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**United States Patent** [19]**Cherpeck**[11] **Patent Number:** **5,366,517**[45] **Date of Patent:** **Nov. 22, 1994**

- [54] **FUEL ADDITIVE COMPOSITIONS CONTAINING POLY(OXYALKYLENE) HYDROXYAROMATIC ETHERS AND POLY(OXYALKYLENE) AMINES**
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- [73] **Assignee:** Chevron Research and Technology Company, San Francisco, Calif.
- [21] **Appl. No.:** 993,219
- [22] **Filed:** Dec. 18, 1992
- [51] **Int. Cl.<sup>5</sup>** ..... C10L 1/22; C10L 1/18
- [52] **U.S. Cl.** ..... 44/387; 44/400; 44/434; 44/442; 44/443; 44/448; 44/450; 560/103; 560/129; 560/144; 568/606; 568/607; 568/608; 568/609; 568/610; 568/611
- [58] **Field of Search** ..... 568/607, 608, 609, 610, 568/611, 606; 560/103, 129, 144; 44/443, 447, 448, 449, 450

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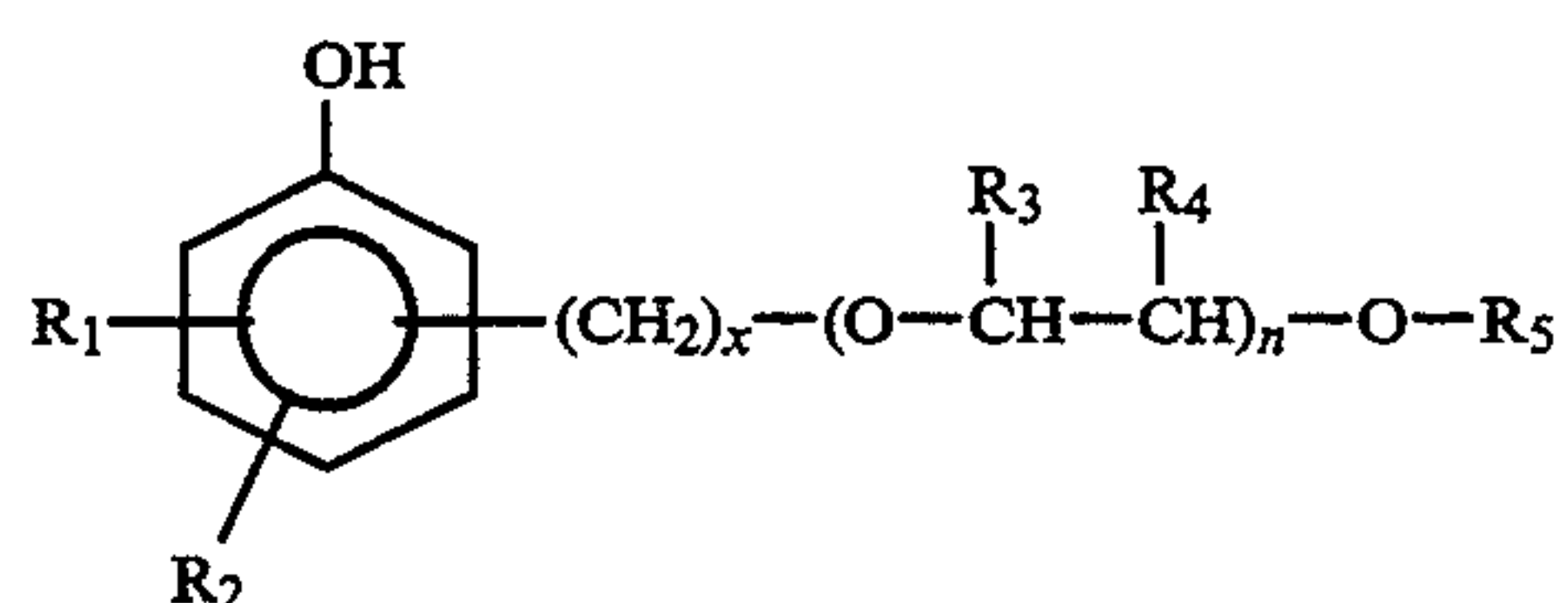
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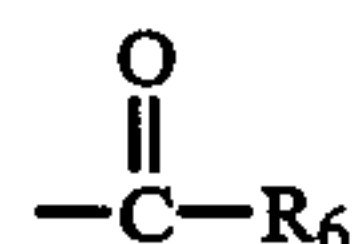
*Primary Examiner*—Ellen M. McAvoy  
*Attorney, Agent, or Firm*—C. J. Caroli[57] **ABSTRACT**

A fuel additive composition comprising:

- (a) a poly(oxyalkylene) hydroxyaromatic ether having the formula:



or a fuel-soluble salt thereof; where R<sub>1</sub> and R<sub>2</sub> are each independently hydrogen, hydroxy, lower alkyl having 1 to 6 carbon atoms, or lower alkoxy having 1 to 6 carbon atoms; R<sub>3</sub> and R<sub>4</sub> are each independently hydrogen or lower alkyl having 1 to 6 carbon atoms; R<sub>5</sub> is hydrogen, alkyl having 1 to 30 carbon atoms, phenyl, aralkyl or alkaryl having 7 to 36 carbon atoms, or an acyl group of the formula:



where R<sub>6</sub> is alkyl having 1 to 30 carbon atoms, phenyl, or aralkyl or alkaryl having 7 to 36 carbon atoms; n is an integer from 5 to 100; and x is an integer from 0 to 10; and

- (b) a poly(oxyalkylene) amine having at least one basic nitrogen atom and a sufficient number of oxyalkylene units to render the poly(oxyalkylene) amine soluble in hydrocarbons boiling in the gasoline or diesel fuel range.

**44 Claims, No Drawings**



# FUEL ADDITIVE COMPOSITIONS CONTAINING POLY(OXYALKYLENE) HYDROXYAROMATIC ETHERS AND POLY(OXYALKYLENE) AMINES

## BACKGROUND OF THE INVENTION

### 1. Field of the Invention

This invention relates to a fuel additive composition. More particularly, this invention relates to a fuel additive composition containing a poly(oxyalkylene) hydroxyaromatic ether and a poly(oxyalkylene) amine.

### 2. Description of the Related Art

It is well known that automobile engines tend to form deposits on the surface of engine components, such as carburetor ports, throttle bodies, fuel injectors, intake ports and intake valves, due to the oxidation and polymerization of hydrocarbon fuel. These deposits, even when present in relatively minor amounts, often cause noticeable driveability problems, such as stalling and poor acceleration. Moreover, engine deposits can significantly increase an automobile's fuel consumption and production of exhaust pollutants. Therefore, the development of effective fuel detergents or "deposit control" additives to prevent or control such deposits is of considerable importance and numerous such materials are known in the art.

For example, aliphatic hydrocarbon-substituted phenols are known to reduce engine deposits when used in fuel compositions. U.S. Pat. No. 3,849,085, issued Nov. 19, 1974 to Kreuz et al., discloses a motor fuel composition comprising a mixture of hydrocarbons in the gasoline boiling range containing about 0.01 to 0.25 volume percent of a high molecular weight aliphatic hydrocarbon-substituted phenol in which the aliphatic hydrocarbon radical has an average molecular weight in the range of about 500 to 3,500. This patent teaches that gasoline compositions containing minor amount of an aliphatic hydrocarbon-substituted phenol not only prevent or inhibit the formation of intake valve and port deposits in a gasoline engine, but also enhance the performance of the fuel composition in engines designed to operate at higher operating temperatures with a minimum of decomposition and deposit formation in the manifold of the engine.

Similarly, U.S. Pat. No. 4,134,846, issued Jan. 16, 1979 to Machleder et al., discloses a fuel additive composition comprising a mixture of (1) the reaction product of an aliphatic hydrocarbon-substituted phenol, epichlorohydrin and a primary or secondary mono- or polyamine, and (2) a polyalkylene phenol. This patent teaches that such compositions show excellent carburetor, induction system and combustion chamber detergency and, in addition, provide effective rust inhibition when used in hydrocarbon fuels at low concentrations.

Poly(oxyalkylene) amines are also well known in the art as fuel additives for the prevention and control of engine deposits. For example, U.S. Pat. No. 4,191,537, issued Mar. 4, 1980 to R. A. Lewis et al., discloses a fuel composition comprising a major portion of hydrocarbons boiling in the gasoline range and from 30 to 2000 ppm of a hydrocarbyl poly(oxyalkylene) aminocarbonate having a molecular weight from about 600 to 10,000, and at least one basic nitrogen atom. The hydrocarbyl poly(oxyalkylene) moiety is composed of oxyalkylene units selected from 2 to 5 carbon oxyalkylene units. These fuel compositions are taught to maintain

the cleanliness of intake systems without contributing to combustion chamber deposits.

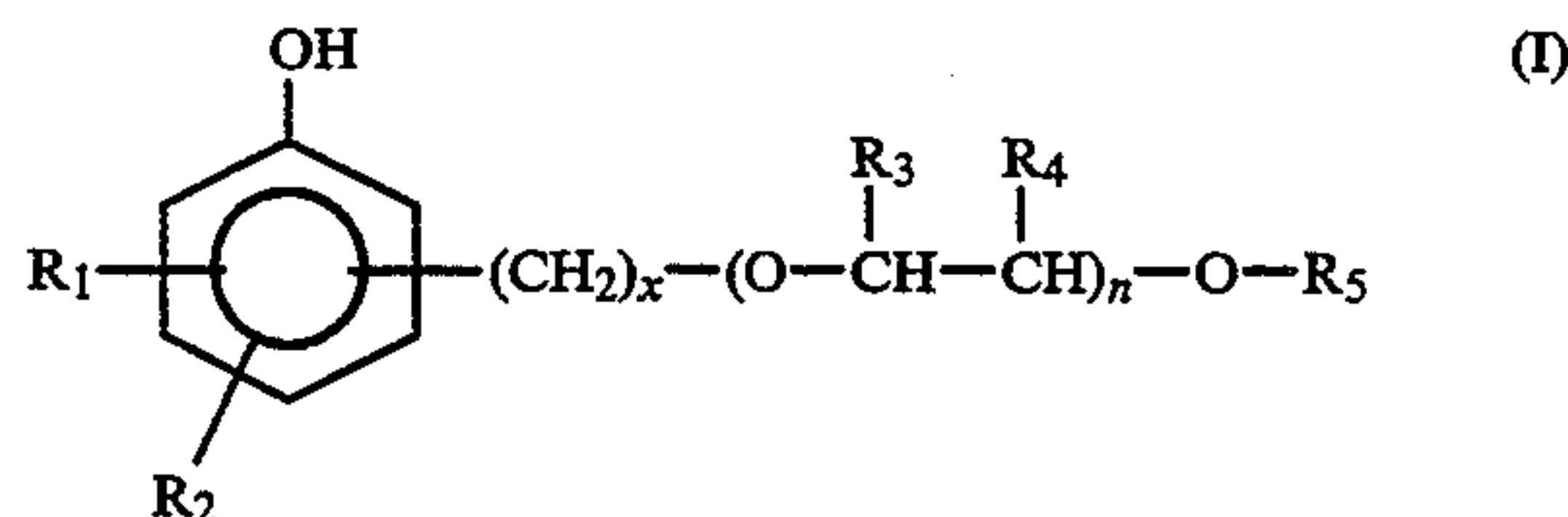
Similar poly(oxyalkylene) amine fuel additives and fuel compositions containing such additives are described in U.S. Pat. Nos. 4,160,648; 4,197,409; 4,233,168; 4,236,020; 4,243,798; 4,247,301; 4,261,704; 4,270,930; 4,274,837; 4,281,199; 4,288,612; 4,329,240; 4,332,595; 4,604,103; 4,778,481; 4,881,945; 5,055,607; 5,094,667; and in PCT International Patent Application Publication No. WO 90/07564, published Jul. 12, 1990.

It has now been discovered that the combination of a poly(oxyalkylene) amine and a novel poly(oxyalkylene) hydroxyaromatic ether affords a unique fuel additive composition that provides unexpectedly superior deposit control performance and fewer combustion chamber deposits than either component individually.

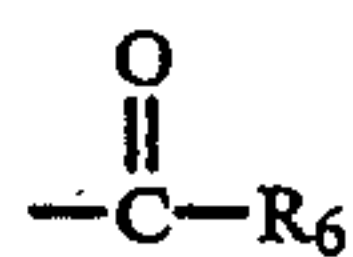
## SUMMARY OF THE INVENTION

The present invention provides a novel fuel additive composition comprising:

- (a) a poly(oxyalkylene) hydroxyaromatic ether having the formula:



or a fuel-soluble salt thereof; wherein  $R_1$  and  $R_2$  are each independently hydrogen, hydroxy, lower alkyl having 1 to 6 carbon atoms, or lower alkoxy having 1 to 6 carbon atoms;  $R_3$  and  $R_4$  are each independently hydrogen or lower alkyl having 1 to 6 carbon atoms;  $R_5$  is hydrogen, alkyl having 1 to 30 carbon atoms, phenyl, aralkyl or alkaryl having 7 to 36 carbon atoms, or an acyl group of the formula:



wherein  $R_6$  is alkyl having 1 to 30 carbon atoms, phenyl, or aralkyl or alkaryl having 7 to 36 carbon atoms;  $n$  is an integer from 5 to 100; and  $x$  is an integer from 0 to 10; and

- (b) a poly(oxyalkylene) amine having at least one basic nitrogen atom and a sufficient number of oxyalkylene units to render the poly(oxyalkylene) amine soluble in hydrocarbons boiling in the gasoline or diesel fuel range.

The present invention further provides a fuel composition comprising a major amount of hydrocarbons boiling in the gasoline or diesel range and an effective deposit-controlling amount of the novel fuel additive composition of the present invention.

The present invention additionally provides a fuel concentrate comprising an inert stable oleophilic organic solvent boiling in the range of from about 150° F. to 400° F. and from about 10 to 70 weight percent of the fuel additive composition of the present invention.

Among other factors, the present invention is based on the surprising discovery that the unique combination of a poly(oxyalkylene) hydroxyaromatic ether and a poly(oxyalkylene) amine provides unexpectedly supe-



rior deposit control performance and fewer combustion chamber deposits than either component individually.

### DETAILED DESCRIPTION OF THE INVENTION

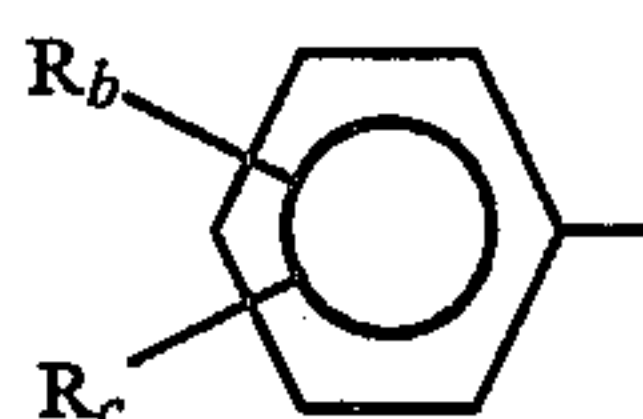
As used herein the following terms have the following meanings unless expressly stated to the contrary.

The term "alkyl" refers to both straight- and branched-chain alkyl groups.

The term "lower alkyl" refers to alkyl groups having 1 to about 6 carbon atoms and includes primary, secondary and tertiary alkyl groups. Typical lower alkyl groups include, for example, methyl, ethyl, n-propyl, isopropyl, n-butyl, sec-butyl, t-butyl, n-pentyl, n-hexyl and the like.

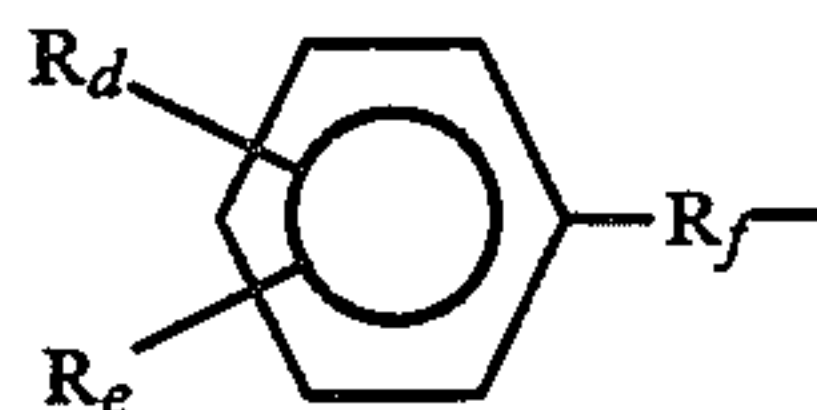
The term "lower alkoxy" refers to the group  $-\text{OR}_a$  wherein  $R_a$  is lower alkyl. Typical lower alkoxy groups include methoxy, ethoxy, and the like.

The term "alkaryl" refers to the group:



wherein  $R_b$  and  $R_c$  are each independently hydrogen or an alkyl group, with the proviso that both  $R_b$  and  $R_c$  are not hydrogen. Typical alkaryl groups include, for example, tolyl, xylyl, cumenyl, ethylphenyl, butylphenyl, dibutylphenyl, hexylphenyl, octylphenyl, dioctylphenyl, nonylphenyl, decylphenyl, didecylphenyl, dodecylphenyl, hexadecylphenyl, octadecylphenyl, icosylphenyl, tricontylphenyl and the like. The term "alkylphenyl" refers to an alkaryl group of the above formula in which  $R_b$  is alkyl and  $R_c$  is hydrogen.

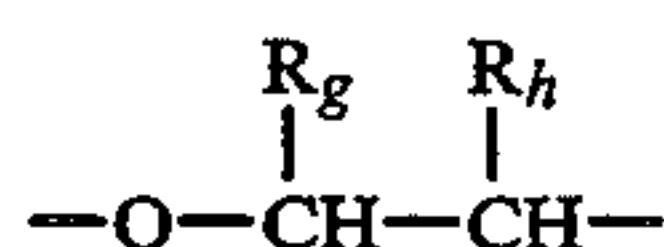
The term "aralkyl" refers to the group:



wherein  $R_d$  and  $R_e$  are each independently hydrogen or an alkyl group; and  $R_f$  is an alkylene group. Typical aralkyl groups include, for example, benzyl, methylbenzyl, dimethylbenzyl, phenethyl, and the like.

