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(54) **INTEGRATED CAPACITOR-LIKE BATTERY AND ASSOCIATED METHOD**

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(73) Assignee: **Inegrated Power Solutions Inc.**

(57) **ABSTRACT**

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(22) Filed: **Mar. 23, 2001**

A method and system for fabricating solid-state energy-storage and energy-conversion devices including fabrication of films for devices without an anneal step, especially for the fabrication of supercapacitors and photovoltaic cells. A film is fabricated by depositing a first material layer to a location. Energy is supplied directly to the material forming the film. The energy can be in the form of energized ions of a second material. Supplying energy directly to the material and/or the film being deposited assists the growth of the crystalline structure of the film and controls stoichiometry.

Related U.S. Application Data

(63) Non-provisional of provisional application No. 60/191,774, filed on Mar. 24, 2000. Non-provisional of provisional application No. 60/225,134, filed on

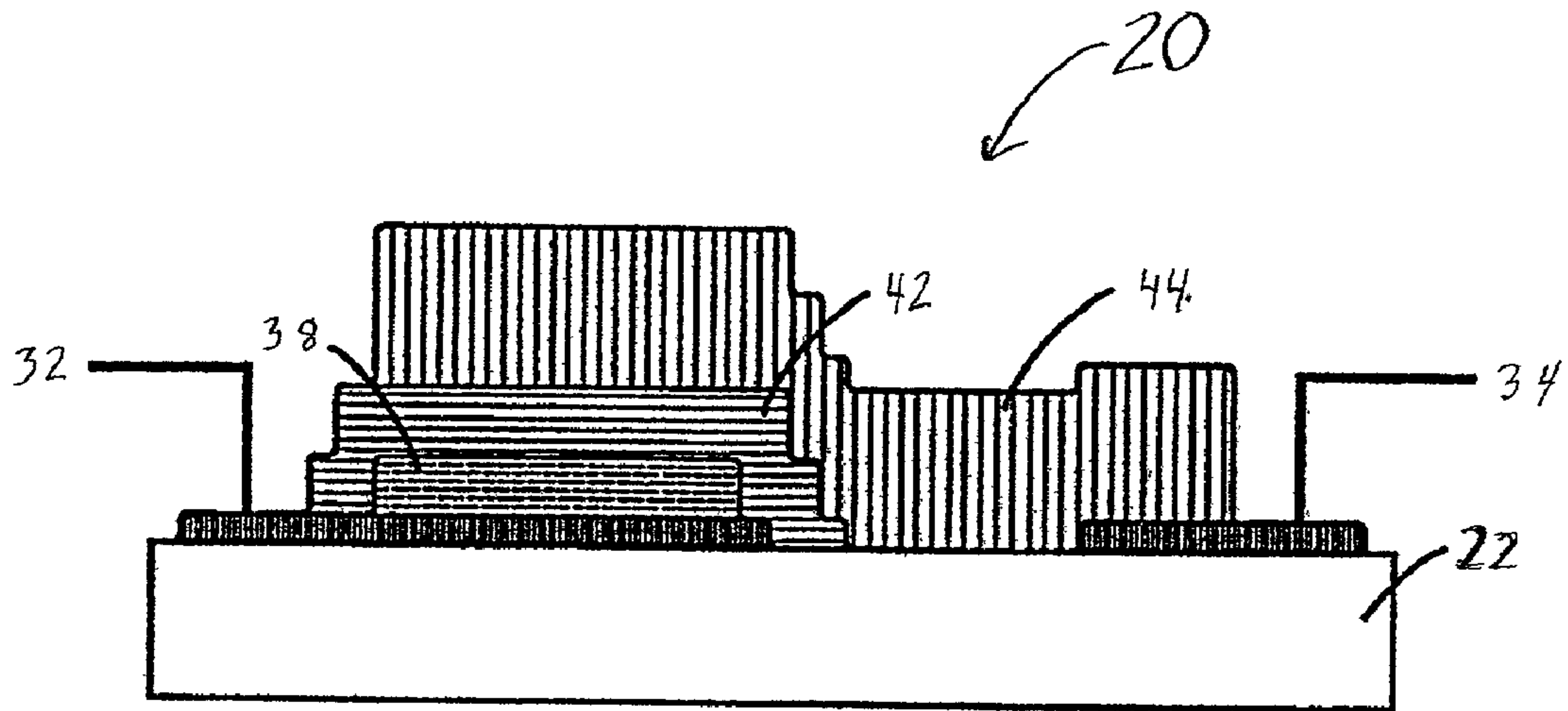


Fig. 1A Prior Art

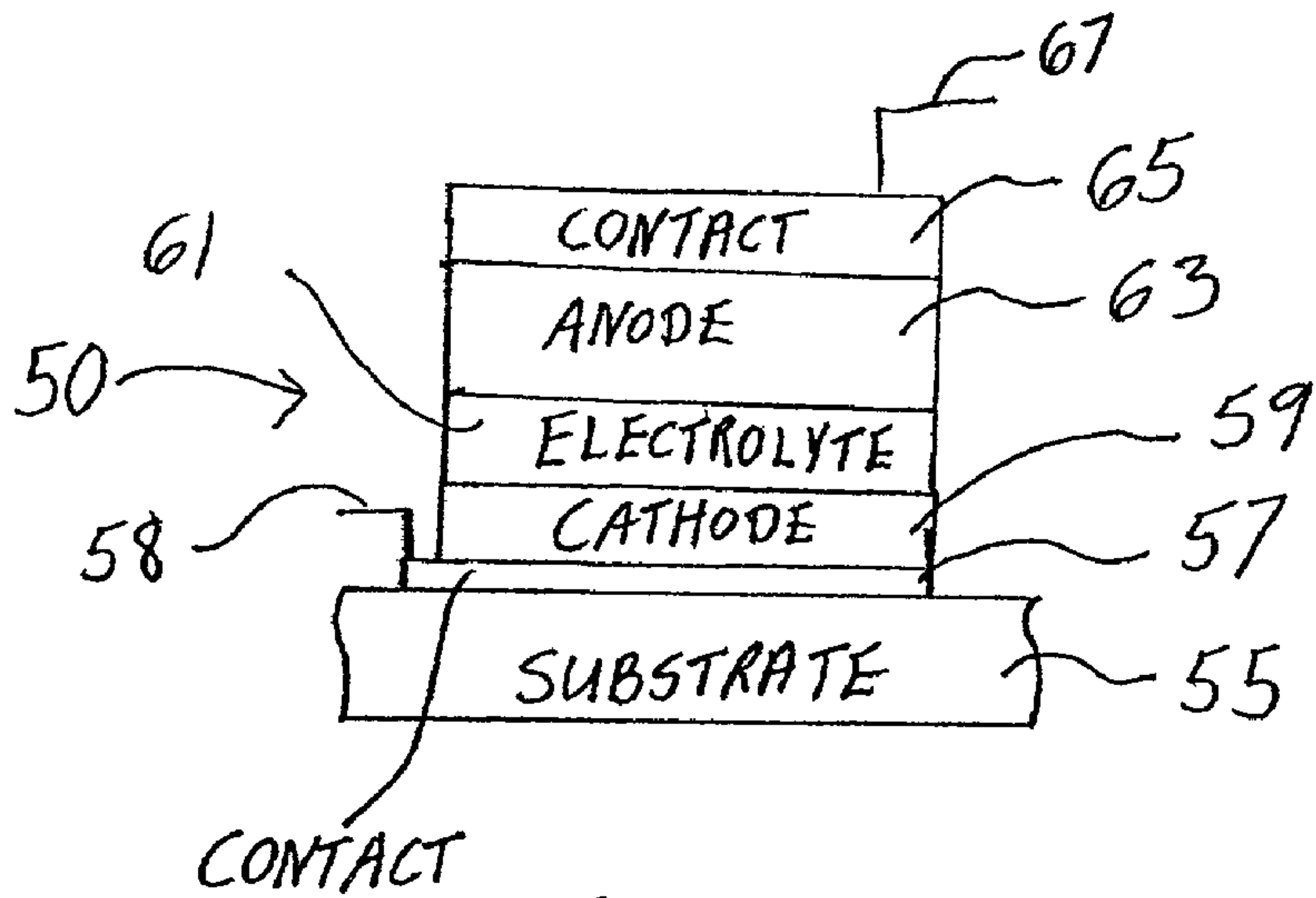


Fig. 1B

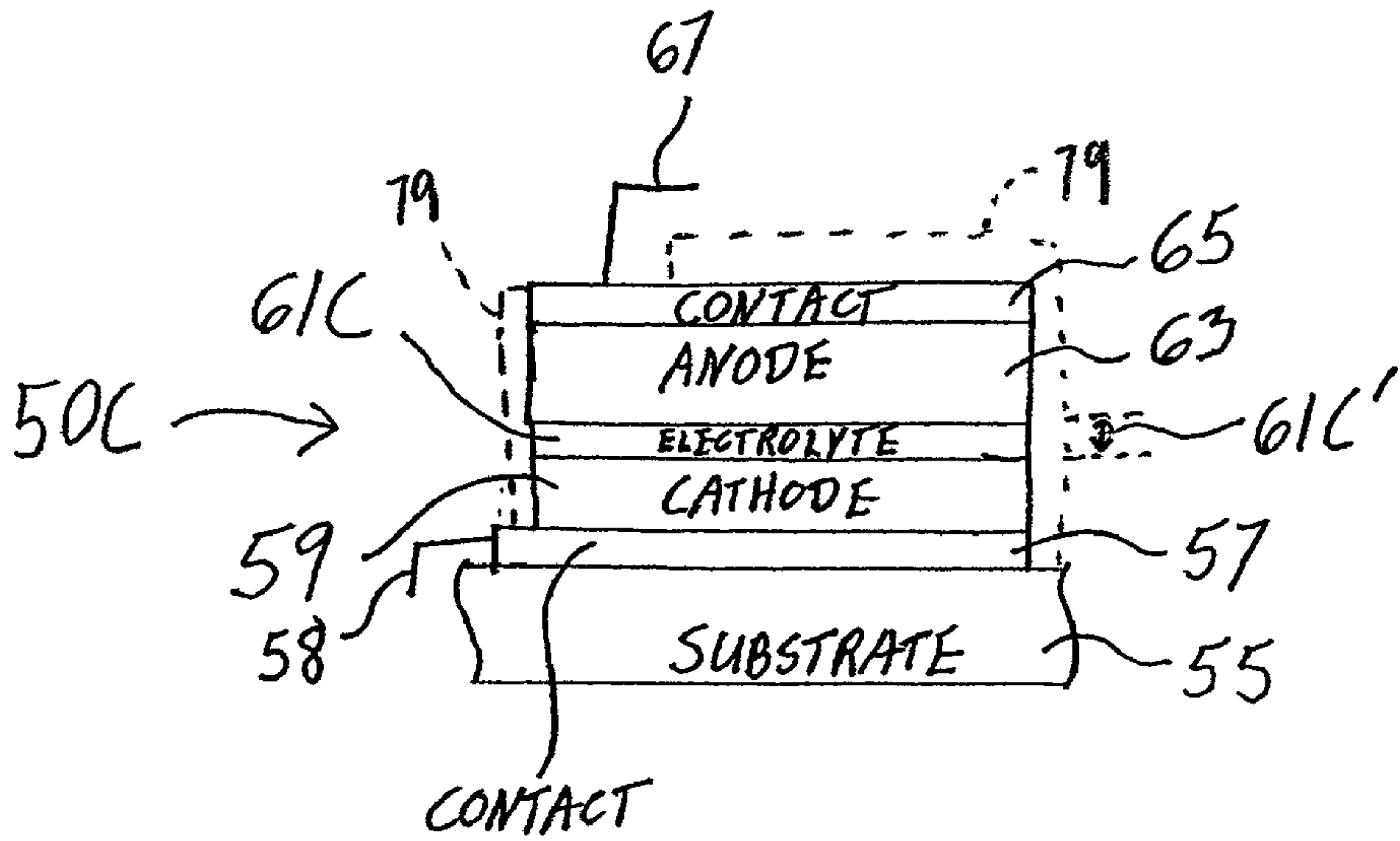


Fig. 1C

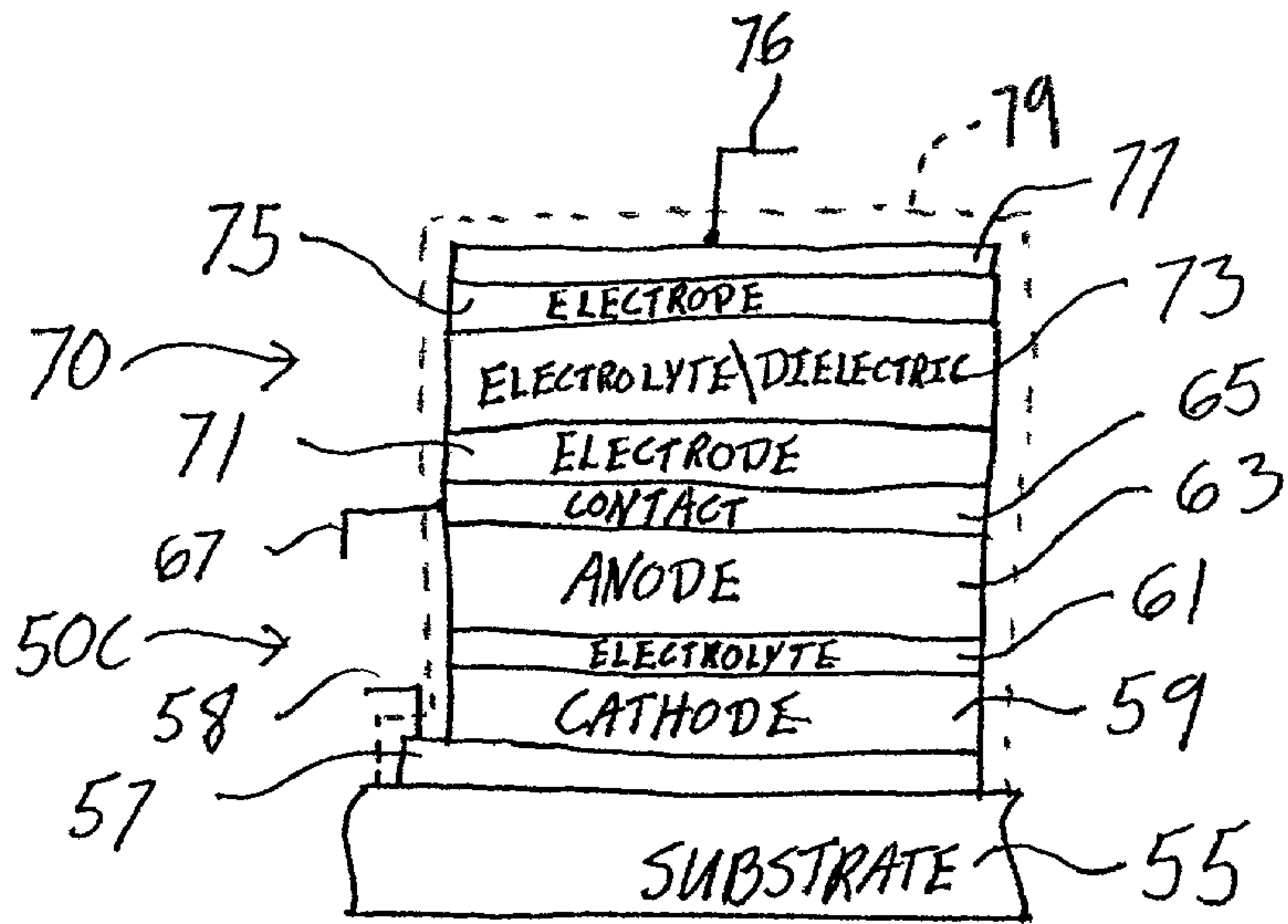


Fig. 1D

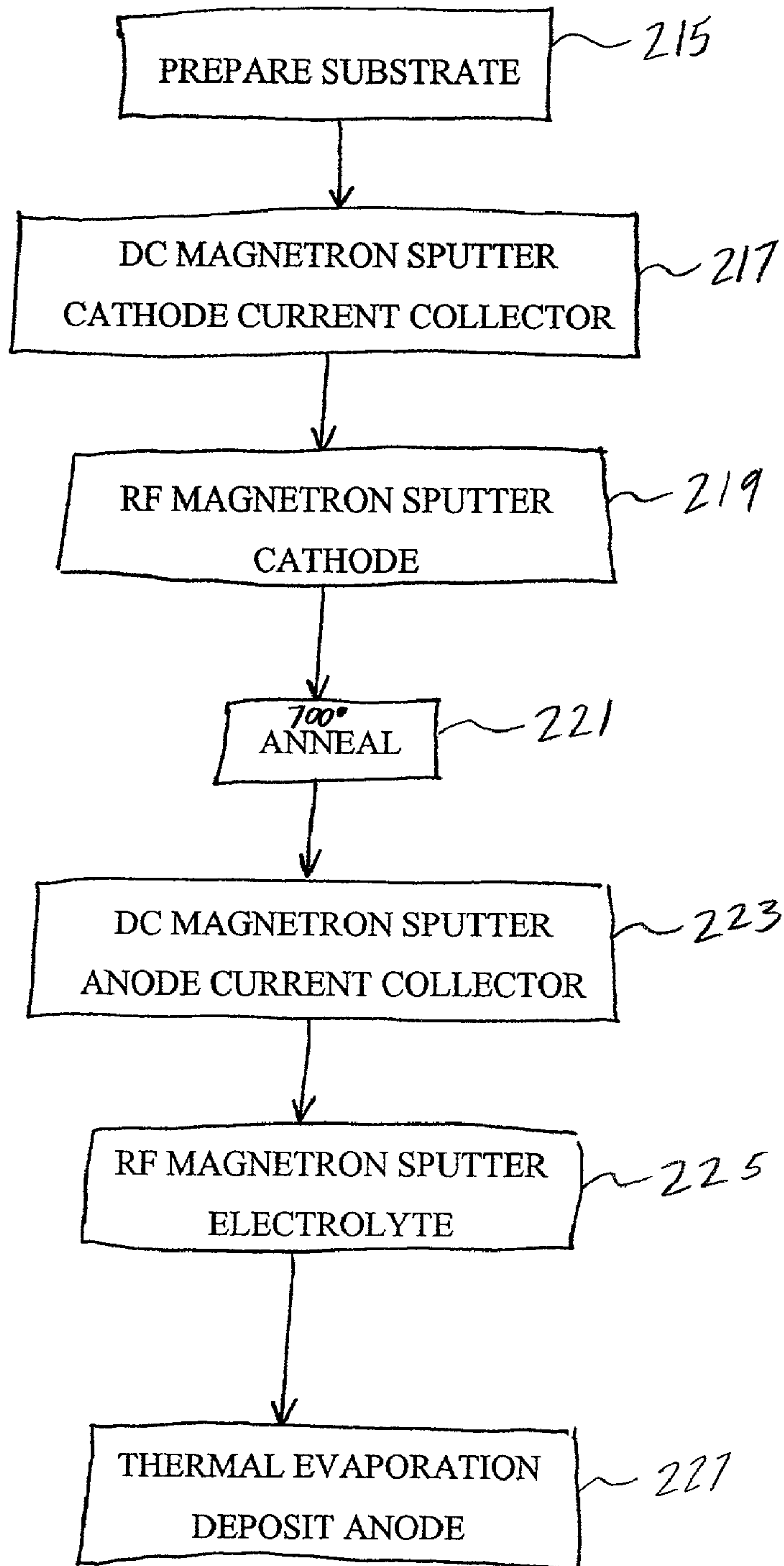


Fig. 2A Prior Art

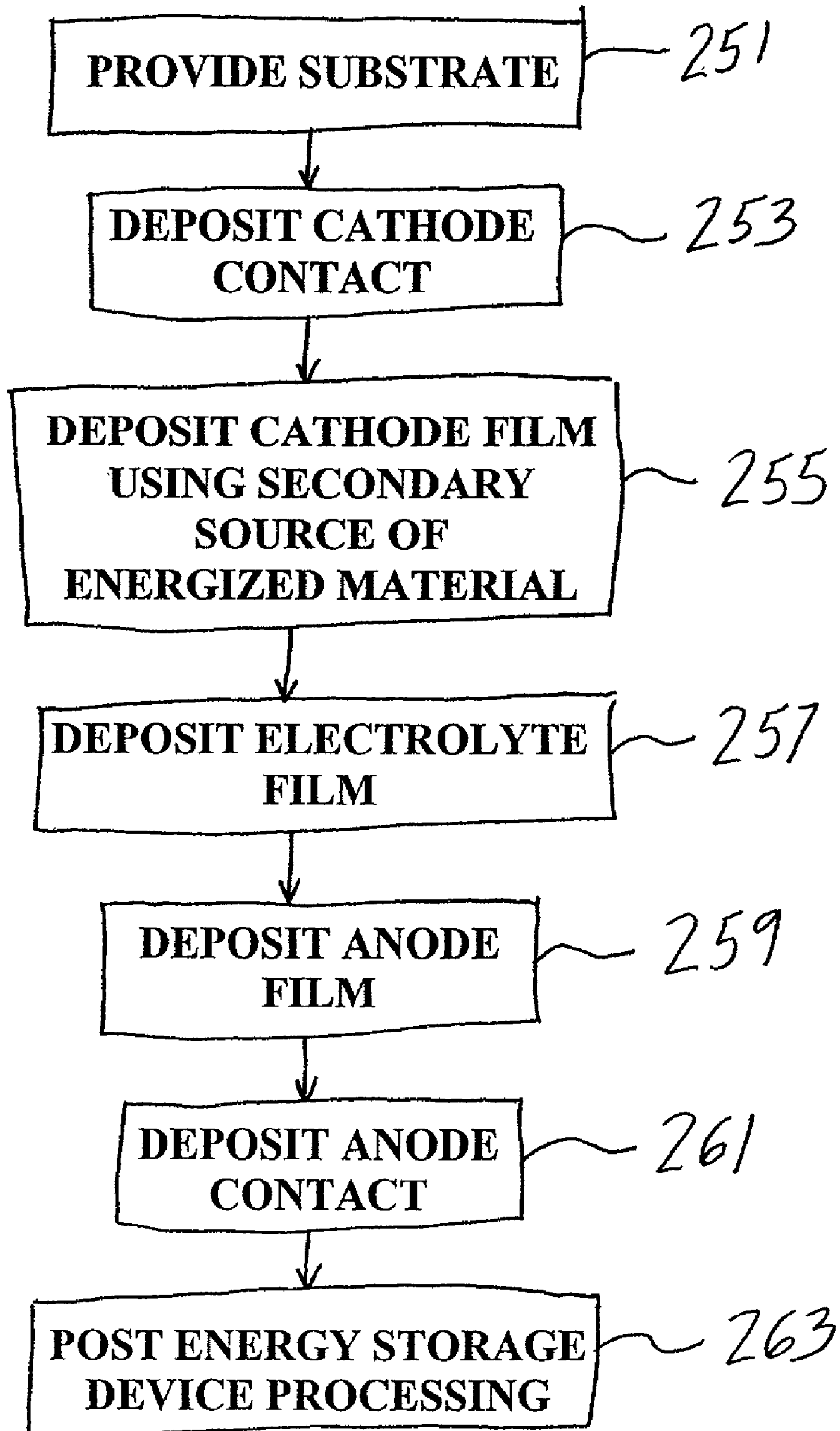


FIGURE 2B

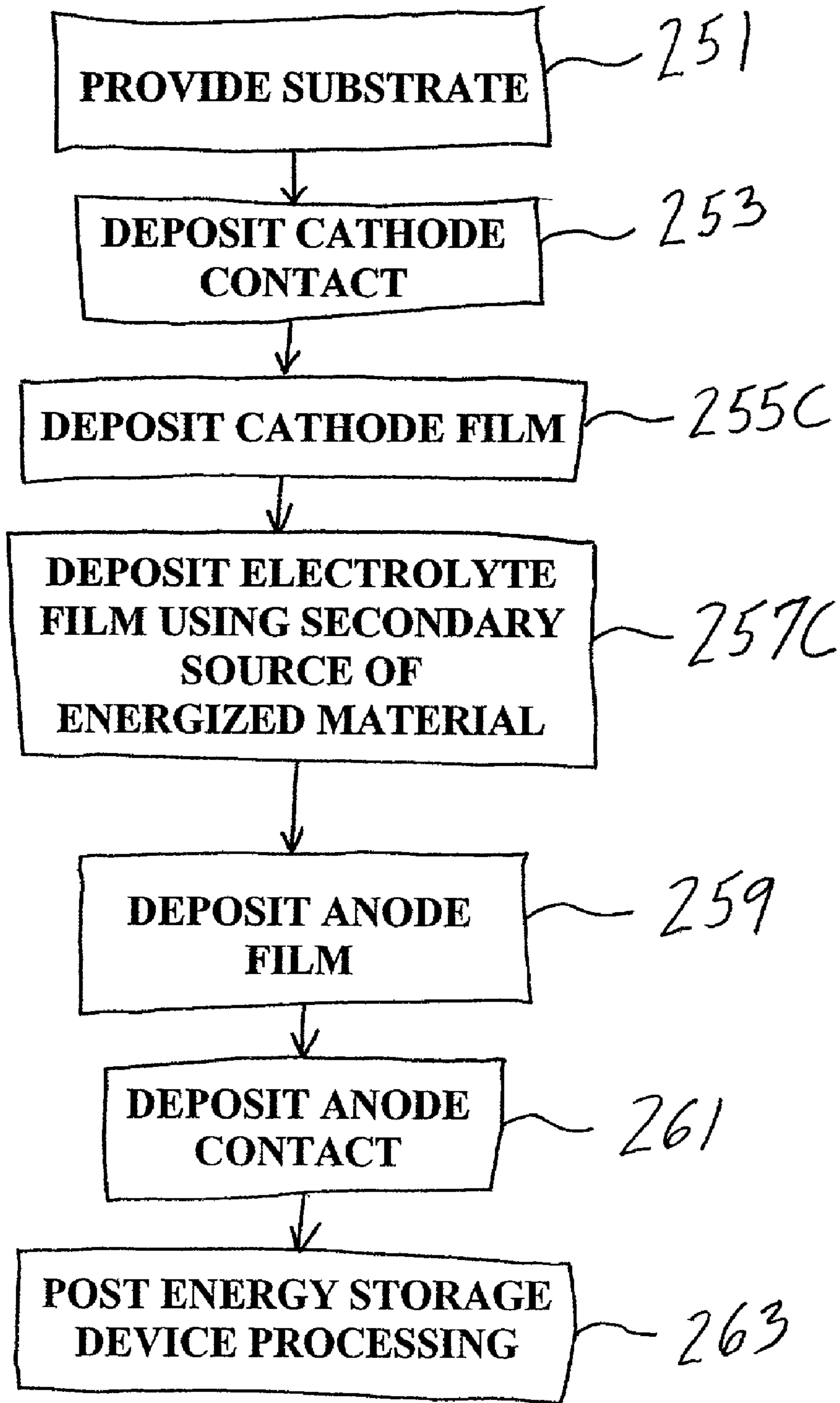


FIGURE 2C

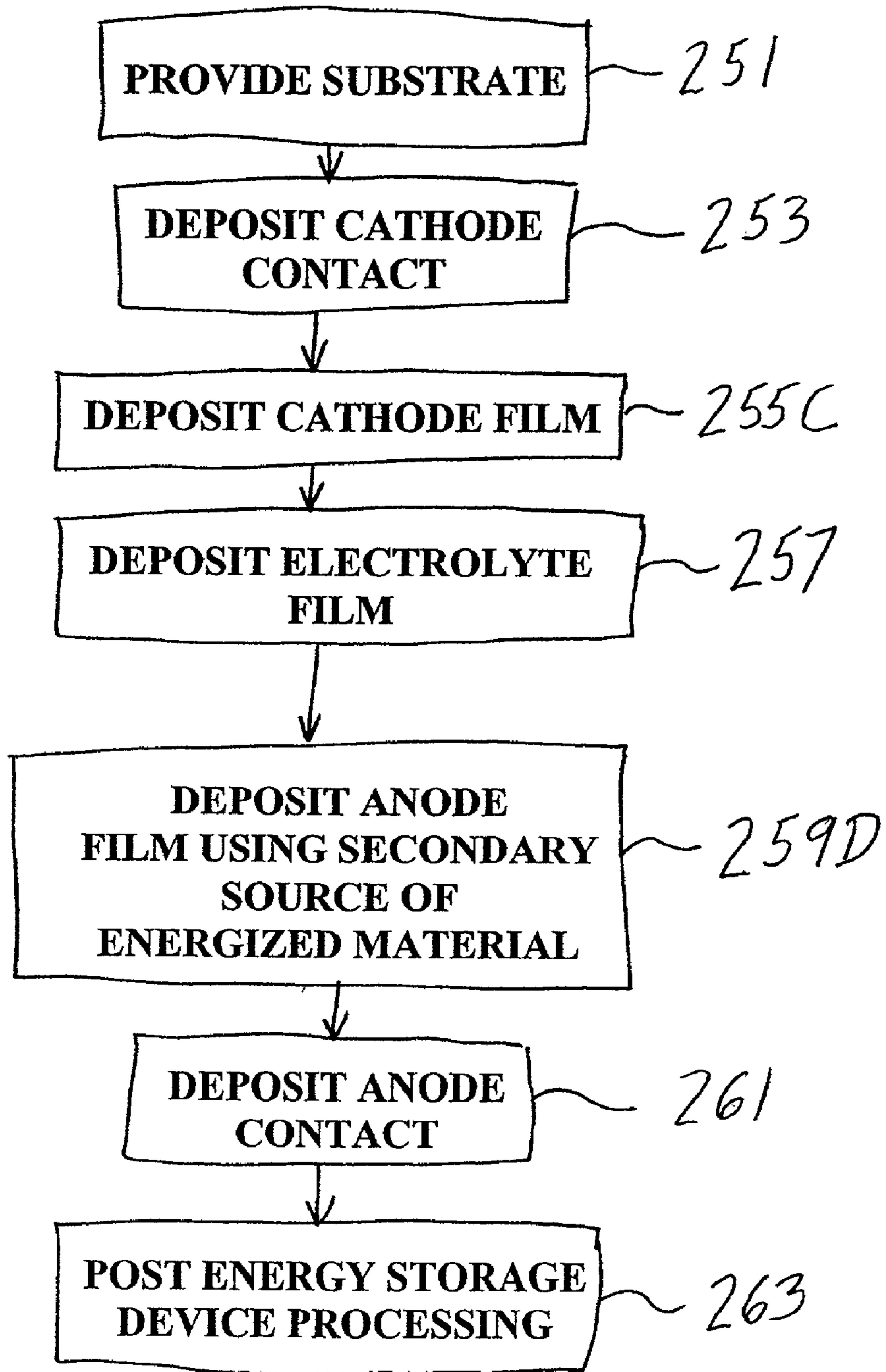
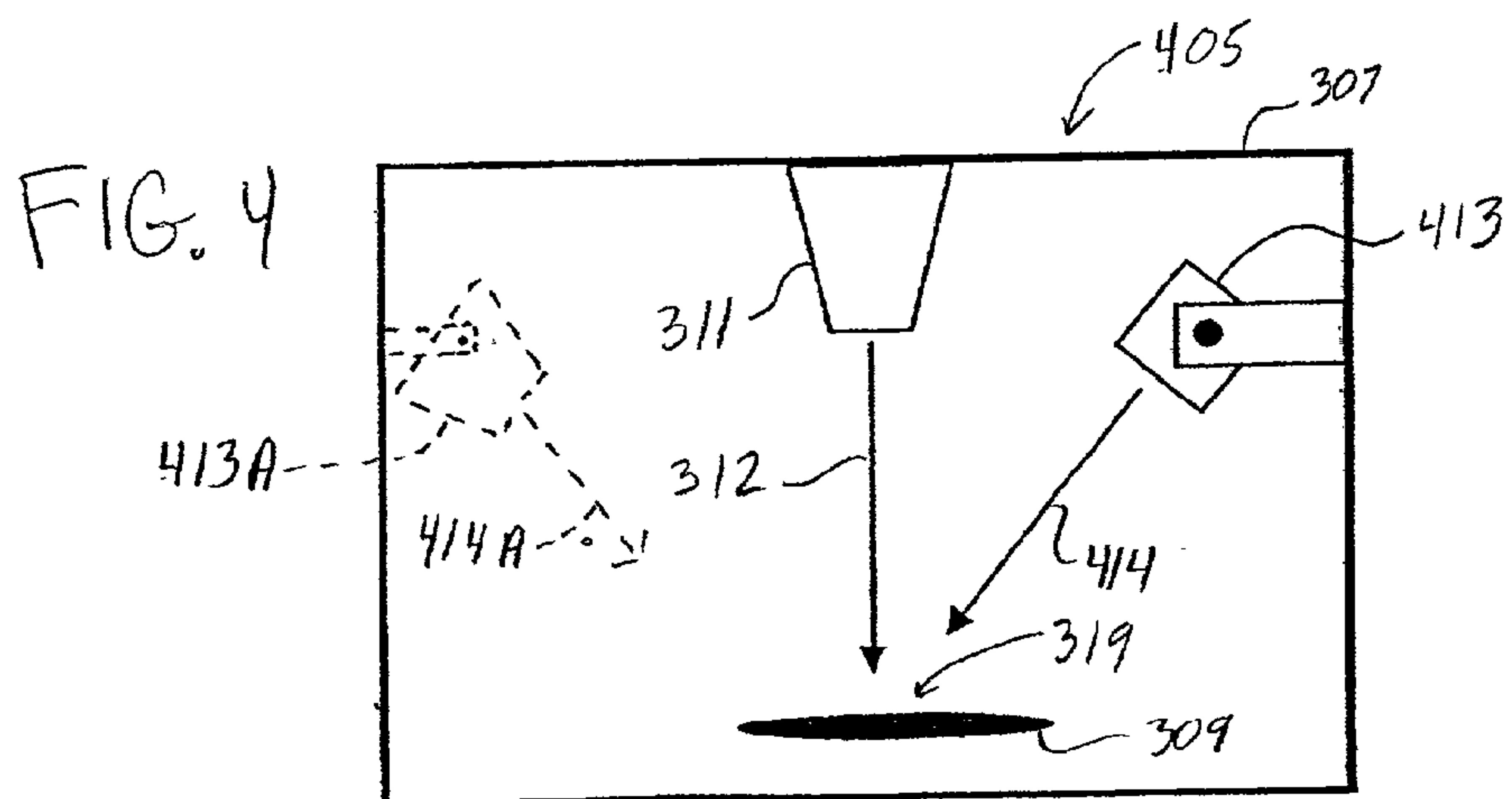
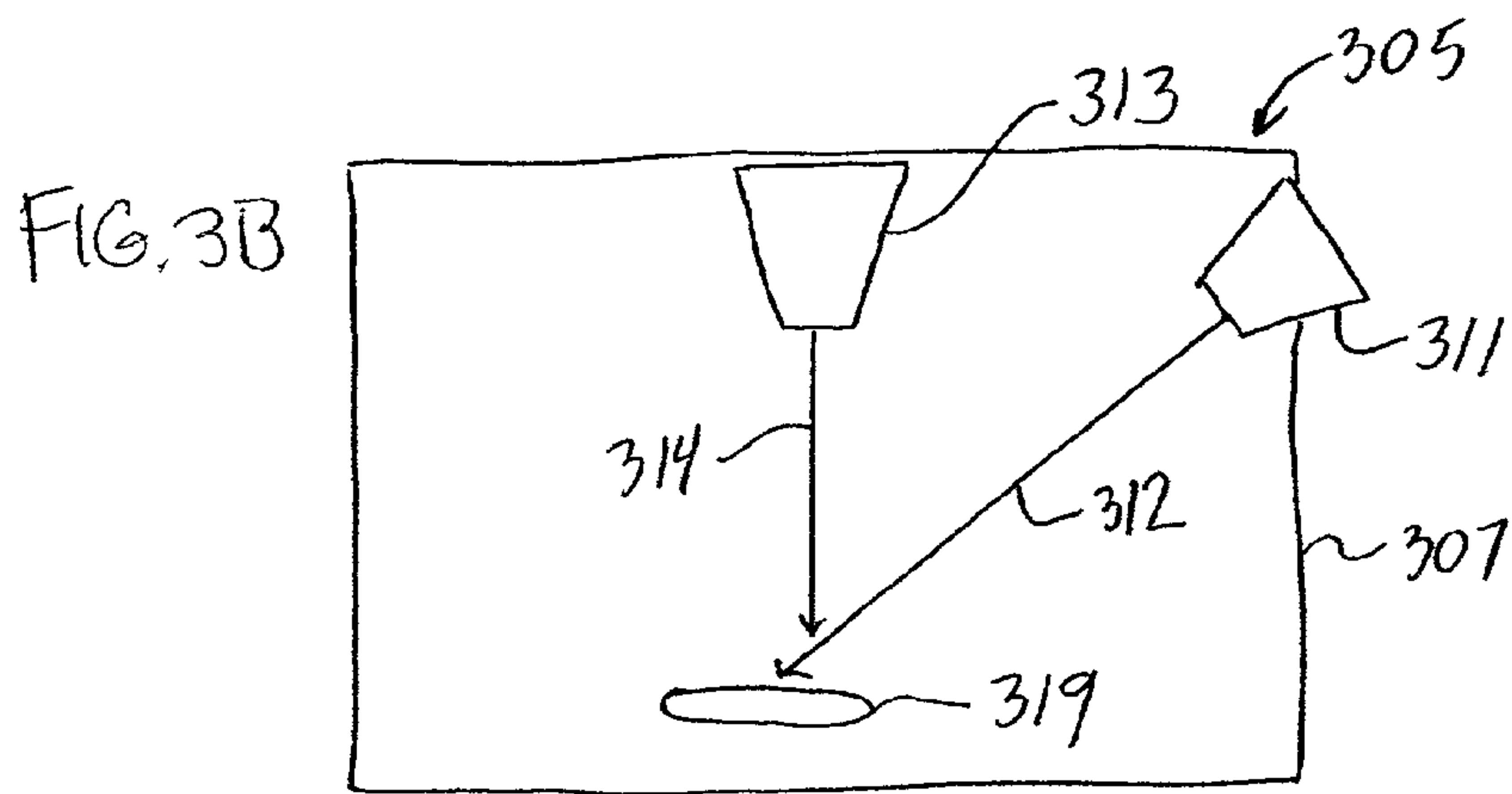
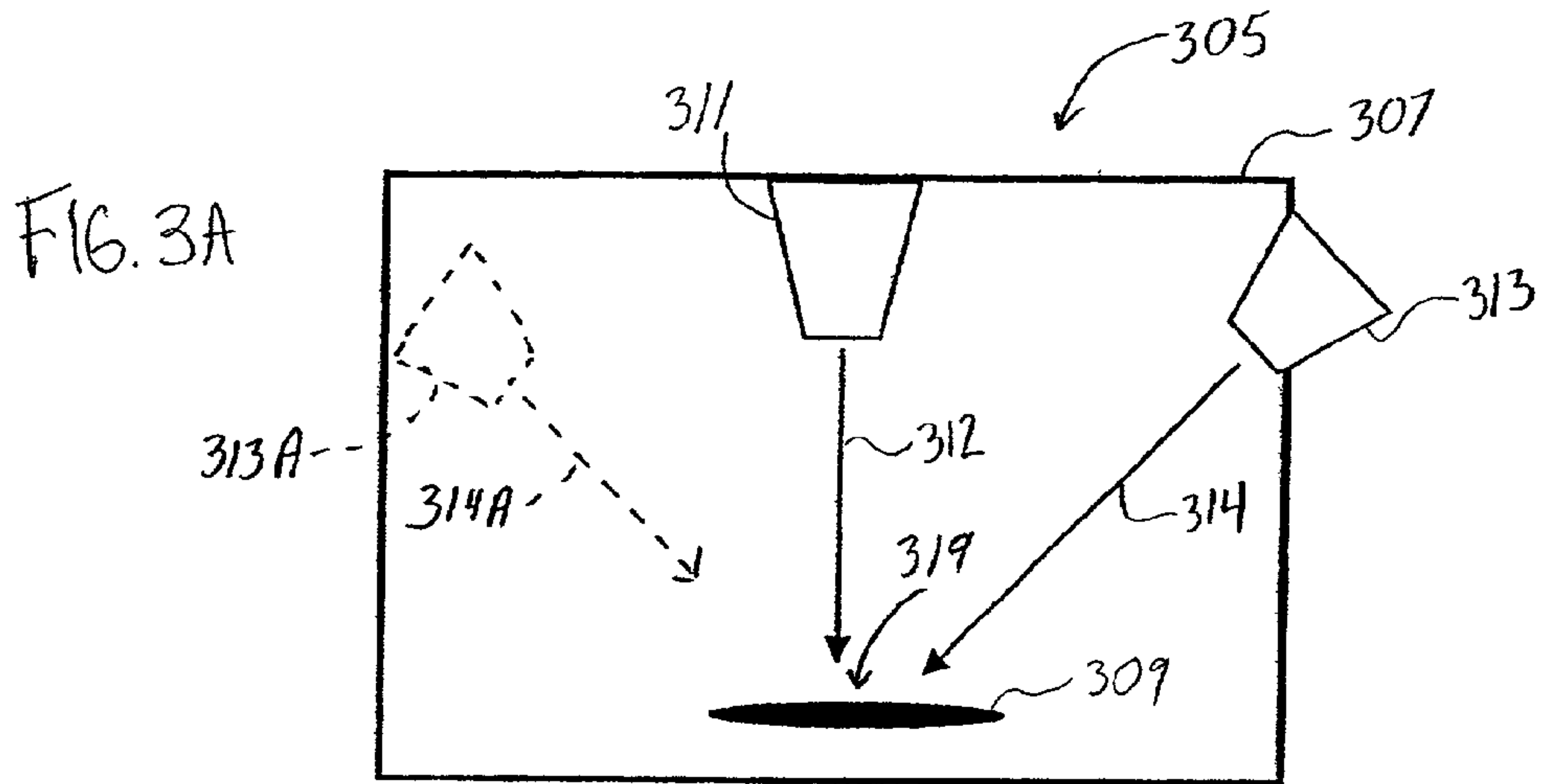
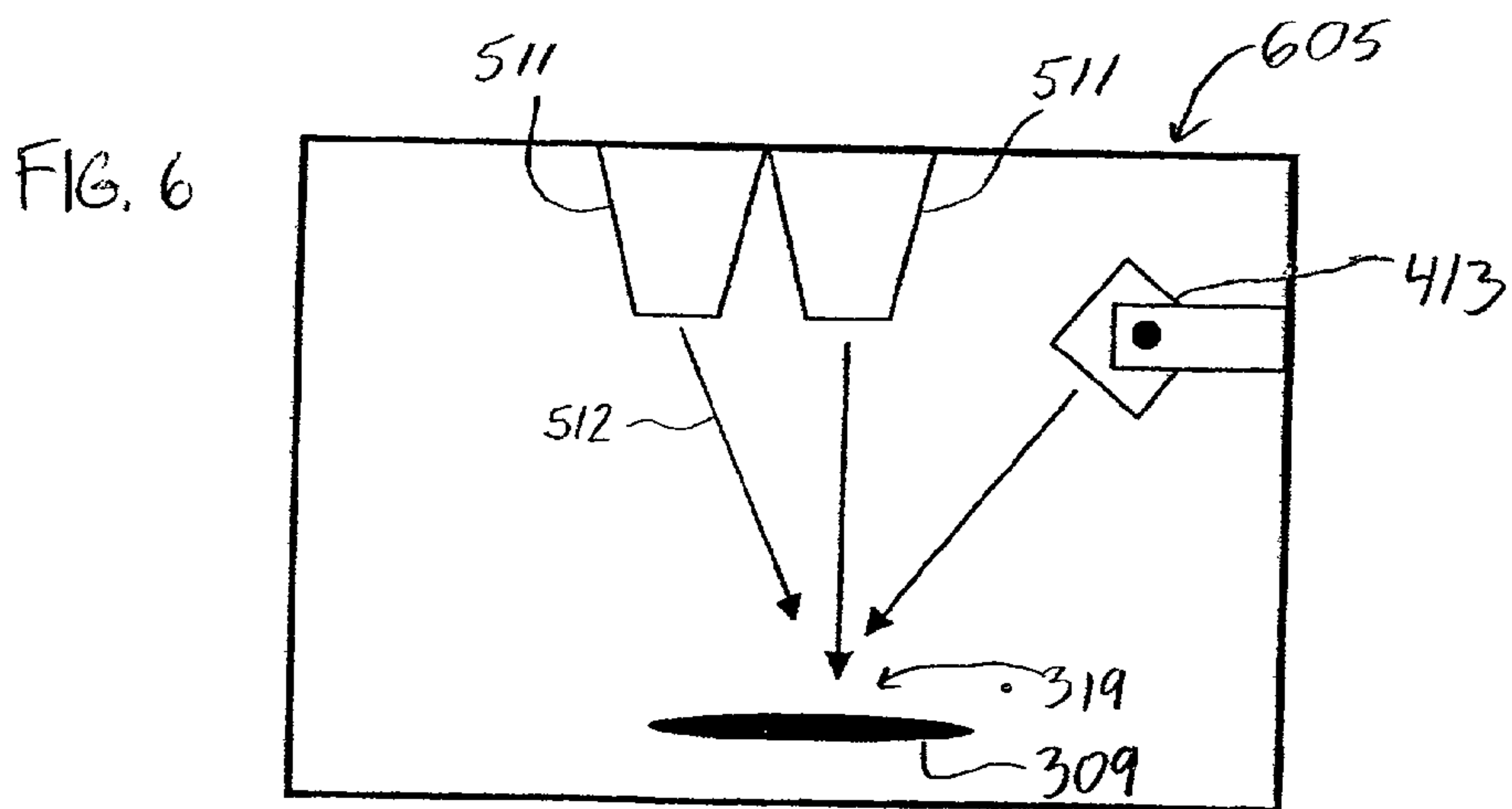
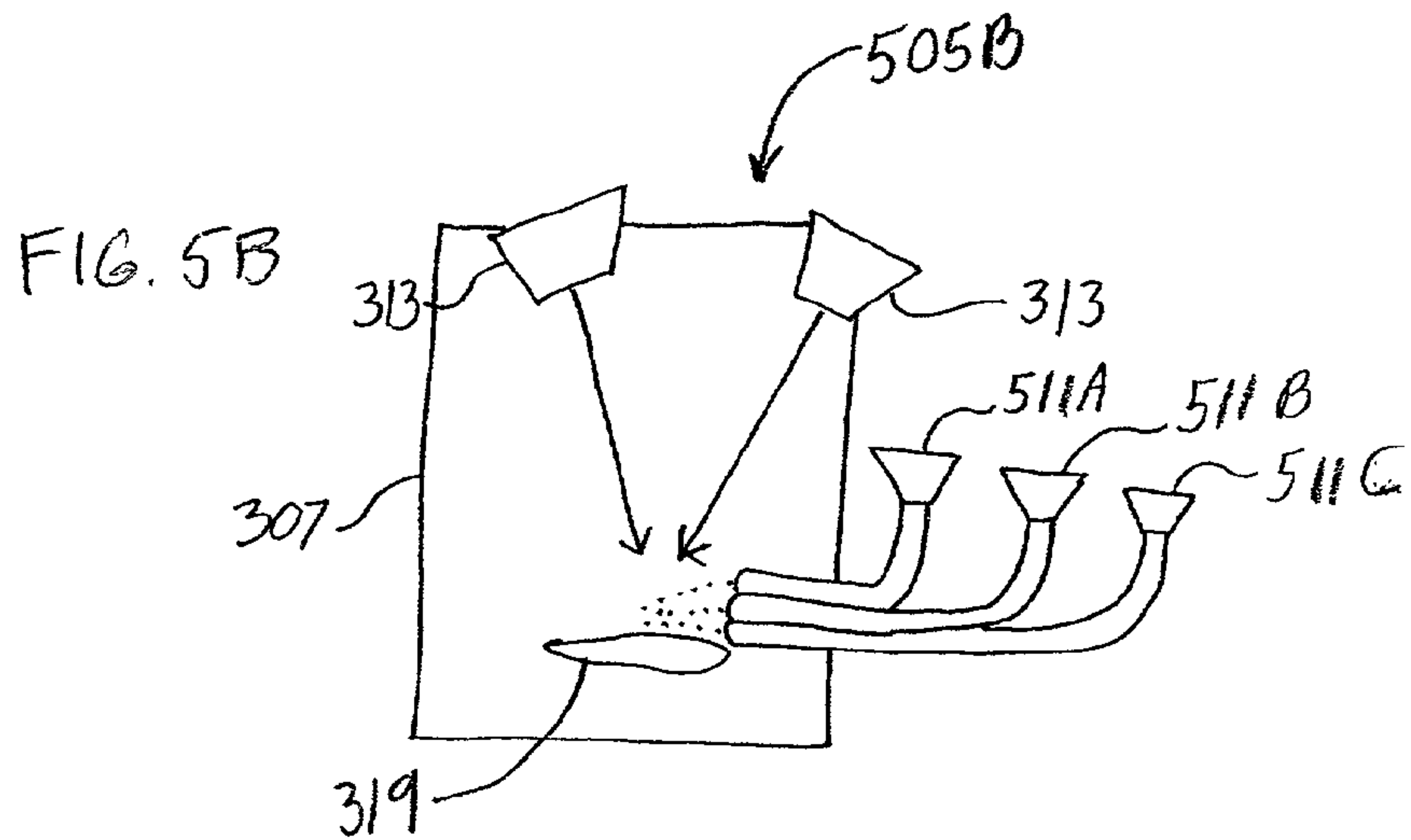
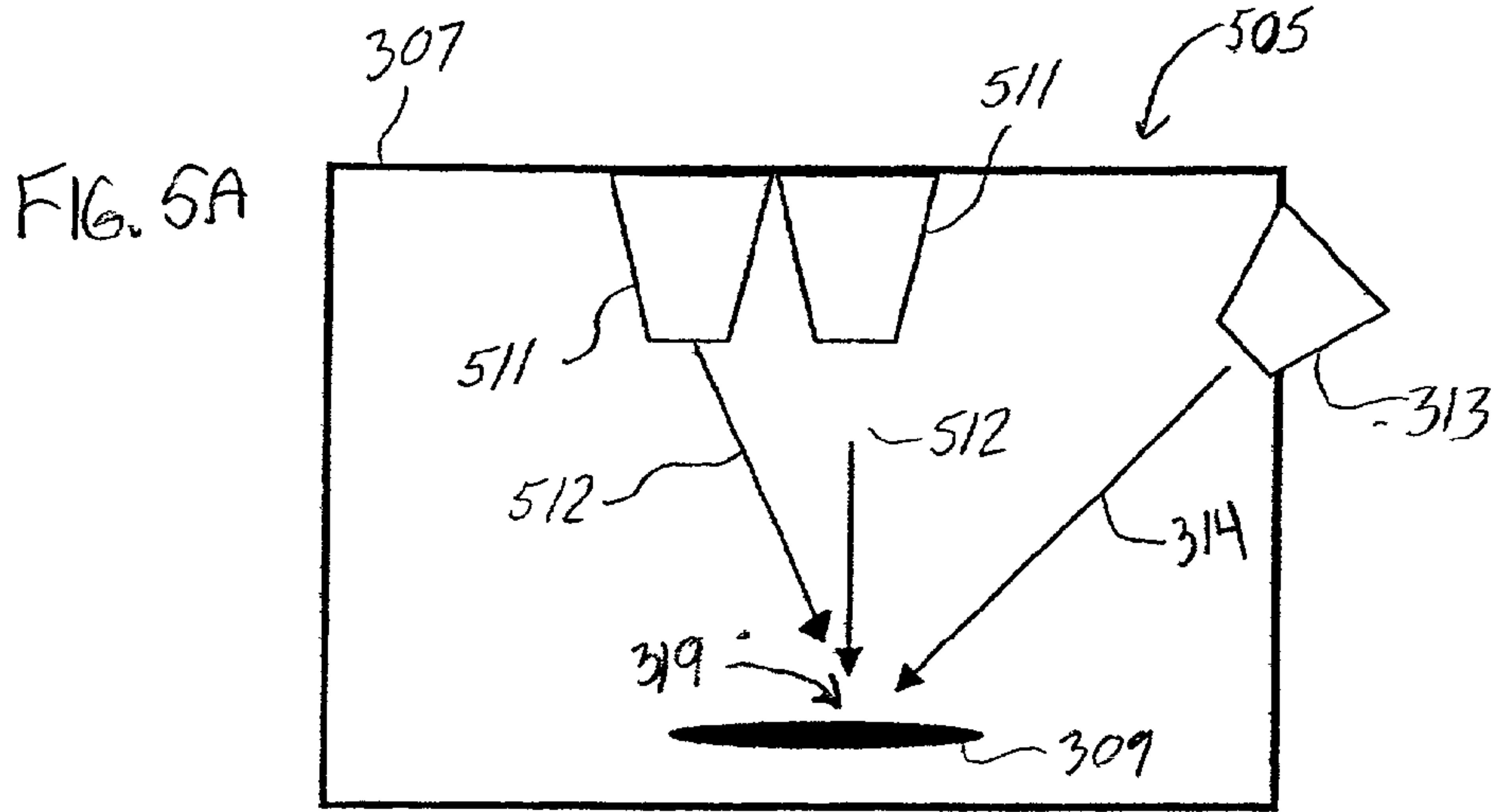


