

[54] OPTICALLY ACTIVE PHENYL ACETATES

[75] Inventors: Keimei Fujimoto, Hyogo; Nobuo Ohno, Osaka; Yoshitoshi Okuno, Osaka; Toshio Mizutani, Osaka; Isao Ohno, Osaka; Masachika Hirano, Osaka; Nobushige Itaya, Osaka; Takashi Matsuo, Hyogo, all of Japan

[73] Assignee: Sumitomo Chemical Co., Ltd., Japan

[*] Notice: The portion of the term of this patent subsequent to Dec. 13, 1994 has been disclaimed.

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Related U.S. Application Data

[60] Continuation of Ser. No. 837,973, Sep. 29, 1977, abandoned, which is a continuation of Ser. No. 596,334, Jul. 16, 1975, Pat. No. 4,062,968, which is a division of Ser. No. 378,301, Jul. 11, 1973, Pat. No. 3,996,244.

[30] Foreign Application Priority Data

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Apr. 19, 1973 [JP] Japan 48-44809

[51] Int. Cl.³ C07D 317/44; C07C 69/76; C07C 121/75

[52] U.S. Cl. 549/447; 260/465 D; 560/105

[58] Field of Search 560/105; 260/465 D; 549/447

[56] References Cited

U.S. PATENT DOCUMENTS

- 3,567,740 3/1971 Matsui et al. 260/347.4
3,666,789 5/1972 Itaya et al. 560/124
3,673,215 6/1972 Vollrath et al. 560/124 X
3,766,218 10/1973 Ueda et al. 560/124 X
3,835,176 9/1974 Matsuo et al. 260/465 D
3,857,863 12/1974 Ohno et al. 260/347.4
4,062,968 12/1977 Fujimoto et al. 424/275

FOREIGN PATENT DOCUMENTS

738112 3/1970 Belgium .

919193 11/1946 France .
919197 11/1946 France .

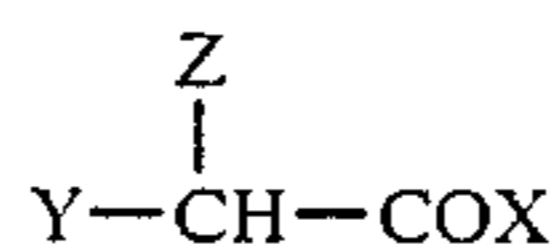
OTHER PUBLICATIONS

- Staudinger and Ruzick, Helvetia Chimica Acta, pp. 393, 404 and 454-458, vol. 7, (1924).
"Synthetic Pyrethroids", Advances in Pest Control Research, Metcalf, Ed., Interscience Publishers, Inc. (1961), pp. 31-74.
W. F. Barthel, "Insecticidal Activity of Barthrins, Dimethrin & Related Compounds", Present at 5th Int'l. Pesticide Congress, 7/17/63, London, England.
J. Baker, "Dimethrin-A New Mosquito Larviciding Material", Pest Control, 33 (1), 32 (1965).
M. Elliot, "New Insecticidal Esters of Chrysanthemic Acid", Proc. Third. Br. Ins. & Fung. Conference, 1965, pp. 432-439.
M. Elliot et al., "New Pyrethrin-Like Esters With High Insecticidal Activity", Nature, vol. 207, No. 5000, pp. 938-940, Aug. 28, 1965.
Berteau et al., "Pyrethroid-Like Biological Activity of Compounds Lacking Cyclopropane & Ester Groupings", Science, vol. 16, pp. 1151-1153, (9/19/68).
M. Elliot, "The Relationship Between the Structure & Activity of Pyrethroids", Bulletin World Health Organization, vol. 44, No. 1, 2, 3, pp. 315-324 (1971).

Primary Examiner—Dolph H. Torrence
Attorney, Agent, or Firm—Sughrue, Mion, Zinn, Macpeak, and Seas

[57] ABSTRACT

Substituted acetate compounds represented by the formula



which are useful as pesticides and pesticidal composition containing the substituted acetate compounds as active ingredients, as well as processes for preparing the compounds and the compositions are disclosed.

30 Claims, No Drawings

OPTICALLY ACTIVE PHENYL ACETATES

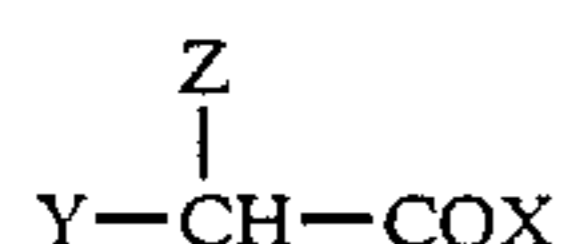
This is a continuation of application Ser. No. 837,973, filed Sept. 29, 1977, abandoned, which is a continuation of application Ser. No. 596,334, filed July 16, 1975, U.S. Pat. No. 4,062,968, which is a division of application Ser. No. 378,301, filed July 11, 1973, U.S. Pat. No. 3,996,244.

SUMMARY OF THE INVENTION

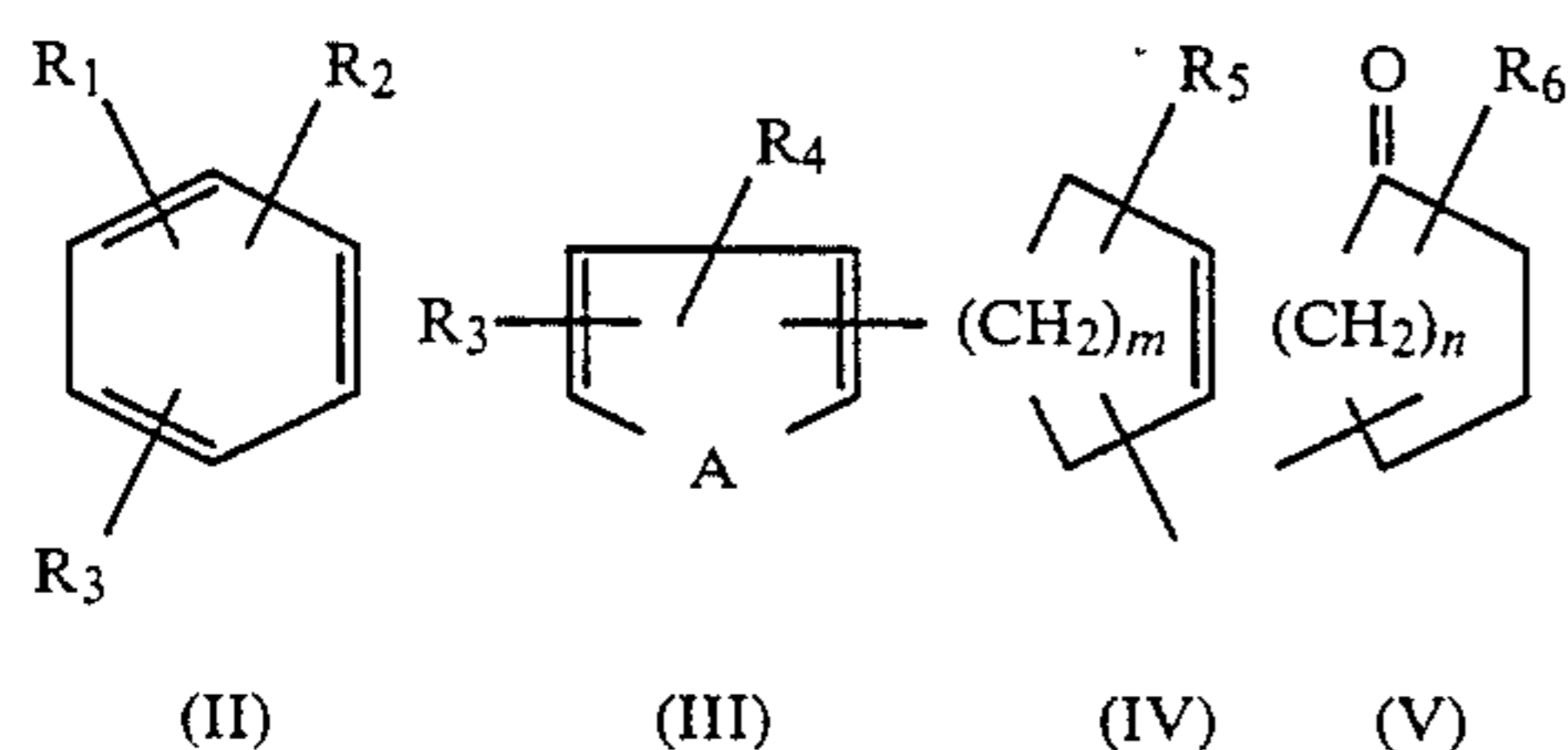
1. Field of the Invention

This invention relates to a pesticide, and more particularly, this invention relates to substituted acetate compounds of the following formula (I) useful for controlling an arthropod, pesticidal compositions containing as active ingredients at least one substituted acetate compound of the formula (I), as well as processes for preparing the substituted acetate compounds and the pesti-

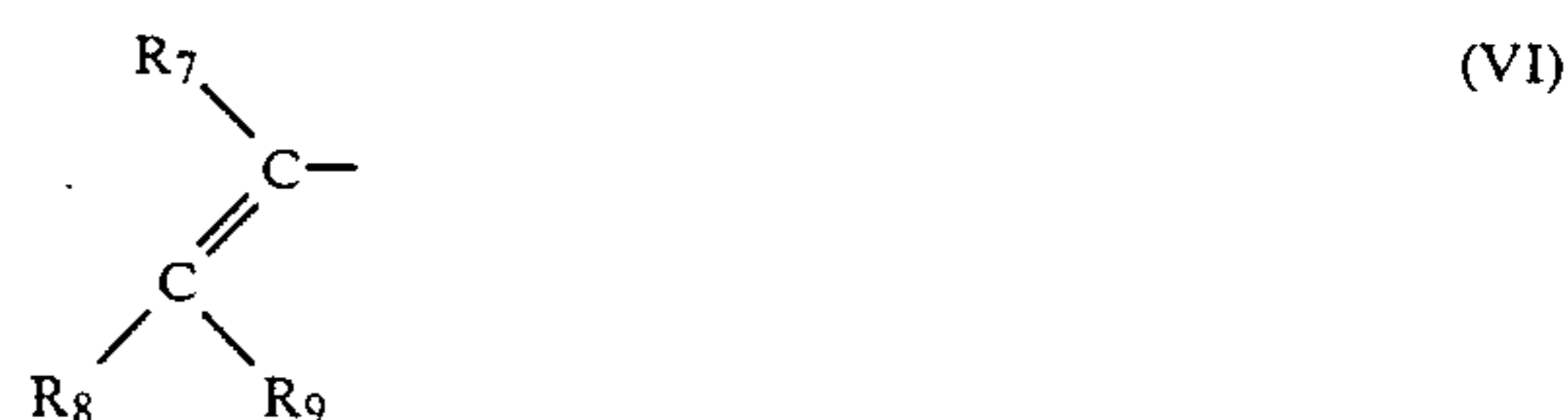
cidal compositions. The substituted acetate compounds of this invention are represented by the formula:



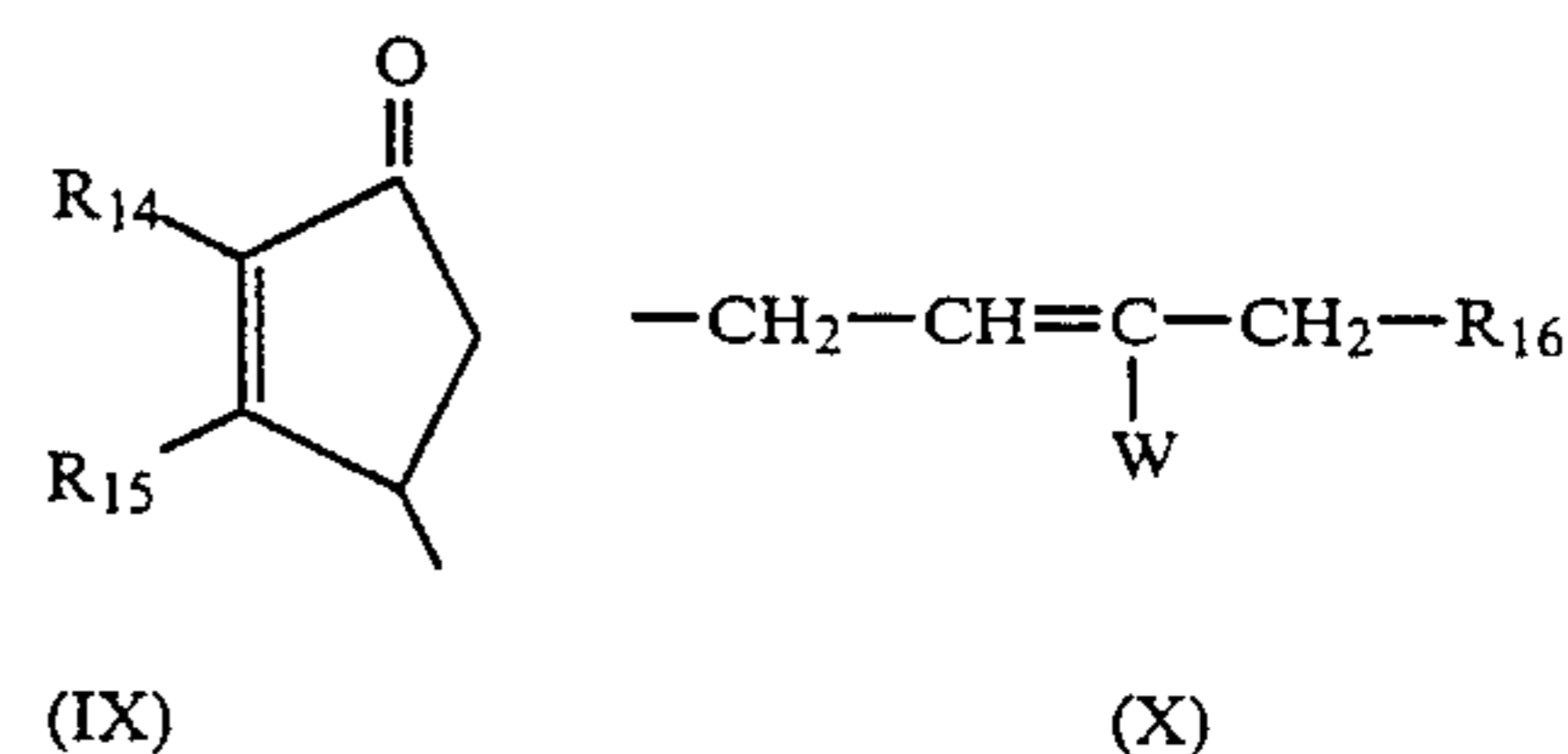
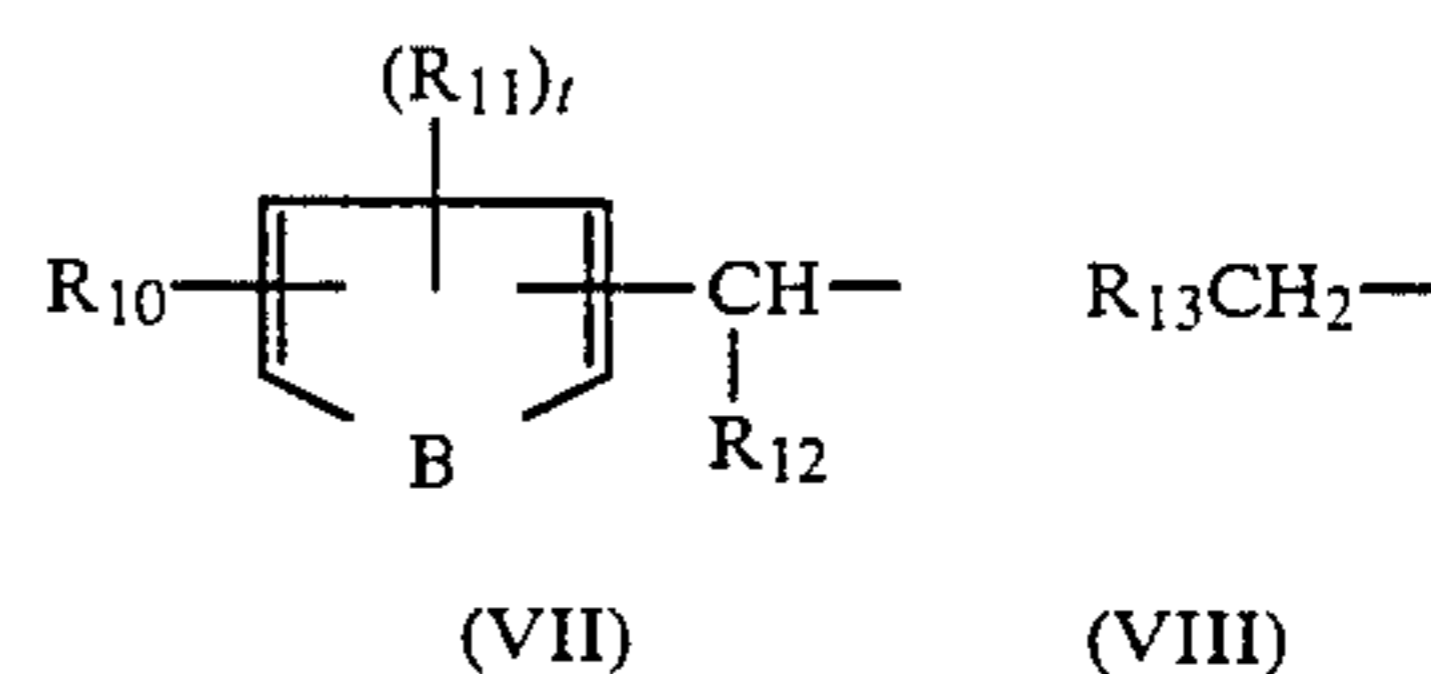
wherein Y represents a substituted aromatic group or an unsaturated Alicyclic group having an appropriate plane and is selected from the group consisting of the formulae



in which R₁ and R₂ are individually hydrogen, halogen, cyano, nitro, lower alkyl, lower alkenyl, lower alkynyl, lower alkoxy, lower alkoxyalkyl, halogen-substituted lower alkyl, halogen-substituted lower alkenyl, halogen-substituted lower alkynyl, lower-alkylthio, lower alkylsulfoxyl, acyl, acyloxy, lower alkoxy carbonyl, lower alkenyloxycarbonyl, lower alkynyloxycarbonyl, methylenedioxy, tetramethylene or trimethylene group; R₃ and R₄ are individually hydrogen, halogen, cyano, nitro, lower alkyl, lower alkenyl, lower alkynyl, lower alkoxy lower alkoxyalkyl, halogen-substituted lower alkyl, halogen-substituted lower alkenyl, halogen-substituted lower alkynyl, lower alkylthio, lower alkylsulfoxyl, acyl, acyloxy, lower alkoxy carbonyl, lower alkenyloxycarbonyl or lower alkynyloxycarbonyl group; A represents an oxygen atom or a sulfur atom; R₅ represents hydrogen, halogen, cyano, nitro or lower alkyl group; m is an integer of from 1 to 3; R₆ represents hydrogen, halogen, cyano, nitro or lower alkyl group; n is an integer of from 1 to 3; and the dotted line in the formula (V) represents a double bond present at a position conjugated or non-conjugated with the ketone (C=O), or Y represents a substituted ethylene group of the formula (VI)

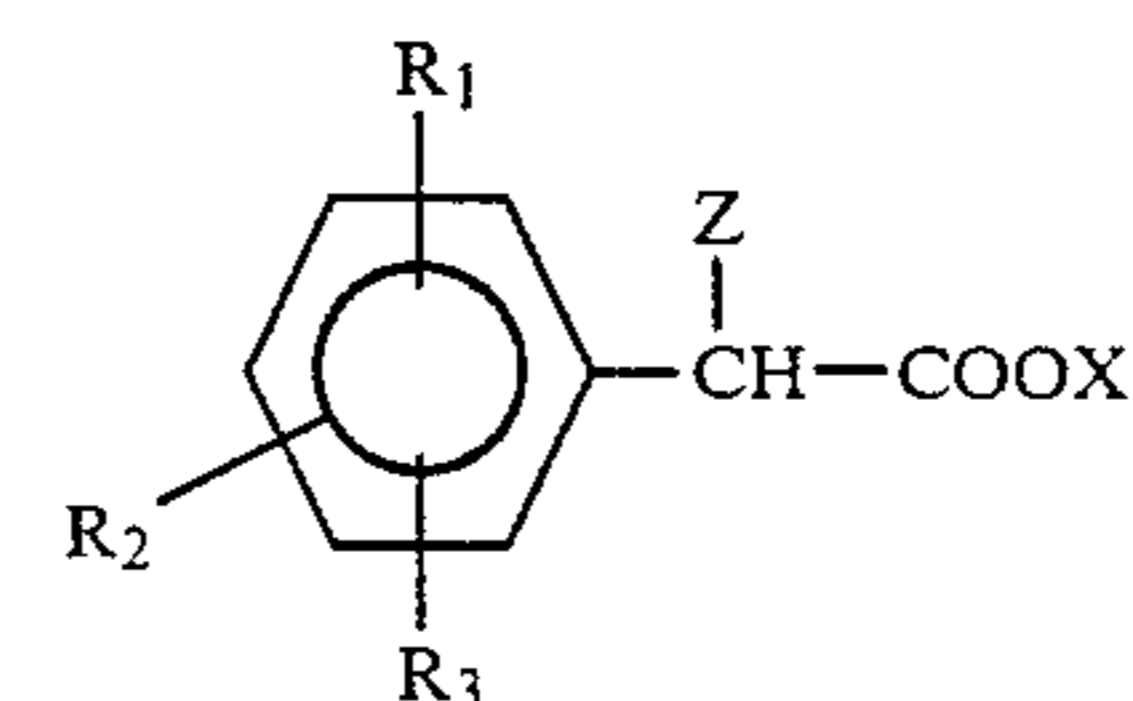


in which R₇, R₈ and R₉ constitutes a plane together with the double bond at β—γ position of the ester group, and are individually hydrogen, lower alkyl, lower alkenyl, lower alkynyl, halogen, acyl or acyloxy; Z represents a straight or branched lower alkyl, lower alkenyl, lower alkynyl, lower alkoxy, cyano, halogen-substituted lower alkyl, halogen-substituted lower alkenyl group or an alicyclic group having 3 to 7 carbon atoms; X represents a group corresponding to an alcohol moiety of the ester-type insecticides which are known as pyrethroid insecticides, and is selected from the group consisting of the formulae



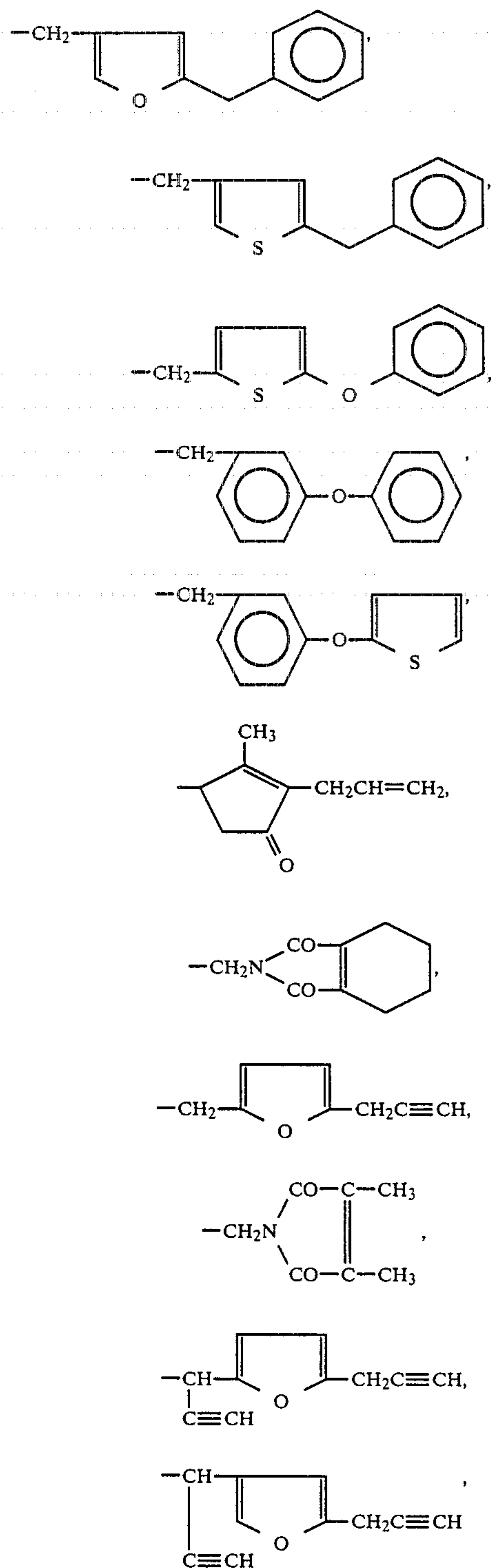
in which R₁₀ represents an allyl, propargyl, benzyl, thenyl, furylmethyl, phenoxy or phenylthio group; R₁₁ represents hydrogen, methyl, trifluoromethyl or halogen, R₁₀ and R₁₁ may, when they are bonded to each other at terminals, form a trimethylene or tetramethylene group; R₁₂ represents hydrogen, ethynyl or cyano; t is an integer of from 1 to 2; B represents an oxygen or sulfur atom or a group —CH=CH—; R₁₃ represents phthalimide, thiophthalimide, di- or tetrahydrophthalimide or dialkylmaleimide; R₁₄ represents allyl, propargyl, benzyl or alkadienyl; R₁₅ represents hydrogen or methyl; R₁₆ represents phenyl, thienyl or furyl; and W represents methyl or halogen. Preferred ranges of compounds within the formula (I) are as follows.

(1) compounds of the formula;

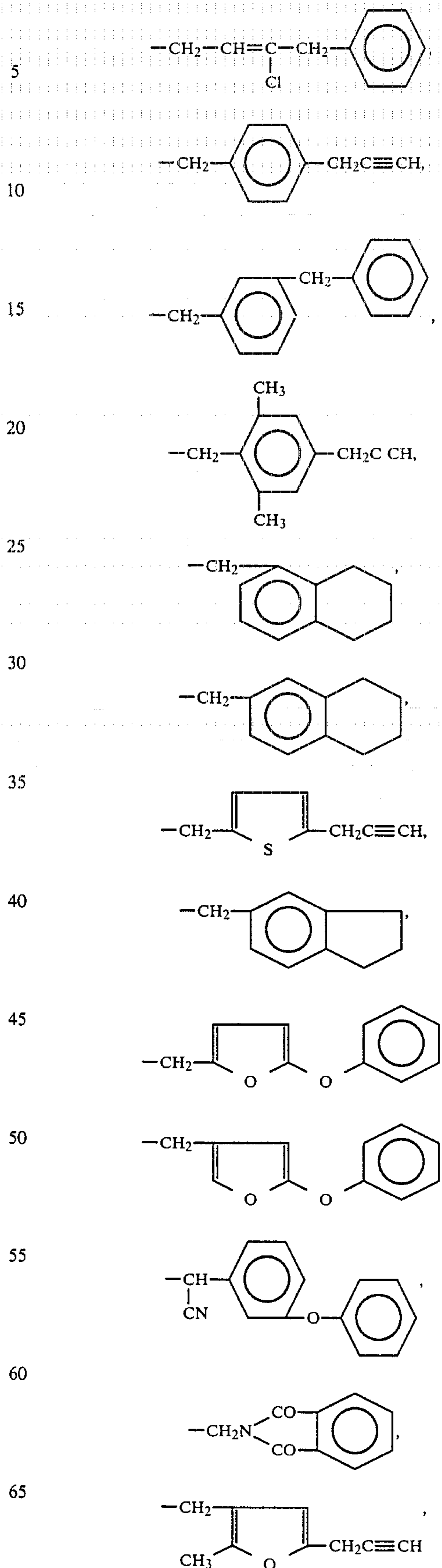


wherein R₁ is a member selected from the group consisting of hydrogen, methoxy, ethoxy, acetoxy, methylsulfinyl, C₁–C₄ alkyl, trifluoromethyl, allyl, acetyl, ethoxycarbonyl, methylenedioxy, methylthio, trimethylene, tetramethylene, chlorine, fluorine, iodine, isopropenyl, propargyl, methoxymethyl, ethoxymethyl, chloroethylene, chloroallyl, butyryl, butylthio, allyloxycar-

bonyl, nitro and methoxycarbonyl; R₂ is a member selected from the group consisting of hydrogen, methyl, methoxy, chlorine and bromine; R₃ is hydrogen or methyl; Z is a member selected from the group consisting of C₁-C₄ alkyl, ethoxy, allyl, bromoethyl, cyclohexyl, cyclopropylmethyl, isopropenyl, propargyl, trifluoromethyl and cyano; and X is a member selected from the group consisting of:

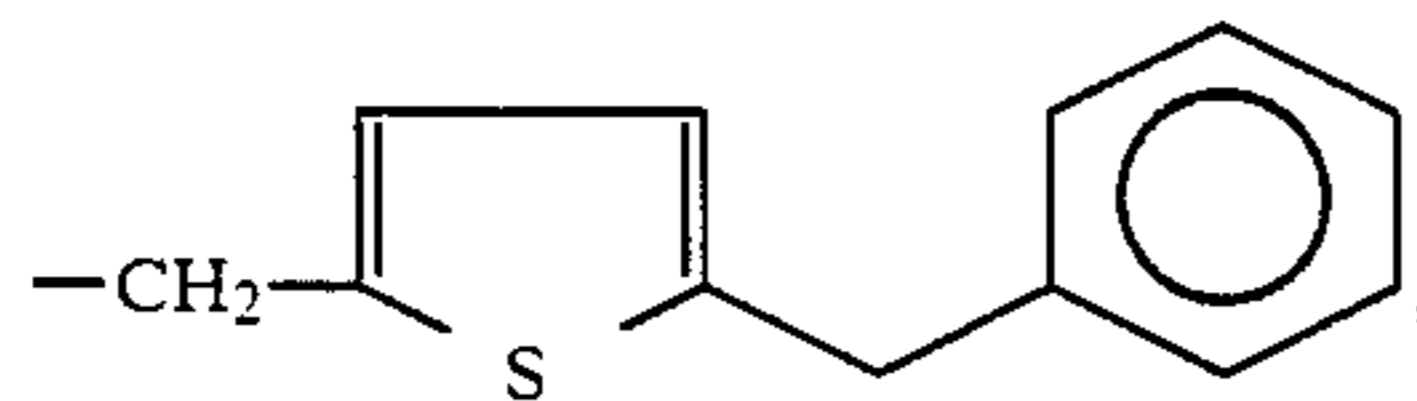
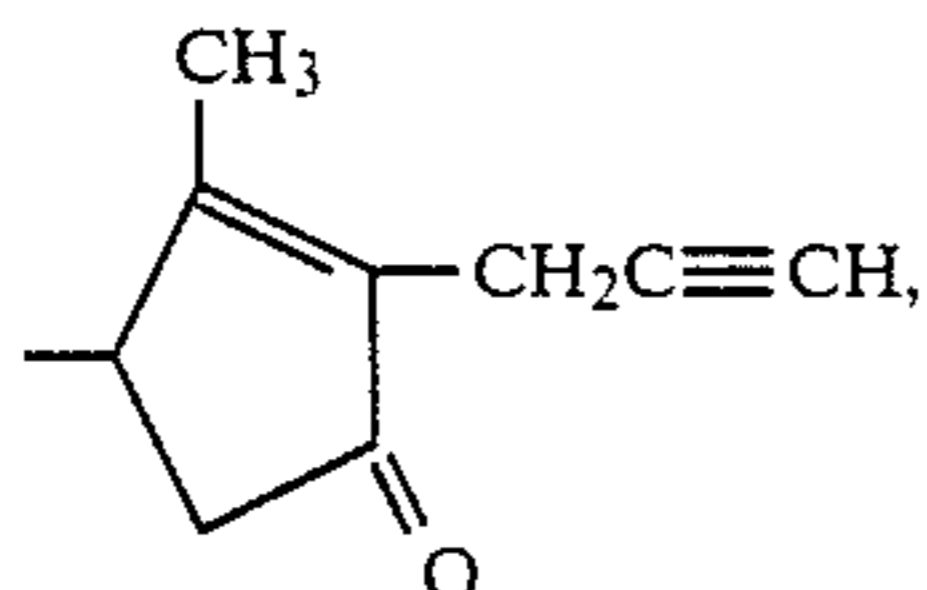
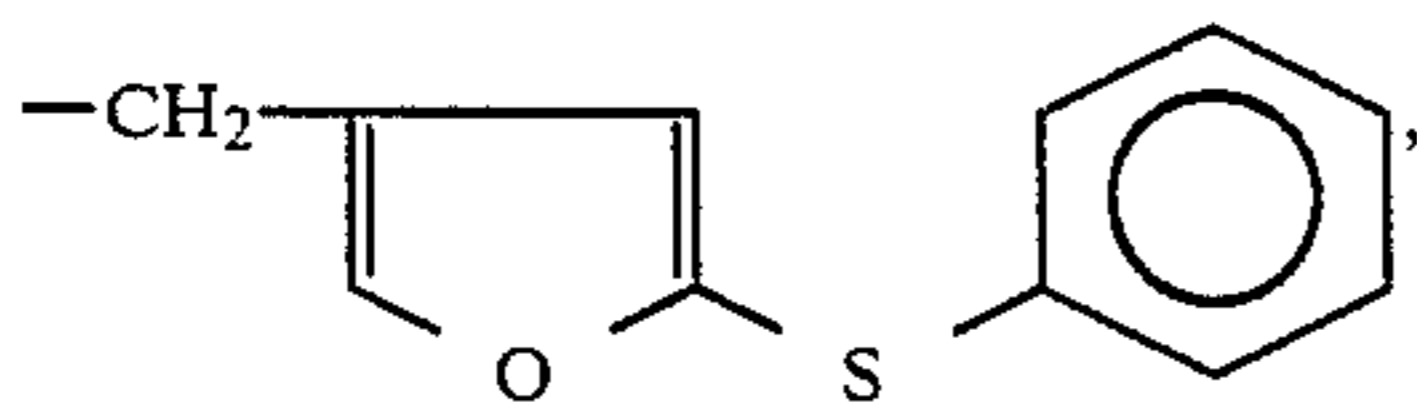
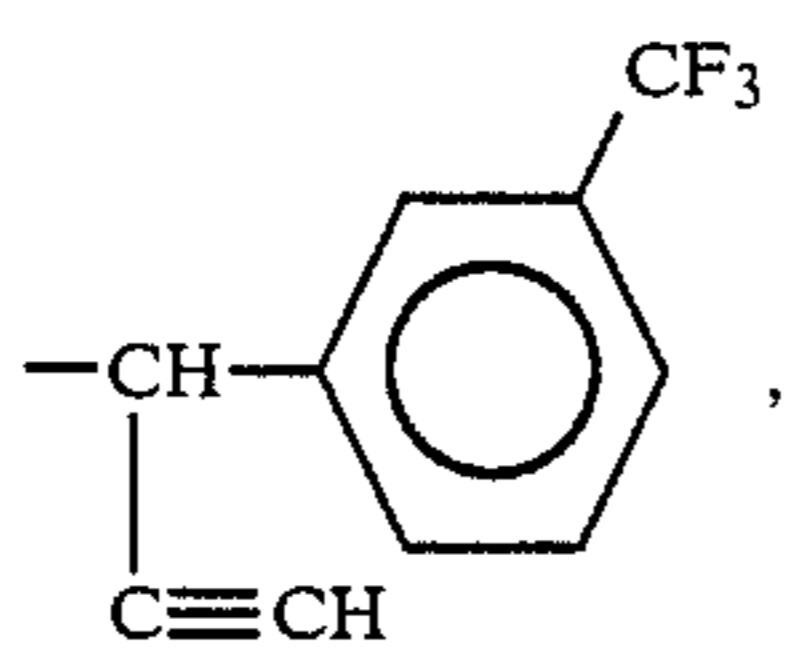
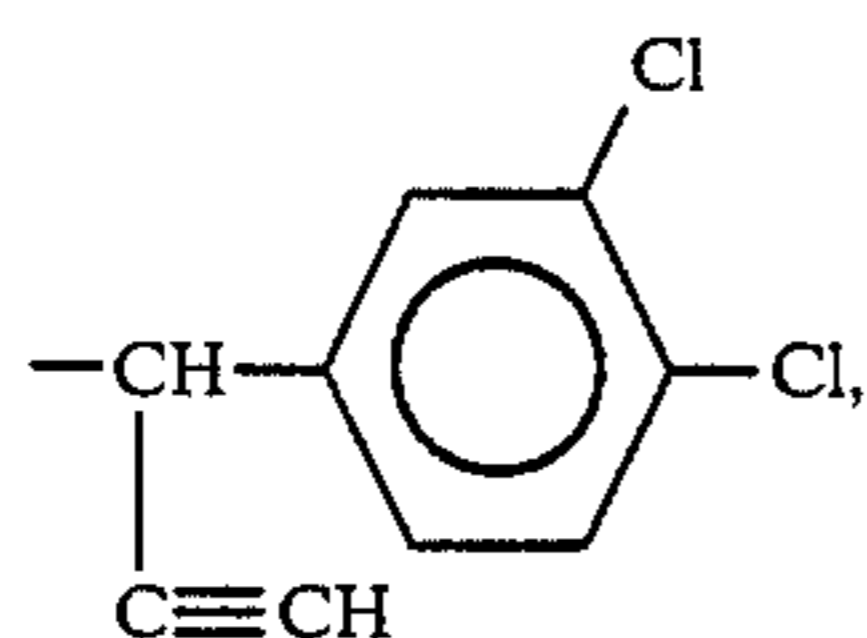
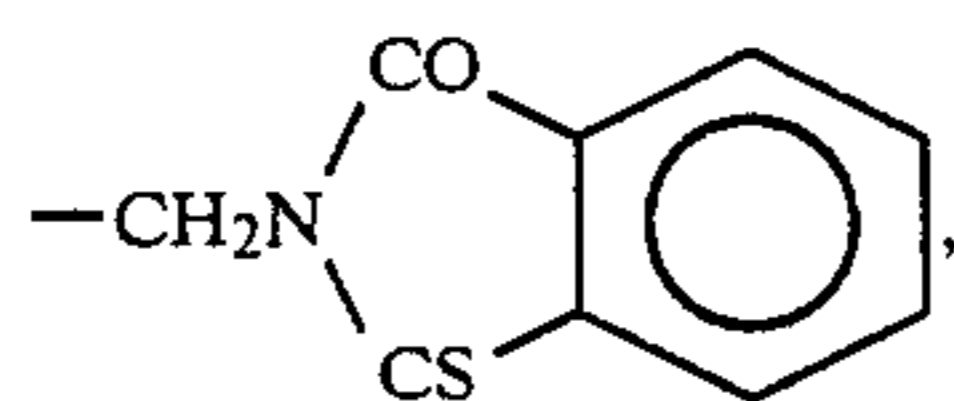
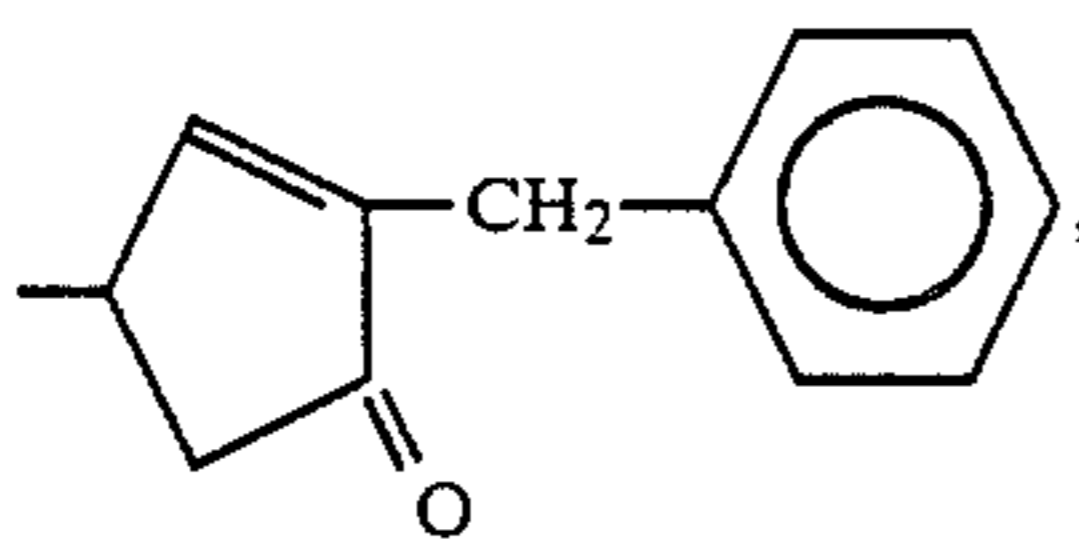
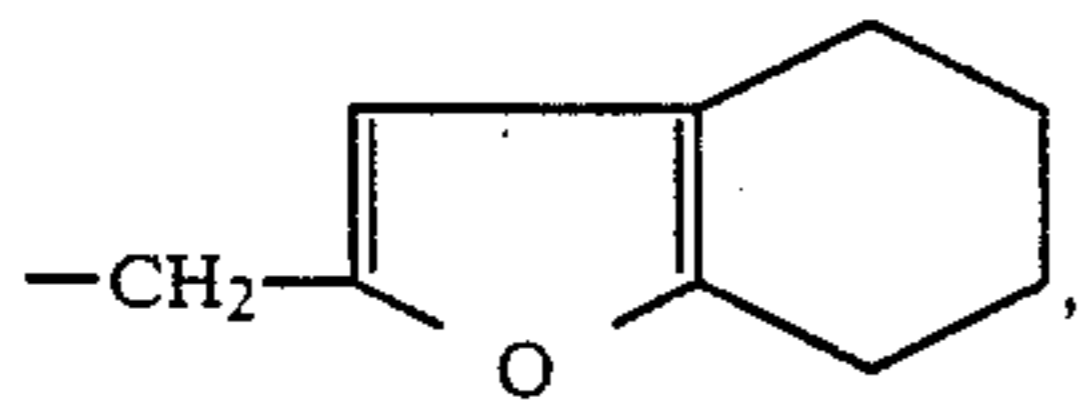
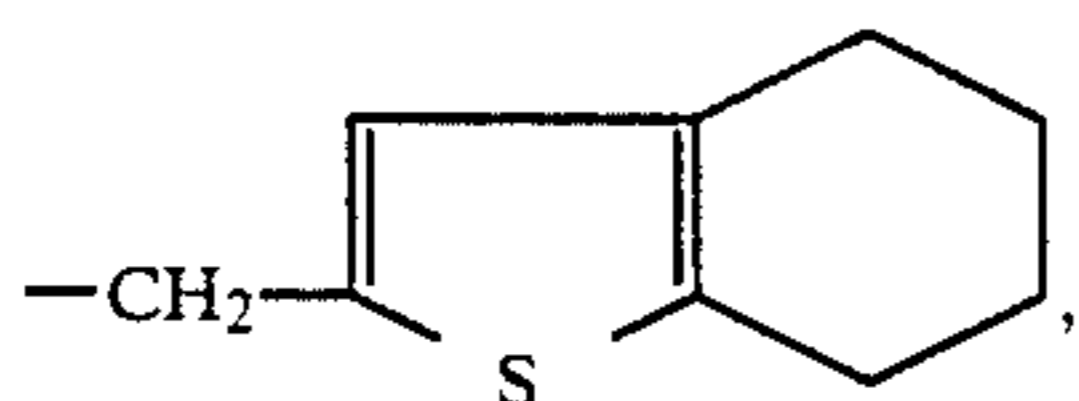
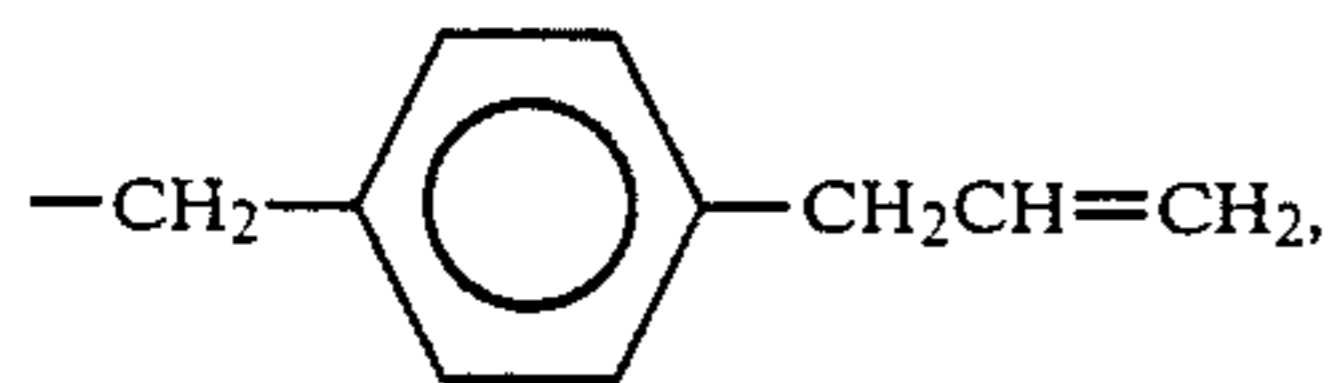
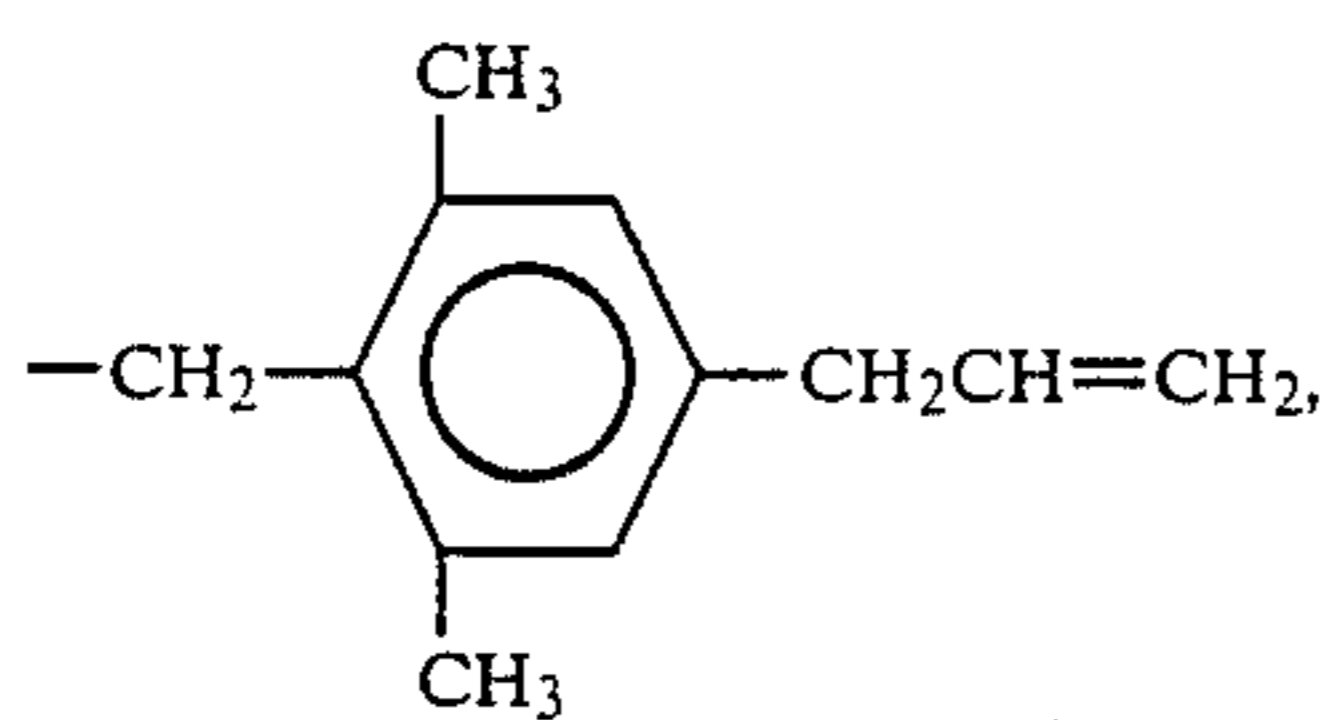


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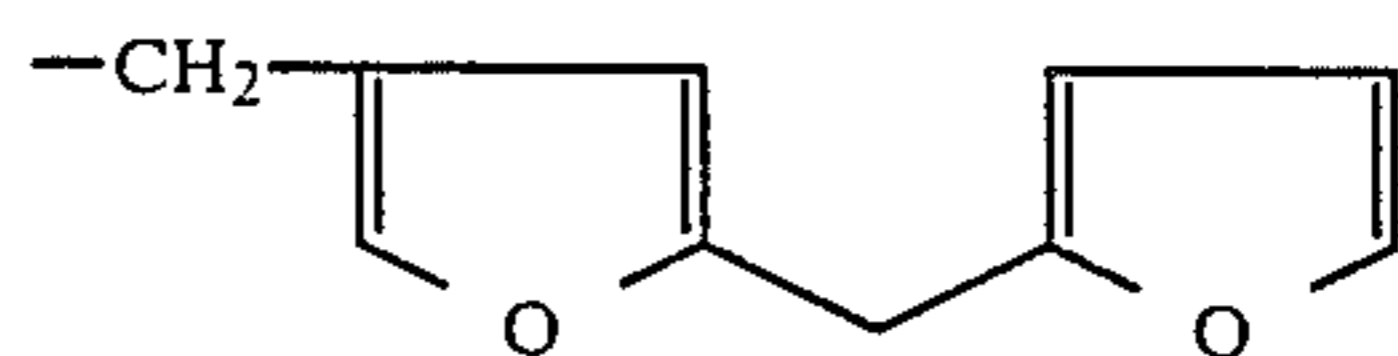
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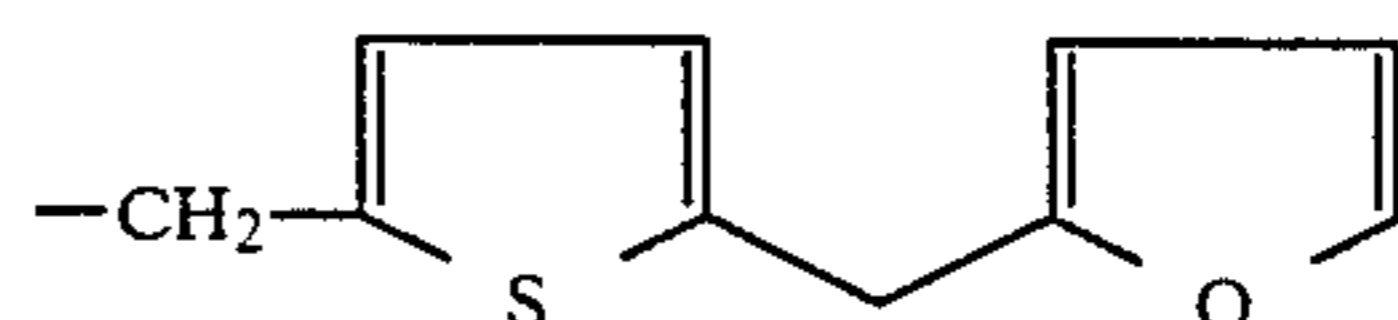


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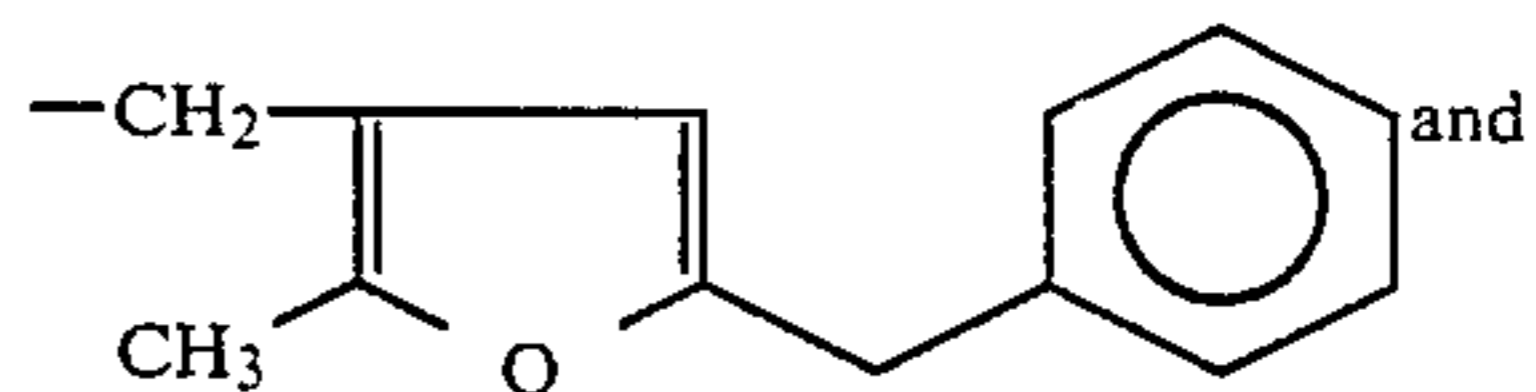
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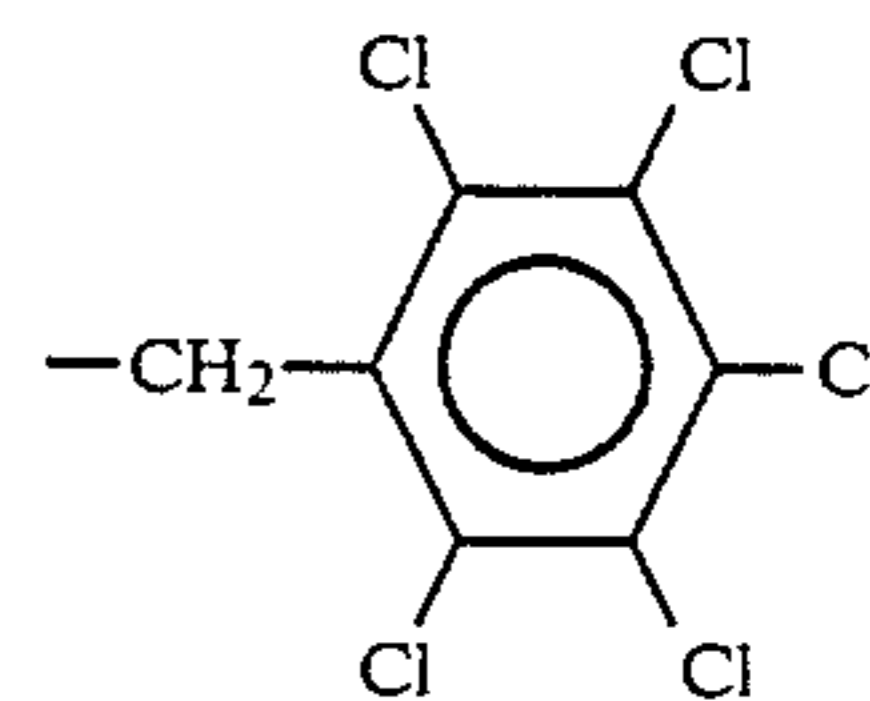
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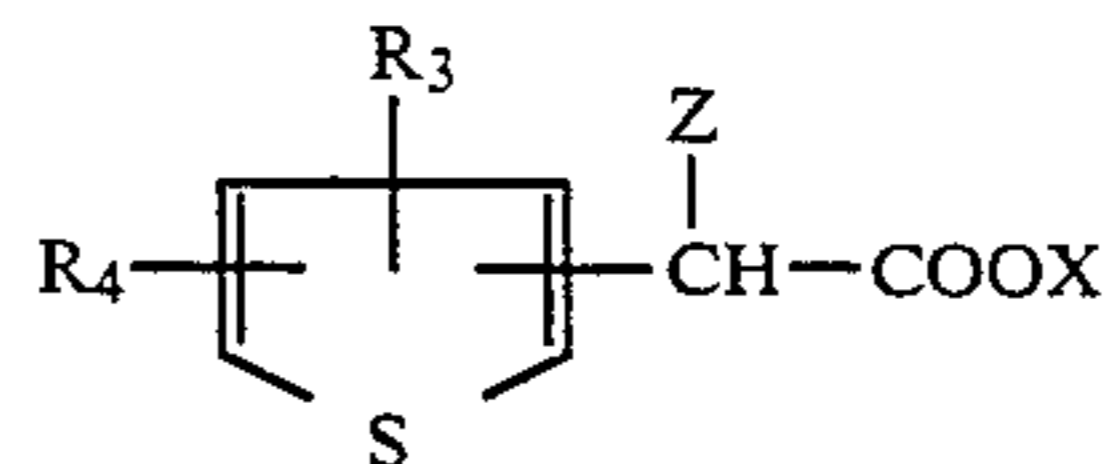
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(2) compounds of the formula;

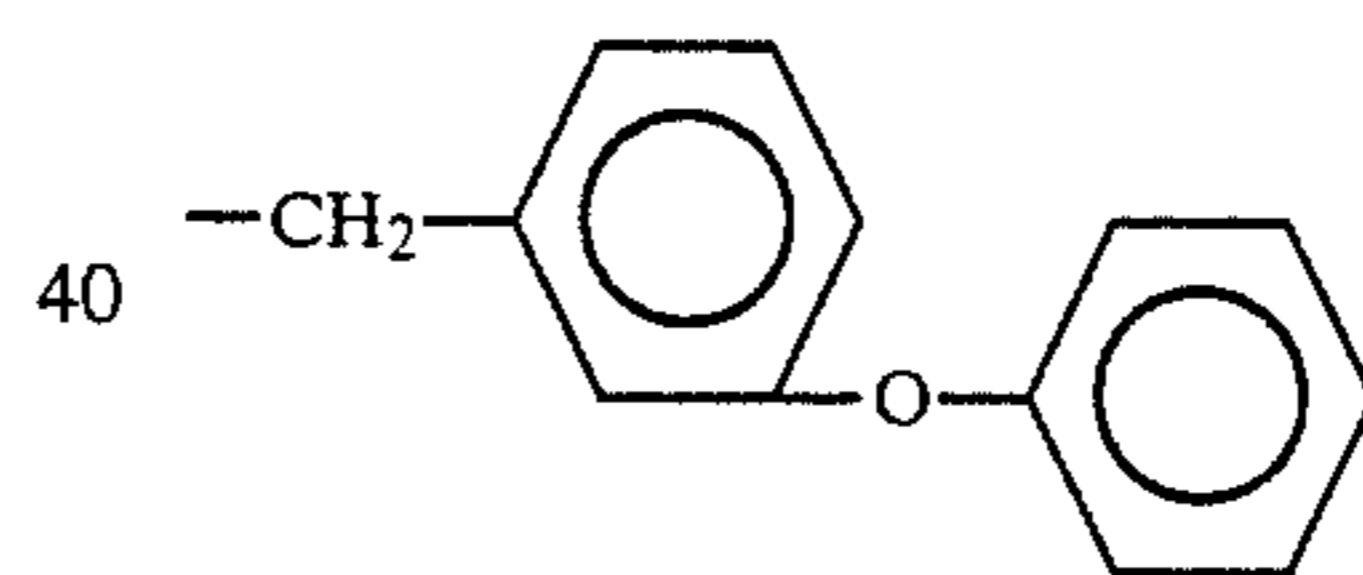
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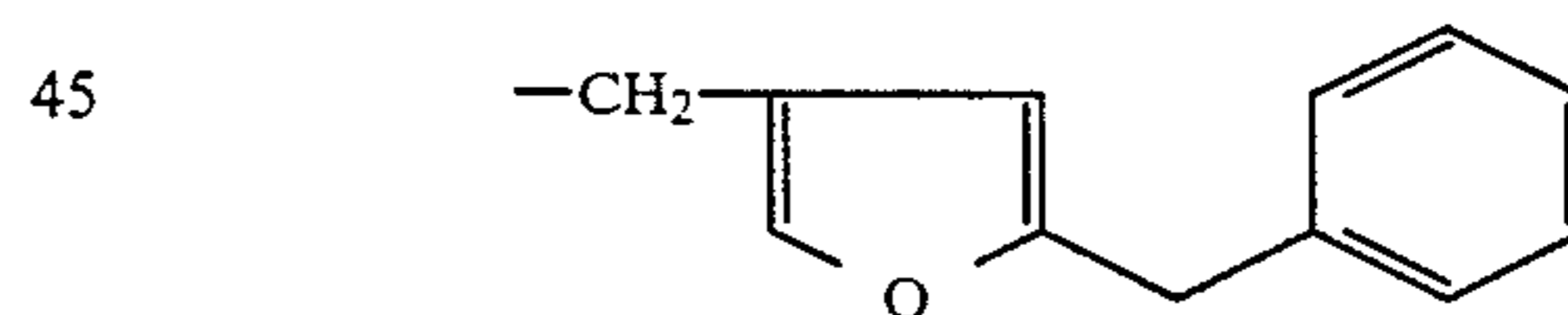
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wherein R_3 is a member selected from the group consisting of hydrogen, methyl, allyl, methoxycarbonyl and acetyl; R_4 is hydrogen; Z is a member selected from the group consisting of ethyl, propyl and allyl; and X is a member selected from the group consisting of

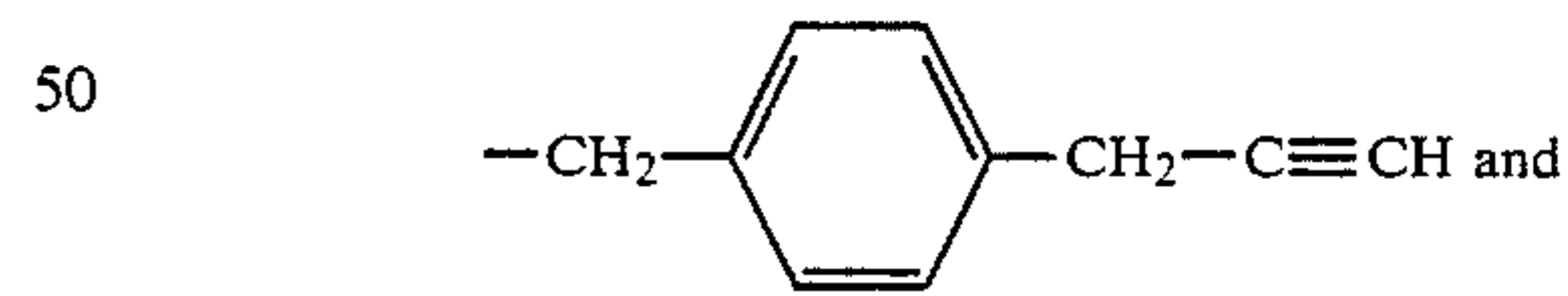
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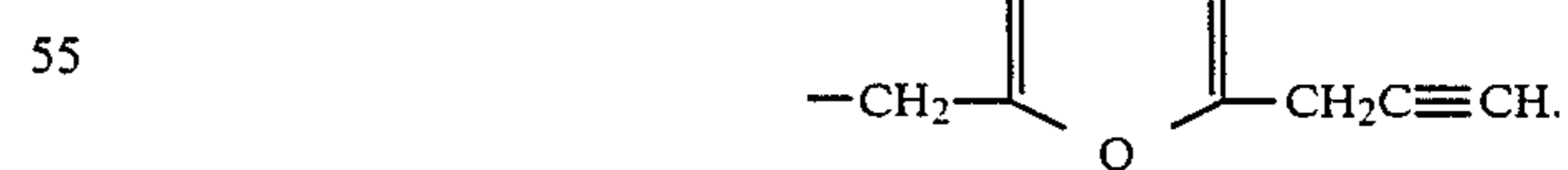
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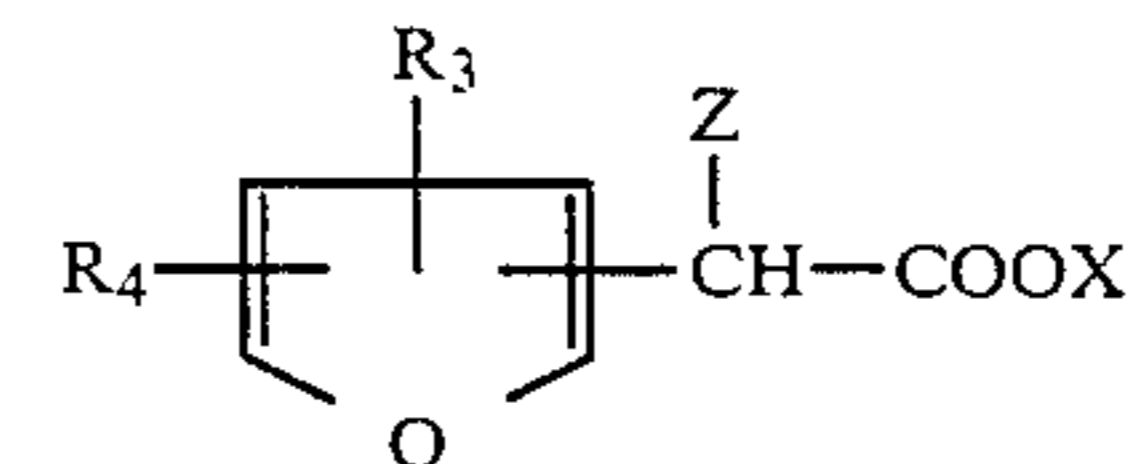
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(3) compounds of the formula;

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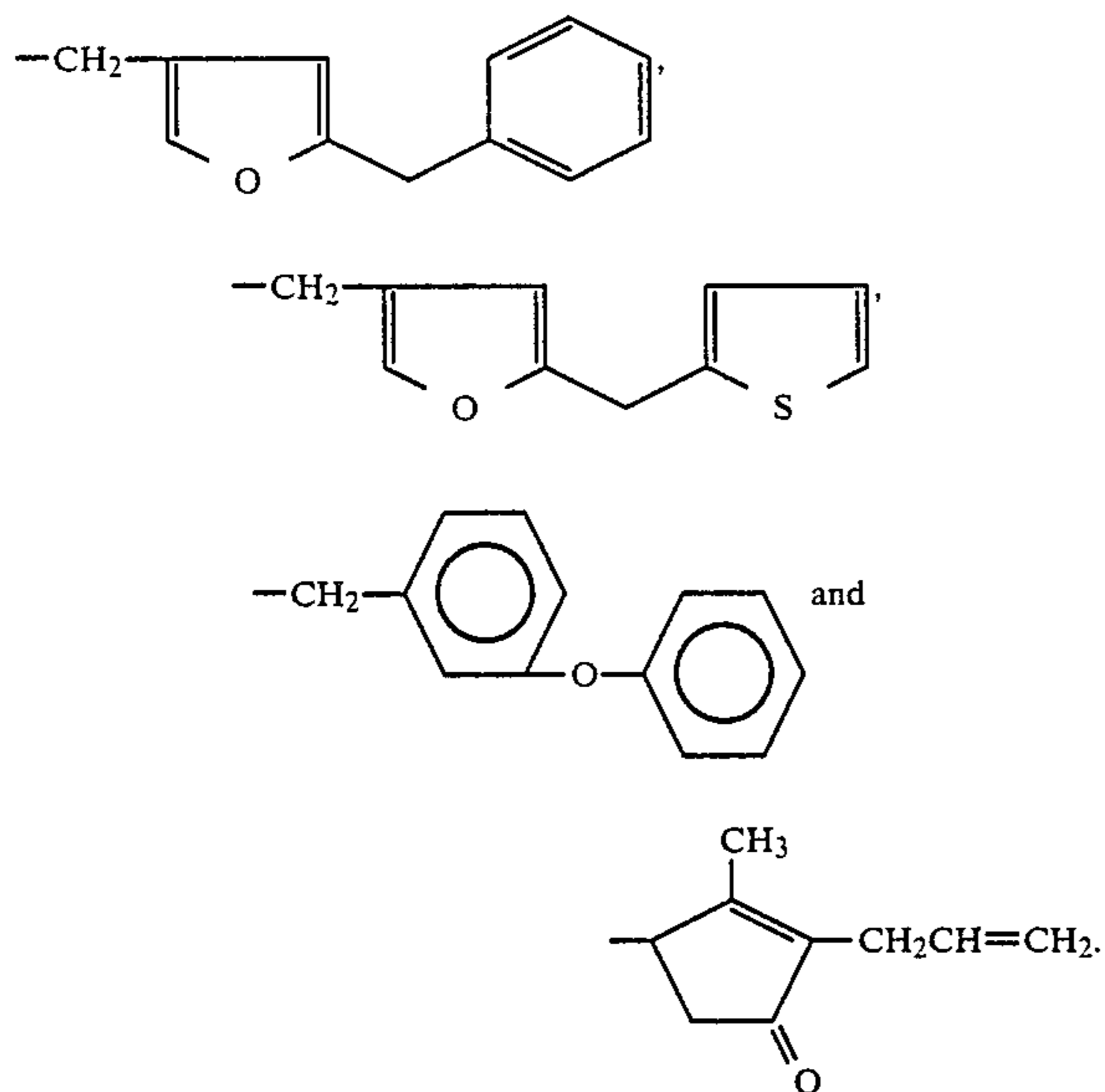


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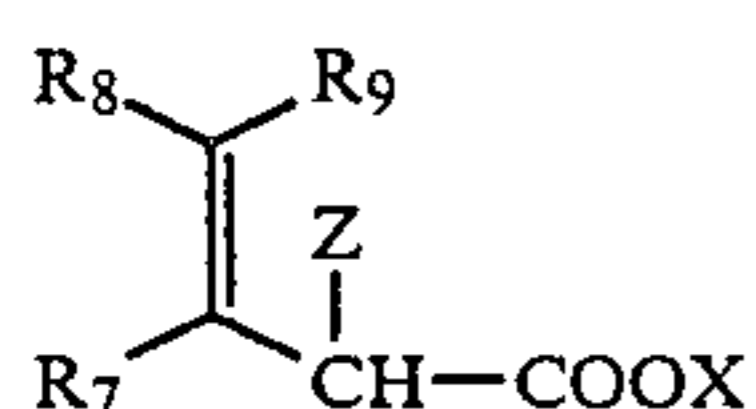
wherein R_3 is a member selected from the group consisting of hydrogen, methyl, allyl, methoxycarbonyl and

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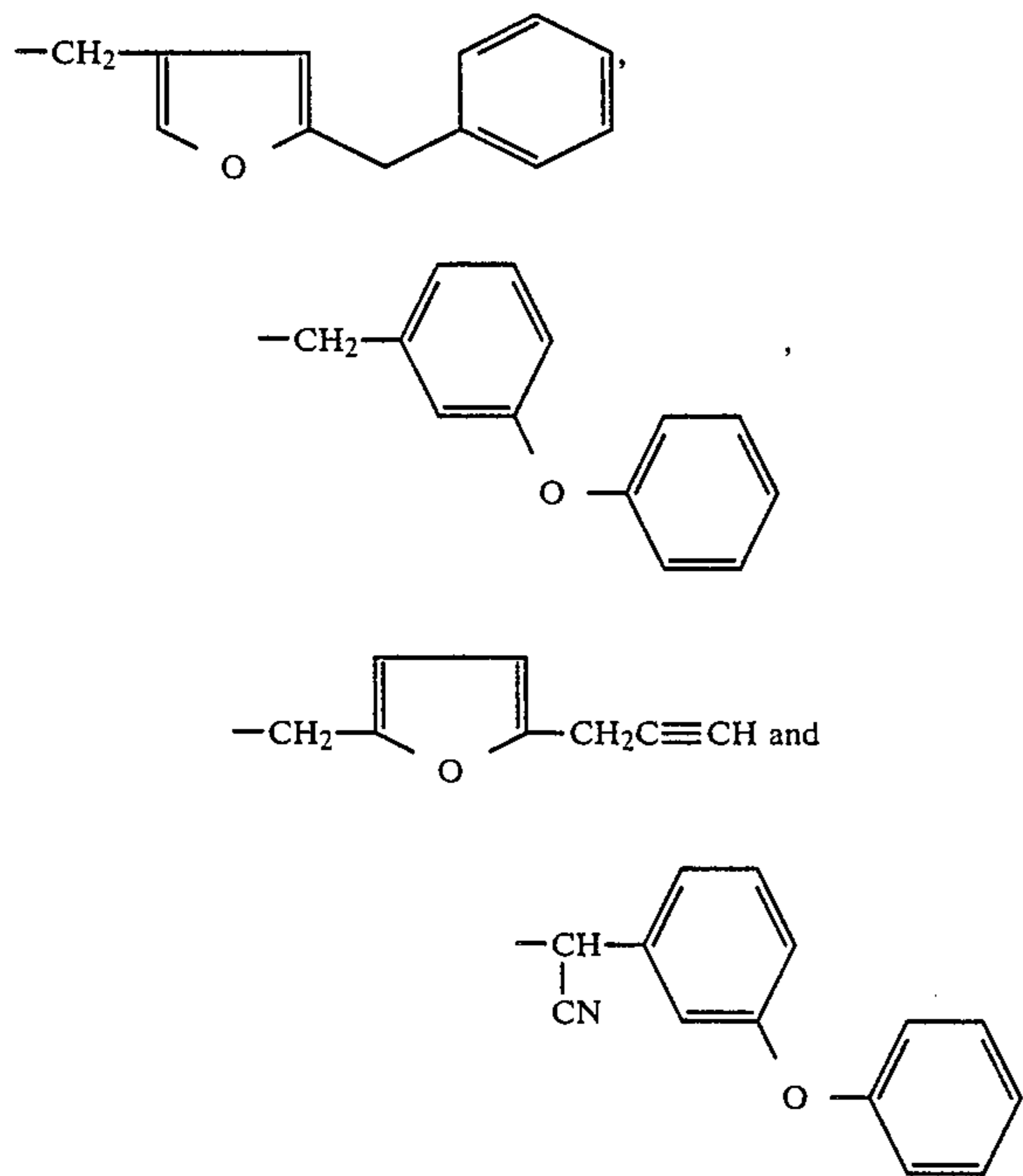
butyryl; R_4 is hydrogen; Z is ethyl or propyl; X is a member selected from the group consisting of



(4) compounds of the formula;

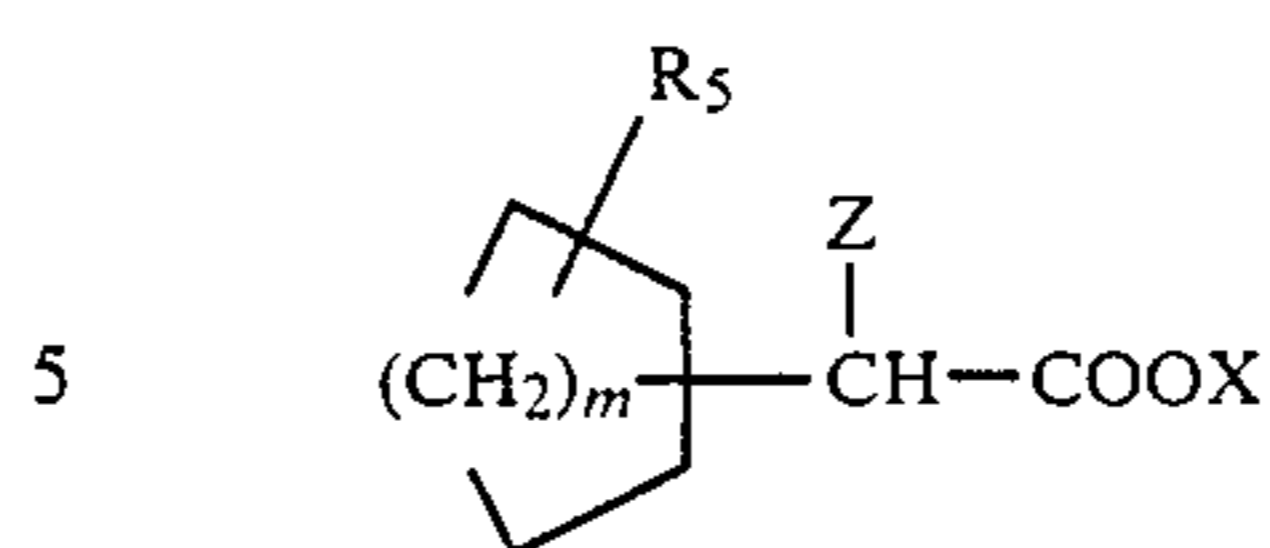


wherein R_7 is a member selected from the group consisting of hydrogen, methyl and ethyl; R_8 is hydrogen or methyl; R_9 is methyl; Z is C_1 - C_3 alkyl; and X is a member selected from the group consisting of

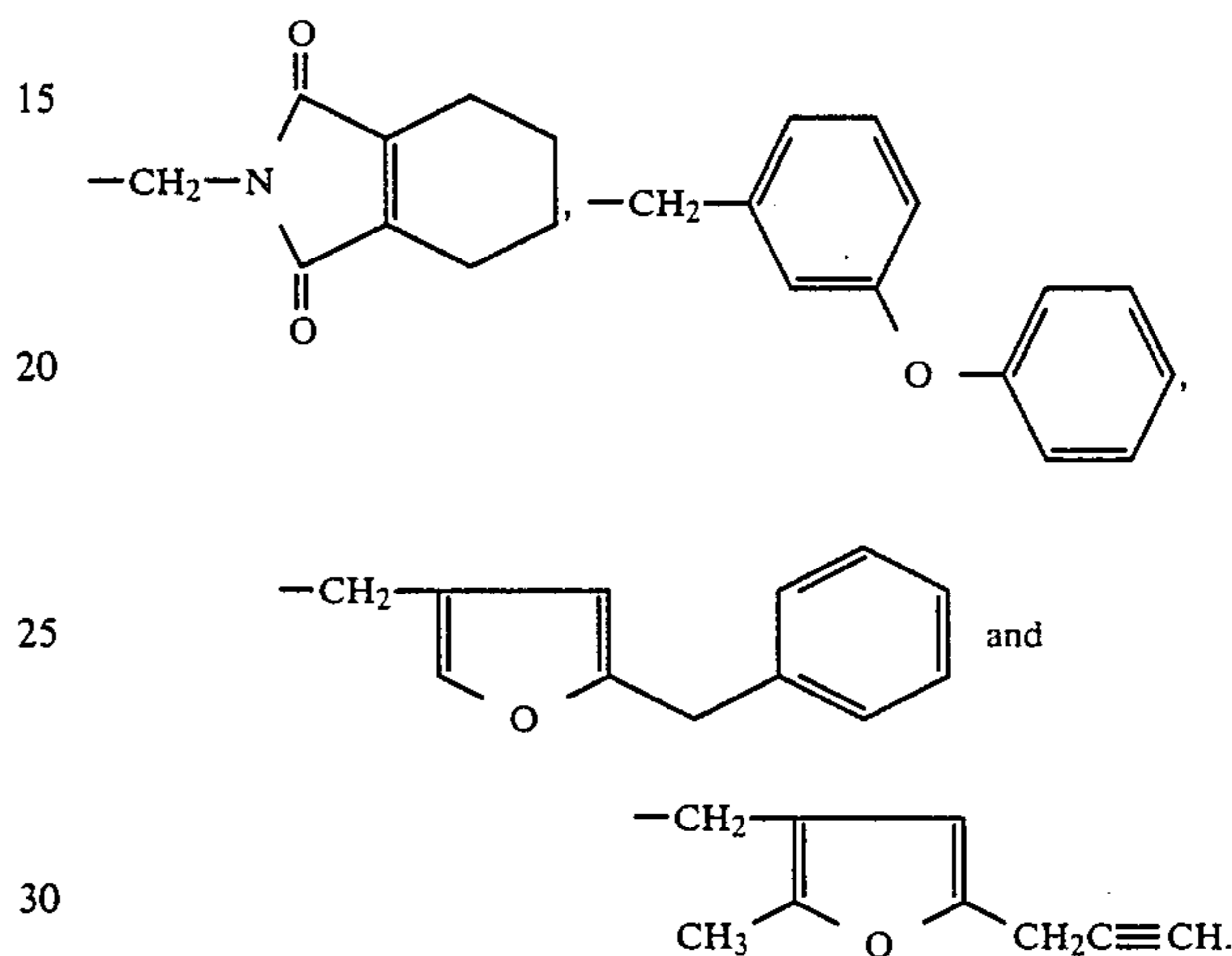


(5) compounds of the formula;

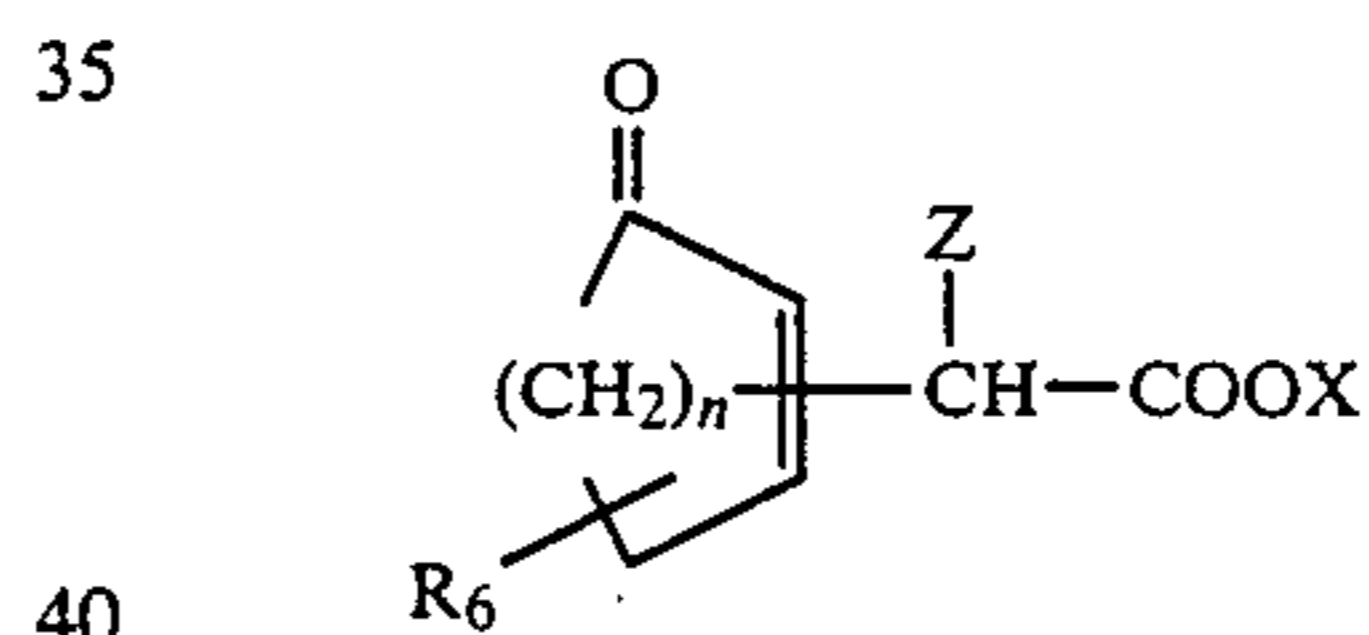
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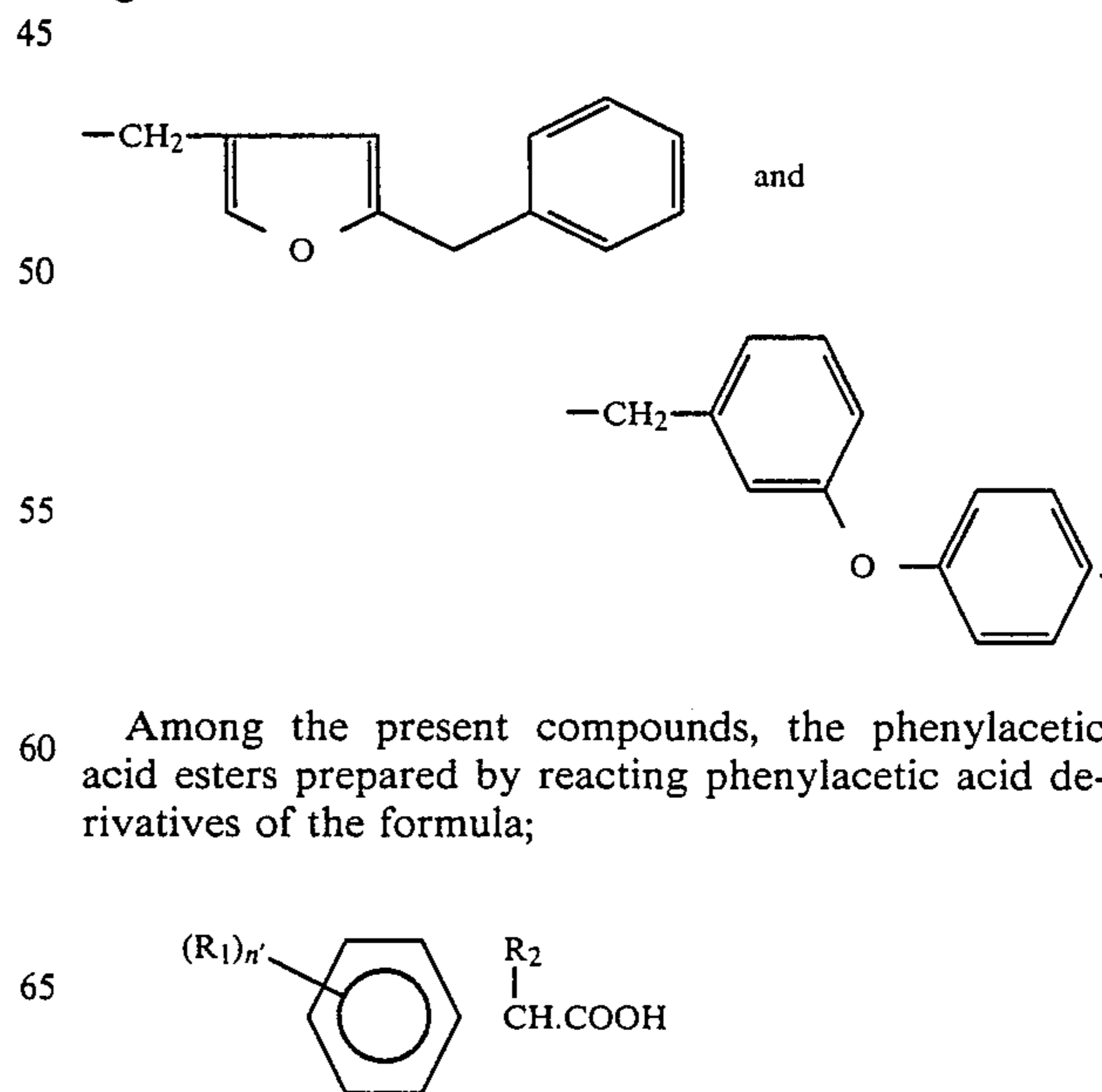
wherein R_5 is hydrogen or methyl; m is an integer of 1 or 2; Z is a member selected from the group consisting of methyl, propyl, allyl and propargyl; X is a member selected from the group consisting of



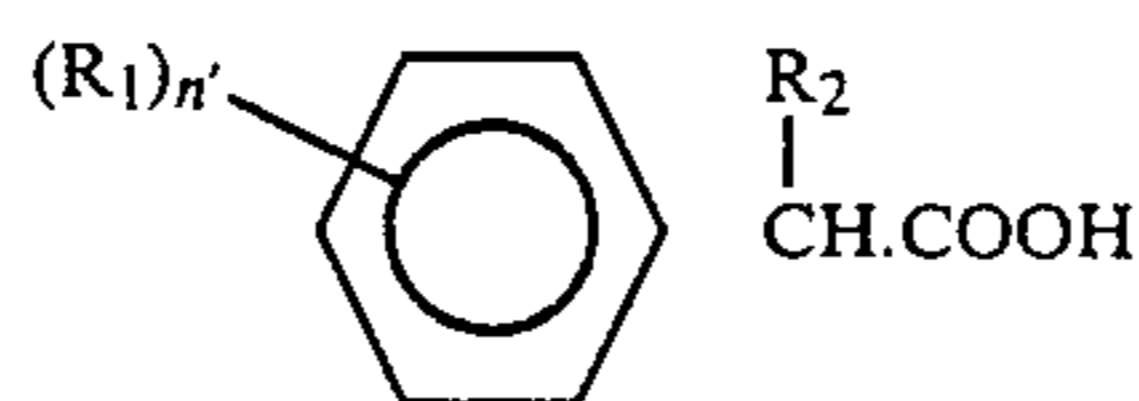
(6) compounds of the formula;



wherein R_6 is methyl, n is an integer of 1, Z is ethyl or propyl; X is a member selected from the group consisting of



Among the present compounds, the phenylacetic acid esters prepared by reacting phenylacetic acid derivatives of the formula;



wherein R₁ represents a straight or branched chain alkyl group having 1 to 5 carbon atoms, a straight or branched chain alkoxy group having 1 to 5 carbon atoms, a halogen atom or a thiomethyl group and two of R₁ may form a trimethylene —(CH₂)₃— or tetramethylene —(CH₂)₄— or a methylenedioxy group —O—CH₂—O—; n' is an integer of from 1 to 4; and R₂ represents a straight or branched alkyl group having 3 to 5 carbon atoms, allyl group, a propargyl group or isopropenyl group; with a pyrethroid alcohol have a prominent effects to injurious insects and low toxicity to mammals. And, therefore, the above mentioned phenylacetic acids are important as intermediates of the present invention.

2. Description of the Prior Art

Since BHC and DDT have been developed, remarkable advances have been made in the organic agricultural agents such as phosphorus agents, carbamates, pyrethroids and the progress of such organic agricultural agents now establishes a certain close correlation between the consumed quantity of pesticides and the harvested quantity of crops. However, on the other hand, these agricultural agents have caused pollutions by the environment contamination and at present there are many problems such as the accumulation of toxicity and the chronic toxicity as well as the generation of a wide variety of pesticide-resistant noxious insects. The desirable requirements for pesticides are, therefore, that the active pesticides exhibit a low toxicity, a low residual and a rapid decomposition and they are converted into non-toxic materials upon entrance into one of the biological cycles of the ecological system. In particular, the compounds containing no heavy metals nor halogen atoms and composed mainly of carbon, hydrogen and oxygen are considered to be advantageous as agricultural agents.

The active groups of the pesticides presently used include organic phosphorus compounds, carbamates, chrysanthem acid esters, organochloric compounds, etc., and most of the researchers in this field now endeavour to develop new types of pesticides by considering the superiority or inferiority of each of the above active groups and screening various synthetic derivatives retaining only the superiority of the known pesticides.

The present inventors considered that each of the above well-known active groups has certain properties as basic pesticidal characteristics inherent to the specific active group and that the disadvantages associated with the specific active group would not be eliminated unless departing from the above specific properties. As a result of extensive researches for active groups having novel chemical structures, the present inventors found the new and useful acetic acid esters represented by the above formula (I).

The compounds of this invention have various advantages broadly and also exhibit strong selectivity, and the chemical structures retaining only properties advantageous as agricultural agents can be considered depending upon the specific type of the compounds of this invention.

Hithertofore, esters which are considered to be pesticidal active groups, for example, phosphoric acid esters and chrysanthem acid esters are believed to have their activities in the basic structures of phosphoric acid and cyclopropanecarboxylic acid, respectively. These esters contain a S (or O) atom bonded to a pentavalent phosphorus by a double bond as an essential structure of the phosphoric acid esters and a three-membered ring as an

essential structure of the chrysanthem acid esters, respectively, and the pesticidal activity tends to be strengthened or weakened depending upon the type of moieties present around the above essential structure as a nucleus.

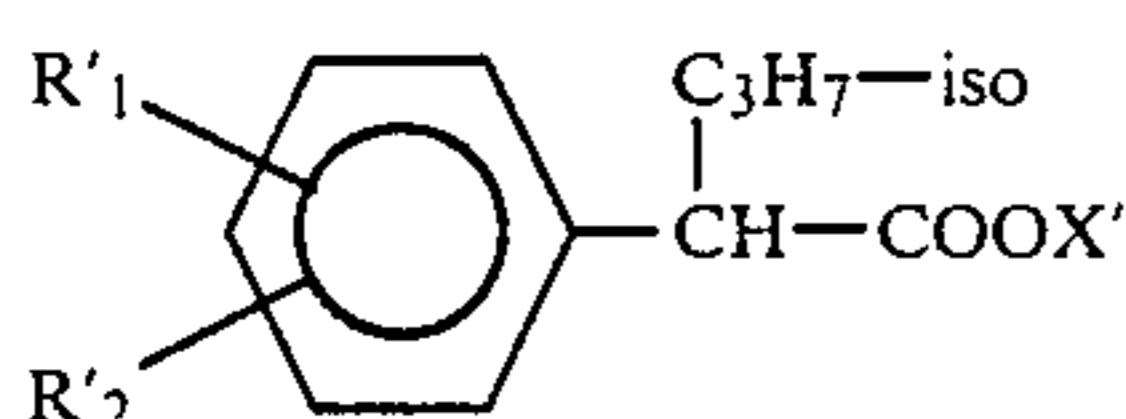
In the compounds of this invention, the acid moiety represented by the formula (XI)



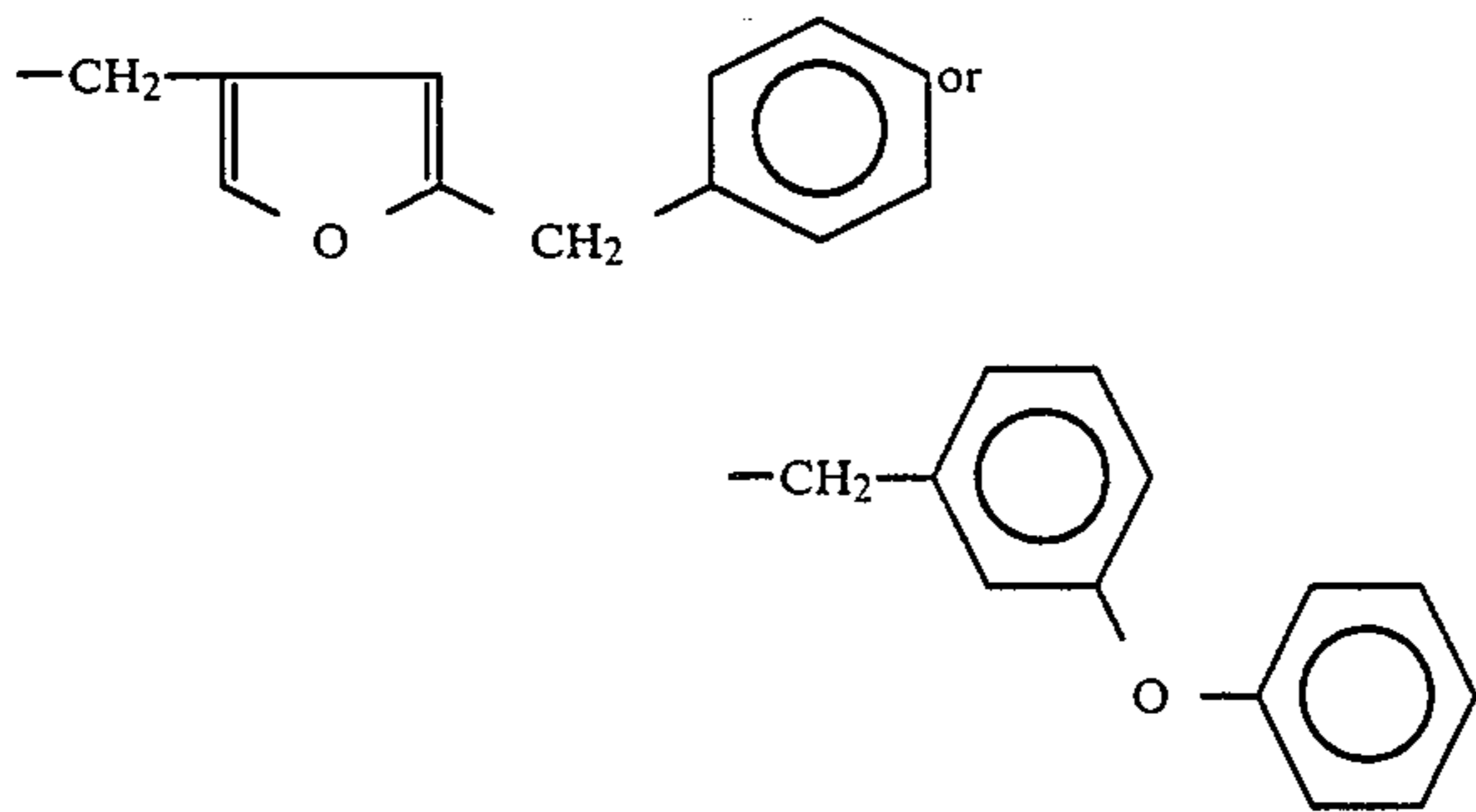
is considered to be an important active basic structure, and its essential structure is completed by a hydrogen atom at α -position of the ester group, an appropriate substituent (Z) and a substituent (Y) which are characterized in its plane structure. On the other hand, the alcohol moiety of the ester corresponds to the alcohol moiety of the well-known chrysanthem acid esters as shown by the examples of compounds described below, but, in accordance with the present invention, it was found that highly active pesticides can be prepared by esterification of the acids represented by the formula (XI) with the above alcohol moiety and that the compounds having a surprisingly high selectivity can be prepared by using an appropriate combination of an acid moiety and an alcohol moiety. However, the order of activity of the ester composed of the known alcohol and the known acid cannot be applied as it is to the evaluation of the activity of the esters in accordance with the present invention since the esters of this invention exhibit activities entirely different those of the known esters. Accordingly, the esters composed of the acid used in the present invention and the alcohol which has been found to form a highly active ester in the known pesticides do not necessarily exhibit an excellent activity.

In addition, in accordance with the present invention, various advantages such as immediate onset of action, permeability, transfer into living plants, residual activity, pesticidal activity, metamorphosis disturbance, sterilization, prohibition of egg-lying can freely be controlled by appropriately selecting a combination of an acid of the present invention and a wide variety of alcohols. Also, with respect to the spectrum of pesticidal activity, the compounds of this invention exhibit a selective or non-selective activity on such order as Coleoptera, Lepidoptera, Diptera, Orthoptera, Hemiptera, Homoptera and Acarus depending upon a specific combination of acid and an alcohol according to the present invention. Further, the compounds of this invention are expected to be useful for controlling noxious insects such as a nematoda. Other important features of the compounds of this invention are that they are active against noxious insects which are resistant to the pesticides presently used because of their basic structure different from those of the known pesticides to which the insects are resistant, and that they are totally low toxic to mammals including man as indicated by Experiments shown below.

As the results of the present invention, the substituted acetate compounds of the formula,



wherein R'₁ represents straight or branched chain alkyl having 1 to 4 carbon atoms, halogen or alkoxy having 1 to 4 carbon atoms; R'₂ represents hydrogen or a group which completes, together with R'₁, methylenedioxy; and X' represents a group of the formula,



are more excellent in insecticidal effect, a killing effect and a knock down effect than the other substituted acetate compounds of the formula (I).

In order to make the above fact clearer, the superiority of the compounds represented by the formula (I) in biological effects is shown below with reference to Experimental Examples. But the other compounds of the formula (I) and their geometrical isomers and optical isomers display more excellent effects as well than the known insecticidal agents.

In preparing the substituted acetic acid esters represented by the formula (I), an acid represented by the formula



wherein Y and Z are as defined above, or a reactive derivative thereof is reacted with an alcohol represented by the formula



wherein X is as defined above, or a halide or sulfoxylate thereof.

The term "reactive derivative" of the acid used herein refers to an acid halide, an acid anhydride, an ester with an alcohol having a low boiling point, and an alkali metal salt, a silver salt or an organic tertiary base salt of the acid.

The substituted acetic acid esters represented by the formula (I) include optical isomers due to the carbon atom at α -position, and it is to be understood that the present invention is contemplated to include optical isomers.

In carrying out the production of the substituted acetic acid esters of the formula (I), the esterification between the acid (XI) and the alcohol (XII) can advantageously be accomplished by reacting the reactants in an appropriate inert solvent at room temperature or an elevated temperature under an appropriate dehydrating condition, e.g., in the presence of cyclohexylcarbodi-

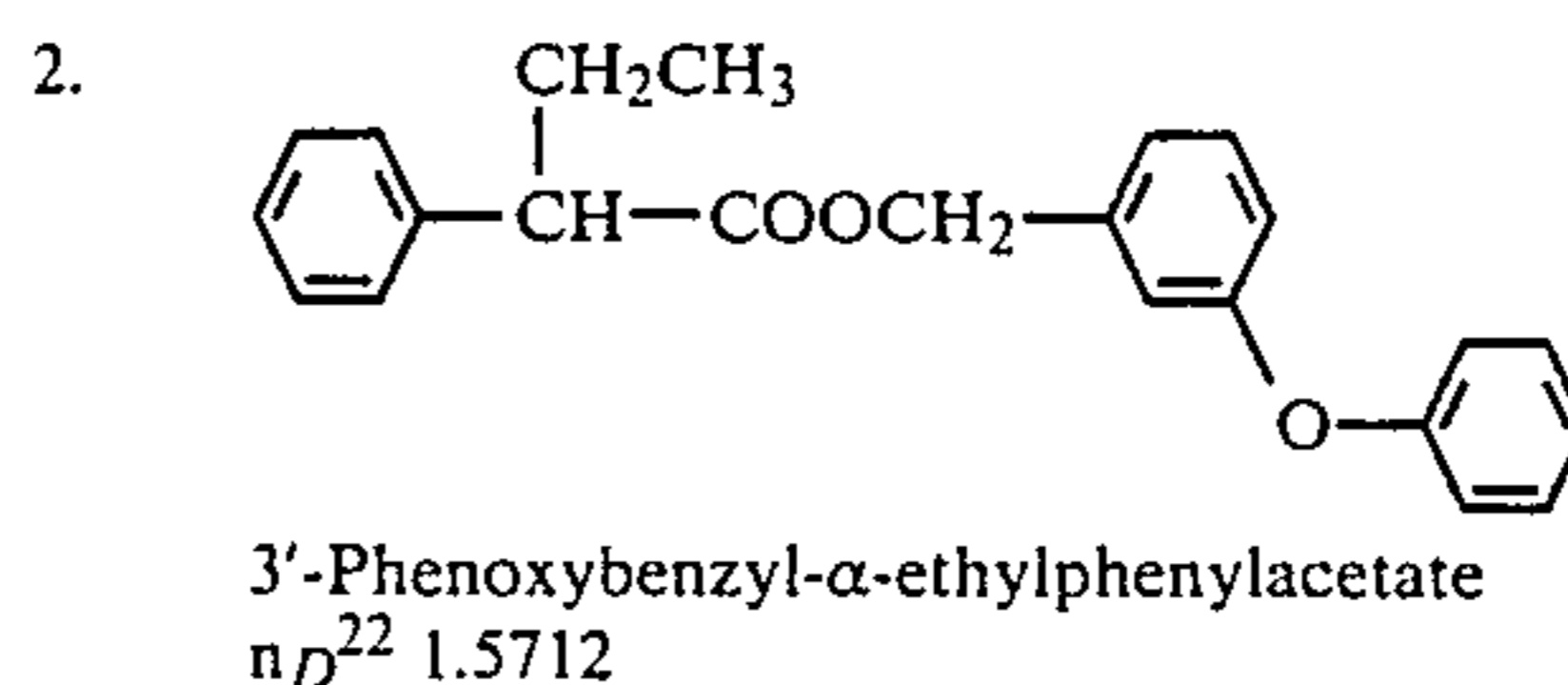
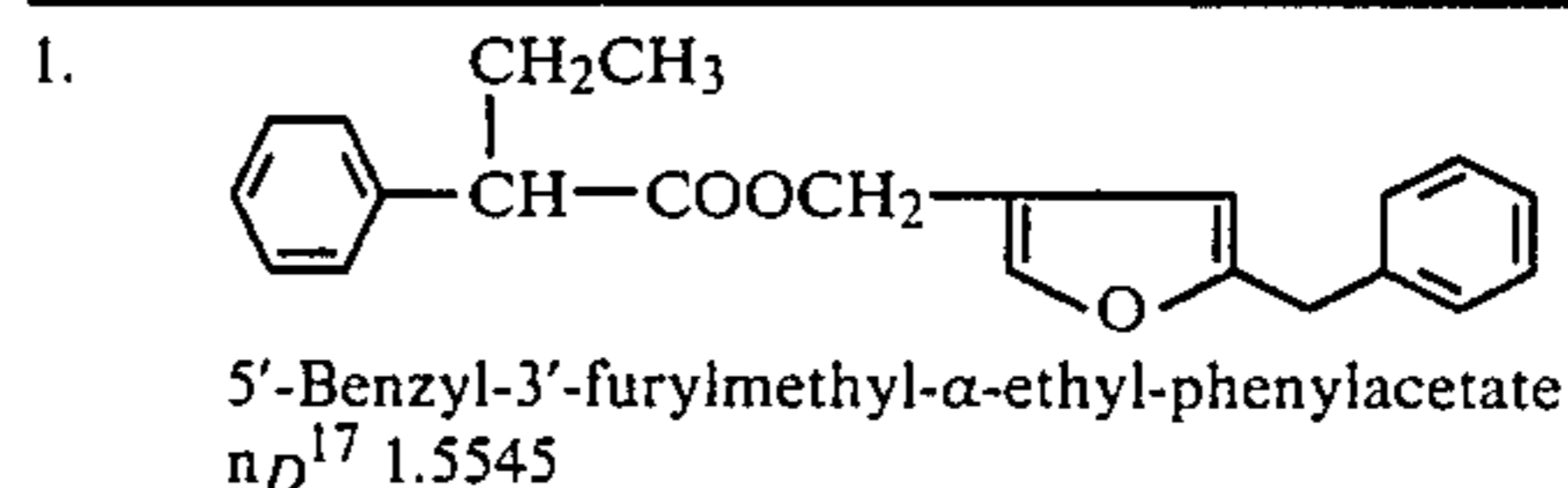
imide. When an acid halide is used as a reactive derivative of the acid (XI), the desired ester can be obtained in high yield by reacting the alcohol (XII) and the acid halide at room temperature using an acid acceptor, for example, an organic tertiary amine such as pyridine, triethylamine and the like. The acid halides which can be used in the process of this invention may be any type of halides within the scope of this invention, but an acid chloride is generally preferred. The presence of an inert solvent is not essential in the esterification reaction, but it is generally preferred to use a solvent in order to assure a smooth reaction. Any solvent which is inert to the reactants and the ester product may be used and a preferred inert solvent includes benzene, toluene and petroleum benzene. When an ester of the acid (XI) and a low boiling point alcohol, e.g., methanol, ethanol and the like is used as a reactive derivative of the acid, the desired ester (I) can be obtained in high yield by heat-reacting the ester and the alcohol (XII) with heating in the presence of an appropriate organic base catalyst, preferably, an alkali metal alkoxide corresponding to the low boiling alcohol of the ester used or sodium hydride in an inert solvent such as toluene while removing a low boiling alcohol liberated during the reaction from the reaction system by a fractional distillation column.

In an alternative esterification procedure using an acid anhydride as a reactive derivative of the acid (XI), the desired ester can be obtained by reacting the acid anhydride with the alcohol (XII) at room temperature without using specific reaction aids. In this case, it is preferred to heat the reaction system and to use a solvent such as toluene, xylene and the like in order to ensure a smooth reaction, but the heating and the use of solvent are not critical in this procedure.

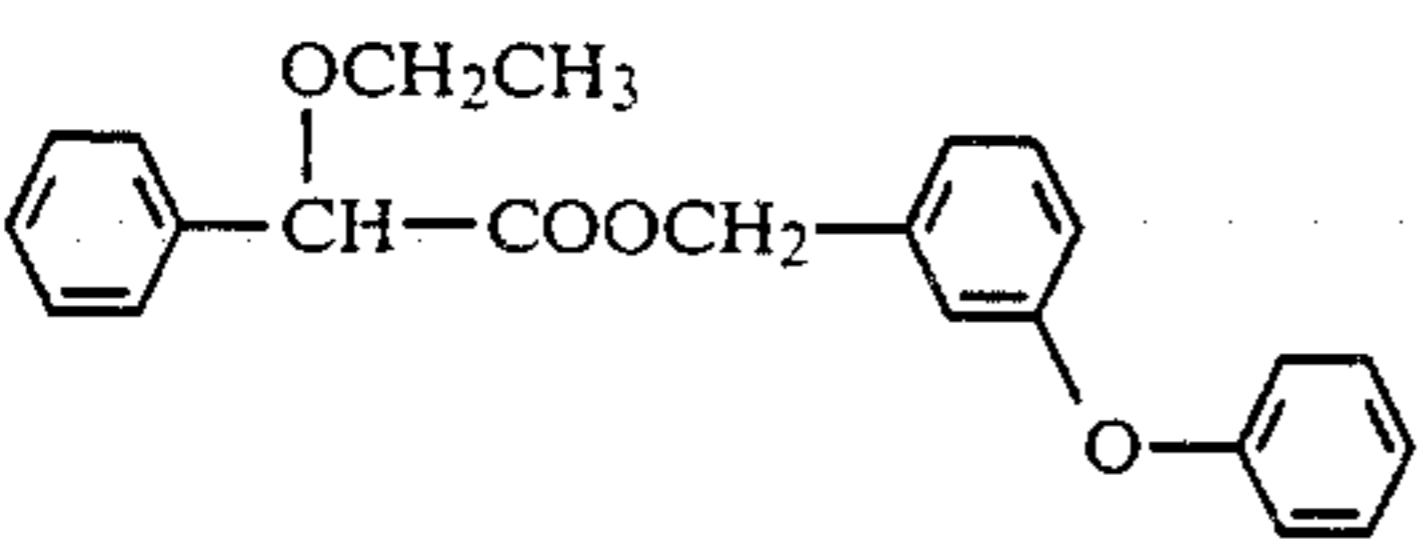
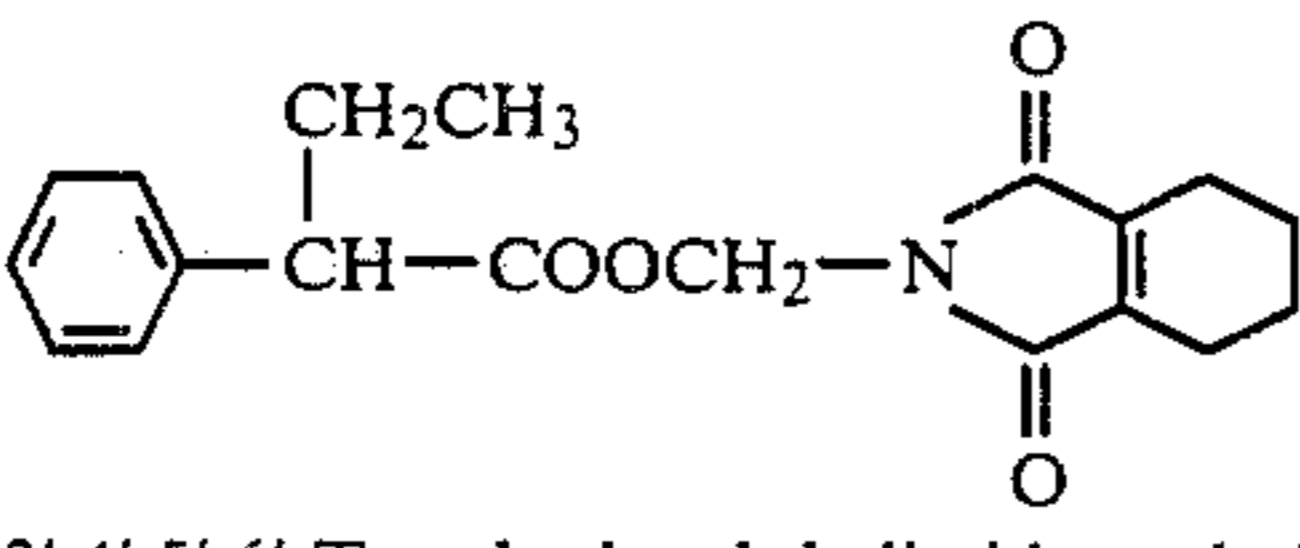
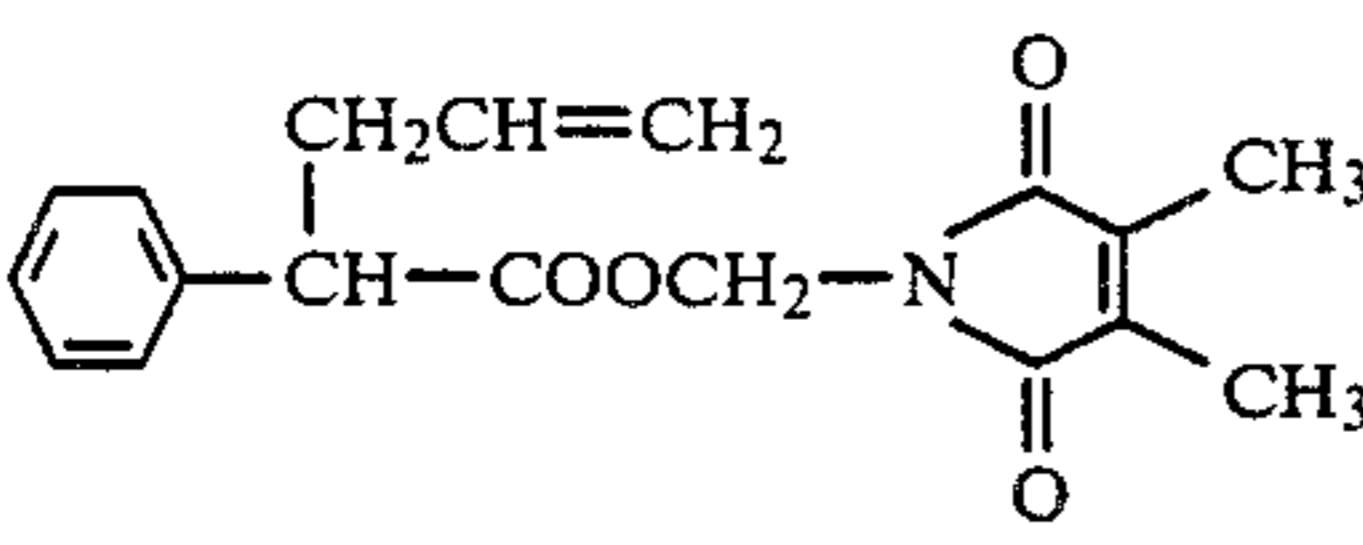
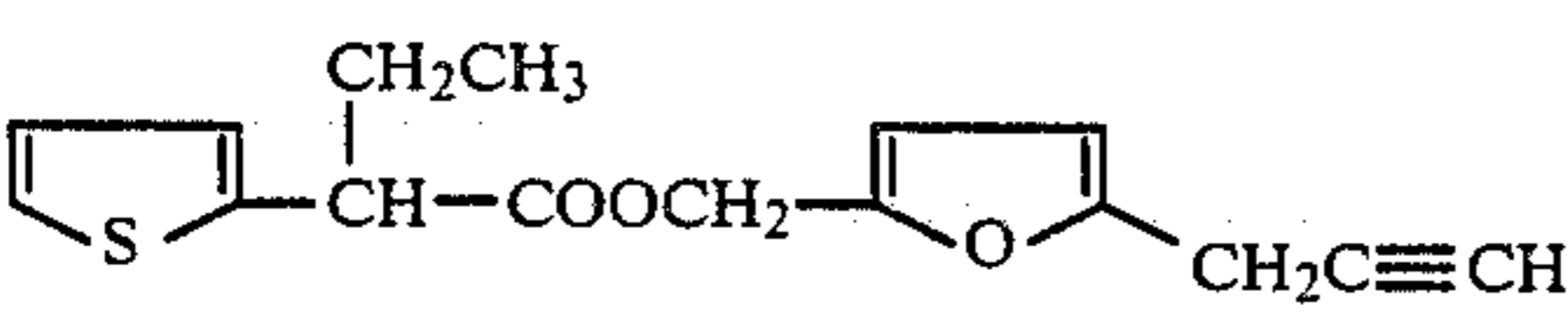
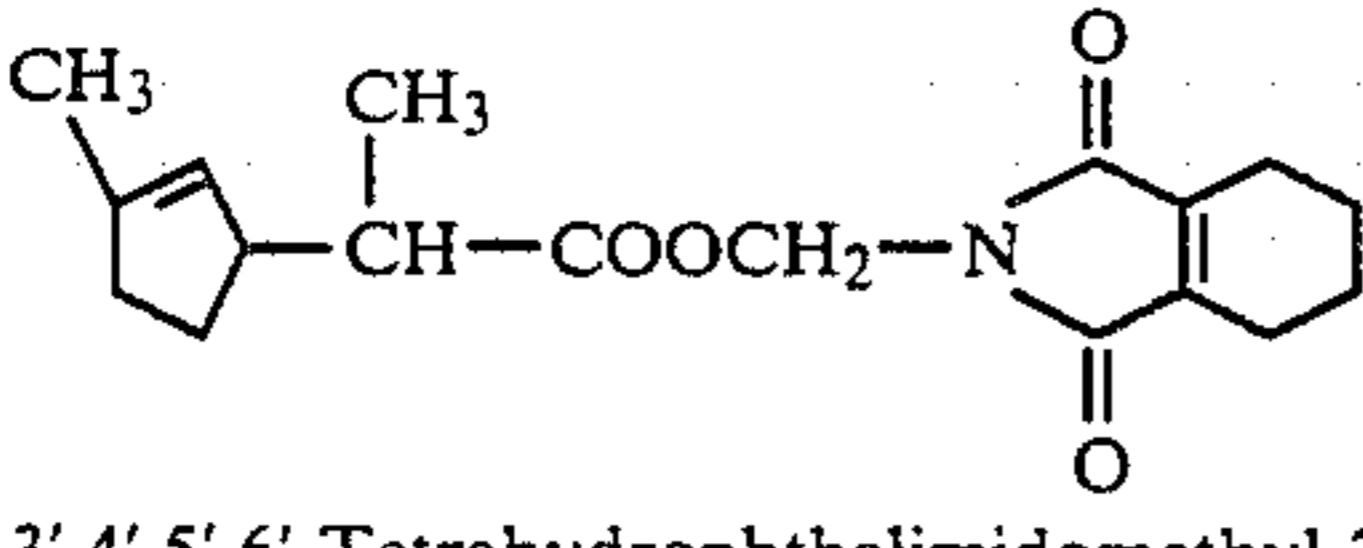
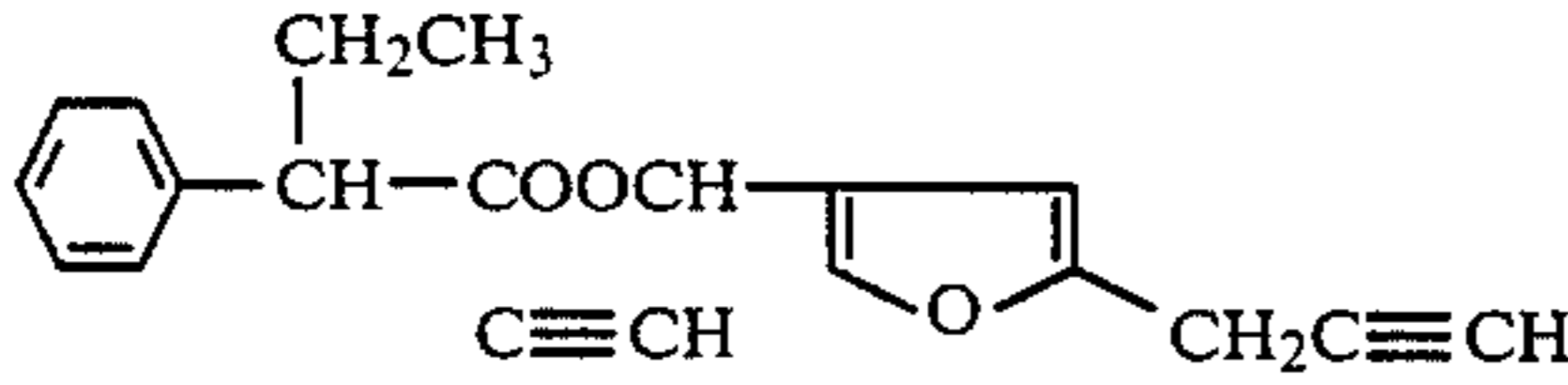
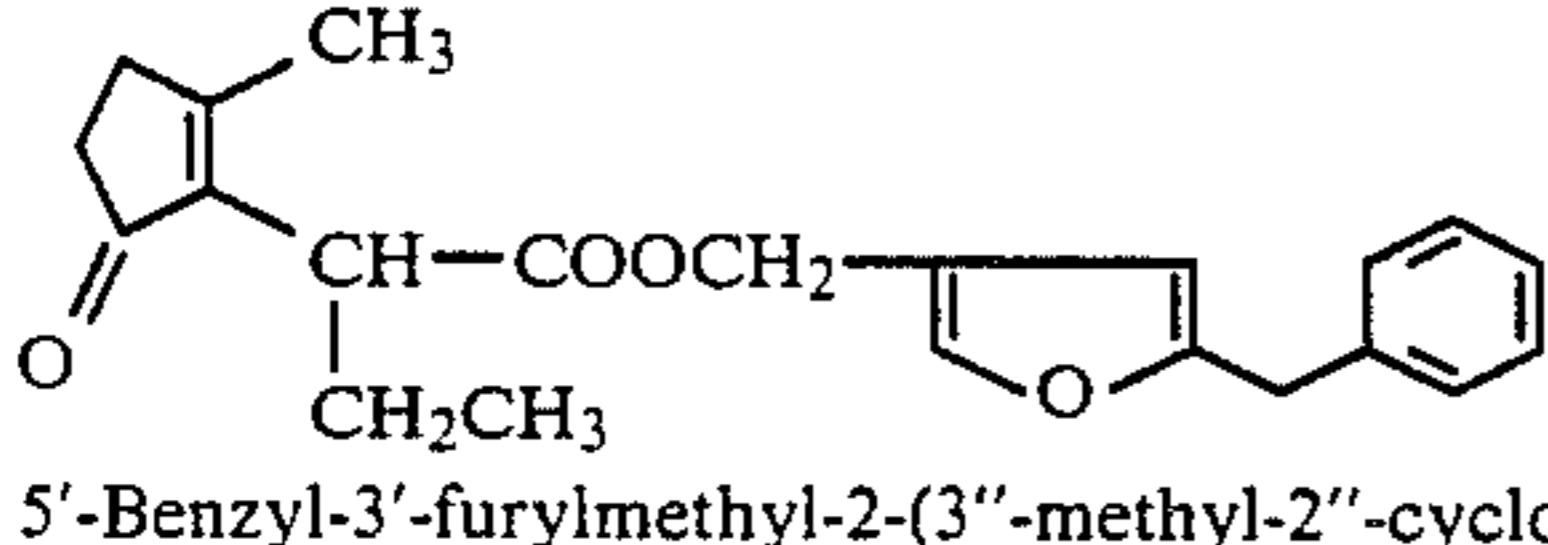
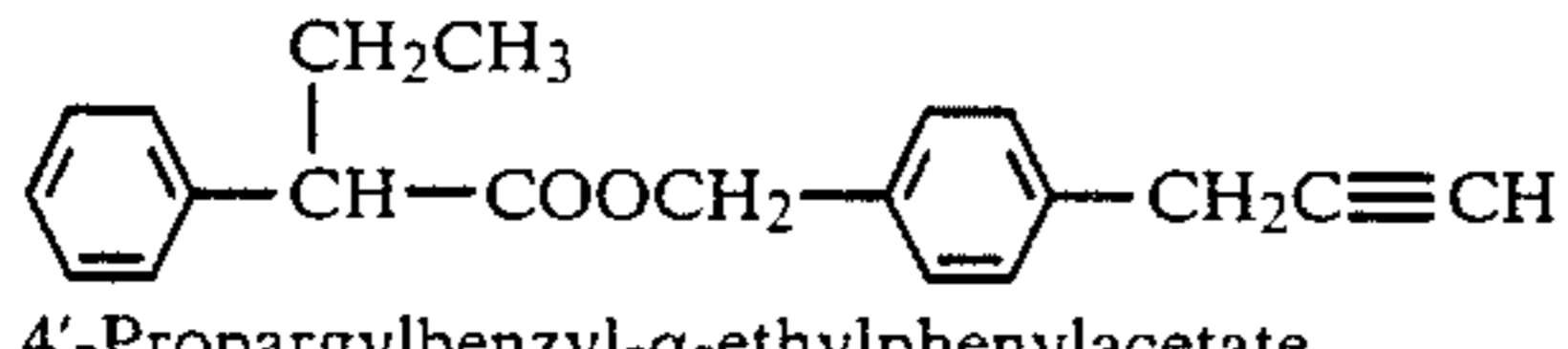
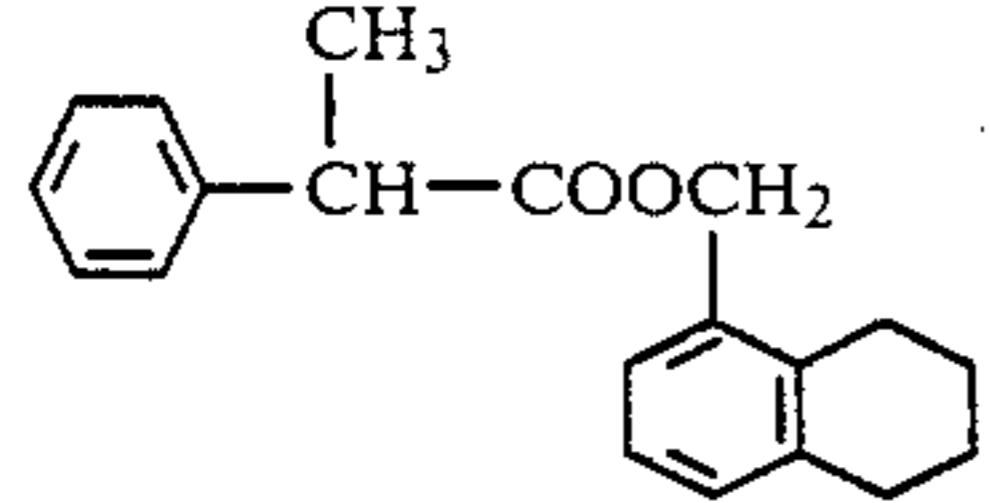
When a halide or sulfoxylate of the above alcohol (XII) is used, the acid (XI) is generally employed in the form of an alkali metal salt, a silver salt or an organic tertiary base salt. Those salts may be formed in situ by adding simultaneously the acid (XII) and the corresponding base to the reaction system. In this case, a solvent such as benzene, acetone, dimethylformamide and the like is preferably used, and the reaction is preferably conducted by heating the reaction system at or below the boiling point of the solvent used. A preferred halogen for the halide of alcohol (XII) includes chlorine and bromine.