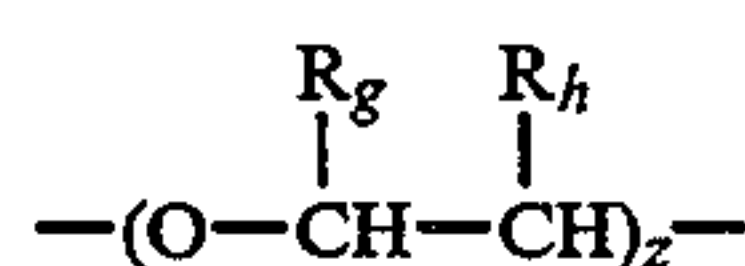
The term "hydrocarbyl" refers to an organic radical composed primarily of carbon and hydrogen which may be aliphatic, alicyclic, aromatic or combinations thereof, e.g., aralkyl or alkaryl. Such hydrocarbyl groups are generally relatively free of aliphatic unsaturation, i.e., olefinic or acetylenic unsaturation.

The term "oxyalkylene unit" refers to an ether moiety having the general formula:



wherein  $R_g$  and  $R_h$  are each independently hydrogen or lower alkyl groups.

The term "poly(oxyalkylene)" refers to a polymer or oligomer having the general formula:



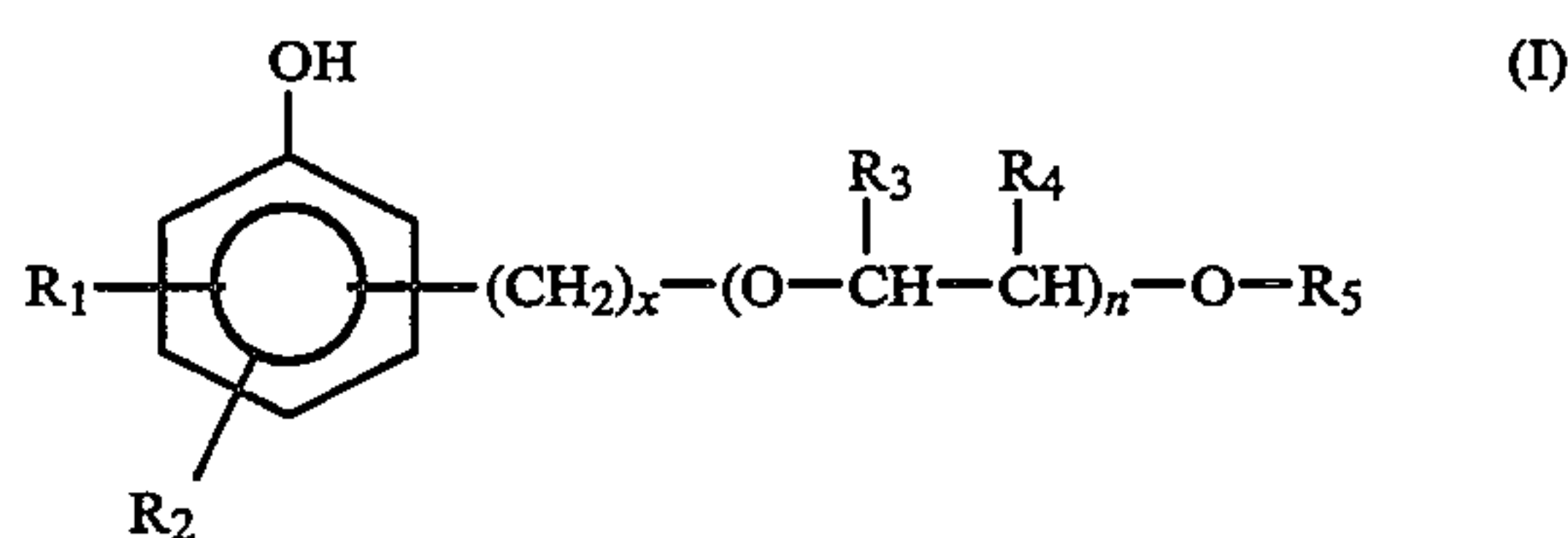
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wherein  $R_g$  and  $R_h$  are as defined above, and  $z$  is an integer greater than 1. When referring herein to the number of poly(oxyalkylene) units in a particular poly(oxyalkylene) compound, it is to be understood that this number refers to the average number of poly(oxyalkylene) units in such compounds unless expressly stated to the contrary.

### The Poly(oxyalkylene) Hydroxyaromatic Ether

The poly(oxyalkylene) hydroxyaromatic ether component of the present invention has the general formula:

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or a fuel-soluble salt thereof; wherein  $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$ ,  $R_5$ ,  $n$  and  $x$  are as defined hereinabove.

Preferably,  $R_1$  is hydrogen, hydroxy, or lower alkyl having 1 to 4 carbon atoms. More preferably,  $R_1$  is hydrogen or hydroxy. Most preferably,  $R_1$  is hydrogen.

$R_2$  is preferably hydrogen.

Preferably, one of  $R_3$  and  $R_4$  is lower alkyl having 1 to 3 carbon atoms and the other is hydrogen. More preferably, one of  $R_3$  and  $R_4$  is methyl or ethyl and the other is hydrogen. Most preferably, one of  $R_3$  and  $R_4$  is ethyl and the other is hydrogen.

$R_5$  is preferably hydrogen, alkyl having 2 to 22 carbon atoms, alkylphenyl having an alkyl group containing 4 to 24 carbon atoms, or an acyl group having the formula:  $-\text{C}(\text{O})\text{R}_7$ , wherein  $R_7$  is alkyl having 4 to 12 carbon atoms. More preferably,  $R_5$  is hydrogen, alkyl having 4 to 12 carbon atoms, or alkylphenyl having an alkyl group containing 4 to 12 carbon atoms. Most preferably,  $R_5$  is hydrogen.

Preferably,  $n$  is an integer from 10 to 50. More preferably,  $n$  is an integer from 15 to 30. Preferably,  $x$  is an integer from 0 to 2. More preferably,  $x$  is 0.

A preferred group of poly(oxyalkylene) hydroxyaromatic ethers for use in this invention are those of formula I wherein  $R_1$  is hydrogen, hydroxy, or lower alkyl having 1 to 4 carbon atoms;  $R_2$  is hydrogen; one of  $R_3$  and  $R_4$  is hydrogen and the other is methyl or ethyl;  $R_5$  is hydrogen, alkyl having 4 to 12 carbon atoms, alkylphenyl having an alkyl group containing 4 to 12 carbon atoms, or an acyl group having the formula:  $-\text{C}(\text{O})\text{R}_7$ , wherein  $R_7$  is alkyl having 4 to 12 carbon atoms;  $n$  is 15 to 30 and  $x$  is 0.

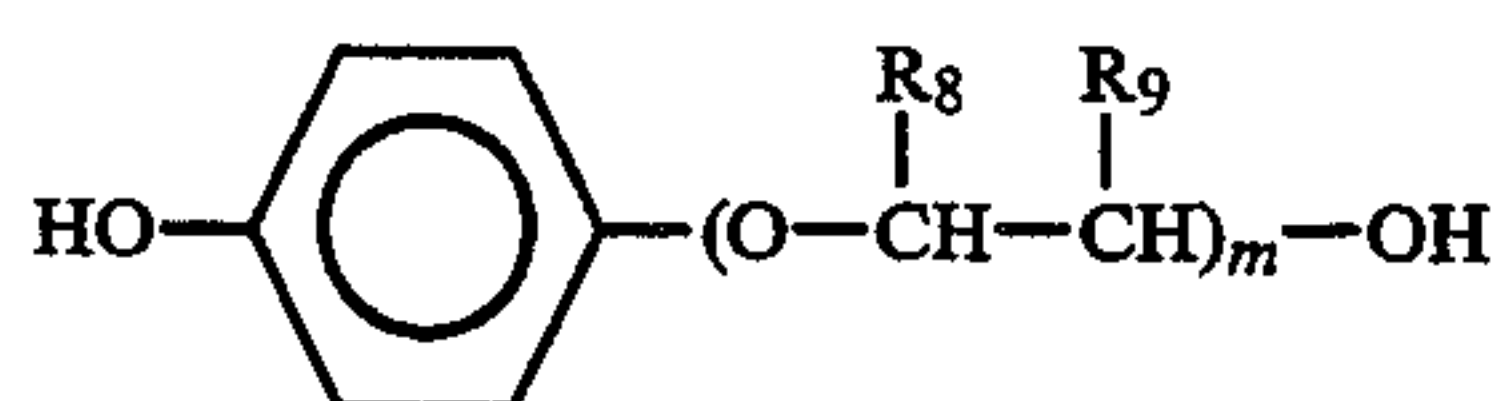
Another preferred group of poly(oxyalkylene) hydroxyaromatic ethers for use in this invention are those of formula I wherein  $R_1$  is hydrogen, hydroxy, or lower alkyl having 1 to 4 carbon atoms;  $R_2$  is hydrogen; one of  $R_3$  and  $R_4$  is hydrogen and the other is methyl or ethyl;  $R_5$  is hydrogen, alkyl having 4 to 12 carbon atoms, alkylphenyl having an alkyl group containing 4 to 12 carbon atoms, or an acyl group having the formula:  $-\text{C}(\text{O})\text{R}_7$ , wherein  $R_7$  is alkyl having 4 to 12 carbon atoms;  $n$  is 15 to 30 and  $x$  is 1 or 2.

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A more preferred group of poly(oxyalkylene) hydroxyaromatic ethers for use in this invention are those of formula I wherein  $R_1$  is hydrogen or hydroxy;  $R_2$  is hydrogen; one of  $R_3$  and  $R_4$  is hydrogen and the other is methyl or ethyl;  $R_5$  is hydrogen, alkyl having 4 to 12 carbon atoms, or alkylphenyl having an alkyl group containing 4 to 12 carbon atoms;  $n$  is 15 to 30; and  $x$  is 0.

A particularly preferred group of poly(oxyalkylene) hydroxyaromatic ethers for use in this invention are those having the formula:



wherein one of  $R_8$  and  $R_9$  is methyl or ethyl and the other is hydrogen; and  $m$  is an integer from 15 to 30.

The poly(oxyalkylene) hydroxyaromatic ether component of the present fuel additive composition will generally have a sufficient molecular weight so as to be non-volatile at normal engine intake valve operating temperatures (about 200°–250° C.). Typically, the molecular weight of the poly(oxyalkylene) hydroxyaromatic ether component will range from about 600 to about 10,000, preferably from 1,000 to 3,000.

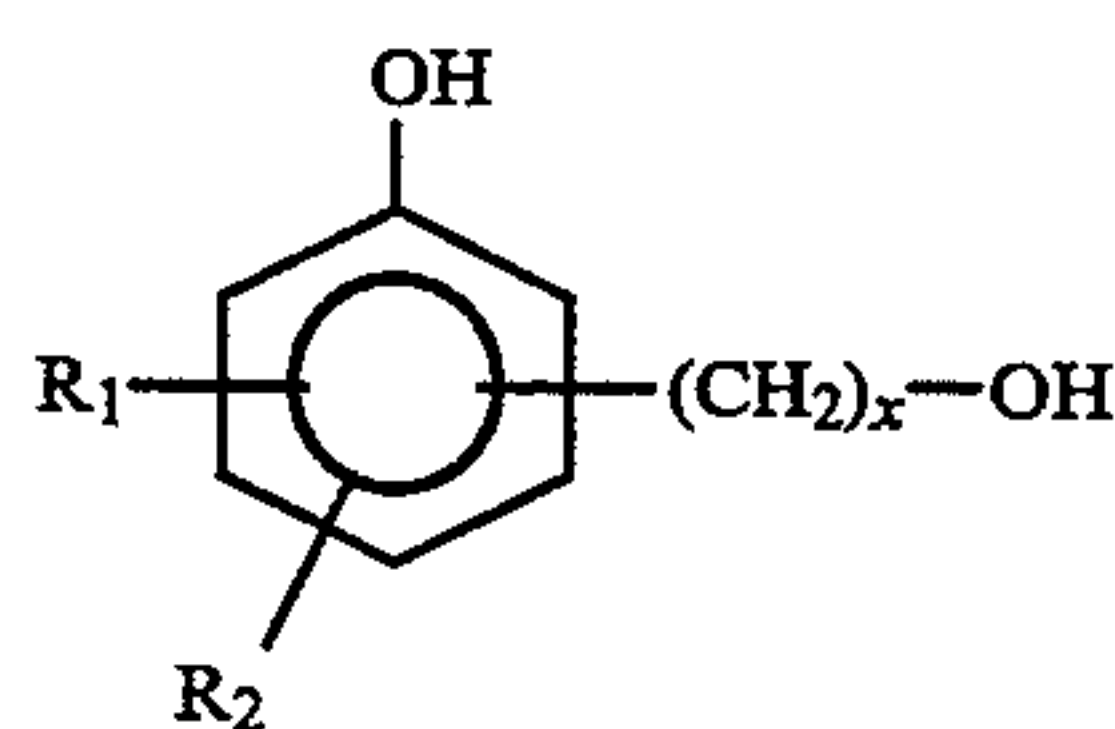
Generally, the poly(oxyalkylene) hydroxyaromatic ethers employed in this invention will contain an average of about 5 to about 100 oxyalkylene units; preferably, 10 to 50 oxyalkylene units; more preferably, 15 to 30 oxyalkylene units.

Fuel-soluble salts of poly(oxyalkylene) hydroxyaromatic ethers are also contemplated to be useful in the fuel additive composition of the present invention. Such salts include alkali metal, alkaline earth metal, ammonium, substituted ammonium and sulfonium salts. Preferred metal salts are the alkali metal salts, particularly the sodium and potassium salts, and the substituted ammonium salts, particularly tetraalkyl-substituted ammonium salts, such as the tetrabutylammonium salts.

#### General Synthetic Procedures

The poly(oxyalkylene) hydroxyaromatic ether component of the present fuel additive composition may be prepared by the following general methods and procedures. It should be appreciated that where typical or preferred process conditions (e.g. reaction temperatures, times, mole ratios of reactants, solvents, pressures, etc.) are given, other process conditions may also be used unless otherwise stated. Optimum reaction conditions may vary with the particular reactants or solvents used, but such conditions can be determined by one skilled in the art by routine optimization procedures.

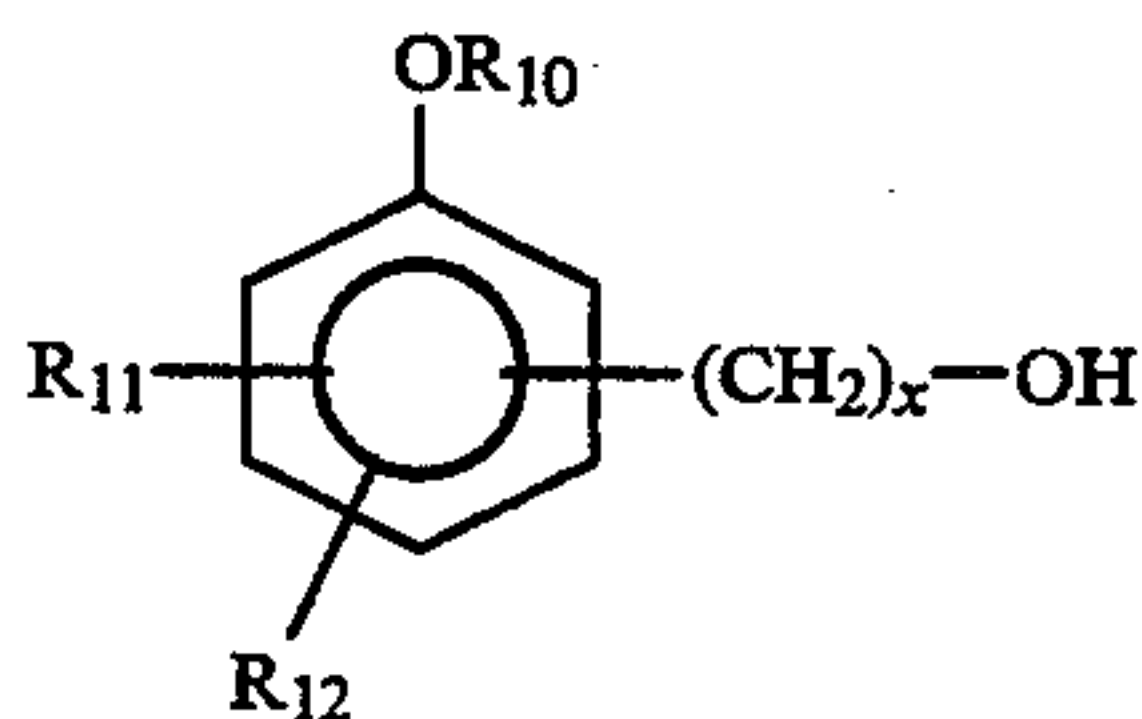
The poly(oxyalkylene) hydroxyaromatic ethers employed in the present fuel additive composition may be prepared from a hydroxyaromatic compound having the formula:



wherein  $R_1$ ,  $R_2$ , and  $x$  are as defined above.

The hydroxyaromatic compounds of formula III are either known compounds or can be prepared from known compounds by conventional procedures. Suitable hydroxyaromatic compounds for use as starting materials in this invention include catechol, resorcinol, hydroquinone, 1,2,3-trihydroxybenzene (pyrogallol), 1,2,4-trihydroxybenzene (hydroquinol), 1,3,5-trihydroxybenzene (phloroglucinol), 1,4-dihydroxy-2-methylbenzene, 1,3-dihydroxy-5-methylbenzene, 2-t-butyl-1,4-dihydroxybenzene, 2,6-di-t-butyl-1,4-dihydroxybenzene, 1,4-dihydroxy-2-methoxybenzene, 1,3-dihydroxy-5-methoxybenzene, 4-hydroxybenzyl alcohol, 4-hydroxyphenethyl alcohol and the like.

In a preferred method of synthesizing the poly(oxyalkylene) hydroxyaromatic ether component of the present fuel additive composition, a hydroxyaromatic compound of formula III is first selectively protected to provide a compound having the formula:



wherein  $R_{10}$  is a suitable hydroxyl protecting group, such as benzyl, tert-butyldimethylsilyl, methoxymethyl, and the like;  $R_{11}$  and  $R_{12}$  are each independently hydrogen, lower alkyl, lower alkoxy, or the group  $-\text{OR}_{13}$ , wherein  $R_{13}$  is a suitable hydroxyl protecting group, such as benzyl, tert-butyldimethylsilyl, methoxymethyl, and the like. Preferably,  $R_{10}$  and  $R_{13}$  are benzyl; except in the case where  $x$  is 1, then  $R_{10}$  and  $R_{13}$  are preferably a tert-butyldimethylsilyl group.

Selective protection of III may be accomplished using conventional procedures. The choice of a suitable protecting group for a particular hydroxyaromatic compound will be apparent to those skilled in the art. Various protecting groups, and their introduction and removal, are described, for example, in T. W. Greene and P. G. M. Wuts, *Protective Groups in Organic Synthesis*, Second Edition, Wiley, New York, 1991, and references cited therein. Alternatively, the protected derivatives IV can be prepared from known starting materials other than the hydroxyaromatic compounds of formula III by conventional procedures. In some cases, the protected derivatives IV are commercially available, e.g. 4-benzyloxyphenol is commercially available from Aldrich Chemical Co., Milwaukee, Wis. 53233.