FIGURE 2D





701 Fig. 7

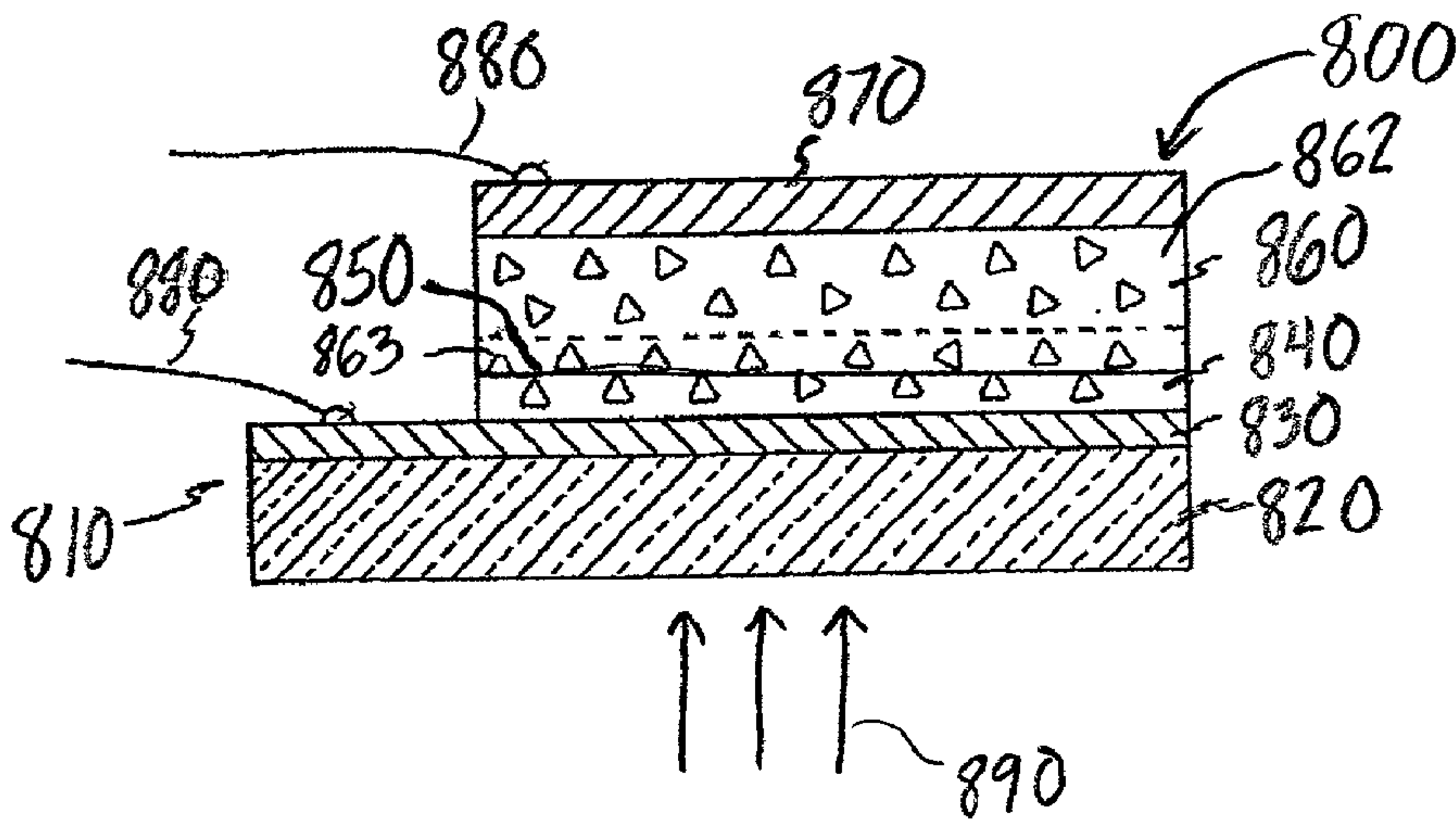
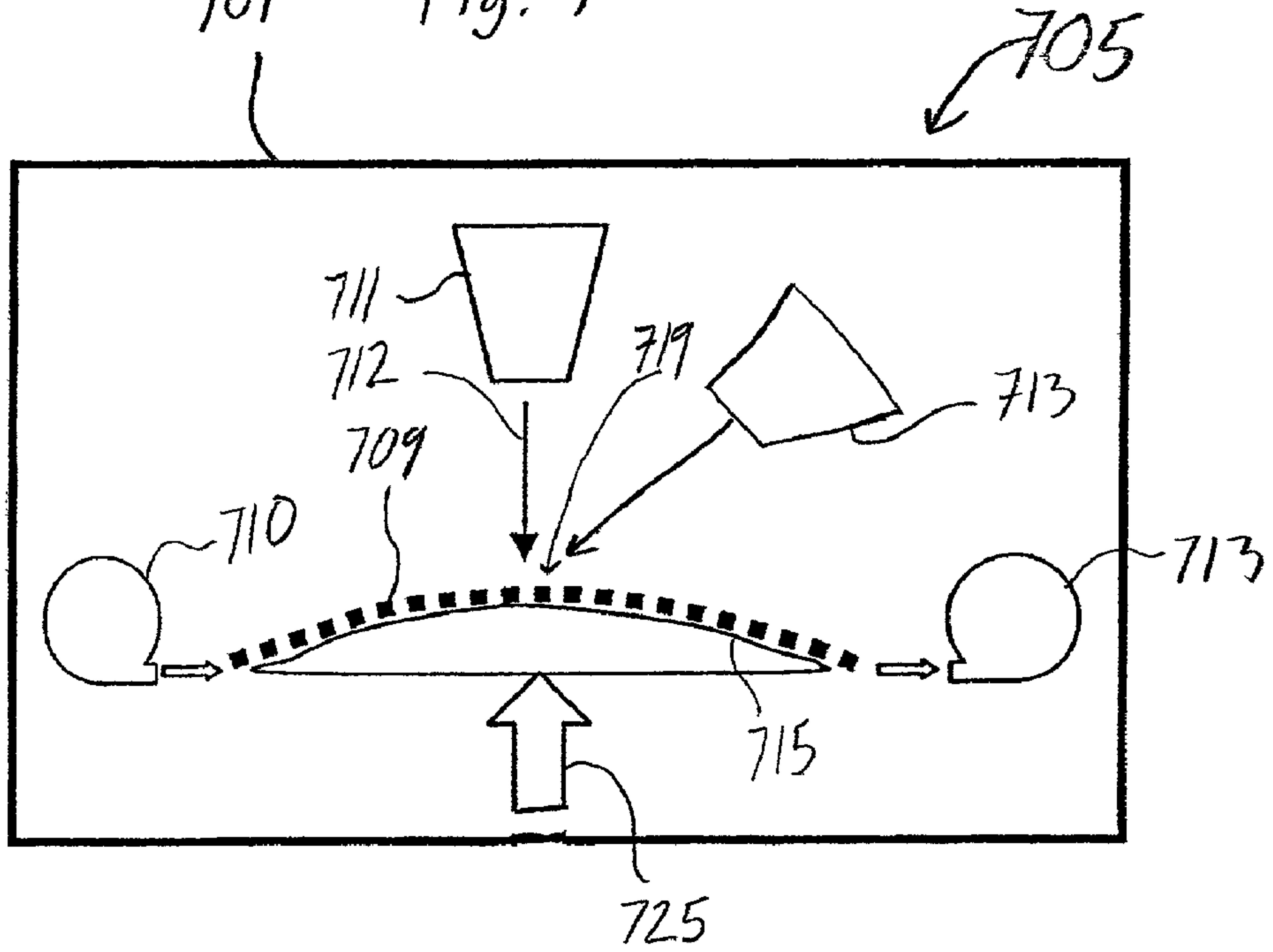
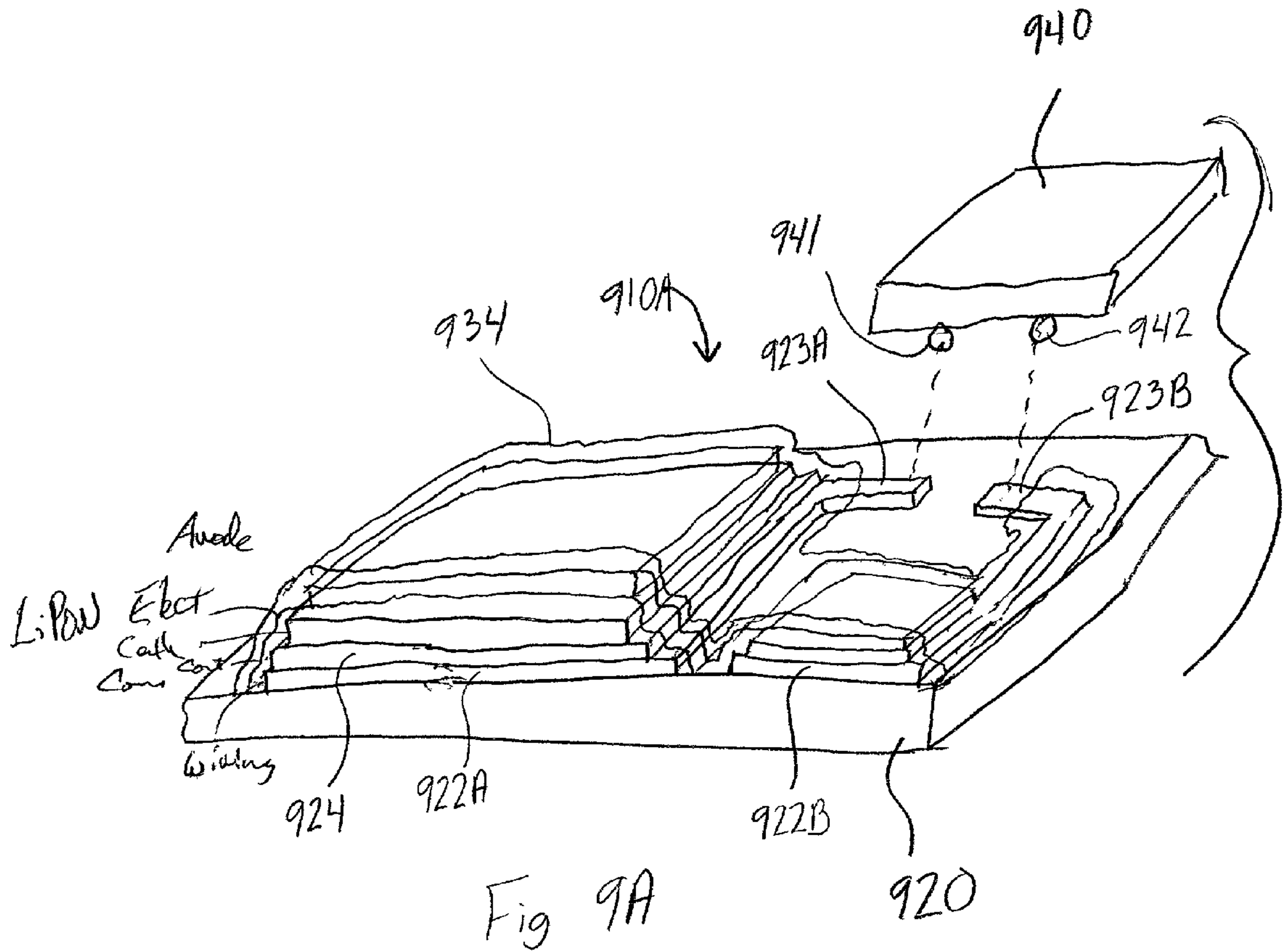


Fig. 8



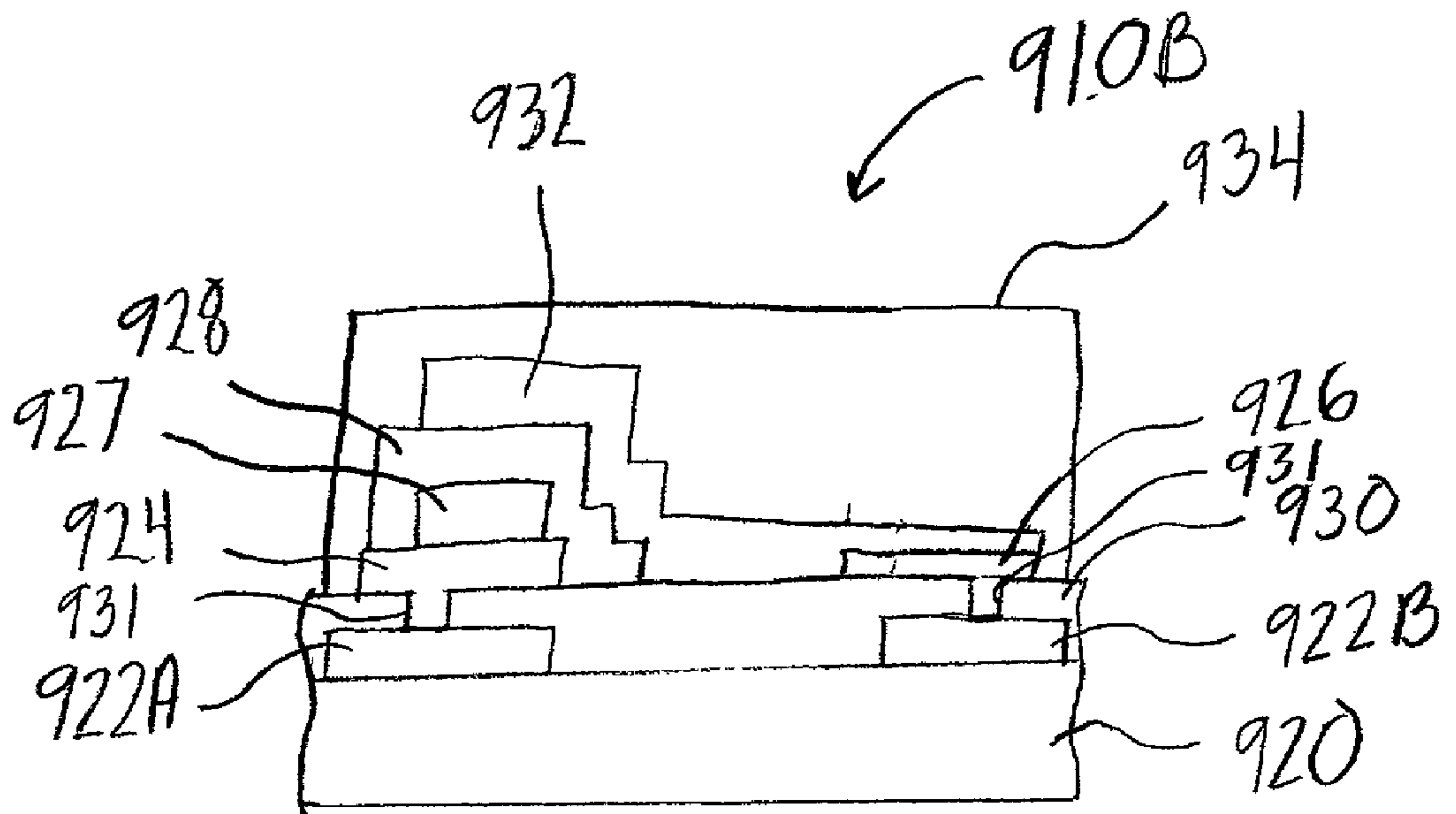
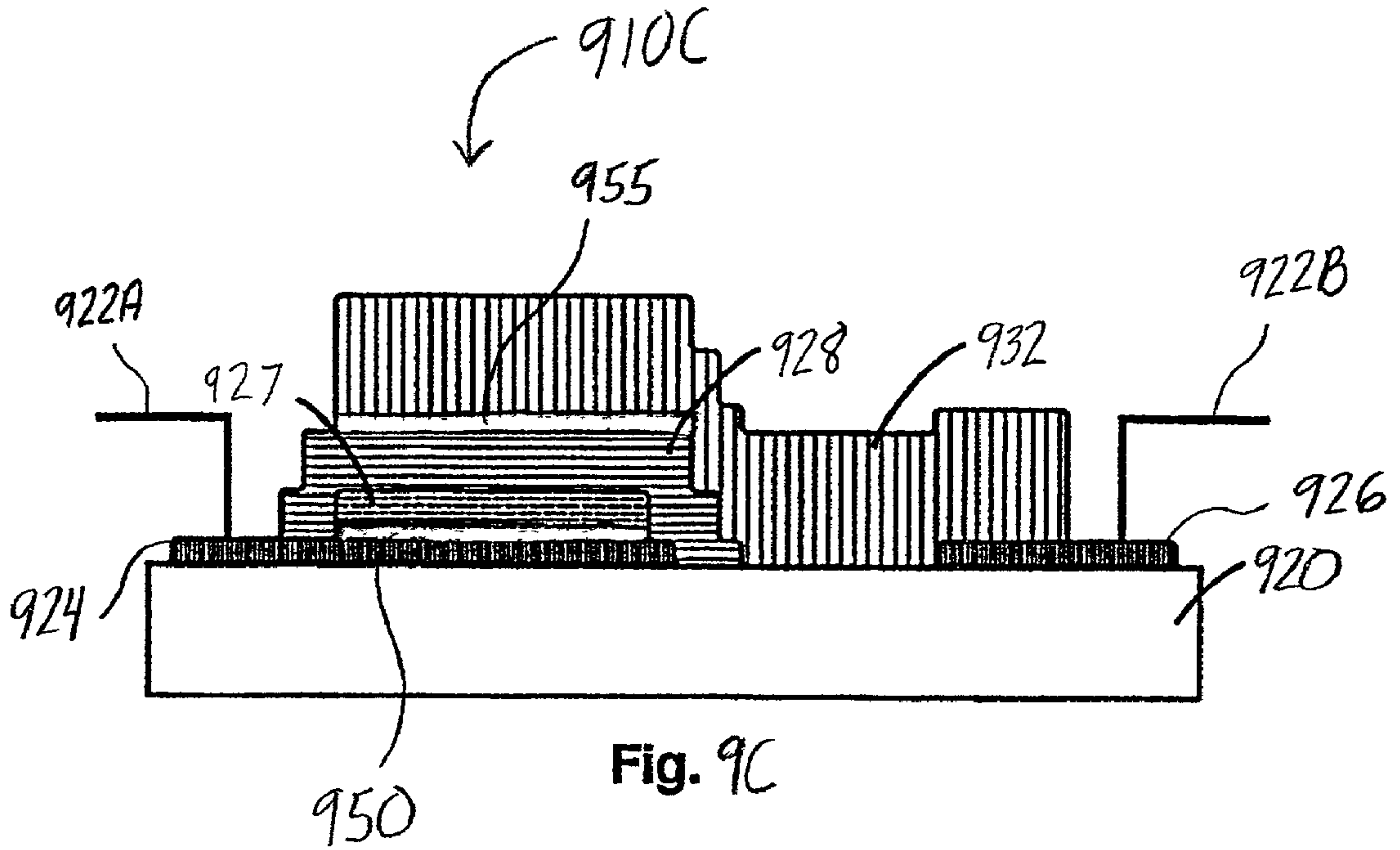


Fig. 9B



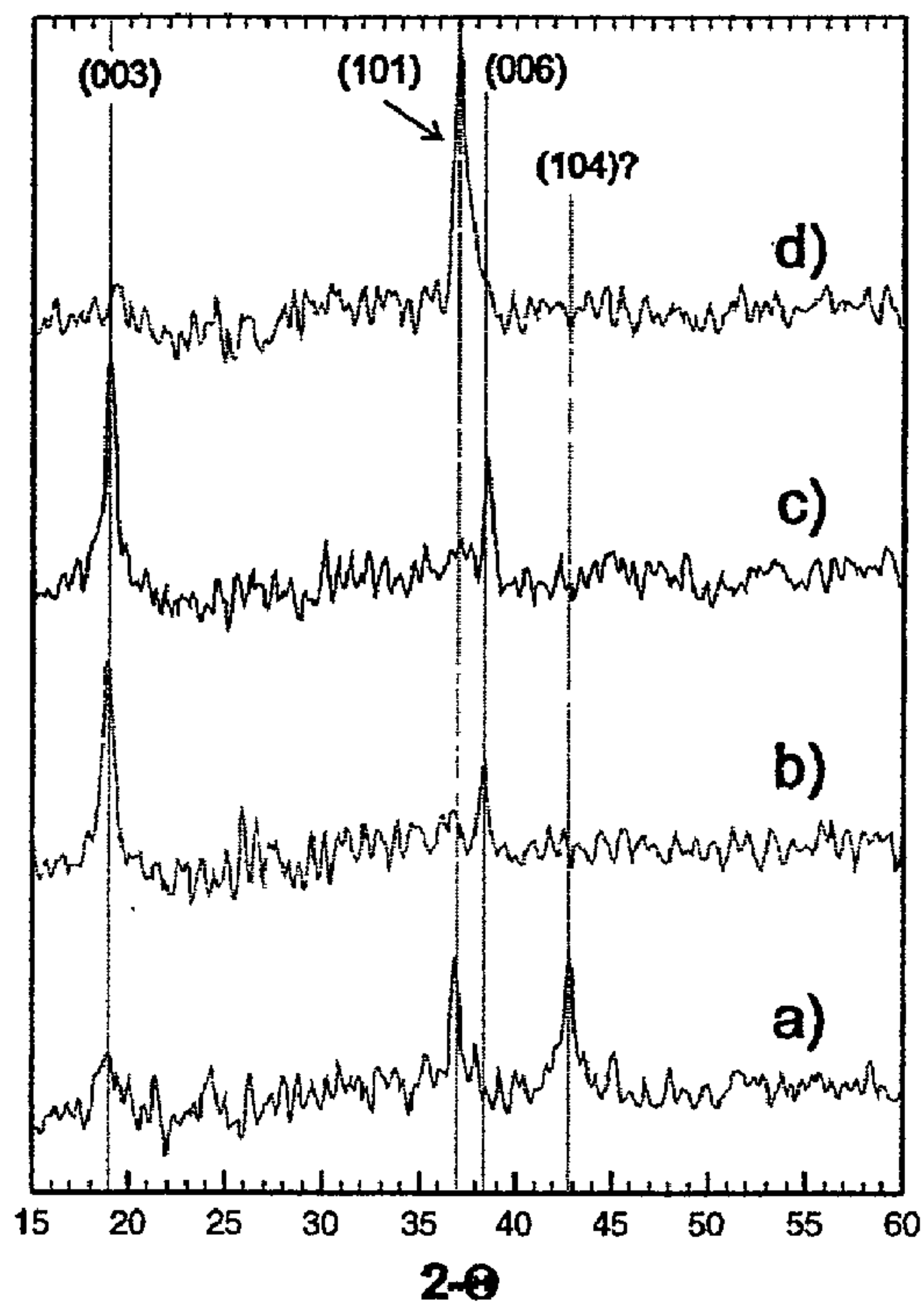


Figure 10

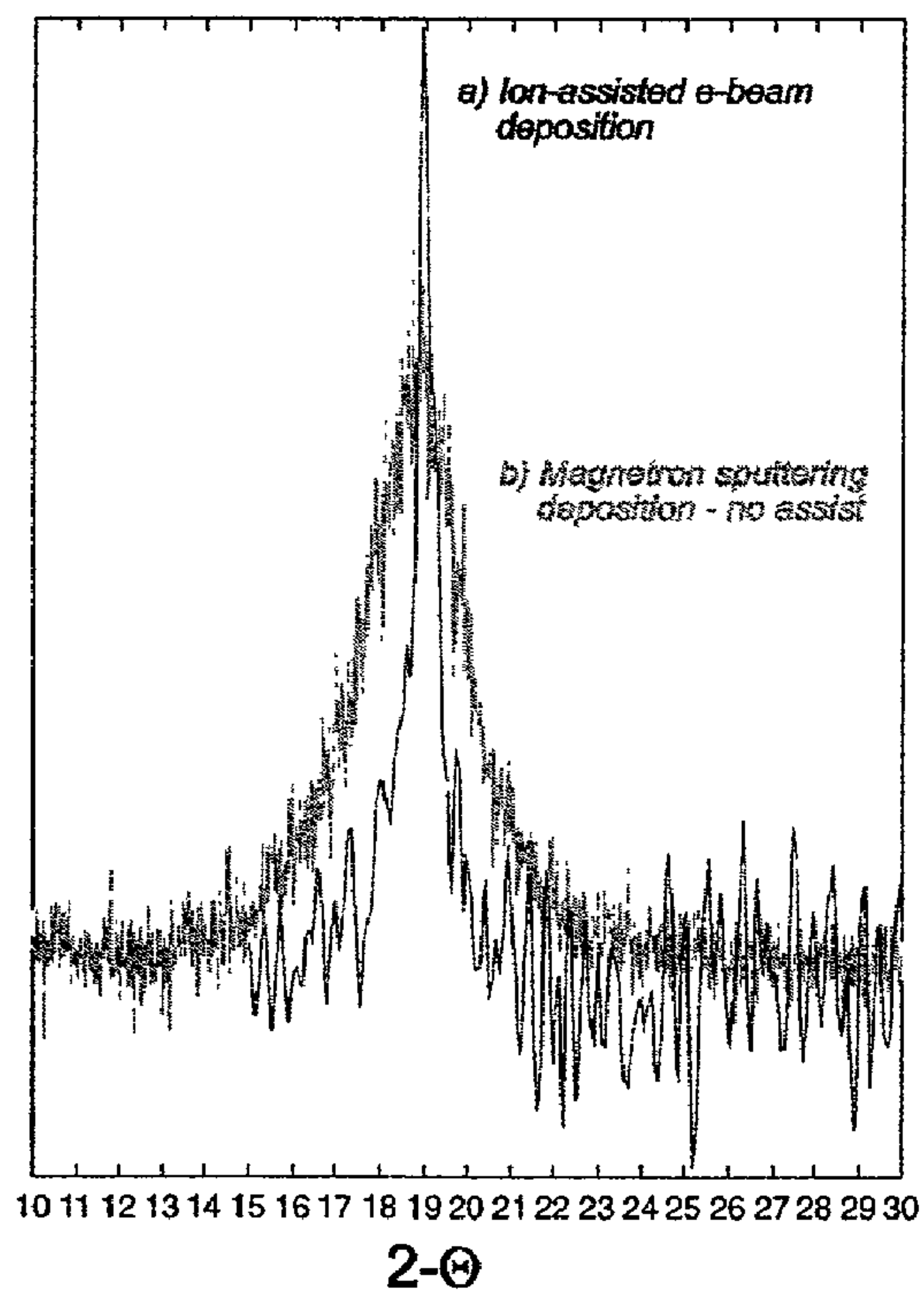


Figure 11

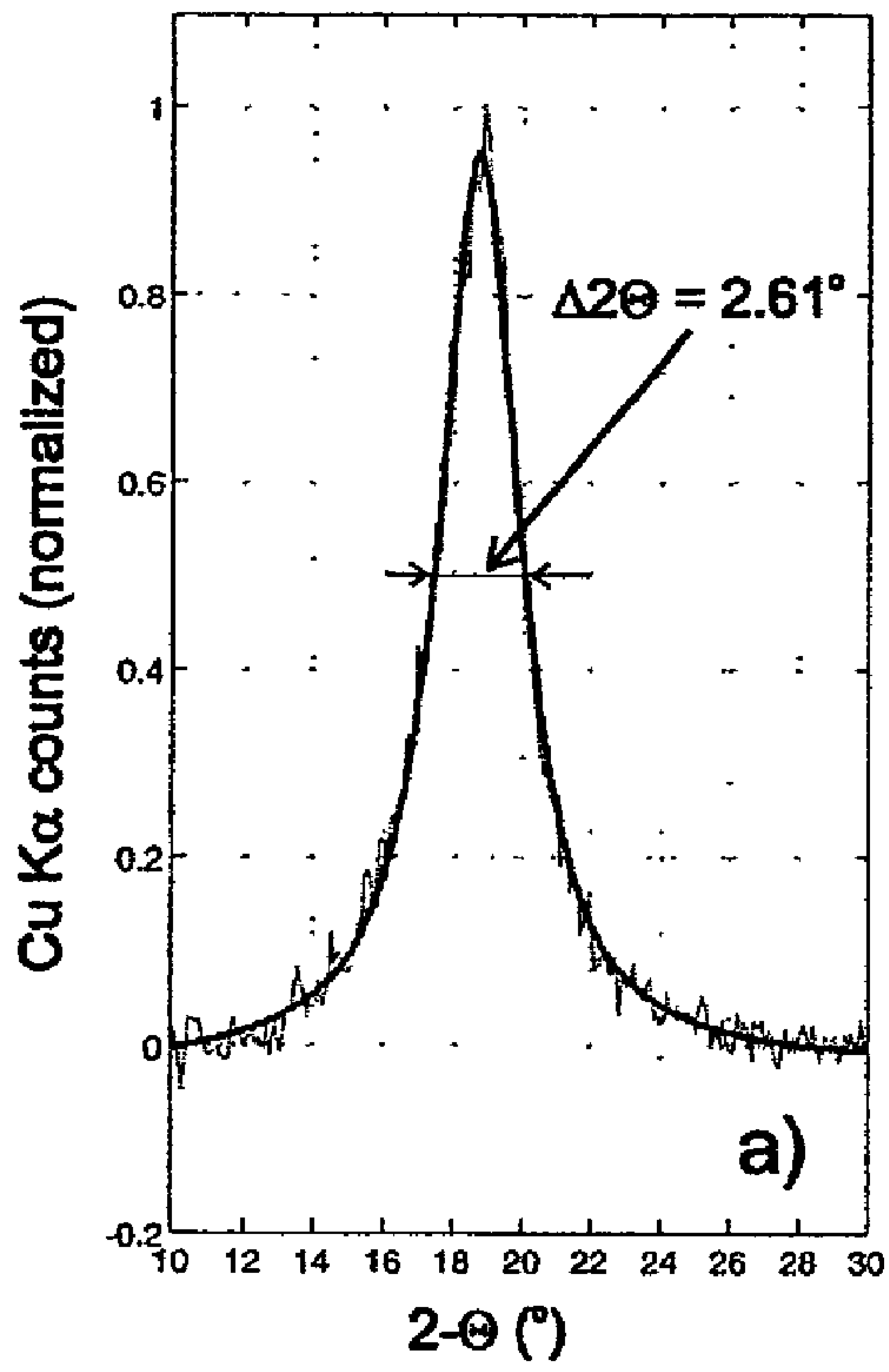


Figure 12A

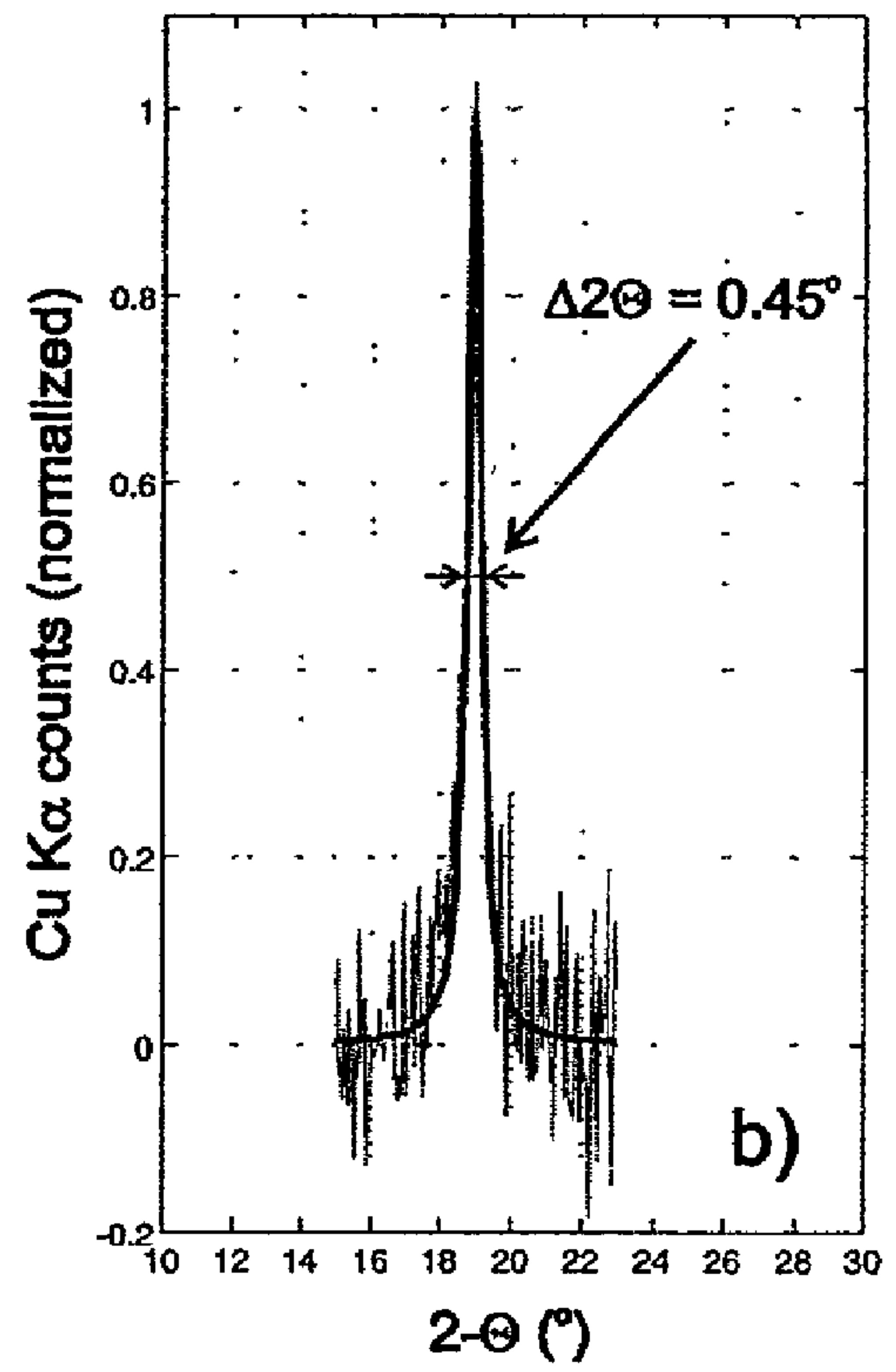


Figure 12B

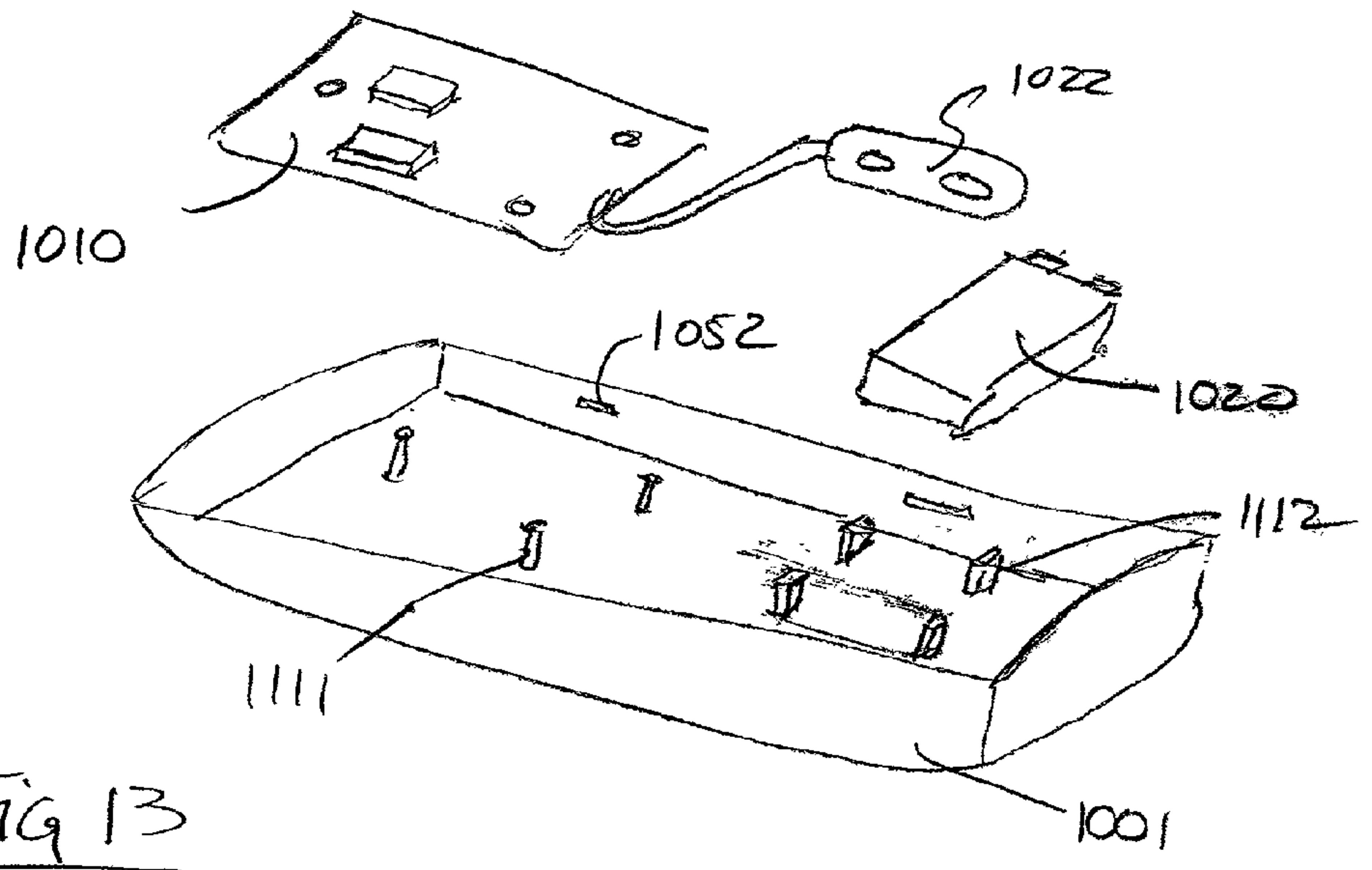
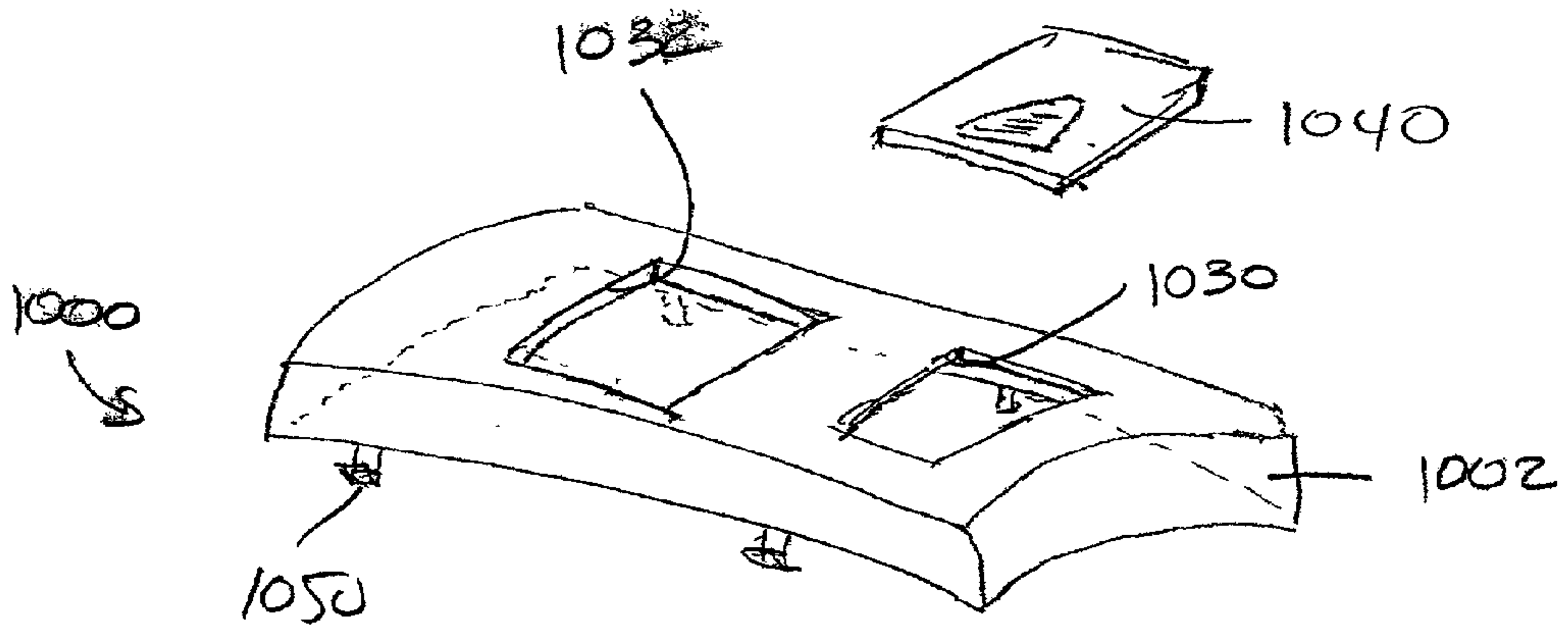