Representative examples of the compounds of this invention are illustrated below, but the present invention is not limited thereto.

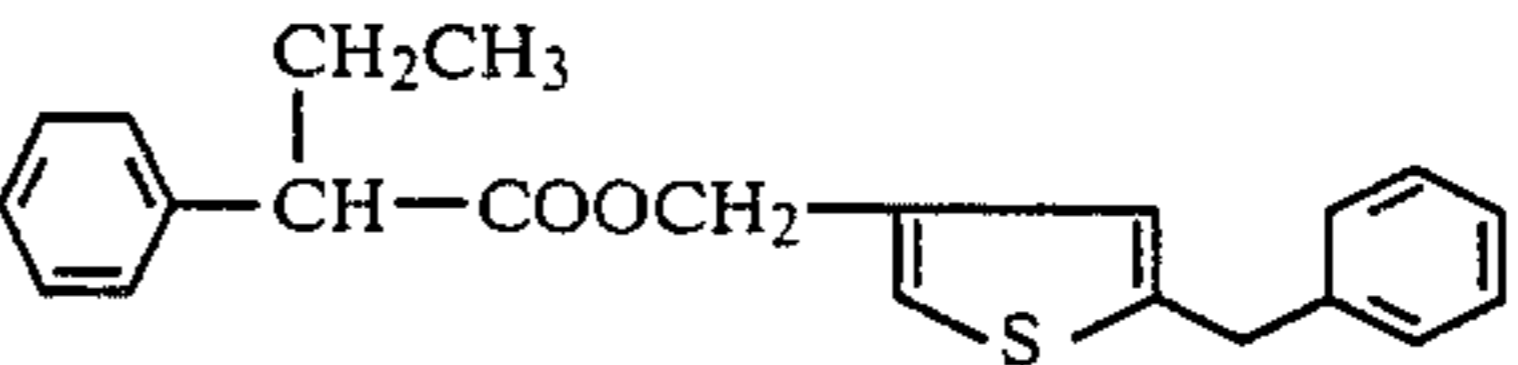

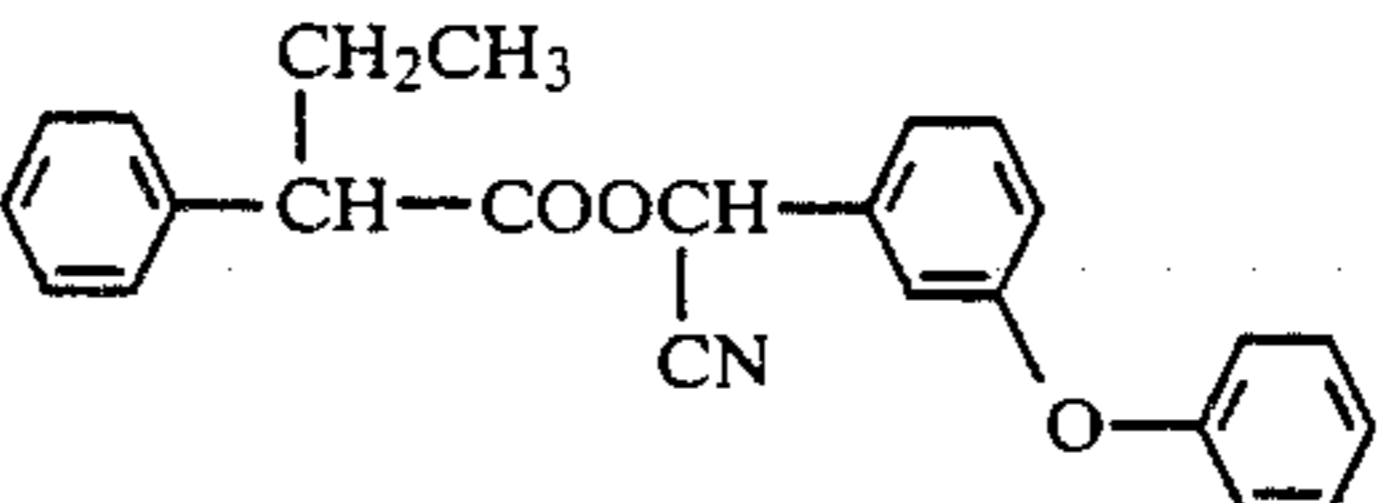
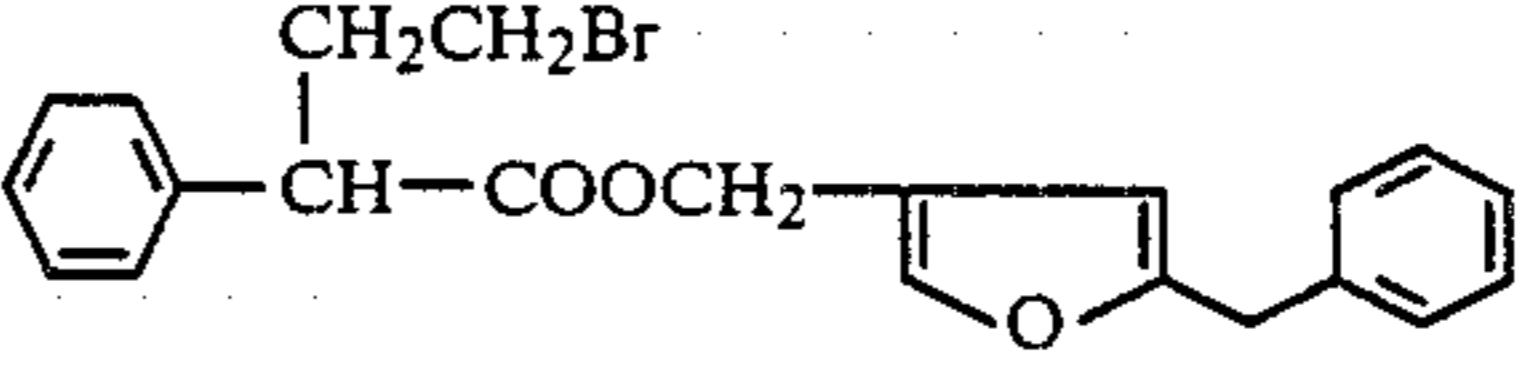
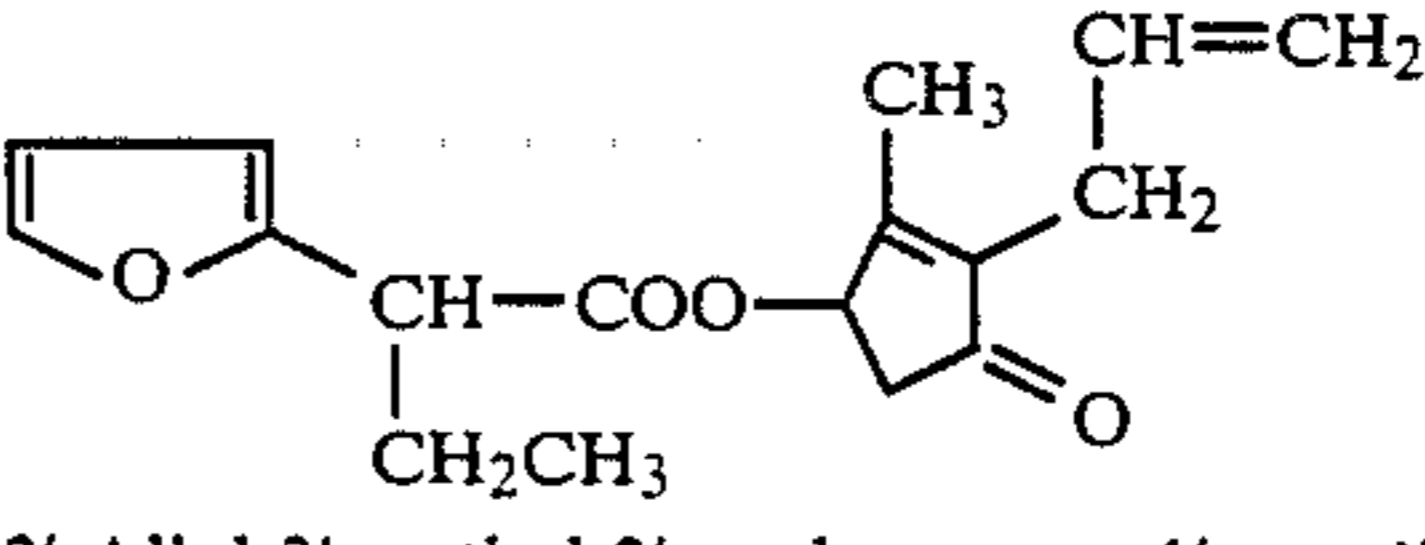
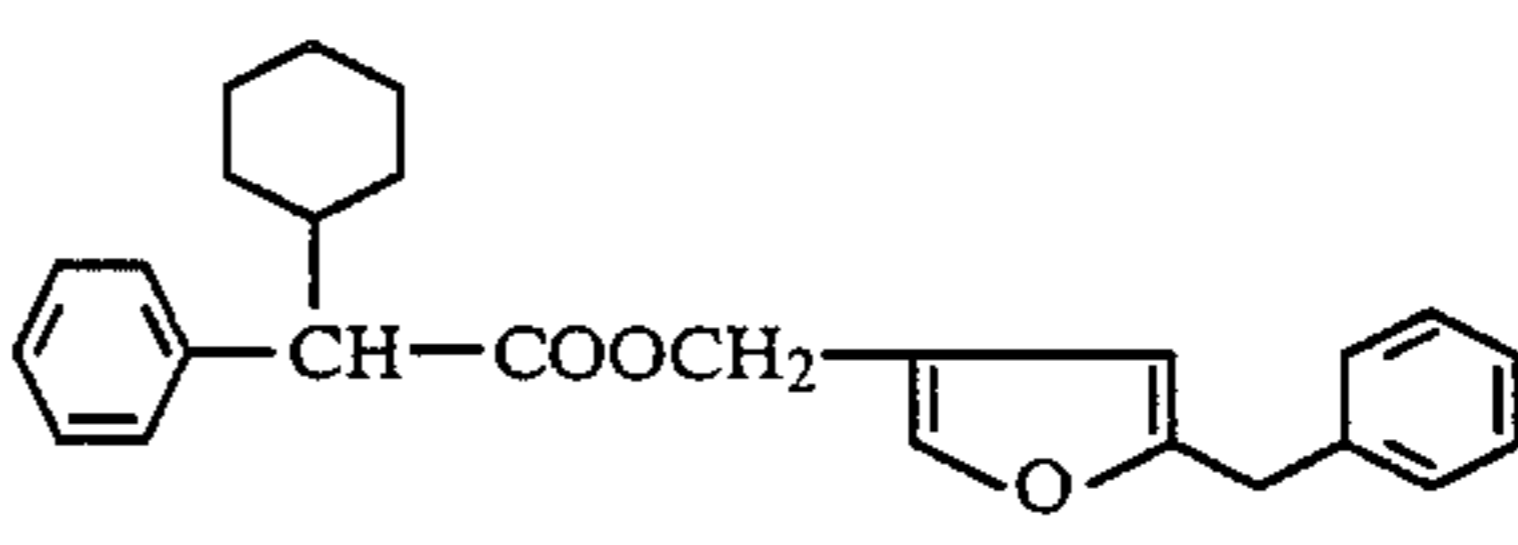
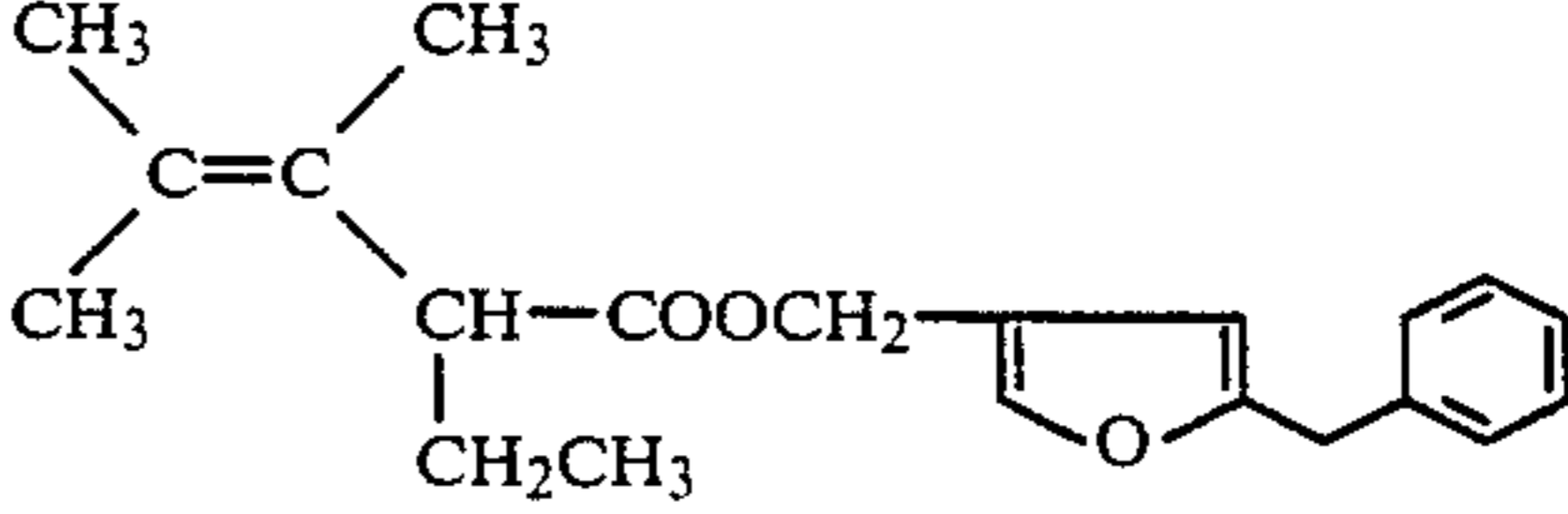
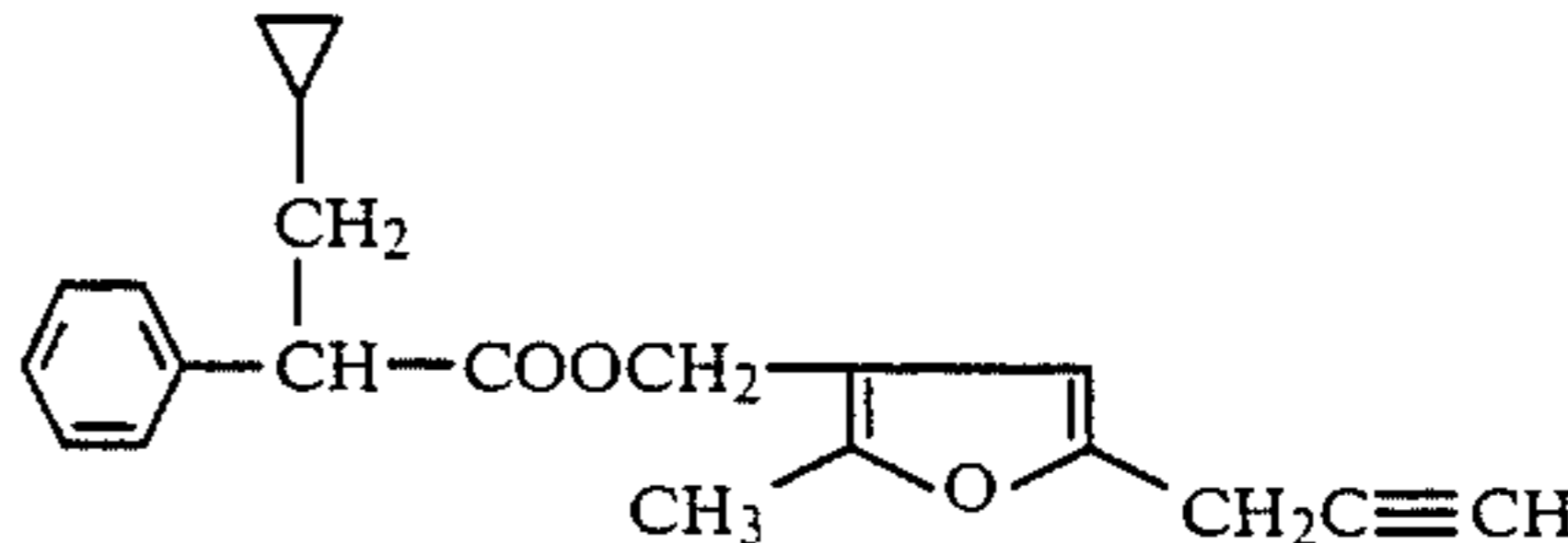
Compound
No.



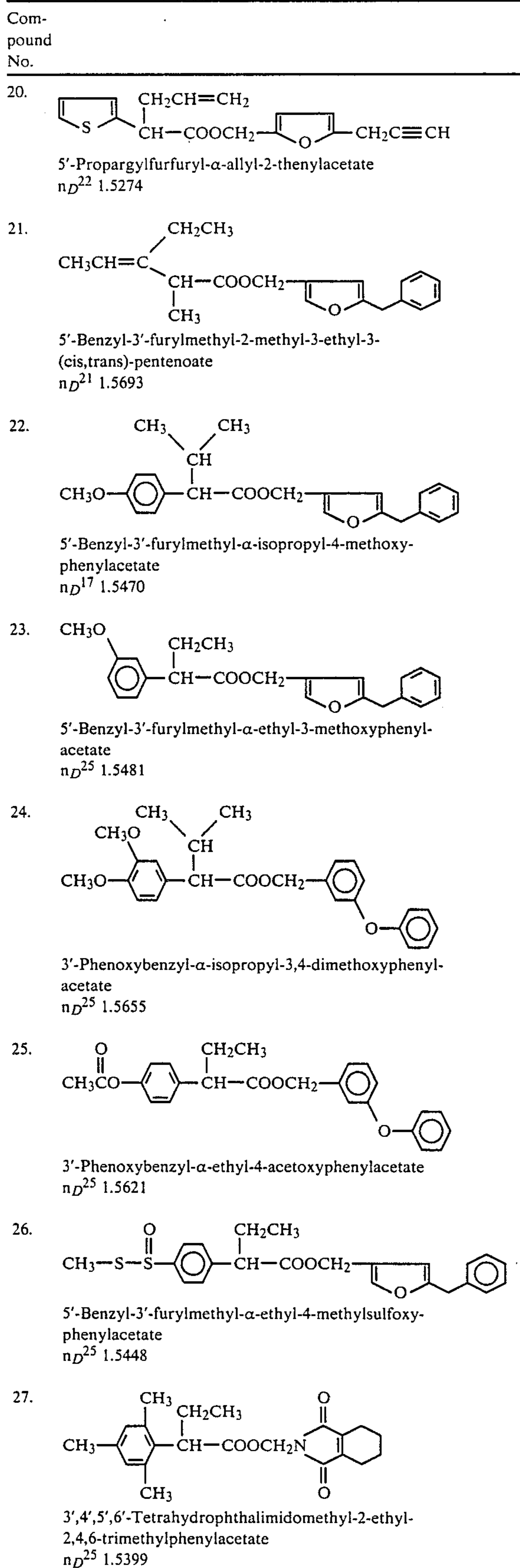
-continued

Compound No.	
3.	 <p>3'-Phenoxybenzyl-α-ethoxyphenylacetate n_D^{24} 1.5275</p>
4.	 <p>3',4',5',6'-Tetrahydrophthalimidomethyl-α-ethylphenylacetate n_D^{22} 1.5403</p>
5.	 <p>Dimethylmaleimidomethyl-α-allylphenylacetate n_D^{22} 1.5349</p>
6.	 <p>5'-Propargylfurfuryl-α-ethyl-2-thienylacetate n_D^{22} 1.5835</p>
7.	 <p>3',4',5',6'-Tetrahydrophthalimidomethyl-2-(3''-methyl-2''-cyclopentene-1''-yl)-propionate n_D^{23} 1.5648</p>
8.	 <p>5'-Propargyl-α'-ethynyl-3'-furylmethyl-α-ethylphenylacetate $n_D^{24.5}$ 1.5310</p>
9.	 <p>5'-Benzyl-3'-furylmethyl-2-(3''-methyl-2''-cyclopentenone-2''-yl)-butyrate n_D^{22} 1.5123</p>
10.	 <p>4'-Propargylbenzyl-α-ethylphenylacetate n_D^{23} 1.5039</p>
11.	 <p>5'-(1',2',3',4'-Tetrahydronaphthalenyl)-methyl-α-methylphenylacetate</p>

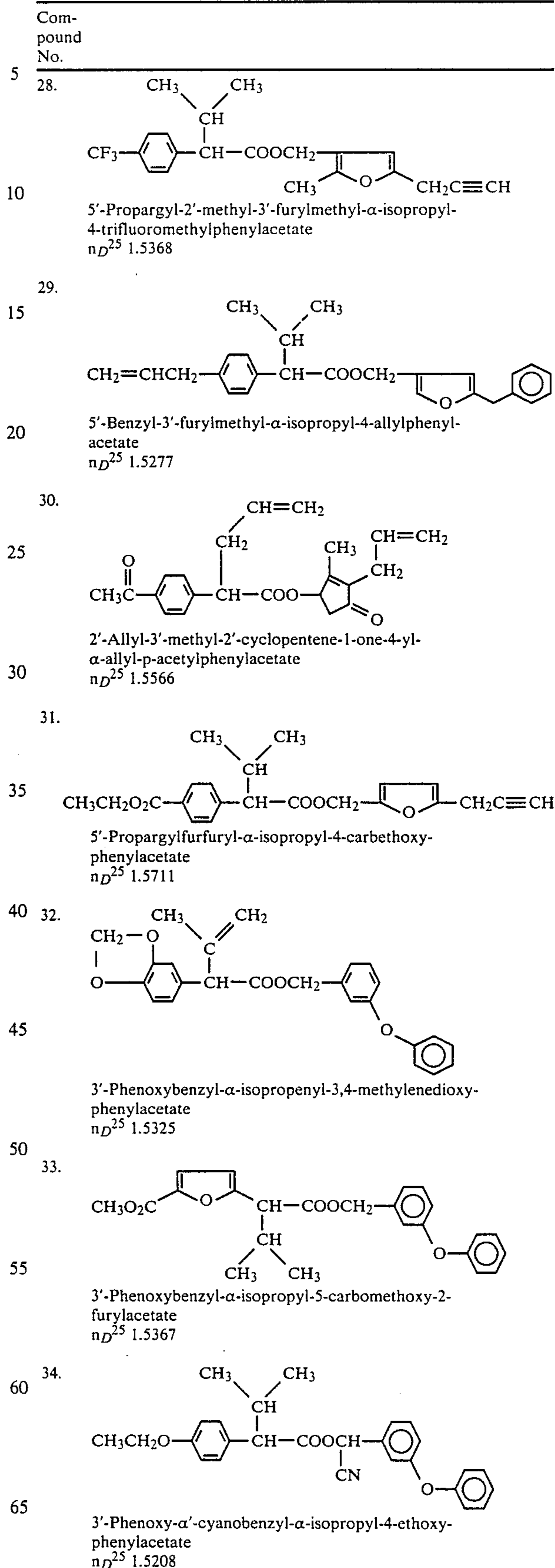
-continued

Compound No.	
5	n_D^{24} 1.5536
12.	 <p>5'-Benzyl-3'-thienyl-α-ethylphenylacetate n_D^{22} 1.5488</p>
13.	 <p>5'-Benzyl-3'-thienyl-α-isopropylphenylacetate n_D^{22} 1.5613</p>
14.	 <p>3'-Phenoxy-α'-cyanobenzyl-α-ethylphenylacetate n_D^{21} 1.5638</p>
15.	 <p>5'-Benzyl-3'-furylmethyl-α-(2''-bromoethyl)phenylacetate n_D^{23} 1.5911</p>
16.	 <p>2'-Allyl-3'-methyl-2'-cyclopentene-1'-one-4''-yl-α-ethyl-2-furylacetate n_D^{22} 1.5118</p>
17.	 <p>5-Benzyl-3'-furylmethyl-α-cyclohexylphenylacetate m.p. 51-53° C.</p>
18.	 <p>5'-Benzyl-3'-furylmethyl-2-ethyl-3,4-dimethyl-3-pentenoate n_D^{23} 1.5417</p>
19.	 <p>5'-Propargyl-2'-methyl-3'-furylmethyl-α-cyclopropylmethylphenylacetate n_D^{24} 1.5632</p>

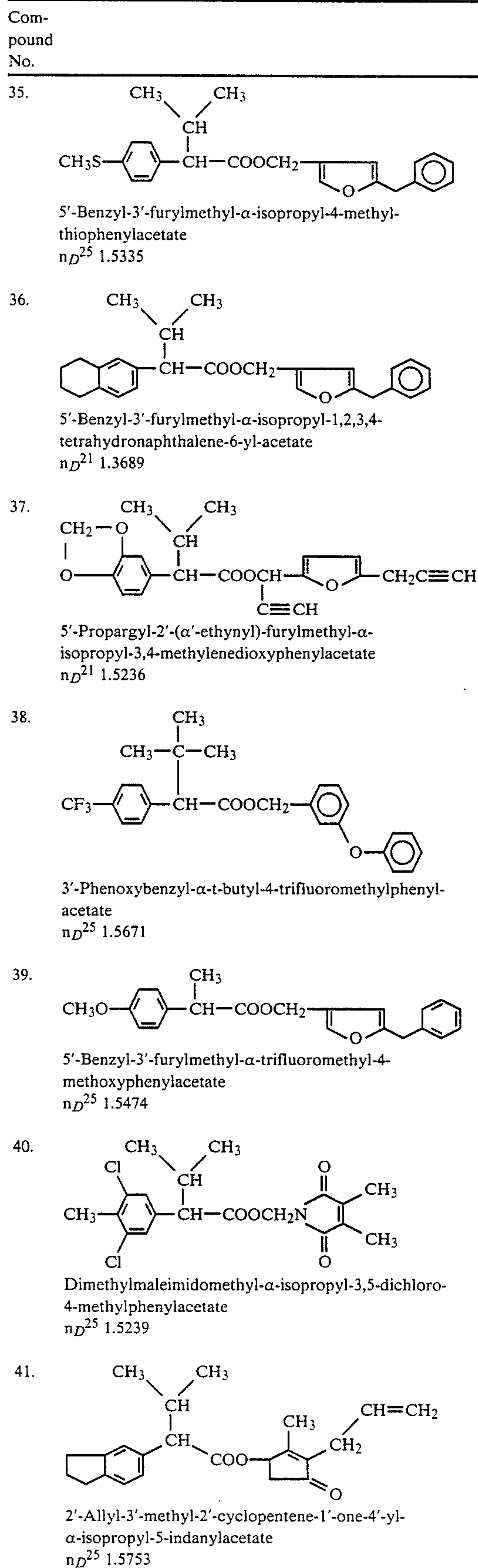
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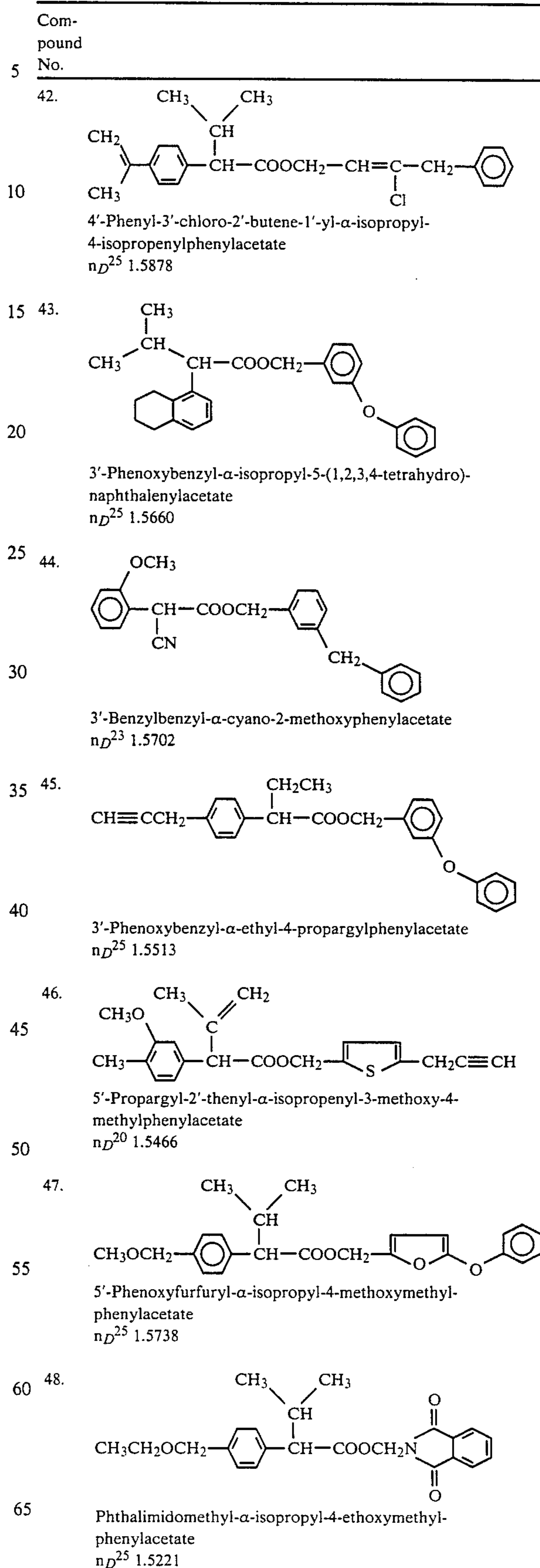
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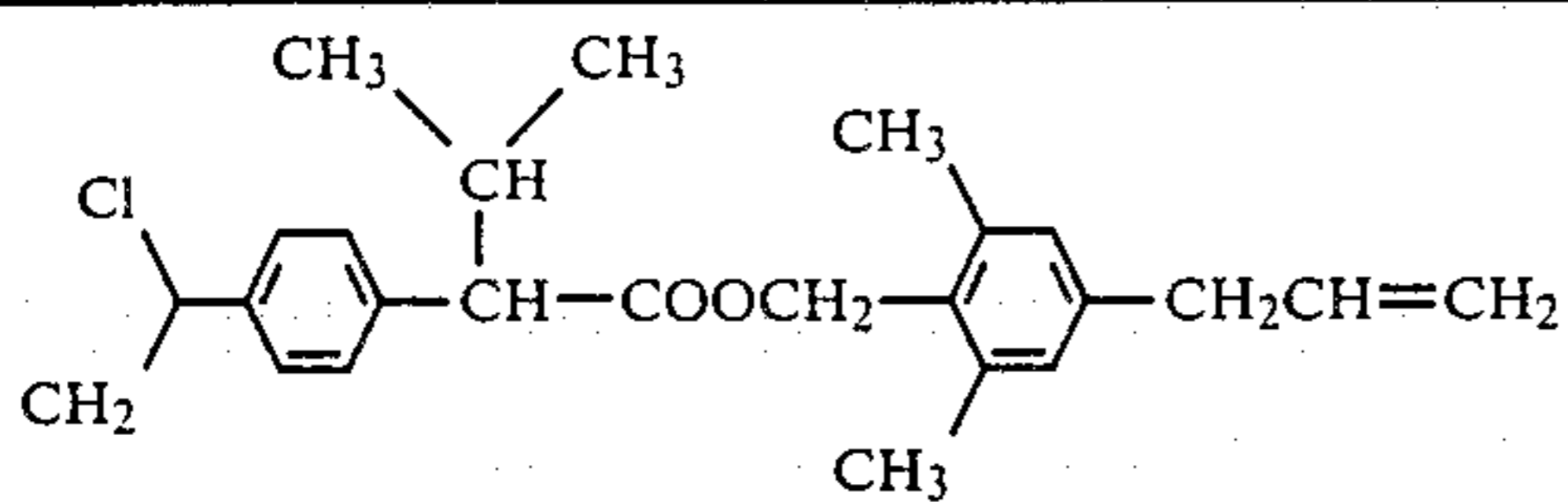
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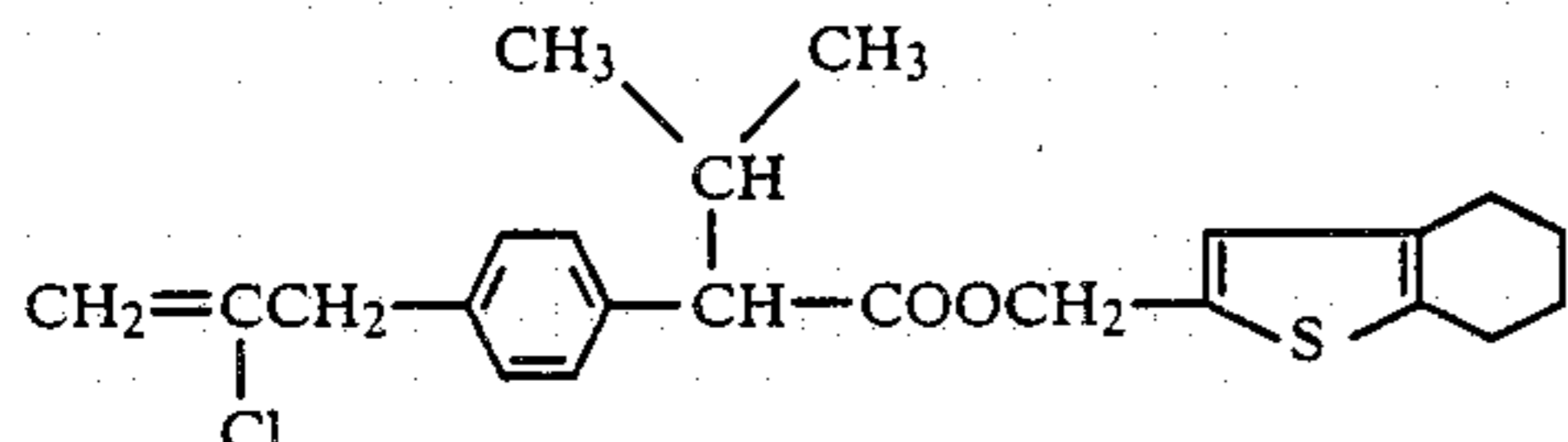
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pound
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49.



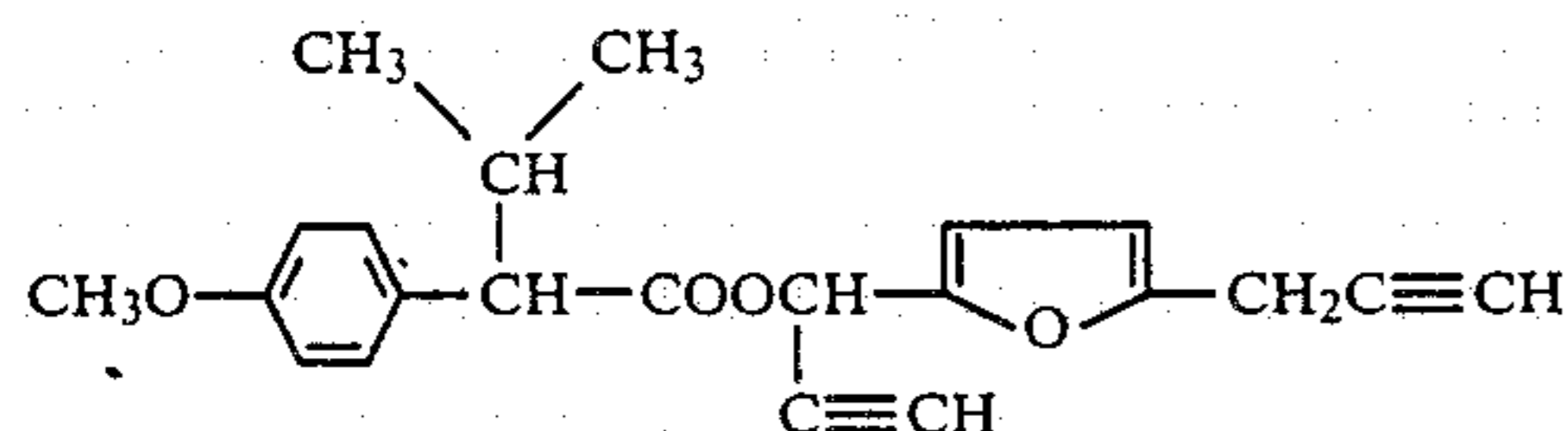
2',6'-Dimethyl-4'-allylbenzyl- α -isopropyl-4-1''-chloroethylene-1''-yl-phenylacetate
 n_D^{25} 1.5413

50.



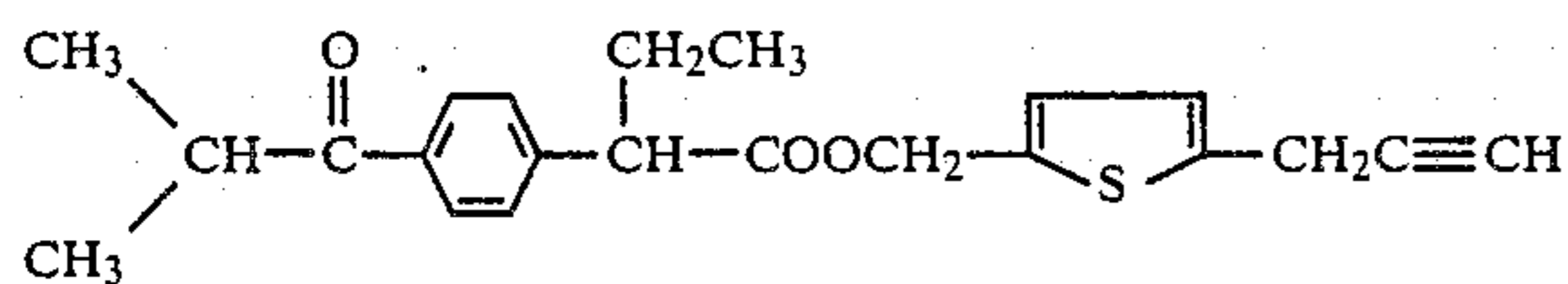
4',5'-Tetramethylene-2-thenyl- α -isopropyl-4-(2''-chloroallyl)-phenylacetate
 n_D^{25} 1.5239

51.



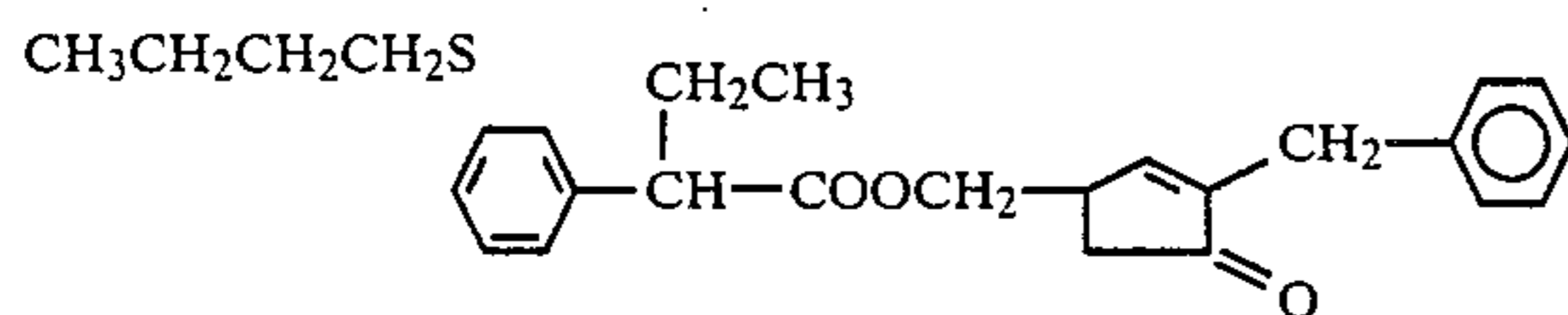
5'-Propargyl-2'-(α' -ethynyl)-furfuryl- α -isopropyl-4-methoxyphenylacetate
 n_D^{26} 1.5253

52.



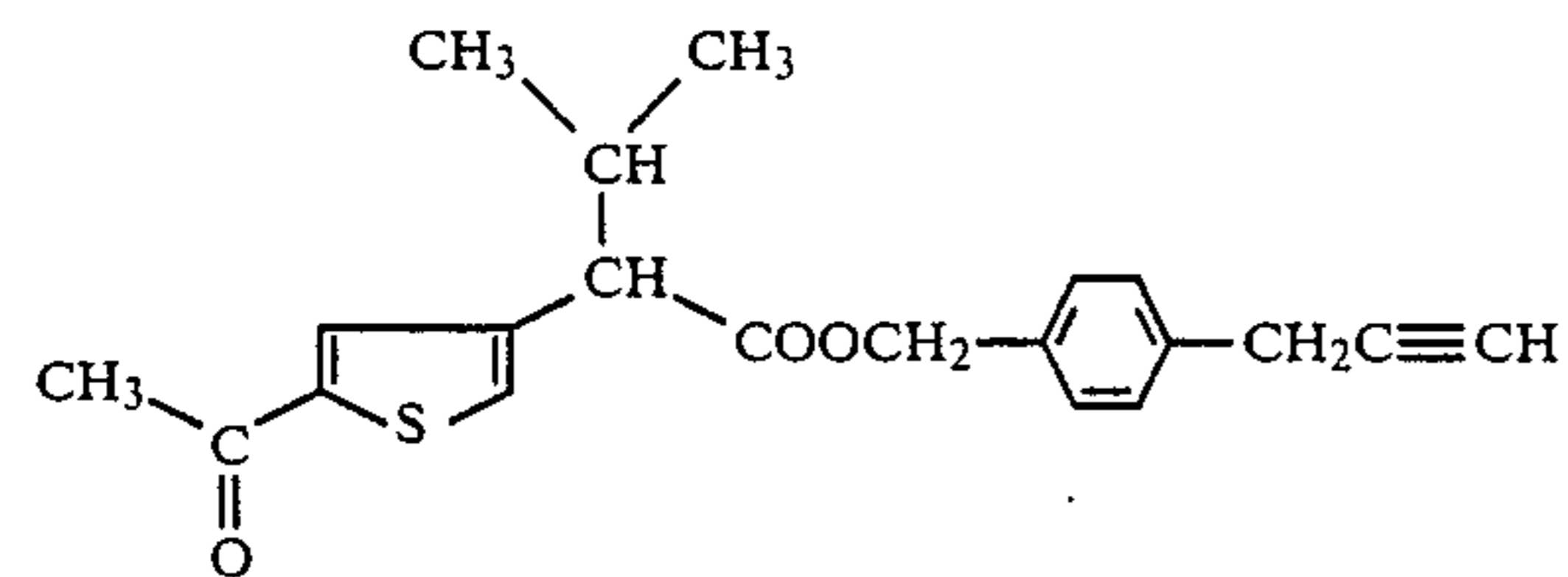
5'-Propargyl-2'-thenyl- α -ethyl-4-isobutyrylphenylacetate
 n_D^{25} 1.5549

53.



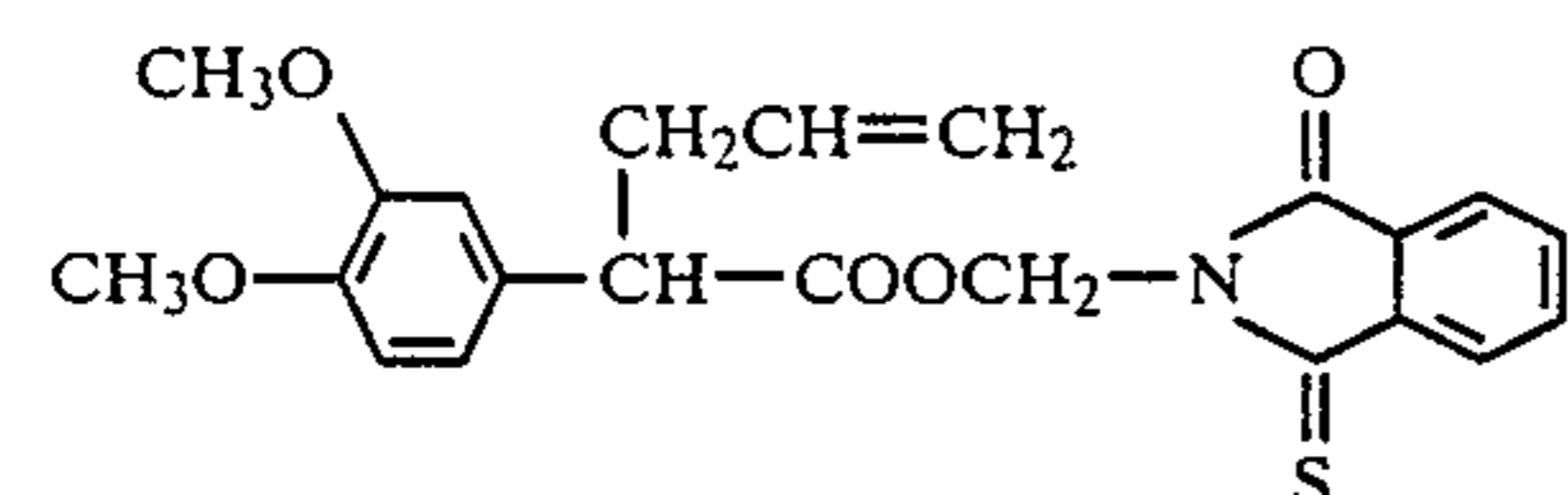
2'-Benzyl-2'-cyclopentene-1'-one-4'-yl- α -ethyl-3-butylthio-phenylacetate
 n_D^{25} 1.5508

54.



4'-Propargylbenzyl- α -isopropyl-5-acetyl-3-thienylacetate
 n_D^{22} 1.5259

55.



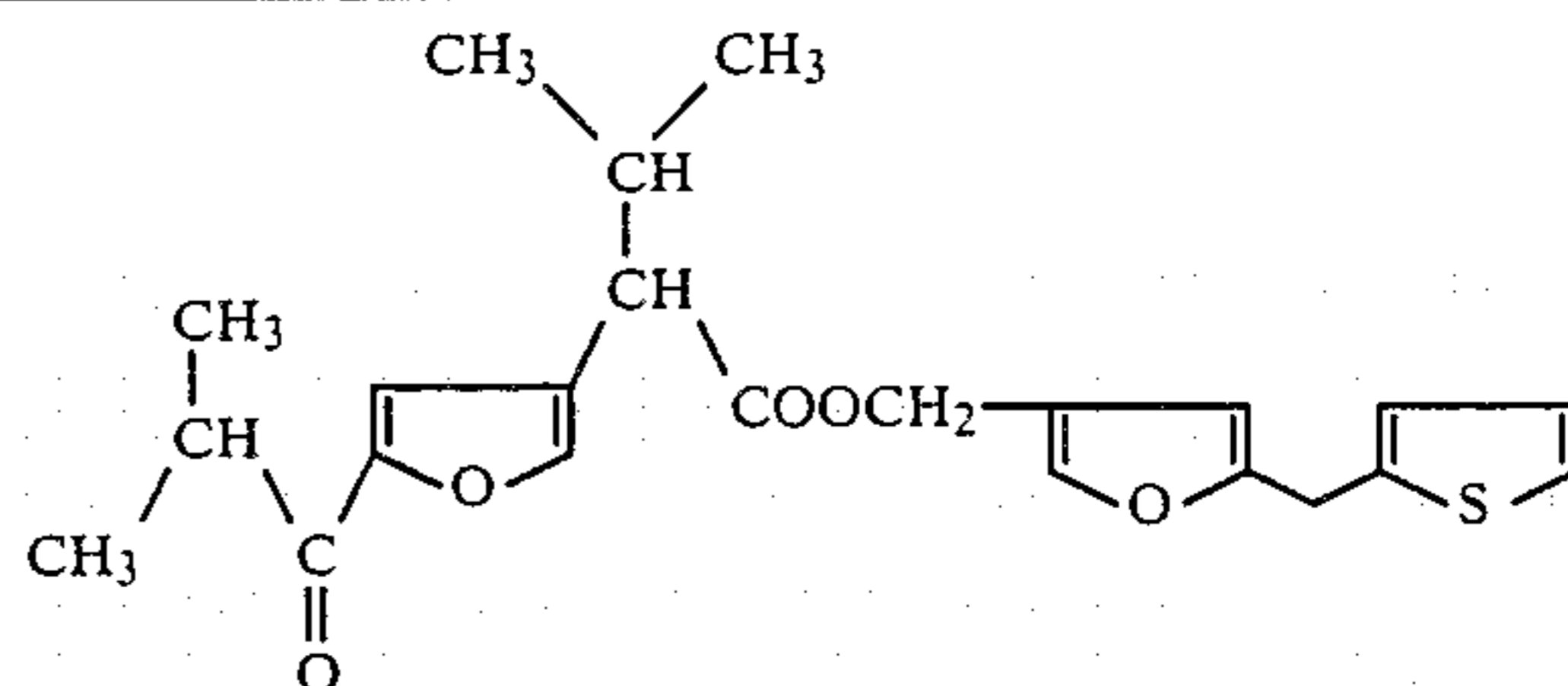
Monothiophthalimidomethyl- α -allyl-3,4-dimethoxyphenylacetate
 n_D^{19} 1.5350

-continued

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pound
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5

56.

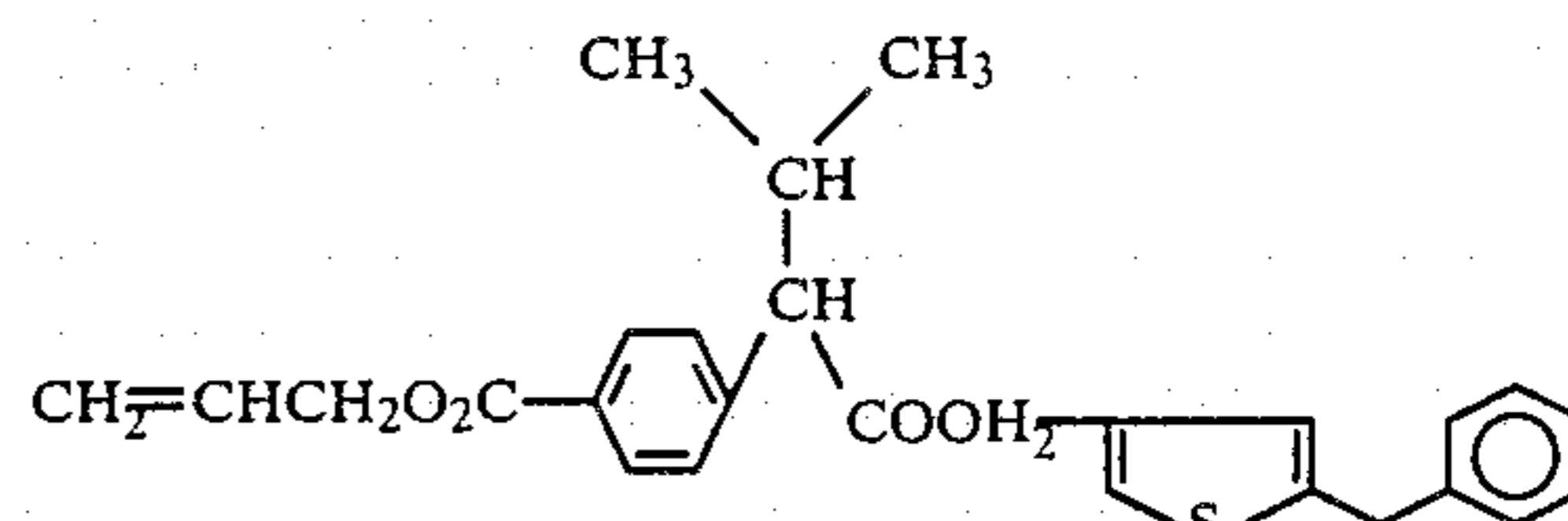


5'-(2''-Thenyl)-3'-furylmethyl- α -isopropyl-3-(5-isobutyryl)-furylacetate
 n_D^{22} 1.5324

10

15

57.

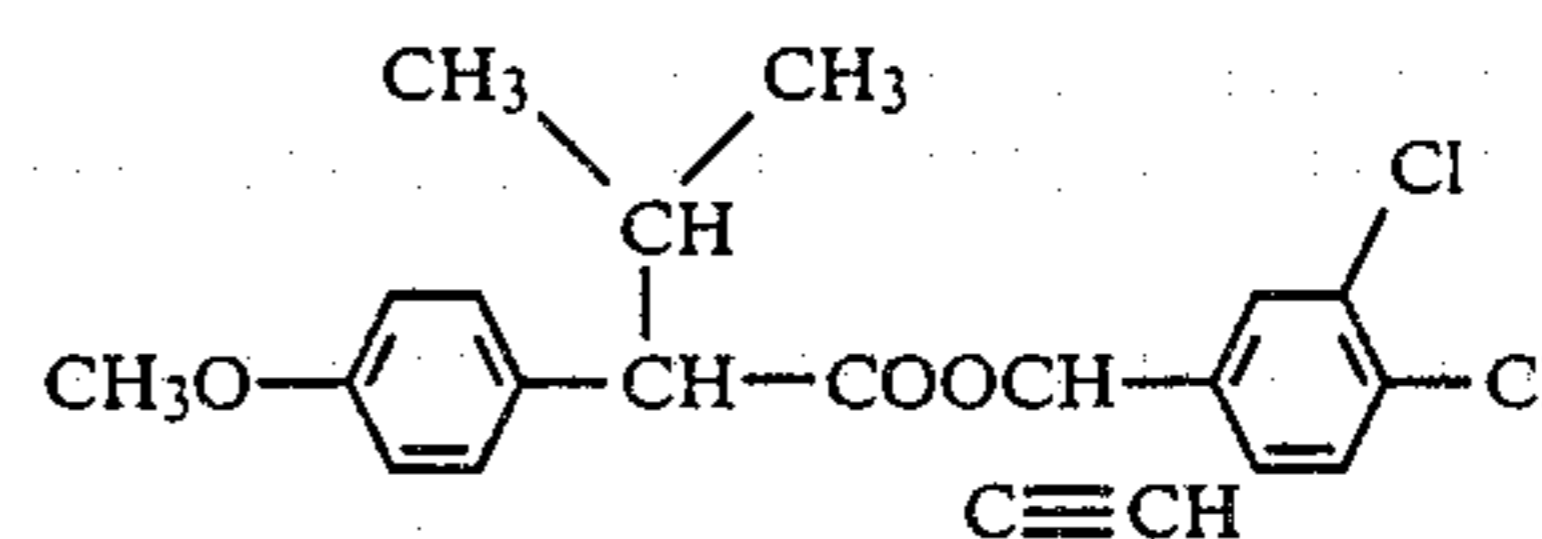


5'-Benzyl-3'-thenyl- α -isopropyl-3-carballyloxyphenylacetate
 n_D^{25} 1.5556

20

25

58.

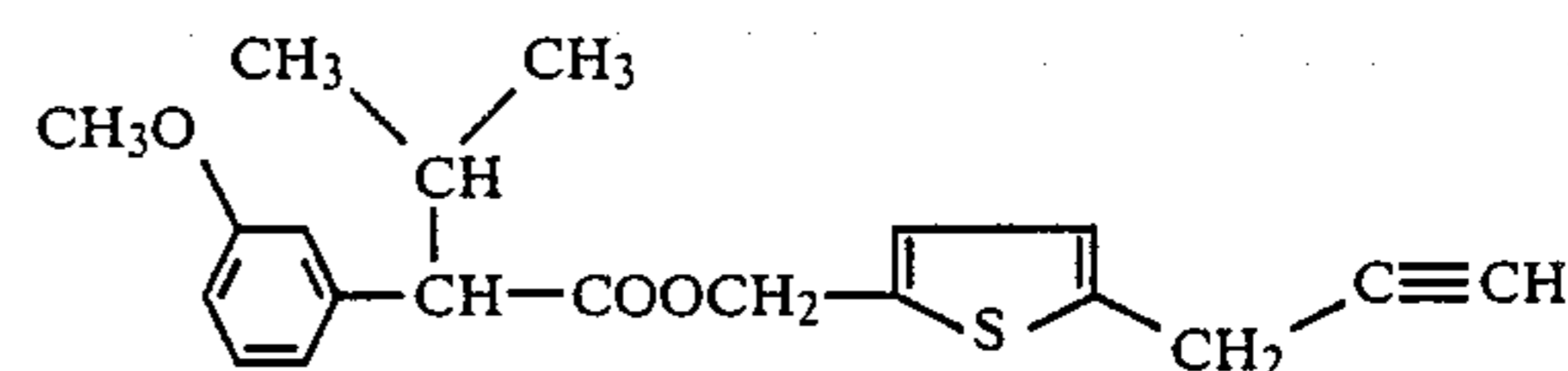


3',4'-Dichloro- α' -ethynylbenzyl- α -isopropyl-4-methoxyphenylacetate
 n_D^{25} 1.5368

30

35

59.

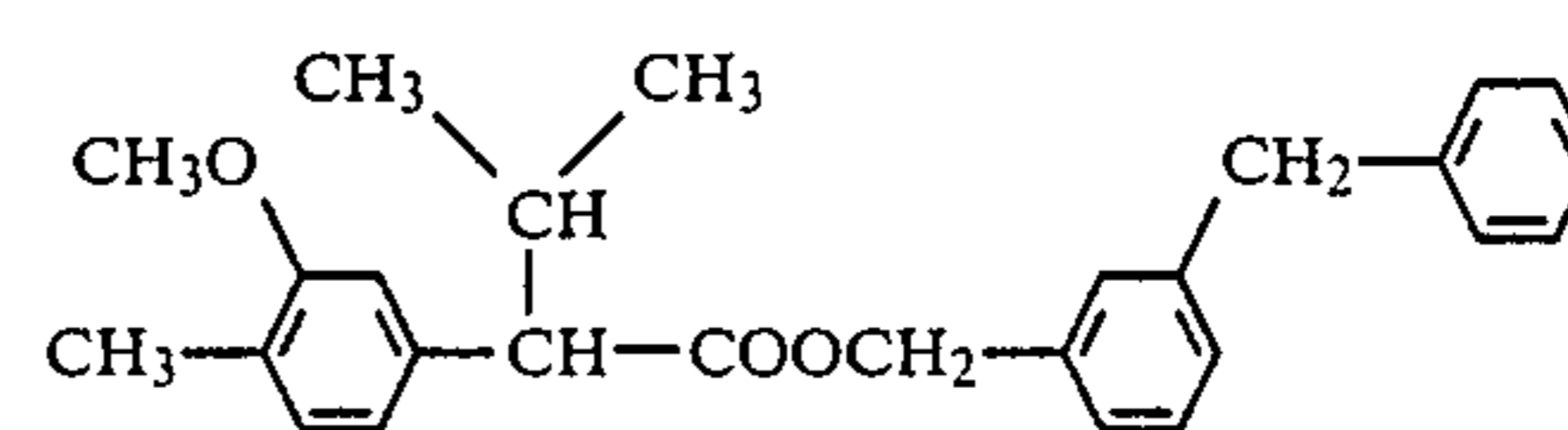


5'-Propargyl-2'-thenyl- α -isopropyl-3-methoxyphenylacetate
 n_D^{25} 1.5552

40

45

60.

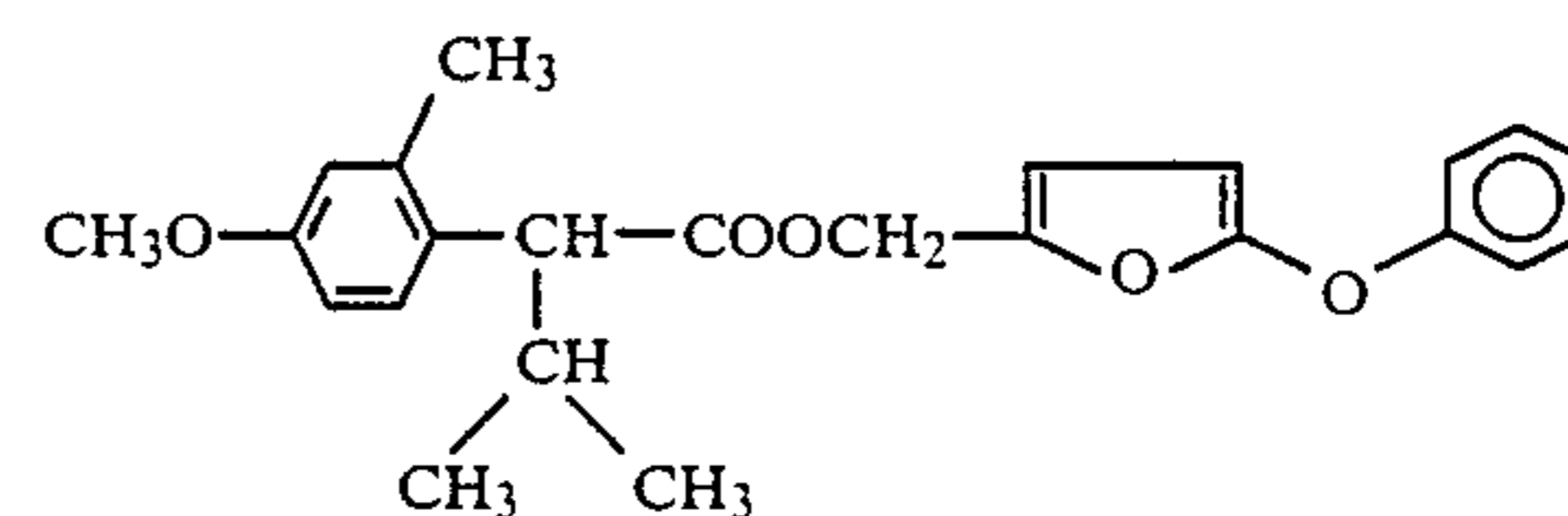


3'-Benzylbenzyl- α -isopropyl-3-methoxy-4-methylphenylacetate
 n_D^{25} 1.5509

50

55

61.

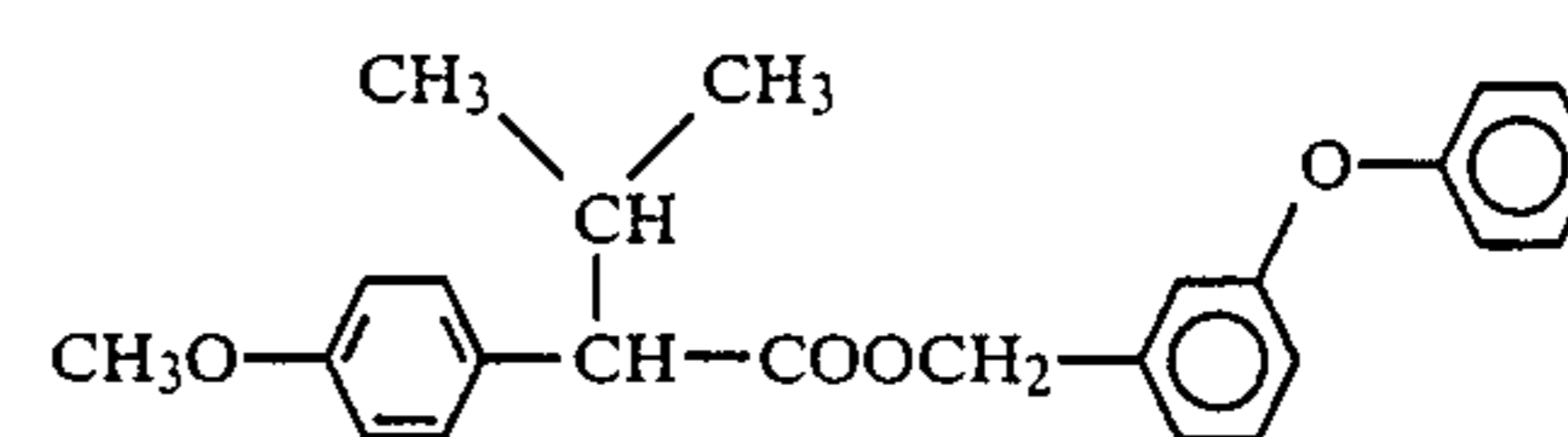


5'-Phenoxyfurfuryl- α -isopropyl-2-methyl-4-methoxyphenylacetate
 n_D^{25} 1.5501

55

60

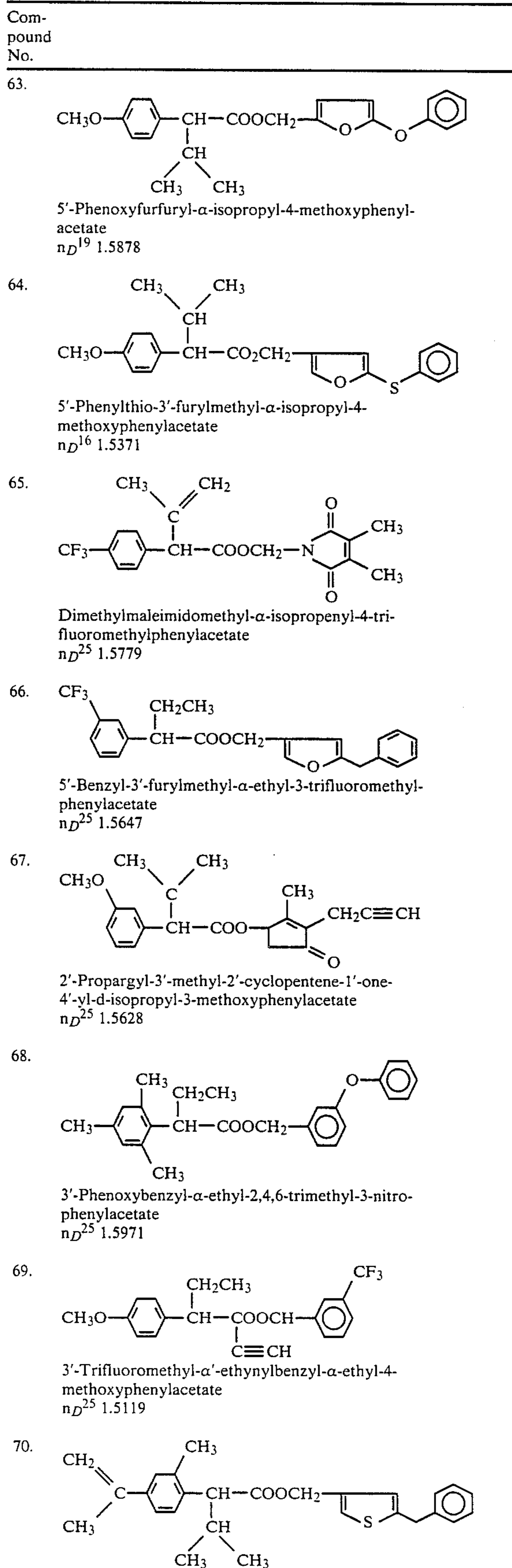
62.



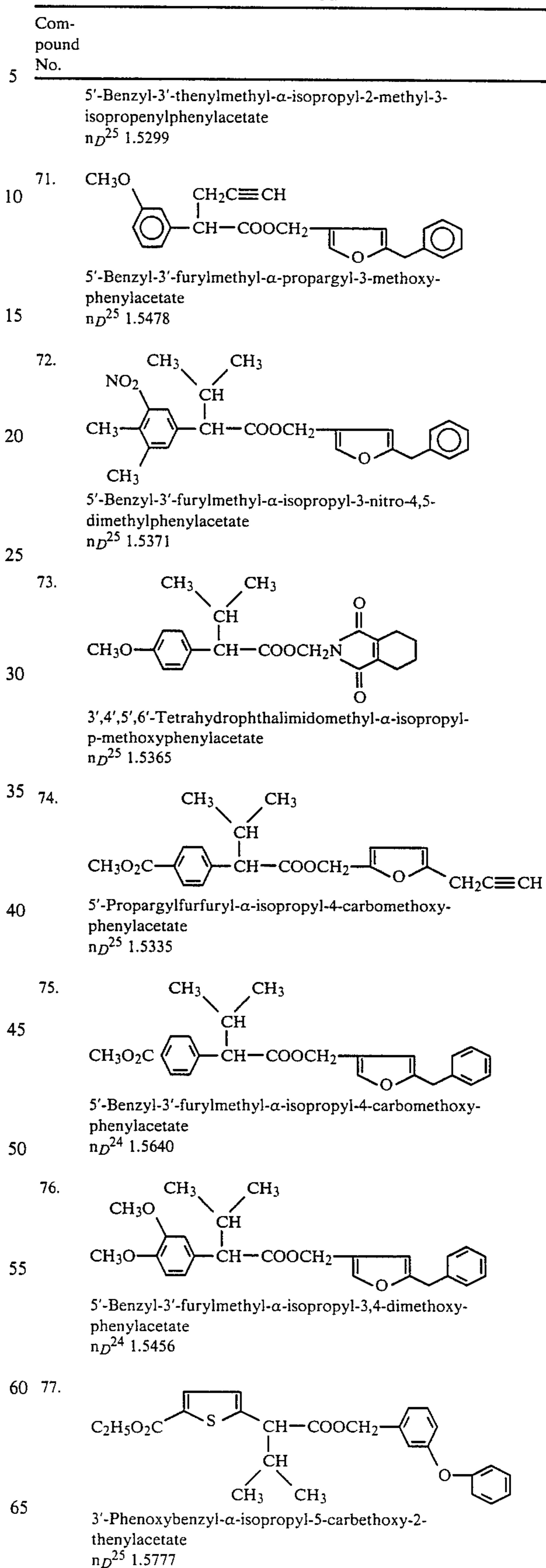
3'-Phenoxybenzyl- α -isopropyl-4-methoxyphenylacetate
 n_D^{19} 1.5878

65

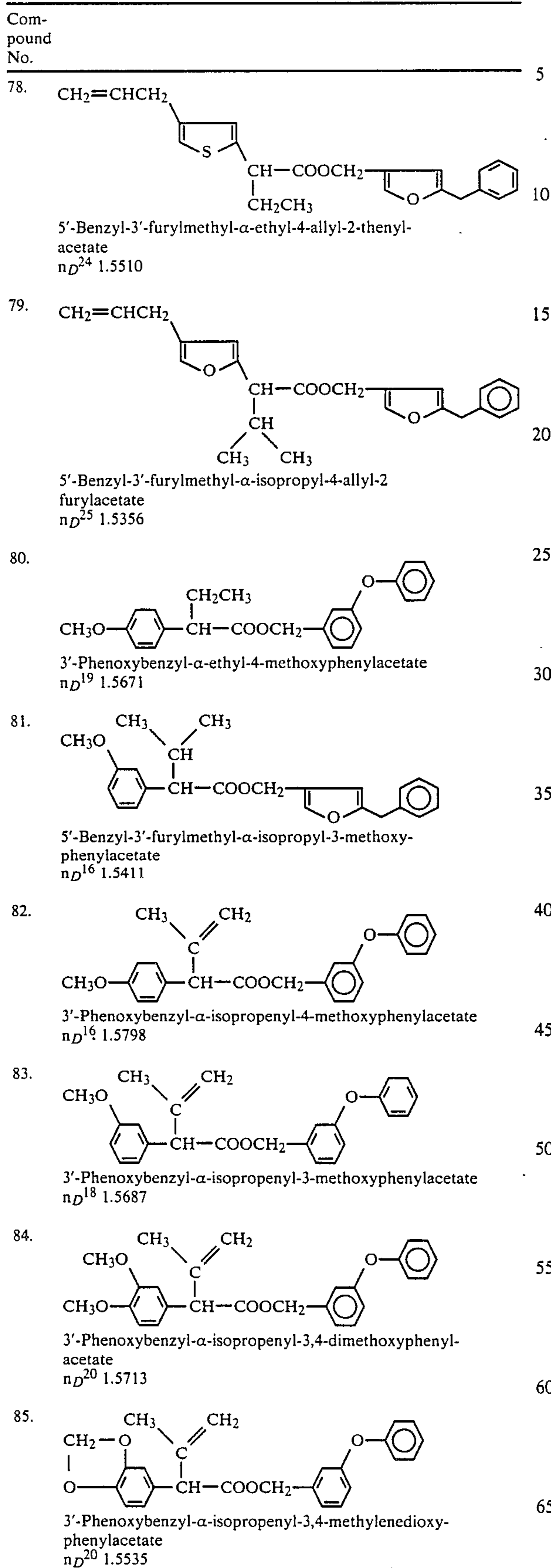
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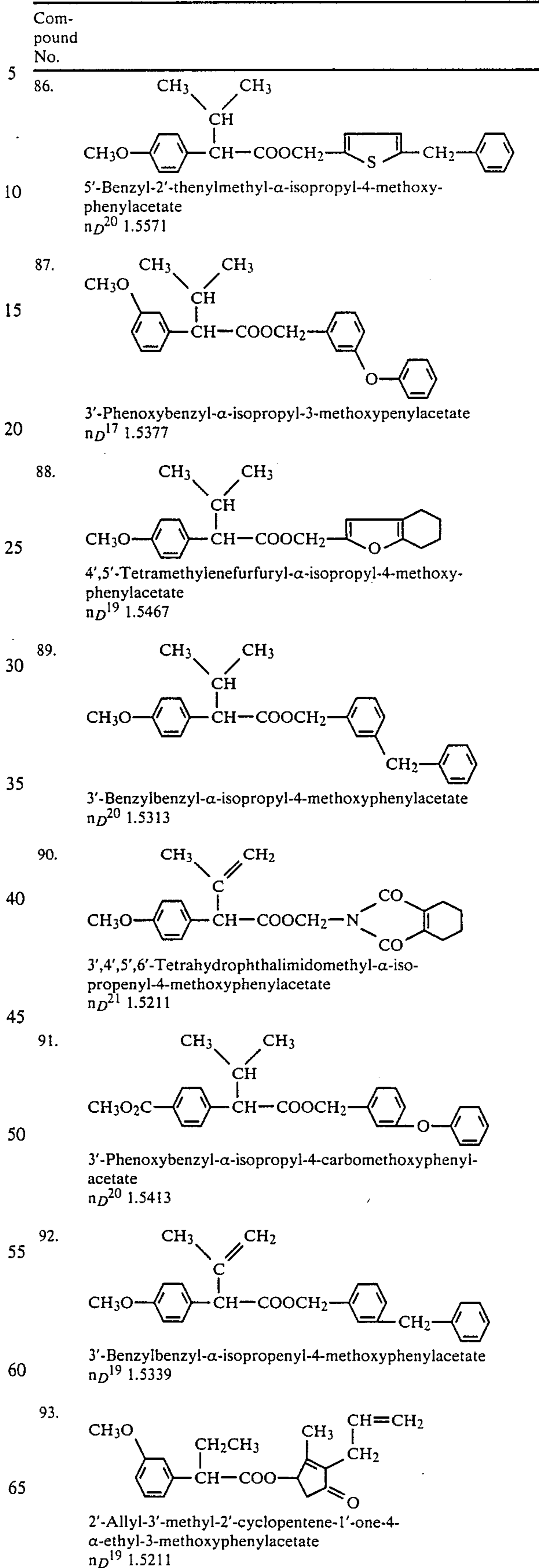
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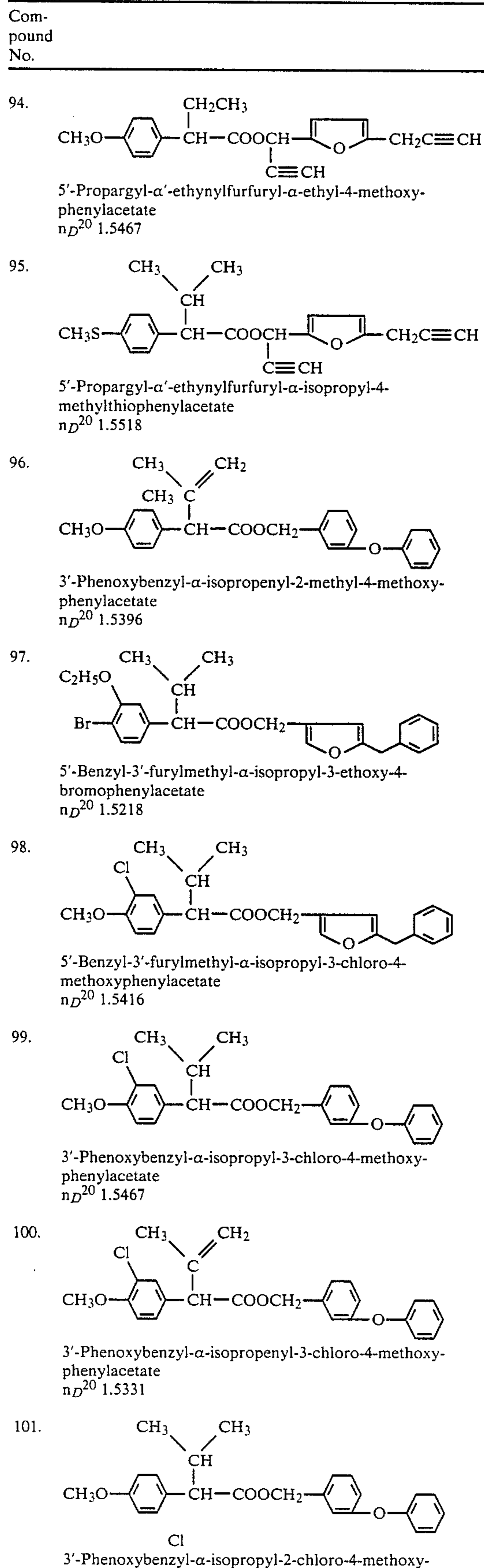
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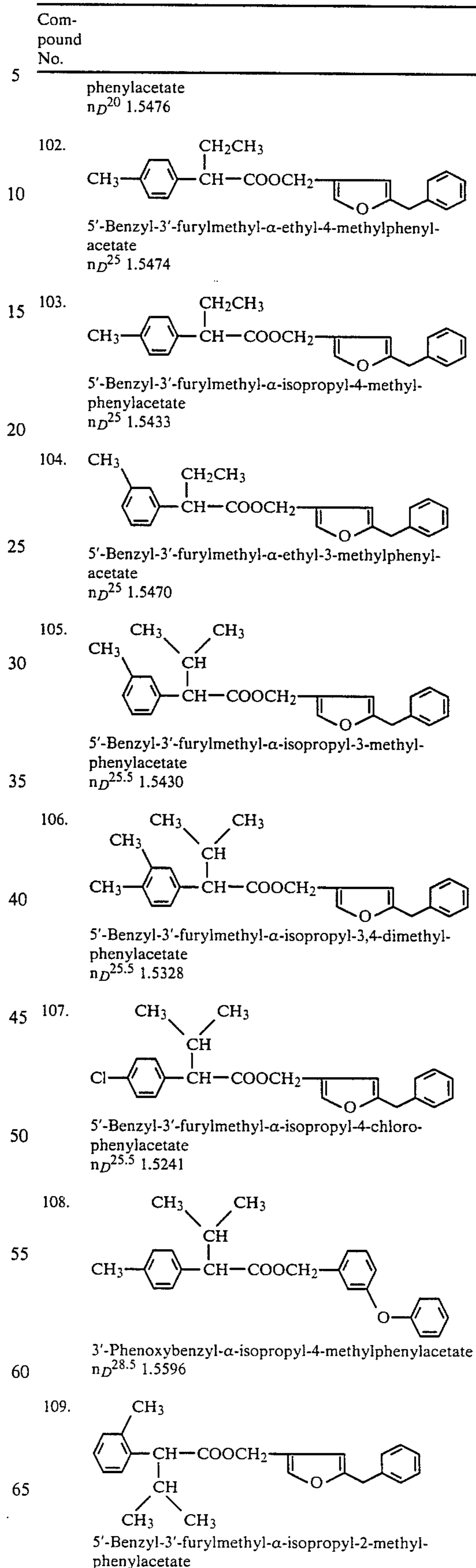
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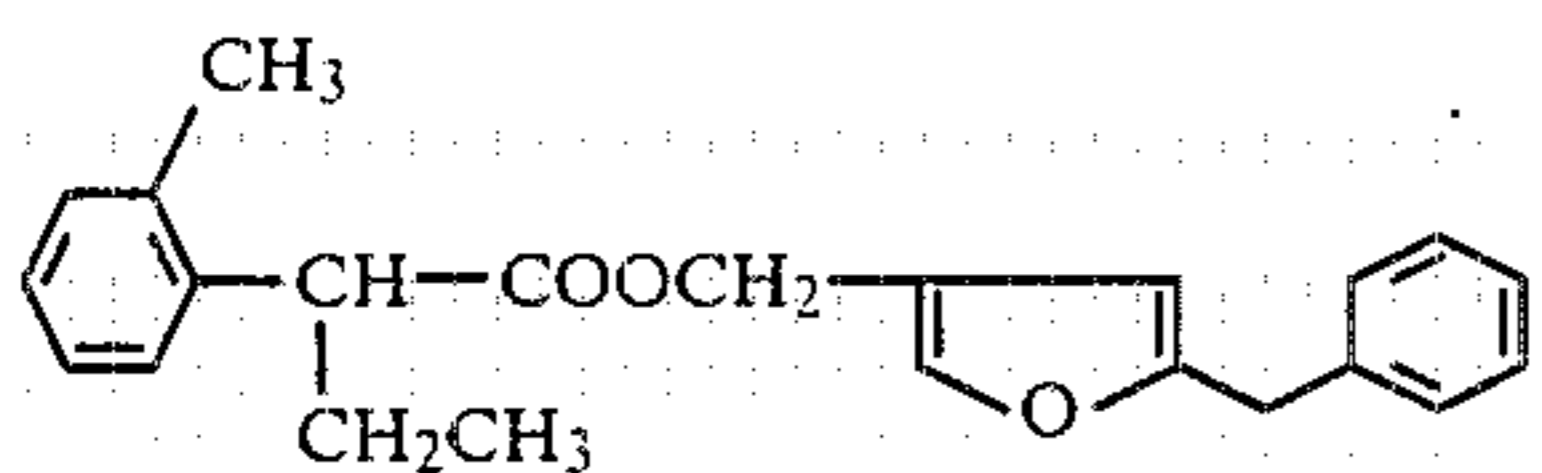
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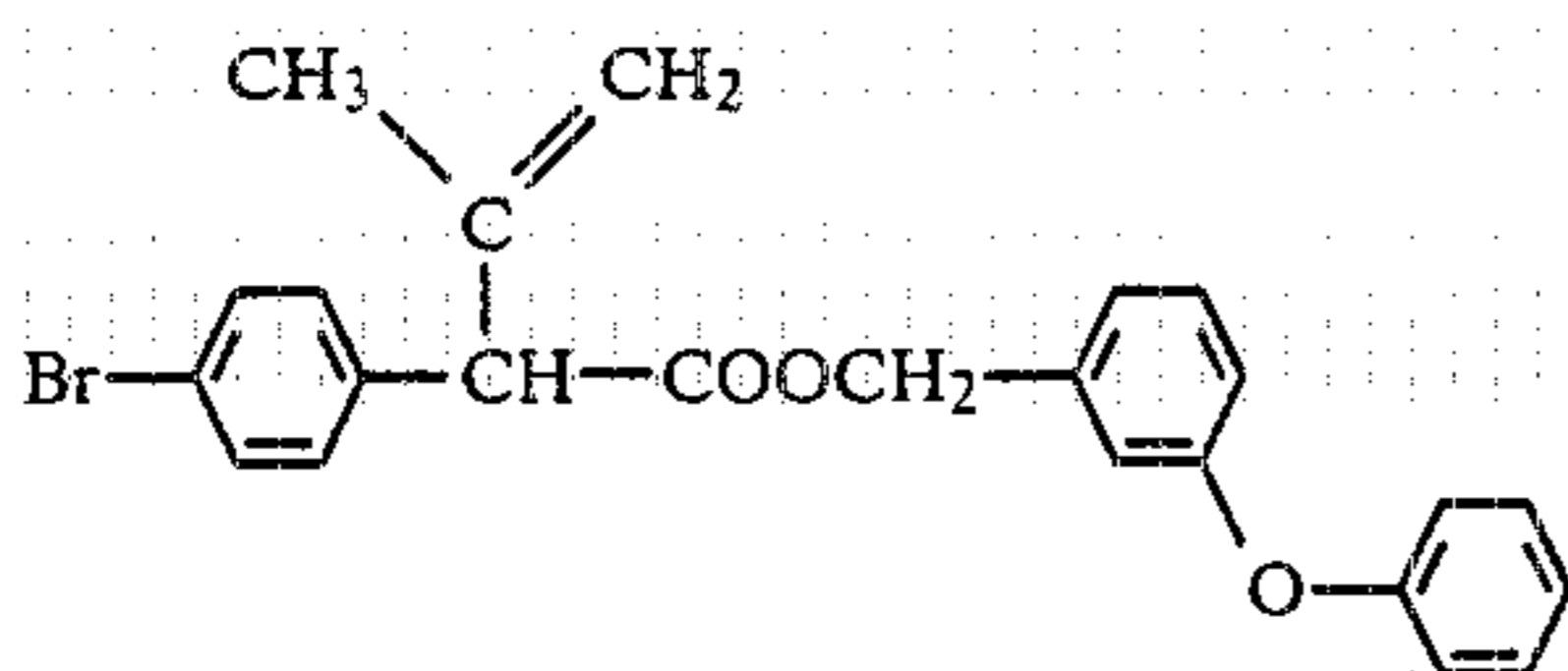
Com-
pound
No. $n_D^{26.5}$ 1.5431

110.



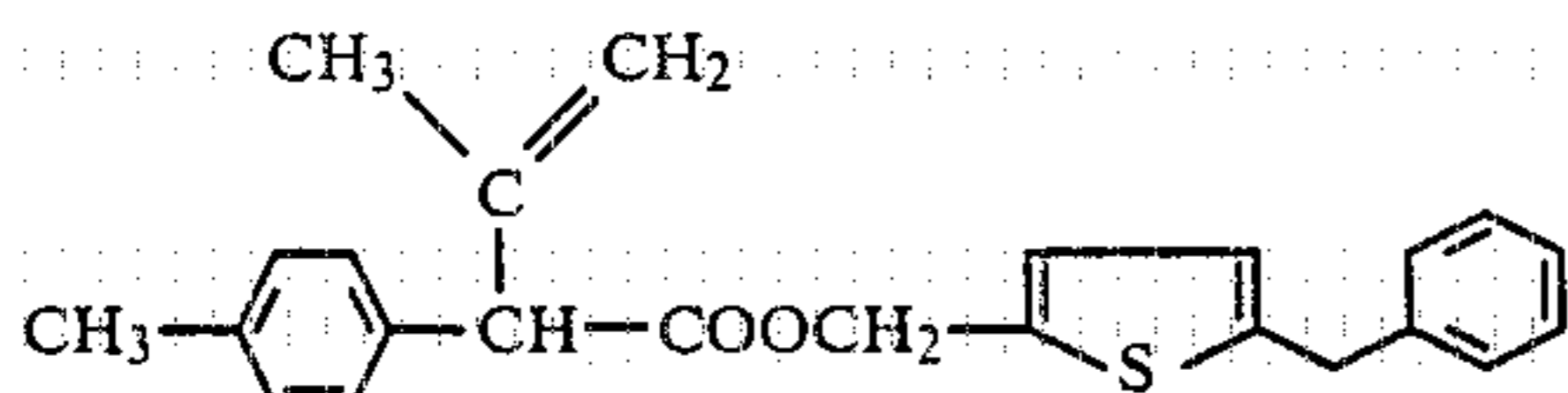
5'-Benzyl-3'-furylmethyl- α -ethyl-2-methylphenyl-
acetate
 $n_D^{26.5}$ 1.5501

111.



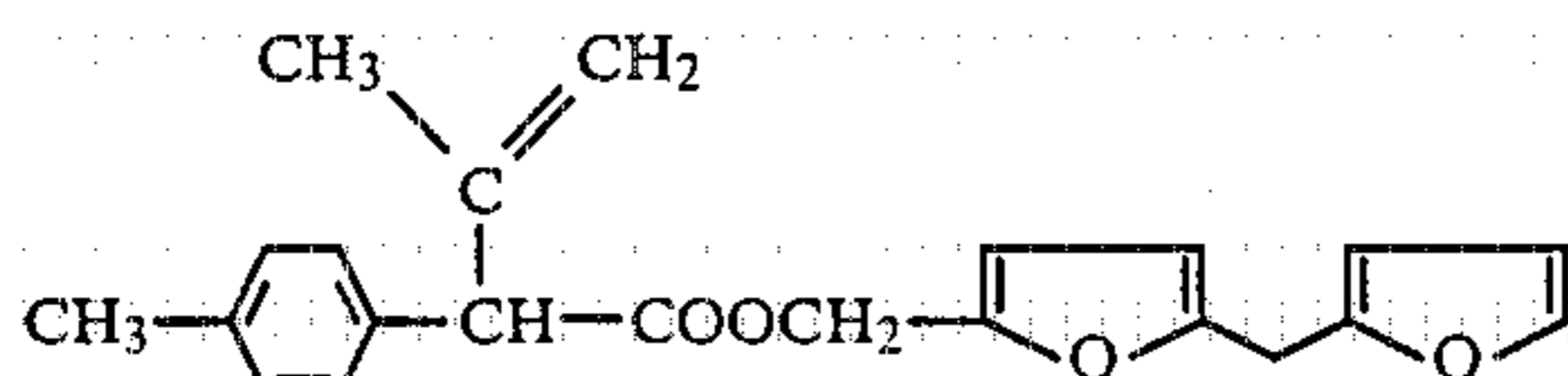
3'-Phenoxybenzyl- α -isopropenyl-4-bromophenylacetate
 n_D^{20} 1.5278

112.



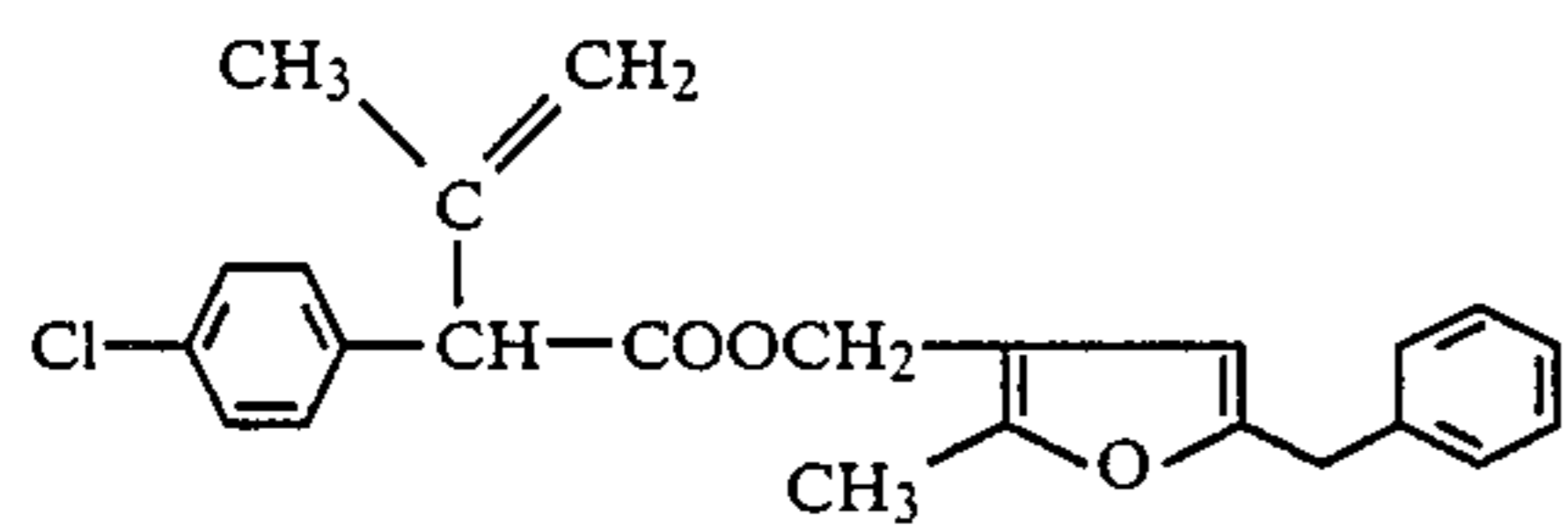
5'-Benzyl-2'-thenyl- α -isopropenyl-4-methylphenyl-
acetate
 n_D^{20} 1.5439

113.



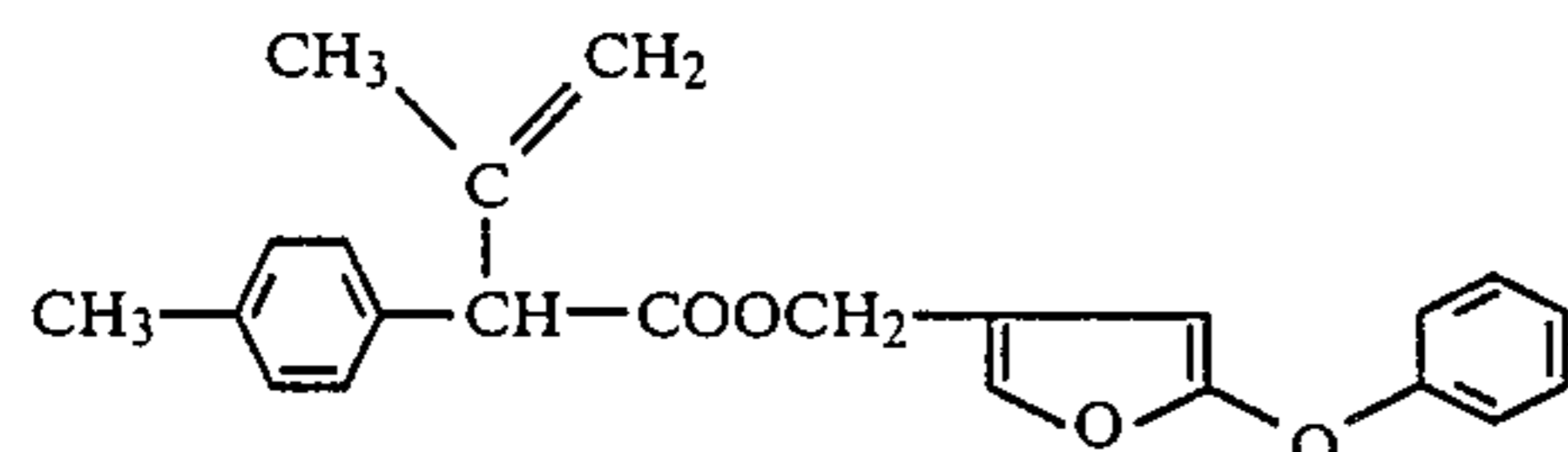
5'-Furfuryl-3'-furylmethyl- α -isopropenyl-4-methyl-
phenylacetate
 n_D^{20} 1.5411

114.



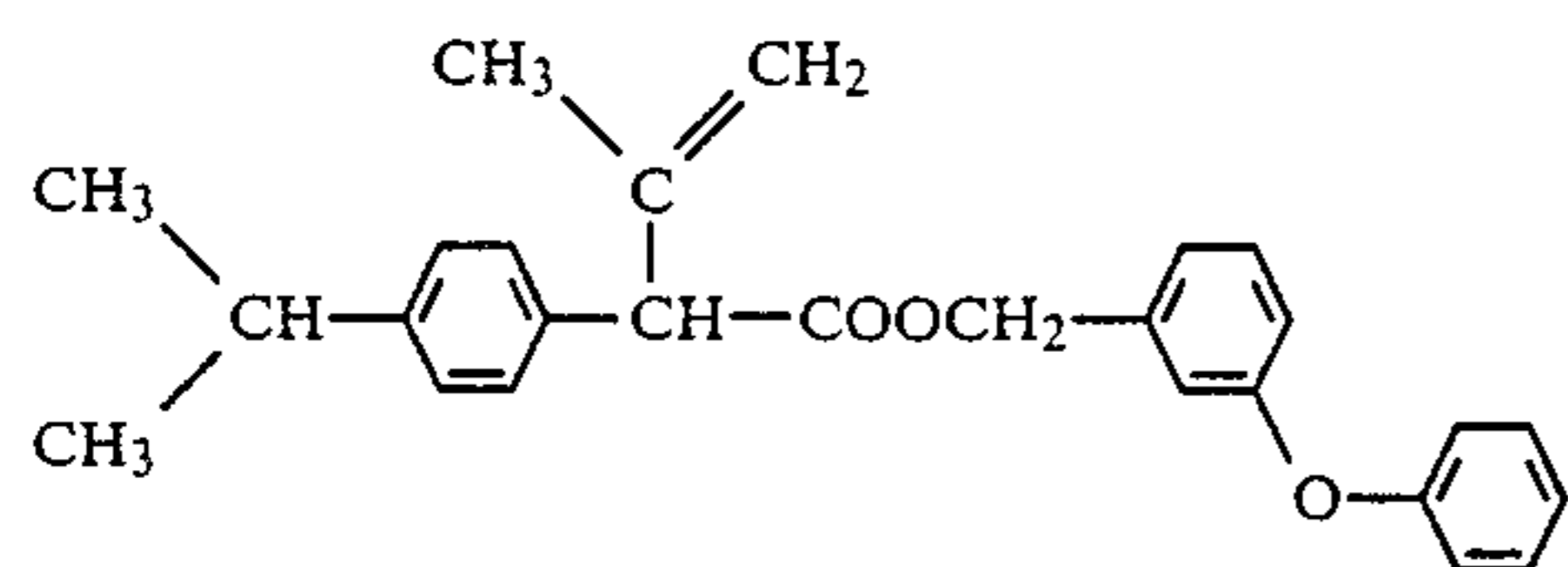
5'-Benzyl-2'-methyl-3'-furylmethyl- α -isopropenyl-4-
chlorophenylacetate
 n_D^{25} 1.5212

115.



5'-Phenoxy-3'-furylmethyl- α -isopropenyl-4-methyl-
phenylacetate
 n_D^{25} 1.5471

116.



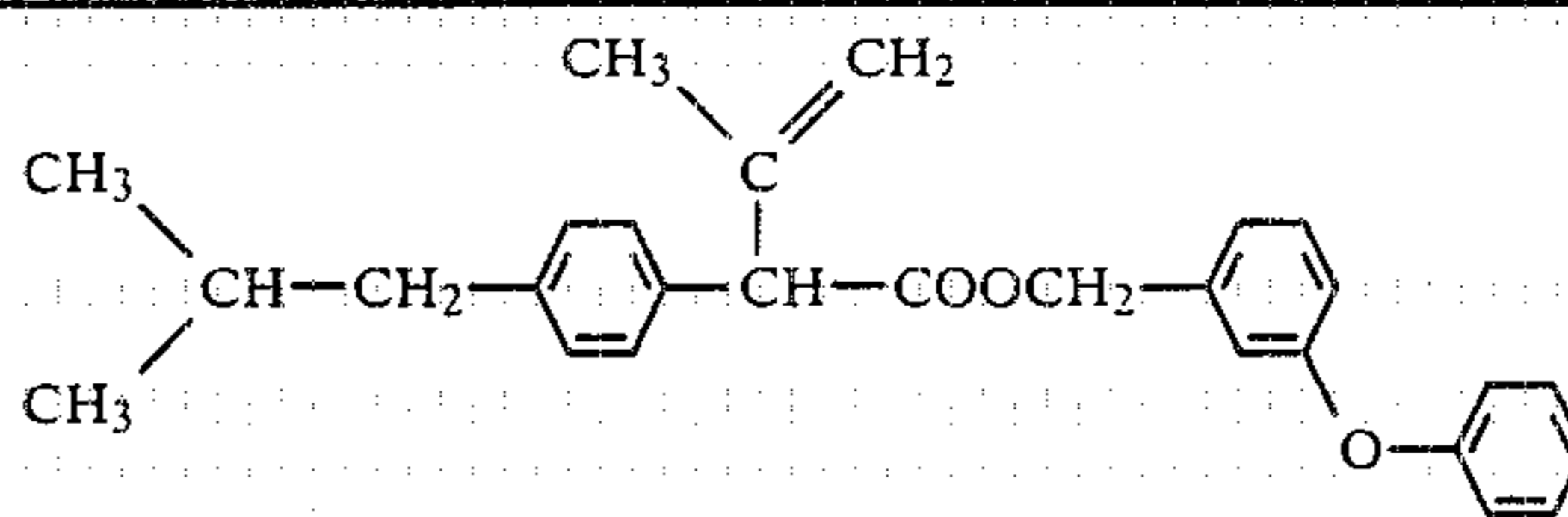
3'-Phenoxybenzyl- α -isopropenyl-4-isopropylphenyl-
acetate
 n_D^{20} 1.5218

-continued

Com-
pound
No.

5

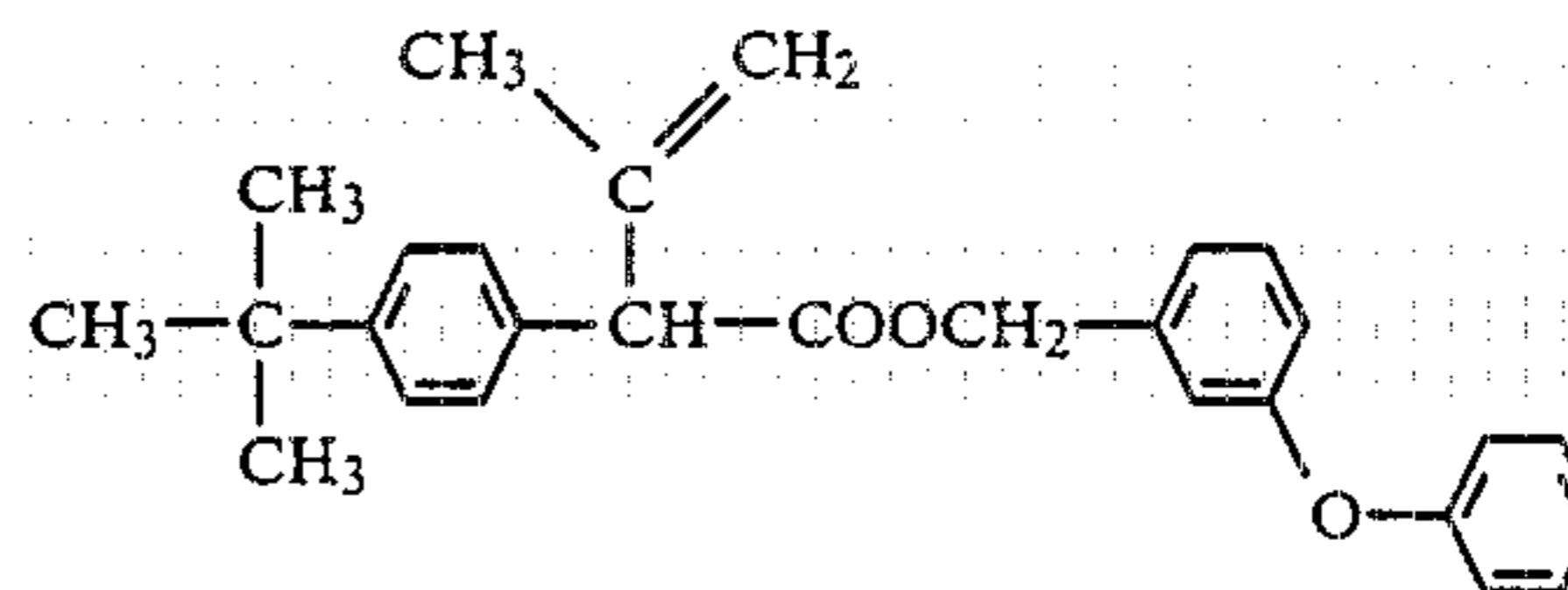
117.



3'-Phenoxybenzyl- α -isopropenyl-4-isobutylphenyl-
acetate
 n_D^{20} 1.5236

15

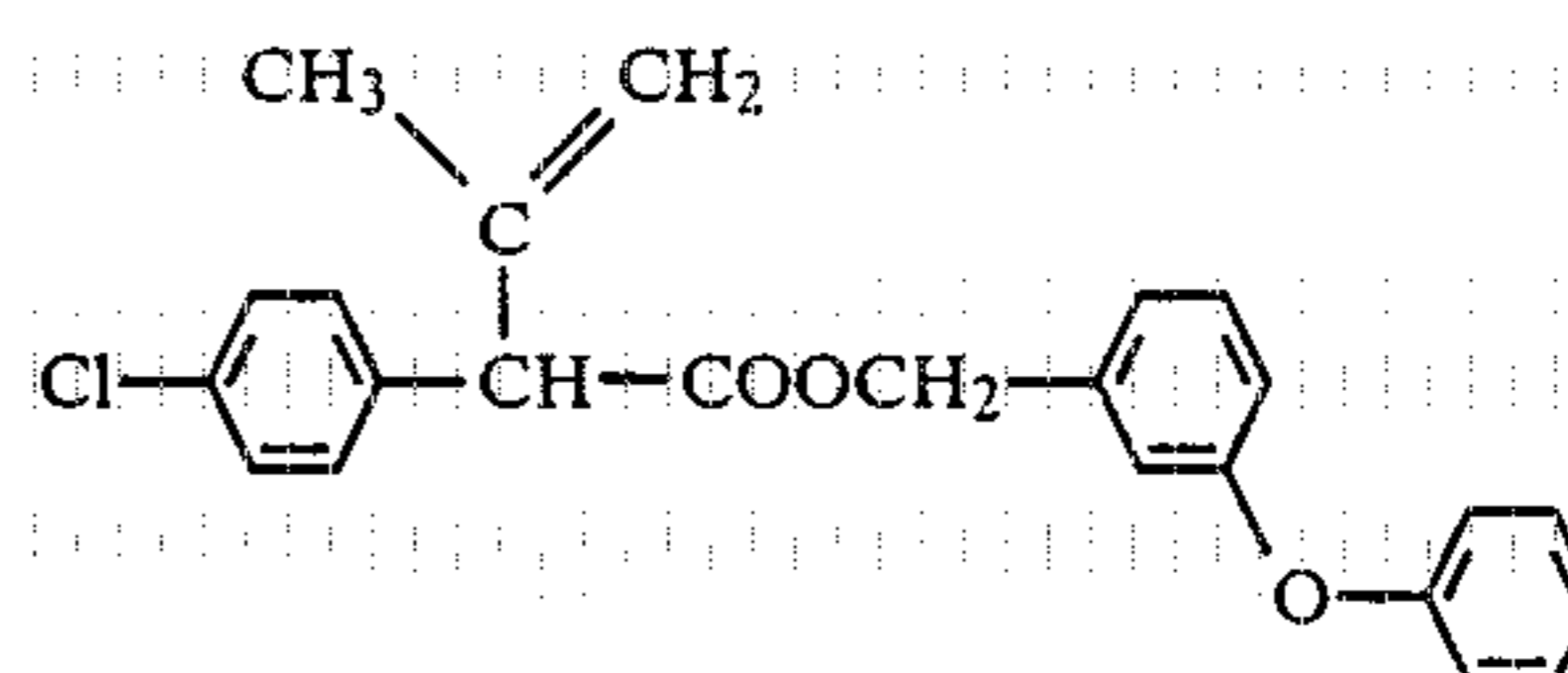
118.



3'-Phenoxybenzyl- α -isopropenyl-4-t-butylphenyl-
acetate
 n_D^{20} 1.5311

25

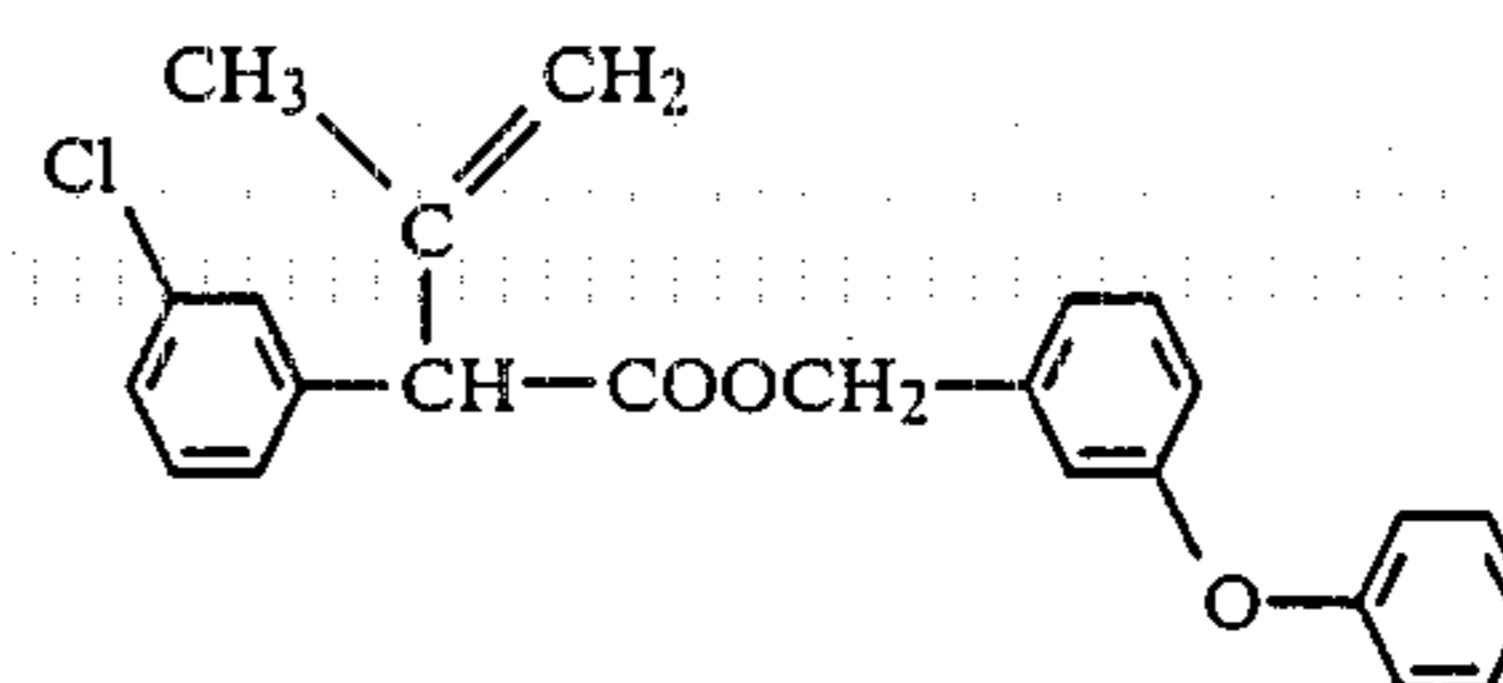
119.



3'-Phenoxybenzyl- α -isopropenyl-4-chlorophenylacetate
 n_D^{21} 1.5322

30

120.

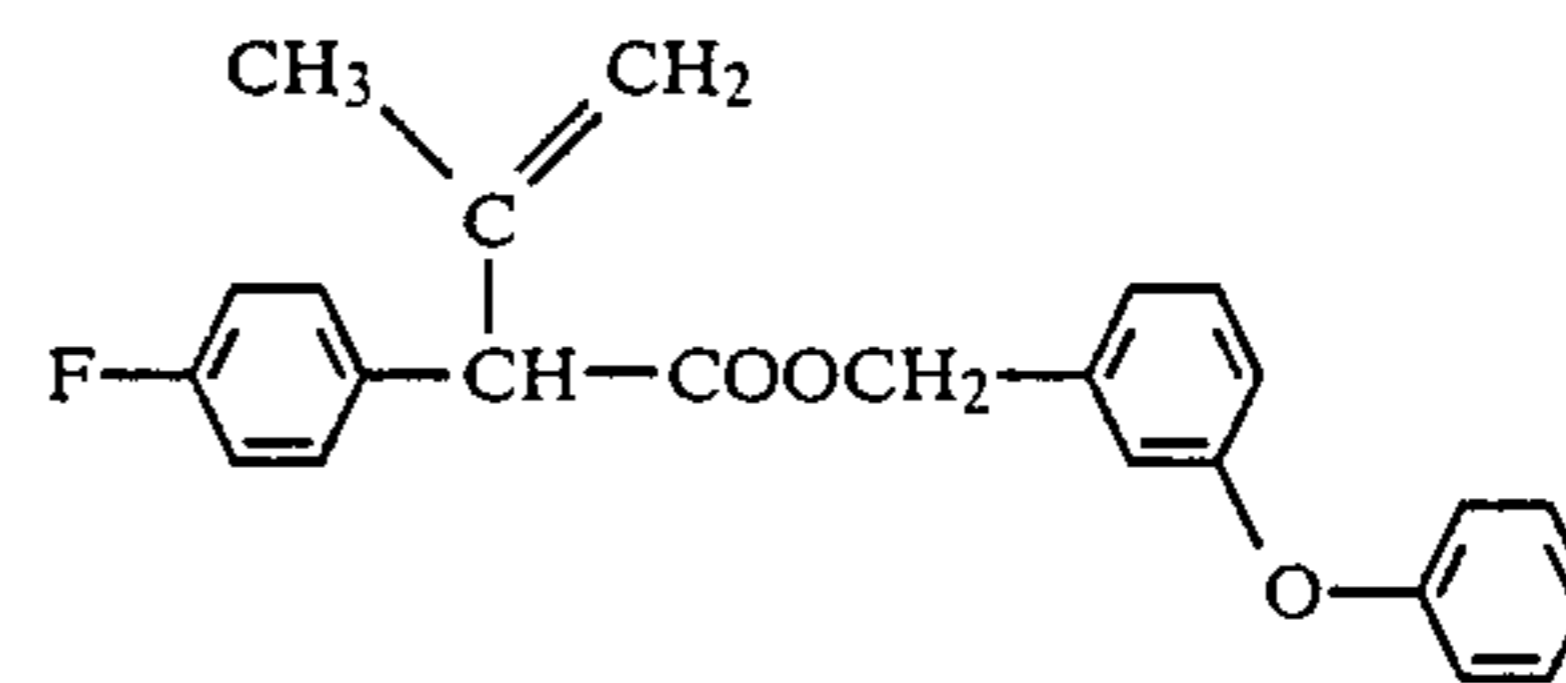


3'-Phenoxybenzyl- α -isopropyl-3-chlorophenylacetate
 n_D^{21} 1.5393

35

40

121.

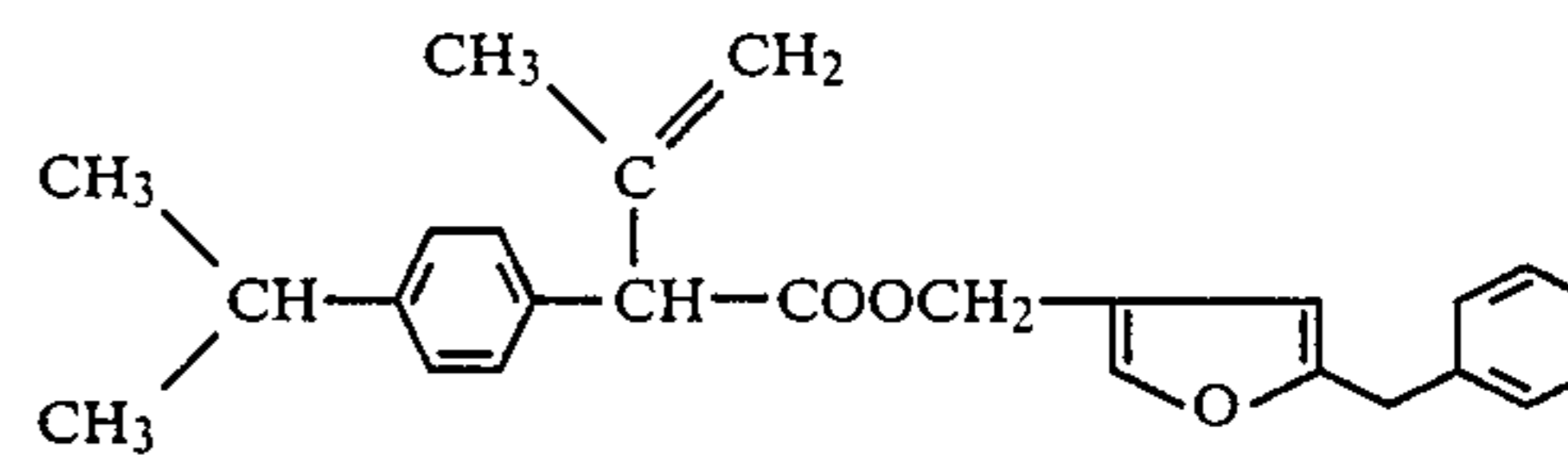


3'-Phenoxybenzyl- α -isopropenyl-4-fluorophenylacetate
 n_D^{20} 1.5363

45

50

122.

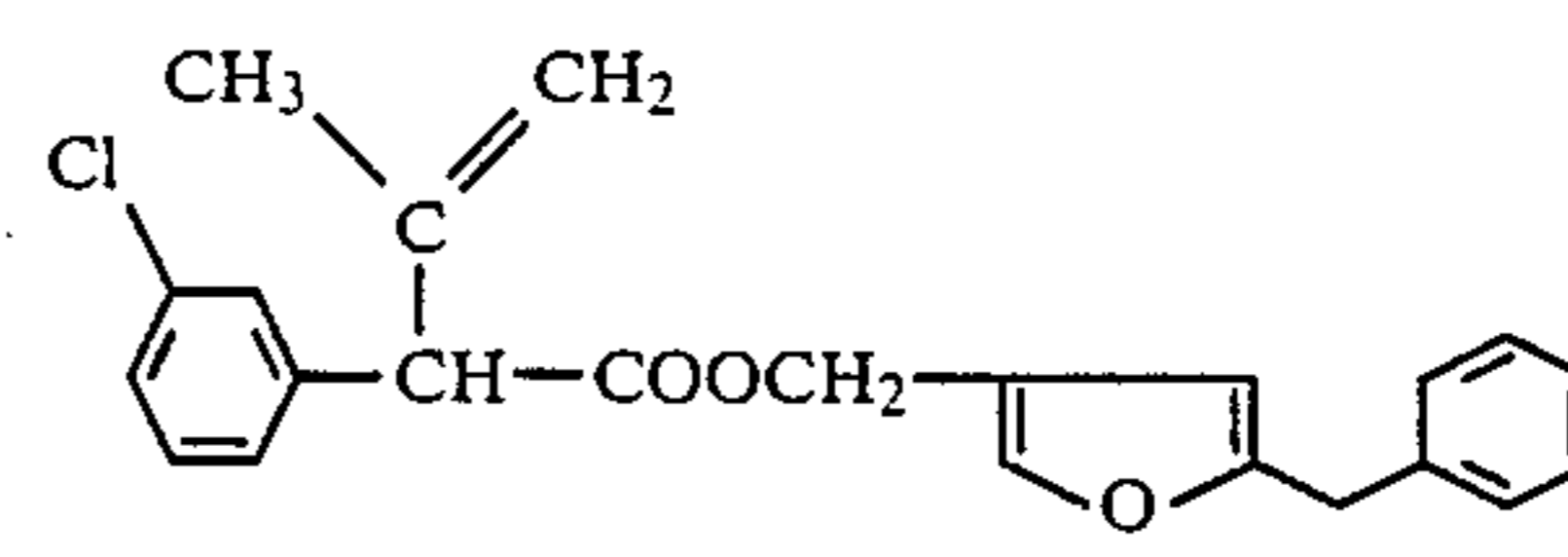


5'-Benzyl-3'-furylmethyl- α -isopropenyl-4-isopropyl-
phenylacetate
 n_D^{19} 1.5428

55

60

123.