The protected hydroxyaromatic compound of formula IV is then deprotonated with a suitable base to provide a metal salt having the formula:







deprotonation reaction will be conducted in an inert solvent, such as toluene, tetrahydrofuran, and the like, under substantially anhydrous conditions at a temperature in the range from  $-10^{\circ}\text{C}$ . to  $120^{\circ}\text{C}$ . for about 0.25 to about 3 hours. The resulting metal salt is then contacted with about 0.90 to about 1.1 molar equivalents of a suitable alkylating agent at a temperature in the range from  $0^{\circ}\text{C}$ . to  $120^{\circ}\text{C}$ . for about 1 to about 50 hours to afford, after neutralization, a poly(oxyalkylene) hydroxyaromatic ether of formula IX.

Suitable alkylating agents for use in this reaction include alkyl and aralkyl halides, such as alkyl chlorides, bromides and iodides and aralkyl chlorides, bromides and iodides; and alkyl and aralkyl sulfonates, such as alkyl mesylates and tosylates, and aralkyl mesylates and tosylates.

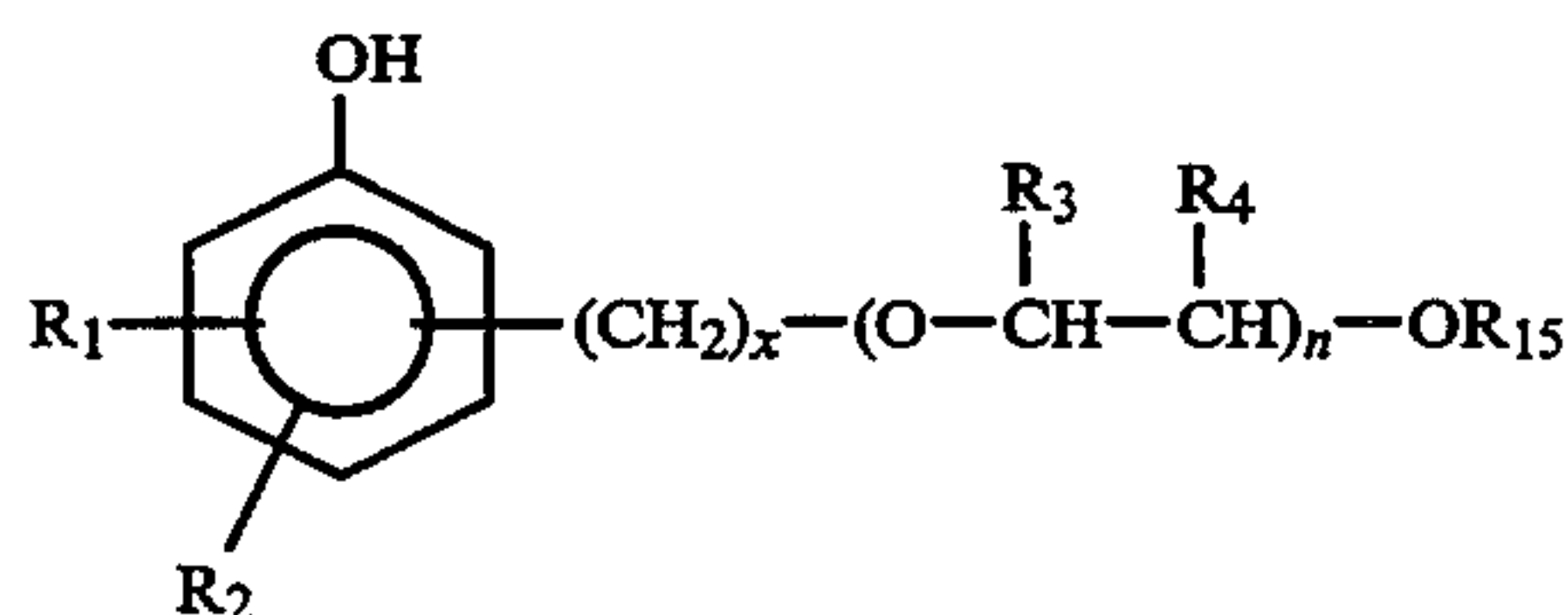
Preferred alkylating agents are primary and secondary alkyl halides having 1 to 30 carbon atoms, and primary and secondary aralkyl halides having 7 to 36 carbon atoms; more preferred alkylating agents are primary alkyl halides having 4 to 12 carbon atoms.

Representative examples of alkylating agents include, but are not limited to, methyl iodide, ethyl iodide, n-propyl bromide, n-butyl bromide, n-pentyl bromide, n-hexyl chloride, n-octyl chloride, n-decyl chloride, benzyl chloride and phenethyl chloride. Particularly preferred alkylating agents are benzyl chloride, n-butyl bromide.

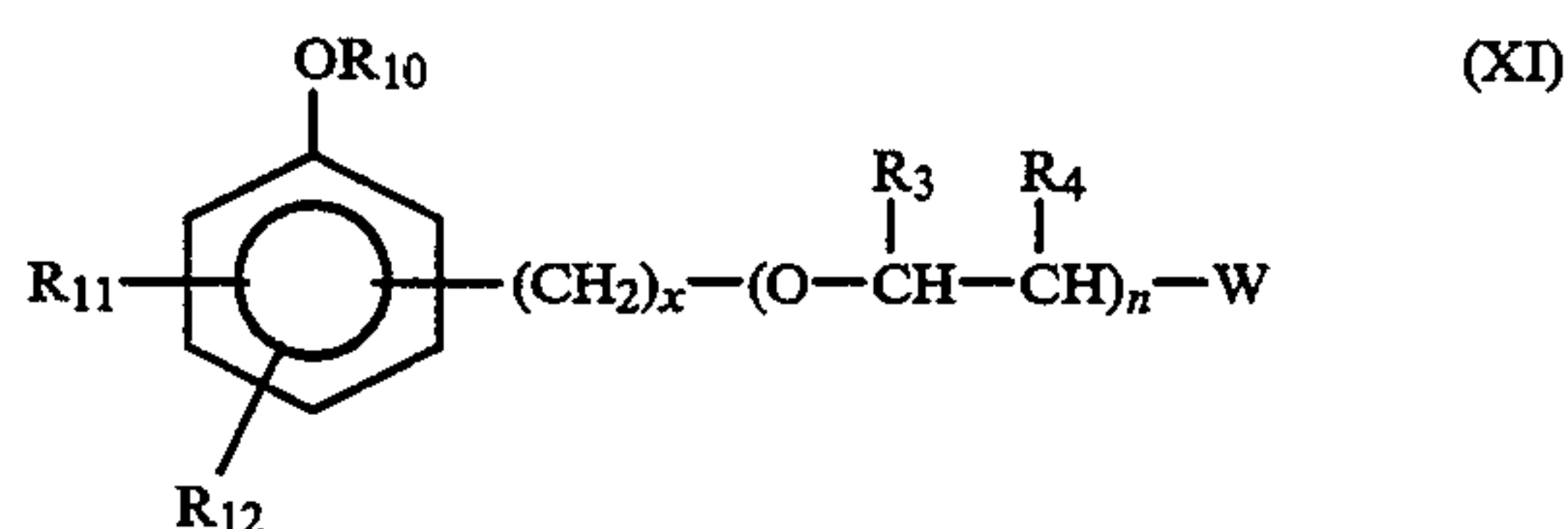
Alternatively, poly(oxyalkylene) hydroxyaromatic ethers of formula IX may be prepared by alkylating the hydroxyl group of the poly(oxyalkylene) moiety of protected intermediate VII, and then deprotecting the resulting product. The conditions for alkylating intermediate VII are essentially the same as those described above; however, a lesser amount of base will be required since the aromatic hydroxyl groups of VII are in a protected form.

Other suitable methods for preparing alkyl and aralkyl ethers from alcohols, and appropriate reaction conditions for such reactions, can be found, for example, in I. T. Harrison and S. Harrison, *Compendium of Organic Synthetic Methods*, Vol. 1, pp. 310-312, Wiley-Interscience, New York (1971) and references cited therein.

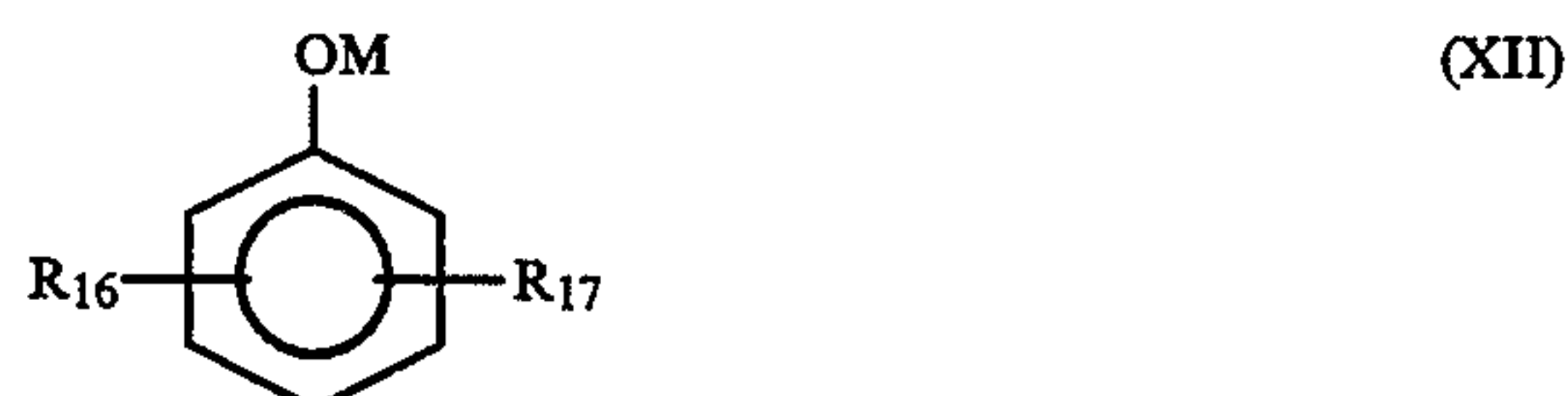
The poly(oxyalkylene) hydroxyaromatic ethers employed in the present fuel additive composition that contain an alkaryl ether moiety, i.e. those having the formula:



wherein  $\text{R}_1$ - $\text{R}_4$ ,  $n$  and  $x$  are as defined above, and  $\text{R}_{15}$  is a phenyl or alkaryl group, may be prepared from intermediate VII in several steps by first converting the hydroxyl group present of the poly(oxyalkylene) moiety of VII into a suitable leaving group, i.e. forming an intermediate having the formula:



wherein  $\text{R}_3$ ,  $\text{R}_4$ ,  $\text{R}_{10}$ ,  $\text{R}_{11}$ ,  $\text{R}_{12}$ ,  $n$  and  $x$  are as defined above, and  $\text{W}$  is a suitable leaving group; and then displacing the leaving group of XI with a metal salt of a phenol having the formula:



wherein  $\text{R}_{16}$  and  $\text{R}_{17}$  are each independently hydrogen or an alkyl group. Subsequent deprotection of the resulting product affords poly(oxyalkylene) hydroxyaromatic ethers of formula X.

The hydroxyl group of the poly(oxyalkylene) moiety of VII may be converted into a suitable leaving group by contacting VII with a sulfonyl chloride to form a sulfonate ester, such as a methanesulfonate (mesylate) or a toluenesulfonate (tosylate). Typically, this reaction is conducted in the presence of a suitable amine, such as triethylamine or pyridine, in an inert solvent, such as dichloromethane, at a temperature in the range of about  $-10^{\circ}\text{C}$ . to about  $30^{\circ}\text{C}$ . Alternatively, the hydroxyl group of the poly(oxyalkylene) moiety of VII can be exchanged for a halide, such chloride or bromide, by contacting VII with a halogenating agent, such as thionyl chloride, oxalyl chloride or phosphorus tribromide. Other suitable methods for preparing sulfonates and halides from alcohols, and appropriate reaction conditions for such reactions, can be found, for example, in I. T. Harrison and S. Harrison, *Compendium of Organic Synthetic Methods*, Vol. 1, pp. 331-337, Wiley-Interscience, New York (1971) and references cited therein.

After forming intermediate XI, the leaving group may be displaced therefrom by contacting XI with metal salt XII. Generally, this reaction will be conducted in an inert solvent, such as toluene, tetrahydrofuran and the like, under substantially anhydrous conditions at a temperature in the range of about  $25^{\circ}\text{C}$ . to about  $150^{\circ}\text{C}$ . for about 1 to about 48 hours. The metal salt XII can be formed by contacting the corresponding phenol with a strong base capable of abstracting the proton from the phenolic hydroxyl group, such as sodium hydride, potassium hydride, sodium amide and the like, in an inert solvent.

Suitable phenolic compounds for use in this reaction include phenol, monoalkyl-substituted phenols and dialkyl-substituted phenols. Monoalkyl-substituted phenols are preferred, especially monoalkylphenols having an alkyl substituent in the para position. Representative examples of suitable phenolic compounds include, but are not limited to, phenol, methylphenol, dimethylphenol, ethylphenol, butylphenol, octylphenol, decylphenol, dodecylphenol, tetradecylphenol, hexadecylphenol, octadecylphenol, eicosylphenol, tetracosylphenol, hexacosylphenol, triacontylphenol and the like. Also, mixtures of alkylphenols may be employed, such



as a mixture of C<sub>14</sub>-C<sub>18</sub> alkylphenols, a mixture of C<sub>18</sub>-C<sub>24</sub> alkylphenols, a mixture of C<sub>20</sub>-C<sub>24</sub> alkylphenols, or a mixture of C<sub>16</sub>-C<sub>26</sub> alkylphenols.

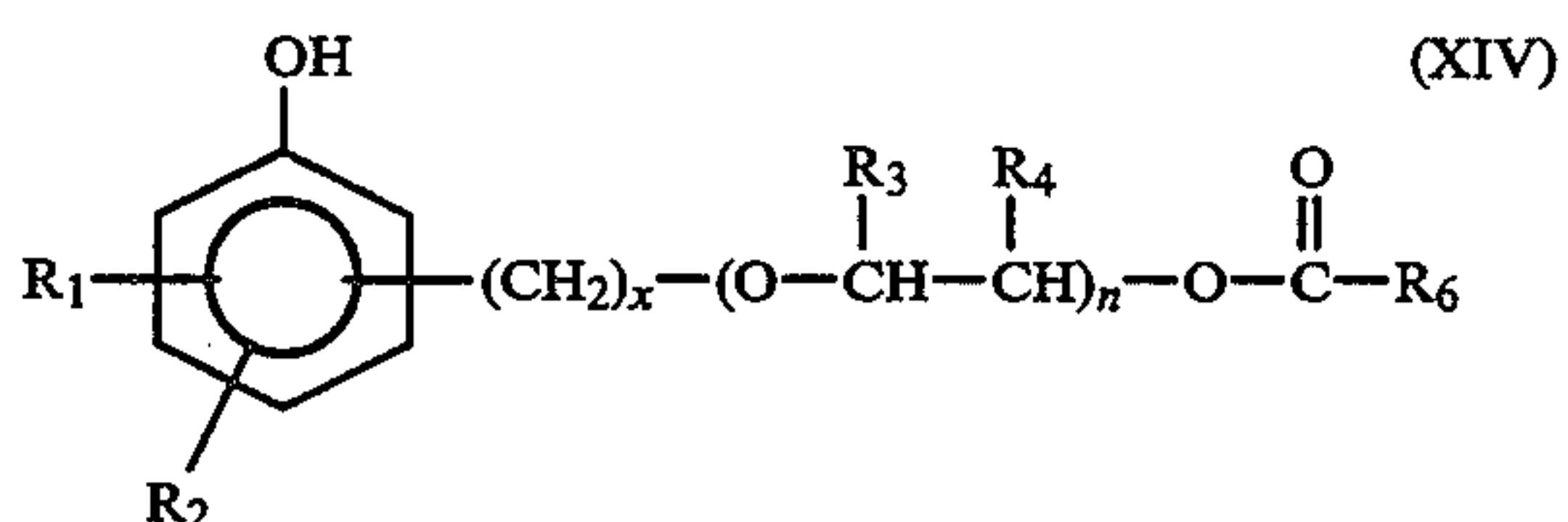
Particularly preferred alkylphenols are those derived from alkylation of phenol with polymers or oligomers of C<sub>3</sub> to C<sub>6</sub> olefins, such as polypropylene or polybutene. These polymers preferably contain 10 to 30 carbon atoms. An especially preferred alkylphenol is prepared by alkylating phenol with a propylene polymer having an average of 4 units. This polymer has the common name of propylene tetramer and is commercially available.

Alternatively, the poly(oxyalkylene) hydroxyaromatic ethers of formula X can be prepared by displacing a leaving group from an intermediate having the formula:



wherein R<sub>3</sub>, R<sub>4</sub>, R<sub>15</sub>, n and x are as defined above, and W is a suitable leaving group, with metal salt V; and then deprotecting the resulting product. Conditions for this reaction are essentially the same as those described above for reaction of XI with XII. Compounds of formula XIII may be prepared from XII and VI using the conditions described above for the preparation of VII, followed by conversion of the hydroxyl group of the poly(oxyalkylene) moiety of the resulting product into a suitable leaving using the procedures described above for the preparation of XI.

The poly(oxyalkylene) hydroxyaromatic ethers employed in the present fuel additive composition that contain an acyl moiety, i.e. those having the formula:



wherein R<sub>1</sub>-R<sub>4</sub>, R<sub>6</sub>, n and x are as defined above; may be prepared from intermediate VII by first acylating the hydroxyl group of the poly(oxyalkylene) moiety of VII to form an ester. Subsequent deprotection of the aromatic hydroxyl group(s) of the resulting ester using conventional procedures then affords poly(oxyalkylene) hydroxyaromatic ethers of formula XIV.

Generally, the acylation reaction will be conducted by contacting intermediate VII with about 0.95 to about 1.2 molar equivalents of a suitable acylating agent. Suitable acylating agents for use in this reaction include acyl halides, such as acyl chlorides and bromides; and carboxylic acid anhydrides. Preferred acylating agents are those having the formula: R<sub>6</sub>C(O)-X, wherein R<sub>6</sub> is alkyl having 1 to 30 carbon atom, phenyl, or aralkyl or alkaryl having 7 to 36 carbon atoms, and X is chloro or bromo. More preferred acylating agents are those having the formula: R<sub>7</sub>C(O)-X, wherein R<sub>7</sub> is alkyl having 4 to 12 carbon atoms. Representative examples of suitable acylating agents include, but are not limited to, acetyl chloride, acetic anhydride, propionyl chloride, butanoyl chloride, pivaloyl chloride, octanoyl chloride, decanoyl chloride 4-butylbenzoyl chloride and the like.

