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[54]	LUBRICANT AND MAGNETIC RECORDING MEDIUM THEREWITH				
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427/128, 131

[56] References Cited U.S. PATENT DOCUMENTS

3,646,112	2/1972	Sterling, Jr	252/54.6
4,096,079	6/1978	Pardee	252/54.6
4,188,434	2/1980	Loran	252/54.6
4,232,072	11/1980	Pardee	252/54.6
4,912,252	3/1990	Dekura et al	252/54.6
5,004,554	4/1991	Tohzuka et al	252/54.6

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[57] ABSTRACT

A lubricant contains ester compound made of perfluoropolyether and long chain alkyl carboxylic acid or alcohol. The perfluoropolyether has hydroxyl or carboxyl group at its end. The ester compound is admixed with long chain alkyl amine of from 0.01 to 100 in mole ratio. The long chain alkyl may be partially fluorined or perfluoro alkyl. At least one of phosphoric ester, phosphorous ester and long chain carboxylate ester can be further admixed. A magnetic recording medium can be coated by the lubricant.

12 Claims, No Drawings

LUBRICANT AND MAGNETIC RECORDING MEDIUM THEREWITH

BACKGROUND OF THE INVENTION

1. Field of the Invention

This invention relates to lubricant with perfluoropolyether and a magnetic recording medium having such a lubricant.

2. Background

There has been developed a magnetic recording medium of the thin metal layer type, in which a magnetic layer is formed on a non-magnetic base film by means of, for example, evaporating ferromagnetic metal material. In such a medium, the surface of magnetic layer is 15 so smooth that substantial contacting area may become large against transporting elements such as magnetic heads and guide rollers. Since the coefficient of friction also becomes large, it causes some problems such as a sticking phenomena inviting unstabilized running char- 20 acteristics and poor durability.

The use of many kinds of lubricant has been considered to improve the problem. It has been tried to decrease the coefficient of friction by coating higher fatty acid and its ester on the surface of magnetic layer of 25 magnetic recording medium.

As the lubricant for use in the magnetic recording medium, there are very severe requirements on its characteristics. Up to the present, no conventional lubricant has completely satisfied the requirements. The follow- 30 ing describes the requirements for the lubricant for use in the magnetic recording medium.

- [1] To have an excellent low temperature characteristic to secure a certain lubricant effect when using in very cold area;
- [2] To be able to coat as an extremely thin film without causing any problem due to the space between a magnetic head, but with a sufficient lubricant characteristics; and
- [3] To have a long life time maintaining its lubricant 40 effect after the use for long hours.

The conventional higher fatty acid and its ester become frozen and solidified in low temperature condition such as below zero centigrade. Therefore, it loses the function as a lubricant and tends to have no durability 45 after the use for long hours.

The practical characteristics such as running characteristic and durability have not been satisfied yet because of the lack of ability of lubricant in use in the field of magnetic recording medium.

SUMMARY OF THE INVENTION

It is an object of the invention to provide a lubricant having an excellent lubricant characteristics for the use under various conditions.

It is another object of the invention to provide a lubricant to maintain its lubricant effect for a long time.

It is a further object of the invention to provide a magnetic recording medium, containing the above lubricant, with excellent running characteristics, wear 60 characteristics and durability.

These and other objects, advantages and features of the invention will become apparent to those skilled in the art upon consideration of the following description of the invention.

In one aspect of the invention, a lubricant contains ester compound made of perfluoropolyether and long chain carboxylic acid or alcohol. The perfluoropolyether has hydroxyl or carboxyl group at its end. The ester compound is admixed with long chain hydrocarbon amine.

In another aspect of the invention, a magnetic record-⁵ ing tape is provided to have the surface of magnetic layer covered by the lubricant containing ester compound made of perfluoropolyether and long chain carboxylic acid or alcohol. The ester compound is admixed with long chain hydrocarbon amine.

In accordance with the present invention, improvement on the solubility can be expected by esterifying perfluoropolyether. It becomes unnecessary to use freon as a diluent. The lubricant has excellent lubricant effect under any condition and keeps it for a long time. The magnetic recording medium with this lubricant therefore has excellent wear characteristics, durability and in particular outstanding running characteristics.

DESCRIPTION OF THE INVENTION

The lubricants as the first to fourth embodiments of the invention have ester compound of perfluoropolyether respectively admixed by long chain hydrocarbon amine (see, general formula [15]).

The ester compound utilized here includes the followings.

- (A) Compound of perfluoropolyether having hydroxyl group at its end and long chain carboxylic acid (see, general formula [11]).
- (B) Compound of perfluoropolyether having hydroxyl group at its end and partially fluorined long chain carboxylic acid (see, general formula [12]).
- (C) Compound of perfluoropolyether having carboxyl group at its end and partially fluorined long chain alcohol (see, general formula [13]).
- (D) Compound of perfluoropolyether having hydroxyl group at its end and perfluoro carboxylic acid (see, general formula [14]).

R^{f1}—CH₂OCOR¹

R¹COOCH₂R^{fl}CH₂OCOR¹

general formula [11]

(where R^{fl} shows perfluoropolyether group and R^l shows long chain hydrocarbon)

R/2—CH₂OCOR²

R²COOCH₂R²CH₂OCOR²

general formula [12]

(where R/2 shows perfluoropolyether group and R2 shows partially fluorined long chain hydrocarbon)

Rf3—COOR3

R³OCOR^{f3}COOR³

general formula [13]

(where R/3 shows perfluoropolyether group and R3 shows partially fluorined long chain hydrocarbon)

R/4—CH2OCOR/5

Rf5COOCH2Rf4CH2OCORf5

general formula

[14]

(where R^{f4} shows perfluoropolyether group and R^{f5} shows perfluoro hydrocarbon)

general formula [15]

(where R shows hydrocarbon)

RNH₂

The lubricants as the fifth and sixth embodiments of the invention have ester compound as explained below of perfluoropolyether admixed by long chain hydrocarbon amine (see, general formula [15]) and further admixed by at least more than one of phosphoric ester, phosphorous ester and long chain carboxylate ester.

The ester compound utilized here includes the compound shown by the general formula [11] and the compound of perfluoropolyether having long chain alcohol (see, general formula [16]).

R/6—COOR6

R6OCOR5COOR6

general formula
[16]

(where R\sigma shows perfluoropolyether group and R\sigma shows perfluoro hydrocarbon)

The lubricant of the invention has ester compound (shown by general formulas [11], [12], [13], [14] and [16]) of perfluoropolyether with respective long chain hydrocarbon group (such as R¹, R², R³and R⁴, etc.). The long chain hydrocarbon group can be selected regardless of the differences on its molecular weight, branch structure, unsaturated coupling, isomer structure and alicyclic structure. However, considering its solubility, it is preferable to have the carbon number more than six. The structures will be shown later, more in detail by the embodiments.

The long chain hydrocarbon group as the admixed long chain hydrocarbon amine (shown by R in the general formula [15]). This alkyl group can be also selected regardless of the differences on its molecular weight, carbon number, branch structure, unsaturated coupling, existence aromatic cycle, isomer structure and alicyclic structure. Considering its coefficient of friction, it is preferable to have the long chain hydrocarbon with the carbon number more than ten.

The structure of perfluoropolyether having hydroxyl group at its end is shown by a general formula [17] as multifunctional and also by a general formulas [18] and [19] as monofunctional. The structure of perfluoropolyether having carboxyl group at its end is shown by a general formula [20] as multifunctional and also by a general formulas [21] and [22] as monofunctional.

HOCH ₂ CF ₂ (OC ₂ F ₄) _p (OCF ₂) _q OCF ₂ CH ₂ OH	general formula [17]
F(CF ₂ CF ₂ CF ₂ O) _n CF ₂ CF ₂ CH ₂ OH	general formula [18] 5
CF ₃ (OCF(CF ₃)CF ₂) _m (OCF ₂) _l CH ₂ OH	general formula [19]
HOOCCF ₂ (OC ₂ F ₄) _p (OCF ₂) _q OCF ₂ COOH	general formula [20] 6
F(CF ₂ CF ₂ CF ₂ O) _n CF ₂ CF ₂ COOH	general formula [21]
CF3(OCF(CF3)CF2)m(OCF2)nCOOH	general formula

In these chemical formulas, the suffix (l), (m) and (n) means the integer numbers more than one. However,

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the chemical structure of perfluoropolyether doesn't have such limitations.

Though the molecular weight of perfluoropolyether doesn't have any particular limitation, it is preferable to have approximately 600 to 5000. When the molecular weight becomes too large, the effect of end group becomes small and more freon is needed to dissolve it. On the other hand, when the molecular weight becomes too small, the effect of perfluoropolyether group becomes very small.

In the lubricant of this invention, the long chain hydrocarbon amine is admixed within the range of molar ratio 0.01 to 100 to the ester compound of each perfluoropolyether.

The magnetic recording medium of this invention has at least one of the lubricant of this invention coated on the surface of its magnetic layer which is formed on its non-magnetic base film.

By coating the above described lubricant on the sur-20 face of thin film magnetic recording medium such as an evaporated tape, the magnetic recording medium of this invention can obtain and maintain the excellent durability under severe conditions such as high temperature and high humidity or low temperature and low humid-25 ity.

This invention can be applied to the magnetic recording medium of the thin metal film type, in which a magnetic film layer can be formed on the non-magnetic base film by means of evaporation. It can be also applied to the magnetic medium of the thin metal film type, in which an under coating layer is formed between the magnetic film layer and the non-magnetic base film.

The materials for the non-magnetic base film and the thin metal magnetic film have no limitations to be applied for the magnetic recording medium of the thin metal film type of this invention. In other words, any conventional material known to public can be utilized.

For example, the materials as follows can be used by the method of plated coat, sputtering or PVD such as vacuum evaporation so as to form a continuous thin magnetic film of metal. They are metal such as Fe, Co or Ni; alloys to become magnetic recording film for surface magnetization such as Co/Ni, Co/Pt, Co/Pt-/Ni, FE/Co, Fe/Ni, Fe/Ni/B, Fe/Co/B, Fe/Co/Ni/B; alloy thin film like Co/Cr.

In particular for the magnetic recording thin metal film, non-magnetic materials with low melting point such as Bi, Sb, Pb, Sn, Ga, In, Ge, Si or Ti can be previously coated on the non-magnetic base film so as to create the under coating layer. Thereafter, the metal magnetic materials as above can be evaporated or sputtered from a perpendicular direction. By doing so, the non-magnetic materials with low melting point is diffused into the metal magnetic thin film so that the surface isotropy can be secured by realigning orientation and retentivity can be improved.

When using the metal magnetic thin film as a hard disk, a hard protective layer may be formed on the surface by a carbon film, a diamond or amorphous carbon film, a chromium oxide film, a zirconium oxide film or a silicon oxide film.

The material for the non-magnetic base film has no limitations and therefore any conventional material known to public can be utilized. For example, when using a hard base plate such as an aluminum alloy plate or a glass plate, an oxide coating by Alumilite procedure or an Ni/P alloy coating may be performed on the surface of base plate so as to make the surface hard.

One of the method for making magnetic recording medium have the above lubricant is making the lubricant layer top coated on the surface of the metal magnetic thin film layer or the above mentioned protective layers. At this time, as the amount for coating of lubricant, it is preferable to have 0.5-100 mg/m² and more preferably 1-20 mg/m².

The lubricant of this invention can be utilized alone as the lubricant for magnetic recording medium, but can be also utilized combined with conventional lubricant.