5'-Benzyl-3'-furylmethyl- α -isopropyl-3-chloro-
phenylacetate
 n_D^{17} 1.5411

65

-continued

Compound No.	Chemical Structure	Properties
124.		3'-Phenoxybenzyl-α-isopropenyl-phenylacetate n_D^{20} 1.5396
125.		3'-Phenoxybenzyl-α-isopropenyl-4-methylphenylacetate n_D^{20} 1.5378
126.		3'-Phenoxybenzyl-α-isopropenyl-4-ethylphenylacetate n_D^{20} 1.5441
127.		3'-Phenoxybenzyl-α-isopropenyl-4-n-propyl-phenylacetate n_D^{20} 1.5466
128.		5'-Benzyl-3'-furylmethyl-α-isopropenyl-4-methyl-phenylacetate n_D^{20} 1.5398

The process for preparing the esters in accordance with the present invention is illustrated by the following Preparations, but the present invention is not limited thereto.

PREPARATION 1

Preparation of 5'-benzyl-3'-furylmethyl-α-ethyl phenylacetate (Compound No. 1)

6.20 g (0.02 mole) of α-ethylphenylacetic anhydride and 3.76 g (0.02 mole) of 5-benzyl-3-furylmethylalcohol were dissolved in 50 ml of dry pyridine and the mixture was stirred at room temperature for 2 hours. The resulting reaction mixture was poured into 100 g of ice water and extracted with three 20 ml portions of ethyl ether. The combined ether layer was then extracted with two 20 ml portions of 5% aqueous sodium hydroxide to remove α-ethylphenylacetic acid as a by-product. The ether layer was washed successively with 10% hydrochloric acid, saturated aqueous sodium bicarbonate and

saturated sodium chloride, dried over anhydrous sodium sulfate, and the ether was removed under reduced pressure to obtain 6.25 g of crude ester as a pale yellow oily substance. The resulting oily substance was then purified by column chromatography using 120 g of activated alumina and a developing solvent of benzene-n-hexane (1:3) to obtain 5.68 g (85% theoretical yield) of the desired ester as a colorless oily substance.

n_D^{17} 1.5545.

ν_{max}^{film} 3005, 1724, 1550, 1407, 1180, 806 cm^{-1} .

$\delta_{ppm}^{CCl_4}$ 7.16 (s. 5H), 7.15 (s. 5H), 7.2 (1H), 5.79 (bs. 1H), 4.78 (s. 2H), 3.78 (s. 2H), 3.31 (t. 1H, J=7 Hz), 1.3-2.3 (m. 2H), 0.83 (t. 3H, J=7 Hz).

PREPARATION 2

Preparation of m-phenoxybenzyl-α-ethylphenylacetate (Compound No. 2)

2.00 g (0.01 mole) of m-phenoxybenzyl alcohol and 1.19 g (0.015 mole) of dry pyridine were dissolved in 20 ml of benzene, and 1.83 g (0.01 mole) of α-ethylphenylacetyl chloride dissolved in 5 ml of dry benzene was added dropwise thereto under ice-cooling. After completion of addition, the mixture was stirred at room temperature for 3 hours to complete the reaction, and the reaction mixture was poured into 30 g of ice-water. The mixture was separated, and the aqueous layer was extracted with two 10 ml portions of benzene. The combined benzene layer was washed successively with 5% hydrochloric acid, saturated aqueous sodium bicarbonate and saturated aqueous sodium chloride, and benzene was removed under reduced pressure. The resulting crude ester was charged onto a column packed with 40 g of activated alumina using a mixture of benzene-n-hexane (1:3) as an eluent to obtain 3.15 g (91% theoretical yield) of the purified ester.

n_D^{22} 1.5712.

ν_{max}^{film} 3030, 1740, 1590, 1470, 1260, 1220, 1160 cm^{-1} .

$\delta_{ppm}^{CCl_4}$ 7.4-5.8 (m. 14H), 4.95 (s. 2H), 3.48 (t. 1H, J=8 Hz), 2.3-1.1 (m. 2H), 0.85 (t. 3H, J=8 Hz).

PREPARATION 17

Preparation of 5'-benzyl-3'-furylmethyl-α-cyclohexylphenylacetate (Compound No. 17)

2.82 g (0.015 mole) of 5-benzyl-3-furylmethyl alcohol and 2.22 g (0.01 mole) of ethyl-α-cyclohexylphenylacetate were dissolved in 50 ml of dry toluene and 0.1 g of sodium hydride was added thereto as a catalyst. The mixture was charged into a Teflon Spinning Band rectification tower and ethanol formed as a by-product was removed while heating and stirring. The reaction was completed in about 1 hour, and the reaction mixture was allowed to stand and cooled to room temperature and poured into 30 g of ice-water. The layers were separated to remove the toluene layer and the toluene layer was washed with saturated aqueous sodium chloride. Toluene was removed under reduced pressure, and the resulting crude ester was subjected to a column packed with 50 g of activated alumina using benzene-n-hexane (1:3) as an eluent to obtain 2.79 g (72% theoretical yield) of the purified ester as white crystals. Melting point, 51°-53° C.

ν_{max}^{nujol} 3035, 1738, 1606, 1500, 1454, 1155 cm^{-1} .

$\delta_{ppm}^{CCl_4}$ 7.3-6.9 (m. 11H), 5.80 (bs. 1H), 4.77 (AB. q 2H), 3.70 (s. 2H), 3.06 (d. 2H, J=10 Hz), 2.1-0.6 (m. 10H).

PREPARATION 5

Preparation of
dimethylmaleimidomethyl- α -allylphenylacetate
(Compound No. 5)

1.74 g (0.01 mole) of N-chloromethyl-dimethylmaleimide and 2.93 g (0.01 mole) of silver α -allylphenylacetate were added to 25 ml of dimethylformamide and the mixture was allowed to stand overnight at room temperature. The precipitated crystals were filtered and dimethylformamide was distilled off under reduced pressure. The resulting crude ester was subjected to a silica gel column (90 g) using a mixture of chloroform-carbon tetrachloride (2:1) as an eluent to obtain 2.43 g (75% theoretical yield) of the desired ester as a white oily substance.

$n_D^{22.0}$ 1.5349.

ν_{max}^{film} 1780, 1740, 1720, 1640, 1600, 1140, 910 cm^{-1} .

$\delta_{ppm}^{CDCl_3}$ 7.15 (s, 5H), 6.0-4.9 (m, 3H), 3.48 (t, 3H, $J=10$ Hz), 3.0-2.0 (m, 2H), 2.0 (s, 6H).

PREPARATION 22

Preparation of
5'-benzyl-3'-furylmethyl- α -isopropyl-4-methoxyphenylacetate (Compound No. 22)

1.90 g (0.01 mole) of 5'-benzyl-3-furfuryl alcohol and 1.58 g (0.02 mole) of pyridine were dissolved in 50 ml of dry benzene, and 2.30 g (0.01 mole) of α -isopropyl-4-methoxyphenylacetyl chloride dissolved in 5 ml of dry benzene was added dropwise to the solution over about 30 minutes. After completion of the addition, the mixture was stirred at room temperature for 3 hours to complete the reaction. The reaction mixture was poured into 50 g of ice-water followed by separating the benzene layer, and the aqueous layer was extracted with two 20 ml portions of benzene. The combined benzene layer was washed successively with 5% hydrochloric acid, saturated aqueous sodium bicarbonate and saturated aqueous sodium chloride. A yellow oily substance obtained after removal of benzene was subjected to a column packed with 25 g of activated alumina using benzene-n-hexane (1:3) as an eluent to obtain 3.33 g (86.0% theoretical yield) of the purified ester as a colorless oily substance.

n_D^{17} 1.5470.

ν_{max}^{film} 1735, 1617, 1515, 1250, 1032, 830, 730 cm^{-1} .

$\delta_{ppm}^{CCl_4}$ 0.65 (d, 3H, $J=7$ Hz), 0.95 (d, 3H, $J=7$ Hz), 1.7-2.5 (m, 1H), 2.97 (d, 1H, $J=11$ Hz), 3.67 (s, 3H), 3.81 (s, 2H), 4.78 (d.d, 2H), 5.83 (b.s, 1H), 6.90 (d.d, 4H), 7.1-7.3 (m, 6H).

PREPARATION 27

Preparation of
3,4,5,6-tetrahydrophthalimidomethyl- α -ethyl-2,4,6-trimethylphenylacetate (Compound No. 27)

10.31 g (0.05 mole) of α -ethyl-2,4,6-trimethylphenylacetic acid and 9.05 g (0.05 mole) of 3,4,5,6-tetrahydrophthalimidomethylol were dissolved in 100 ml of dry benzene, and 16.51 g (0.08 mole) of dicyclohexylcarbodiimide was added to the solution. The mixture was then allowed to stand overnight in a sealed vessel. On next day, the mixture was heated at reflux to complete the reaction, and, after allowing to cool, the precipitated dicyclohexylurea was separated by filtration. A viscous oily substance obtained by concentration of the filtrate was subjected to a silica gel column (360 g)

to obtain 11.3 g (61.18% theoretical yield) of the purified ester as a colorless viscous oily substance.

n_D^{25} 1.5399.

ν_{max}^{film} 1780, 1740, 1720, 1511, 1405, 1140 cm^{-1} .

$\delta_{ppm}^{CCl_4}$ 0.85 (t, 3H, $J=7$ Hz), 1.5-2.3 (m, 2H), 2.3 (b.s, 9H), 3.32 (t, 3H, $J=8$ Hz), 1.5-2.0 (m, 4H), 2.0-2.5 (m, 4H), 5.34 (d.d, 2H), 7.0-7.3 (m, 2H).

PREPARATION 47

Preparation of
2'-allyl-3'-methyl-2'-cyclopentene-1'-one-4'-yl- α -ethyl-3-methoxyphenylacetate (Compound No. 93)

7.41 g (0.02 mole) of α -ethyl-3-methoxyphenylacetic anhydride and 1.52 g (0.01 mole) of 2-allyl-3-methyl-4-hydroxy-2-cyclopentene-1-one were dissolved in 50 ml of dry pyridine, and the mixture was stirred overnight at room temperature. On next day, the reaction mixture was poured into 100 g of ice-water and extracted with three 20 ml portions of ethyl ether. The combined ether layer was then extracted with two 30 ml portions of 5% aqueous sodium hydroxide to remove a carboxylic acid formed as a by-product. The ether layer was washed successively with 10% aqueous hydrochloric acid, saturated aqueous sodium bicarbonate and saturated aqueous sodium chloride and dried over anhydrous sodium sulfate. Removal of ether under reduced pressure yielded 3.2 g of a crude ester. The product was then subjected to a column packed with 20 g of activated alumina using benzene-n-hexane (1:3) as an eluent to obtain 2.73 g (83.1% theoretical yield) of the desired ester.

n_D^{19} 1.5211.

ν_{max}^{film} 1730, 1710, 1150, 1140, 1000, 910 cm^{-1} .

$\delta_{ppm}^{CCl_4}$ 0.90 (t, 3H, $J=8$ Hz), 1.5-3.0 (m, 11H), 3.36 (t, 1H, $J=9$ Hz), 3.67 (s, 1H), 4.7-6.0 (m, 3H), 6.5-6.9 (m, 4H).

PREPARATION 46

Preparation of
3'-phenoxybenzyl- α -isopropyl-3-methoxyphenylacetate (Compound No. 87)

8.01 g (0.04 mole) of m-phenoxybenzyl alcohol and 7.09 g (0.03 mole) of ethyl α -isopropyl-3-methoxyphenylacetate were dissolved in 100 ml of dry toluene, and 0.1 g of sodium hydride was added thereto as a catalyst. The mixture was then charged in a Teflon Spinning Band rectifier tower having a length of 50 cm and heated while stirring. The reaction was completed in about 3 hours by removing ethanol as an azeotropic distillation with toluene at the top of the tower. After allowing to stand and cooling the mixture was poured into cold water to separate layers. Removal of toluene by distillation under reduced pressure yielded 11.5 g of a crude ester. The product was then subjected to a column packed with 55 g of activated alumina using benzene-n-hexane (1:3) as an eluent to obtain 10.12 g (86.4% theoretical yield based on the starting ethyl ester of the desired ester).

n_D^{17} 1.5377.

ν_{max}^{film} 3060, 1738, 1590, 1490, 1255, 1145, 775, 690 cm^{-1} .

$\delta_{ppm}^{CCl_4}$ 0.70 (d, 3H, $J=7$ Hz), 0.97 (d, 3H, $J=7$ Hz), 0.9-2.6 (m, 1H), 3.08 (d, 1H, $J=11$ Hz), 3.65 (s, 3H), 4.95 (AB Type d.d, 2H), 6.5-7.4 (m, 13H).

PREPARATION 48

Preparation of

3'-phenoxybenzyl-2-(1',2',3',4'-tetrahydronaphthalene-8'-yl)isovalerate (Compound No. 43)

3.95 g (0.015 mole) of m-phenoxybenzyl bromide, 2.32 g (0.01 mole) of 2-(1',2',3',4'-tetrahydronaphthalene-8'-yl)isovaleric acid and 2.02 g (0.02 mole) of triethylamine were added to 50 ml of dimethylformamide and the mixture was stirred overnight at room temperature. The reaction mixture was poured into 50 g of ice-water and extracted with three 30 ml portions of ethyl ether. The combined ether layer was washed successively with 5% hydrochloric acid, saturated aqueous sodium bicarbonate and saturated aqueous sodium chloride and dried over anhydrous sodium sulfate. Removal of ethyl ether by distillation under reduced pressure yielded a crude ester which was then subjected to a column packed with 20 g of activated alumina using benzene-n-hexane (1:3) as an eluent to obtain 3.45 g (83.2% theoretical yield based on the carboxylic acid

stating material) of the desired ester as a colorless oily substance.

n_D^{19} 1.5660.

ν_{max}^{film} 3060, 1736, 1588, 1490, 1255, 770, 690 cm^{-1} .

$\delta_{ppm}^{CCl_4}$ 0.68 (d. 3H, J=7 Hz), 1.03 (d. 3H, J=7 Hz), 1.5-2.0 (m. 4H), 2.1-2.6 (m. 1H), 2.5-3.0 (m. 4H), 2.48 (d. 1H, J=11 Hz), 4.74 (S. H), 6.6-7.4 (m. 12H).

In a similar manner as described in Preparations 1, 2, 5, 17, 22, 27, 46 and 48, the compounds listed in Table below were obtained. In the table, the symbols "a", "b", "c", "d" and "e" stand for the following esterification methods.

a: Esterification using an acid chloride.

b: Esterification using an acid anhydride.

c: Dehydration esterification of an acid and an alcohol with dicyclohexylcarbodiimide in an inert solvent.

d: Esterification by transesterification using sodium hydride as a catalyst in toluene.

e: Esterification using a silver or a triethylammonium salt of acid and a halide of an alcohol.

Representative compounds of this invention prepared by the above methods are shown below, but they are not to be construed as limiting the present invention.

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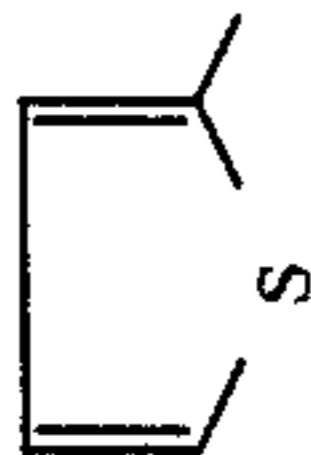

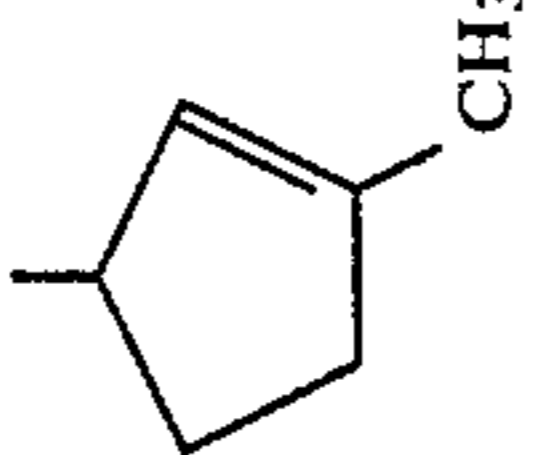
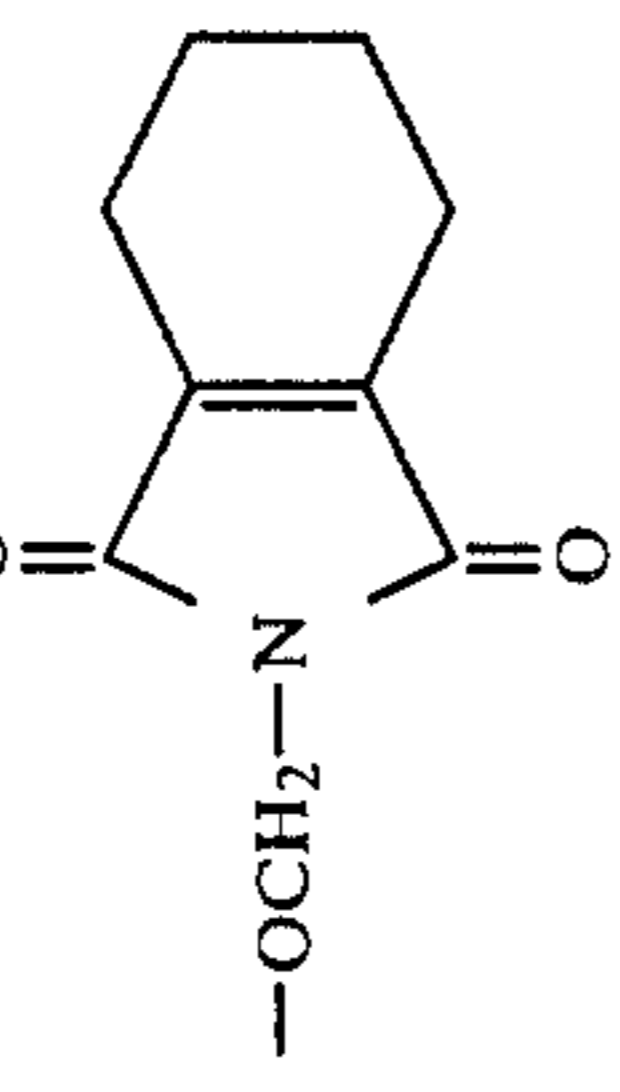
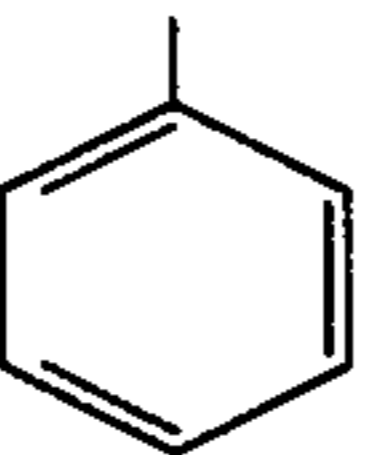
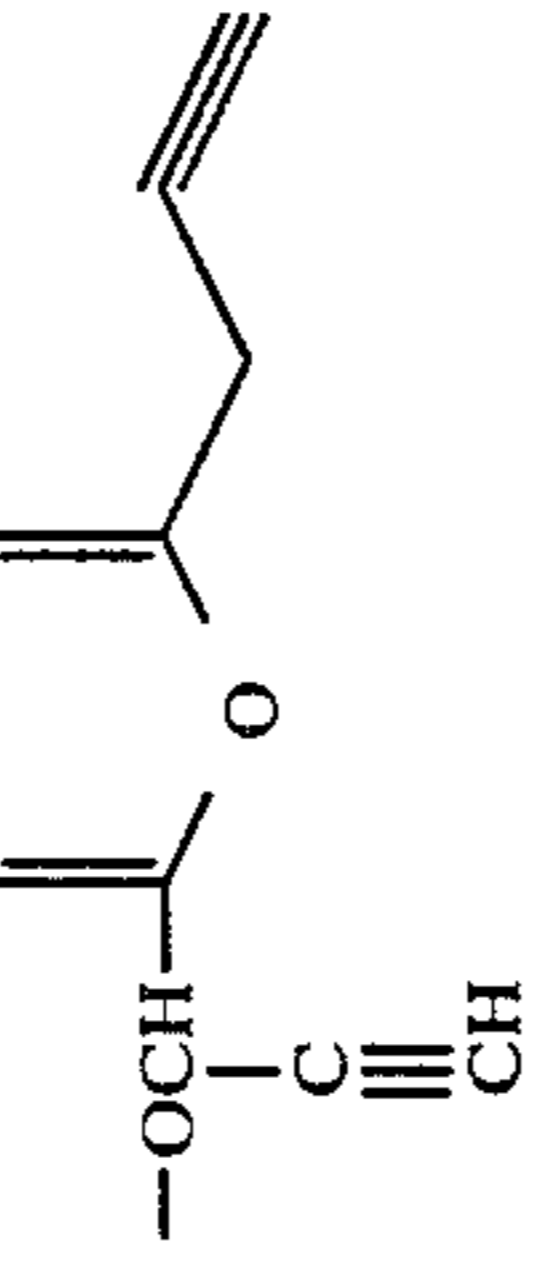
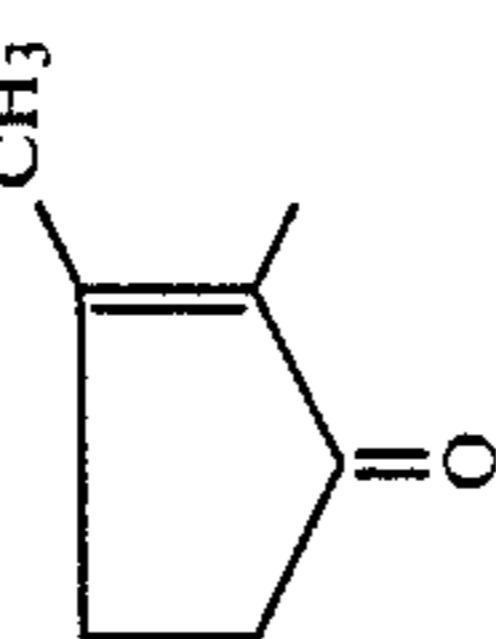
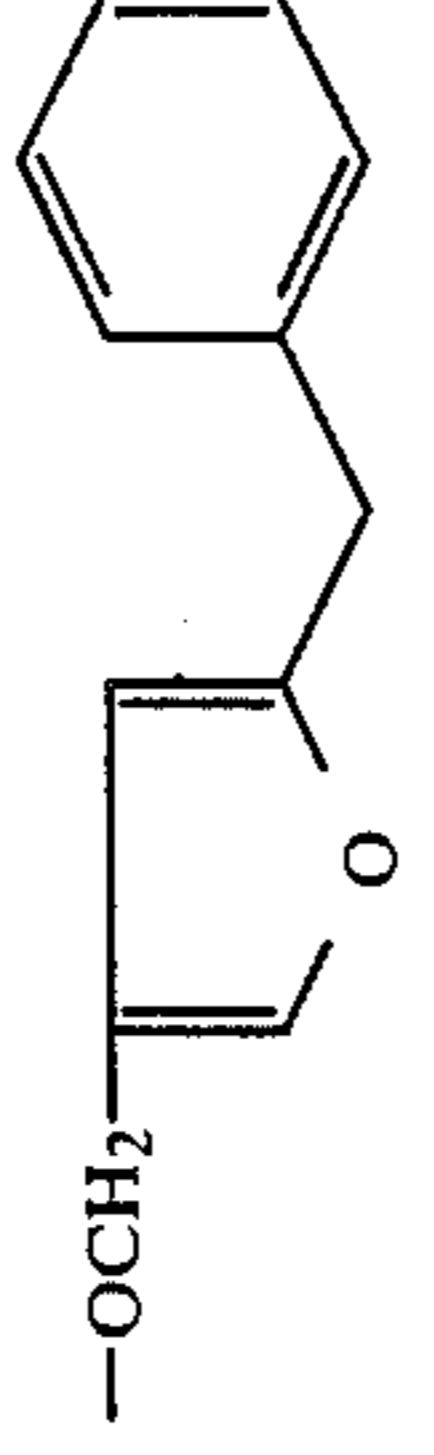
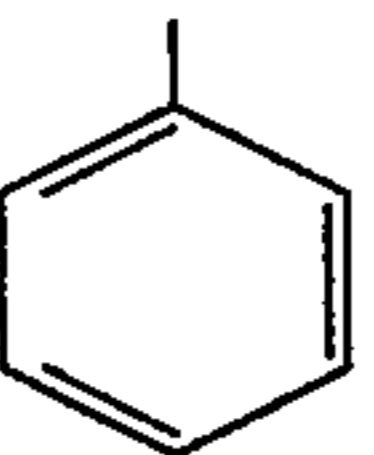
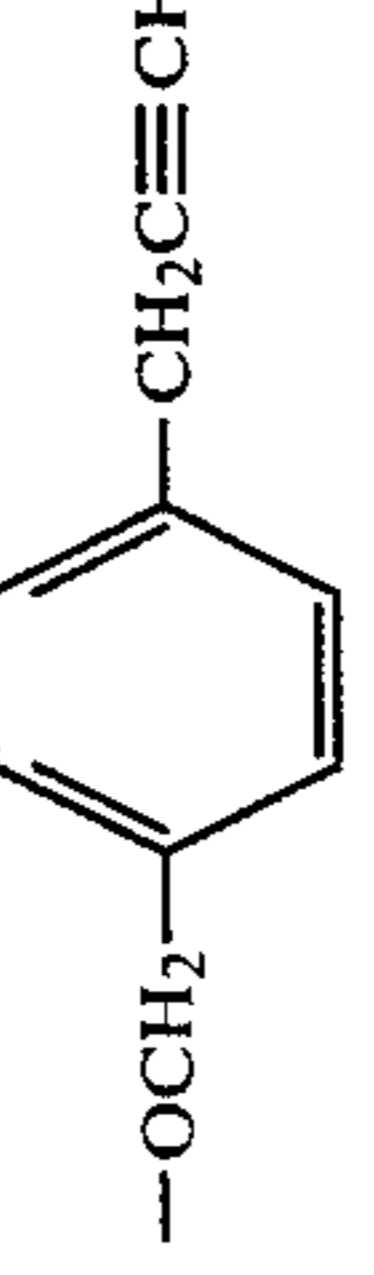
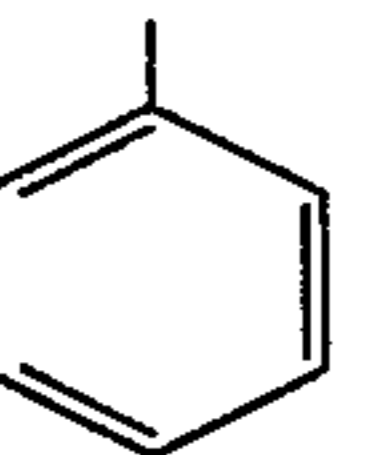
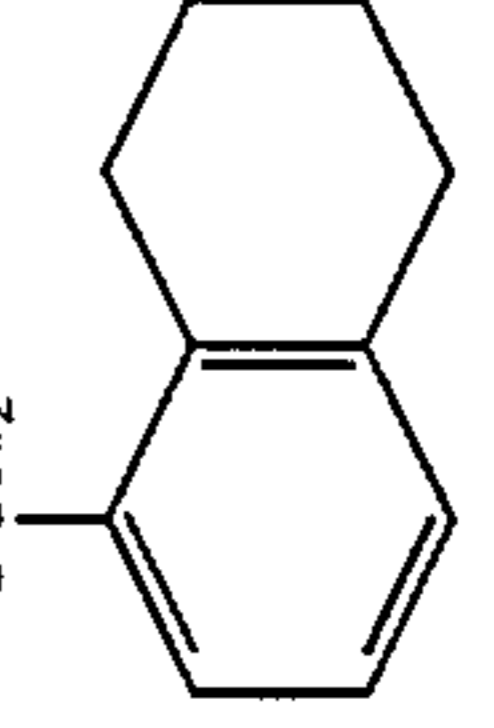
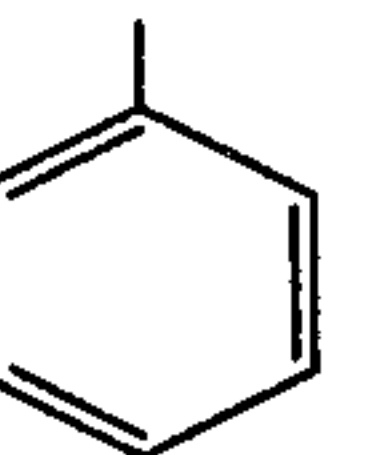
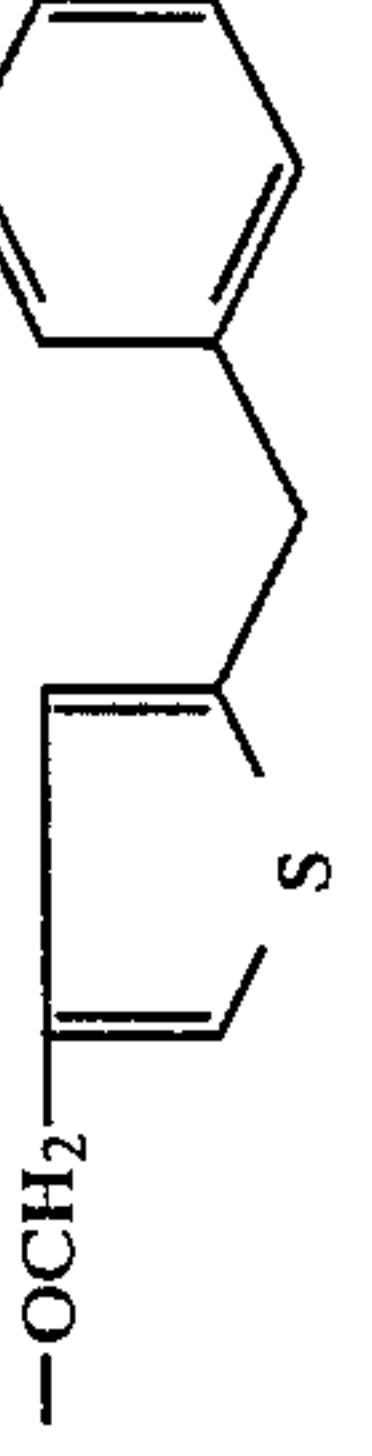
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Exp. No.	Preparation Comp. No.	Z	Y	X	Chemical Structure	Esterification		Physical Properties	
						Method	Yield (%)	Calcd. (%)	Found (%)
1	1	CH ₃ CH ₂ —			b	85	C ₂₂ H ₂₂ O ₃ C: 79.01 H: 6.63	n _D ¹⁷ 1.5545 78.91 6.54	
2	2	CH ₃ CH ₂ —			a	91	C ₂₃ H ₂₂ O ₃ C: 79.74 H: 6.40	n _D ²² 1.5712 79.65 6.41	
3	3	CH ₃ CH ₂ O—			e	80	C ₂₃ H ₂₂ O ₄ C: 76.88 H: 6.12	n _D ²⁴ 1.5275 76.19 6.17	
4	4	CH ₃ CH ₂ —			a	92	C ₁₉ H ₁₇ O ₄ N C: 70.57 H: 5.30 N: 4.33	n _D ²² 1.5403 70.61 5.28 4.25	
5	5	CH ₂ =CH—CH ₂ —			e	75	C ₁₈ H ₁₉ O ₄ N C: 68.99 H: 6.11 N: 4.47	n _D ²² 1.5349 67.01 6.09 4.42	

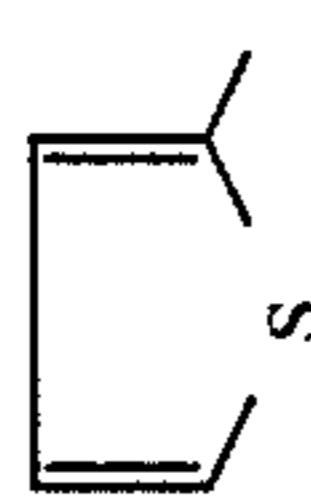
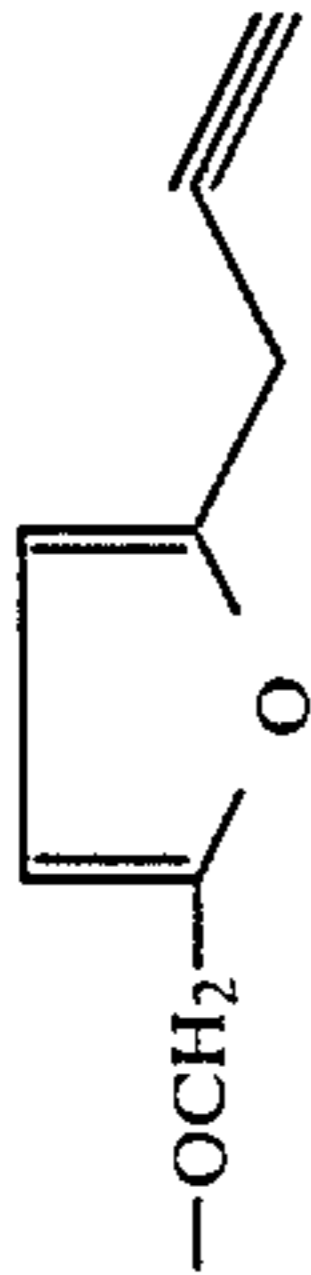
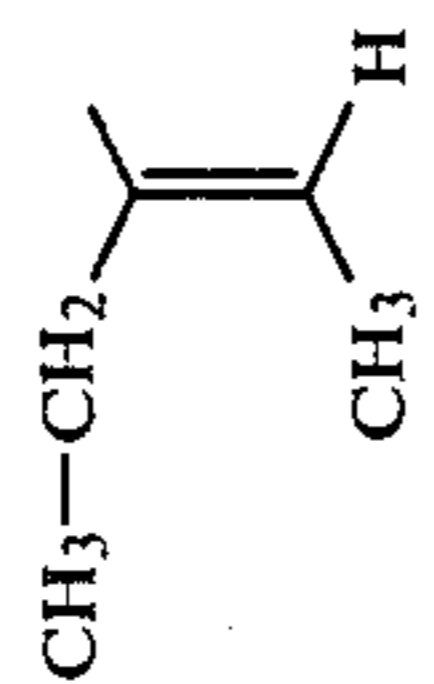
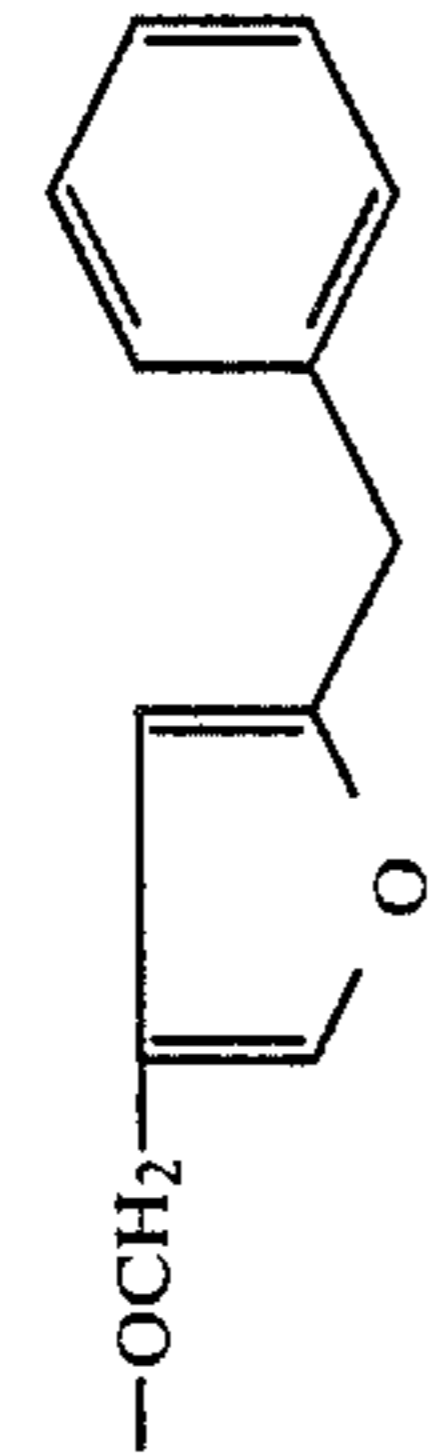
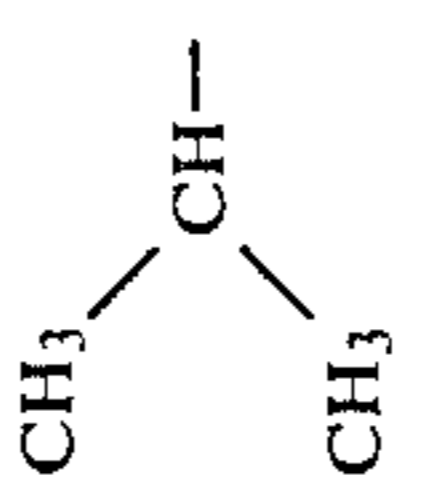
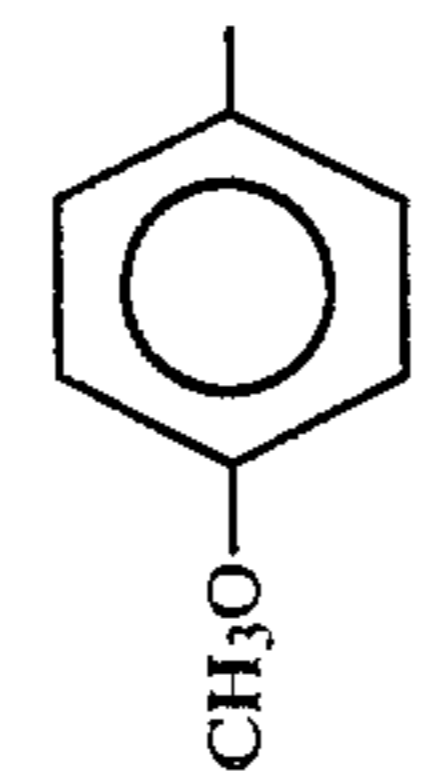
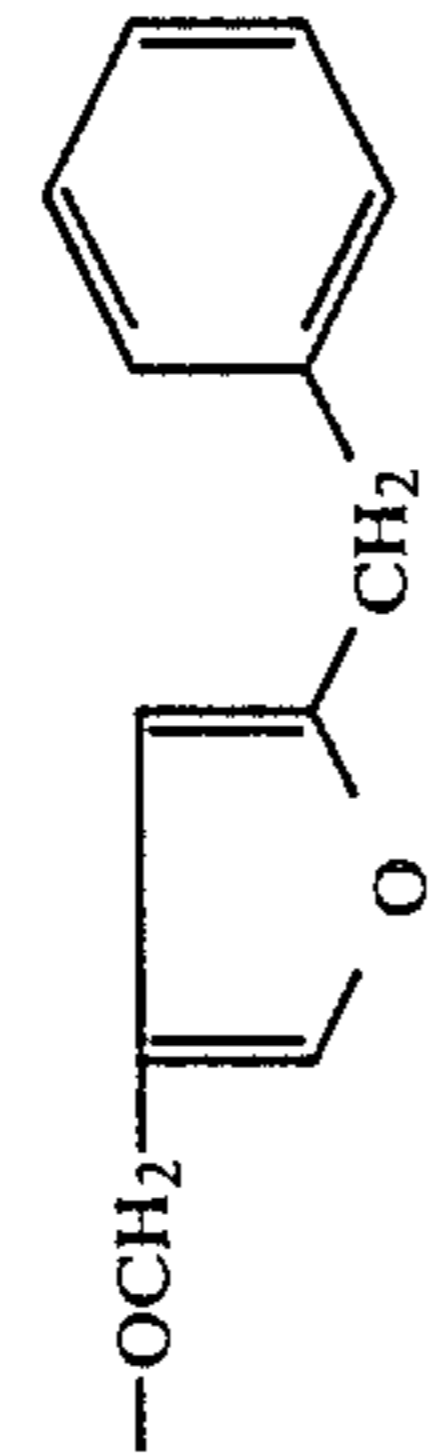
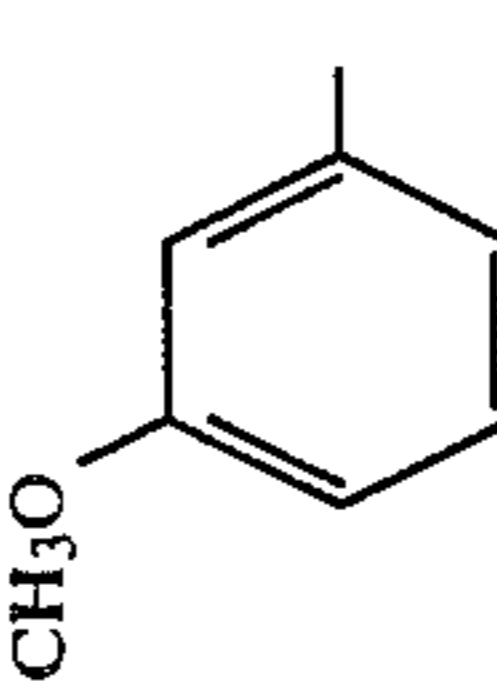
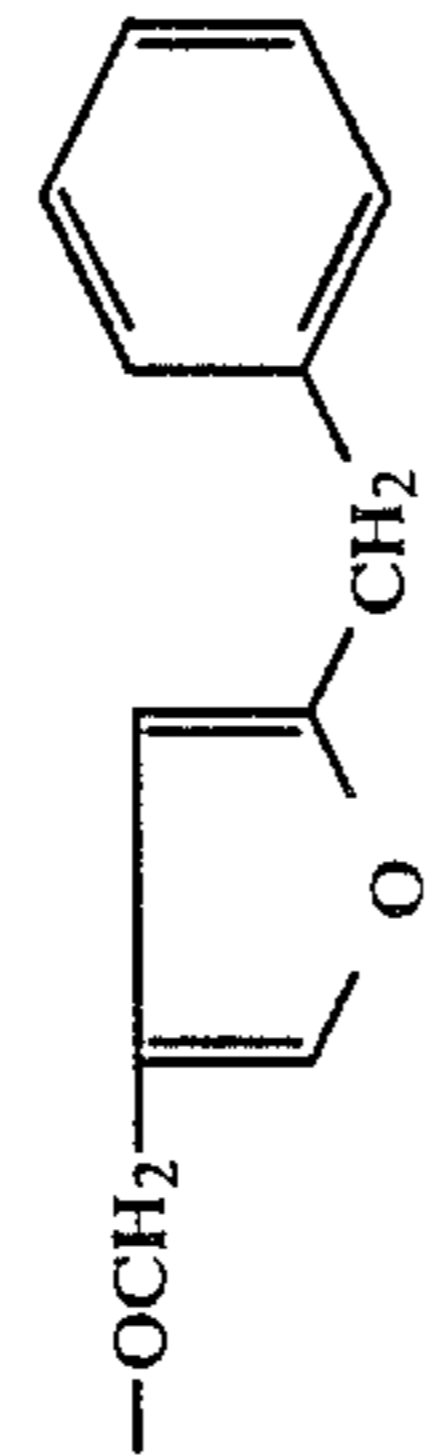
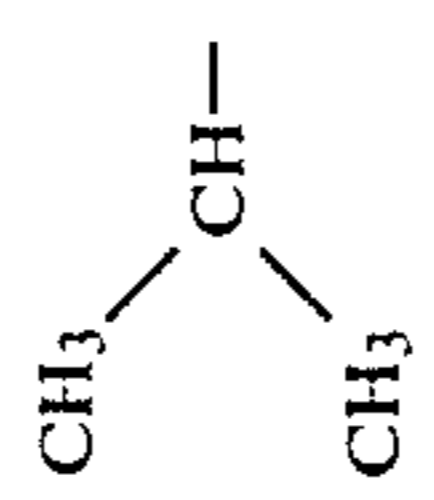
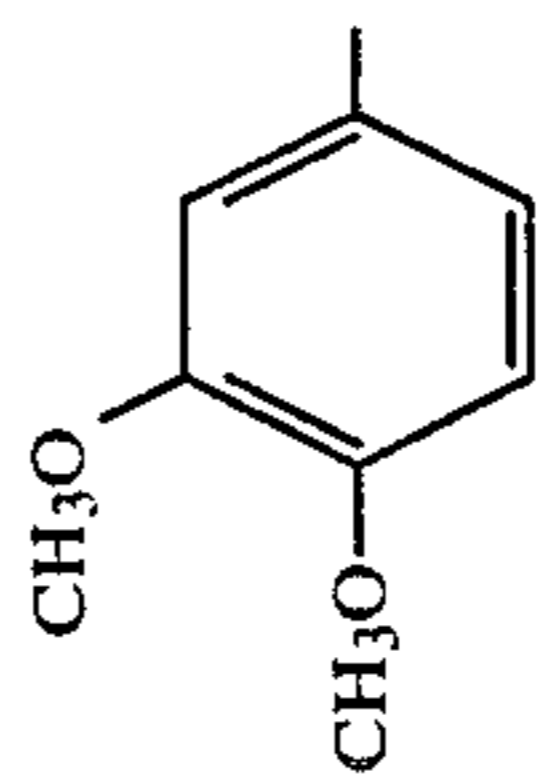
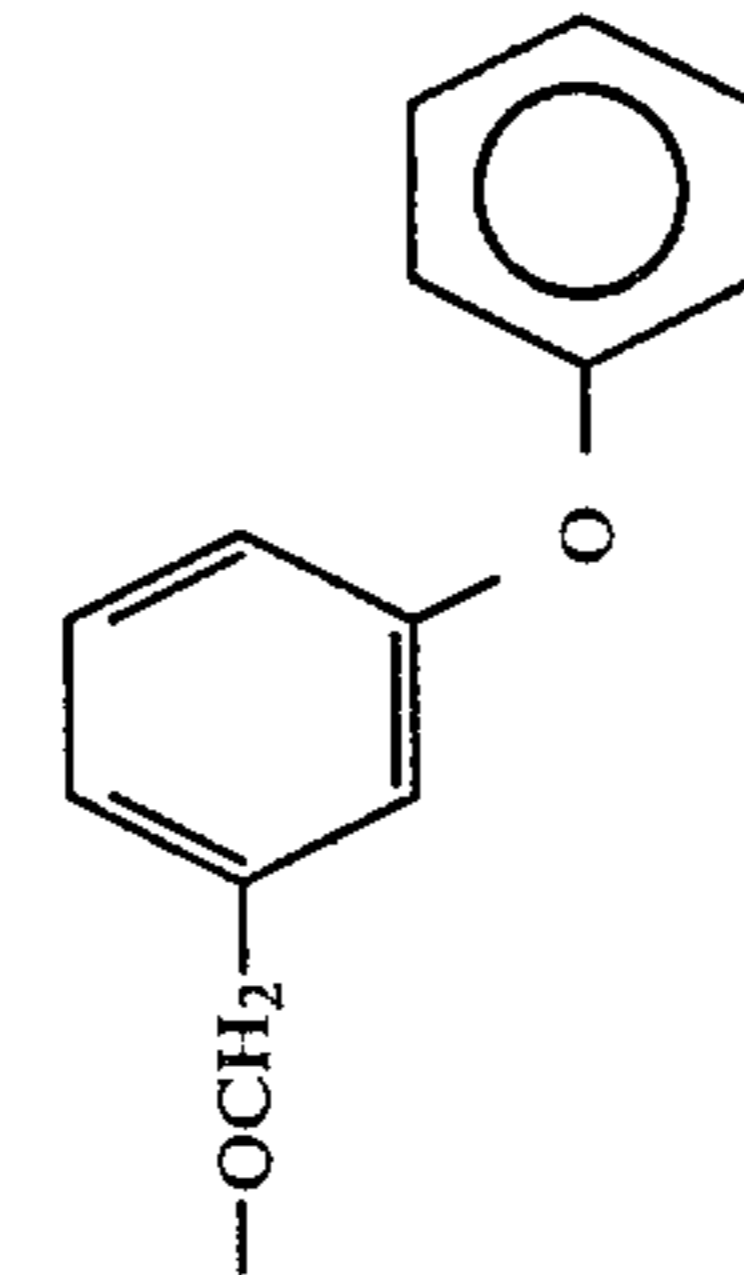
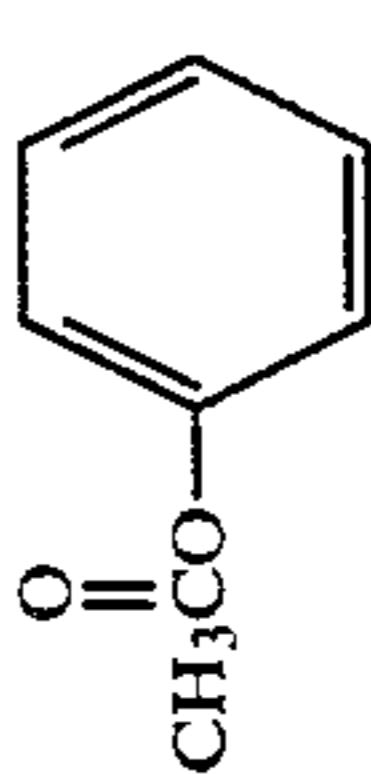
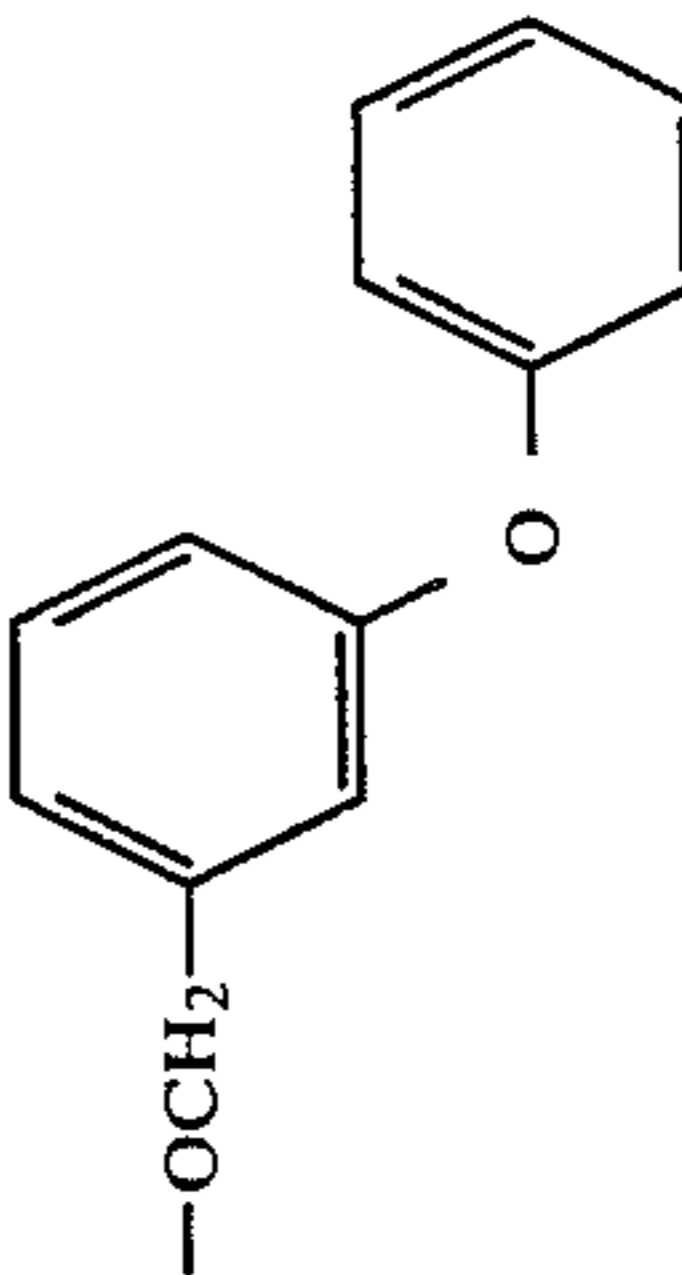
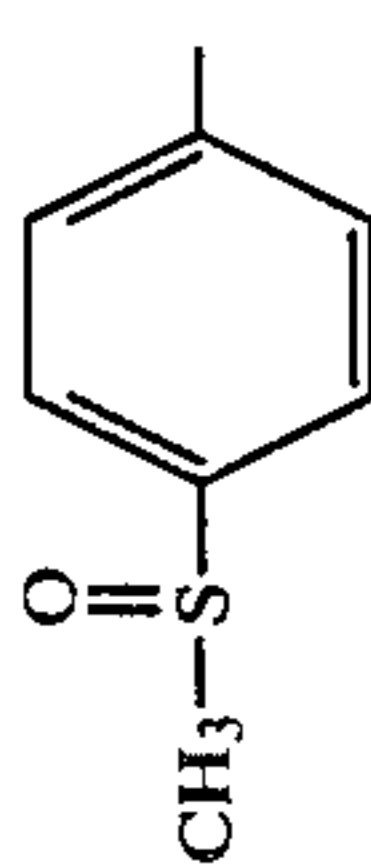

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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
6	6	CH ₃ CH ₂ —			a	81	C ₁₆ H ₁₆ O ₃ S C: 66.64 H: 5.59 S: 11.12	<i>n</i> _D ²² 1.5835 66.59 5.50 11.18
7	7	CH ₃ —			e	85	C ₁₈ H ₂₃ O ₄ N C: 68.12 H: 7.31 N: 4.41	<i>n</i> _D ²³ 1.5648 68.09 7.22 4.50
8	8	CH ₃ CH ₂ —			a	83	C ₂₀ H ₁₈ O ₃ C: 78.41 H: 5.92	<i>n</i> _D ^{24.5} 1.5310 78.3 5.86
9	9	CH ₃ CH ₂ —			a	84	C ₂₂ H ₂₄ O ₄ C: 74.97 H: 6.86	<i>n</i> _D ²² 1.5123 75.06 6.80
10	10	CH ₃ CH ₂ —			a	90	C ₂₀ H ₂₀ O ₂ C: 82.15 H: 6.89	<i>n</i> _D ²³ 1.5039 82.05 6.93
11	11	CH ₃ —			a	93	C ₂₀ H ₂₂ O ₂ C: 81.60 H: 7.53	<i>n</i> _D ²⁴ 1.5536 81.52 7.78
12	12	CH ₃ CH ₂ —			a	91	C ₂₂ H ₂₂ O ₂ S C: 75.39 H: 6.32 S: 9.15	<i>n</i> _D ²² 1.5488 75.29 6.27 9.20

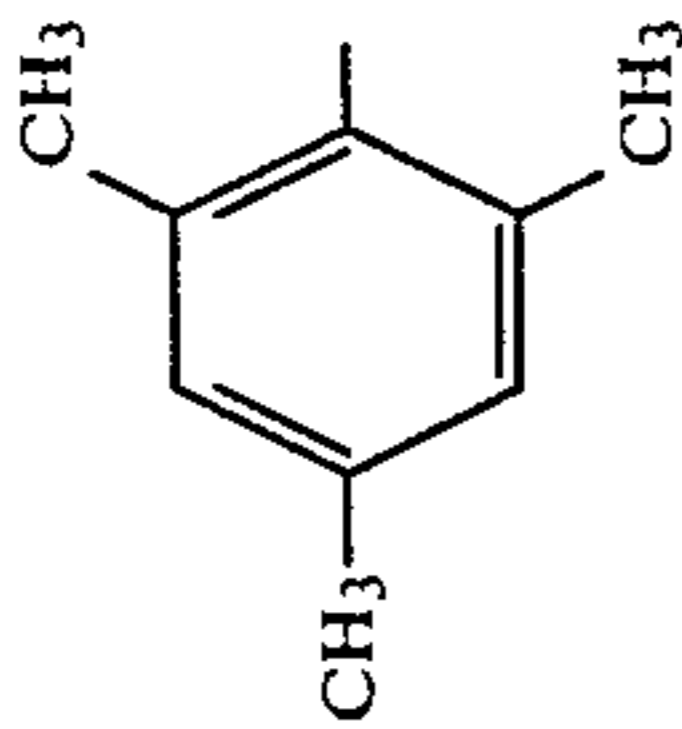
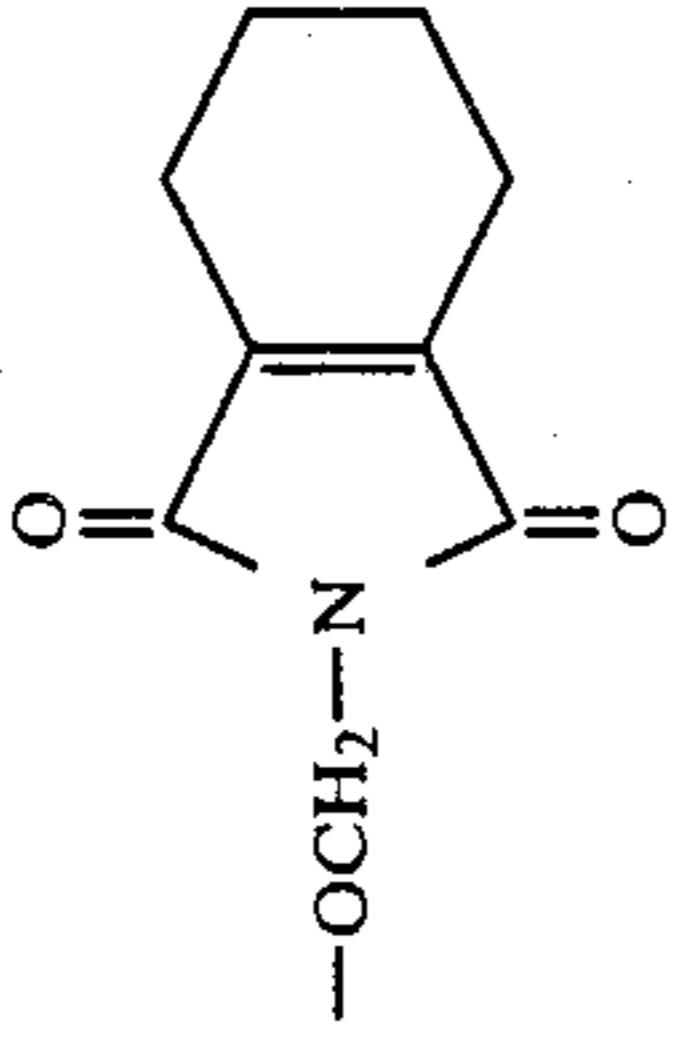
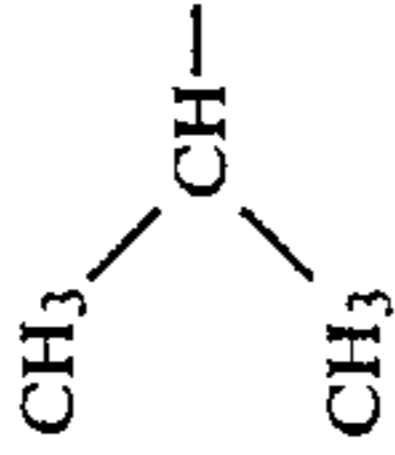
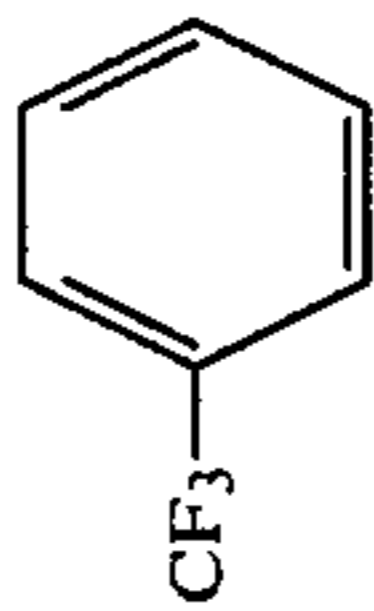
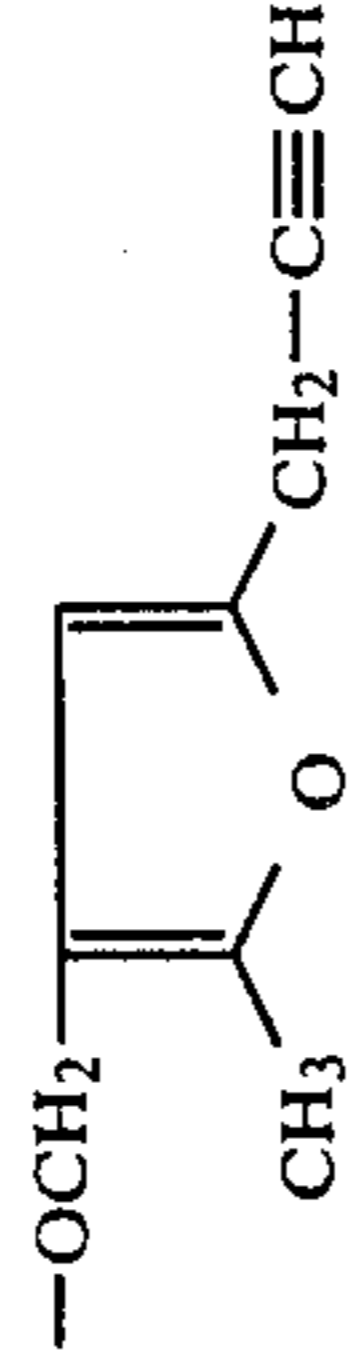
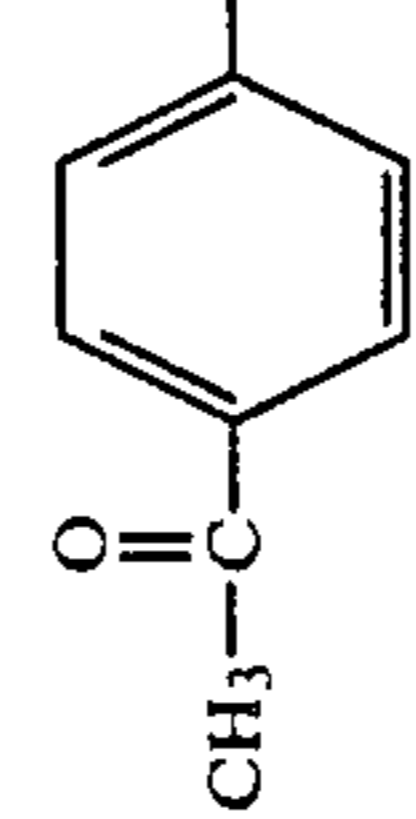
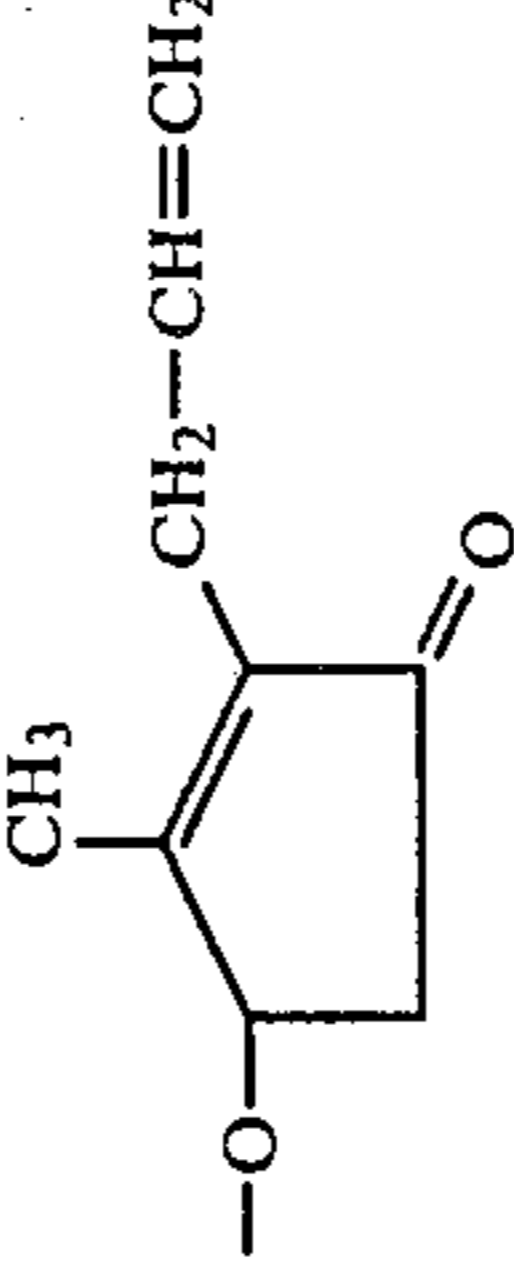
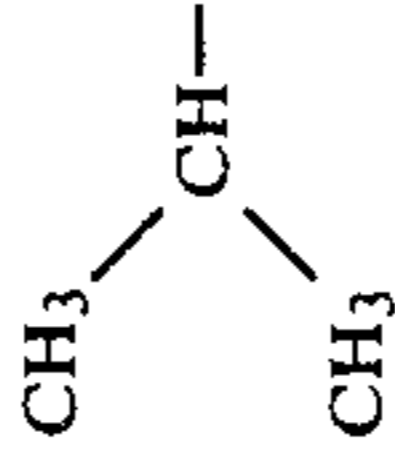
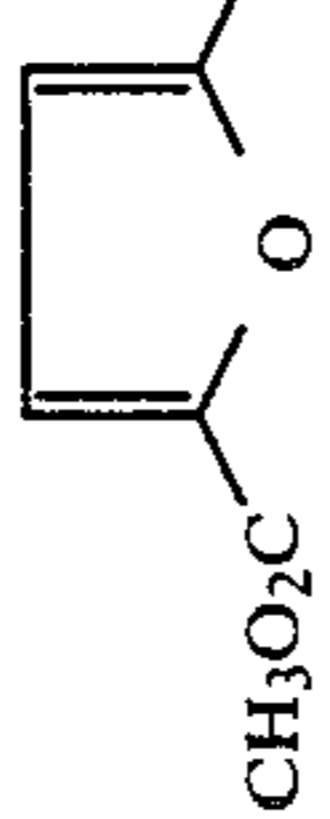
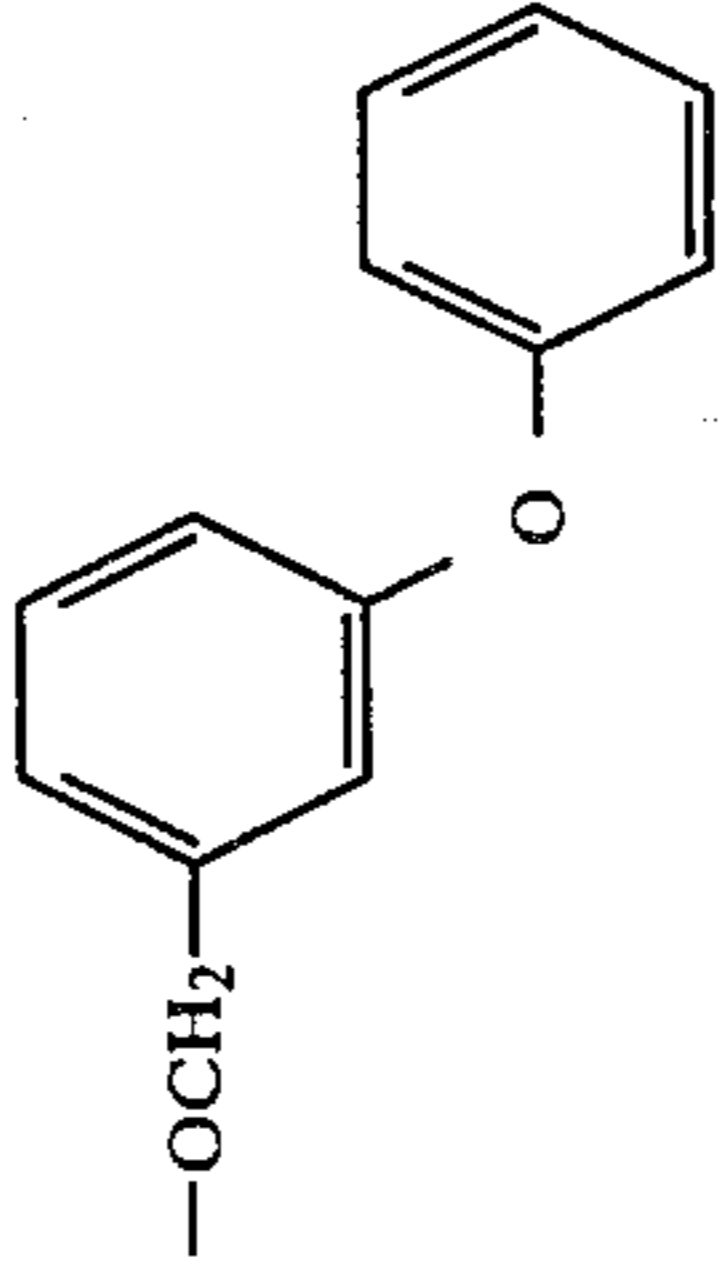
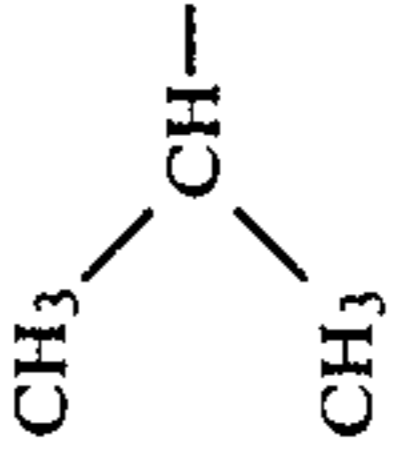
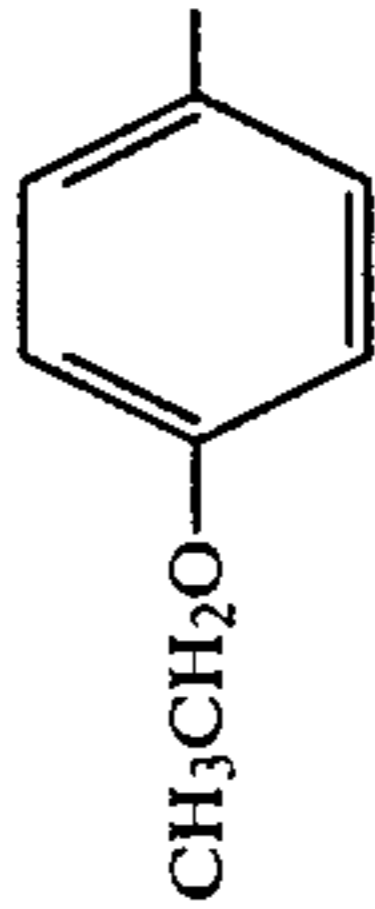
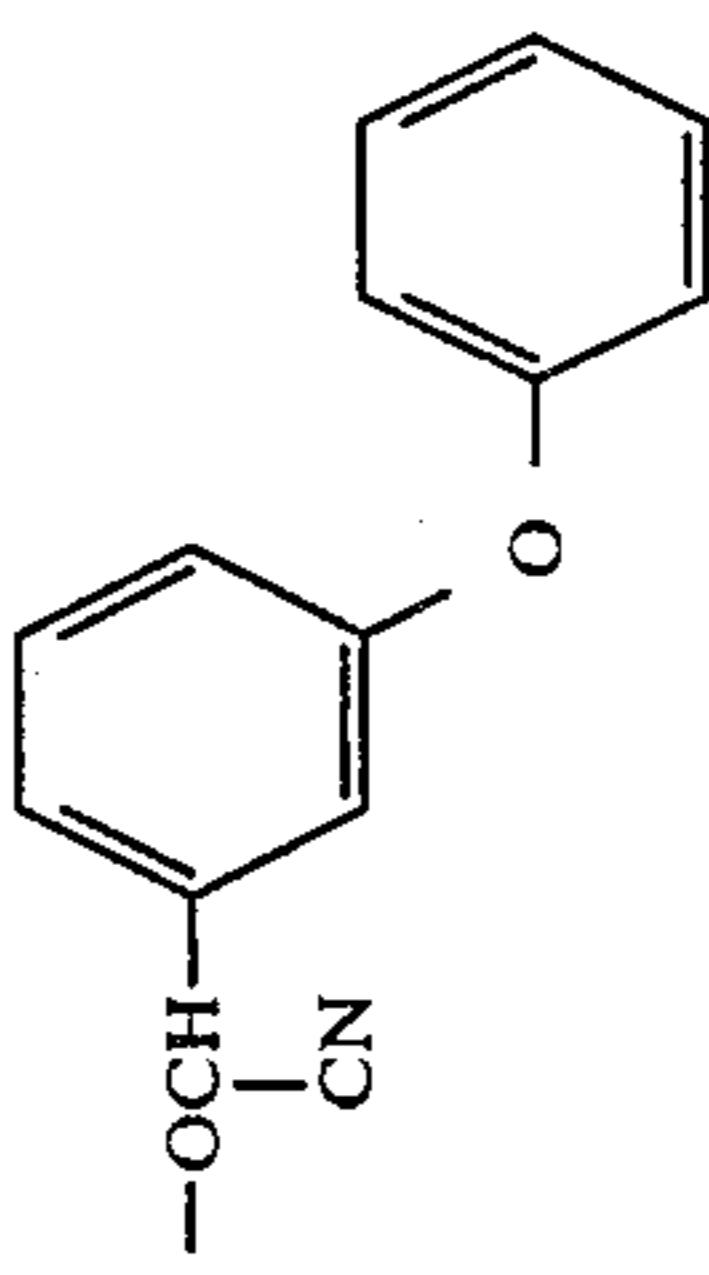
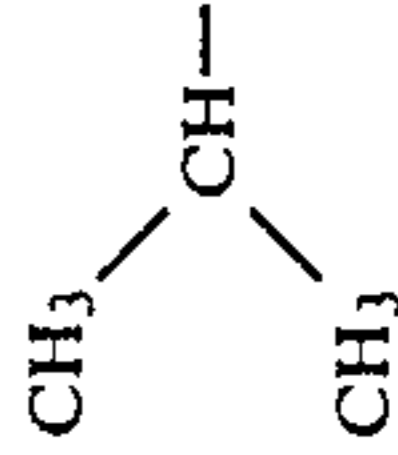
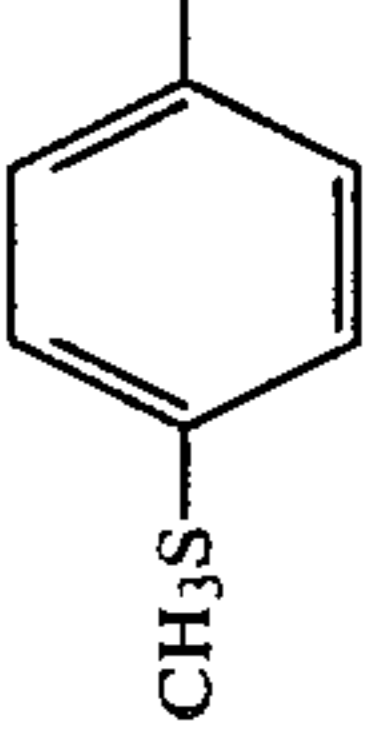
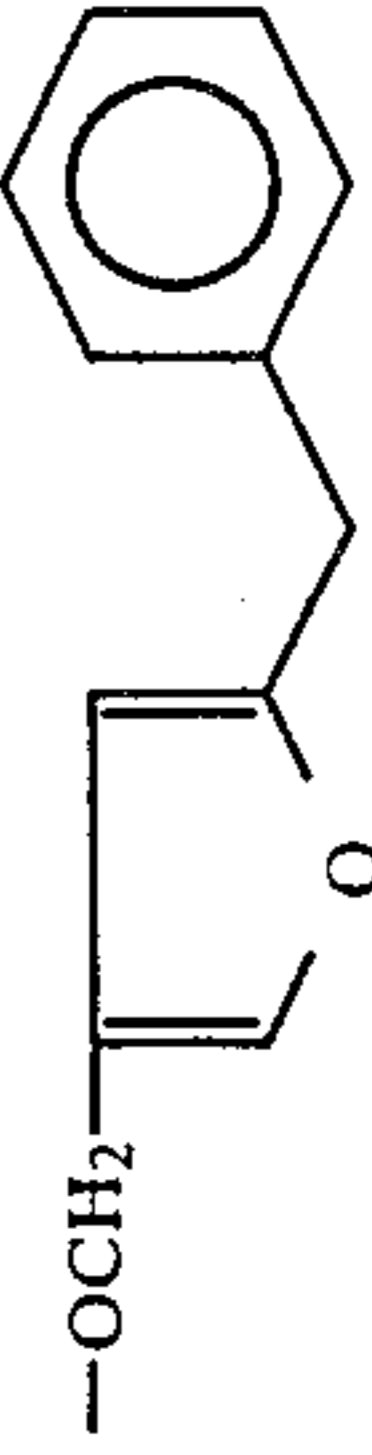
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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
13	13				a	90	C ₂₃ H ₂₄ O ₂ S C: 75.78 H: 6.64 S: 8.79	n_D^{22} 1.5613 75.70 6.69 8.84
14	14	CH ₃ CH ₂ —			a	71	C ₂₄ H ₂₁ O ₃ N C: 77.60 H: 5.70 N: 3.77	n_D^{21} 1.5638 77.49 5.97 3.79
15	15	BrCH ₂ CH ₂ —			a	89	C ₂₂ H ₂₁ O ₃ Br C: 64.08 H: 5.13 Br: 19.38	n_D^{23} 1.5911 64.11 5.11 19.29
16	16	CH ₃ CH ₂ —			a	90	C ₁₇ H ₂₀ O ₃ C: 70.81 H: 6.99	n_D^{22} 1.5118 70.89 6.77
17	17				d	72	C ₂₆ H ₂₈ O ₃ C: 80.38 H: 7.27	mp. 51–53° C. 80.41 7.24
18	18	CH ₃ CH ₂ —			a	84	C ₂₁ H ₂₆ O ₃ C: 77.27 H: 8.03	n_D^{23} 1.5417 77.41 7.96
19	19				a	80	C ₂₁ H ₂₂ O ₃ C: 78.23 H: 6.88	n_D^{24} 1.5632 78.19 6.94

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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
20	20	$\text{CH}_2=\text{CH}-\text{CH}_2-$			a	90	$\text{C}_{17}\text{H}_{16}\text{O}_3\text{S}$ C: 67.98 H: 5.37 S: 10.64	n_D^{22} 1.5274 68.07 5.36 10.64
21	21	CH_3-			a	88	$\text{C}_{20}\text{H}_{24}\text{O}_3$ C: 76.89 H: 7.74	n_D^{21} 1.5693 76.80 7.71
22	22				a	88	$\text{C}_{24}\text{H}_{26}\text{O}_4$ C: 76.16 H: 6.93	n_D^{17} 1.5470 76.20 6.89
23	23	CH_3CH_2-			b	83.2	$\text{C}_{23}\text{H}_{24}\text{O}_4$ C: 75.80 H: 6.64	n_D^{25} 1.5481 75.71 6.69
24	24				e	79.8	$\text{C}_{26}\text{H}_{28}\text{O}_5$ C: 74.26 H: 6.71	n_D^{25} 1.5655 74.25 6.77
25	25	CH_3CH_2-			e	81.5	$\text{C}_{25}\text{H}_{24}\text{O}_5$ C: 74.24 H: 5.98	n_D^{25} 1.5621 74.11 6.09
26	26	CH_3CH_2-			a	90.1	$\text{C}_{23}\text{H}_{24}\text{O}_4\text{S}$ C: 69.67 H: 6.10 S: 8.09	n_D^{25} 1.5448 69.55 6.11 8.11

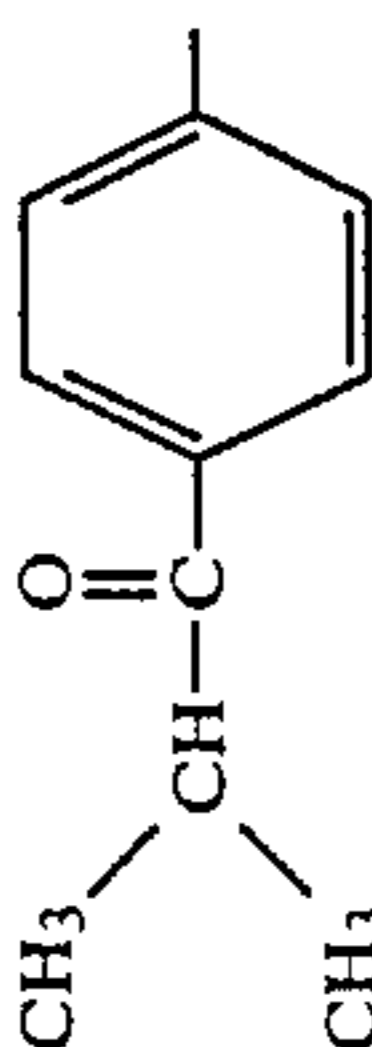
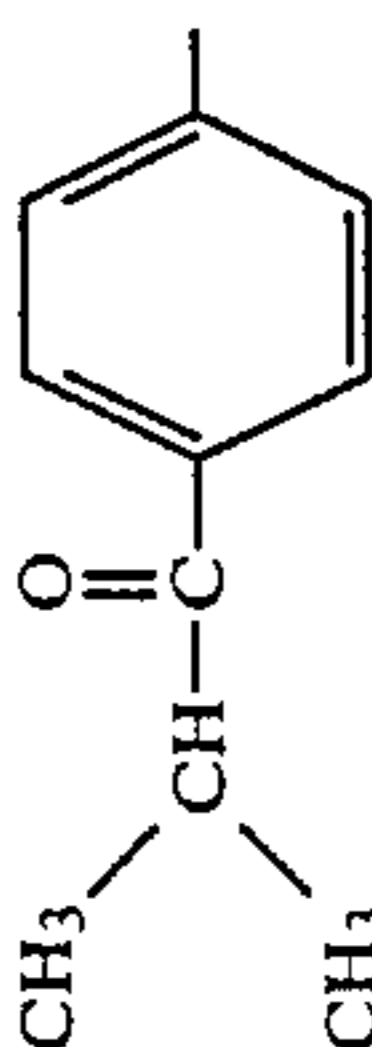
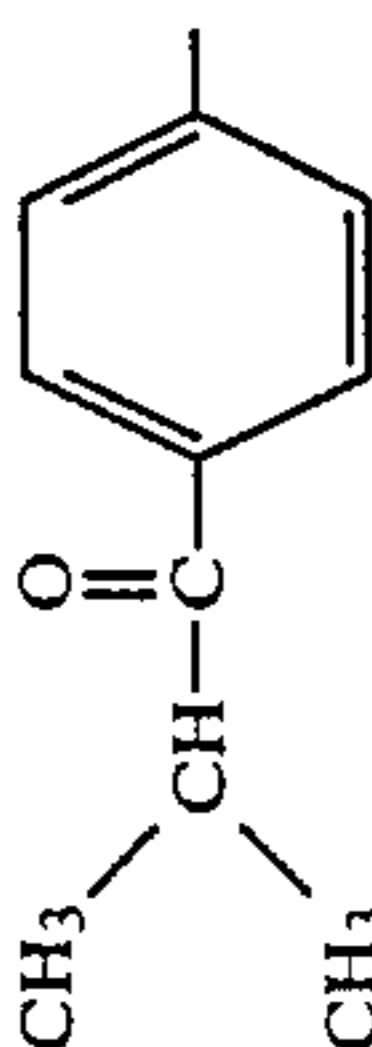
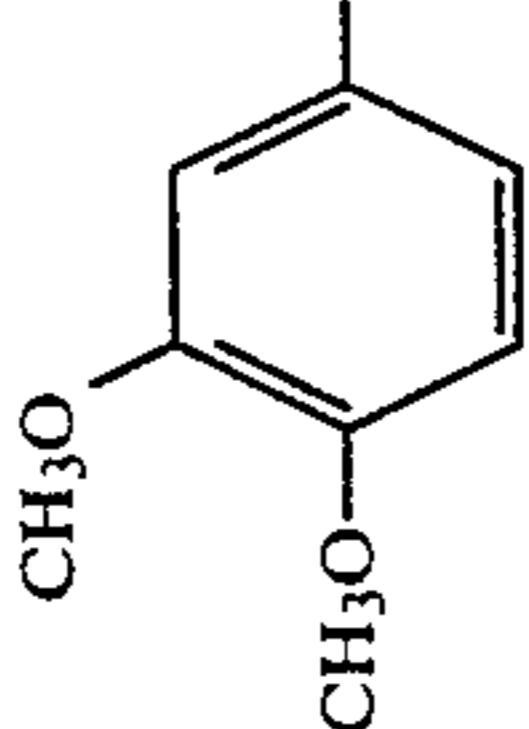
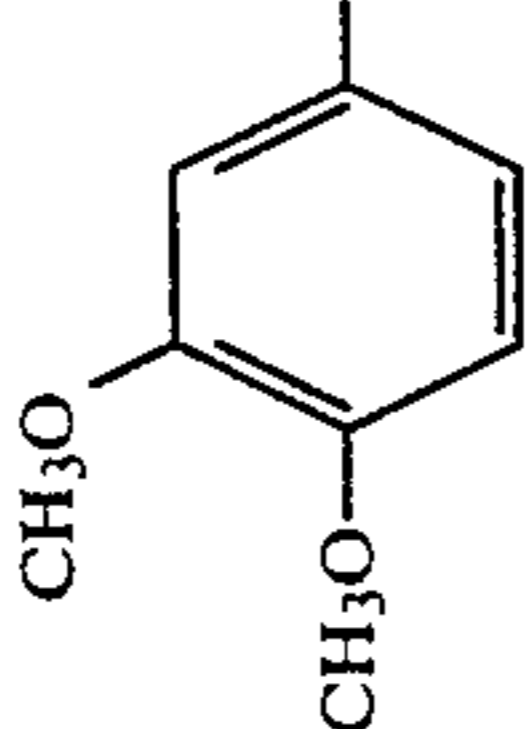
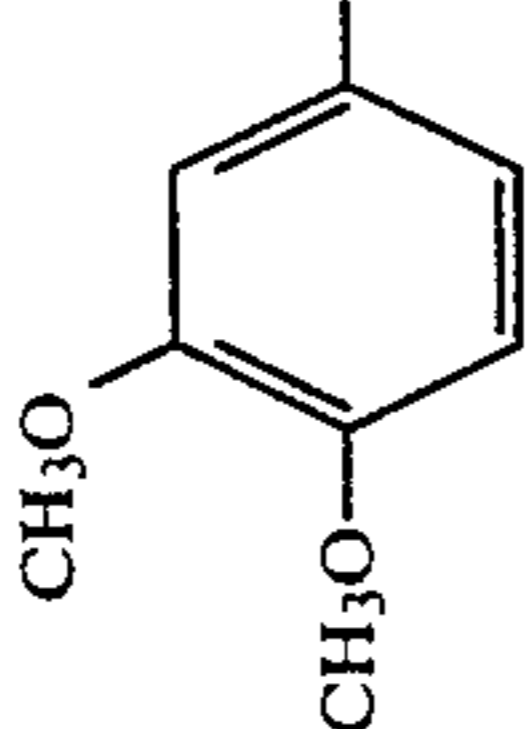
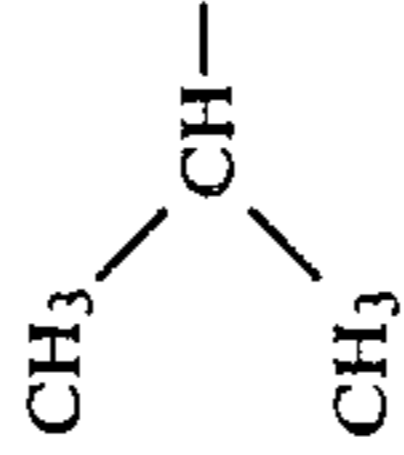
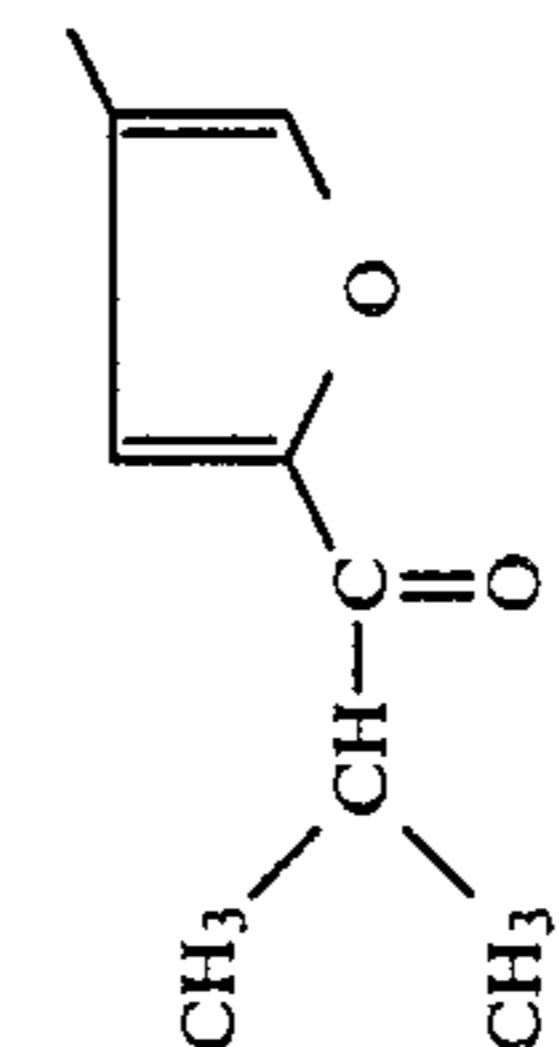
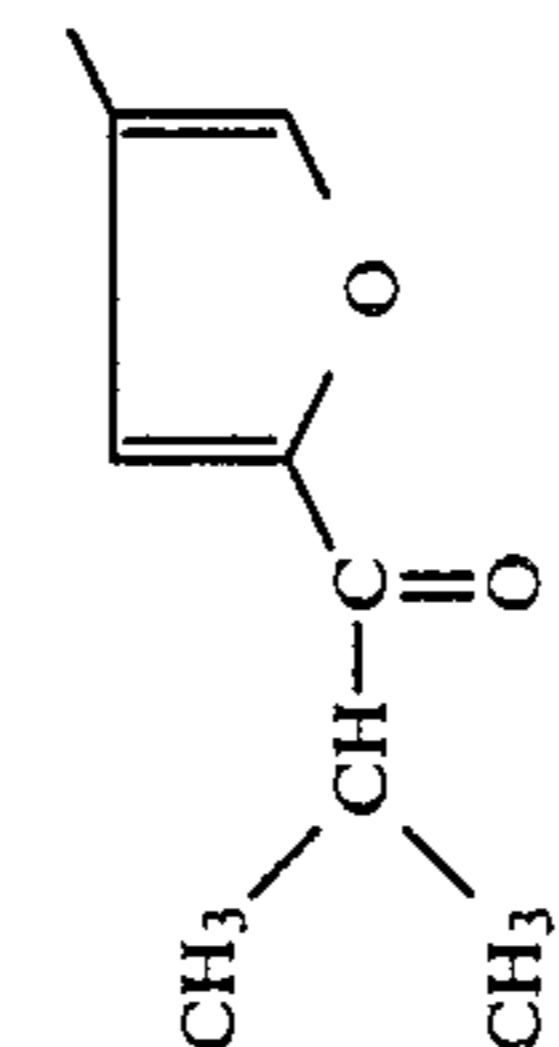
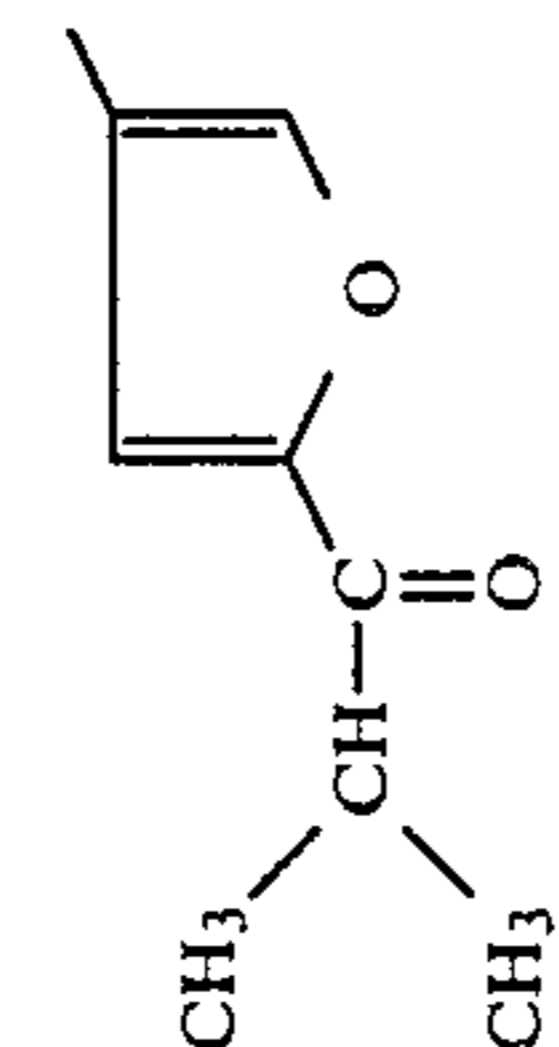
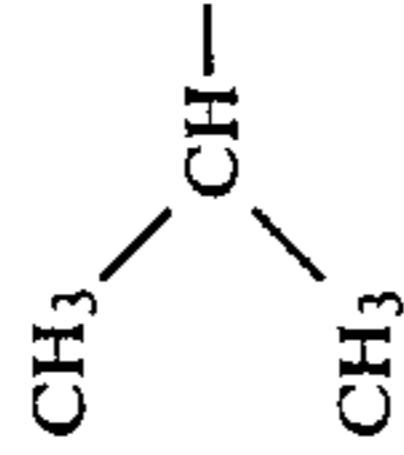
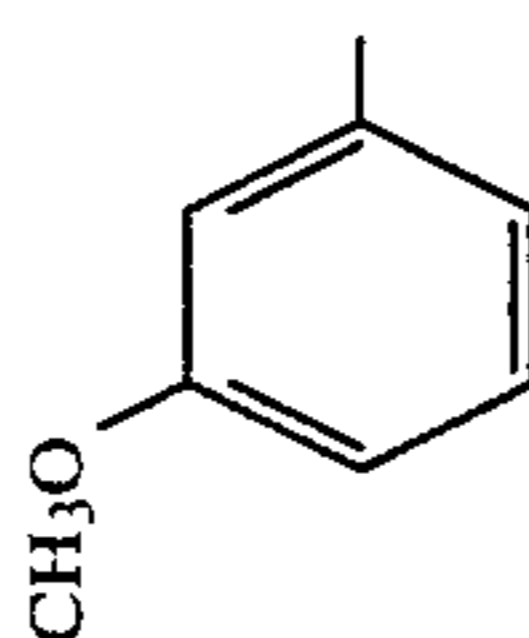
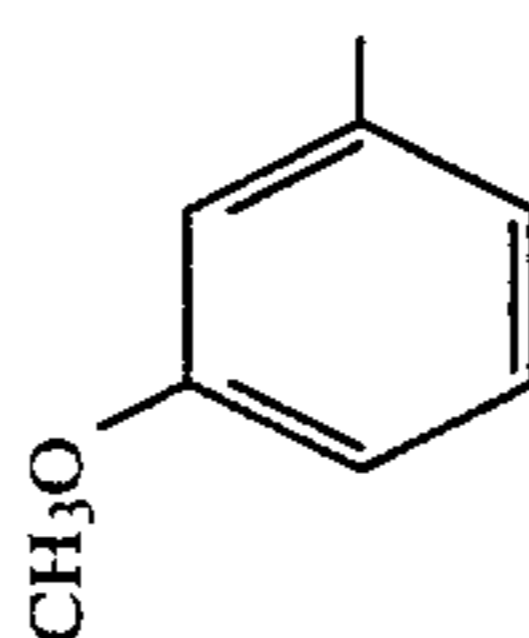
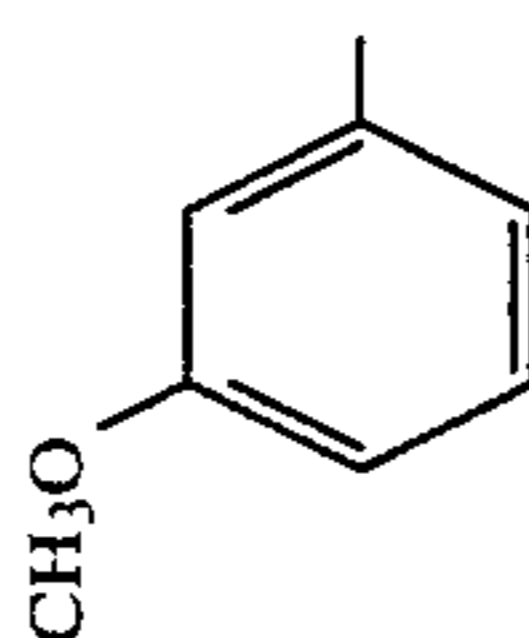
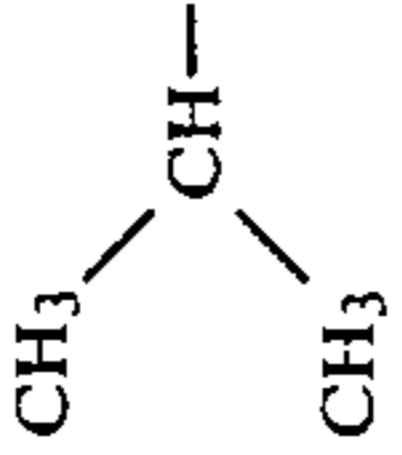
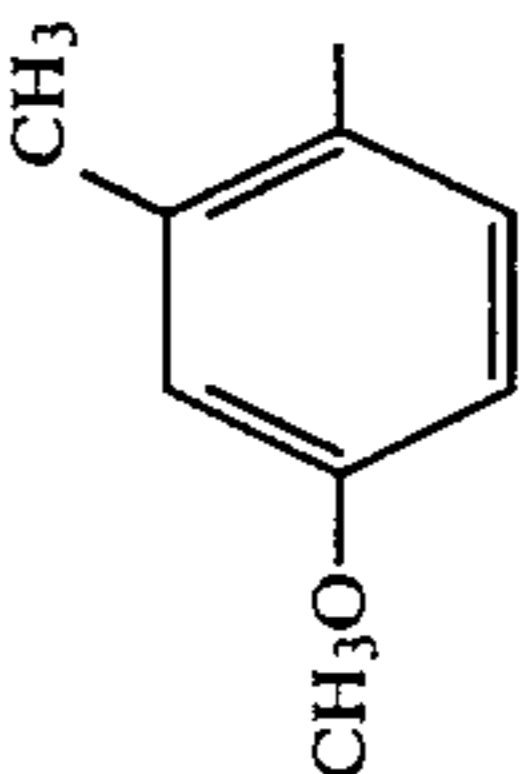
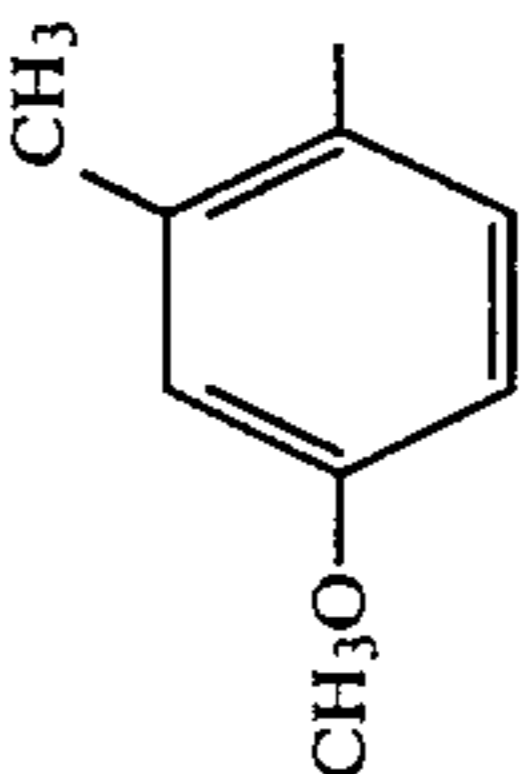
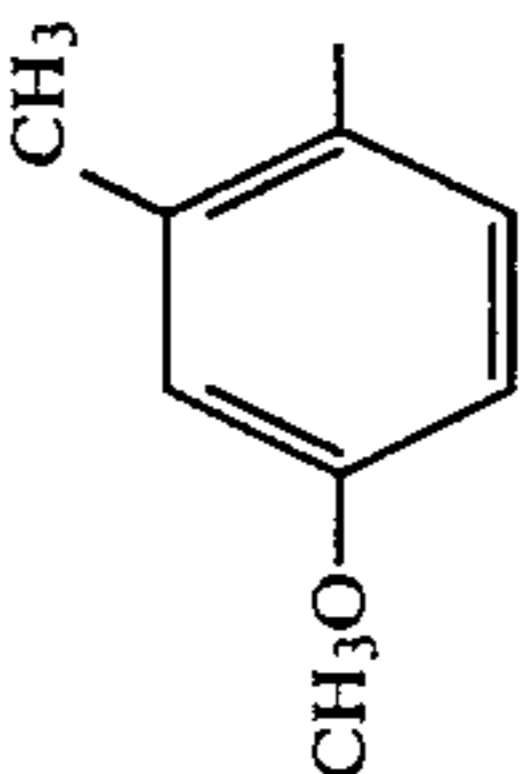
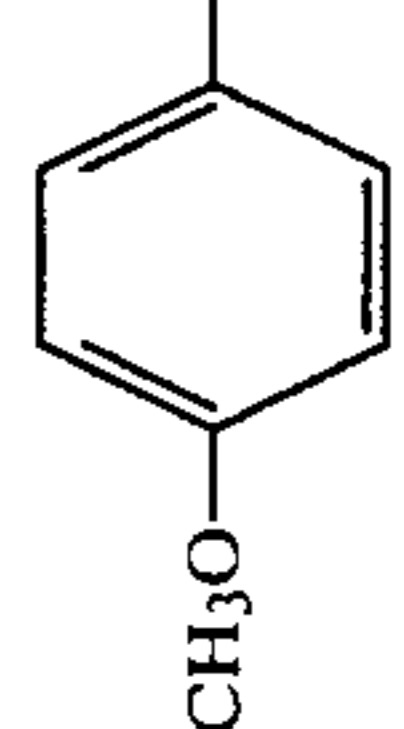
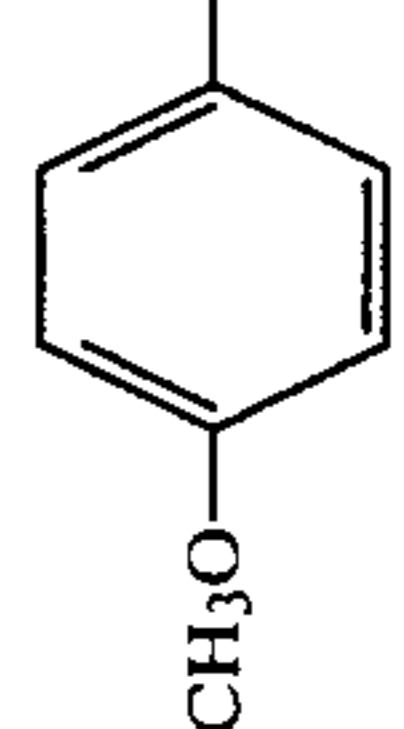
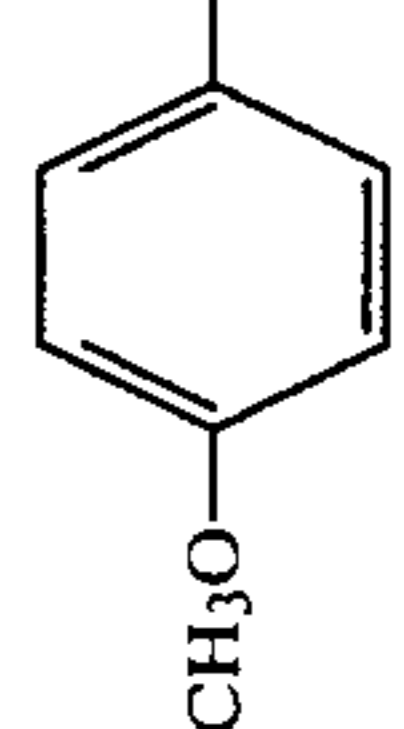
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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
27	27	CH ₃ CH ₂ —			c	61.2	C ₂₂ H ₂₇ O ₄ N C: 71.52 H: 7.37 N: 3.79	<i>n</i> _D ²⁵ 1.5399 71.60 7.29 3.85
28	28				a	92.2	C H ₂₁ O ₃ F ₃ C: 66.66 H: 5.59 F: 15.06	<i>n</i> _D ²⁵ 1.5368 66.61 5.62 15.11
29	30	CH ₂ =CH—CH ₂ —			a	82.5	C ₂₂ H ₂₄ O ₄ C: 74.97 H: 6.86	<i>n</i> _D ²⁵ 1.5566 75.02 6.04
30	33				e	81.8	C ₂₄ H ₂₄ O ₆ C: 70.57 H: 5.92	<i>n</i> _D ²⁵ 1.5367 70.56 5.89
31	34				a	92.5	C ₂₇ H ₂₇ O ₄ N C: 75.50 H: 6.34 N: 3.26	<i>n</i> _D ²⁵ 1.5208 75.19 6.30 3.31
32	35				a	95.3	C ₂₄ H ₂₆ O ₃ S C: 73.06 H: 6.64 S: 8.13	<i>n</i> _D ²⁵ 1.5335 73.10 6.64 8.15

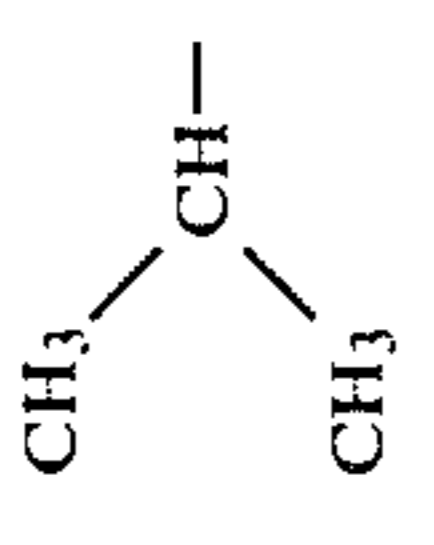
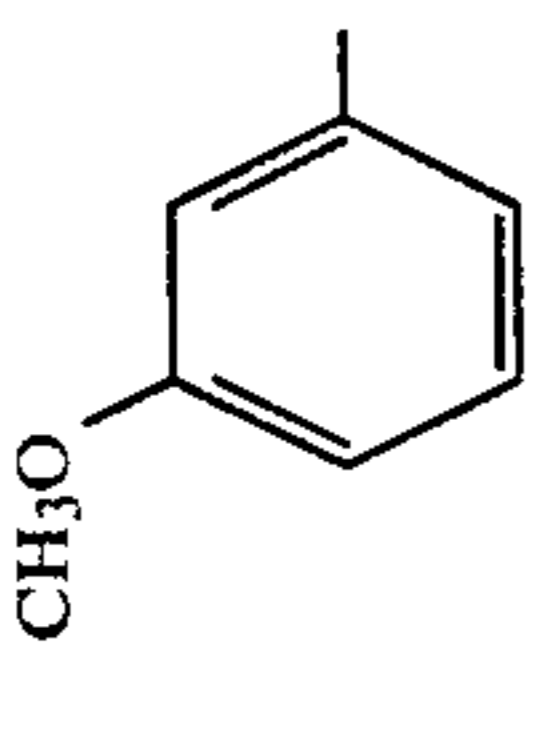
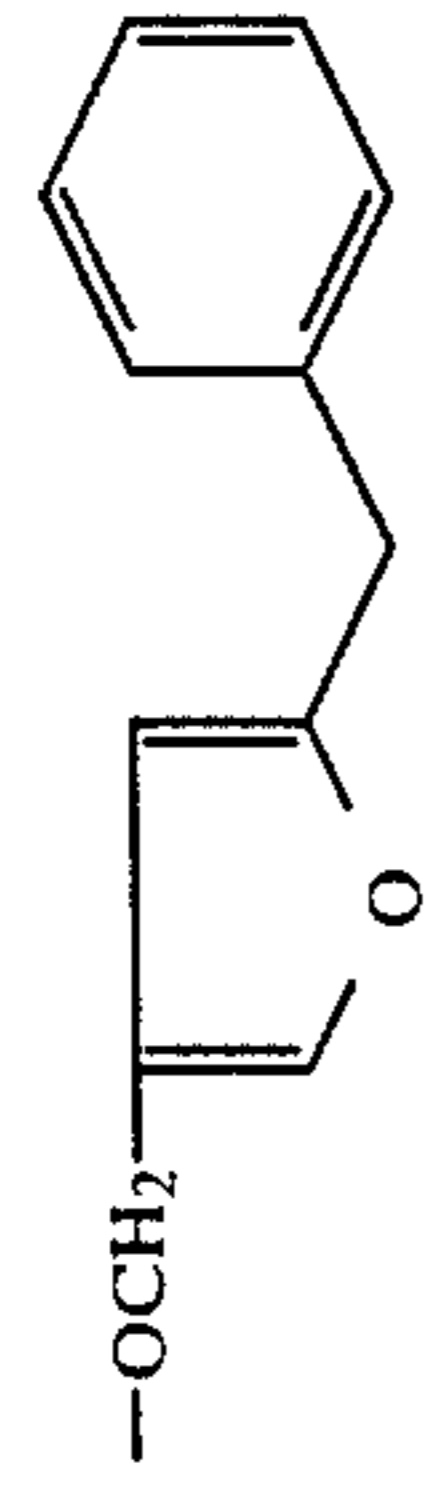
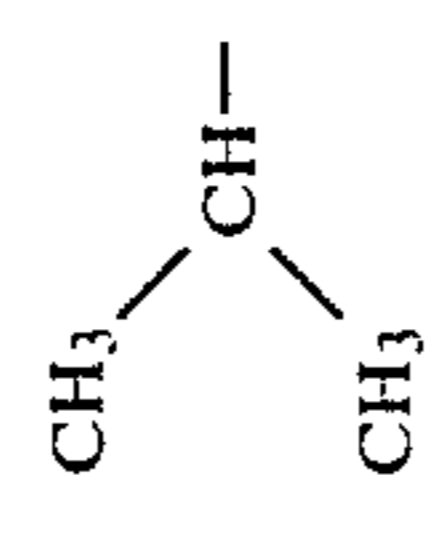
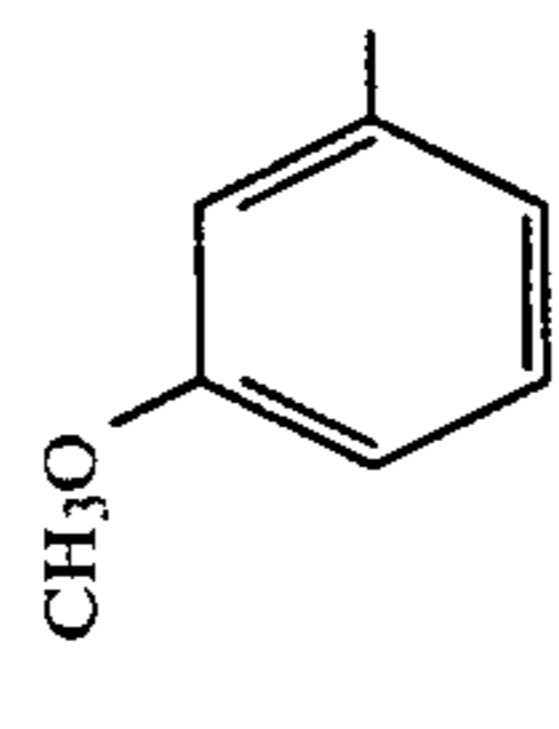
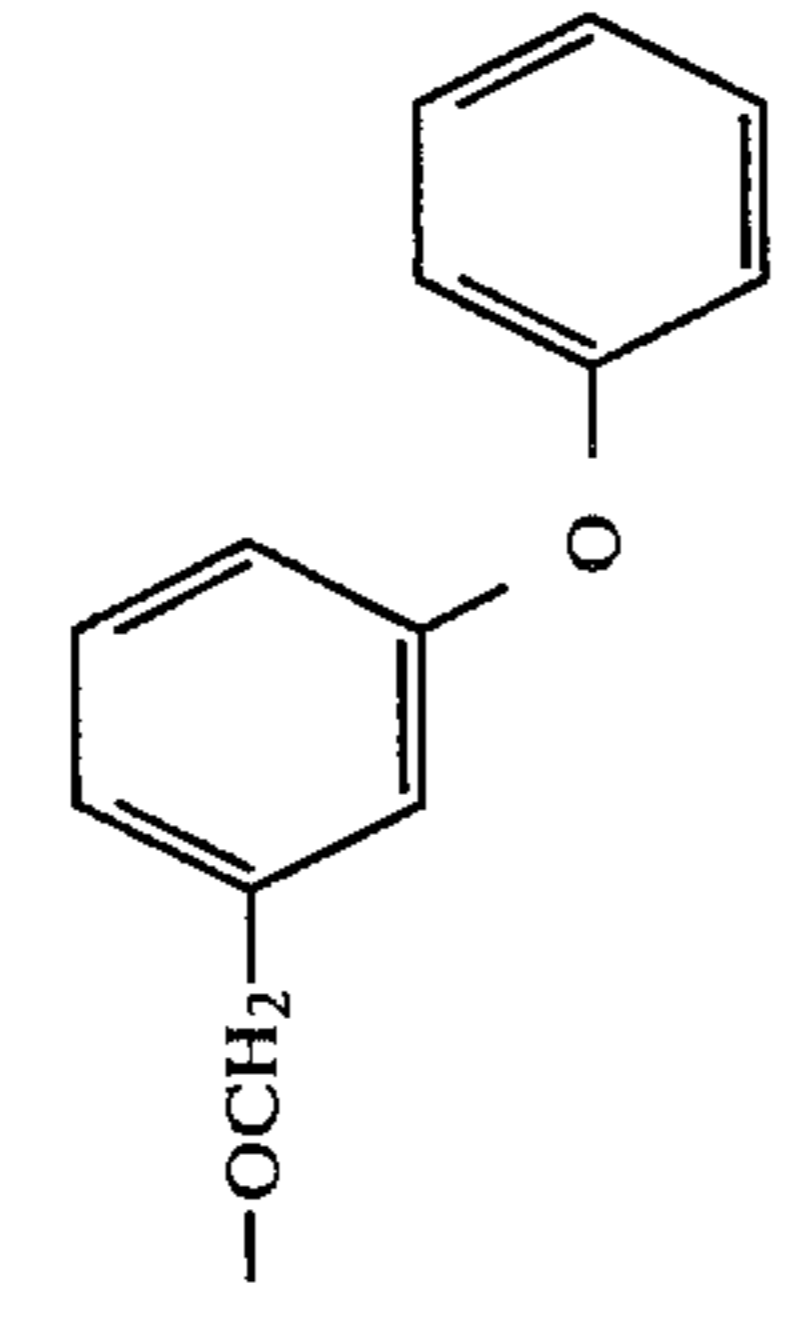

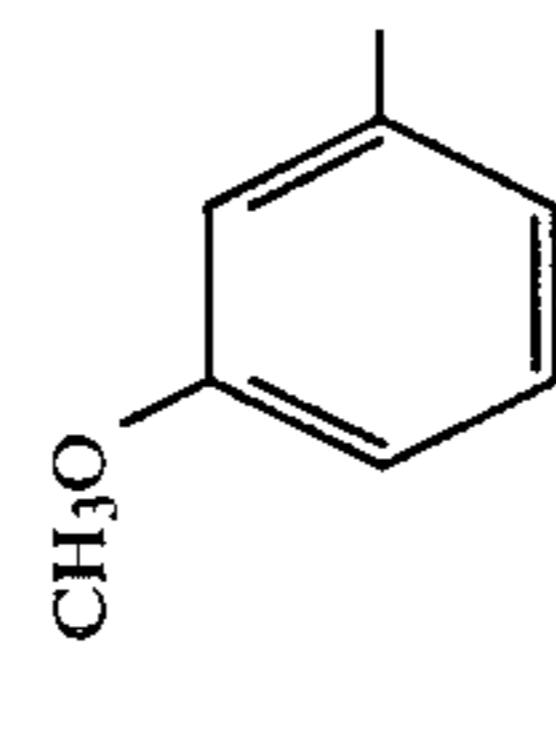
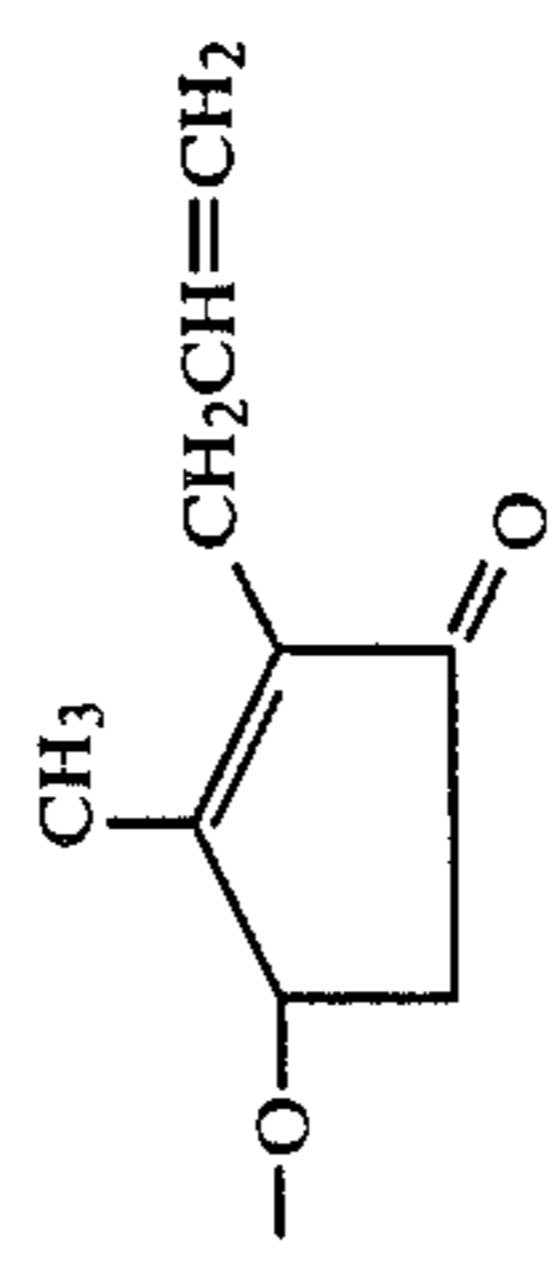
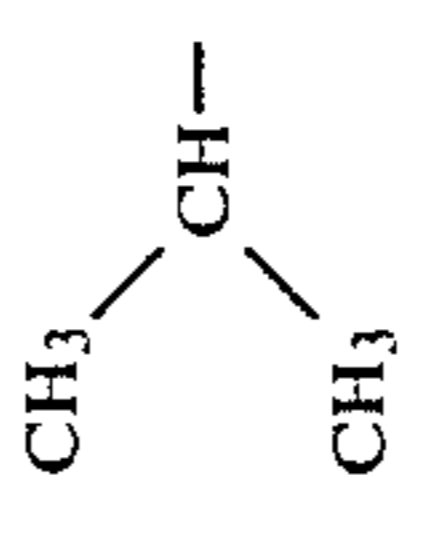
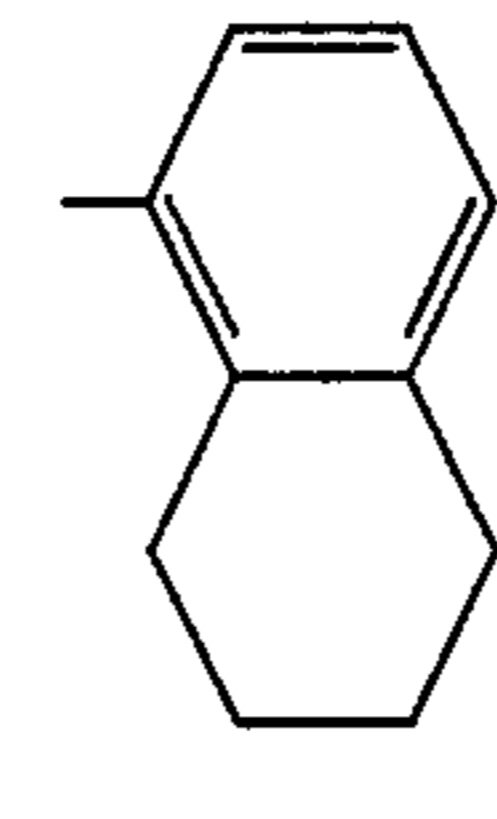
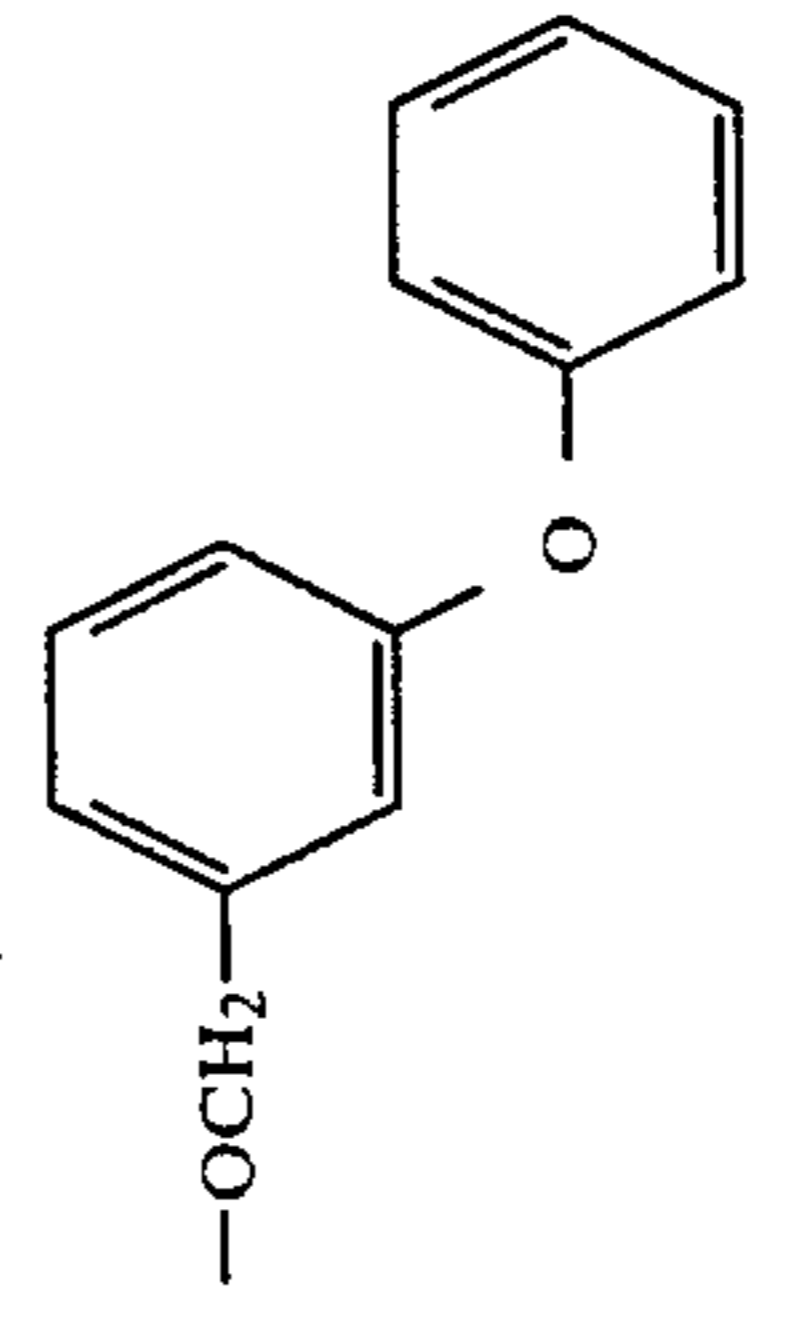