Generally, this reaction is conducted in an inert solvent, such as toluene, dichloromethane, diethyl ether

and the like, at a temperature in the range of about 25° C. to about 150° C., and is generally complete in about 0.5 to about 48 hours. When an acyl halide is employed as the acylating agent, this reaction is preferably conducted in the presence of a sufficient amount of an amine capable of neutralizing the acid generated during the reaction, such as triethylamine, di(isopropyl)ethylamine, pyridine or 4-dimethylaminopyridine.

Additional methods for preparing esters from alcohols, and suitable reaction conditions for such reactions, can be found, for example, in I. T. Harrison and S. Harrison, *Compendium of Organic Synthetic Methods*, Vol. 1, pp. 273-276 and 280-283, Wiley-Interscience, New York (1971) and references cited therein.

#### The Poly(oxyalkylene) Amine

The poly(oxyalkylene) amine component of the present fuel additive composition is a poly(oxyalkylene) amine having at least one basic nitrogen atom and a sufficient number of oxyalkylene units to render the poly(oxyalkylene) amine soluble in hydrocarbons boiling in the gasoline or diesel range.

Preferably, such poly(oxyalkylene) amines will also be of sufficient molecular weight so as to be nonvolatile at normal engine intake valve operating temperatures, which are generally in the range of about 200° C. to 250° C.

Generally, the poly(oxyalkylene) amines suitable for use in the present invention will contain at least about 5 oxyalkylene units, preferably about 5 to 100, more preferably about 8 to 100, and even more preferably about 10 to 100. Especially preferred poly(oxyalkylene) amines will contain about 10 to 25 oxyalkylene units.

The molecular weight of the presently employed poly(oxyalkylene) amines will generally range from about 500 to about 10,000, preferably from about 500 to about 5,000.

Suitable poly(oxyalkylene) amine compounds for use in the present invention include hydrocarbyl poly(oxyalkylene) polyamines as disclosed, for example, in U.S. Pat. No. 4,247,301, issued Jan. 27, 1981 to Honnen, the disclosure of which is incorporated herein by reference. These compounds are hydrocarbyl poly(oxyalkylene) polyamines wherein the poly(oxyalkylene) moiety comprises at least one hydrocarbyl-terminated poly(oxyalkylene) chain of 2 to 5 carbon atom oxyalkylene units, and wherein the poly(oxyalkylene) chain is bonded through a terminal carbon atom to a nitrogen atom of a polyamine having from 2 to about 12 amine nitrogen atoms and from 2 to about 40 carbon atoms with a carbon-to-nitrogen ratio between about 1:1 and 10:1. The hydrocarbyl group on these hydrocarbyl poly(oxyalkylene) polyamines will contain from about 1 to 30 carbon atoms. These compounds generally have molecular weights in the range of about 500 to 10,000, preferably from about 500 to 5,000 and more preferably from about 800 to 5,000.

The above-described hydrocarbyl poly(oxyalkylene) polyamines are prepared by conventional procedures known in the art, as taught, for example, in U.S. Pat. No. 4,247,301.

Other poly(oxyalkylene) amines suitable for use in the present invention are the poly(oxyalkylene) polyamines wherein the poly(oxyalkylene) moiety is connected to the polyamine moiety through an oxyalkylene hydroxy-type linkage derived from an epihalohydrin, such as epichlorohydrin or epibromohydrin. This type



of poly(oxyalkylene) amine having an epihalohydrin-derived linkage is described, for example, in U.S. Pat. No. 4,261,704, issued Apr. 14, 1981 to Langdon, the disclosure of which is incorporated herein by reference.

Useful polyamines for preparing the epihalohydrin-derived poly(oxyalkylene) polyamines include, for example, alkylene polyamines, polyalkylene polyamines, cyclic amines, such as piperazines, and amino-substituted amines. The poly(oxyalkylene) polyamines having an epihalohydrin-derived linkage between the poly(oxyalkylene) and polyamine moieties are prepared using known procedures as taught, for example, in U.S. Pat. No. 4,261,704.

Another type of poly(oxyalkylene) amine useful in the present invention is a highly branched alkyl poly(oxyalkylene) monoamine as described, for example in U.S. Pat. No. 5,094,667, issued Mar. 10, 1992 to Schilowitz et al., the disclosure of which is incorporated herein by reference. These highly branched alkyl poly(oxyalkylene) monoamines have the general formula:



wherein  $R_{18}$  is a highly branched alkyl group containing from 12 to 40 carbon atoms, preferably an alkyl group having 20 carbon atoms which is derived from a Guerbet condensation reaction, and  $p$  is a number up to 30, preferably 4 to 8. The preferred alkyl group is derived from a Guerbet alcohol containing 20 carbon atoms having the formula:



wherein  $R_{19}$  is a hydrocarbyl chain.

The above highly branched alkyl poly(oxyalkylene) monoamines are prepared by using known methods as disclosed, for example, in U.S. Pat. No. 5,094,667.

A particularly preferred class of poly(oxyalkylene) amine for use in the fuel additive composition of the present invention are the hydrocarbyl-substituted poly(oxyalkylene) aminocarbamates disclosed, for example, in U.S. Pat. Nos. 4,288,612; 4,236,020; 4,160,648; 4,191,537; 4,270,930; 4,233,168; 4,197,409; 4,243,798 and 4,881,945, the disclosure of each of which are incorporated herein by reference.

These hydrocarbyl poly(oxyalkylene) aminocarbamates contain at least one basic nitrogen atom and have an average molecular weight of about 500 to 10,000, preferably about 500 to 5,000, and more preferably about 1,000 to 3,000. As described more fully hereinbelow, these hydrocarbyl poly(oxyalkylene) aminocarbamates contain a (a) poly(oxyalkylene) moiety, (b) an amine moiety and (c) a carbamate connecting group.

#### A. The Poly(oxyalkylene) Moiety

The hydrocarbyl-terminated poly(oxyalkylene) polymers which are utilized in preparing the hydrocarbyl poly(oxyalkylene) aminocarbamates employed in the present invention are monohydroxy compounds, e.g., alcohols, often termed monohydroxy polyethers, or polyalkylene glycol monocarbyl ethers, or "capped" poly(oxyalkylene) glycols, and are to be distinguished from the poly(oxyalkylene) glycols (diols), or polyols, which are not hydrocarbyl-terminated, i.e., are not capped. These hydrocarbyl poly(oxyalkylene) alcohols may be produced under conditions essentially the same as those described above for the preparation of VII, i.e.

by the addition of lower alkylene oxides, such as ethylene oxide, propylene oxide, butylene oxide, etc. to a hydroxy compound,  $R_{20}OH$ , under polymerization conditions, wherein  $R_{20}$  is the hydrocarbyl group which caps the poly(oxyalkylene) chain.

In the hydrocarbyl poly(oxyalkylene) aminocarbamates employed in the present invention, the group  $R_{20}$  will generally contain from 1 to about 30 carbon atoms, preferably from 2 to about 20 carbon atoms and is preferably aliphatic or aromatic, i.e., an alkyl or alkyl phenyl wherein the alkyl is a straight or branched-chain of from 1 to about 24 carbon atoms. More preferably,  $R_{20}$  is alkylphenyl wherein the alkyl group is a branched-chain of 12 carbon atoms, derived from propylene tetramer, and commonly referred to as tetrapropenyl.

The oxyalkylene units in the poly(oxyalkylene) moiety preferably contain from 2 to about 5 carbon atoms but one or more units of a larger carbon number may also be present. Generally, each poly(oxyalkylene) polymer contains at least about 5 oxyalkylene units, preferably about 5 to about 100 oxyalkylene units, more preferably about 8 to about 100 units, even more preferably about 10 to 100 units, and most preferably 10 to about 25 such units. The poly(oxyalkylene) moiety of the hydrocarbyl poly(oxyalkylene) aminocarbamates employed in the present invention is more fully described and exemplified in U.S. Pat. No. 4,191,537, issued Mar. 4, 1980 to Lewis, the disclosure of which is incorporated herein by reference.

Although the hydrocarbyl group on the hydrocarbyl poly(oxyalkylene) moiety will preferably contain from 1 to about 30 carbon atoms, longer hydrocarbyl groups, particularly longer chain alkyl phenyl groups, may also be employed. For example, alkylphenyl poly(oxyalkylene) aminocarbamates wherein the alkyl group contains at least 40 carbon atoms, as described in U.S. Pat. No. 4,881,945, issued Nov. 21, 1989 to Buckley, are also contemplated for use in the present invention. The alkyl phenyl group on the aminocarbamates of U.S. Pat. No. 4,881,945 will preferably contain an alkyl group of 50 to 200 carbon atoms, and more preferably, an alkyl group of 60 to 100 carbon atoms. The disclosure of U.S. Pat. No. 4,881,945 is incorporated herein by reference.

Also contemplated for use in the present invention are alkylphenyl poly(oxypropylene) aminocarbamates wherein the alkyl group is a substantially straight-chain alkyl group of about 25 to 50 carbon atoms derived from an alpha olefin oligomer of  $C_8$  to  $C_{20}$  alpha olefins, as described in PCT International Patent Application Publication No. WO 90/07564, published Jul. 12, 1990, the disclosure of which is incorporated herein by reference.

#### B. The Amine Moiety

The amine moiety of the hydrocarbyl poly(oxyalkylene) aminocarbamate is preferably derived from a polyamine having from 2 to about 12 amine nitrogen atoms and from 2 to about 40 carbon atoms.

The polyamine is preferably reacted with a hydrocarbyl poly(oxyalkylene) chloroformate to produce the hydrocarbyl poly(oxyalkylene) aminocarbamate fuel additive finding use within the scope of the present invention. The chloroformate is itself derived from hydrocarbyl poly(oxyalkylene) alcohol by reaction with phosgene.

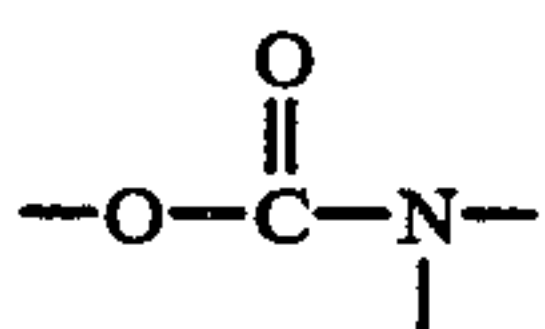


The polyamine provides the hydrocarbyl poly(oxyalkylene) aminocarbamate with, on the average, at least about one basic nitrogen atom per carbamate molecule, i.e., a nitrogen atom titratable by strong acid. The polyamine preferably has a carbon-to-nitrogen ratio of from about 1:1 to about 10:1. The polyamine may be substituted with substituents selected from hydrogen, hydrocarbyl groups of from 1 to about 10 carbon atoms, acyl groups of from 2 to about 10 carbon atoms, and monoketone, monohydroxy, mononitro, monocyano, alkyl and alkoxy derivatives of hydrocarbyl groups of from 1 to 10 carbon atoms. It is preferred that at least one of the basic nitrogen atoms of the polyamine is a primary or secondary amino nitrogen. The amine moiety of the hydrocarbyl poly(oxyalkylene) aminocarbamates employed in the present invention has been described and exemplified more fully in U.S. Pat. No. 4,191,537.

A more preferred polyamine for use in preparing the hydrocarbyl poly(oxyalkylene) aminocarbamates finding use within the scope of the present invention is a polyalkylene polyamine, including alkylenediamine, and including substituted polyamines, e.g., alkyl and hydroxyalkyl-substituted polyalkylene polyamine. Preferably, the alkylene group contains from 2 to 6 carbon atoms, there being preferably from 2 to 3 carbon atoms between the nitrogen atoms. Examples of such polyamines include ethylenediamine, diethylenetriamine, triethylenetetramine, di(trimethylene)triamine, dipropylenetriamine, tetraethylenepentamine, etc. Among the polyalkylene polyamines, polyethylene polyamine and polypropylene polyamine containing 2 to about 12 amine nitrogen atoms and 2 to about 24 carbon atoms are especially preferred and in particular, the lower polyalkylene polyamines, e.g., ethylenediamine, diethylenetriamine, propylenediamine, dipropylenetriamine, etc., are most preferred.

### C. The Aminocarbamate Connecting Group

The hydrocarbyl poly(oxyalkylene) aminocarbamate employed as the poly(oxyalkylene) amine component of the fuel additive composition of the present invention is obtained by linking the polyamine and the hydrocarbyl poly(oxyalkylene) alcohol together through a carbamate linkage, i.e.,



wherein the oxygen may be regarded as the terminal hydroxyl oxygen of the poly(oxyalkylene) alcohol, the nitrogen is derived from the polyamine and the carbonyl group  $\text{---C(O)---}$ , is preferably provided by a coupling agent, such as phosgene.

In a preferred method of preparation, the hydrocarbyl poly(oxyalkylene) alcohol is reacted with phosgene to produce a chloroformate and the chloroformate is reacted with the polyamine. Since there may be more than one nitrogen atom of the polyamine which is capable of reacting with the chloroformate, the carbamate product may contain more than one hydrocarbyl poly(oxyalkylene) moiety. It is preferred that the hydrocarbyl poly(oxyalkylene) aminocarbamate product contains on the average, about one poly(oxyalkylene) moiety per molecule (i.e., is a monocarbamate), although it is understood that this reaction route may lead to mix-

tures containing appreciable amounts of di- or higher poly(oxyalkylene) chain substitution on a polyamine containing several reactive nitrogen atoms.

A particularly preferred aminocarbamate is alkylphenyl poly(oxybutylene) aminocarbamate, wherein the amine moiety is derived from ethylene diamine or diethylene triamine. Synthetic methods to avoid higher degrees of substitution, methods of preparation, and other characteristics of the aminocarbamates used in the present invention are more fully described and exemplified in U.S. Pat. No. 4,191,537.

### Fuel Compositions

The fuel additive composition of the present invention will generally be employed in hydrocarbon fuels to prevent and control engine deposits, particularly intake valve deposits. The proper concentration of the additive composition necessary to achieve the desired level of deposit control varies depending upon the type of fuel employed, the type of engine, and the presence of other fuel additives.

Generally, the present fuel additive composition will be employed in hydrocarbon fuel in a concentration ranging from about 75 to about 5,000 parts per million (ppm) by weight, preferably from 200 to 2,500 ppm.

In terms of individual components, hydrocarbon fuel containing the fuel additive composition of this invention will generally contain about 50 to 2,500 ppm of the poly(oxyalkylene) hydroxyaromatic ether component and about 25 to 1,000 ppm of the poly(oxyalkylene) amine component. The ratio of the poly(oxyalkylene) hydroxyaromatic ether to poly(oxyalkylene) amine will generally range from about 0.5:1 to about 10:1, and will preferably be about 1:1 or greater.

The fuel additive composition of the present invention may be formulated as a concentrate using an inert stable oleophilic (i.e., dissolves in gasoline) organic solvent boiling in the range of about 150° F. to 400° F. (about 65° C. to 205° C.). Preferably, an aliphatic or an aromatic hydrocarbon solvent is used, such as benzene, toluene, xylene or higher-boiling aromatics or aromatic thinners. Aliphatic alcohols containing about 3 to 8 carbon atoms, such as isopropanol, isobutylcarbinol, n-butanol and the like, in combination with hydrocarbon solvents are also suitable for use with the present additives. In the concentrate, the amount of the additive composition will generally range from about 10 to about 70 weight percent, preferably 10 to 50 weight percent, more preferably from 20 to 40 weight percent.

In gasoline fuels, other fuel additives may be employed with the additives of the present invention, including, for example, oxygenates, such as t-butyl methyl ether, antiknock agents, such as methylcyclopentadienyl manganese tricarbonyl, and other dispersants/detergents, such as hydrocarbyl amines or succinimides. Additionally, antioxidants, metal deactivators and demulsifiers may be present.

In diesel fuels, other well-known additives can be employed, such as pour point depressants, flow improvers, cetane improvers, and the like.

A fuel-soluble, nonvolatile carrier fluid or oil may also be used with the fuel additive composition of this invention. The carrier fluid is a chemically inert hydrocarbon-soluble liquid vehicle which substantially increases the nonvolatile residue (NVR), or solvent-free liquid fraction of the fuel additive composition while not overwhelmingly contributing to octane require-



ment increase. The carrier fluid may be a natural or synthetic oil, such as mineral oil, refined petroleum oils, synthetic polyalkanes and alkenes, including hydrogenated and unhydrogenated polyalphaolefins, and synthetic Poly(oxyalkylene)-derived oils, such as those described, for example, in U.S. Pat. No. 4,191,537 to Lewis.

These carrier fluids are believed to act as a carrier for the fuel additive composition of the present invention and to assist in removing and retarding deposits. The carrier fluid may also exhibit synergistic deposit control properties when used in combination with the fuel additive composition of this invention.

The carrier fluids are typically employed in amounts ranging from about 100 to about 5000 ppm by weight of the hydrocarbon fuel, preferably from 400 to 3000 ppm of the fuel. Preferably, the ratio of carrier fluid to deposit control additive will range from about 0.5:1 to about 10:1, more preferably from 1:1 to 4:1, most preferably about 2:1.

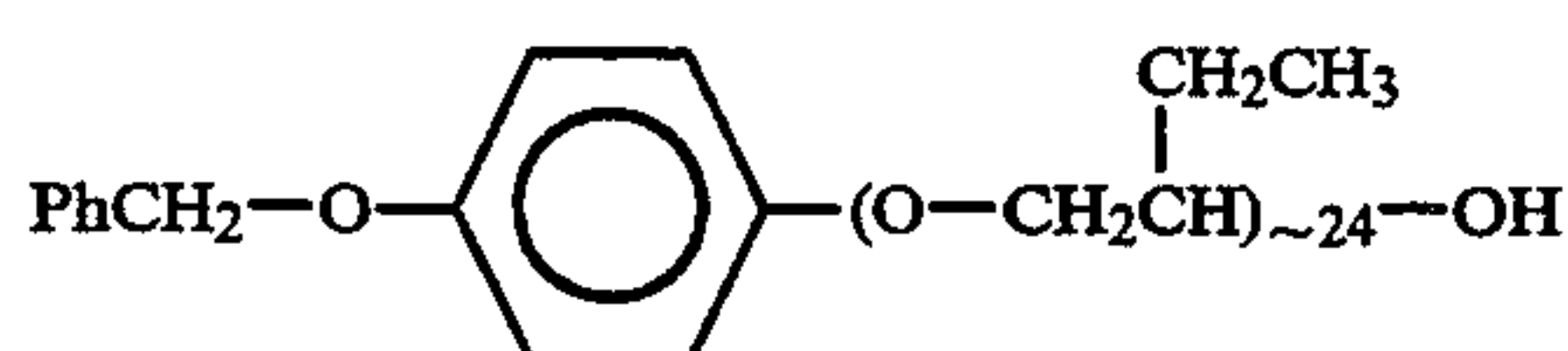
When employed in a fuel concentrate, carrier fluids will generally be present in amounts ranging from about 20 to about 60 weight percent, preferably from 30 to 50 weight percent.