Further, to maintain the lubricant effect under more severe condition, an extreme pressure agent may be used together as a combination of 30:70–70:30 in weight ratio.

The extreme pressure agent is reacted with the metal 15 surface by frictional heat being generated when metal contact partially occurs in the boundary lubrication region. The reactively generated coating is therefore formed for preventing friction and wear. The extreme pressure agent of phosphorus group, sulfur group, halo-20 gen group, organometallic type and composite type can be respectively available.

Furthermore, an anticorrosion agent can be used together if necessary as well as .the above lubricant and extreme pressure agent. The anticorrosion available for 25 this kind of magnetic recording medium has been well known, such as heterocyclic compounds of phenols group, naphthol group, quinoues group, nitrogen, oxygen and sulfur.

In the magnetic recording medium of thin metal film 30 type, a back coat layer or an undercoat layer may be also formed depending on the necessity in addition to the metal magnetic thin film as the magnetic layer.

For example, similar to the magnetic coating film, carbon group fine powder can be added to give conduc- 35 tivity of resin binding agent. Also, inorganic pigment

freon. The use of freon group solvent can be therefore decreased or omitted by the lubricant of the invention.

Since the magnetic recording medium of the invention has the lubricant with higher lubricant effect, it avoids friction against the transport elements such as a magnetic head and guide rollers and therefore has excellent wear characteristics, durability and running characteristics.

Characteristics of the lubricant and the magnetic recording medium therewith will be explained below in accordance with the first to fifth embodiments in order.

EXPERIMENTATION A

The lubricant of the first embodiment contains ester compound made of perfluoropolyether with hydroxyl group at its end and long chain carboxylic acid, which is admixed with long chain hydrocarbon amine.

The ester compound is composed of a solution of perfluoropolyether with hydroxyl group at its end and third grade amine added by dropping long chain carboxylic acid chloride. The reaction has been completed under reflux and heat after dropping. The solution was then cooled to the room temperature to generate amine salt. It was washed by distilled water and then purified by silica gel column chromatography.

The ester compound was thus composed of and thereafter added with long chain hydrocarbon amine so as to become the lubricants A1-A14. The long chain hydrocarbon amine was added two times mole than perfluoropolyether, in other words, equal mole to carbonyl group. However, equal mole amine was added to the monofunctional perfluoropolyether.

The composition and molecular weight of the functional groups \mathbb{R}^{f_1} , \mathbb{R}^1 and \mathbb{R} in the general formulas [11] to [15] are described for respective lubricants in the following Table 1.

TABLE 1

No.	R/I	MW	\mathbb{R}^1	R
A1	CF ₂ (OC ₂ F ₄) _p (OCF ₂) _q OCF ₂	2000	C ₁₈ H ₃₇	C ₁₈ H ₃₇
A2	CF ₂ (OC ₂ F ₄) _p (OCF ₂) _q OCF ₂	2000	C ₁₆ H ₃₃	C ₁₆ H ₃₃
A 3	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	2000	C ₁₄ H ₂₉	C ₁₄ H ₂₉
A4	CF ₂ (OC ₂ F ₄) _p (OCF ₂) _q OCF ₂	2000	C ₁₂ H ₂₅	$C_{10}H_{21}$
A 5	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	2000	C ₁₈ H ₃₅	C ₁₈ H ₃₇
A 6	CF ₂ (OC ₂ F ₄) _p (OCF ₂) _q OCF ₂	2000	iso-C ₁₈ H ₃₇	C ₁₈ H ₃₇
A 7	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	4000	C ₁₈ H ₃₁	C ₁₈ H ₃₁
A 8	F(CF ₂ CF ₂ CF ₂ O) _n CF ₂ CF ₂	3500	C ₁₈ H ₃₇	C ₁₈ H ₃₇
A9	F(CF ₂ CF ₂ CF ₂ O) _n CF ₂ CF ₂	3500	C ₁₈ H ₃₇ C ₆ H ₄	C ₁₈ H ₃₇
A10	F(CF ₂ CF ₂ CF ₂ O) _n CF ₂ CF ₂	3500	C ₆ H ₁₁	C ₆ H ₁₁
			(Cyclo hexyl)	(Cyclo hexyl)
A11	F(CF ₂ CF ₂ CF ₂ O) _n CF ₂ CF ₂	3500	C24H49	C ₁₈ H ₃₇ C ₆ H ₄
A12	CF ₃ (OCF(CF ₃)CF ₂) _m (OCF ₂) ₁	650	C ₁₈ H ₃₇	iso-C ₁₈ H ₃₇
A13	$CF_3(OCF(CF_3)CF_2)_m(OCF_2)_1$	650	$CH_2 = CHC_{16}H_{32}$	C ₁₈ H ₃₇
A14	CF ₃ (OCF(CF ₃)CF ₂) _m (OCF ₂) ₁	650	iso-C ₁₈ H ₃₇	$CH_2 = CHC_{16}H_{32}$

can be added to control the roughness of surface to 55 form the back coat layer.

In accordance with this invention, the above lubricant can be added into the back coat layer. Of course, it can be contained by the top coating. Various combination can be also possible for example adding the lubri- 60 cant of the invention into all of the magnetic coating film, metal magnetic thin film and back coat layer in addition to the top coating.

The lubricant of the invention has an excellent lubricant effect to reduce the coefficient of friction. This 65 lubricant effect is not lost under severe condition like low temperature. By esterifying perfluoropolyether, the lubricant becomes easy to dissolve by solvent other than

The above lubricants are coated on the surface of magnetic layer to make the magnetic recording medium. The example will be explained below, which apply to the magnetic recording medium of metal thin film type (evaporated tape).

The cobalt adheres to polyethylene terephthalate film with 14 µm thickness by means of oblique evaporation method so that the ferromagnetic metal thin film are formed having 200 nm thickness. The lubricant A1 shown in Table 1 is dissolved into n-hexane and coated on the surface of the metal magnetic thin film so that the amount of coating can become 5 mg/m². The sample tape of the first example A1 is so made after cutting it to have 8 mm width. The sample tapes of examples A2-A14 are made by the same way as one of the exam-

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ple A1 except for using the lubricants A2-A14 rather than the lubricant A1.

With respect to each sample tape, the coefficient of friction, still durability and shuttle durability are measured respectively at [a] temperature 25 centigrade, 5 humidity 60%, [b] temperature —5 centigrade and [c] temperature 40 centigrade, humidity 80%. The still durability is judged by evaluating the time when the output disappears during a pause mode. The shuttle durability is judged by an evaluation after the repeated playback (shuttle running) for two minutes. In particular by evaluating the numbers of shuttle running up to when the output decreases 3 dB.

These results are shown in Tables 2 and 3.

TABLE 2

IADLE Z					_
	Condition	Coefficient of Friction	Still Dura- bility (Min.)	Shuttle Dura- bility (Times)	- 20
Example A1	25° C., 60% RH	. 0.15	>120	>150	
(Lubricant A1)	40° C., 80% RH	0.19	>120	>150	
	−5° C.	0.16	>120	>150	
Example A2	25° C., 60% RH	0.16	>120	>150	
(Lubricant A2)	40° C., 80% RH	0.20	>120	>150	
	−5° C.	0.17	>120	>150	25
Example A3	25° C., 60% RH	0.16	>120	>150	
(Lubricant A3)	40° C., 80% RH	0.20	>120	>150	
	−5° C.	0.17	>120	>150	
Example A4	25° C., 60% RH	0.16	>120	>150	
(Lubricant A4)	40° C., 80% RH	0.19	> 120	>150	
	−5° C.	0.17	>120	>150	30
Example A5	25° C., 60% RH	0.16	>120	>150	
(Lubricant A5)	40° C., 80% RH	0.21	>120	>150	
	−5° C.	0.17	> 120	>150	
Example A6	25° C., 60% RH	0.17	>120	>150	
(Lubricant A6)	40° C., 80% RH	0.22	>120	>150	
	−5° C.	0.18	>120	>150	35
Example A7	25° C., 60% RH	0.17	> 120	>150	
(Lubricant A7)	40° C., 80% RH	0.22	>120	>150	
	−5° C.	0.17	>120	>150	

TABLE 3					40
	Condition	Coefficient of Friction	Still Dura- bility (Min.)	Shuttle Dura- bility (Times)	_
Example A8	25° C., 60% RH	0.16	>120	>150	45
(Lubricant A8)	40° C., 80% RH	0.22	>120	>150	
	−5° C.	0.17	>120	>150	
Example A9	25° C., 60% RH	0.19	>120	>150	
(Lubricant A9)	40° C., 80% RH	0.23	>120	>150	
	−5° C.	0.19	>120	>150	
Example A10	25° C., 60% RH	0.17	>120	>150	50
(Lubricant A10)	40° C., 80% RH	0.21	>120	>150	
	−5° C.	0.18	>120	>150	
Example A11	25° C., 60% RH	0.16	> 120	>150	
(Lubricant A11)	40° C., 80% RH	0.19	>120	>150	
	−5° C.	0.16	>120	>150	
Example A12	25° C., 60% RH	0.16	>120	>150	55
(Lubricant A12)	40° C., 80% RH	0.20	>120	>150	
	−5° C.	0.16	>120	>150	
Example A13	25° C., 60% RH	0.16	> 120	>150	
(Lubricant A13)	40° C., 80% RH	0.21	>120	>150	
	−5° C.	0.17	> 120	>150	
Example A14	25° C., 60% RH	0.18	>120	>150	60
(Lubricant A14)	40° C., 80% RH	0.23	>120	>150	
	−5° C.	0.18	>120	>150	_

For the purpose of comparison, a sample tape (comparison example A1) is prepared without using any 65 lubricant. The sample tapes (comparison example A2, A3 and A4) are prepared with a lubricant made by perfluoropolyether alone having carboxyl group at its

end (shown by general formulas [20], [21] and [22]). The result of durability measurement is shown in Table 4.

TABLE 4

	Condition	Coefficient of Friction	Still Dura- bility (Min.)	Shuttle Dura- bility (Times)
Comparison	25° C., 60% RH	0.9	2	3
Example A1	40° C., 80% RH			_
(without Lubricant)	−5° C.	<u></u>	—	-
Comparison	25° C., 60% RH	0.20	>120	>150
Example A2	40° C., 80% RH	0.30	91	55
	−5° C.	0.21	51	60
Comparison	25° C., 60% RH	0.24	>120	120
Example A3	40° C., 80% RH	0.32	76	46
_	−5° C.	0.25	29	40
Comparison	25° C., 60% RH	0.20	95	120
Example A4	40° C., 80% RH	0.27	67	35
_	−5° C.	0.21	20	30

The sample tapes as examples of the invention have the lubricant containing ester compound made of perfluoropolyether with carboxyl group at its end and long chain alcohol, which is admixed with long chain alkyl amine. With these sample tapes, the relationship between the added amount of amine and its coefficient of friction is examined.

With respect to the lubricant A1, different amount of amine are added to the ester compound by changing them from 0.01 to 100 mole ratio. The coefficients of friction are measured for these sample tapes. Tables 5 and 6 show the coefficients of friction depending on mole ratio of amine and three different conditions.