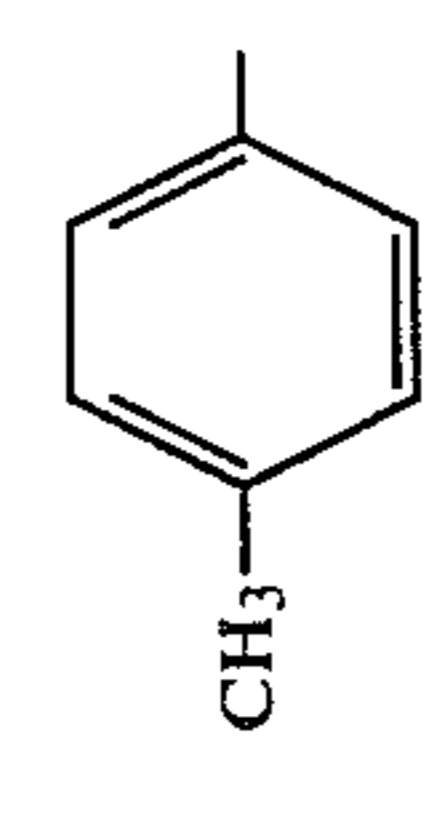
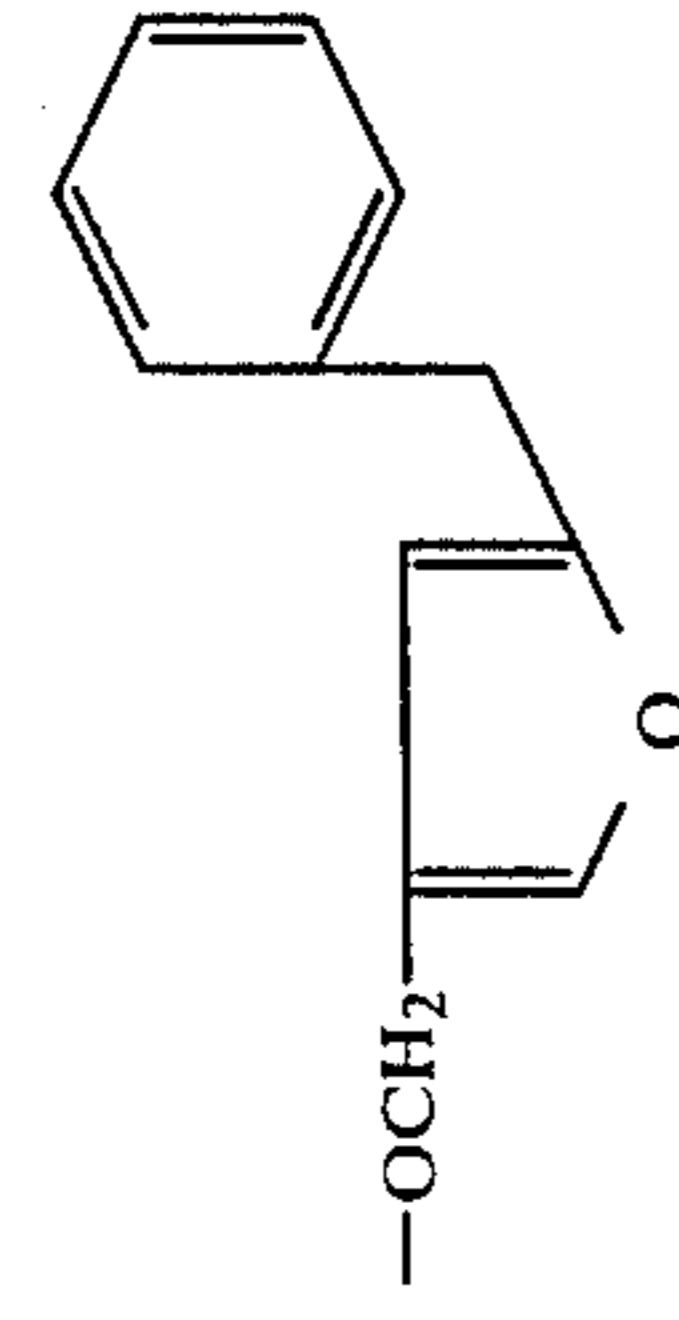
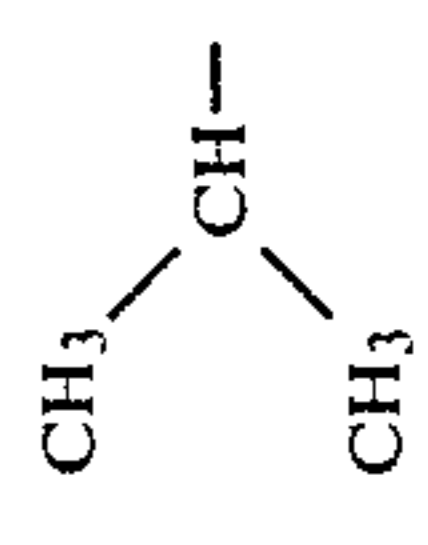
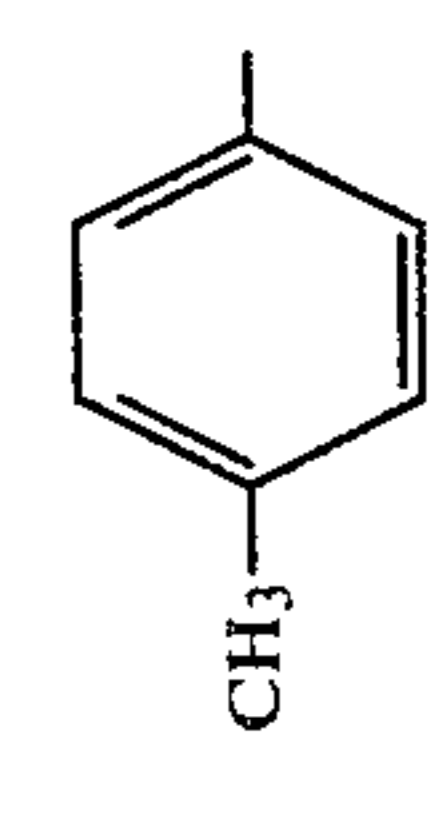
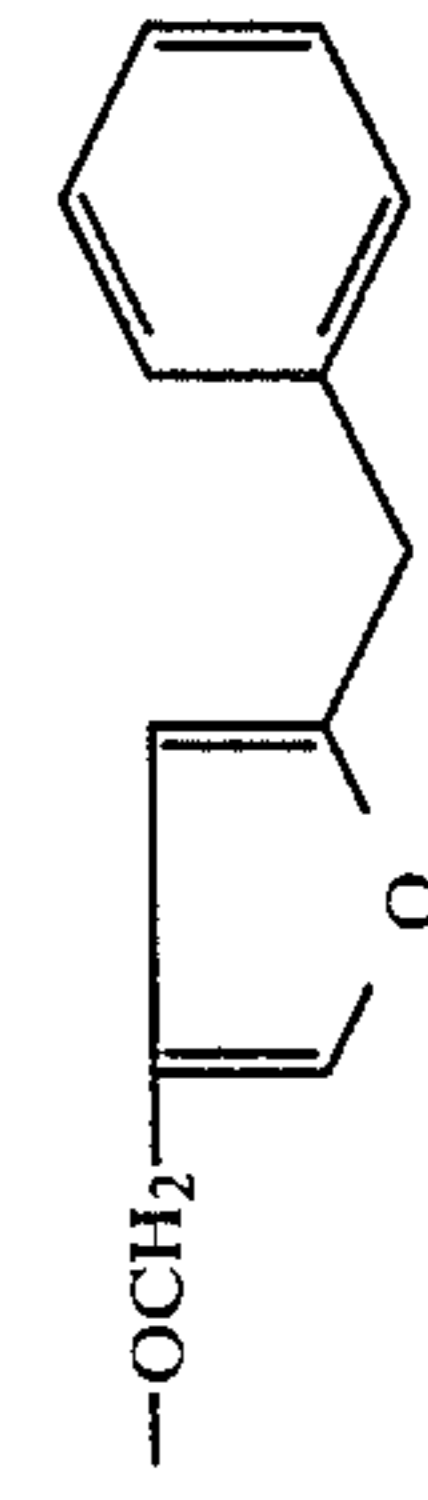
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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
33	37				a	73.6	C ₂₂ H ₂₀ O ₅ C: 72.51 H: 5.53	n_D^{25} 1.5236 72.55 5.61
34	36				d	86.2	C ₂₇ H ₃₀ O ₃ C: 80.56 H: 7.51	80.61 7.46
35	45				e	78.1	C ₂₆ H ₂₄ O ₃ C: 81.22 H: 6.29	81.20 6.31
36	48				a	92.7	C ₂₃ H ₂₅ O ₅ N C: 69.85 H: 6.37 N: 3.54	n_D^{25} 1.5221 69.78 6.39 3.55
37	49				a	91.2	C ₂₅ H ₂₉ O ₂ Cl C: 75.64 H: 7.36 Cl: 8.93	n_D^{25} 1.5413 75.66 7.39 8.87
38	51				a	84.6	C ₂₂ H ₂₂ O ₄ C: 75.41 H: 6.33	n_D^{26} 1.5253 75.40 6.22

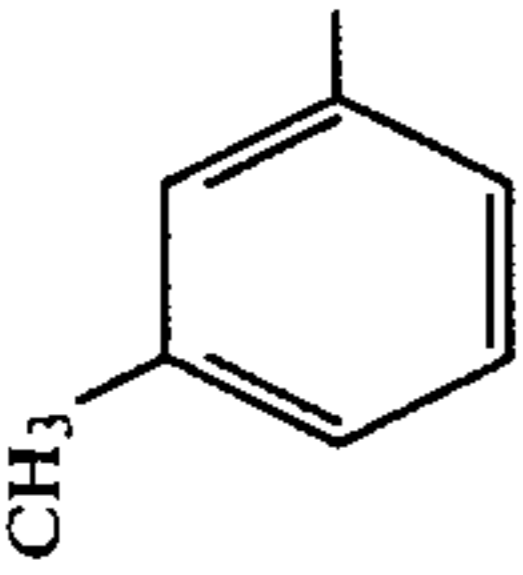
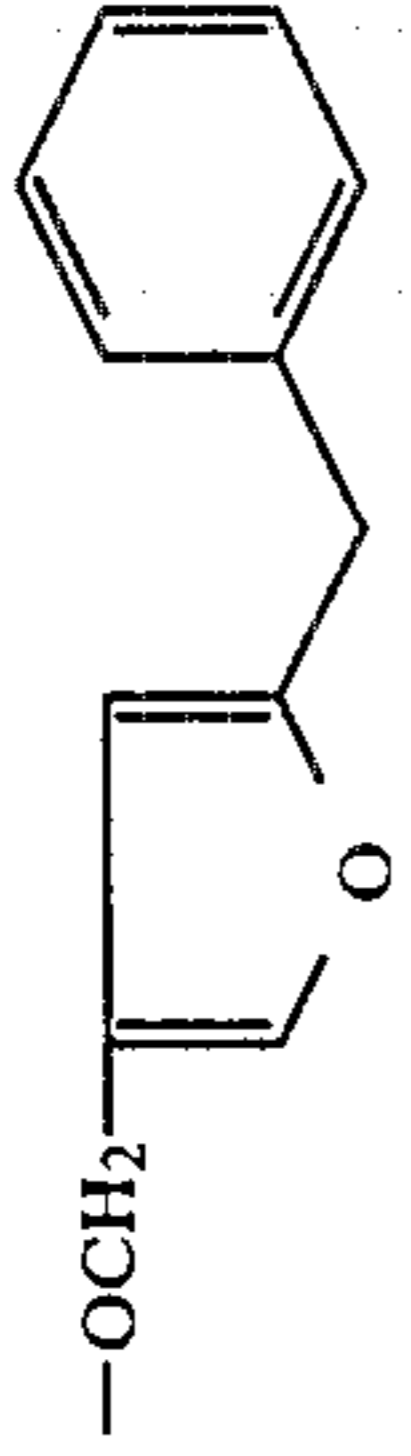
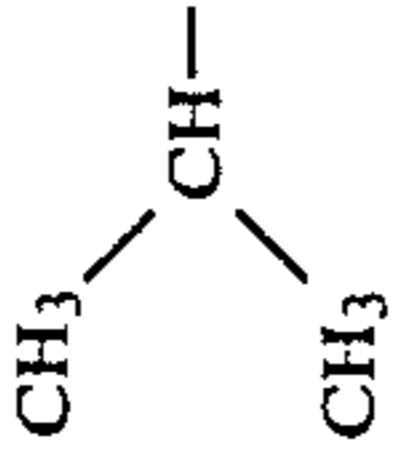
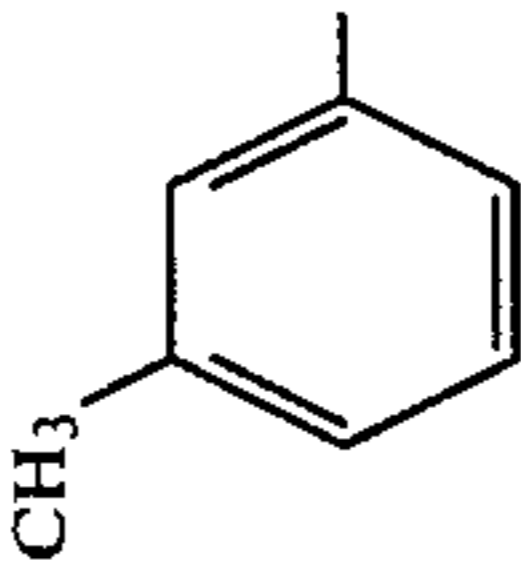
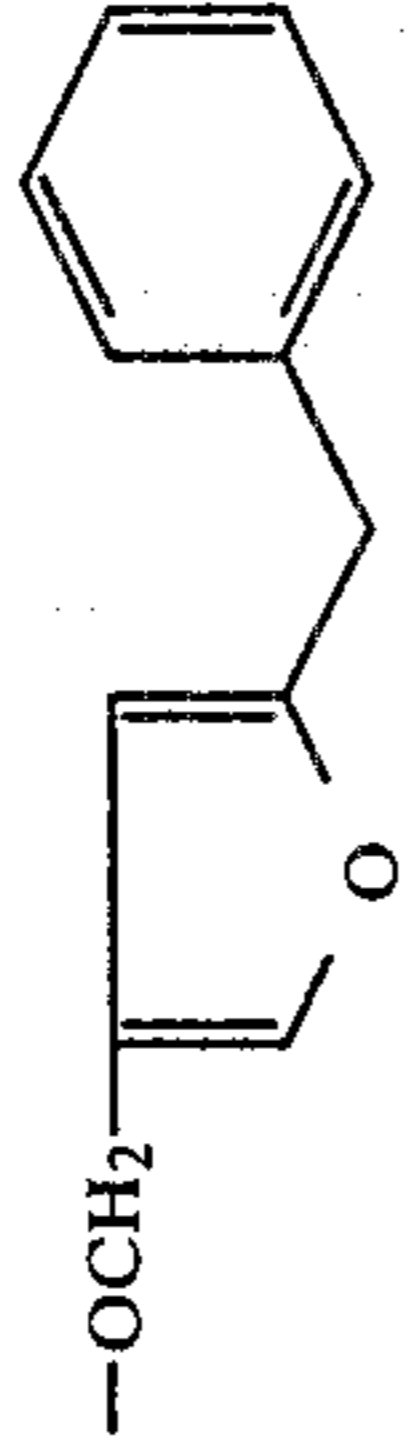
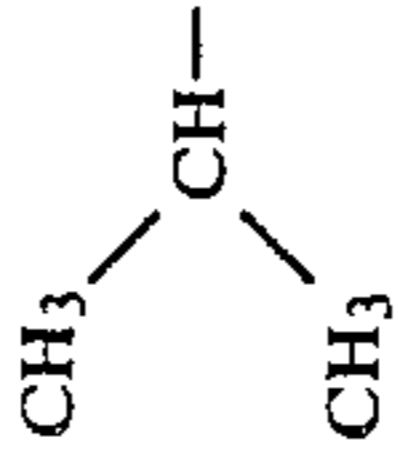
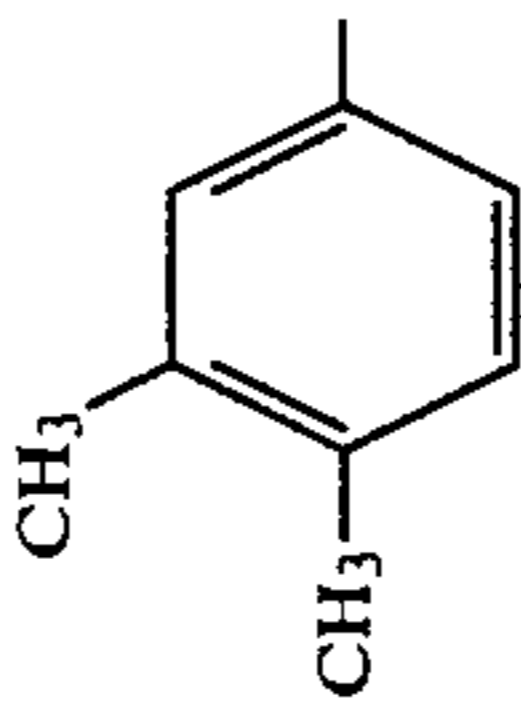

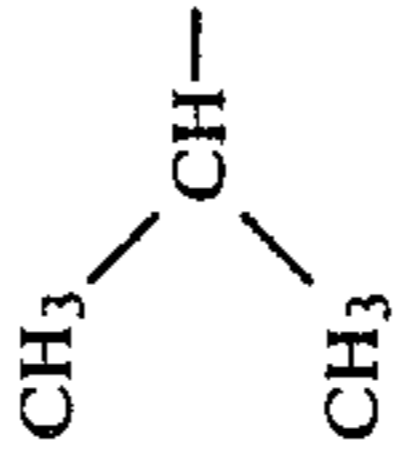
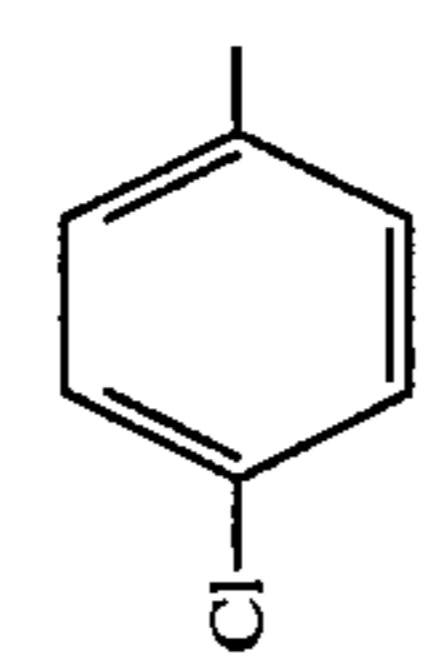
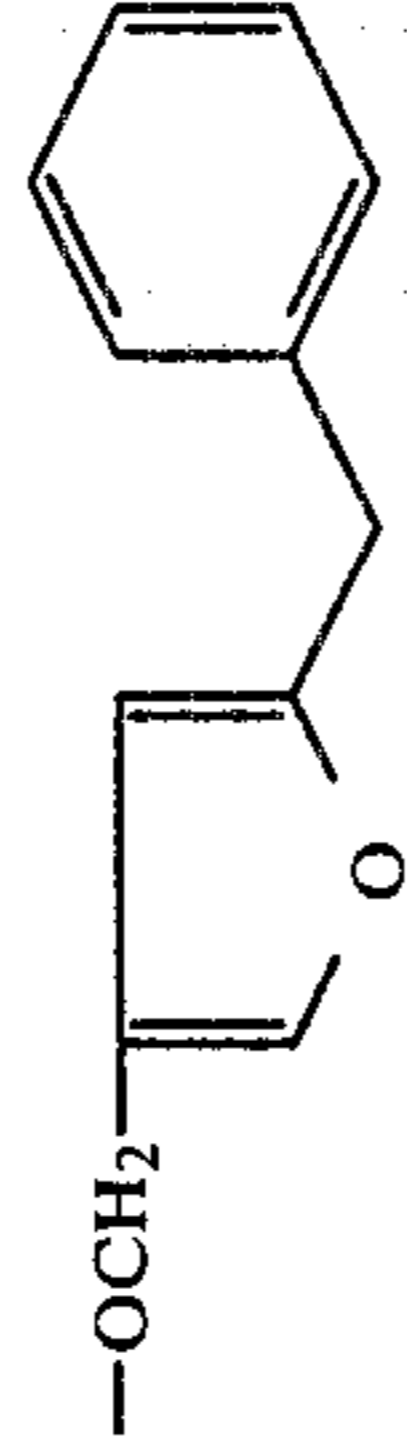
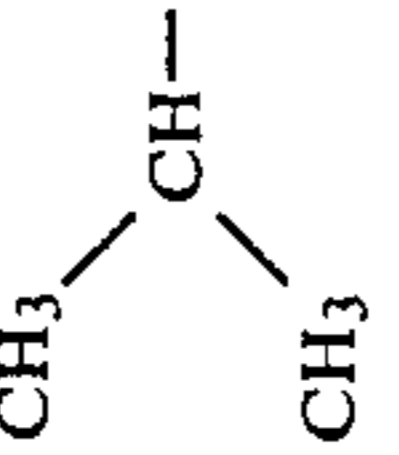
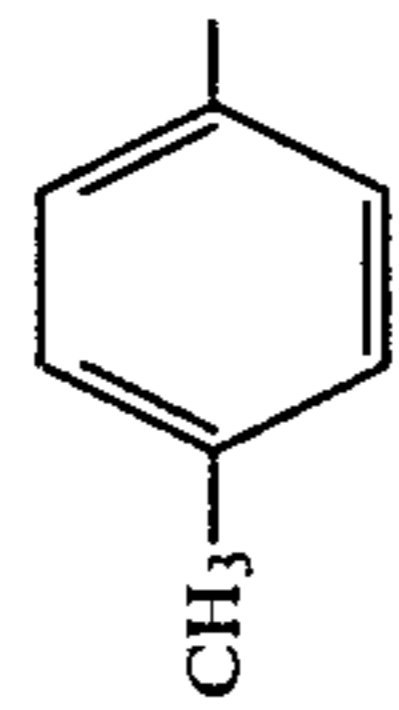
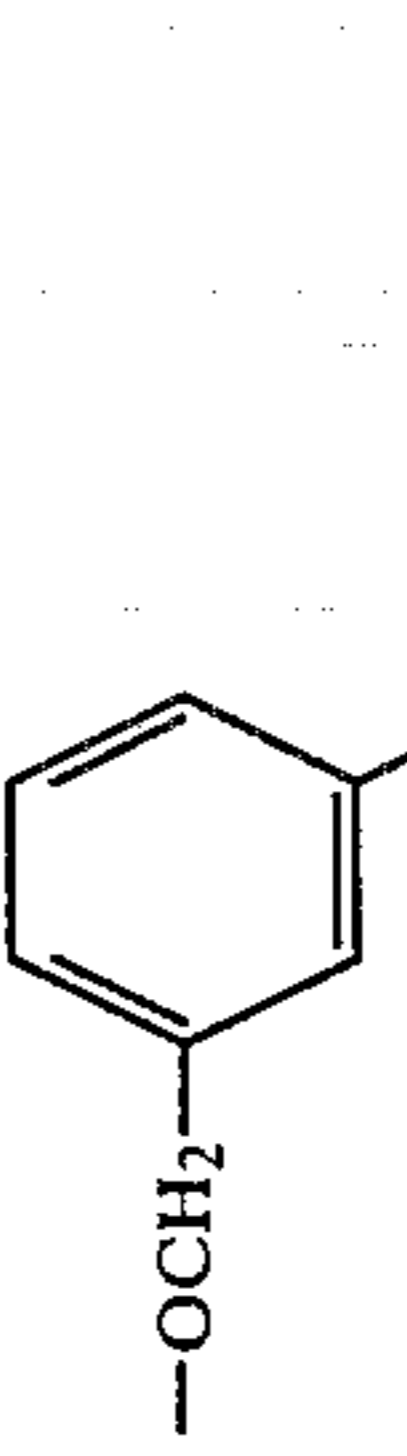

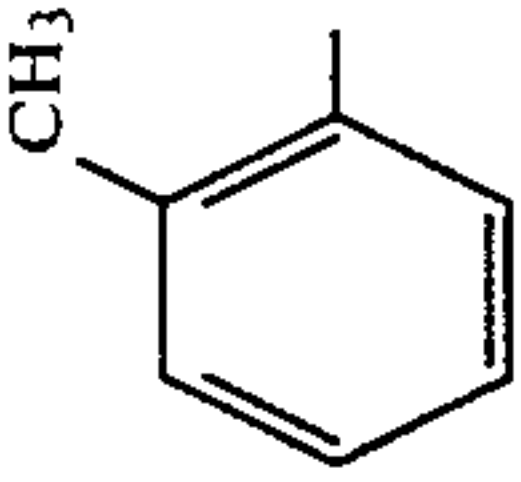
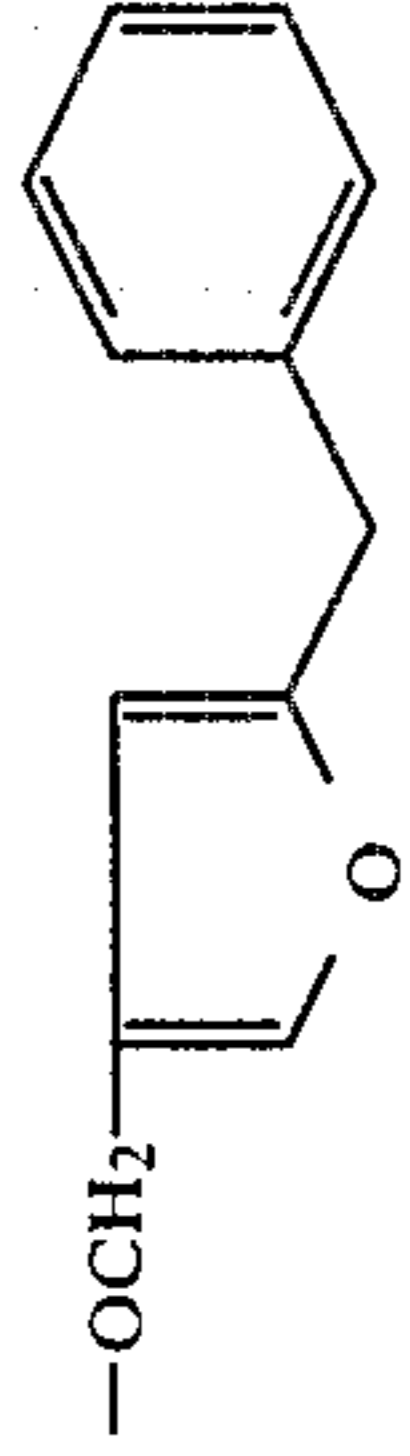
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Preparation Exp. No.	Comp. No.	Z	Y	X	Chemical Structure	Method	Yield (%)	Physical Properties		
								Calcd. (%)	Found (%)	n_D^{25}
39	52	CH ₃ CH ₂ -				a	81.2	C ₂₂ H ₂₄ O ₃ S C: 71.71 H: 6.56 S: 8.70	n _D ²⁵ 1.5549 71.80 6.59 8.78	
40	55	CH ₂ =CH-CH ₂ -				a	86.9	C ₂₂ H ₂₁ O ₅ NS C: 64.21 H: 5.15 N: 3.40 S: 7.79	n _D ¹⁹ 1.5350 64.30 5.19 3.29 7.72	
41	56					c	72.1	C ₂₃ H ₂₆ O ₅ S C: 66.64 H: 6.32 S: 7.14	n _D ²² 1.5324 66.57 6.40 7.71	
42	67					a	93.7	C ₂₁ H ₂₄ O ₄ C: 74.09 H: 7.11	n _D ²⁵ 1.5628 74.06 7.22	
43	61					a	93.2	C ₂₄ H ₂₆ O ₅ C: 73.07 H: 6.64	n _D ²⁵ 1.5501 73.11 6.57	
44	69	CH ₃ CH ₂ -				b	78.9	C ₂₁ H ₁₉ O ₃ F ₃ C: 67.01 H: 5.09 F: 15.15	n _D ²⁵ 1.5119 66.95 5.14 15.20	

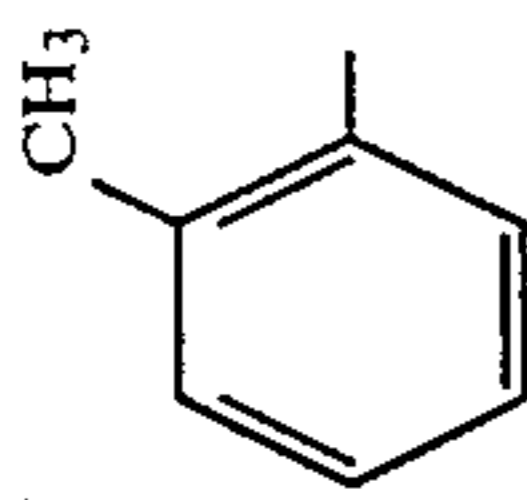
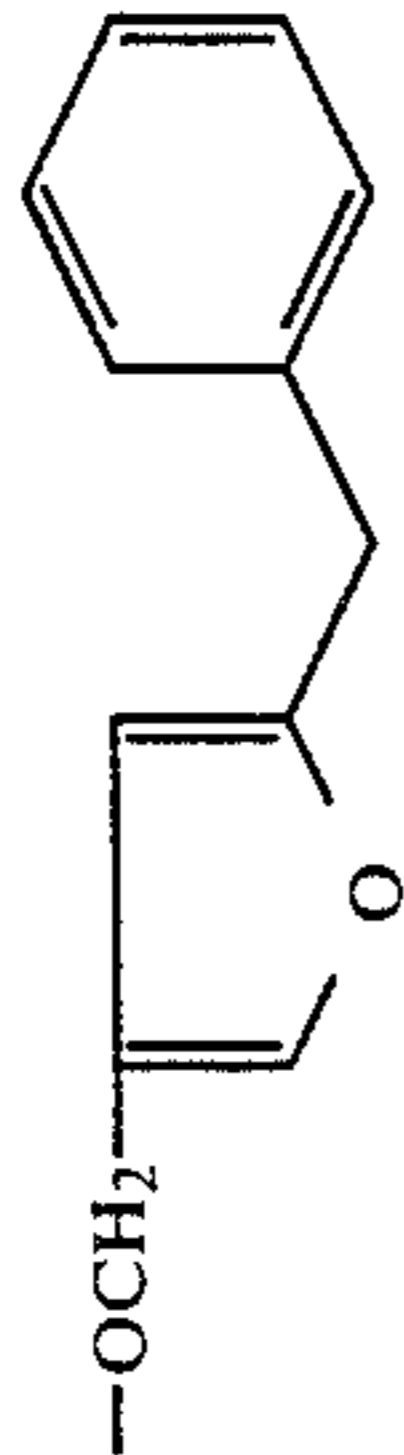
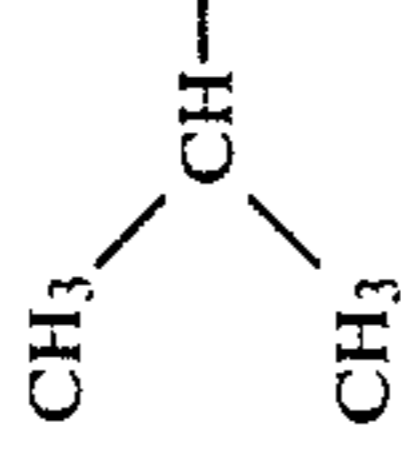
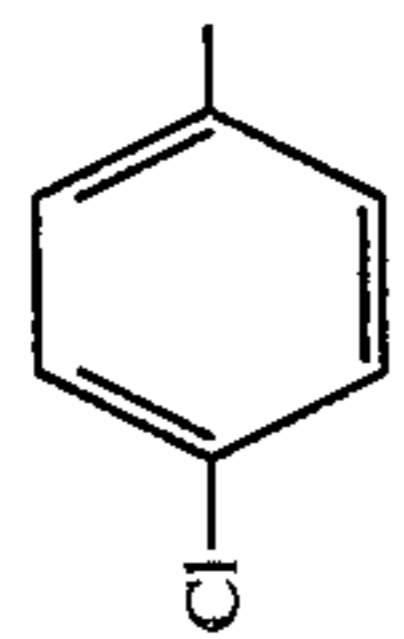
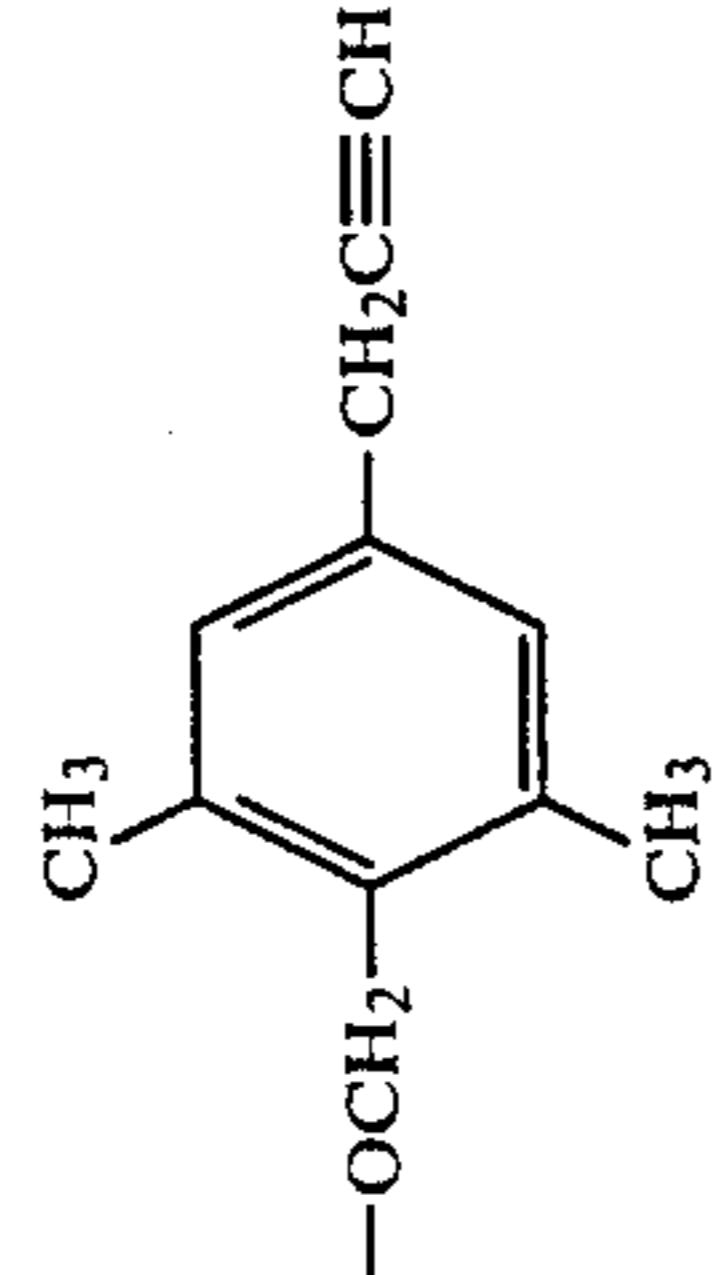
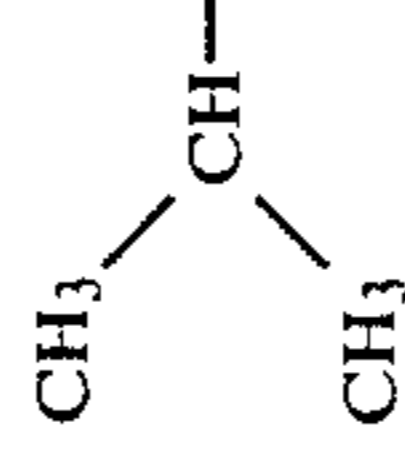
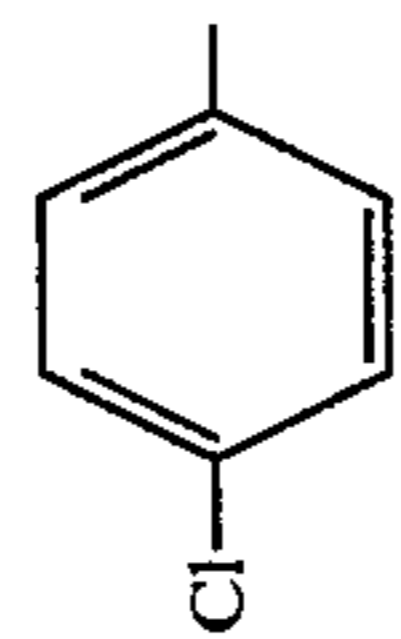
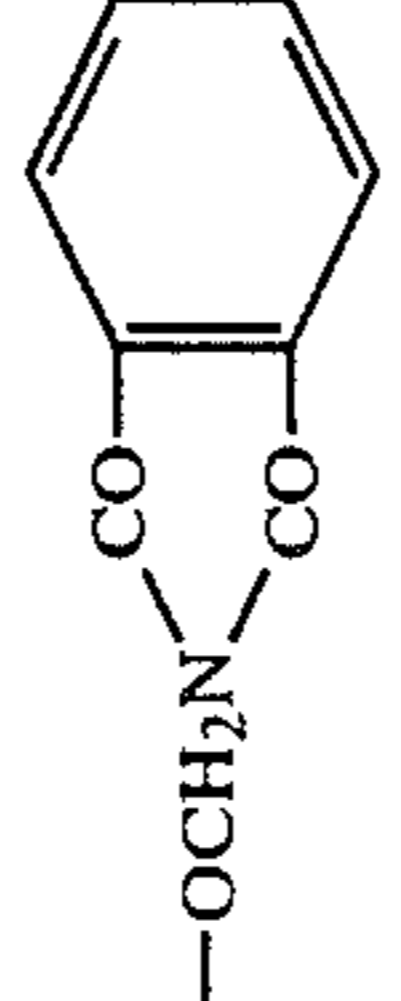
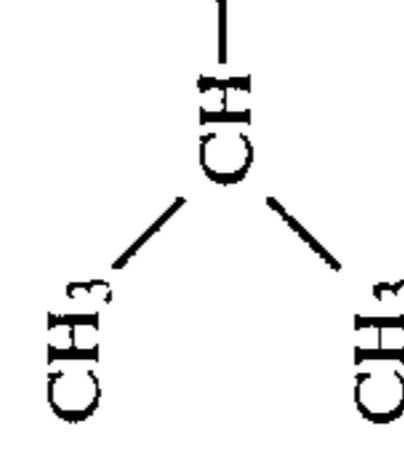
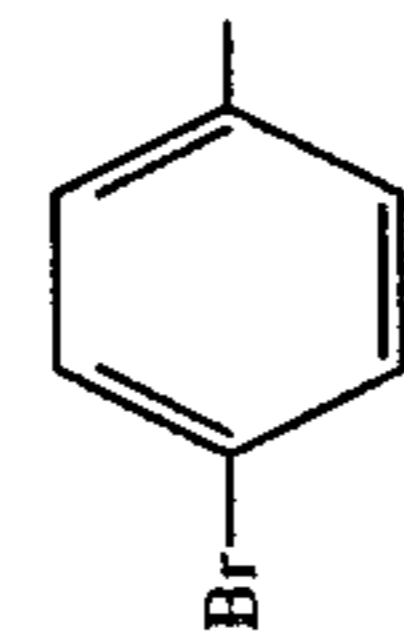
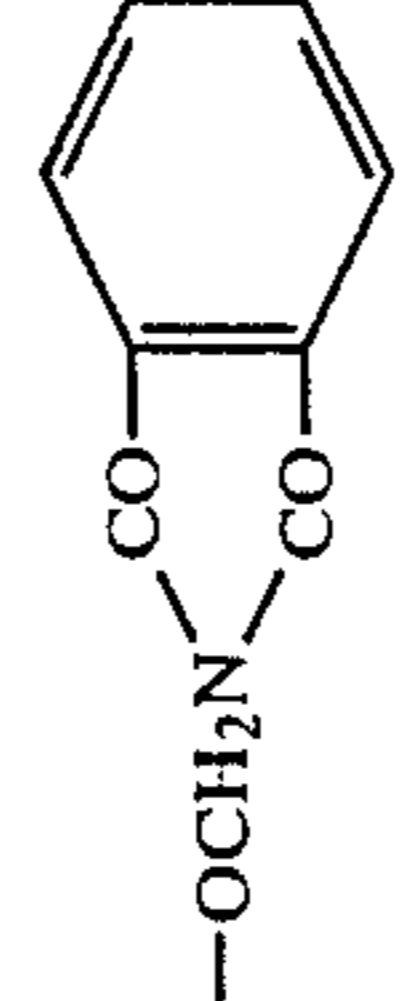
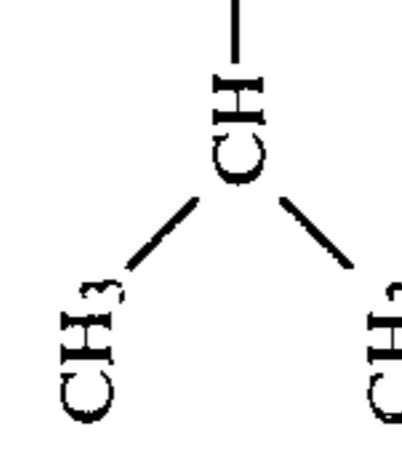
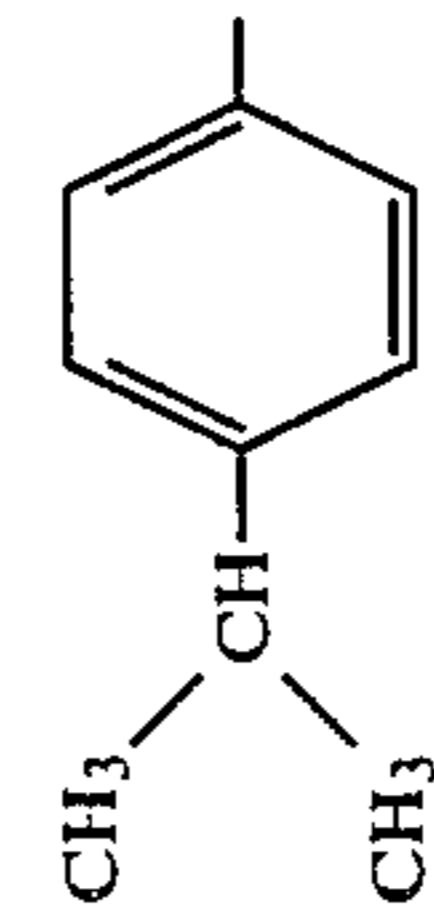
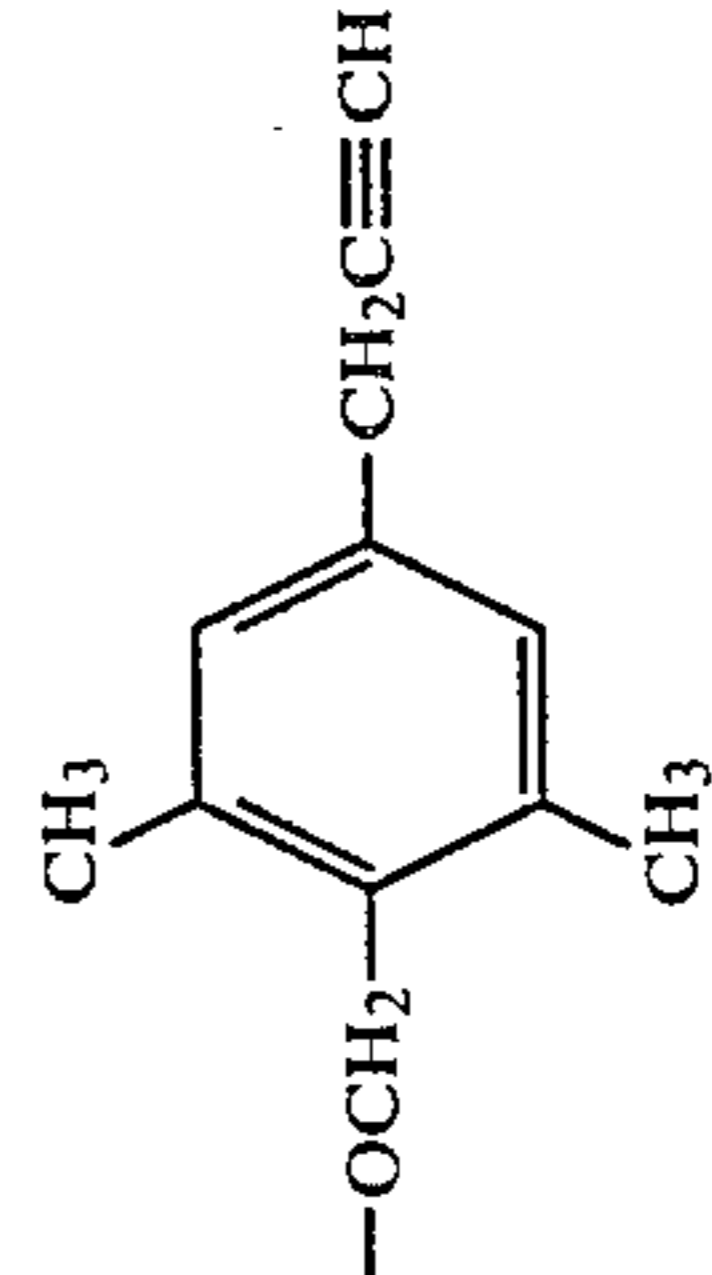
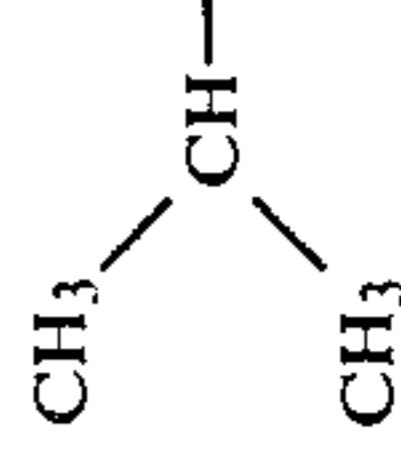
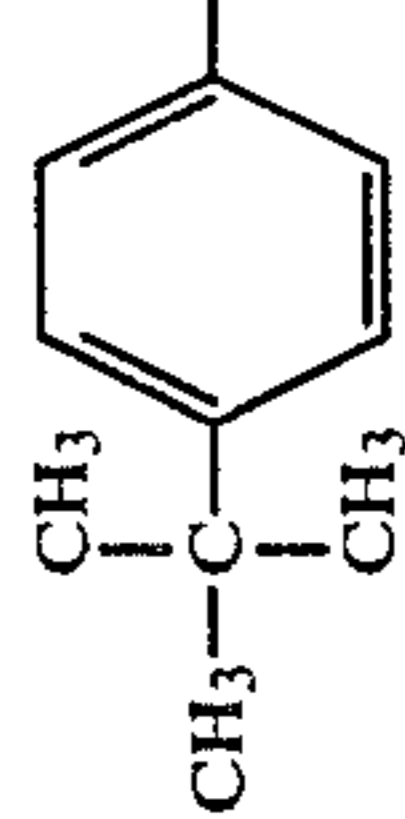
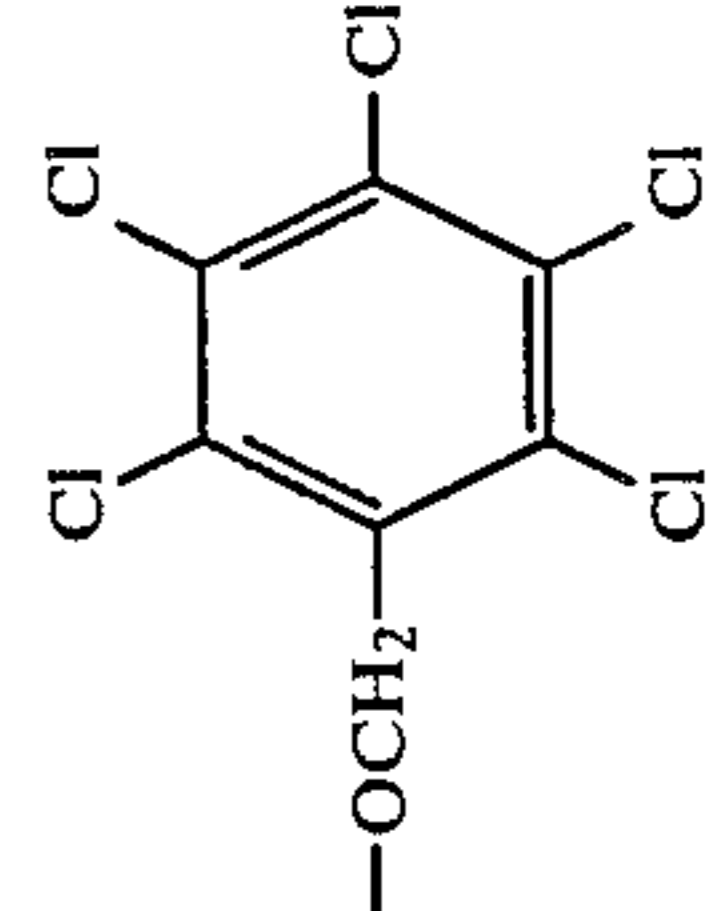
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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
45	81				d	84.8	C ₂₄ H ₂₆ O ₄ C: 76.16 H: 6.93	n_D^{16} 1.5411 76.22 6.94
46	87				d	86.4	C ₂₅ H ₂₆ O ₄ C: 76.90 H: 6.71	n_D^{17} 1.5377 76.85 6.76
47	93				b	83.1	C ₂₀ H ₂₄ O ₄ C: 73.14 H: 7.37	n_D^{19} 1.5211 73.19 7.30
48	43				e	83.2	C ₂₈ H ₃₀ O ₃ C: 81.12 H: 7.30	n_D^{20} 1.5660 81.20 7.27
49	102				a	91	C ₂₃ H ₂₄ O ₃ C: 79.28 H: 6.94	n_D^{25} 1.5474 79.32 7.10
50	103				a	90	C ₂₄ H ₂₆ O ₃ C: 79.53 H: 7.23	n_D^{25} 1.5433 79.60 7.18

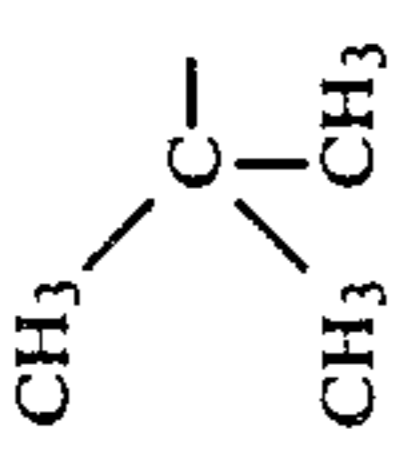
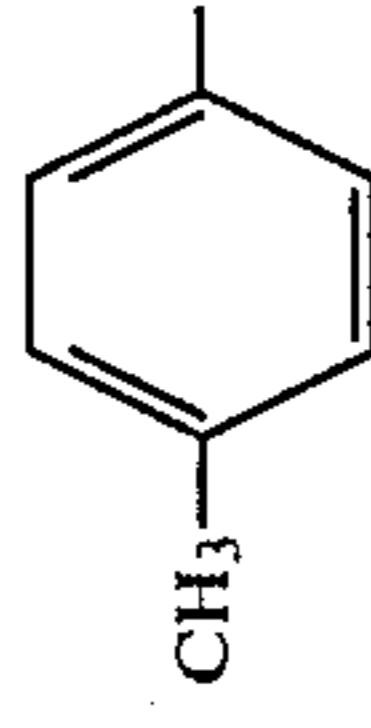
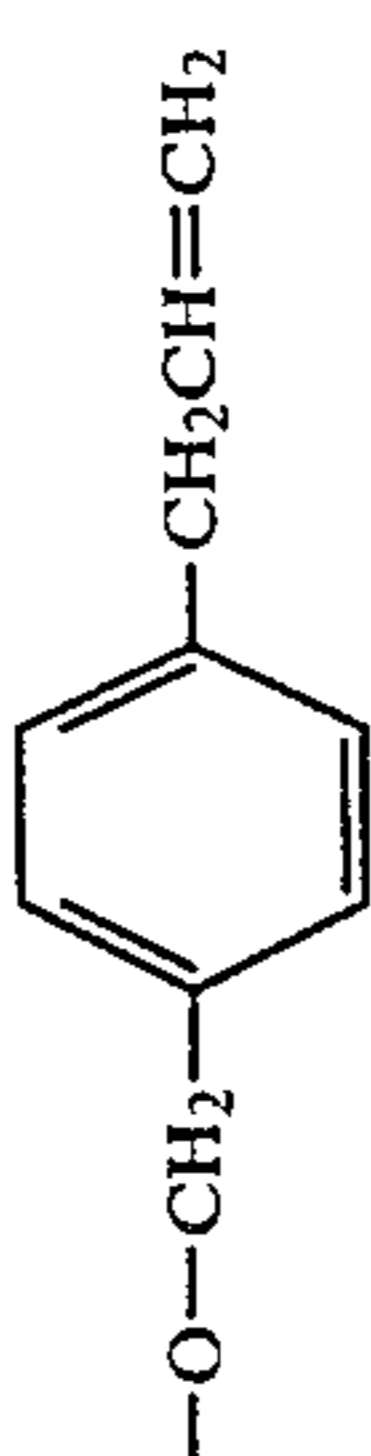
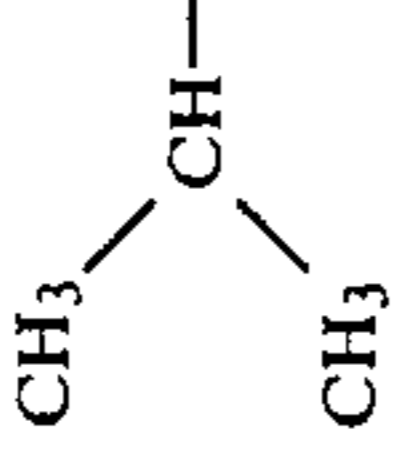
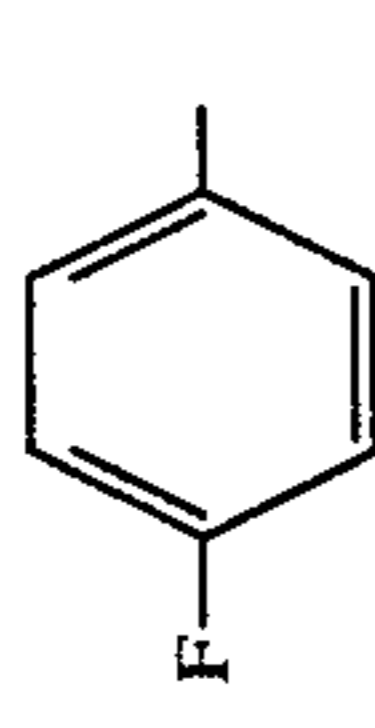
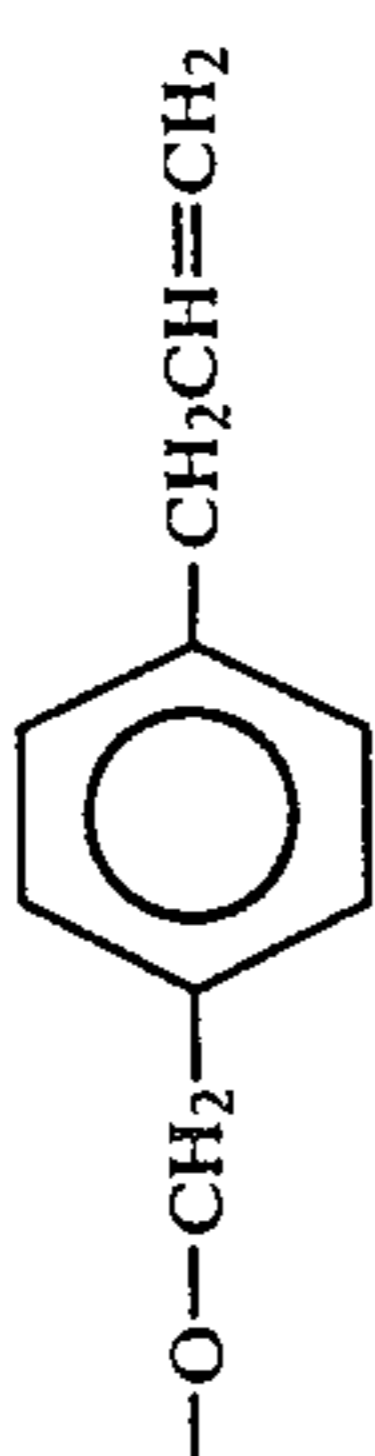
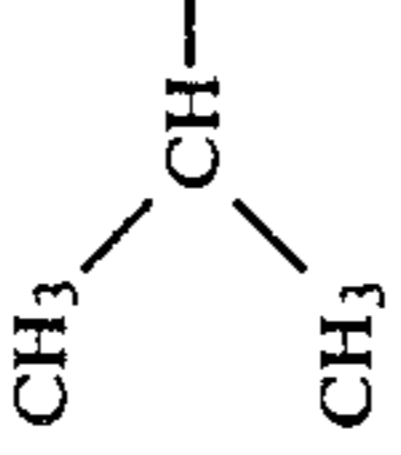
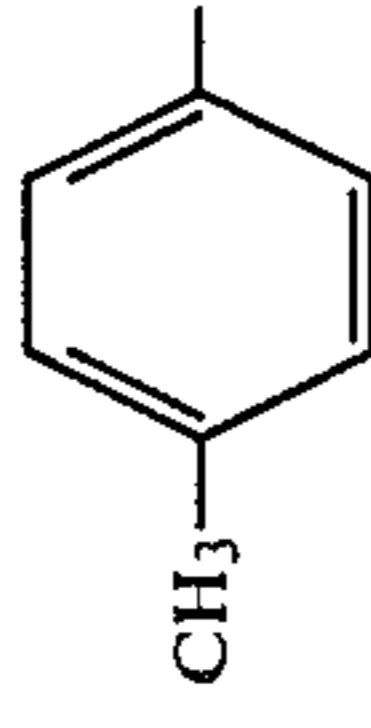
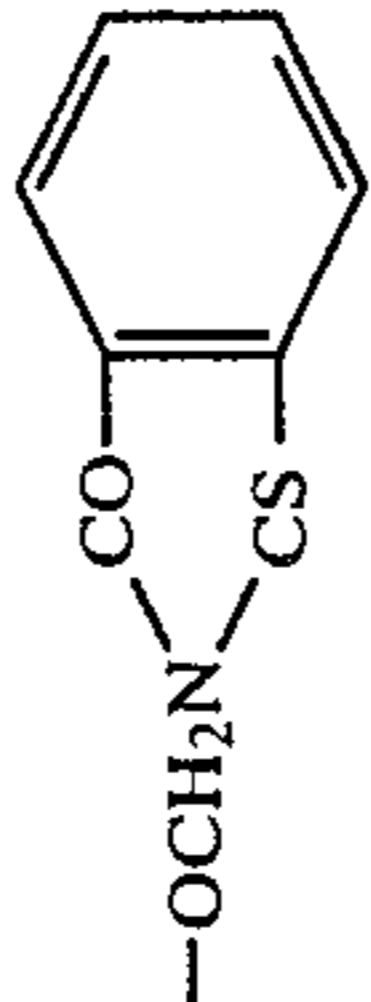
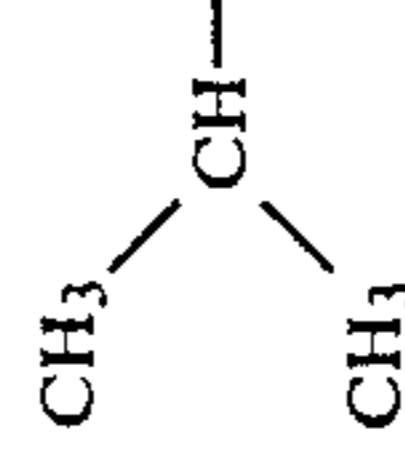
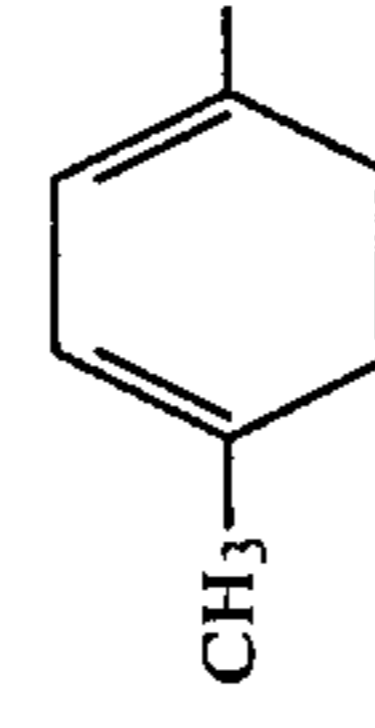
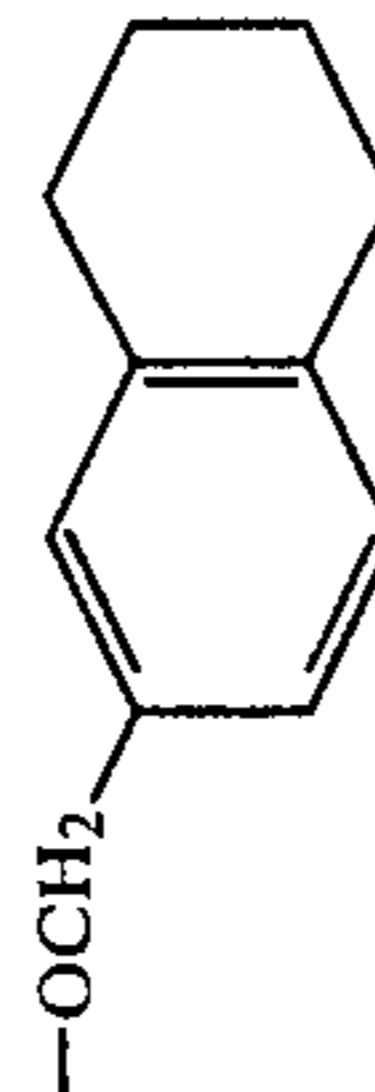
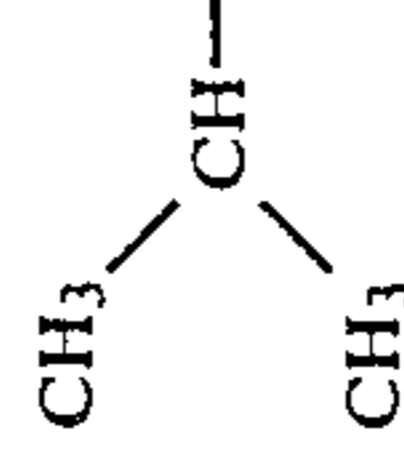
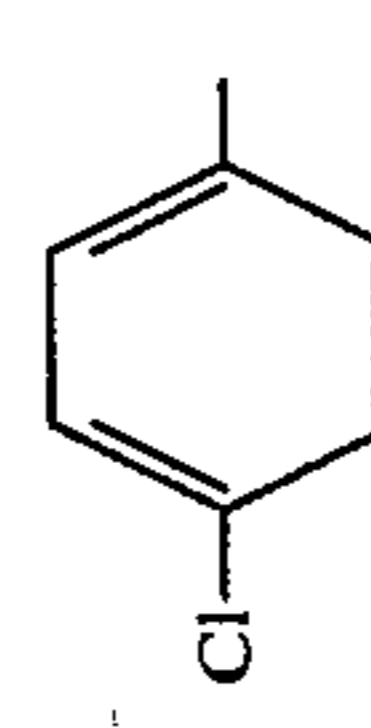
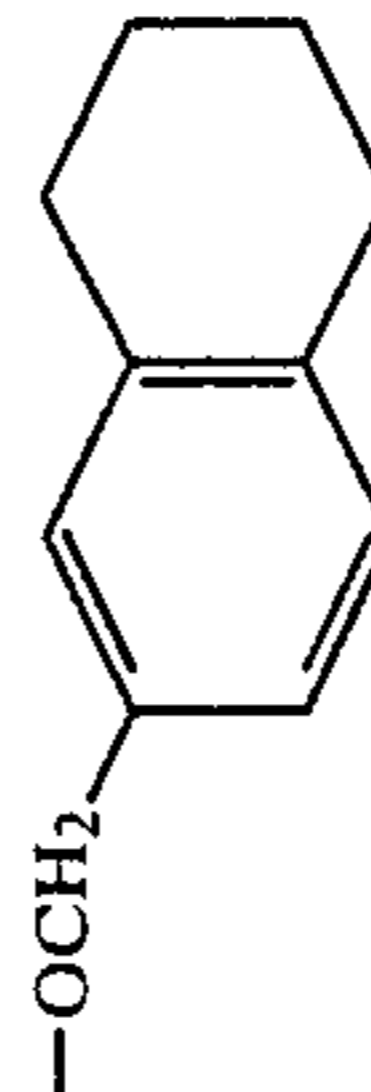
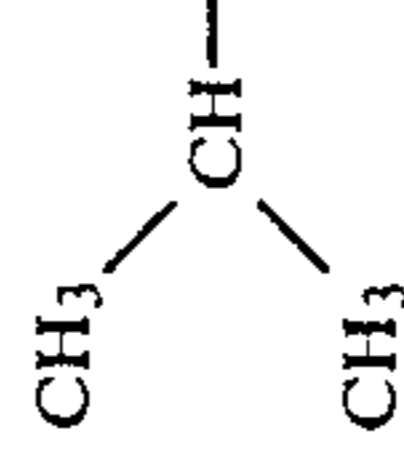
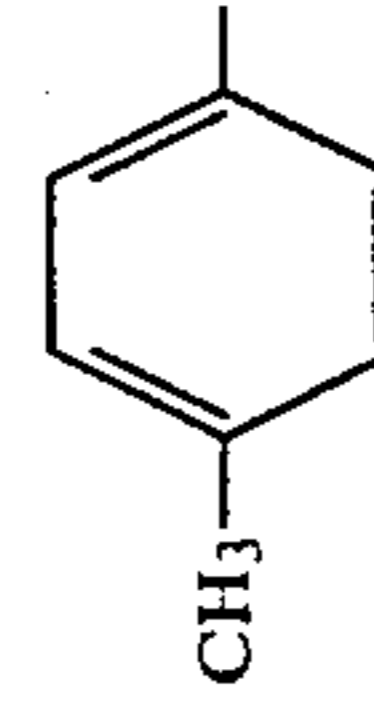
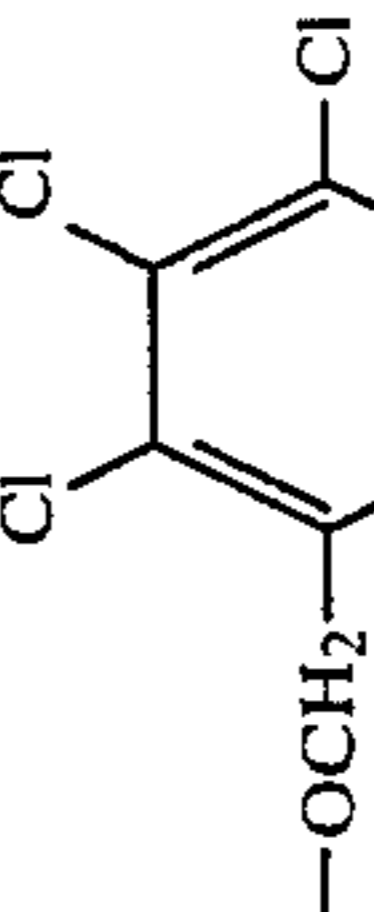
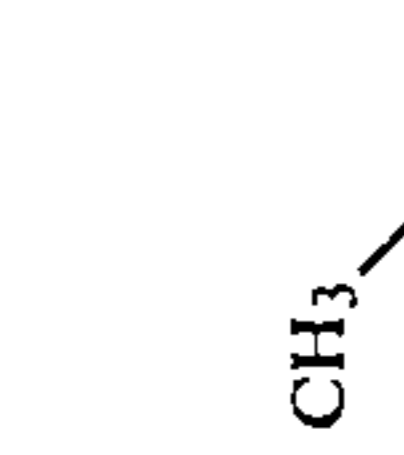

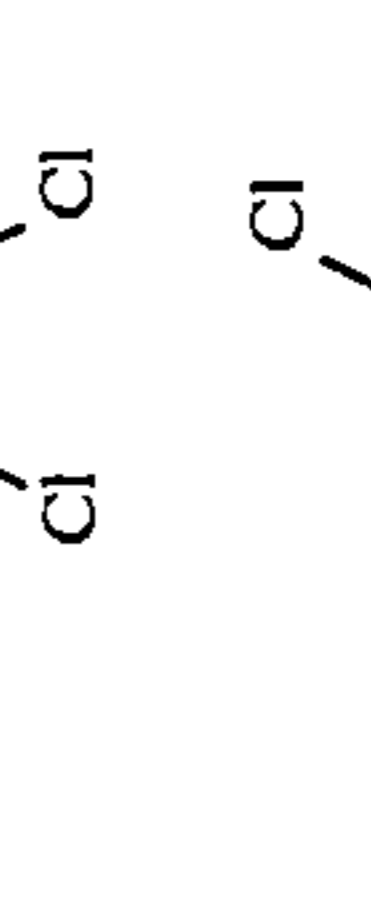
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Preparation Exp. No.	Comp. No.	Z	Y	X	Chemical Structure	Esterification Method	Yield (%)	Physical Properties	
								Calcd. (%)	Found (%)
51	104	CH ₃ CH ₂ —		—OCH ₂ —		a	89	C ₂₃ H ₂₄ O ₃ C: 79.28 H: 6.94	<i>n</i> _D ²⁵ 1.5470 79.23 6.99
52	105			—OCH ₂ —		a	93	C ₂₄ H ₂₆ O ₃ C: 79.53 H: 7.23	<i>n</i> _D ^{25.5} 1.5430 79.50 7.30
53	106			—OCH ₂ —		a	92	C ₂₅ H ₂₈ O ₃ C: 79.75 H: 7.50	<i>n</i> _D ^{25.5} 1.5238 79.79 7.49
54	107			—OCH ₂ —		a	91	C ₂₃ H ₂₃ O ₃ Cl C: 72.15 H: 6.05 Cl: 9.26	<i>n</i> _D ^{25.5} 1.5241 72.18 6.10 9.29
55	108			—OCH ₂ —		a	90	C ₂₅ H ₂₆ O ₃ C: 80.18 H: 7.00	<i>n</i> _D ^{28.5} 1.5596 80.21 6.92
56	109			—OCH ₂ —		a	94	C ₂₄ H ₂₆ O ₃ C: 77.53 H: 7.23	<i>n</i> _D ^{26.5} 1.5431 79.60 7.21

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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification		Physical Properties	
					Method	Yield (%)	Calcd. (%)	Found (%)
57	110	CH ₃ CH ₂ -			a	89	C ₂₃ H ₂₄ O ₃ C: 79.28 H: 6.94	n _D ^{26.5} 1.5501 79.40 7.11
58	129				b	85	C ₂₃ H ₂₅ O ₂ Cl C: 74.88 H: 6.83 Cl: 9.61	n _D ²⁰ 1.5423 74.79 6.81 9.63
59	130				e	68	C ₂₀ H ₁₈ O ₄ NCl C: 64.60 H: 4.88 N: 3.77 Cl: 9.54	n _D ²⁰ 1.5501 64.63 4.86 3.79 9.53
60	131				e	69	C ₂₀ H ₁₈ O ₄ NBr C: 57.70 H: 4.36 N: 3.37 Br: 19.20	n _D ²⁰ 1.5383 57.80 4.39 3.37 19.28
61	132				a	88	C ₂₆ H ₃₂ O ₂ C: 82.93 H: 8.57	n _D ²⁰ 1.5433 82.88 8.63
62	133				a	92	C ₂₂ H ₂₃ O ₂ Cl ₅ C: 53.20 H: 4.67 Cl: 35.69	n _D ²⁰ 1.5211 53.38 4.68 35.71

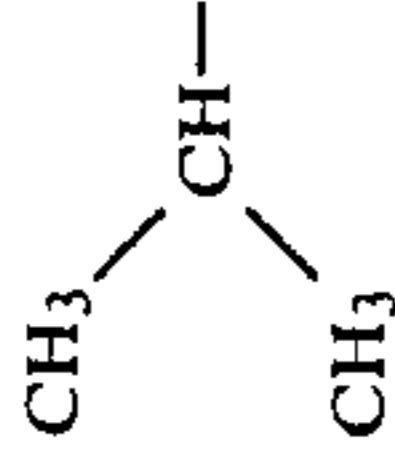
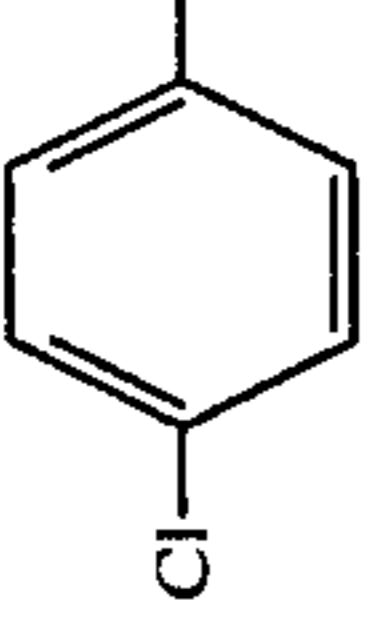
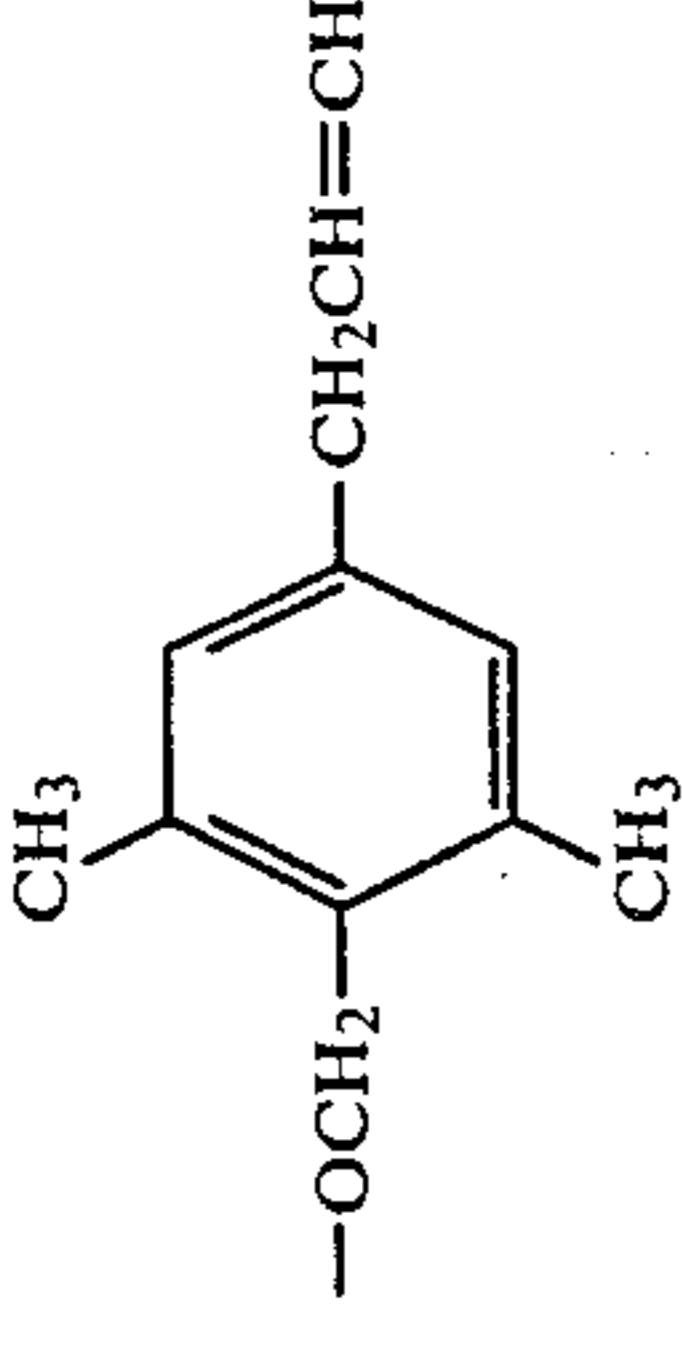
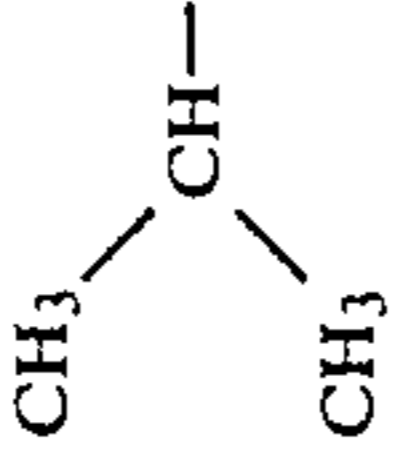
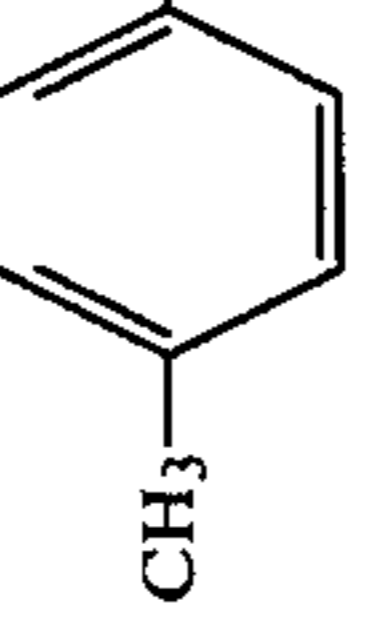

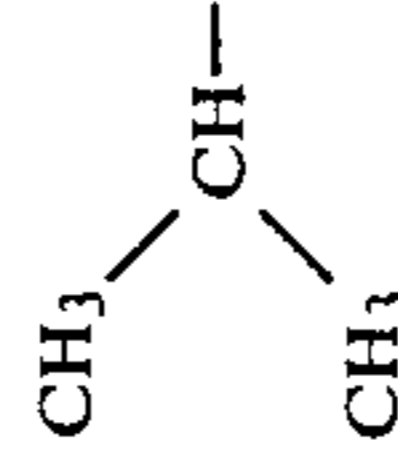
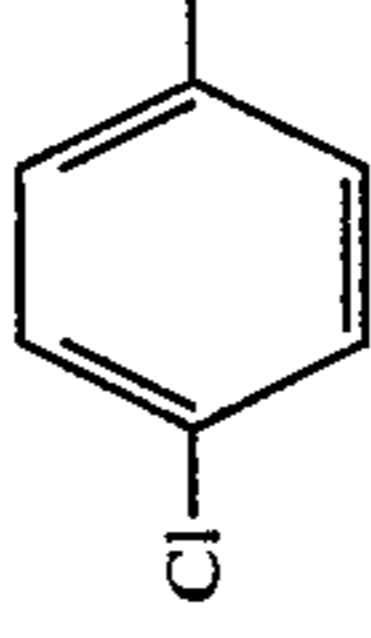
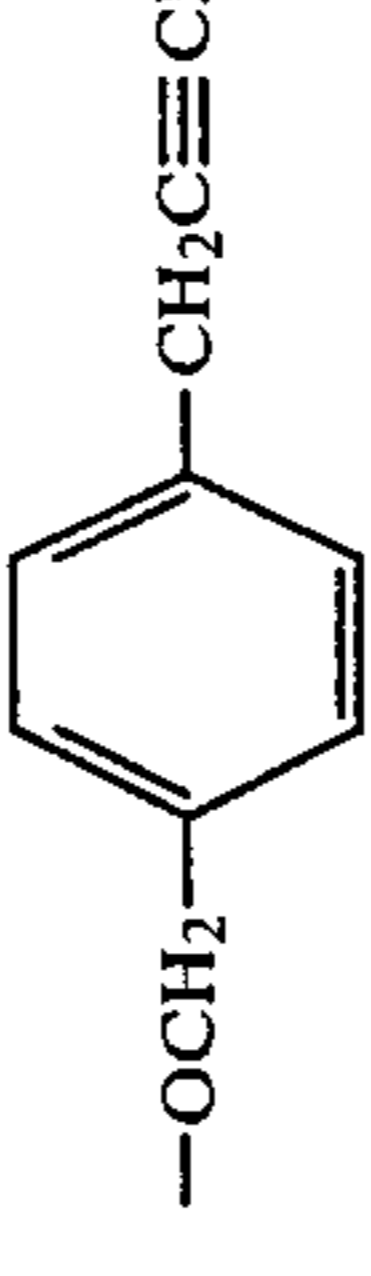
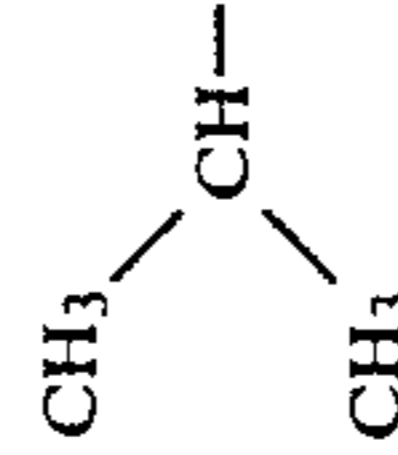
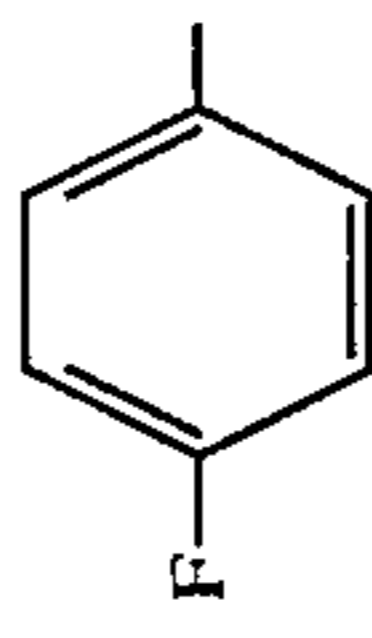
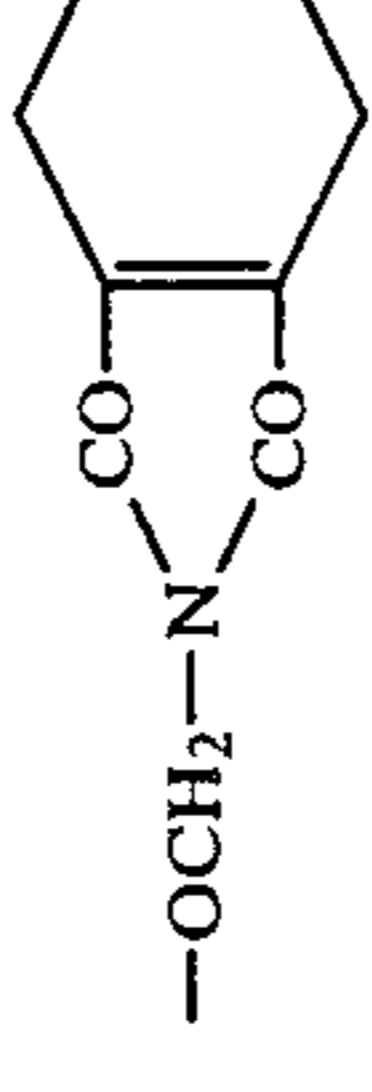
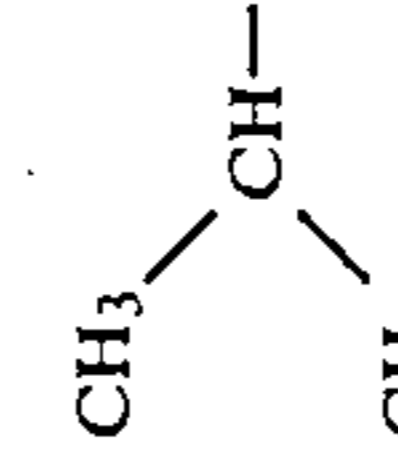
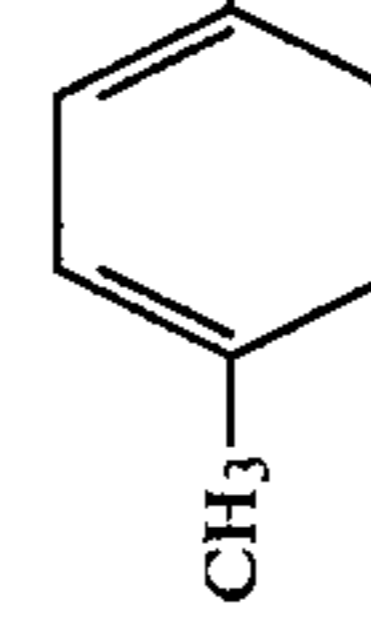
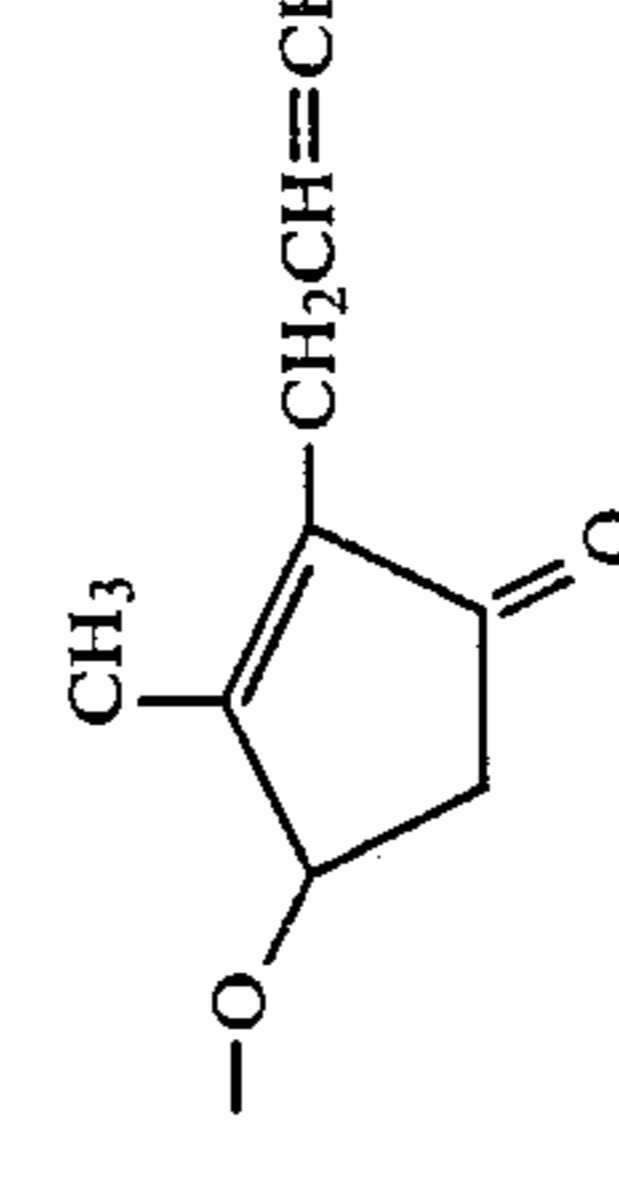
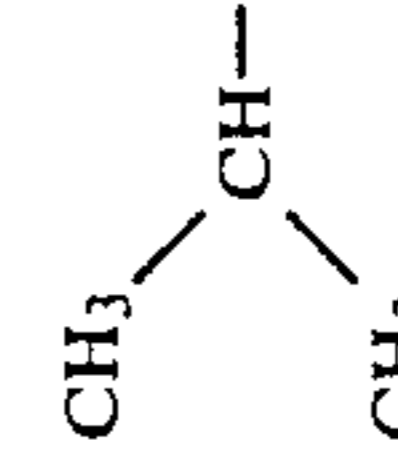
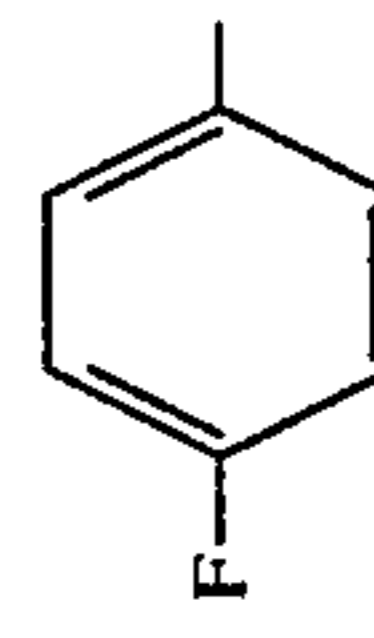
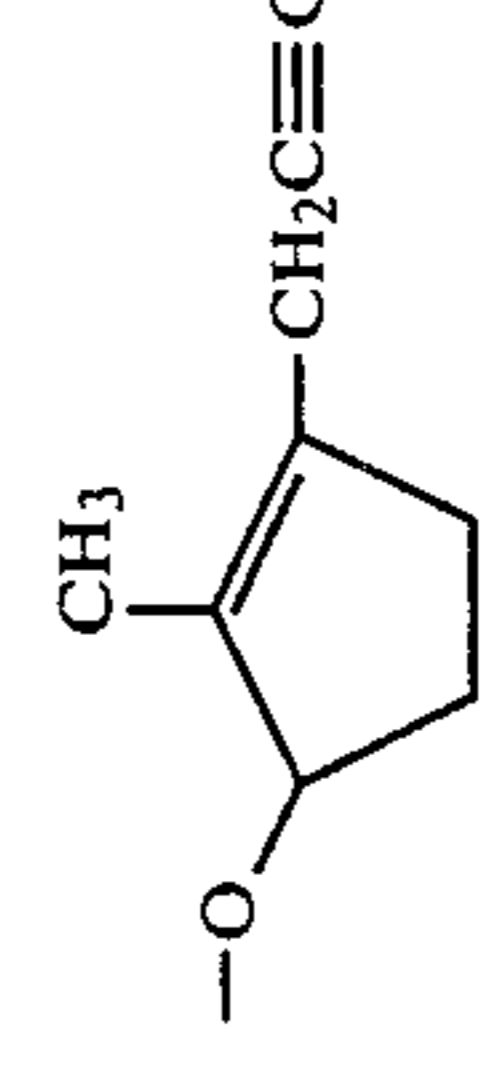
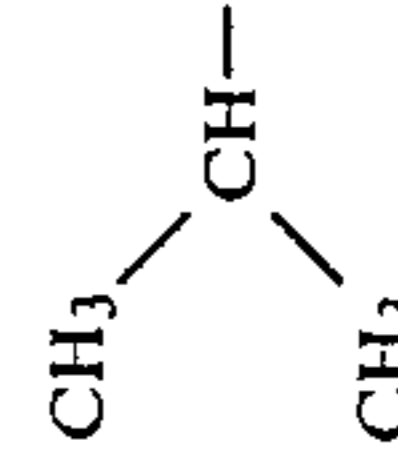
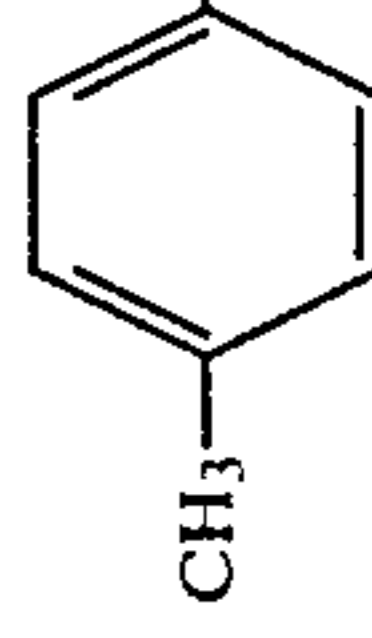
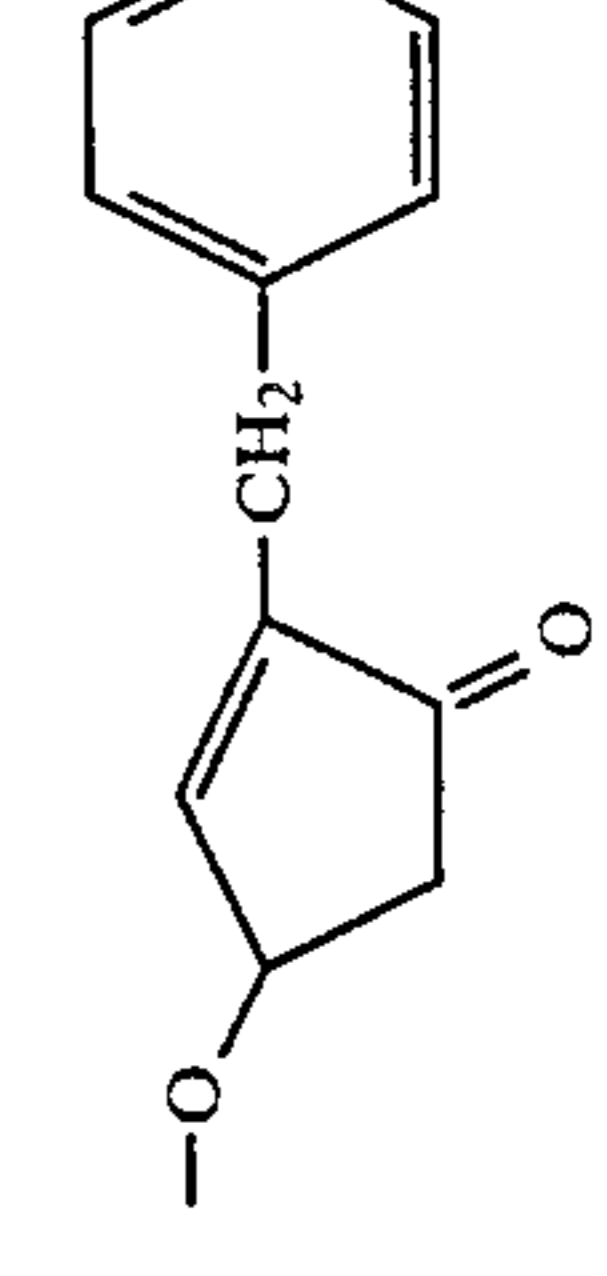
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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification		Physical Properties	
					Method	Yield (%)	Calcd. (%)	Found (%)
63	134				a	93	C ₂₃ H ₂₈ O ₂ C: 82.10 H: 8.39	82.19 8.41
64	135				a	89	C ₂₁ H ₂₃ O ₂ F C: 77.27 H: 7.10 F: 5.82	77.32 7.03 5.91
65	136				a	89	C ₂₁ H ₂₁ O ₃ NS C: 68.64 H: 5.76 N: 3.81 S: 8.73	n _D ²⁰ 1.5335 68.67 5.81 3.80 8.79
66	137				d	81	C ₂₃ H ₂₈ O ₂ C: 82.10 H: 8.39	n _D ²⁰ 1.5299 82.15 8.30
67	138				d	87	C ₂₂ H ₂₅ O ₂ Cl C: 74.04 H: 7.06 Cl: 9.94	n _D ²⁰ 1.5414 74.08 7.10 10.03
68	139				b	79	C ₁₉ H ₁₇ O ₂ Cl ₅ C: 50.19 H: 3.77 Cl: 39.01	n _D ²⁰ 1.5382 50.22 3.79 39.13
69	140				a	69	C ₂₀ H ₁₇ O ₂ FCl ₂ C: 63.34 H: 4.52 F: 5.01 Cl: 18.70	n _D ²⁰ 1.5288 63.39 4.46 5.08 18.80

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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
70	141				a	78	C ₂₂ H ₂₀ O ₂ F ₃ C: 70.76 H: 5.40 F: 15.27	n_D^{20} 1.5385 70.80 5.39 15.27
71	142				a	82	C ₂₂ H ₂₆ O ₂ C: 81.95 H: 8.13	n_D^{20} 1.5236 82.06 8.12
72	143				a	80	C ₂₄ H ₃₀ O ₂ C: 82.24 H: 8.63	n_D^{20} 1.5341 82.19 8.66
73	144				a	83	C ₂₁ H ₂₃ O ₂ Cl C: 73.56 H: 6.76 Cl: 10.34	n_D^{20} 1.5491 73.51 6.81 10.29
74	145				b	86	C ₂₂ H ₂₄ O ₃ S C: 71.71 H: 6.56 S: 8.70	n_D^{20} 1.5372 71.73 6.60 8.79
75	146				a	92	C ₂₁ H ₂₄ O ₃ C: 77.75 H: 7.46	n_D^{20} 1.5377 77.77 7.40
76	147				a	95	C ₁₈ H ₂₀ O ₄ NCl C: 61.80 H: 5.76 N: 4.01 Cl: 10.14	n_D^{20} 1.5469 61.90 5.73 4.02 10.19
77	148				b	87	C ₂₂ H ₂₅ O ₂ Cl C: 74.04 H: 7.06 Cl: 9.94	n_D^{20} 1.5471 74.11 7.92 9.90

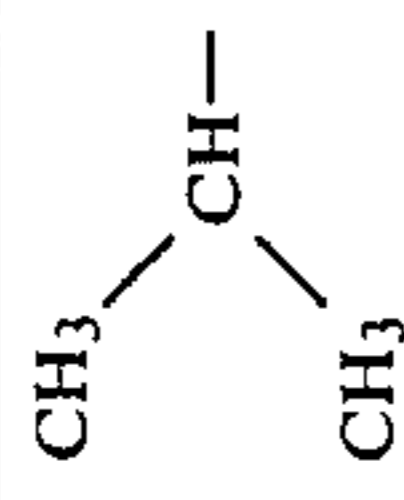
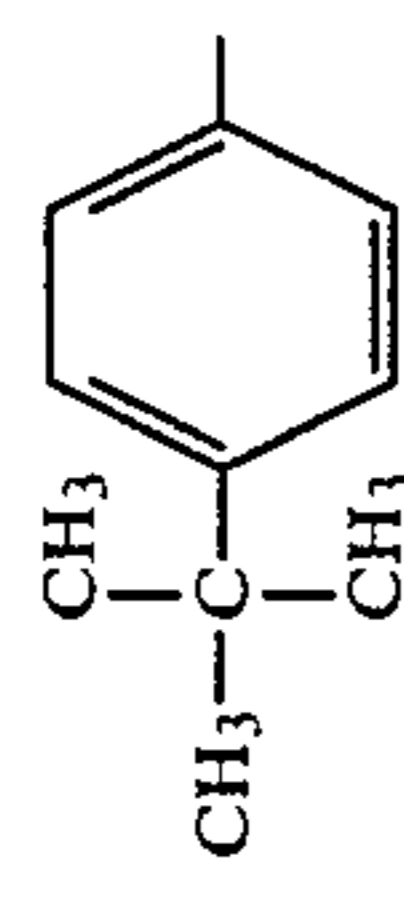
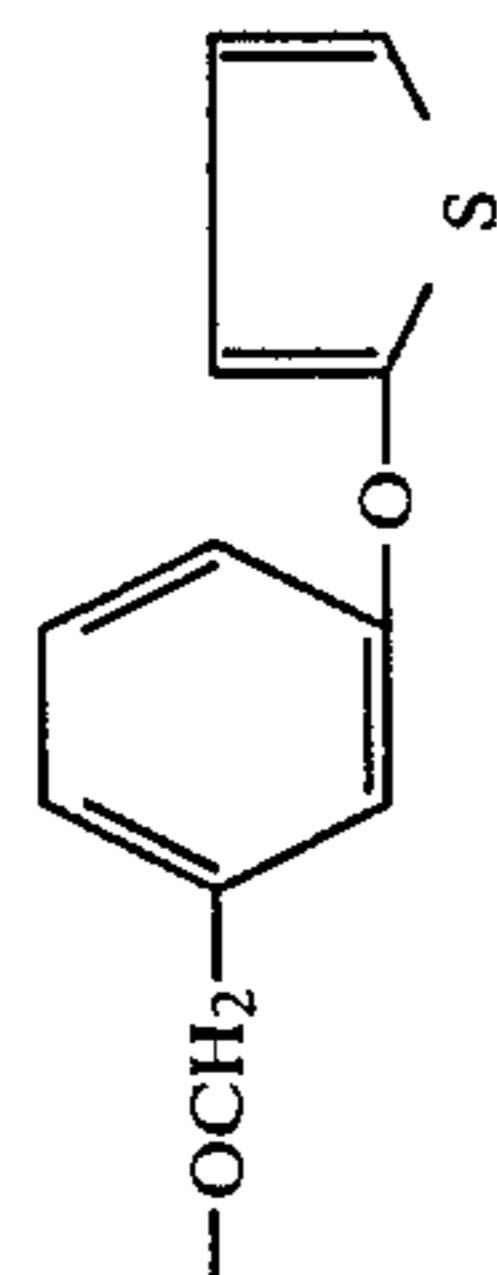
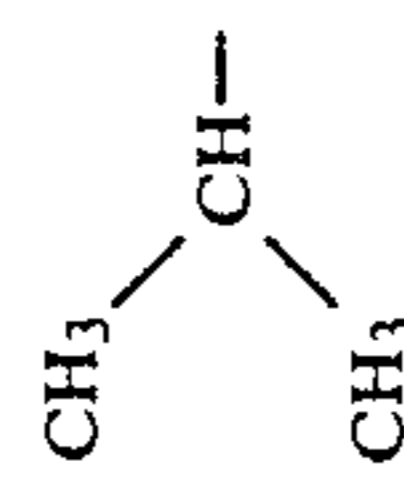
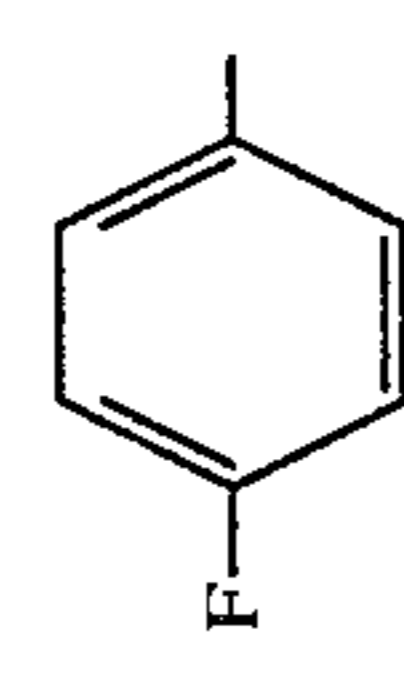
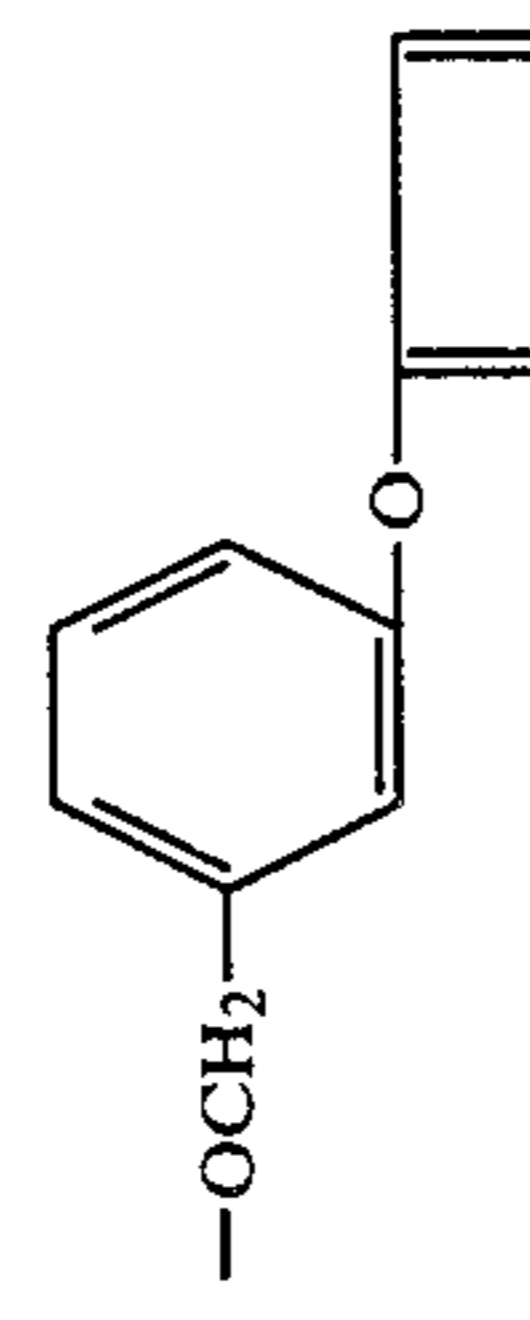
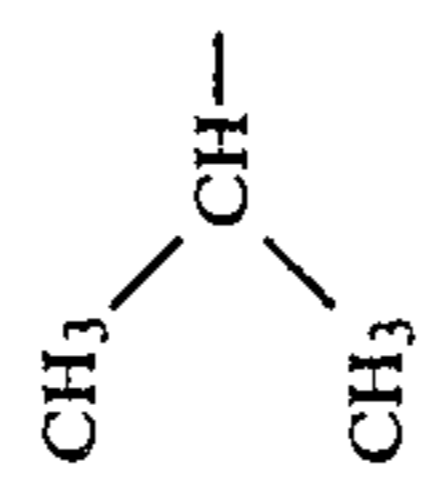
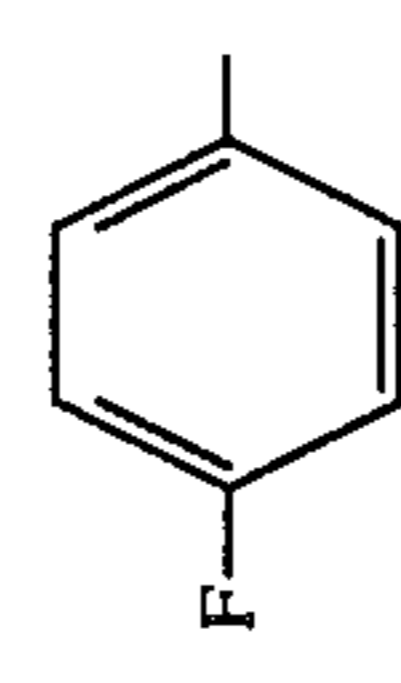
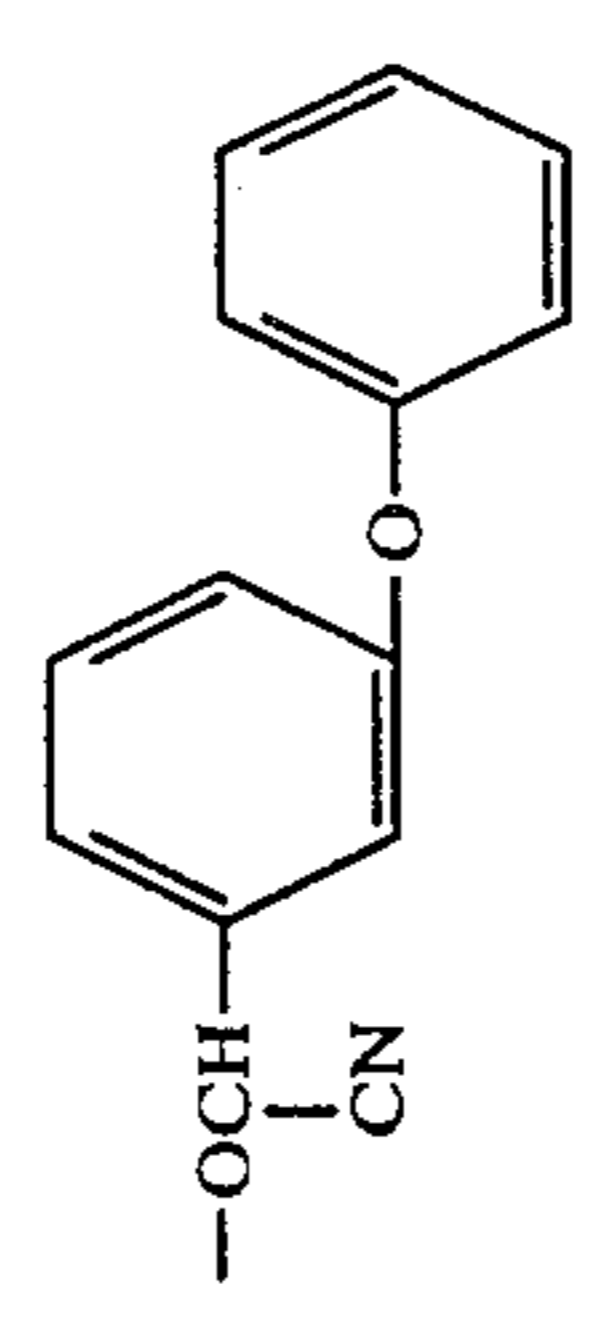
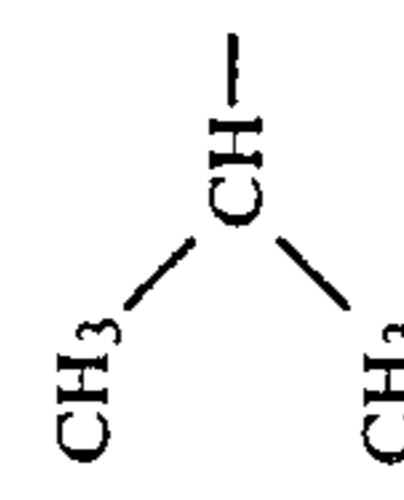
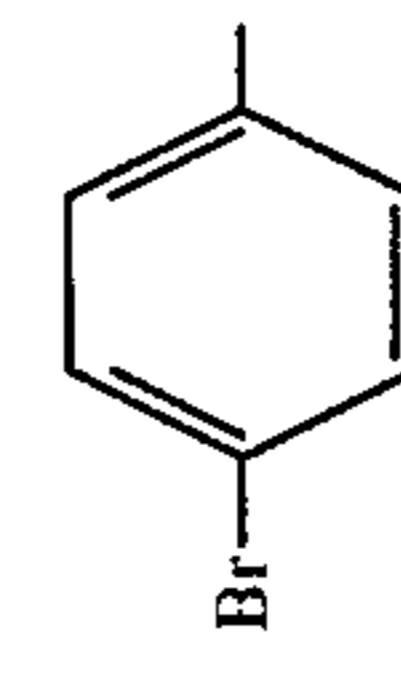
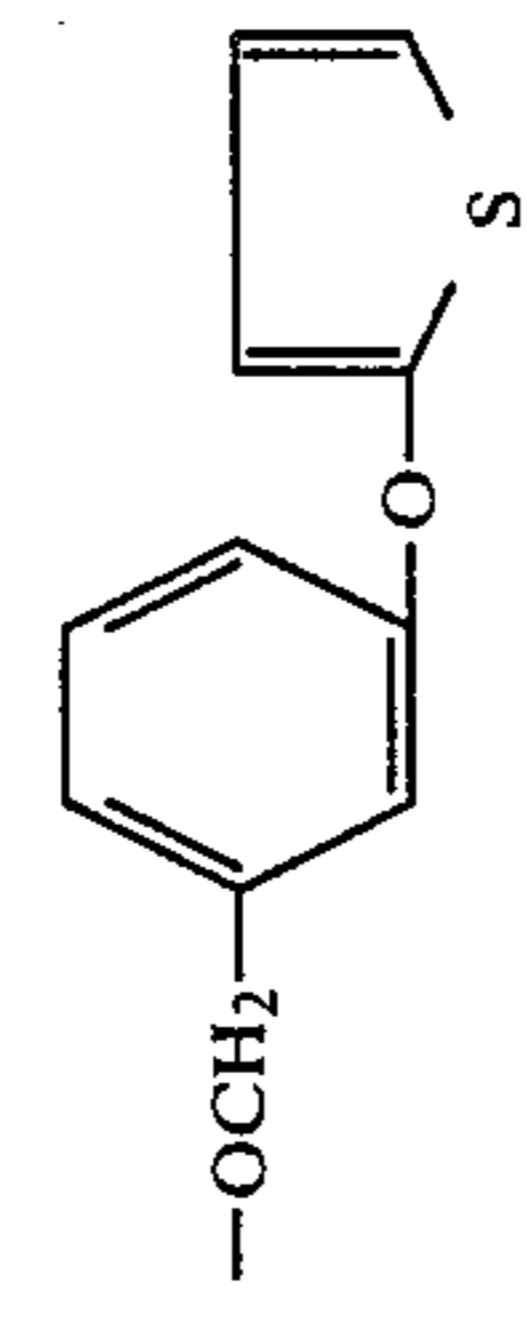
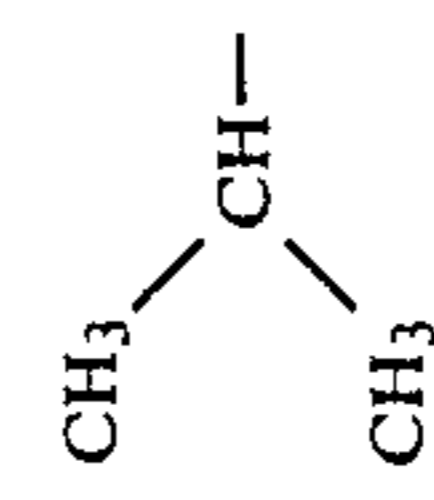
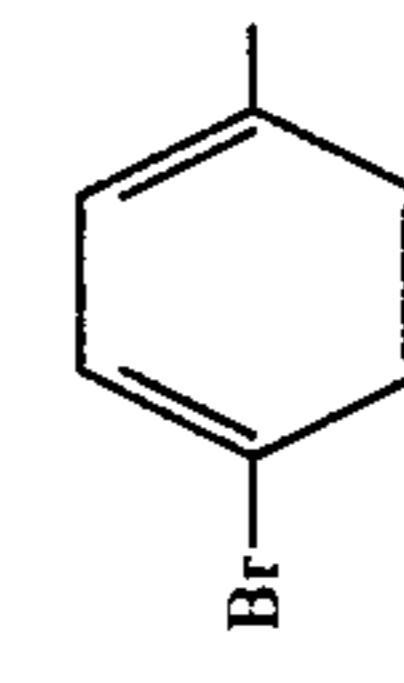
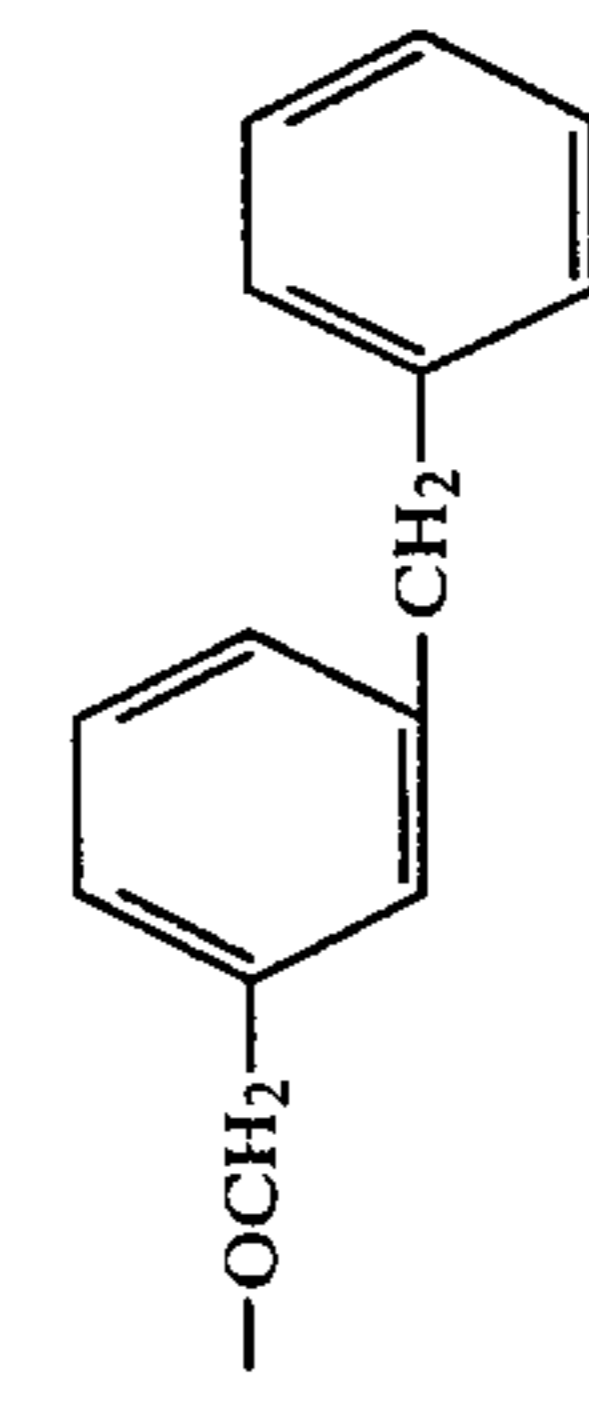
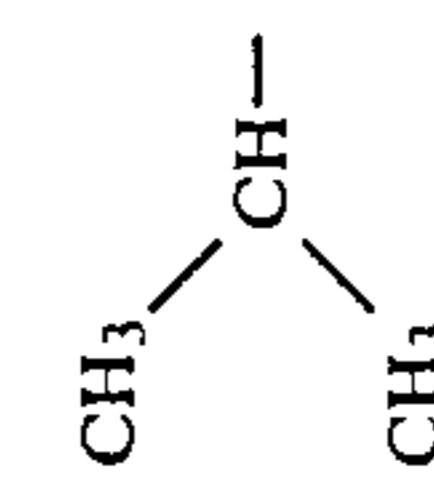
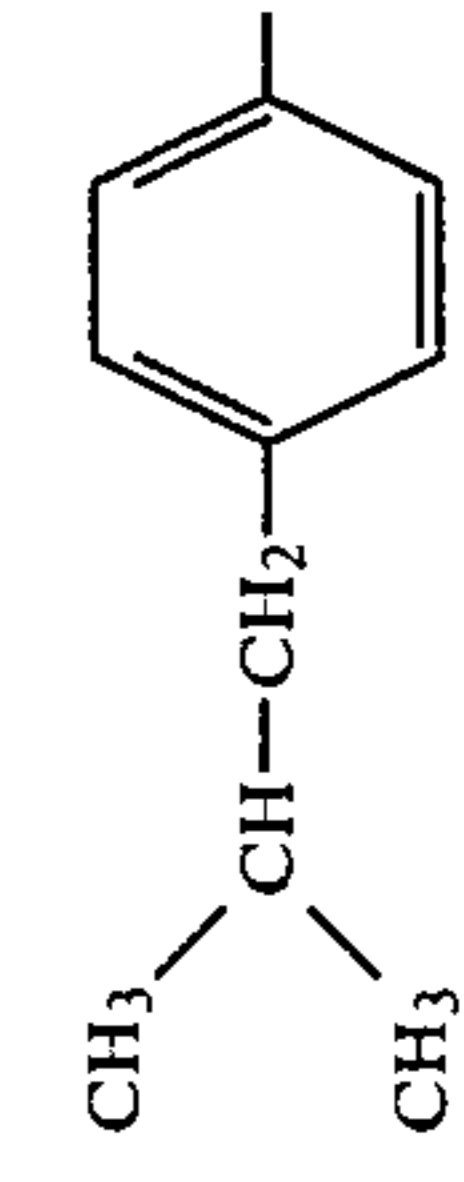
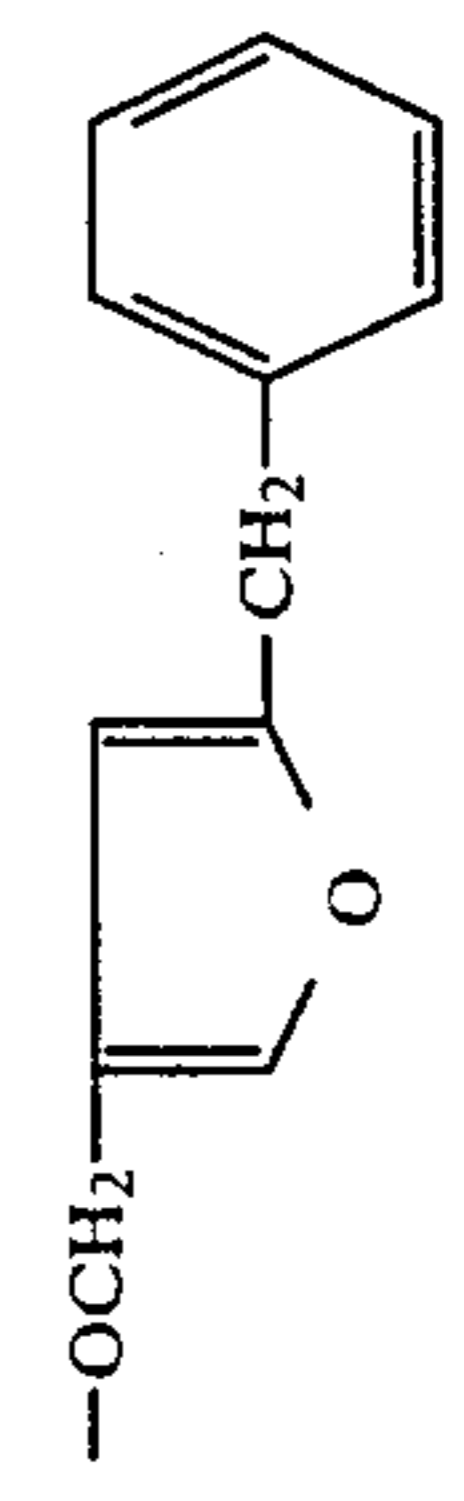
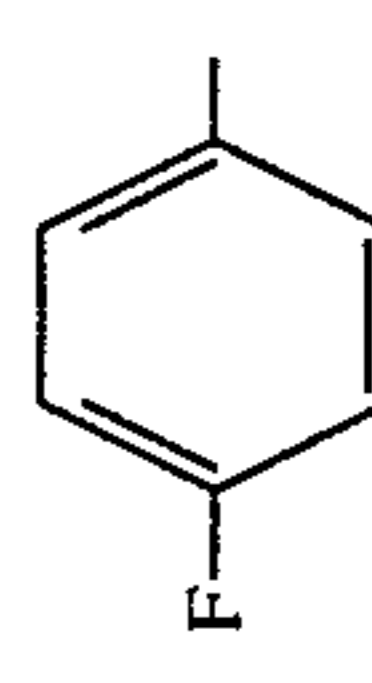
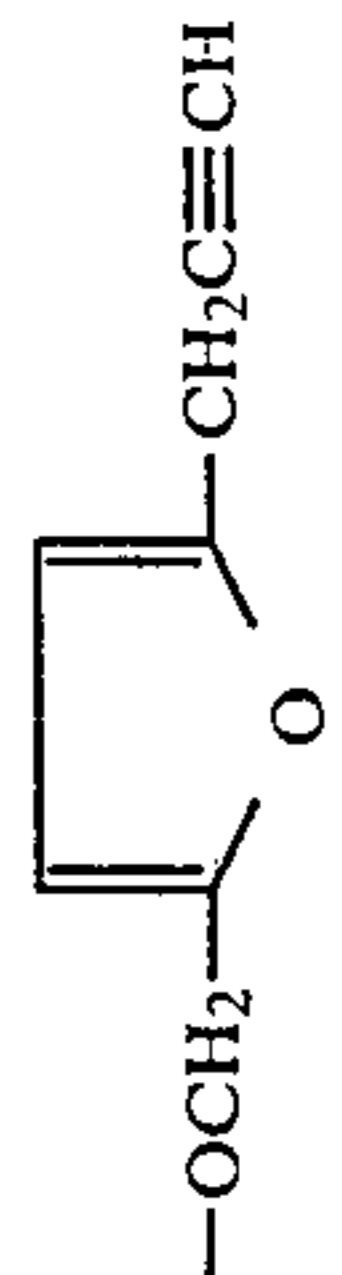
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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
78	149				a	84	C ₂₃ H ₂₆ O ₂ Cl C: 74.68 H: 7.09 Cl: 9.59	<i>n</i> _D ²⁰ 1.5483 74.77 7.12 9.52
79	150				a	89	C ₂₁ H ₂₆ O ₂ S C: 73.64 H: 7.65 S: 9.36	<i>n</i> _D ²⁰ 1.5396 73.66 7.64 9.50
80	151				a	95	C ₂₁ H ₂₁ O ₂ Cl C: 74.00 H: 6.21 Cl: 10.40	<i>n</i> _D ²⁰ 1.5611 74.11 6.23 10.36
81	152				e	82	C ₂₀ H ₂₂ O ₄ NF C: 66.84 H: 6.17 N: 3.90 F: 5.29	<i>n</i> _D ²⁰ 1.5229 66.88 6.18 4.01 5.30
82	153				b	80	C ₂₁ H ₂₆ O ₃ C: 77.27 H: 8.03	<i>n</i> _D ²⁰ 1.5387 77.20 8.09
83	154				a	89	C ₂₀ H ₂₁ O ₃ F C: 73.15 H: 6.45 F: 5.79	<i>n</i> _D ²⁰ 1.5316 73.19 6.50 5.81
84	155				a	93	C ₂₄ H ₂₆ O ₃ C: 79.53 H: 7.23	<i>n</i> _D ²⁰ 1.5452 79.55 7.33