#### EXAMPLES

The following examples are presented to illustrate specific embodiments of the present invention and synthetic preparations thereof; and should not be interpreted as limitations upon the scope of the invention.

##### EXAMPLE 1

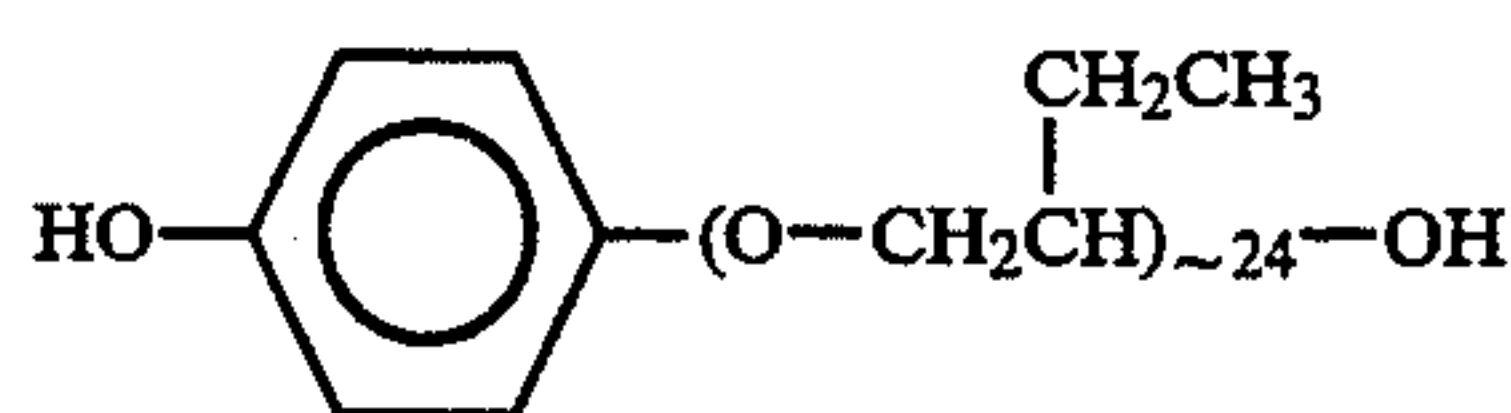
Preparation of  
 $\alpha$ -(4-Benzyloxyphenyl)- $\omega$ -hydroxypoly(oxybutylene)



To a flask equipped with a magnetic stirrer, thermometer, addition funnel, reflux condenser and nitrogen inlet was added 6.88 grams of a 35 wt % dispersion of potassium hydride in mineral oil. Forty grams of 4-benzyloxyphenol dissolved in 500 mL of anhydrous toluene was added dropwise and the resulting mixture was stirred at room temperature for ten minutes. The temperature of the reaction mixture, a thick white suspension, was raised to 90° C. and 430.8 mL of 1,2-epoxybutane was added dropwise. The reaction mixture was refluxed until the pot temperature reached 110° C. (approximately 48 hours) at which time the reaction mixture was a light brown clear solution. The reaction was cooled to room temperature, quenched with 50 mL of methanol and diluted with 1 liter of diethyl ether. The resulting mixture was washed with saturated aqueous ammonium chloride, followed by water and saturated aqueous sodium chloride. The organic layer was dried over anhydrous magnesium sulfate, filtered and the solvents removed in vacuo to yield 390 grams of a yellow oil. The oil was chromatographed on silica gel, eluting with hexane: diethyl ether (1:1), to yield 339.3 grams of the desired product as a colorless oil.

##### EXAMPLE 2

Preparation of  
 $\alpha$ -(4-Hydroxyphenyl)- $\omega$ -hydroxypoly(oxybutylene)



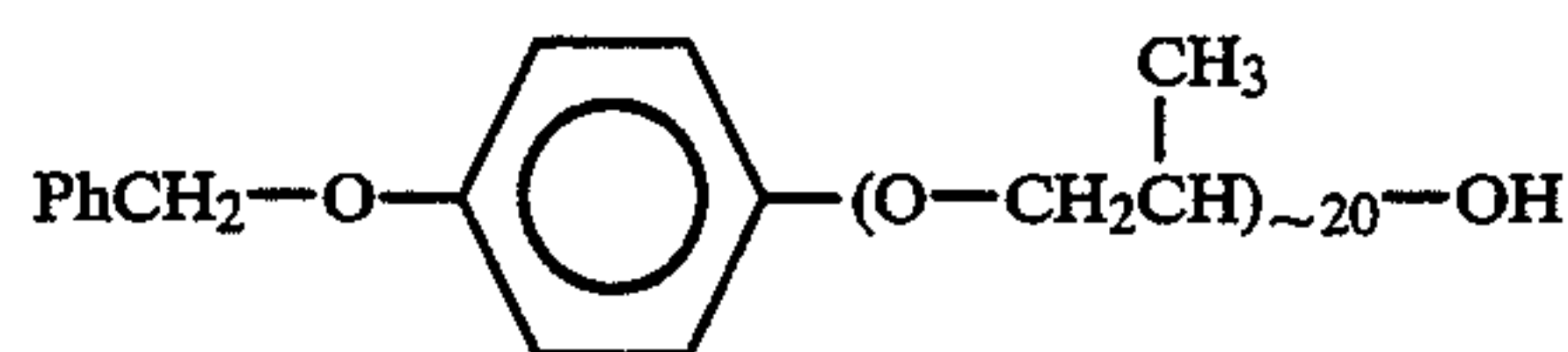
A solution of 54.10 grams of the product from Example 1 in 100 mL of ethyl acetate and 100 mL of acetic acid containing 5.86 grams of 10% palladium on charcoal was hydrogenolyzed at 35–40 psi for 16 hours on a Parr low-pressure hydrogenator. Catalyst filtration and removal of solvent in vacuo followed by azeotropic removal of residual acetic acid with toluene under vacuum yielded 48.1 grams of the desired product as a colorless oil. The product had an average of 24 oxybutylene units. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ 7.2 (broad s, 2H), 6.7 (s, 4H), 3.1–4.0 (m, 72H), 1.2–1.8 (m, 48H), 0.8 (t, 72H).

Similarly, by using the above procedures and the appropriate starting materials and reagents, the following compounds can be prepared:

$\alpha$ -(2-hydroxyphenyl)- $\omega$ -hydroxypoly(oxybutylene);  
 $\alpha$ -(3-hydroxyphenyl)- $\omega$ -hydroxypoly(oxybutylene);  
 $\alpha$ -(3-t-butyl-4-hydroxyphenyl)- $\omega$ -hydroxypoly(oxybutylene);  
 $\alpha$ -(4-hydroxy-3-methoxyphenyl)- $\omega$ -hydroxypoly(oxybutylene);  
 $\alpha$ -(3,4-dihydroxyphenyl)- $\omega$ -hydroxypoly(oxybutylene);  
 $\alpha$ -(3,4-hydroxy-5-methylphenyl)- $\omega$ -hydroxypoly(oxybutylene);  
 $\alpha$ -(3,5-di-t-butyl-4-hydroxyphenyl)- $\omega$ -hydroxypoly(oxybutylene); and  
 $\alpha$ -(3,4,5-trihydroxyphenyl)- $\omega$ -hydroxypoly(oxybutylene).

##### EXAMPLE 3

Preparation of  
 $\alpha$ -(4-Benzyloxyphenyl)- $\omega$ -hydroxypoly(oxypropylene)



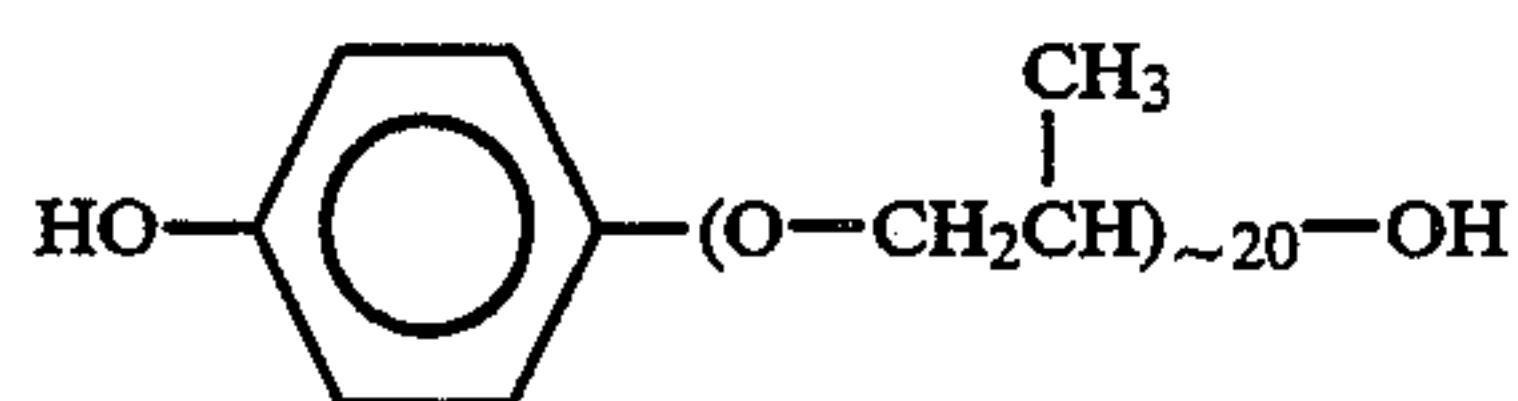
To a flask equipped with magnetic stirrer, thermometer, addition funnel, reflux condenser and nitrogen inlet was added 6.88 grams of a 35 wt % dispersion of potassium hydride in mineral oil. 4-Benzyloxyphenol (40 grams) dissolved in 500 mL of anhydrous toluene was added dropwise and then stirred at room temperature for ten minutes. The temperature of the reaction mixture, a thick white suspension, was raised to 110° C. and stirred for 3 hours. The reaction was cooled to room temperature and 349.9 mL of 1,2-epoxypropane was added dropwise. The reaction mixture was refluxed until the pot temperature reached 110° C. (approximately 96 hours) at which time the reaction mixture was a light brown clear solution. The reaction was cooled to room temperature, quenched with 50 mL of methanol and diluted with 1 liter of diethyl ether. The reaction was washed with saturated aqueous ammonium chloride, followed by water and saturated aqueous sodium chloride. The organic layer was dried over anhydrous



sulfate, filtered and the solvents removed under vacuum to yield 212.2 grams of the desired product as a light yellow oil.

#### EXAMPLE 4

Preparation of  
 $\alpha$ -(4-Hydroxyphenyl)- $\omega$ -hydroxypoly(oxypropylene)



A solution of 60.0 grams of the product from Example 3 in 100 mL of ethyl acetate and 100 mL of acetic acid containing 7.0 grams of 10% palladium on charcoal was hydrogenolyzed at 35–40 psi for 16 hours on a Parr low-pressure hydrogenator. Catalyst filtration and removal of solvent in vacuo followed by azeotropic removal of the residual acetic acid with toluene under vacuum yielded 31.7 grams of the desired product as a brown oil. The product had an average of 20 oxypropylene units.  $^1\text{H NMR}$  ( $\text{CDCl}_3$ )  $\delta$ 6.7 (s, 4H), 5.4–6.0 (broad s, 2H), 3.0–4.0 (m, 60H), 0.8–1.4 (m, 60H).

Similarly, by using the above procedures and the appropriate starting materials and reagents, the following compounds can be prepared:

- $\alpha$ -(2-hydroxyphenyl)- $\omega$ -hydroxypoly(oxypropylene);
- $\alpha$ -(3-hydroxyphenyl)- $\omega$ -hydroxypoly(oxypropylene);
- $\alpha$ -(4-hydroxy-3-methylphenyl)- $\omega$ -hydroxypoly(oxypropylene);
- $\alpha$ -(3,5-dimethoxy-4-hydroxyphenyl)- $\omega$ -hydroxypoly(oxypropylene);
- $\alpha$ -(3,4-dihydroxyphenyl)- $\omega$ -hydroxypoly(oxypropylene);
- $\alpha$ -(3,5-di-*t*-butyl-4-hydroxyphenyl)- $\omega$ -hydroxypoly(oxypropylene); and
- $\alpha$ -(3,4,5-trihydroxyphenyl)- $\omega$ -hydroxypoly(oxypropylene).

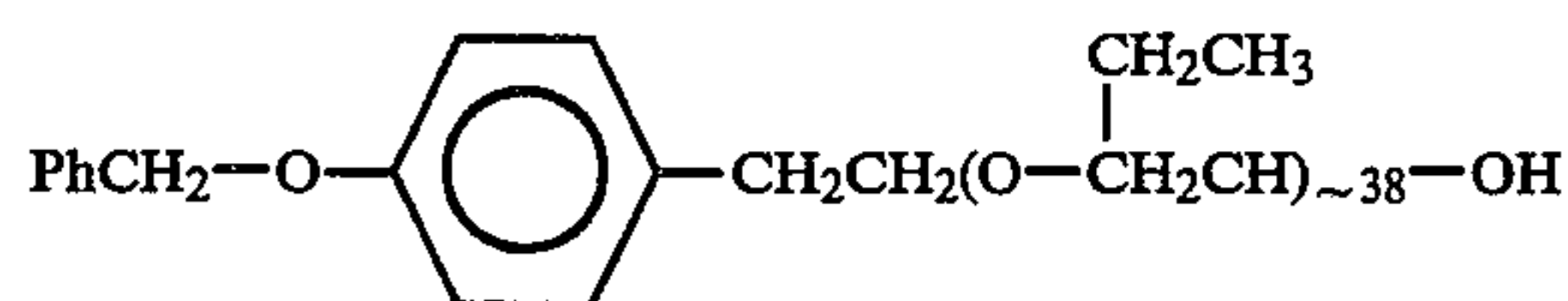
#### EXAMPLE 5

Preparation of 2-(4-Benzyloxyphenyl)ethanol

To a flask equipped with a magnetic stirrer, reflux condenser and nitrogen inlet was added 13.8 grams of 2-(4-hydroxyphenyl)ethanol, 14.5 grams of anhydrous potassium carbonate, 33.0 grams of tetrabutylammonium bromide, 12 mL of benzyl chloride and 200 mL of acetone. The reaction mixture was heated at reflux for 3 days, and then cooled to room temperature and filtered. The filtrate was concentrated in vacuo, diluted with 500 mL of dichloromethane, and washed with 2% aqueous sodium hydroxide and then with saturated brine. The organic layer was dried over anhydrous magnesium sulfate, filtered, and concentrated in vacuo. The resulting product was purified by chromatography on silica gel, eluting with dichloromethane, to yield 20.0 grams of the desired product as a white solid.

#### EXAMPLE 6

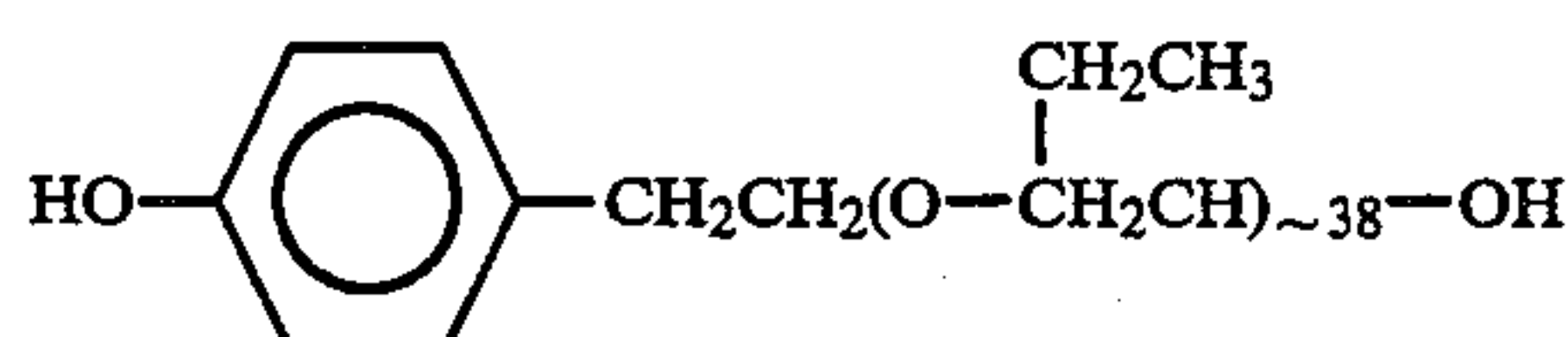
Preparation of  
 $\alpha$ -[2-(4-Benzyloxyphenyl)ethyl]- $\omega$ -hydroxypoly(oxybutylene)



To a flask equipped with a magnetic stirrer, thermometer, addition funnel, reflux condenser and nitrogen inlet was added 1.05 grams of a 35 weight percent dispersion of potassium hydride in mineral oil and 50 mL of toluene. 2-(4-Benzyloxyphenyl)ethanol (6.8 grams) from Example 5, dissolved in 7.5 mL of toluene, was added dropwise and the mixture was heated at reflux for two hours. The reaction was cooled to room temperature and 65 mL of 1,2-epoxybutane were added dropwise. The reaction mixture was then refluxed until the pot temperature reached 110° C. (approximately 16 hours). The reaction was then cooled to room temperature, quenched with 50 mL of methanol and diluted with diethyl ether (300 mL). The organic layer was washed with water (2 times), saturated aqueous ammonium chloride (2 times), dried over anhydrous magnesium sulfate, filtered and concentrated in vacuo. The resulting product was chromatographed on silica gel, eluting with hexane/diethyl ether, followed by hexane/diethyl ether/ethanol (7.5:2.5:0.5) to yield 26.0 grams of the desired product as a colorless oil.