TABLE 5

Amine Molar Ratio	Condition	Coefficient of Friction
100	25° C., 60% RH	0.23
	40° C., 80% RH	0.26
	−5° C.	0.24
50	25° C., 60% RH	0.22
	40° C., 80% RH	0.25
	−5° C.	0.23
9	25° C., 60% RH	0.21
	40° C., 80% RH	0.25
	−5° C.	0.22
4	25° C., 60% RH	0.18
	40° C., 80% RH	0.21
	−5° C.	0.19
2.33	25° C., 60% RH	0.17
-	40° C., 80% RH	0.20
	−5° C.	0.17
1.5	25° C., 60% RH	0.16
	40° C., 80% RH	0.19
•	−5° C.	0.16
1	25° C., 60% RH	0.15
	40° C., 80% RH	0.19
	−5° C.	0.16

TABLE 6

Amine Molar Ratio	Condition	Coefficient of Friction
0.67	25° C., 60% RH	0.15
	40° C., 80% RH	0.18
	5° C.	0.15
0.43	25° C., 60% RH	0.15
	40° C., 80% RH	0.18
	−5° C.	0.16
0.25	25° C., 60% RH	0.16
	40° C., 80% RH	0.18
	5° C.	0.16
0.10	25° C., 60% RH	0.16
	40° C., 80% RH	0.19
	−5° C.	0.17

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TABLE 6-continued

Amine Molar Ratio	Condition	Coefficient of Friction
0.05	25° C., 60% RH	0.17
	40° C., 80% RH	0.20
	−5° C.	0.18
0.02	25° C., 60% RH	0.18
	40° C., 80% RH	0.21
	−5° C.	0.18
0.01	25° C., 60% RH	0.19
	40° C., 80% RH	0.23
	−5° C.	0.20
0	25° C., 60% RH	0.29
	40° C., 80% RH	0.43
	−5° C.	0.29

Similarly, with respect to the lubricant AS, different amount of amine are added to the ester compound by changing them from 0.01 to 100 mole ratio. The coefficients of friction are measured for these sample tapes. Tables 7 and 8 show the coefficients of friction depending on mole ratio of amine and three different conditions.

TABLE 7

Amine Molar Ratio	Condition	Coefficient of Friction
100	25° C., 60% RH	0.21
	40° C., 80% RH	0.22
	−5° C.	0.21
50	25° C., 60% RH	0.20
	40° C., 80% RH	0.21
	−5° C.	0.21
9	25° C., 60% RH	0.19
	40° C., 80% RH	0.21
	−5° C.	0.20
4	25° C., 60% RH	0.19
	40° C., 80% RH	0.21
	−5° C.	0.19
2.33	25° C., 60% RH	0.18
	40° C., 80% RH	0.21
•	−5° C.	0.19
1.5	25° C., 60% RH	0.17
	40° C., 80% RH	0.21
	−5° C.	0.18
1	25° C., 60% RH	0.16
	40° C., 80% RH	0.22
	−5° C.	0.17

TABLE 8

Amine Molar Ratio	Condition	Coefficient of Friction	
0.67	25° C., 60% RH	0.16	_ 5
	40° C., 80% RH	0.20	
	−5° C.	0.17	
0.43	25° C., 60% RH	0.17	
	40° C., 80% RH	0.20	
	−5° C.	0.17	
0.25	25° C., 60% RH	0.17	5
	40° C., 80% RH	0.21	
	−5° C.	0.18	
0.10	25° C., 60% RH	0.18	
	40° C., 80% RH	0.21	
	−5° C.	0.18	
0.05	25° C., 60% RH	0.19	6
	40° C., 80% RH	0.21	
	−5° C.	0.19	
0.02	25° C., 60% RH	0.19	
	40° C., 80% RH	0.21	
	−5° C.	0.19	
0.01	25° C., 60% RH	0.19	6
•	40° C., 80% RH	0.22	
	−5° C.	0.20	

The lubricants have been compared between the perfluoropolyether alone with carboxyl group at its end and the ester compound of perfluoropolyether with carboxyl group and long chain alcohol admixed with long chain hydrocarbon amine of from 0.01 to 100 mole ratio. It was found that the latter could obtain the better result under various conditions such as friction coefficient, still durability and shuttle durability.

The solubility has been also considered with respect to the ester compound of perfluoropolyether with carboxyl group at its both ends and long chain alcohol (utilized as the lubricant A1). The ester is soluble in toluene, hexane as well as freon. As the long chain hydrocarbon amine is also soluble in toluene and hexane the lubricant A1 is soluble in toluene and hexane.

Therefore, the ester compound of perfluoropolyether with carboxyl group at its end and long chain alcohol admixed with long chain hydrocarbon amine can be utilized as a lubricant with the excellent lubricant effect without using freon.

The identification for the composed ester of perfluoropolyether with carboxyl group at the end and long chain alcohol has been examined by the infrared absorption spectrum. The stretching vibration of the hydroxyl group of 3600 cm⁻¹ disappeared. The stretching vibration of the CH can be found at 2920 cm⁻¹ and 2850 cm⁻¹. The stretching vibration of CO moves from 1760 cm⁻¹ to 1800 cm⁻¹. And the stretching vibration of the CF can be found from 1300 cm⁻¹ to 1060 cm⁻¹. So, the structure can be identified by those measurement.

EXPERIMENTATION B

The lubricant of the second embodiment contains ester compound made of perfluoropolyether having hydroxyl group at its end and partially fluorined long chain carboxylic acid admixed with long chain hydrocarbon amine.

The ester compound is composed of a solution of perfluoropolyether with hydroxyl group at its end and third grade amine added by dropping partially fluorined long chain carboxylic acid chloride. The reaction has been completed under reflux and heat after dropping. The solution was then cooled to the room temperature. The generated amine salt is washed by distilled water and then purified by silica gel column chromatography.

The partially fluorined long chain carboxylic acid chloride can be obtained by the method for example proposed by Brace and others (see, Journal of Organic Chemistry, Vol. 27, pp. 4491). By using chlorination agent such as thionyl chloride, the compound of partially fluorined long chain carboxylic acid can become acid chloride.

The ester compound was thus composed of and thereafter added with long chain hydrocarbon amine so as to become the lubricants B1-B14. The long chain hydrocarbon amine was added two times mole than two functional perfluoropolyether and equal to monofunctional perfluoropolyether, in other words, equal mole to carbonyl group.

The composition and molecular weight of the functional groups R/2, R² and R in the general formulas [12] and [15] are described for respective lubricants in the following Table 9.

TABLE 9

No.	R/2	MW	R ²	R	
B1	CF ₂ (OC ₂ F ₄) _p (OCF ₂) _q OCF ₂	2000	C ₈ F ₁₇ C ₁₀ H ₂₀	C ₁₈ H ₃₇	
B 2	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	2000	$C_7F_{15}C_{10}H_{20}$	C ₁₆ H ₃₃	
B 3	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	2000	$C_{12}F_{25}C_{10}H_{20}$	C ₁₄ H ₂₉	
B 4	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	2000	$C_2F_5C_{10}H_{20}$	$C_{10}H_{21}$	
B 5	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	2000	iso-C ₃ F ₇ C ₁₀ H ₂₀	$C_{18}H_{37}$	
B 6	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	2000	iso-C ₃ F ₇ C ₉ H ₁₈	$C_{18}H_{37}$	
B 7	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	4000	$C_{12}F_{25}C_6H_{12}$	$C_{18}H_{31}$	
B 8	F(CF ₂ CF ₂ CF ₂ O) _n CF ₂ CF ₂	3500	$C_3F_7CH=CHC_8H_{16}$	C ₁₈ H ₃₇	
B 9	F(CF ₂ CF ₂ CF ₂ O) _n CF ₂ CF ₂	3500	C7F15C16H32	C ₁₈ H ₃₇	
B10	F(CF ₂ CF ₂ CF ₂ O) _n CF ₂ CF ₂	3500	$(CF_3)_2CFC_6F_{12}C_{16}H_{32}$	C ₆ H ₁₁	
			•	(Cyclo hexyl)	
B11	F(CF ₂ CF ₂ CF ₂ O) _n CF ₂ CF ₂	3500	$C_8F_{17}C_{10}H_{20}$	C ₁₈ H ₃₇ C ₆ H ₄	
B12	$CF_3(OCF(CF_3)CF_2)_m(OCF_2)_I$	650	$C_8F_{17}C_{10}H_{20}$	iso-C ₁₈ H ₃₇	
B13	$CF_3(OCF(CF_3)CF_2)_m(OCF_2)_I$	650	$C_3F_7CH=CHC_8H_{16}$	C ₁₈ H ₃₇	
B14	$CF_3(OCF(CF_3)CF_2)_m(OCF_2)_l$	650	C ₇ F ₁₅ C ₁₆ H ₃₂	CH2=CHC16H32	

With these lubricants B1 to B14, the sample tapes as examples B1 to B14 are prepared by the same method as explained in the experimentation A. Regarding each sample tape, the friction coefficient, still and shuttle durabilities are measured by the same method as explained in the experimentation A. The results are shown in Tables 10 and 11.

TABLE 10

					_ 2
	Condition	Coefficient of Friction	Still Dura- bility (Min.)	Shuttle Dura- bility (Times)	
Example B1 (Lubricant B1)	25° C., 60% RH 40° C., 80% RH -5° C.	0.15 0.19 0.15	>120 >120 >120 >120	>150 >150 >150	3

TABLE 10-continued

	Condition	Coefficient of Friction	Still Dura- bility (Min.)	Shuttle Dura- bility (Times)
	−5° C.	0.18	>120	>150
Example B5	25° C., 60% RH	0.16	>120	>150
(Lubricant B5)	40° C., 80% RH	0.22	>120	>150
	−5° C.	0.17	>120	>150
Example B6	25° C., 60% RH	0.16	>120	>150
(Lubricant B6)	40° C., 80% RH	0.22	>120	>150
	−5° C.	0.18	>120	>150
Example B7	25° C., 60% RH	0.17	>120	>150
(Lubricant B7)	40° C., 80% RH	0.23	>120	>150
	−5° C.	0.17	>120	>150

TABLE 11

	Condition	Coefficient of Friction	Still Durability (Min.)	Shuttle Durability (Times)
Example B8	25° C., 60% RH	0.16	>120	>150
(Lubricant B8)	40° C., 80% RH	0.22	>120	>150
•	−5° C.	0.17	>120	>150
Example B9	25° C., 60% RH	0.16	>120	>150
(Lubricant B9)	40° C., 80% RH	0.19	>120	>150
	5° C.	0.16	>120	>150
Example B10	25° C., 60% RH	0.17	>120	>150
(Lubricant B10)	40° C., 80% RH	0.22	>120	>150
•	−5° C.	0.18	>120	>150
Example B11	25° C., 60% RH	0.17	>120	>150
(Lubricant B11)	40° C., 80% RH	0.20	>120	>150
	−5° C.	0.17	>120	>150
Example B12	25° C., 60% RH	0.17	>120	>150
(Lubricant B12)	40° C., 80% RH	0.21	>120	>150
	−5° C.	0.18	>120	>150
Example B13	25° C., 60% RH	0.17	>120	>150
(Lubricant B13)	40° C., 80% RH	0.22	>120	>150
	5° C.	0.18	>120	>150
Example B14	25° C., 60% RH	0.17	>120	>150
(Lubricant B14)	40° C., 80% RH	0.21	>120	>150
	5° C.	0.17	>120	>150

Example B2	25° C., 60% RH	0.16	>120	>150
(Lubricant B2)	40° C., 80% RH	0.20	>120	>150
	−5° C.	0.16	>120	>150
Example B3	25° C., 60% RH	0.16	>120	>150
(Lubricant B3)	40° C., 80% RH	0.20	>120	>150
	−5° C.	0.17	>120	>150
Example B4	25° C., 60% RH	0.17	>120	>150
(Lubricant B4)	40° C., 80% RH	0.22	>120	>150

For the purpose of comparison, a sample tape (comparison example B1) is prepared without using any lubricant. The sample tapes (comparison example B2, B3 and B4) are prepared with lubricants made by perfluoropolyether alone having hydroxyl group at its end (shown in general formula [17], [18] and [19]). The result of durability measurement is shown in Table 12.