FIG 13

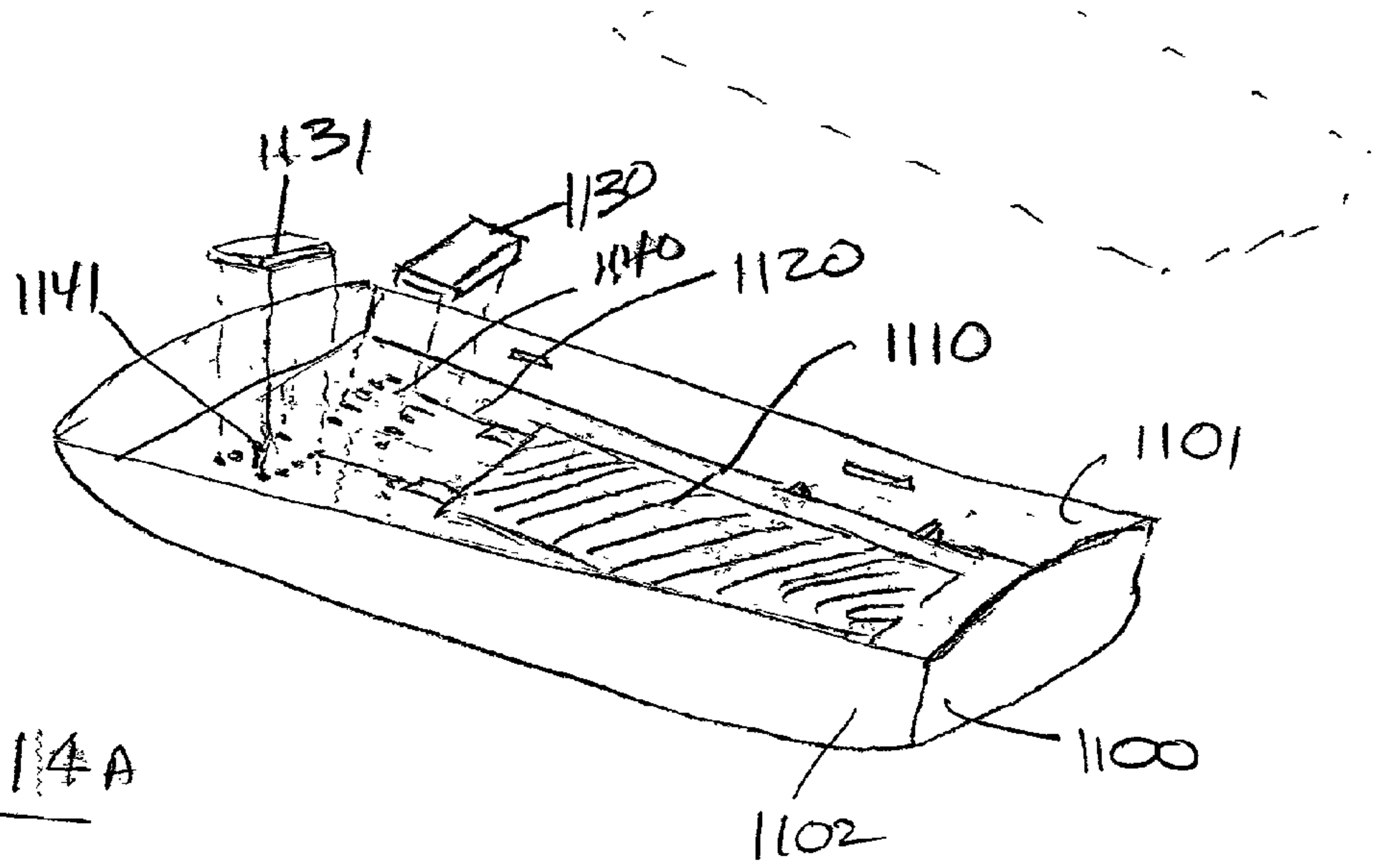


FIG 14A

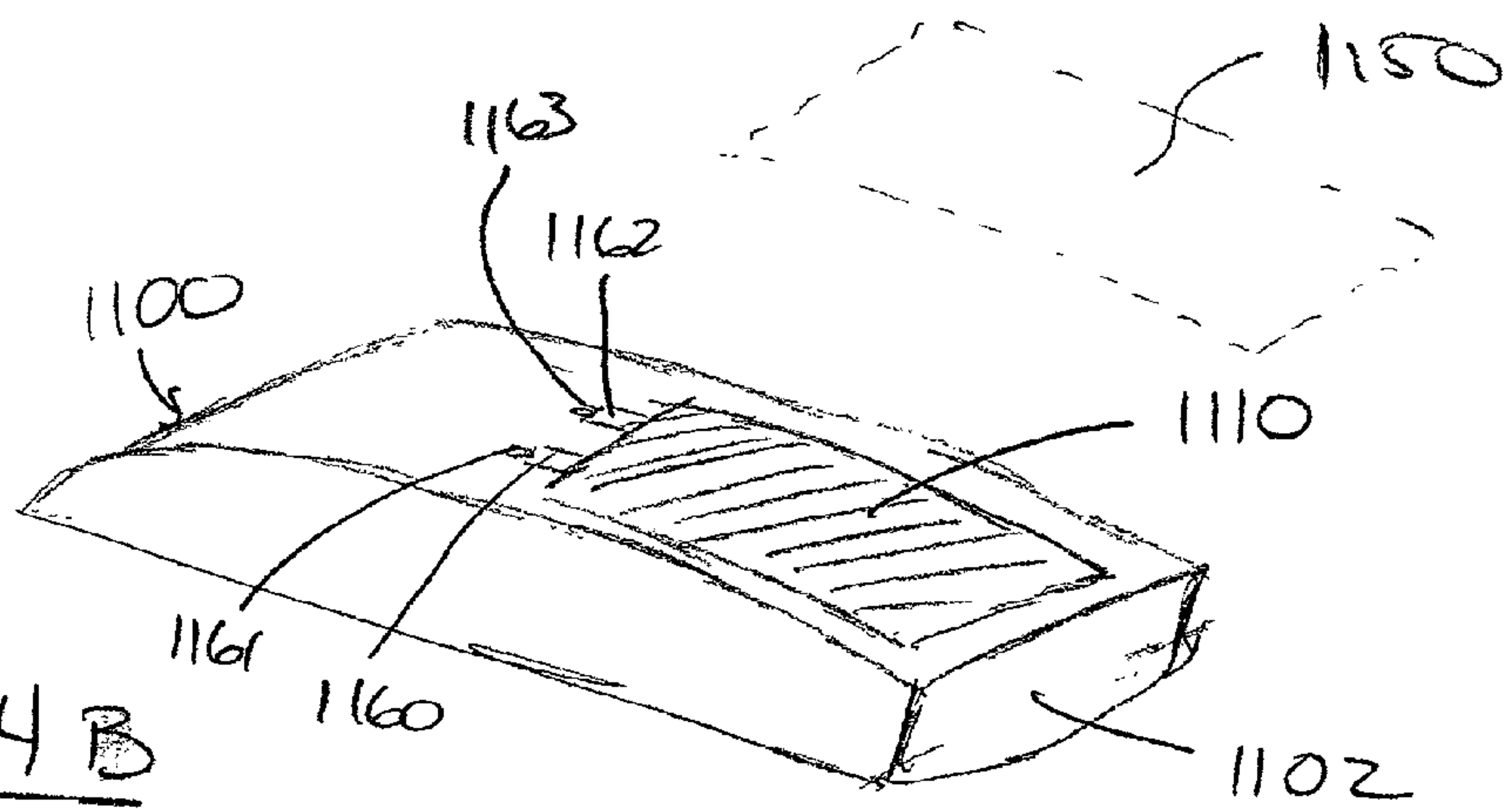
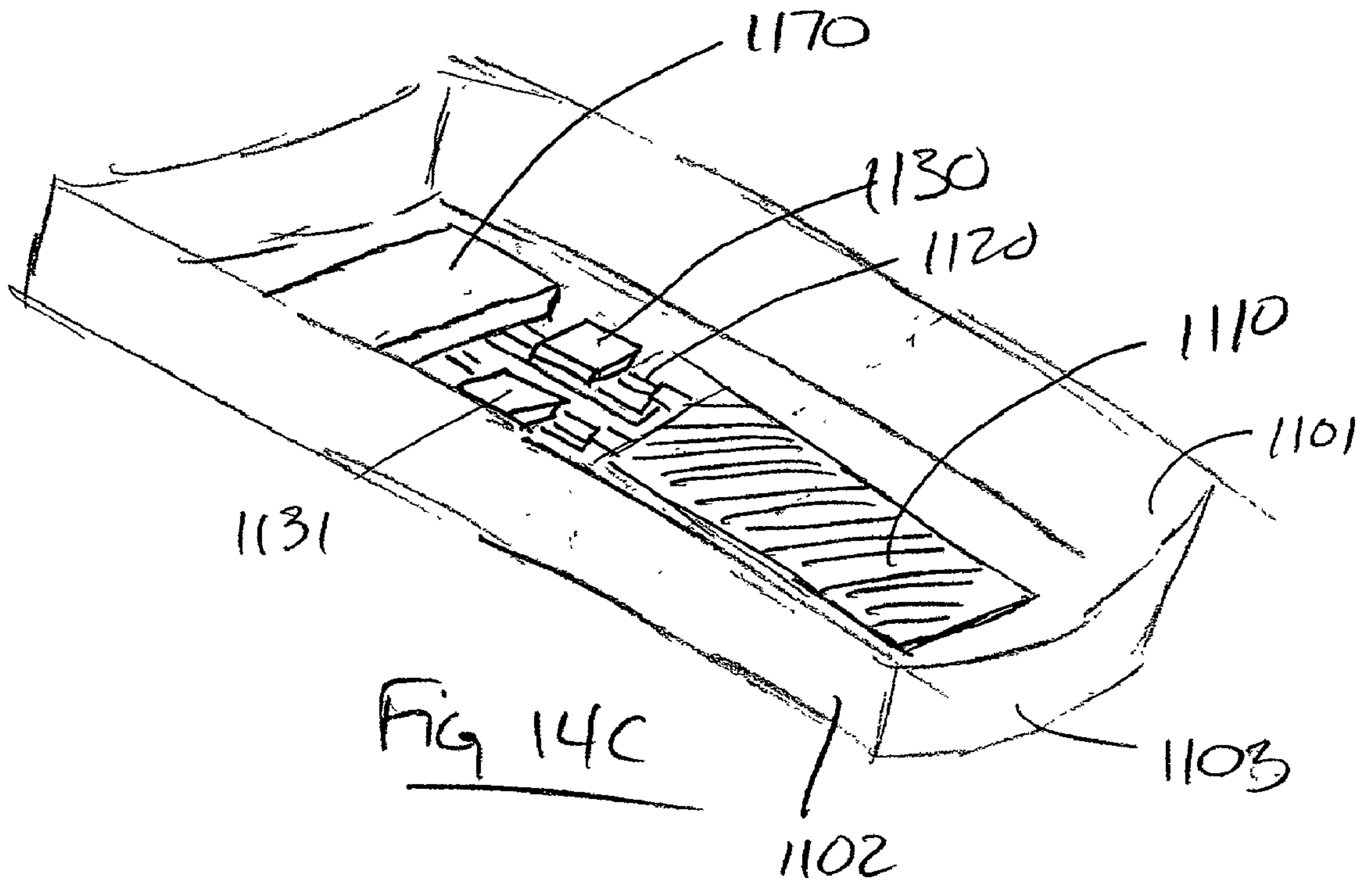
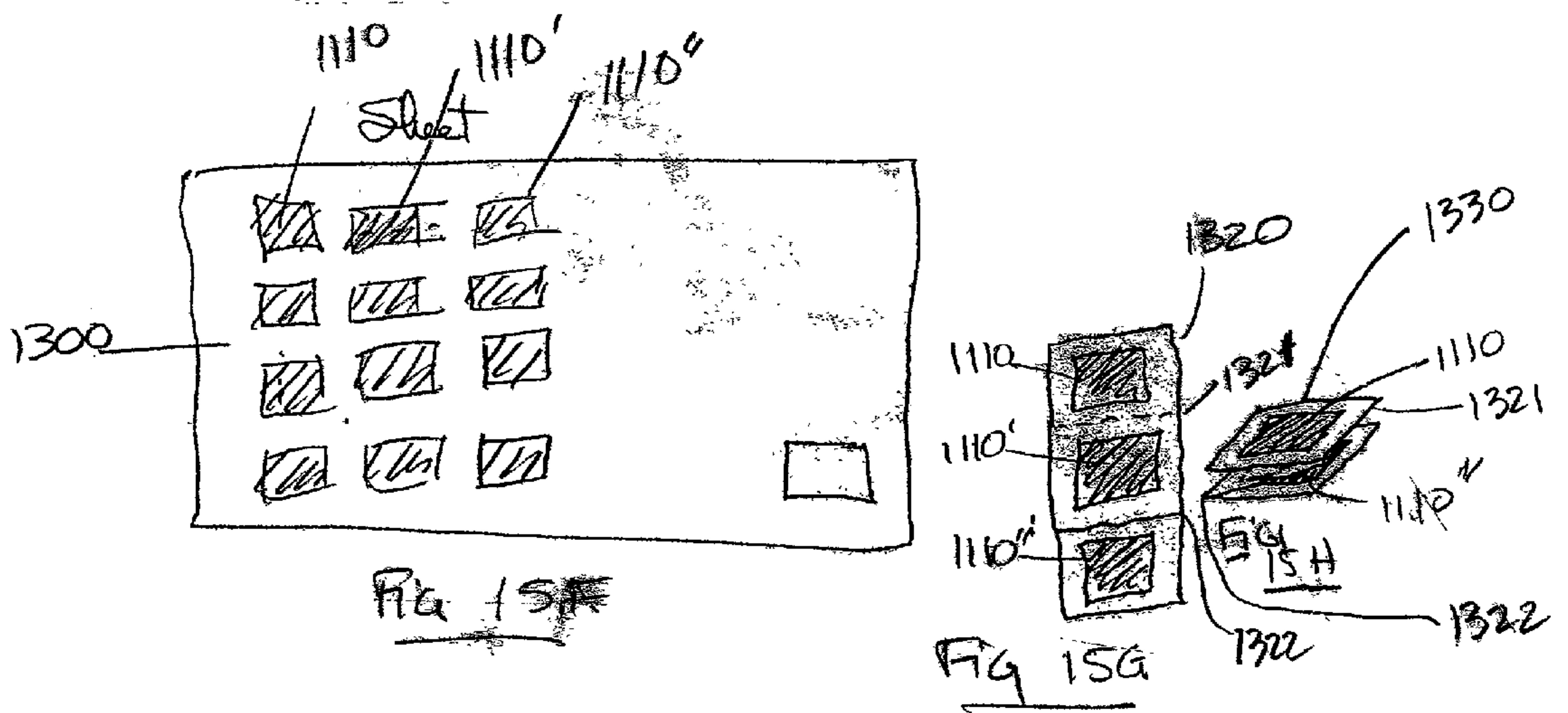
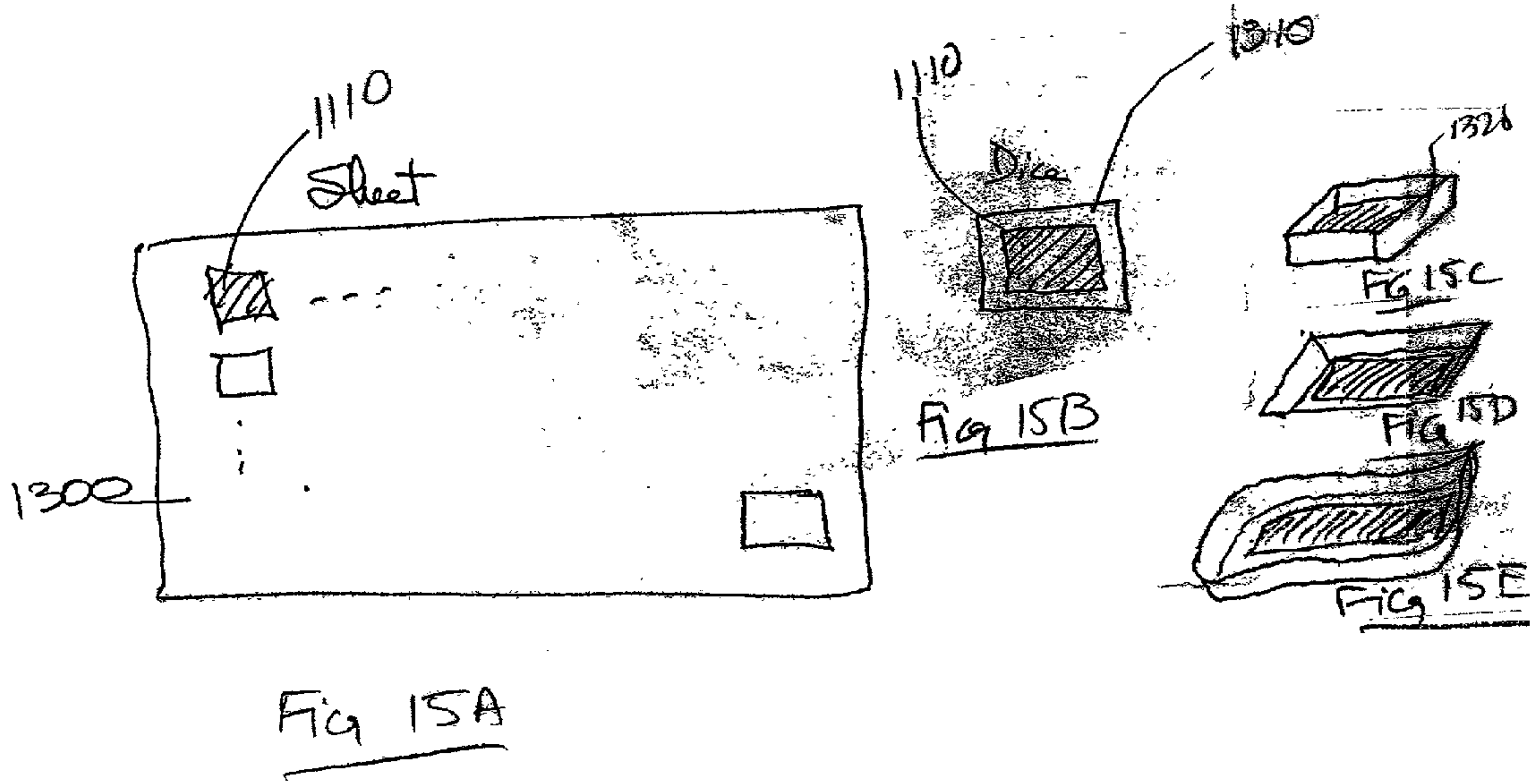


FIG 14B





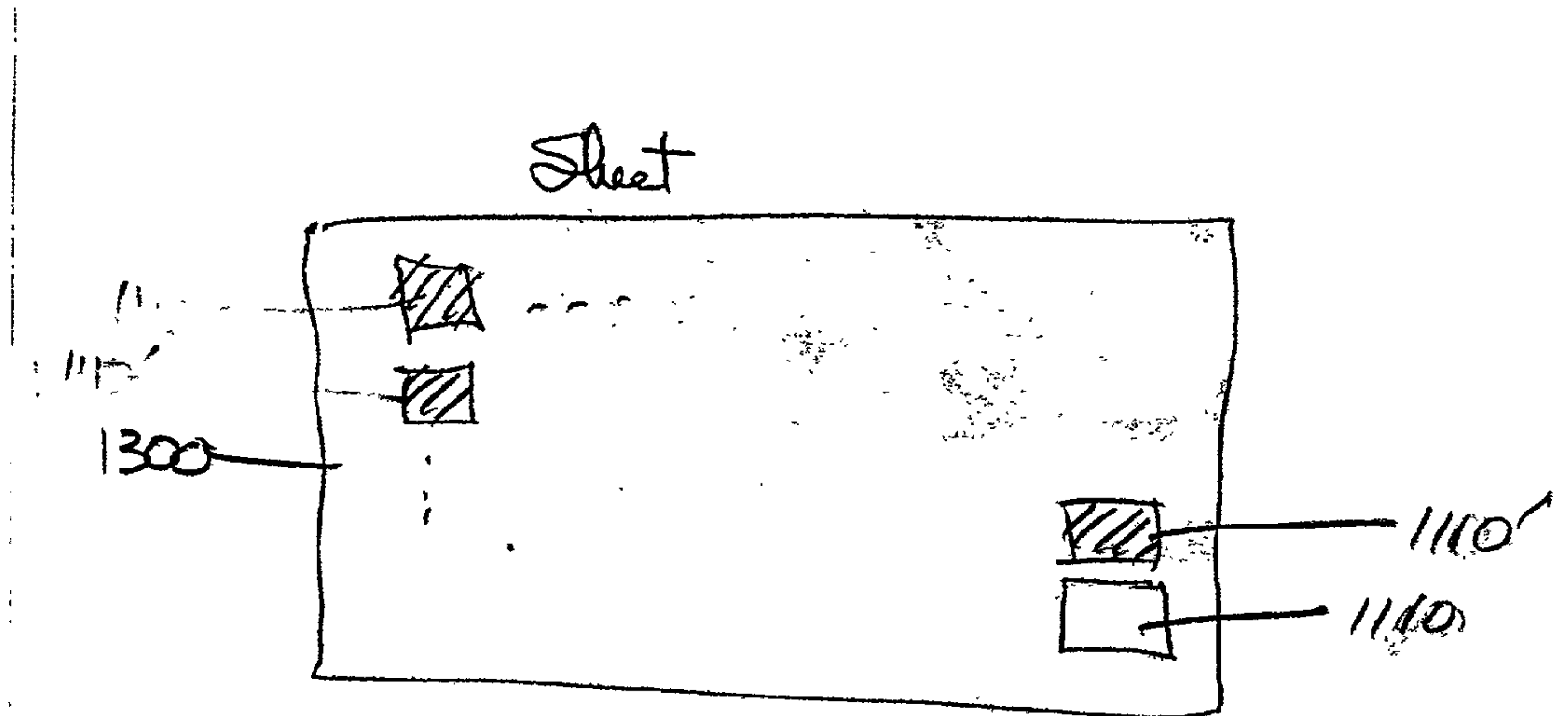


FIG 15I

or egg carton vacuum form onto form

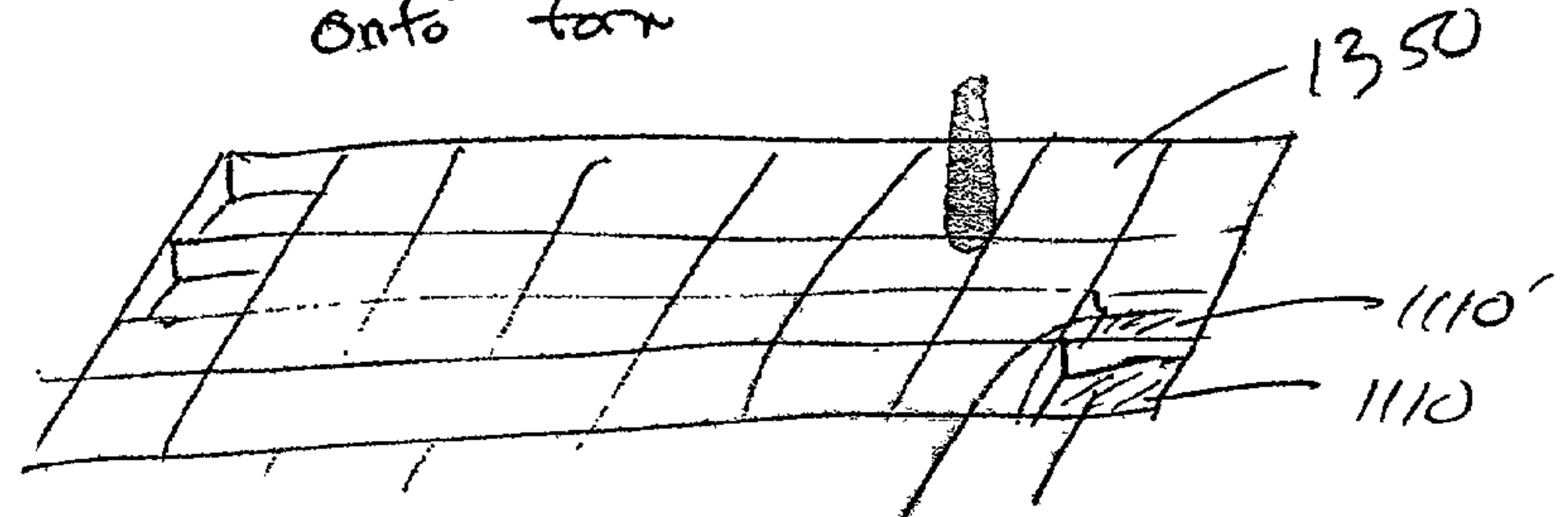


FIG 15J

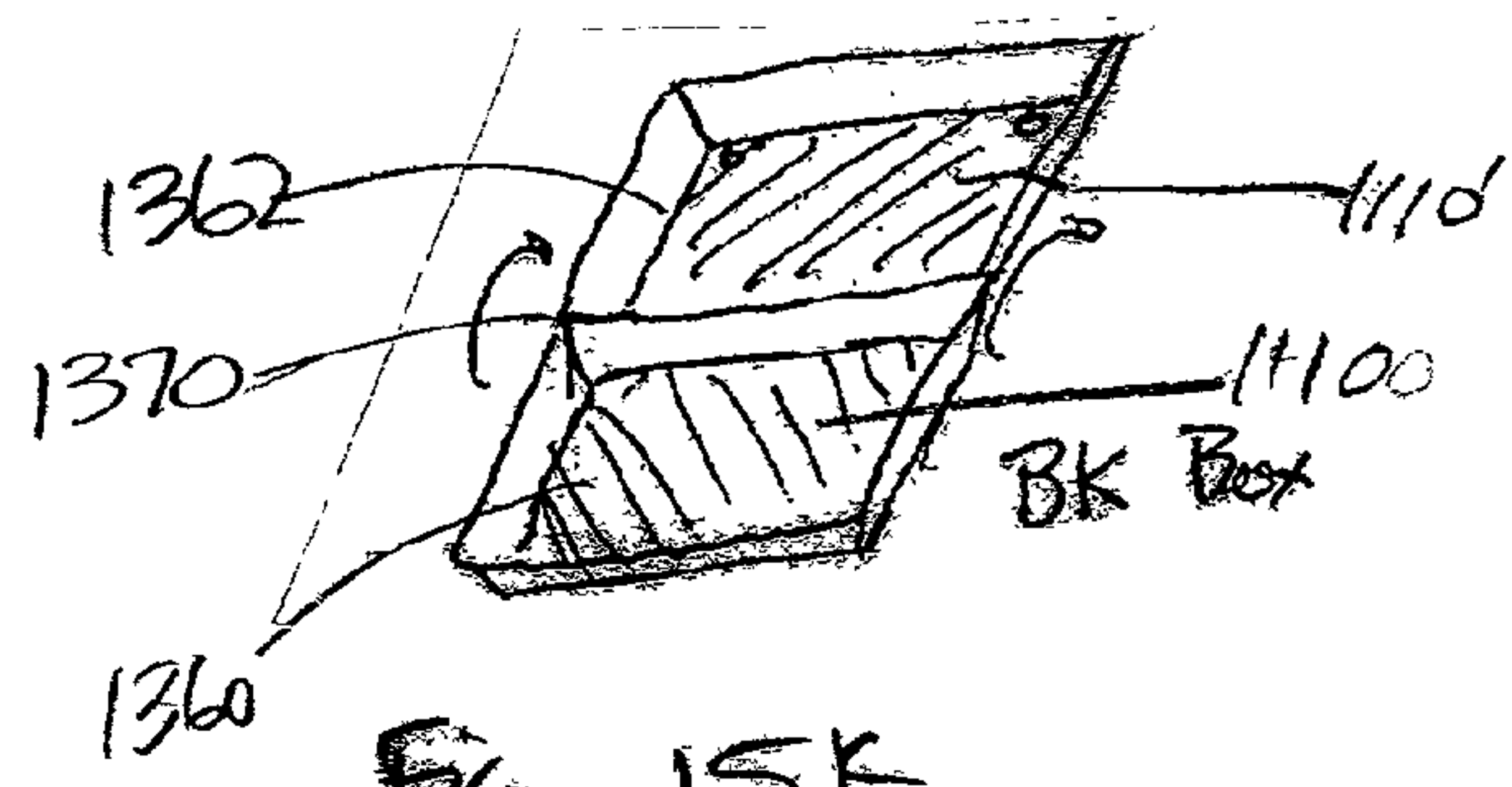


FIG 15K

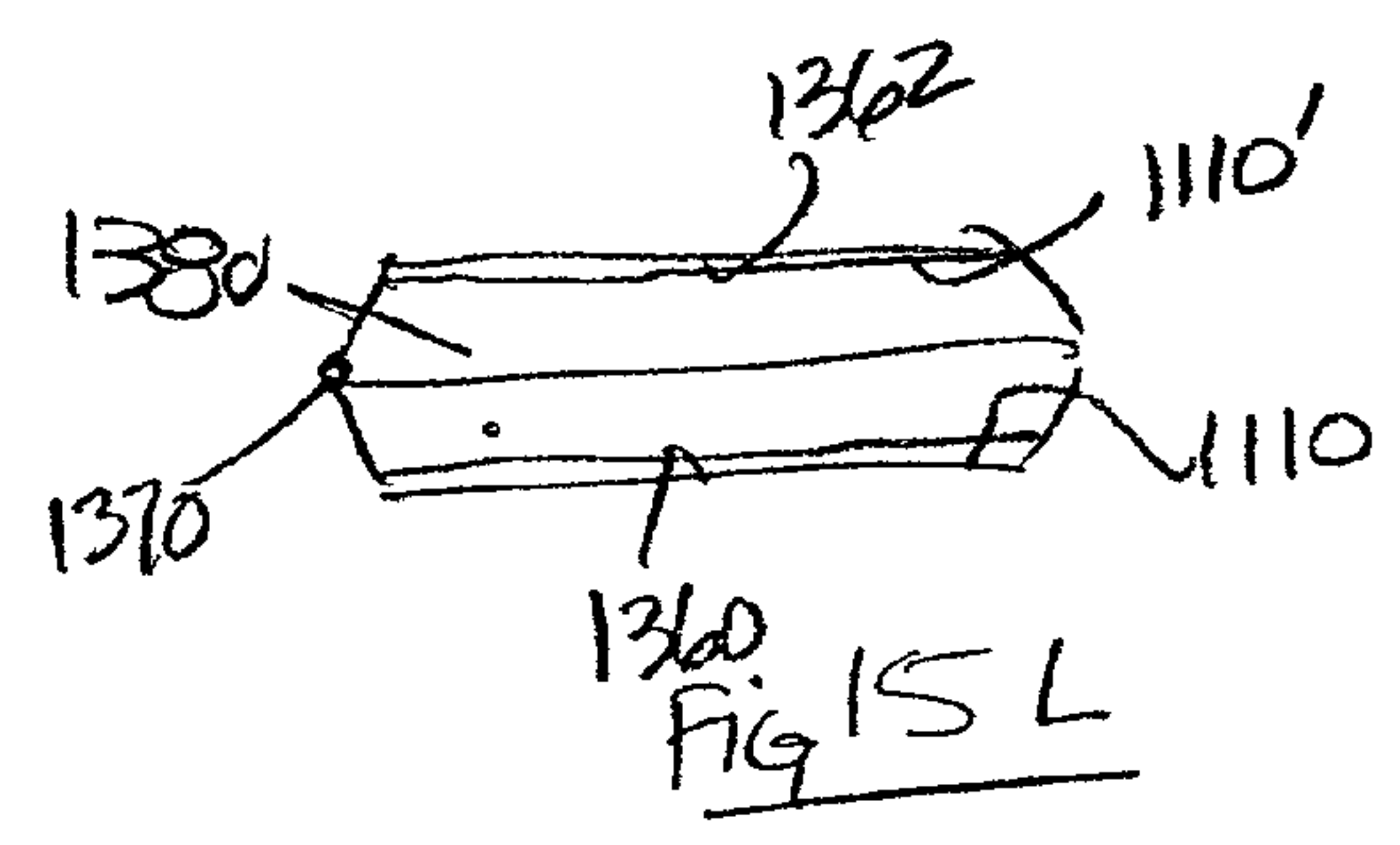


FIG 15L

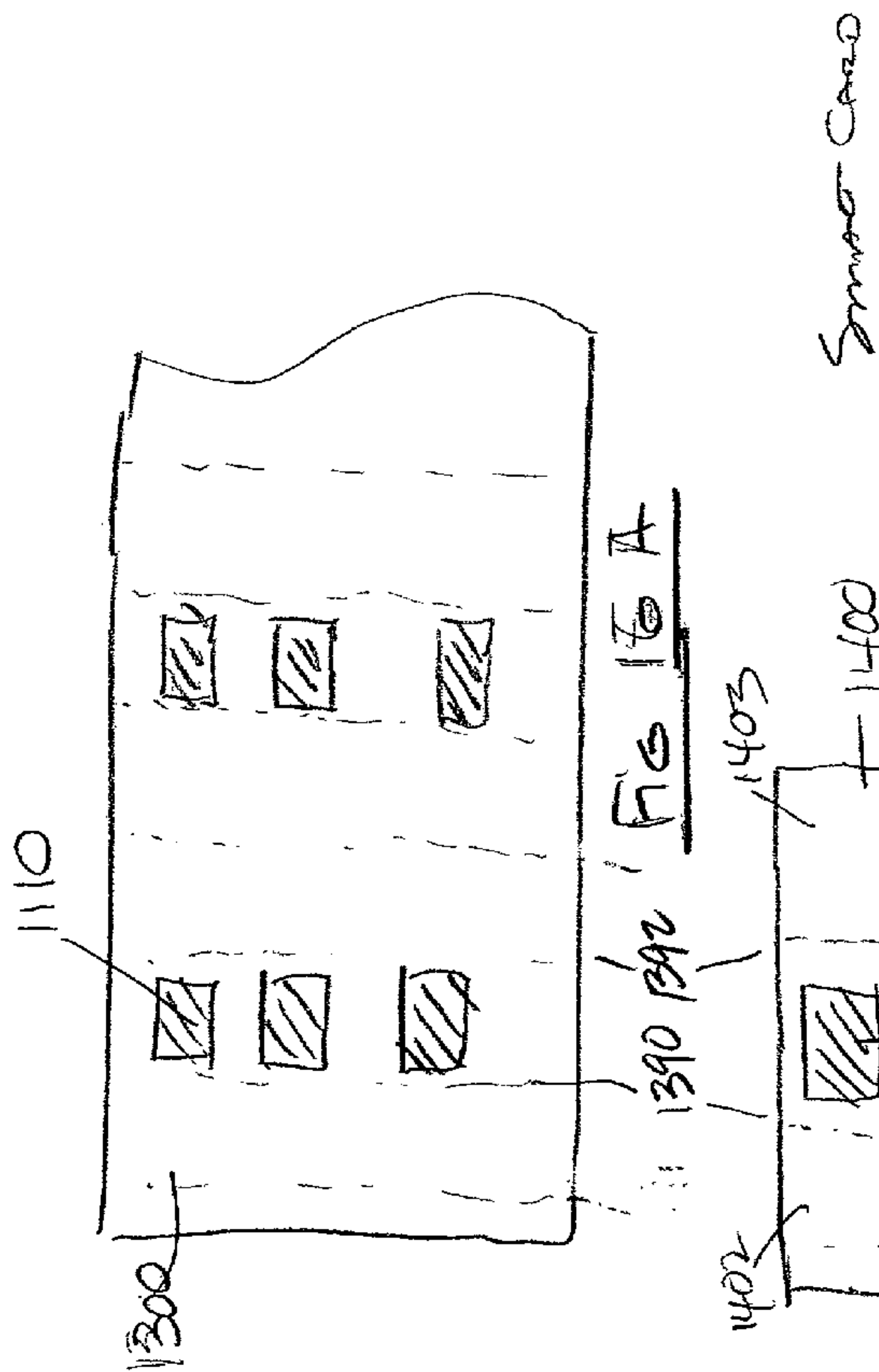


Fig 16A

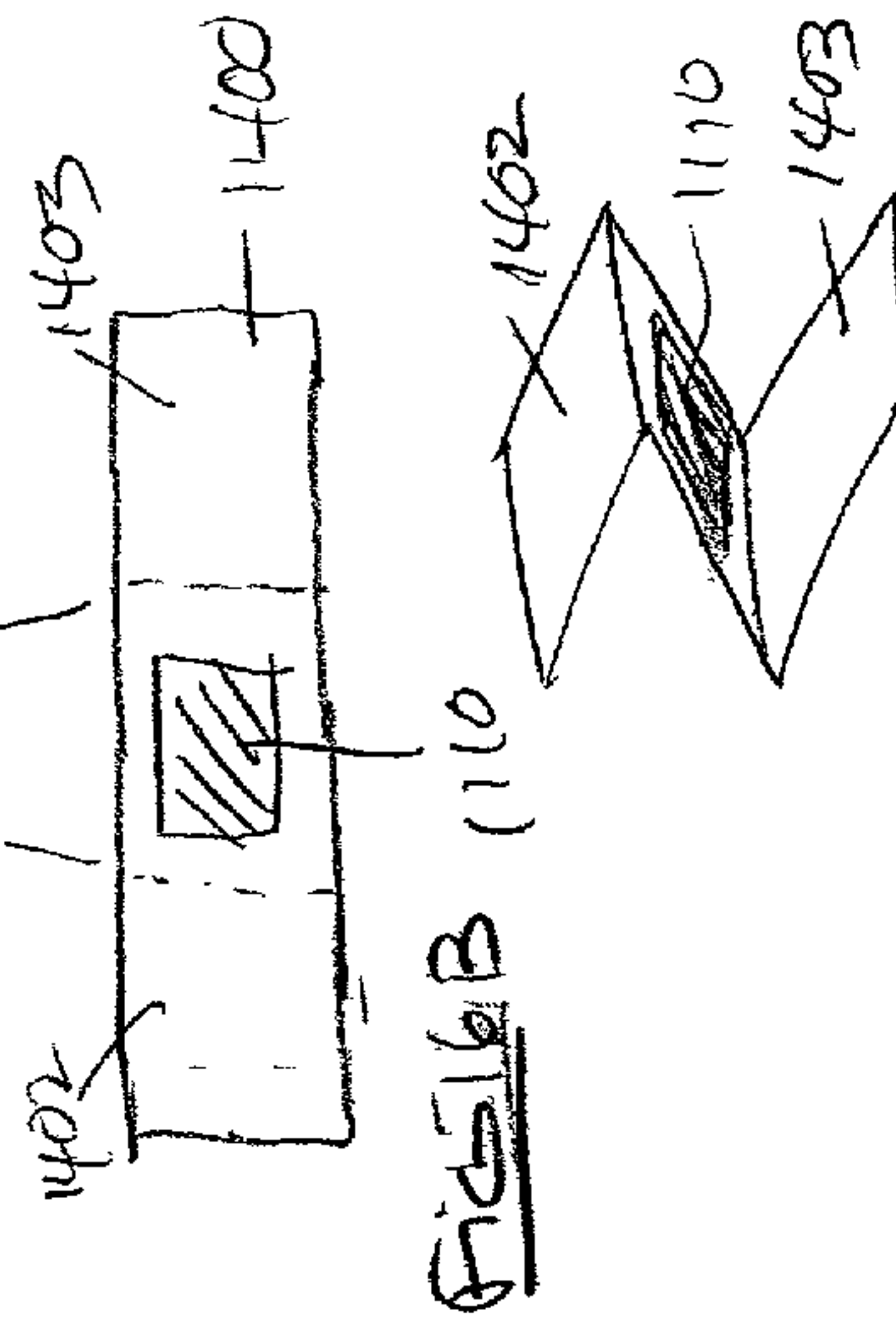


Fig 16B

Fig 16C

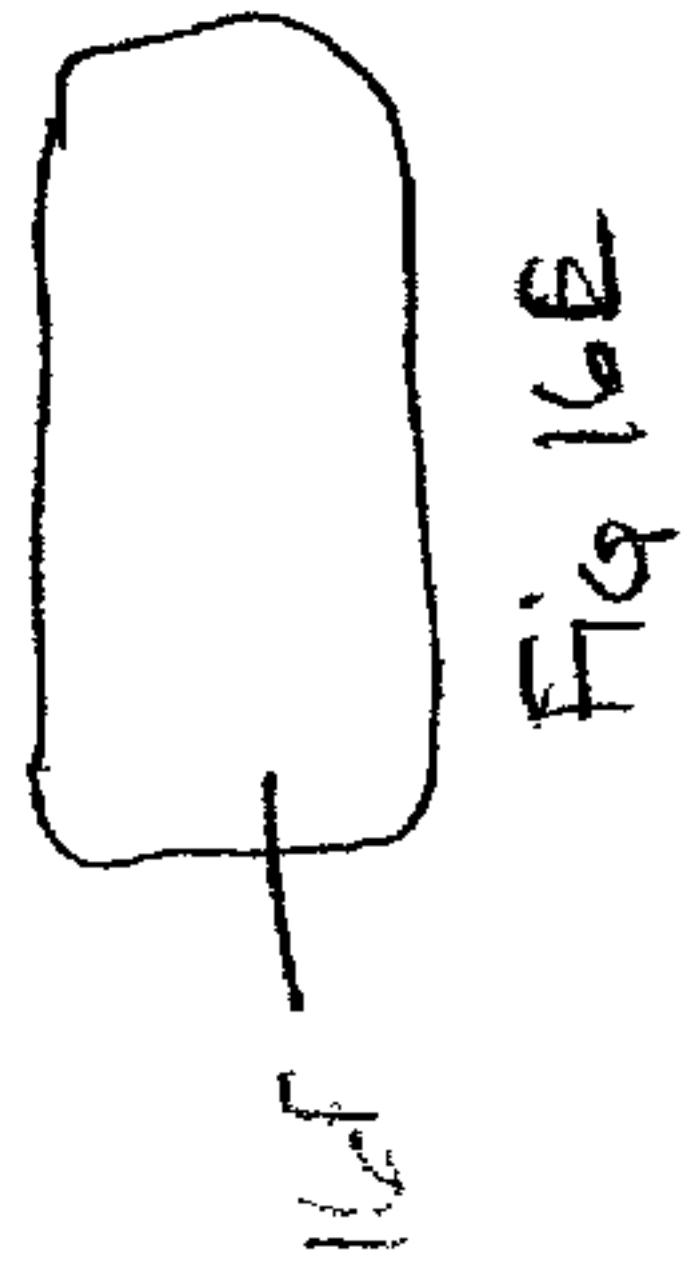
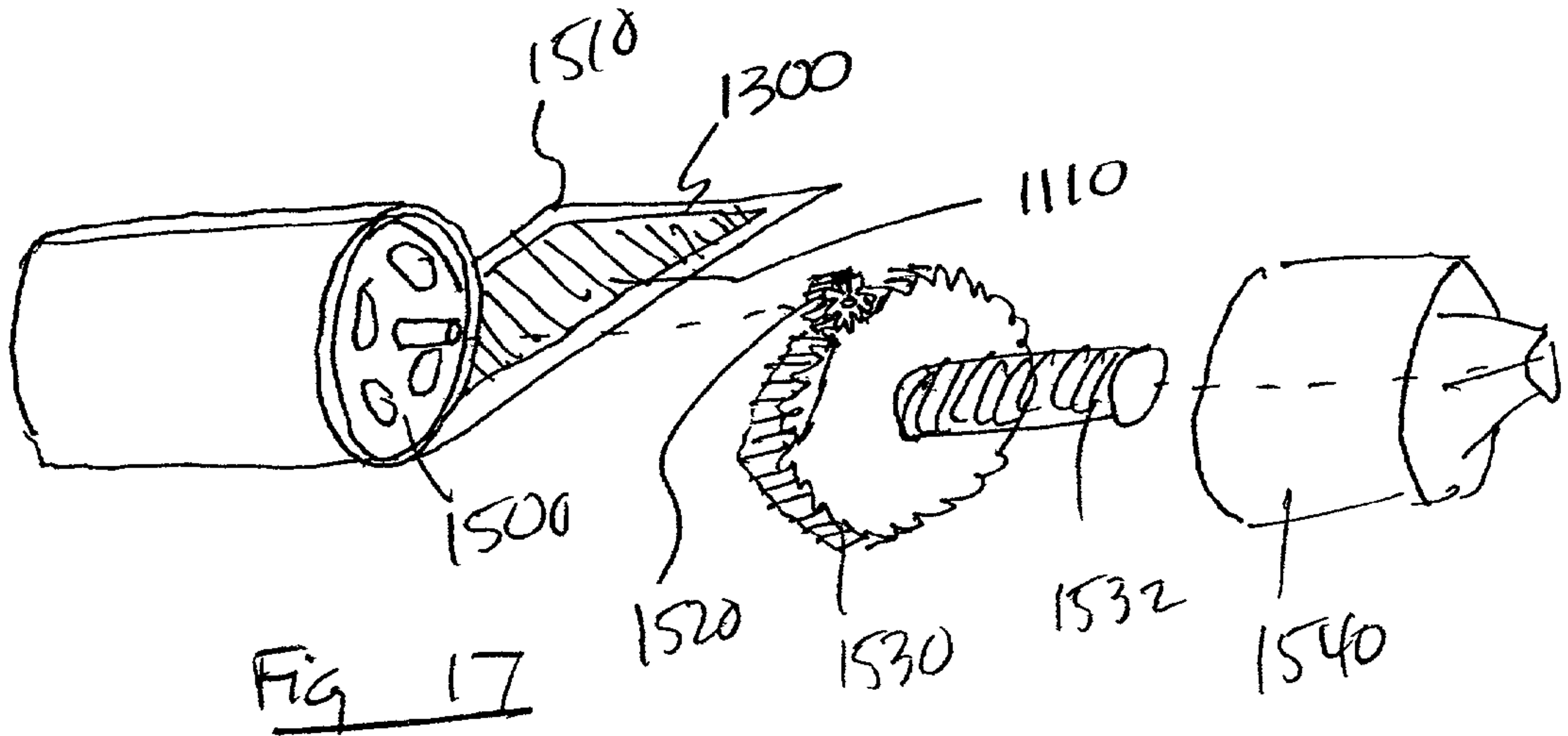
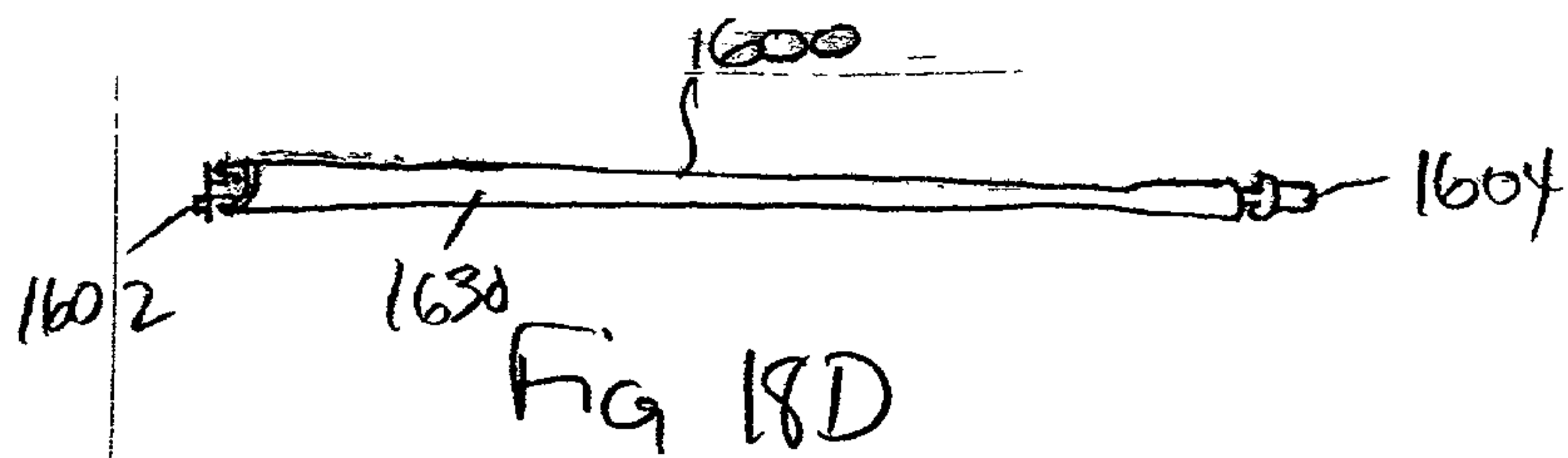
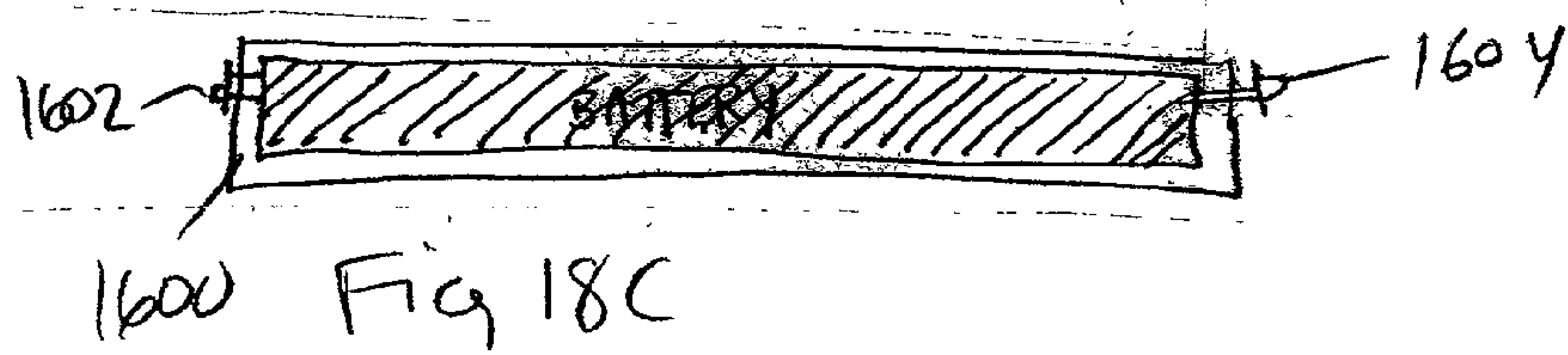
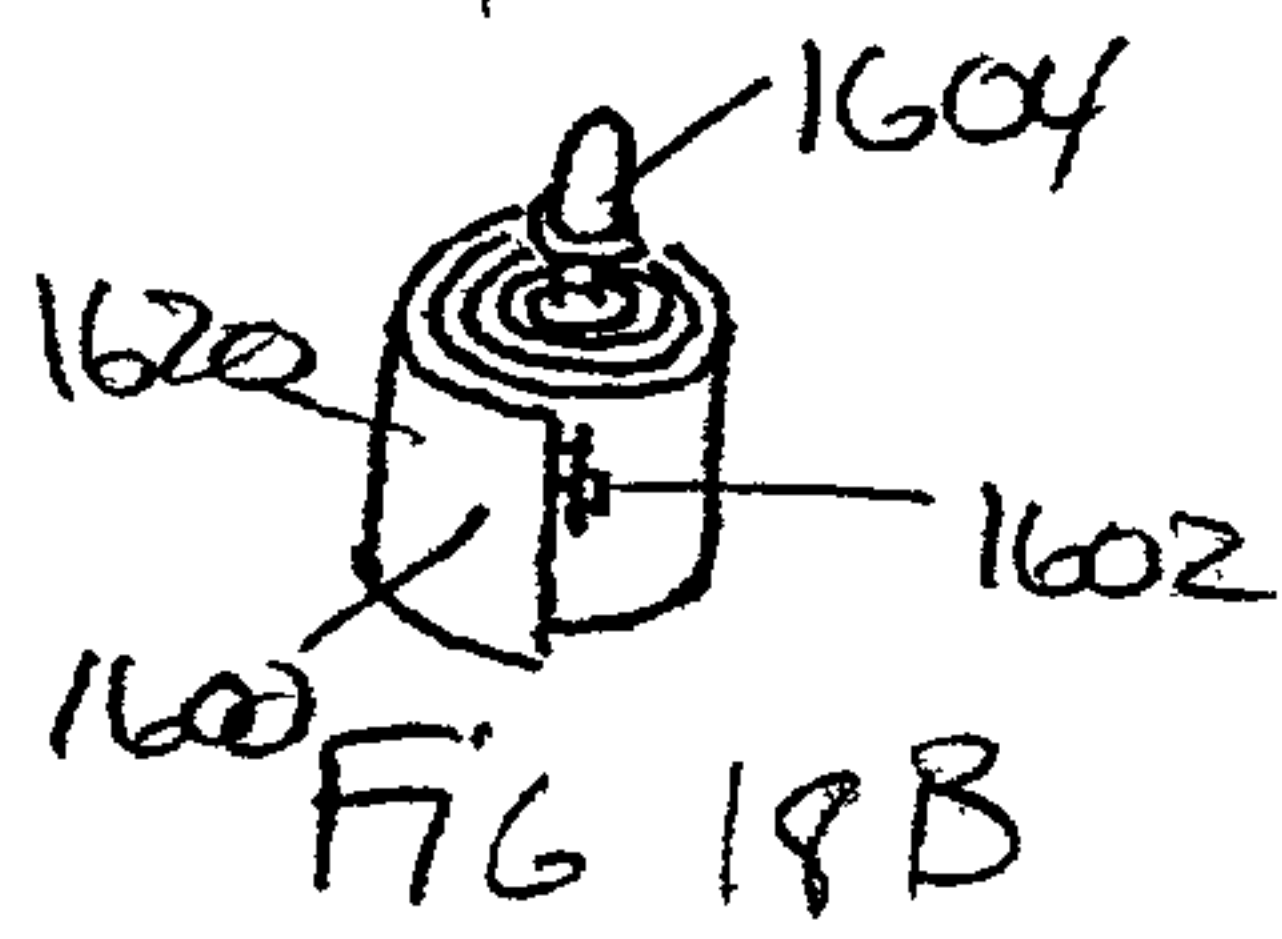
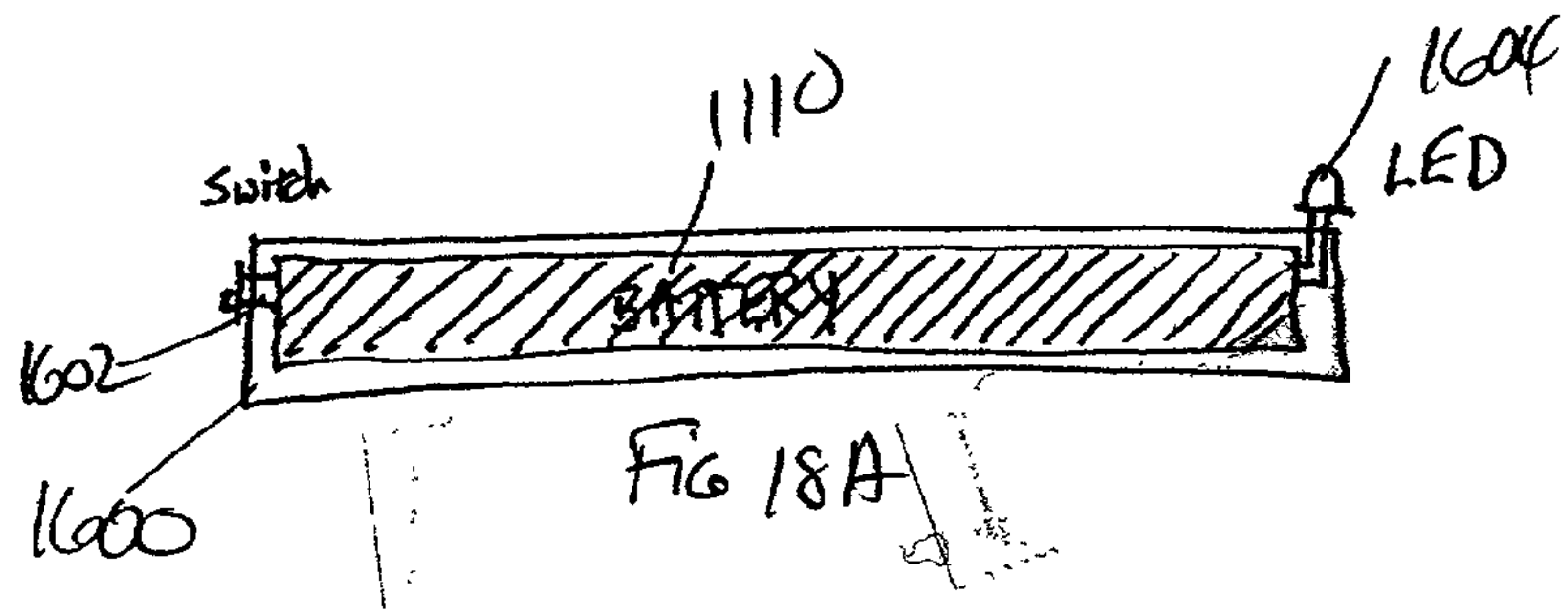


