAAGGGA														1274
TCCTCTCATG AGCAGGCTTG TCATTGTATA GCATGGCATC CTGGACCAAG TGTTCGACCC																1334
TTGTTGTACA TAGTATATCT TCATTGTATA TATTTAGACA CATAGATAGC CTCTTGTCAG																1394
CGACAACTGG CTACAAAAGA CTTGGCAGGC TTGTTCAATA TTGACACAGT TTCCTCCATA																1454
AAAAAAAAAA AAAAAAAAAA																1473

(2) INFORMATION FOR SEQ ID NO:4:

(i) SEQUENCE CHARACTERISTICS:

(A) LENGTH: 376 amino acids

(B) TYPE: amino acid

(D) TOPOLOGY: linear

(i i) MOLECULE TYPE: protein

(x i) SEQUENCE DESCRIPTION: SEQ ID NO:4:

Met 1	Arg	Ser	Tyr	Thr 5	Leu	Leu	Ala	Leu	Ala 10	Gly	Pro	Leu	Ala	Val 15	Ser
Ala	Ala	Ser	Gly 20	Ser	Gly	His	Ser	Thr 25	Arg	Tyr	Trp	Asp	Cys 30	Cys	Lys
Pro	Ser	Cys 35	Ser	Trp	Ser	Gly	Lys 40	Ala	Ala	Val	Asn	Ala 45	Pro	Ala	Leu
Thr	Cys 50	Asp	Lys	Asn	Asp	Asn 55	Pro	Ile	Ser	Asn	Thr 60	Asn	Ala	Val	Asn
Gly 65	Cys	Glu	Gly	Gly	Gly 70	Ser	Ala	Tyr	Ala	Cys 75	Thr	Asn	Tyr	Ser	Pro 80
Trp	Ala	Val	Asn	Asp 85	Glu	Leu	Ala	Tyr	Gly 90	Phe	Ala	Ala	Thr	Lys 95	Ile
Ser	Gly	Gly	Ser 100	Glu	Ala	Ser	Trp	Cys 105	Cys	Ala	Cys	Tyr	Ala 110	Leu	Thr
Phe	Thr	Thr 115	Gly	Pro	Val	Lys	Gly 120	Lys	Lys	Met	Ile	Val 125	Gln	Ser	Thr

-continued

Asn	Thr	Gly	Gly	Asp	Leu	Gly	Asp	Asn	His	Phe	Asp	Leu	Met	Met	Pro
	130					135					140				
Gly	Gly	Gly	Val	Gly	Ile	Phe	Asp	Gly	Cys	Thr	Ser	Glu	Phe	Gly	Lys
145					150					155					160
Ala	Leu	Gly	Gly	Ala	Gln	Tyr	Gly	Gly	Ile	Ser	Ser	Arg	Ser	Glu	Cys
				165					170					175	
Asp	Ser	Tyr	Pro	Glu	Leu	Leu	Lys	Asp	Gly	Cys	His	Trp	Arg	Phe	Asp
			180					185					190		
Trp	Phe	Glu	Asn	Ala	Asp	Asn	Pro	Asp	Phe	Thr	Phe	Glu	Gln	Val	Gln
		195					200					205			
Cys	Pro	Lys	Ala	Leu	Leu	Asp	Ile	Ser	Gly	Cys	Lys	Arg	Asp	Asp	Asp
	210					215					220				
Ser	Ser	Phe	Pro	Ala	Phe	Lys	Val	Asp	Thr	Ser	Ala	Ser	Lys	Pro	Gln
225					230					235					240
Pro	Ser	Ser	Ser	Ala	Lys	Lys	Thr	Thr	Ser	Ala	Ala	Ala	Ala	Ala	Gln
				245					250					255	
Pro	Gln	Lys	Thr	Lys	Asp	Ser	Ala	Pro	Val	Val	Gln	Lys	Ser	Ser	Thr
			260					265					270		
Lys	Pro	Ala	Ala	Gln	Pro	Glu	Pro	Thr	Lys	Pro	Ala	Asp	Lys	Pro	Gln
		275					280					285			
Thr	Asp	Lys	Pro	Val	Ala	Thr	Lys	Pro	Ala	Ala	Thr	Lys	Pro	Val	Gln
	290					295					300				
Pro	Val	Asn	Lys	Pro	Lys	Thr	Thr	Gln	Lys	Val	Arg	Gly	Thr	Lys	Thr
305					310					315					320
Arg	Gly	Ser	Cys	Pro	Ala	Lys	Thr	Asp	Ala	Thr	Ala	Lys	Ala	Ser	Val
				325					330					335	
Val	Pro	Ala	Tyr	Tyr	Gln	Cys	Gly	Gly	Ser	Lys	Ser	Ala	Tyr	Pro	Asn
			340					345					350		
Gly	Asn	Leu	Ala	Cys	Ala	Thr	Gly	Ser	Lys	Cys	Val	Lys	Gln	Asn	Glu
		355					360					365			
Tyr	Tyr	Ser	Gln	Cys	Val	Pro	Asn								
	370					375									