TABLE 12

	Condition	Coefficient of Friction	Still Durability (Min.)	Shuttle Durability (Times)
Comparison	25° C., 60% RH	0.9	2	3
Example B1	40° C., 80% RH	41-211111-		
(without	−5° C.			

TABLE 12-continued

	Condition	Coefficient of Friction	Still Durability (Min.)	Shuttle Durability (Times)
Lubricant)				
Comparison	25° C., 60% RH	0.20	>120	>150
Example B2	40° C., 80% RH	0.30	91	55
	5° C.	0.21	51	60
Comparison	25° C., 60% RH	0.24	>120	120
Example B3	40° C., 80% RH	0.32	76	46
	−5° C.	0.25	29	40
Comparison	25° C., 60% RH	0.20	95	120
Example B4	40° C., 80% RH	0.27	67	35
	−5° C.	0.21	20	30

The sample tapes as examples of the invention have the lubricant containing ester compound made of perfluoropolyether with hydroxyl group at its end and partially fluorined long chain carboxylic acid admixed with long chain hydrocarbon amine. With these sample 20 tapes, the relationship between the added amount of amine and its coefficient of friction is examined.

With respect to the lubricant B1, different amount of amine are added to the ester compound by changing them from 0.01 to 100 mole ratio. The coefficients of 25 friction are measured for these sample tapes. Tables 13 and 14 show the coefficients of friction depending on mole ratio of amine and three different conditions.

TARLE 13

Amine Molar Ratio	Condition	Coefficient of Friction	2
100	25° C., 60% RH	0.23	
	40° C., 80% RH	0.26	
	−5° C.	0.24	
50	25° C., 60% RH	0.22	3
	40° C., 80% RH	0.25	•
•	5° C.	0.23	
9	25° C., 60% RH	0.21	
	40° C., 80% RH	0.25	
	−5° C.	0.22	
4	25° C., 60% RH	0.18	4
	40° C., 80% RH	0.21	•
•	−5° C.	0.19	
2.33	25° C., 60% RH	0.17	
	40° C., 80% RH	0.20	
	−5° C.	0.17	
1.5	25° C., 60% RH	0.16	
	40° C., 80% RH	0.19	4
	−5° C.	0.16	
1	25° C., 60% RH	0.15	
	40° C., 80% RH	0.19	
	−5° C.	0.15	

TABLE 14

Coefficient of

Amine Molar Ratio	Condition	Friction
0.67	25° C., 60% RH	0.15
	40° C., 80% RH	0.18
	5° C.	0.15
0.43	25° C., 60% RH	0.15
	40° C., 80% RH	0.18
	−5° C.	0.16
0.25	25° C., 60% RH	0.16
	40° C., 80% RH	0.18
·	−5° C.	0.16
0.10	25° C., 60% RH	0.16
	40° C., 80% RH	0.19
	−5° C.	0.17
0.05	25° C., 60% RH	0.17
	40° C., 80% RH	0.20
	−5° C.	0.18
0.02	25° C., 60% RH	0.18
	40° C., 80% RH	0.21
	−5° C.	0.18

.

TABLE 14-continued

Amine Molar Ratio	Condition	Coefficient of Friction
0.01	25° C., 60% RH	0.19
	40° C., 80% RH	0.23
	5° C.	0.20
0	25° C., 60% RH	0.29
	40° C., 80% RH	0.43
	−5° C.	0.29

Similarly, with respect to the lubricant B11, different amount of amine are added to the ester compound by changing them from 0.01 to 100 mole ratio. The coefficients of friction are measured for these sample tapes. 30 Tables 15 and 16 show the coefficients of friction depending on mole ratio of amine and three different conditions.

TABLE 15

35	Amine Molar Ratio	Condition	Coefficient of Friction
	100	25° C., 60% RH	0.21
		40° C., 80% RH	0.22
		−5° C.	0.21
	50	25° C., 60% RH	0.20
40		40° C., 80% RH	0.21
T U		−5° C.	0.21
	. 9	25° C., 60% RH	0.19
		40° C., 80% RH	0.21
		−5° C.	0.20
	4	25° C., 60% RH	0.19
4.5		40° C., 80% RH	0.21
45		5° C.	0.19
	2.33	25° C., 60% RH	0.18
		40° C., 80% RH	0.21
		−5° C.	0.19
	1.5	25° C., 60% RH	0.17
		40° C., 80% RH	0.21
50		−5° C.	0.18
	1	25° C., 60% RH	0.17
		40° C., 80% RH	0.20
		−5° C.	0.17

55	TABLE 16				
	Amine Molar Ratio	Condition	Coefficient of Friction		
	0.67	25° C., 60% RH	0.16		
60	•	40° C., 80% RH	0.20		
60		5° C.	0.17		
	0.43	25° C., 60% RH	0.17		
		40° C., 80% RH	0.20		
		−5° C.	0.17		
	0.25	25° C., 60% RH	0.17		
		40° C., 80% RH	0.21		
65		−5° C.	0.18		
	0.10	25° C., 60% RH	0.18		
		40° C., 80% RH	0.21		
		−5° C.	0.18		
	0.05	25° C., 60% RH	0.19		

TABLE 16-continued

Amine Molar Ratio	Condition	Coefficient of Friction
	40° C., 80% RH	0.21
	−5° C.	0.19
0.02	25° C., 60% RH	0.19
	40° C., 80% RH	0.21
	−5° C.	0.19
0.01	25° C., 60% RH	0.19
	40° C., 80% RH	0.22
	5° C.	0.20

The lubricants have been compared between the perfluoropolyether alone with hydroxyl group at its end and the ester compound of perfluoropolyether with hydroxyl group and partially fluorined long chain carboxylic acid admixed with long chain hydrocarbon amine of from 0.01 to 100 mole ratio. It was found that the latter could obtain the better result under various conditions such as friction coefficient, still durability and shuttle durability.

The solubility has been also considered with respect to the ester compound of perfluoropolyether with hydroxyl group at its both ends and partially fluorined long chain carboxylic acid (utilized as the lubricant B1). 25

The ester compound is soluble in hexane as well as freon. As the long chain hydrocarbon amine is also soluble in hexane, the lubricant B1 is soluble in hexane.

EXPERIMENTATION C

The lubricant of the third embodiment contains ester compound made of perfluoropolyether having carboxyl group at its end and partially fluorined long chain alcohol admixed with long chain hydrocarbon amine.

The ester compound is composed of perfluoropolyether with carboxyl group at its end and partially fluorined long chain alcohol with p-toluensul-10 fonic acid or concentrated sulfuric acid as a catalyst. The reaction has been completed under reflux and heat in anhydrous toluene by removing the generated water. The solution was then cooled to the room temperature. After removing toluene, it is purified by silica gel col-15 umn chromatography.

The partially fluorined long chain alcohol can be obtained by the method for example proposed by Brace and others (see, Journal of Organic Chemistry, Vol. 27, pp. 4491). By using chlorination agent such as thionyl chloride, the compound of partially fluorined long chain carboxylic acid can become acid chloride which is thereafter reduced.

The ester compound was thus composed of and thereafter added with long chain hydrocarbon amine so as to become the lubricants C1-C14. The composition and molecular weight of the functional groups R/3, R³ and R in the general formulas [5] and [10] are described for respective lubricants in the following Table 17.

TABLE 17

No.	R ^{f2}	MW	R ³	R
C1	CF ₂ (OC ₂ F ₄) _p (OCF ₂) _q OCF ₂	2000	C ₈ F ₁₇ C ₁₀ H ₂₀	C ₁₈ H ₃₇
C2	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	2000	$C_7F_{15}C_{10}H_{20}$	C ₁₆ H ₃₃
C 3	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	2000	$C_{12}F_{25}C_{10}H_{20}$	C ₁₄ H ₂₉
C4	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	2000	$C_2F_5C_{10}H_{20}$	$C_{10}H_{21}$
C 5	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	2000	iso—C ₃ F ₇ C ₁₀ H ₂₀	C ₁₈ H ₃₇
C 6	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	2000	iso-C ₃ F ₇ C ₉ H ₁₈	C ₁₈ H ₃₇
C7	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	4000	$C_{12}F_{25}C_6H_{12}$	$C_{18}H_{31}$
C 8	F(CF ₂ CF ₂ CF ₂ O) _n CF ₂ CF ₂	3500	$C_3F_7CH=CHC_8H_{16}$	C ₁₈ H ₃₇
C 9	F(CF ₂ CF ₂ CF ₂ O) _n CF ₂ CF ₂	3500	C7F15C16H32	C ₁₈ H ₃₇
C10	F(CF ₂ CF ₂ CF ₂ O) _n CF ₂ CF ₂	3500	$(CF_3)_2CFC_6F_{12}C_{16}H_{32}$	C ₆ H ₁₁
				(Cyclo hexyl)
Cll	F(CF ₂ CF ₂ CF ₂ O) _n CF ₂ CF ₂	3500	$C_8F_{17}C_{10}H_{20}$	C ₁₈ H ₃₇ C ₆ H ₄
C12	$CF_3(OCF(CF_3)CF_2)_m(OCF_2)_1$	650	C ₈ F ₁₇ C ₁₀ H ₂₀	iso-C ₁₈ H ₃₇
C13	$CF_3(OCF(CF_3)CF_2)_m(OCF_2)_1$	650	C ₃ F ₇ CH=CHC ₈ H ₁₆	C ₁₈ H ₃₇
C14	$CF_3(OCF(CF_3)CF_2)_m(OCF_2)_1$	650	C7F15C16H32	$CH_2 = CHC_{16}H_{32}$

Therefore, the ester compound of perfluoropolyether with hydroxyl group at its end and partially fluorined long chain carboxylic acid admixed with long chain hydrocarbon amine can be utilized as a lubricant with the excellent lubricant effect without using freon.

With these lubricants C1 to C14, the sample tapes as examples C1 to C14 are prepared by the same method as explained in the experimentation A. Regarding each sample tape, the friction coefficient, still and shuttle durabilities are measured by the same method as explained in the experimentation A. The results are shown in Tables 18 and 19.