#### EXAMPLE 7

Preparation of  
 $\alpha$ -[2-(4-Hydroxyphenyl)ethyl]- $\omega$ -hydroxypoly(oxybutylene)



A solution of 26.0 grams of the product from Example 6 in 50 mL of ethyl acetate and 50 mL of acetic acid containing 3.0 grams of 10% palladium on charcoal was hydrogenolyzed at 35–40 psi for 16 hours on a Parr low-pressure hydrogenator. Catalyst filtration and removal of solvent in vacuo followed by azeotropic removal of residual acetic acid with toluene under vacuum yielded 21.0 grams of the desired product as a light yellow oil. The product had an average of 38 oxybutylene units.  $^1\text{H NMR}$  ( $\text{CDCl}_3$ )  $\delta$ 6.7, 6.9 (AB quartet, 4H), 3.0–3.8 (m, 116H), 2.75 (t, 2H), 0.6–1.8 (m, 190H).

Similarly, by using the above procedures and the appropriate starting materials and reagents, the following compounds can be prepared:

- $\alpha$ -[2-(2-hydroxyphenyl)ethyl]- $\omega$ -hydroxypoly(oxybutylene);
- $\alpha$ -[2-(3-hydroxyphenyl)ethyl]- $\omega$ -hydroxypoly(oxybutylene);
- $\alpha$ -[3-(4-hydroxyphenyl)propyl]- $\omega$ -hydroxypoly(oxybutylene);
- $\alpha$ -[2-(3,4-dihydroxyphenyl)ethyl]- $\omega$ -hydroxypoly(oxybutylene);
- $\alpha$ -[3-(3,4-dihydroxyphenyl)propyl]- $\omega$ -hydroxypoly(oxybutylene);

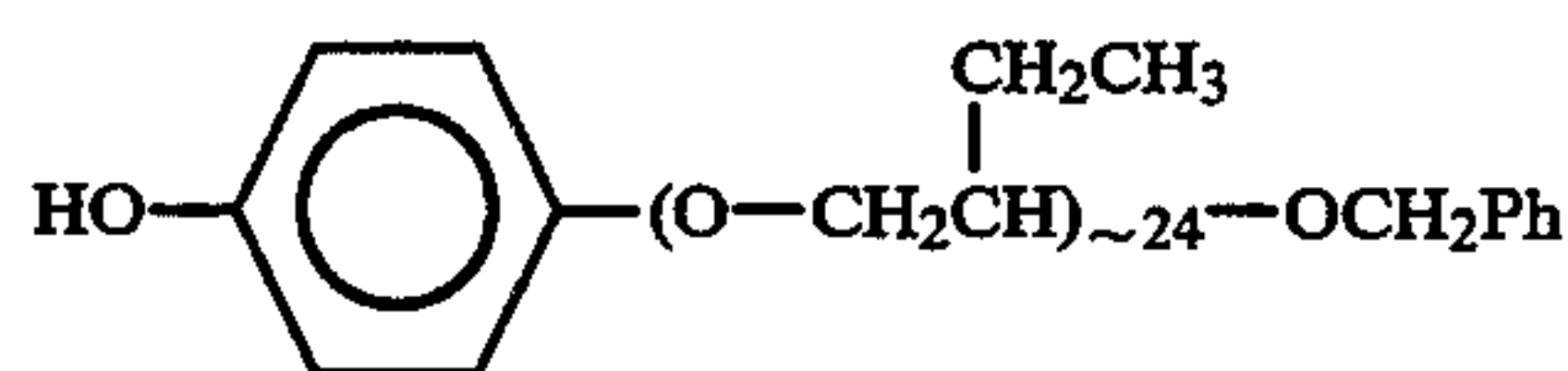


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$\alpha$ -[2-(3,5-di-*t*-butyl-4-hydroxyphenyl)ethyl]- $\omega$ -hydroxypoly(oxybutylene); and  
 $\alpha$ -[2-(3,4,5-trihydroxyphenyl)ethyl]- $\omega$ -hydroxypoly(oxybutylene).

## EXAMPLE 8

Preparation of  
 $\alpha$ -(4-Hydroxyphenyl)- $\omega$ -benzyloxypoly(oxybutylene)



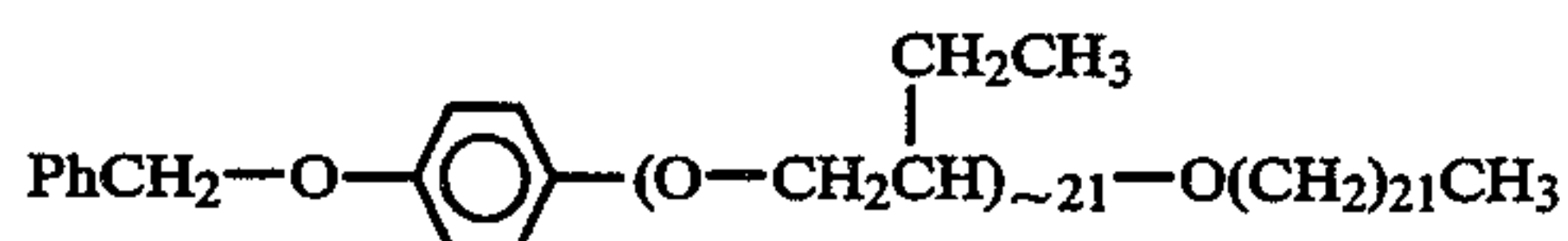
To a flask equipped with a magnetic stirrer, thermometer, reflux condenser and nitrogen inlet was added 0.8 grams of a 35 wt % dispersion of potassium hydride in mineral oil. The oil was removed by trituration with anhydrous toluene. The product from Example 2 (6.0 grams) was dissolved in 50 mL of anhydrous tetrahydrofuran and added dropwise to the potassium hydride. The reaction mixture was heated to reflux for 45 minutes and then cooled to room temperature. Benzyl chloride (0.36 mL) was added dropwise and the reaction was then heated to reflux for 12 hours, cooled to room temperature and quenched with 2 mL of isopropanol. The solvent was removed in vacuo and the residue dissolved in 200 mL of diethyl ether, washed with 5% aqueous hydrochloric acid followed by saturated aqueous sodium chloride. The organic layer was dried over anhydrous magnesium sulfate, filtered and the solvents removed under vacuum. The oil was chromatographed on silica gel, eluting with hexane/ethyl acetate (7:3), to yield 3.8 grams of the desired product as a colorless oil. The product had an average of 24 oxybutylene units.  $^1\text{H NMR}$  ( $\text{CDCl}_3$ )  $\delta$ 7.2–7.4 (m, 6H), 6.7 (s, 4H), 4.4–4.7 (m, 2H), 3.1–4.0 (m, 72H), 1.2–1.8 (m, 48H), 0.8 (t, 72H).

Similarly, by using the above procedures and the appropriate starting materials and reagents, the following compounds can be prepared:

$\alpha$ -(2-hydroxyphenyl)- $\omega$ -benzyloxypoly(oxybutylene);  
 $\alpha$ -(3-hydroxyphenyl)- $\omega$ -benzyloxypoly(oxybutylene);  
 $\alpha$ -(3,4-dihydroxyphenyl)- $\omega$ -benzyloxypoly(oxybutylene);  
 $\alpha$ -(3,5-di-*t*-butyl-4-hydroxyphenyl)- $\omega$ -benzyloxypoly(oxybutylene);  
 $\alpha$ -(4-hydroxy-3-methoxyphenyl)- $\omega$ -benzyloxypoly(oxybutylene); and  
 $\alpha$ -[2-(4-hydroxyphenyl)ethyl]- $\omega$ -benzyloxypoly(oxybutylene).

## EXAMPLE 9

Preparation of  
 $\alpha$ -(4-Benzyloxyphenyl)- $\omega$ -docosanoxypoly(oxybutylene)



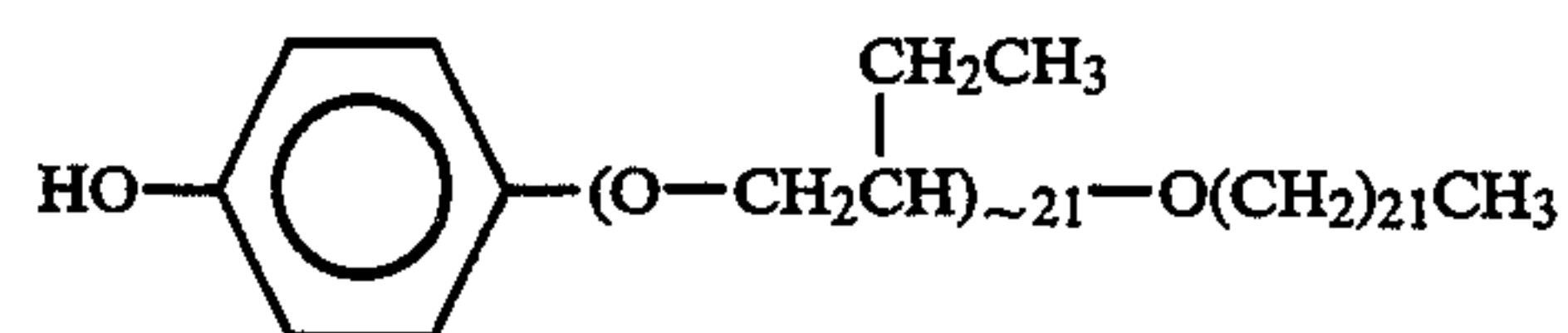
To a flask equipped with a magnetic stirrer, addition funnel, reflux condenser and nitrogen inlet was added 7.26 grams of a 35 wt % dispersion of potassium hydride in mineral oil. The oil was removed by trituration with anhydrous hexane, and 500 milliliters of anhydrous tetrahydrofuran were added.  $\alpha$ -(4-Benzyloxyphenyl)- $\omega$ -

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hydroxypoly(oxybutylene) (104.0 grams) containing an average of 21 oxybutylene units (prepared essentially as described in Example 1), dissolved in 100 milliliters of anhydrous tetrahydrofuran, was added dropwise and the resulting mixture was heated to reflux for two hours. The reaction was then cooled to room temperature and 24.0 grams of 1-bromodocosane were added. The reaction was refluxed for sixteen hours, cooled to room temperature, diluted with 1200 mL of diethyl ether, and washed with 5% aqueous hydrochloric acid, followed by brine. The organic layers were dried over anhydrous magnesium sulfate, filtered and concentrated in vacuo to give an oil. The oil was chromatographed on silica gel, eluting with hexane/ethyl acetate (7:3) to yield 11.0 grams of the desired product as a yellow oil.

## EXAMPLE 10

Preparation of  
 $\alpha$ -(4-Hydroxyphenyl)- $\omega$ -docosanoxypoly(oxybutylene)



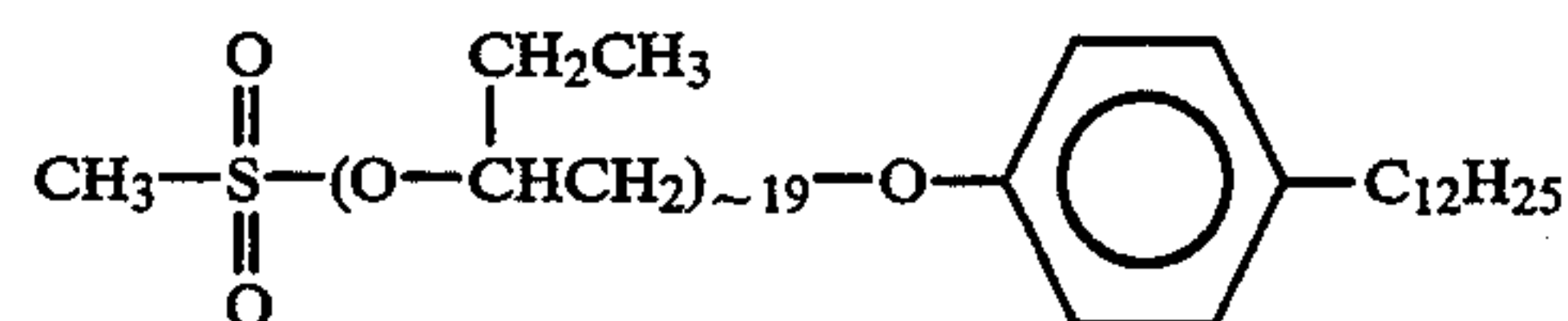
A solution of 11.0 grams of the product from Example 9 in 50 mL of ethyl acetate and 50 mL of acetic acid containing 1.5 grams of 10% palladium on charcoal was hydrogenolyzed at 35–40 psi for 14 hours on a Parr low-pressure hydrogenator. Catalyst filtration and removal of solvent in vacuo followed by azeotropic removal of the residual acetic acid with toluene under vacuum yielded 10.2 grams of the desired product. The product had an average of 21 oxybutylene units.  $^1\text{H NMR}$  ( $\text{CDCl}_3$ )  $\delta$ 6.7 (s, 4H), 3.1–4.0 (m, 62H), 0.6–1.8 (m, 148H).

Similarly, by using the above procedures and the appropriate starting materials and reagents, the following compounds can be prepared:

$\alpha$ -(4-hydroxyphenyl)- $\omega$ -*n*-butoxypoly(oxybutylene);  
 $\alpha$ -(4-hydroxyphenyl)- $\omega$ -*n*-octyloxypoly(oxybutylene);  
 $\alpha$ -(4-hydroxyphenyl)- $\omega$ -*n*-dodecyloxypoly(oxybutylene);  
 $\alpha$ -(3,5-di-*t*-butyl-4-hydroxyphenyl)- $\omega$ -*n*-pentyloxypoly(oxybutylene);  
 $\alpha$ -(4-hydroxy-3-methoxyphenyl)- $\omega$ -*n*-hexyloxypoly(oxybutylene);  
 $\alpha$ -(3,4-hydroxyphenyl)- $\omega$ -nonyloxypoly(oxybutylene); and  
 $\alpha$ -[2-(4-hydroxyphenyl)ethyl]- $\omega$ -octyloxypoly(oxybutylene).

## EXAMPLE 11

Preparation of  
 $\alpha$ -(Methanesulfonyl)- $\omega$ -4-dodecylphenoxy poly(oxybutylene)



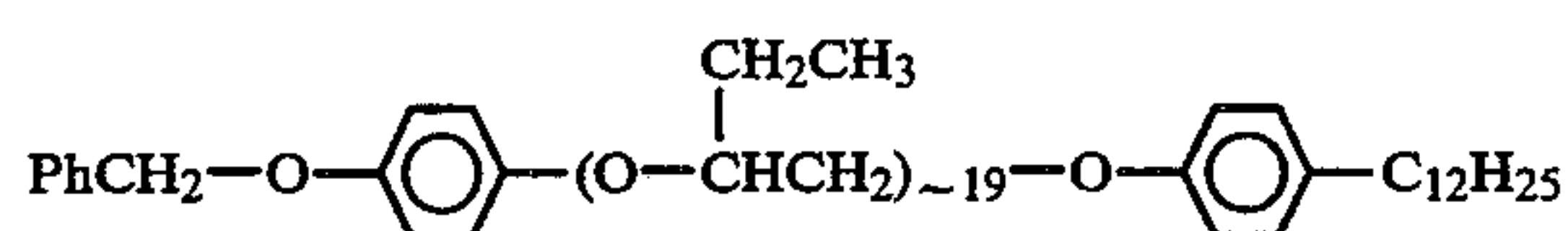
To a flask equipped with a magnetic stirrer, septa and a nitrogen inlet was added 35.0 grams of  $\alpha$ -hydroxy- $\omega$ -



4-dodecylphenoxy poly(oxybutylene) having an average of 19 oxybutylene units (prepared essentially as described in Example 6 of U.S. Pat. No. 4,160,648), 440 mL of dichloromethane and 3.6 mL of triethylamine. The flask was cooled in an ice bath and 1.8 mL of methanesulfonyl chloride were added dropwise. The ice bath was removed and the reaction was stirred at room temperature for 16 hours. Dichloromethane (800 mL) was added and the organic phase was washed two times with saturated aqueous sodium bicarbonate, and then once with brine. The organic layer was dried over anhydrous magnesium sulfate, filtered and concentrated in vacuo to yield 35.04 grams of the desired product as a yellow oil.

#### EXAMPLE 12

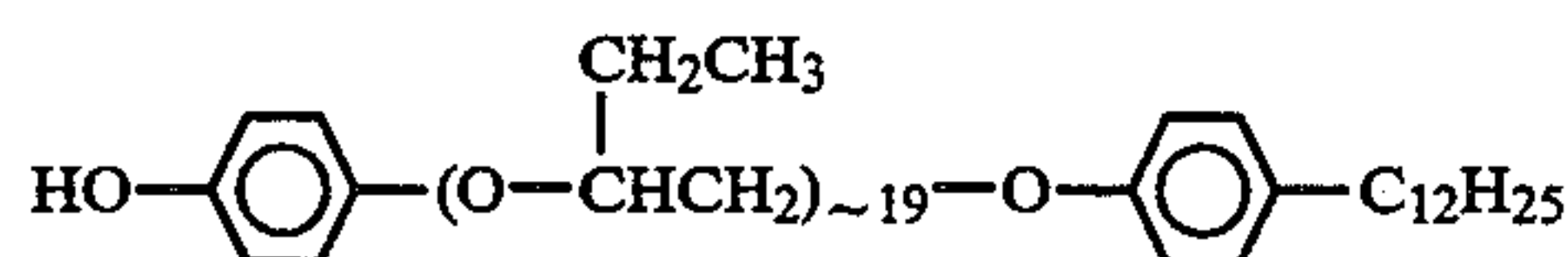
Preparation of  
 $\alpha$ -(4-Benzoyloxyphenyl)- $\omega$ -4-dodecylphenoxy poly(oxybutylene)



To a flask equipped magnetic stirrer, reflux condenser, nitrogen inlet and septa was added 2.59 grams of a 35 wt % dispersion of potassium hydride in mineral oil. The mineral oil was removed by trituration with hexane and the flask was cooled in an ice bath. 4-Benzoyloxyphenol (4.11 grams) dissolved in 150 mL of tetrahydrofuran was added dropwise. The ice bath was removed and the reaction was allowed to stir for 45 minutes at room temperature. The mesylate from Example 11 was dissolved in 275 mL of anhydrous tetrahydrofuran and added to the reaction mixture. The resulting solution was refluxed for 16 hours, cooled to room temperature and 10 mL of methanol were added. The reaction was diluted with 1 liter of diethyl ether, washed with water (1 time), brine (1 time), dried over anhydrous magnesium sulfate, filtered and concentrated in vacuo to 36.04 grams of a yellow oil. The oil was chromatographed on silica gel, eluting with hexane/diethyl ether/ethanol (8:1.8:0.2) to yield 18.88 grams of the desired product as a light yellow oil.