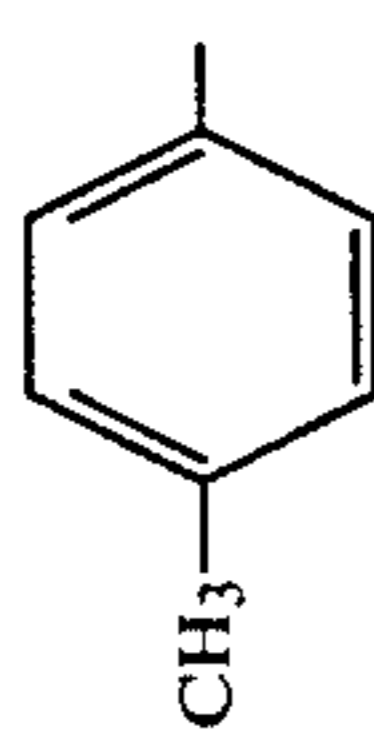
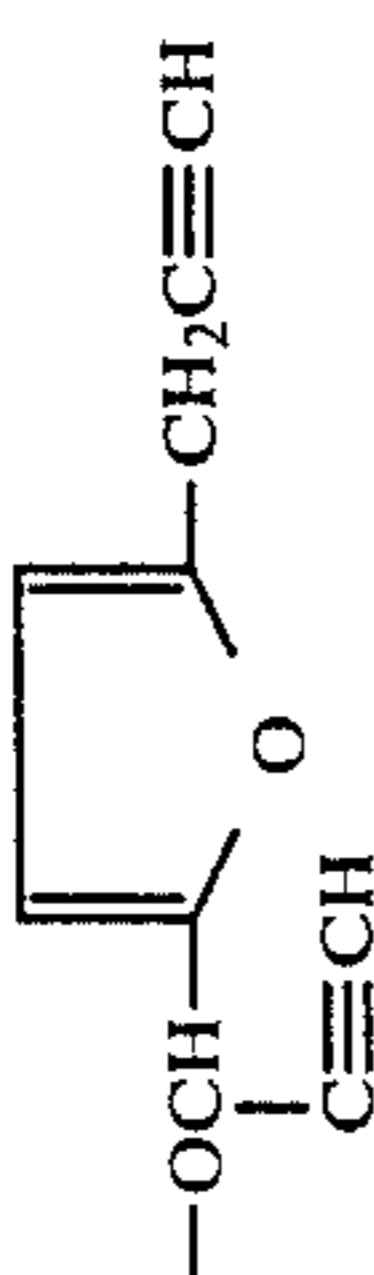
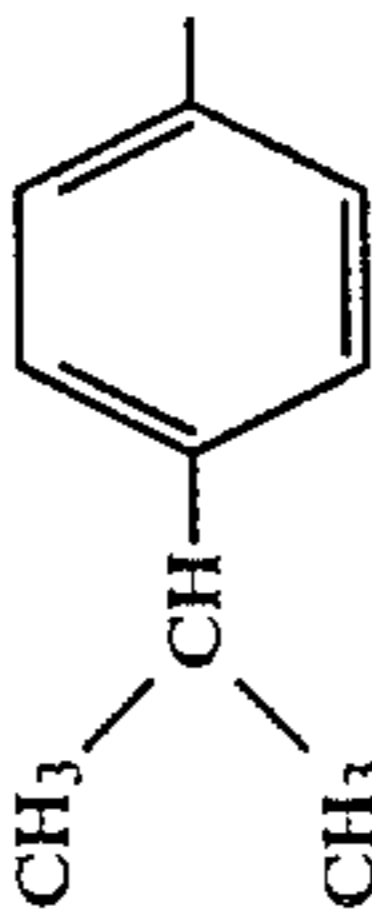
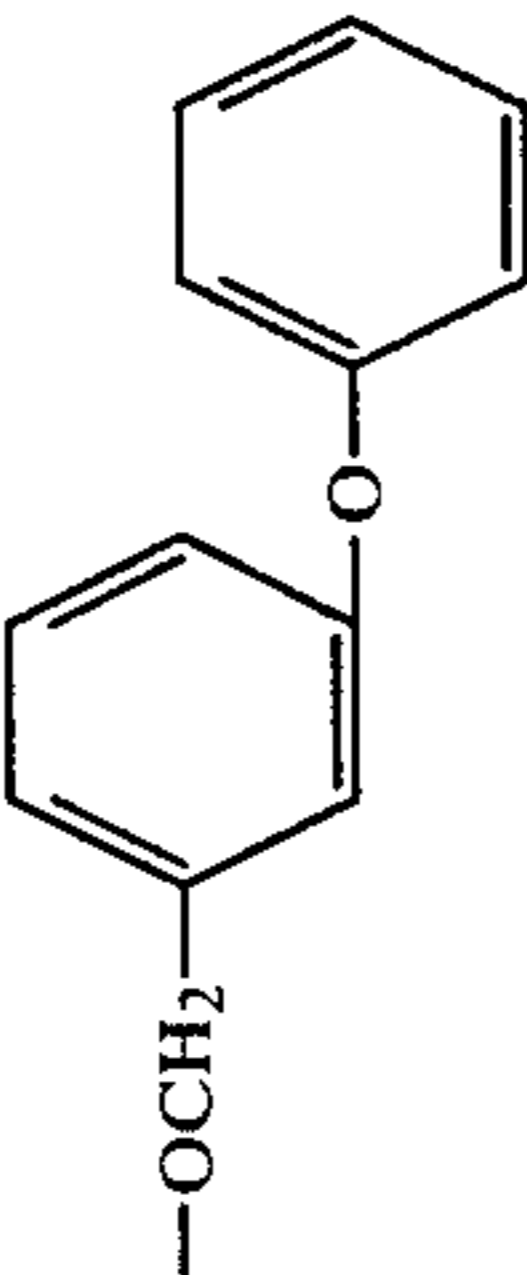
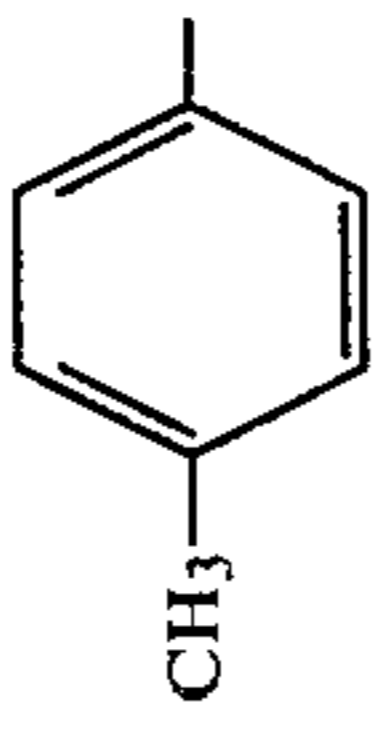
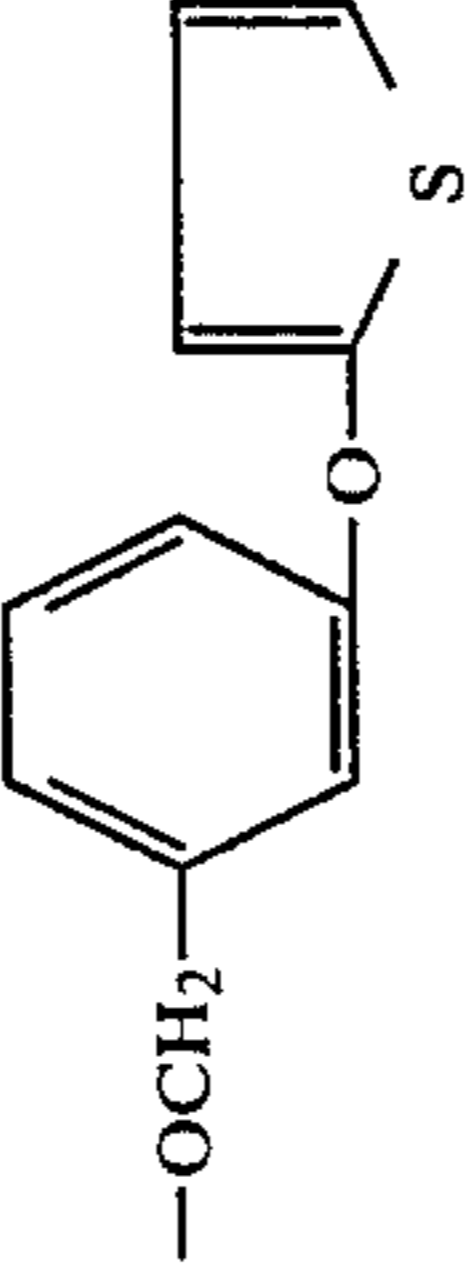
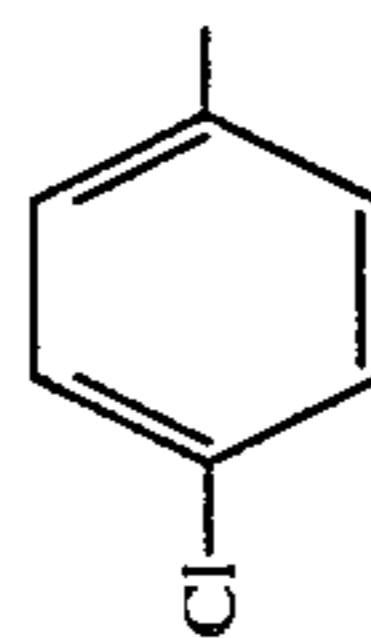
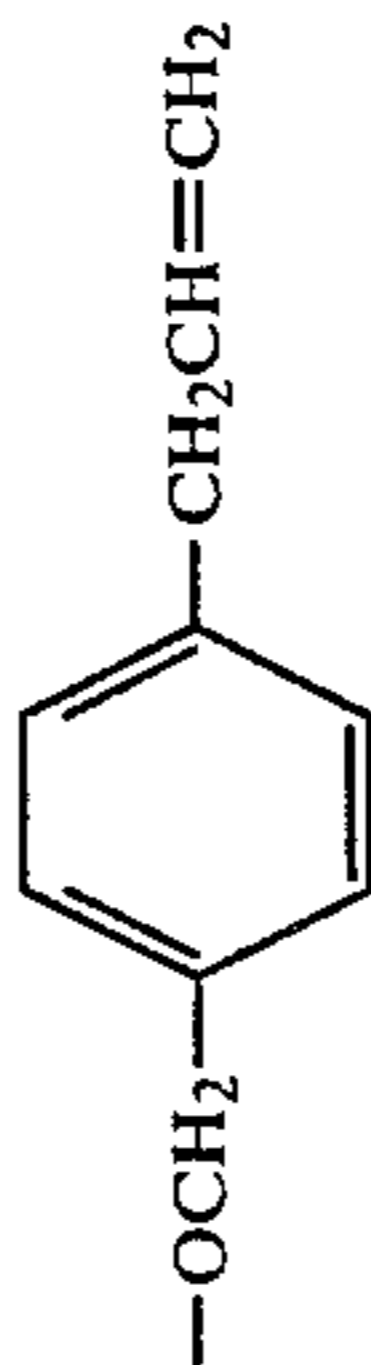
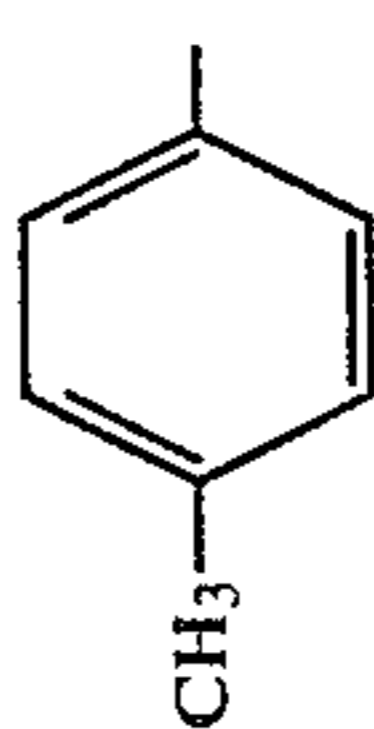
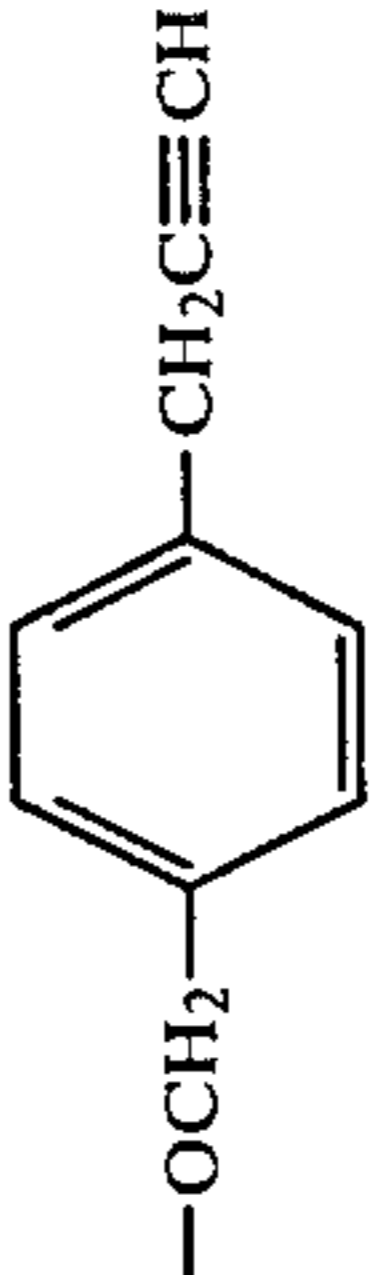
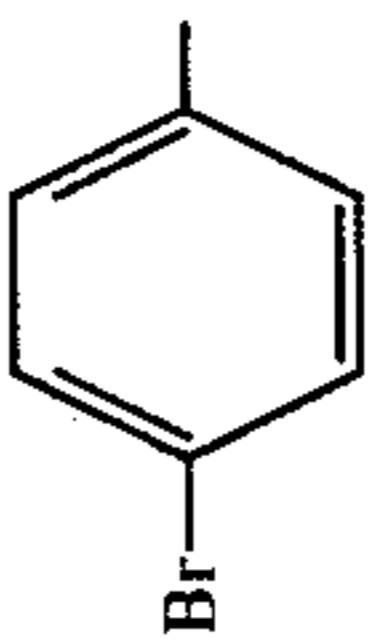
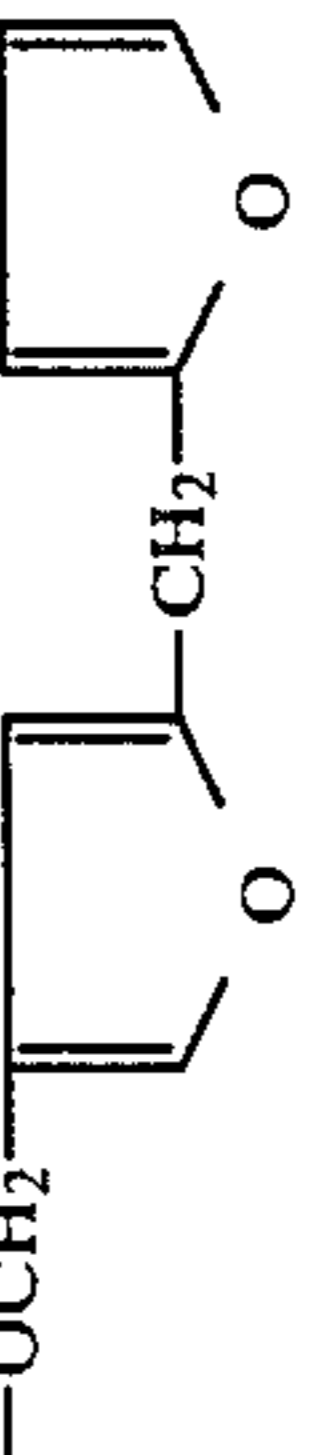
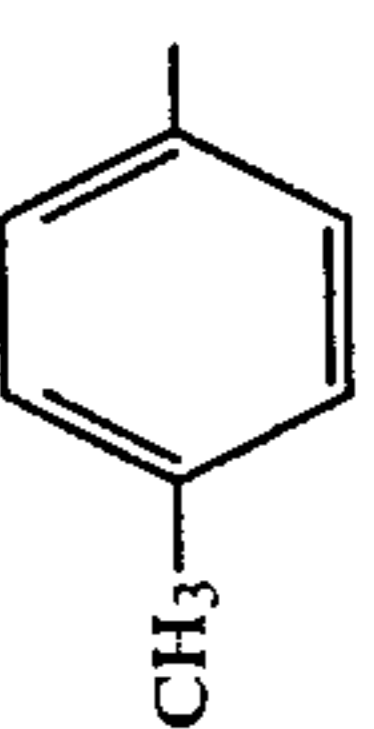
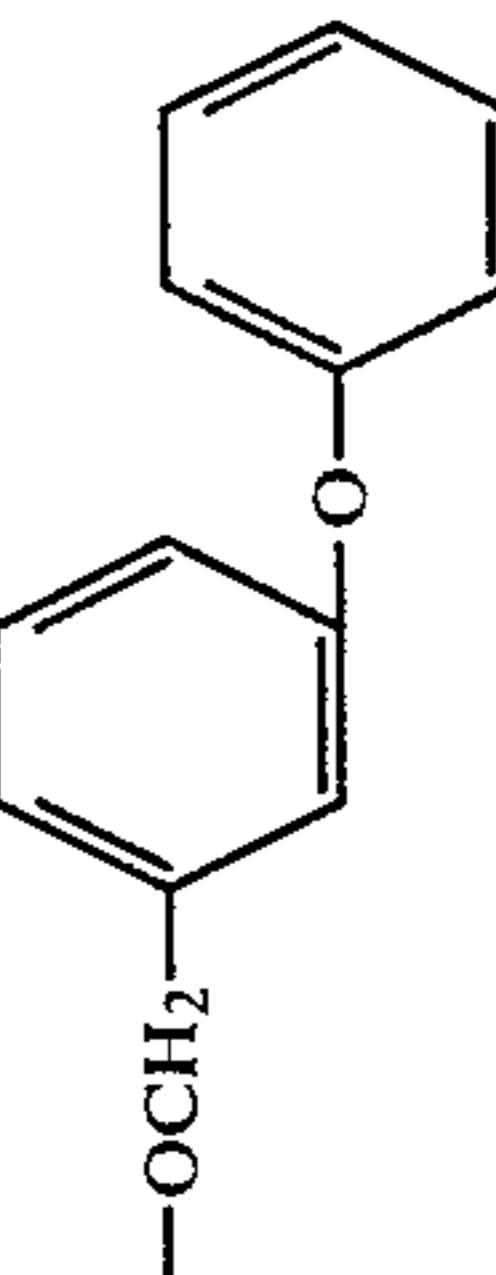
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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
85	156				a	92	C ₂₃ H ₂₃ O ₃ Cl C: 72.15 H: 6.05 Cl: 9.26	n _D ²⁰ 1.5448 72.18 6.00 9.32
86	157				a	92	C ₂₃ H ₂₃ O ₃ I C: 58.24 H: 4.89 I: 26.76	n _D ²⁰ 1.5352 58.22 4.93 26.88
87	158				a	96	C ₂₄ H ₂₆ O ₂ S C: 76.15 H: 6.92 S: 8.47	n _D ¹⁹ 1.5311 76.16 7.03 8.46
88	159				a	86	C ₂₆ H ₃₀ O ₂ S C: 76.80 H: 7.44 S: 7.89	n _D ¹⁹ 1.5342 76.77 7.51 7.82
89	160				a	81	C ₂₂ H ₂₂ O ₃ C: 79.01 H: 6.63	n _D ²⁶ 1.5262 79.22 6.58
90	161				a	76	C ₂₃ H ₂₄ O ₃ C: 79.28 H: 6.94	n _D ²² 1.5291 79.33 7.10
91	162				a	92	C ₂₂ H ₂₆ O ₂ C: 81.95 H: 8.13	n _D ²⁰ 1.5366 82.00 8.18

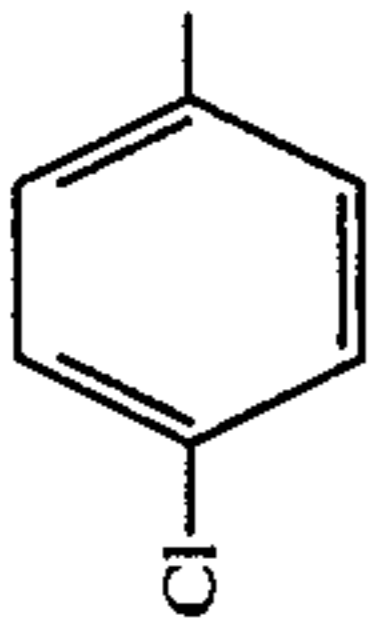
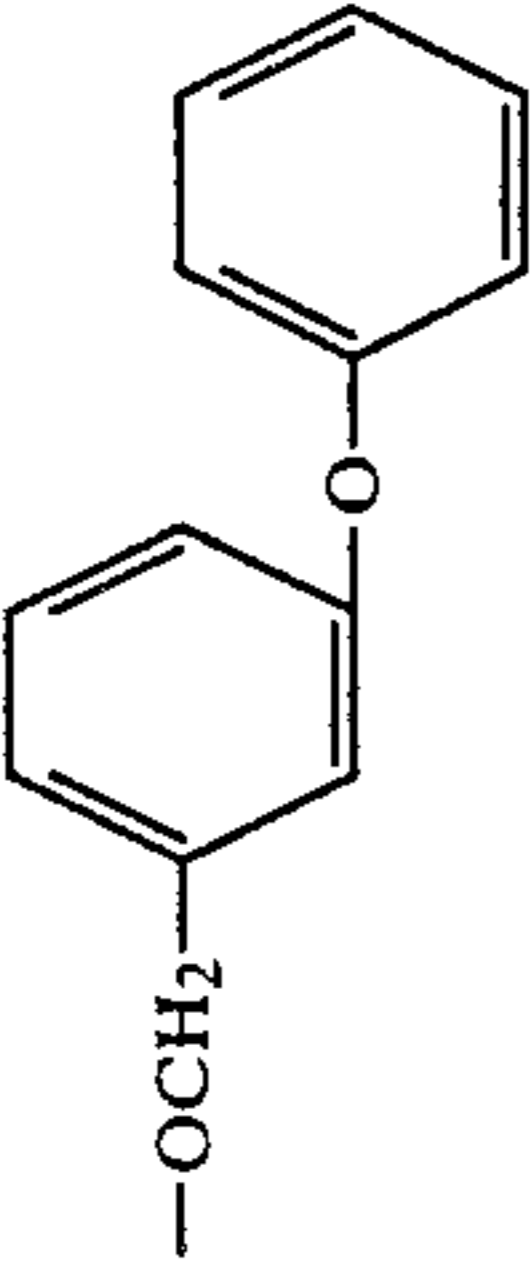
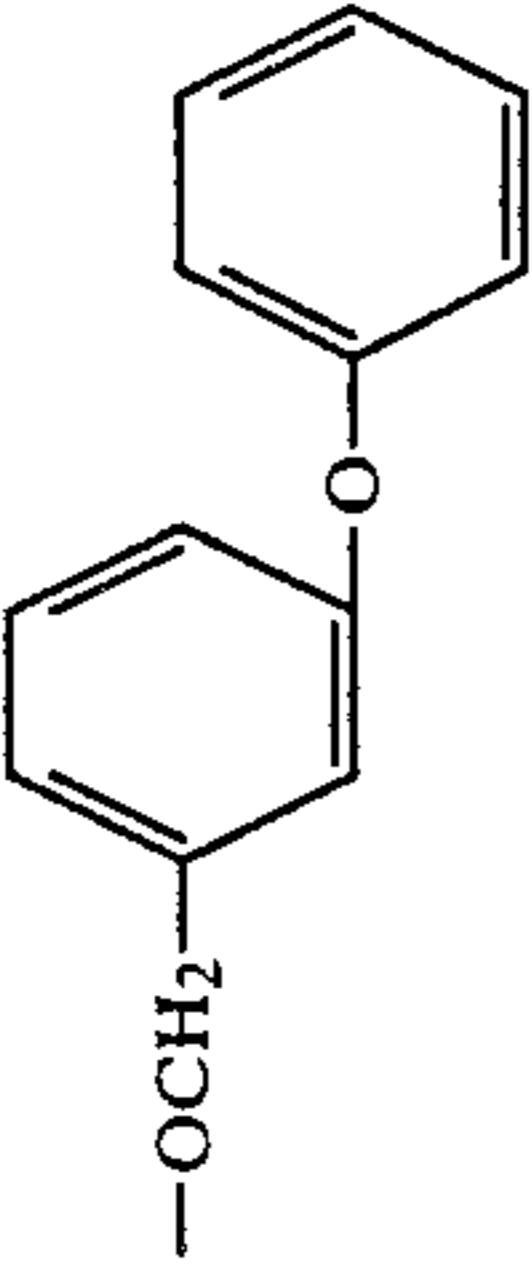
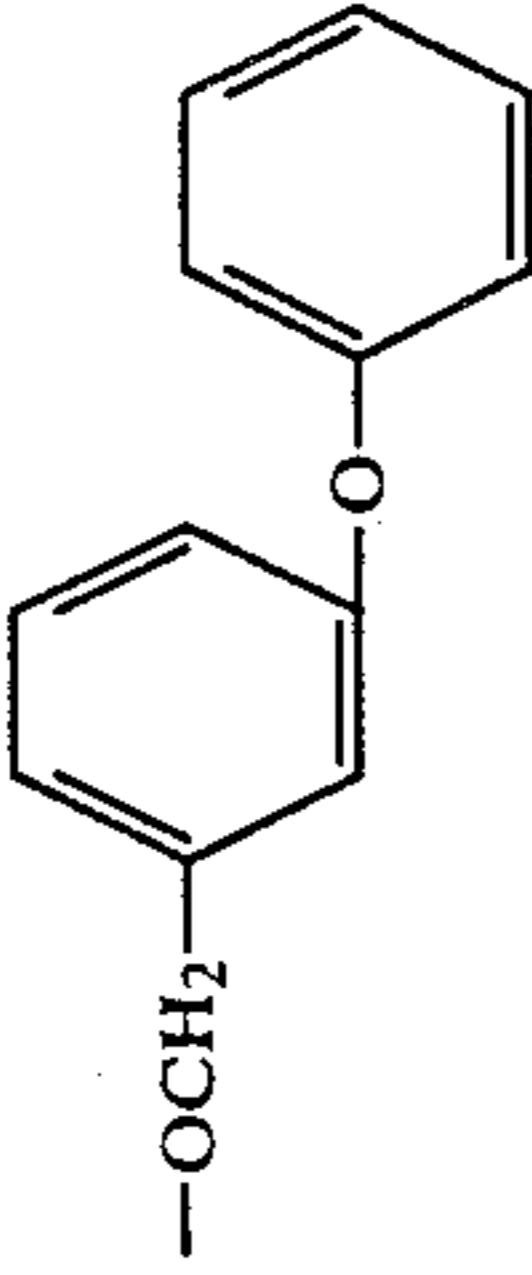
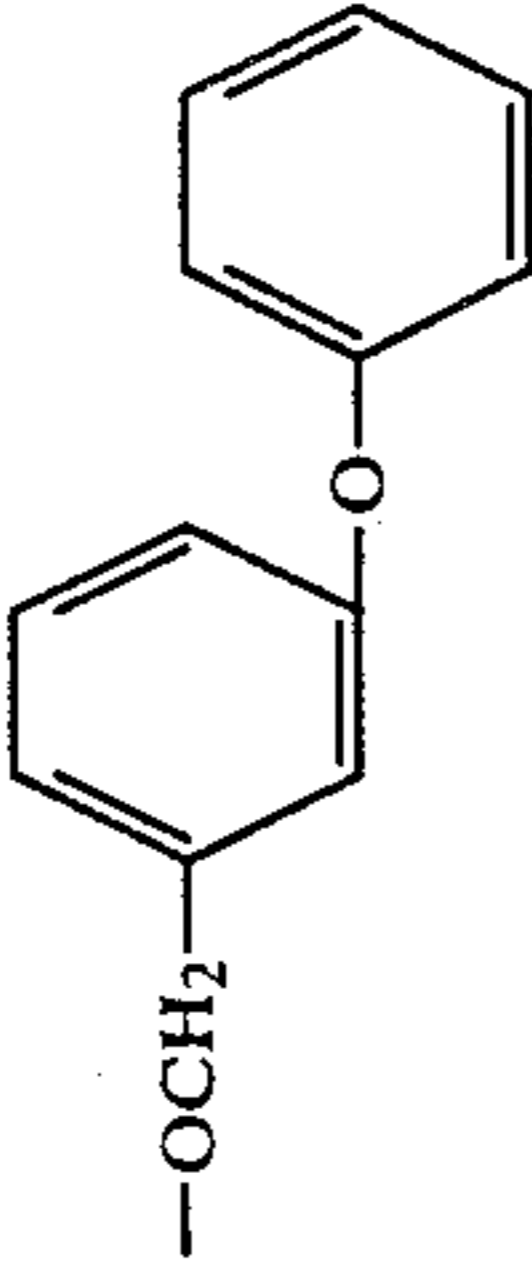
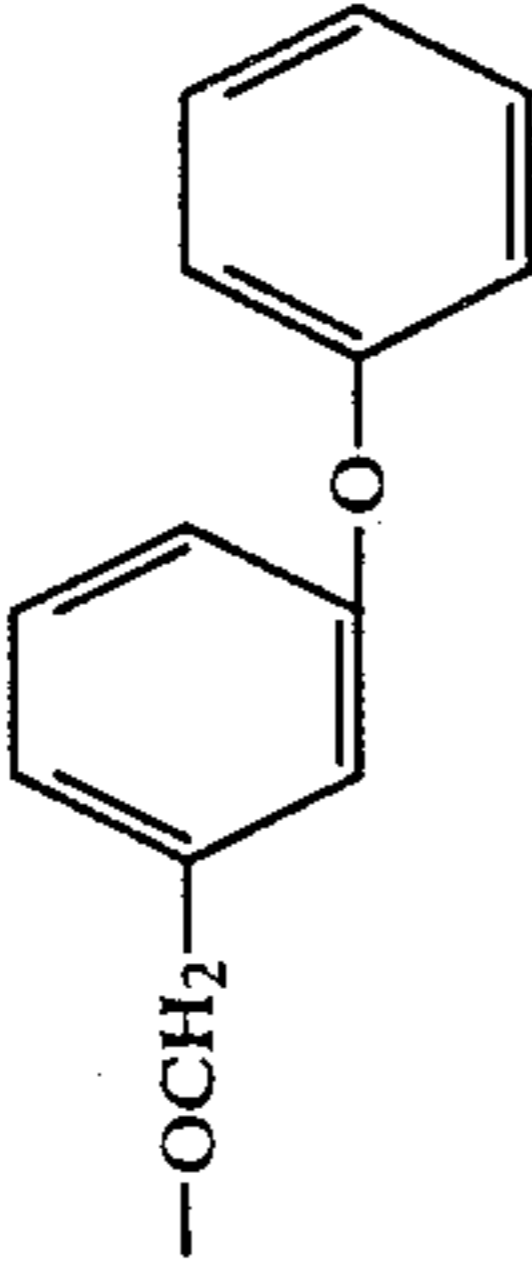
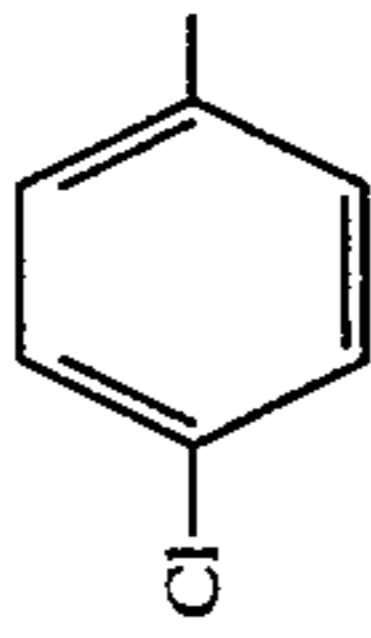
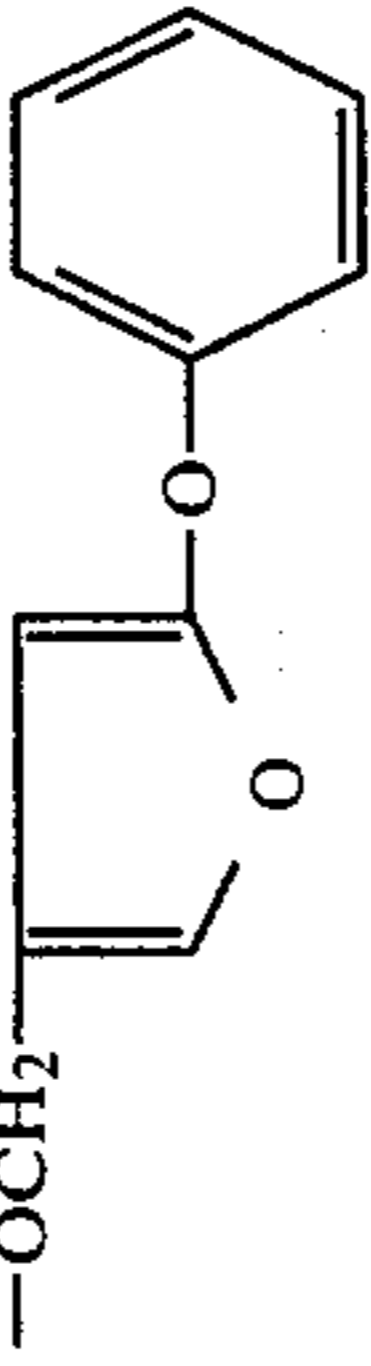
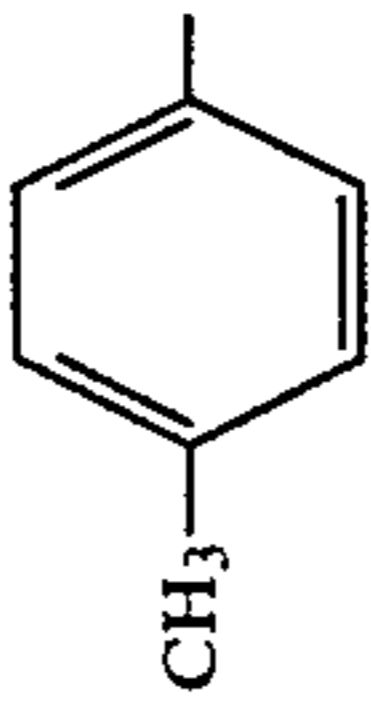
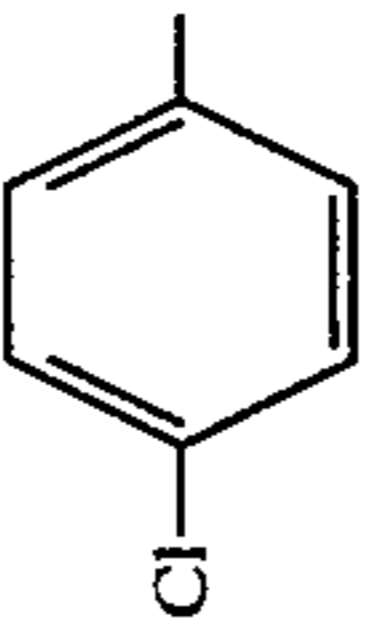
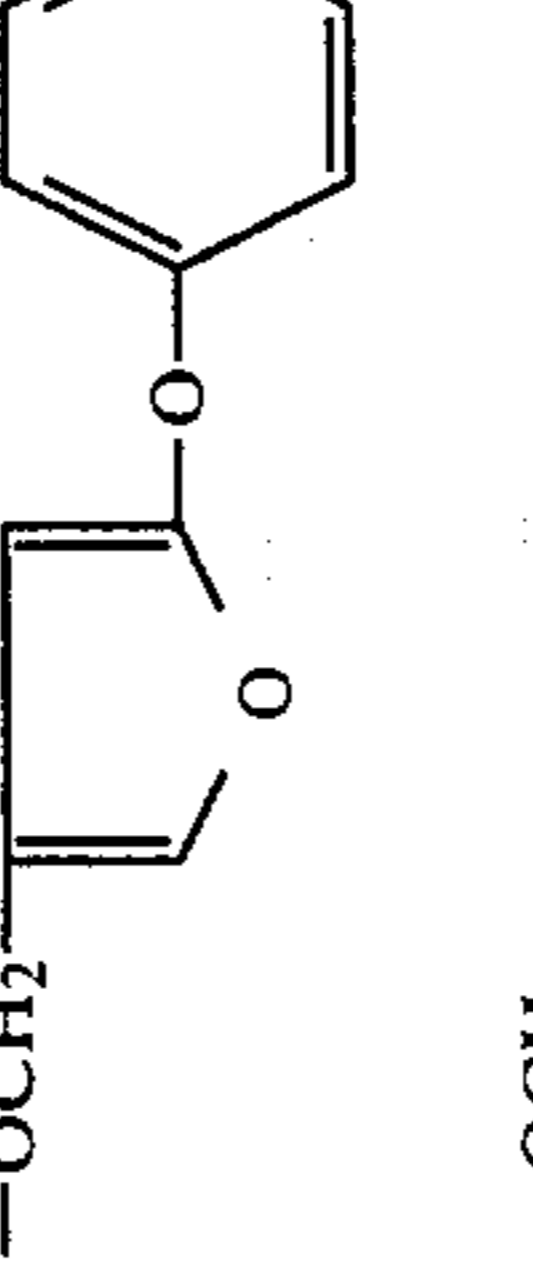
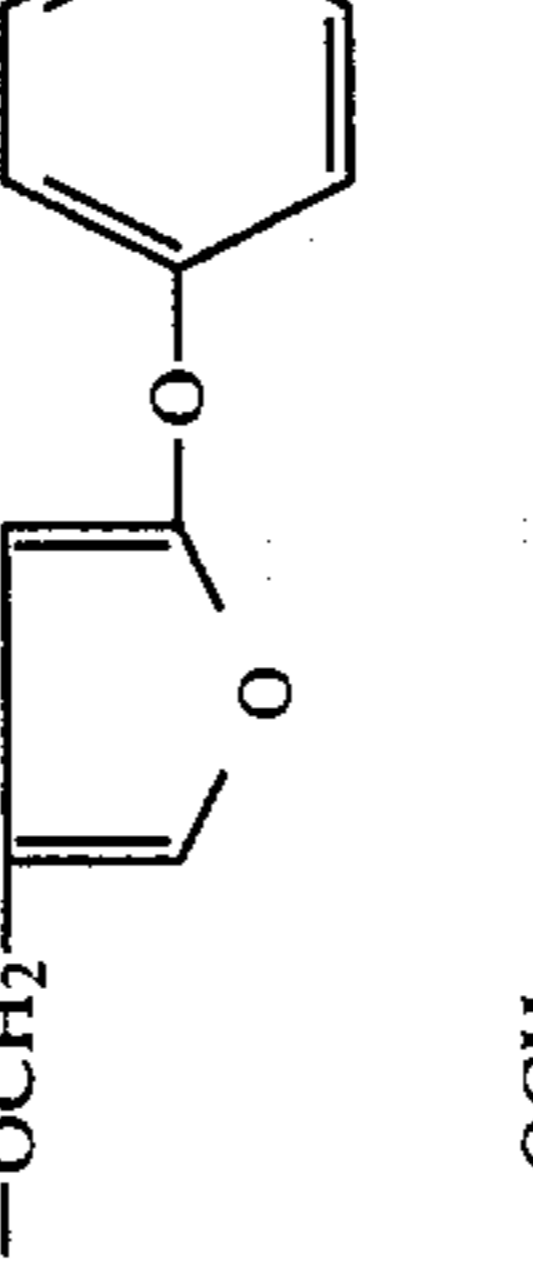
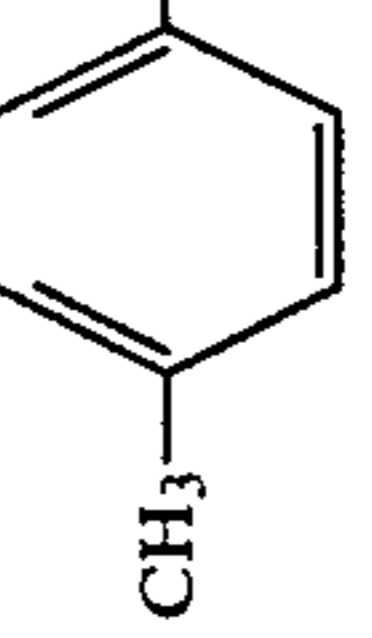
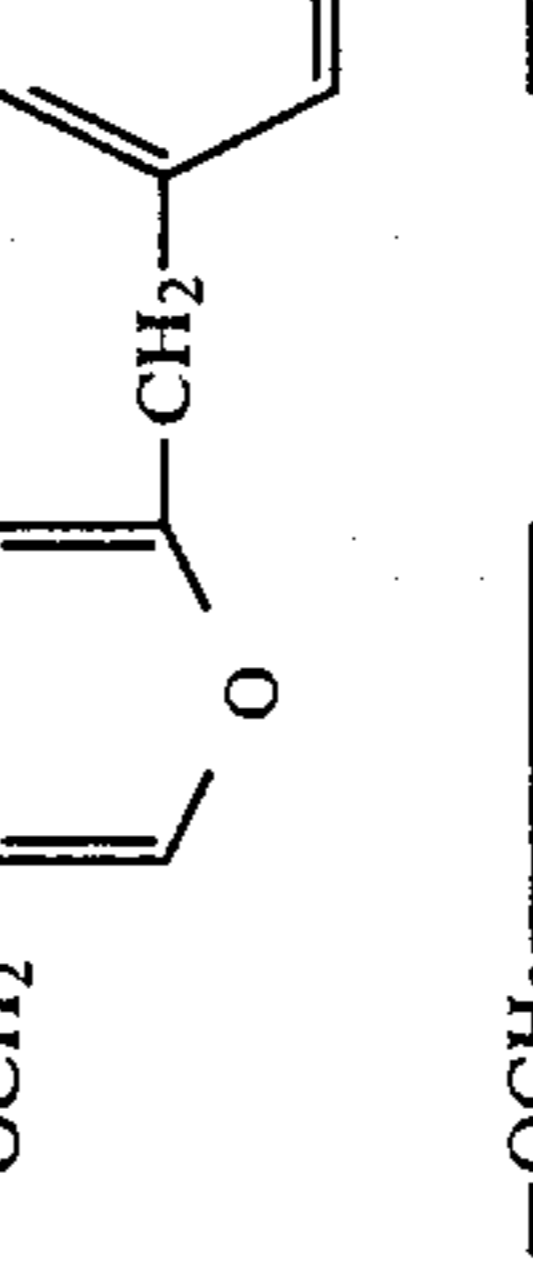
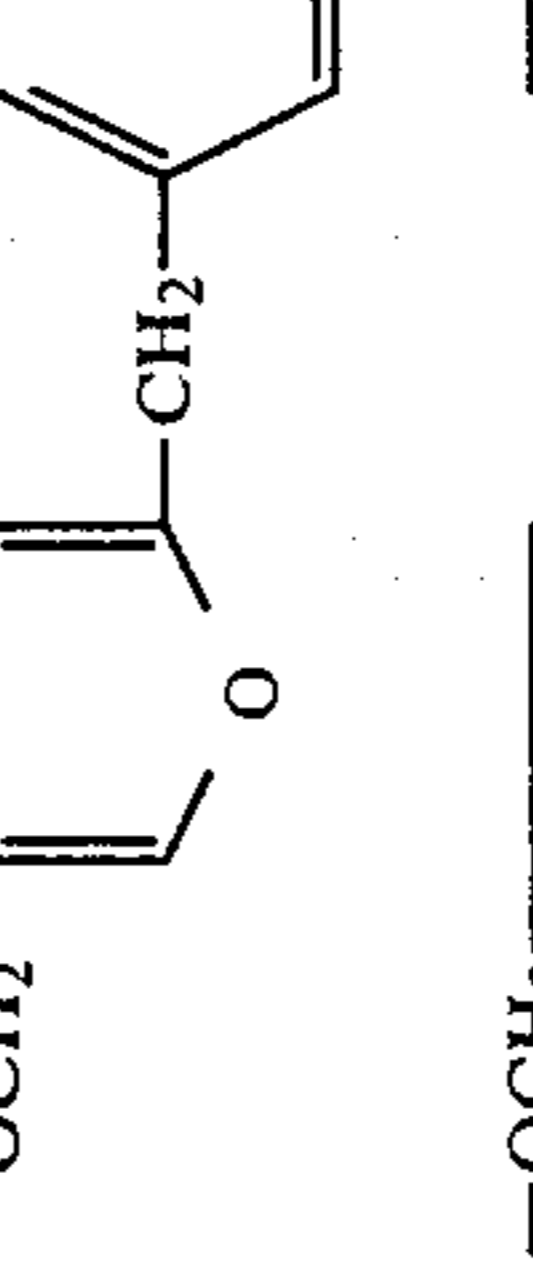
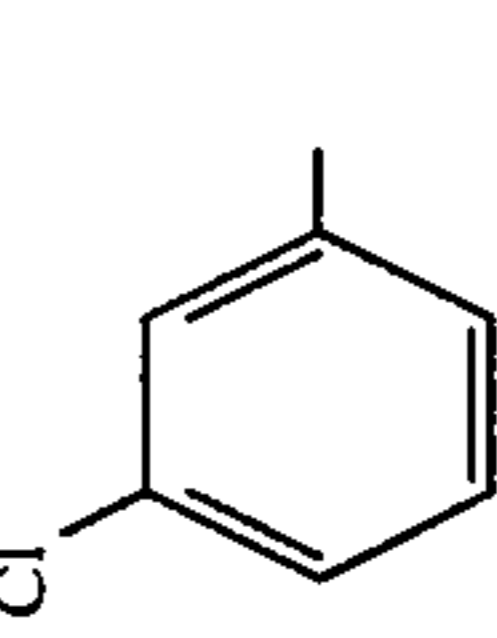
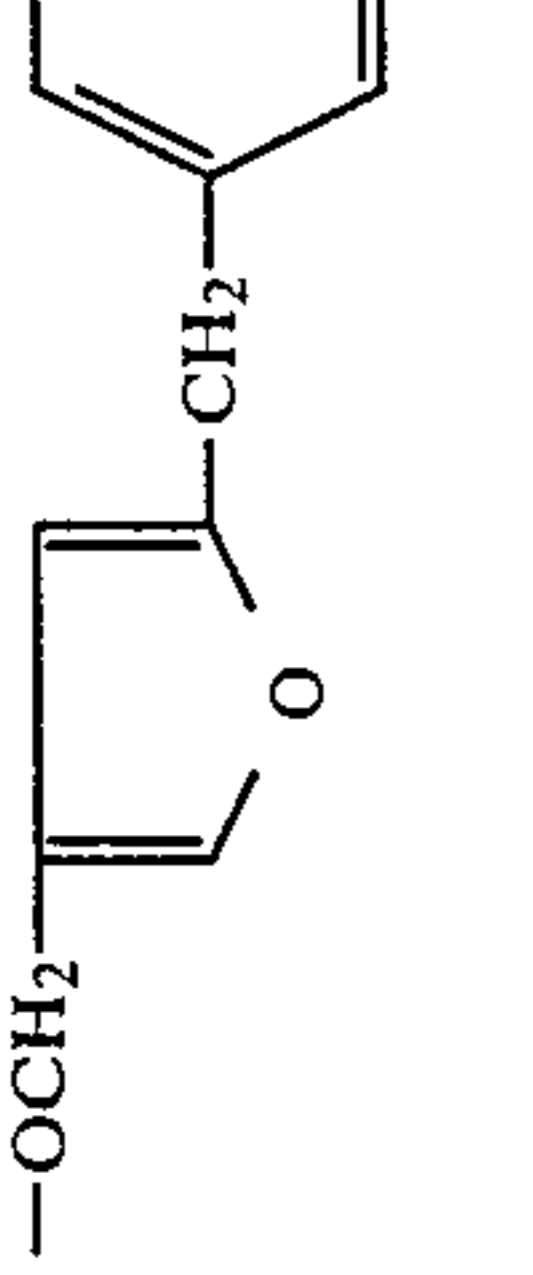
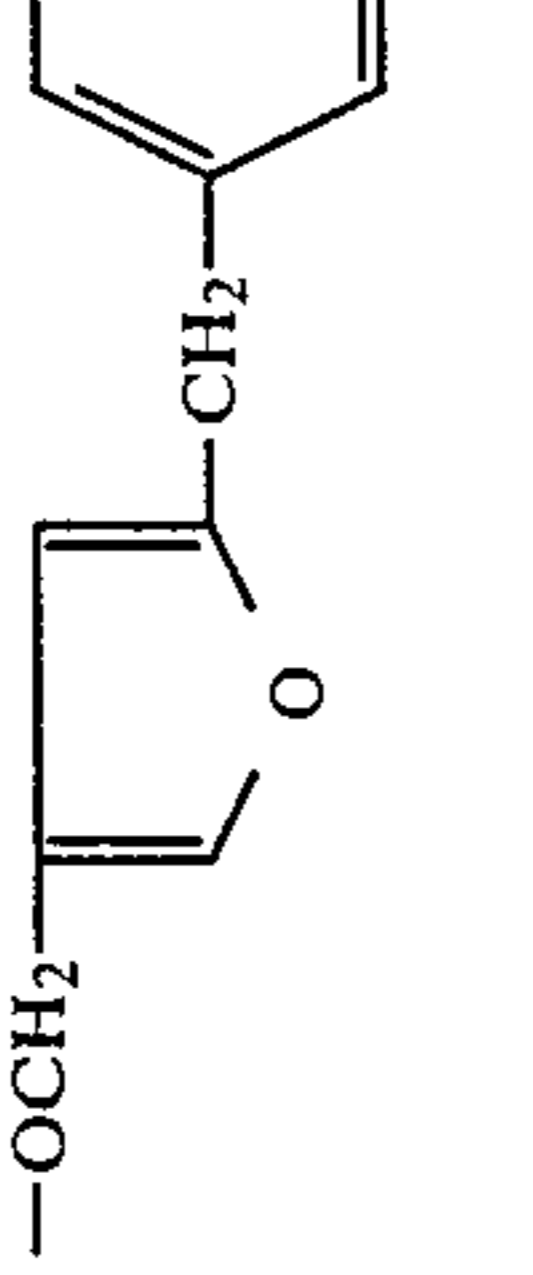
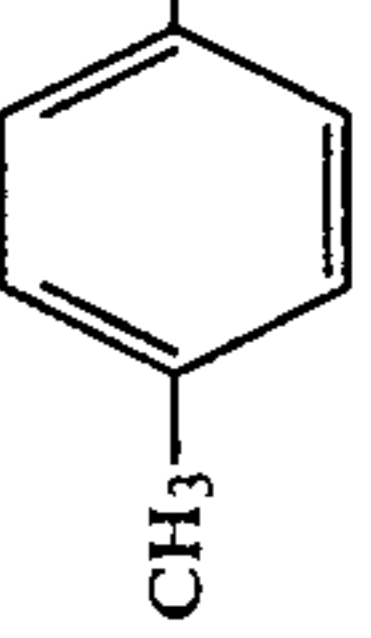
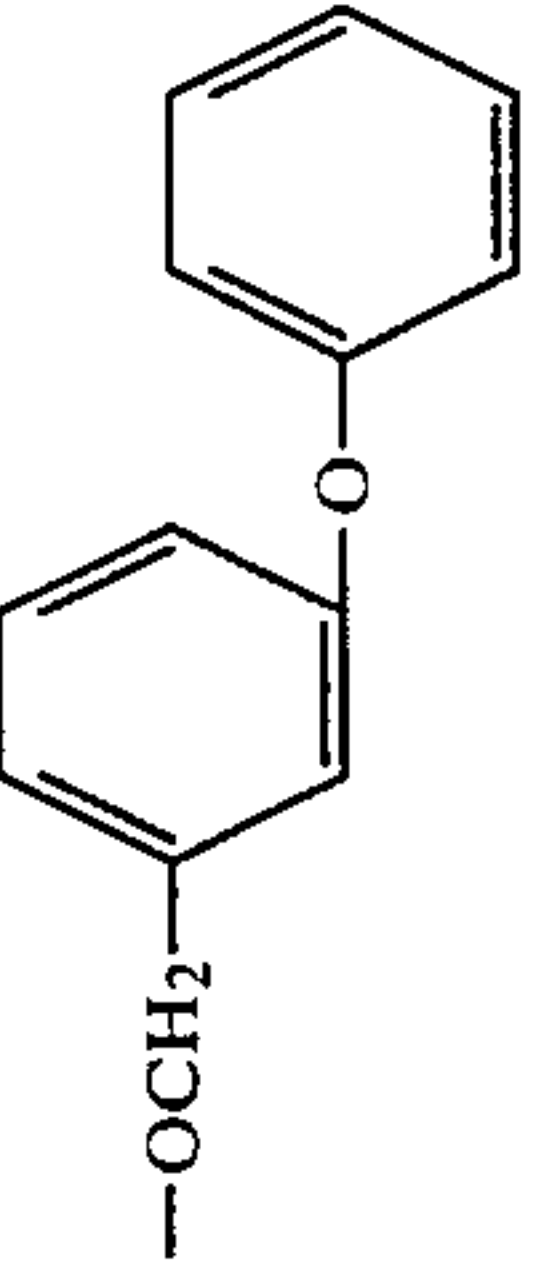
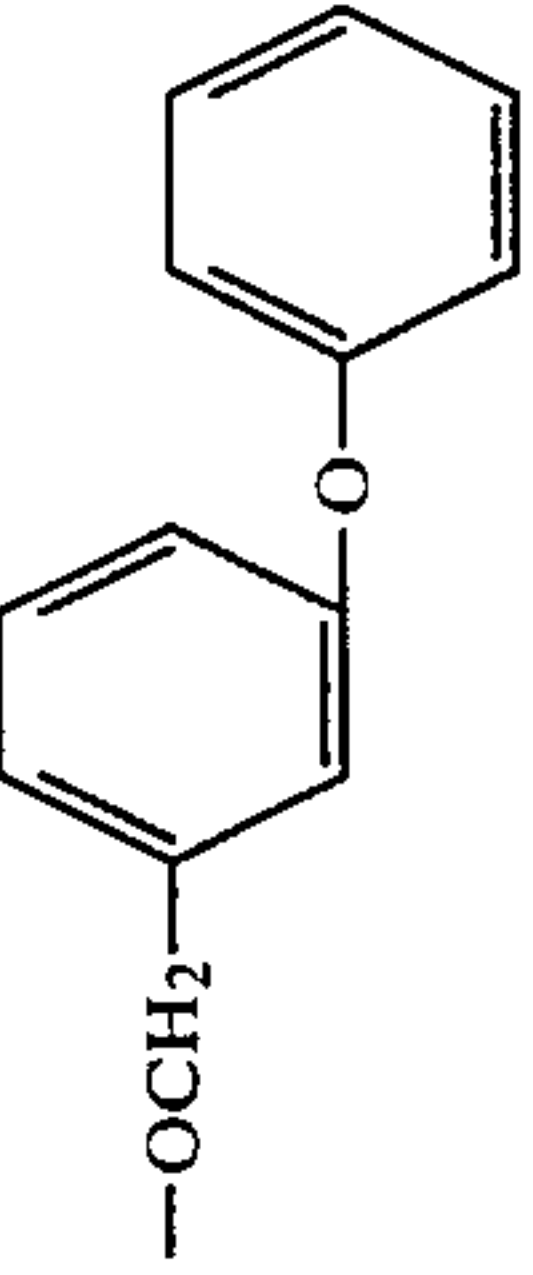
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Preparation Exp. No.	Comp. No.	Z	Y	X	Chemical Structure	Esterification Method	Yield (%)	Physical Properties	
								Calcd. (%)	Found (%)
92	163			X		a	87	C ₂₆ H ₃₀ O ₃ S C: 73.90 H: 7.16 S: 7.59	n_D^{20} 1.5361 73.92 7.17 7.63
93	164					a	88	C ₂₂ H ₂₁ O ₃ SF C: 68.73 H: 5.51 S: 8.34 F: 4.94	n_D^{20} 1.4837 68.77 5.48 8.36 4.90
94	165					a	88	C ₂₂ H ₂₂ O ₃ NF C: 74.42 H: 5.50 N: 3.47 F: 4.71	n_D^{20} 1.5233 74.49 5.56 3.49 4.79
95	166					a	92	C ₂₂ H ₂₁ O ₃ SBr C: 59.33 H: 4.75 S: 7.20 Br: 17.94	n_D^{20} 1.5811 59.36 4.79 7.11 17.92
96	167					d	91	C ₂₅ H ₂₃ O ₂ Br C: 68.97 H: 5.32 Br: 18.36	n_D^{20} 1.5802 68.96 5.33 18.40
97	168					a	93	C ₂₇ H ₃₂ O ₃ C: 80.16 H: 7.97	n_D^{20} 1.5387 80.15 7.96
98	169	C ₂ H ₅ -				a	79	C ₁₈ H ₁₇ O ₃ F C: 71.98 H: 5.71 F: 6.33	n_D^{20} 1.5218 71.99 5.76 6.27

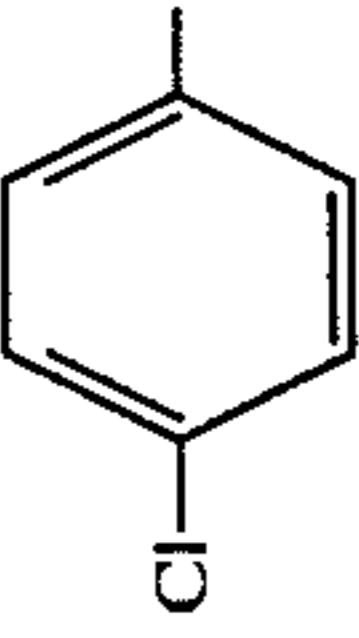

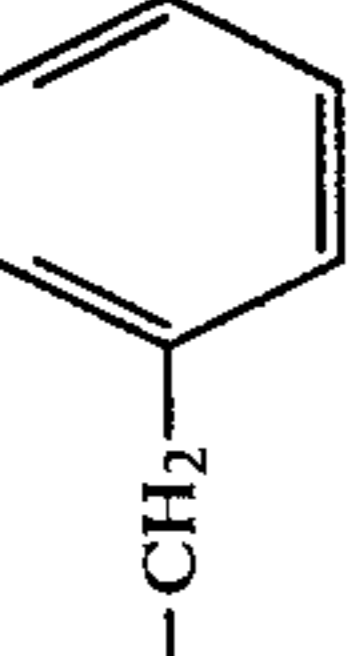
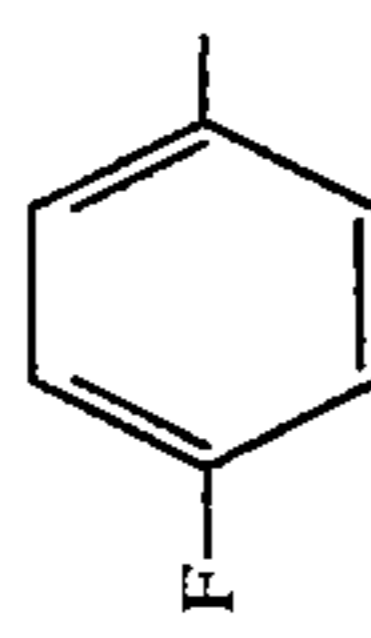

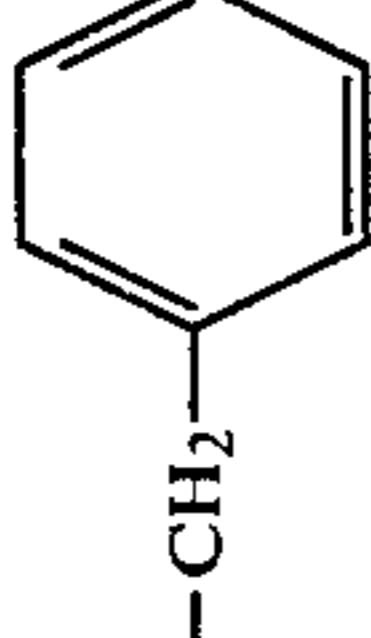
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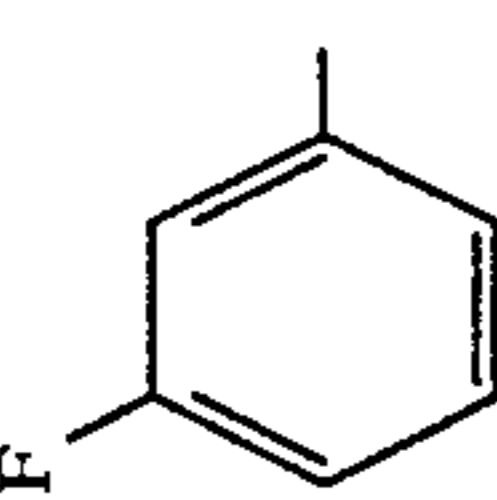
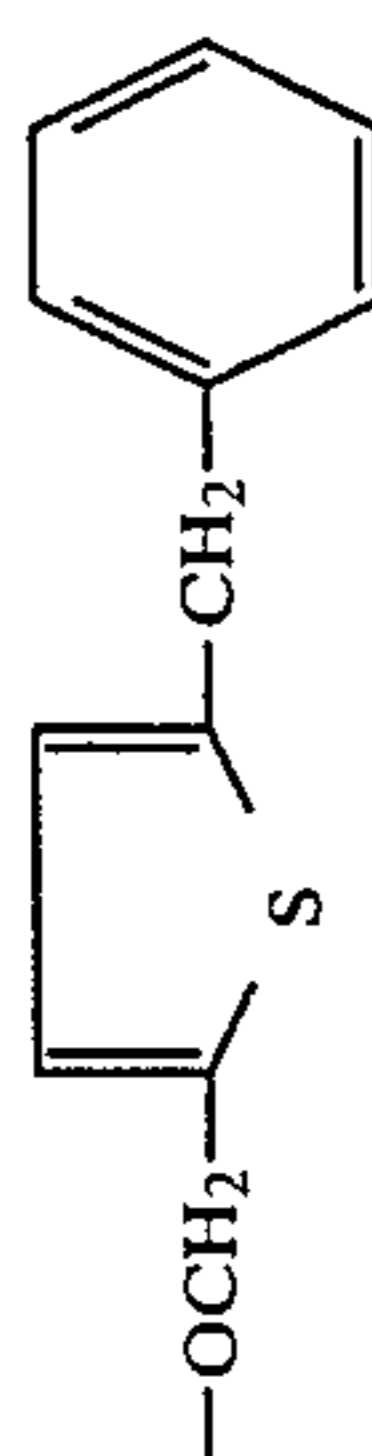
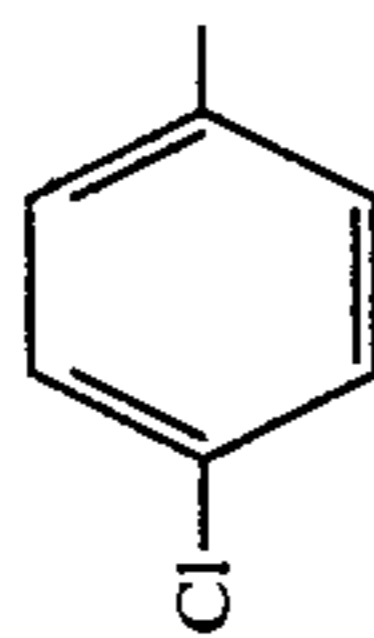
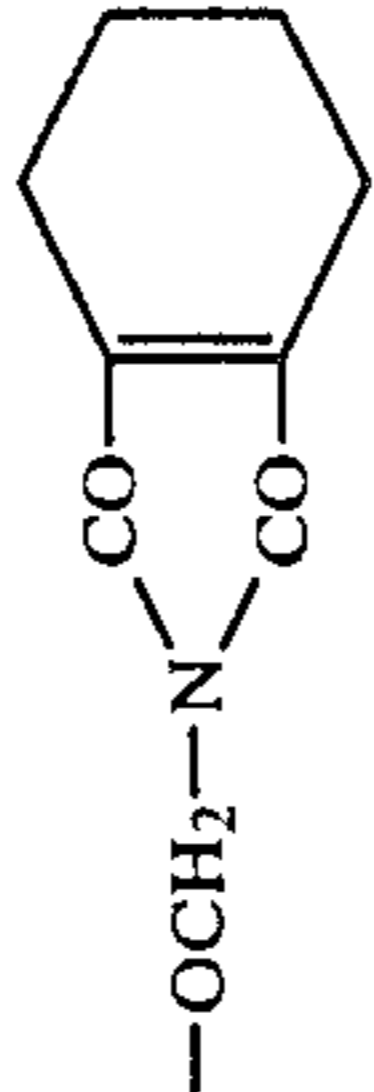
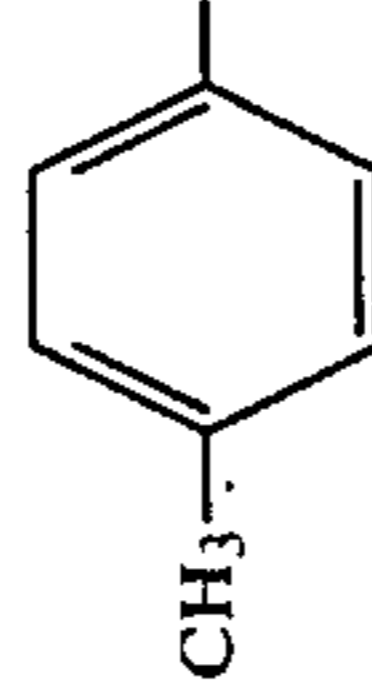
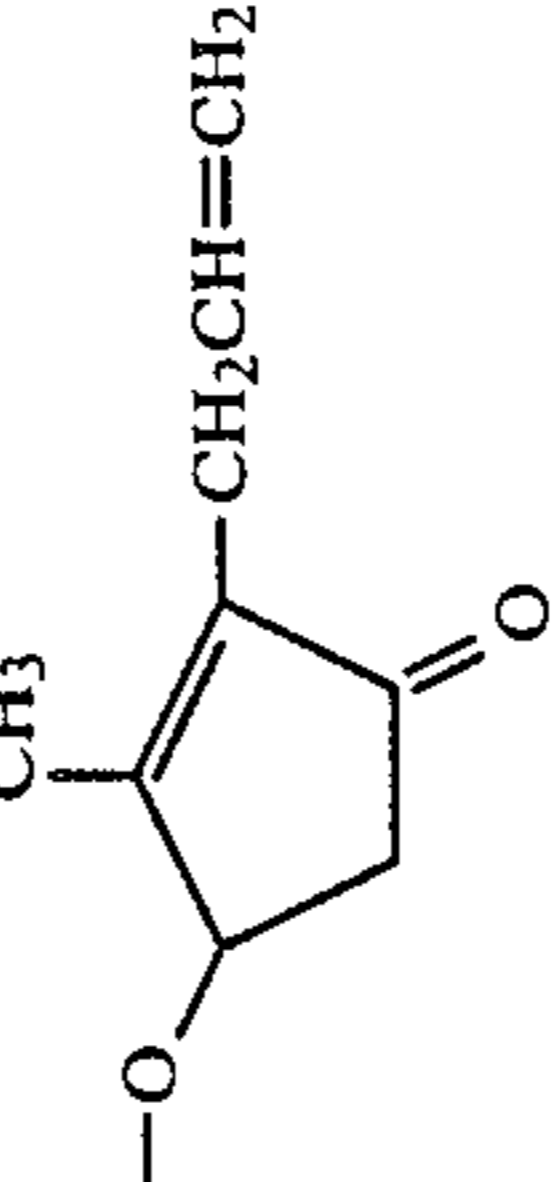
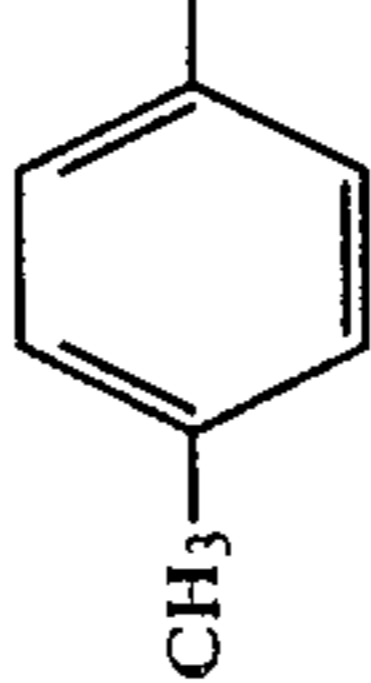
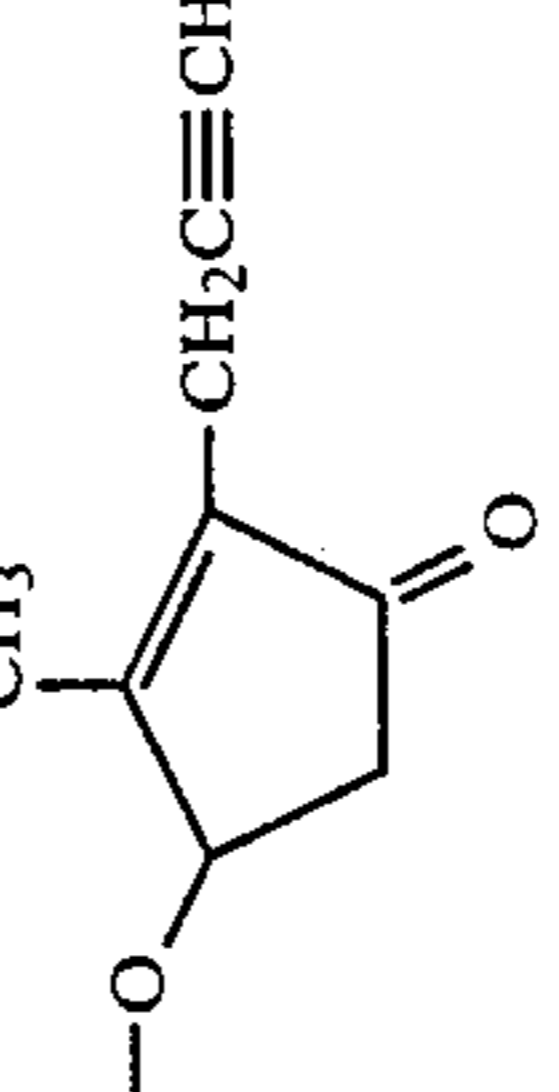
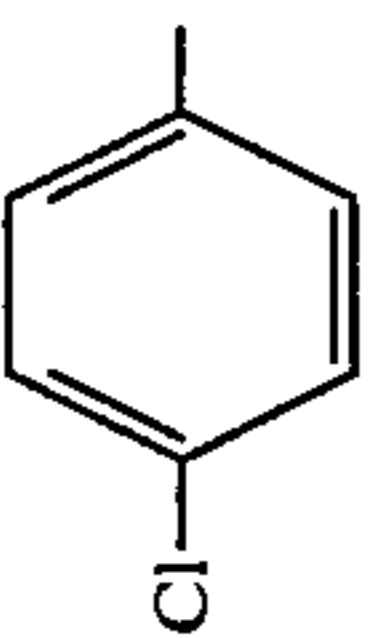
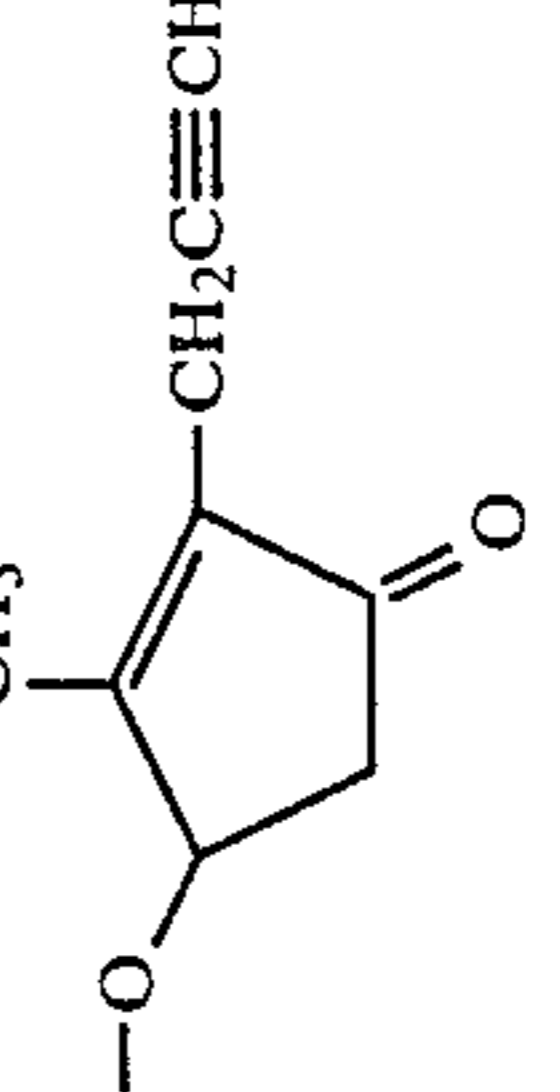
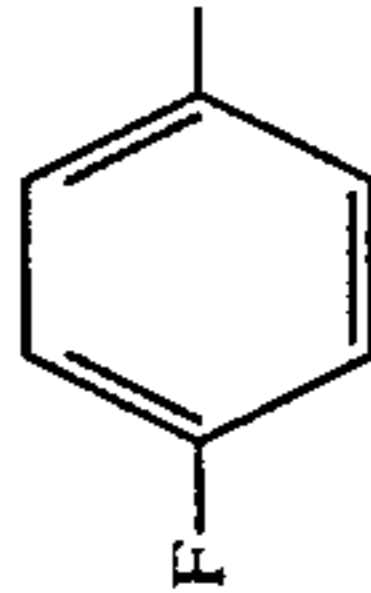
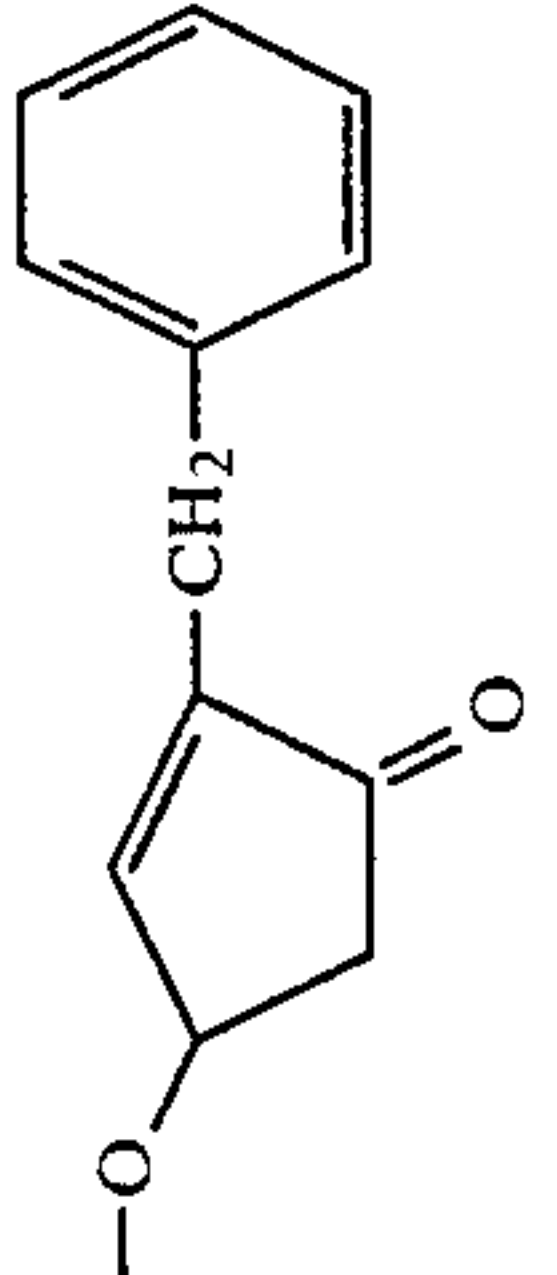
Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
99	170	C ₂ H ₅			b	82	C ₂₁ H ₂₀ O ₃ C: 78.72 H: 6.29	n _D ²⁶ 1.5261 78.77 6.22
100	171	C ₂ H ₅			c	89	C ₂₆ H ₂₈ O ₃ C: 80.38 H: 7.27	n _D ¹⁹ 1.5536 80.29 7.33
101	172	C ₂ H ₅			e	82	C ₂₂ H ₂₂ O ₃ S C: 72.10 H: 6.05 S: 8.75	n _D ¹⁹ 1.5499 72.16 6.07 8.82
102	173	C ₂ H ₅			a	93	C ₂₀ H ₂₁ O ₂ Cl C: 73.05 H: 6.44 Cl: 10.78	n _D ²⁰ 1.5236 73.09 6.46 10.69
103	174	C ₂ H ₅			a	92	C ₂₁ H ₂₂ O ₂ C: 83.32 H: 7.24	n _D ²⁰ 1.5252 82.40 7.26
104	175	C ₂ H ₅			a	86	C ₂₀ H ₁₉ O ₄ Br C: 59.56 H: 4.75 Br: 19.82	n _D ²⁰ 1.5233 59.61 4.77 20.91
105	176	C ₂ H ₅			e	92	C ₂₄ H ₂₄ O ₃ C: 79.97 H: 6.71	n _D ²⁰ 1.5711 80.05 6.77

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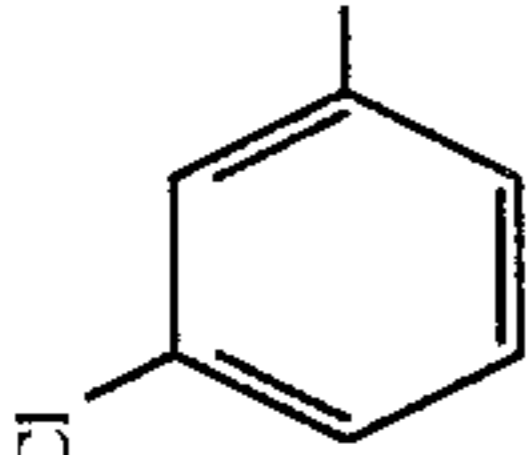
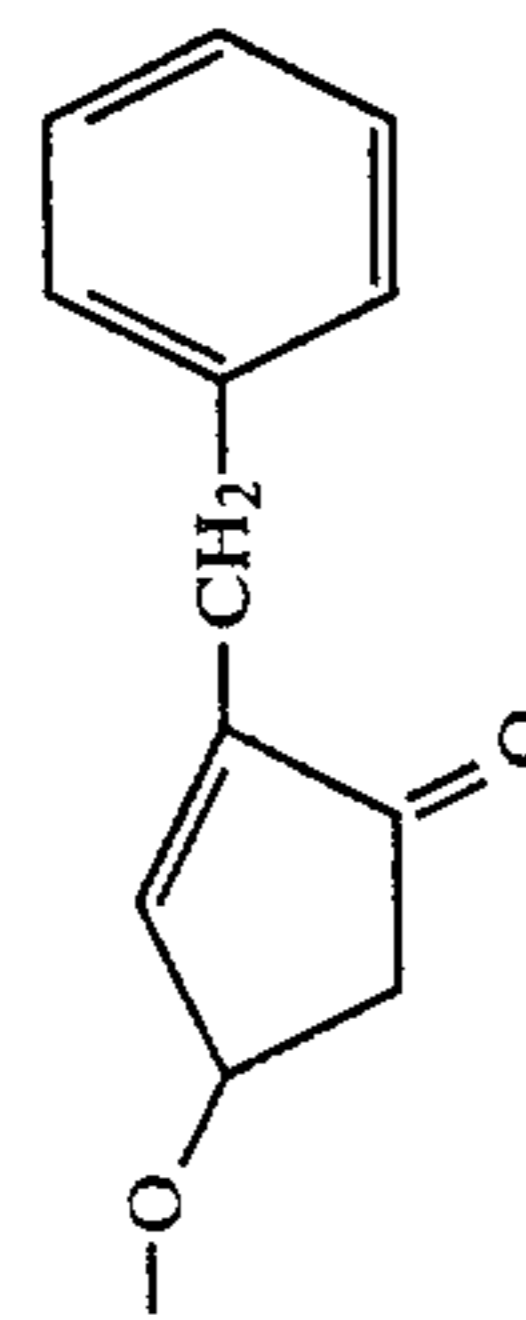
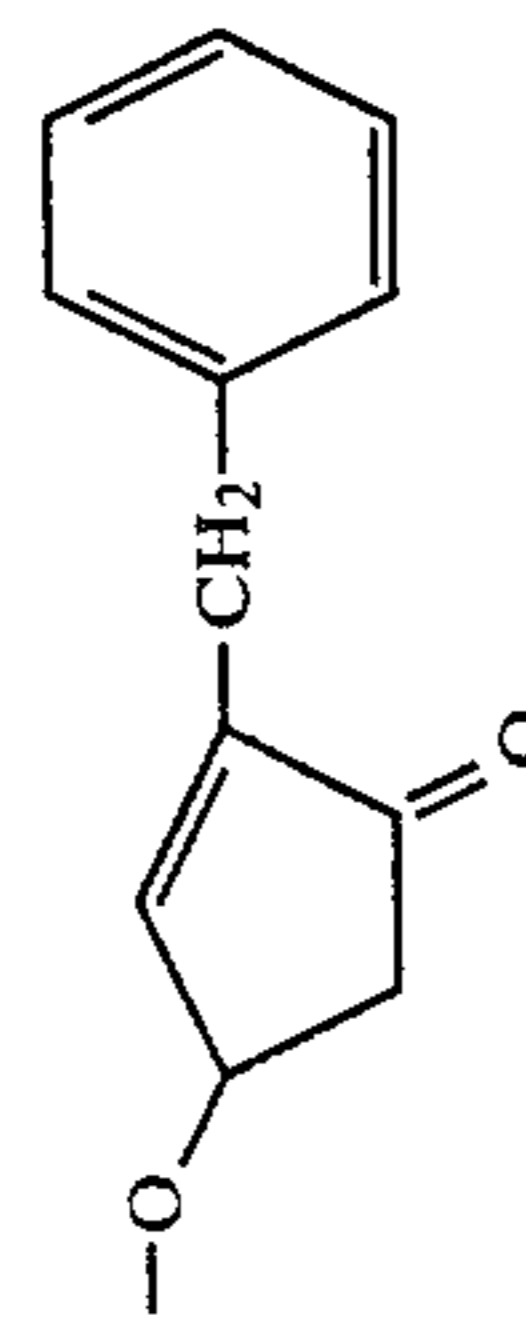
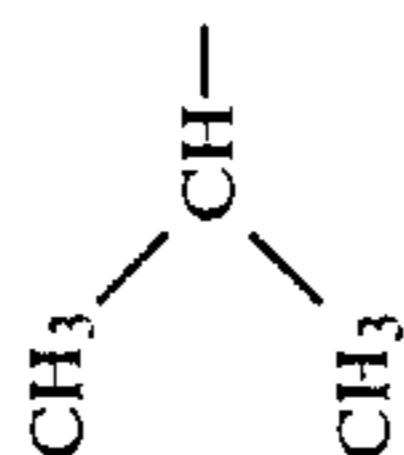
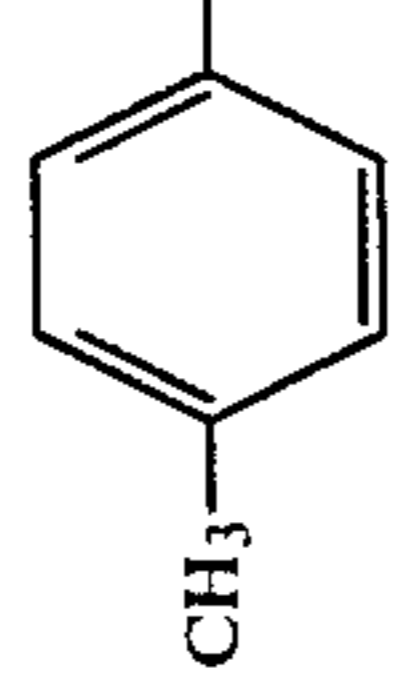
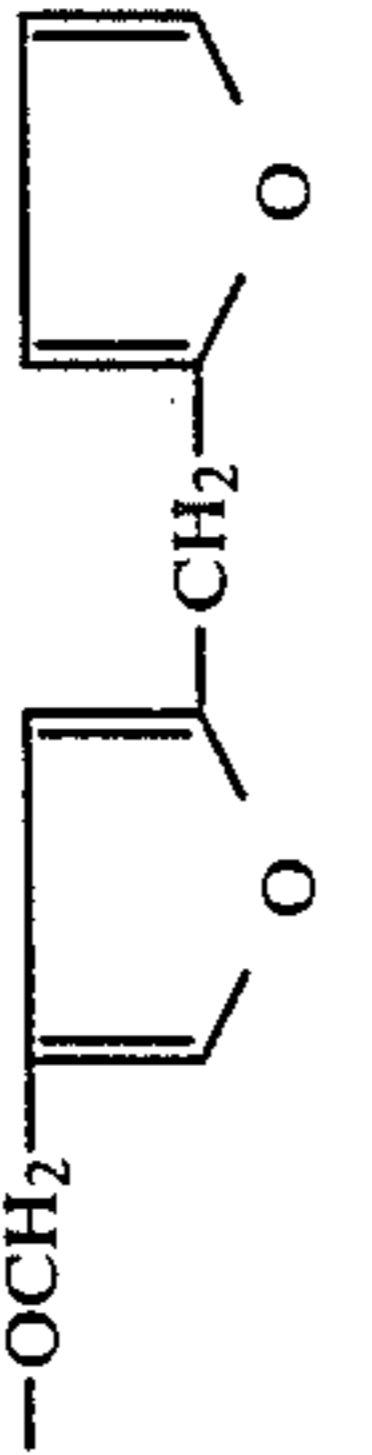
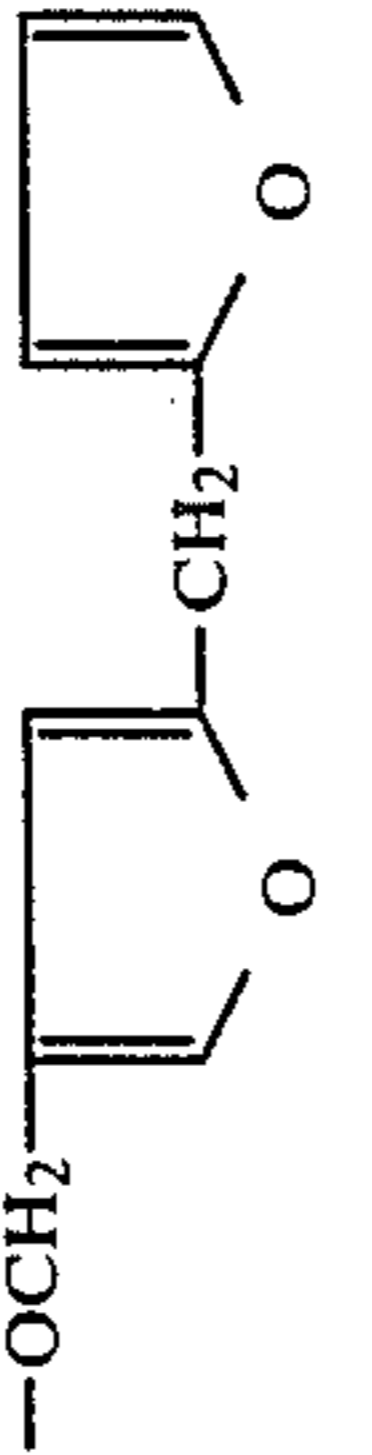
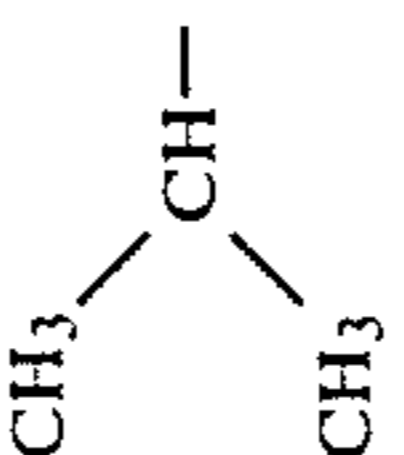
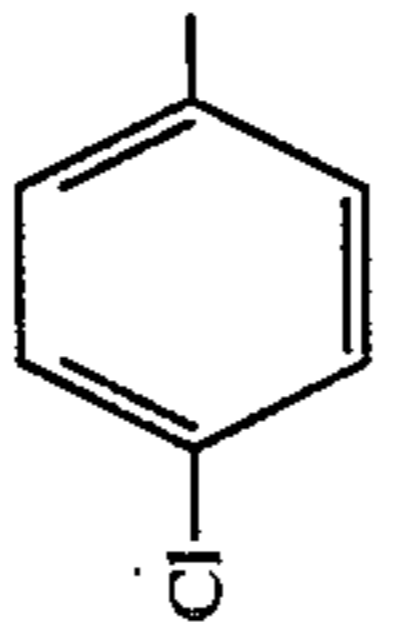
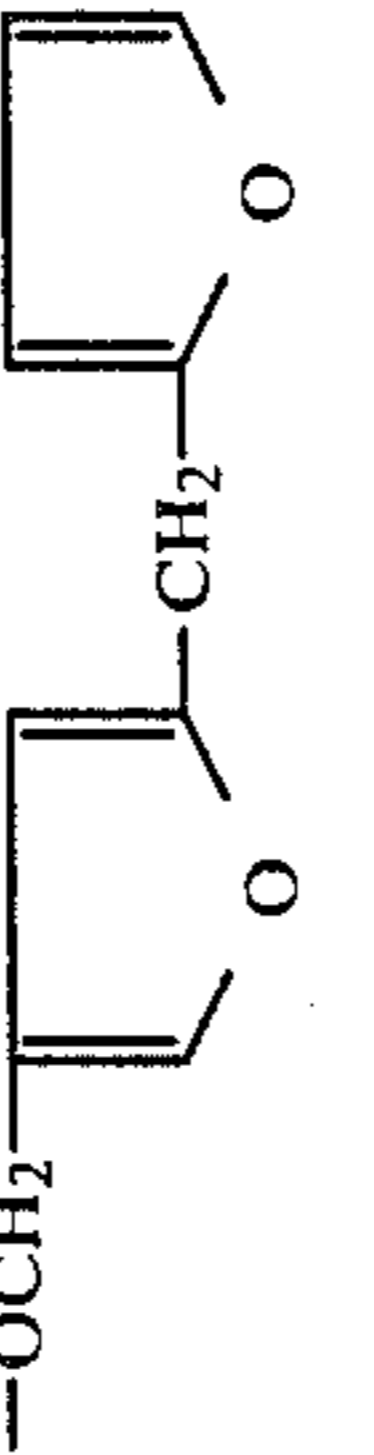
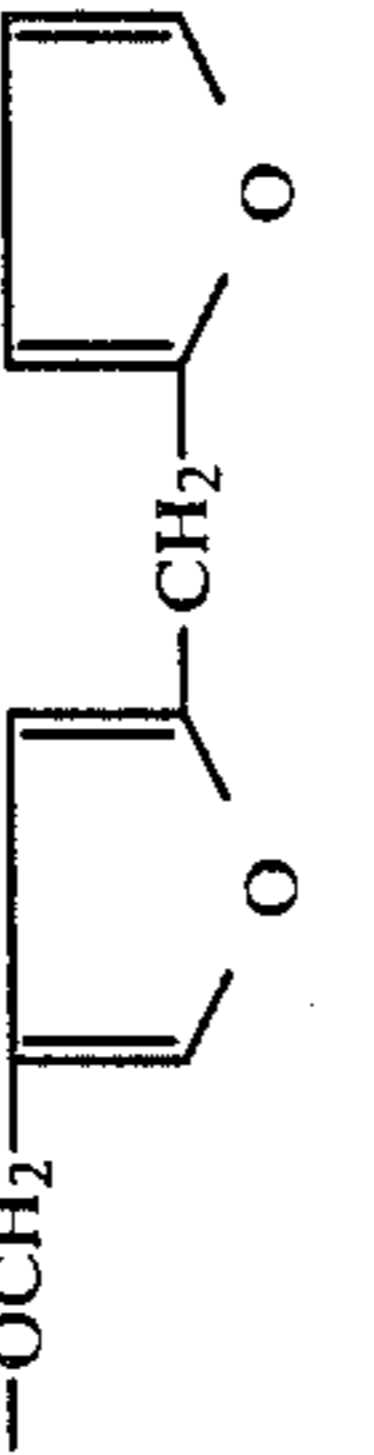
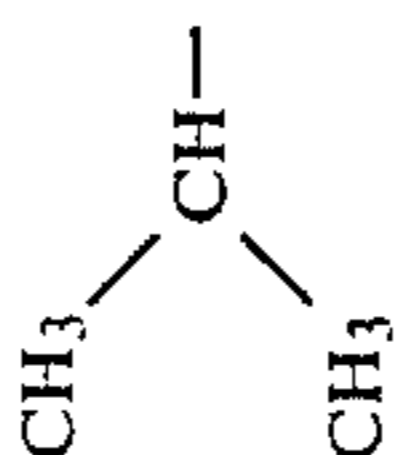
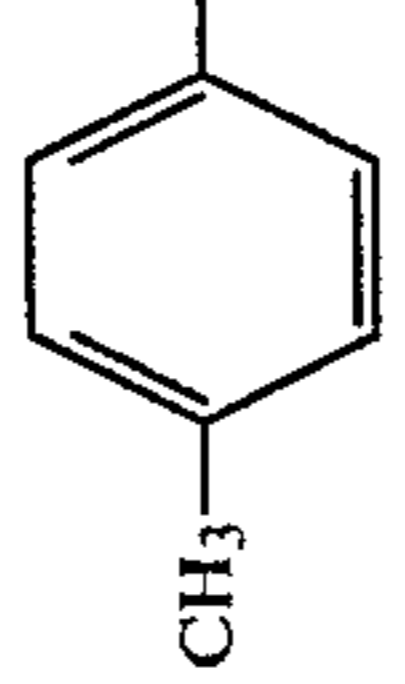
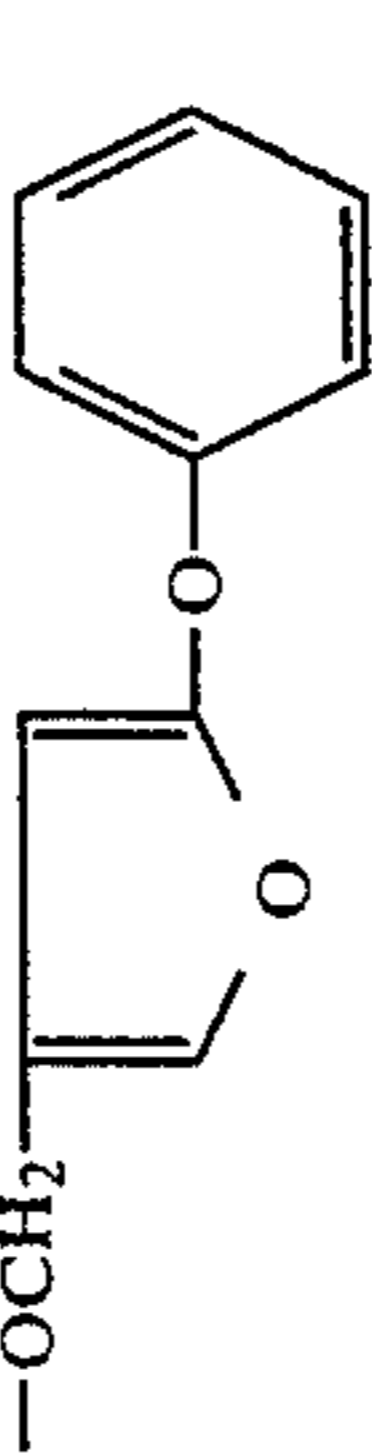
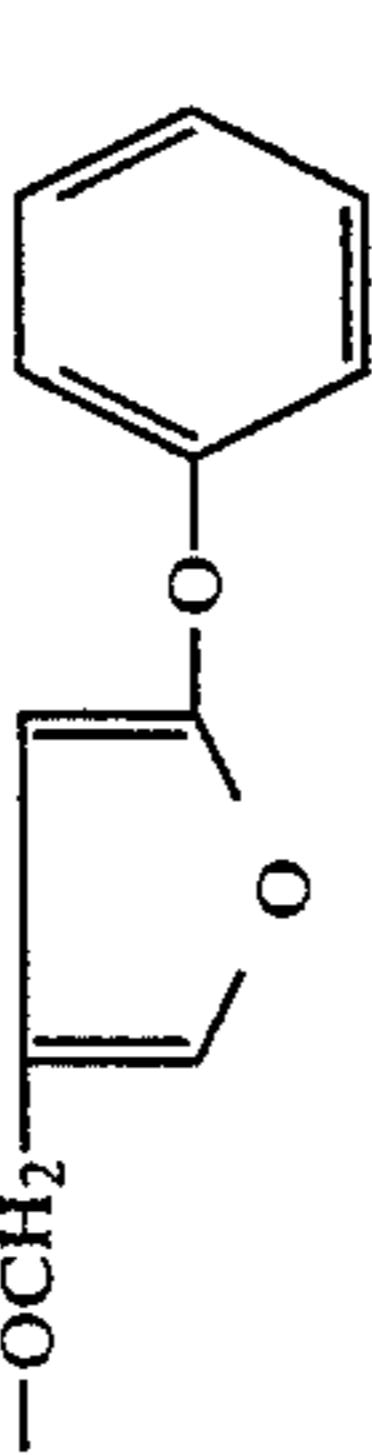
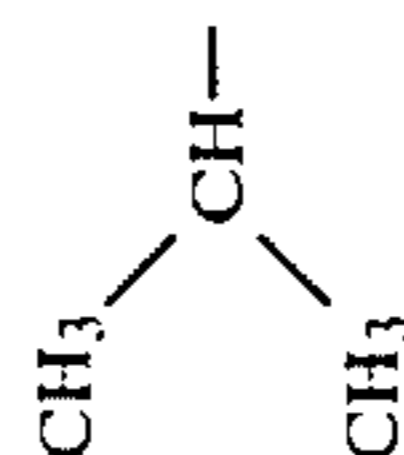
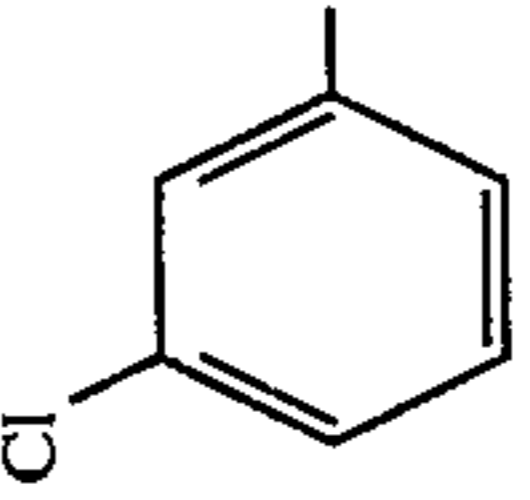
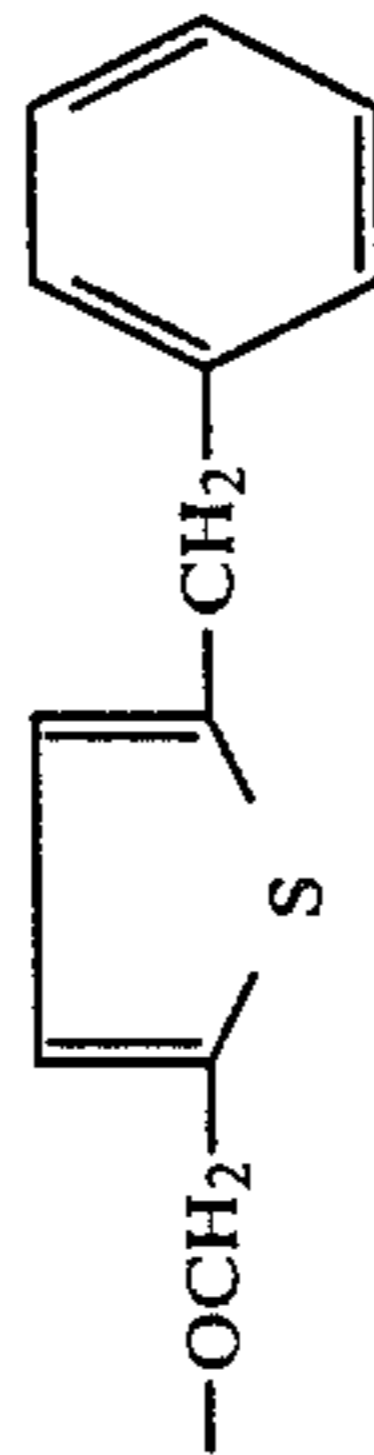
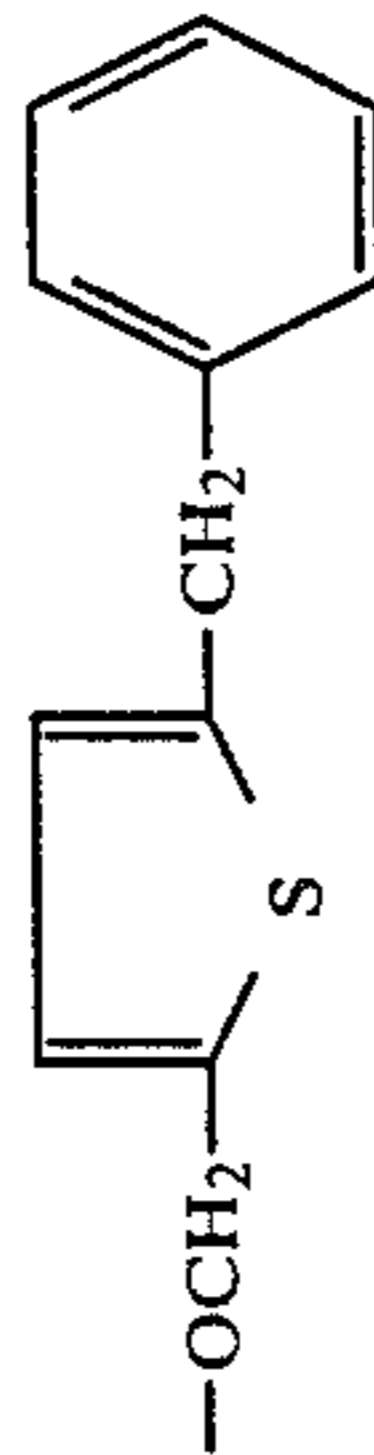
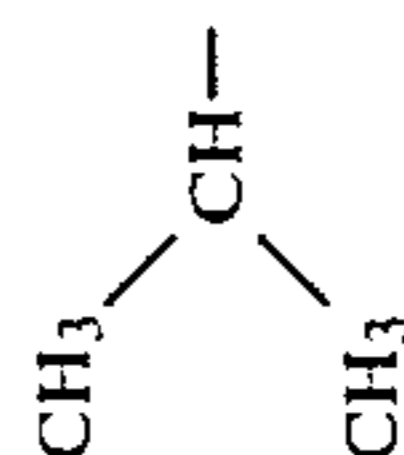
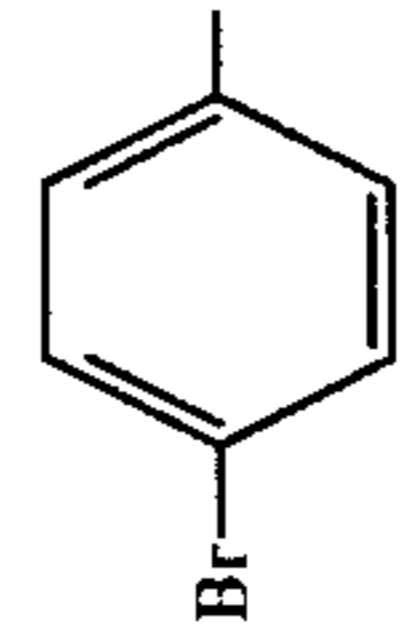
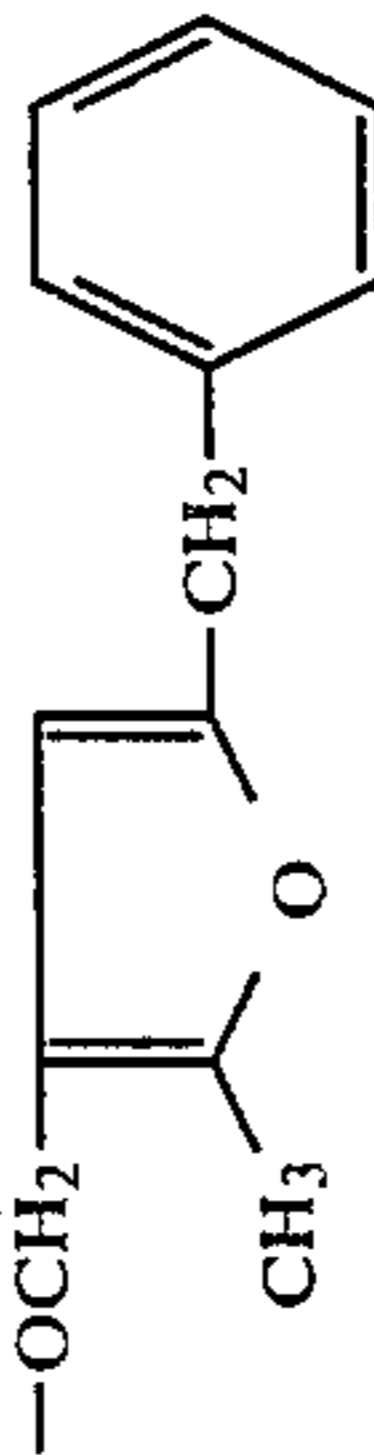
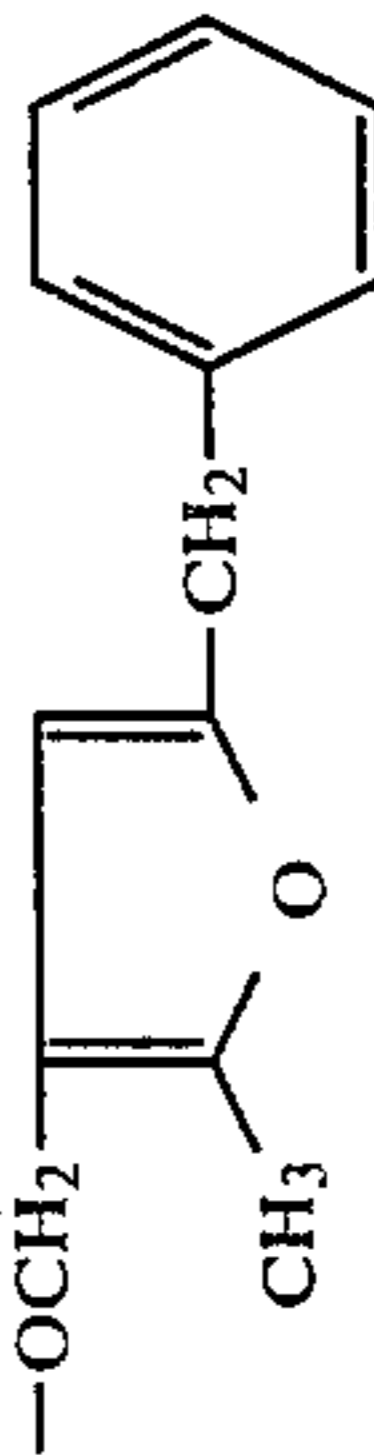
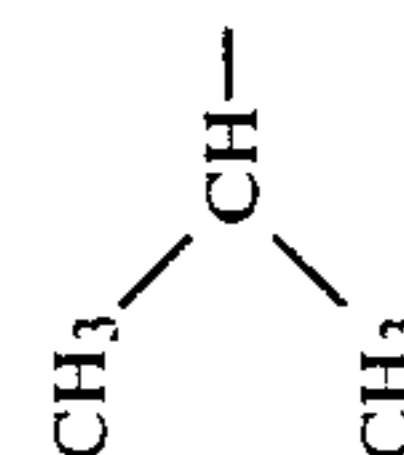
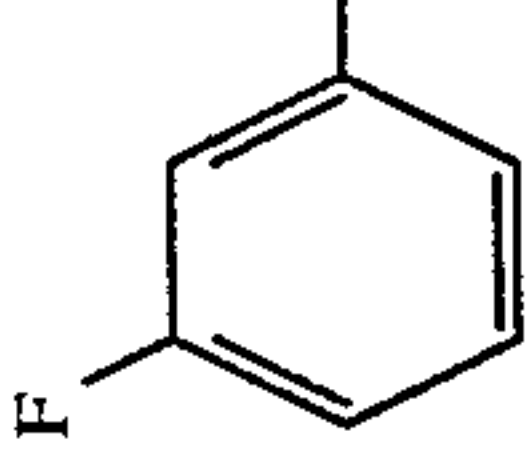
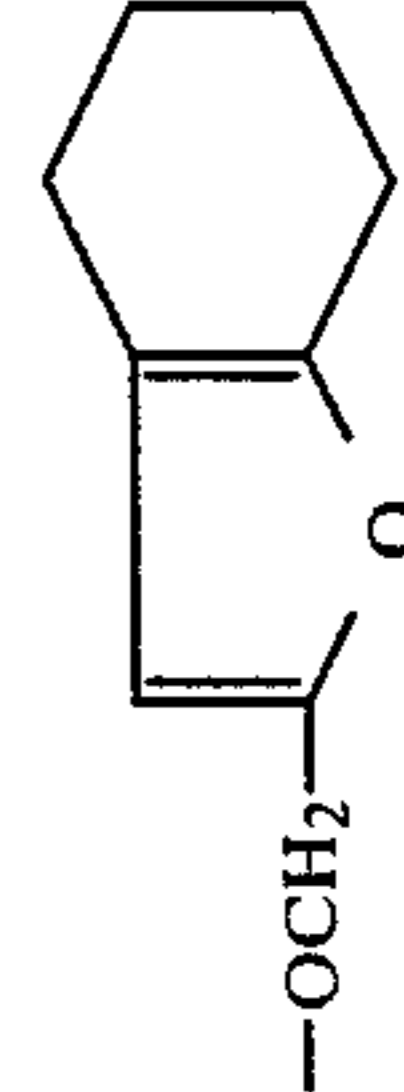
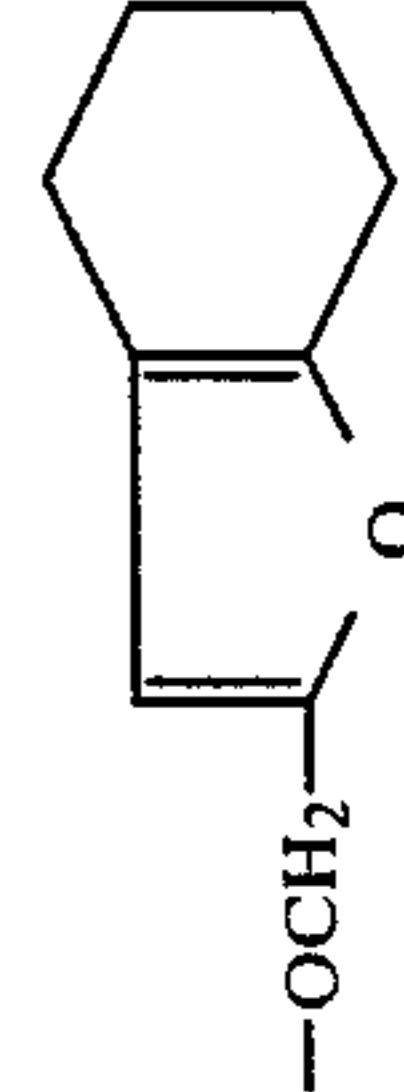
Preparation Exp. No.	Comp. No.	Z	Y	X	Chemical Structure	Esterification Method	Yield (%)	Physical Properties	
								Calcd. (%)	Found (%)
106	177	C ₂ H ₅				e	96	C ₂₃ H ₂₁ O ₃ Cl C: 72.53 H: 5.56 Cl: 9.31	$n_D^{21.5}$ 1.5732 72.51 5.46 9.33
107	178	C ₂ H ₅				e	89	C ₂₃ H ₂₁ O ₃ Br C: 64.95 H: 4.98 Br: 18.79	n_D^{17} 1.5861 64.92 4.86 18.82
108	179	C ₂ H ₅				a	82	C ₂₃ H ₂₃ O ₃ I C: 58.24 H: 4.89 I: 26.76	n_D^{20} 1.5733 58.25 4.92 28.81
109	180	C ₂ H ₅				b	76	C ₂₁ H ₁₉ O ₄ Cl C: 68.01 H: 5.16 Cl: 9.56	n_D^{20} 1.5625 66.06 5.22 9.62
110	181	CH ₃				a	90	C ₂₂ H ₂₂ O ₃ C: 79.01 H: 6.63	n_D^{25} 1.5471 79.07 6.58
111	182	CH ₃				a	91	C ₂₁ H ₁₉ O ₃ Cl C: 71.08 H: 5.40 Cl: 9.99	n_D^{22} 1.5518 71.11 5.32 9.89
112	183	CH ₃				e	87	C ₂₃ H ₂₂ O ₃ C: 79.74 H: 6.40	n_D^{28} 1.5592 79.77 6.36

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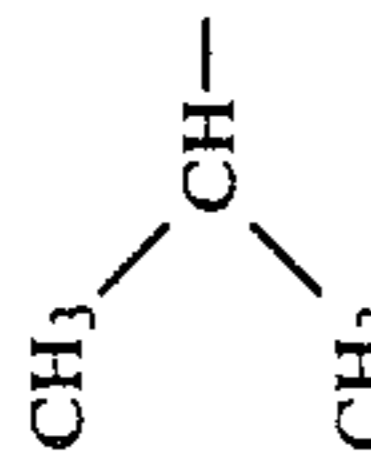
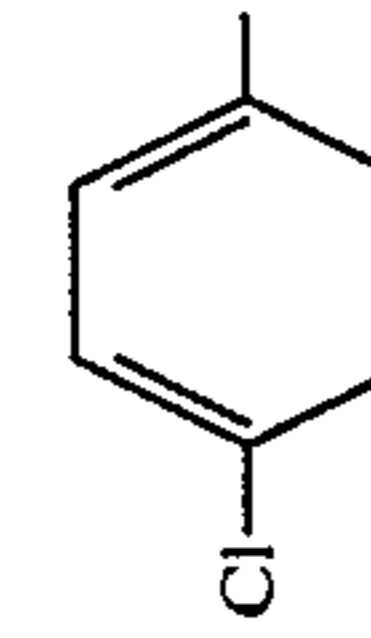
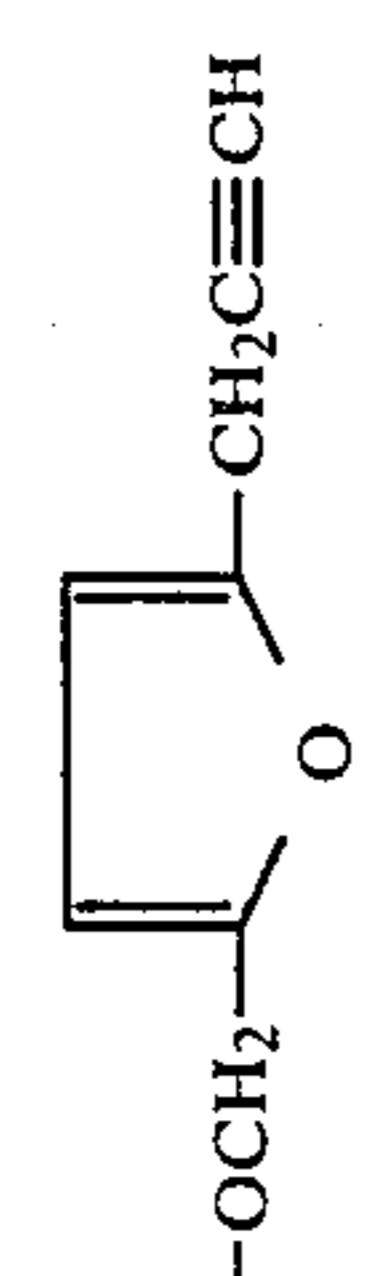
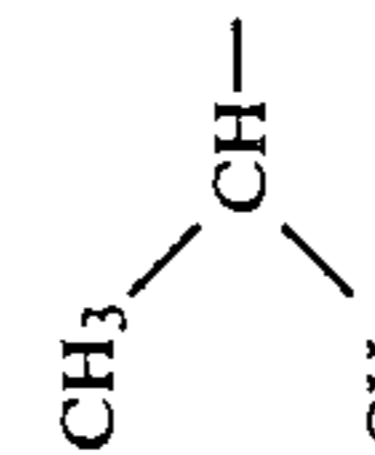
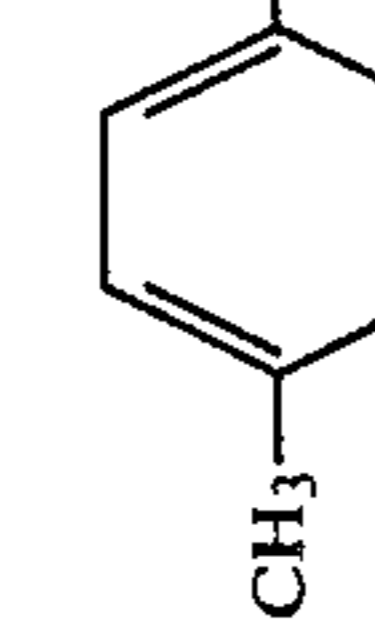
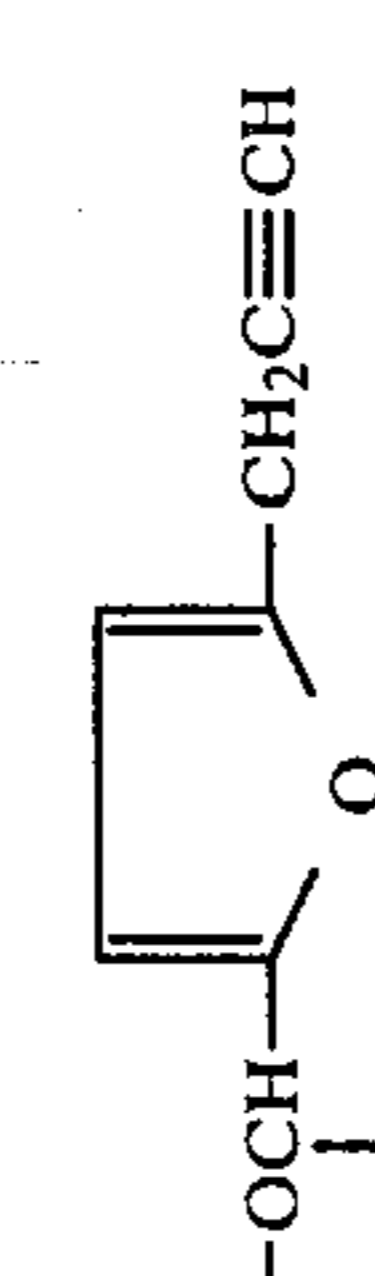
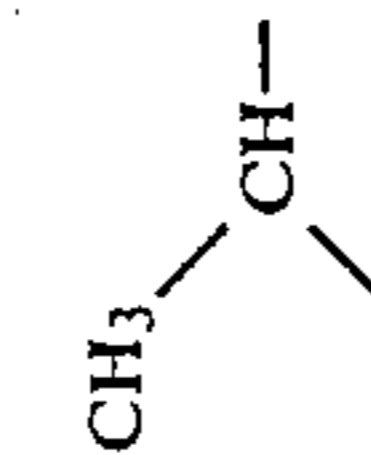
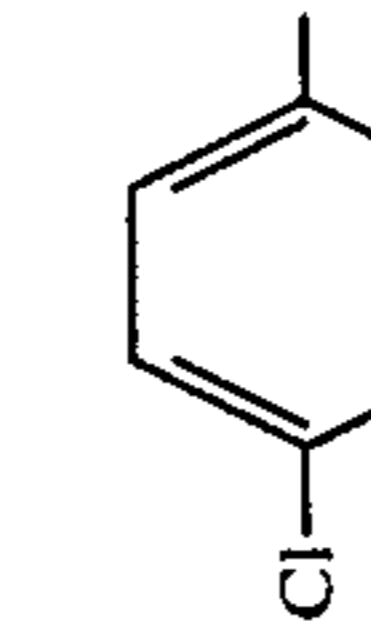
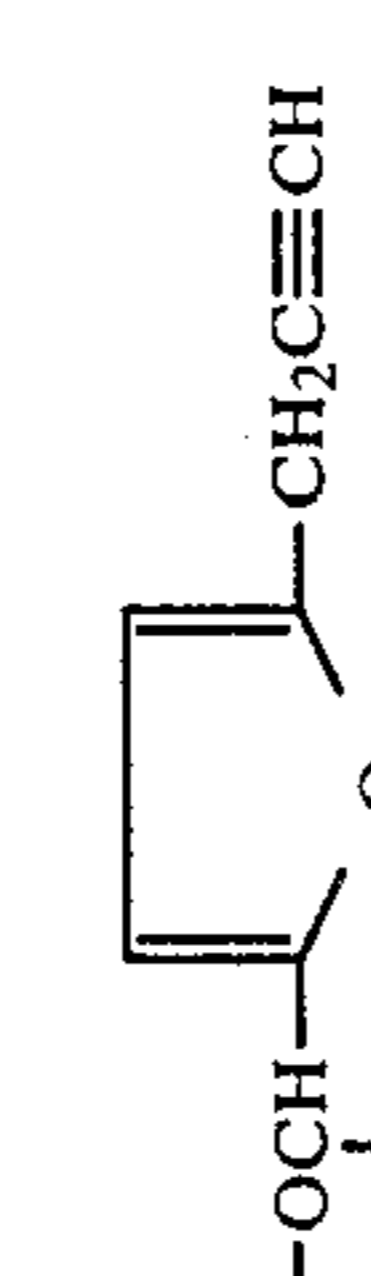
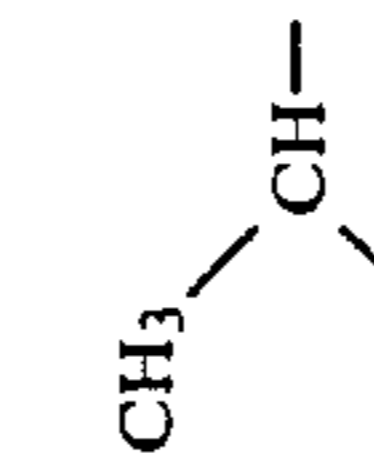
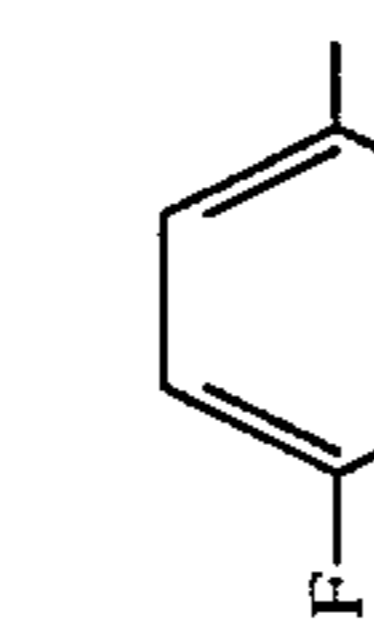
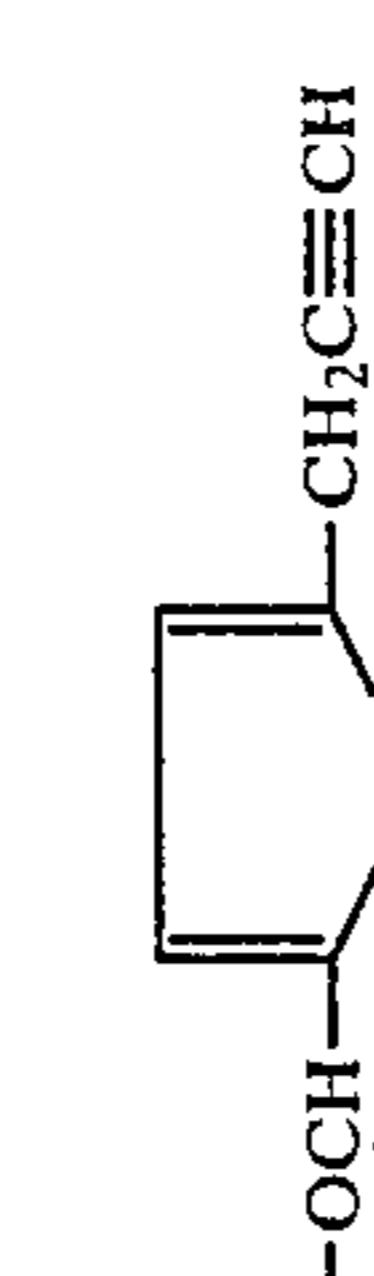
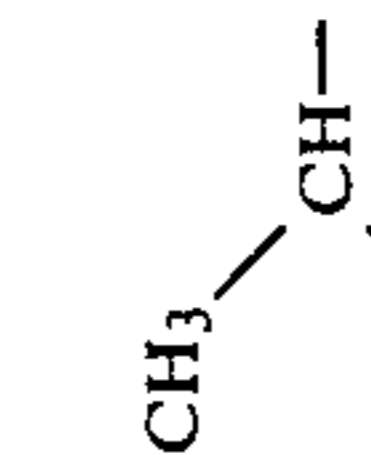
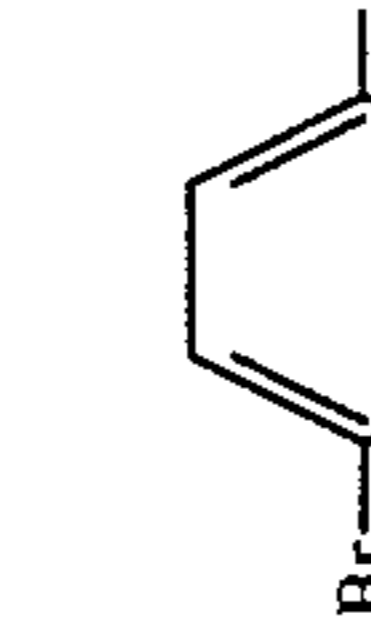