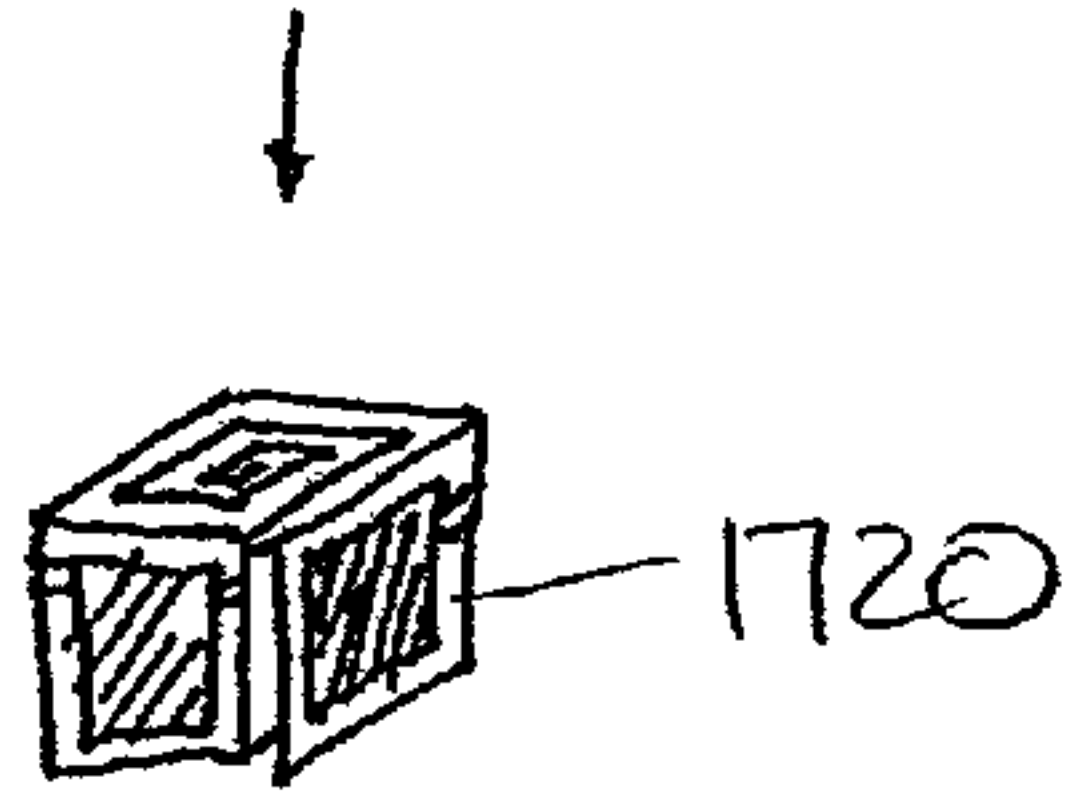
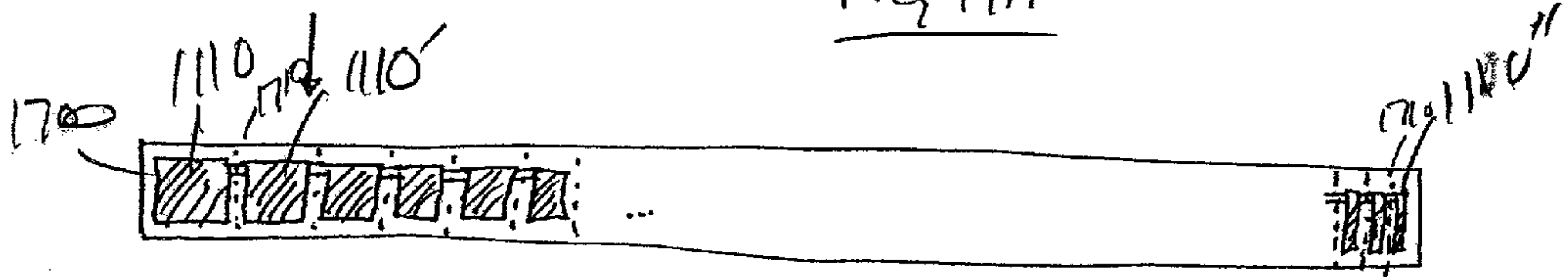
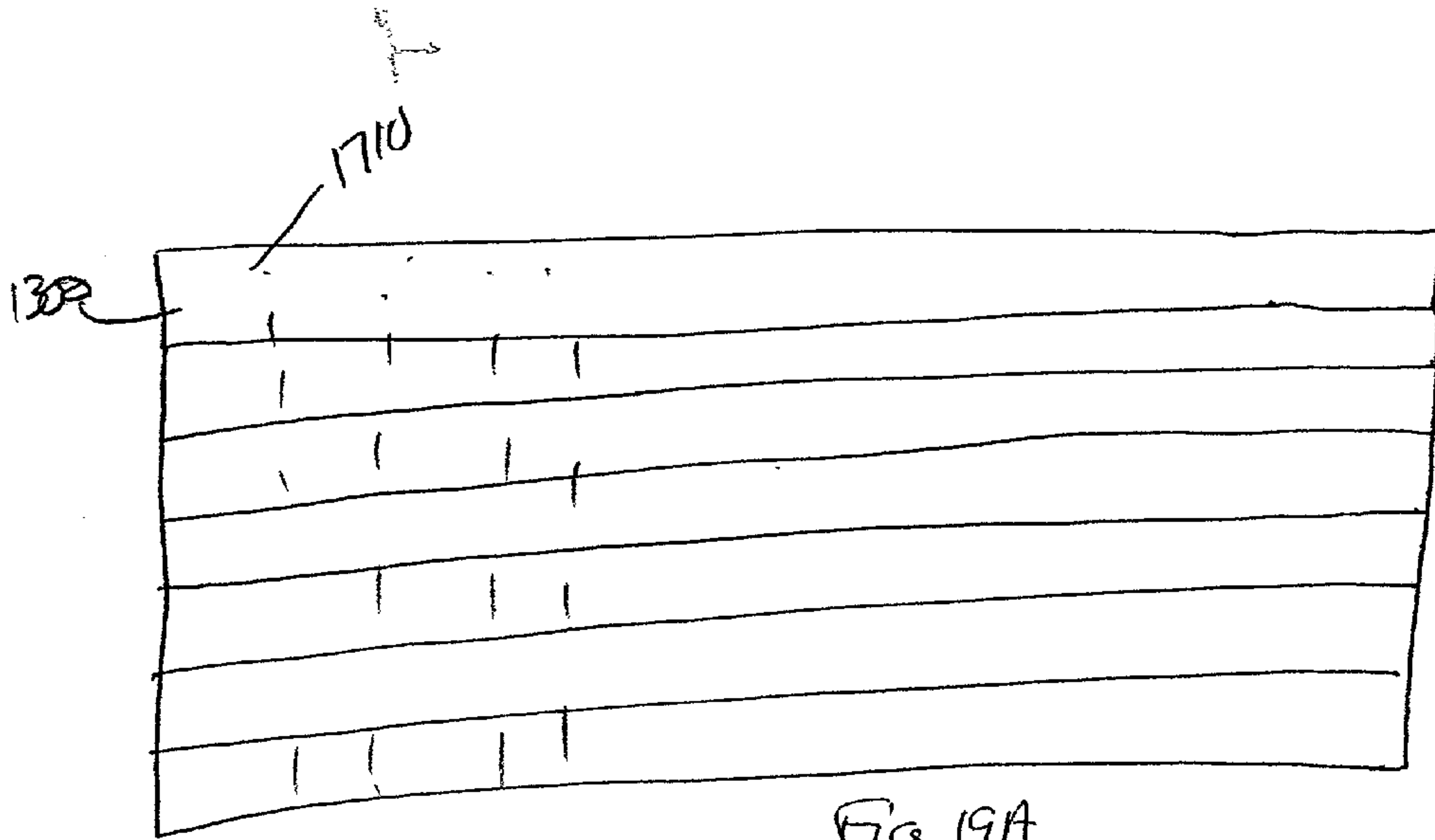
Fig 16E