#### EXAMPLE 13

Preparation of  
 $\alpha$ -(4-Hydroxyphenyl)- $\omega$ -4-dodecylphenoxy poly(oxybutylene)



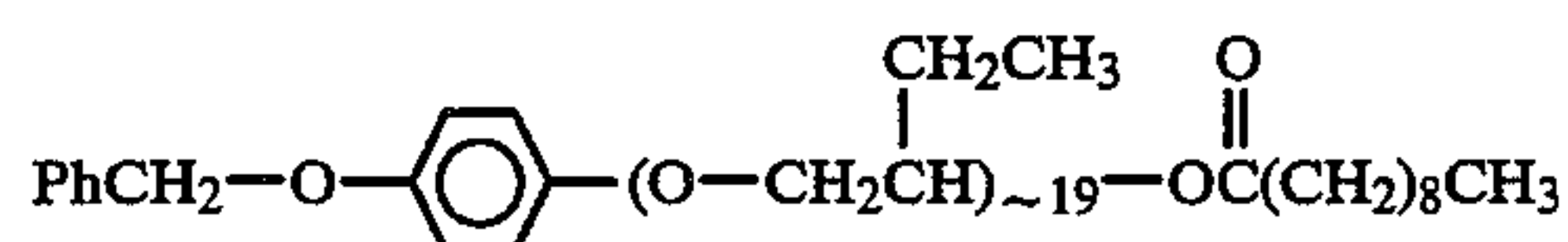
A solution of 18.88 grams of the product from Example 12 in 80 mL of ethyl acetate and 80 mL of acetic acid containing 2.08 grams of 10% palladium on charcoal was hydrogenolyzed at 35-40 psi for 6 hours on a Parr low-pressure hydrogenator. Filtration of the catalyst and removal of solvent in vacuo, followed by azeotropic removal of residual acetic acid with toluene under vacuum yielded 17.63 grams of the desired product as a yellow oil. The product had an average of 19 oxybutylene units.  $^1\text{H NMR}$  ( $\text{CDCl}_3$ )  $\delta$ 7.0-7.3 (m, 2H), 6.6-6.9 (m, 6H), 4.0-4.2 (m, 1H), 3.8-4.0 (m, 2H), 3.0-3.8 (m, 54H), 0.5-1.8 (m, 120H).

Similarly, by using the above procedures and the appropriate starting materials and reagents, the following compounds can be prepared:

$\alpha$ -(2-hydroxyphenyl)- $\omega$ -4-dodecylphenoxy poly(oxybutylene);  
 $\alpha$ -(3-hydroxyphenyl)- $\omega$ -4-dodecylphenoxy poly(oxybutylene);  
 $\alpha$ -(3,4-dihydroxyphenyl)- $\omega$ -4-dodecylphenoxy poly(oxybutylene);  
 $\alpha$ -(4-hydroxyphenyl)- $\omega$ -phenoxy poly(oxybutylene);  
 $\alpha$ -(4-hydroxyphenyl)- $\omega$ -4-t-butylphenoxy poly(oxybutylene);  
 $\alpha$ -(4-hydroxyphenyl)- $\omega$ -4-decylphenoxy poly(oxybutylene); and  
 $\alpha$ -(4-hydroxyphenyl)- $\omega$ -4-octadecylphenoxy poly(oxybutylene).

#### EXAMPLE 14

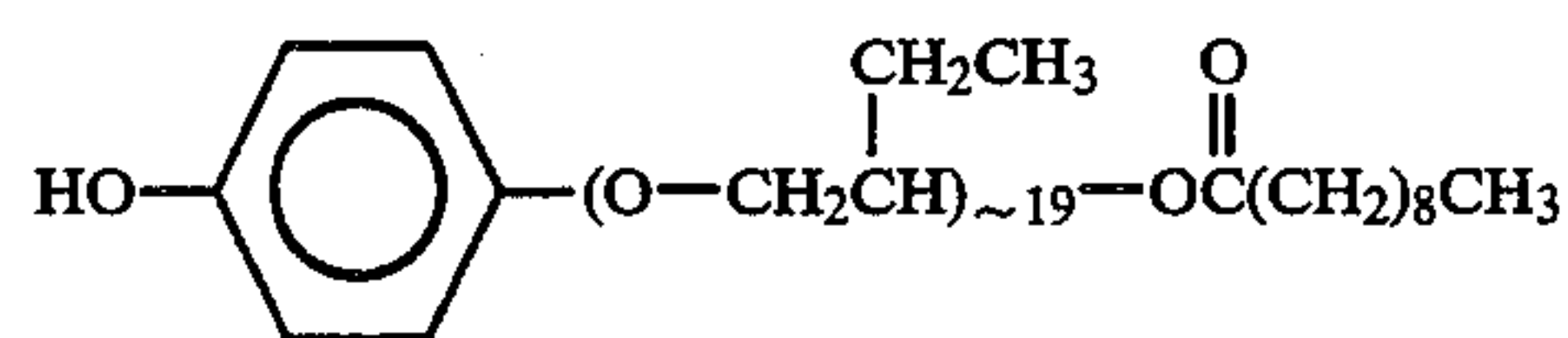
Preparation of  
 $\alpha$ -(4-Benzoyloxyphenyl)- $\omega$ -decanoyloxy poly(oxybutylene)



$\alpha$ -(4-Benzoyloxyphenyl)- $\omega$ -hydroxy poly(oxybutylene) (40.75 grams) containing an average of 19 oxybutylene units (prepared essentially as described in Example 1) was combined with 200 mL of toluene, 3.9 mL of triethylamine, 1.5 grams of 4-dimethylamine pyridine and 5.2 mL of n-decanoyl chloride in a flask equipped with a thermometer, magnetic stirrer, reflux condenser and nitrogen inlet. The contents were refluxed for 16 hours, cooled to room temperature and diluted with 400 mL of hexane. The organic layers were washed with water (2 times), saturated aqueous sodium bicarbonate (2 times), saturated aqueous sodium chloride (2 times), dried over anhydrous magnesium sulfate, filtered and concentrated to yield 40 grams of a yellow oil. The oil was chromatographed on silica gel, eluting with hexane/diethyl ether (1:1) to yield 23.3 grams of the product as a yellow oil.

#### EXAMPLE 15

Preparation of  
 $\alpha$ -(4-Hydroxyphenyl)- $\omega$ -decanoyloxy poly(oxybutylene)



A solution of the ester from Example 14 (23.3 grams) in 50 mL of ethyl acetate and 50 mL of acetic acid containing 2.5 grams of 10% palladium on charcoal was hydrogenolyzed at 35-40 psi for 16 hours on a Parr low-pressure hydrogenator. Filtration of the catalyst and removal of solvent in vacuo followed by azeotropic removal of residual acetic acid with toluene under vacuum yielded 16.0 grams of the desired product as a yellow oil. The product had an average of 19 oxybutylene units. IR (neat)  $1735\text{ cm}^{-1}$ ;  $^1\text{H NMR}$  ( $\text{CDCl}_3$ )  $\delta$ 6.7 (s, 4H), 4.8-4.9 (m, 1H), 3.1-4.0 (m, 56H), 2.3 (t, 2H), 0.7-1.8 (m, 112H).



Similarly, by using the above procedures and the appropriate starting materials and reagents, the following compounds can be prepared:

- $\alpha$ -(2-hydroxyphenyl)- $\omega$ -decanoyloxypoly(oxybutylene);
- $\alpha$ -(3-hydroxyphenyl)- $\omega$ -decanoyloxypoly(oxybutylene);
- $\alpha$ -(4-hydroxyphenyl)- $\omega$ -dodecanoyloxypoly(oxybutylene);
- $\alpha$ -(4-hydroxyphenyl)- $\omega$ -octanoyloxypoly(oxybutylene);
- $\alpha$ -(4-hydroxyphenyl)- $\omega$ -butanoyloxypoly(oxybutylene);
- $\alpha$ -(4-hydroxyphenyl)- $\omega$ -benzoyloxypoly(oxybutylene);
- $\alpha$ -(3,4-dihydroxyphenyl)- $\omega$ -hexanoyloxypoly(oxybutylene);
- $\alpha$ -(3,4-hydroxyphenyl)- $\omega$ -2-ethylhexanoyloxypoly(oxybutylene);
- $\alpha$ -(3,5-di-*t*-butyl-4-hydroxyphenyl)- $\omega$ -nonanoyloxypoly(oxybutylene);
- $\alpha$ -(3,4,5-trihydroxyphenyl)- $\omega$ -decanoyloxypoly(oxybutylene); and
- $\alpha$ -[2-(4-hydroxyphenyl)ethyl]- $\omega$ -decanoyloxypoly(oxybutylene).

#### EXAMPLE 16

##### Single-Cylinder Engine Test

The test compounds were blended in gasoline and their deposit reducing capacity determined in an ASTM/CFR single-cylinder engine test.

A Waukesha CFR single-cylinder engine was used. Each run was carried out for 15 hours, at the end of which time the intake valve was removed, washed with hexane and weighed. The previously determined weight of the clean valve was subtracted from the weight of the valve at the end of the run. The differences between the two weights is the weight of the deposit. A lesser amount of deposit indicates a superior additive. The operating conditions of the test were as follows: water jacket temperature 200° F.; vacuum of 12 in Hg, air-fuel ratio of 12, ignition spark timing of 40° BTC; engine speed is 1800 rpm; the crankcase oil is a commercial 30 W oil.

The amount of carbonaceous deposit in milligrams on the intake valves is reported for each of the test compounds in Table I.

TABLE I

Single-Cylinder Engine Test Results			
Sample <sup>1</sup>	Intake Valve Deposit Weight (in milligrams)		
	Run 1	Run 2	Average
Base Fuel	214.7	193.7	204.2
Example 2	12.7	26.5	19.6
Example 4	59.6	73.8	66.7
Example 7	44.3	54.0	42.9
Example 8	52.8	75.9	64.4
Example 10	53.9	47.9	50.9
Example 13	32.2	32.3	32.3
Example 15	32.5	31.1	31.8

<sup>1</sup>At 200 parts per million actives (ppma).

The base fuel employed in the above single-cylinder engine tests was a regular octane unleaded gasoline containing no fuel detergent. The test compounds were admixed with the base fuel to give a concentration of 200 ppma (parts per million actives).

The data in Table I illustrates the significant reduction in intake valve deposits provided by the poly(oxy-

alkylene) hydroxyaromatic ether component of the present fuel additive composition (Examples 2, 4, 7, 8, 10, 13, 15) compared to the base fuel.

#### EXAMPLE 17

##### Multicylinder Engine Test

The fuel additive composition of the present invention was tested in a laboratory multicylinder engine to evaluate its intake valve and combustion chamber deposit control performance. The test engine was a 4.3 liter, TBI (throttle body injected), V6 engine manufactured by General Motors Corporation. The major engine dimensions are set forth in Table II:

TABLE II

Engine Dimensions	
Bore	10.16 cm
Stroke	8.84 cm
Displacement Volume	4.3 liter
Compression Ratio	9.3:1

The test engine was operated for 40 hours (24 hours a day) on a prescribed load and speed schedule representative of typical driving conditions. The cycle for engine operation during the test is set forth in Table III.

TABLE III

Engine Driving Cycle				
Step	Mode	Time in Mode [Sec] <sup>1</sup>	Dynamometer Load [kg]	Engine Speed [RPM]
1	Idle	60	0	800
2	City Cruise	150	10	1,500
3	Acceleration	40	25	2,800
4	Heavy HWY Cruise	210	15	2,200
5	Light HWY Cruise	60	10	2,200
6	Idle	60	0	800
7	City Cruise	180	10	1,500
8	Idle	60	0	800

<sup>1</sup>All steps, except step number 3, include a 15 second transition ramp. Step 3 includes a 20 second transition ramp.

All of the test runs were made with the same base gasoline, which was representative of commercial unleaded fuel. The results are set forth in Table IV.

TABLE IV

Multicylinder Engine Test Results			
Sample	Conc. (ppma)	Intake Valve Deposits <sup>1</sup>	Combustion Chamber Deposits <sup>1</sup>
Base Fuel	—	972	1902
Poly(oxyalkylene) Hydroxyaromatic Ether <sup>2</sup>	400	283	2547
Poly(oxyalkylene) Amine <sup>3</sup>	200	340	2282
Poly(oxyalkylene) Hydroxyaromatic Ether/ Poly(oxyalkylene) Amine <sup>4</sup>	400/200	75	2177

<sup>1</sup>Average of two runs, in milligrams (mg).

<sup>2</sup> $\alpha$ -(4-Hydroxyphenyl)- $\omega$ -hydroxypoly(oxybutylene) prepared as described in Example 2.

<sup>3</sup>Dodecylphenyl poly(oxyalkylene) ethylene diamine carbamate prepared essentially as described in Examples 6-8 of U.S. Pat. No. 4,160,648.

<sup>4</sup>Mixture of 400 ppm of  $\alpha$ -(4-Hydroxyphenyl)- $\omega$ -hydroxypoly(oxybutylene) and 200 ppm of dodecylphenyl poly(oxyalkylene) ethylene diamine carbamate.

The base fuel employed in the above multicylinder engine tests contained no fuel detergent. The test com-



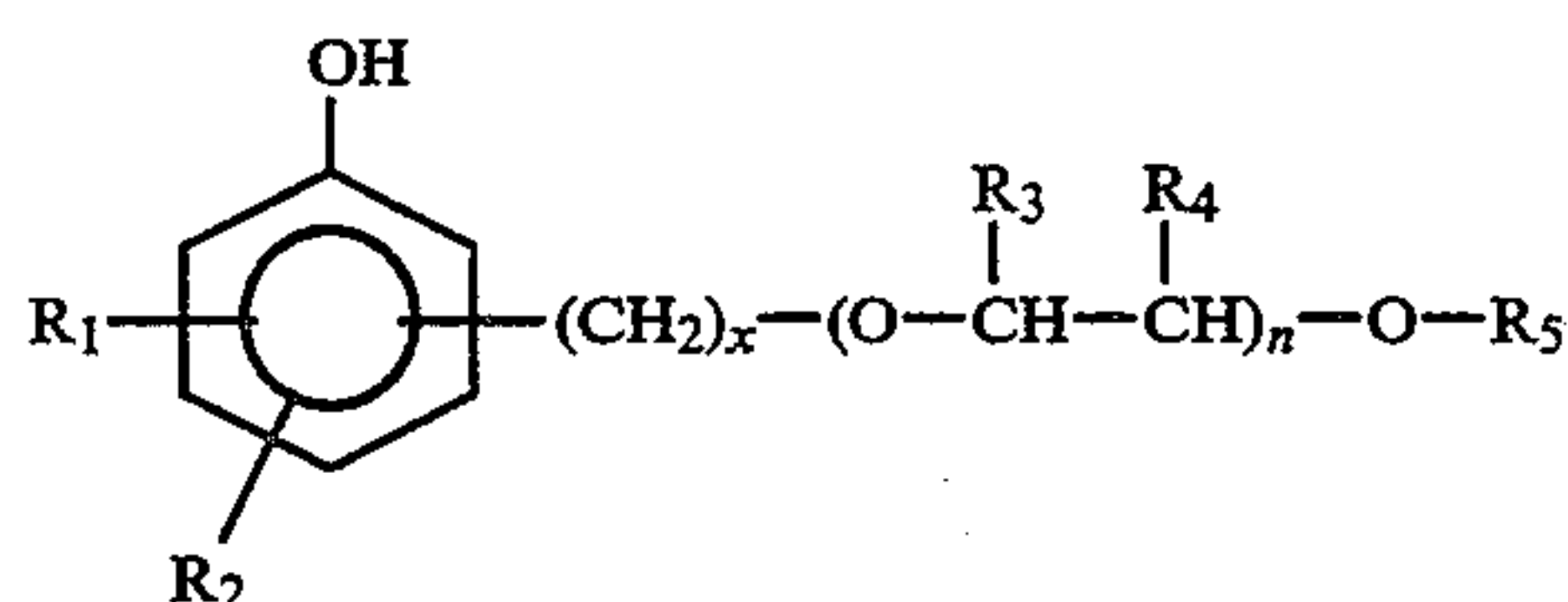
pounds were admixed with the base fuel at the indicated concentrations.

The data in Table IV demonstrates that the combination of a poly(oxyalkylene) hydroxyaromatic ether and a poly(oxyalkylene) amine has a synergistic effect and gives significantly better intake valve deposit control than either component individually. Moreover, the data in Table IV further demonstrates that the combination produces fewer combustion chamber deposits than each individual component.

What is claimed is:

1. A fuel additive composition comprising:

(a) a poly(oxyalkylene) hydroxyaromatic ether having the formula:

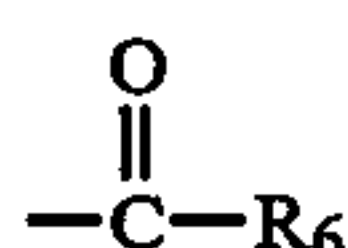


or a fuel-soluble salt thereof; wherein

R<sub>1</sub> and R<sub>2</sub> are each independently hydrogen, hydroxy, lower alkyl having 1 to 6 carbon atoms, or lower alkoxy having 1 to 6 carbon atoms;

R<sub>3</sub> and R<sub>4</sub> are each independently hydrogen or lower alkyl having 1 to 6 carbon atoms;

R<sub>5</sub> is hydrogen, alkyl having 1 to 30 carbon atoms, phenyl, aralkyl or alkaryl having 7 to 36 carbon atoms, or an acyl group of the formula:



wherein R<sub>6</sub> is alkyl having 1 to 30 carbon atoms, phenyl, or aralkyl or alkaryl having 7 to 36 carbon atoms; n is an integer from 5 to 100; and x is an integer from 0 to 10; and

(b) a poly(oxyalkylene) amine having at least one basic nitrogen atom and a sufficient number of oxyalkylene units to render the poly(oxyalkylene) amine soluble in hydrocarbons boiling in the gasoline or diesel range.

2. The fuel additive composition according to claim 1, wherein n of said poly(oxyalkylene) hydroxyaromatic ether is an integer ranging from 10 to 50.

3. The fuel additive composition according to claim 2, wherein n of said poly(oxyalkylene) hydroxyaromatic ether is an integer ranging from 15 to 30.

4. The fuel additive composition according to claim 2, wherein R<sub>1</sub> of said poly(oxyalkylene) hydroxyaromatic ether is hydrogen, hydroxy, or lower alkyl having 1 to 4 carbon atoms; and R<sub>2</sub> is hydrogen.

5. The fuel additive composition according to claim 4, wherein R<sub>5</sub> of said poly(oxyalkylene) hydroxyaromatic ether is hydrogen, alkyl having 2 to 22 carbon atoms, alkylphenyl having an alkyl group containing 4 to 24 carbon atoms, or an acyl group having the formula: -C(O)R<sub>7</sub>, wherein R<sub>7</sub> is alkyl having 4 to 12 carbon atoms.

6. The fuel additive composition according to claim 5, wherein R<sub>1</sub> of said poly(oxyalkylene) hydroxyaromatic ether is hydrogen or hydroxy.