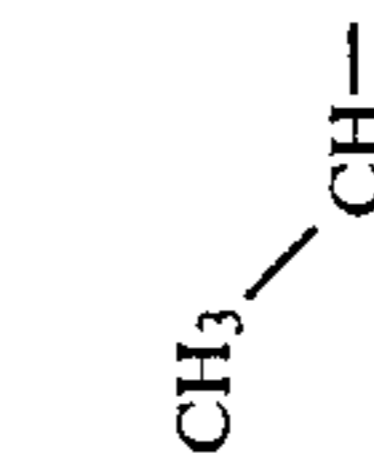
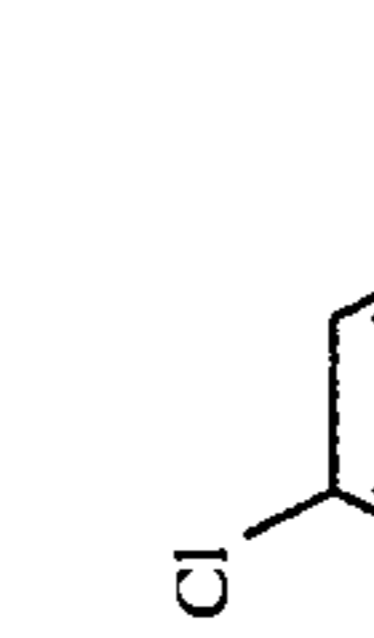
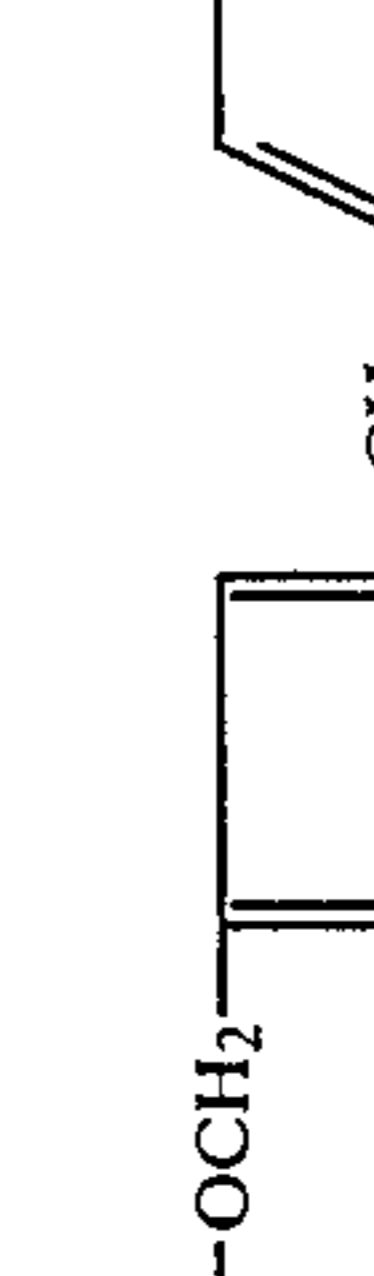
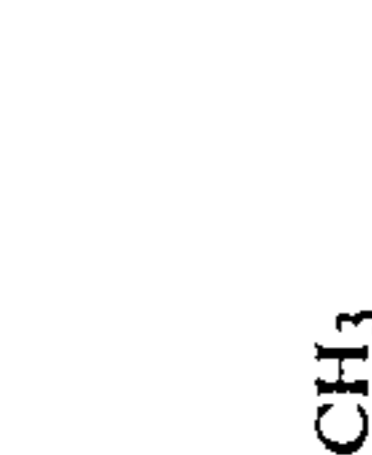
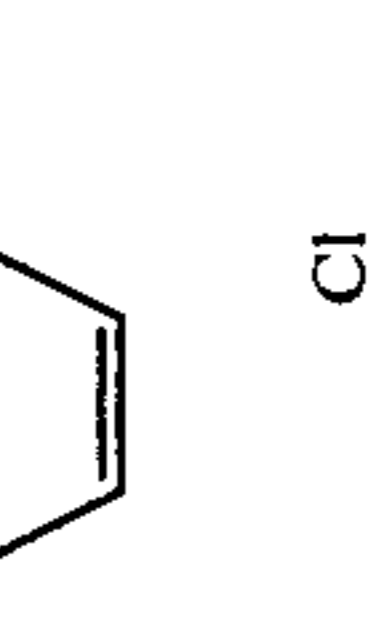

Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
113	184	C ₂ H ₅ —				88	C ₂₂ H ₂₁ O ₃ Cl C: 71.63 H: 5.74 Cl: 9.61	<i>n</i> _D ²⁵ 1.5549 71.66 5.81 9.70
114	185	C ₂ H ₅ —				87	C ₂₂ H ₂₁ O ₃ F C: 74.98 H: 6.00 F: 5.39	<i>n</i> _D ²⁰ 1.4437 75.03 6.02 5.29

Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification		Yield		Physical Properties	
					Method	(%)	Calcd. (%)	Found (%)		
115	186	C ₂ H ₅ —			a	83	C ₂₂ H ₂₁ O ₂ SF C: 71.71 H: 5.75 S: 8.70 F: 5.16	n _D ²⁰ 1.4933 71.77 5.76 8.81 5.13		
116	187	C ₂ H ₅ —			e	76	C ₁₉ H ₂₀ O ₄ NCl C: 63.07 H: 5.57 N: 3.87 Cl: 9.80	n _D ²⁰ 1.5316 62.98 5.54 3.92 9.69		
117	188	C ₂ H ₅ —			a	83	C ₂₀ H ₂₄ O ₃ C: 76.89 H: 7.74	n _D ²⁰ 1.5224 76.79 7.81		
118	189	C ₂ H ₅ —			a	80	C ₂₀ H ₂₂ O ₃ C: 77.39 H: 7.14	n _D ²⁰ 1.5456 77.44 7.14		
119	190	C ₂ H ₅ —			a	82	C ₁₉ H ₁₉ O ₃ Cl C: 68.98 H: 5.79 Cl: 10.72	n _D ²⁰ 1.5445 68.92 5.78 10.85		
120	191	C ₂ H ₅ —			a	90	C ₂₂ H ₂₁ O ₃ F C: 74.98 H: 6.01 F: 5.39	n _D ²⁰ 1.5267 74.87 6.02 5.41		

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Exp. No.	Preparation Comp. No.	Z	Y	X	Chemical Structure	Esterification Method	Yield (%)	Physical Properties	
								Calcd. (%)	Found (%)
121	192	C ₂ H ₅				a	92	C ₂₂ H ₂₁ O ₃ Cl C: 71.63 H: 5.74 Cl: 9.61	n_D^{20} 1.5352 71.66 5.78 9.62
122	193					a	90	C ₂₂ H ₂₄ O ₄ C: 74.97 H: 6.86	n_D^{19} 1.5377 75.10 6.86
123	194					a	91	C ₂₁ H ₂₁ O ₄ Cl C: 67.65 H: 5.68 Cl: 9.51	n_D^{19} 1.5421 67.62 5.72 9.61
124	195					a	89	C ₂₃ H ₂₄ O ₄ C: 75.80 H: 6.64	n_D^{20} 1.5235 75.83 6.70
125	196					a	91	C ₂₃ H ₂₃ O ₂ S C: 69.24 H: 5.81 S: 8.04 Cl: 8.89	n_D^{20} 1.5551 69.26 5.84 8.11 8.92
126	197					a	88	C ₂₄ H ₂₅ O ₃ Br C: 65.31 H: 5.71 Br: 18.11	n_D^{20} 1.5349 65.33 5.83 18.20
127	198					a	92	C ₂₀ H ₂₃ O ₃ F C: 72.70 H: 7.02 F: 5.75	n_D^{20} 1.5466 72.69 7.11 5.72

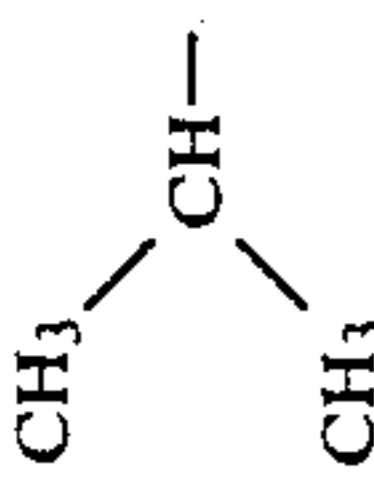
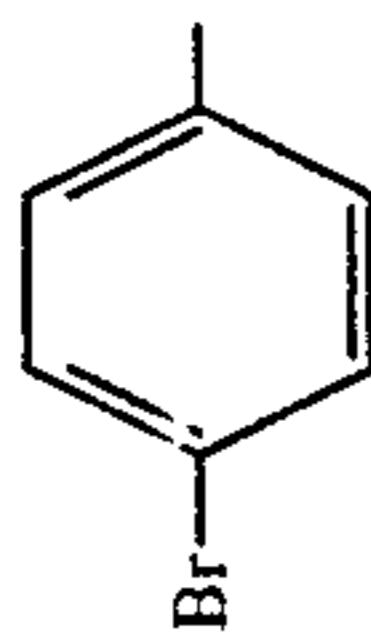
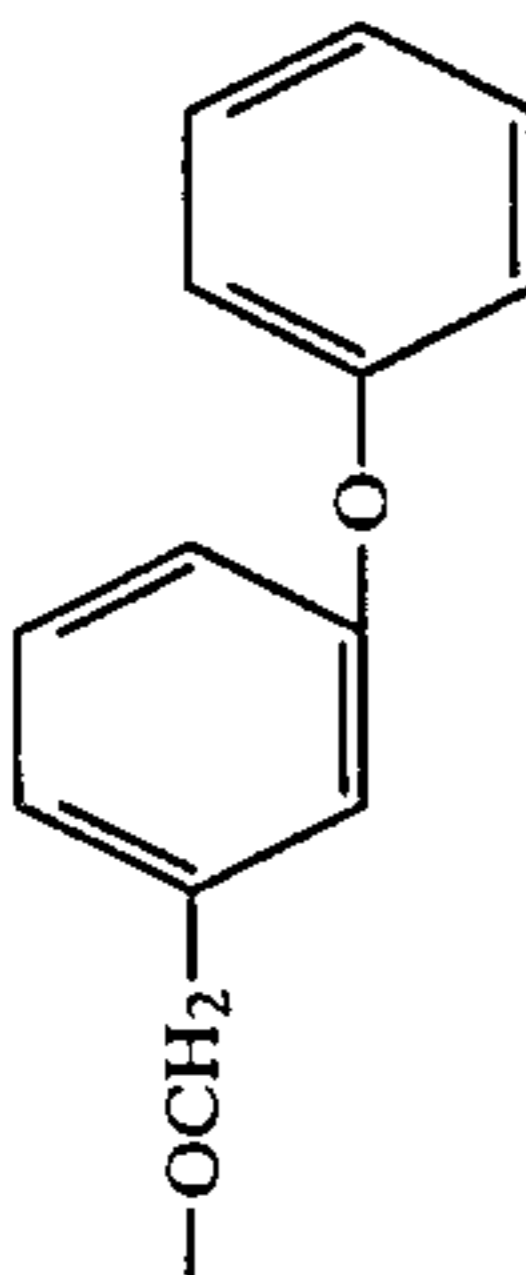
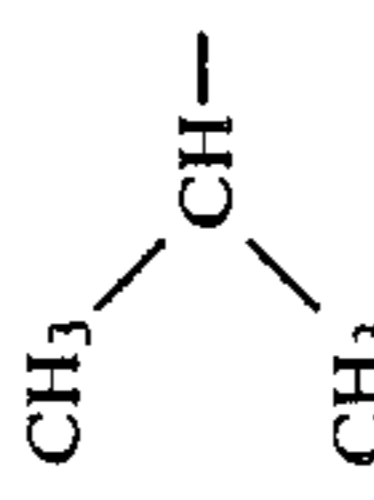
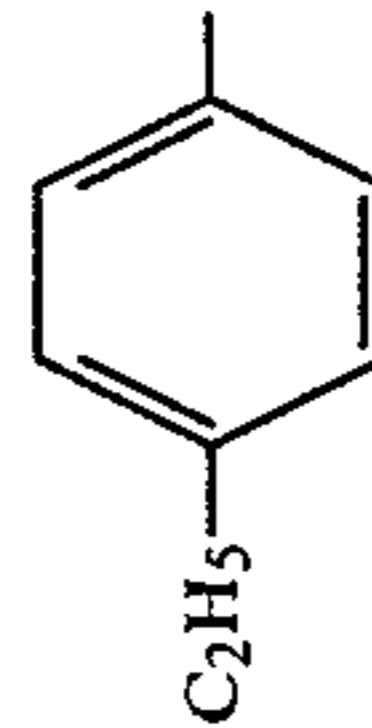
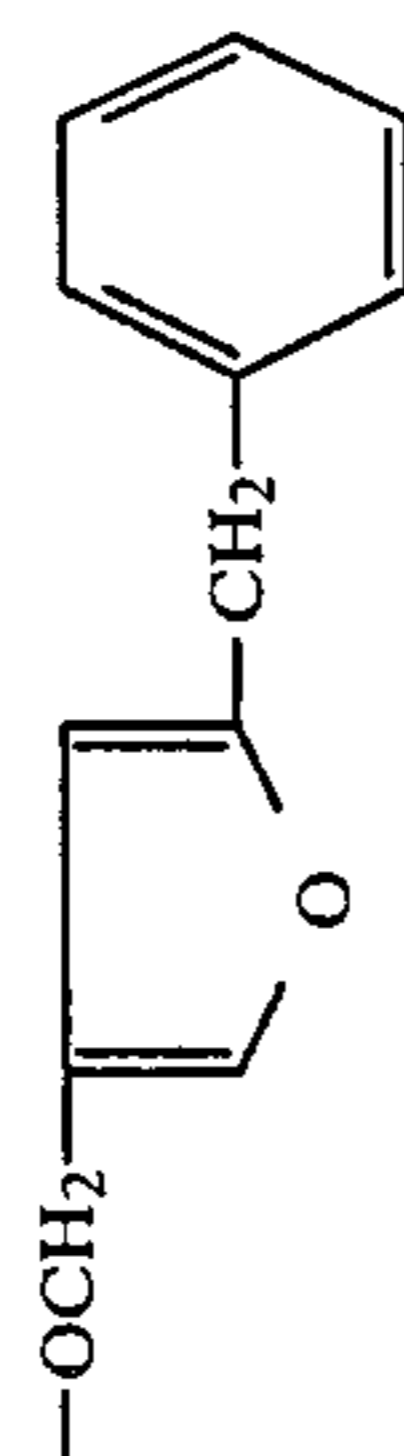
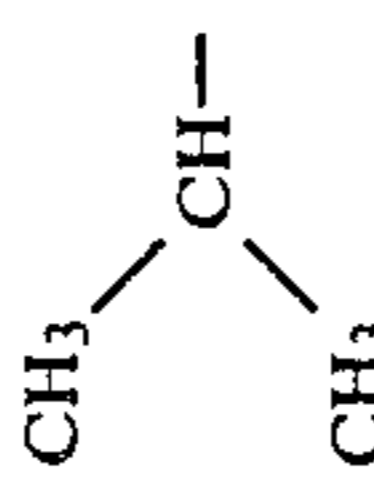
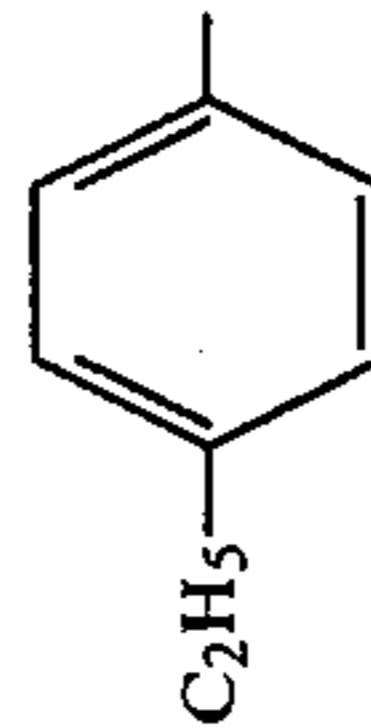
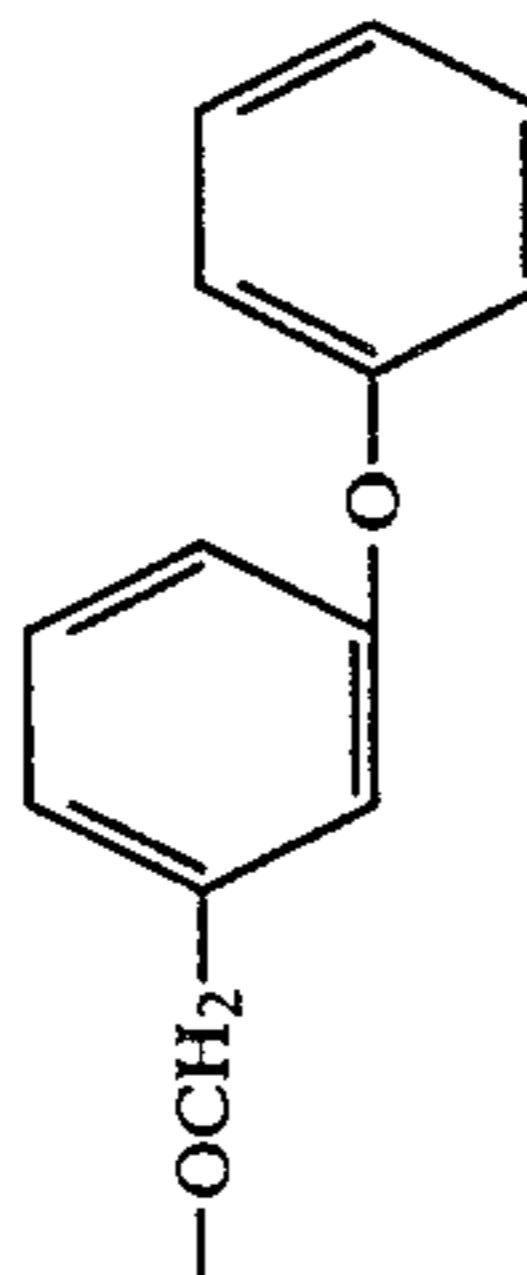
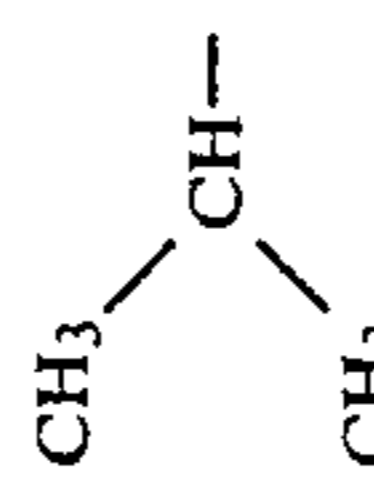
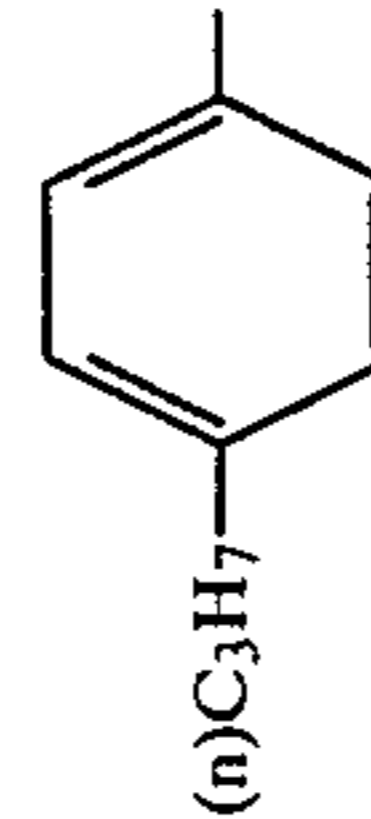
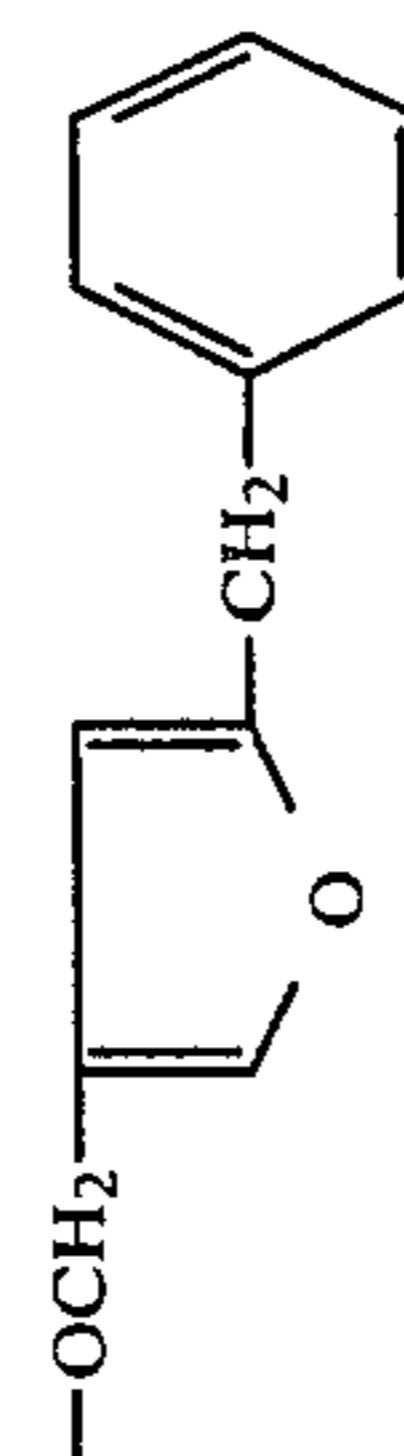
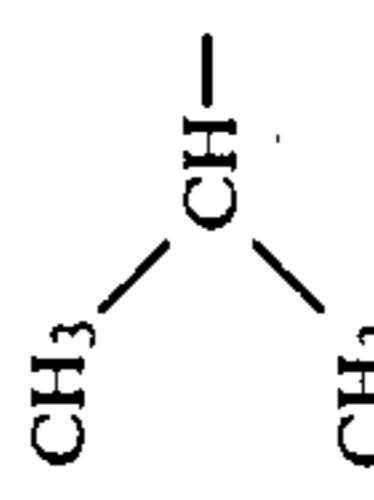
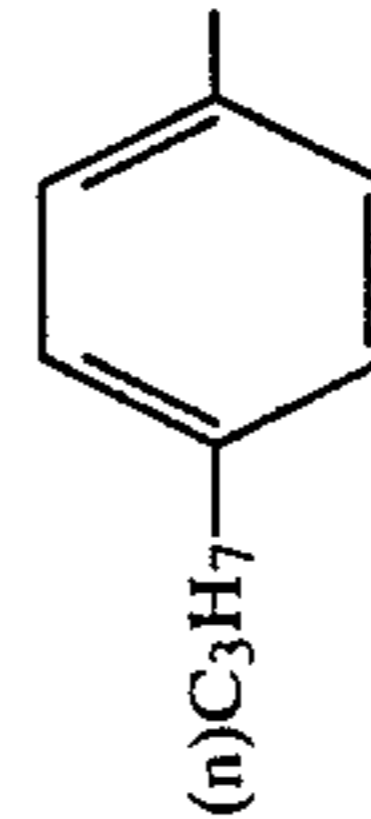
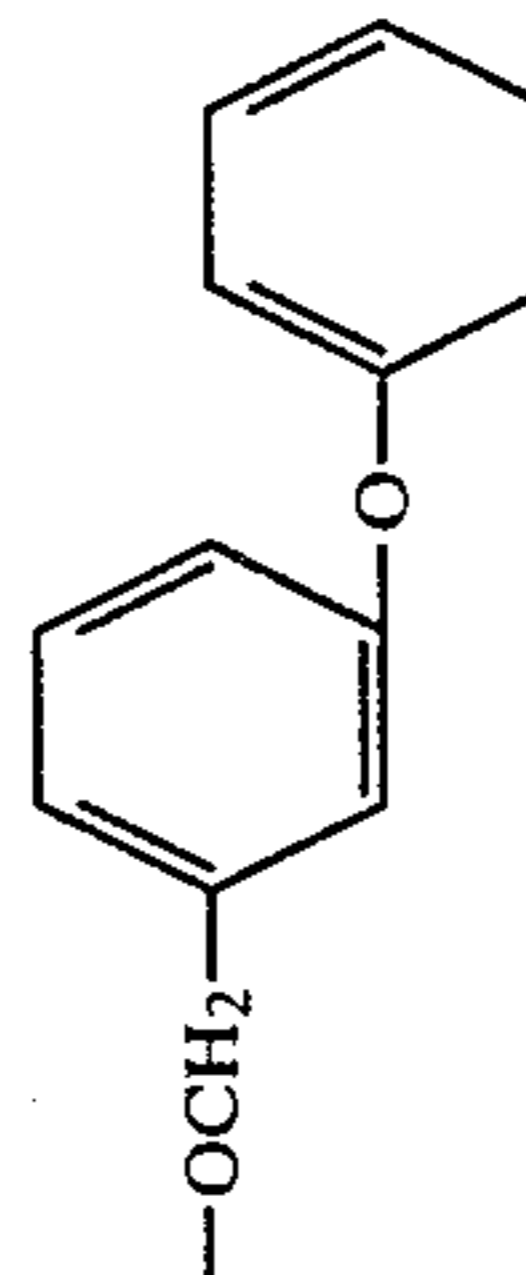
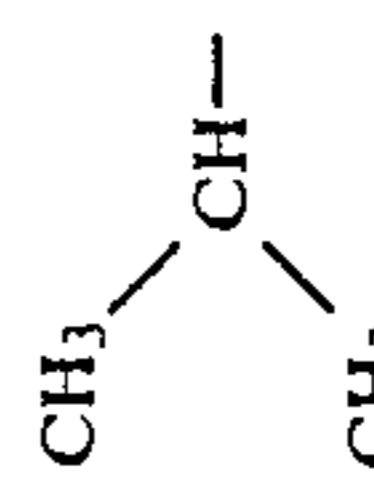
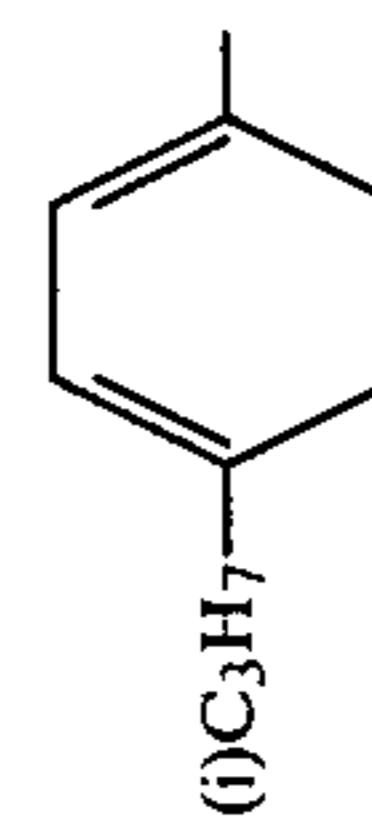
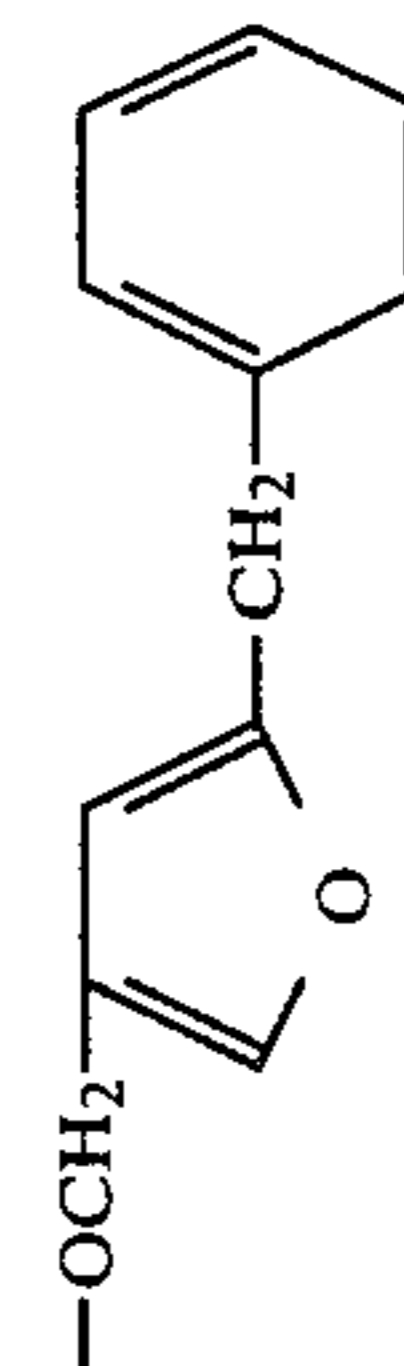
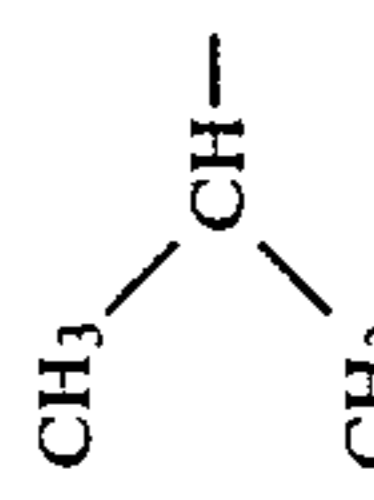
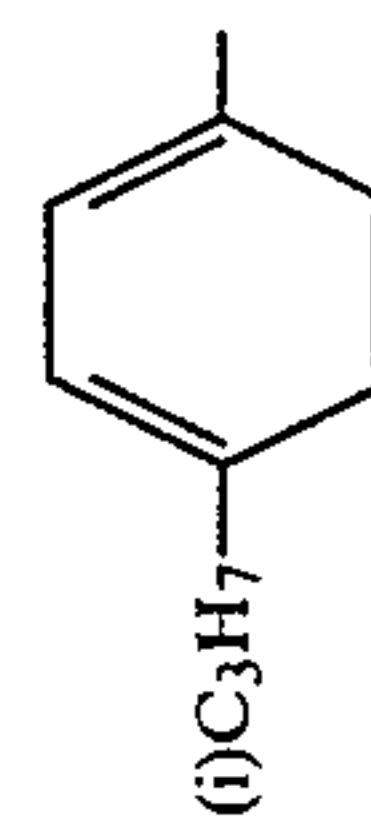
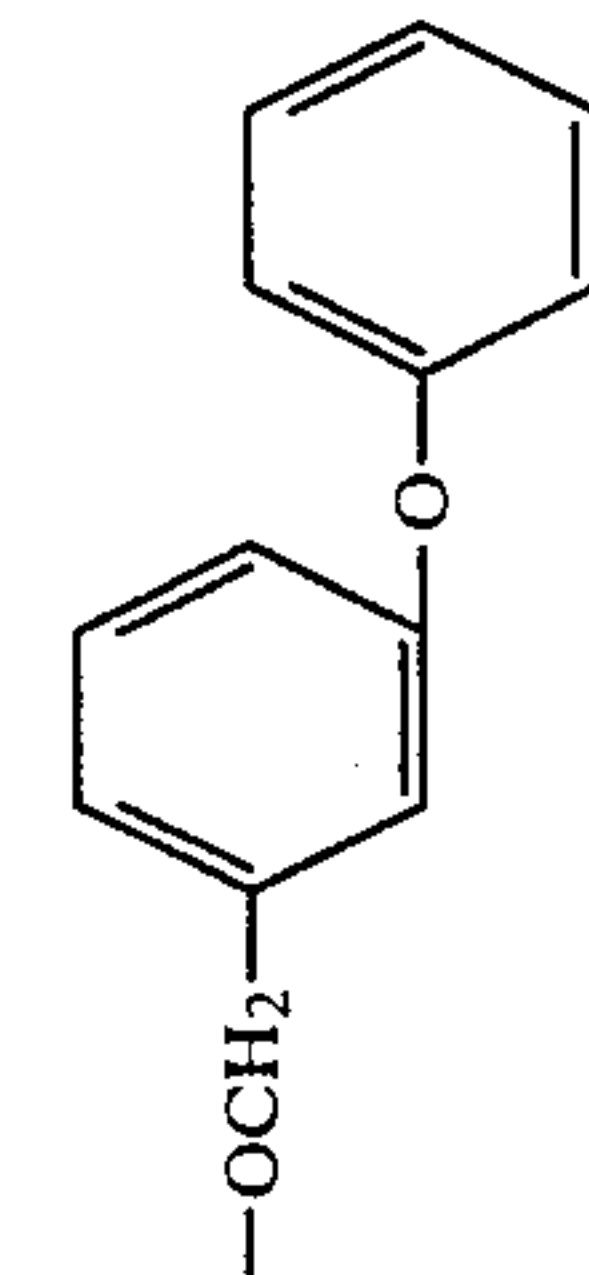
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Exp. No.	Preparation Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
128	199				a	80	C ₁₉ H ₁₉ O ₃ Cl C: 68.98 H: 5.79 Cl: 10.72	n_D^{19} 1.5238 69.01 5.72 10.77
129	200				a	82	C ₂₂ H ₂₂ O ₃ C: 79.01 H: 6.63	n_D^{20} 1.5265 79.10 6.58
130	201				a	79	C ₂₁ H ₁₉ O ₃ Cl C: 71.08 H: 5.40 Cl: 9.99	n_D^{20} 1.5233 71.00 5.46 9.89
131	202				a	75	C ₂₁ H ₁₉ O ₃ F C: 74.54 H: 5.66 F: 5.62	n_D^{20} 1.5249 74.53 5.77 5.66
132	203				a	83	C ₂₁ H ₁₉ O ₃ Br C: 63.17 H: 4.80 Br: 20.02	n_D^{20} 1.5289 63.18 4.79 20.04
133	204				a	90	C ₂₃ H ₂₃ O ₃ Cl C: 72.15 H: 6.05 Cl: 9.26	n_D^{21} 1.5521 72.16 6.11 9.39
134	205				a	89	C ₂₃ H ₂₃ O ₃ Cl C: 72.15 H: 6.05 Cl: 9.26	n_D^{21} 1.5576 72.14 6.09 9.31

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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
135	206				d	83	C ₂₄ H ₂₃ O ₃ Cl C: 72.99 H: 5.87 Cl: 8.98	n_D^{21} 1.5655 73.04 5.87 8.99
136	207				e	92	C ₂₄ H ₂₃ O ₃ Cl C: 72.99 H: 5.87 Cl: 8.98	n_D^{17} 1.5722 72.93 5.89 9.00
137	208				d	84	C ₂₃ H ₂₃ O ₃ F C: 75.39 H: 6.33 F: 5.19	n_D^{21} 1.4637 75.42 6.31 5.22
138	209				a	92	C ₂₃ H ₂₃ O ₃ F C: 75.39 H: 6.33 F: 5.19	n_D^{21} 1.5330 75.44 6.26 5.14
139	210				d	93	C ₂₄ H ₂₃ O ₃ F C: 76.17 H: 6.13 F: 5.02	n_D^{23} 1.5544 76.20 6.14 5.10
140	211				e	92	C ₂₄ H ₂₃ O ₃ F C: 76.17 H: 6.13 F: 5.02	$n_D^{21.5}$ 1.5543 76.15 6.13 5.11
141	212				a	89	C ₂₃ H ₂₃ O ₃ Br C: 64.64 H: 5.42 Br: 18.70	n_D^{17} 1.5651 64.58 5.44 18.96

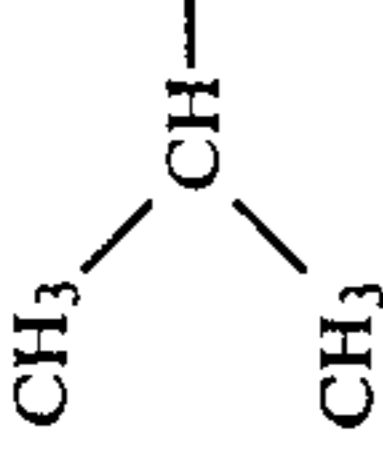
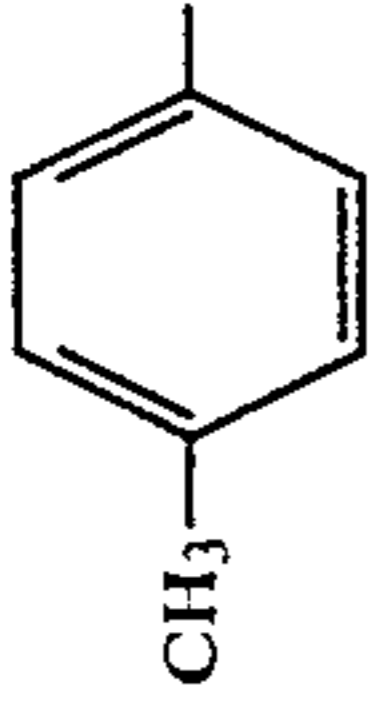
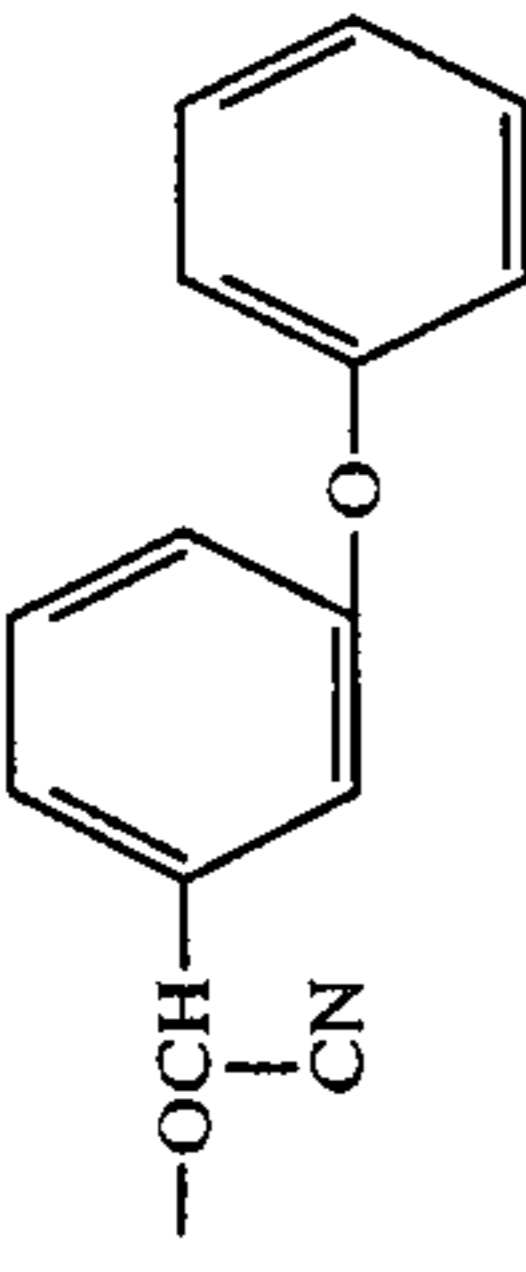
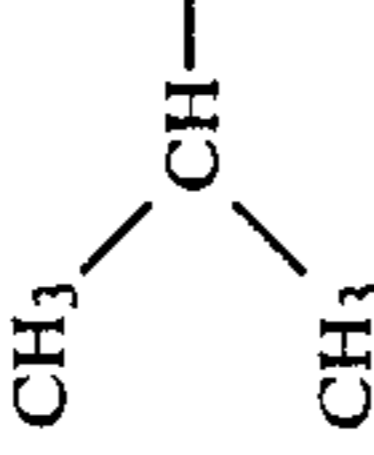
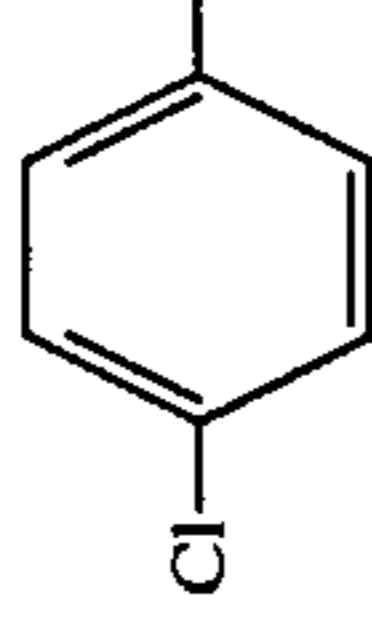
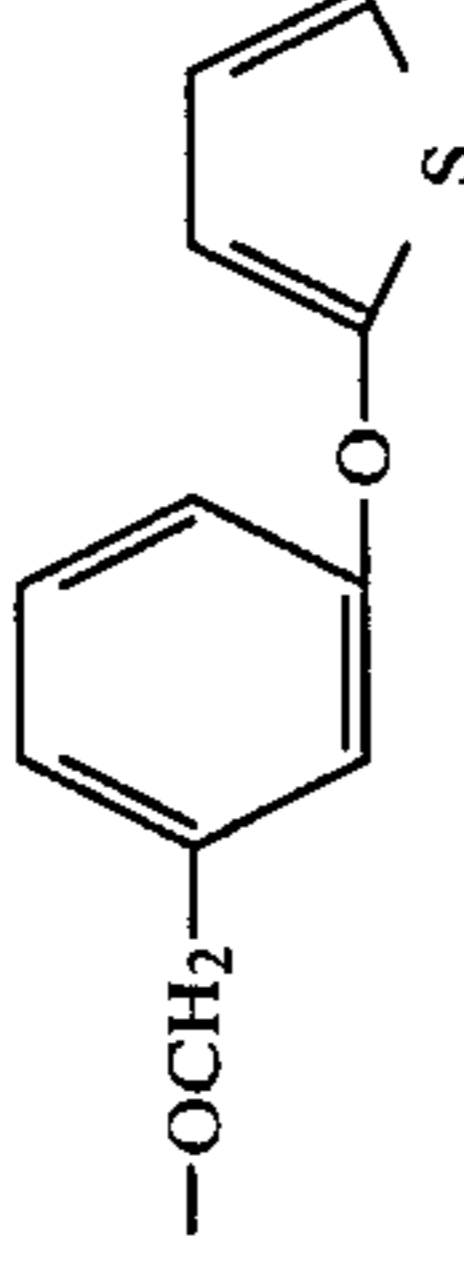
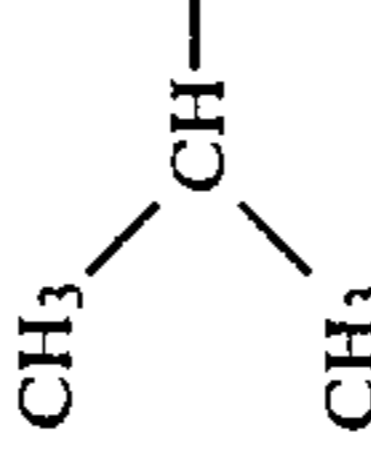
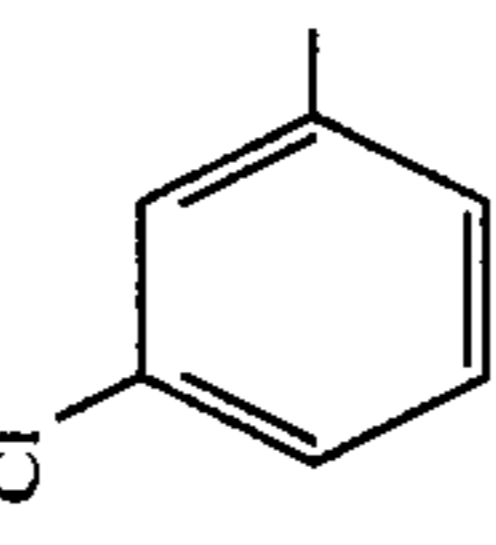
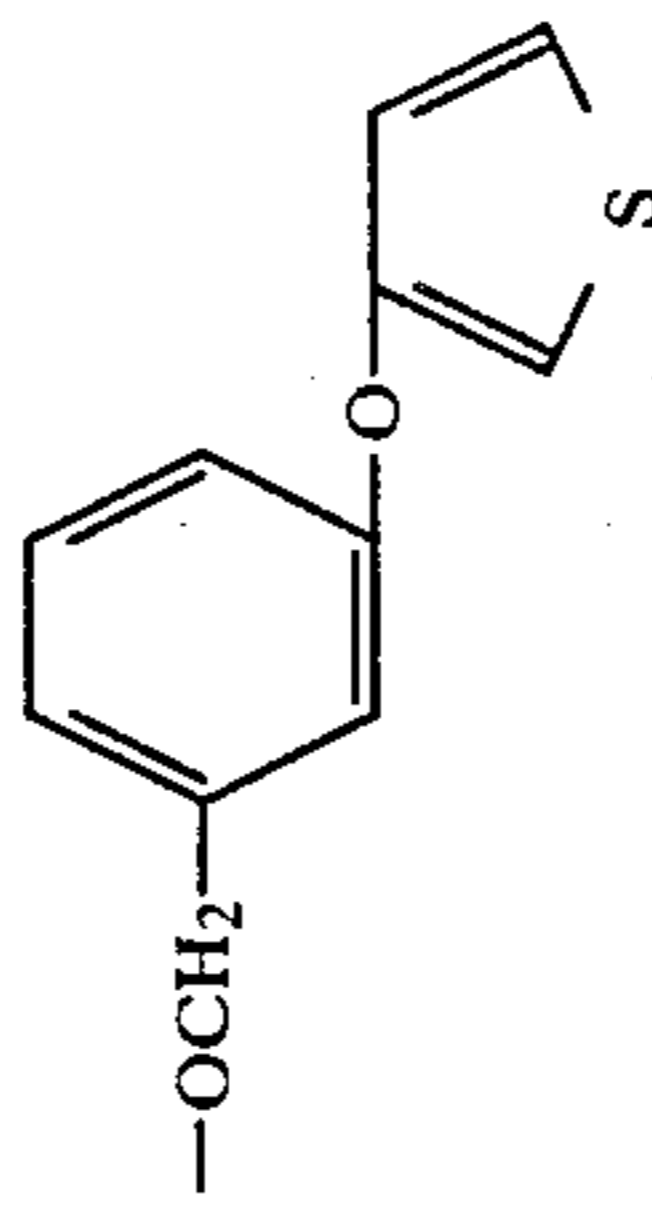
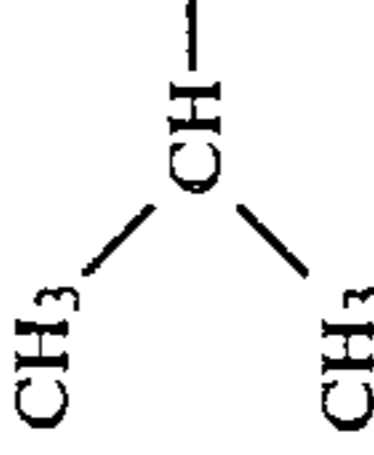
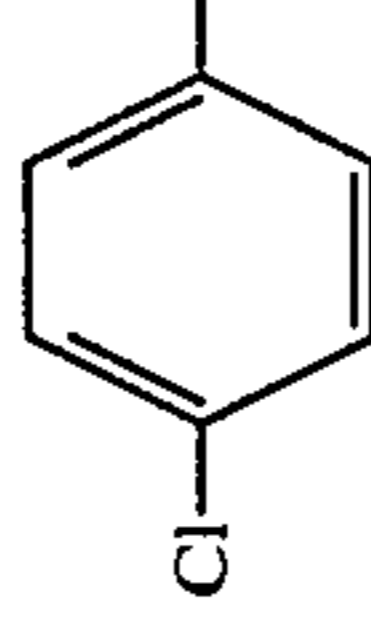
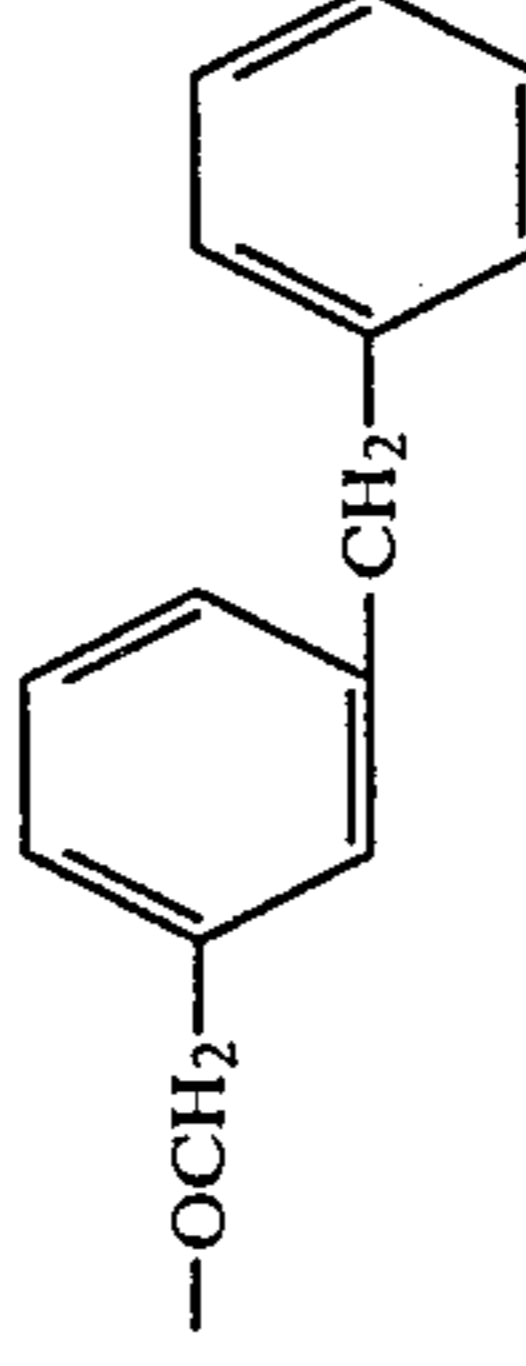
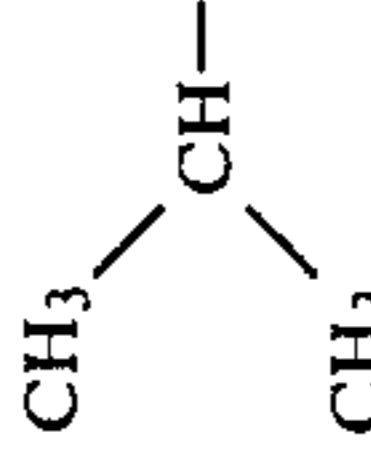
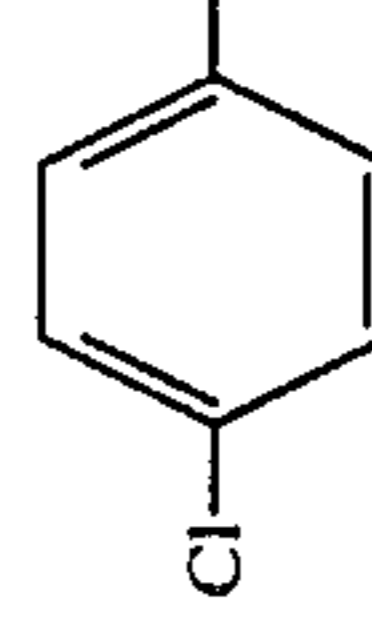
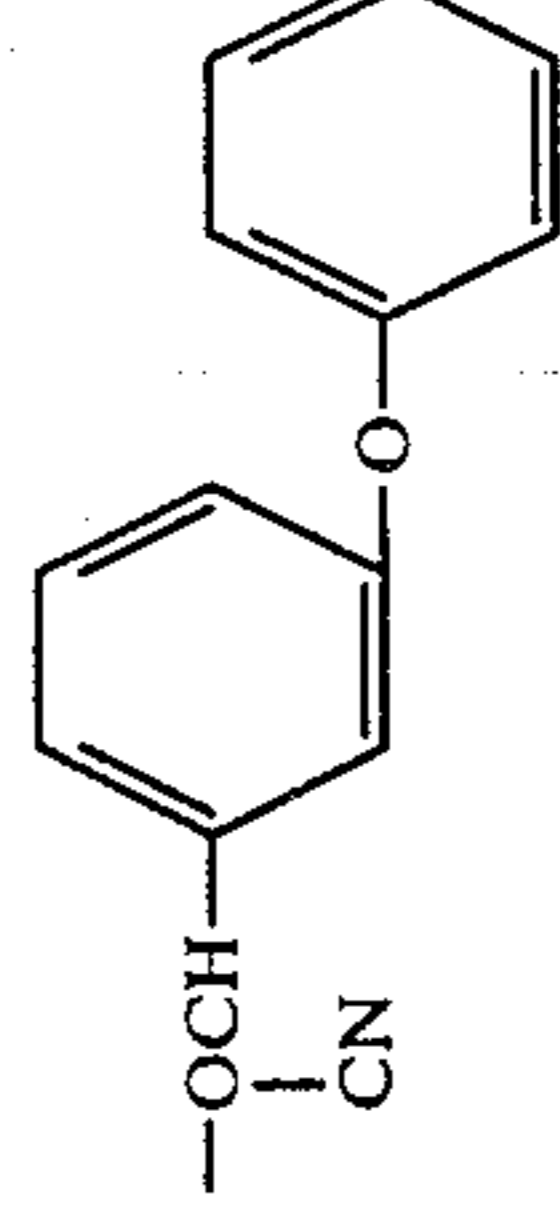
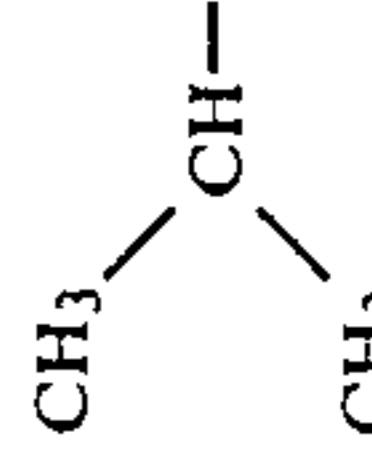
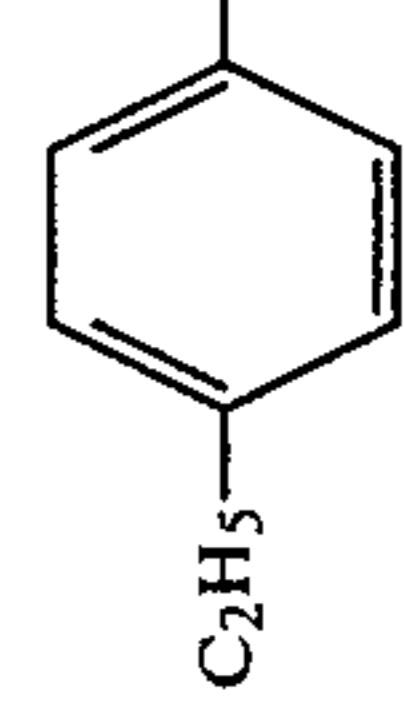
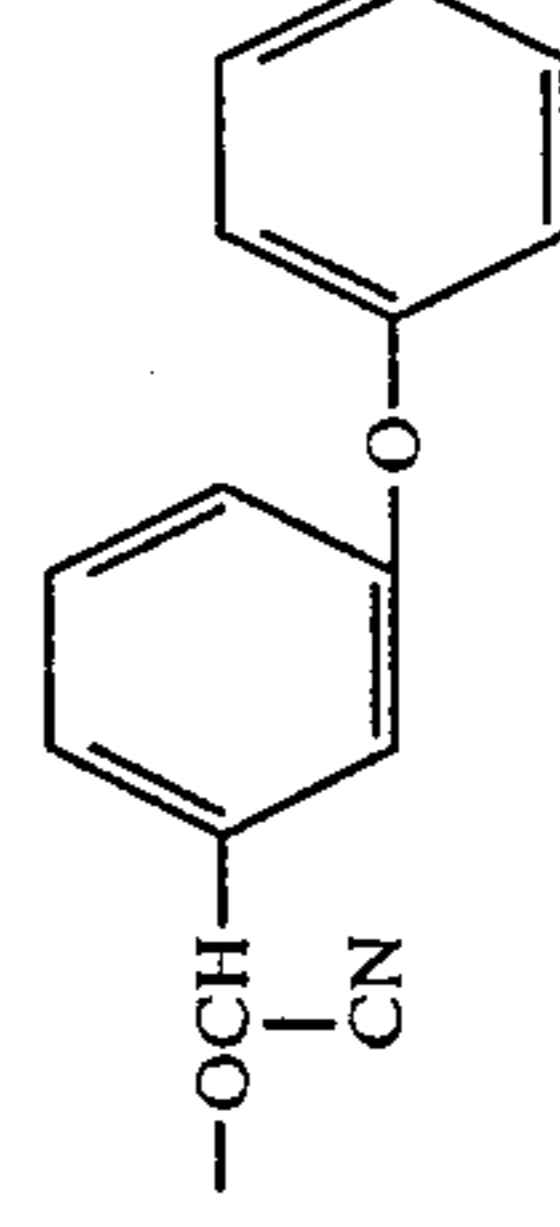
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Preparation Exp. No.	Comp. No.	Z	Y	X	Chemical Structure	Esterification Method	Yield (%)	Physical Properties	
								Calcd. (%)	Found (%)
142	213					e	91	C ₂₄ H ₂₃ O ₃ Br C: 65.61 H: 5.28 Br: 18.19	<i>n</i> _D ^{17.5} 1.5802 65.62 5.30 18.23
143	214					a	82	C ₂₅ H ₂₈ O ₃ C: 79.75 H: 7.50	<i>n</i> _D ^{18.5} 1.5347 79.81 7.51
144	215					a	94	C ₂₆ H ₂₈ O ₃ C: 80.38 H: 7.27	<i>n</i> _D ^{21.5} 1.5590 80.34 7.29
145	216					a	92	C ₂₆ H ₃₀ O ₃ C: 79.96 H: 7.74	<i>n</i> _D ¹⁸ 1.5387 80.01 7.70
146	217					a	90	C ₂₇ H ₃₀ O ₃ C: 80.56 H: 7.51	<i>n</i> _D ¹⁸ 1.5542 80.52 7.52
147	218					a	92	C ₂₆ H ₃₀ O ₃ C: 79.96 H: 7.74	<i>n</i> _D ²² 1.5385 79.94 7.83
148	219					a	93	C ₂₇ H ₃₀ O ₃ C: 80.56 H: 7.51	<i>n</i> _D ¹⁹ 1.5548 80.58 7.53

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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
149	220				a	90	C ₂₇ H ₃₂ O ₃ C: 80.16 H: 7.97	n_D^{24} 1.5302 80.21 8.04
150	221				a	94	C ₂₈ H ₃₂ O ₃ C: 80.73 H: 7.74	n_D^{20} 1.5327 80.67 7.71
151	222				a	82	C ₂₁ H ₂₁ O ₃ SCl C: 64.85 H: 5.44 S: 8.25 Cl: 9.12	n_D^{20} 1.5377 64.88 5.42 8.26 9.13
152	223				a	77	C ₂₂ H ₂₁ O ₃ SF C: 68.73 H: 5.51 S: 8.34 F: 4.94	n_D^{20} 1.5467 68.70 5.59 8.32 4.92
153	224				a	86	C ₂₃ H ₂₄ O ₃ S C: 72.60 H: 6.36 S: 8.43	n_D^{19} 1.5366 72.54 6.33 8.50
154	225				a	79	C ₂₃ H ₂₄ O ₃ S C: 72.60 H: 6.36 S: 8.43	n_D^{19} 1.5277 72.61 6.38 8.49
155	226				a	81	C ₂₆ H ₂₈ O ₂ C: 81.62 H: 10.01	n_D^{20} 1.5235 81.66 10.03

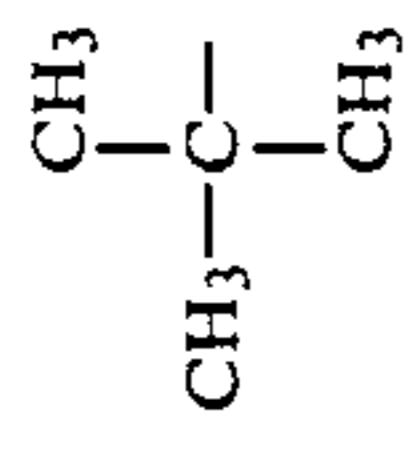
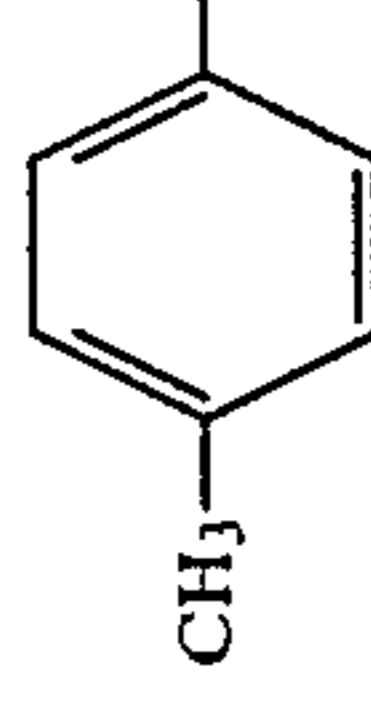
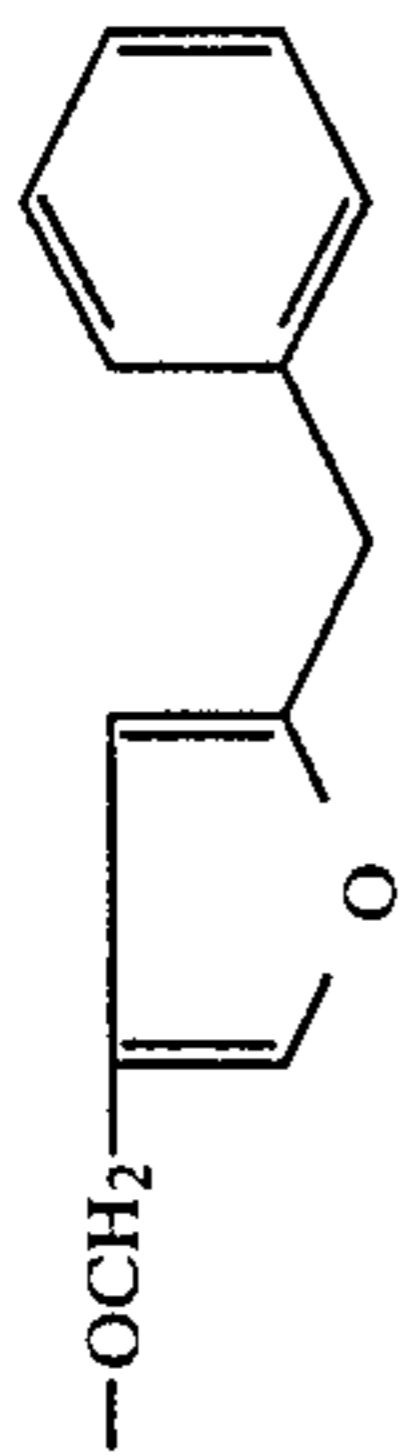
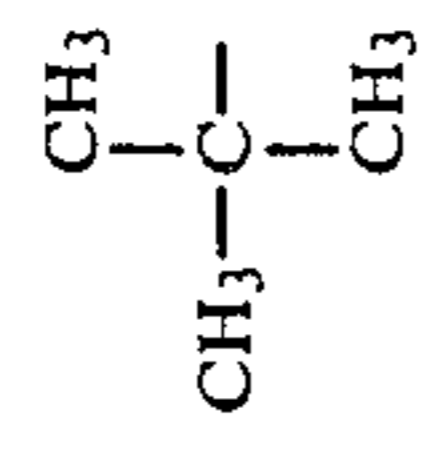
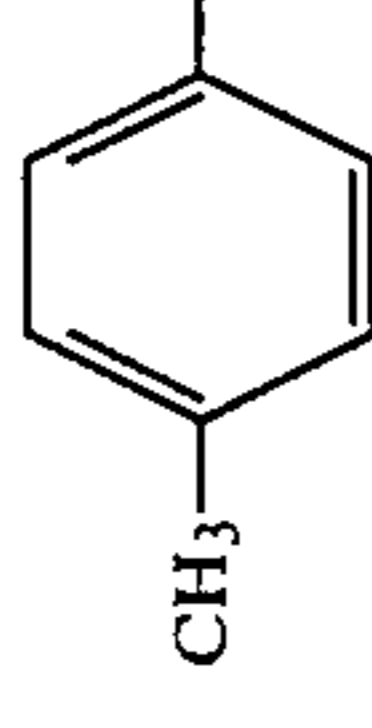
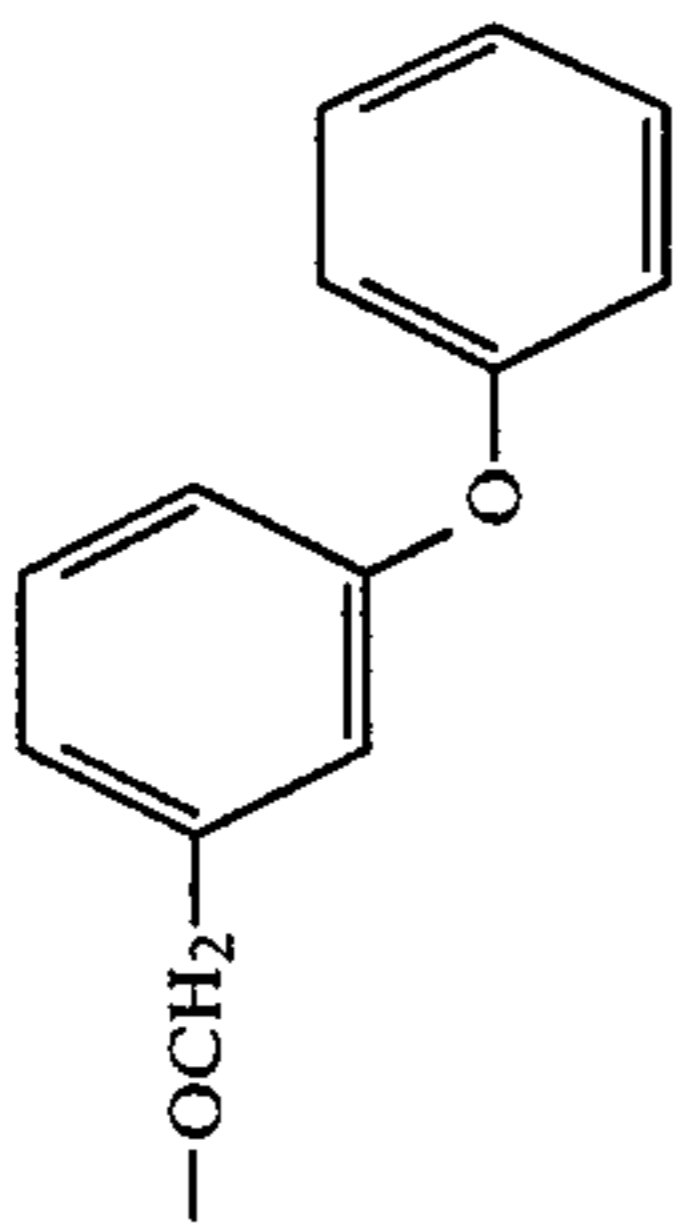
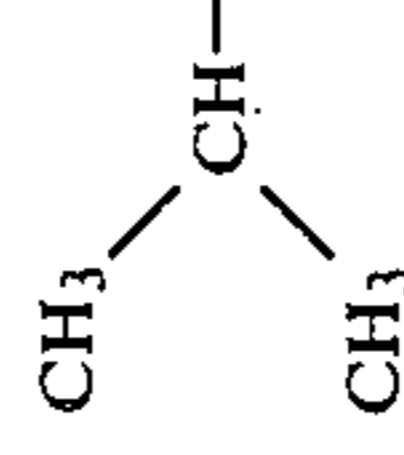
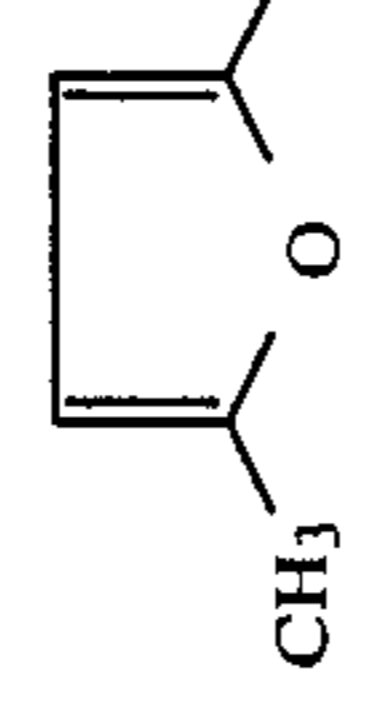
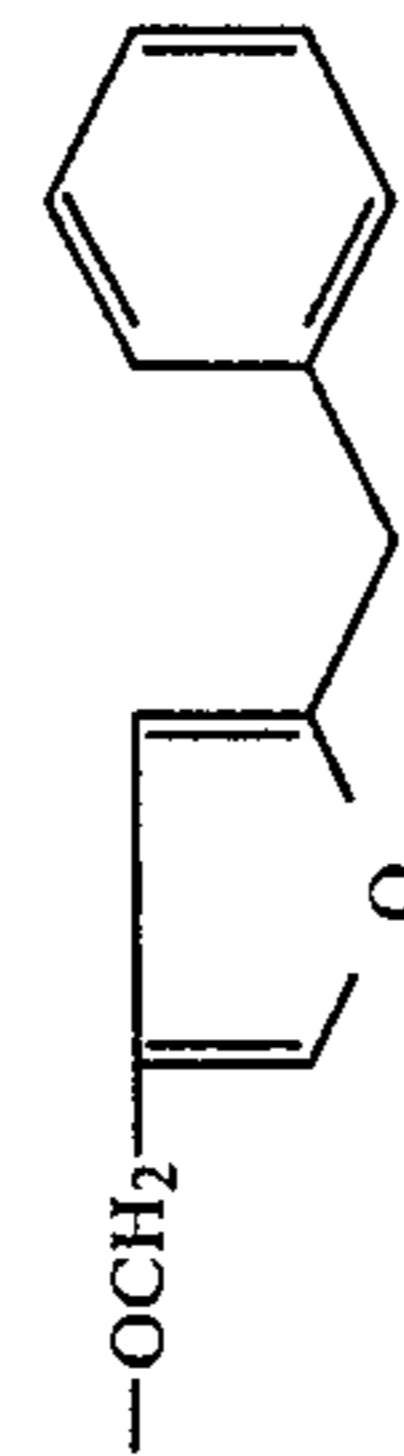
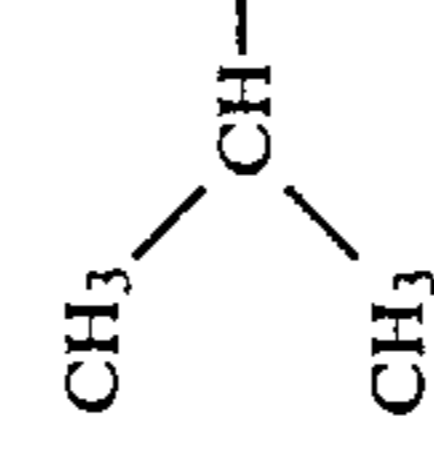
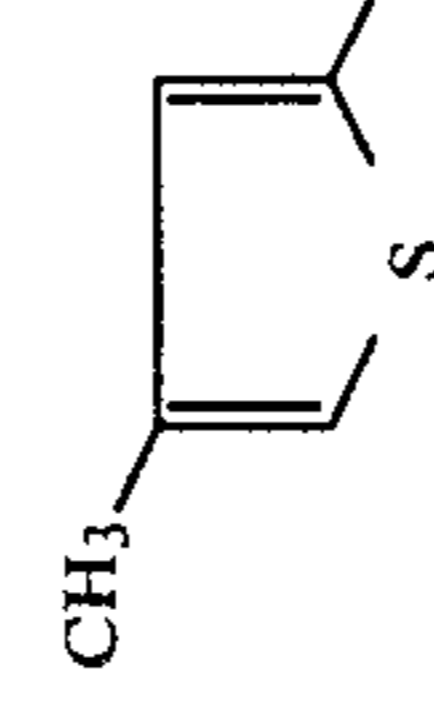
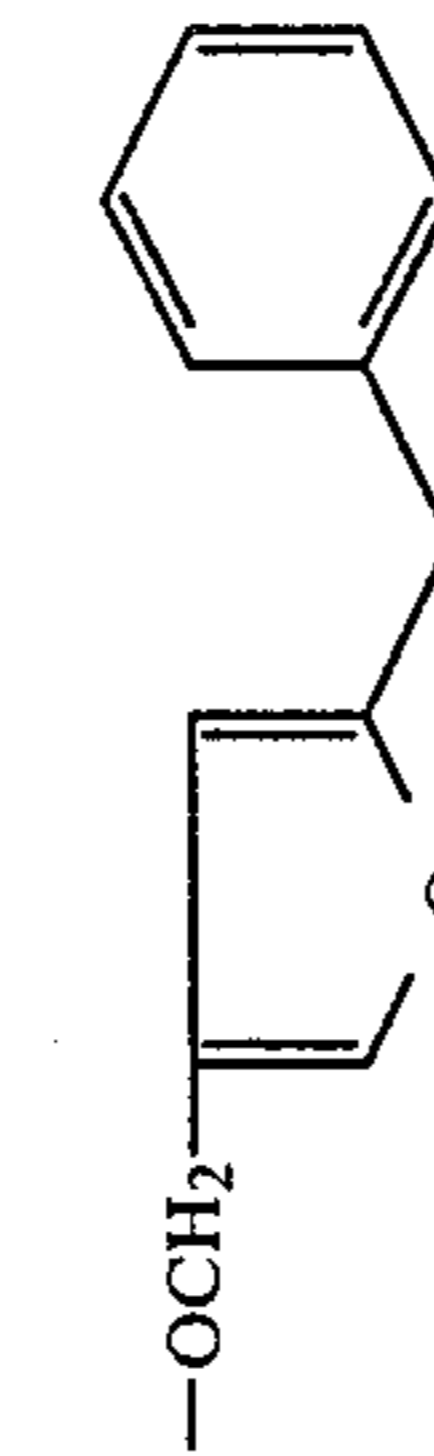
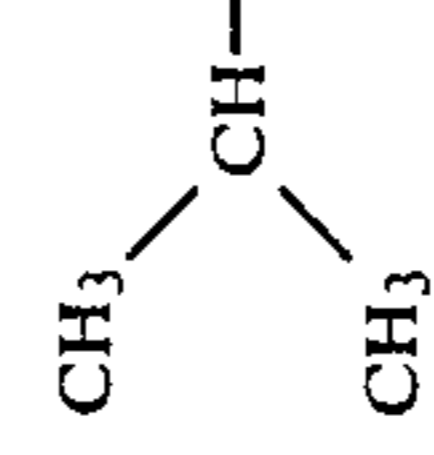
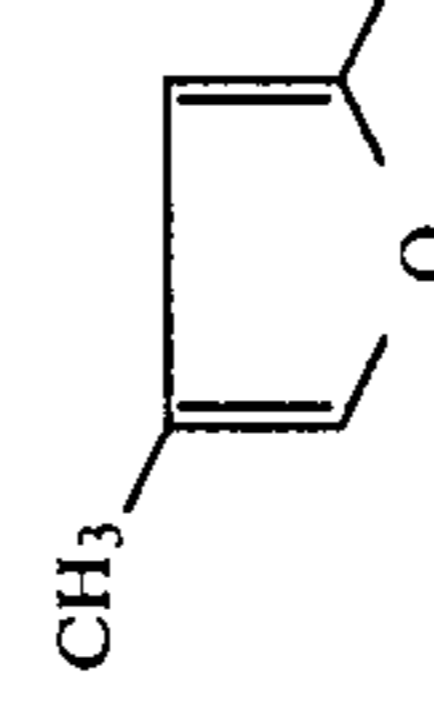
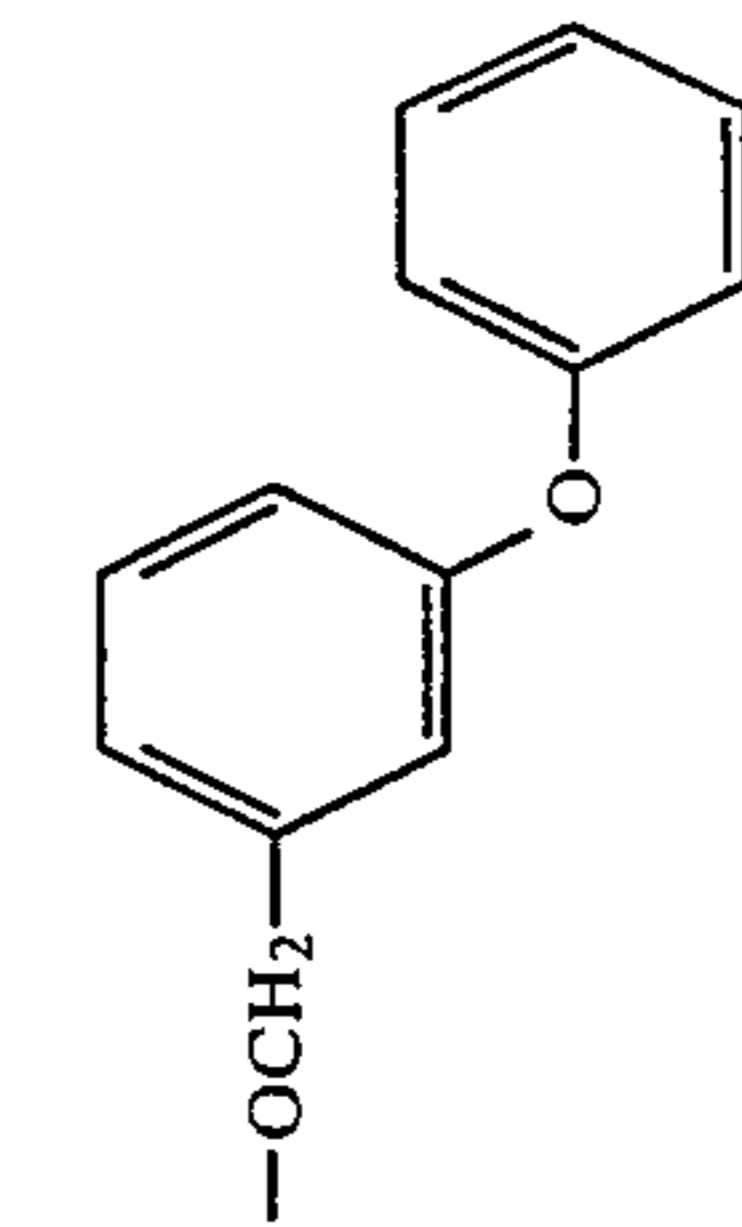
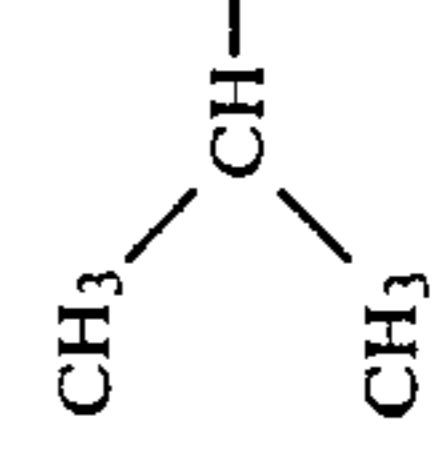
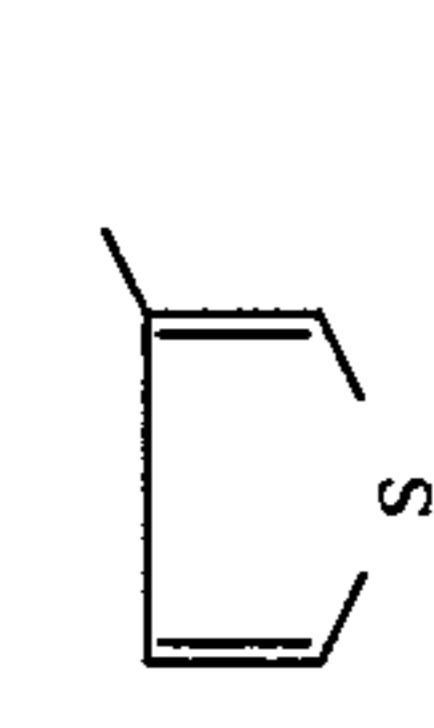
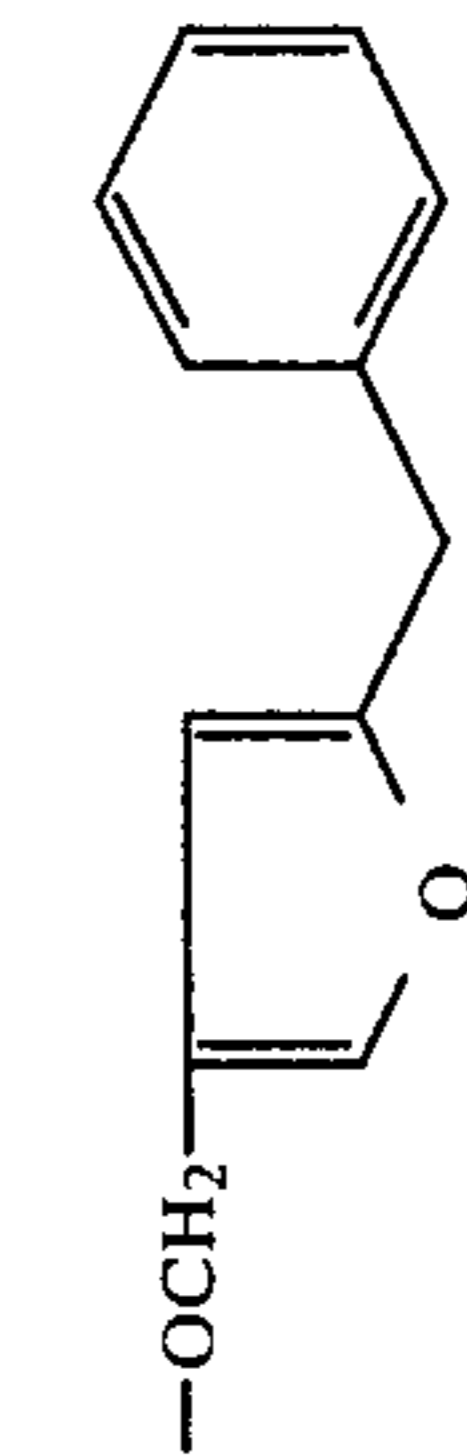
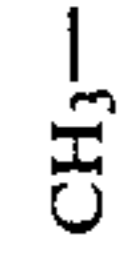
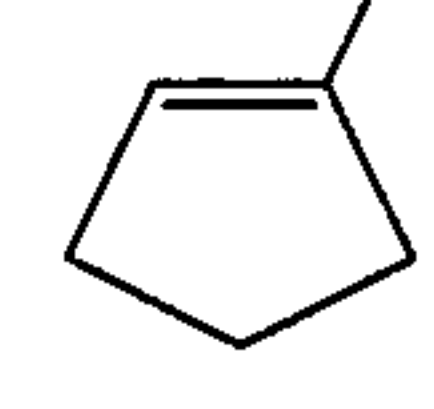
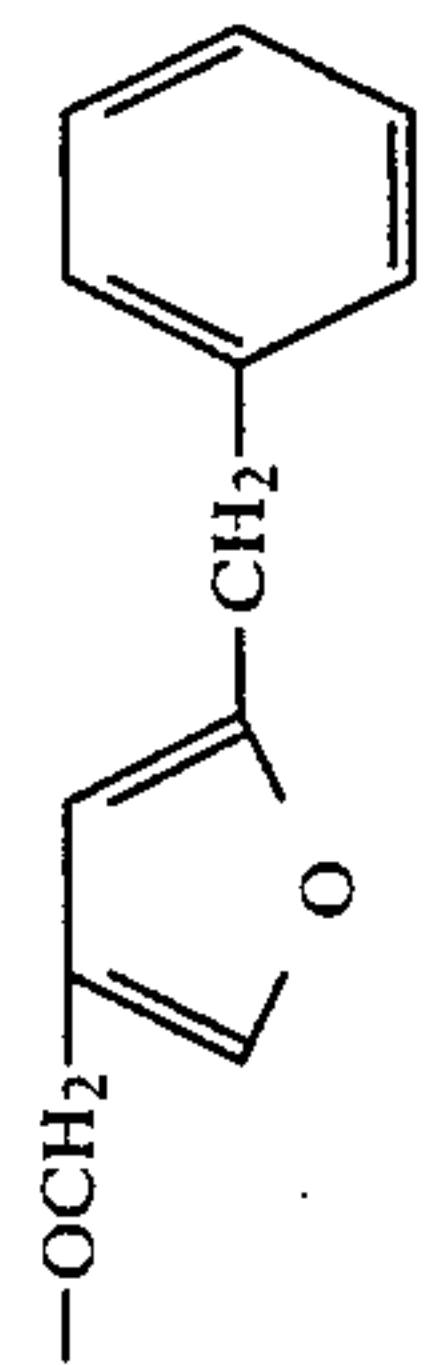
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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
156	227				a	72	C ₂₆ H ₂₅ O ₃ N C: 78.17 H: 6.31 N: 3.51	n_D^{20} 1.5624 78.19 6.28 3.61
157	228				a	83	C ₂₂ H ₂₁ O ₃ SCl C: 65.90 H: 5.28 S: 8.00 Cl: 8.84	n_D^{19} 1.5433 65.99 5.27 8.04 8.82
158	229				a	85	C ₂₂ H ₂₁ O ₃ SCl C: 65.90 H: 5.28 S: 8.00 Cl: 8.84	n_D^{19} 1.5436 65.91 5.29 8.16 8.83
159	230				a	90	C ₂₅ H ₂₅ O ₂ Cl C: 76.42 H: 6.41 Cl: 9.02	n_D^{20} 1.5329 76.44 6.43 9.09
160	231				a	74	C ₂₅ H ₂₂ O ₃ NCl C: 71.51 H: 5.28 N: 3.34 Cl: 8.44	n_D^{20} 1.5533 71.54 5.30 3.31 8.46
161	232				a	75	C ₂₇ H ₂₇ O ₃ N C: 78.42 H: 6.58 N: 3.39	n_D^{20} 1.5239 78.44 6.61 3.44

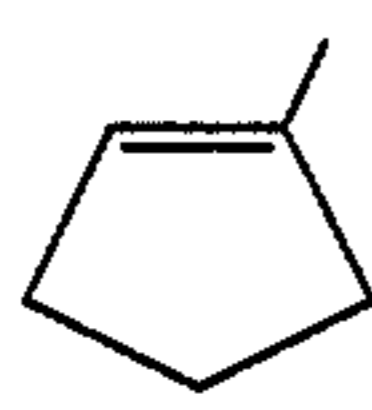
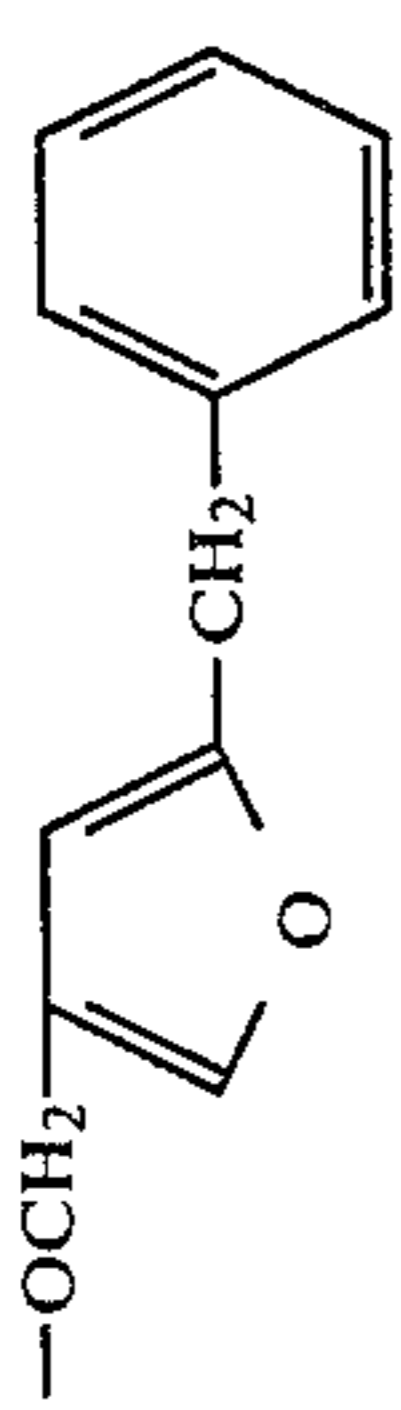
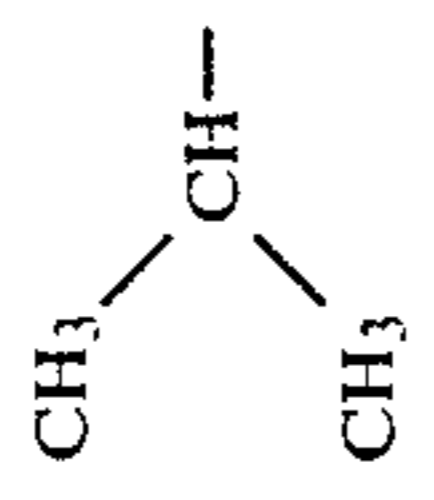

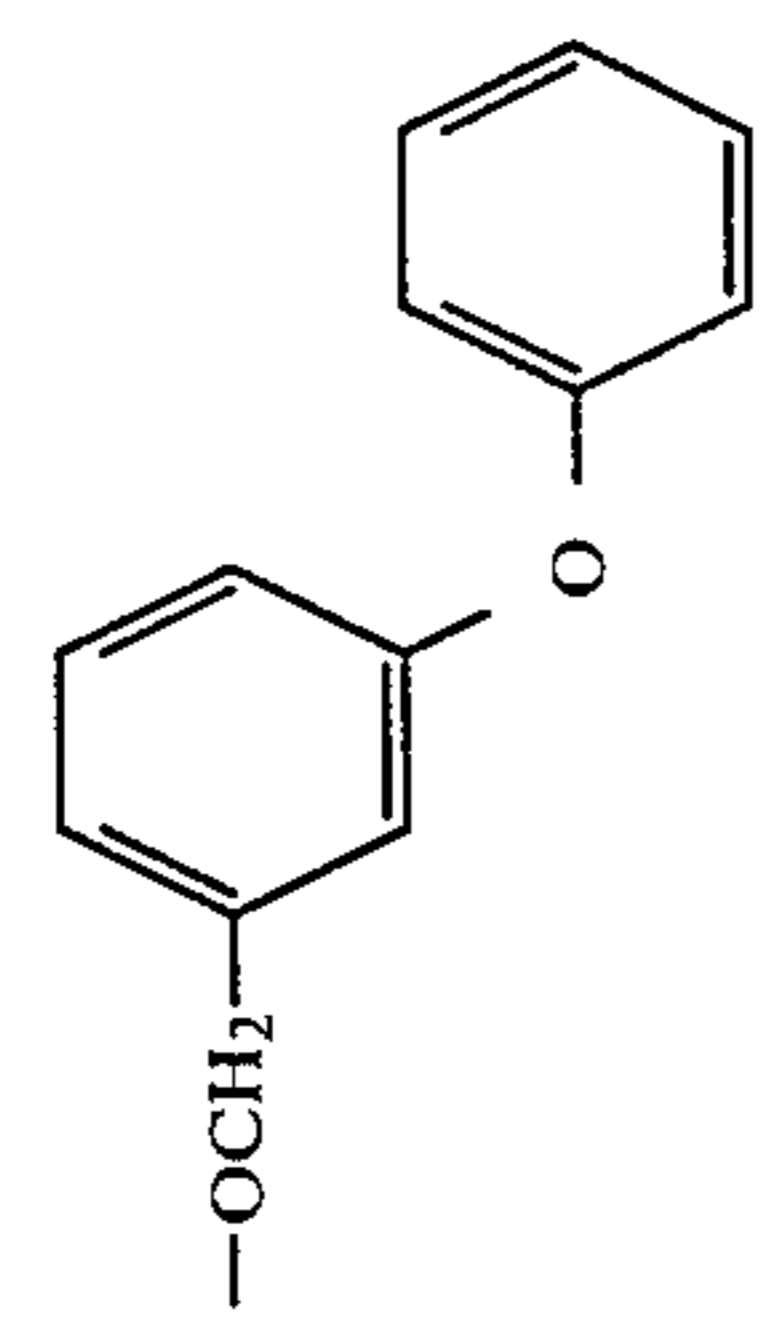
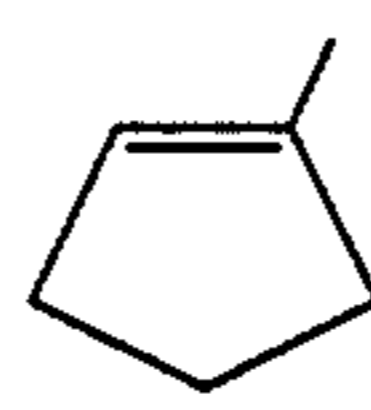
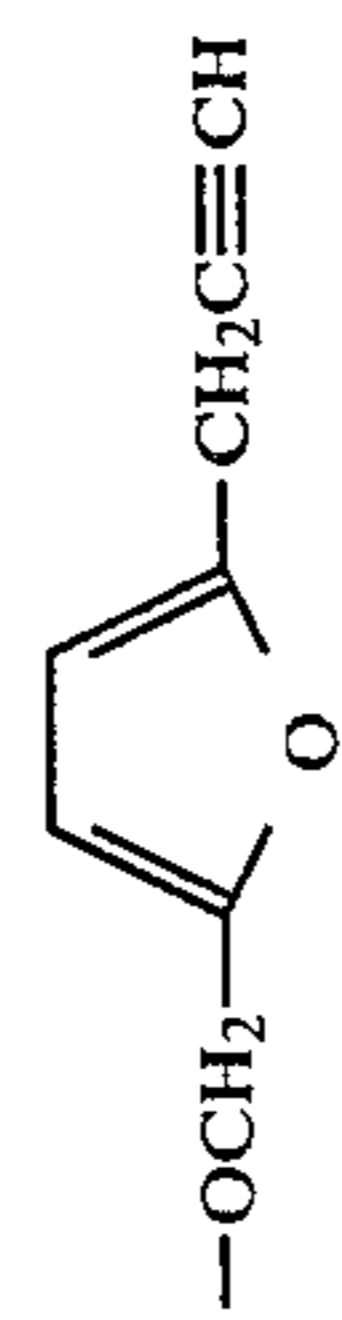
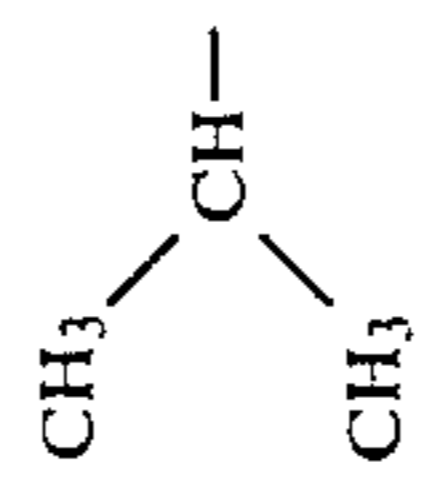
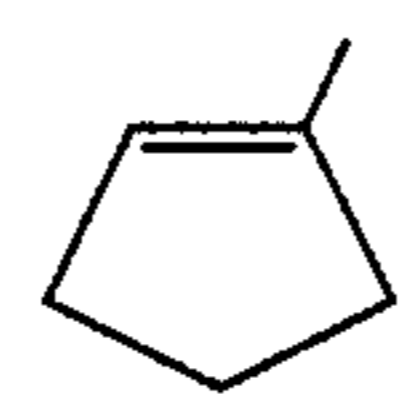
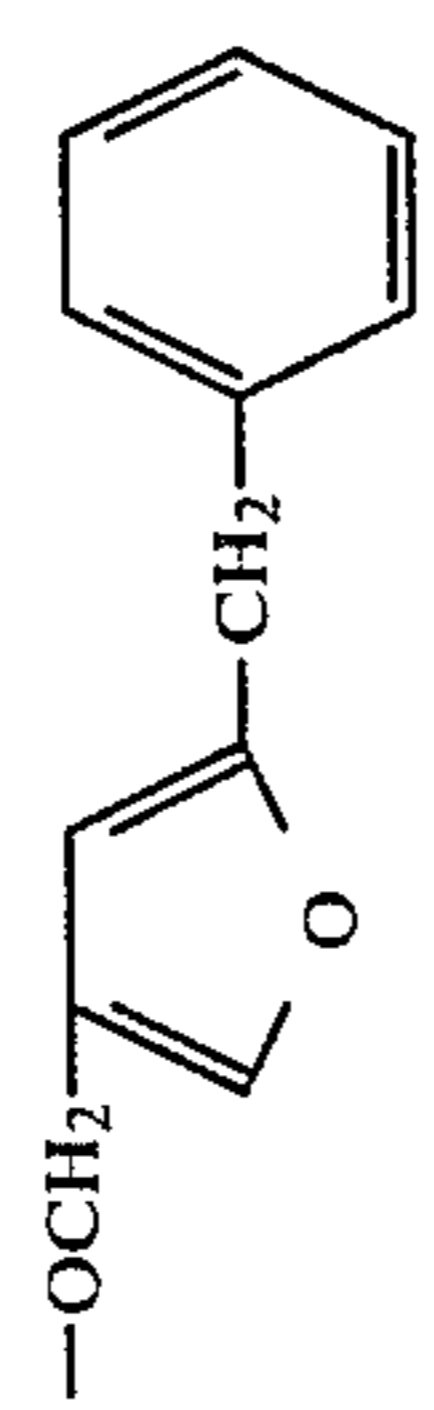
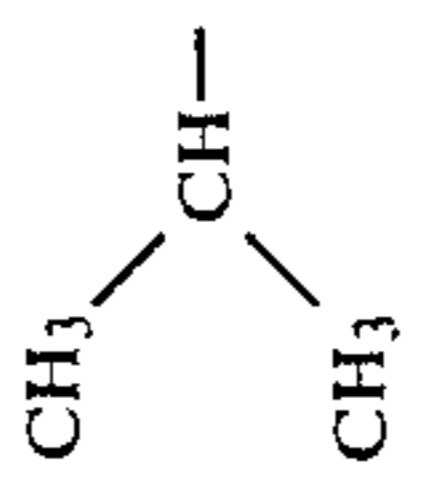
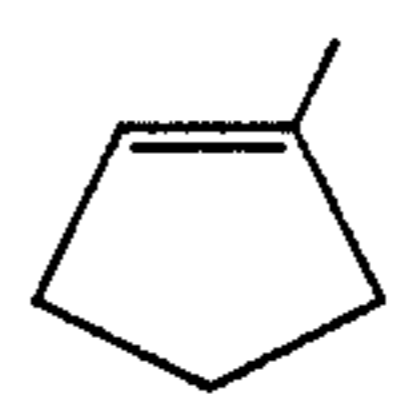
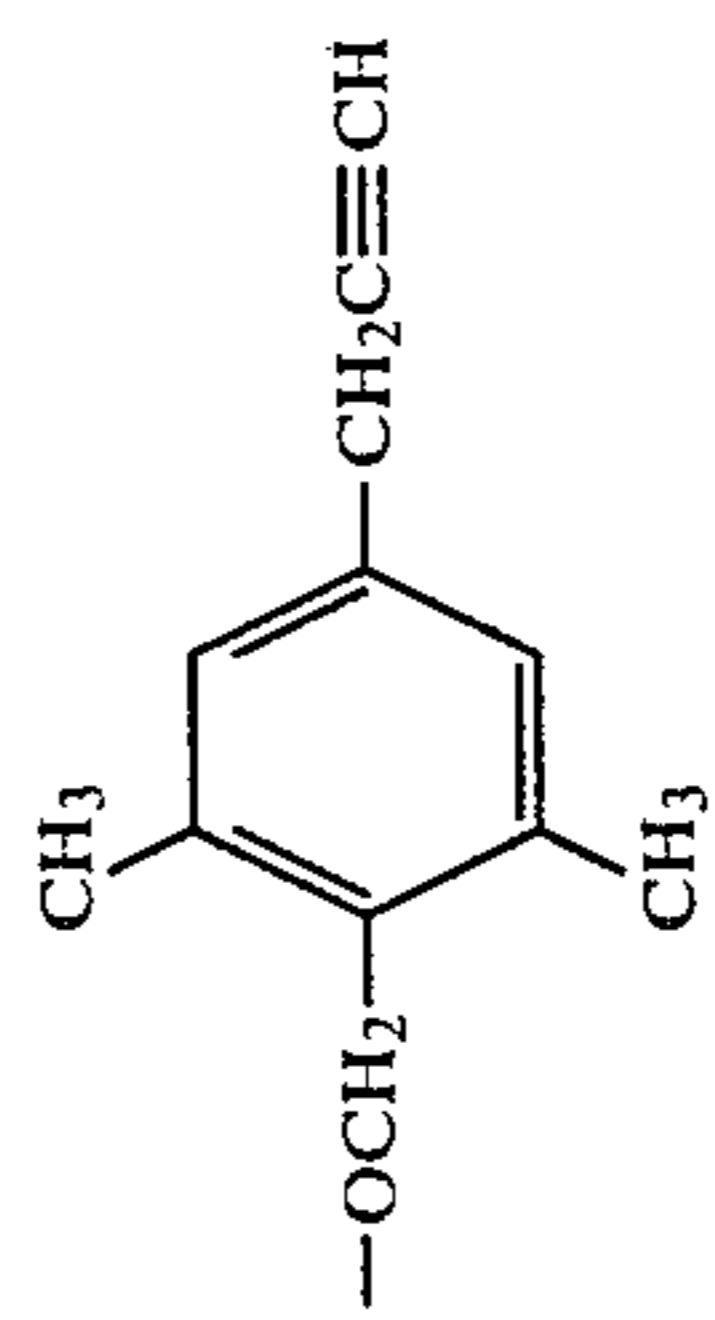
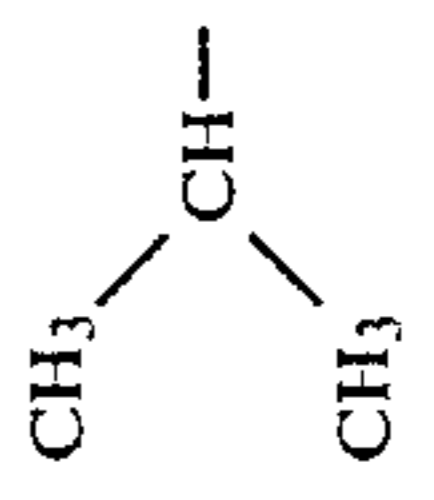
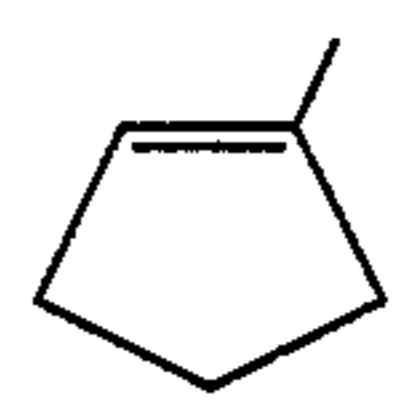
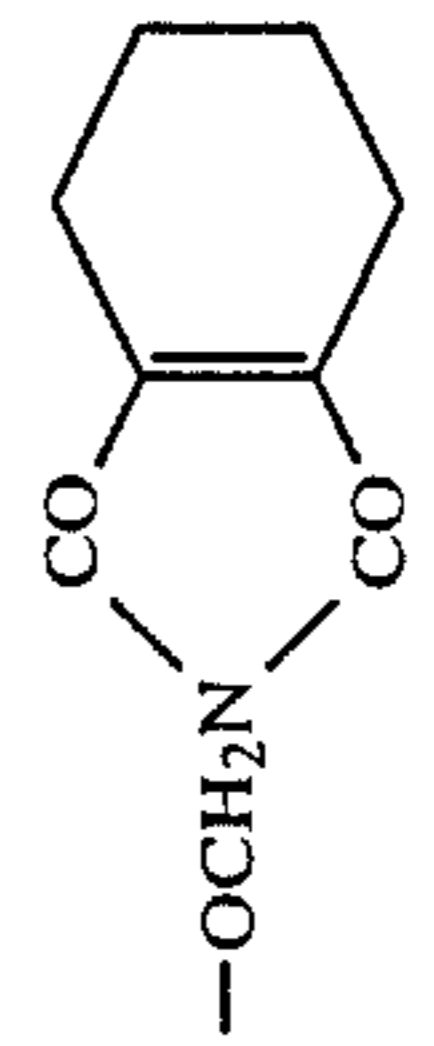
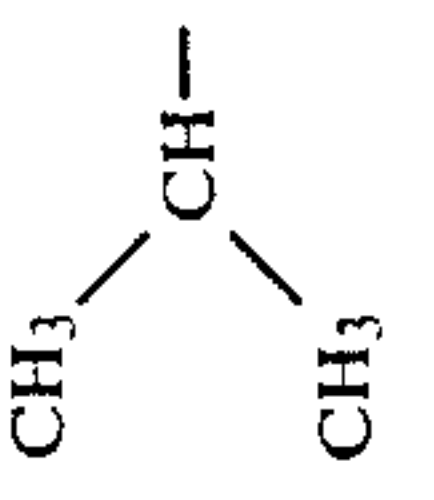
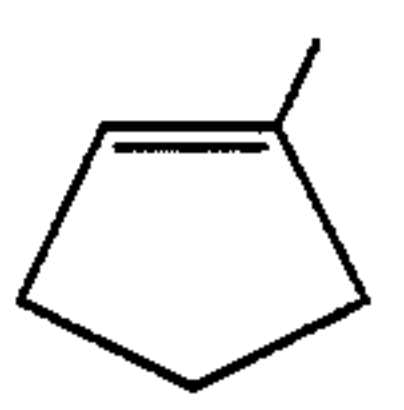
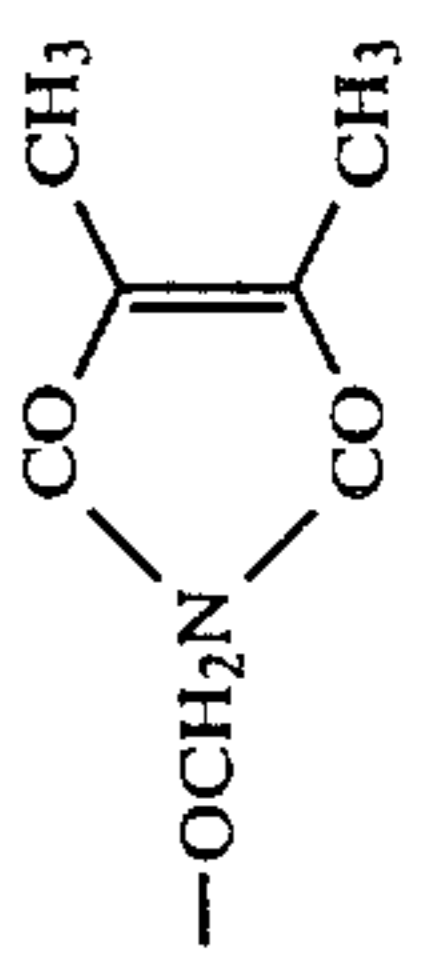
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Exp. No.	Preparation No.	Comp. Z	Y	X	Chemical Structure	Esterification Method	Yield (%)	Physical Properties	
								Cacl. (%)	Found (%)
162	233			X		a	86	C ₂₅ H ₂₈ O ₃ S C: 73.49 H: 6.91 S: 7.85	n_D^{20} 1.5567 73.51 6.94 7.86
163	234					a	89	C ₂₈ H ₃₂ O ₂ C: 83.96 H: 8.05	n_D^{20} 1.5341 83.99 8.06
164	235					a	80	C ₂₈ H ₂₉ O ₃ N C: 78.66 H: 6.84 N: 3.28	n_D^{20} 1.5226 78.67 6.90 3.11
165	236					a	79	C ₂₉ H ₃₁ O ₃ N C: 78.88 H: 7.08 N: 3.17	n_D^{20} 1.5451 78.79 7.06 3.24
166	237					a	87	C ₂₄ H ₂₆ O ₃ C: 79.53 H: 7.23	n_D^{20} 1.5335 79.56 7.20
167	238					a	89	C ₂₅ H ₂₅ O ₃ Cl C: 73.43 H: 6.16 Cl: 8.67	n_D^{19} 1.5427 73.49 6.20 8.60