TABLE 18

	· -			
	Condition	Coefficient of Friction	Still Durability (Min.)	Shuttle Durability (Times)
Example C1	25° C., 60% RH	. 0.16	>120	>150
(Lubricant C1)	40° C., 80% RH	0.19	>120	>150
	−5° C.	0.16	>120	>150
Example C2	25° C., 60% RH	0.16	>120	>150
(Lubricant C2)	40° C., 80% RH	0.20	>120	>150
	−5° C.	0.17	>120	>150
Example C3	25° C., 60% RH	0.17	>120	>150
(Lubricant C3)	40° C., 80% RH	0.20	>120	>150
	−5° C.	0.17	>120	>150
Example C4	25° C., 60% RH	0.18	>120	>150
(Lubricant C4)	40° C., 80% RH	0.22	>120	>150
	−5° C.	0.18	>120	>150
Example C5	25° C., 60% RH	0.17	>120	>150
(Lubricant C5)	40° C., 80% RH	0.22	>120	>150
•	5° C.	0.17	>120	>150

TABLE 18-continued

	Condition	Coefficient of Friction	Still Durability (Min.)	Shuttle Durability (Times)
Example C6	25° C., 60% RH	0.17	>120	>150
(Lubricant C6)	40° C., 80% RH	0.23	>120	>150
	−5° C.	0.17	>120	>150
Example C7	25° C., 60% RH	0.17	>120	>150
(Lubricant C7)	40° C., 80% RH	0.24	>120	>150
	−5° C.	0.17	>120	>150

TABLE 19

	Condition	Coefficient of Friction	Still Durability (Min.)	Shuttle Durability (Times)
Example C8	25° C., 60% RH	0.17	>120	>150
(Lubricant C8)	40° C., 80% RH	0.22	>120	>150
	−5° C.	0.17	>120	>150
Example C9	25° C., 60% RH	0.17	>120	>150
(Lubricant C9)	40° C., 80% RH	0.20	>120	>150
	−5° C.	0.17	>120	>150
Example C10	25° C., 60% RH	0.18	>120	>150
(Lubricant C10)	40° C., 80% RH	0.23	>120	>150
	−5° C.	0.18	>120	>150
Example C11	25° C., 60% RH	0.17	>120	>150
(Lubricant C11)	40° C., 80% RH	0.22	>120	>150
	−5° C.	0.17	>120	>150
Example C12	25° C., 60% RH	0.18	>120	>150
(Lubricant C12)	40° C., 80% RH	0.22	>120	>150
	−5° C.	0.18	>120	>150
Example C13	25° C., 60% RH	0.17	>120	>150
(Lubricant C13)	40° C., 80% RH	0.23	>120	>150
	−5° C.	0.18	>120	>150
Example C14	25° C., 60% RH	0.18	>120	>150
(Lubricant C14)	40° C., 80% RH	0.23	>120	>150
	−5° C.	0.18	>120	>150

For the purpose of comparison, a sample tape (com- 35 and 22 show the coefficients of friction depending on parison example C1) is prepared without using any mole ratio of amine and three different conditions. lubricant. The sample tapes (comparison example C2, C3 and C4) are prepared with lubricants made by perfluoropolyether alone having carboxyl group at its end (shown in general formula [20], [21] and [22]). The result of durability measurement is shown in Table 20.

TABLE 21

Amine Molar Ratio	Condition	Coefficient of Friction
100	25° C., 60% RH	0.23
	40° C., 80% RH	0.26

TABLE 20

	Condition	Coefficient of Friction	Still Durability (Min.)	Shuttle Durability (Times)
Comparison	25° C., 60% RH	0.9	2	3
Example C1	40° C., 80% RH		<u></u>	
(without Lubricant)	–5° C.	<u></u>		
Comparison	25° C., 60% RH	0.24	>120	>150
Example C2	40° C., 80% RH	0.27	91	55
_	−5° C.	0.25	51	60
Comparison	25° C., 60% RH	0.22	>120	120
Example C3	40° C., 80% RH	0.23	76	46
_	−5° C.	0.22	29	40
Comparison	25° C., 60% RH	0.24	95	120
Example C4	40° C., 80% RH	0.26	67	35
_	−5° C.	0.25	20	30

The sample tapes as examples of the invention have the lubricant containing ester compound made of perfluoropolyether with carboxyl group at its end and 60 partially fluorined long chain alcohol admixed with long chain hydrocarbon amine. With these sample tapes, the relationship between the added amount of amine and its coefficient of friction is examined.

With respect to the lubricant C1, different amount of 65 amine are added to the ester compound by changing them from 0.01 to 100 mole ratio. The coefficients of friction are measured for these sample tapes. Tables 21

	5° C.	0.24
50	25° C., 60% RH	0.22
	40° C., 80% RH	0.25
	−5° C.	0.22
9	25° C., 60% RH	0.21
	40° C., 80% RH	0.25
	−5° C.	0.21
4	25° C., 60% RH	0.19
	40° C., 80% RH	0.23
	−5° C.	0.19
2.33	25° C., 60% RH	0.18
	40° C., 80% RH	0.22
	−5° C.	0.18

Amine Molar Ratio

Coefficient of

Friction

TABLE 21-continued

Amine Molar Ratio	Condition	Coefficient of Friction	
1.5	25° C., 60% RH	0.17	
	40° C., 80% RH	0.21	
	−5° C.	0.17	
1	25° C., 60% RH	0.16	
	40° C., 80% RH	0.19	
	−5° C.	0.16	

Condition	Coefficient of Friction	_
25° C., 60% RH	0.16	_
40° C., 80% RH	0.19	
−5° C.	0.16	
25° C., 60% RH	0.16	
40° C., 80% RH	0.18	
−5° C.	0.16	
25° C., 60% RH	0.16	
40° C., 80% RH	0.19	
−5° C.	0.16	
25° C., 60% RH	0.17	
40° C., 80% RH	0.19	
−5° C.	0.17	
25° C., 60% RH	0.17	
40° C., 80% RH	0.20	
−5° C.	0.18	
25° C., 60% RH	0.18	
40° C., 80% RH	0.22	
− 5° C.	0.19	ı
25° C., 60% RH	0.22	
40° C., 80% RH	0.25	
5° C.	0.22	
25° C., 60% RH	0.24	
	25° C., 60% RH 40° C., 80% RH -5° C. 25° C., 60% RH 40° C., 80% RH -5° C. 25° C., 60% RH 40° C., 80% RH -5° C. 25° C., 60% RH 40° C., 80% RH -5° C. 25° C., 60% RH 40° C., 80% RH -5° C. 25° C., 60% RH 40° C., 80% RH -5° C. 25° C., 60% RH 40° C., 80% RH -5° C.	Condition Friction 25° C., 60% RH 0.16 40° C., 80% RH 0.19 -5° C. 0.16 25° C., 60% RH 0.16 40° C., 80% RH 0.16 25° C., 60% RH 0.16 40° C., 80% RH 0.19 -5° C. 0.16 25° C., 60% RH 0.17 40° C., 80% RH 0.17 40° C., 80% RH 0.17 40° C., 80% RH 0.18 40° C., 80% RH 0.18 40° C., 80% RH 0.22 -5° C. 0.19 25° C., 60% RH 0.22 40° C., 80% RH 0.25 -5° C. 0.19

Similarly, with respect to the lubricant C11, different amount of amine are added to the ester compound by changing them from 0.01 to 100 mole ratio. The coefficients of friction are measured for these sample tapes. Tables 23 and 24 show the coefficients of friction depending on mole ratio of amine and three different conditions.

40° C., 80% RH

−5° C.

0.27

0.25

45

TABLE 23

Amine Molar Ratio	Condition	Coefficient of Friction	
100	25° C., 60% RH	0.21	
•	40° C., 80% RH	0.23	5
	5° C.	0.21	
50	25° C., 60% RH	0.21	
	40° C., 80% RH	0.23	
	−5° C.	0.21	
9	25° C., 60% RH	0.20	_
	40° C., 80% RH	0.23	3
	−5° C.	0.20	
4	25° C., 60% RH	0.19	
	40° C., 80% RH	0.23	
	5° C.	0.19	
2.33	25° C., 60% RH	0.18	6
	40° C., 80% RH	0.23	`
	5° C.	0.19	
1.5	25° C., 60% RH	0.17	
	40° C., 80% RH	0.23	
	5° C.	0.18	
1	25° C., 60% RH	0.17	6
	40° C., 80% RH	0.22	
	−5° C.	0.17	

TABLE 24

Condition

5	0.67	25° C., 60% RH	0.17	
J		40° C., 80% RH	0.21	
		−5° C.	0.17	
	0.43	25° C., 60% RH	0.17	
		40° C., 80% RH	0.22	
		5° C .	0.17	
0	0.25	25° C., 60% RH	0.18	
·		40° C., 80% RH	0.23	
		5° C.	0.18	
	0.10	25° C., 60% RH	0.19	
		40° C., 80% RH	0.23	
		−5° C.	0.19	
5	0.05	25° C., 60% RH	0.19	
.5		40° C., 80% RH	0.23	
		−5° C.	0.19	
	0.02	25° C., 60% RH	0.20	
		40° C., 80% RH	0.24	
		−5° C.	0.21	
	0.01	25° C., 60% RH	0.22	
20		40° C., 80% RH	0.26	
		−5° C .	0.23	

The lubricants have been compared between the perfluoropolyether alone with carboxyl group at its end and the ester compound of perfluoropolyether with carboxyl group and partially fluorined long chain alcohol admixed with long chain hydrocarbon amine of from 0.01 to 100 mole ratio. It was found that the latter could obtain the better result under various conditions such as friction coefficient, still durability and shuttle durability.

The solubility has been also considered with respect to the ester compound of perfluoropolyether with carboxyl group at its both ends and partially fluorined long chain alcohol (utilized as the lubricant C1).

The ester compound is soluble in hexane as well as freon. As the long chain hydrocarbon amine is also soluble in hexane, the lubricant C1 is soluble in hexane.

Therefore, the ester compound of perfluoropolyether with carboxyl group at its end and partially fluorined long chain alcohol admixed with long chain hydrocarbon amine can be utilized as a lubricant with the excellent lubricant effect without using freon.

EXPERIMENTATION D

The lubricant of the forth embodiment contains ester compound of perfluoropolyether having hydroxyl group at its end and perfluoro carboxylic acid admixed with long chain hydrocarbon amine.

The ester compound is composed of a solution of perfluoropolyether with hydroxyl group at its end and third grade amine added by dropping perfluoro carboxylic acid chloride. The reaction has been completed under reflux and heat after dropping. The solution was then cooled to the room temperature. The generated amine salt is washed by distilled water and then purified by silica gel column chromatography.

The perfluoro carboxylic acid chloride can be obtained by using chlorination agent such as thionyl chloride and catalyst such as dimethylformamide. The perfluoro carboxylic acid can easily become acid chloride.

The ester compound was thus composed of and thereafter added with long chain hydrocarbon amine so as to become the lubricants D1-D14. The composition and molecular weight of the functional groups R/4, R/5 and R in the general formulas [14] and [15] are described for respective lubricants in the following Table 25.

TABLE 25

No.	R/I	MW	R ²	R	
D1	CF ₂ (OC ₂ F ₄) _p (OCF ₂) _q OCF ₂	2000	C ₈ F ₁₇	C ₁₈ H ₃₇	
D2	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	2000	C7F15	C ₁₆ H ₃₃	
D3	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	2000	$C_{12}F_{25}$	C ₁₄ H ₂₉	
D 4	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	2000	C9F19	$C_{10}H_{21}$	
\mathbf{D} 5	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	2000	C ₆ F ₁₃	C ₁₈ H ₃₇	
$\mathbf{D}6$	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	2000	C ₉ F ₁₉	C ₁₈ H ₃₇	
$\mathbf{D}7$	$CF_2(OC_2F_4)_p(OCF_2)_qOCF_2$	4000	C ₁₂ F ₂₅	C ₁₈ H ₃₁	
$\mathbf{D}8$	F(CF ₂ CF ₂ CF ₂ O) _n CF ₂ CF ₂	3500	C ₈ F ₁₇	C ₁₈ H ₃₇	
$\mathbf{D}9$	F(CF ₂ CF ₂ CF ₂ O) _n CF ₂ CF ₂	3500	C ₉ F ₁₉	C ₁₈ H ₃₇	
D 10	F(CF ₂ CF ₂ CF ₂ O) _n CF ₂ CF ₂	3500	$(CF_3)_2CFC_6F_{12}$	C ₆ H ₁₁	
			. ,,,	(Cyclo hexyl)	
D 11	F(CF ₂ CF ₂ CF ₂ O) _n CF ₂ CF ₂	3500	C ₇ F ₁₅	C ₁₈ H ₃₇ C ₆ H ₄	
D 12	$CF_3(OCF(CF_3)CF_2)_m(OCF_2)_1$	650	C ₈ F ₁₇	iso—C ₁₈ H ₃₇	
$\mathbf{D}13$	$CF_3(OCF(CF_3)CF_2)_m(OCF_2)_1$	650	C7F15	C ₁₈ H ₃₇	
D 14	$CF_3(OCF(CF_3)CF_2)_m(OCF_2)_1$	650	C ₉ F ₁₉	$CH_2 = CHC_{16}H_{32}$	

With these lubricants D1 to D14, the sample tapes as examples D1 to D14 are prepared by the same method

plained in the experimentation A. The results are shown in Tables 26 and 27.