7. The fuel additive composition according to claim 6, wherein R<sub>5</sub> of said poly(oxyalkylene) hydroxyaromatic ether is hydrogen, alkyl having 4 to 12 carbon

atoms, or alkylphenyl having an alkyl group containing 4 to 12 carbon atoms.

8. The fuel additive composition according to claim 7, wherein one of R<sub>3</sub> and R<sub>4</sub> of said poly(oxyalkylene) hydroxyaromatic ether is lower alkyl having 1 to 3 carbon atoms and the other is hydrogen.

9. The fuel additive composition according to claim 8, wherein one of R<sub>3</sub> and R<sub>4</sub> of said poly(oxyalkylene) hydroxyaromatic ether is methyl or ethyl and the other is hydrogen.

10. The fuel additive composition according to claim 9, wherein x of said poly(oxyalkylene) hydroxyaromatic ether is 0, 1 or 2.

11. The fuel additive composition according to claim 10, wherein R<sub>1</sub> and R<sub>5</sub> of said poly(oxyalkylene) hydroxyaromatic ether are both hydrogen, and x is 0.

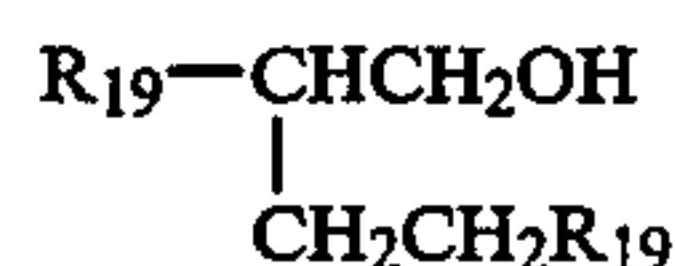
12. The fuel additive composition according to claim 1, wherein said poly(oxyalkylene) amine has a molecular weight in the range of about 500 to about 10,000.

13. The fuel additive composition according to claim 1, wherein said poly(oxyalkylene) amine contains at least about 5 oxyalkylene units.

14. The fuel additive composition according to claim 1, wherein said poly(oxyalkylene) amine is a hydrocarbyl poly(oxyalkylene) polyamine.

15. The fuel additive composition according to claim 1, wherein said poly(oxyalkylene) amine is a poly(oxyalkylene) polyamine, wherein the poly(oxyalkylene) moiety is connected to the polyamine moiety through an oxyalkylene hydroxy linkage.

16. The fuel additive composition according to claim 1, wherein said poly(oxyalkylene) amine is a branched alkyl poly(oxyalkylene) monoamine, wherein the branched alkyl group is derived from the alcohol product of a Guerbet condensation reaction, said alcohol product having the formula:



wherein R<sub>19</sub> is a hydrocarbyl chain.

17. The fuel additive composition according to claim 1, wherein said poly(oxyalkylene) amine is a hydrocarbyl poly(oxyalkylene) aminocarbamate.

18. The fuel additive composition according to claim 17, wherein the hydrocarbyl group of said hydrocarbyl poly(oxyalkylene) aminocarbamate contains from 1 to about 30 carbon atoms.

19. The fuel additive composition according to claim 18, wherein said hydrocarbyl group of said hydrocarbyl poly(oxyalkylene) aminocarbamate is an alkylphenyl group.

20. The fuel additive composition according to claim 19, wherein the alkyl moiety of said alkylphenyl group is tetrapropenyl.

21. The fuel additive composition according to claim 17, wherein the amine moiety of said hydrocarbyl poly(oxyalkylene) aminocarbamate is derived from a polyamine having from 2 to 12 amine nitrogen atoms and from 2 to 40 carbon atoms.

22. The fuel additive composition according to claim 21, wherein said polyamine is a polyalkylene polyamine having 2 to 12 amine nitrogen atoms and 2 to 24 carbon atoms.

23. The fuel additive composition according to claim 22, wherein said polyalkylene polyamine is selected



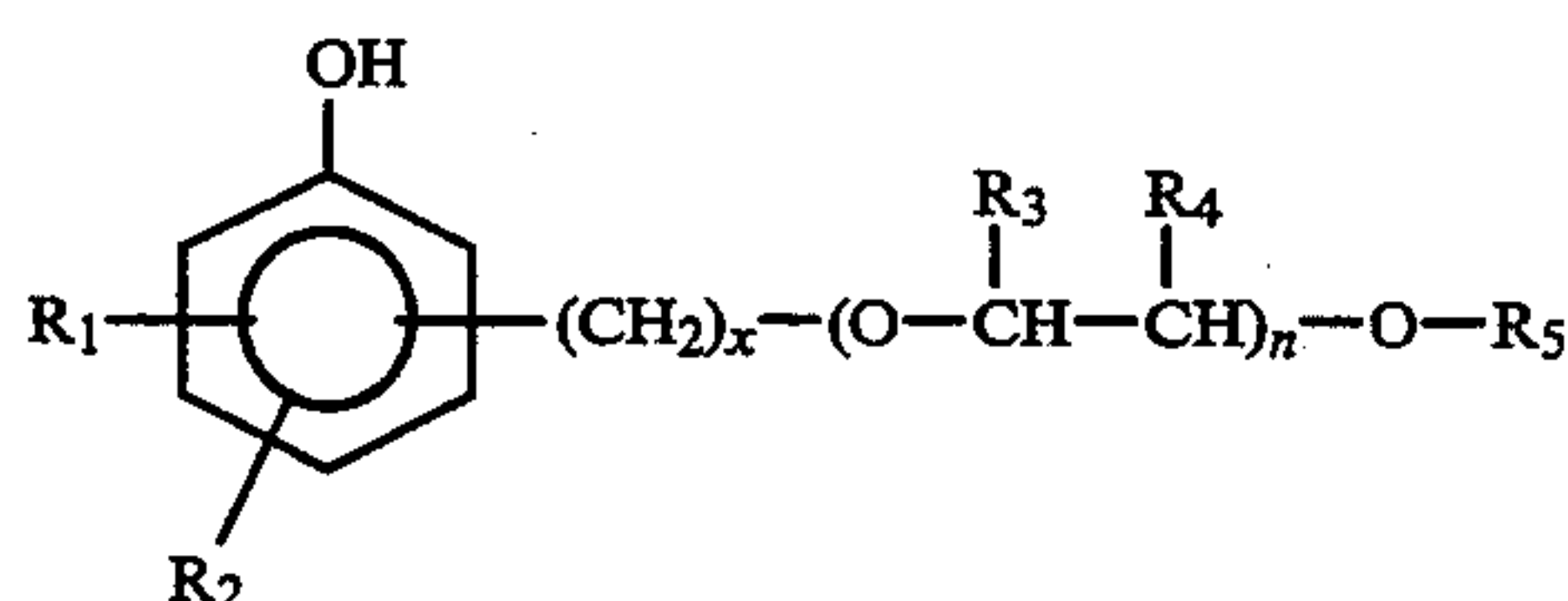
from the group consisting of ethylenediamine, propylenediamine, diethylenetriamine and dipropylenetriamine.

24. The fuel additive composition according to claim 17, wherein the poly(oxyalkylene) moiety of said hydrocarbyl poly(oxyalkylene) aminocarbamate is derived from C<sub>2</sub> to C<sub>5</sub> oxyalkylene units.

25. The fuel additive composition according to claim 17, wherein said hydrocarbyl poly(oxyalkylene) aminocarbamate is an alkylphenyl poly(oxybutylene) aminocarbamate, wherein the amine moiety is derived from ethylenediamine or diethylenetriamine.

26. A fuel composition comprising a major amount of hydrocarbons boiling in the gasoline or diesel range and an effective detergent amount of a fuel additive composition comprising:

(a) a poly(oxyalkylene) hydroxyaromatic ether having the formula:

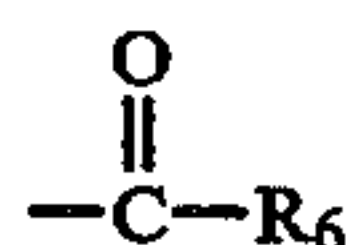


or a fuel-soluble salt thereof; wherein

R<sub>1</sub> and R<sub>2</sub> are each independently hydrogen, hydroxy, lower alkyl having 1 to 6 carbon atoms, or lower alkoxy having 1 to 6 carbon atoms;

R<sub>3</sub> and R<sub>4</sub> are each independently hydrogen or lower alkyl having 1 to 6 carbon atoms;

R<sub>5</sub> is hydrogen, alkyl having 1 to 30 carbon atoms, phenyl, aralkyl or alkaryl having 7 to 36 carbon atoms, or an acyl group of the formula:



wherein R<sub>6</sub> is alkyl having 1 to 30 carbon atoms, phenyl, or aralkyl or alkaryl having 7 to 36 carbon atoms; n is an integer from 5 to 100; and x is an integer from 0 to 10; and

(b) a poly(oxyalkylene) amine having at least one basic nitrogen atom and a sufficient number of oxyalkylene units to render the poly(oxyalkylene) amine soluble in hydrocarbons boiling in the gasoline or diesel range.

27. The fuel composition according to claim 26, wherein R<sub>1</sub> of said poly(oxyalkylene) hydroxyaromatic ether is hydrogen, hydroxy, or lower alkyl having 1 to 4 carbon atoms; R<sub>2</sub> is hydrogen; one of R<sub>3</sub> and R<sub>4</sub> is hydrogen and the other is methyl or ethyl; R<sub>5</sub> is hydrogen, alkyl having 2 to 22 carbon atoms, alkylphenyl having an alkyl group containing 4 to 24 carbon atoms, or an acyl group having the formula: -C(O)R<sub>7</sub>, wherein R<sub>7</sub> is alkyl having 4 to 12 carbon atoms; n is 15 to 30; and x is 0, 1 or 2.

28. The fuel composition according to claim 27, wherein R<sub>1</sub> of said poly(oxyalkylene) hydroxyaromatic ether is hydrogen or hydroxy; R<sub>5</sub> is hydrogen, alkyl having 4 to 12 carbon atoms, or alkylphenyl having an alkyl group containing 4 to 12 carbon atoms; and x is 0.

29. The fuel composition according to claim 28, wherein R<sub>1</sub> and R<sub>5</sub> of said poly(oxyalkylene) hydroxyaromatic ether are both hydrogen.

30. The fuel composition according to claim 26, wherein said poly(oxyalkylene) amine is a hydrocarbyl poly(oxyalkylene) aminocarbamate.

31. The fuel composition according to claim 30, wherein the hydrocarbyl group of said hydrocarbyl poly(oxyalkylene) aminocarbamate contains from 1 to about 30 carbon atoms; and wherein the amine moiety of said hydrocarbyl poly(oxyalkylene) aminocarbamate is derived from a polyamine having from 2 to 12 amine nitrogen atoms and from 2 to 40 carbon atoms.

32. The fuel composition according to claim 31, wherein said hydrocarbyl group of said hydrocarbyl poly(oxyalkylene) aminocarbamate is an alkylphenyl group; and wherein said polyalkylene polyamine is selected from the group consisting of ethylenediamine, propylenediamine, diethylenetriamine and dipropylenetriamine.

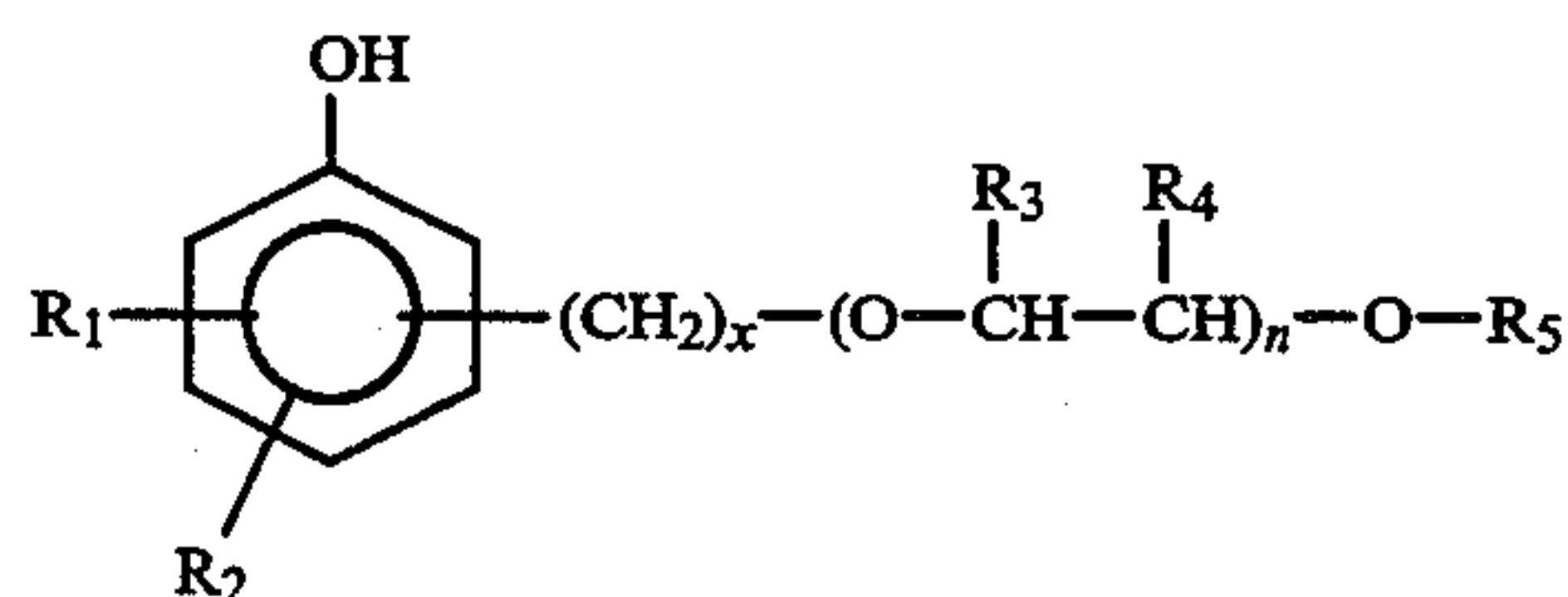
33. The fuel composition according to claim 32, wherein the alkyl moiety of said alkylphenyl group is tetrapropenyl.

34. The fuel composition according to claim 30, wherein said hydrocarbyl poly(oxyalkylene) aminocarbamate is an alkylphenyl poly(oxybutylene) aminocarbamate, wherein the amine moiety is derived from ethylenediamine or diethylenetriamine.

35. The fuel composition according to claim 26, wherein said composition contains about 50 to about 2,500 parts per million by weight of said poly(oxyalkylene) hydroxyaromatic ether and about 25 to about 1,000 parts per million of said poly(oxyalkylene) amine.

36. A fuel concentrate comprising an inert stable oleophilic organic solvent boiling in the range of from about 150° F. to 400° F. and from about 10 to about 70 weight percent of a fuel additive composition comprising:

(a) a poly(oxyalkylene) hydroxyaromatic ether having the formula:

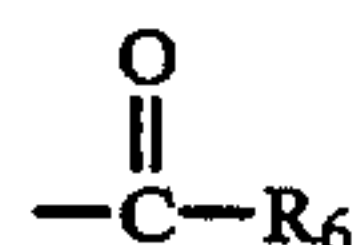


or a fuel-soluble salt thereof; wherein

R<sub>1</sub> and R<sub>2</sub> are each independently hydrogen, hydroxy, lower alkyl having 1 to 6 carbon atoms, or lower alkoxy having 1 to 6 carbon atoms;

R<sub>3</sub> and R<sub>4</sub> are each independently hydrogen or lower alkyl having 1 to 6 carbon atoms;

R<sub>5</sub> is hydrogen, alkyl having 1 to 30 carbon atoms, phenyl, aralkyl or alkaryl having 7 to 36 carbon atoms, or an acyl group of the formula:



wherein R<sub>6</sub> is alkyl having 1 to 30 carbon atoms, phenyl, or aralkyl or alkaryl having 7 to 36 carbon atoms; n is an integer from 5 to 100; and x is an integer from 0 to 10; and

(b) a poly(oxyalkylene) amine having at least one basic nitrogen atom and a sufficient number of oxyalkylene units to render the poly(oxyalkylene)



amine soluble in hydrocarbons boiling in the gasoline or diesel range.

37. The fuel concentrate according to claim 36, wherein  $R_1$  of said poly(oxyalkylene) hydroxyaromatic ether is hydrogen, hydroxy, or lower alkyl having 1 to 4 carbon atoms;  $R_2$  is hydrogen; one of  $R_3$  and  $R_4$  is hydrogen and the other is methyl or ethyl;  $R_5$  is hydrogen, alkyl having 2 to 22 carbon atoms, alkylphenyl having an alkyl group containing 4 to 24 carbon atoms, or an acyl group having the formula:  $-C(O)R_7$ , wherein  $R_7$  is alkyl having 4 to 12 carbon atoms;  $n$  is 15 to 30; and  $x$  is 0, 1 or 2.

38. The fuel concentrate according to claim 37, wherein  $R_1$  of said poly(oxyalkylene) hydroxyaromatic ether is hydrogen or hydroxy;  $R_5$  is hydrogen, alkyl having 4 to 12 carbon atoms, or alkylphenyl having an alkyl group containing 4 to 12 carbon atoms; and  $x$  is 0.

39. The fuel concentrate according to claim 38, wherein  $R_1$  and  $R_5$  of said poly(oxyalkylene) hydroxyaromatic ether are both hydrogen.

40. The fuel concentrate according to claim 36, wherein said poly(oxyalkylene) amine is a hydrocarbyl poly(oxyalkylene) aminocarbamate.

41. The fuel concentrate according to claim 40, wherein the hydrocarbyl group of said hydrocarbyl poly(oxyalkylene) aminocarbamate contains from 1 to about 30 carbon atoms; and wherein the amine moiety of said hydrocarbyl poly(oxyalkylene) aminocarbamate is derived from a polyamine having from 2 to 12 amine nitrogen atoms and from 2 to 40 carbon atoms.

42. The fuel concentrate according to claim 41, wherein said hydrocarbyl group of said hydrocarbyl poly(oxyalkylene) aminocarbamate is an alkylphenyl group; and wherein said polyalkylene polyamine is selected from the group consisting of ethylenediamine, propylenediamine, diethylenetriamine and dipropylenetriamine.

43. The fuel concentrate according to claim 42, wherein the alkyl moiety of said alkylphenyl group is tetrapropenyl.

44. The fuel concentrate according to claim 40, wherein said hydrocarbyl poly(oxyalkylene) aminocarbamate is an alkylphenyl poly(oxybutylene) aminocarbamate, wherein the amine moiety is derived from ethylenediamine or diethylenetriamine.

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