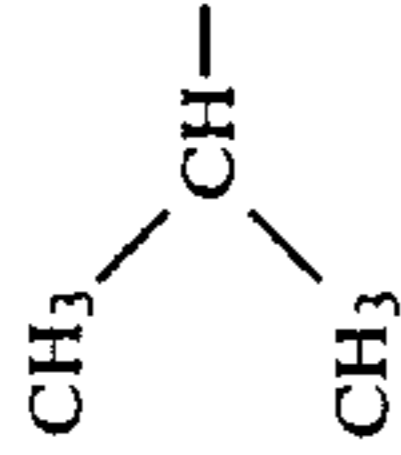
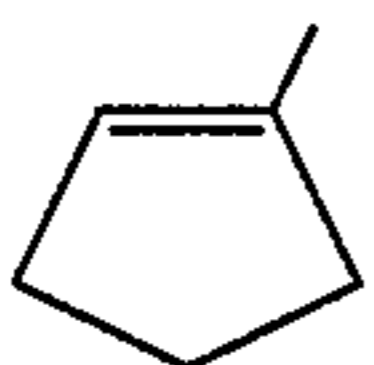
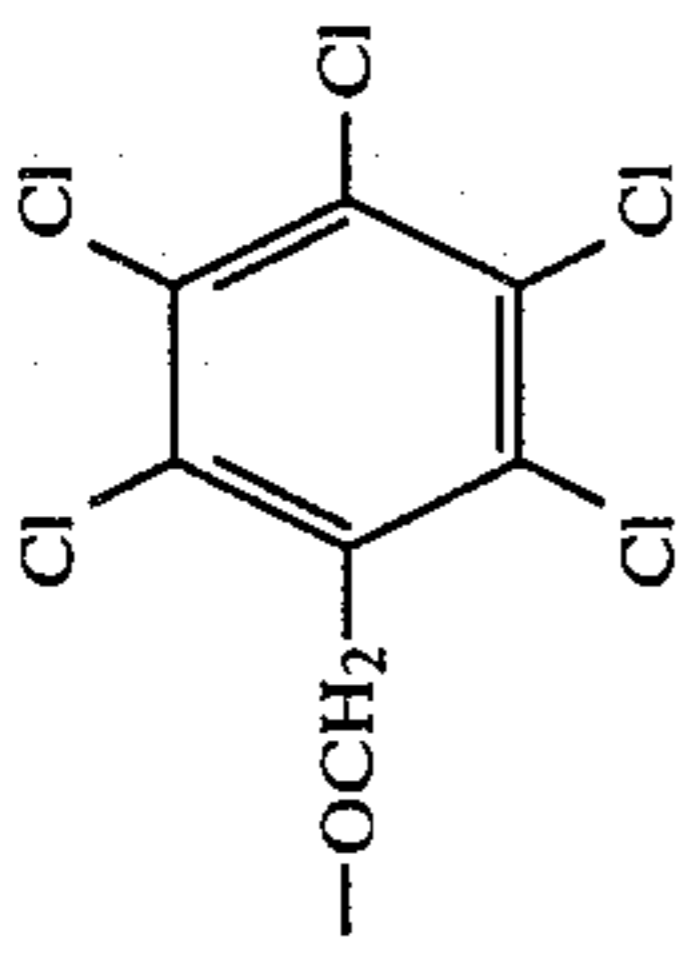
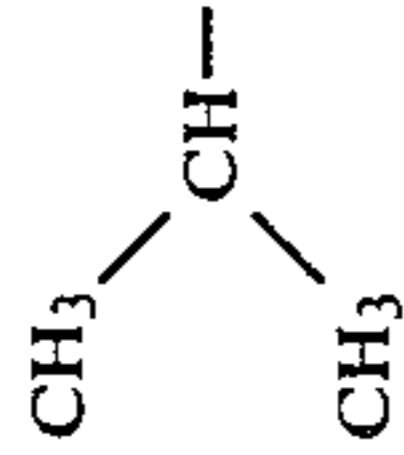
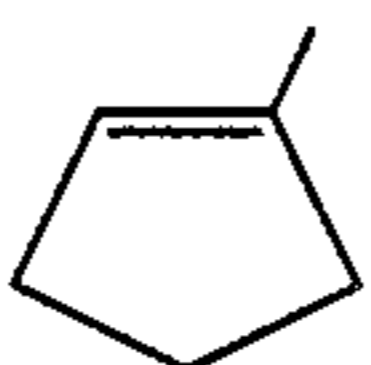
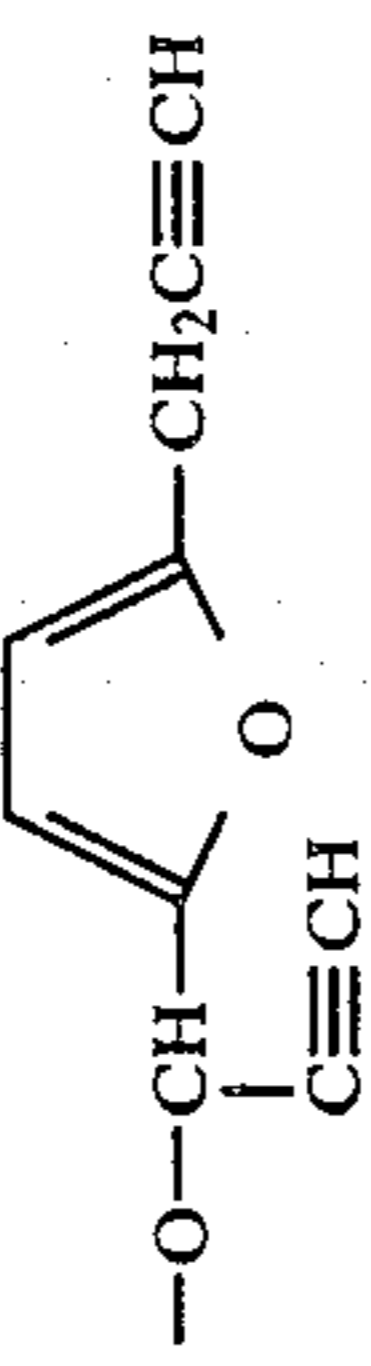
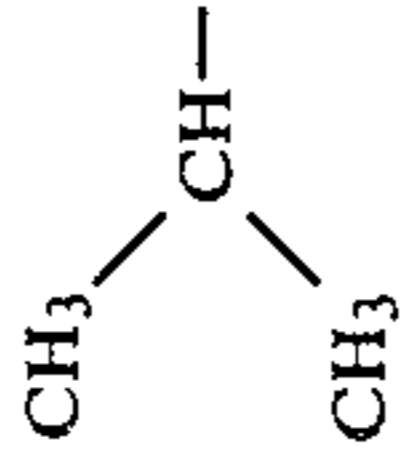
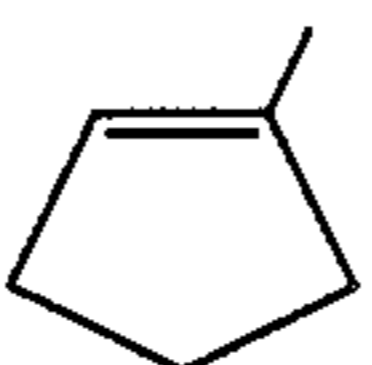
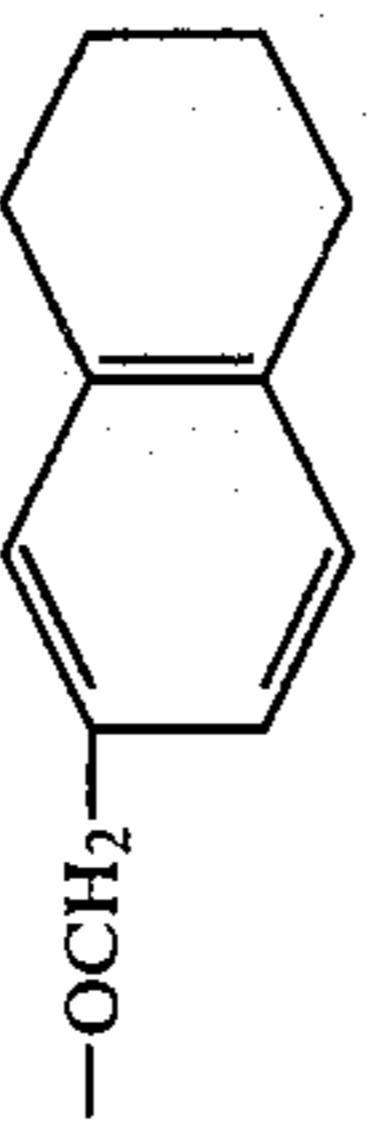
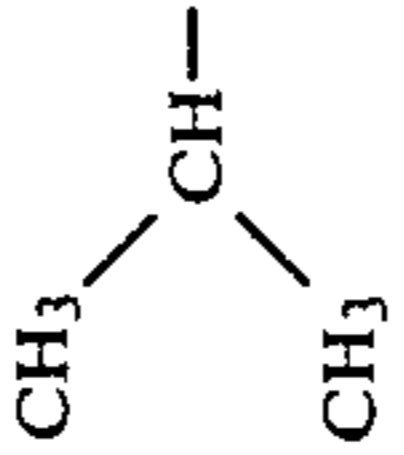
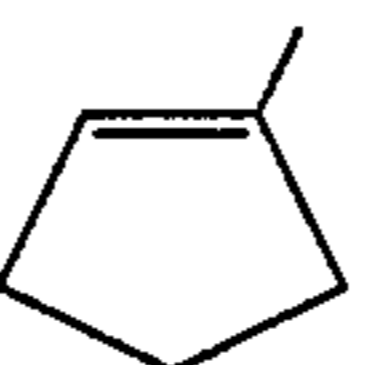
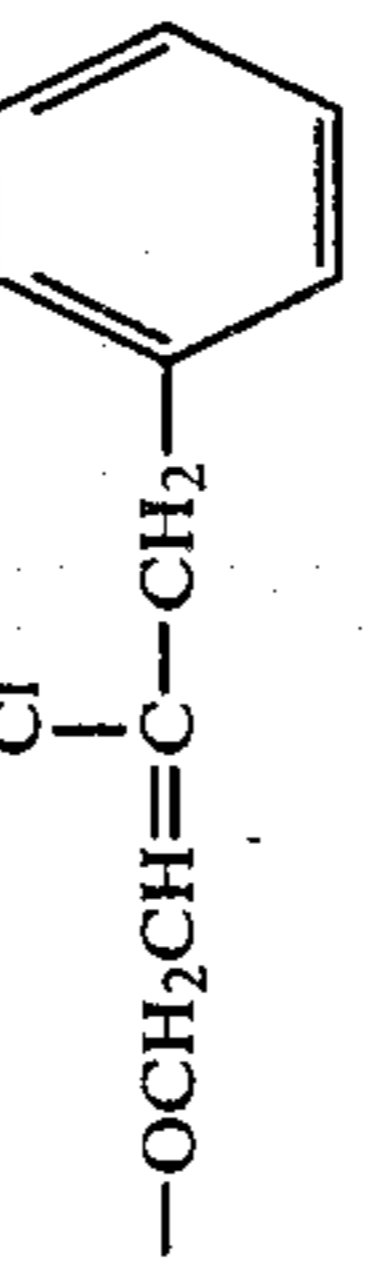
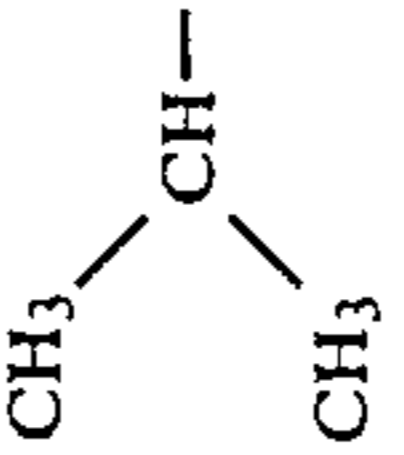
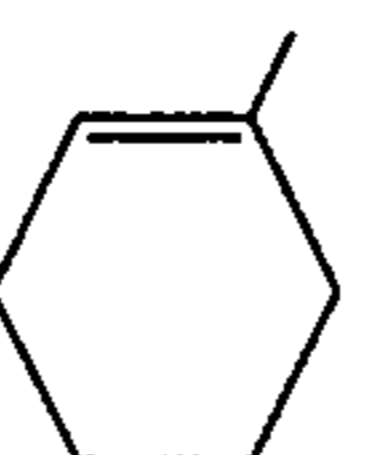
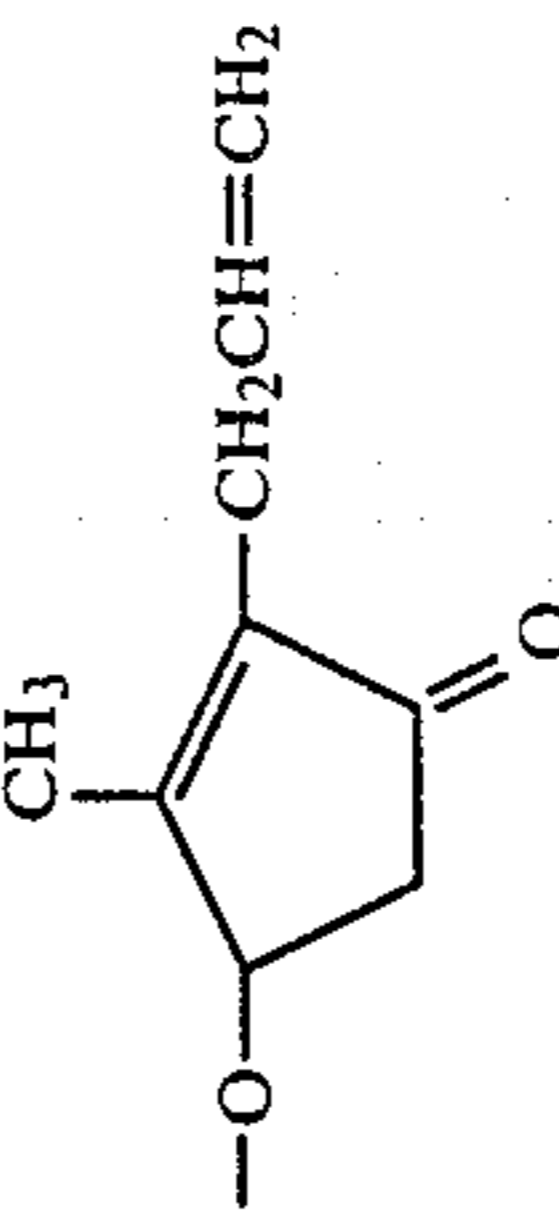
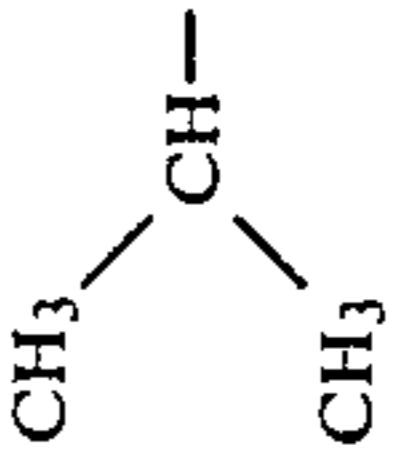
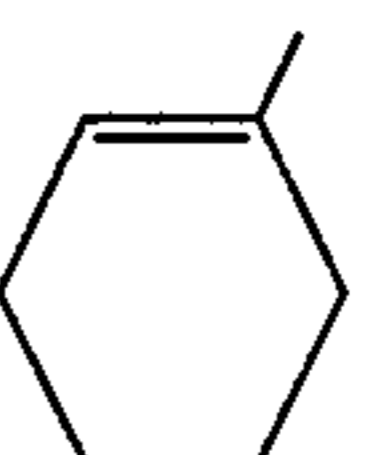
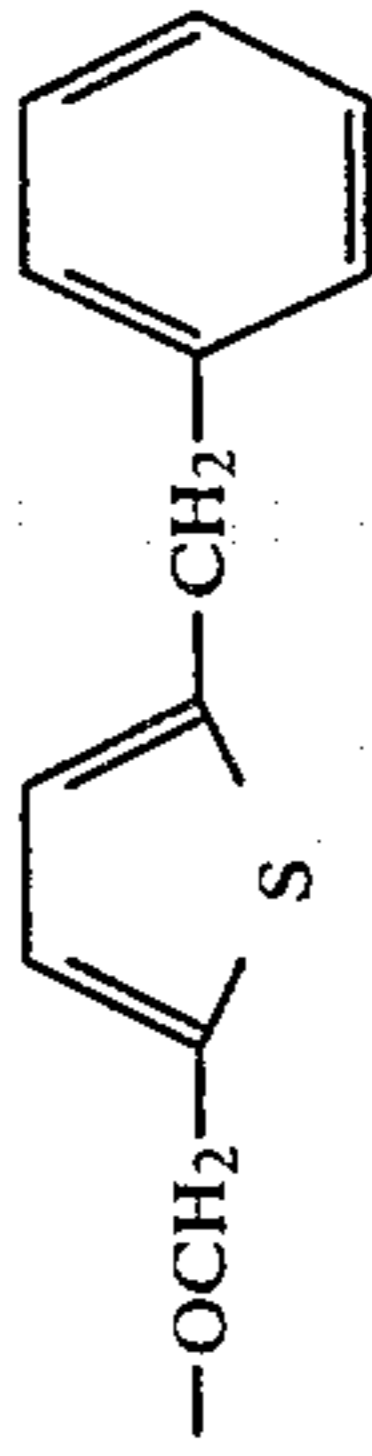
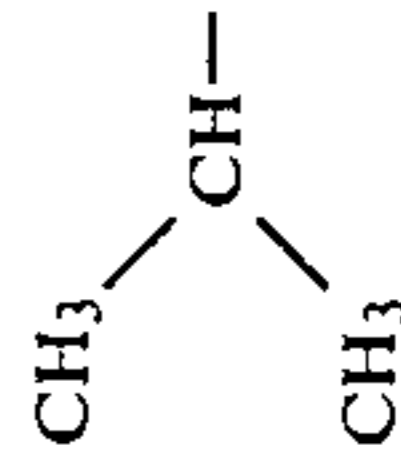
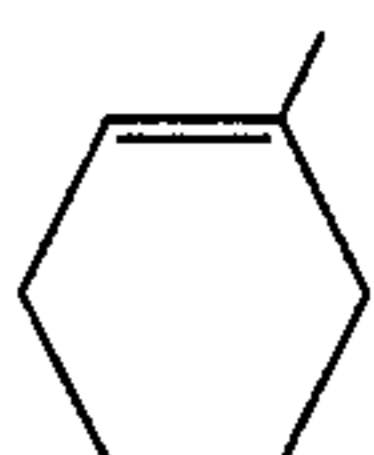
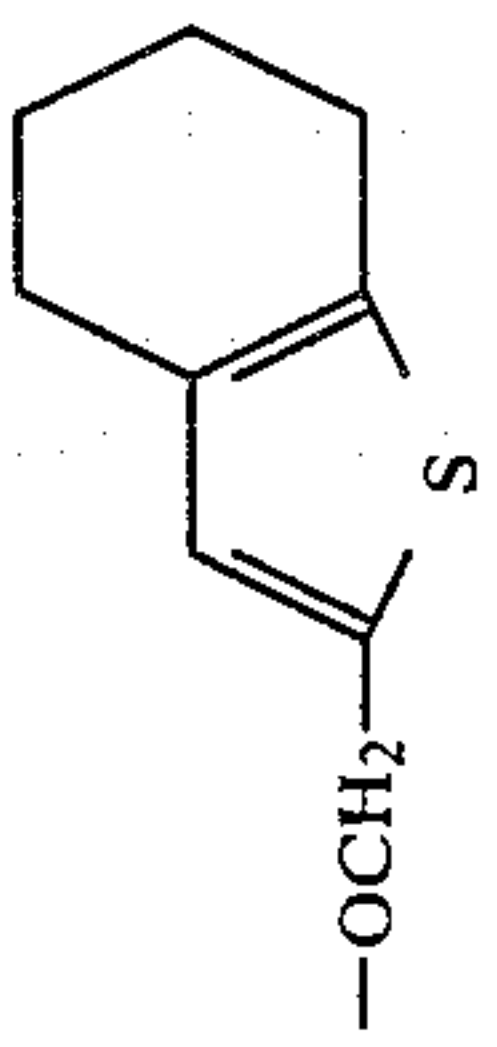
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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
168	239				a	89	C ₂₅ H ₂₈ O ₃ C: 79.75 H: 7.49	n_D^{20} 1.5366 79.72 7.52
169	240				a	91	C ₂₆ H ₂₈ O ₃ C: 80.38 H: 7.26	n_D^{21} 1.5225 80.33 7.24
170	241				a	76	C ₂₂ H ₂₄ O ₄ C: 74.98 H: 6.86	n_D^{22} 1.5236 75.02 6.80
171	242				a	79	C ₂₂ H ₂₄ O ₃ S C: 71.71 H: 6.56 S: 8.70	n_D^{20} 1.5475 71.77 6.52 8.72
172	243				a	79	C ₂₃ H ₂₄ O ₄ C: 75.80 H: 6.64	n_D^{20} 1.5344 75.76 6.62
173	244				a	82	C ₂₁ H ₂₂ O ₃ S C: 71.15 H: 6.26 S: 9.04	n_D^{19} 1.5226 71.09 6.22 9.11
174	245				a	89	C ₂₀ H ₂₂ O ₃ C: 77.37 H: 7.14	n_D^{28} 1.5293 77.29 7.16

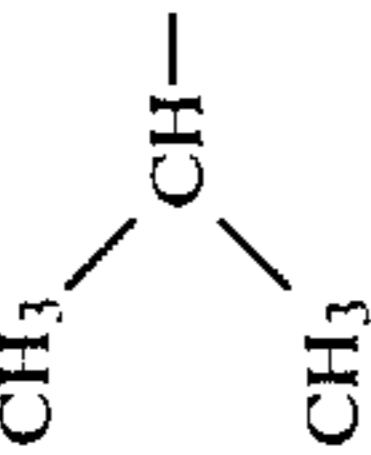
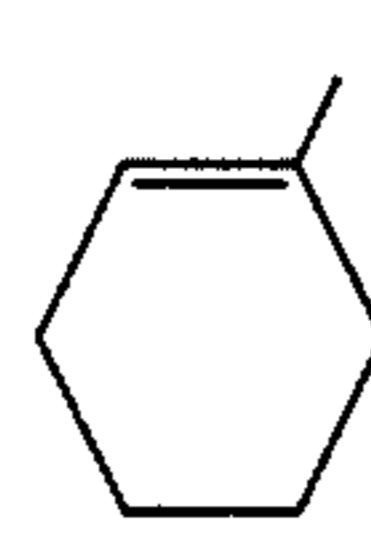
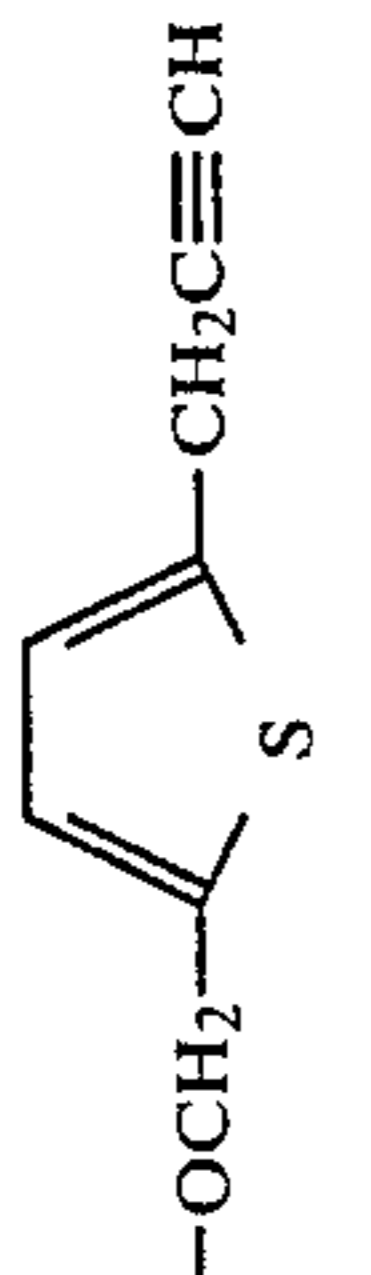
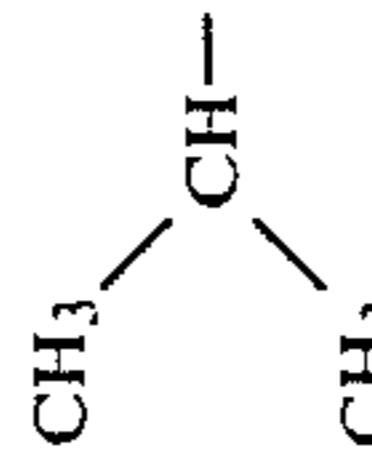

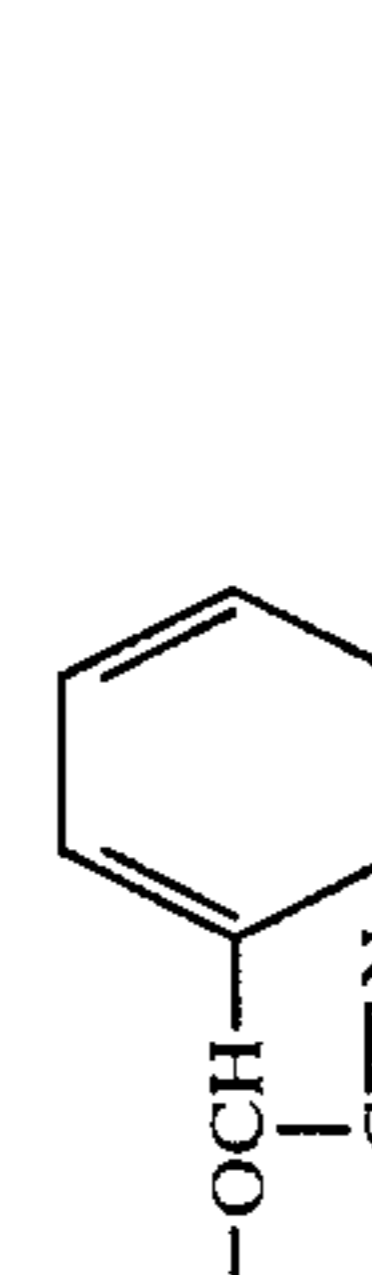
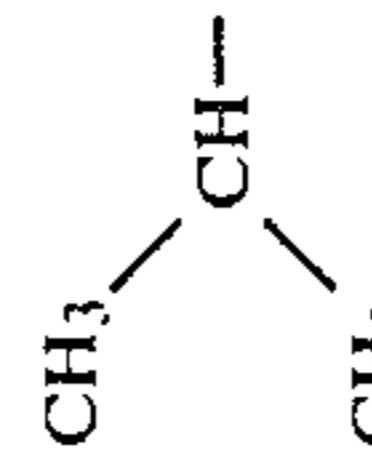
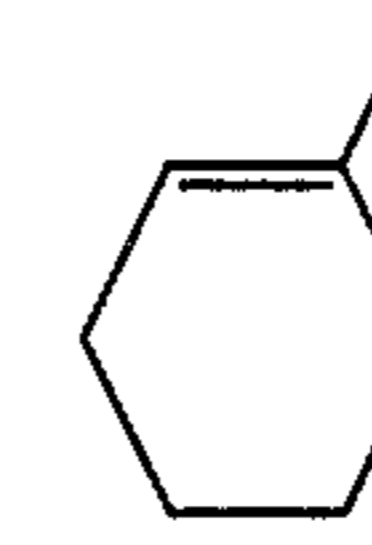
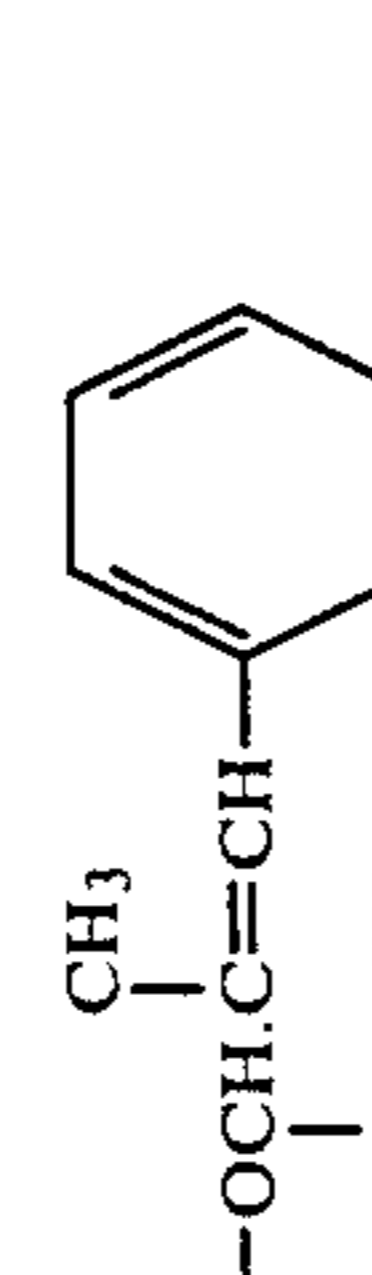
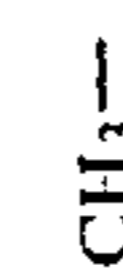
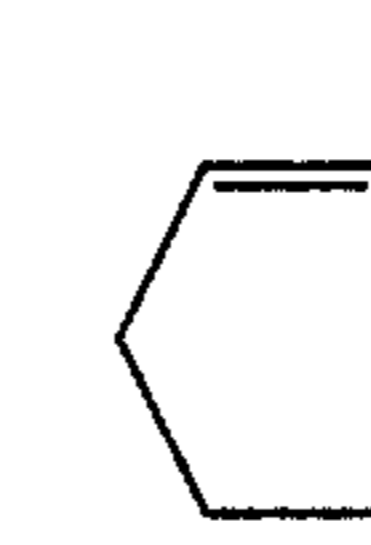
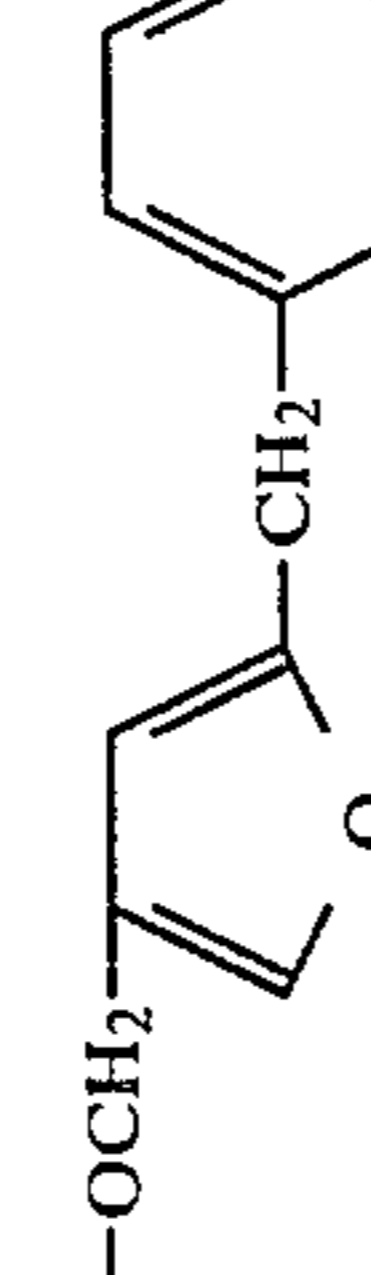
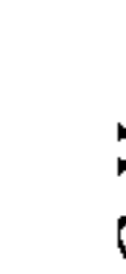
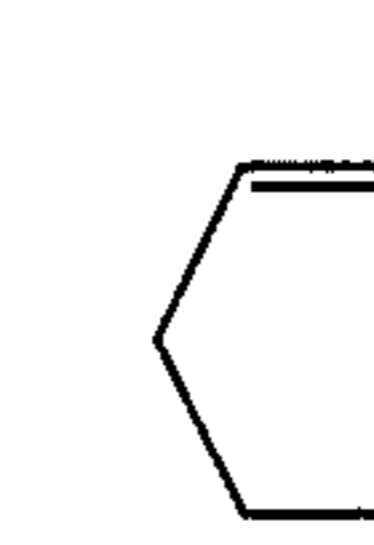
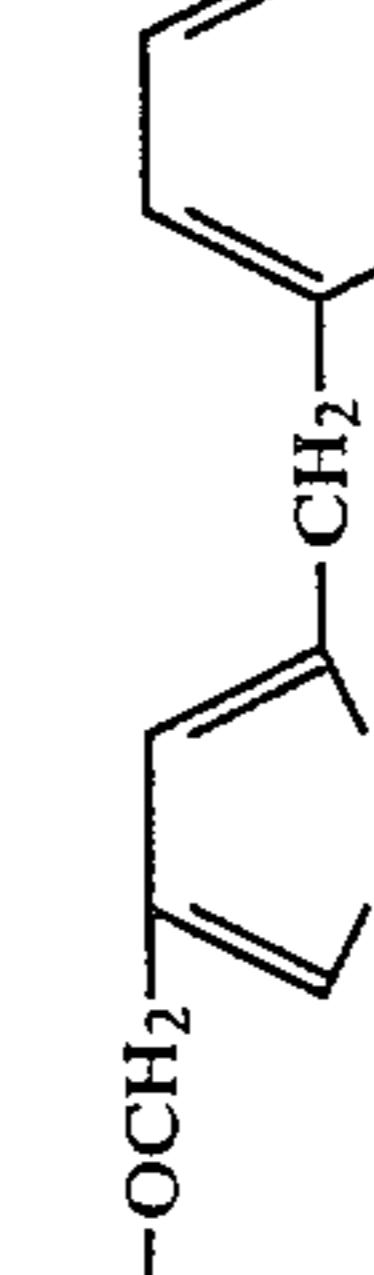
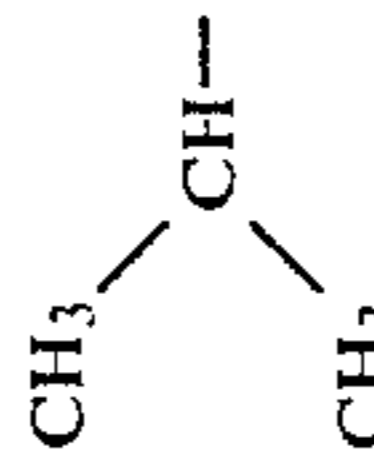
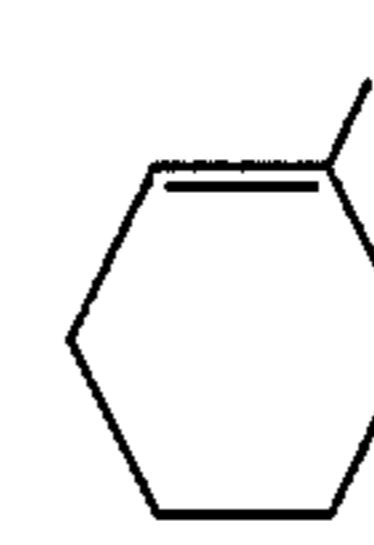
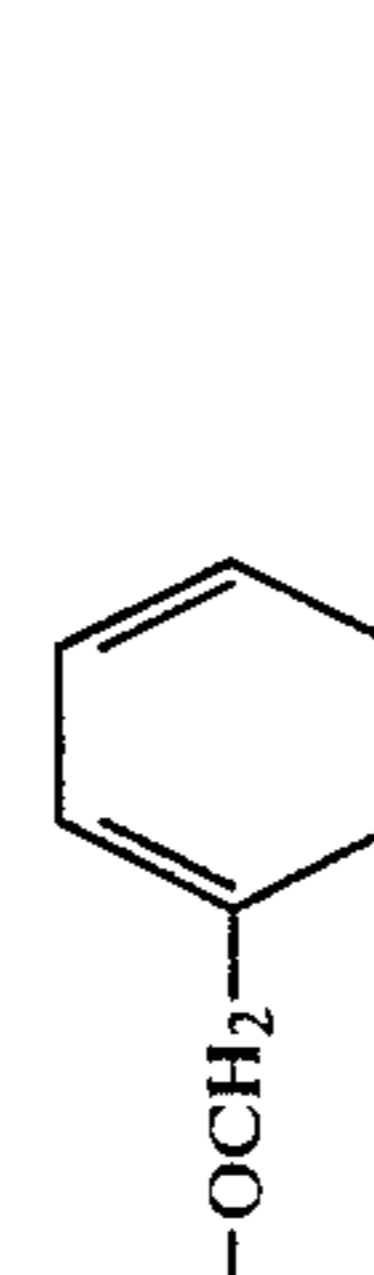
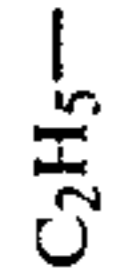
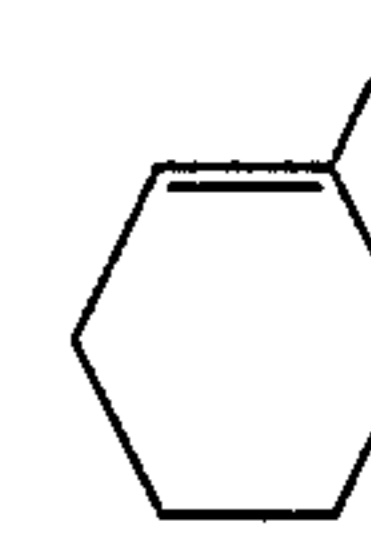
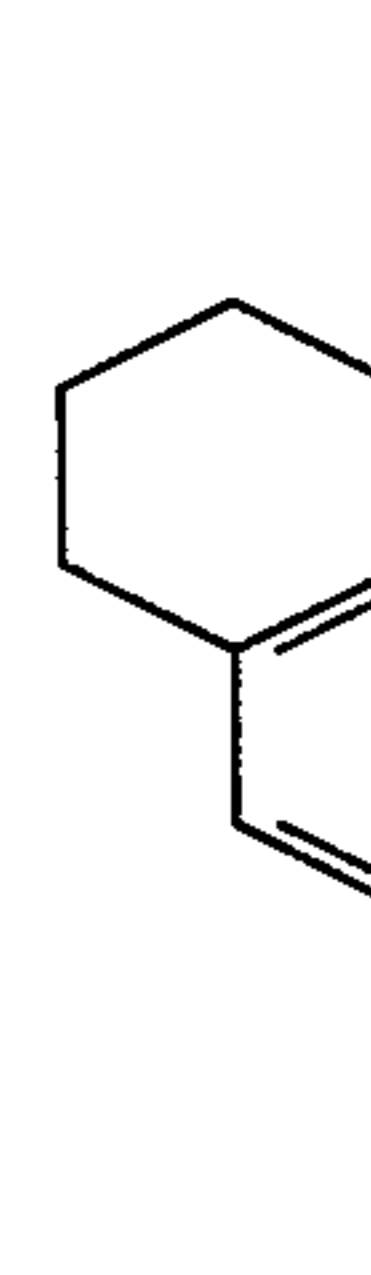
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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
175	246	C ₂ H ₅			a	89	C ₂₁ H ₂₄ O ₃ C: 77.75 H: 7.46	n _D ^{24.5} 1.5276 77.78 7.44
176	247				a	86	C ₂₃ H ₂₆ O ₃ C: 78.83 H: 7.48	n _D ²⁸ 1.5483 78.81 7.40
177	248	C ₂ H ₅			a	90	C ₁₈ H ₂₂ O ₃ C: 75.49 H: 7.74	n _D ^{24.5} 1.4965 75.53 7.69
178	249				a	90	C ₂₂ H ₂₆ O ₃ C: 78.07 H: 7.74	n _D ^{24.5} 1.5248 78.10 7.74
179	250				a	85	C ₂₂ H ₂₈ O ₂ C: 81.44 H: 8.70	n _D ²⁸ 1.5615 81.64 8.53
180	251				a	91	C ₁₉ H ₂₅ O ₄ N C: 68.86 H: 7.60 N: 4.23	n _D ²⁵ 1.5380 68.69 7.64 4.10
181	252				a	88	C ₁₇ H ₂₃ O ₄ N C: 66.86 H: 7.59 N: 4.59	n _D ²⁵ 1.5237 66.97 7.50 4.45

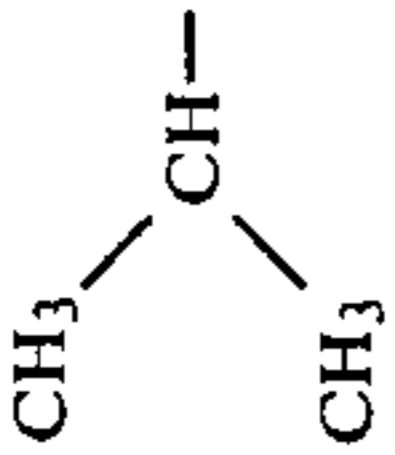
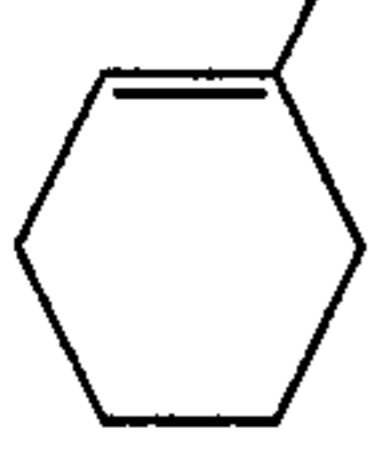
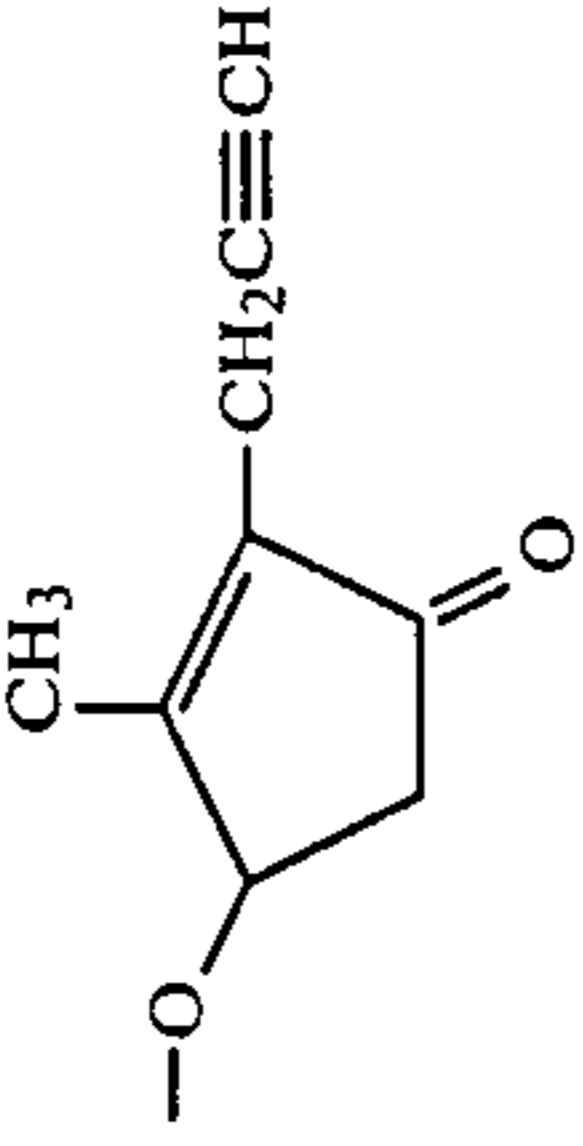
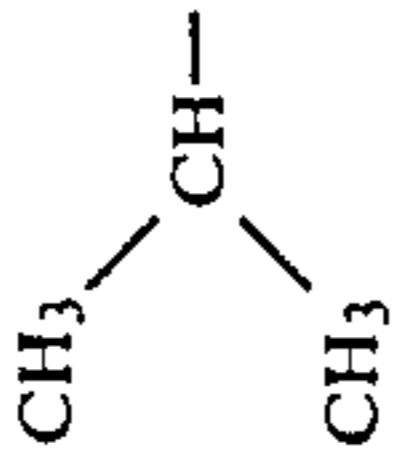
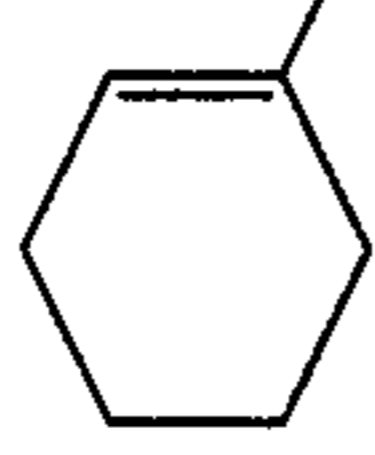
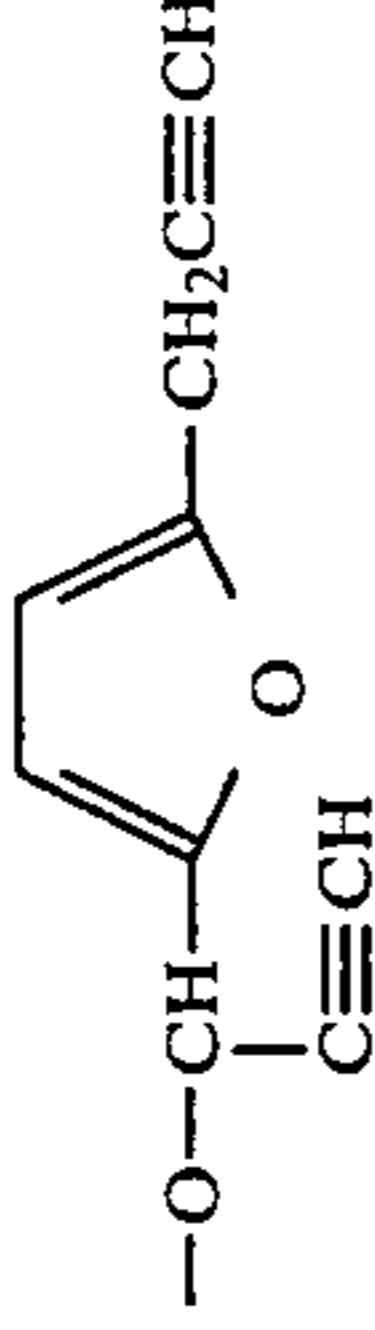
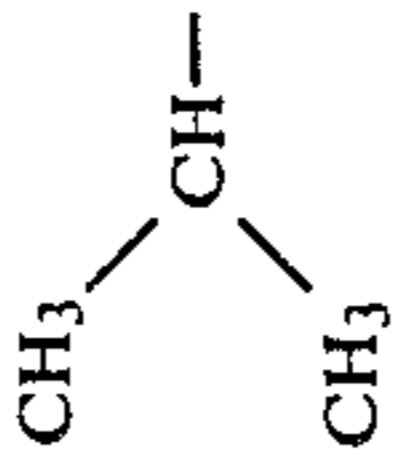
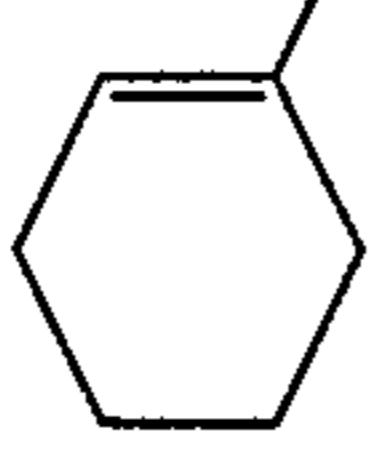
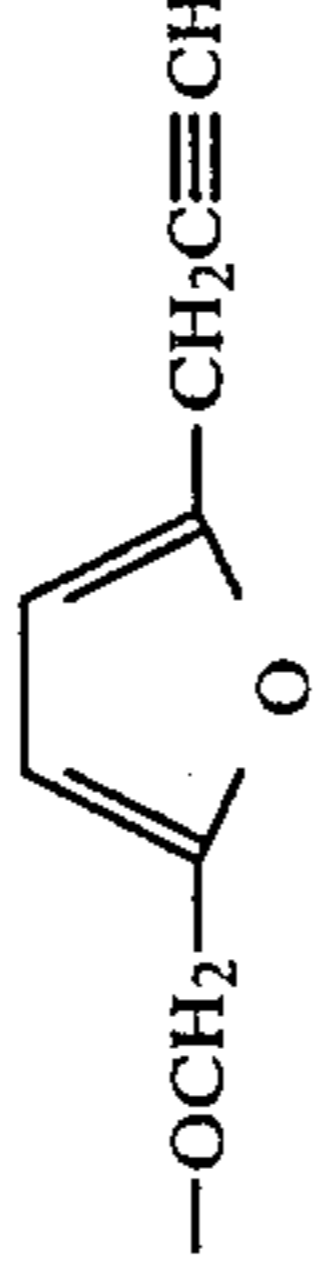
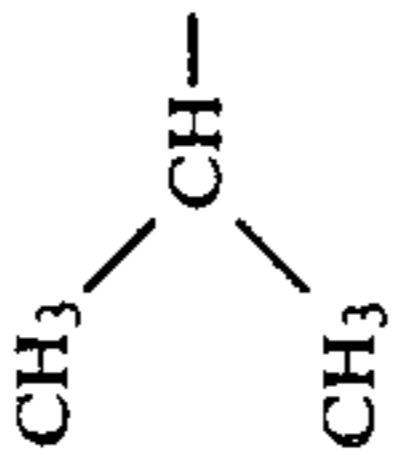
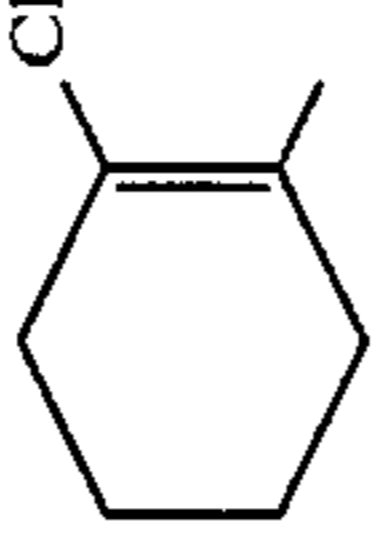
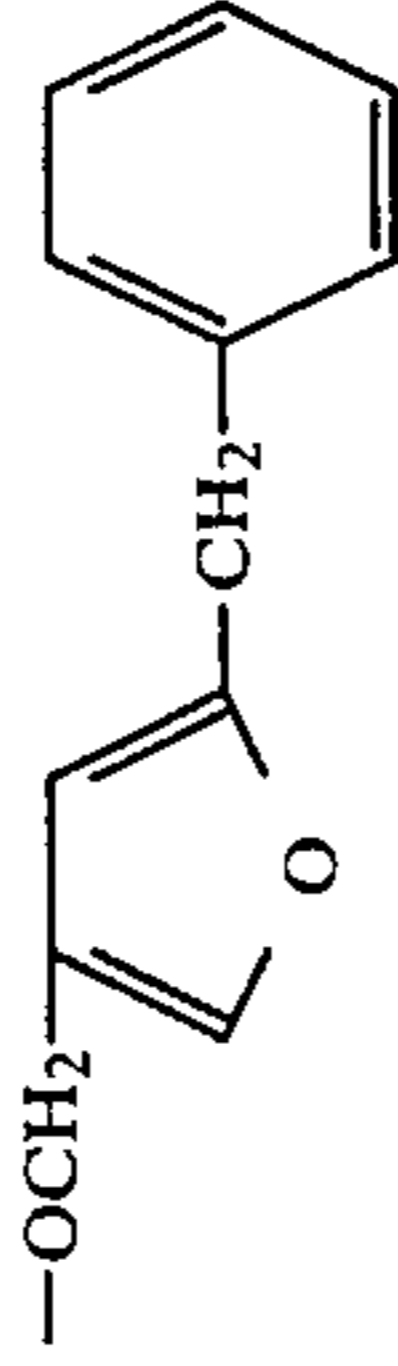
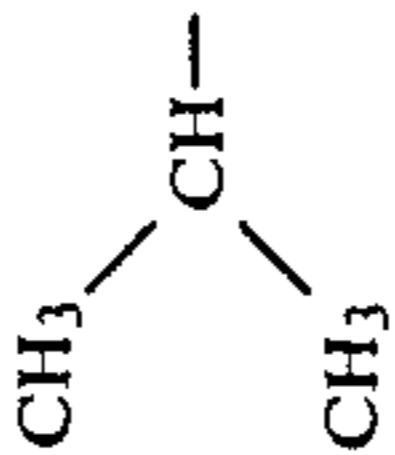
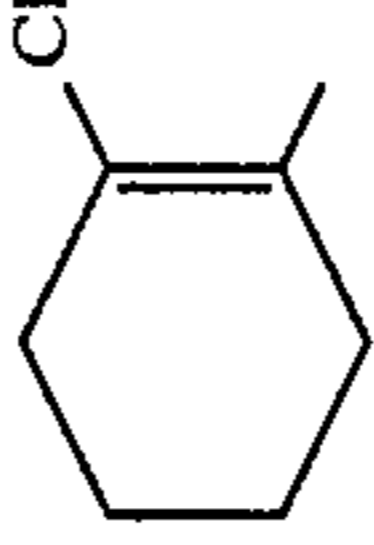
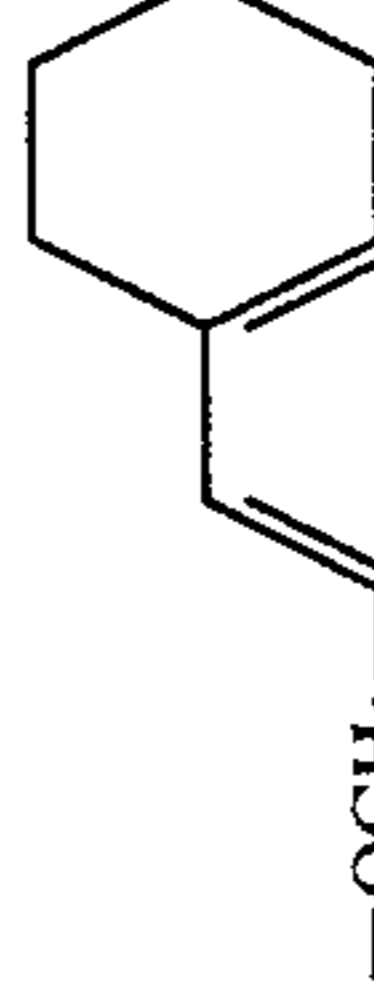
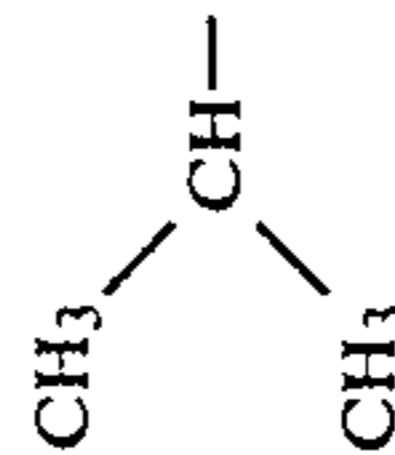
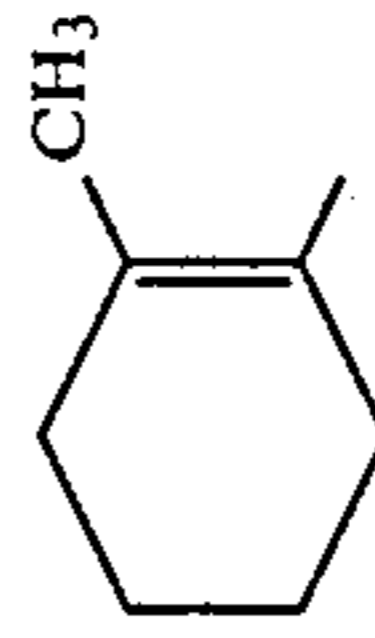
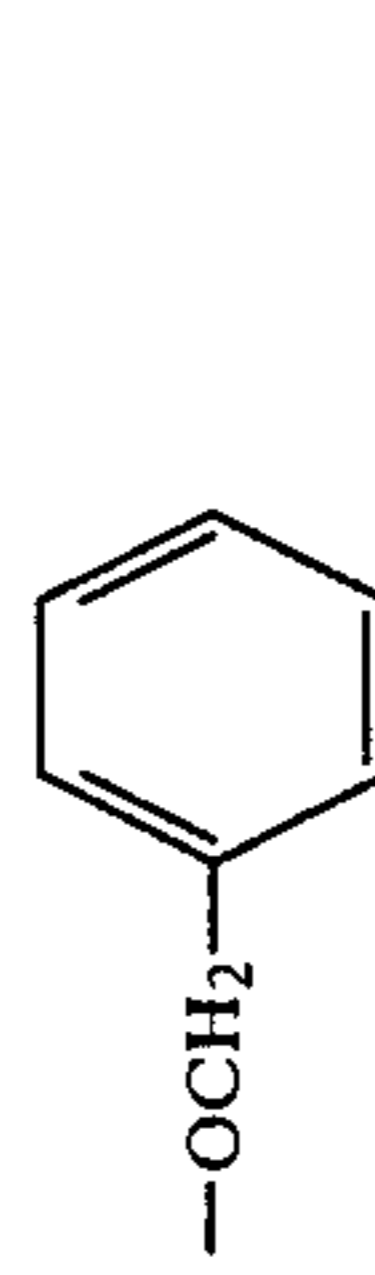
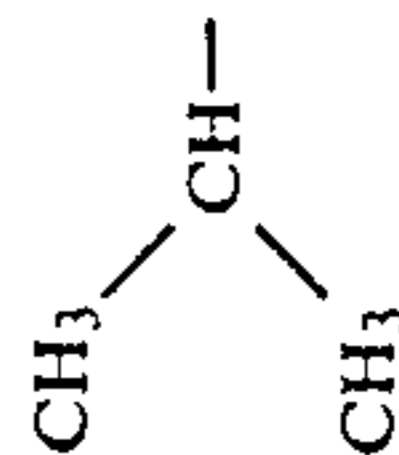
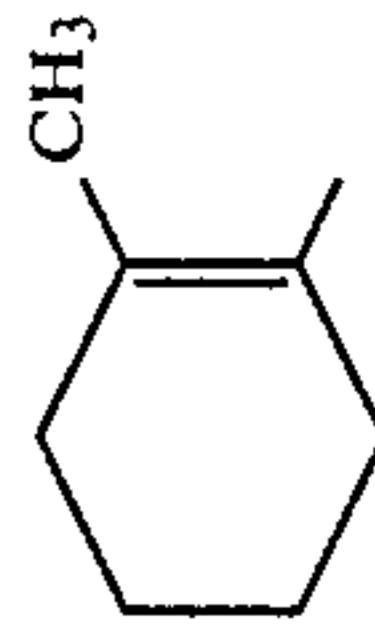
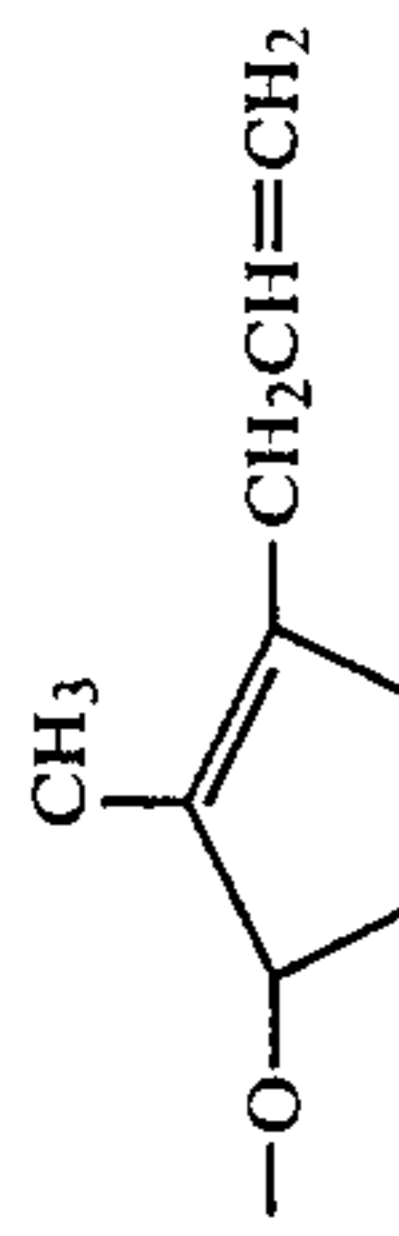
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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
182	253				a	90	C ₁₇ H ₁₇ O ₂ Cl ₅ C: 47.42 H: 3.98 Cl: 41.17	n_D^{28} 1.5395 47.88 3.70 41.53
183	254				a	85	C ₂₀ H ₂₂ O ₃ C: 77.37 H: 7.14	$n_D^{24.5}$ 1.5088 77.35 7.17
184	255				a	90	C ₂₁ H ₂₈ O ₂ C: 80.73 H: 9.03	n_D^{25} 1.5132 80.70 8.98
185	256				a	89	C ₂₀ H ₂₅ O ₂ Cl C: 72.16 H: 7.59 Cl: 10.65	n_D^{25} 1.5205 72.05 7.55 10.92
186	257				a	86	C ₂₀ H ₂₈ O ₃ C: 75.91 H: 8.92	$n_D^{24.5}$ 1.5085 75.86 8.90
187	258				a	90	C ₂₃ H ₂₈ O ₂ S C: 74.96 H: 7.66 S: 8.70	n_D^{23} 1.5370 74.78 7.62 8.58
188	259				a	89	C ₁₉ H ₂₆ O ₂ S C: 71.66 H: 8.23 S: 10.07	n_D^{23} 1.5352 71.69 8.24 9.80

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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
189	260				a	92	C ₁₉ H ₂₄ O ₂ S C: 72.11 H: 7.64 S: 10.13	n_D^{23} 1.5323 72.15 7.68 10.05
190	261				a	85	C ₂₅ H ₂₇ O ₃ N C: 77.09 H: 6.99 N: 3.60	n_D^{28} 1.5486 76.93 6.92 3.52
191	262				a	86	C ₂₃ H ₂₈ O ₂ C: 82.10 H: 8.39	n_D^{28} 1.5213 82.15 8.39
192	263				a	86	C ₂₁ H ₂₄ O ₃ C: 77.75 H: 7.46	n_D^{28} 1.5309 77.60 7.48
193	264				a	88	C ₂₂ H ₂₆ O ₃ C: 78.07 H: 7.74	n_D^{24} 1.5308 78.12 7.61
194	265				a	86	C ₂₄ H ₂₈ O ₃ C: 70.09 H: 7.74	n_D^{24} 1.5467 78.80 7.76
195	266				a	86	C ₁₉ H ₂₆ O ₃ C: 75.46 H: 8.67	n_D^{28} 1.5088 75.40 8.66

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Preparation Exp. No.	Comp. No.	Z	Y	X	Chemical Structure	Esterification Method	Yield (%)	Physical Properties	
								Calcd. (%)	Found (%)
196	267					a	90	C ₂₀ H ₂₆ O ₃ C: 76.40 H: 8.34	<i>n</i> _D ^{24.5} 1.5140 76.58 8.29
197	268					a	85	C ₂₁ H ₂₄ O ₃ C: 77.75 H: 7.46	<i>n</i> _D ^{24.5} 1.5133 77.81 7.40
198	269					a	87	C ₁₉ H ₂₄ O ₃ C: 75.97 H: 8.05	<i>n</i> _D ^{24.5} 1.5064 75.86 8.04
199	270					a	92	C ₂₂ H ₂₅ O ₃ Cl C: 70.86 H: 6.76 Cl: 9.51	<i>n</i> _D ^{24.5} 1.5358 70.66 6.73 9.30
200	271					a	88	C ₁₉ H ₂₅ O ₂ ClS C: 64.66 H: 7.14 Cl: 10.05 S: 9.09	<i>n</i> _D ²³ 1.5430 64.74 7.15 9.77 9.13
201	272					a	90	C ₂₅ H ₃₀ O ₃ C: 79.33 H: 7.99	<i>n</i> _D ²³ 1.5474 79.25 7.90
202	273					a	87	C ₂₁ H ₃₀ O ₃ C: 76.32 H: 9.15	<i>n</i> _D ²⁸ 1.5030 76.55 9.13

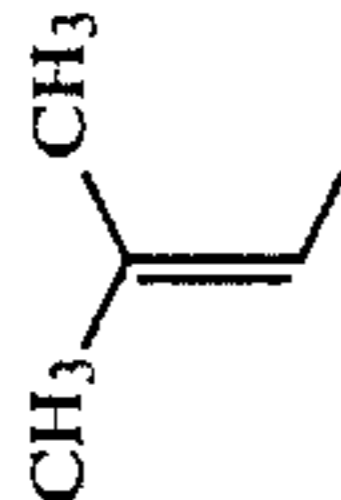
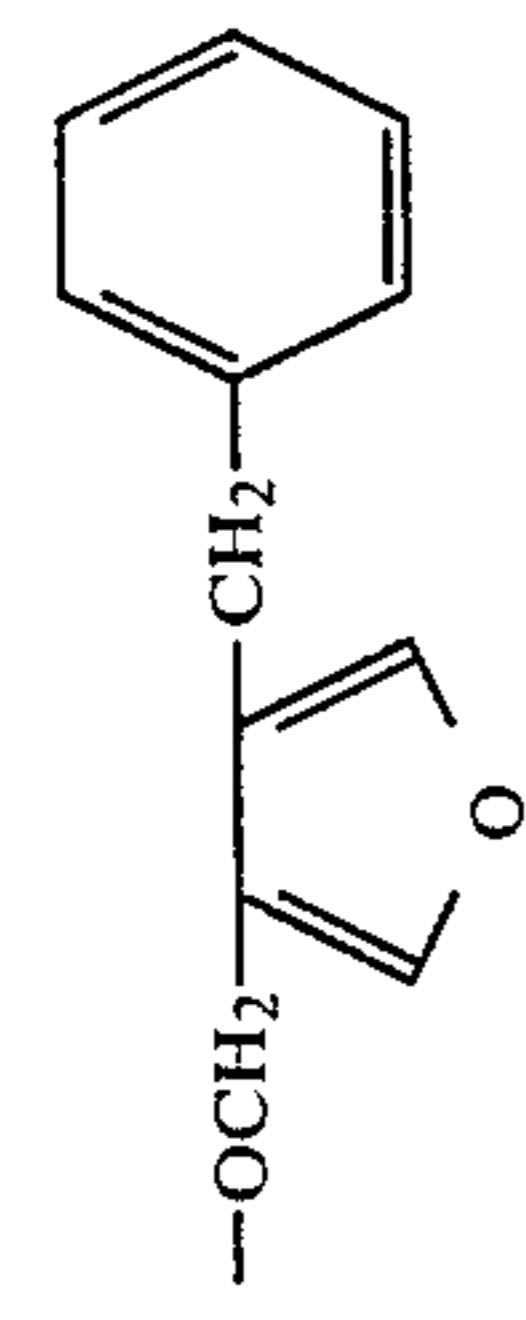
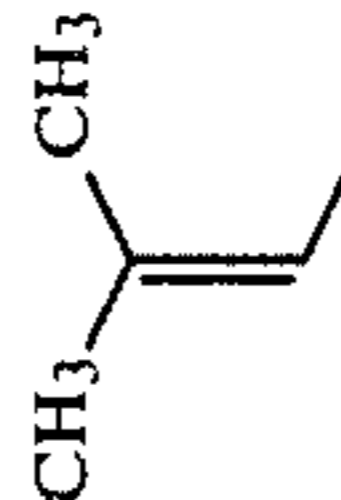
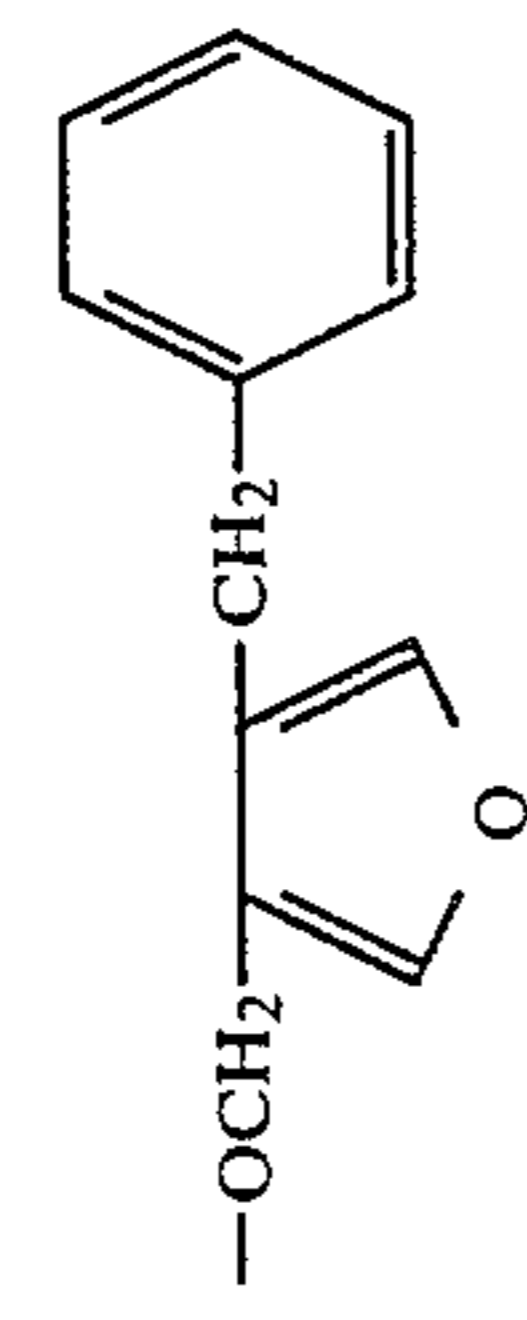
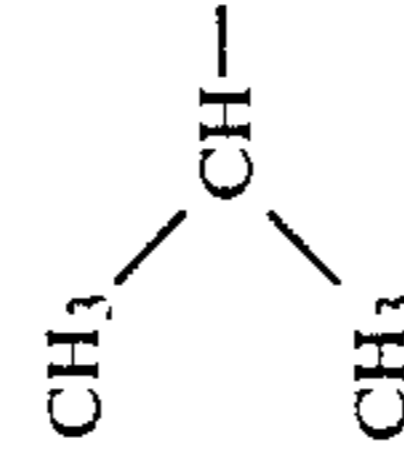
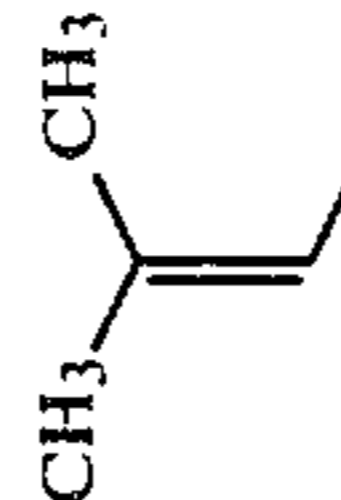
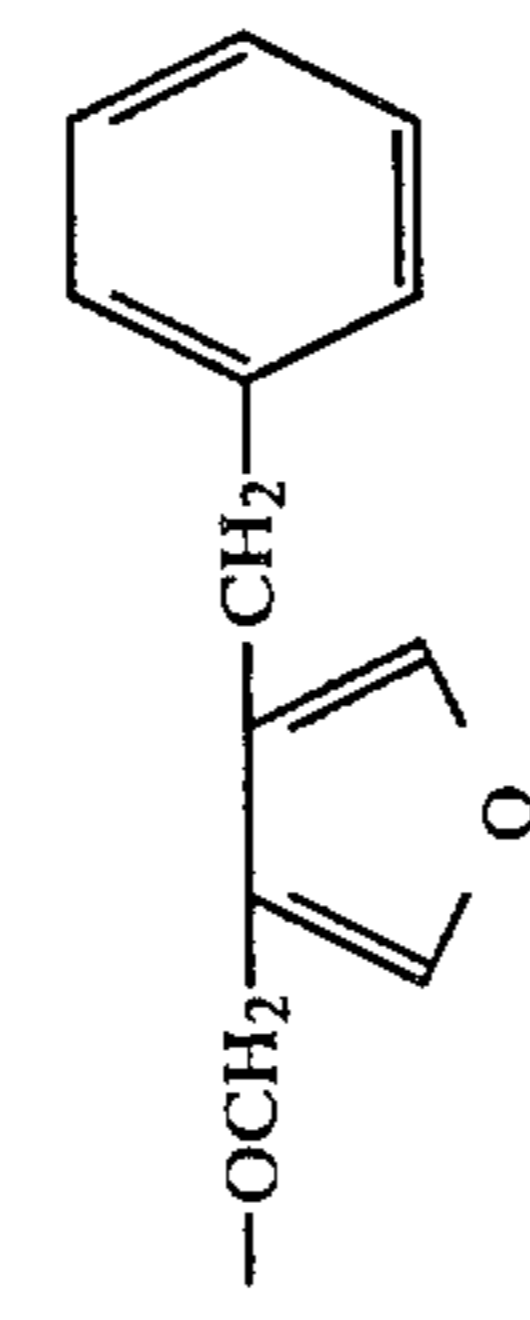
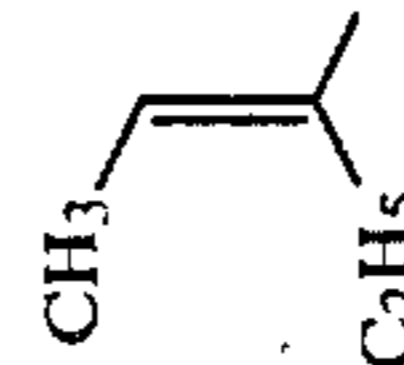
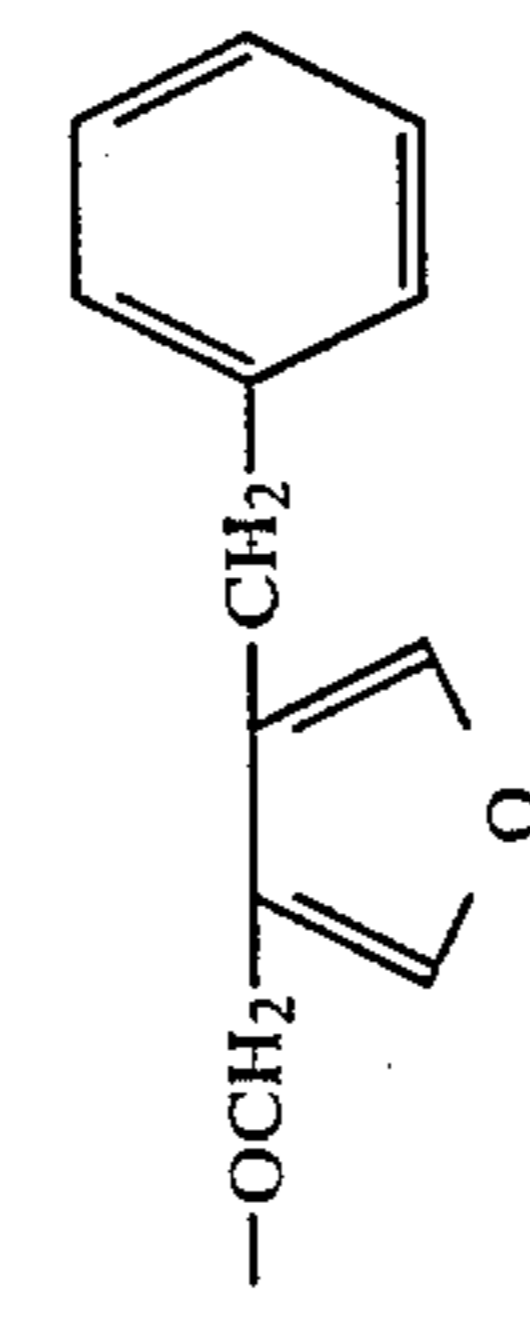
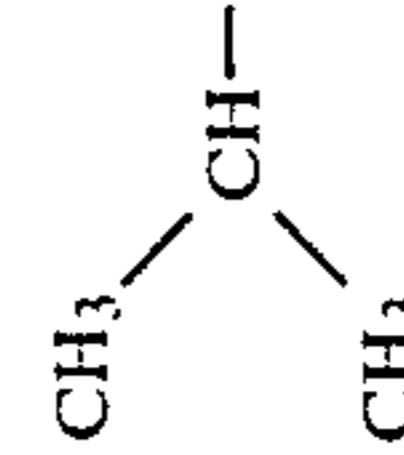
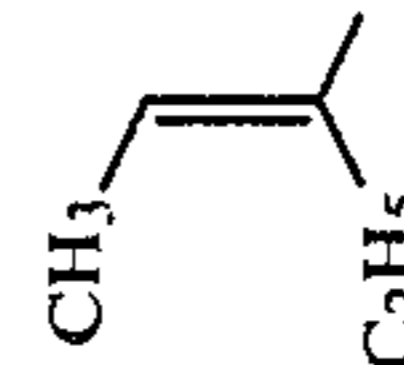
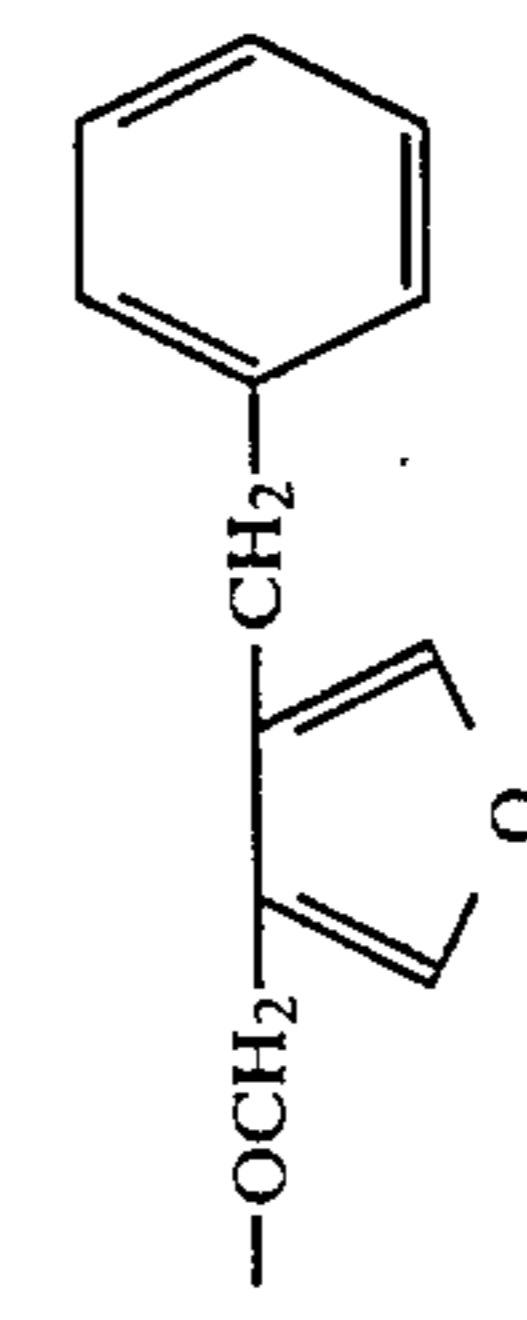
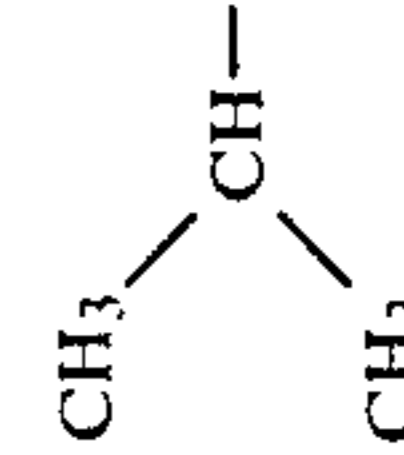
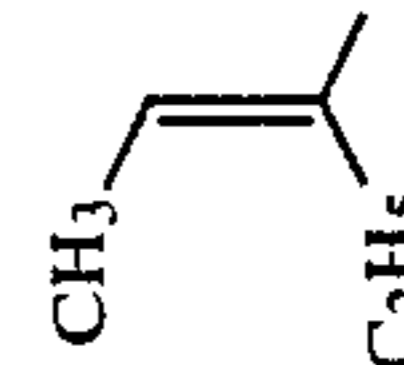
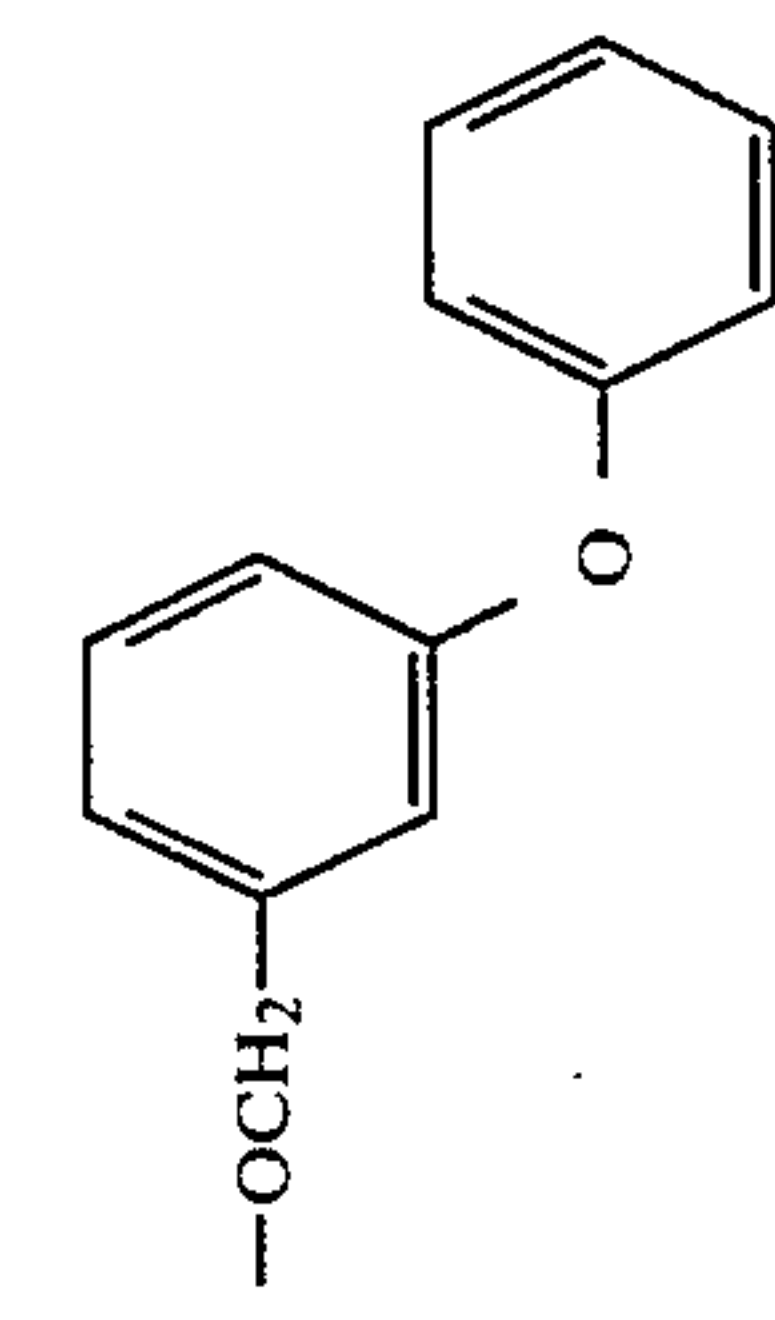
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Exp. No.	Preparation Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
203	274				a	87	C ₂₅ H ₃₀ O ₃ C: 79.33 H: 7.99	n_D^{28} 1.5243 79.47 7.95
204	275				a	89	C ₂₄ H ₂₈ O ₃ C: 79.09 H: 7.74	n_D^{24} 1.5252 79.05 7.73
205	276				a	86	C ₂₁ H ₂₄ O ₃ C: 77.75 H: 7.46	n_D^{24} 1.5097 77.62 7.49
206	277				a	89	C ₂₅ H ₂₇ O ₃ N C: 77.09 H: 6.99 N: 3.60	n_D^{24} 1.5470 77.11 6.80 3.53
207	278				a	89	C ₁₉ H ₂₀ O ₄ S C: 66.26 H: 5.85 S: 9.31	n_D^{23} 1.5335 66.07 5.88 9.24
208	279				a	83	C ₂₃ H ₂₆ O ₄ C: 75.38 H: 7.15	n_D^{23} 1.5088 75.15 7.02

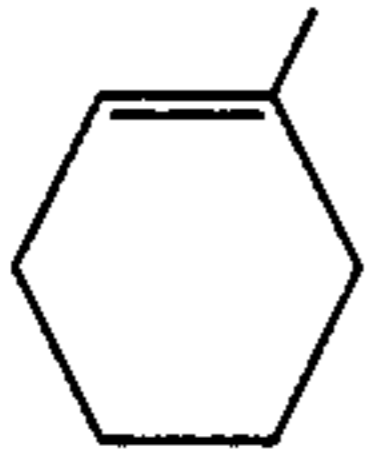
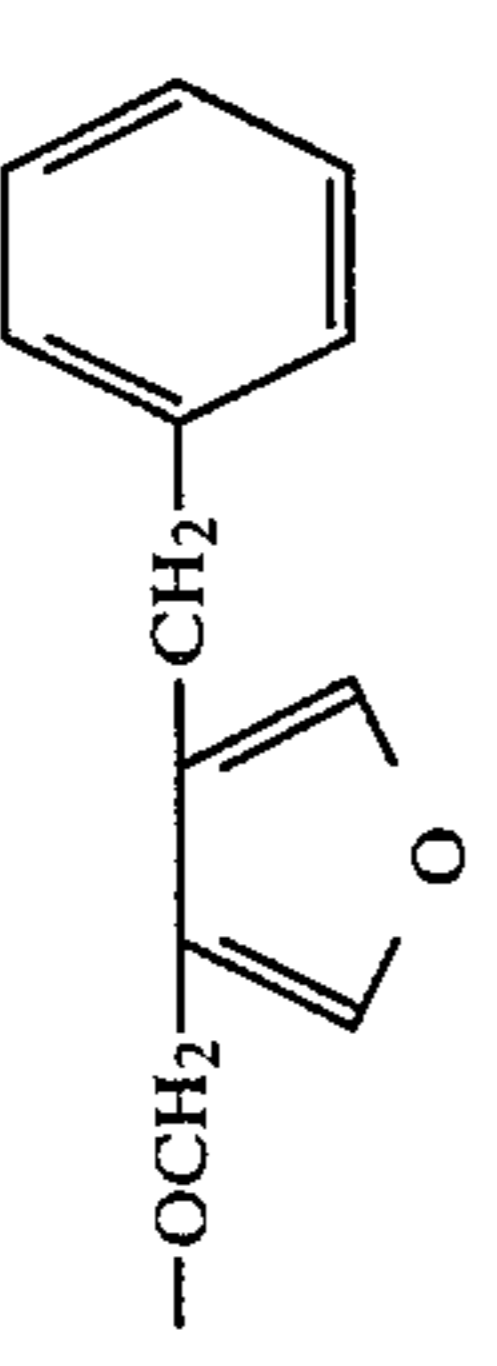
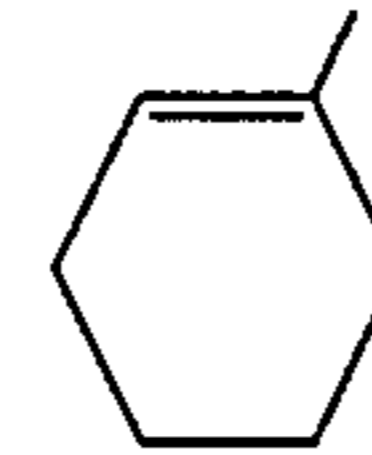
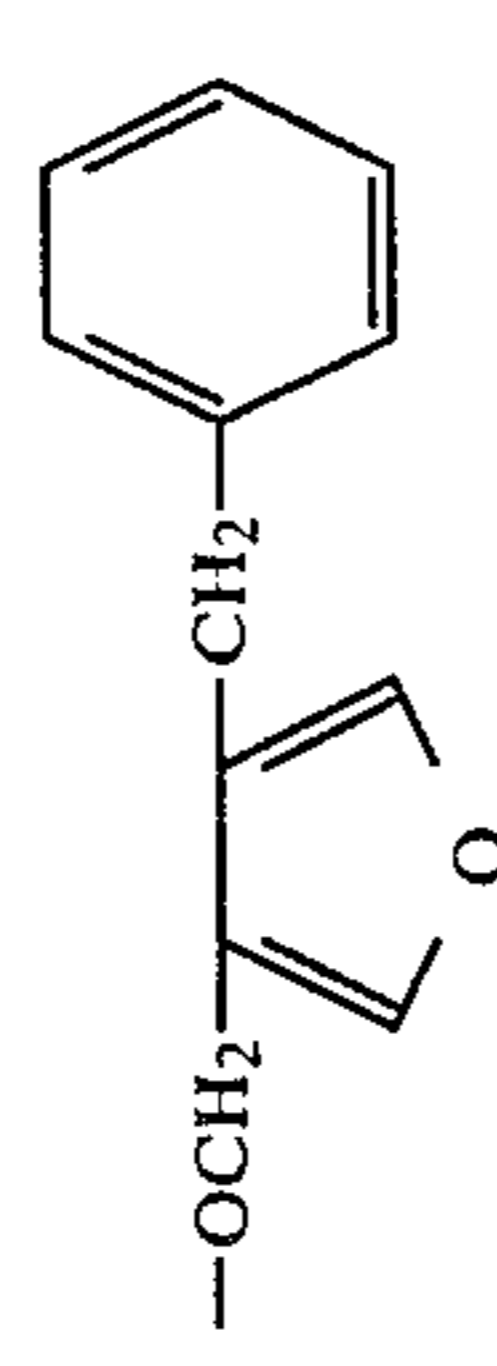
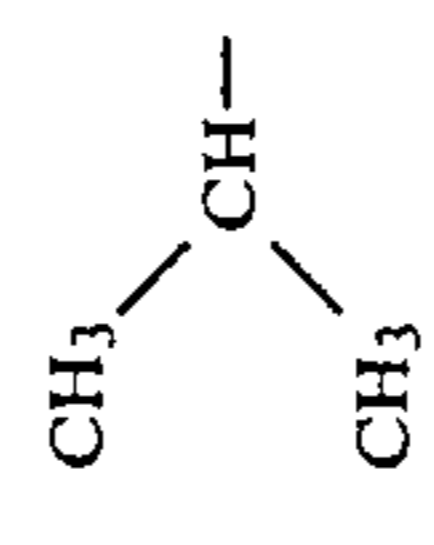
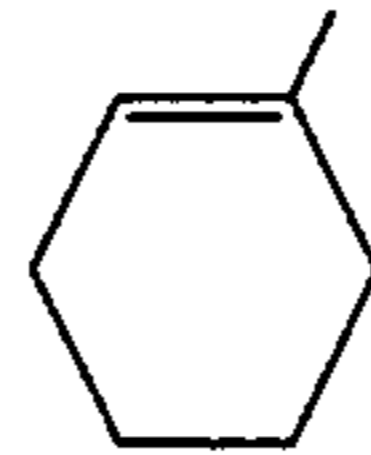
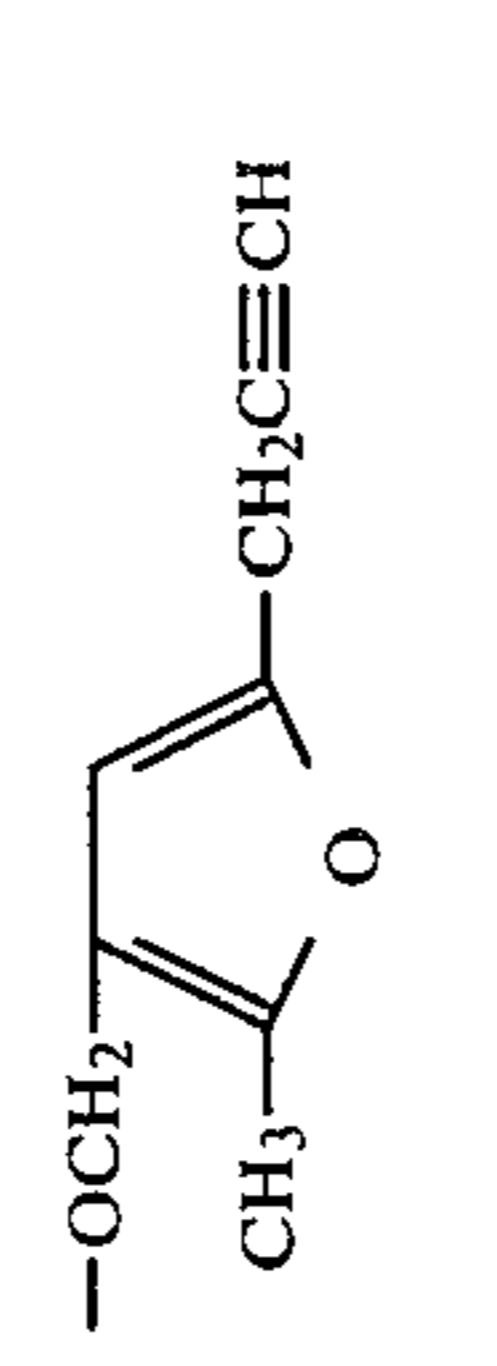
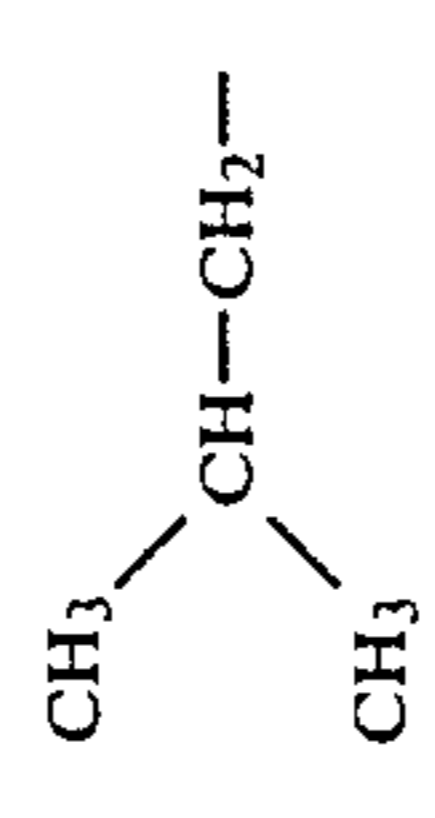
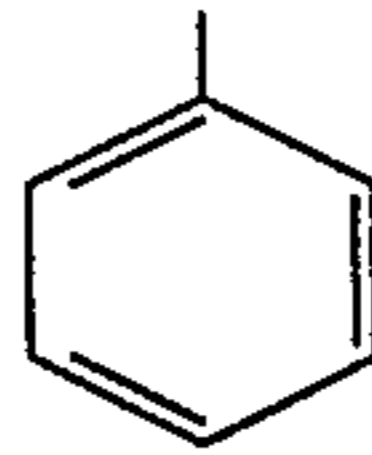
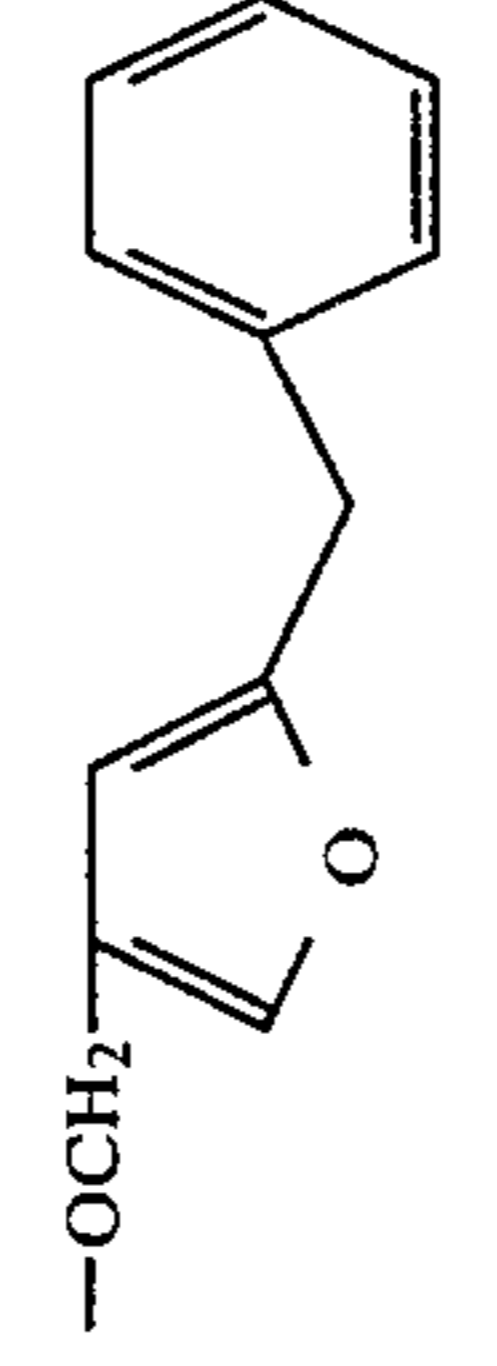
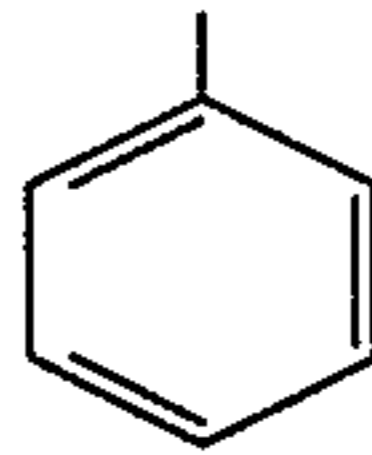
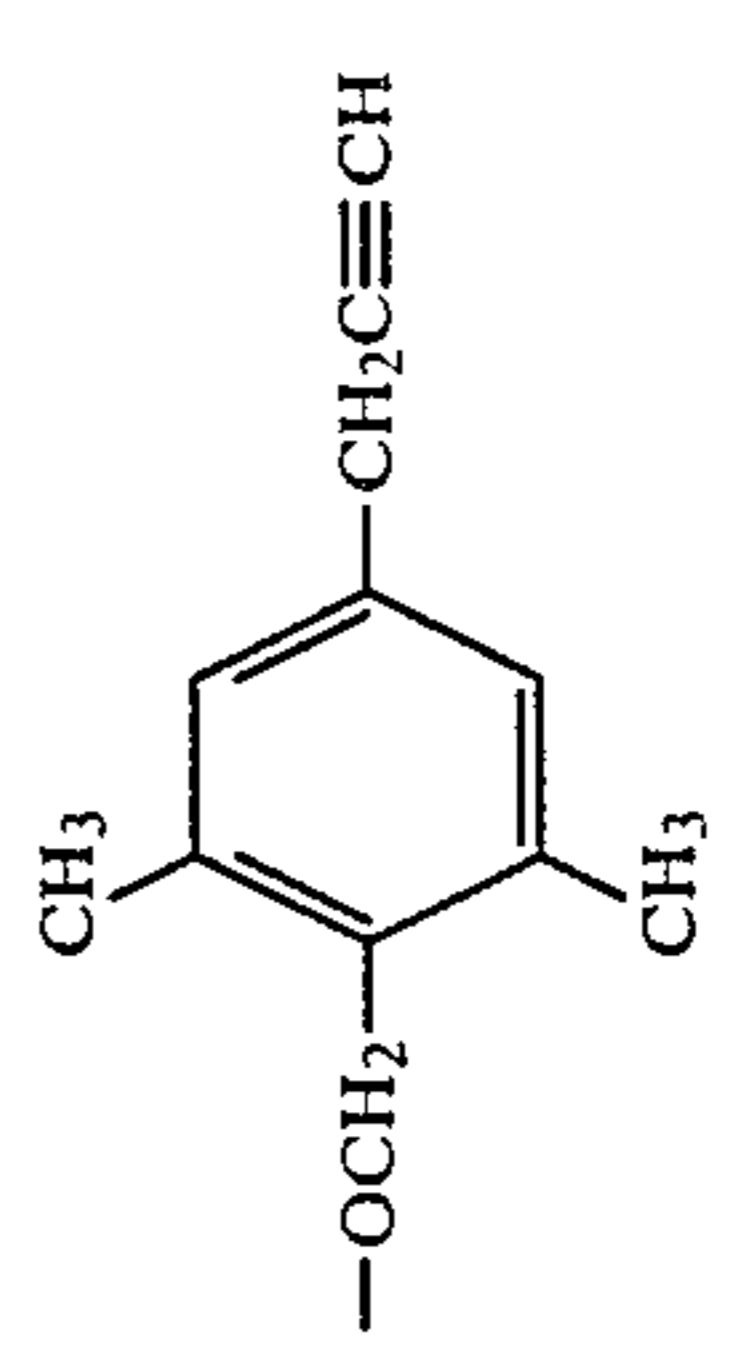
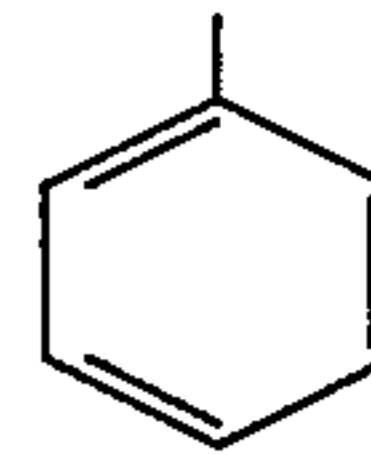
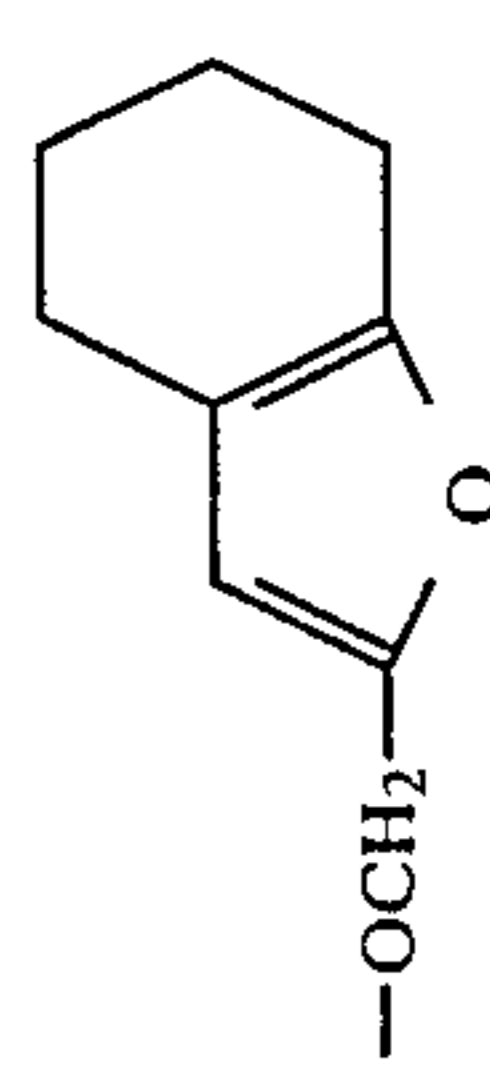
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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification		Physical Properties	
					Method	Yield (%)	Calcd. (%)	Found (%)
209	280				a	87	C ₂₃ H ₂₆ O ₄ C: 75.38 H: 7.15	n_D^{23} 1.5075 75.36 7.18
210	281	C ₂ H ₅ -			a	86	C ₂₀ H ₂₄ O ₃ C: 76.89 H: 7.74	$n_D^{28.5}$ 1.5178 76.54 7.70
211	282				a	86	C ₂₁ H ₂₆ O ₃ C: 77.27 H: 8.03	n_D^{25} 1.5185 77.34 8.06
212	283				a	90	C ₂₂ H ₂₆ O ₃ C: 78.07 H: 7.74	n_D^{25} 1.5413 77.95 7.90
213	284				a	89	C ₁₇ H ₂₂ O ₃ C: 74.42 H: 8.08	n_D^{25} 1.4960 74.84 8.15
214	285				a	85	C ₂₃ H ₂₅ O ₃ N C: 76.01 H: 6.93 N: 3.85	n_D^{25} 1.5407 76.10 6.88 3.62

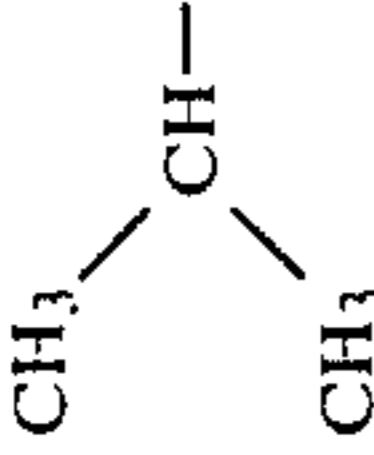
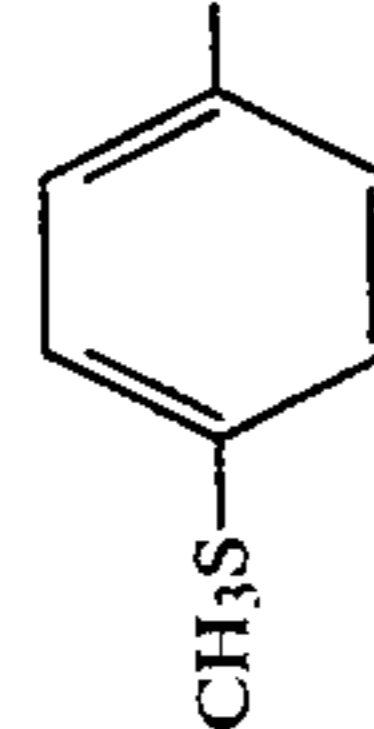
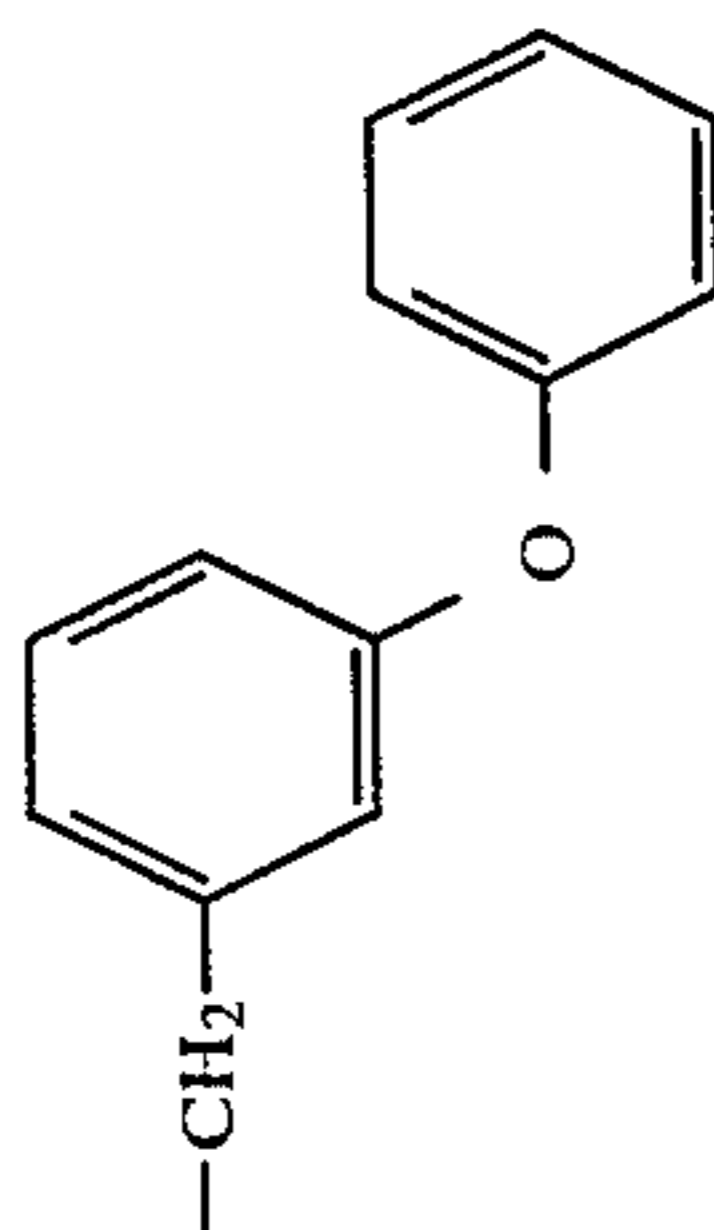
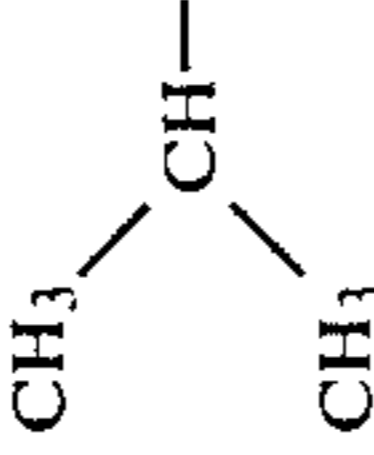
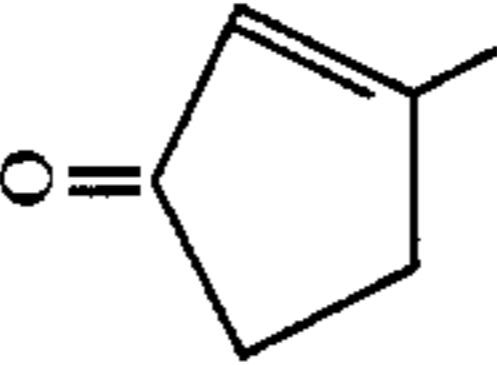
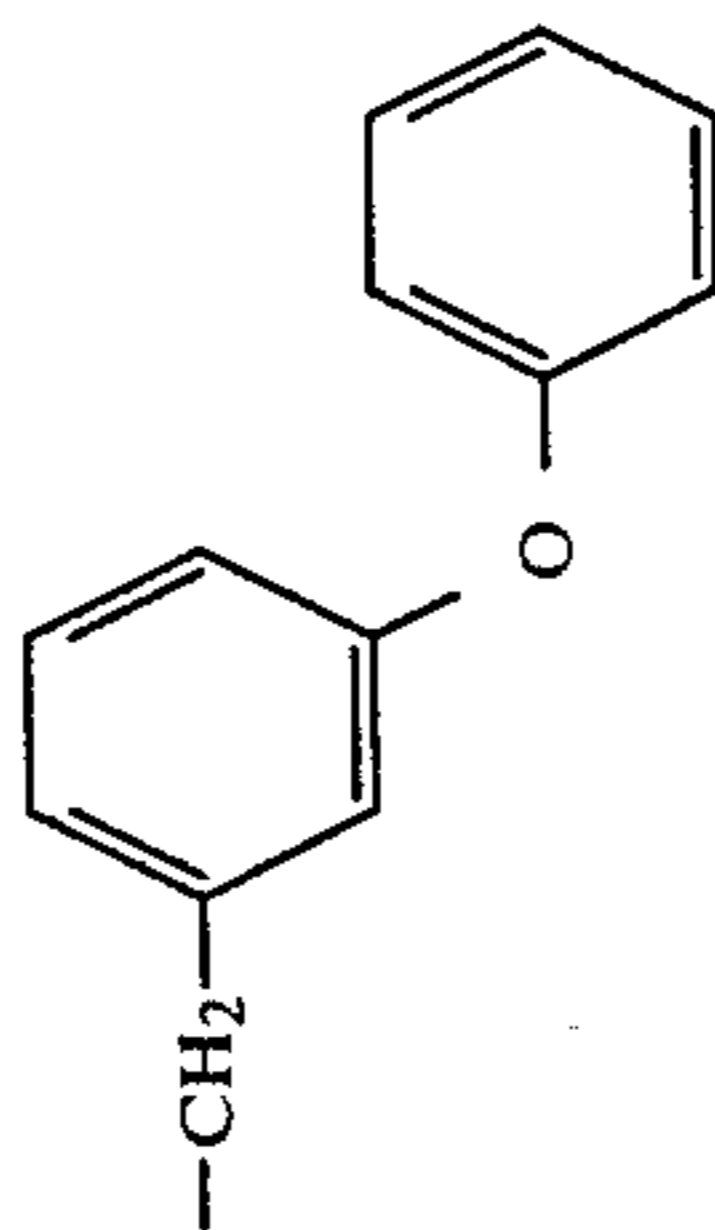
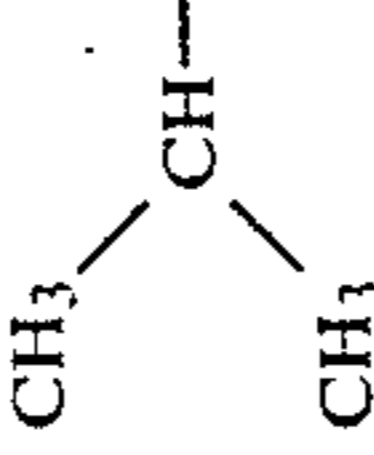
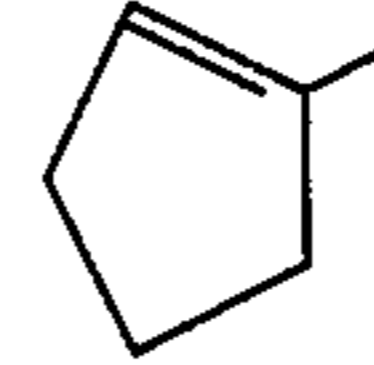
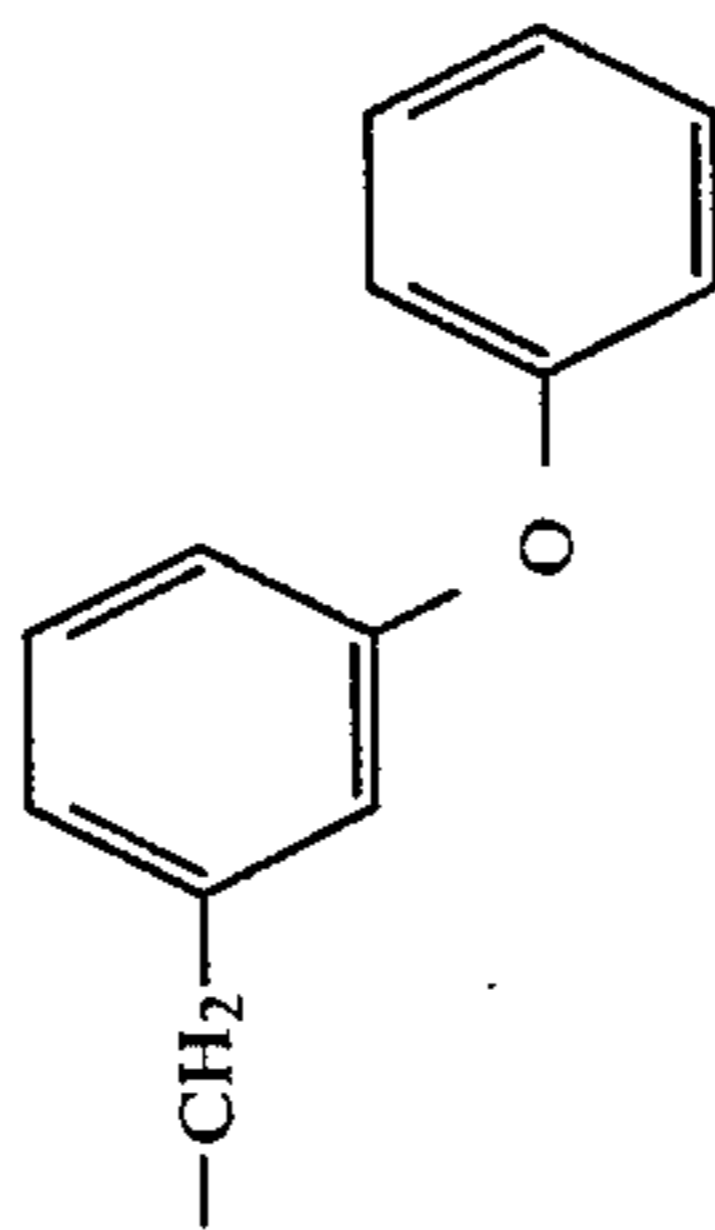
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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification		Physical Properties	
					Method	Yield (%)	Calcd. (%)	Found (%)
215	286	CH ₃ —			a	91	C ₁₉ H ₂₂ O ₃ C: 76.48 H: 7.43	$n_D^{28.5}$ 1.5202 76.41 7.55
216	287	C ₂ H ₅ —			a	87	C ₂₀ H ₂₄ O ₃ C: 76.89 H: 7.74	$n_D^{28.5}$ 1.5165 76.71 7.85
217	288				a	88	C ₂₁ H ₂₆ O ₃ C: 77.27 H: 8.03	n_D^{23} 1.5153 77.25 8.09
218	289	C ₂ H ₅ —			a	88	C ₂₁ H ₂₆ O ₃ C: 77.27 H: 8.03	n_D^{27} 1.5150 77.45 8.01
219	290				a	92	C ₂₂ H ₂₈ O ₃ C: 77.61 H: 8.29	n_D^{27} 1.5134 77.76 8.25
220	291				a	89	C ₂₃ H ₂₈ O ₃ C: 78.37 H: 8.01	n_D^{27} 1.5363 78.43 8.08

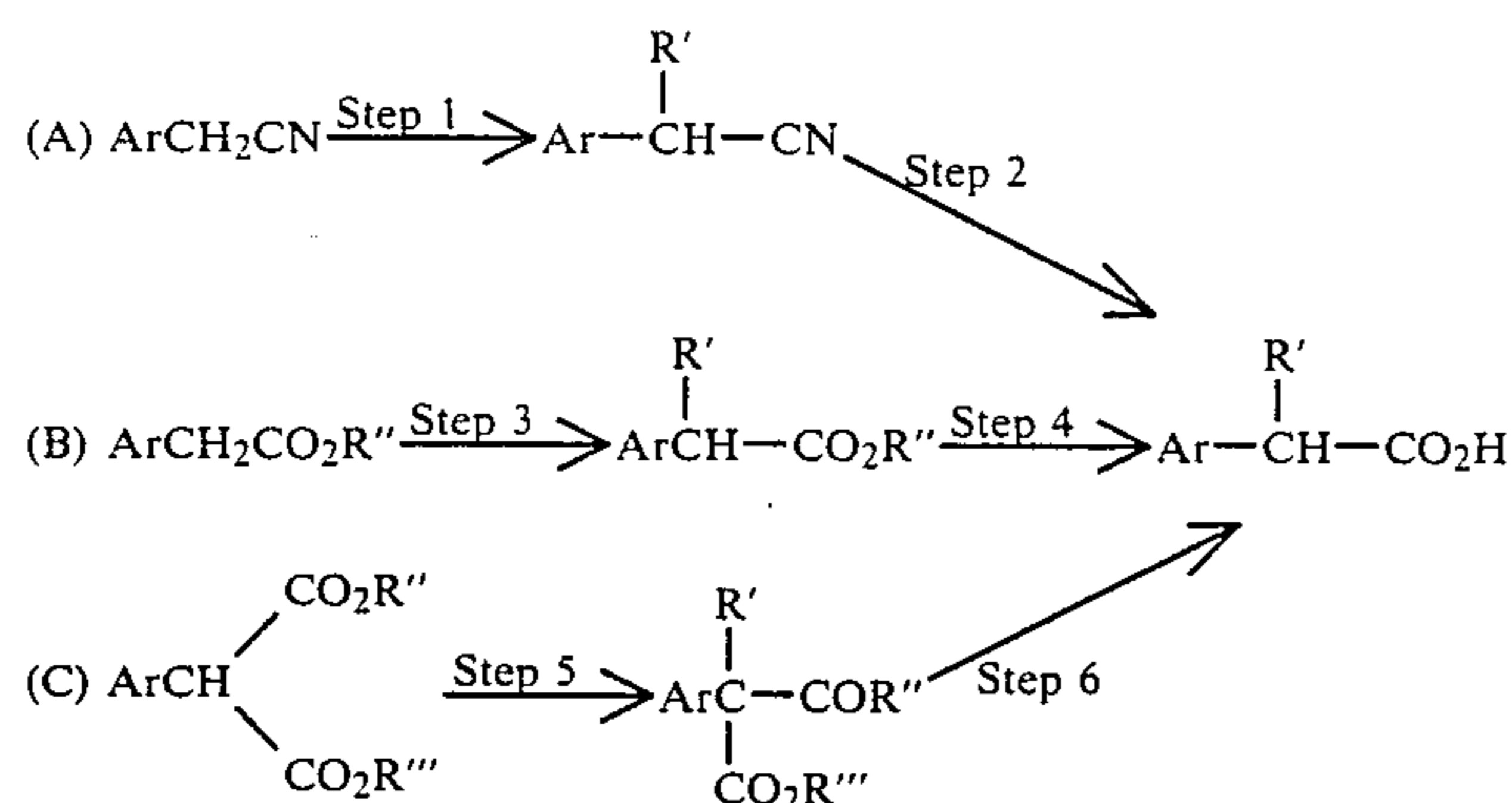
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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification		Yield		Physical Properties	
					Method	(%)	Calcd. (%)	Found (%)		
221	292	$\text{CH}_2=\text{CHCH}_2-$			a	89	$\text{C}_{23}\text{H}_{26}\text{O}_3$ C: 78.83 H: 7.48	n_D^{25} 1.5345 78.85 7.48		
222	293	$\text{CH}\equiv\text{CCH}_2-$			a	90	$\text{C}_{23}\text{H}_{24}\text{O}_3$ C: 79.28 H: 6.94	n_D^{25} 1.5363 79.20 7.02		
223	294				a	86	$\text{C}_{20}\text{H}_{26}\text{O}_3$ C: 76.40 H: 8.34	n_D^{28} 1.4995 76.49 8.55		
224	295				a	82	$\text{C}_{24}\text{H}_{26}\text{O}_3$ C: 79.53 H: 7.23	n_D^{22} 1.5356 78.91 7.26		
225	296	CH_3CH_2-			a	89	$\text{C}_{22}\text{H}_{24}\text{O}_2$ C: 82.46 H: 7.55	n_D^{19} 1.5238 82.41 7.53		
226	297	CH_3CH_2-			a	78	$\text{C}_{19}\text{H}_{22}\text{O}_3$ C: 76.48 H: 7.45	n_D^{20} 1.5542 76.55 7.37		

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Preparation Exp. No.	Comp. No.	Z	Y	X	Esterification Method	Yield (%)	Physical Properties	
							Calcd. (%)	Found (%)
227	298				a	84	C ₂₅ H ₂₆ O ₃ S C: 73.86 H: 6.45 S: 7.89	n_D^{20} 1.5387 73.92 6.40 7.81
228	299				a	87	C ₂₃ H ₂₄ O ₄ C: 75.80 H: 6.64	n_D^{20} 1.5366 75.77 6.72
229	300				a	91	C ₂₃ H ₂₆ O ₃ C: 78.82 H: 7.48	n_D^{20} 1.5412 78.79 7.55

Most of the important intermediates of the present invention, i.e., disubstituted acetic acid, can effectively be prepared in high yields via one of the alternative procedures (A), (B) and (C) given below.



wherein Ar represents an aryl group, R' represents an alkyl, alkenyl or alkynyl group, R'' represents hydrogen or a lower alkyl group, and R''' represents a lower alkyl group.

Step 1:

This procedure can be accomplished by alkylating the corresponding arylacetonitrile with an appropriate halide or sulfoxylate of the formula R'-X in an inert solvent (for example, ethers, tetrahydrofuran, benzene, toluene, etc. including liquid ammonia when NaNH₂ and the like is used as a base described below) in the presence of a base such as alkali metals, alkali metal hydrides, alkali metal amides and the like at room temperature or an elevated temperature.

Step 2:

This procedure can be accomplished under the well-known conditions commonly used for hydrolysis of a nitrile group, i.e., by heating the nitrile with a mineral acid or an alkali hydroxide.

Steps 3 and 4:

This procedure can be accomplished by alkylating the corresponding phenylacetic acid or a lower-alkyl ester thereof with an appropriate halide or sulfoxylate of the formula R'-X in an inert solvent in the presence of a base such as alkali metals or hydrides thereof, n-butyl lithium and the like at room temperature or an elevated temperature, and, when the lower-alkyl ester is used for the alkylation, hydrolyzing the resulting ester with an appropriate acid or alkali in the usual manner.

Step 5:

This procedure can be accomplished by reacting an arylmalonic acid ester (which is easily obtainable by carbalkoxylation of the corresponding arylacetic ester) with an appropriate halide or sulfoxylate of the formula R'-X in the presence of an alkali metal alkoxide or an alkali metal hydride as a base in an inert solvent or an alcohol corresponding to the alkali metal alkoxide.

Step 6:

This procedure can be accomplished under the well-known conditions commonly used for hydrolysis of a usual ester, i.e., by hydrolyzing the ester with an acid or alkali to obtain the corresponding dicarboxylic acid followed by heat-decarbonization to obtain the desired carboxylic acid, or can be accomplished by heating the diester with an alkali metal alkoxide in an alcohol whereby the diester is easily converted into a monoester by decarboxylation and hydrolyzing the monoester by the well-known procedure to obtain the corresponding carboxylic acid in high yields.

The preparation of the intermediates, carboxylic acids, can be illustrated by the following Synthesis Examples, but they are not to be construed as limiting the processes or intermediates of the present invention.

SYNTHESIS EXAMPLE 1

α -Isopropyl-4-methoxyphenylacetonitrile

14.72 g (0.1 mole) of 4-methoxyphenylacetonitrile in 10 ml of dry toluene was added to a stirred solution of sodium hydride (2.64 g, 0.11 mole) and isopropylbromide (18.45 g, 0.15 mole) in 100 ml of dry toluene and 10 ml of dry dimethylformamide which was heated at 70° C., in 30 minutes. After addition was completed, the mixture was heated at a temperature of from 80° to 85° C. for 3 hours, and cooled to room temperature then poured into 200 g of ice-water.

Organic layer was separated and the aqueous layer was extracted with ether twice. Combined organic layer was washed successively with water and saturated sodium chloride solution, and dried over anhydrous sodium sulfate. The solvent was removed in vacuo and the residual oil was distilled in vacuo to afford 15.70 g (b.p. 95°-96°/0.15 mmHg, 83.0%) of α -isopropyl-4-methoxyphenylacetonitrile.

In similar manners, following α -isopropylphenylacetonitriles were obtained.

	b.p.°C./mmHg	Yield (%)
4-bromo- α -isopropylphenyl-acetonitrile	(1) 90-93°/0.10	82.1
4-ethyl-d-isopropylphenyl-acetonitrile	(2) 96-93°/0.10	85.1
4-isopropyl- α -isopropylphenyl-acetonitrile	(3) 100-102°/0.40	83.5
3-chloro- α -isopropylphenyl-acetonitrile	(4) 104-106°/0.30	80.2
4-fluoro- α -isopropylphenyl-acetonitrile	(5) 89-90°/5.0	80.7
3-fluoro- α -isopropylphenyl-acetonitrile	(6) 89-90°/5.0	81.1
4-thiomethyl- α -isopropylphenyl-acetonitrile	(7) 105°/0.10	75.1
2,4,6-trimethyl- α -isopropylphenyl-acetonitrile	(8) 87-88°/0.20	53.1
3-trifluoromethyl- α -isopropylphenyl-acetonitrile	(9) 95-96°/5.0	87.9
3,4-methylenedioxy- α -isopropylphenyl-acetonitrile	(10) 95-100°/0.2	78.2
3-phenoxy- α -isopropylphenyl-acetonitrile	(11) 147-148°/0.3	85.2

(1) Starting 4bromophenylacetonitrile was synthesized from 4-bromotoluene which was brominated at 180° C. with bromine and converted into nitrile with potassium cyanide in DMSO-H₂O.

(2) Ethylbenzene was chloromethylated with Kosolopoff method (JACS, 68, 1670 (1946)) and the chloride was purified by fractional distillation, then converted into nitrile in similar manner described above.

(3) Isopropylbenzene was chloromethylated and converted into a nitrile.

(4) Commercially available 3-chlorobenzylchloride was converted into a nitrile similarly.

(5, 6) 3-fluorotoluene and 4-fluorotoluene were brominated with NBS in carbon tetrachloride then converted into nitriles.

(7) Thioanisole was chloromethylated with para-formaldehyde and zinc chloride in poor conversion, then converted into nitrile.

(8) Mesitylene was chloromethylated and converted into a nitrile. In this case, alkylation proceeded very slowly even though an elevated temperature (110° C.) and longer reaction times (15 hours) were tried.

(9) 3-bromobenzotrifluoride was converted to 3-hydroxymethylbenzotrifluoride via Grignard method (R. Filler and H. Novar, J.O.C., 25, 733 (1960)) and converted into a bromide (HBr—H₂SO₄) and a nitrile (KON-DMSO), successively.

(10) Piperonyl alcohol was converted into a chloride with thionyl chloride at 0° C., followed by KON/DMSO to afford the nitrile.

(11) m-phenoxytoluene which was obtained from m-cresol and bromotoluene was brominated at 230° C. with bromine, and the corresponding bromide was converted into a nitrile similarly.

SYNTHESIS EXAMPLE 2

4-Bromo- α -isopropyl-phenylacetic acid

20 g of 4-bromo- α -isopropylphenylacetone nitrile was heated with 140 ml of sulfuric acid (50 V/V %) at 145° C. for 6 hours. The cooled reaction mixture was poured onto 150 g of ice, and extracted with ether (100 ml \times 3). The ether layers were combined and extracted with three 70 ml portions of cold 5% sodium hydroxide solution, and the aqueous extracts were acidified with concentrated hydrochloric acid, and then extracted with ether. The ether extract was washed successively with water, saturated sodium chloride solution, dried over anhydrous sodium sulfate. Removal of the solvent gave 16.9 g (78.2%) of white crystals, which were sufficiently pure for use in next step. A portion of the crystals was recrystallized from benzene, melting point: 97°–98° C.

SYNTHESIS EXAMPLE 3

4-Methoxy- α -isopropylphenylacetic acid

12.8 g of 4-methoxy- α -isopropylphenylacetone nitrile was heated under reflux (140° C.) with 200 ml of ethylene-glycol and 40 g of 50% aqueous potassium hydroxide solution for 8 hours. The cooled reaction mixture was poured into 200 g of ice-water, which was washed with ether. The aqueous layer was acidified with concentrated hydrochloric acid, and extracted with ether (150 ml \times 3). The combined ether extract was washed successively with water, saturated sodium chloride, then dried over anhydrous sodium sulfate. Removal of the solvent gave 11.30 g (80.4%) of white crystals. The acid recrystallized from benzene melted at 146°–148° C.

In similar manners following acids were prepared.