TABLE 26

	Condition	Coefficient of Friction	Still Durability (Min.)	Shuttle Durability (Times)
Example D1	25° C., 60% RH	0.17	>120	>150
(Lubricant D1)	40° C., 80% RH	0.23	>120	>150
	−5° C.	0.17	>120	>150
Example D2	25° C., 60% RH	0.17	>120	>150
(Lubricant D2)	40° C., 80% RH	0.23	>120	>150
	−5° C.	0.18	>120	>150
Example D3	25° C., 60% RH	0.18	>120	>150
(Lubricant D3)	40° C., 80% RH	0.23	>120	>150
	−5° C.	0.18	>120	>150
Example D4	25° C., 60% RH	0.18	>120	>150
(Lubricant D4)	40° C., 80% RH	0.23	>120	>150
	−5° C.	0.18	>120	>150
Example D5	25° C., 60% RH	0.17	>120	>150
(Lubricant D5)	40° C., 80% RH	0.22	>120	>150
	−5° C.	0.17	>120	>150
Example D6	25° C., 60% RH	0.16	>120	~ >150
(Lubricant D6)	40° C., 80% RH	0.22	>120	>150
	−5° C.	0.17	>120	>150
Example D7	25° C., 60% RH	0.17	>120	>150
(Lubricant D7)	40° C., 80% RH	0.23	>120	>150
	−5° C.	0.18	>120	>150

TABLE 27

	Condition	Coefficient of Friction	Still Durability (Min.)	Shuttle Durability (Times)
Example D8	25° C., 60% RH	0.17	>120	>150
(Lubricant D8)	40° C., 80% RH	0.22	>120	>150
	−5° C.	0.17	>120	>150
Example D9	25° C., 60% RH	0.17	>120	>150
(Lubricant D9)	40° C., 80% RH	0.22	>120	>150
	5° C.⁻	0.17	>120	>150
Example D10	25° C., 60% RH	0.19	>120	>150
(Lubricant D10)	40° C., 80% RH	0.24	>120	>150
	−5° C.	0.20	>120	>150
Example D11	25° C., 60% RH	0.17	>120	>150
(Lubricant D11)	40° C., 80% RH	0.22	>120	>150
	−5° C.	0.17	>120	>150
Example D12	25° C., 60% RH	0.19	>120	>150
(Lubricant D12)	40° C., 80% RH	0.24	>120	>150
	−5° C.	0.19	>120	>150
Example D13	25° C., 60% RH	0.17	>120	>150
(Lubricant D13)	40° C., 80% RH	0.22	>120	>150
	−5° C.	0.18	>120	>150
Example D14	25° C., 60% RH	0.18	>120	>150
(Lubricant D14)		0.23	>120	>150
	−5° C.	0.19	>120	>150

as explained in the experimentation A. Regarding each sample tape, the friction coefficient, still and shuttle durabilities are measured by the same method as ex-

For the purpose of comparison, a sample tape (comparison example D1) is prepared without using any lubricant. The sample tapes (comparison example D2, D3 and D4) are prepared with lubricants made by per-

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fluoropolyether alone having hydroxyl group at its end (shown in general formula [17], [18] and [19]). The result of durability measurement is shown in Table 28.

TABLE 30-continued

		Coefficient of
Amine Molar Ratio	Condition	Friction

		TABLE 28		
	Condition	Coefficient of Friction	Still Durability (Min.)	Shuttle Durability (Times)
Comparison	25° C., 60% RH	0.9	2	3
Example D1	40° C., 80% RH		_	
(without Lubricant)	5° C.	<u></u> .		
Comparison	25° C., 60% RH	0.20	>120	>150
Example D2	40° C., 80% RH	0.30	91	55
	−5° C.	0.21	51	60
Comparison	25° C., 60% RH	0.24	>120	120
Example D3	40° C., 80% RH	0.32	76	46
	−5° C.	0.25	29	40
Comparison	25° C., 60% RH	0.20	95	120
Example D4	40° C., 80% RH	0.27	67	35

0.21

20

The sample tapes as examples of the invention have the lubricant containing ester compound made of perfluoropolyether admixed with long chain hydrocarbon amine. With these sample tapes, the relationship between the added amount of amine and its coefficient of 25 friction is examined.

−5° C.

With respect to the lubricant D1, different amount of amine are added to the ester compound by changing them from 0.01 to 100 mole ratio. The coefficients of friction are measured for these sample tapes. Tables 29 30 and 30 show the coefficients of friction depending on mole ratio of amine and three different conditions.

TABLE 29

Amine Molar Ratio	Condition	Coefficient of Friction	3
100	25° C., 60% RH	0.24	
	40° C., 80% RH	0.26	
	−5° C.	0.24	
50	25° C., 60% RH	0.22	
	40° C., 80% RH	0.25	A
	−5° C.	0.23	4
9	25° C., 60% RH	0.21	
	40° C., 80% RH	0.25	
	−5° C.	0.22	
4	25° C., 60% RH	0.18	
•	40° C., 80% RH	0.24	
	−5° C.	0.19	4
2.33	25° C., 60% RH	0.18	
	40° C., 80% RH	0.23	
	−5° C.	0.18	
1.5	25° C., 60% RH	0.17	
	40° C., 80% RH	0.23	
	−5° C.	0.17	5
1	25° C., 60% RH	0.16	
	40° C., 80% RH	0.22	
	5° C.	0.17	

					
TO A DIT TO 200	2Δ		TOT		راك
TABLE 30	11	.H.	P% I	\boldsymbol{A}	

Amine Molar Ratio	Condition	Coefficient of Friction
0.67	25° C., 60% RH	0.16
	40° C., 80% RH	0.22
	−5° C.	0.17
0.43	25° C., 60% RH	0.17
	40° C., 80% RH	0.23
	−5° C.	0.18
0.25	25° C., 60% RH	0.17
•	40° C., 80% RH	0.23
	−5° C.	0.18
0.10	25° C., 60% RH	0.19
	40° C., 80% RH	0.23
	−5° C.	0.19
0.05	25° C., 60% RH	0.19

	40° C., 80% RH	0.24
	5° C.	0.21
0.02	25° C., 60% RH	0.20
	40° C., 80% RH	0.25
	−5° C.	0.21
0.01	25° C., 60% RH	0.21
	40° C., 80% RH	0.26
	−5° C.	0.22
0	25° C., 60% RH	0.24
	40° C., 80% RH	0.27
	−5° C.	0.25

Similarly, with respect to the lubricant D11, different amount of amine are added to the ester compound by changing them from 0.01 to 100 mole ratio. The coefficients of friction are measured for these sample tapes. Tables 31 and 32 show the coefficients of friction depending on mole ratio of amine and three different conditions.

TABLE 31

Amine Molar Ratio	Condition	Coefficient of Friction
100	25° C., 60% RH	0.22
	40° C., 80% RH	0.26
	−5° C.	0.21
50	25° C., 60% RH	0.21
	40° C., 80% RH	0.25
	−5° C.	0.21
9	25° C., 60% RH	0.19
	40° C., 80% RH	0.21
	−5° C.	0.20
4	25° C., 60% RH	0.19
	40° C., 80% RH	0.21
	−5° C.	0.19
2.33	25° C., 60% RH	0.18
	40° C., 80% RH	0.21
	−5° C.	0.19
1.5	25° C., 60% RH	0.17
	40° C., 80% RH	0.23
	−5° C.	0.18
1	25° C., 60% RH	0.17
	40° C., 80% RH	0.22
· ····································	−5° C.	0.17

TABLE 32

	Amine Molar Ratio	Condition	Coefficient of Friction
5 [–]	0.67	25° C., 60% RH	0.16
		40° C., 80% RH	0.22
		5° C.	0.17
	0.43	25° C., 60% RH	0.17
		40° C., 80% RH	0.23

TABLE 32-continued

Amine Molar Ratio	Condition	Coefficient of Friction	_
	−5° C.	0.17	
0.25	25° C., 60% RH	0.17	
	40° C., 80% RH	0.24	
	5° C .	0.18	
0.10	25° C., 60% RH	0.18	
	40° C., 80% RH	0.25	
-	−5° C.	0.18	
0.05	25° C., 60% RH	0.19	
	40° C., 80% RH	0.25	
	−5° C.	0.19	
0.02	25° C., 60% RH	0.19	
	40° C., 80% RH	0.25	
	−5° C.	0.19	
0.01	25° C., 60% RH	0.19	
	40° C., 80% RH	. 0.25	
	5° C.	0.20	

The lubricants have been compared between the perfluoropolyether alone with hydroxyl group at its 20 end and the ester compound of perfluoropolyether with hydroxyl group and perfluoro carboxylic acid admixed with long chain hydrocarbon amine of from 0.01 to 100 mole ratio. It was found that the latter could obtain the better result under various conditions such as friction 25 coefficient, still durability and shuttle durability.

The solubility has been also considered with respect to the ester compound of perfluoropolyether with hydroxyl group at its end and perfluoro carboxylic acid (utilized as the lubricant D1).

The ester compound is soluble in hexane as well as freon. As the long chain alkyl amine is also soluble in hexane, the lubricant D1 is soluble in hexane.

Therefore, the ester compound of perfluoropolyether with hydroxyl group at its end and perfluoro carboxylic 35

mixed with long chain hydrocarbon amine. The ester compound is further admixed by at least selected one of phosphoric ester, phosphorous ester and carboxylate ester.

The ester compound of perfluoropolyether having hydroxyl group at its end and long chain carboxylic acid (shown by the general formula [11]) is composed of a solution of perfluoropolyether with hydroxyl group at its end and third grade amine added by dropping long chain carboxylic acid chloride. The reaction has been completed under reflux and heat after dropping. The solution was then cooled to the room temperature. The generated amine salt was washed by distilled water and then purified by silica gel column chromatography.

The ester compound of perfluoropolyether having carboxyl group at its end and long chain alcohol (shown by the general formula [16]) is performed in anhydrous toluene by removing water out of the perfluoropolyether having carboxyl group at its end and the long chain alcohol. After having the reaction completed, toluene is removed and it is purified by silica gel column chromatography.

The ester compound shown by the general formula [11] is more in detail by the formula [23]. The lubricants E1 to E14 are shown by the formula [23A] with molecular weight of 2000 and two functional or shown by the formula [23B] with molecular weight of 2000 and monofunctional. Table 33 shows respective examples of alkyl group R, phosphorus group extreme pressure agent and long chain carboxylic acid ester as the long chain hydrocarbon amine.