We claim:

1. A granular detergent composition comprising surface-active agent, builder and cellulase wherein said cellulase consists essentially of a homogeneous endoglucanase component which is immunoreactive with a monoclonal antibody raised against a partially purified about 43 kD cellulase derived from *Humicola insolens*, DSM 1800;

said granular detergent composition comprising no more than about 15% by weight of inorganic filler salt, and said granular detergent composition having a density of about 550 to about 950 g/liter of composition.

2. A detergent composition according to claim 1 wherein the endoglucanase component of said cellulase has an isoelectric point of about 5.1.

3. A detergent composition according to claim 1, wherein said endoglucanase component is produced by a method comprising cultivating a host cell transformed with a recombinant DNA vector carrying a DNA sequence encoding said endoglucanase component or a precursor of said endoglucanase component as well as DNA sequences encoding functions permitting the expression of the DNA sequence encoding the endoglucanase component, or a precursor thereof, in a culture medium under conditions permitting the expression of the endoglucanase component or precursor

thereof and recovering the endoglucanase component from the culture.

4. A detergent composition according to claim 2 wherein said endoglucanase component is produced by a method comprising cultivating a host cell transformed with a recombinant DNA vector carrying a DNA sequence encoding said endoglucanase component or a precursor of said endoglucanase component as well as DNA sequences encoding functions permitting the expression of the DNA sequence encoding the endoglucanase component, or a precursor thereof, in a culture medium under conditions permitting the expression of the endoglucanase component or precursor thereof and recovering the endoglucanase component from the culture.

5. A detergent composition in accordance with claim 1, 2, 3, or 4 wherein the level of the cellulase is such that the amount of enzyme protein to be delivered in the wash solution is from 0.005 to 40 mg/liter of wash solution.

6. A detergent composition according to claim 1 wherein said inorganic filler salt is selected from alkali and alkaline-earth metal salts of sulphate and chloride.

7. A detergent composition in accordance with claim 1 which does not contain more than 10% by wt of inorganic filler salt.

8. A detergent composition in accordance with claim 5 which does not contain more than 10% by wt of inorganic filler salt.

9. A detergent composition in accordance with claim 1 which does not contain more than 5% by wt of inorganic filler salt.

10. A detergent composition according to claim 1 which has a density of 650 to 850 g/liter.

11. A detergent composition according to claim 1, 2, 3 or 4 which is substantially free of phosphate compounds, and wherein said builder is selected from the group consisting of aluminosilicate ion exchangers, citrates, carbonates and mixtures thereof.

12. A granular detergent composition comprising surface-active agent, builder and cellulase wherein said cellulase is an endoglucanase enzyme having the amino acid sequence shown in the appended sequence listing ID#2;

said granular detergent composition comprising no more than about 15% by weight of inorganic filler salt, and

said granular detergent composition having a density of about 550 to about 950 g/liter of composition.

13. A detergent composition according to claim 12 wherein said endoglucanase enzyme is produced by a species of *Humicola*, e.g. *Humicola insolens*.

14. A granular detergent composition comprising surface-active agent, builder and cellulase wherein said cellulase is an endoglucanase enzyme having the amino acid sequence shown in the appended sequence listing ID#4;

said granular detergent composition comprising no more than about 15% by weight of inorganic filler salt, and

said granular detergent composition having a density of about 550 to about 950 liter of composition.

15. A detergent composition according to claim 14 wherein said endoglucanase enzyme is produced by a species of *Fusarium*.