	Method of Hydrolysis	Yields	N.M.R. signals (CHCl ₃ , in ppm)
5	A	25.0	0.74 (3H, d.J=7Hz), 1.07 (3H, d.J=7Hz), 1.9–2.6 (1H, m.), 3.22 (1H, d.J=10Hz), 7.4–7.7 (4H, m.), 10.6 (1H, b.S)
10			
15	A	85.2	0.69 (3H, d.J=7Hz), 1.05 (3H, d.J=7Hz), 1.9–2.6 (1H, m.) 3.10 (1H, d.J=12Hz), 6.9–7.4 (4H, m.), 9.70 (1H, b.S.)
20			
25	B	78.2	0.69 (3H, d.J=6.5Hz), 1.07 (3H, d.J=6.5Hz), 2.42 (3H, S), 3.07 (1H, d.J=10Hz), 7.19 (4H, S.), 9.64 (1H, b.S.)
30			
35	A	34.5	0.61 (3H, d.J=7Hz), 1.15 (3H, d.J=7Hz), 1.22 (3H, S.), 1.32 (6H, S.), 2.25–2.95 (1H, m.), 3.68 (1H, d.J=10Hz), 6.82 (2H, b.S.), 9.50 (1H, b.S.)
40			
45	B	87.2	0.65 (3H, S.J.=7Hz), 1.05 (3H, S.J.=7Hz), 1.7–2.6 (1H, m.), 3.07 (1H, d.J=11Hz), 3.75 (3H, S.), 6.5–7.1 (4H, m.), 6.2 (1H, b.S.)
50	A	88.2	0.68 (3H, d.J=7Hz), 1.05 (3H, d.J=7Hz), 1.95–2.50 (1H, m.), 2.29 (3H, S.), 3.05 (1H, d.J=11Hz), 7.11 (centered, 4H, AB type 8), 9.52 (1H, b.S.)

*Signal patterns were designated as follows. S: singlet, d: doublet, m: multiplet, bS: broad Singlet

*Hydrolysis Method A: 50% H₂SO₄. Method B: Ethyleneglycol-aqueous KOH

SYNTHESIS EXAMPLE 4

Diethyl-2-phenyl-2-allyl-malonate

Diethyl-2-phenylmalonate (23.60 g, 0.10 mole, prepared from ethylphenylacetate, Org. Synth. Call, Vol.

2, 288) in 20 ml of dry benzene was added to a stirred solution of sodium hydride (2.38 g, 0.12 mole) in 150 ml of dry benzene at 30°–40° C., and the mixture was kept at 40° C. for 30 minutes. To the white enolate generated was added a solution of allyl bromide (24.2 g, 0.10 mole) in 20 ml of dry benzene at 5° C., thereafter stirring continued at room temperature for one hour.

The reaction mixture was poured into 200 ml of ice-water, and the organic layer was separated. The aqueous layer was extracted with ether (two 100 ml portions). The organic layers were combined, and washed successively with 5% hydrochloric acid, water and saturated sodium chloride solution, then dried over anhydrous sodium sulfate. The solvent was removed in vacuo, and the residual oil was distilled at reduced pressure to afford 24.27 g (87.90%, b.p. 115°–120°/0.25 mmHg) of diethyl-2-phenyl-2-allyl-malonate.

SYNTHESIS EXAMPLE 5

Ethyl- α -allylphenylacetate

Diethyl allylphenyl malonate (20.0 g, 0.072 mole) in 250 ml of dry EtOH was heated under reflux with 7.4 g of sodium ethoxide for 5 hours.

Ethanol was removed in vacuo, and the residue was poured to 75 g of ice-water. The mixture was extracted with ether (three 50 ml portions), ether layer was washed with saturated sodium chloride solution, dried over anhydrous sodium sulfate. Removal of the solvent, and distillation at reduced pressure afforded 12.54 g (b.p. 60°–63°/0.13 mmHg, 85.3%) of the phenyl acetate.

SYNTHESIS EXAMPLE 6

α -Allyl-phenyl acetic acid

Above ethylester was hydrolyzed with 10% potassium hydroxide in methanol at room temperature in a usual manner to give the acid (yield 94.6%, m.p. 34° C.).

SYNTHESIS EXAMPLE 7

Diethyl-2-phenyl-2-propargyl malonate

In a similar manner to Example 4, the malonate was prepared in 81.7% yield (b.p. 115°–120°/0.15 mmHg).

SYNTHESIS EXAMPLE 8

α -Propargyl phenyl acetic acid

20.0 g (0.073 mole) of diethyl-2-phenyl-2-propargyl malonate was hydrolyzed in 330 g of 5% methanolic potassium hydroxide at reflux temperature for 2 hours. Methanol was removed in vacuo, and the residue was dissolved in 150 ml of cold water, acidification with concentrated HCl caused a spontaneous decarboxylation to give the monocarboxylic acid, which was collected by ether extraction. (12.4 g, 97.6%, m.p. 90°–93° C.)

SYNTHESIS EXAMPLE 9

Although the compound which bears an isopropenyl group on α -position such as compounds No. 83, No. 85, No. 125, can be prepared in the usual esterification method using corresponding acid chloride, the double bond of the isopropenyl group readily isomerized to α – β position to make the acid or acid chloride into a mixture of double bond positions. A low yield was attained after chromatographic purification.

Following routes seemed to be better than the usual methods.

3'-Phenoxybenzyl- α -isopropenyl-4-methoxyphenylacetate (Compound No. 82)

(1) To a solution of isopropylmagnesium bromide, which was prepared from 38.1 g of isopropylbromide and 7.3 g of magnesium in 90 ml of dry ether, was added a solution of 4-methoxyphenylacetic acid (16.6 g) in 100 ml of dry tetrahydrofuran at an ice-bath temperature, thereafter the solution was stirred at room temperature for 2 hours. Dry acetone (7.0 g), diluted with 20 ml of dry toluene, was added to the solution, and it was heated to reflux for 5 hours. Cooled solution was acidified with 15% sulfuric acid, the aqueous phase separated, which was extracted with ether. Combined organic layers were extracted with 10% sodium carbonate solution. Carbonate extract was acidified with concentrated hydrochloric acid and extracted with ether. Ether extract was dried over anhydrous sodium sulfate. Removal of the solvent and a recrystallization from ethyl acetate afforded 12.4 g of 4-methoxy- β , β -dimethylatropic acid (55.2%, m.p. 124°–125° C.).

(2) To a solution of 4-methoxy- β , β -dimethylatropic acid (5.0 g) and triethylamine (4.5 g) in 50 ml of dimethylformamide, was added 7.0 g of 3-phenoxybenzylbromide gradually at an ice-bath temperature, and stirred overnight at room temperature. The mixture was poured into cold 10% H₂SO₄ and extracted with ether. The ether layer was washed with 10% sodium carbonate solution and dried over anhydrous sodium sulfate. Removal of the solvent gave 4.8 g of crude hydroxyester, which was dehydrated with P₂O₅ in benzene at 80° C. for 13 hours. Filtration, and removal of the solvent gave crude isopropenyl-ester as a dark oil. Purification with silica-gel chromatography afforded 1.9 g of pure ester. (40.5%, *n*_D¹⁶ 1.5798)

SYNTHESIS EXAMPLE 10

Ethyl-2-(cyclohexane-1'-ol-1'-yl)-butyrate

This was prepared from ethyl-2-bromobutyrate (29.3 g), cyclohexanone (14.7 g) and zinc (9.51 g) in benzene (50 ml) and toluene (25 ml) under usual Reformatsky conditions (b.p. 166°–117°/4 mmHg, yield 16.4 g, 51.0%).

SYNTHESIS EXAMPLE 11

Ethyl-2(cyclohexene-1-yl)-butyrate

Above hydroxyester (16.4 g) was heated under reflux with phosphorous pentoxide (11 g) in 60 ml of dry benzene for 3 hours. Cooled mixture was poured onto 80 ml of cold water, organic layer was separated and washed with saturated sodium chloride solution. Solvent was removed in vacuo, and the residue was distilled under reduced pressure to afford 11.6 g of pale yellow oil. (76.5%, b.p. 88°–92°/4 mmHg)

SYNTHESIS EXAMPLE 12

Ethyl-2,3-diethyl-3-hydroxy-pentanoate

In similar manners described in Example 10, this was prepared from 3-pentanone and ethyl-2-bromobutyrate. (yield 52.8%, b.p. 113°–121°/20 mmHg)

SYNTHESIS EXAMPLE 13

The hydroxyester obtained in Example 12, was dehydrated with phosphorous pentoxide in benzene in a manner described above. (yield 68.2%, b.p. 95°–98°/22 mmHg)

The compounds of the type designated as Compound Nos. 18, 21, 245, 246 and 248 were prepared from the corresponding carboxylic acid or an acid chloride derived therefrom which is obtained by hydrolyzing the ethyl ester prepared in the similar manner as described in Synthesis Examples 10-13. The hydrolysis can be effected by the conventional procedure using sodium hydroxide or potassium hydroxide in methanol at room temperature. The carboxylic acid thus obtained was found to contain a small amount of isomers with respect to double bonds (up to about 10%), but the crude carboxylic acid was subjected to esterification without isolating impurities (isomers) and the product was purified by silica gel chromatography in the final step.

The substituted ester compounds represented by the formula (I) exhibit an excellent pesticidal activity and are expected to have a repelling activity against mites as well as a synergistic activity with other biologically active agents, and can be employed in a wide variety of applications at low cost as pesticidal compositions for use in agricultural and horticultural, forest, sanitary fields as well as in a stockroom of cereals and a composition for controlling mites.

In practical use of these compounds, an appropriate amount of one or more solvents, fillers, diluents, active agents, dispersing agents, surface active agents, spreaders, pressure-imparting agents, emulsifying agents and attractants can be incorporated into the ester compounds of this invention to form an emulsifiable concentrate, a wettable powder, a dust, a granule, a fine grain, a coating agent, a powder, an oil preparation, an aerosol, a mosquito coil, a smoking agent, a fumigating agent, a mosquito incense agent for electrical vaporization, and the like.

The effect of the pesticidal compositions of this invention is further illustrated by the following experiments.

EXPERIMENTAL EXAMPLE 1

An emulsifiable concentrate was prepared by blending each of the compounds (1) to (300) of the present invention, xylol and Sorpol SM-200 (trade name; manufactured by Toho Chemical Co., Ltd.) in a proportion of 25%, 50% and 25% (by weight), respectively. 20 to 25 day-old rice plants after seeding were grown in pots and sprayed with each 300-fold dilute solution of the emulsifiable concentrates thus prepared and a 300-fold dilute solution of carbaryl 30% wettable powder diluted with water as a control commercial agent in an amount of 10 ml/pot in each case and then the pots were covered by a cylindrical metal net. 15 smaller brown planthoppers (*Delphacodes striatella* Fallén) per group were released in each of the pots treated with the carbaryl wettable powder, and compound (1) to (21) and (102) to (110), while 15 green rice leafhopper (*Nephotettix bipunctatus cincticeps* Uhler) per group were released in each of the pots treated with the carbaryl wettable powder and compounds (22) to (101) and (111) to (300). Observation on survival one day after the treatment revealed that each of the compounds of this invention exhibits more than 90% killing effect as in the case of carbaryl.

EXPERIMENTAL EXAMPLE 2

Of the emulsifiable concentrates prepared in Experimental Example 1, each of those prepared from the compounds of Table 1 below was diluted with water to each of test concentrations and a 200 ml portion thereof

was added to a 300 ml volume beaker. 30 larvae of mosquito per group were released in each of the beaker and allowed to stand for 1 day. Survival was observed and a LC₅₀ (50% lethal concentration) was determined by calculating a percent killing in each case. The results obtained are also shown in Table 1. As a control, lindane emulsifiable concentrate was used.

TABLE 1

Insecticidal Activity on Larvae of Mosquito (<i>Culex pipiens</i>)			
Test Emulsifiable Concentrate	LC ₅₀ (ppm)	Test Emulsifiable Concentrate	LC ₅₀ (ppm)
Compound (1)	0.0015	Compound (2)	0.0034
Compound (10)	0.0030	Compound (14)	0.0026
Compound (16)	0.0014	Compound (102)	0.0095
Compound (103)	0.0092	Compound (104)	0.0120
Compound (105)	0.0105	Compound (110)	0.0125
Compound (111)	0.0025	Compound (112)	0.0046
Compound (113)	0.0053	Compound (115)	0.0050
Compound (119)	0.0044	Compound (120)	0.0040
Compound (121)	0.0037	Compound (125)	0.0032
Compound (126)	0.0056	Compound (128)	0.0047
Compound (145)	0.0039	Compound (165)	0.0012
Compound (167)	0.0043	Compound (177)	0.0074
Compound (180)	0.0072	Compound (181)	0.0820
Compound (183)	0.0885	Compound (184)	0.0078
Compound (204)	0.0075	Compound (206)	0.0032
Compound (210)	0.0035	Compound (212)	0.0037
Compound (214)	0.0127	Compound (215)	0.0132
Compound (226)	0.0135	Compound (227)	0.0019
Compound (231)	0.0014	Compound (298)	0.0097
Lindane	0.22		

EXPERIMENTAL EXAMPLE 3

A 2000-fold dilute solution of each of the emulsifiable concentrates prepared in Experimental Example 1 from the compounds of Table 2 below was sprayed on a 12 day-old seedling of mottled kidney beans after seeding grown in a pot in an amount of 10 ml/pot. The seedling thus treated was cut off and put in a wide opening bottle. In the same manner, an untreated seedling of mottled kidney beans in the same stage was cut off and put in another wide opening bottle. Another fresh seedling was fixed with pins bridging on the former treated seedling and the latter untreated seedling. A leaf of mottled kidney beans on which a number of spider mites (*Tetranychus telarius*) were parasitic was placed on the middle of the bridging seedling and allowed to stand for 2 days. Activity on repellency was assessed by investigating the numbers of mites swarming on the treated seedling and the untreated seedling. The assessment standards are as follows:

-	The ratio of the number of mites on the untreated seedling to that on the treated seedling	1:1
+	The ratio of the number of mites on the untreated seedling to that on the treated seedling	up to 4:1
++	The ratio of the number of mites on the untreated seedling to that on the treated seedling	greater than 4:1

The results obtained was shown in Table 2 below.

TABLE 2

Repellent Activity on Spider Mites			
Test Emulsifiable Concentrate	Repellency	Test Emulsifiable Concentrate	Repellency
Compound (1)	++	Compound (3)	+
Compound (4)	++	Compound (10)	++

TABLE 2-continued

Repellent Activity on Spider Mites			
Test Emulsifiable Concentrate	Repellency	Test Emulsifiable Concentrate	Repellency
Compound (18)	+	Compound (23)	+
Compound (29)	++	Compound (35)	+
Compound (45)	++	Compound (65)	+
Compound (73)	+	Compound (82)	++
Compound (92)	++	Compound (98)	++
Compound (116)	++	Compound (117)	++
Compound (118)	++	Compound (121)	++
Compound (124)	+	Compound (127)	++
Compound (130)	+	Compound (165)	++
Compound (168)	++	Compound (175)	++
Compound (176)	++	Compound (177)	++
Compound (178)	++	Compound (200)	++
Compound (204)	++	Compound (206)	++
Compound (210)	++	Compound (216)	++
Compound (217)	++	Compound (218)	++
Compound (219)	++	Compound (220)	++
Compound (221)	++	Compound (227)	++
Compound (234)	++	Compound (235)	++

EXPERIMENTAL EXAMPLE 4

0.2% oil preparation was prepared from each of the compounds (1), (2), (8), (14), (22), (25), (26), (27), (51), (62), (73), (82), (86), (88) to (90), (92), (94), (96), (111), (119), (121), (165), (175), (180), (184), (185), (195), (200), (201), and (208) and allethrin as a control using deodorized kerosene. About 50 adult mosquitoes (*Culex pipiens*) were released in a (70 Cm)³ volume glass box and sprayed with each of the oil preparations obtained above in an amount of 0.7 ml under a pressure of 1.5 Kg/cm². It was found that more than 80% of mosquitoes was knocked down by both allethrin and the compounds of this invention.

EXPERIMENTAL EXAMPLE 5

20 oriental house flies per group were released in a (70 cm)³ volume glass box and sprayed with each of the oil preparations prepared in Experimental Example 4 from the compounds of Table 3 below and allethrin in an amount of 0.7 ml under a pressure of 1.5 Kg/cm². The test was repeated several times and the knock down of the flies was observed to determine KT₅₀ (50% knock-down time). The results obtained are shown in Table 3.

TABLE 3

Activity on Knock-Down Against Oriental House Flies			
Test Oil Preparation	KT ₅₀ (sec.)	Test Oil Preparation	KT ₅₀ (sec.)
Compound (1)	167	Compound (8)	128
Compound (22)	136	Compound (51)	104
Compound (62)	154	Compound (63)	140
Compound (82)	132	Compound (88)	147
Compound (89)	155	Compound (96)	92
Compound (111)	138	Compound (119)	132
Compound (121)	130	Compound (165)	145
Compound (175)	152	Compound (180)	157
Compound (184)	142	Compound (185)	137
Compound (195)	140	Compound (200)	114
Compound (201)	105	Compound (208)	150
Allethrin	186		

EXPERIMENTAL EXAMPLE 6

To each of the compounds (22), (51), (62), (63), (82), (86), (125), (150), (160), (176), (177), (178), (184), (185), (191), (195), (200), (201), (204), (208), (210), (212) and (226) was added piperonyl butoxide in an amount of 5

times that of the active ingredient, respectively to prepare an acetone solution having a given concentration. Each of the thus prepared acetone solutions was applied onto a thoracic dosal plate of oriental house flies using a microshringe to observe a synergistic effect on insecticidal activity by piperonyl butoxide. Potentiation magnification of insecticidal activity by the addition of piperonyl butoxide is shown in Table 4.

TABLE 4

Potentiation Effect on Insecticidal Activity by the Addition of Piperonyl Butoxide			
Test Compound	Potentiation Magnification	Test Compound	Potentiation Magnification
Compound (2)	5.0	Compound (51)	7.2
Compound (62)	5.3	Compound (63)	4.8
Compound (82)	6.7	Compound (86)	4.5
Compound (119)	5.5	Compound (125)	6.0
Compound (150)	4.2	Compound (160)	7.2
Compound (176)	5.2	Compound (177)	5.8
Compound (178)	5.4	Compound (184)	5.6
Compound (185)	5.0	Compound (191)	4.3
Compound (195)	5.7	Compound (200)	7.5
Compound (201)	7.0	Compound (204)	4.7
Compound (206)	5.1	Compound (208)	5.2
Compound (210)	4.8	Compound (212)	5.0
Compound (226)	4.5		

As is apparent from the above Experimental Examples, the compounds of this invention exhibit an excellent biological activity on various noxious insects and mites and also are of low toxicity.

Oral toxicity to mice for representative compounds is shown in the Table below.

TABLE

Test Compound	LD ₅₀ (mg/Kg)	Test Compound	LD ₅₀ (mg/Kg)
Compound (1)	>1000	Compound (2)	>1000
Compound (8)	750	Compound (22)	900
Compound (62)	>1000	Compound (63)	>1000
Compound (80)	>1000	Compound (82)	>1000
Compound (108)	>1000	Compound (119)	>1000
Compound (125)	>1000	Compound (129)	>1000
Compound (130)	>1000	Compound (145)	>1000
Compound (150)	>1000	Compound (176)	>1000
Compound (184)	940	Compound (206)	>1000

The compositions according to the present invention are very useful for controlling insanitary insects such as mosquitoes, flies and cockroaches, grain insects such as rice weevil (*Calandra oryzae*) and mites as well as agricultural noxious insects such as plant-hoppers, green rice leafhopper (*Nephotettix bipunctatus cincticeps* Uhler), cabbage army worms (*Barathra brassicae* Linne), diamond-back moths (*Plutella maculipennis* Curtis), noctuidae, cabbage worm (*Pieris rapae* Linne), rice stem borer (*Chilo suppressalis* Walker), aphids, tortrixes, leaf-miners and the like.

They are also superior whenever they may be freely used for harvested crops, horticultural application, cultures in green house, and packaging materials for food-stuffs.

In preparing the compositions according to the present invention, it is possible to obtain more excellent controlling effects by employing a combination of two or more compounds of this invention, and further, it is possible to obtain various compositions which are useful for a wide variety of applications by combining one or more compounds of this invention with other pesticidal compounds, for example, organochloric compounds

such as DDT, BHC, Methoxychlor and the like; organic phosphorus compounds such as Sumithion (trade name of Sumitomo Chemical Company Limited), DDVP, diazinon, phenthion, Cyanox (trade name of Sumitomo Chemical Company Limited) and the like; carbamates such as 1-naphthyl-N-methylcarbamate, 3,4-dimethylphenyl N-methylcarbamate, 3,5-dimethylphenyl N-methylcarbamate and the like; cyclopropanecarboxylic acid esters such as pyrethrin, allethrin, Neopynamin (trade name of Sumitomo Chemical Company Limited), Chrysron (trade name of Sumitomo Chemical Company Limited), 3-phenoxybenzyl chrysanthemate, 5-propargylfurfuryl chrysanthemate and their geometrical or optical isomers and the like, as well as piperonyl butoxide, sulfoxide, safroxane, isobornyl thiocyanate (I.B.T.A.) and octachloro dipropyl ester (S-421) which potentiate the effect of the esters synergetically; terephthalic acid, isophthalic acid and PHT which are generally used in fumigating preparations; phenol derivatives used as a stabilizer; bisphenol derivatives; arylamines such as phenyl- α -naphthylamine, phenyl- β -naphthylamine, a condensate of phenetidin and acetone and the like; other insecticidal agents or miticidal agents such as Padan, Galecron, antimicrobial agents, nematocidal agents, herbicidal agents, fertilizers and other agricultural agents. These compositions are expected to be useful for saving man power for application of these agricultural agents and for obtaining synergetic activities between the active ingredients of the composition.

The preparation of compositions and the effect of compounds of this invention are further illustrated by the following formulations and Examples, but they are not to be construed as limiting the scope of this invention. The parts used in formulations are by weight unless otherwise indicated.

Formulation 1

To 20 parts of each of Compound Nos. 1 to 110 of this invention, there were added 10 parts of Sumithion (as described previously), 10 parts of Sorpol SM-200 (as described previously), 60 parts of xylol, and the mixture was dissolved by thoroughly stirring and mixing to obtain an emulsifiable concentrate in each case.

Formulation 2

To 10 parts of each of Compound Nos. 111 to 192 of this invention, there were added 10 parts of Cyanox (as described previously), 10 parts of Sorpol SM-200 (as described previously) and 70 parts of xylol, and the mixture was dissolved by thoroughly stirring and mixing to obtain an emulsifiable concentrate in each case.

Formulation 3

To 15 parts of each of Compound Nos. 193 to 232 of this invention, there were added 30 parts of piperonyl butoxide, 15 parts of Sorpol SM-200 (as described previously) and 40 parts of xylol, and the mixture was dissolved by thoroughly stirring and mixing to obtain an emulsifiable concentrate in each case.

Formulation 4

To 15 parts of each of Compound Nos. 233 to 300 of this invention, there were added 20 parts of a 25% pyrethrin extract, 20 parts of piperonyl butoxide, 15 parts of Sorpol SM-200 (as described previously) and 30 parts of xylol, and the mixture was dissolved by thor-

oughly stirring and mixing to obtain an emulsifiable concentrate in each case.

Formulation 5

0.5 parts of each of Compound Nos. 1, 10, 102, 103, 108 and 111 to 115 was dissolved in kerosene in an amount sufficient to make 100 parts thereby obtaining an oil preparation in each case.

Formulation 6

To 0.2 parts of each of Compound Nos. 116 to 192 of this invention, there were added 0.1 parts of allethrin α -trans chrysanthemate and 1.5 parts of safroxane, and the mixture was dissolved in kerosene to prepare 100 parts of an oil preparation in each case.

Formulation 7

To 0.2 parts of each of Compound Nos. 193 to 300 of this invention, there were added 0.1 parts of Neopynamine and 1.5 parts of S-421, and the mixture was dissolved in kerosene to prepare 100 parts of an oil preparation in each case.

Formulation 8

To 10 parts of each of Compound Nos. 1 to 110 of this invention, there were added 10 parts of 1-naphthyl-N-methylcarbamate and 5 parts of Sorpol SM-200 (as described previously) followed by mixing. 75 parts of 300-mesh talc was then added to the mixture and the resulting blend was thoroughly stirred in a triturator to obtain a wettable powder in each case.

Formulation 9

The active pesticidal ingredients having a composition shown in Table below were dissolved in 20 ml of methanol and the resulting solution was mixed with 100 g of a carrier for mosquito coil (a mixture of tabu powder, pyrethrum marc and wood powder in a proportion of 3:5:1, respectively), followed by blending uniformly. After methanol was evaporated, 150 ml of water was added to the blend, and the mixture was thoroughly kneaded, molded and dried to prepare a mosquito coil in each case.

Formulations of Mosquito Coil

No.	Composition	
1.	Compound No. 1	0.3 g
	Allethrin	0.3 g
2.	Compound No. 8	0.3 g
	Allethrin	0.3 g
3.	Compound No. 22	0.3 g
	Allethrin	0.2 g
	BHT	0.3 g
4.	Compound No. 37	0.3 g
	Allethrin	0.3 g
5.	Compound No. 51	0.3 g
	Allethrin α -trans chrysanthemate	0.1 g
	BHT	0.4 g
6.	Compound No. 62	0.2 g
	5-Propargylfurfuryl chrysanthemate	0.2 g
	BHT	0.8 g
7.	Compound No. 82	0.3 g
	5-Propargyl-2-methyl-3-furylmethyl chrysanthemate	0.1 g
	BHT	0.4 g
8.	Compound No. 146	0.3 g
	Allethrin d-cis,trans chrysanthemate	0.1 g
	BHT	0.5 g
9.	Compound No. 169	0.3 g
	Allethrin	0.2 g

-continued

Formulations of Mosquito Coil		
No.	Composition	
10.	BHT	0.5 g
	Compound No. 184	0.3 g
	Allethrin	0.3 g
	BHT	0.3 g

Formulation 10

The active pesticidal ingredients having a composition shown in Table below were dissolved in a mixed solvent comprising xylol and refined kerosene (1:1 by volume) to prepare 15 parts of solution and the solution was filled in an aerosol dispensing container. After a valve was mounted to the container, the aerosol dispenser was pressurized by charging 85 parts of a propellant (for example, freon, vinyl chloride monomer, liquefied petroleum gas, etc.) to obtain an aerosol preparations.

Aerosol Formulations		
No.	Composition	
1.	Compound No. 22	0.3 (parts)
	3-Phenoxybenzyl-d-cis,trans chrysanthemate	0.1
2.	Compound No. 22	0.2
	Neopynamin	0.2
	I.B.T.A.	1
3.	Compound No. 62	0.2
	Allethrin d-trans chrysanthemate	0.2
	Safroxane	2
4.	Compound No. 62	0.3
	DDVP	0.3
5.	Compound No. 82	0.2
	Sumiothin	0.3
	Neopynamin	0.2
6.	Compound No. 82	0.4
	piperonyl butoxide	2.0
7.	Compound No. 88	0.2
	Neopynamin	0.2
	Chrysron	0.1
8.	Compound No. 89	0.3
	Allethrin d-trans chrysanthemate	0.2
9.	Compound No. 96	0.2 (parts)
	Allethrin d-cis, trans chrysanthemate	0.2
	Safroxane	2
10.	Compound No. 102	0.4
	Neopynamin	0.1
	Piperonyl butoxide	1.5
11.	Compound No. 103	0.4
	Allethrin	0.1
	S-421	1.5
12.	Compound No. 108	0.4
	25% Pyrethrin extract	0.5
	Piperonyl butoxide	1.0
13.	Compound No. 125	0.4
	I.B.T.A.	2

Formulation 11

The active pesticidal ingredients having a composition shown in Table below were dissolved in an appropriate amount of chloroform, and the solution was adsorbed uniformly on the surface of asbestos having a size of 2.5 cm × 1.5 cm and a thickness of 0.3 mm. Another piece of asbestos having the same size and thickness was then placed on the treated surface of the above asbestos to prepare a fibrous fumigation pesticidal composition (mat) for heating on a hot plate. As a fibrous carrier, those having the same effect as that of asbestos

such as a pulp board, etc. may also be used in place of asbestos.

Formulation of Mat for Heating on Hot Plate		
No.	Composition	
1.	Compound No. 51	0.05 g
	Allethrin	0.02 g
	Piperonyl butoxide	0.07 g
2.	Compound No. 63	0.07 g
	Allethrin d-trans chrysanthemate	0.01 g
	Piperonyl butoxide	0.1 g
3.	Compound No. 80	0.5 g
	5-Propargyl furylmethyl chrysanthemate	0.02 g
	Piperonyl butoxide	0.15 g
4.	Compound No. 82	0.04 g
	Allethrin	0.04 g
	Piperonyl butoxide	0.08 g
5.	Compound No. 88	0.05 g
	5-Propargyl-2-methyl-3-furylmethyl chrysanthemate	0.02 g
	Piperonyl butoxide	0.15 g
6.	Compound No. 103	0.05 g
	Allethrin	0.02 g
	Piperonyl butoxide	0.15 g
7.	Compound No. 107	0.05 g
	5-Phenoxybenzyl chrysanthemate	0.03 g
	Compound No. 125	0.05 g
8.	Compound No. 125	0.05 g
	Allethrin	0.01 g

Formulation 12

To 1 part of each of Compound Nos. 22, 51, 62, 63, 83, 86, 94, 103, 108 and 125 of this invention was added 5 parts of piperonyl butoxide and the mixture was dissolved in 20 parts of acetone. After 94 parts of 300-mesh diatomaceous earth was added, the resulting blend was thoroughly stirred and mixed in a triturator, and acetone was removed by evaporation to obtain a powder in each case.

EXAMPLE 1

10 third inster larvae of tobacco cutworm (*Spodoptera lituna* Fabricius) were placed in a glass tray having a high side-wall and a diameter of 14 cm, and 1 ml of a 200-fold dilute solution of each of the emulsifiable concentrates obtained in Formulations 1 and 3 diluted with water was sprayed on the larvae by a sprayer and the larvae thus treated were then transferred into a glass tray containing feeds followed by allowing to stand. Two days after the treatment, greater than 80% of the larvae was found to be killed in each case.

EXAMPLE 2

In an area of the radish field in 5-6 leaf stage, a number of green peach aphid (*Myzus persicae* Sulza) was atomized with a 200-fold dilute solution of each of the emulsifiable concentrates obtained in Formulations 1, 2 and 4 diluted with water at a level of 424 l/acre. Investigation two days after treatment revealed that the density of aphids is reduced to less than 1/10 the density before treatment in each case.

EXAMPLE 3

A piece of plywood board (15 cm × 15 cm) was applied with a 200fold dilute solution of each of the emulsifiable concentrates obtained in Formulations 1 and 2 diluted with water in an amount of 50 ml/m² followed by being dried. The adult german cockroaches (*Bla-*

tuella germanica Linne) were then brought into contact with the plywood board thus treated for 1 hour, and it was found that more than 80% cockroaches was killed three days after contacting with the board.

EXAMPLE 4

A field of egg-plant (ripe plant) growing a number of larvae of potato lady beetle (*Epilachna spersa orientalis* Dieke) was atomized with a 200-fold dilute solution of each of the emulsifiable concentrates obtained in Formulation 1 in an amount of 424 l/acre. Investigation on knock-down thirty minutes after the treatment revealed that 90% larvae fell to the ground. A further investigation on density of larvae 24 hours after the treatment revealed that almost 100% larvae had been controlled as compared with the density of larvae before the treatment.

EXAMPLE 5

Forty-five day-old rice plants after seeding were grown in Wagner pots (1/50000) and sprayed with a 400-fold dilute solution of each of the wettable powders obtained in Formulation 8 in an amount of 10 ml/pot. The pots were then covered with a cylindrical metal net and about 20 larvae of leaf hopper (*Cicadula sexnotata*) were released therein. In each case, more than 80% larvae were found to be killed one day after the release.

EXAMPLE 6

Each of the dusts obtained in Formulation 12 was placed uniformly at the bottom of glass trays having a diameter of 14 cm and a high side-wall in an amount of 2 g/m² and butter was applied to the glass wall except for about 1 cm portion of the bottom. 10 adult german cockroaches per group were then released in the glass trays and, after contacting 30 minutes, the larvae were transferred to a fresh glass container. In each case, more than 80% of the larvae was found to be killed three days after the treatment.

EXAMPLE 7

Each of the oil preparations obtained in Formulation 5 was atomized using Campel's turn table apparatus [Soap and Sanitary Chemicals, Vol. 14, No. 6, 119 (1938)] in an amount of 5 ml. 20 seconds after atomizing, the shutter was opened and 100 adult oriental house flies (*Musca domestica vicina* Maquant) per group were exposed to the falling atomized fumes for a period of 10 minutes. The flies thus treated were then fed and followed by allowing to stand. More than 80% of the flies were found to be killed in each case.

EXAMPLE 8

Each of the oil preparations obtained in Formulation 7 was sprayed in a (70 cm)³ volume glass box containing about 50 adult oriental house flies in an amount of 0.7 ml under a pressure of 1.5 Kg/cm² and it was found that more than 80% of the flies was knocked down in 10 minutes after the treatment in each case.

EXAMPLE 9

About 50 adult mosquitoes (*Culex pipiens*) per group were released in a (70 cm)³ volume glass box and sprayed with each of the oil preparations obtained in Formulation 6 in an amount of 0.7 ml under a pressure of 1.5 Kg/cm². It was found that more than 80% of the flies was knocked down in 10 minutes after the treatment in each case.

EXAMPLE 10

A nylon net of about 15 mesh was sprayed on the lower part of a glass cylinder having a diameter of 20 cm and a height of 20 cm and butter was applied on the upper part of the glass cylinder in a width of about 3 cm. Then 20 adult german cockroaches were released therein. The glass cylinder containing the cockroaches was plied on another glass cylinder of the same size, and further another glass cylinder having a diameter of 20 cm and a height of 40 cm was placed thereon. 0.5 ml of each of the oil preparations obtained in Formulation 7 was sprayed from the upper cylinder using a glass atomizer under a pressure of 0.75 Kg/cm² and then sealed followed by allowing to stand. It was found that more than 90% of the cockroaches were knocked down in 30 minutes after the treatment and that more than 90% of the cockroaches were killed 3 days after the treatment.

EXAMPLE 11

Insecticidal activity of each of the aerosols obtained in Formulation 10 on oriental house flies was tested by an aerosol test method using a Peet grady chamber [a method reported in Soap and Chemical Specialities Bluebook (1965)]. The test results revealed that knock-down of more than 80% of the flies was observed in 15 minutes after the treatment and more than 70% of the flies was killed one day after the treatment in each case.

EXAMPLE 12

About 50 adult mosquitoes (*Culex pipiens*) were released in a (70 cm)³ volume glass box and a mosquito coil both ends thereof having been ignited, was placed in the center of the box bottom. After 20 minutes more than 80% of the flies was found to be knocked down.

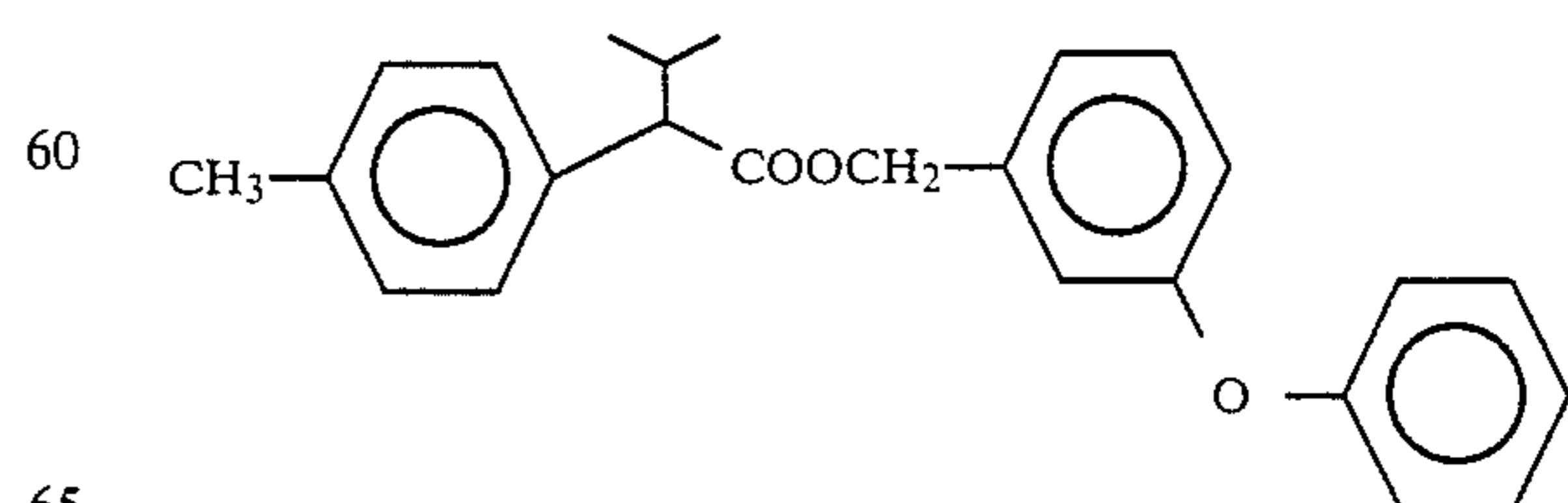
EXAMPLE 13

About 50 adult mosquitoes (*Culex pipiens*) were released in a (70 cm)³ volume glass box and a heating mat obtained in Formulation 11 fixed on an electrothermic plate was placed in the glass box followed by heating to smoke. Within 20 minutes, more than 90% of the mosquitoes was found to be knocked down.

While the invention has been described in detail and with reference to specific embodiments thereof, it will be apparent to one skilled in the art that various changes and modifications can be made therein without departing from the spirit and scope thereof.

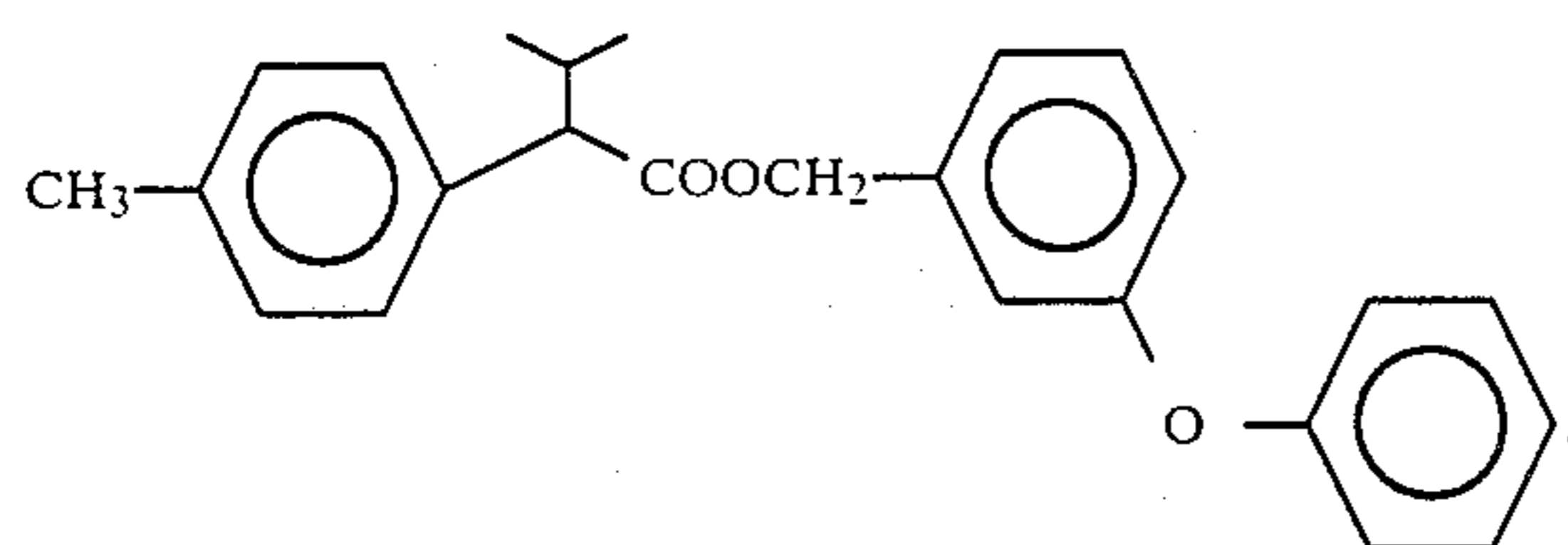
What is claimed is:

1. An insecticidally or miticidally effective optically active phenyl acetate of the formula

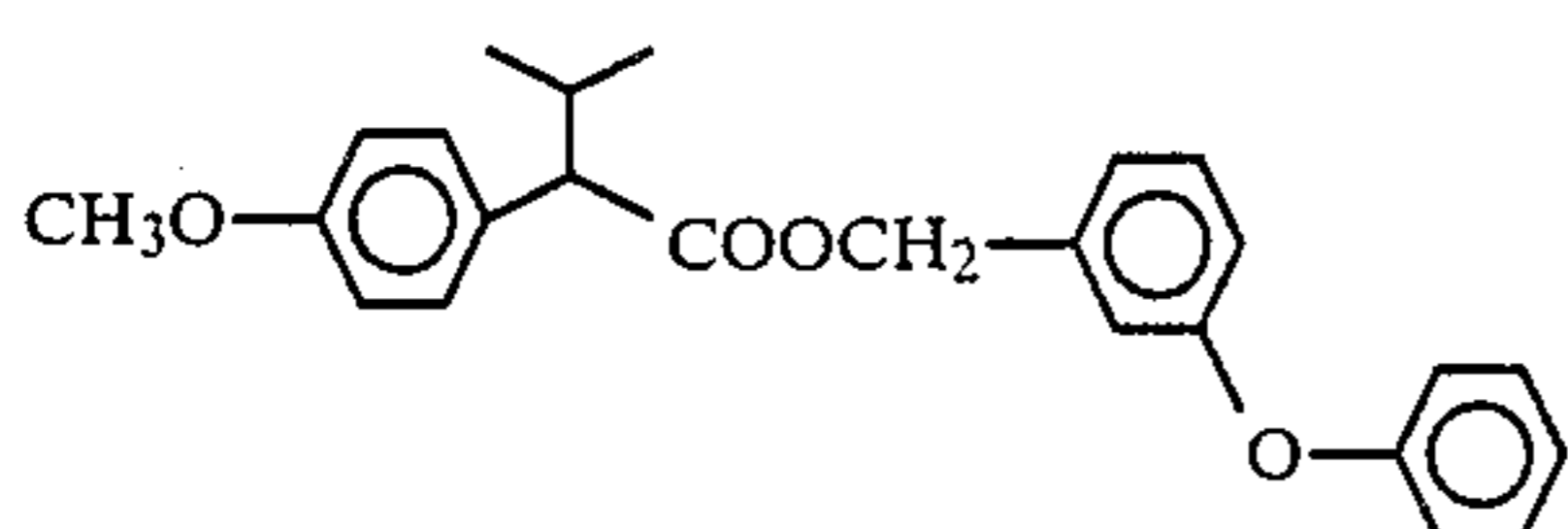


2. An insecticidally or miticidally effective mixture of optical isomers of the formula

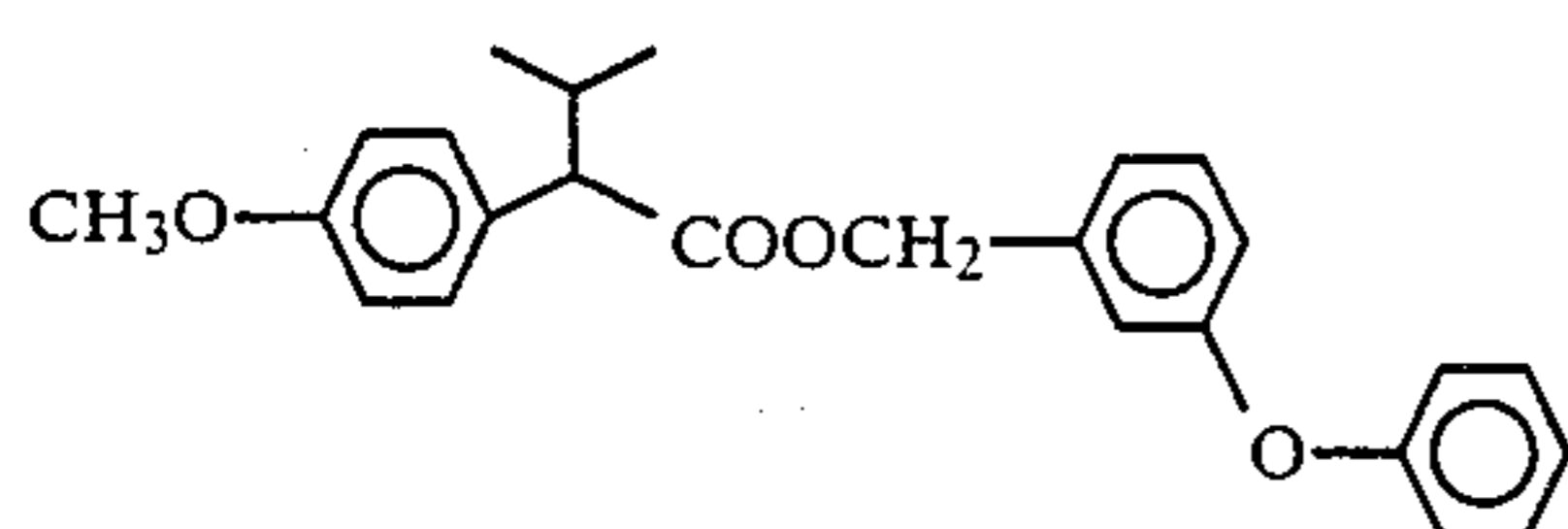
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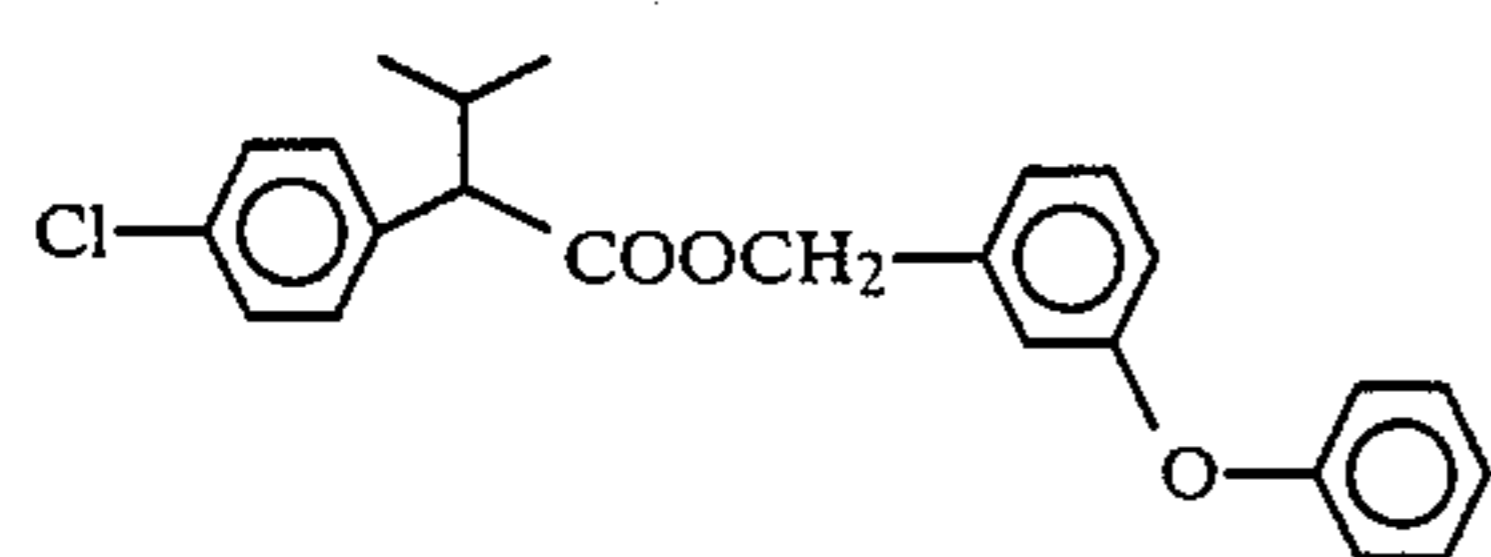
3. An insecticidally or miticidally optically active phenyl effective acetate of the formula



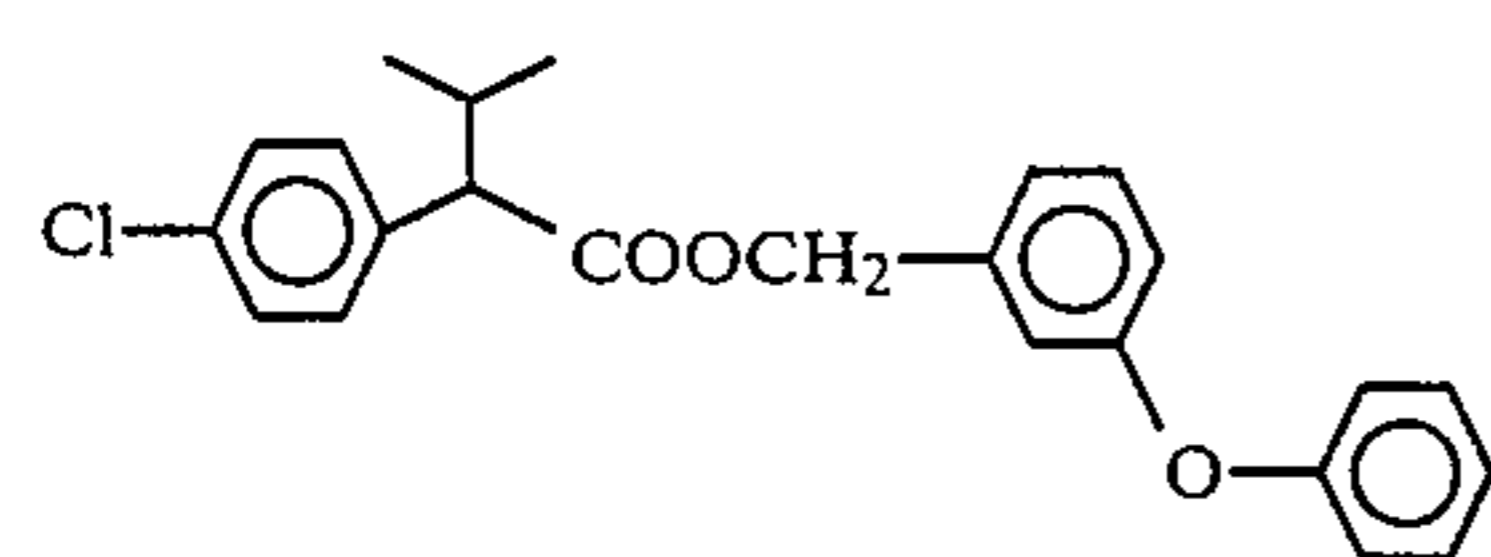
4. An insecticidally or miticidally effective mixture of optical isomers of the formula



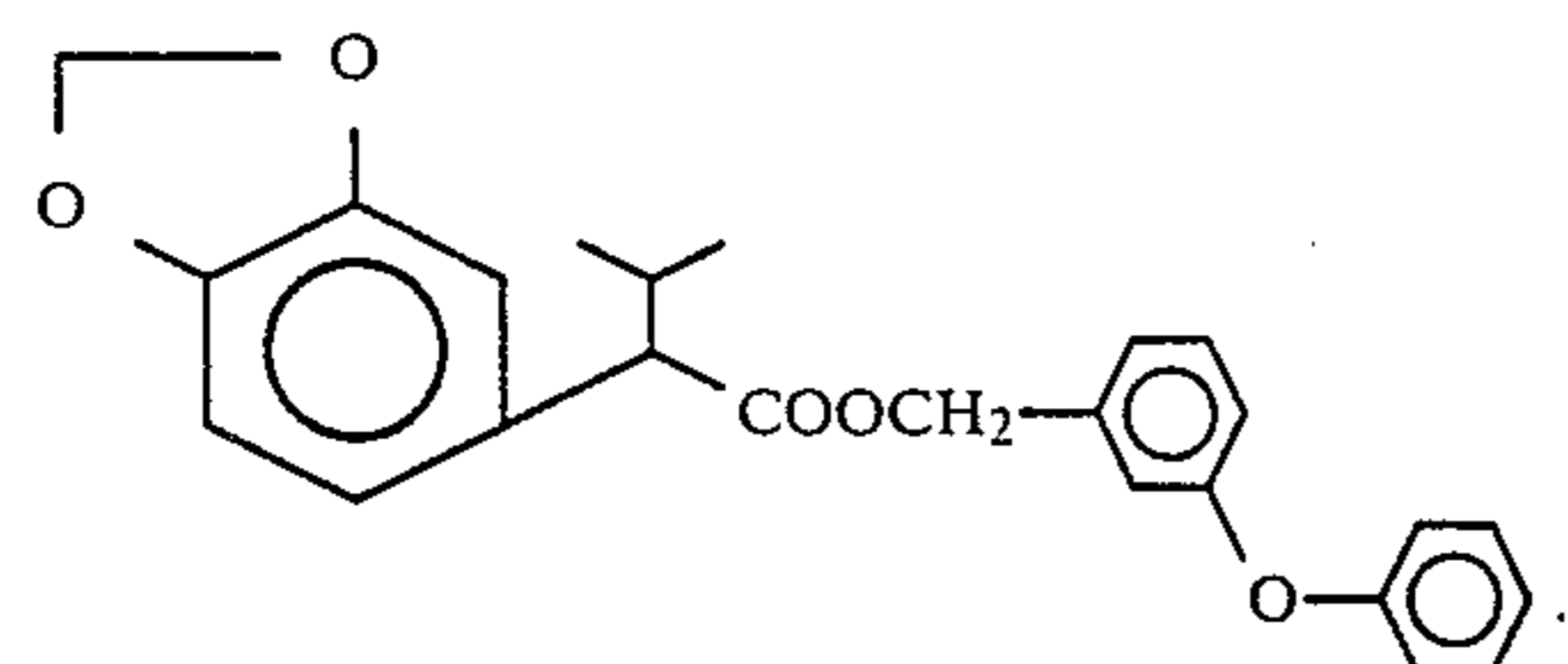
5. An insecticidally or miticidally effective optically active phenyl acetate of the formula



6. An insecticidally or miticidally effective mixture of optical isomers of the formula

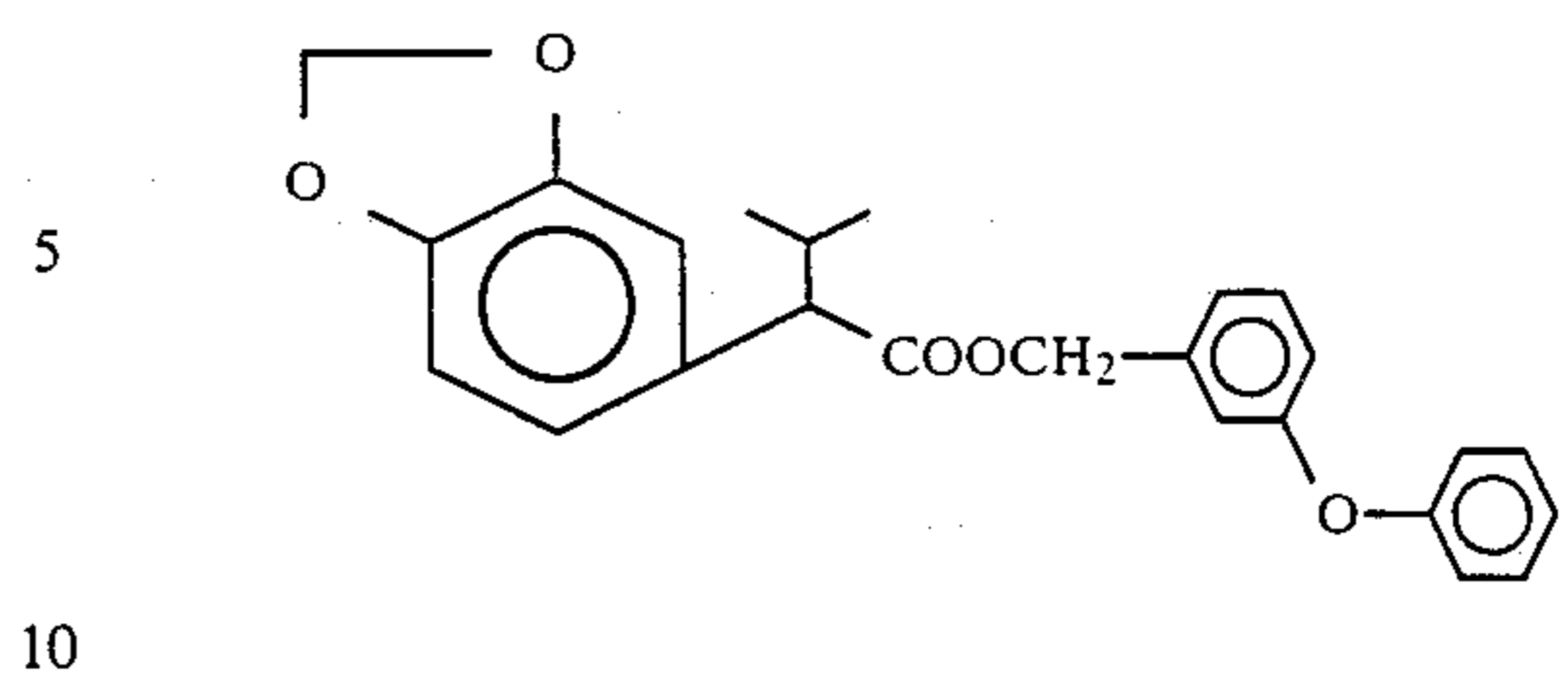


7. An insecticidally or miticidally effective optically active phenyl acetate of the formula

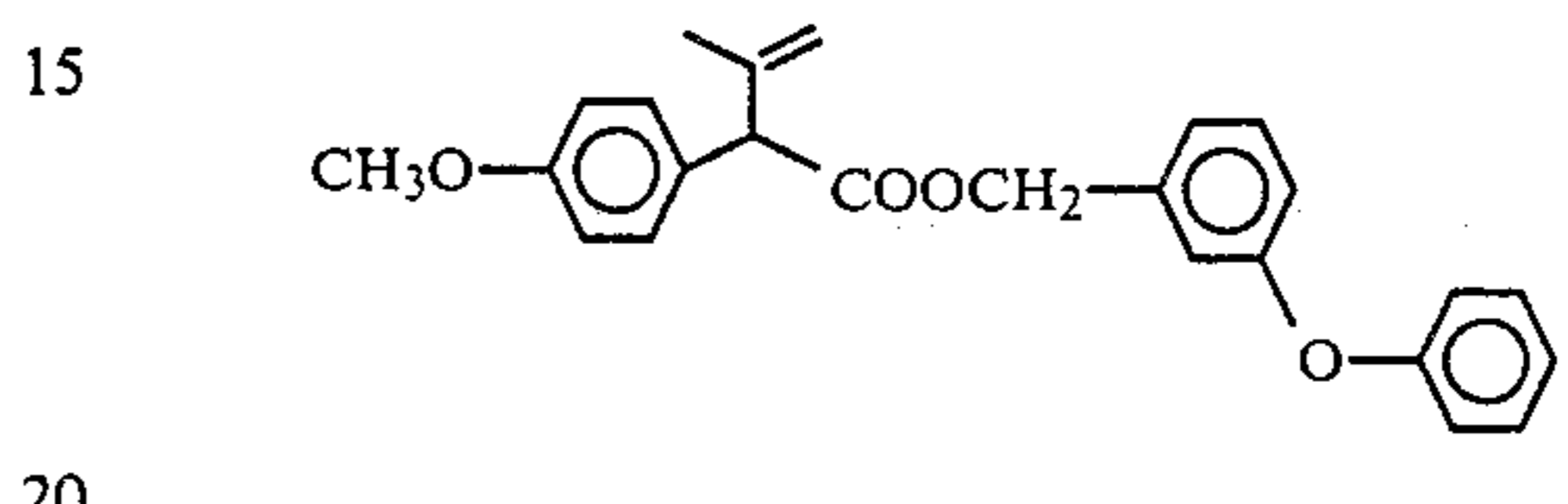


8. An insecticidally or miticidally effective mixture of optical isomers of the formula

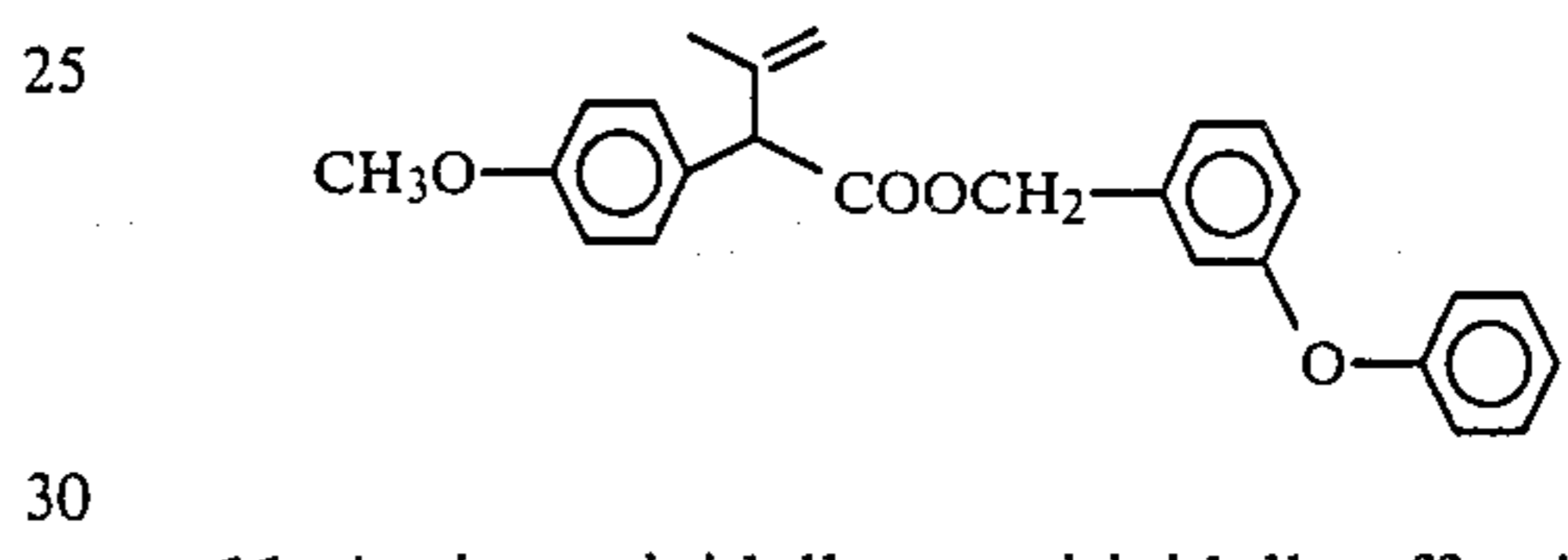
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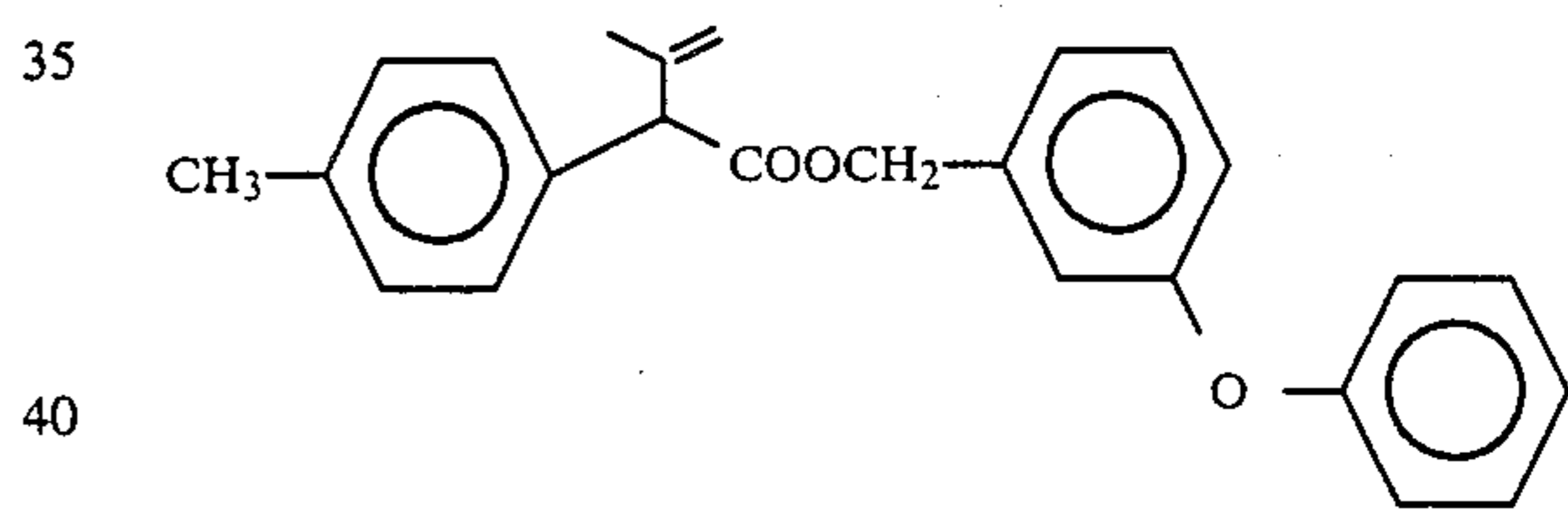
9. An insecticidally or miticidally effective optically active phenyl acetate of the formula



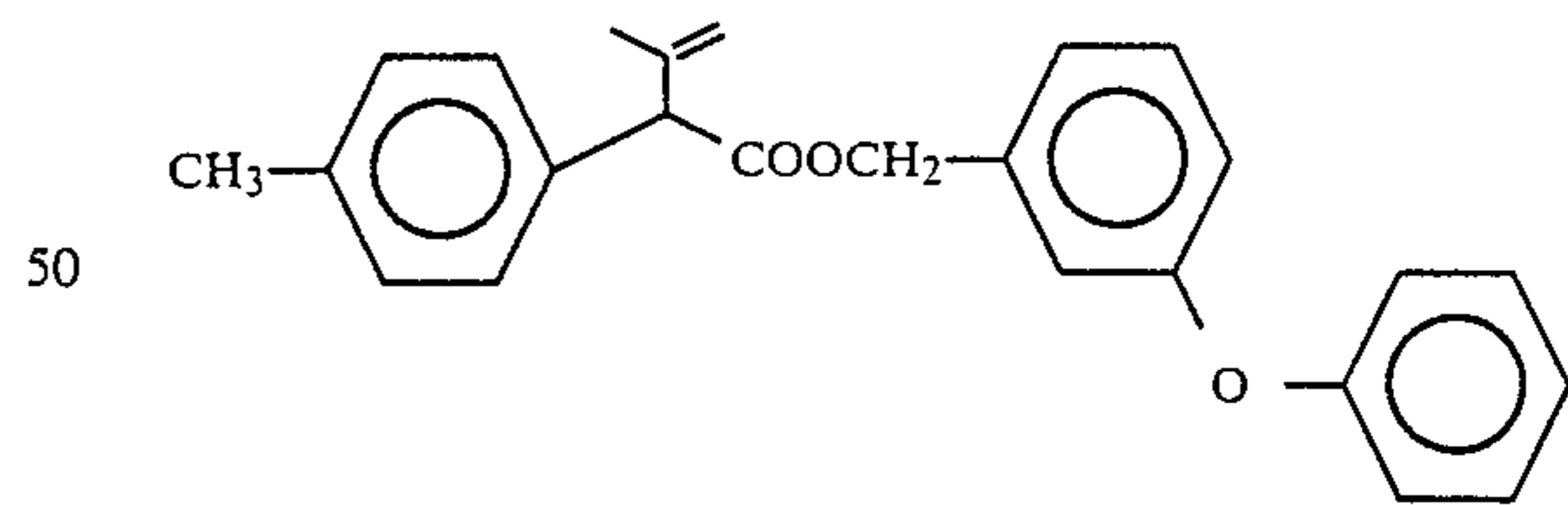
10. An insecticidally or miticidally effective mixture of optical isomers of the formula



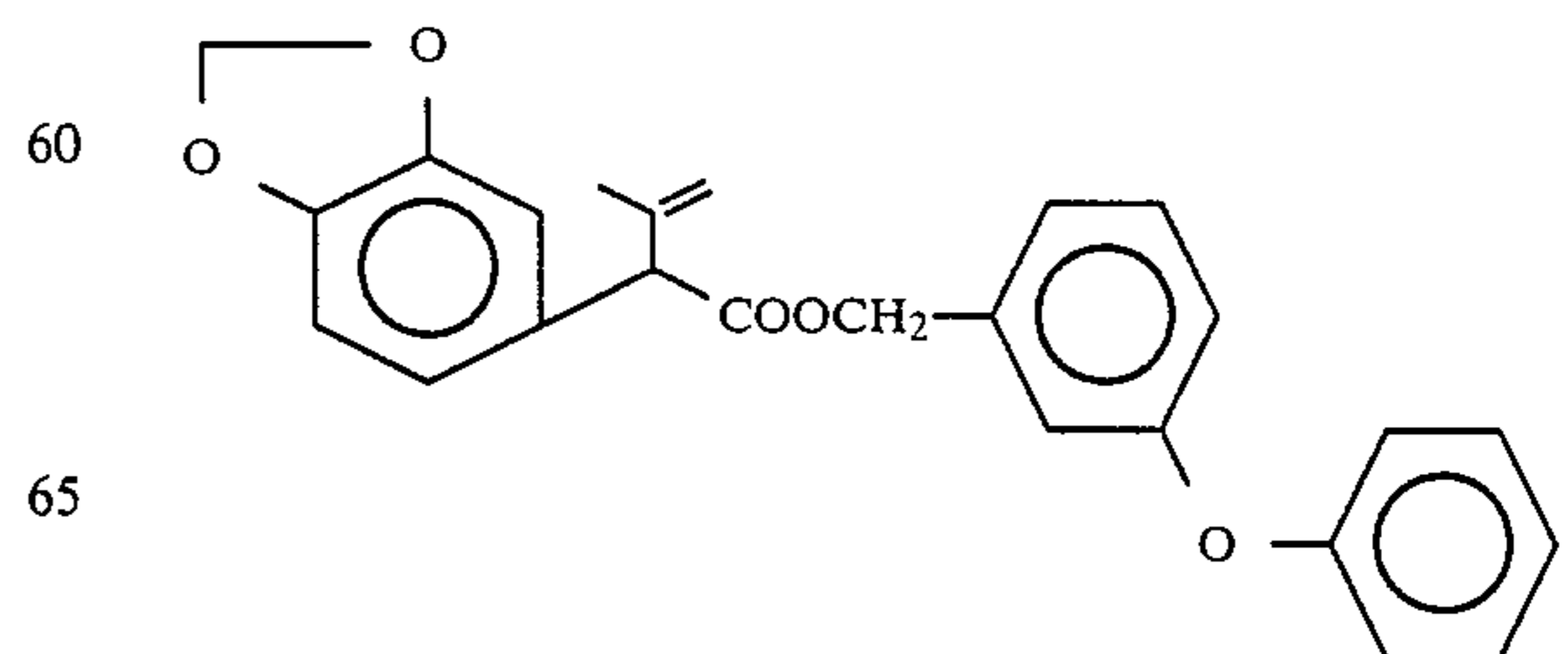
11. An insecticidally or miticidally effective optically active phenyl acetate of the formula



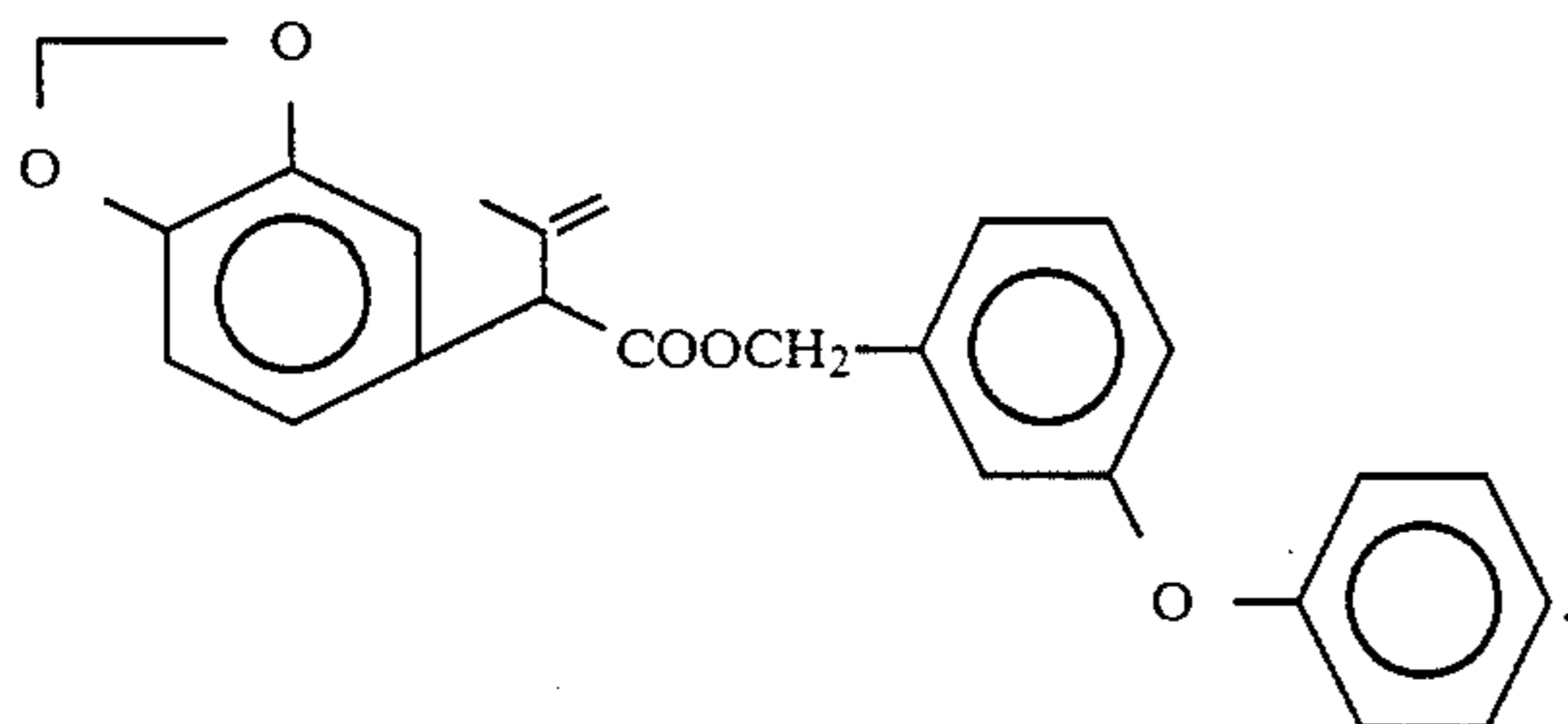
12. An insecticidally or miticidally effective mixture of optical isomers of the formula



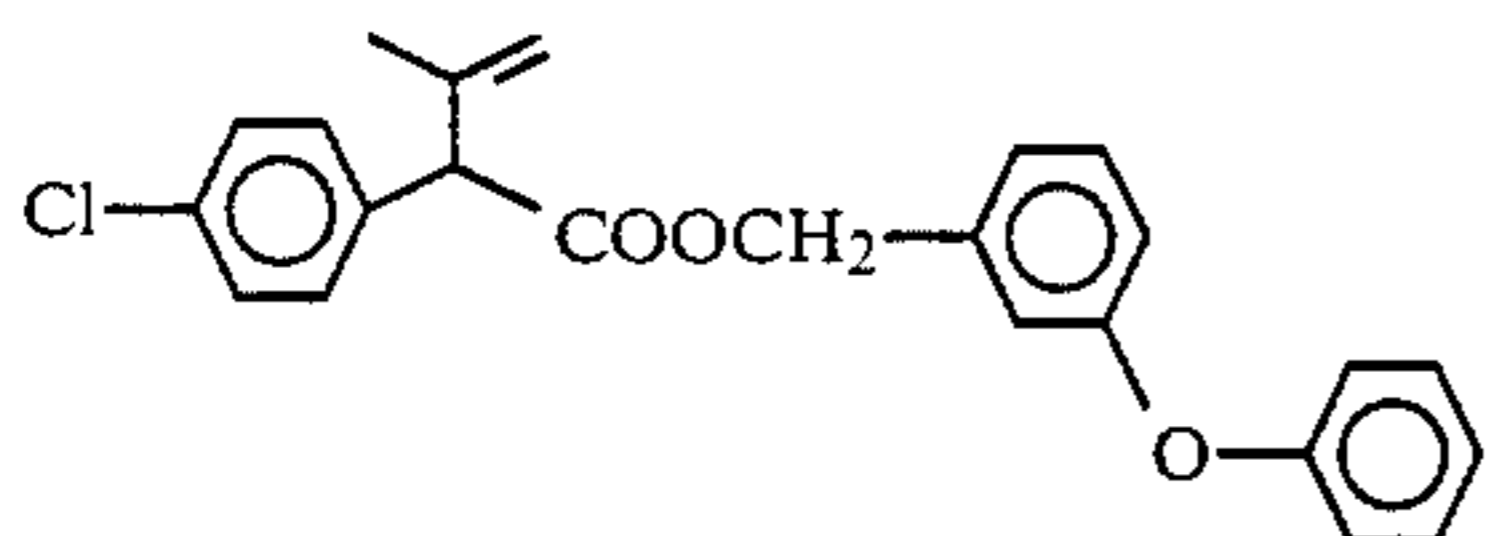
13. An insecticidally or miticidally effective optically active phenyl acetate of the formula



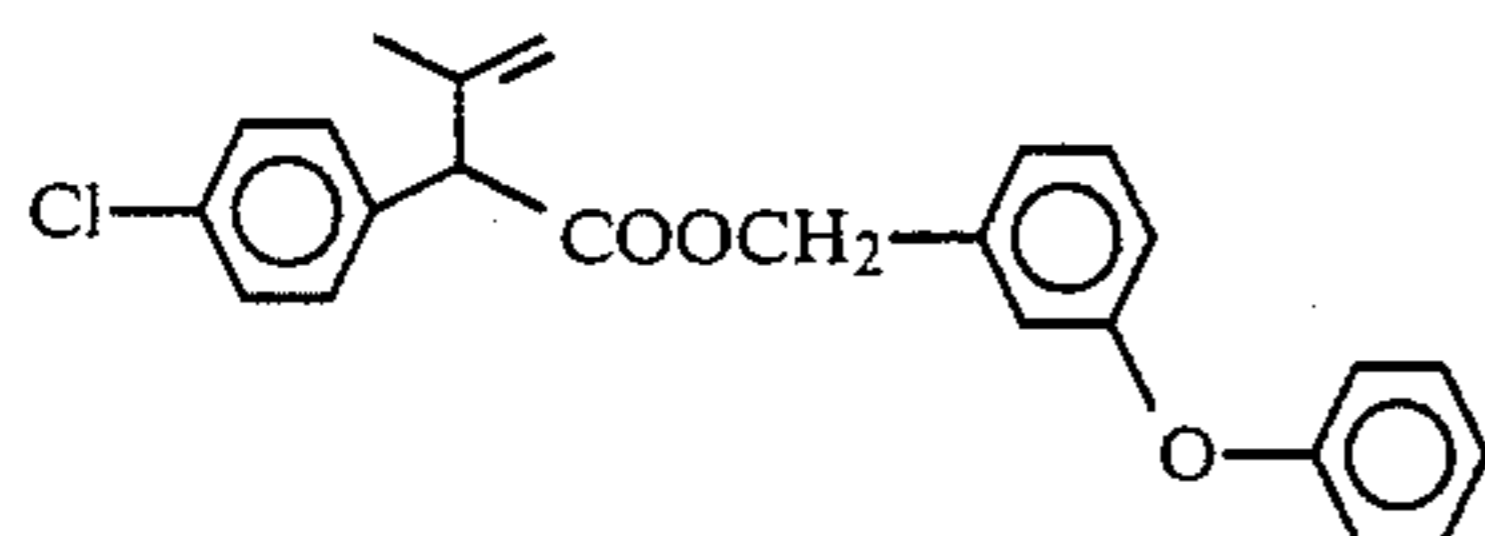
14. An insecticidally or miticidally effective mixture of optical isomers of the formula



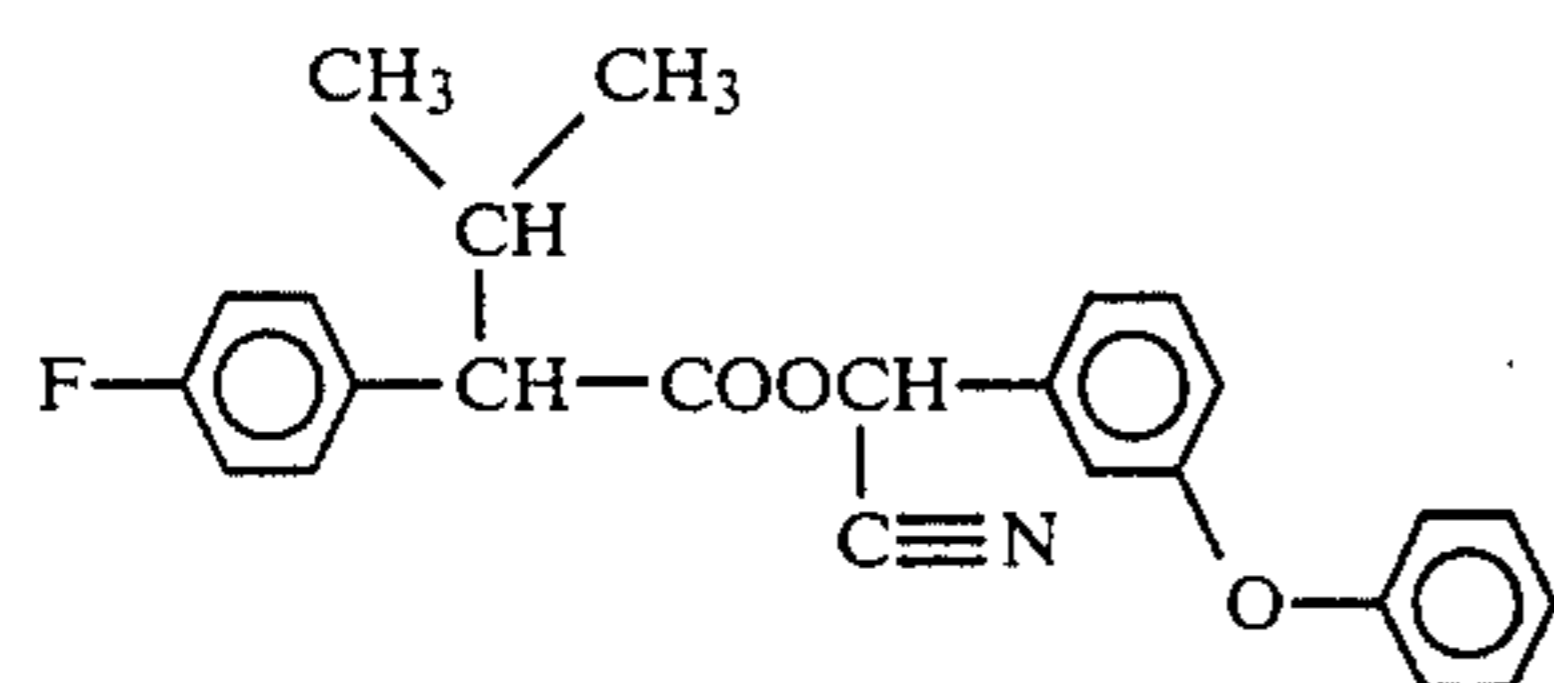
15. An insecticidally or miticidally effective optically active phenyl acetate of the formula



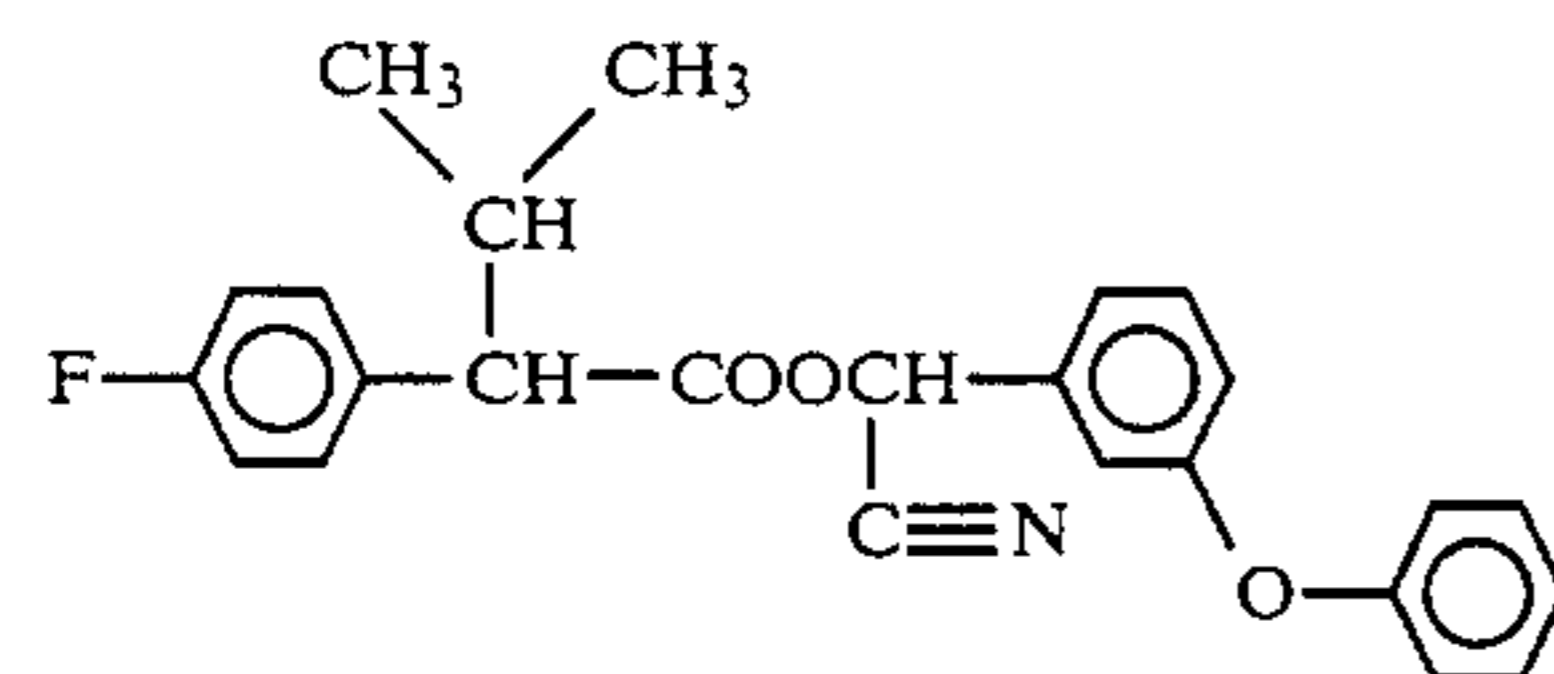
16. An insecticidally or miticidally effective mixture of optical isomers of the formula



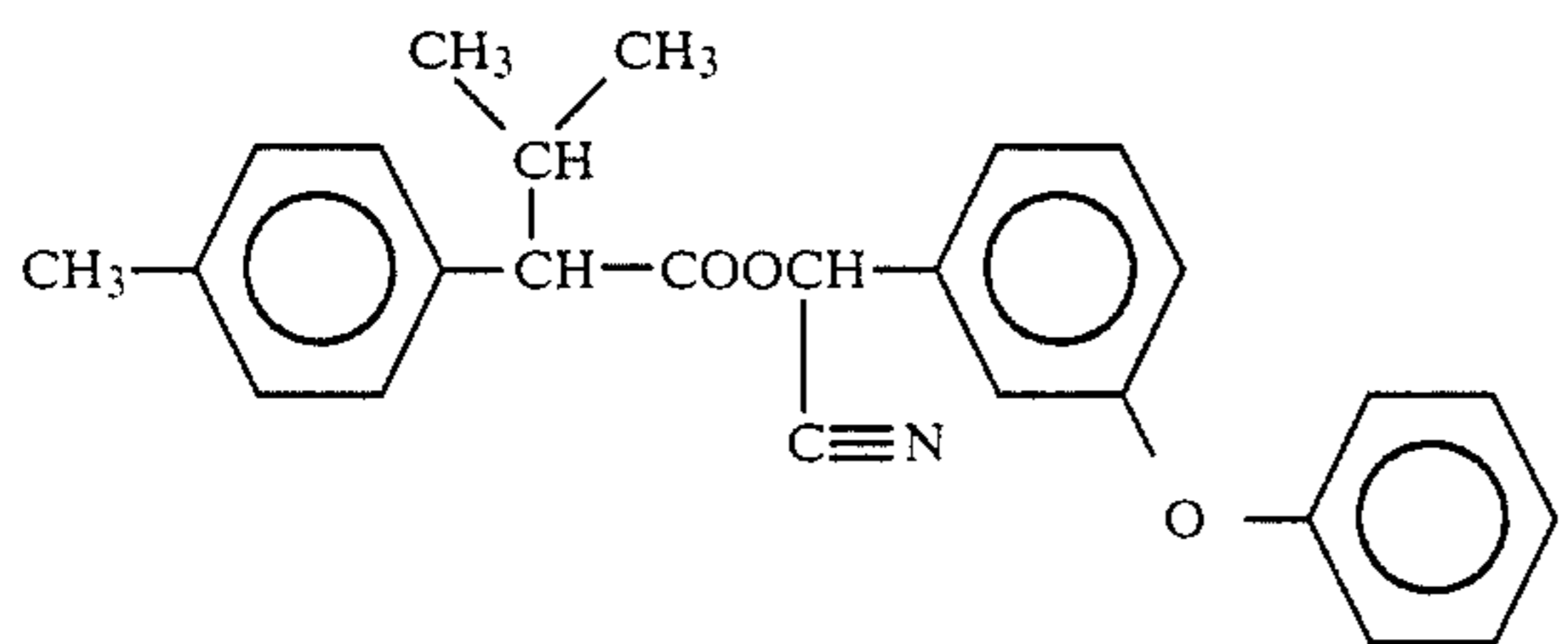
17. An insecticidally or miticidally effective optically active phenyl acetate of the formula



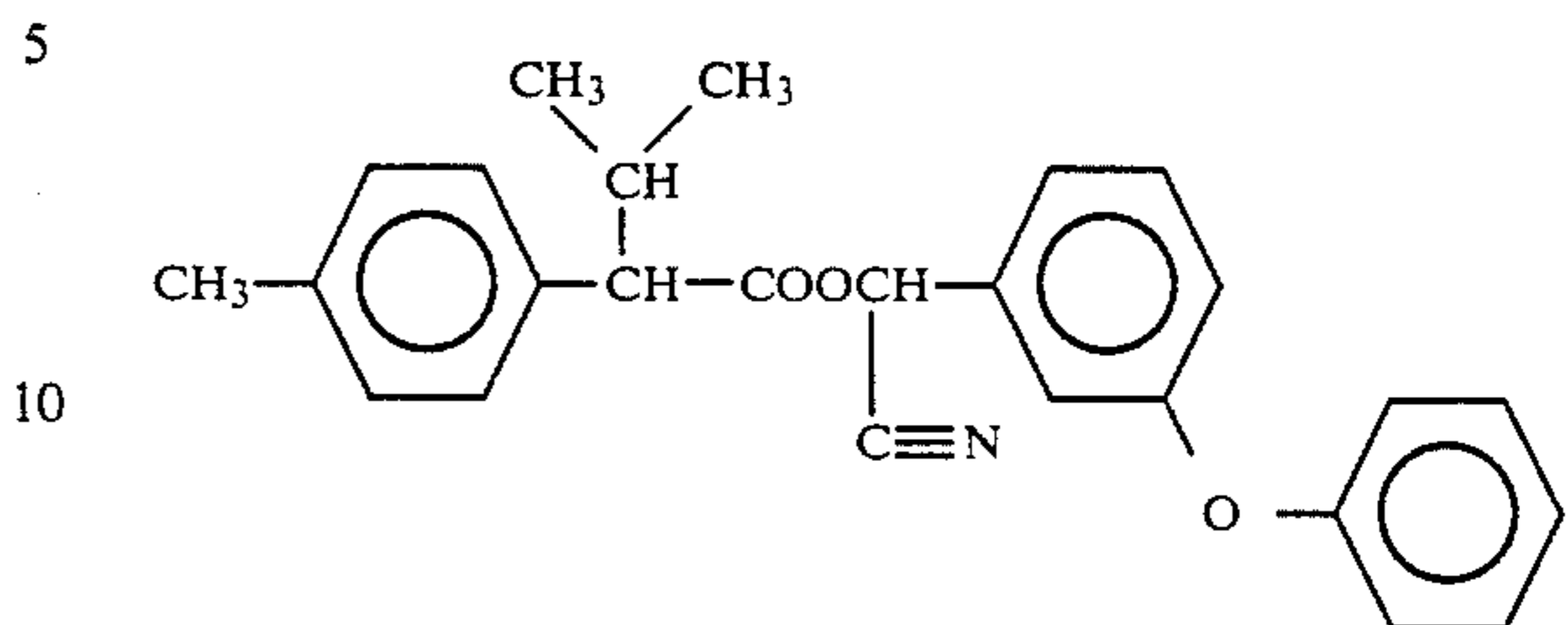
18. An insecticidally or miticidally effective mixture of optical isomers of the formula



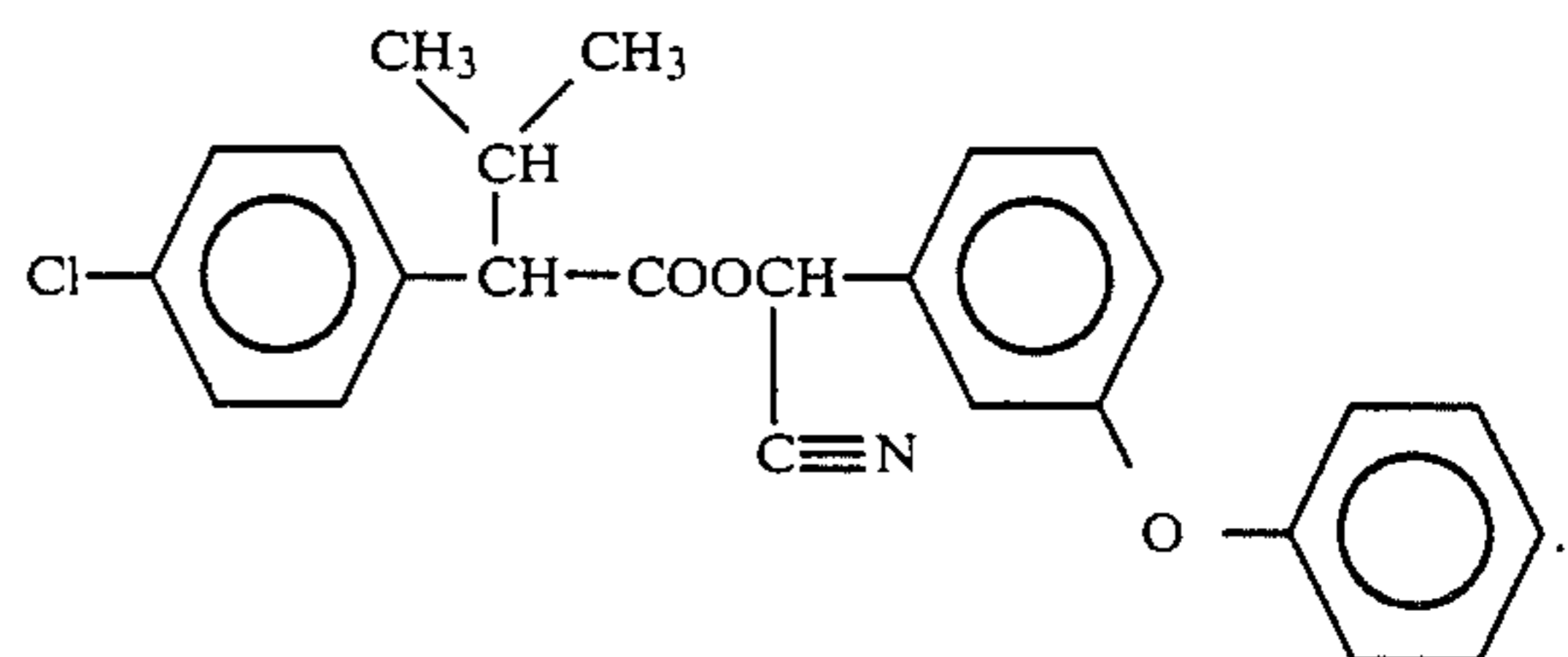
19. An insecticidally or miticidally effective optically active phenyl acetate of the formula



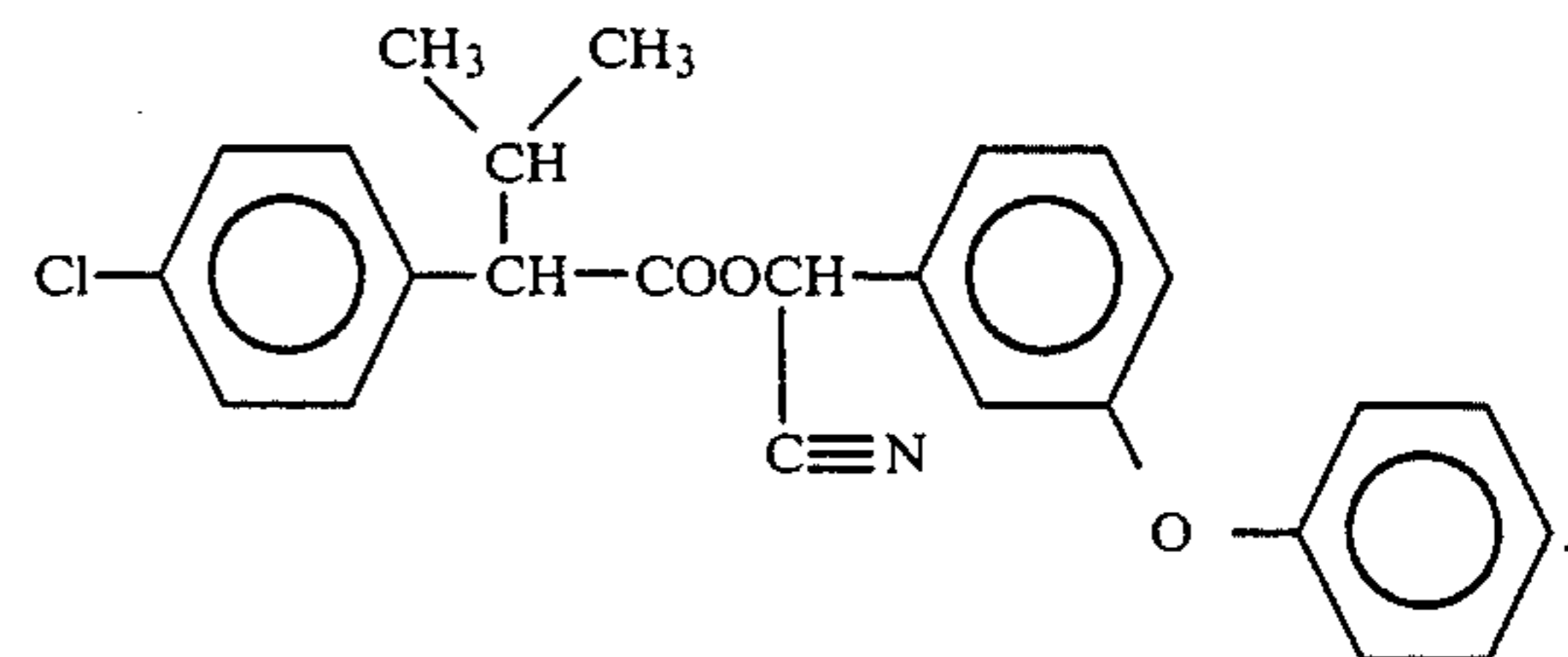
20. An insecticidally or miticidally effective mixture of optical isomers of the formula



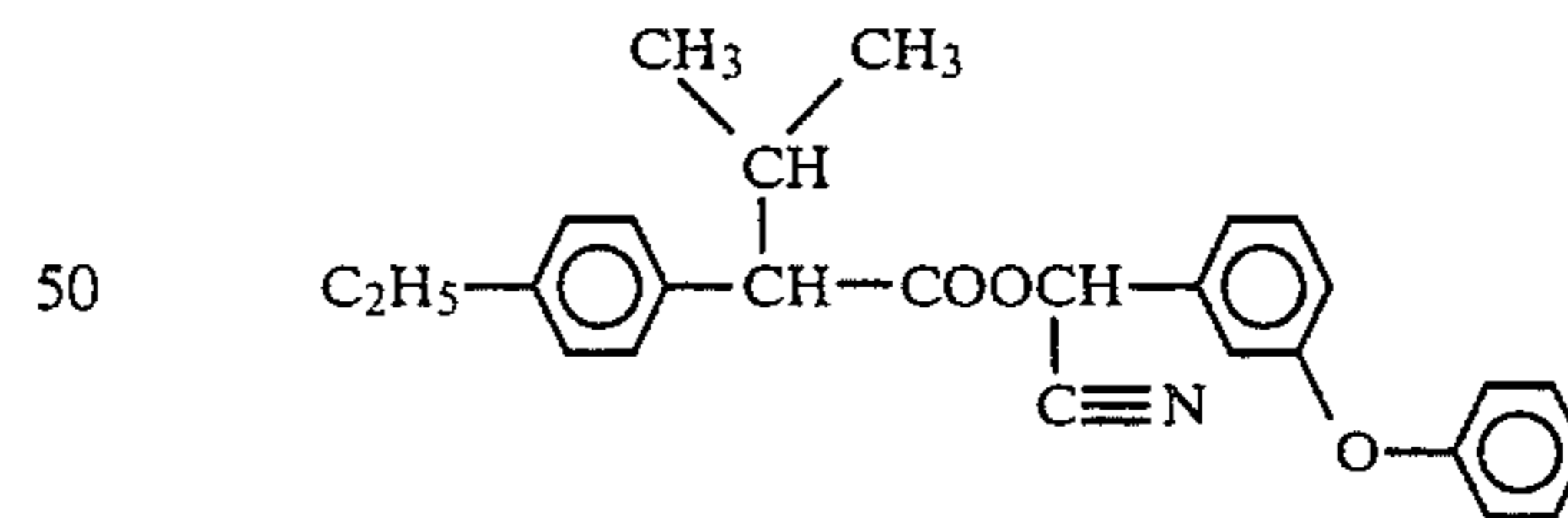
21. An insecticidally or miticidally effective optically active phenyl acetate of the formula



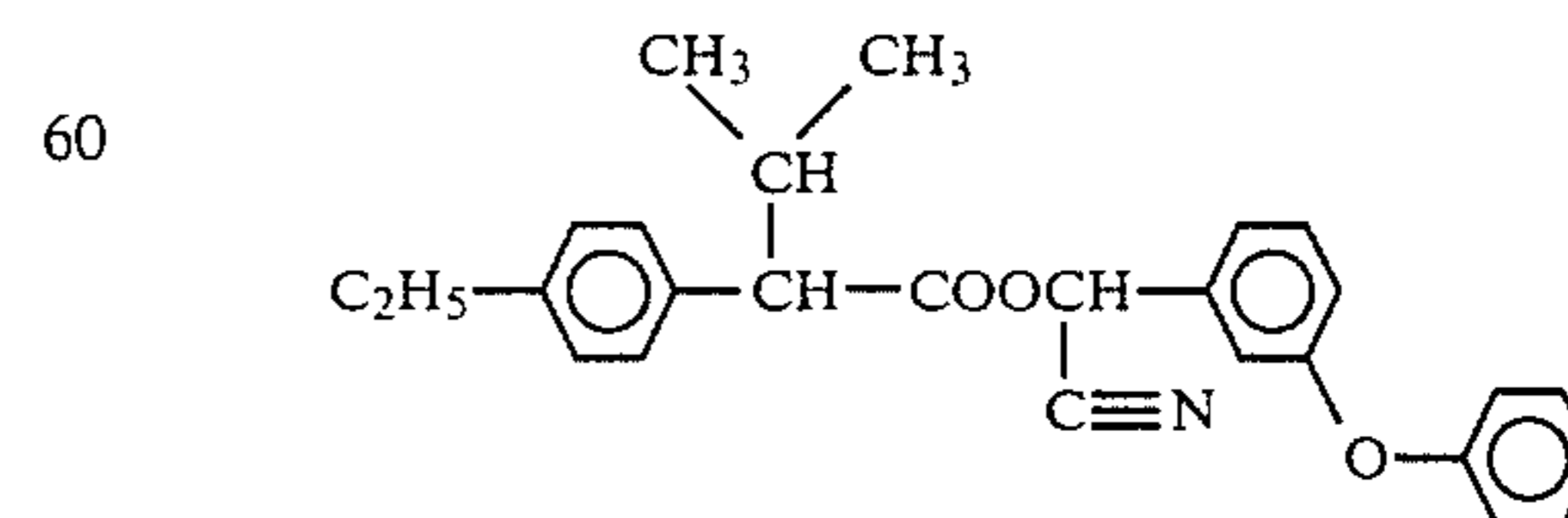
22. An insecticidally or miticidally effective mixture of optical isomers of the formula



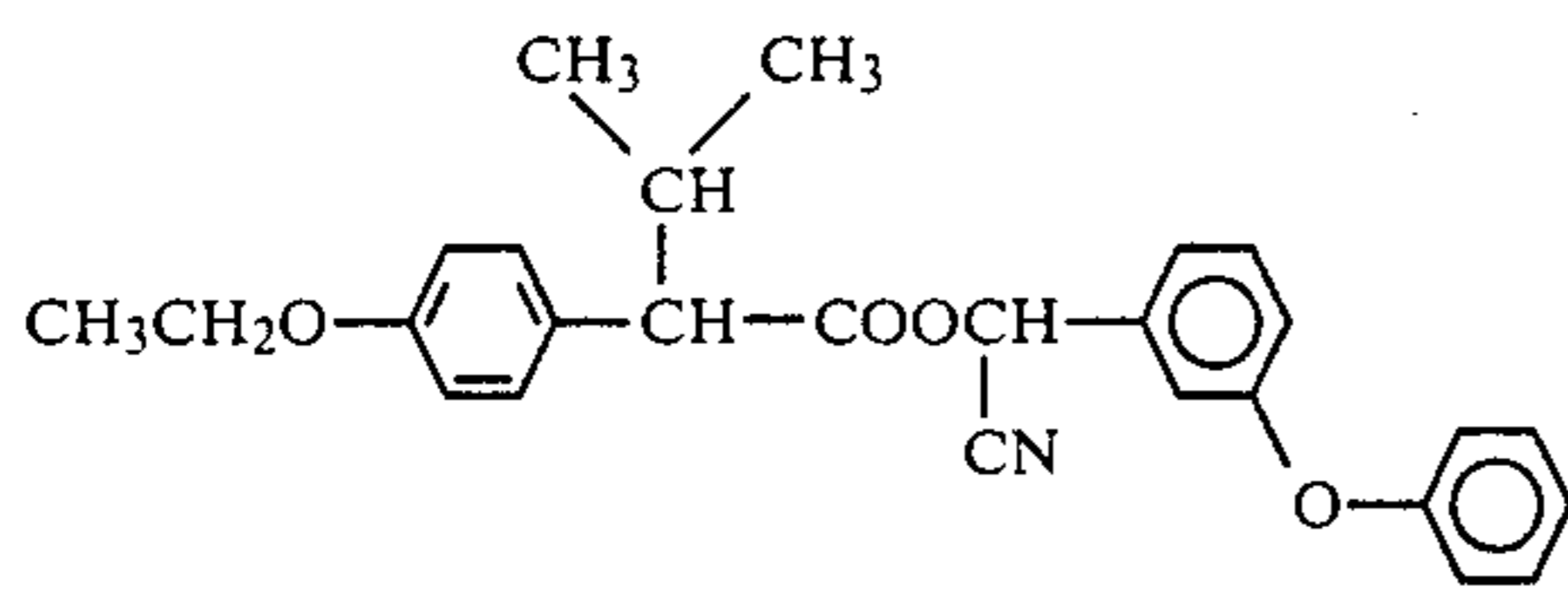
23. An insecticidally or miticidally effective optically active phenyl acetate of the formula



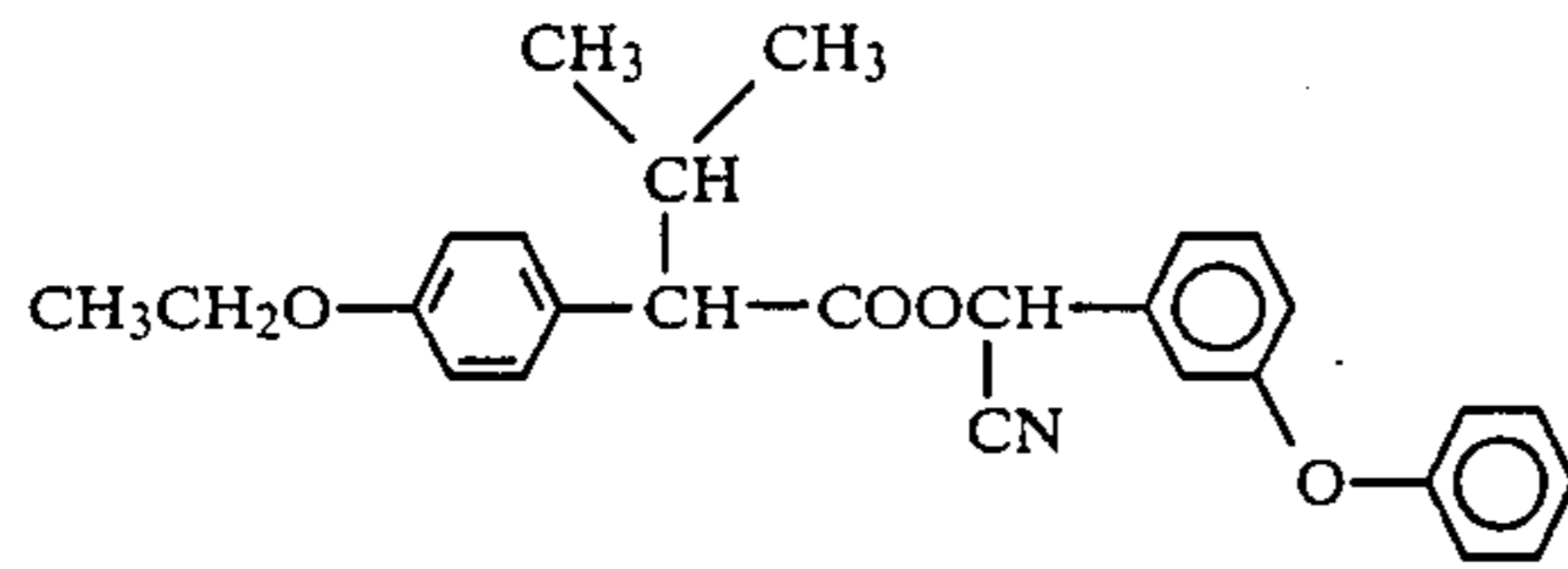
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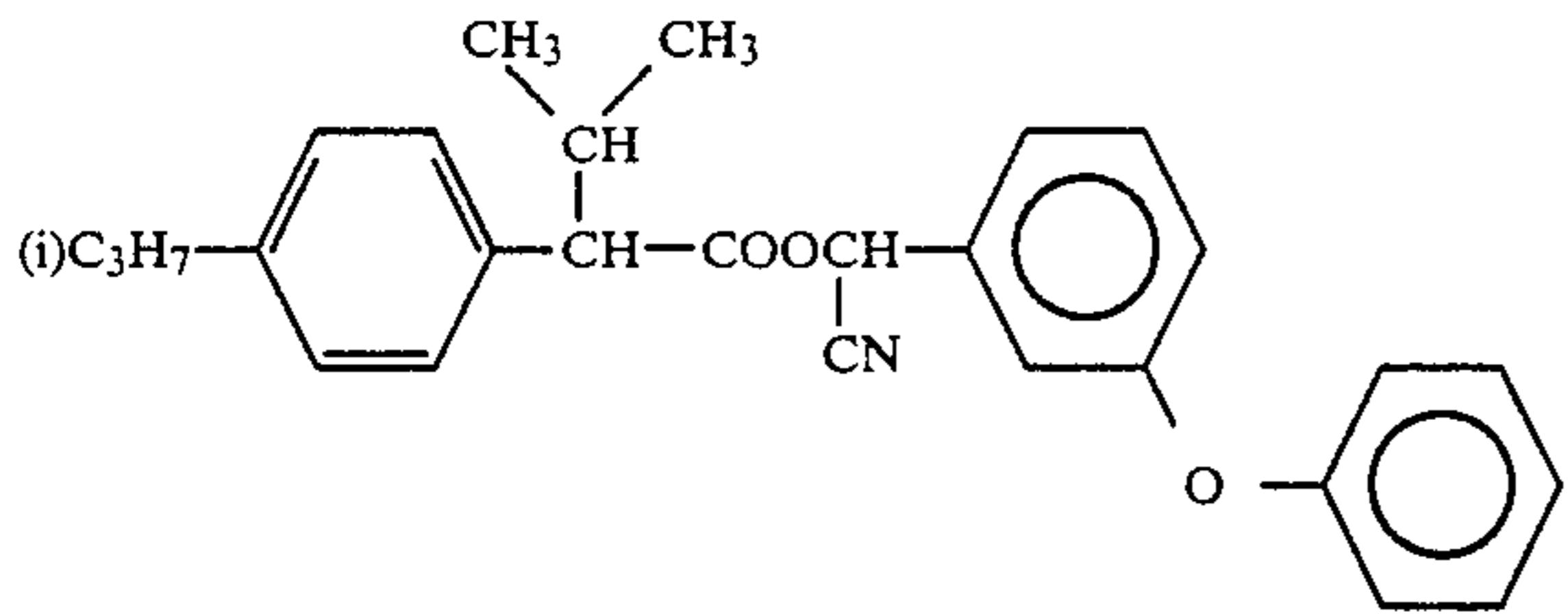
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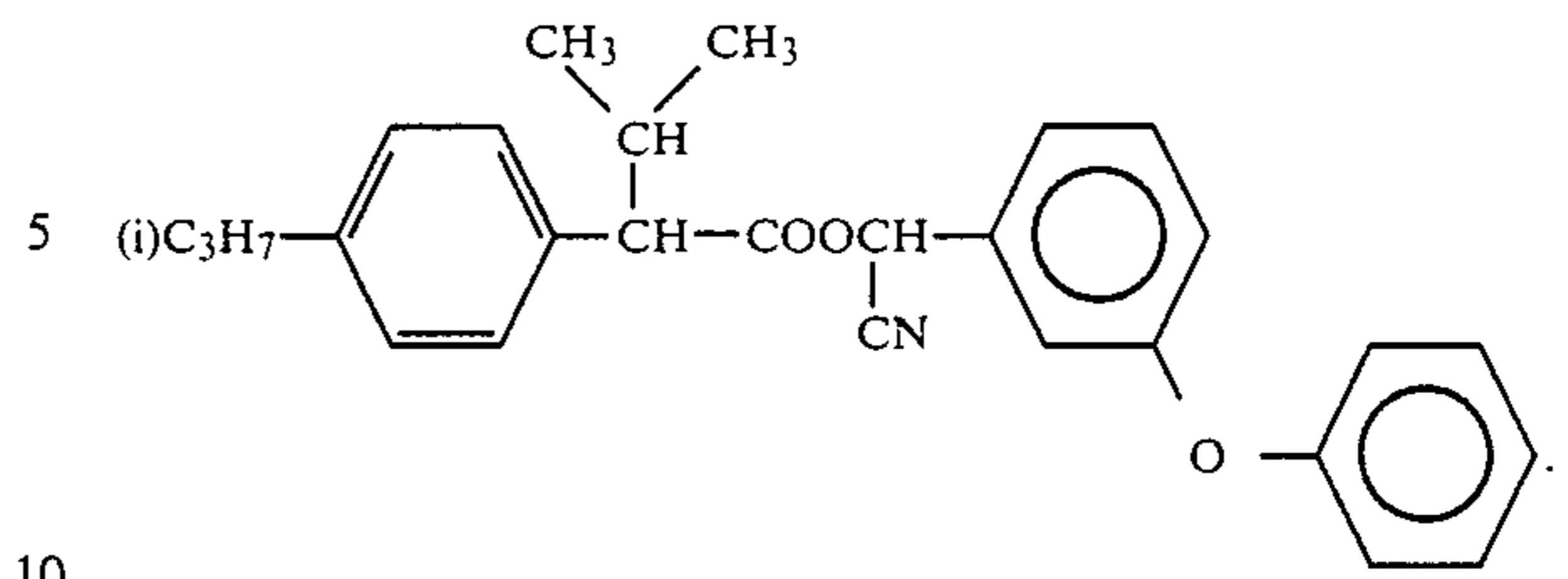
26. An insecticidally or miticidally effective mixture of optical isomers of the formula



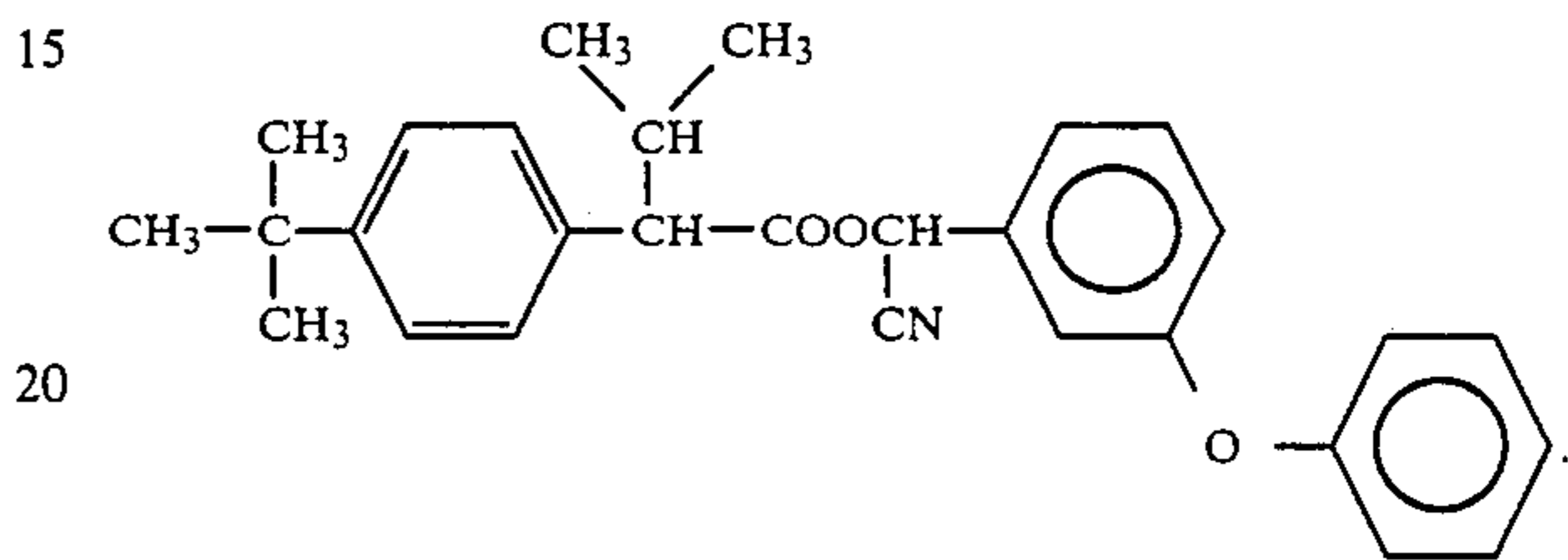
27. An insecticidally or miticidally effective optically active acetate of the formula



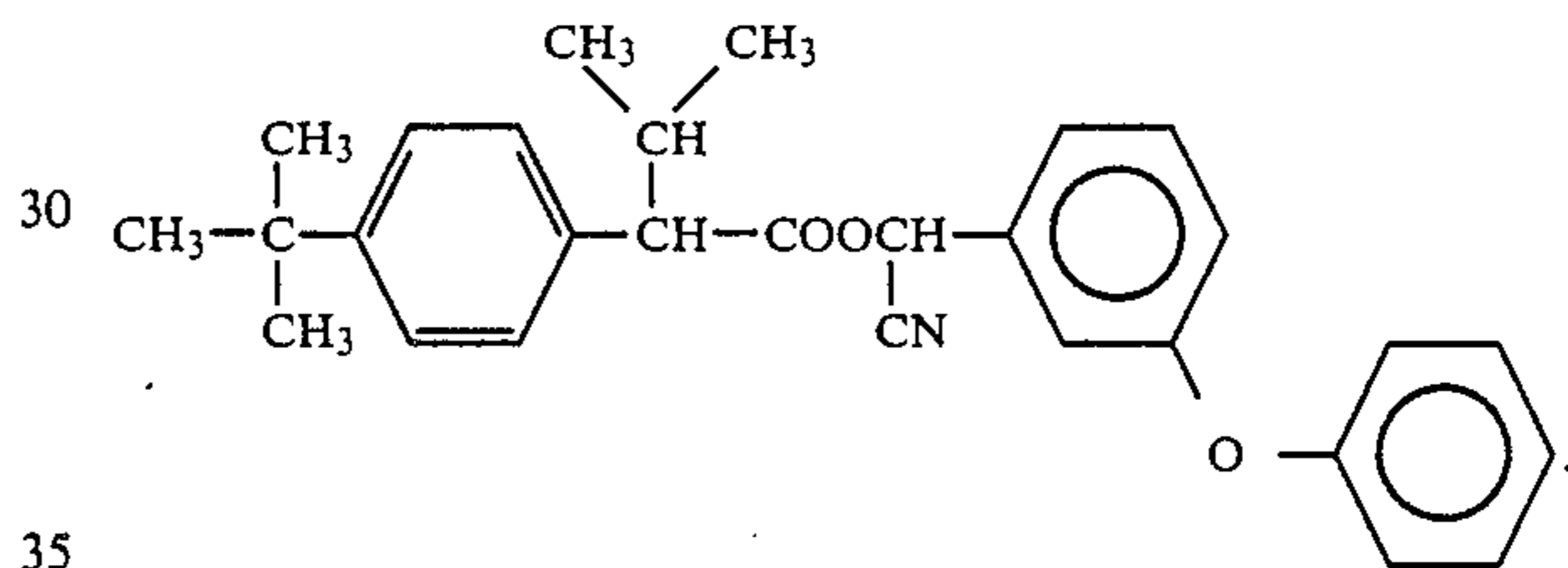
28. An insecticidally or miticidally effective mixture of optical isomers of the formula



29. An insecticidally or miticidally effective optically active phenyl acetate of the formula



30. An insecticidally or miticidally effective mixture of optical isomers of the formula



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