C₁₇H₃₅COOCH₂CF₂(OC₂F₄)_p(OCF₂)_qOCF₂C-H₂OCOC₁₇H₃₅ for

formula [23A]

F(CF₂CF₂CF₂O)_nCF₂CF₂CH₂OCOC₁₇H₃₅

formula [23B]

TABLE 33

No.	Ester	R	Phosphorus Group Extreme Pressure Agent	Long Chain Carboxylic Acid Ester
E1	(a)	C ₁₈ H ₃₇	(C ₁₈ H ₃₇ O) ₃ P	C ₁₇ F ₃₅ COOC ₂ H ₅
E2	(a)	C ₁₆ H ₃₃	(C ₁₆ H ₃₃ O) ₃ P	C ₁₅ F ₃₁ COOC ₂ H ₅
E 3	(a)	C ₁₄ H ₂₉	$(C_{14}H_{29}O)_3P$	$C_{11}F_{23}COOC_2H_5$
E 4	(a)	$C_{10}H_{21}$	$C_{12}H_{25}OP(OH)_2$	C ₁₇ F ₃₁ COOC ₂ H ₅
E5	(a)	C ₁₈ H ₃₇	$(C_{12}H_{25}O)_2POH$	C ₁₇ F ₂₉ COOC ₂ H ₅
E 6	(a)	C ₁₈ H ₃₇	$(C_{12}H_{25}S)_3P$	C ₁₇ F ₂₉ COOCH ₂ C ₇ F ₁₅
E7	(a)	C ₁₈ H ₃₁	$(C_{18}H_{31}S)_3P=O$	iso-C ₁₇ H ₃₅ COOC ₂ H ₅
E8	(a)	C ₁₈ H ₃₇	$(C_{12}H_{25}S)_3P = S$	C9F19COOC18H37
E9	(b)	C ₁₈ H ₃₇	$(C_{12}H_{25}O)_2POOH$	C9H19COOC2H5
E10	(b)	C ₆ H ₁₁ (Cyclo hexyl)	$(C_6H_{11}O)_3P$	C ₁₇ H ₃₅ COOC ₆ H ₁₁
E11	(b)	C ₁₈ H ₃₇ C ₆ H ₄	(C ₁₈ H ₃₇ C ₆ H ₄ O) ₃ P	C ₁₇ H ₃₅ COOC ₂ H ₅
E12	(b)	iso-C ₁₈ H ₃₇	(iso-C ₁₈ H ₃₇ O) ₃ P	C ₁₇ H ₃₅ COOC ₂ H ₅
E13	(b)	C ₁₈ H ₃₇	$(C_{18}H_{37}O)_3P$	C ₁₇ H ₃₅ COOC ₂ H ₅
E14	(b)	$CH_2 = CHC_{16}H_{32}$	(C ₁₈ H ₃₇ O) ₃ P	C ₁₇ H ₃₅ COOC ₂ H ₅

acid admixed with long chain hydrocarbon amine can be utilized as a lubricant with the excellent lubricant effect without using freon.

EXPERIMENTATION E

The lubricant of the fifth embodiment contains ester 60 compound made of perfluoropolyether having hydroxyl group at its end and long chain carboxylic acid admixed with long chain hydrocarbon amine. The ester compound is further admixed by at least selected one of phosphoric ester, phosphorous ester and carboxylate 65 ester. The lubricant of the sixth embodiment contains ester compound made of perfluoropolyether having carboxyl group at its end and long chain alcohol ad-

The ester compound shown by the general formula [16] is more in detail by the formula [24]. The lubricants E15 to E28 are shown by the formula [23C] with molecular weight of 2 000 and two functional or shown by the formula [23D] with molecular weight of 2000 and monofunctional, Table 34 shows respective examples of alkyl group R, phosphorus group extreme pressure agent and long chain carboxylic acid ester as the long chain hydrocarbon amine,

C₁₈H₃₇OCOCF₂(OC₂F₄)_p(OCF₂)_qOCF₋₂COOC₁₈H₃₇

formula [23C]

F(CF₂CF₂CF₂O)_nCF₂CF₂COOC₁₈H₃₇

formula [23D]

TABLE 34

No.	Ester	R	Phosphorus Group Extreme Pressure Agent	Long Chain Carboxylic Acid Ester
E15	(c)	C ₁₈ H ₃₇	(C ₁₈ H ₃₇ O) ₃ P	C ₁₇ F ₃₅ COOC ₂ H ₅
E16	(c)	$C_{16}H_{33}$	$(C_{16}H_{33}O)_3P$	C ₁₅ F ₃₁ COOC ₂ H ₅
E17	(c)	C ₁₄ H ₂₉	$(C_{14}H_{29}O)_3P$	C ₁₁ F ₂₃ COOC ₂ H ₅
E18	(c)	$C_{10}H_{21}$	$C_{12}H_{25}OP(OH)_2$	C ₁₇ F ₃₁ COOC ₂ H ₅
E19	(c)	C ₁₈ H ₃₇	$(C_{12}H_{25}O)_2POH$	C ₁₇ F ₂₉ COOC ₂ H ₅
E20	(c)	$C_{18}H_{31}$	$(C_{12}H_{25}S)_3P$	C ₁₇ F ₂₉ COOCH ₂ C ₇ F ₁₅
E21	(c)	C ₁₈ H ₃₇	$(C_{18}H_{31}S)_3P = O$	iso-C ₁₇ H ₃₅ COOC ₂ H ₅
E22	(c)	$C_{18}H_{37}$	$(C_{12}H_{25}S)_3P = S$	C ₉ F ₁₉ COOC ₁₈ H ₃₇
E23	(d)	C ₁₈ H ₃₇	$(C_{12}H_{25}O)_2POOH$	C ₉ H ₁₉ COOC ₂ H ₅
E24	(d)	C_6H_{11}	$(C_6H_{11}O)_3P$	C ₁₇ H ₃₅ COOC ₆ H ₁₁
		(Cyclo hexyl)		
E25	(d)	C ₁₈ H ₃₇ C ₆ H ₄	$(C_{18}H_{37}C_6H_4O)_3P$	C ₁₇ H ₃₅ COOC ₂ H ₅
E26	(d)	isoC ₁₈ H ₃₇	(iso-C ₁₈ H ₃₇ O) ₃ P	C ₁₇ H ₃₅ COOC ₂ H ₅
E27	(d)	C ₁₈ H ₃₇	$(C_{18}H_{37}O)_3P$	C ₁₇ H ₃₅ COOC ₂ H ₅
E28	(d)	$CH_2 = CHC_{16}H_{32}$	$(C_{18}H_{37}O)_3P$	C ₁₇ H ₃₅ COOC ₄ H ₉

With these lubricants E1 to E28, the sample tapes are prepared by the same method as explained in the experiples E1 to 14 with the lubricants E1 to E14 and also in mentation A. Regarding each sample tape, the friction coefficient, still and shuttle durabilities are measured by

Tables 37 and 38 for the examples E15 to 28 with the lubricants E15 to E28.

TABLE 35

	Condition	Coefficient of Friction	Still Durability (Min.)	Shuttle Durability (Times)
Example E1	25° C., 60% RH	0.16	>150	>150
(Lubricant E1)	40° C., 80% RH	0.19	>150	>150
	−5° C.	0.16	>150	>150
Example E2	25° C., 60% RH	0.16	>150	>150
(Lubricant E2)	40° C., 80% RH	0.20	>150	>150
	−5° C.	0.17	>150	>150
Example E3	25° C., 60% RH	0.16	>150	>150
(Lubricant E3)	40° C., 80% RH	0.21	>150	>150
	−5° C.	0.17	>150	>150
Example E4	25° C., 60% RH	0.16	>150	- >150
(Lubricant E4)	40° C., 80% RH	0.19	>150	>150
	−5° C.	0.16	>150	>150
Example E5	25° C., 60% RH	0.16	>150	>150
(Lubricant E5)	40° C., 80% RH	0.19	>150	>150
	−5° C.	0.17	>150	>150
Example E6	25° C., 60% RH	0.16	>150	>150
(Lubricant E6)	40° C., 80% RH	0.19	>150	>150
	−5° C.	0.17	>150	>150
Example E7	25° C., 60% RH	0.17	>150	>150
(Lubricant E7)	40° C., 80% RH	0.20	>150	>150
	−5° C.	0.17	>150	>150

the same method as explained in the experimentation A.

TABLE 36				
	Condition	Coefficient of Friction	Still Durability (Min.)	Shuttle Durability (Times)
Example E8	25° C., 60% RH	0.17	>150	>150
(Lubricant E8)	40° C., 80% RH	0.22	>150	>150
	−5° C.	0.18	>150	>150
Example E9	25° C., 60% RH	0.18	>150	>150
(Lubricant E9)	40° C., 80% RH	0.22	>150	>150
	−5° C.	0.19	>150	>150
Example E10	25° C., 60% RH	0.19	>150	>150
(Lubricant E10)	40° C., 80% RH	0.23	>150	>150
	−5° C.	0.20	>150	>150
Example E11	25° C., 60% RH	0.16	>150	>150
(Lubricant E11)	40° C., 80% RH	0.20	>150	>150
	−5° C.	0.17	>150	>150
Example E12	25° C., 60% RH	0.18	>150	>150
(Lubricant E12)	40° C., 80% RH	0.22	>150	>150
	−5° C.	0.19	>150	>150
Example E13	25° C., 60% RH	0.16	>150	>150
(Lubricant E13)	40° C., 80% RH	0.19	>150	>150
	−5° C.	0.17	>150	>150
Example E14	25° C., 60% RH	0.17	>150	>150
(Lubricant E14)	40° C., 80% RH	0.20	>150	>150

TABLE 36-continued

Condition	Coefficient of Friction	Still Durability (Min.)	Shuttle Durability (Times)
 −5° C.	0.18	>150	>150

TABLE 37

	Condition	Coefficient of Friction	Still Durability (Min.)	Shuttle Durability (Times)
Example E15	25° C., 60% RH	0.15	>150	>150
(Lubricant E15)	40° C., 80% RH	0.19	>150	>150
	−5° C.	0.16	>150	>150
Example E16	25° C., 60% RH	0.16	>150	>150
(Lubricant E16)	40° C., 80% RH	0.20	>150	>150
	−5° C.	0.17	>150	>150
Example E17	25° C., 60% RH	0.16	>150	>150
(Lubricant E17)	40° C., 80% RH	0.22	>150	>150
	5° C.	0.17	>150	>150
Example E18	25° C., 60% RH	0.16	>150	>150
(Lubricant E18)	40° C., 80% RH	0.19	>150	>150
	−5° C.	0.17	>150	>150
Example E19	25° C., 60% RH	0.16	>150	>150
(Lubricant E19)	40° C., 80% RH	0.21	>150	>150
	−5° C.	0.17	>150	>150
Example E20	25° C., 60% RH	0.17	>150	>150
(Lubricant E20)	40° C., 80% RH	0.20	>150	>150
	−5° C.	0.18	>150	>150
Example E21	25° C., 60% RH	0.18	>150	>150
(Lubricant E21)	40° C., 80% RH	0.20	>150	>150
	−5° C.	0.18	>150	>150

TABLE 38

	Condition	Coefficient of Friction	Still Durability (Min.)	Shuttle Durability (Times)
Example E22	25° C., 60% RH	0.17	>150	>150
(Lubricant E22)	40° C., 80% RH	0.20	>150	>150
•	−5° C.	0.18	>150	>150
Example E23	25° C., 60% RH	0.19	>150	>150
(Lubricant E23)	40° C., 80% RH	0.23	>150	>150
	−5° C.	0.19	>150	>150
Example E24	25° C., 60% RH	0.19	>150	>150
(Lubricant E24)	40° C., 80% RH	0.24	>150	>150
	−5° C.	0.19	>150	>150
Example E25	25° C., 60% RH	0.16	>150	>150
(Lubricant E25)	40° C., 80% RH	0.19	>150	>150
	5° C.	0.17	>150	>150
Example E26	25° C., 60% RH	0.18	>150	>150
(Lubricant E26)	40° C., 80% RH	0.23	>150	>150
	−5° C.	0.18	>150	>150
Example E27	25° C., 60% RH	0.17	>150	>150
(Lubricant E27)	40° C., 80% RH	0.20	>150	>150
	5° C.	0.18	>150	>150
Example E28	25° C., 60% RH	0.17	>150	>150
(Lubricant E28)	40° C., 80% RH	0.22	>150	>150
	−5° C.	0.18	>150	>150

For the purpose of comparison, a sample tape (comparison example E1) is prepared without using any lubricant. The sample tapes (comparison example E2 and E3) are prepared with lubricants made by perfluoropolyether alone having hydroxyl group at its end (shown in general formula [17] and [18]). The sample

tapes (comparison example E4 and E5) are prepared with lubricants made by perfluoropolyether alone having carboxyl group at its end (shown in general formula [20] and [21]). The result of durability measurement is shown in Table 39.