Fig 16D







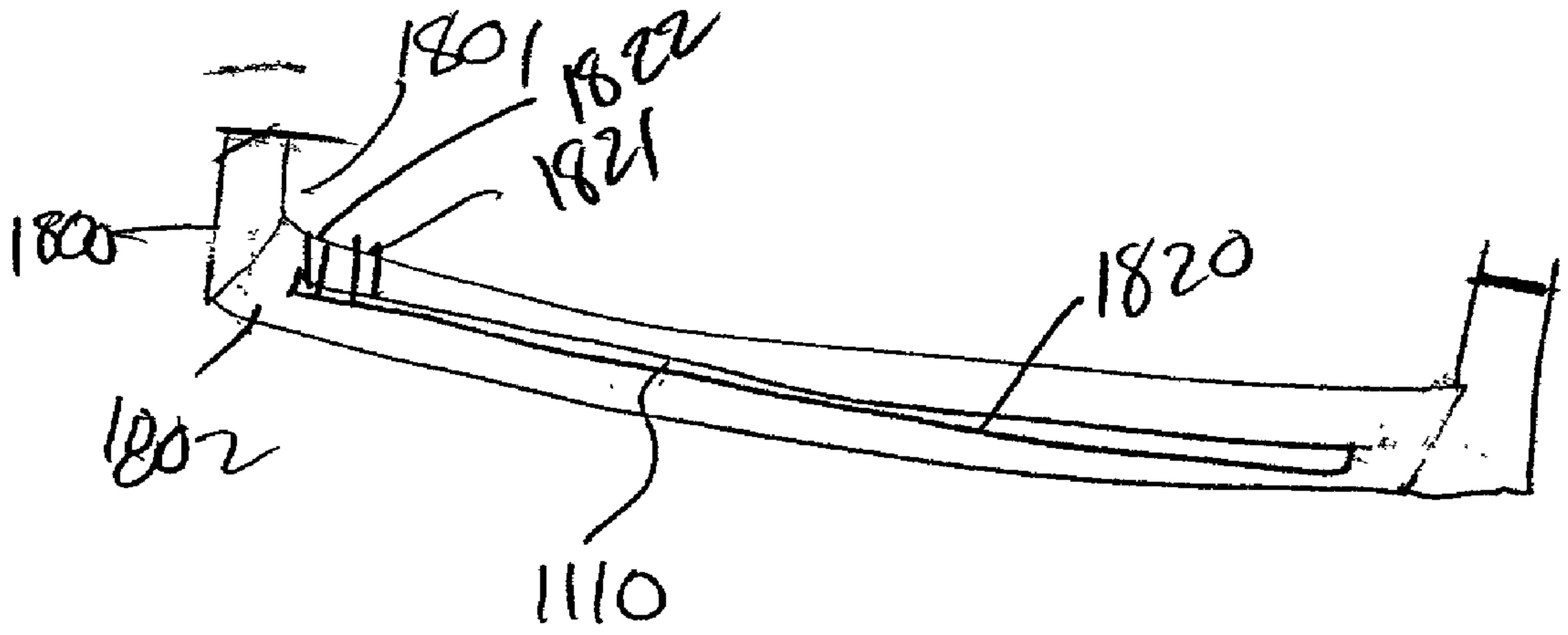


Fig 20

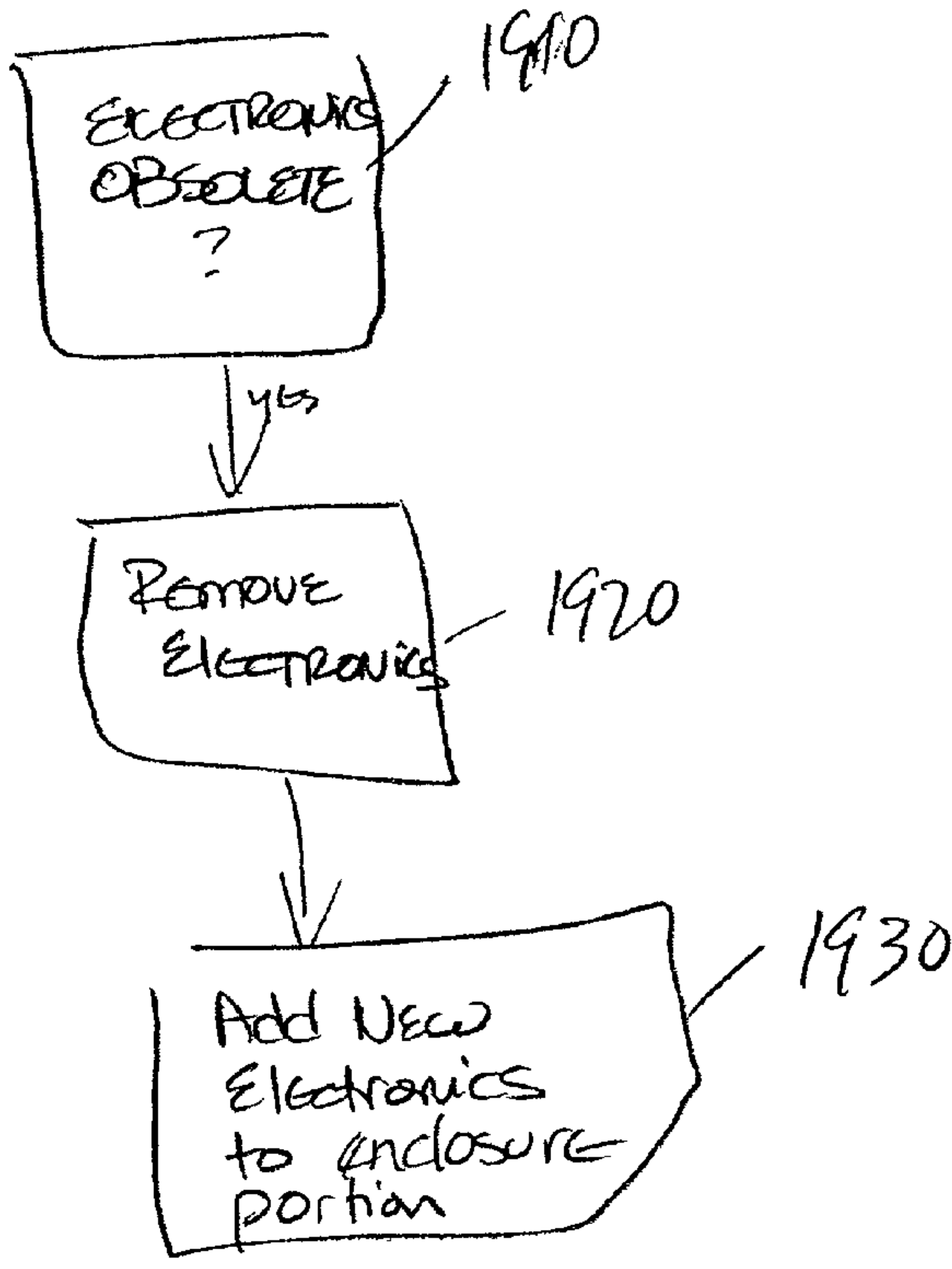


Fig 21 A

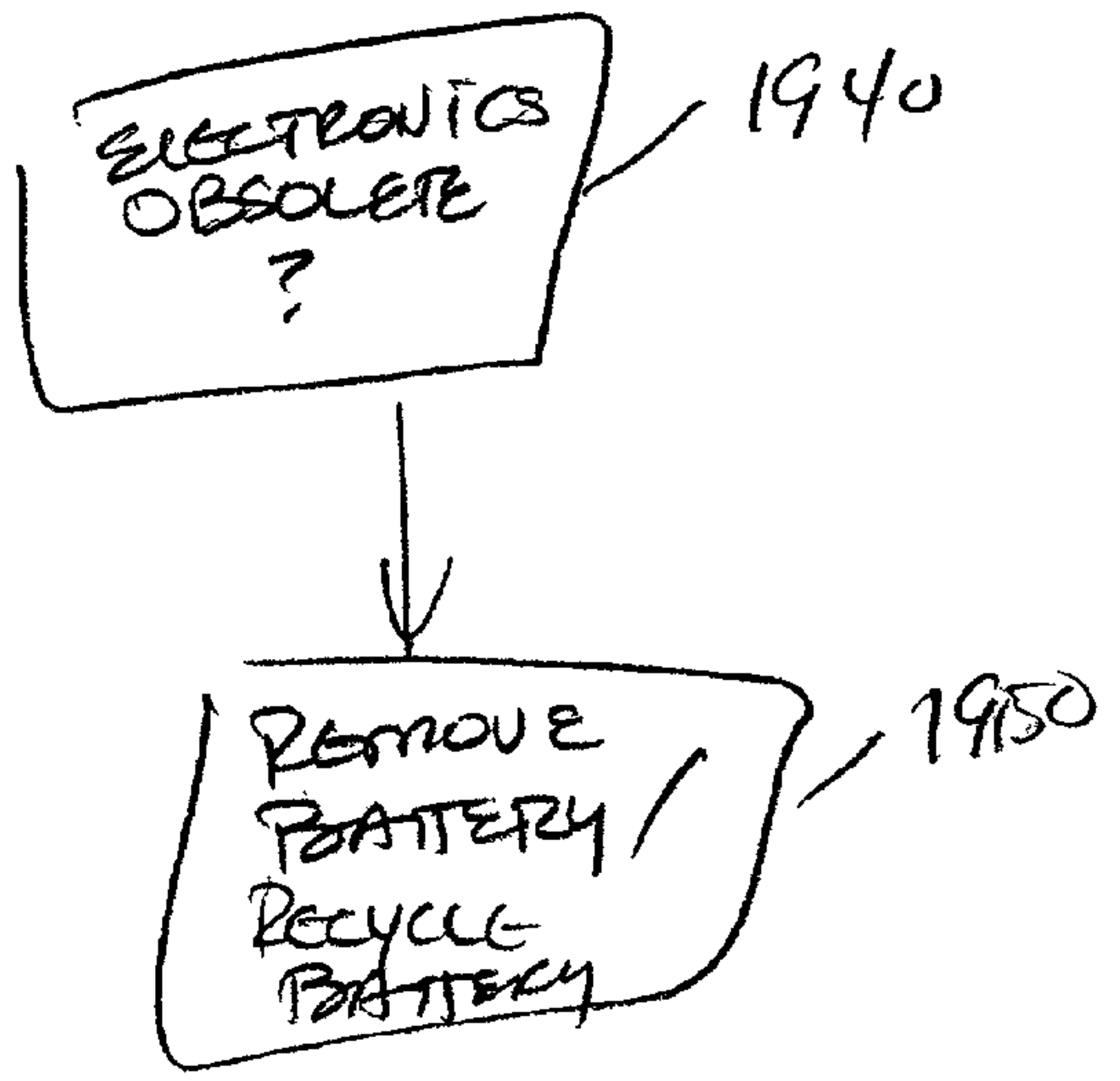
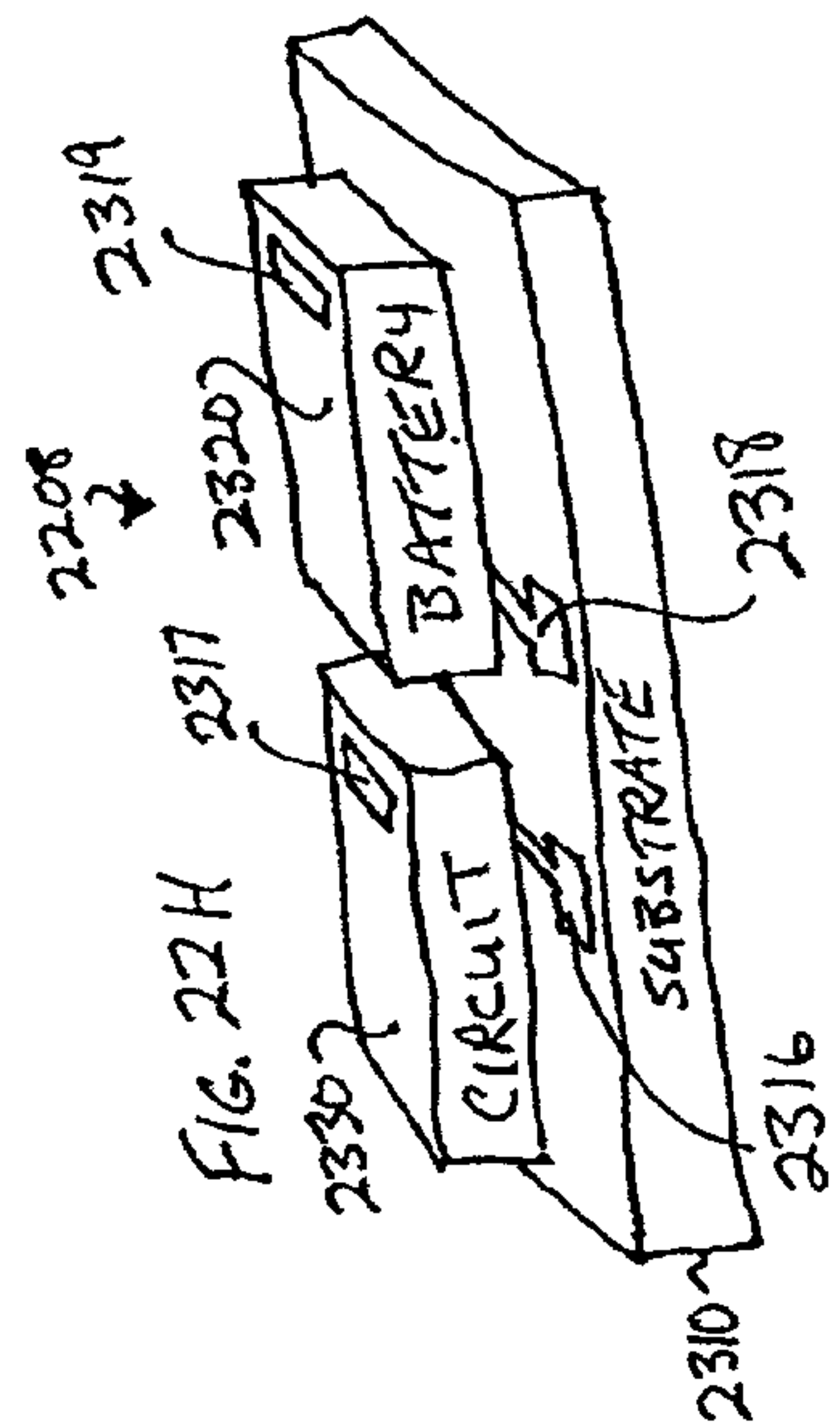
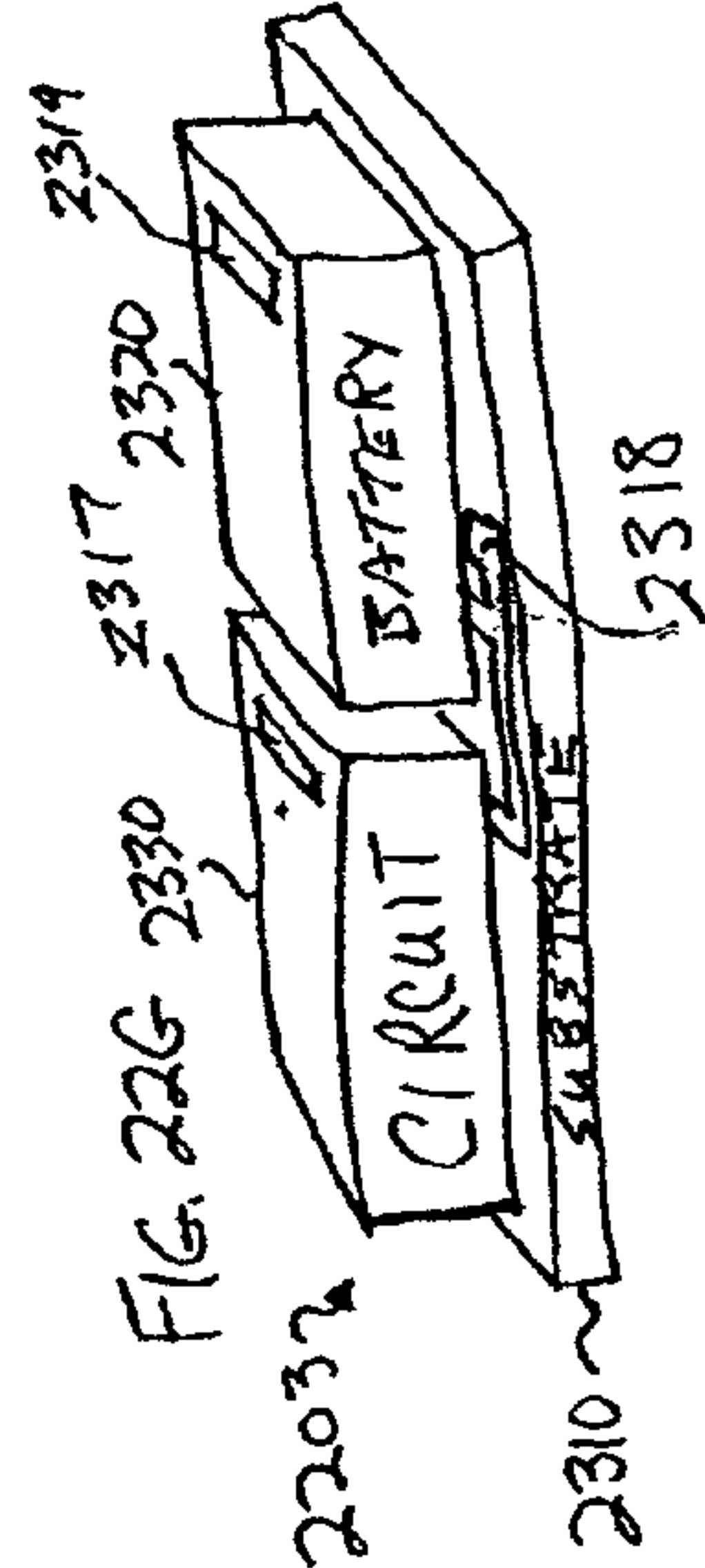
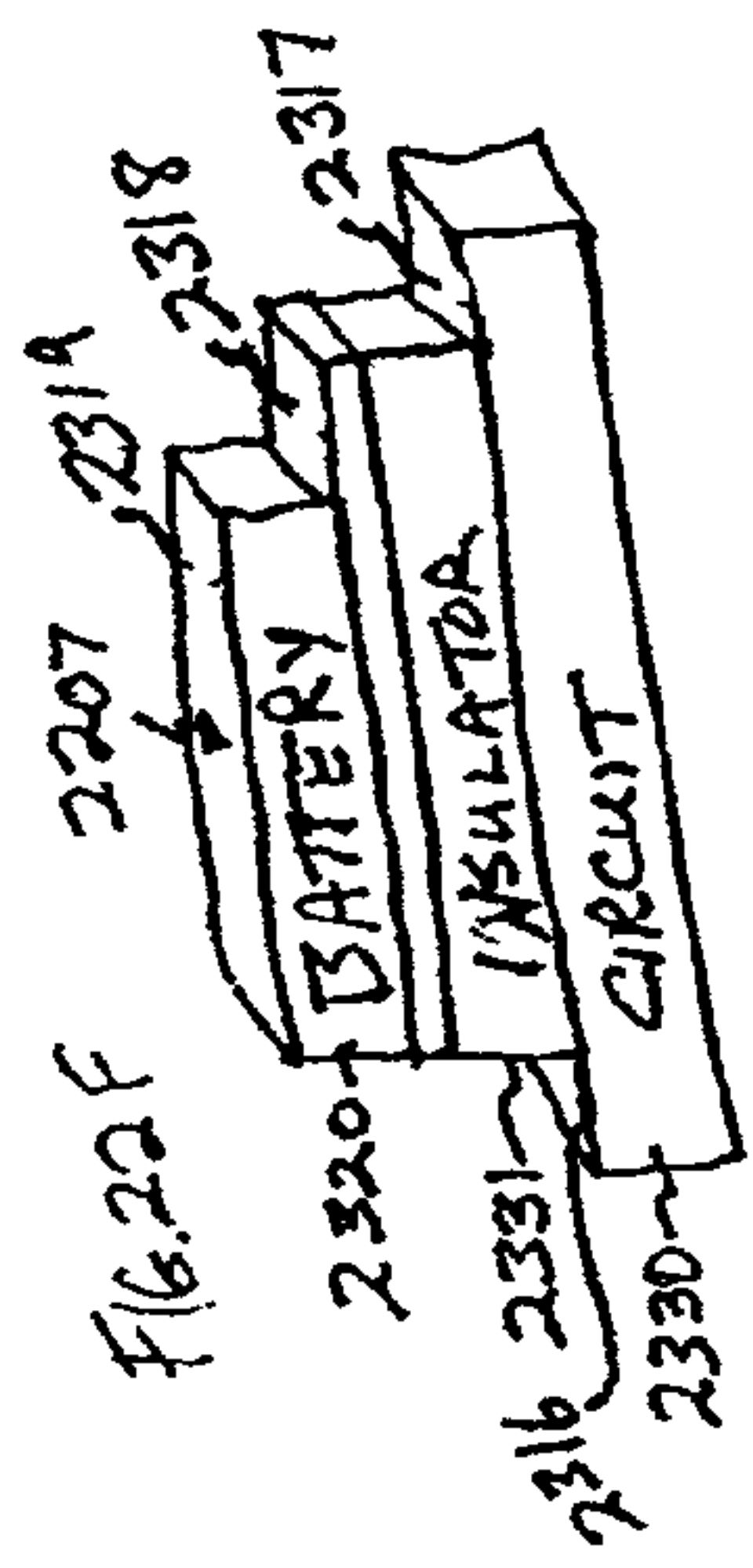
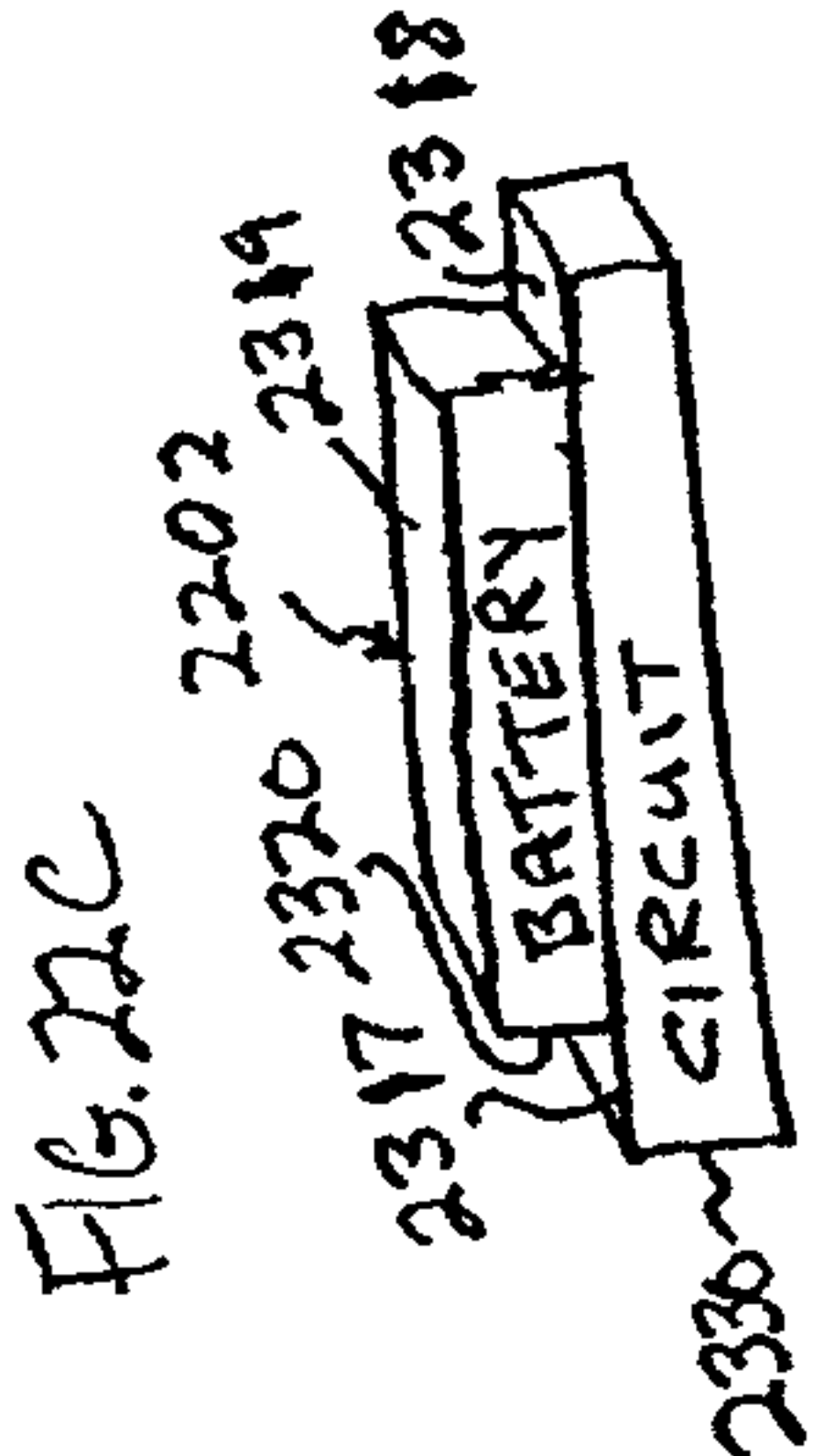
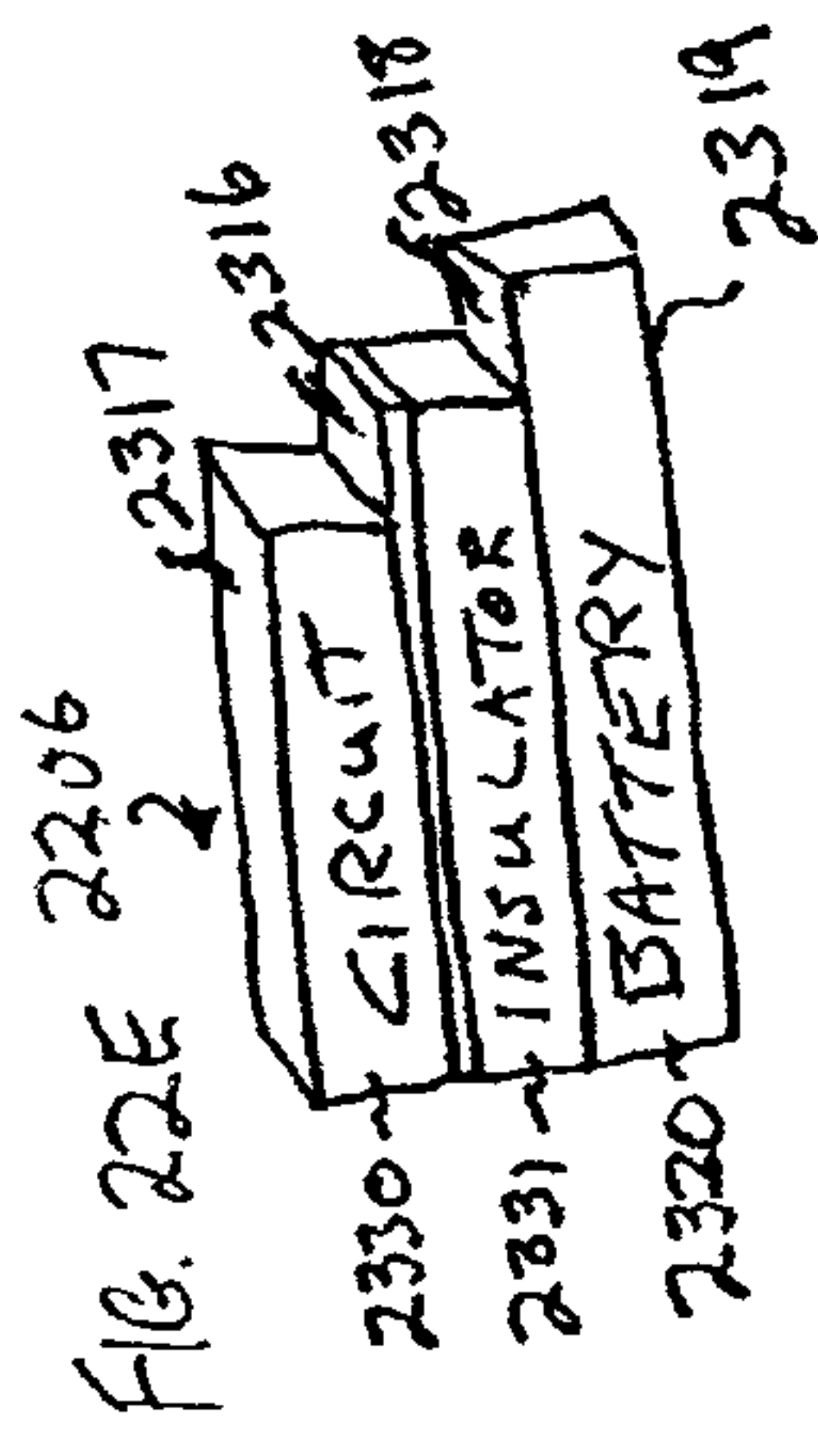
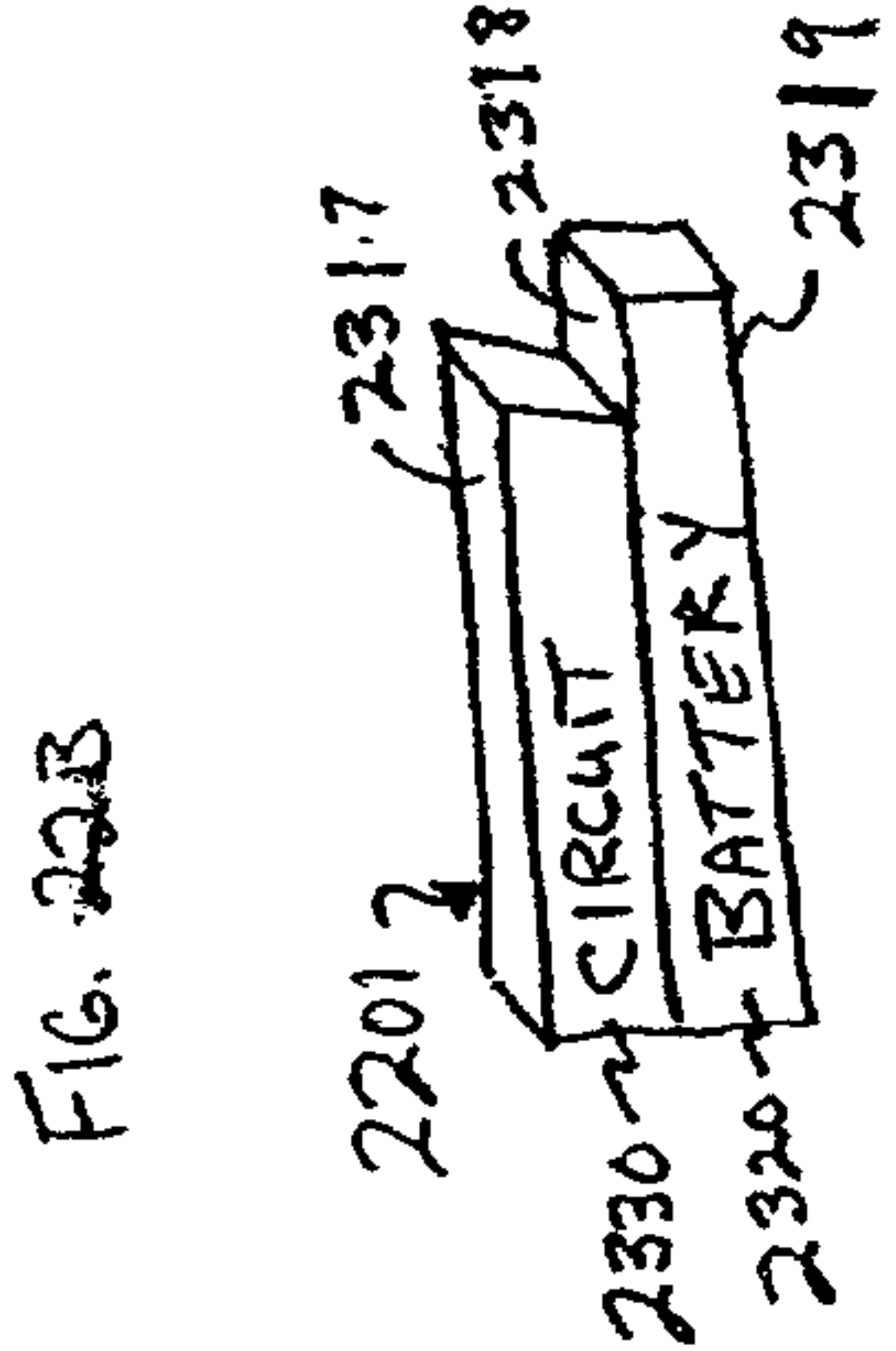
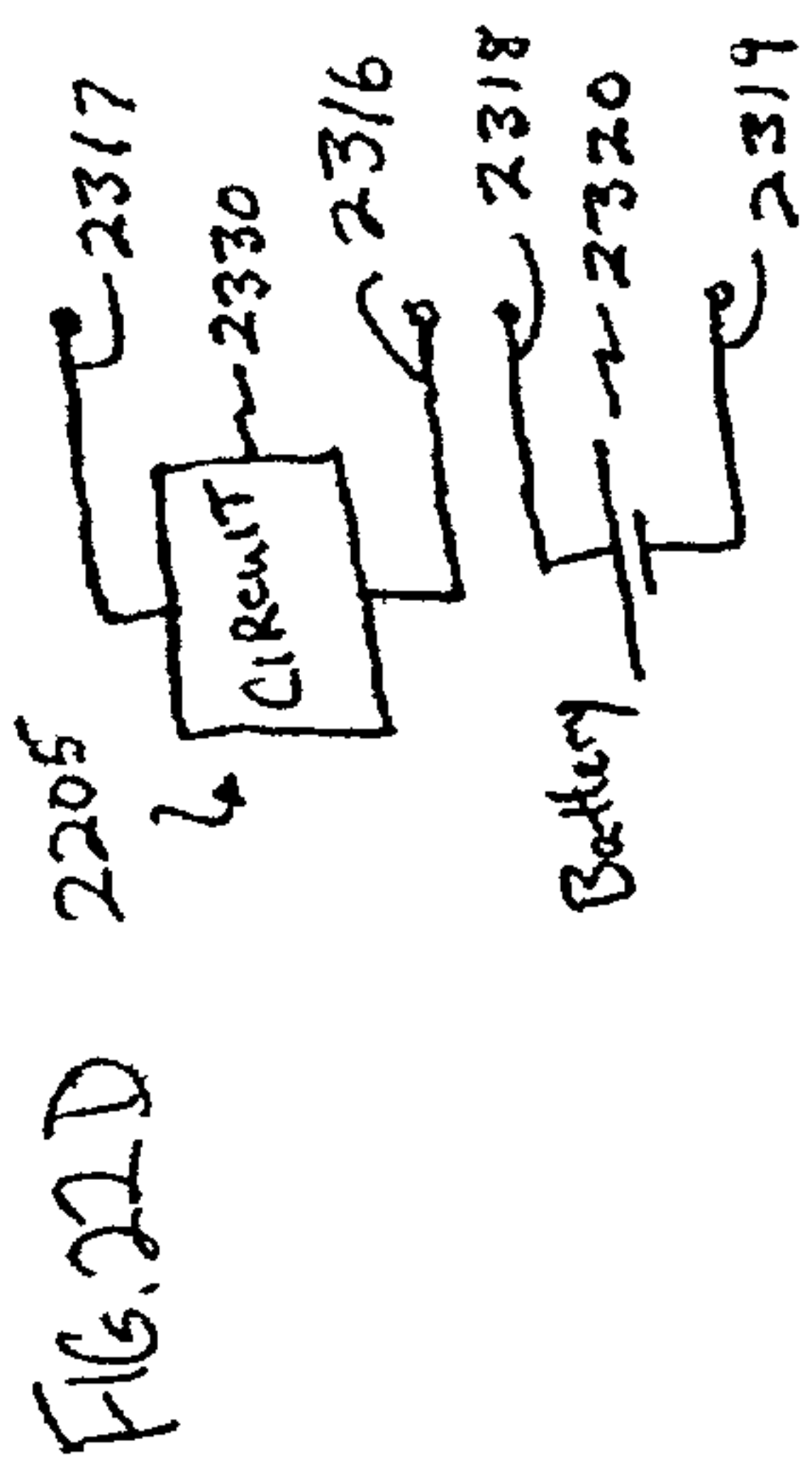
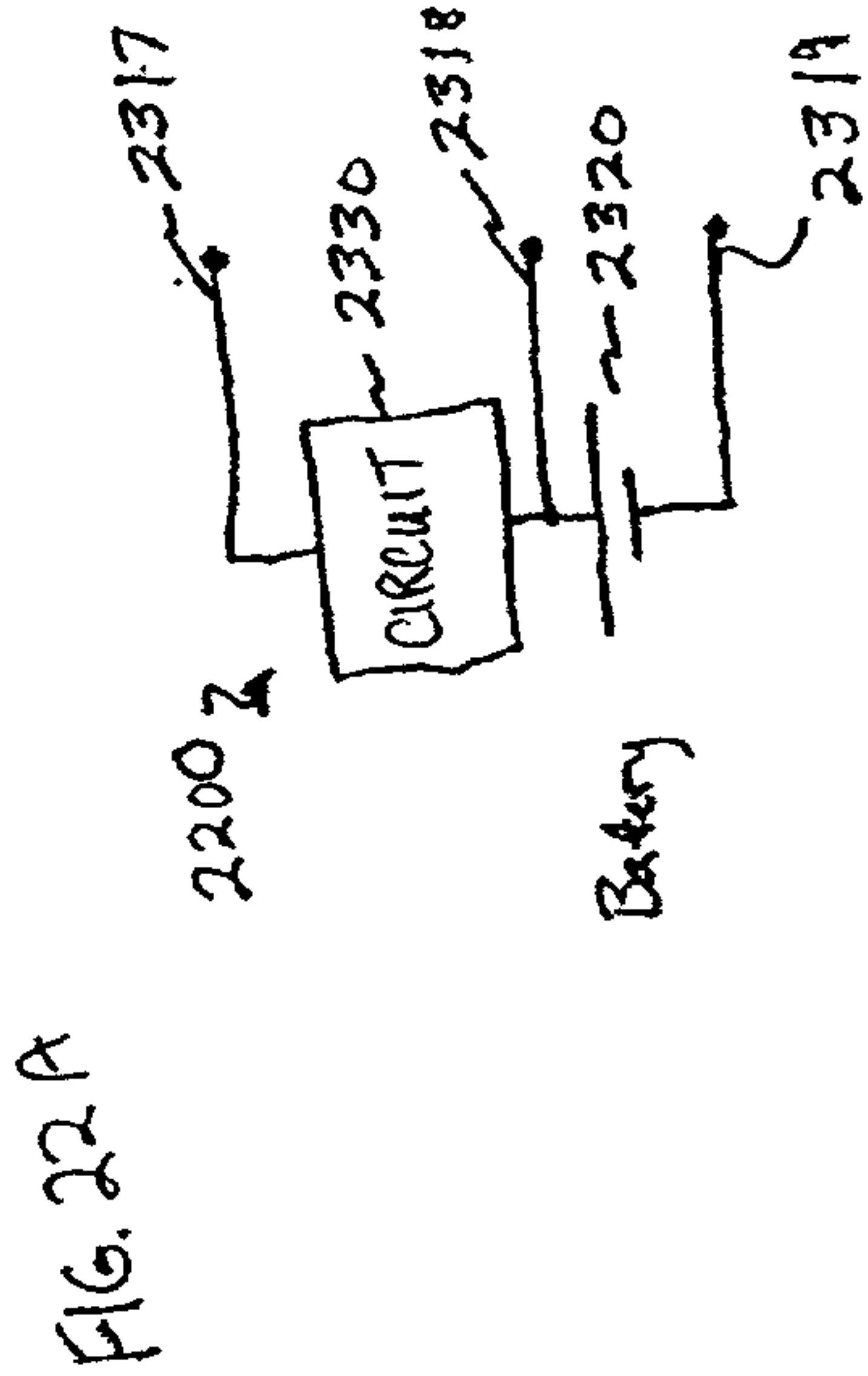
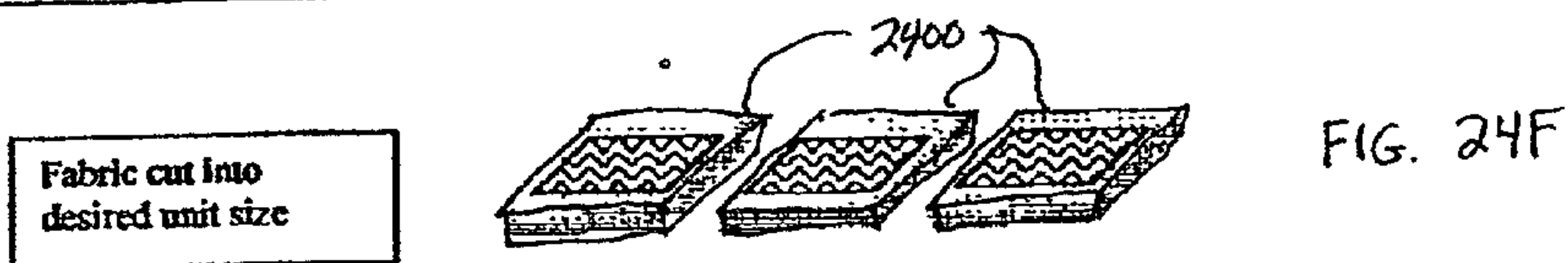
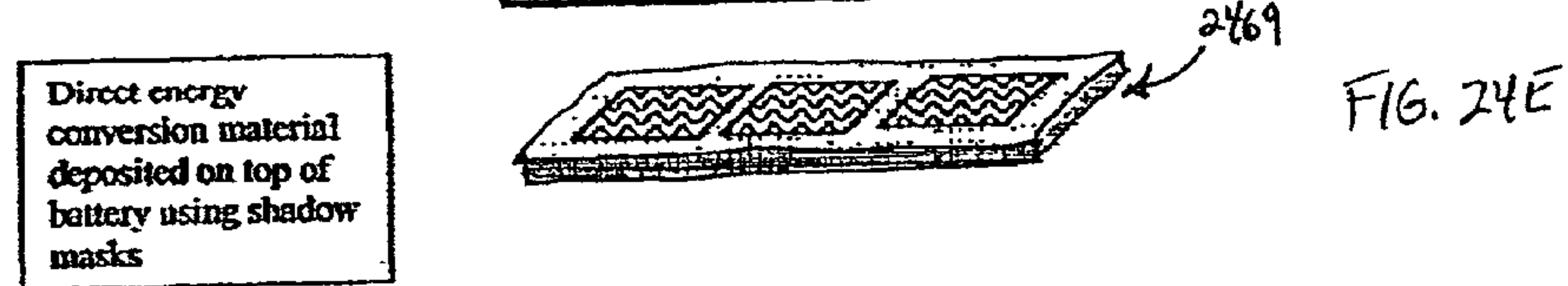
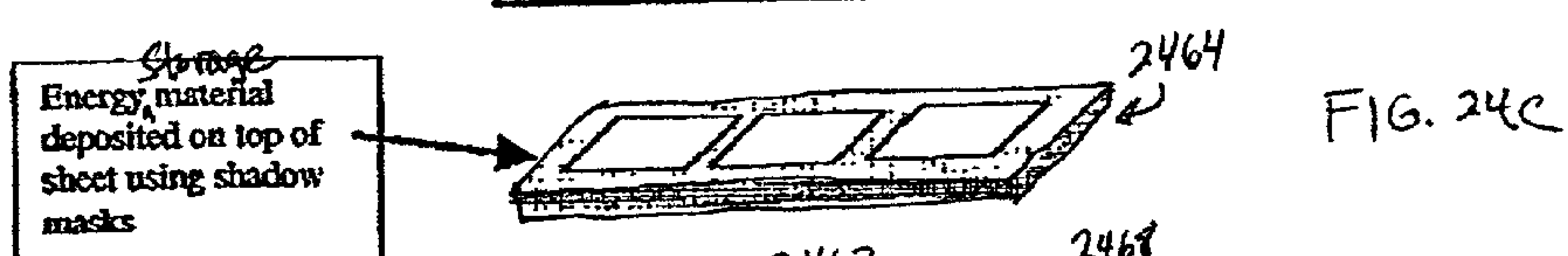
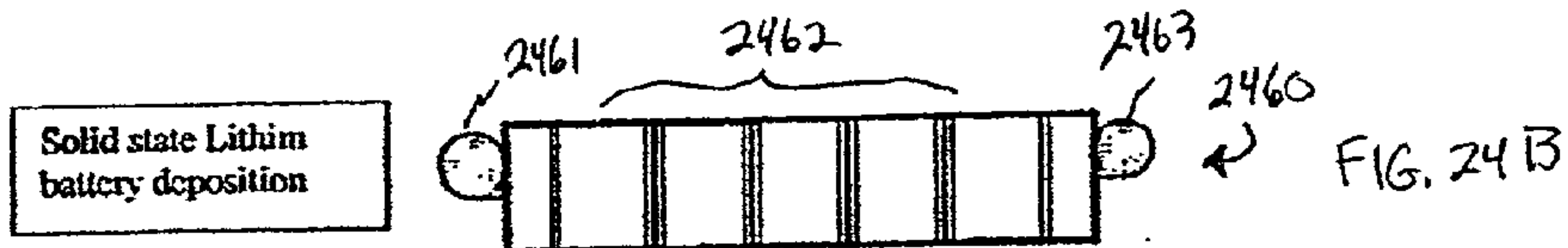
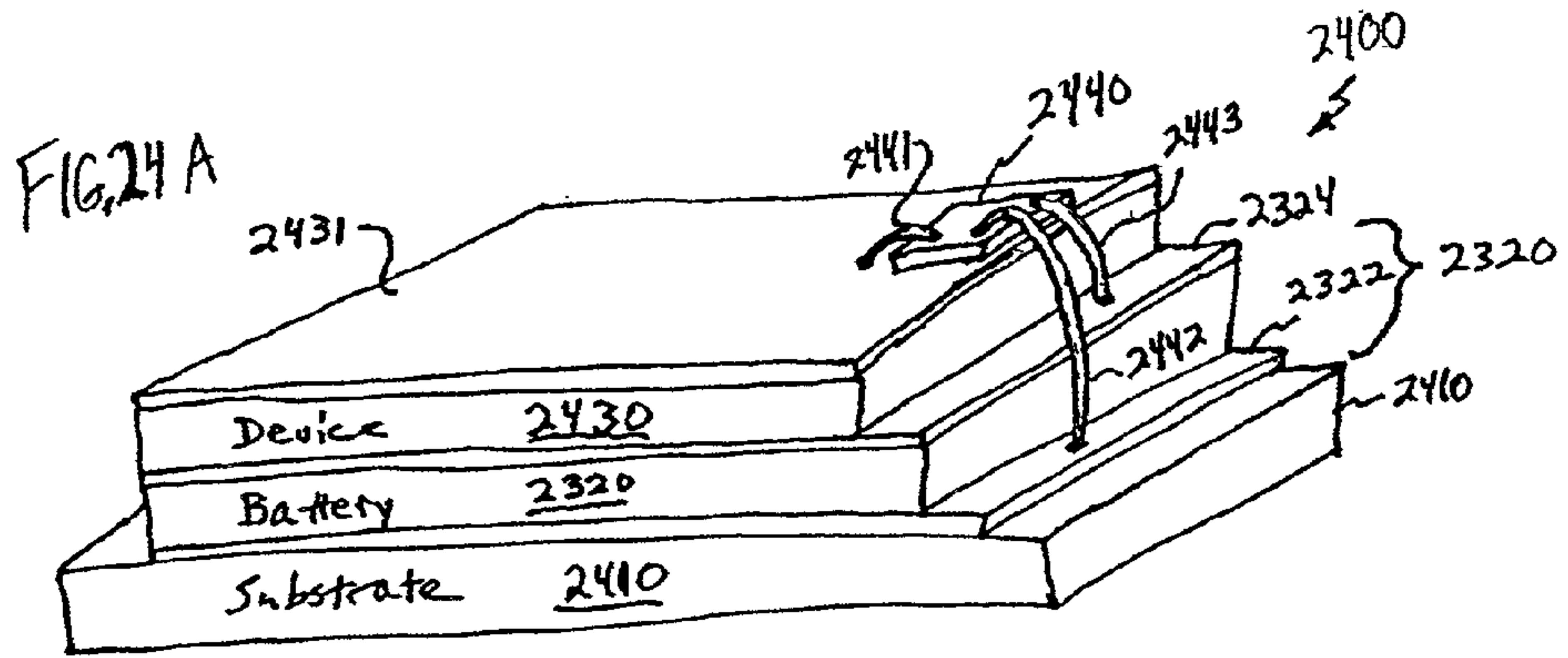
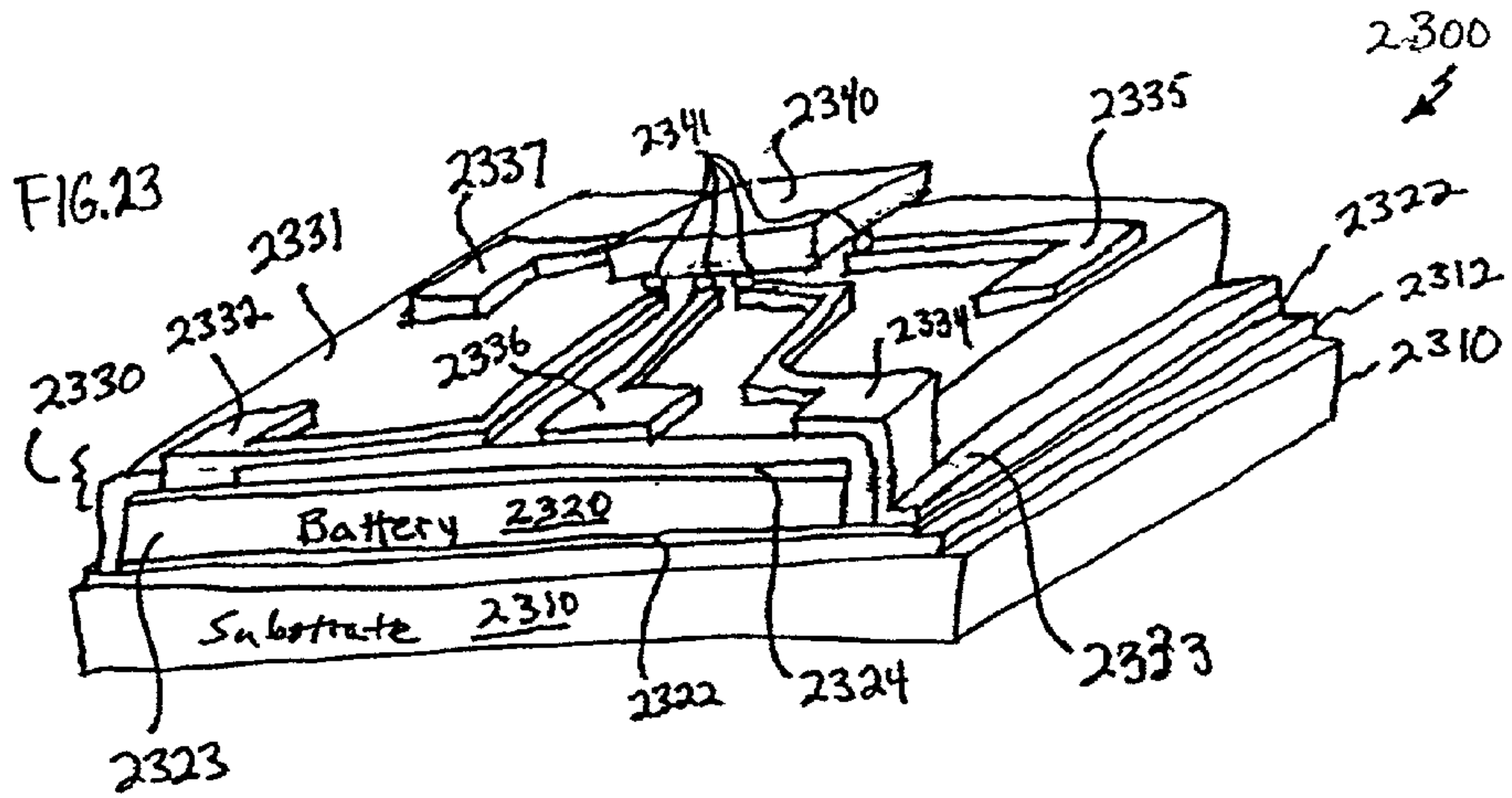
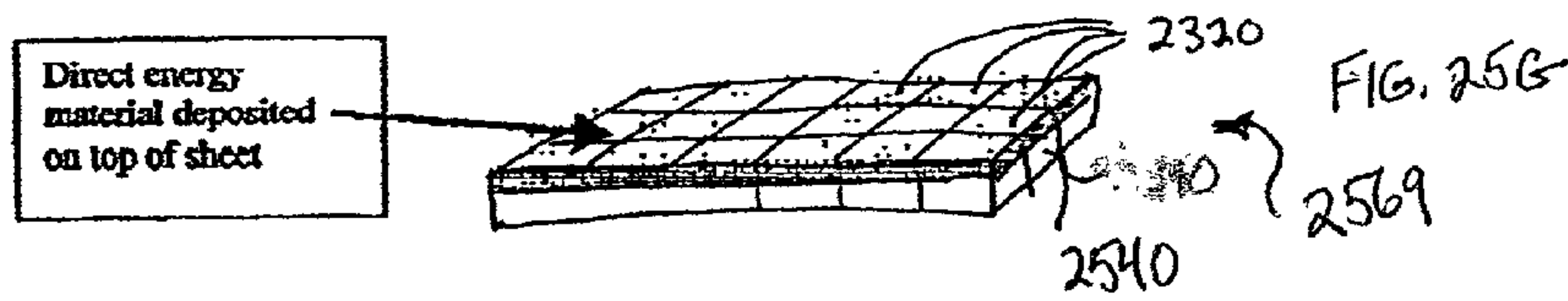
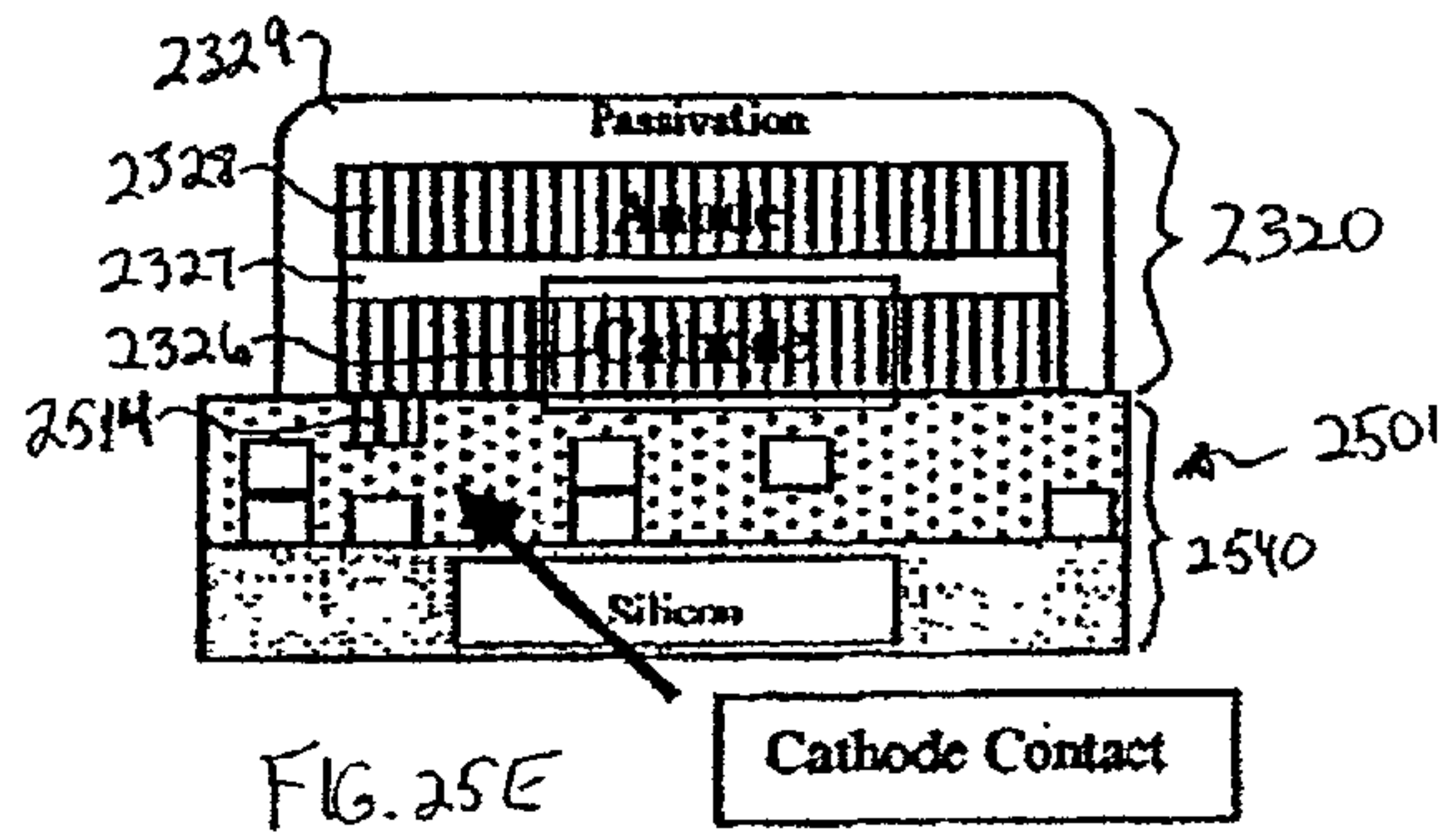
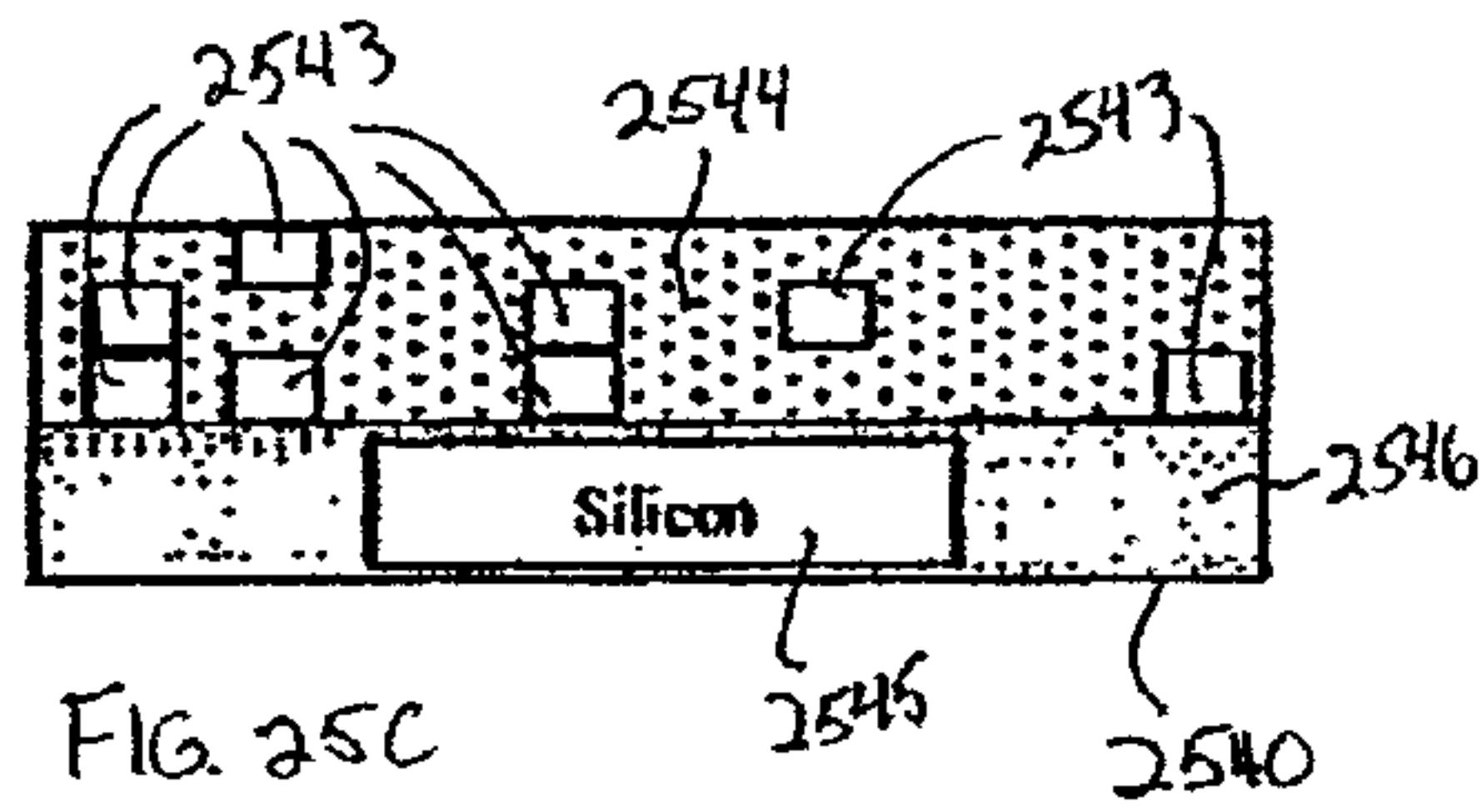
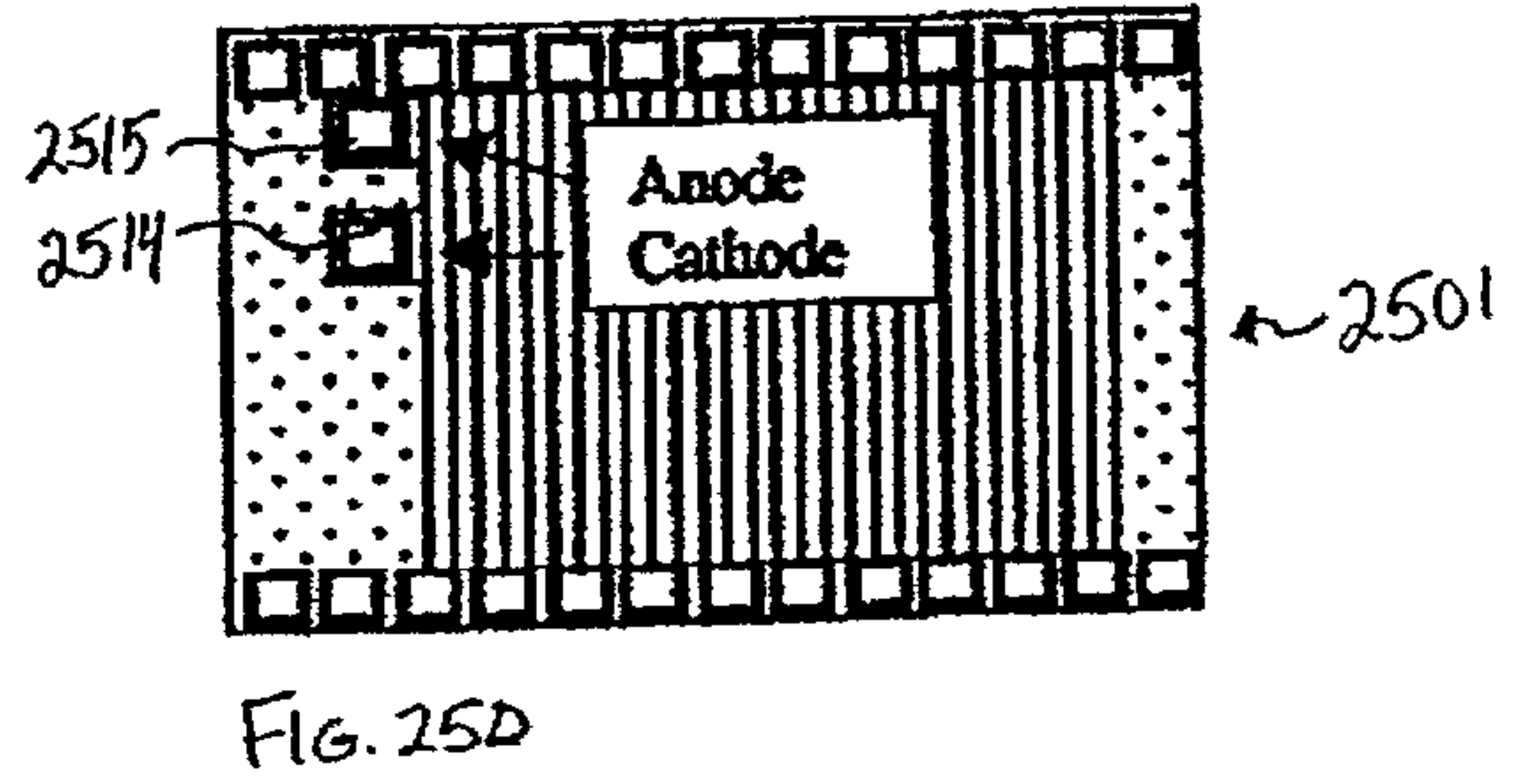
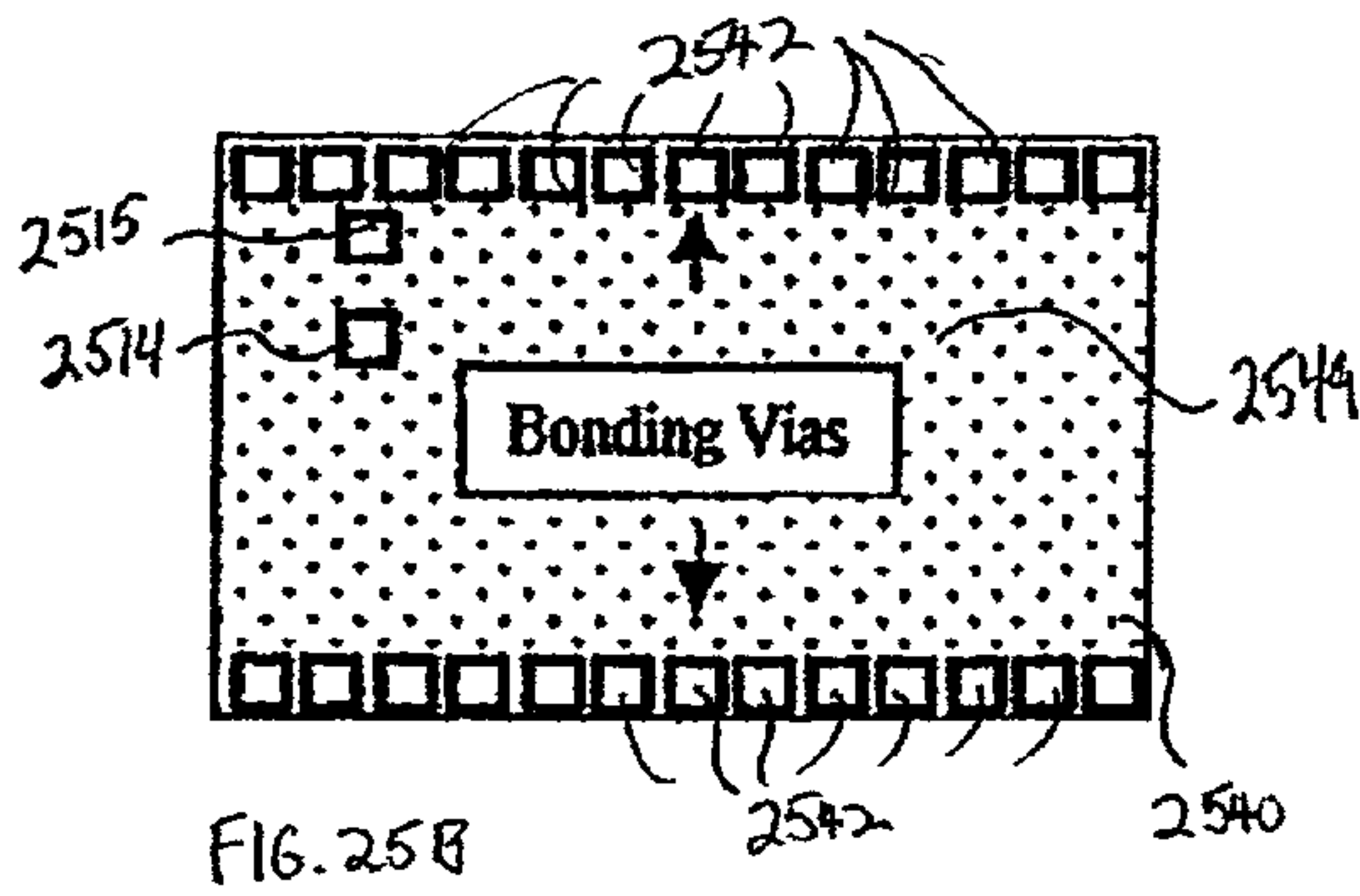
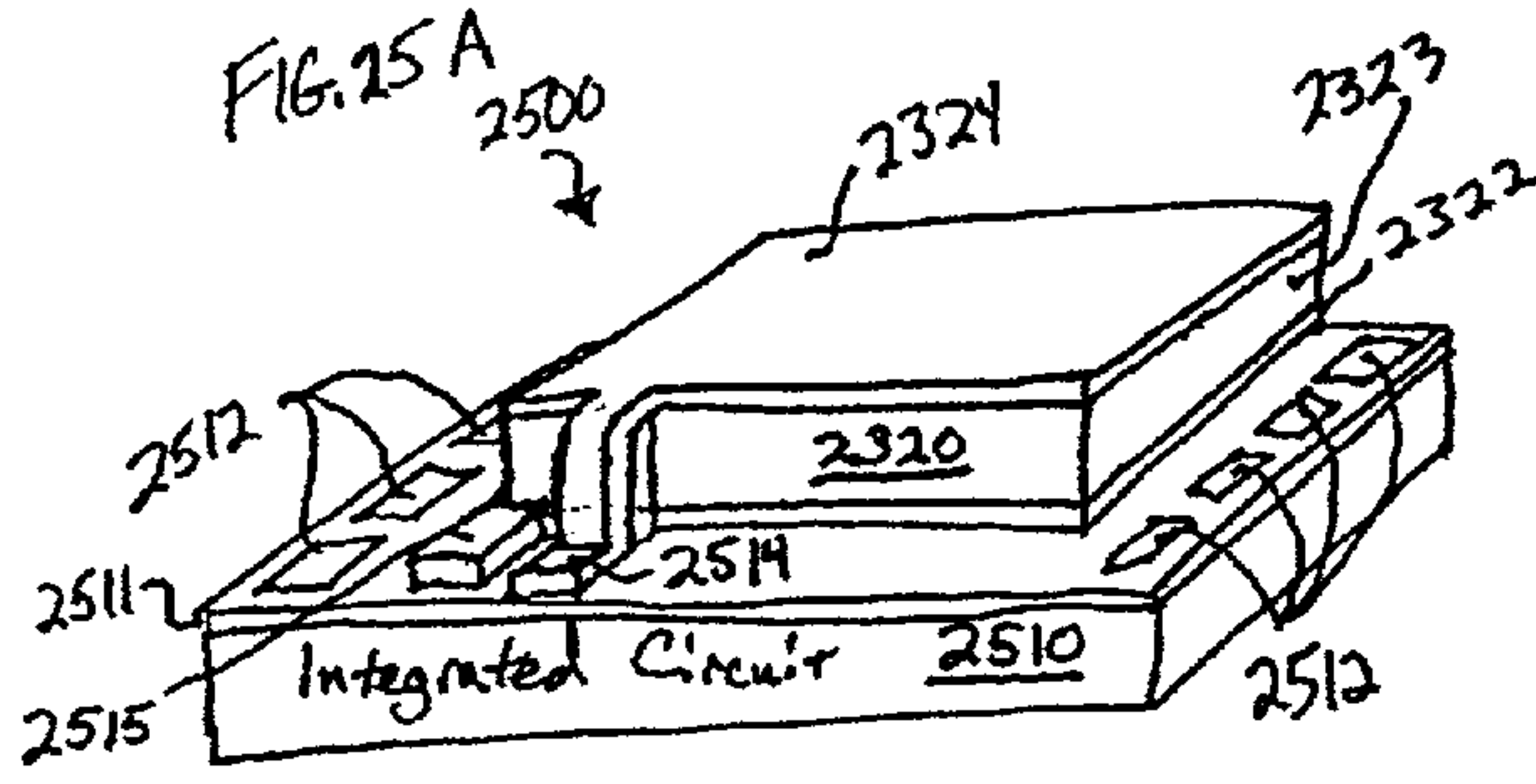
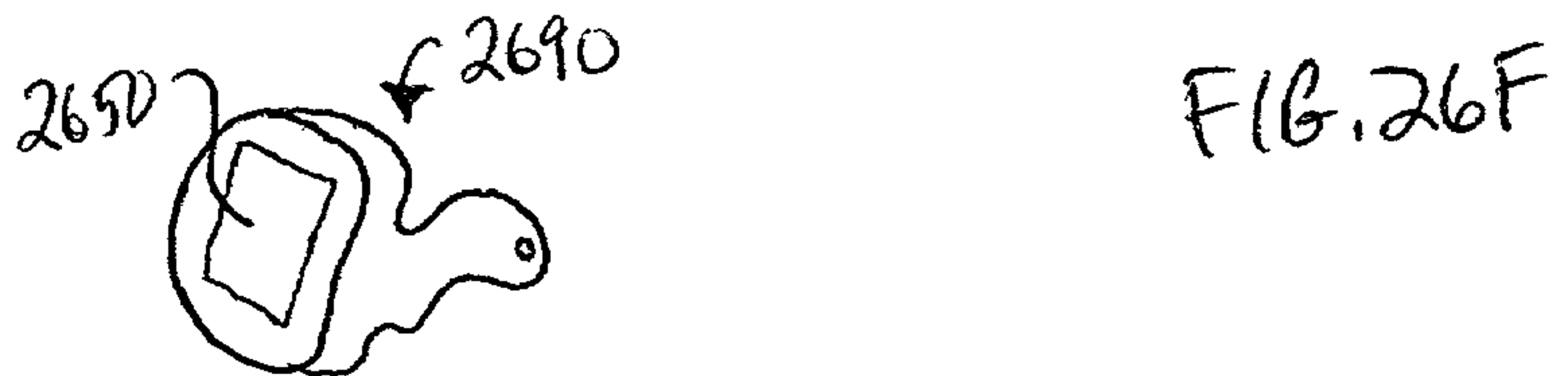
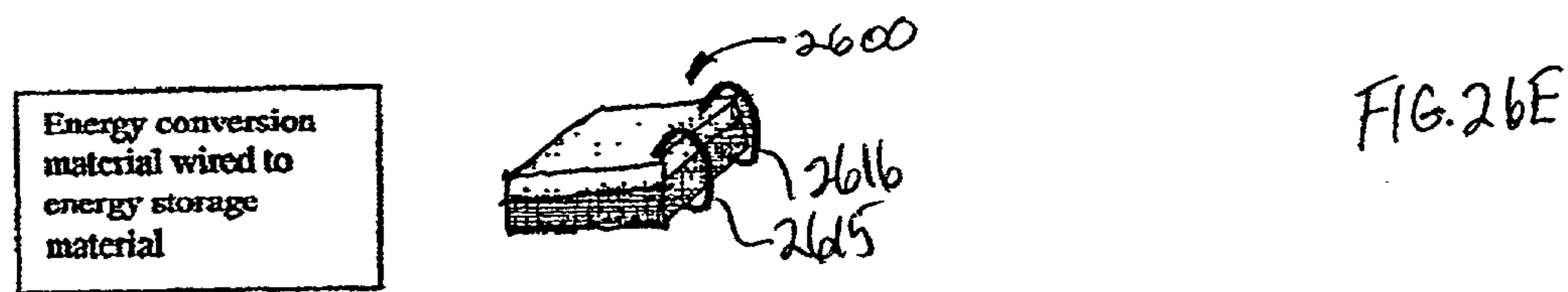
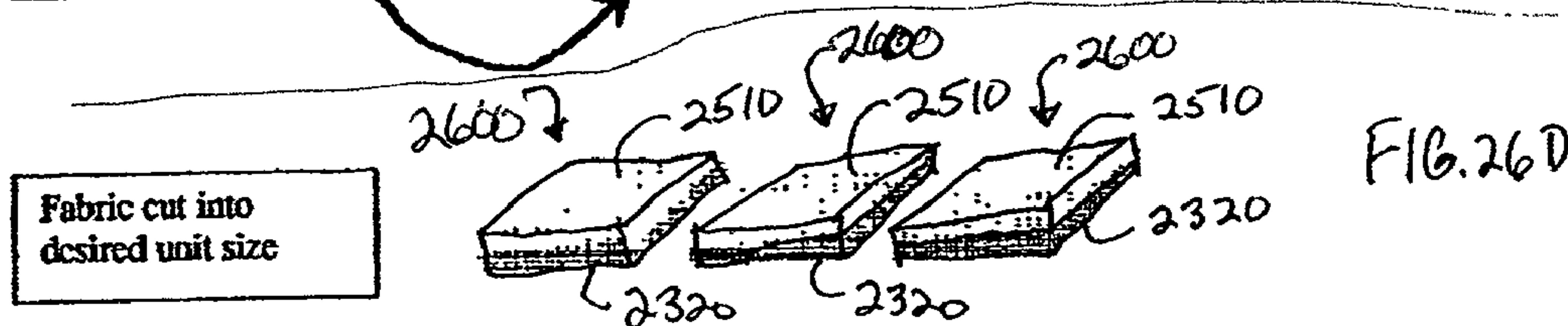
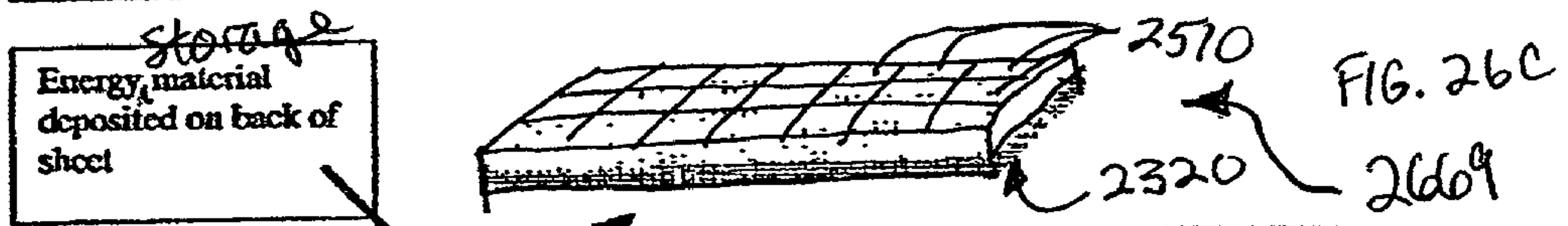
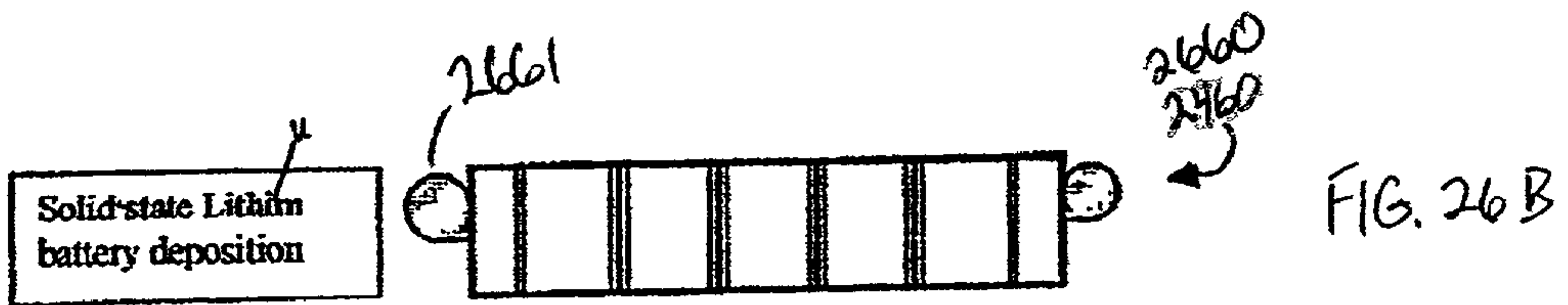
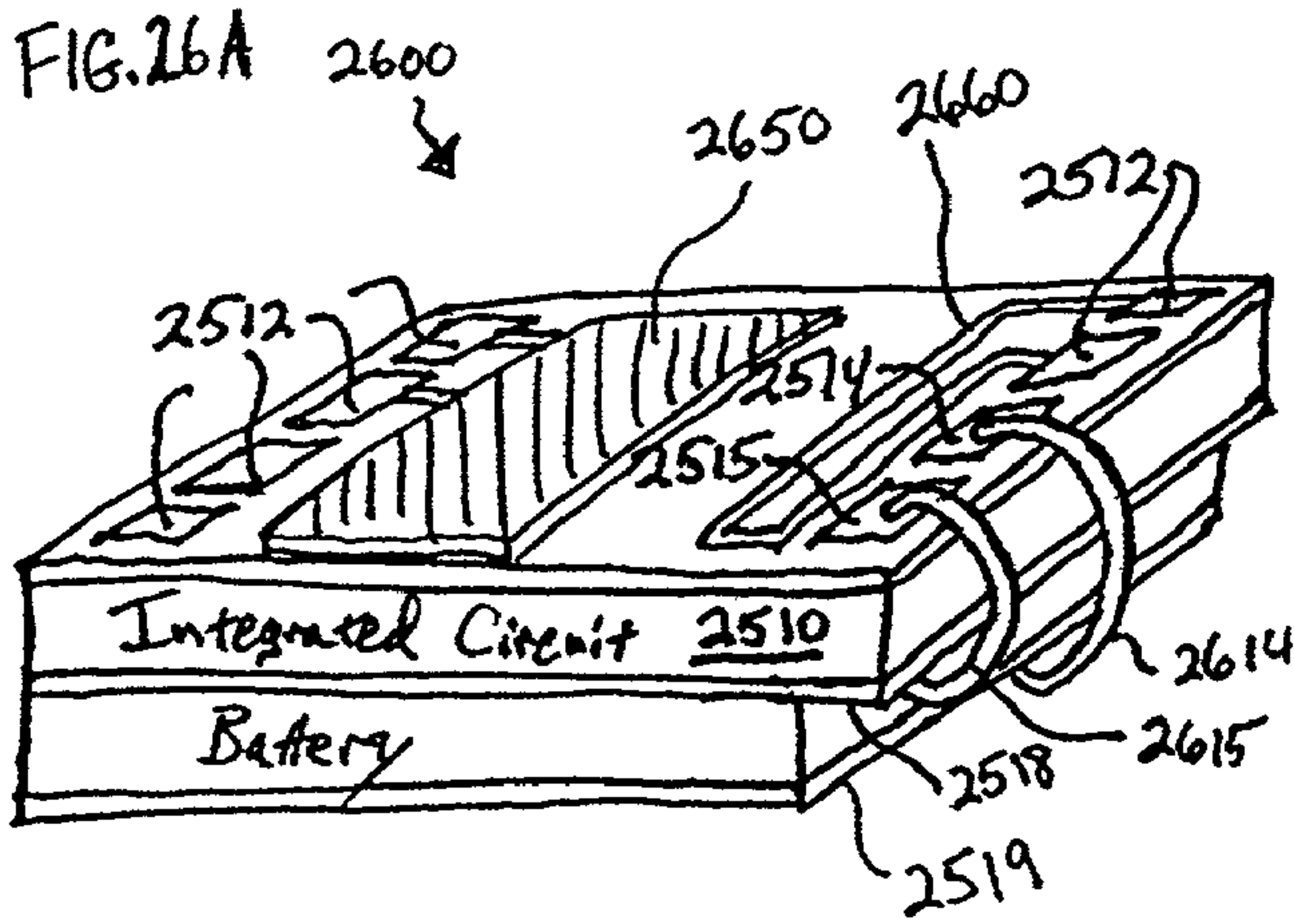