16. A detergent composition according to claim 1, 2, 3 or 4 wherein said enzyme is produced by a DNA construct comprising a DNA sequence encoding the enzyme.

17. A detergent composition according to claim 15 wherein the DNA sequence is as shown in the appended sequence listings ID #1 or ID #3.

18. A detergent composition according to claim 3 or 4 wherein said host cell is a strain of the fungus such as *Trichoderma* or *Aspergillus*, or a yeast cell belonging to a strain of *Hansenula* or *Saccharomyces*, e.g. a strain of *Saccharomyces cerevisiae*.

19. A detergent composition according to claim 3 or 4 wherein said host cell is a strain of a bacterium, e.g. *Bacillus*, *Streptomyces* or *E. coli*.

20. A process for washing fabrics in a washing machine wherein an amount of from 15 to 170 g of a detergent composition according to claim 1 is used for the main wash cycle.

21. A process for washing fabrics according to claim 20 wherein said amount of detergent composition is put in a container able to release the composition at the start of the wash cycle, and said container is placed in the drum of the washing machine, together with the fabrics to be washed.

* * * * *

**UNITED STATES PATENT AND TRADEMARK OFFICE
CERTIFICATE OF CORRECTION**

PATENT NO. : 5,520,838

Page 1 of 9

DATED : May 28, 1996

INVENTOR(S) : Andre C. Baeck et al.

It is certified that error appears in the above-identified patent and that said Letters Patent is hereby corrected as shown below:

Column 1, line 14 "cellulase-is" should read --cellulase is--.

Column 3, line 4 "oppositions" should read --compositions--.

Column 3, line 10 "no" should read --not--.

Column 5, line 5 "hardhess/g" should read --hardness--.

Column 6, line 3 delete duplicate "and".

Column 6, line 33 "expected used" should read --expected in use--.

Column 11, line 23 "mofidications" should read --modifications--.

Column 12, line 11 "wich" should read --which--.

Column 14, line 20 "mofidiers" should read --modifiers--.

Column 16, line 30 "so s" should read --so as--.

Column 16, line 62 "products" should read --product--.

Column 17, line 26 "Phosphonate: sodium salt of" should be moved to line 27

Column 17, line 33 "Silicate (R=n): SiO2/" should be moved to line 34

Column 18, line 46 Example II is incomplete; please insert before Sequence Listing:

COMPOSITIONS:

(all levels in % by weight)

	Compact	Non-compact
	Detergent	Detergent
Silicate (R - 2)	2.90	1.93
Protease	1.62	1.08

**UNITED STATES PATENT AND TRADEMARK OFFICE
CERTIFICATE OF CORRECTION**

PATENT NO. : 5,520,838

Page 2 of 9

DATED : May 28, 1996

INVENTOR(S) : Andre C. Baeck et al.

It is certified that error appears in the above-identified patent and that said Letters Patent is hereby corrected as shown below:

Sulfate	4.50	30.00
SSS	0.40	0.27
Minors - water	balance to 100%	
Density : g/L at 20°C	680	415
Recommended product usage (g/wash)	120	180

Color Rejuvenation Testing

Test conditions :

Lauderometer equipment

Washing temperature : 40°C

Washing time : 3h

Number of wash cycles : 2

pH - 8.2 non-compact detergent

8.5 compact detergent

Water hardness : 15gr./US gal.

Detergent concentration :

0.75% for non-compact detergent

0.66% for compact detergent

Test fabric : worn blue pyjama cotton

(90/10 cotton/Polyester)

UNITED STATES PATENT AND TRADEMARK OFFICE
CERTIFICATE OF CORRECTION

PATENT NO. : 5,520,838

Page 3 of 9

DATED : May 28, 1996

INVENTOR(S) : Andre C. Baeck et al.

It is certified that error appears in the above-identified patent and that said Letters Patent is hereby corrected as shown below:

Cellulases : 1) Celluzyme^R supplied by Novo Nordisk
(= reference) .
2) 43kD endoglucanase - cellulase
according to the present invention

Wash test : Swatches of 8g of worn blue pyjama fabric were treated with the different wash solutions. After tumble drying, the fabrics were graded for colour clarification effects by direct comparison of the two different detergent matrices at equal cellulase level. Visual grading by expert judges using a 0 to 4 scale was preferred. (0 stands for no difference and 4 stands for very big difference.)