TABLE 39

	Condition	Coefficient of Friction	Still Durability (Min.)	Shuttle Durability (Times)
Comparison	25° C., 60% RH	0.9	2	3
Example E1	40° C., 80% RH			
(without Lubricant)	5° C.			
Comparison	25° C., 60% RH	0.20	>120	>150
Example E2	40° C., 80% RH	0.30	91	55

TABLE 39-continued

			Still	Shuttle
<u></u>	Condition	Coefficient of Friction	Durability (Min.)	Durability (Times)
	5° C.	0.21	51	60
Comparison	25° C., 60% RH	0.24	>120	120
Example E3	40° C., 80% RH	0.27	76	46
	−5° C.	0.25	29	40
Comparison	25° C., 60% RH	0.24	>120	>150
Example E4	40° C., 80% RH	0.27	91	55
-	−5° C.	0.25	51	60
Comparison	25° C., 60% RH	0.22	>120	120
Example E5	40° C., 80% RH	0.23	76	46
-	−5° C.	0.22	29	40

The sample tapes as examples of the invention have the lubricant containing ester compound made of perfluoropolyether admixed with long chain hydrocarbon amine. With these sample tapes, the relationship between the added amount of amine and its coefficient of friction is examined.

With respect to the lubricant E1, different amount of stearic amine are added to the ester compound by changing them from 0.01 to 100 mole ratio. The coefficients of friction are measured for these sample tapes. As the phosphoric ester, $(C_{12}H_{25}O)_3P$ is utilized and as the long chain carboxylic acid, $C_{17}H_{35}COOC_4H_9$ is utilized. Tables 40 and 41 show the coefficients of friction depending on mole ratio of amine and three different conditions.

TABLE 40

· .	IABLE 40		
Amine Molar Ratio	Condition	Coefficient of Friction	 -
100	25° C., 60% RH	0.24	
	40° C., 80% RH	0.28	•
	−5° Ć.	0.24	•
50	25° C., 60% RH	0.23	
	40° C., 80% RH	0.26	
	−5° C.	0.23	
9	25° C., 60% RH	0.22	•
	40° C., 80% RH	0.25	
	−5° C.	0.22	4
4	25° C., 60% RH	0.20	
	40° C., 80% RH	0.25	
	−5° C.	0.21	
2.33	25° C., 60% RH	0.18	
	40° C., 80% RH	0.23	
	−5° C.	0.18	4
1.5	25° C., 60% RH	0.17	
	40° C., 80% RH	0.21	
	−5° C.	0.17	
1	25° C., 60% RH	0.16	
	40° C., 80% RH	0.19	
	5° C.	0.16	4

TABLE 41

Amine Molar Ratio	Condition	Coefficient of Friction	-
0.67	25° C., 60% RH	0.16	
•	40° C., 80% RH	0.20	
	−5° C.	0.16	
0.43	25° C., 60% RH	0.17	
	40° C., 80% RH	0.21	
	−5° C.	0.17	(
0.25	25° C., 60% RH	0.17	
	40° C., 80% RH	0.22	
	−5° C.	0.17	
0.10	25° C., 60% RH	0.17	
	40° C., 80% RH	0.22	
	−5° C.	0.17	(
0.05	25° C., 60% RH	0.19	
	40° C., 80% RH	0.24	
	− 5° C .	0.19	
0.02	25° C., 60% RH	0.21	

TABLE 41-continued

	Amine Molar Ratio	Condition	Coefficient of Friction			
•••••		40° C., 80% RH	0.25			
)		−5° C.	0.21			
	0.01	25° C., 60% RH	0.23			
		40° C., 80% RH	0.27			
		−5° C.	0.23			
	0	25° C., 60% RH	0.29			
		40° C., 80% RH	0.43			
		−5° C.	0.29			

Regarding the above lubricant, when it has the amount of amine of zero mole ratio, in other words no amine is admixed, the following data has obtained by the still and shuttle durabilities measurement.

25° C., 60% RH: still durability: 57 mins.; shuttle durability: 63 mins.

40° C., 80% RH: still durability: 23 mins.; shuttle durability: 31 mins.

35 -5° C.: still durability: 32 mins.; shuttle durability: 33 mins.

These data show relatively unsatisfactory values.

The lubricants have been compared between the perfluoropolyether alone with hydroxyl group or car40 boxyl group at its end and the ester compound admixed with long chain hydrocarbon amine of from 0.01 to 100 mole ratio, further admixed with phosphoric ester, phosphorous ester or long chain carboxylate ester. It was found that the latter could obtain the better result under various conditions such as friction coefficient, still durability and shuttle durability.

The solubility has been also considered with respect to the ester compound of perfluoropolyether with hydroxyl group at its end and long chain carboxylate acid and the ester compound of perfluoropolyether with carboxyl group at its end and long chain alcohol. The ester compound are soluble in hexane as well as freon. As the long chain hydrocarbon amine, phosphoric ester, phosphorous ester and long chain carboxylate ester etc. are also soluble in hexane, the above lubricant is soluble in hexane.

Therefore, the ester compound of perfluoropolyether with hydroxyl group at its end and long chain carboxylate acid or the ester compound of perfluoropolyether with carboxyl group at its end and long chain alcohol admixed with long chain hydrocarbon amine, phosphoric ester, phosphorous ester and long chain carboxylate ester etc. can be utilized as a lubricant with the excellent lubricant effect without using freon.

As clearly understood by the above explanation, the lubricant is made of the ester compound of per-fluoropolyether and long chain alcohol admixed by long chain hydrocarbon amine of from 0.01 to 100 mole

ratio. The lubricant has excellent lubricant effect and maintain the effect for a long time under any conditions.

By esterifying perfluoropolyether, The solubility can be improved and it becomes unnecessary to use freon as a diluent.

The magnetic recording medium with these lubricant therefore has the same lubricant effect at the surface of the medium for a long time under any conditions. The magnetic recording medium can have excellent wear characteristics, durability and in particular outstanding 10 running characteristics.

Thus it is apparent that in accordance with the present invention, an apparatus that fully satisfies the objectives, aims and advantages is set forth above. While the invention has been described in conjunction with specific embodiments, it is evident that many alternatives, modifications, permutations and variations will become apparent to those skilled in the art in light of the foregoing description. Accordingly, it is intended that the present invention embrace all such alternatives, modifications and variations as fall within the scope of the appended claims.

What is claimed is:

1. A lubricant comprising:

- an ester compound including a perfluoropolyether with an hydroxyl group at its end and a long chain carboxylic acid, and
- a long chain hydrocarbon amine, wherein said ester compound is represented by a formula chosen from the group consisting of:

 R^{f_1} — CH_2OCOR^1 ;

and

R¹COOCH₂R^fCH₂OCOR¹

wherein R^{fl} is a perfluoropolyether group and R^l is a long chain hydrocarbon group; and said long chain hydrocarbon amine is represented by the 40 formula:

 RNH_2

in which R is a long chain hydrocarbon group.

- 2. A magnetic recording medium comprising a magnetic layer on non-magnetic base member and the lubricant in accordance with claim 1 on the surface of said magnetic layer.
 - 3. A lubricant comprising:
 - an ester compound including a perfluoropolyether with a hydroxyl group at its end and a partially fluorined long chain carboxylic acid, and
 - a long chain hydrocarbon amine, wherein said ester compound is represented by a formula chosen from 55 the group consisting of:

R/2—CH₂OCOR²;

and

R²COOCH₂R^{/2}CH₂OCOR²

wherein Rf² is a perfluoropolyether group and R² is a partially fluorined long chain hydrocarbon group; and said long chain hydrocarbon amine is 65 represented by the formula:

RNH₂

in which R is a long chain hydrocarbon group.

- 4. A magnetic recording medium comprising a magnetic layer on non-magnetic base member and the lubricant in accordance with claim 3 on the surface of said magnetic layer.
 - 5. A lubricant comprising:
 - an ester compound including a perfluoropolyether with a carboxyl group at its end and a partially fluorined long chain alcohol, and

long chain hydrocarbon amine, wherein said ester compound is represented by a formula chosen from the group consisting of:

 $R^{/3}$ —COOR³;

and

RªOCOR^{f3}COOR³

wherein R/3 is a perfluoropolyether group and R3 is a partially fluorined long chain hydrocarbon group; and said long chain hydrocarbon amine is represented by the formula:

RNH₂

in which R is a long chain group hydrocarbon.

- 6. A magnetic recording medium comprising a magnetic layer on non-magnetic base member and the lubricant in accordance with claim 5 on the surface of said magnetic layer.
 - 7. A lubricant comprising:
 - an ester compound including a perfluoropolyether with a hydroxyl group at its end and a perfluoro carboxylic acid, and
 - a long chain hydrocarbon amine, wherein said ester compound is represented by a formula chosen from the group consisting of:

 R^{f4} — CH_2OCOR^{f5} ;

and

Rf5COOCH2Rf4CH2OCORf5

wherein R⁴ is a perfluoropolyether group and R⁵ is a perfluoro hydrocarbon group; and said long chain hydrocarbon amine is represented by the formula;

RNH₂

in which R is a long chain hydrocarbon group.

- 8. A magnetic recording medium comprising a magnetic layer on non-magnetic base member and the lubricant in accordance with claim 7 on the surface of said magnetic layer.
- 9. A lubricant in accordance with claim 1, further admixed with at least one of phosphoric ester, phosphorous ester and long chain carboxylate ester.
 - 10. A lubricant comprising:
 - an ester compound including a perfluoropolyether with a carboxyl group at its end and a long chain alcohol,
 - a long chain hydrocarbon amine, and at least one of phosphoric ester, phosphorous ester and long chain carboxylate ester, wherein said ester compound is

represented by a formula chosen from the group consisting of:

R/6—COOR6;

and

R6OCOR6COOR6

wherein R^{f6} is a perfluoropolyether group and R⁶ is a long chain hydrocarbon group; and said long chain amine is represented by the formula;

RNH₂

in which R is a long chain hydrocarbon group.

11. A magnetic recording medium comprising a magnetic layer on non-magnetic base member and the lubricant in accordance with claim 10 on the surface of said magnetic layer.

12. A lubricant in accordance with claim 1, wherein said long chain alkyl amine is admixed in said ester compound with an amount of from 0.01 to 100 in mole ratio.

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