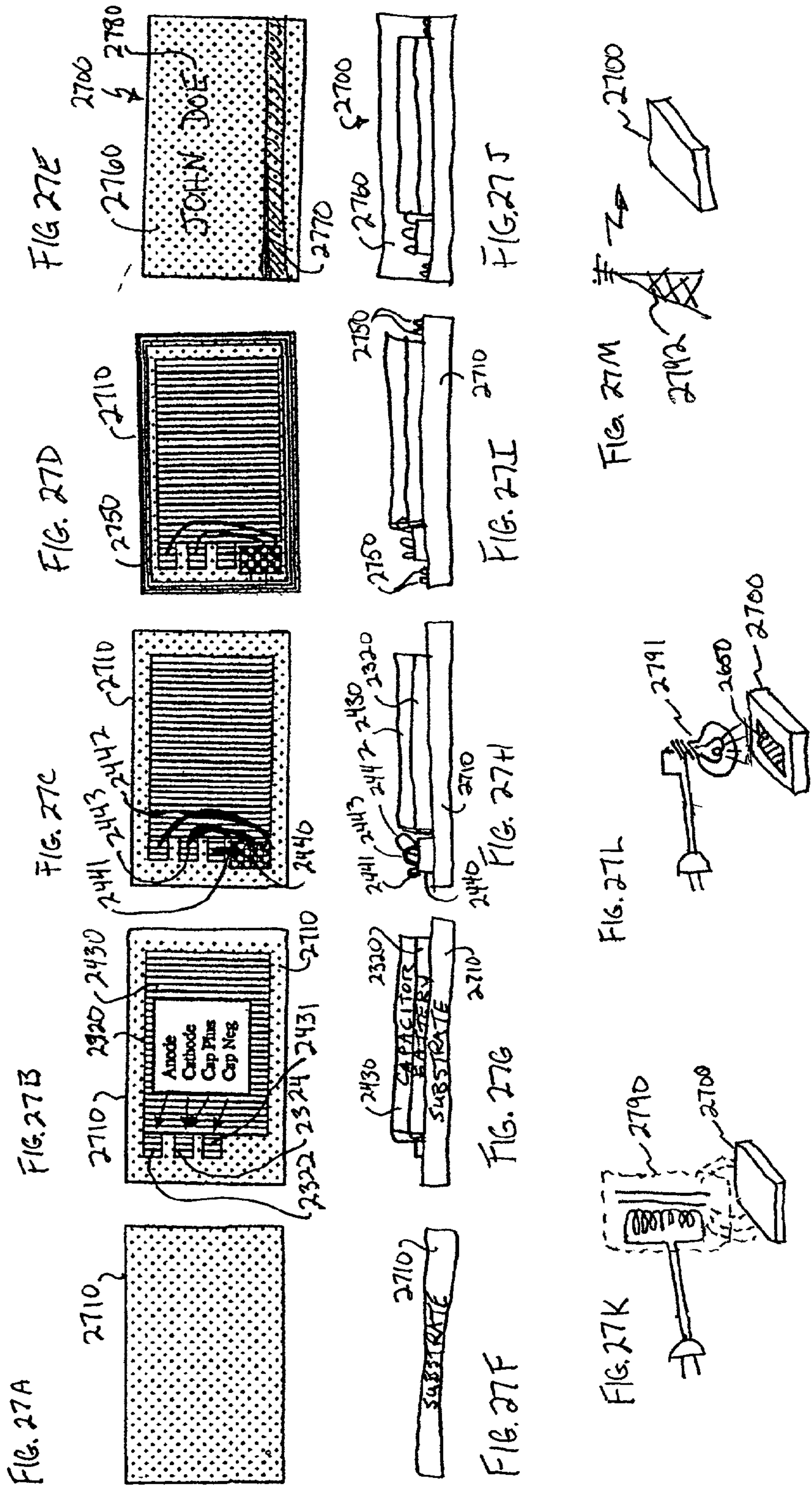
Fig 21 B











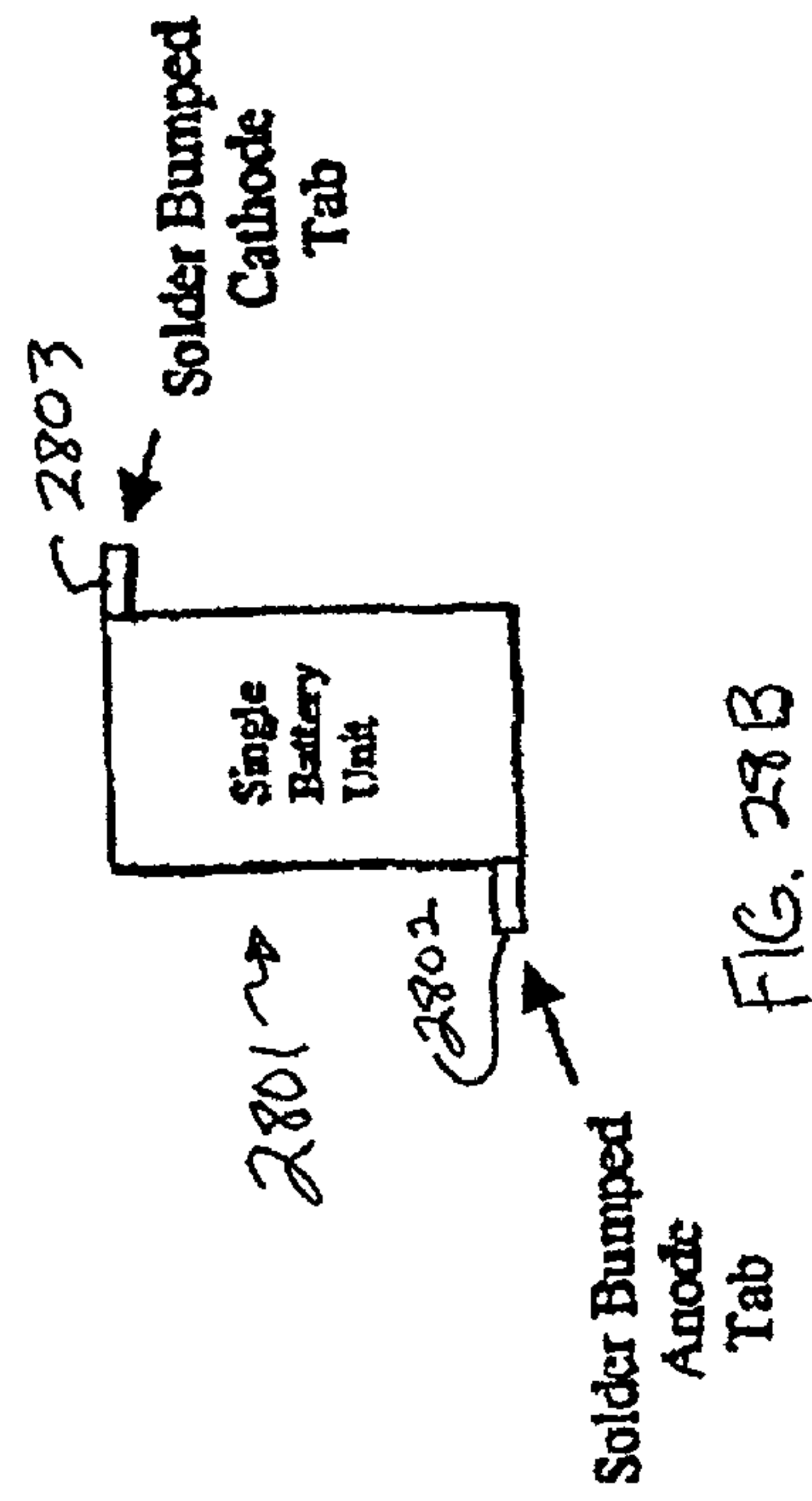


FIG. 28A

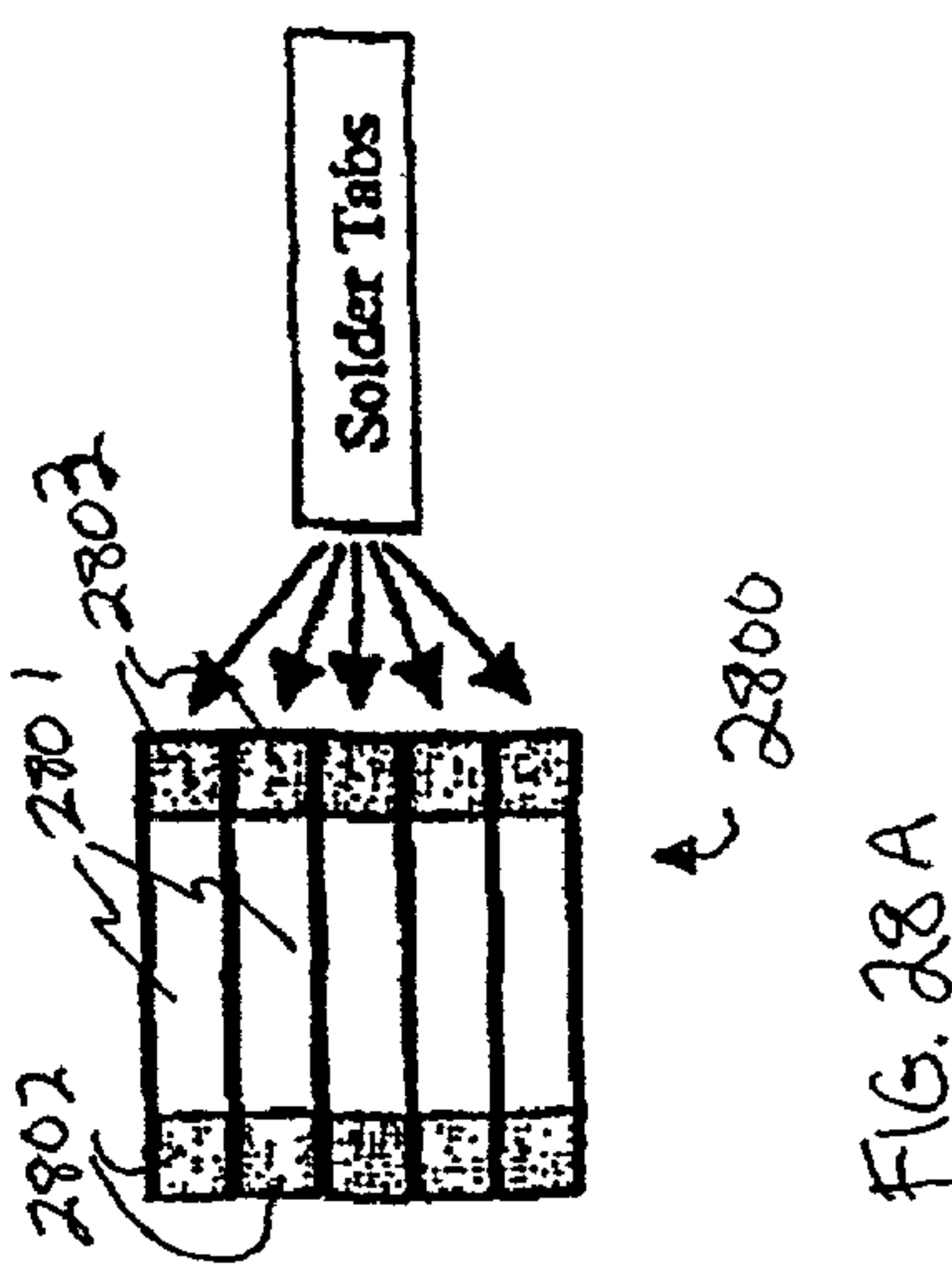


FIG. 28B

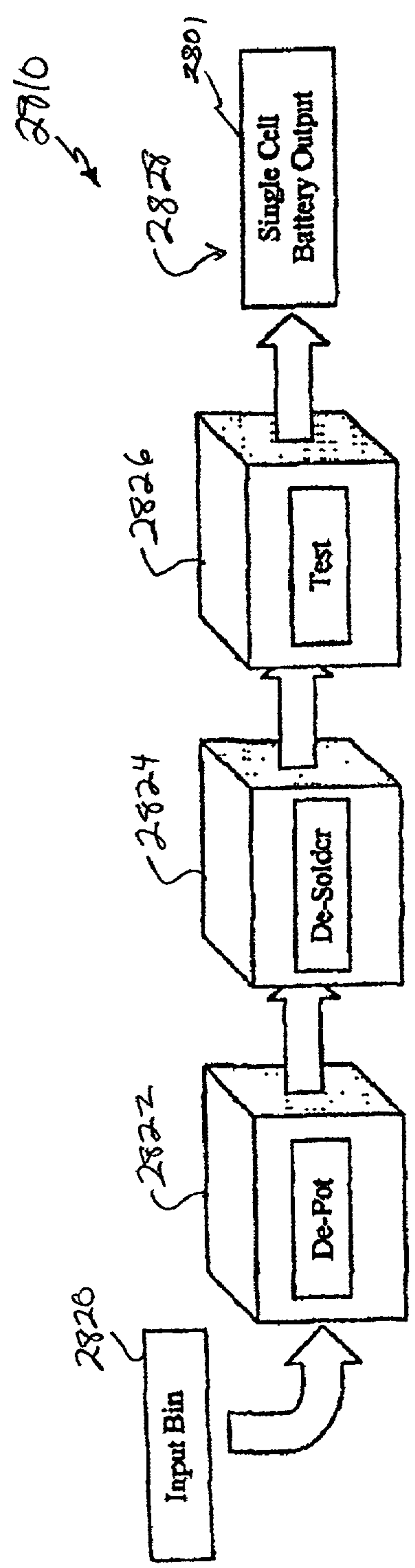
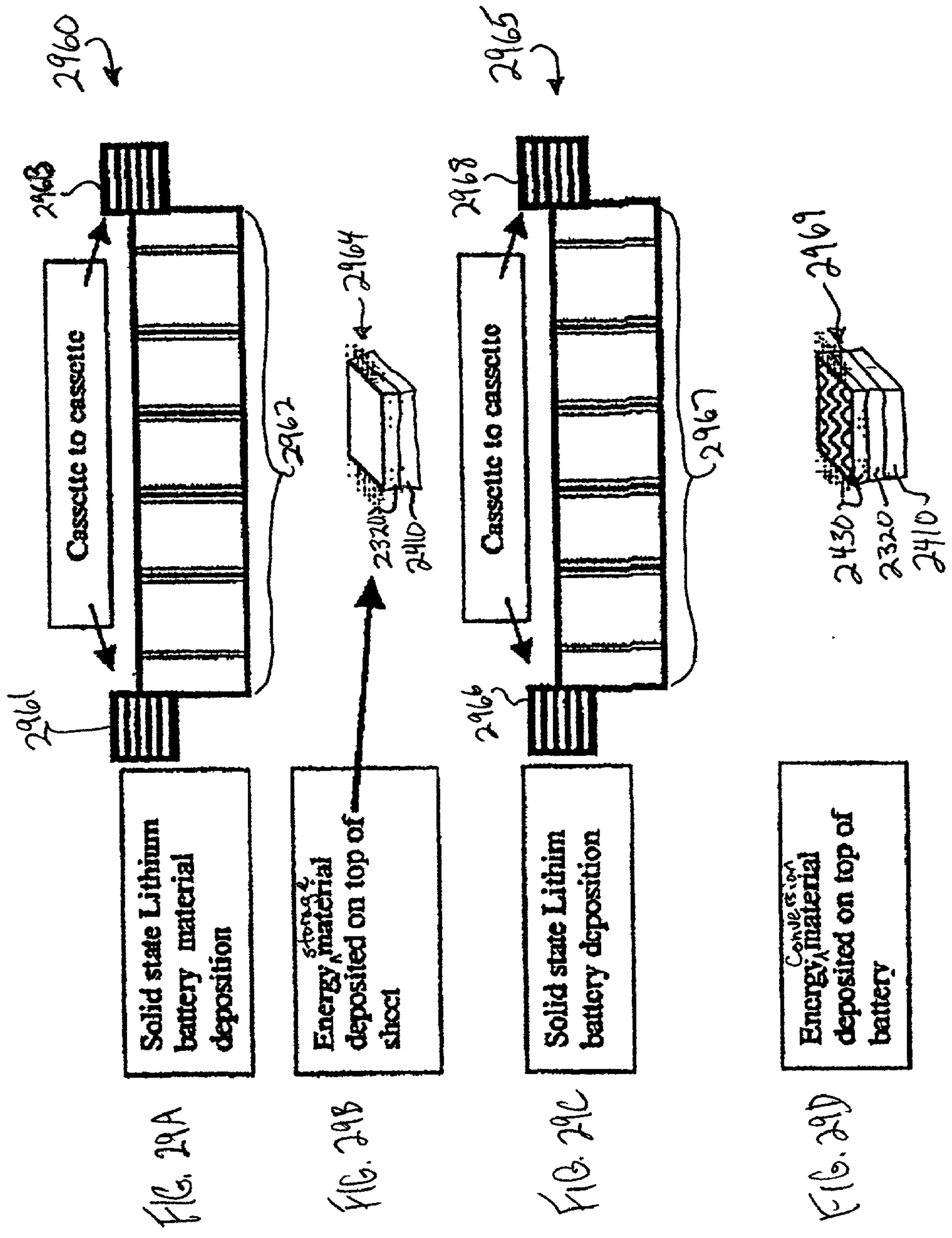
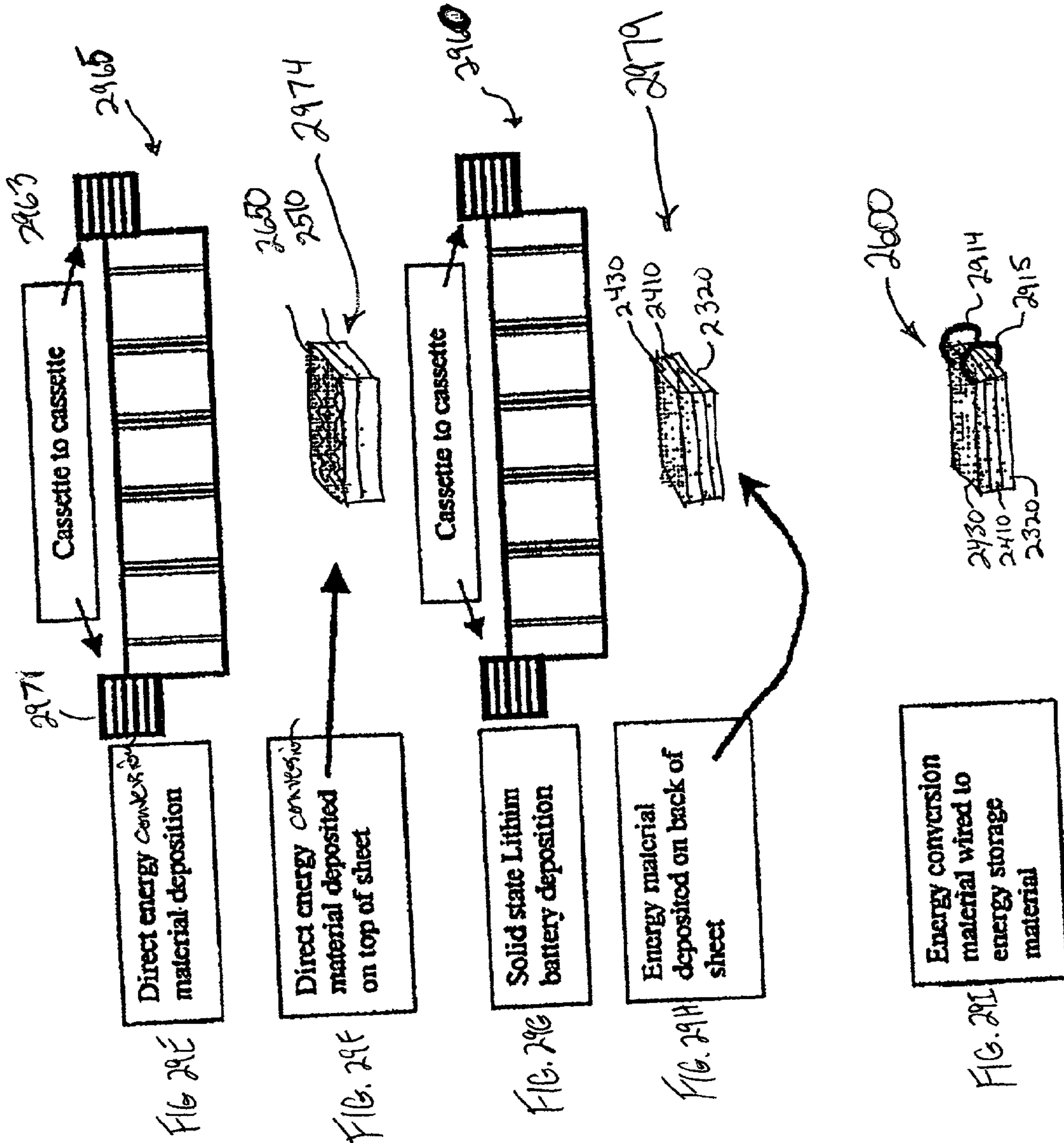


FIG. 28C





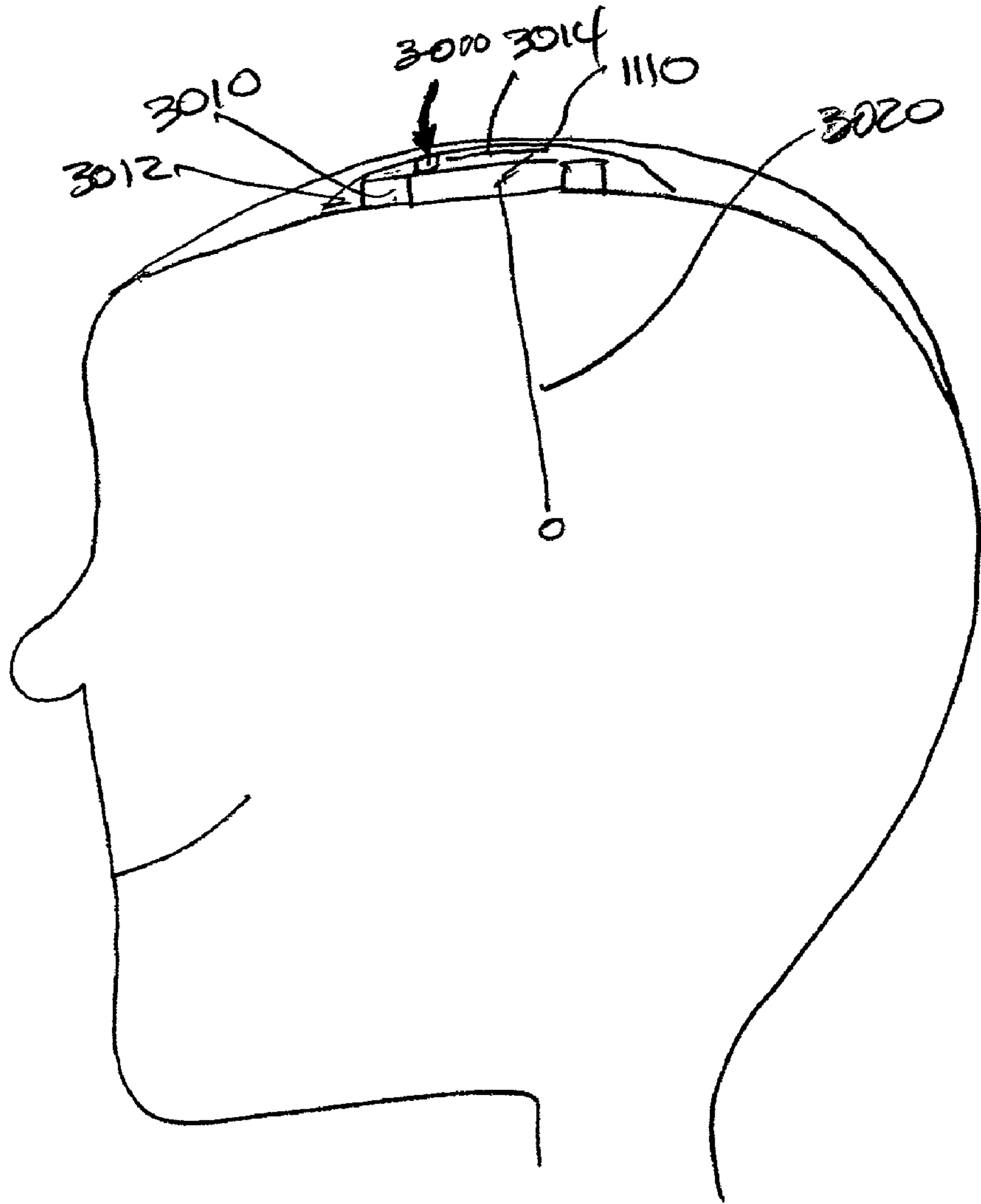


Fig 30

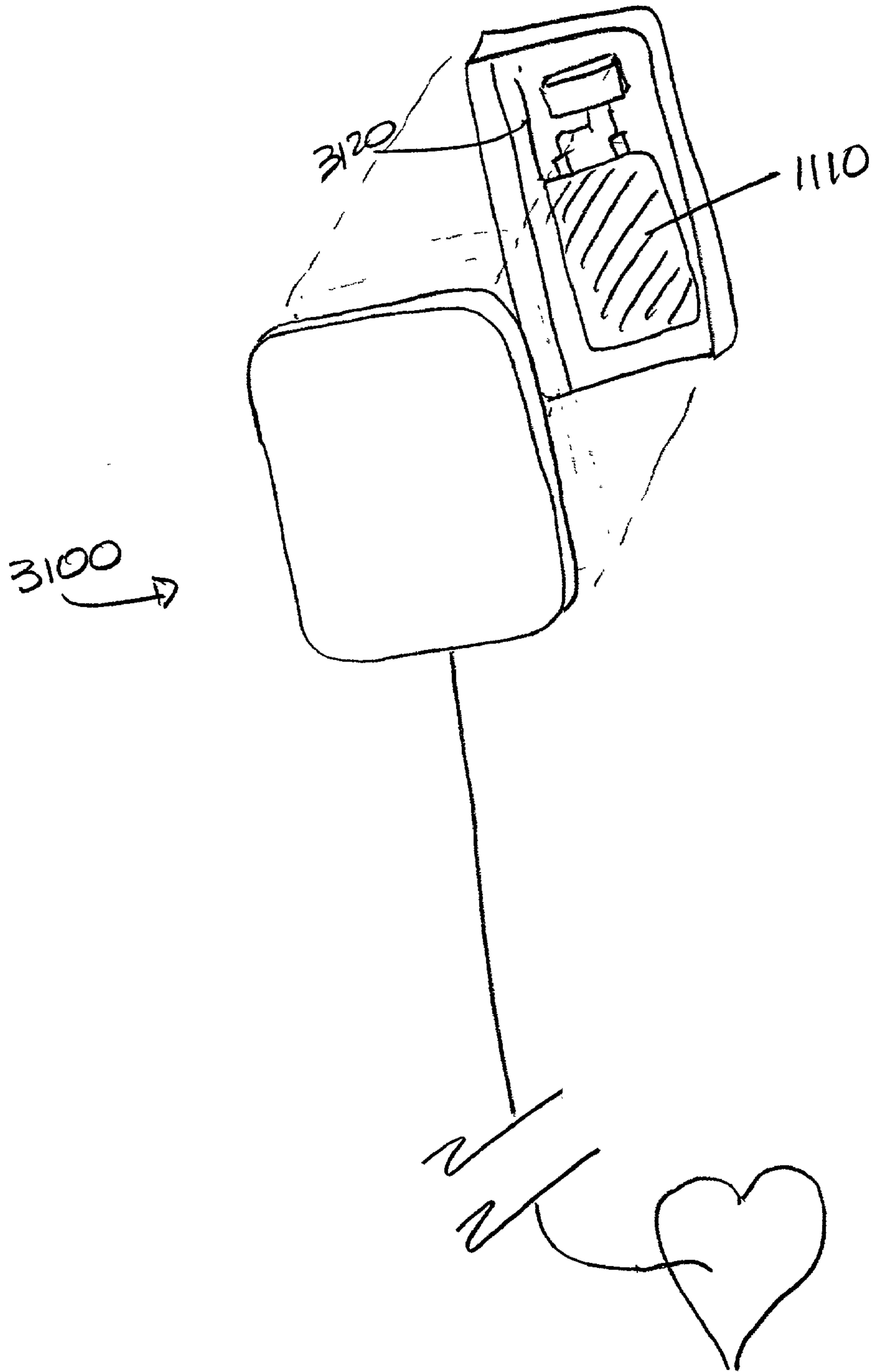


Fig 31A

3101
↓

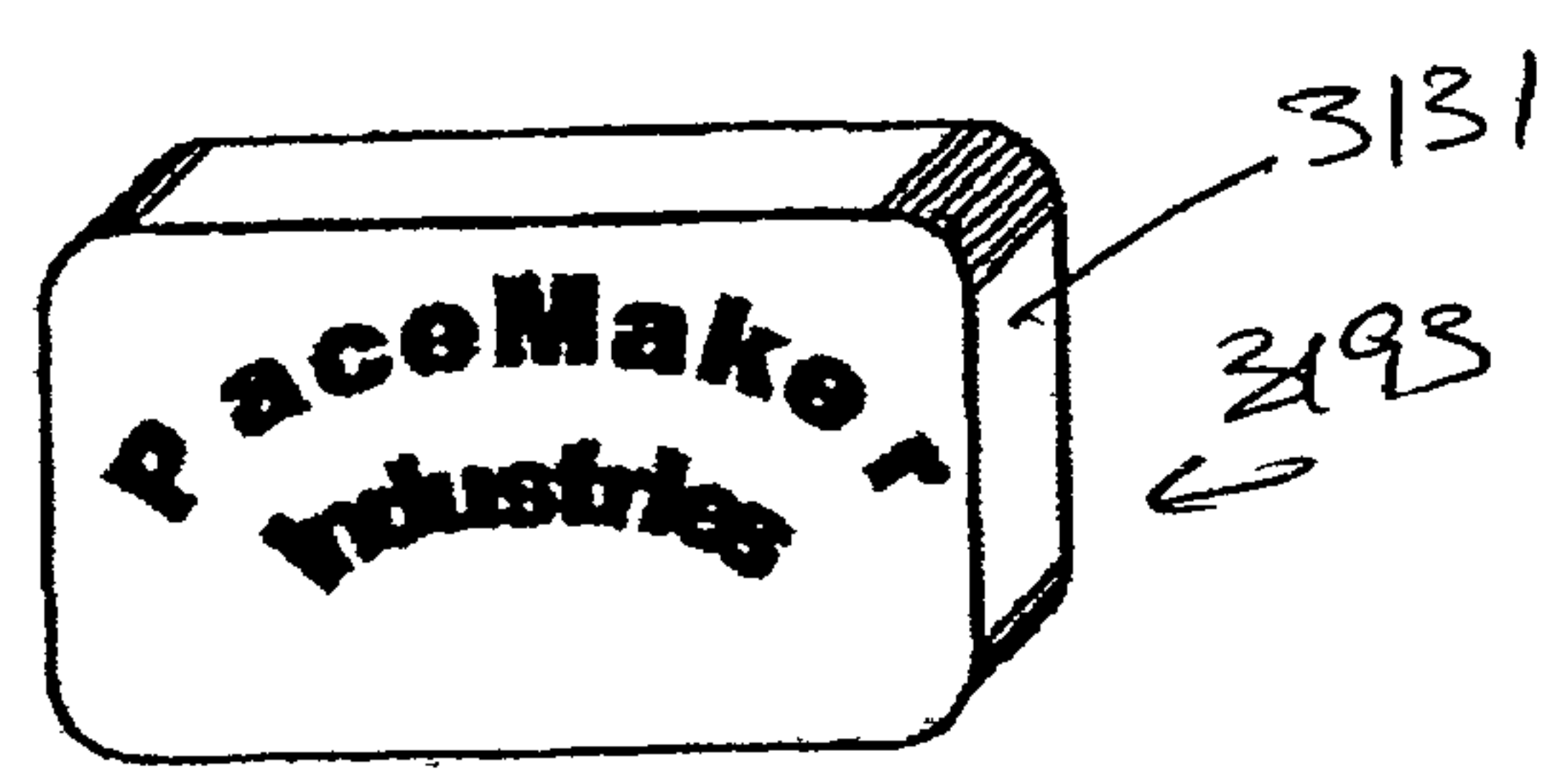
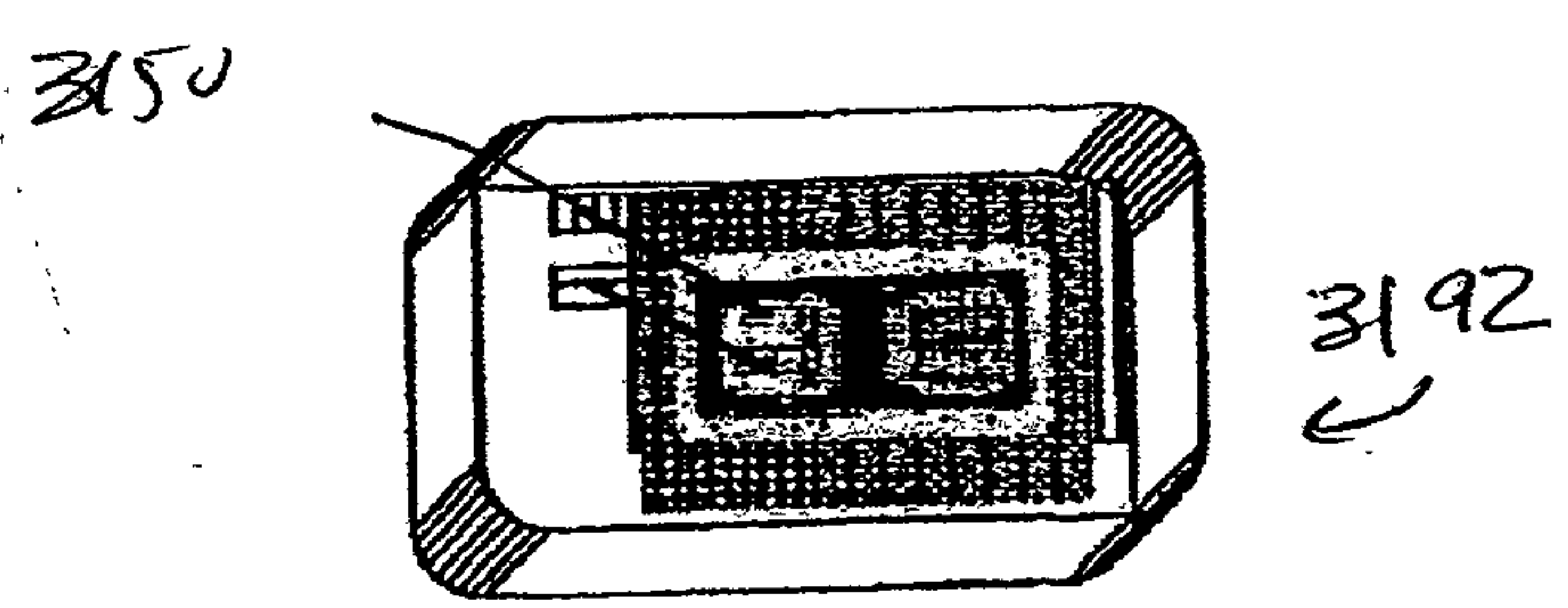
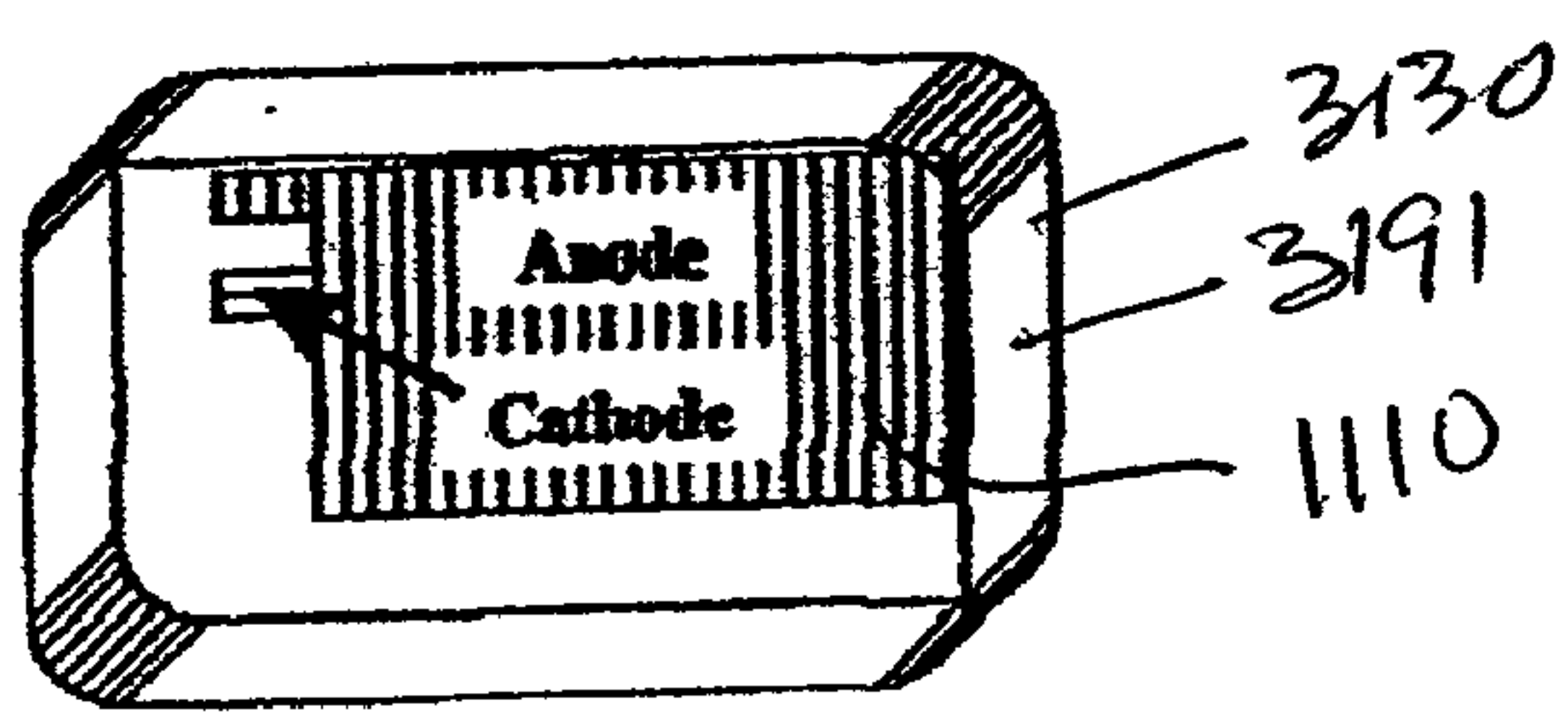
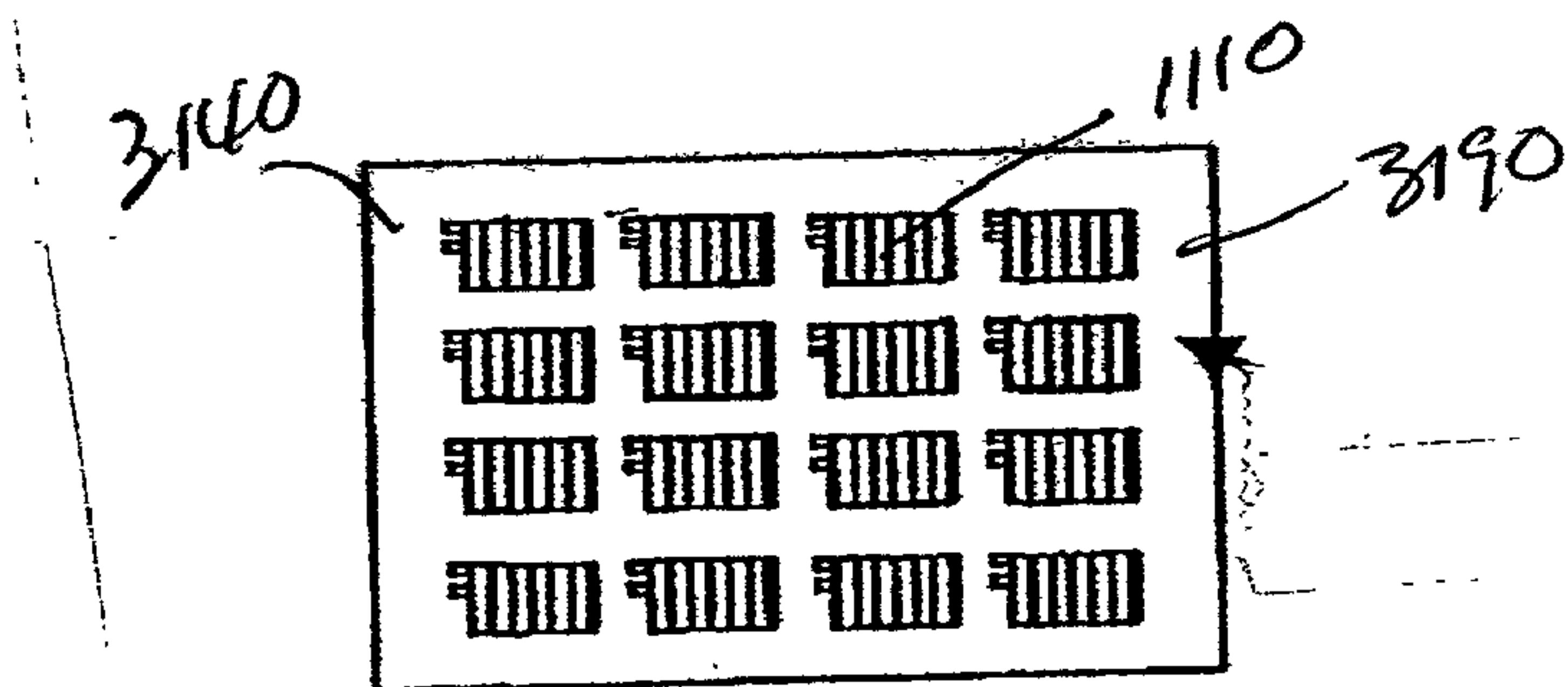


Fig 31 B

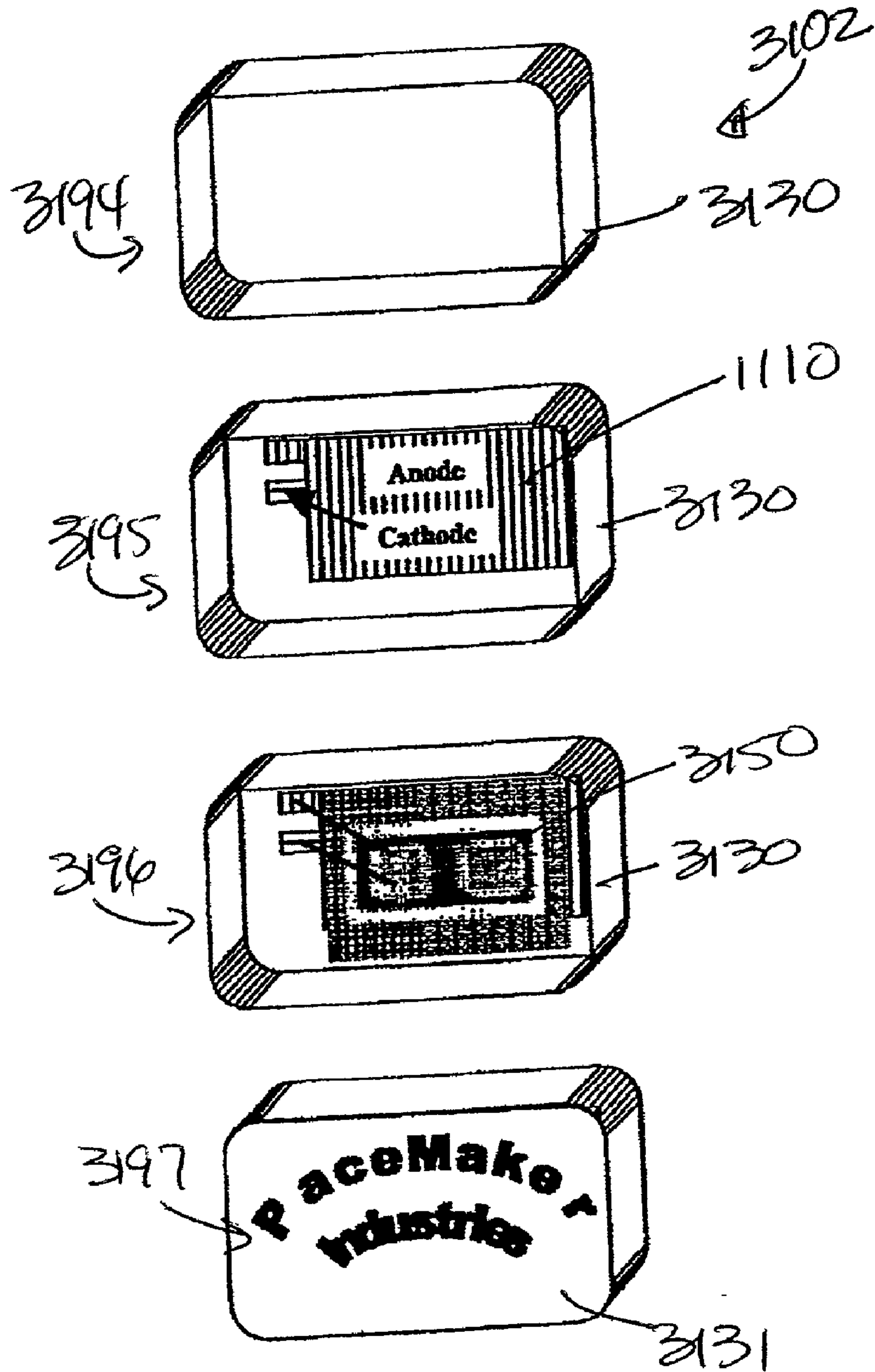


FIG 31C

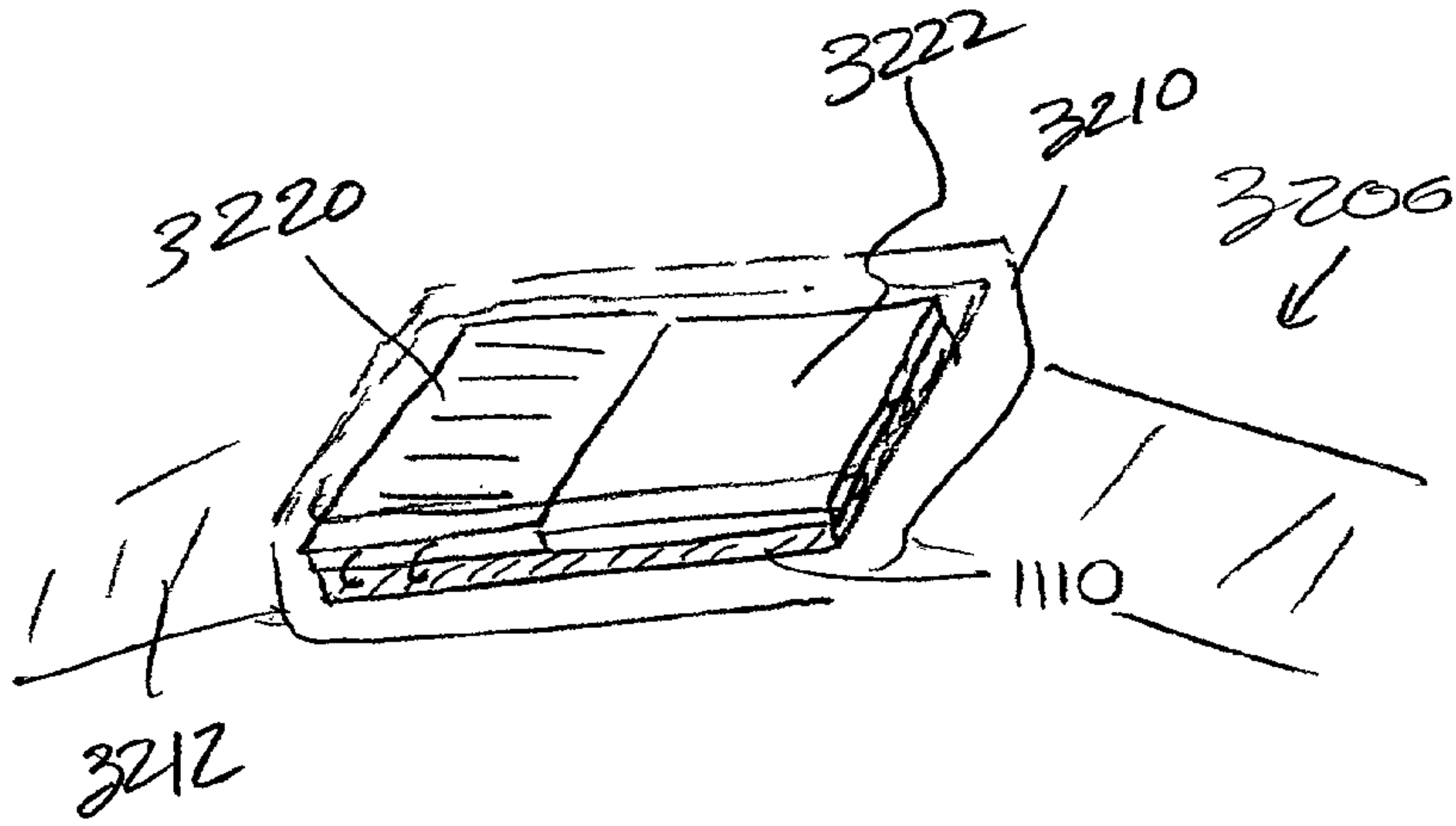


Fig 32 A

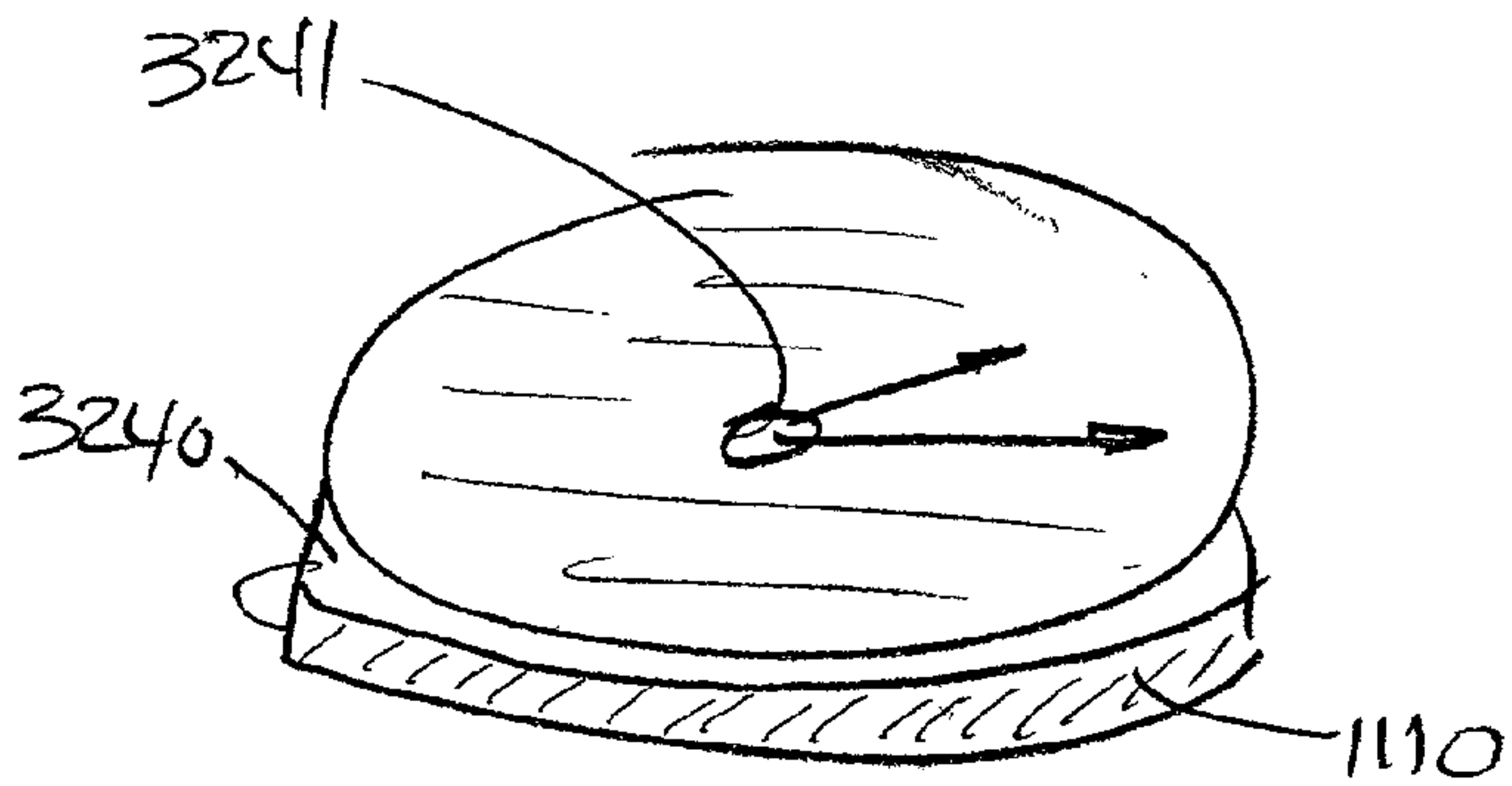


Fig 32 B

INTEGRATED CAPACITOR-LIKE BATTERY AND ASSOCIATED METHOD

CROSS-REFERENCES TO RELATED INVENTIONS

[0001] This invention claims priority to the following three provisional U.S. patent applications: attorney docket SLWK 1327.001prv, filed: Mar. 24, 2000, Ser. No. 60/191,774, titled "Comprehensive Patent for the Fabrication of a High Volume, Low Cost Energy Products Such as Solid State Lithium Ion Rechargeable Battery, Supercapacitors and Fuel Cells," attorney docket SLWK 1327.003prv, filed: Aug. 14, 2000, Ser. No. 60/225,134, titled "Apparatus and Method for Rechargeable Batteries and for Making and Using Batteries" and attorney docket SLWK 1327.005prv, filed: Oct. 6, 2000, Ser. No. 60/238,673, Titled "Battery Having Ultrathin Electrolyte," each of which is incorporated by reference.

[0002] This invention also is related to patent applications for the following six attorney dockets each filed on even date herewith: 1327.003us1 titled "Low-Temperature Fabrication of Thin-Film Energy-Storage Devices," 1327.005us1 titled "Thin-Film Battery Having Ultra-Thin Electrolyte and Associated Method," 1327.008us1 titled "Method and Apparatus for Integrated-Battery Devices and Like Devices," 1327.009us1 titled "Continuous Processing of Thin-Film Batteries and Like Devices," 1327.010us1 titled "Device Enclosures and Devices with Integrated Battery," and 1327.011us1 titled "Battery-Operated Wireless-Communication Apparatus and Method," each of which is incorporated by reference.

FIELD OF THE INVENTION

[0003] The present invention relates to solid-state energy-storage devices and solid-state energy-conversion devices. More particularly, this invention relates to methods and systems for fabricating such devices and the resulting devices, such as batteries, capacitors, supercapacitors, and photovoltaic cells.

BACKGROUND OF THE INVENTION

[0004] Electronics have been incorporated into many portable devices such as computers, mobile phones, tracking systems, scanners, etc. One drawback to portable devices is the need to include the power supply with the device. Portable devices typically use batteries as power supplies. Batteries must have sufficient capacity to power the device for at least the length of time the device is in use. Sufficient battery capacity can result in a power supply that is quite heavy or large compared to the rest of the device. Accordingly, smaller and lighter batteries (i.e., power supplies) with sufficient energy storage are desired. Other energy storage devices, such as supercapacitors, and energy conversion devices, such as photovoltaics and fuel cells, are alternatives to batteries for use as power supplies in portable electronics and non-portable electrical applications.

[0005] Another drawback of conventional batteries is the fact that some are fabricated from potentially toxic materials that may leak and be subject to governmental regulation. Accordingly, it is desired to provide an electrical power source that is safe, solid-state and rechargeable over many charge/discharge life cycles.

[0006] One type of an energy-storage device is a solid-state, thin-film battery. Examples of thin-film batteries are described in U.S. Pat. Nos. 5,314,765; 5,338,625; 5,445,126; 5,445,906; 5,512,147; 5,561,004; 5,567,210; 5,569,520; 5,597,660; 5,612,152; 5,654,084; and 5,705,293, each of which is herein incorporated by reference. U.S. Pat. No. 5,338,625 describes a thin-film battery, especially a thin-film microbattery, and a method for making same having application as a backup or first integrated power source for electronic devices. U.S. Pat. No. 5,445,906 describes a method and system for manufacturing a thin-film battery structure formed with the method that utilizes a plurality of deposition stations at which thin battery component films are built up in sequence upon a web-like substrate as the substrate is automatically moved through the stations.

[0007] FIG. 1A shows a prior art thin-film battery 20 formed on substrate 22. The battery includes a cathode current collector 32 and an anode current collector 34 formed on the substrate 22. A cathode layer 38 is formed on the cathode current collector 32. An electrolyte layer 42 is formed on the cathode layer 38. An anode layer 44 is formed on the electrolyte layer 42, the substrate 22 and the anode current collector 34. The current collectors 32 and 34 are connected to external circuitry to provide electrical power to the same. In a discharge operation, ions in the anode layer 44 travel through the electrolyte layer 42 and are stored in the cathode layer 38. Thereby, creating current flowing from the anode current collector 34 to the cathode current collector 32. In a charge operation, an external electrical charge is applied to the current collectors 32 and 34. Thereby, ions in the cathode layer 38 are forced to travel through the electrolyte layer 42 and are stored in the anode layer 44.