Test Results :

I) Non-Compact Detergent

	<u>PSU</u>	<u>mg protein/PSU</u>
NO cellulase	0	
Celluzyme		
138 mg protein/L + 2.3		60
43kD endoglucanase		
18.6 mg protein/L + 2.2		8.5

UNITED STATES PATENT AND TRADEMARK OFFICE
CERTIFICATE OF CORRECTION

PATENT NO. : 5,520,838

Page 4 of 9

DATED : May 28, 1996

INVENTOR(S) : Andre C. Baeck et al.

It is certified that error appears in the above-identified patent and that said Letters Patent is hereby corrected as shown below:

II) Compact Detergent

	<u>PSU</u>	<u>mg protein/PSU</u>
NO cellulase	0	
Celluzyme		
165 mg protein/L	+ 3.8	43
43kD endoglucanase		
3.4 mg protein/	+ 3.4	1.0

LSD (Least Significant Difference) = 0.5 PSU

From the mg protein/PSU result, the following efficiency factors were calculated :

Efficiency factor of 43kD endoglucanase versus Celluzyme :

in Non Compact Detergent

in Compact Detergent

$$60/8.5 = 7$$

$$43/1.0 = 43$$

Efficiency factor in Compact Detergent versus in Non Compact Detergent

of Celluzyme

of 43kD endoglucanase

$$60/43 = 1.4$$

$$8.5/1 = 8.5$$

**UNITED STATES PATENT AND TRADEMARK OFFICE
CERTIFICATE OF CORRECTION**

PATENT NO. : 5,520,838

Page 5 of 9

DATED : May 28, 1996

INVENTOR(S) : Andre C. Baeck et al.

It is certified that error appears in the above-identified patent and that said Letters Patent is hereby corrected as shown below:

Conclusions :

The above results show a cellulase selected according to the present invention is 43 times more effective than a state-of-the-art cellulase in the claimed compact matrix. Furthermore, the above results show that the performance enhancement due to the claimed compact matrix seen with the selected cellulases is surprisingly much higher than what can be obtained with a state-of-the-art cellulase.

EXAMPLE III

CLAY SOIL REMOVAL TESTING

Cellulase enzymes also are very efficient in removing clay stains from fabrics. This particular performance characteristic has been checked for a 43kD endoglucanase in the two detergent compositions given in example II.

Conditions:

Linitest equipment

60C wash (heat up cycle)

Wash time: 40 min.

Water hardness: Brussels city water

Detergent concentrations:

0.66% for the Compact detergent

1.0% for the non compact detergent

**UNITED STATES PATENT AND TRADEMARK OFFICE
CERTIFICATE OF CORRECTION**

PATENT NO. : 5,520,838

Page 6 of 9

DATED : May 28, 1996

INVENTOR(S) : Andre C. Baeck et al.

It is certified that error appears in the above-identified patent and that said Letters Patent is hereby corrected as shown below:

Cellulase concentrations: 1.55, 3.10, 4.65 and 6.2mg enzyme protein / L wash liquor.

Wash test:

Muslin cotton fabric was soiled with naturally-derived clays of two different locations (US, UK). Cellulase performance was evaluated by comparing the clay stains washed at equal cellulase level in the two different detergent compositions. The visual grading scale used in example II was again preferred.

Results:

<u>Cellulase level:</u>	<u>1.55</u>	<u>3.1</u>	<u>4.7</u>	<u>6.2</u>
(mg enz. prot. / L wash liquor)				

Compact detergent

US clay	+ 1.50	+ 2.50	+ 2.00	+ 1.50
UK clay	+ 0.50	+ 1.00	+ 1.50	+ 2.50

Non compact detergent

(-reference)	0	0	0	0
--------------	---	---	---	---

LSD (least significant difference) = 0.42 at 95% confidence.

**UNITED STATES PATENT AND TRADEMARK OFFICE
CERTIFICATE OF CORRECTION**

PATENT NO. : 5,520,838

Page 7 of 9

DATED : May 28, 1996

INVENTOR(S) : Andre C. Baeck et al.