[0008] FIG. 2A shows a prior art method for fabricating the thin-film battery 20. First, the substrate is prepared for deposition of the thin-film battery (step 215). The cathode current collector is deposited on the substrate using DC-magnetron sputtering (step 217). The cathode is deposited on the cathode current collector by RF-magnetron sputtering (step 219). In this method, the magnetron source provides sputtered material having energy of about 1-3 eV, which is insufficient to crystallize the cathode material to form desirable crystal structures that encourage ion movement into and out of the cathode material. The cathode must be annealed to produce a crystalline lattice structure in the cathode, which is necessary to produce an energy-storage device that has the required electrical performance characteristics. In some embodiments, a desired electrical characteristic of a battery is a discharge curve that has a relatively constant voltage (small delta) over a range of capacity and then the voltage decreases rapidly as remaining capacity is exhausted (large delta). Accordingly, the stack of the substrate, cathode current collector and the cathode are annealed at a temperature of 700 degrees Celsius (step 221 of FIG. 2A). The anneal step 221 complicates and adds cost to the fabrication of this type of solid-state battery. Further, the anneal step 221 precludes the use of any material as the substrate or other part of the battery thus formed that is unable to withstand the high anneal temperature. The anode current collector is deposited on the substrate by DC-magnetron sputtering (step 223). The electrolyte layer is deposited by RF-magnetron sputtering (step 225). The anode is deposited by thermal evaporation (step 227).

[0009] Commercial capacitors can be traditionally classed in two groups: electrostatic and electrolytic. Electrostatic capacitors generally have two flat metallic electrodes separated by a layer of dielectric material. Paper, polymer film, mica, and ceramic are common dielectric materials that have been developed and commercialized for electrostatic capacitors. Electrolytic capacitors usually have one metallic electrode with an electrochemically formed oxide layer as the dielectric, and an electrolyte as the opposite electrode. Aluminum and tantalum capacitors are two commercialized electrolytic capacitors.

[0010] A different group of capacitors, supercapacitors, have received a great attention for about two decades. Examples of supercapacitors are found in U.S. Pat. Nos. 5,151,848; 5,585,999; 6,059,847; 5,426,561 and 6,181,545, incorporated herein by reference. As the name says, supercapacitors have "super" properties. One such property is a "relatively" high energy density. Regarding performance, supercapacitors fall in the middle between capacitors and batteries. In general, the energy density of supercapacitors (3-30 Wh/kg) is an order of magnitude lower than that of batteries (30-300 Wh/kg), but the power density (10E3-10E6 vs. 100-500 W/kg) is several orders higher. The energy density of supercapacitors is much better than conventional capacitors (0.03-0.3 Wh/kg) but the power density is not as high as that of conventional capacitors (10E4-10E7 W/kg). This is because the working voltage of supercapacitors is limited by electrolyte decomposition potential rather than the dielectric thickness as in electrolytic and electrostatic conventional capacitors. The electrolyte decomposition potential is about 1.2 V for aqueous systems and about 2 to 5 V for non-aqueous systems. Because of this reason, supercapacitors are more suitable for relatively long pulses (milliseconds to seconds) and low to intermediate power applications.

[0011] There are two main groups of supercapacitors: double layer capacitors (DLC) and electrochemical capacitors (EC), sometimes termed pseudocapacitors, or redox capacitors. In DLC, electric charge is stored in and released from a double layer that is on the electrode/electrolyte interface. In EC, it is not only through a double-layer discharge, which is a non-faradaic process, but also by a charge transfer, which is a faradaic process. However, unlike in a battery, the charge transfer has an extremely fast kinetics to allow a quick discharge. One commercially produced DLC is a carbon-based electrode with a very high surface area. Two types of EC have drawn the greatest attention: metal oxide and polymer. A common oxide is RuOx and it has been proven most promising. Other electrode oxides include IrOx, TaOx, MnOx, or mixtures thereof.

[0012] Attempts have been made to reduce the internal electrical resistance in supercapacitor electrodes by means other than directly applied physical pressure, notably through some manner of unifying the particulate carbon electrode composition and conductive collectors. A process of high-temperature sintering of the elements to achieve this end is described in U.S. Pat. No. 5,115,378. Other means for minimizing the internal resistance of supercapacitor structures have, for example, attempted to combine pyrolyzed aerogel, carbon foam electrodes with high-temperature soldering of conductive collector elements, as described in U.S. Pat. No. 5,260,855. However, such extensive processing steps and high-energy consumption lead to economic unde-

sirability of this approach. Moreover, high temperature annealing limits the applicability of such a process to a narrow array of substrates and devices.

[0013] Another area of interest with regard to the present invention includes the fabrication of energy conversion devices. Energy conversion devices that use renewable energy source to generate electricity as opposed to fossil fuels are highly desirable. An example of an energy conversion device is a photovoltaic cell such as is described in U.S. Pat. No. 4,207,119. However, photovoltaic cells have not significantly displaced fossil fuels as an electricity producing means. Two factors that have retarded the wide spread commercial acceptance of photovoltaics are costs of mass producing such cells and the low efficiency of energy conversion by the cells. Improvement of either of these two factors would create wider acceptance and use of photovoltaic cells.

[0014] Accordingly, there is a need for solid-state energy-storage devices, e.g., thin-film batteries and capacitors, and energy conversion devices, e.g., photovoltaic cells, that can be economically fabricated and that have acceptable electrical properties for use in a variety of electrical devices. More specifically, there is a need for a fabrication method and system that does not require a high-temperature anneal to form films having acceptable properties for such devices.

SUMMARY OF THE INVENTION

[0015] A method and system is described for fabricating solid-state energy-storage devices. Some embodiments include fabricating films for an energy-storage device without an anneal step. The present invention provides energy focused at the location where it is needed to form a film having certain structural characteristics. Accordingly, the total energy applied to the film and its substrate is less than if the energy was applied to the entire device, e.g., high temperature annealing the entire device. Thus, the present invention provides methods for applying energy to cure defects and/or create desired crystal structure at the time of depositing a film. In some embodiments, the energy is applied essentially at a few atomic layers at a time during deposition of a film.

[0016] In an embodiment as described herein, a film of an energy-storage device is fabricated by depositing a first material layer to a location on a substrate. Energized ions of a second material are directed to the first material to supply energy thereto, thereby assisting the growth of the crystalline structure of the first material. In some embodiments, the first material includes an intercalation material, which releasably stores ions therein. In one embodiment, the intercalation material is a lithium intercalation material. In energy-storage devices, and specifically in thin-film batteries, it is desirable to have large crystal size and a specific crystal orientation to improve electrical characteristics of the energy-storage device.

[0017] Another aspect of the invention includes fabricating a layer of an energy-storage device using a first source for supplying component material for the layer and a second source for supplying energy to a location on the substrate at which the layer is to be deposited. In one embodiment, the fabricated layer is the cathode for a thin-film battery. In another embodiment, the thin-film battery is a rechargeable, solid-state lithium-ion battery.

[0018] Another feature of the present invention is fabricating the electrolyte film by depositing a first material layer to a location on a substrate. Energized ions of a second material are directed to supply energy to the first material, thus assisting the growth of the crystalline structure of the first material. Another feature of the present invention includes fabricating the anode film by depositing a first intercalation material layer to a location on a substrate. Energized ions of a second material are directed to supply energy to the first material, thereby assisting the growth of the crystalline structure of the first material and controlling stoichiometry of the crystalline structure of the first material.

[0019] Another aspect of the present invention includes controlling the energy provided by the second source energized ions. The energized ions provide energy to the intercalation material of about 5 eV or greater.

[0020] It is yet another aspect of the present invention to provide a seed layer on which an intercalation film is grown. The seed layer assists the formation of desired crystal structures to improve energy-storage device performance.

[0021] Another feature of the present invention includes fabricating an energy-storage device on any one of a plurality of different substrates. Some of the substrates have thermal degradation temperatures that are less than temperatures used for processes conventionally used for making thin-film batteries. It is another feature to provide systems and fabrication techniques that do not cause temperature degradation of a substrate or other layers thereon.

[0022] Another aspect of the present invention is overcoming the limitations of prior supercapacitor structures and fabrication procedures, the present invention provides, in particular, means for fabricating low internal resistance supercapacitor structures under economical manufacturing conditions. Such simple fabrication procedures enable the expanded use of these devices in a wide variety of configurations and applications, including combinations with integrated rechargeable battery energy sources of compatible composition and structure.

BRIEF DESCRIPTION OF THE DRAWINGS

[0023] FIG. 1A is a cross-sectional view of a conventional lithium-ion battery.

[0024] FIG. 1B is a cross-sectional view of an energy-storage device according to the present invention.

[0025] FIG. 1C is a cross-sectional view of an energy-storage device according to the present invention.

[0026] FIG. 1D is a cross-sectional view of an energy-storage device and a supercapacitor according to the present invention.

[0027] FIG. 2A is a flowchart of a conventional method for manufacturing the lithium-ion battery of FIG. 1A.

[0028] FIG. 2B is a flowchart of one embodiment of a fabrication process according to the teachings of the present invention.

[0029] FIG. 2C is a flowchart of one embodiment of a fabrication process according to the teachings of the present invention.

[0030] FIG. 2D is a flowchart of one embodiment of a fabrication process according to the teachings of the present invention.

[0031] FIG. 3A is a diagram of a device for fabricating a thin-film battery according to the teachings of the present invention.

[0032] FIG. 3B is a diagram of a device for fabricating a thin-film battery according to the teachings of the present invention.

[0033] FIG. 4 is a diagram of another embodiment of a device for fabricating a thin-film battery according to the teachings of the present invention.

[0034] FIG. 4A is a diagram of another embodiment of a device for fabricating a thin-film battery according to the teachings of the present invention.

[0035] FIG. 5B is a diagram of another embodiment of a device for fabricating a thin-film battery according to the teachings of the present invention.

[0036] FIG. 6 is a diagram of another embodiment of a device for fabricating a thin-film battery according to the teachings of the present invention.

[0037] FIG. 7 is a diagram of another embodiment of a device for fabricating a thin-film battery according to the teachings of the present invention.

[0038] FIG. 8 is a cross-sectional view of a photovoltaic cell according to the teachings of the present invention.

[0039] FIG. 9A is a perspective view of a thin-film energy-storage device according to the teachings of the present invention.

[0040] FIG. 9B is a view of another embodiment of a thin-film energy-storage device according to the teachings of the present invention.

[0041] FIG. 9C is a view of another embodiment of a thin-film energy-storage device according to the teachings of the present invention.

[0042] FIG. 10 shows X-ray diffraction spectra of cathode films for thin-film batteries.

[0043] FIG. 11 shows X-ray diffraction spectra of both a conventional cathode layer and a cathode film according to the teachings of the present invention.

[0044] FIG. 12A shows X-ray diffraction spectra for a conventional magnetron sputtered cathode layer.

[0045] FIG. 12B shows X-ray diffraction spectra for a thin film for an energy-storage device according to the present disclosure.

[0046] FIG. 13 is an exploded perspective view of an electronic device with a separate printed circuit board and battery.

[0047] FIG. 14A is an exploded perspective view of a portion of an enclosure for an electronic device according to one embodiment of this invention.

[0048] FIG. 14B is an exploded perspective view of a portion of an enclosure for an electronic device according to another embodiment of this invention.

- [0049] FIG. 14C is an exploded perspective view of a portion of an enclosure for an electronic device according to yet another embodiment of this invention.
- [0050] FIG. 15A is a plan view of a sheet including a plurality of battery cells.
- [0051] FIG. 15B is a plan view of a diced battery cell before forming.
- [0052] FIG. 15C is a perspective view of a battery cell after forming.
- [0053] FIG. 15D is a perspective view of a battery cell after forming.
- [0054] FIG. 15E is a perspective view of a battery cell after forming.
- [0055] FIG. 15F is a plan view of a sheet including a plurality of battery cells.
- [0056] FIG. 15G is a plan view of a plurality of diced battery cells before forming.
- [0057] FIG. 15H is a perspective view of a fan folded plurality of diced battery cells before forming.
- [0058] FIG. 15I is a plan view of a sheet including a plurality of battery cells.
- [0059] FIG. 15J is a perspective view of a sheet including a plurality of battery cells formed on the sheet according to this invention.
- [0060] FIG. 15K is a perspective view of a sheet including two battery cells formed into a case from the sheet shown in FIG. 15J according to this invention.
- [0061] FIG. 15L is a side view of an electronic device enclosure formed from a sheet.
- [0062] FIG. 16A is a plan view of a sheet including a plurality of battery cells.
- [0063] FIG. 16B is a plan view of a diced battery cell before forming.
- [0064] FIG. 16C is a perspective view of a fan folded diced battery cell before forming.
- [0065] FIG. 16D shows a fan folded cord being truncated.
- [0066] FIG. 16E shows a finished cord.
- [0067] FIG. 17 is an exploded perspective view of a sheet including at least one battery cell rolled around an electrical motor in accordance with this invention.
- [0068] FIG. 18A is a plan view of a diced battery cell and LED before forming.
- [0069] FIG. 18B is a perspective view of a diced battery cell and LED after forming.
- [0070] FIG. 18C is a plan view of a diced battery cell and LED before forming.
- [0071] FIG. 18D is a perspective view of a diced battery cell and LED after forming.
- [0072] FIG. 19A is a plan view of a sheet including a plurality of battery cells according to another embodiment of this invention.
- [0073] FIG. 19B is a plan view of a plurality of diced battery cells before forming.
- [0074] FIG. 19C is a perspective view of a formed battery including a plurality of cells.
- [0075] FIG. 20 is a cutaway side view of a sheet including a plurality of battery cells, which are embedded in an enclosure portion.
- [0076] FIG. 21A is a flow chart for a first recycling method using the inventive battery and enclosure.
- [0077] FIG. 21B is a flow chart for a first recycling method using the inventive battery and enclosure.
- [0078] FIG. 22A shows a schematic circuit of an embodiment of an integrated battery and circuit sharing a common terminal.
- [0079] FIG. 22B shows a block diagram perspective view of an integrated device implementing the circuit of FIG. 22A having the circuit built on the battery.
- [0080] FIG. 22C shows a block diagram perspective view of an integrated device implementing the circuit of FIG. 22A having the battery built on the circuit.
- [0081] FIG. 22D shows a schematic circuit of an embodiment 2202 of an integrated battery and circuit each having separate terminals.
- [0082] FIG. 22E shows a block diagram perspective view of an integrated device implementing the circuit of FIG. 22D having the circuit built on the battery.
- [0083] FIG. 22F shows a block diagram perspective view of an integrated device implementing the circuit of FIG. 22D having the battery built on the circuit.
- [0084] FIG. 22G shows a block diagram perspective view of an integrated device implementing the circuit of FIG. 22A having the battery and the circuit built side-by-side on a substrate.
- [0085] FIG. 22H shows a block diagram perspective view of an integrated device implementing the circuit of FIG. 22D having the battery and the circuit built side-by-side on a substrate.
- [0086] FIG. 23 shows a perspective view of an embodiment 2300 of the present invention having a battery overlaid with circuitry.
- [0087] FIG. 24A shows a perspective view of an embodiment 2400 of the present invention having a battery overlaid with an integrated device.
- [0088] FIG. 24B shows a block diagram of a layer-deposition system 2460.
- [0089] FIG. 24C shows a perspective view of a partially processed sheet 2464.
- [0090] FIG. 24D shows a block diagram of a layer-deposition system 2465.
- [0091] FIG. 24E shows a perspective view of a processed sheet 2469.
- [0092] FIG. 24F shows a perspective view of a diced final device 2400.

- [0093] FIG. 25A shows a perspective view of an embodiment 2500 of the present invention having an integrated circuit overlaid with a battery.
- [0094] FIG. 25B shows a plan view of IC 2540.
- [0095] FIG. 25C shows an elevational view of IC 2540.
- [0096] FIG. 25D shows a plan view integrated battery-IC 2501.
- [0097] FIG. 25E shows an elevational view of integrated battery-IC 2501.
- [0098] FIG. 25F shows a block diagram of a layer-deposition system 2560.
- [0099] FIG. 25G shows a perspective view of a processed sheet 2569.
- [0100] FIG. 26A shows a perspective view of an embodiment 2600 of the present invention having an integrated circuit overlaid on its back with a battery.
- [0101] FIG. 26B shows a block diagram of a layer-deposition system 2660.
- [0102] FIG. 26C shows a perspective view of a processed sheet 2669.
- [0103] FIG. 26D shows a perspective view of diced final devices 2600.
- [0104] FIG. 26E shows a perspective view of wired diced final device 2600.
- [0105] FIG. 26F shows a perspective view of a hearing aid 2690 incorporating a wired diced final device 2600.
- [0106] FIG. 27A shows a plan view of a starting substrate of an embodiment that will have an integrated battery and device sharing a common terminal.
- [0107] FIG. 27B shows a plan view of the substrate of FIG. 27A after deposition of the integrated battery and device sharing a common terminal.
- [0108] FIG. 27C shows a plan view of the substrate of FIG. 27B after placing and wiring a separately fabricated chip connected to the integrated battery and device sharing a common terminal.
- [0109] FIG. 27D shows a plan view of the substrate of FIG. 27C after placing and wiring a loop antenna.
- [0110] FIG. 27E shows a plan view of the substrate of FIG. 27D after a top encapsulation layer has been deposited.
- [0111] FIG. 27F shows an elevation view of the starting substrate of FIG. 27A.
- [0112] FIG. 27G shows an elevation view of the partially built device of FIG. 27B.
- [0113] FIG. 27H shows an elevation view of the partially built device of FIG. 27C.
- [0114] FIG. 27I shows an elevation view of the partially built device of FIG. 27D.
- [0115] FIG. 27J shows an elevation view of the device of FIG. 27E.
- [0116] FIG. 27K shows a perspective view of the device of FIG. 27E at a magnetic-recharging station.
- [0117] FIG. 27L shows a perspective view of the device of FIG. 27E at a light-recharging station.
- [0118] FIG. 27M shows a schematic of the device of FIG. 27E at a radio-wave-recharging station.
- [0119] FIG. 28A shows an elevation view of a battery 2800 having stacked cells.
- [0120] FIG. 28B shows a plan view of a single battery cell after recycling.
- [0121] FIG. 28C shows a process 2810 used for recycling.
- [0122] FIG. 29A shows a block diagram of a layer-deposition system 2960.
- [0123] FIG. 29B shows a perspective view of a partially processed wafer 2964.
- [0124] FIG. 29C shows a block diagram of a layer-deposition system 2965.
- [0125] FIG. 29D shows a perspective view of a processed wafer 2969.
- [0126] FIG. 29E shows a block diagram of a layer-deposition system 2965.
- [0127] FIG. 29F shows a perspective view of a partially processed wafer 2974.
- [0128] FIG. 29G shows a block diagram of a layer-deposition system 2960.
- [0129] FIG. 29H shows a perspective view of a processed wafer 2979.
- [0130] FIG. 29I shows a perspective view of wired diced final device 2600.
- [0131] FIG. 30 is a perspective view of an implantable device according to this invention.
- [0132] FIG. 31A is an exploded perspective view of a pacemaker according to this invention.
- [0133] FIG. 31B is an exploded perspective view of one pacemaker as it is being formed.
- [0134] FIG. 31C is an exploded perspective view of another pacemaker as it is being formed.
- [0135] FIG. 32A is a perspective view of a first embodiment of a watch of the invention.
- [0136] FIG. 32B is a perspective view of a second embodiment of a watch.
- [0137] In the drawings, like numerals describe substantially similar components throughout the several views. Signals and connections may be referred to by the same reference number, and the meaning will be clear from the context of the description.

DETAILED DESCRIPTION

[0138] In the following detailed description of the preferred embodiments, reference is made to the accompanying drawings that form a part hereof, and in which are shown, by way of illustration, specific embodiments in which the invention may be practiced. It is to be understood that other embodiments may be utilized and structural changes may be made without departing from the scope of the present invention.

[0139] It is to be understood that in different embodiments of the invention, each battery in the Figures or the description can be implemented using one or more cells, and if a plurality of cells is implemented, the cells can be wired in parallel or in series. Thus, where a battery or more than one cell is shown or described, other embodiments use a single cell, and where a single cell is shown or described, other embodiments use a battery or more than one cell. Further, the references to relative terms such as top, bottom, upper, lower, etc. refer to an example orientation such as used in the Figures, and not necessarily an orientation used during fabrication or use.

[0140] The terms wafer and substrate as used herein include any structure having an exposed surface onto which a film or layer is deposited, for example, to form an integrated circuit (IC) structure or an energy-storage device. The term substrate is understood to include semiconductor wafers, plastic film, metal foil, and other structures on which an energy-storage device may be fabricated according to the teachings of the present disclosure. The term substrate is also used to refer to structures during processing that include other layers that have been fabricated thereupon. Both wafer and substrate include doped and undoped semiconductors, epitaxial semiconductor layers supported by a base semiconductor or insulator, as well as other semiconductor structures well known to one skilled in the art. Substrate is also used herein as describing any starting material that is useable with the fabrication method as described herein.

[0141] The term battery used herein refers to one example of an energy-storage device. A battery may be formed of a single cell or a plurality of cells connected in series or in parallel. A cell is a galvanic unit that converts chemical energy, e.g., ionic energy, to electrical energy. The cell typically includes two electrodes of dissimilar material isolated from each other by an electrolyte through which ions can move.

[0142] The term adatom as used herein refers to a particle, molecule, or ion of material that has not yet been formed into a structure or film.

[0143] The term intercalation as used herein refers to a property of a material that allows ions to readily move in and out of the material without the material changing its phase. Accordingly, a solid-state intercalation film remains in a solid state during discharging and charging of an energy-storage device.

[0144] FIG. 1B shows an embodiment of an energy-storage device 50 according to the present invention. A substrate 55 is provided on which is formed a contact film 57. Contact film 57 acts as a current collector and is connected to a lead 58, which connects one pole of the energy storage device 50 to an external circuit. An electrode film 59 is formed on the contact film 57. In some embodiments, the electrode film 59 substantially covers a surface of the contact film 57 to as to minimize resistance by maximizing the area of the interface between the films. In some embodiments, the electrode film 59 is a cathode for a thin-film battery. In other embodiments, electrode film 59 is an electrode of a supercapacitor. An electrolyte film 61 is formed on the electrode film 59. An electrode film 63 is formed on the electrolyte film 61. The electrolyte film 61 isolates electrode film 59 from electrode film 63. A contact film 65 is formed on electrode film 63. Contact film 65 acts

as a current collector and is connected to a lead 67, which connects one pole of the energy storage device 50 to an external circuit. In some embodiments, the contact film 65 substantially covers a surface of the electrode film 63 to as to minimize resistance by maximizing the area of the interface between these films. In some embodiments, the electrode film 63 is an anode for a thin-film battery. In other embodiments, electrode film 63 is an electrode of a supercapacitor.

[0145] FIG. 1C shows a cross sectional view of an embodiment of an energy-storage device 50C. A substrate 55 is provided and, in some embodiments, includes additional layers and/or devices formed therewith. In some embodiments, the substrate 55 includes a substrate as described herein. Contact films 57 and 59 are formed on the substrate 55 according to the methods described herein. In some embodiments, contact films 57 and 59 are metal films deposited on the substrate according to other methods as known in the art. Contact films 57 and 59 act as contacts for connecting the energy-storage device 50C to other circuit elements (not shown).

[0146] An electrode first film 59 is formed on contact 57. Electrode first film 59 includes a metal or intercalation material in some embodiments, for example, thin-film battery embodiments in which the electrode first film 59 functions as a cathode. In some such embodiments, the electrode first film 59 includes lithium metal and/or a lithium-intercalation material. In other embodiments, such as supercapacitors, electrode first film 59 is a metal oxide. It is desirable to maximize the contact interface between the electrode first film 59 and contact film 57. Accordingly, in some embodiments, the electrode first film 59 substantially covers contact film 57 except for a portion reserved for connection to external circuits.

[0147] An electrolyte film 61C is formed on, or at least partially on, the electrode first film 59. The electrolyte film 61C, in some embodiments, completely encloses the electrode first film 59. The electrolyte film 61C is formed using the systems and methods described herein. In one embodiment, a first material of the electrolyte film 61C is deposited using a first source, which directs a first electrolyte material (adatoms) to the location on the substrate or, as shown in FIG. 1C, to a location on the electrode first film 59.

[0148] An electrode second film 63 is formed on electrolyte film 61C and contact film 59. Electrolyte film 61C completely separates the electrode first film 59 from the electrode second film 63. The electrode second film 63 includes a metal or intercalation material in some embodiments, for example, thin-film battery embodiments in which the electrode second film is an anode. In other embodiments, such as supercapacitor embodiments, electrode second film 63 is a metal oxide. Electrode second film 63, in some embodiments is deposited according to the methods described herein. In other embodiments, electrode second film 63 is formed according to methods known in the art.

[0149] The electrolyte film 61C as deposited includes the electrolyte material. A first source (e.g., sources 311, 511, 511A, and 711 as described herein) of the electrolyte material, in one embodiment, is a physical vapor deposition source. In another embodiment, the first source is a chemical vapor deposition source. A second source provides energized particles to the location. The energized particles

impinge on the electrolyte material and assist in forming a desired structure of the electrolyte film **61C**. In some embodiments, the second source provides energized particles simultaneously with the first source supplying the electrolyte material. The use of the energized particles conforms the electrolyte film **61C** to electrode first film **59** such that the electrolyte film provides the necessary insulative property, namely preventing electrons from travelling directly between the electrode first film **59** and the electrode second film **63**, i.e., shorting the electrodes. In some embodiments, the second source is an ion source as described herein, e.g., sources **313**, **413**, or **713**. The second source provides energized ions that supply energy to the electrolyte material from the first source. The energy that is supplied by the ions assists in conforming the formed electrolyte film **61C** to the electrode first layer **59**. It is believed that the use of the energized particles in the energy range referenced herein provides the growing electrolyte material an extended period of mobility upon the previous film surface, and this extended period of mobility allows the electrolyte material to grow in a more defect-free manner.

[0150] In some embodiments, it is desired to form the electrolyte film **61C** as thin as possible to lower its contribution to the internal resistance of the energy-storage device. It is also desired to maintain the electrolyte's property of blocking the flow of electrons (which would result in a short of the cathode to the anode) while permitting the flow of the ions that provide the battery function across the electrolyte. Using the methods and systems described herein, the electrolyte film **61C** is formed to a thickness **61C'** of less than about 5000 Angstroms. In some embodiments, the electrolyte film **61C** has a thickness **61C'** of less than about 2500 Angstroms. In some embodiments, the electrolyte film **61C** has a thickness **61C'** of less than about 1000 Angstroms. In some embodiments, the electrolyte film **61C** has a thickness **61C'** of less than about 500 Angstroms. In some embodiments, the electrolyte film **61C** has a thickness **61C'** of less than about 250 Angstroms. In some embodiments, the electrolyte film **61C** has a thickness **61C'** of less than about 100 Angstroms. In some embodiments, the electrolyte film **61C** has a thickness **61C'** in a range of about 10 Angstroms to about 200 Angstroms. In some embodiments, the electrolyte film **61C** has a thickness **61C'** in a range of about 10 Angstroms to about 100 Angstroms.

[0151] In one embodiment, the electrolyte film **61C** includes LiPON and is formed using the first source **311** with the second source **313** or **413**. As used herein, LiPON refers generally to lithium phosphorus oxynitride materials. One example is $\text{Li}_3\text{PO}_4\text{N}$. Other examples incorporate higher ratios of nitrogen in order to increase lithium ion mobility across the electrolyte. In some embodiments, the first source **311** provides Li_3PO_4 in a nitrogen atmosphere. In other embodiments, the first source **311** provides Li_3PO_4 in a vacuum environment wherein the background pressure is less than $1\text{E-}3$ Torr. The second source **313** or **413** provides energized particles from a source gas. In some embodiments, the secondary source is an ion source supplying energetic ions from a source gas comprising oxygen (e.g., O_2) or nitrogen (e.g., N_2). The source gas, in other embodiments, comprises a noble gas, e.g., argon, xenon, helium, neon, and krypton. The energized particles and/or ions increase the energy of the material forming the electrolyte

film **61C**, thus enhancing layer-by-layer growth. Accordingly, the electrolyte film is of a higher quality than conventional electrolyte layers.

[0152] An embodiment for forming a LiPON electrolyte film **61C** includes the first source providing Li_3PO_4 at or to the location where the LiPON electrolyte film is to be formed and second source providing energized nitrogen particles to or near the same location. The energized nitrogen particles react with Li_3PO_4 provided at the location for forming the electrolyte film. This increases the amount of nitrogen in the LiPON electrolyte film. Increasing the nitrogen content is desirable to increase lithium ion mobility across the electrolyte.

[0153] In a further embodiment, the chamber in which the substrate **55** is positioned has a nitrogen enhanced atmosphere. A LiPON electrolyte film **61C** is formed by the Li_3PO_4 supplied by the first source reacting with the nitrogen in the chamber. The second source provides energized particles assisting in the formation of the electrolyte film. In another embodiment, the second source also provides nitrogen to the Li_3PO_4 at the location. Thus, the Li_3PO_4 reacts with both the nitrogen in the chamber and with energized, nitrogen containing particles supplied by the second source. This increases the nitrogen content of the electrolyte film **61C**. In some embodiments, increasing the nitrogen content in the electrolyte film **61C** is desirable since published data from the Department of Energy lab at Oak Ridge, Tenn. indicates an increase in nitrogen content increases the ion conductivity or mobility in the electrolyte film.

[0154] As will be understood by reading the present invention, the systems shown herein for depositing films are adaptable to form the electrolyte film **61C** according to the present invention. Examples of some such systems are shown in FIGS. 3-7.

[0155] FIG. 1D shows another embodiment of an energy storage device according to the teachings of the present invention. A supercapacitor **70** is formed on the energy-storage device **50C** having the ultra-thin electrolyte film **61**. The energy-storage device **50C** being formed on the substrate prior to forming the supercapacitor **70** represents an embodiment of layer/devices being formed on the substrate prior to applying the techniques described herein to form energy-storage and/or energy conversion devices. The supercapacitor **70** includes an intermediate film **73** formed in physical contact with electrode films **71** and **75**. In some embodiments, the intermediate film **73** is an electrolyte for storing and discharging electrical charge by a faradaic process. In some embodiments, the intermediate film **73** includes a dielectric material. The contact film **65** is in physical and electrical contact with electrode **71**. Thus, in this embodiment contact film **65** is a shared contact film for both the energy storage device **50C** and supercapacitor **70**. In other embodiments, energy storage device **50C** and supercapacitor **70** have separate contact films. In some embodiments, the intermediate film **73** includes LiPON. In some embodiments, the electrolyte film **73** includes TaO. In some embodiments, the electrode films are RuO_2 . A contact film **77** is formed on the electrode film **75**. A lead **76** extends from the contact film **77** to contact one plate of the supercapacitor to an external circuit.

[0156] A method for fabricating the solid-state energy-storage device **50** will now be described with reference to

FIGS. 1B and 2B. The method includes providing a substrate **55** (step **251**) and depositing a cathode contact film **57** on the substrate **55** (step **253**). In some embodiments, step **251** includes providing a substrate having insulator layers or other layers/devices formed thereon. The method further includes a step **255** of depositing an electrode material to a location on the substrate, while simultaneously supplying energized particles to the electrode material at the substrate. In one embodiment, an assist source provides the energized particles. In some such embodiments, the energized particle beam is directed to the same location on the substrate as the electrode material. In an embodiment, the energized particles are energized ions. The energized ions, in an embodiment, include a material that is different than the electrode material. The energized particles or the ion beam assist in controlling growth of the structure of the electrode material at the location. In some embodiments, step **255** is used to form a cathode film or layer **59** for a solid-state, thin-film battery. The cathode film **59** is in electrical and physical contact with the cathode contact. An electrolyte film **61** is deposited, step **257**, on the cathode film **59**. An anode film **63** is deposited, step **259**, on the electrolyte film. The electrolyte film **61** separates the cathode and anode films **59** and **63** to prevent shorting the energy-storage device **50**, e.g., battery. An anode contact is formed, step **261**, in electrical and physical contact with the anode film. The thin-film battery according to the present invention is now formed and is subjected to post energy-storage device fabrication steps **263**.

[0157] The deposition of the cathode film includes directing a first material (e.g., adatoms) to a location on the substrate, while simultaneously supplying energized particles (e.g., ions) of a second material to the location on the substrate. In some embodiments, the second material is different from the first material. The energized particles supply energy to the first material to assist in the growth of a desirable crystal structure in the cathode film. Moreover, this controls the stoichiometry of the growing film at the location on the substrate. In one embodiment, the first material is a lithium-intercalation material used as a solid-state, thin-film battery cathode. The assist source provides ions that provide energy in a range of 5 eV to 3000 eV to the lithium-intercalation material. Control of the energy in the ions produced by the assist source provides in situ control for growing a lithium-intercalation film having a crystalline structure. The energy from the ions assists the formation of lithium-intercalation materials into a crystalline structure at the time of deposition. In one embodiment, the gas used to form the ions is used to control the stoichiometry of the growing, crystalline film. For example, an ionized, assist beam of O_2 is used to control the growth and stoichiometry of a $LiCoO_2$ intercalation material. In some such embodiments, the O_2 in the ion assist beam combines with $LiCo$ at the location to form the $LiCoO_2$ intercalation material.

[0158] The crystalline structure of a thin film formed according to the teachings herein has a higher order than those achieved by conventional cathode film forming techniques. Conventional techniques rely on a high-temperature, post-cathode-deposition anneal to reorder and crystallize the structure of a conventional cathode film. Unfortunately, such conventional techniques anneal the entire structure to the same temperatures, which is undesirable in that the substrate must withstand such temperatures which eliminates many otherwise suitable substrate materials from consideration.

Further, different layers cannot be provided with different anneals suited to their different requirements. A highly ordered crystalline cathode film is desirably achieved according to the teachings described herein by providing the required energy to form the desired, high-order and appropriately oriented crystal structure without subjecting the substrate, and other layers formed on the substrate including the cathode-contact film to a high-temperature anneal. Further, each layer can be annealed using a different anneal process (such as using ion assist beams having different energies for different layers, or depositing and annealing at different rates or for different durations). Further, by annealing the surface layer of the previous layer, a subsequent layer can be deposited onto a surface that has been ordered in a specific way (for example, to achieve a specific crystal orientation, or a specific ion-bonding surface) that enhances the quality of that subsequent layer.

[0159] FIG. 2C shows one embodiment of a method for fabricating an energy-storage device. Steps **251**, **253**, **259**, **261**, and **263** are the substantially similar to the steps described above with reference to FIG. 2B. Step **255C** is a step for depositing a cathode film at least partially on the cathode contact film. In an embodiment, the cathode film is deposited as described above in step **255**. In other embodiments, the cathode film is deposited according to other deposition processes known in the art. The electrolyte film is formed by depositing an electrolyte material to a location at least partially in contact with the cathode film (step **257B**). In a preferred embodiment, the electrolyte material is in contact with a substantial portion, if not all of, a surface of the cathode film. In some embodiments, an assist source simultaneously supplies energized particles to the electrolyte material as it forms the electrolyte film. In an embodiment, the assist source supplies a beam of energized ions of an assist material different than the electrolyte material. In one embodiment, the second material beam is directed to the same location on the substrate as the electrolyte material. The energized ion beam assists in controlling growth of the structure of the electrolyte film. The ion beam is unfocused in one embodiment. The ion beam is focussed in another embodiment.

[0160] The deposition of the electrolyte film includes directing an electrolyte material to a location at least partially in contact with the cathode film, while simultaneously supplying energy to the electrolyte material. In one embodiment, the energy is supplied by energized particles. In some such embodiments, the energized particles are energized ions. In some such embodiments, the energized particles from the assist source are of a different material than the electrolyte material. The energized particles supply energy to the electrolyte first material to assist in the growth of a desirable, solid electrolyte-film structure. Moreover, this controls the stoichiometry of the growing electrolyte film.

[0161] In one example, the electrolyte material is a lithium phosphorus oxynitride. In some embodiments, the assist source provides ions that provide energy in a range of about 5 eV to about 5000 eV to the lithium phosphorus oxynitride ("LiPON"). Control of the energy in the ions produced by the assist source provides in situ control for growing a lithium phosphorus oxynitride structure at the location. The energy from the ions assists the formation of the lithium phosphorus oxynitride material into a desirable structure at the time of deposition. In one embodiment, the gas used to

form the ions is used to control the stoichiometry of the growing electrolyte film. For example, an ionized assist beam of O_2 is used to control the growth and stoichiometry of a lithium phosphorus oxynitride material. In another embodiment, an ionized assist beam of N_2 is used. In this embodiment, the N_2 not only controls growth and stoichiometry of the electrolyte film, but also injects additional nitrogen into the electrolyte film. This is desirable due to the ionic transportivity of a LiPON electrolyte film is dependant on the amount of nitrogen in the film.

[0162] FIG. 2D shows one embodiment of a method for fabricating an energy-storage device. Steps 251, 253, 257, 261, and 263 are substantially similar to the steps described above with reference to FIG. 2B. Step 255C is a step for depositing a cathode film at least partially on the cathode contact film. In an embodiment, the cathode film is deposited as described above with reference to FIG. 2B. In other embodiments, the cathode film is deposited according to other deposition processes known in the art. Step 259D is a step for depositing an electrode material to a location at least partially on the electrolyte film, while simultaneously supplying energized particles to the electrode material. In one embodiment, the energized particles are directed to the same location as the electrode material. In an embodiment, the energized particles are energized ions. The energized ions, in an embodiment, include a second material that is different than the first material. The energized particles or the ion beam assist in controlling growth of the structure of the electrode material. Step 259D, in some embodiments, is used to form an anode film for a solid-state thin-film battery. The anode film is in electrical and physical contact with the anode contact and electrolyte films.

[0163] The deposition of the anode film includes directing an electrode material to a location at least partially in contact with the electrolyte film, while simultaneously supplying energized particles of a second material. The energized particles supply energy to the electrode material to assist in the growth of a desirable crystal structure in the anode film. Moreover, this controls the stoichiometry of the growing film. In one embodiment, the electrode material includes a lithium-intercalation material used as a battery anode. In an embodiment, the anode includes is a lithium metal or a lithium alloy. In another embodiment, the anode includes a carbonaceous material, such as graphite or diamond-like carbon. In another embodiment, the anode includes a metal oxide, for example, RuO or VaO. In another embodiment, the anode includes a nitride material. A secondary source provides particles, which are ions in some embodiments that provide energy in a range of about 5 eV to about 3000 eV to the lithium-intercalation material. Control of the energy in the ions produced by the secondary source provides in situ control for growing a lithium-intercalation crystalline structure at the location. The energy from the ions assists the formation of lithium-intercalation materials into a crystalline structure at the time of deposition. In one embodiment, the gas used to form the ions is used to control the stoichiometry of the growing, crystalline film.

[0164] The crystalline structure of an electrode thin film formed according to the teachings herein has a higher order than those achieved by conventional film forming techniques. Conventional techniques rely on a high-temperature, post-deposition anneal that affects the substrate and other layers as well as the film intended to reorder and crystallize

the structure of that film. In contrast, the present invention provides a controlled energy source at the time of deposition or after the time of deposition that reorders the surface of the deposition film without substantially heating the underlying layers or substrate. In some embodiments, the energy is provided while depositing each atomic layer of a film such that each atomic layer is ordered as crystallizes into the film. Examples of such energy sources include an ion beam that either react with the adatoms being deposited and/or provide kinetic energy to assist in deposition of the film. Other examples of energy sources include high temperature, short duration heat sources, short duration plasma sources, lasers, other high intensity photo sources that reorder the crystal structure adjacent the surface of the film without effecting other layers or the substrate. A highly ordered crystalline cathode or anode is desirably achieved according to the teachings described herein.

[0165] While the above fabrication process describes forming cathode and anode films in a certain order, other embodiments reverse the order of the cathode film and anode film. Moreover, the fabrication process describes forming cathode and anode films, for example in a battery. In some embodiments, the cathode and anode films are electrodes of a battery. Other embodiments include films forming various layers of supercapacitors. Supercapacitors operate In these embodiments, at least one of the films forming the supercapacitor, e.g., electrode films 71, 75 and electrolyte and/or dielectric film 73, have improved crystalline structure, crystallite size, or fewer defects without resorting to a high temperature anneal of the entire structure to provide these properties. Accordingly, techniques and systems for fabricating thin films for use in an energy-storage device as described herein are applicable to both solid-state batteries and solid-state capacitors.

[0166] In another embodiment, the thin-film energy-storage device is formed on a substrate. A contact film, which is electrically conductive and does not react with a subsequently deposited, adjacent cathode film, is formed on the substrate. The contact film acts as a barrier between the substrate and the cathode film. The contact film further acts as a current collector and as a connection between the cathode film and circuits that are external to the energy-storage device. In an embodiment, the contact film has a thickness of greater than 0.3 microns.

[0167] FIG. 3A shows a deposition apparatus 305 including a reaction chamber 307 in which is positioned a substrate 309 on which an energy-storage device is to be fabricated. Reaction chamber 307, in one embodiment, is a sealed chamber that holds gases for the reaction and that provides a sub-atmospheric pressure. In some embodiments, it is desirable to hold the pressure in the chamber less than about 1×10^{-3} Torr. A first material source 311 is provided in the chamber 307. The first source 311 produces a beam of adatoms 312 of a first material to be deposited on the substrate 309. In one embodiment, the first material source 311 is a physical vapor deposition source. In one such embodiment, the material source 311 is an e-beam source. In another such embodiment, the first source 311 is an arc source including, for example, a cathodic-arc source, an anodic-arc source, and a CAVAD arc source. Arc sources are particularly suited for use as a source as they effectively operate in a chamber that is operated at low temperatures. In another embodiment, the first source 311 is a physical

deposition source including, for example, a sputtering source. In another embodiment, the source **311** is a chemical vapor deposition source including, for example, a direct ion source using a hydrocarbon precursor gas. Beam **312** is focused on a location **319** on the substrate **309** whereat the material of the beam **312** is deposited to form a film of an energy-storage device. An assist source **313** is provided in the chamber **307** and produces a beam of energized particles **314** directed at least adjacent to the location **319** on the substrate **309**. In some embodiments, the assist source is an energized ion-producing source. In some embodiment, the assist source **313** is offset from the first source **311** such that the beams from these sources are not coincident. The energized particle beam **314** provides the energy that is required to control the growth and stoichiometry of the material in the first beam **312** into a crystalline structure on the substrate **309** as is explained in greater detail herein. In one embodiment, the energized particle beam **314** also provides elements that are required in the film being deposited. In another embodiment, beam **314** is directed at least near location **319** such that sufficient energy to form the desired crystal structure and stoichiometry of the film being deposited is supplied by beam **314** to the material in first beam **312**. In some embodiments, the deposition system **305** includes at least one additional assist source **313A**. In some embodiments, each of the sources **313A** provides an additional assist beam **314A** that provides energy to arriving adatoms at the substrate. Various embodiments of assist beams **314** are described below.

[0168] FIG. 3B shows another embodiment of a deposition apparatus **305**. The assist source **313** produces an energy beam **314** that travels along a path that is essentially normal to the substrate **319**. The source of material to be deposited **311** is offset from assist source **313**. In some embodiments, source **311** produces a beam of adatoms **312** that travels along a path that is non-normal to the substrate **319**. The energy beam supplies energy to the adatoms from beam **312** as described herein.

[0169] FIG. 4 is a view substantially similar to FIG. 3A, except that depositing apparatus **405** includes an assist source **413** for producing the energized beam that is pivotally mounted to a bracket fixed in the chamber **307**. The assist source **413** pivots to direct the energized particle beam **414** at a desired impingement angle to the surface of the substrate **309**. In an embodiment, the impingement angle is in the range of about 15 degrees to about 70 degrees from normal to the substrate. Accordingly, in some embodiments, the impingement angle is variable. In one embodiment, the impingement angle is about 45 degrees. In some embodiments, the deposition system **405** includes at least one additional assist source **413A**. In some embodiments, each of the sources **413A** provides an additional assist beam **414A** that provides energy to arriving adatoms at the substrate. In some embodiments, the energy provided by assist beam **414** differs from the energy provided by at least one of assist beams **414A**. In some embodiments, the assist beam **414** and **414A** need not simultaneously transmit energy to the adatoms. In some embodiments, the means by which the beams **414** and **414A** transmit energy are different. In some embodiments, the material in beams **414** and **414A** are different.

[0170] FIG. 5A is a view substantially similar to FIG. 3 except that depositing apparatus **505** includes a plurality of

first deposition sources **511**. In one embodiment, each one of the first deposition sources **511** directs its respective beam **512** to the location **319** on the substrate **309**. In some embodiments, every one of the first sources **511** produces a beam **512** including the same material. In other embodiments, at least of the first sources **511** produces a beam **512** of a material that is different than that of another of the first sources **511**. In some embodiments, the materials from the plurality of first beams **512** combine at the location **319** to form the desired film. In other embodiments, the materials in first beams **512** combine with material from assist beam **314** to form the desired film. In one embodiment, one of the first sources **511** directs its beam **512** to the substrate **319** but away from the location **319**. In some embodiments, a plurality of assist sources **313** provide energy to the adatoms of beams **512**.

[0171] FIG. 5B shows another embodiment of a depositing apparatus **505B**. A plurality of assist sources **313** is positioned to provide energy to a forming film at the substrate **319**. A plurality of material sources **511A**, **511B**, and **511C** supply material to the chamber **307** and adjacent the surface of the substrate **319**. In some embodiments, each of the material sources **511A**, **511B**, and **511C** provide a same material and, thus, have the ability to provide a greater quantity than one of the sources alone. In some embodiments, at least one of the material sources **511A**, **511B**, and **511C** provides a material different than another of the material sources. In some embodiments, these different materials react at the in chamber **307** to create the adatom material that will form a film on the substrate **319**. In some embodiments, at least one of the material sources **511A**, **511B**, and **511C** provides a precursor material into chamber **307** and another of the material sources provides a reactant material into the chamber. The precursor and reactant material react together to create the material that will form the film. In some embodiments, at least one of the material sources **511A**, **511B**, and **511C** includes a chemical reactor in which chemicals react. This source then injects the resultant material into the chamber. The resultant material is included in the film fabrication process.

[0172] FIG. 6 is a view substantially similar to FIG. 5A except that depositing apparatus **605** includes a plurality of first deposition sources **511** and a pivotable assist source **413**. In some embodiments, this provides more material to a given deposition location. In some embodiments, this provides deposition at multiple locations. In still other embodiments, this allows different materials from different sources to be combined.

[0173] FIG. 7 shows another embodiment of a depositing apparatus **705** according to the teachings of the present invention. Depositing apparatus **705** includes a reaction chamber **707** in which is positioned an elongate, flexible substrate **709** on which an energy-storage device is to be fabricated. The substrate **709** is fed from a source roll **710** over an arched thermal control surface **715** and taken up by an end roll **713**. A first material source **711** is provided in the chamber **707** and is a physical deposition source. First source **711** produces a beam of adatoms **712** of a material to be deposited on the substrate **709**. In one embodiment, the first source **711** is an arc source including, for example, a cathodic arc source, an anodic arc source, and a CAVAD arc source. In another embodiment, the first source **711** is a physical vapor deposition source including, for example, a

sputtering source. In another embodiment, source 711 is a chemical vapor deposition source. Moreover, source 711, in some embodiments, represents a plurality of different material sources. Beam 712 is focused on a location 719 on the substrate 709 whereat the adatoms in the beam are deposited to form a film layer of an energy-storage device. An assist source 713 is provided in the chamber 707 and produces a beam of energized particles 714 directed at the substrate 709. In an embodiment, the assist source 713 produces a beam of energized ions 714. The energized particle beam 714 provides the energy required to control growth and stoichiometry of the deposited material of the first beam 712. Thus, a crystalline structure is formed on the substrate 709 as is explained in greater detail herein. The substrate 709, in one embodiment, is an elastomer, polymer, or plastic web or sheet on which the energy-storage device is fabricated. Substrate 709 being elongate allows a plurality of energy-storage devices to be deposited on successive locations of the substrate, thereby improving the rate of energy device production. Moreover, a plurality of deposition apparatuses 705 or sources 711, in some embodiments, are provided for simultaneously depositing a plurality of films at different locations on the substrate 709.

[0174] The thermal control surface 715 is connected to a thermal source 725, which controls the temperature of surface 715. The substrate 709 is in thermodynamic contact with surface 715 to thereby control the temperature of the substrate as needed for a particular deposition process on a particular substrate. In one embodiment, the thermal source is a coolant source, for example a cryogenic vacuum pump that releases compressed helium toward the surface 715 to cool it. The use of a thermally controlled surface 715 in direct contact with the substrate 709, especially when the direct contact is aligned or coincident with the location whereat a thin film is being formed, allows the use of substrates that have lower thermal degradation temperatures than are possible using conventional solid-state thin-film battery fabrication processes

[0175] The above provides descriptions of various embodiments of systems in which the present invention is performed to produce energy-storage devices or energy-conversion devices. It is within the scope of the present invention to combine the elements of the systems in different ways than shown and described as long as the methods described herein are performable with such a system. For example, in some embodiments, the flexible substrate 709 and rolls 710, 713 can be combined with any of the embodiments shown in FIGS. 3A-6. In some embodiments, the thermal source 725 is also combinable with any of the embodiments of FIGS. 3A-6. In some embodiments, the pivotable assist sources 413 are combinable with any of the embodiments of FIGS. 3A, 3B, 5A, 5B, and 7. In some embodiments, the material sources 511A, 511B, and 511C are combinable with embodiments of FIGS. 3A-5A and 6-7.