It is certified that error appears in the above-identified patent and that said Letters Patent is hereby corrected as shown below:

The clay stain removal performance of the cellulase selected according to the present invention, in the compact detergent composition of the invention is significantly superior to the performance of the same cellulase in the conventional, non compact detergent composition.

EXAMPLES IV-XI

The following compact detergent compositions are also prepared :

COMPACT DETERGENT COMPOSITIONS:

(all levels in % by weight)

<u>EXAMPLE</u>	IV	V	VI	VII	VIII	IX	X	XI	XII
LAS	9.40	12.50	11.00	—	7.58	7.58	8.20	6.50	—
TAS	3.00	—	—	—	2.43	2.43	2.65	3.25	3.90
AS	—	—	4.80	12.00	—	—	—	—	—
FA45E7	2.65	2.00	4.00	1.00	5.11	5.11	3.15	2.20	6.00
CAT	—	—	—	—	—	—	—	—	2.45
Coconut glucose amide	—	11.00	—	—	—	—	—	—	—
Tallow glucose amide	—	—	—	10.00	—	—	—	—	—
Na citrate/citric acid	20.50	29.50	18.00	18.00	—	5.00	23.50	12.00	15.00
Zeolite 4A	33.65	—	32.00	32.50	23.80	15.65	—	16.00	20.00
SKS-6	—	—	—	—	—	12.50	—	—	—
Copolymer AA/MA	4.90	—	4.10	5.00	5.60	2.90	3.50	3.45	3.45
PAA	—	5.70	—	—	—	—	1.50	—	—
Phosphonate	0.19	0.23	0.19	1.00	0.57	0.43	0.30	—	—
EDTA	—	—	—	—	0.25	—	—	0.32	0.32
Na carbonate / bicarbonate	2.00	12.00	3.28	2.50	17.30	8.00	2.50	9.90	9.90
Silicate (R = 2)	3.00	4.20	3.00	2.00	2.00	2.50	2.30	2.50	2.50
CMC	—	0.15	—	—	0.48	0.34	0.25	—	—
Clay	—	—	—	—	—	—	12.00	8.60	8.60

UNITED STATES PATENT AND TRADEMARK OFFICE
CERTIFICATE OF CORRECTION

Page 8 of 9

PATENT NO. : 5,520,838
 DATED : May 28, 1996
 INVENTOR(S) : Andre C. Baeck et al.

It is certified that error appears in the above-identified patent and that said Letters Patent is hereby corrected as shown below:

PB1	—	—	—	—	13.12	13.12	11.47	11.50	—
PB4	—	—	—	—	—	—	3.55	—	—
Percarbonate	—	—	—	—	—	—	—	—	12.00
TAED	—	—	—	—	5.70	5.70	2.47	3.20	—
NOBS	—	—	—	—	—	—	2.00	—	—
P.A.	—	—	—	—	0.002	0.002	—	0.003	0.003
Protease	1.62	1.30	1.20	1.60	1.35	1.35	1.05	1.40	1.40
Lipolase	—	—	0.40	0.30	—	0.20	—	0.30	0.30
Amylase	0.15	—	0.20	0.30	—	0.10	—	—	—
Sulfate	2.54	3.79	2.38	2.45	1.50	1.50	2.23	3.45	3.45
Brightener	—	0.27	0.27	0.27	0.24	0.24	0.24	0.24	0.24
SSS	0.40	0.40	0.40	0.40	0.65	0.65	0.50	0.50	0.50
Minors + water					balance to 100%				

Cellulase

at levels so as to deliver $0.01 < X < 10$ mg enzyme protein / wash liquor

UNITED STATES PATENT AND TRADEMARK OFFICE
CERTIFICATE OF CORRECTION

PATENT NO. : 5,520,838
DATED : May 28, 1996
INVENTOR(S) : Andre C. Baeck, et al.

Page 9 of 9

It is certified that error appears in the above-identified patent and that said Letters Patent is hereby corrected as shown below:

Column 30, line 2 "950 liter" should be --950g/litre--.

Column 30, lines 16 and 17, delete ", e.g. a strain of *Saccharomyces cerevisae*"

Signed and Sealed this
Third Day of December, 1996

Attest:



BRUCE LEHMAN

Attesting Officer

Commissioner of Patents and Trademarks