[0176] In one embodiment, the electrode second film, e.g., films 59 or 71 is a lithium-intercalation material which overlays at least part of the first film, e.g., contact films 57 or 63, but does not extend beyond the boundary of the first film. Thus, the intercalation second film remains in a solid state during discharging and charging of the energy-storage device. In some embodiments, the second film is deposited using the first deposition source simultaneously with the secondary source supplying energetic ions to the growing

second film. In some embodiments, the first deposition source is a physical vapor deposition source. In some embodiments, the secondary source is an ion source supplying energetic ions from a source gas comprising oxygen (e.g., O₂) or nitrogen (e.g., N₂). The source gas, in another embodiment, comprises a noble gas, e.g., argon, xenon, helium, neon, and krypton. The source gas, in yet another embodiment, comprises a hydrocarbon material such as a hydrocarbon precursor. Selection of the secondary source gas is based on the desired effect on the stoichiometry of the deposited film. The secondary source, in one embodiment, provides a focused beam of energized ions. The secondary source, in one embodiment, provides an unfocused beam of energized ions. The energized ions provide energy to the lithium-intercalation material in the range of about 5 eV to about 3,000 eV. In one embodiment, the energy range of is about 5 eV to about 1,000 eV. The energy range in a further embodiment is about 10 eV to about 500 eV. The energy range in a further embodiment is about 30 eV to about 300 eV. In another embodiment, the energy range is in the range of about 60 eV to 150 eV. In another embodiment, the energy range is about 140 eV. In an embodiment, the second film has a thickness of greater than 10 microns. In one embodiment, the second film has a thickness in the range of about 10 to 20 microns. In one embodiment, the second film has a thickness in the range of about 1 to 5 microns.

[0177] An electrolyte third film, e.g., films 61, 61C or 73, having ionic transport qualities but not being electrically conductive (an electrolyte) is deposited so as to completely overlay the second deposited film. In one embodiment, the third film is deposited using a first deposition source and a secondary source supplying energetic ions to the growing film. In some embodiments, the first deposition source is a physical vapor deposition source. In some embodiments, the secondary source is an ion source with the capability of supplying energetic ions having an energy greater than 5 eV. In another embodiment, the energy range is about 5 eV to about 3,000 eV. In one embodiment, the energy range of is about 5 eV to about 1,000 eV. The energy range in a further embodiment is about 10 eV to about 500 eV. The energy range in a further embodiment is about 30 eV to about 300 eV. In another embodiment, the energy range is in the range of about 60 eV to 150 eV. In another embodiment, the energy of the ions from the secondary source is about 140 eV. In some embodiments, the secondary source includes oxygen (e.g., O₂) or nitrogen (e.g., N₂) gas. The secondary source gas, in another embodiment, includes a noble gas, e.g., argon, xenon, helium, neon, and krypton. The secondary source gas, in another embodiment, includes a hydrocarbon material such as a hydrocarbon precursor. Selection of the secondary source gas is based on the desired effect on the stoichiometry of the deposited film. The secondary source, in one embodiment, provides a focused beam of energized ions. The secondary source, in one embodiment, provides a non-focused beam of energized ions. It is desirable to make the electrolyte, third layer as thin as possible and prevent the cathode and anode layers from shorting. In an embodiment, the third film has a thickness of less than 1 micron. In one embodiment, the third film has a thickness in of less than 5,000 Angstroms. In another embodiment, the third film has a thickness of less than 1,000 Angstroms. In another embodiment, the third film has a range of about 10 Angstroms to about 100 Angstroms.

[0178] In another embodiment, the third film is deposited using a first source supplying energetic ions (5 to 3000 eV) to a material source (target) at an impingement angle of 15 to 70 degrees and a second source supplying energetic ions to the growing film. The first deposition source includes a beam of focused energetic ions from a source gas. The source gas includes one of the sources gases described herein.

[0179] An anode, fourth film, e.g., film 65 or 75 includes from a lithium-intercalation material that is deposited on and overlays the third film but not contacting first film (barrier) or second film (cathode). In one embodiment, the fourth film is deposited using a first deposition source simultaneously with a secondary source supplying energetic ions to the growing fourth film. In some embodiments, first deposition source is a physical vapor deposition source. In some embodiments, the secondary source is an ion source supplying energetic ions from a source gas that includes oxygen (e.g., O₂) or nitrogen (e.g., N₂). The source gas, in another embodiment, includes a noble gas, e.g., argon, xenon, helium, neon, and krypton. The source gas, in another embodiment, includes a hydrocarbon material such as a hydrocarbon precursor. Selection of the secondary source gas is based on the desired effect on the stoichiometry of the deposited film. The secondary source, in one embodiment, provides a focused beam of energized ions. The secondary source, in another embodiment, provides an unfocused beam of energized ions. The energized ions provide energy to the lithium-intercalation material in the range of about 5 eV to about 3,000 eV. In one embodiment, the energy range of is about 5 eV to about 1,000 eV. The energy range in a further embodiment is about 10 eV to about 500 eV. The energy range in a further embodiment is about 30 eV to about 00 eV. In another embodiment, the energy range is in the range of about 60 eV to 150 eV. In another embodiment, the energy range of the ions from the secondary source is about 140 eV. In an embodiment, the fourth film has a thickness of greater than 10 microns. In one embodiment, the fourth film has a thickness in the range of about 10 to 40 microns.

[0180] In another embodiment, the fourth film is deposited by plasma decomposition of hydrocarbon precursor(s) at the surface of the substrate thereby forming a lithium-intercalation anode. In some embodiments, deposition is performed by plasma enhanced CVD using hydrocarbon precursors. In one embodiment, the deposition includes dopants such as N₂. In one embodiment, a secondary source provides energized ions to assist in the deposition of the fourth film. The energized ions provide energy in the range as described herein. In some embodiments, the secondary source is the same as any described herein.

[0181] In another embodiment, the anode, fourth film is deposited by direct ion beam deposition of a lithium-intercalation material using hydrocarbon precursors. The first deposition source provides a beam of focused energetic ions (5 to 3000 eV) from a source gas hydrocarbon precursor directed at the target material. In one embodiment, a secondary source supplies energetic ions to assist in growing the fourth film and is a secondary source as described herein.

[0182] A contact, fifth film, e.g., film 65 or 77, which is electrically conductive and does not react with the fourth film is formed in contact with at least part of the fourth film. The fifth film does not contact the second film (cathode). In

an embodiment, the fifth film has a thickness of greater than 0.5 microns. The fifth film acts as an anode current collector for contact to external circuitry.

[0183] In some embodiments, a passivation, sixth film 79, which is electrically non-conductive and chemically inert, essentially overlays the energy-storage device as formed thus far, i.e., all the second, third, and fourth films, so that same are packaged and free from environmental contaminants that may react with these films and degrade performance of the energy-storage device. Environmental contaminants may include further fabrication materials for devices with the energy-storage device integrated therewith. In some embodiments, the first and fifth contact films are partially exposed outside the sixth film for connection to circuitry outside the energy-storage device.

[0184] The substrate 55, 309 or 709, on which the films described herein are deposited, includes any material capable of supporting a thin film and being able to withstand the deposition process described herein. In one embodiment, the substrate is formed of a material having a temperature at which it will begin to degrade due to thermal effects of less than 700 degrees Celsius. A further embodiment includes a substrate having such a temperature at which it experiences thermal degradation of less than or equal to about 300 degrees Celsius. Thermal degradation of the substrate includes loss of shape of the substrate, loss of sufficient rigidity to support an energy-storage device, chemical breakdown of the substrate, cross-linking of materials on the substrate and/or films, melting, and combustion. Examples of substrates include silicon wafers and silicon on insulator structures. Other examples of substrate materials include metals on which an insulator layer is formed prior to formation of the energy-storage device as described herein. In another example, the metal may act as a contact for the energy-storage device with insulator layers electrically separating the electrolyte film, the anode film and the anode contact from the metal substrate. Examples of other materials that have a low thermal degradation temperature that are suitable for fabricating an energy-storage device as disclosed herein include paper, fabrics (natural and synthetic), polymers, plastics, glasses, and ceramics.

[0185] The substrate 55, 309, or 709 has a form that is applicable to the type of apparatus used to fabricate the energy-storage device according to the teachings herein. One example of the substrate shape is a semiconductor wafer. Other forms of the substrate include elongate webs, weaves, foils, and sheets. It is within the scope of the present invention to provide a substrate having sufficient size on which a plurality of energy-storage devices and/or a plurality of energy conversion devices are fabricated.

[0186] One embodiment of the substrate 55, 309, or 709 includes a substrate that retains its support characteristics during an in situ temperature treatment. In the in situ temperature treatment, the substrate is placed in intimate contact with a thermally controlled surface, e.g., surface 715. In one embodiment, the thermally controlled surface is a cooled surface such that heat associated with deposition of any of the films described herein are thermally balanced so as not to thermally degrade the substrate or any other structural element previously formed on the substrate. Thus, in some embodiments, substrates having low thermal degradation temperatures, such as low melting points or low

combustion temperatures, are used as substrates in the present fabrication methods. For example, substrates include ceramics, glasses, polymers, plastics and paper based materials. In an embodiment according to the teachings herein, the substrate is a plastic or metal substrate on which a plurality of energy-storage devices is deposited. The substrate is then divided into separate dies having at least one energy-storage device thereon. The dies then can be worked, e.g., cold worked, into a desired shape as dictated by the energy-storage device application.

[0187] In another embodiment, the substrate is made of a flexible material, e.g., substrate **709**. The flexible substrate is formed into an elongate roll that is caused to pass over a curved object, which forces the material into intimate contact with the surface of the curved object. The curved object is a thermally controlled device (e.g., device **725** as shown in **FIG. 7**) to control the temperature of the substrate and balance the effect of heat generated on the substrate and films thereon during deposition. For example, the object is hollow and sealed from the environment of the deposition vessel. In some embodiments, the hollow space is filled with a coolant, e.g., cryogenic gas such as gas obtained from LN₂ or liquid helium, with the coolant being constantly replenished. An area of intimate contact between the substrate and object is coincident and opposite the location of material impingement on the substrate from the deposition source. In another embodiment, the coolant is chilled water that is constantly being replenished. In another embodiment, the curved object is thermally controlled by an electro-thermal cooling apparatus. In another embodiment, the curved object is a drum, which is either stationary or rotatable about its axis in the direction of substrate movement.

[0188] In another embodiment, the substrate **55** or **309** is formed of a strip of rigid material. The rigid substrate is made to pass over a cooled, thermally controlled surface. Examples of the cooled surface are described herein. One such example is a cooled surface that is cooled by the release of cryogenic fluid such as liquid N₂ or liquid helium into passages within the body of object having the surface but sealed from the environment of the deposition chamber. Other coolant sources include chilled water, cryogenic gas, and electro-thermal devices.

[0189] **FIG. 8** shows a photovoltaic cell **800**, e.g., solar cell, that includes a transparent electrode **810**. Transparent electrode **810** includes a transparent supporting film **820** and a transparent, electrically conductive film **830** formed on film **820**. Examples of supporting film **820** include glass and transparent plastics. In some embodiments, conductive film **830** includes indium tin oxide or tin oxide. In use, light **890**, enters solar cell **800** through the transparent electrode **810**. In some uses of embodiments, light **890** is solar light. A first semiconductor film **840** is positioned in contact with the transparent electrode **810**. A second semiconductor film **860** is positioned in contact with the first semiconductor film **840**, thereby, forming a semiconductor junction **850**. In some embodiments, second semiconductor film **860** includes a bulk, highly doped region **862** and a high quality region **863** adjacent the first semiconductor film **840**. In this embodiment, the junction is formed by the first semiconductor film **840** and region **863**. An electrical contact film **870** contacts the second semiconductor film **860**. First and second conductive leads **880** respectively contact the trans-

parent, electrically conductive film **830** and the electrical contact film **870** to carry power away from the cell.

[0190] In some embodiments, the materials and compositions of photovoltaic cell **800** are conventional CdS/CdTe materials such as is described in U.S. Pat. No. 4,207,119, which is incorporated by reference; with the additional processing according to the present invention to anneal or treat the surface (e.g., by ion-assist beam) of the films as they are deposited using. In other embodiments, the compositions used are as described in the following publications, each of which is incorporated by reference: R. W. Birkmire et al, "Polycrystalline Thin Film Solar Cells: Present Status and Future Potential," *Annu. Rev. Mater. Sci.* 1997.27:625-653 (1997); T. L. Chu et al, "13.4% Efficient thin-film CdS/CdTe Solar Cells," *J. Appl. Phys.* 70 (12) (Dec. 15, 1991); T. Yoshida, "Photovoltaic Properties of Screen-Printed CdTe/CdS Solar Cells on Indium-Tin-Oxide Coated Glass Substrates," *J. Electrochem. Soc.*, Vol. 142, No. 9, (September 1995); T. Aramoto et al., "16% Efficient Thin-Film CdS/CdTe Solar Cells," *Jpn. J. Appl. Phys.* Vol. 36 pp 6304-6305 (October 1997); R. B. King, ed. "Encyclopedia of Inorganic Chemistry" Vol 3., pp 1556-1602, John Wiley & Sons Ltd., (1994).

[0191] The brief description of the operation of a heterojunction, photovoltaic solar cell that follows is to illustrate how the methodology of the present invention is applied to the fabrication of heterojunction, photovoltaic solar cells. It is believed that the present invention provides means and methods for fabricating photovoltaic cells having superior efficiency.

[0192] In a heterojunction photovoltaic cell, the semiconductor films are formed of different materials. For a rectifying junction, the semiconductor films must also be of different type, that is p or n type. The junction between the two semiconductor films is both a pn junction and a heterojunction. The first semiconductor film on which solar light is incident has a band gap higher than that of the second semiconductor film. The band gap of a semiconductor is the energy separation between the semiconductor valance band and the conduction band. The band gap of this first semiconductor film is chosen so that it corresponds to light in the short wavelength region of the solar spectrum. Photons of light having energy equal to or greater than the band gap of the first semiconductor film are strongly absorbed, but photons of light of energy less than the band gap of the first semiconductor pass through the first semiconductor and enter the second semiconductor film. Examples of materials used for the first semiconductor film include CdS, ZnS, CdZnS, CdO, ZnO, CdZnO, or other wide band gap semiconductors like SiC, GaN, InGaN, and AlGaN. The second semiconductor film is chosen from materials that have band gaps that correspond well to the long wavelength onset of solar radiation. Materials such as CdTe, CuInSe₂, InP, GaAs, InGaAs, InGaP, and Si are examples of materials for the second semiconductor film.

[0193] A "built in" electric field exists at the junction between the two semiconductor films due to the migration of majority carriers from one semiconductor type into the other. That is, electrons from the n-type semiconductor migrate into the p-type semiconductor leaving a net positive charge on the n-semiconductor side of the junction. The converse happens to the p-type semiconductor. Holes from

the p-type semiconductor migrate into n-type semiconductor leaving a net negative charge on the p-semiconductor side of the junction. Absorption of a photon in one of the semiconductor films **840**, **860** results in the creation of an electron and a hole. When the photon is absorbed in the vicinity of the pn junction, the built in electric field separates the two carriers in opposite directions, electrons are driven to the n-type material and holes are driven to the p-type film. The separated charges result in a potential difference between the two semiconductor films **840**, **860**. This potential difference is used to drive a current through an external circuit thereby converting solar energy (photons) into electrical energy.

[0194] One embodiment of a heterojunction, photovoltaic solar cell is an n-type, polycrystalline CdS film as the first semiconductor film **840** and a p-type, polycrystalline CdTe film as the second semiconductor film **860**. CdS has a band gap of 2.43 eV that corresponds to 510 nm. CdTe has a band gap of 1.44 eV that corresponds to 860 nm. Solar radiation shorter than 860 nm and longer than 510 nm is absorbed in the p-type CdTe semiconductor film **860**. Each absorbed photon creates an electron hole pair. If the minority carrier, the electron in p-type CdTe, has a lifetime sufficiently long so that it can drift to the pn junction and be swept across the junction to the n-type CdS film, the absorbed photon contributes to solar cell photocurrent. Minority carrier lifetimes in p-type CdTe are long, which results in high quantum efficiencies (number of electrons created per number of photons absorbed at a particular wavelength) of ~90% between 860 nm and 510 nm. Most photons absorbed in the CdTe film contribute to the solar cell photocurrent.

[0195] Solar light at wavelengths shorter than 510 nm is absorbed in the n-type CdS film and creates an electron-hole pair. Minority carriers in n-type CdS, holes, have short lifetimes. Most photogenerated holes recombine with electrons in the n-type CdS film before they can be swept across the junction to the p-type CdTe film. Recombined electron-hole pairs do not contribute to the solar cell photocurrent. Creation of electron-hole pairs by absorption of solar radiation in the CdS film is detrimental to the overall efficiency of the solar cell. High-efficiency solar cells make the CdS film as thin as possible, ~50 nm, so that some fraction of solar radiation shorter than 510 nm can pass through the CdS film and be absorbed in the CdTe film where the photogenerated electron-hole pairs can be efficiently collected. A problem with this procedure is that, in some embodiments, thinning the n-type CdS film increases the series resistance of the cell, which also decreases the efficiency. Additionally, the CdS film must have some reasonable thickness, ~50 nm, to form a stable pn junction.

[0196] The deposition methods according to the present invention are used to enhance the performance of heterojunction solar cells by creating higher quality semiconductor films **840**, **860**. In some embodiments, semiconductor films **840**, **860** have structures that provide sufficiently long minority-carrier lifetimes to allow the minority carriers to be swept across the junction and contribute to the solar cell photocurrent. In some embodiments, higher quality films **840**, **860** are produced by providing energy focused at the surface where a film is being formed. In some embodiments, the energy is supplied simultaneously with the material to be deposited on a substrate. In some embodiments, higher quality films are created by depositing the primary material, for example, CdS in the film **840**, using a physical vapor

deposition technique while impinging energized particles from a second source on the film surface during the deposition. In some embodiments, the second source includes an ion source. In some embodiments, the ion source provides a beam of ions. In some embodiments, the beam of ions includes argon or xenon. In some embodiments, the beam of ions includes sulfur for depositing sulfide materials. In some embodiments, the beam of ions includes oxygen for depositing oxide materials. The effect of supplying focused energy is to increase the extent of crystallinity of the material being deposited. Another effect of supplying focused energy is to decrease defects that provide sites for electron-hole recombination. A further enhancement of the solar cell efficiency is achieved by using the focused energy to control the quality of the physical interface between the first semiconductor film **840** and the second semiconductor film **860**.

[0197] In an embodiment, the first film **840** is fabricated by providing energy to the material being deposited so that the material has fewer defects. With fewer defects the minority carriers will have longer lifetimes in film **840** as there will be fewer recombination sites. In some embodiments, first film **840** includes an n-type CdS material. In some embodiments, the first film **840** is formed in a range of about 40 nanometers to about 100 nanometers. In some embodiments, the first film **840** has a thickness of about 50 nanometers. In some embodiments, the first film **840** is formed in a range of about 40 nanometers to about 100 nanometers.

[0198] In some embodiments, the second film **860** includes two regions **862**, **863**. Region **863** is a high-quality region formed according to the teachings of the present invention. In some embodiments, region **862** is grown in a faster manner using conventional methods. In other embodiments, film **862** is merely a further growth of film **863** using the teachings of the present invention. High quality includes, among other things, fewer defects, larger crystal size, or certain structures being formed. Specifically, energy is supplied to the material of region **863** as the material is formed on the first film **840**. The energy is supplied according to the teachings herein, for example, by an ion-assist beam. In some embodiments, the energy is supplied by energized particles. In some embodiments, the energy is supplied by energized ions. In some embodiments, the energy is supplied by light or heat, e.g., a brief laser sweep of the surface. Due to the application of energy while the region **863** is being formed, a post-deposition high-temperature anneal is not required.

[0199] In some embodiments, the high quality region **863** has fewer defects than p-type regions of other photovoltaics. In some embodiments, region **863** has a thickness of at least about 50 nanometers. In some embodiments, region **863** has a thickness in a range of about 50 nanometers to about 100 nanometers.

[0200] In some embodiments, region **862** is larger than region **863**. In some embodiments, region **862** has a thickness of greater than 500 nanometers. In some embodiments, region **862** has a thickness in a range of 1 micron to 5 microns. In some embodiments, region **862** has a thickness of greater than 3 microns. In addition, region **862** is a highly doped p-type material.

[0201] In some embodiments, a chamber in which the films **840**, **860** are being deposited is held at a temperature

of less than 650 degrees Celsius. In some embodiments, the temperature of the chamber is less than about 300 degrees Celsius. In some embodiments, the temperature is between about 30 degrees Celsius and about 275 degrees Celsius. In some embodiments, the temperature is between about 100 degrees Celsius and about 200 degrees Celsius. In an embodiment, the substrate, e.g., glass layer 820 and conductor layer 830 for depositing film 840; glass layer 820, conductor layer 830, and film 840 for depositing region 863; and glass layer 820, conductor layer 830, film 840, and region 863 for depositing region 862, is not externally heated. Thus, the temperature of the substrate is generally equal to the temperature of the chamber plus minor heating effects of depositing the film. In contrast to prior methods for fabricating layers having sufficient quality such that the cell approaches about 10 percent efficiency, an embodiment of the present invention does not heat the substrate. Accordingly, manufacturing efficiencies are achieved while maintaining sufficient efficiency.

[0202] It is believed that some embodiments of the present invention will have conversion efficiencies of greater than about 5 percent. It is believed that some embodiments of the present invention will have conversion efficiencies of greater than about 6 percent. It is believed that some embodiments of the present invention will have conversion efficiencies of greater than about 7 percent. It is believed that some embodiments of the present invention will have conversion efficiencies of greater than about 8 percent. It is believed that some embodiments of the present invention will have conversion efficiencies of greater than about 9 percent. It is believed that some embodiments of the present invention will have conversion efficiencies of greater than about 10 percent. It is believed that some embodiments of the present invention will have conversion efficiencies of greater than about 11 percent.

[0203] Other embodiments for fabricating energy conversion devices, such as a photovoltaic cell 800, are fabricated according to many of the embodiments described herein with reference to energy storage devices. The thin films of the energy conversion devices are improved in a similar manner as described herein for the thin films of energy storage devices.

[0204] In contrast to some conventional methods for improving performance of a photovoltaic cell, the present methods can produce photovoltaic cells having an enhanced conversion efficiency without heat treating during deposition, e.g., heating the substrate, or a post-deposition high temperature anneal.

[0205] FIG. 9A shows a thin-film energy-storage device 910A according to the teachings of the present disclosure and an integrated circuit 940, here shown as a "flip chip". Energy-storage device 910A includes substrate 920 on which is formed a patterned wiring layer 922. The wiring layer 922 is an electrically conductive layer for connecting energy-storage device 920 to the integrated circuit 940. In some embodiments, layer 922 is formed of a metal. In one embodiment, the wiring layer 922 is patterned copper. In another embodiment, the wiring layer is formed of nickel. In other embodiments, the wiring layer is formed of a noble metal. Wiring layer 922 includes a cathode wiring pattern 922A and an anode wiring pattern 922B, which are separate from each other and form opposite polarity connectors 923A

and 923B to external circuitry, such as integrated circuit 940. Device 910A further includes a cathode contact film 924 formed on at least a portion of cathode wiring pattern 922A and an anode contact film 926 formed on at least a portion of the anode wiring pattern 922B. A cathode film 927 is formed on the cathode contact film 924 according to the teachings herein. An electrolyte film 928 is formed over the cathode film 926, cathode contact film 924 and a portion of the cathode-wiring pattern 922A. Electrolyte film 928 separates the cathode films 922A, 924 and 927 from respective anode films 922B, 926 and 932. Anode film 932 is formed on the electrolyte film and in contact with the anode contact film 926 according to the teachings herein. It will be appreciated that, in one embodiment, cathode contact film 924 and cathode wiring pattern 922A are formed as a single layer. It will be further appreciated that, in one embodiment, anode contact film 926 and anode wiring pattern 922B are formed as a single layer. A passivation layer 934 is formed over all of the films except portions 923A and 923B of the wiring patterns 922A and 922B, which portions are left exposed. Passivation layer 934 protects the films from contact to other layers, which may be formed on substrate 920, and the environment, which may include elements that may react with and damage the films of the energy-storage device 910A.

[0206] In some embodiments, the cathode materials and other materials used in the batteries above include materials discussed more in N. J. Dudney et al, "Nanocrystalline $\text{LiMn}_{1-y}\text{O}_4$ Cathodes for Solid-State Thin-Film Rechargeable Lithium Batteries," *Journal of the Electrochemical Society*, 146(7) 2455-2464 (1999) which is incorporated by reference.

[0207] The integrated circuit 940 includes a first ball contact 941 and a second ball contact 942 both extending outside a package. The first ball contact 941 aligns with the exposed portion 923A of the cathode wiring pattern 922A. The second ball contact 942 aligns with the exposed portion 923B of the anode wiring pattern 922B. Integrated circuit 940 is positioned so that the ball contacts 941 and 942 physically and electrically contact the wiring contacts 923A and 923B, respectively. Integrated circuit 940 is fixed in position relative to the device 910A such that device 910A provides electrical energy to circuit 940. In some embodiments, circuit 940 is provided with circuitry for recharging energy-storage device 910A. It will be recognized that the present invention is not limited to only integrated circuit 940 being connected to wiring contacts 923A and 923B. Other circuits, including integrated circuits fabricated on substrate 920 and circuits with leads connected to wiring contacts 923A and 923B, are within the scope of the present invention.

[0208] FIG. 9B shows another embodiment of the thin-film energy-storage device 910B, substantially similar elements to those described above are designated by the same reference numerals. After forming wiring patterns 922A, 922B, an insulator layer 930 is formed on the substrate 920. Insulator layer separates the thin-film energy-storage device 910 from other layers that may be included with substrate 920. Insulator layer 930 includes vias 931 through which cathode contact film 924 and anode contact film 926 extend downward to connect to cathode contact wiring pattern 922A and anode wiring pattern 922B, respectively.

[0209] In one example of an energy-storage device 910 according to the present invention, the cathode film 927 is a LiCoO_2 deposited using a first source of LiCoO with a secondary source of oxygen. The electrolyte film 928 is LiPON deposited using a first source of LiPO (such as Li_3PO_4) and an assist of nitrogen. The anode film 932 is a metal, e.g., copper, and is deposited by a first source of copper and a secondary source of an inert material, e.g., xenon. In another embodiment, the anode film includes carbon. In yet another embodiment, the anode is formed of pure lithium. In some embodiments, the anode is a lithium alloy. In some embodiments, the anode includes an oxide.

[0210] FIG. 9C shows a further embodiment of a thin-film energy-storage device 910C. This device 910C includes a seed layer 950 formed on the cathode contact 924. Seed layer 950 is formed on the cathode contact 924 prior to forming the cathode film 927, as described herein, on the seed layer 950 and substrate 920. Seed layer 950 is formed using deposition techniques as described herein, e.g., physical vapor deposition such as arc source deposition. Seed layer 950 is a very thin, electrically conductive layer and has a small crystal size. The seed layer 950 also has a high sheet resistance and is non-reactive with the materials of adjacent films. In an embodiment, seed layer 950 has a thickness that is substantially thinner than the adjacent electrode film 927. The material of the seed layer 950 is chosen such that the arriving adatoms of the subsequent material (e.g., in some embodiments, the material from the first source 311, 511 or 711) would have sufficient mobility to allow a period of activity once the adatom contacts the seed layer surface. This improves nucleation of the first few molecular layers of arriving material, minimizes strain associated with lattice mismatch and assists the arriving material to grow in a manner consistent with the desired crystal structure for cathode film 927.

[0211] In some embodiments, a seed layer 955 is formed on the electrolyte layer 928 prior to forming anode film 932, as described herein, on the seed layer 955. Seed layer 955 improves nucleation of the first few molecular layers of arriving material, minimizes strain associated with lattice mismatch and assists the arriving material to grow in a manner consistent with the desired crystal structure for anode film 932.

[0212] The ion transport properties of the materials used in the fabrication of energy-storage devices 910C, e.g., rechargeable batteries, greatly influence the operation and quality of the device. For example, the total energy-storage capability of solid-state, lithium-ion batteries of a given area is limited by a depletion region that forms at or near the cathode/electrolyte interface. The depletion of this region and the inability for additional lithium ions to be transported out of the bulk of the cathode film 927 results in limited capacity and, thus, more frequent recharges. Additionally, the efficiency of the lithium ion transport through the electrolyte film 928 controls and dictates the maximum discharge rate that can be achieved for a given structure. The seed layer 950 improves the crystalline structure of the materials subsequently deposited, i.e., a cathode film 927 or an anode film. The growth of the first few atomic layers of a material significantly impacts its overall structure even when the final film is very thick relative to the initial few atomic layers. If the "seed" material is chosen such that the surface energy kinetics are conducive to pseudo-epitaxial

growth of the subsequent material, high quality cathode and anode (electrode) films 927 and 932 are achieved. Examples of materials for seed layer 950 include chromium, chromium nitride, tantalum, tantalum nitride, tungsten, tungsten nitride, ruthenium and ruthenium nitride.

[0213] The thin-film energy-storage device fabricated according to the present teachings stores electrical energy by introducing ions into a storage layer and removing the ions from the storage layer to create an electrical potential at the contacts. In one embodiment, lithium ions are stored in an anode formed of a lithium-intercalation material with the battery in a charged state. In some embodiments, the anode is formed of a metal or a carbonaceous material. The lithium ions travel from the anode through the electrolyte layer to a cathode, which is also formed of a lithium-intercalation material, to discharge electrical energy from the battery. In order to achieve sufficient energy density to operate external circuitry, the lithium-intercalation material cathode and anode must intercalate (i.e., add) and de-intercalate (i.e., remove) of a substantial mole fraction of lithium ions. It has been found that the choice of intercalation material and fabrication techniques for the cathode determine many operating parameters of a solid-state, thin-film battery. The operating parameters include, but are not limited to, operating voltage range, capacity, specific power, and specific energy. One method of measuring the transport properties of ions in a battery is diffusivity, which is measured by a diffusion coefficient. The diffusion coefficient is a measure of how well a particular material allows ions to diffuse into and out of the material.

[0214] FIG. 10 shows comparative data for LiCoO_2 cathode films in the form of X-ray diffraction spectra. The LiCoO_2 cathode films were created according to the teachings herein and according to a control process that did not include a secondary source assist. A first source supplied a LiCoO_2 material using an electron-beam evaporation process. An assist, second source provided energy in the form of oxygen ions impinging at the location on the substrate whereat it is desired to grow a thin electrode (cathode) film from the LiCoO_2 first material. The beam of oxygen ions from the second source is not co-incident with the LiCoO_2 material from the first source. Four samples of LiCoO_2 thin films were grown according to the data in Table I.

TABLE I

Deposition Parameters	Film a	Film b	Film c	Film d
Electron beam power* (W)	500	500	500	500
Total gas flow, $\text{O}_2 + \text{Ar}$, (sccm)	0	10.8	10.8	10.8
O_2 gas ratio, $\text{O}_2/(\text{O}_2 + \text{Ar})$, (%)	NA	0.48	0.48	0.48
Chamber pressure (Torr)	9.2×10^{-7}	2.6×10^{-6}	3.7×10^{-6}	5.0×10^{-6}
Ion source power (W)	0	123	128	135
Ion source acceleration voltage	0	41	64	135

[0215] The electron beam voltage for each first source used in forming films a-d is 5 kV with an emission current of 100 mA.

[0216] FIG. 10 shows that the LiCoO_2 films deposited with lower energy oxygen ions from the second source,

samples "b" and "c", enhanced the formation of the desirable crystallite structure of the grown film relative to the non-assisted sample "a". Specifically, a more distinct (003) orientation of the crystal structure is found in ion-assisted samples "b" and "c" than in non-assisted sample "a". A strong (003) X-ray diffraction peak indicates one desired crystal orientation of the LiCoO_2 thin film. The (003) X-ray diffraction peak indicates that the film has lattice planes parallel to the substrate, e.g., layer on which the film was deposited. The (003) peak width, full width at half maximum ("FWHM") decreases and the X-ray peak increases in this series of samples as the energy of the oxygen ions impinging the deposited material increases. These examples indicate an increasing crystallite grain size and a larger fraction of ordered grains for sample films "b" and "c" than are found in sample film "a". The (003) orientation of samples "b" and "c" is preferable over an essentially non-ordered, non-crystallized structure of sample "a".

[0217] FIG. 10 further shows a sample "d" that was deposited using the highest energy secondary source of this example. Sample "d" was deposited using a secondary source energy of 135 eV. X-ray diffraction of sample "d" shows it has the most distinct (101) orientation of all the samples described herein. The desired (101) orientation has lattice planes, which contain lithium ions in a LiCoO electrode material, nearly perpendicular to the substrate. In this orientation, the lattice planes are essentially parallel to the direction of travel of the ions and in the direction nearly perpendicular to the substrate. As this is the direction lithium ions must travel in a lithium battery fabricated according to the embodiments described herein, the preferential (101) orientation leads to superior charging and discharging characteristics. As lithium transport through the LiCoO_2 film in the (101) preferential orientation does not rely on diffusion along grain boundaries, which can trap lithium ions and prevent their utilization, the preferential (101) orientation also leads to greater capacity and cycle lifetime. Consequently, this preferred orientation of the LiCoO_2 thin film is produced without additional anneal fabrication steps and the internal resistance is lower with lower capacity loss at high discharge rates.

[0218] FIG. 11 shows a comparison of the (003) x-ray diffraction peak of an ion assisted LiCoO_2 film fabricated according to the teachings of the present invention and a conventionally magnetron sputtered LiCoO_2 film. Both spectra are for as-deposited films. The ion assisted LiCoO_2 film in this spectrum is the same as the "c" sample shown in FIG. 10. The sputtered LiCoO_2 film was fabricated in an MRC 8667 using 1200 watts RF power, 10% O_2 in Argon, 80 sccm total gas flow (8 sccm O_2 and 72 sccm Ar), 20 mTorr pressure, with the substrate table grounded. Film thickness of the sputtered LiCoO_2 film is 5460 Angstroms. The significantly sharper peak for the ion-assisted film indicates the higher degree of long range order in this film. The peak width for this film is approaching that obtained by high temperature annealing of a similar conventionally magnetron sputtered film and exceeds that achieved for 300 degree Celsius annealed films of LiMn_2O_4 . Accordingly, the LiCoO_2 film fabricated according to the teachings of the present invention provides a higher degree of order than conventional LiCoO_2 films without resorting to a post deposition anneal step to provide the desired crystal structure in the film. This results in significant manufacturing efficiencies.

[0219] FIG. 12A shows a X-ray diffraction spectra of a LiCoO_2 layer fabricated according to a conventional method of magnetron sputtering without the subsequent anneal step. The magnetron sputter was performed in MRC 8667 sputter, with 1200 W of RF power, in an environment of argon with 10% oxygen and 80 sccm total gas flow, at a pressure of 20 mTorr. The resulting film thickness is 5460 Angstroms. The x-ray peak full width at half maximum ("half height width") of the peak at 19 degrees of this conventional sample is 2.61 degrees. The half height width is a measure of the crystallite size, which can be calculated from this data as according to known formulas. The crystallite size for this conventionally magnetron sputtered film is 34 Angstroms. This conventional film must be annealed at high temperature to achieve sufficient crystallite size to have adequate electrical properties such that the film is part of a functional and practical battery.

[0220] In other conventional film materials, like LiMn_2O_4 , nanocrystalline structures have been sputtered into films and prior to their anneal they have a crystallite size of about 40 Angstroms to about 50 Angstroms. Annealing this film at a temperature of about 300 degrees Celsius produces a crystallite size of about 130 Angstroms to about 160 Angstroms. In some embodiments of the present invention, these crystallite sizes are achieved at the time of deposition. Moreover, in some embodiments, superior crystallite sizes are achieved at the time of deposition.

[0221] FIG. 12B shows an X-ray diffraction spectrum for a LiCoO_2 film fabricated according to the teachings of the present disclosure. Specifically, this film was deposited using a first source and a secondary source of energized ions as discussed above with respect to samples "b" and "c" of FIG. 11. The peak for the ion assisted deposition film is significantly higher than the non-assisted spectra of FIG. 12A. This indicates a higher degree of long range order in the ion assisted deposition film. The half height width of the peak of the ion-assisted film at 19 degrees is 0.45 degrees. The crystallite size is 242 Angstroms. Accordingly, the present fabrication techniques yield an as deposited film having a crystallite size of greater than seven times that of the conventional deposition methods without post-deposition anneal. Moreover, the present fabrication techniques yield a superior crystallite size even when compared to the conventional film after it has been annealed. The present fabrication technique yields a factor of crystallite size improvement of about a 1.8 to about a 2.6 over the conventional technique. Consequently, the present fabrication method can thus achieve superior crystallite size in the film as they are deposited resulting in faster, more efficient fabrication of thin-film batteries. Such an improved crystallite structure is highly desirable in the cathode film due to the limitations imposed on energy storage in thin-film batteries due to cathode film performance.

[0222] Another aspect of the present fabrication method is the ability to fabricate thin films at essentially room temperature with a crystalline orientation that is essentially perpendicular to a boundary with adjacent films and crystallite size. Ions must travel through these boundaries to charge and discharge the battery. The boundaries include a first boundary that is between the cathode film and the electrolyte film and a second boundary that is between the electrolyte film and the anode film. The crystallite orientation is preferably perpendicular to the boundary planes. That

is, the lithium ion lattice planes are parallel to the lithium ion direction of travel during charging and discharging the thin-film battery. This orientation lowers the internal battery resistance and lowers capacity loss at high discharge rates. The crystallite size is preferably large, e.g., over 100 Å, and more preferably over 200 Å. The larger the crystallite size improves electrical properties. Crystallite size is strongly correlated to the ion diffusion coefficient, a measure of how freely lithium ions can be added to, or extracted from the intercalation material.

[0223] While the above-described embodiments focus on lithium-intercalation materials and, more specifically, LiCoO_2 , it will be recognized that the some embodiments are adaptable to other intercalation materials for producing energy-storage devices. Other types of intercalation material include LiMn_2O_4 , V_2O_5 , and carbonaceous materials, lithium, lithium alloys, oxides, and nitrides

[0224] Using the fundamental teachings herein, i.e., the in situ assist of the growing film with appropriate energy and/or species of ionized gasses, processes involving the manufacture of photovoltaic panels, supercapacitors/ultracapacitors, and fuel cells could be made more robust and efficient. A corresponding cost, fabrication efficiency, and performance advantage can be gained.

[0225] For example, Solid Oxide Fuel Cells (SOFC) require the manufacturer to deposit a ceramic material on a support structure. See U.S. Pat. No. 6,007,683, incorporated herein by reference. This ceramic is then coated with a conductive material such as platinum, which is the catalyst for the fuel cell. The cost of these materials and the efficiency with which they conduct the appropriate ions from one side of the cell to the other determines, in large measure, the cost of manufacture and operation of the fuel cell. The application of the techniques described herein to a fuel cell manufacturing process would yield substantially higher quality catalyst with higher ionic transport capability. Moreover, the present techniques further provide the ability to produce a thinner catalyst by virtue of the structural properties of materials deposited via the methods described herein. This allows lower temperature operation of the fuel cell, thus, widening product latitude.

[0226] Supercapacitor/ultracapacitor performance is also enhanced by the application of the present techniques. See e.g., U.S. Pat. No. 5,426,561, incorporated herein by reference. High energy density and high power density ultracapacitors and supercapacitors are improved by reduction in crystalline defects and improvement in the growth mechanism such that the electrolyte layer could be significantly thinned. This thinning improves the volumetric energy density of the device. The improved crystal structure enhances the voltage stability of the electrolyte.

[0227] While some of the above embodiments include an ion source for providing the focused energy to arriving adatoms at a surface of a substrate to form films having fewer defects and/or certain crystal properties, other source of the focused energy are within the scope of some embodiments of the present invention. Examples of such other sources include high intensity photo sources, lasers, short duration, high intensity (flash) heat sources, short duration plasma sources. Each of these sources provides the required energy to a film and does not harm previously deposited layers, previously connected devices, or the substrate. In

some embodiments, these sources provide the energy to the adatoms as they arrive at the surface on which the adatoms will form a film.

[0228] By way of introduction, one aspect of the invention deals with the field of batteries and, more specifically, to the use of a thin film battery for enclosures for devices and also for devices which include an integrated battery.

[0229] FIG. 13 is an exploded perspective view of an electronic device 1000 having a separate printed circuit board 1010 and a separate battery 1020. The enclosure 1000 typically includes a first portion 1001 and a second portion 1002. The first portion 1001 may also be termed as a bottom portion and typically may include pegs 1011 upon which the printed circuit card 1010 rests. The pegs 1111 are also used to position the printed circuit card 1010 with respect to the bottom portion 1001 of the enclosure 1000. In addition, there are typically several other sets of stops 1112, which are used to position the battery 1020 with respect to the bottom portion 1001 of the enclosure 1000. The second portion 1002 will include openings 1030 and 1032. The opening 1032 may be for a display such as an LCD or liquid crystal display (not shown in FIG. 13). The opening 1030 is typically for an access panel 1040, which fits within the opening 1030. The access panel 1040 provides access to the battery 1020. The printed circuit board 1010 includes a battery connector 1022, which fits over the terminals of the battery 1020. The battery connector 1022 provides an appropriate amount of current to the electrical components on the printed circuit board 1010. The second portion 1002 of the enclosure 1000 includes several plastic hooks, which are used to mate the second portion 1002 with the first portion 1001 to form the enclosure 1000. The prongs or hooks 1050 fit within corresponding slots 1052 on the first portion 1001 of the enclosure 1000.

[0230] These enclosures are typically made of plastic, and housed within the enclosure 1000 is a separate battery 1020 and a separate printed circuit board 1010. These particular types of devices have several problems. First of all, the whole housing or enclosure, or at least a portion of it, has to be removed in order to replace a battery or in order to recharge a battery. The batteries 1020 typically include a gel-type electrolyte which can be very toxic and dangerous and, for that reason, difficult to dispose. From a manufacturing standpoint, there is a need to assemble many parts, including the separate circuit board 1010 and a battery 1020 and an LCD (not shown). These also must be accurately placed within the first portion 1001 to produce a quality-looking enclosure 1000 for the entire electrical device. Each time a separate component must be placed together or into one portion or a first portion of the device requires an additional process step. In addition, mating the second portion 1002 of the enclosure 1000 with the first portion 1001 is still a further process device. From a manufacturing point, it would be advantageous if there were less process steps involved in manufacturing an electronic device such as the one shown. With less manufacturing steps, the device can be made more simply and more cost effective.

[0231] Still a further disadvantage is that the separate components, such as the separate printed circuit card 1010 and the separate battery 1020, require a lot of space in terms of the enclosure. The tendency these days is to form electronic products or electronic devices that save on space. In

most instances, a smaller electronic device is better than a larger electronic device. Therefore, there is a need for a process that can reduce the number of process steps and save on space and yet produce a reliable battery and circuit for an enclosure.

[0232] The above-described method (see FIGS. 1-12) for placing a battery onto a substrate can be used in many different ways in devices to produce a more compact and reliable electronic package having a battery which is capable of being recharged a very large number of times. The batteries and electronics could be placed directly onto an enclosure portion therefore saving space. As a result, the design of the various electronic devices could be smaller than corresponding devices that are currently used.

[0233] FIG. 14A is an exploded perspective view of a portion of an enclosure that includes both a battery 1110, which is deposited directly onto the enclosure portion 1100. The enclosure portion 1100 includes an interior surface 1101 and an exterior surface 1102. In this particular embodiment, the battery 1110 is deposited onto the interior surface 1101 of the enclosure portion 1100. It should be noted that the enclosure portion 1100 resembles the first portion 1101 or the bottom portion, as shown in FIG. 13. The interior 1101 of the enclosure portion 1100 also includes a plurality of traces 1120 for electrically coupling the battery 1110 to various electronic components 1130, 1131, which are attached to sites 1140 and 1141. The sites 1140 and 1141 include the electrical contact pads for electrically connecting the electrical components 1130 and 1131 to the sites 1140 and 1141. The pads associated with the sites 1140, 1141 are also directly deposited onto the interior surface 1101 of the enclosure portion 1100. Advantageously, the battery 1110 can be deposited onto the interior portion 1101 of the enclosure portion 1100 as well as the traces 1120 and the pads associated with the sites 1140 and 1141. Advantageously, in order to complete an electronic circuit, the only process steps that need to be accomplished are to add the electronic components 1130 and 1131. In some instances it also may be possible to produce some of the electronic components during the manufacturing steps required to place the thin film battery 1110 onto the interior portion 1101 of the enclosure 1100. Optionally, a protective layer 1150 may be placed over the battery 1110 or other select portions deposited on the interior surface 1101 of the device enclosure 1100. The optional protective layer is shown in phantom and is referenced by reference numeral 1150.

[0234] FIG. 14B is an exploded perspective view of a portion 1100 of an enclosure for an electronic device according to another embodiment of this invention. The enclosure portion 1100 includes an interior and an exterior surface 1102. In this particular embodiment, the battery 1110 is deposited on the exterior surface 1102 of the enclosure portion 1100. The battery 1110 includes a post 1160 for the cathode and another post 1162 for the anode. The posts 1160 and 1162 terminate or attach to through holes 1161 and 1163. The through holes 1161 and 1163 provide electrical communication to various components located inside the enclosure portion 1100. In essence, the chief difference between the embodiment shown in FIG. 14A and the embodiment shown in this FIG. 14B is that the battery portion 1110 is deposited on the exterior surface 1102 of the enclosure portion 1100. A protective coating 1150 may be placed over the battery portion 1110 and, more specifically, over the

battery portion 1110 and the electrical posts 1160 and 1162 and the through holes 1161 and 1163. The protective layer 1150 may be translucent or may be colored to match the exterior surface 1102 of the enclosure portion 1100.

[0235] FIG. 14C is an exploded perspective view of a portion of enclosure 1103 for an electronic device according to yet another embodiment of this invention. The enclosure portion 1103 includes a battery 1110 that is deposited on the interior surface of the enclosure portion 1103. The enclosure portion 1103 includes an interior portion 1101 and an exterior portion 1102. The enclosure portion 1103 corresponds to a top portion including a display that can be viewed by the consumer during use. The battery 1110 is deposited on the interior surface 1101 of the enclosure portion 1103. Also included are traces 1120 as well as electronic components 1130 and 1131. Completing the circuit is an LCD or liquid crystal display 1170. The LCD is positioned near or at an opening in the enclosure device 1103 so that the readable portion of the LCD 1170 can be viewed from the exterior surface 1102 of the enclosure portion 1103. Enclosure portion 1103 roughly corresponds to the second enclosure portion 1102 or on top of the electronic device shown in FIG. 13.

[0236] An addition to depositing a device or a battery device or energy device 1110 onto the surface of an enclosure, another embodiment of this invention is to produce a sheet including multiple cells or batteries 1110. The batteries 1110 are formed on a sheet of flexible or plastic material 1300. It should be noted that the size of the cells 1110 and the placement of the cells or individual batteries 1110 can be varied for producing various different sizes and styles of formed batteries.

[0237] FIGS. 15A through 15E disclose a method whereby the battery is formed into a conformed or conformable sheet having roughly the same shape as either the interior or exterior surface of an electronic device. The conformed sheet can then be placed or adhered directly to the interior surface or exterior surface of an electronic device. The sheet is produced with a number or plurality of cells 1110, as will be discussed later in this application. Once the sheet is formed as described later in this application, the sheet 1300 is diced into individual cells or individual battery portions 1310. In other words, a battery 1110 will be formed on a dice sheet 1310 from the main sheet 1300. The individually diced battery portion 1310 can then be formed into a variety of shapes, as shown by FIGS. 15C, 15D and 15E. These shapes can be any desired shapes. In some embodiments or in most embodiments, the shape of the sheet will conform or will be able to be placed on the interior or exterior surface of an electronic device. FIG. 15C, for example, shows a roughly square battery that has folded up sides or vacuum formed sides 1320. This particular device could be placed on the interior surface of an electronic device such as a garage door opener or any other like device.

[0238] FIG. 15D shows a more rectangular portion or diced sheet which resulted from a more rectangular battery laid down upon a sheet and diced into an individual battery portion 1310. This more rectangular formation may be glued or adhered to the inner surface of an electronic enclosure for a personal data assistant. In the alternative, the form shown in FIG. 15D may also be suitable for placement on the exterior surface of an electronic device, such as a portable data assistant.

[0239] FIG. 15E shows a more formed device that might be found on a cell phone or similar device. FIG. 15E may be formed to fit on the interior surface of a cell phone or the exterior surface of a cell phone or calculator. In other words, a diced sheet 1310 is used as a starting point for vacuum forming or for otherwise forming a battery that can be attached either to the interior or exterior surface of an electronic device. An electronic device to which it is attached can be anything including hearing aides, calculators, personal data assistants, smart cards or other credit card, watches, laser pens, power tools, surgical devices or even catheters. The list above is not exhaustive but is merely set forth as examples of the type of the devices that may include a battery shown and formed in FIGS. 15A through 15E.

[0240] In some instances, it may be advantageous to include a battery having multiple cells 1110, 1110' and 1110". In this particular instance, a dice is made 1320 that includes cells 1110, 1110' and 1110". The sheet can also be formed with fold lines 1321 and 1322, as shown in FIG. 15G.

[0241] FIG. 15H shows that the batteries have been folded along the fold lines to form a stack of three batteries 1100, 1110, 1110' and 1110". The folds shown in FIG. 15H are a fan fold. Once the fan fold is formed, as shown in FIG. 15H, the fan folded battery, including three cells 1330, can be formed in any desired shape, such as those shown in FIG. 15C, 15D and 15E. The three-celled or multi-celled unit 1330 can be adhered to the interior or exterior surface of any electronic device, as discussed above. It should be noted that the fan fold can include more than three batteries or less than three batteries. The inventive aspect is that it includes a plurality of batteries. The cells 1110, 1110' and 1110" can be attached to one another so that the cells are in series after they are diced. Another possibility is that the electrical contacts for each of these could be put in contact with one another as a result of fan folding the multi-celled unit 1330.

[0242] FIGS. 15I, 15J and 15K show yet another embodiment of the invention. In this particular embodiment of the invention, the sheet of electrical cells 1300 includes a plurality of cells including 1110 and 1110'. The entire sheet 1300 is then vacuum formed to form more or less an egg carton 1350 with individual battery cells 1110 and 1110' being formed within well 1360 and 1362 in the sheet 1300. Between the wells 1360 and 1362 is a living hinge 1370. The batteries 1110 and 1110' are at the bottom of each well 1160 and 1162, as shown in FIG. 15K. The living hinge 1370 is positioned between the two wells 1360 and 1362. The first cell 1360 can be folded on top of the second well 1362 to form an electronic device enclosure 1380, as shown in FIG. 15L. It should be noted that the size of the battery portions 1110 and 1110' can be limited or placed so that other traces and room for other electronic devices can be added so that a total circuit can be formed within a disc enclosure. This provides for an advantage that wherein the electronic component could be directly placed into the wells 1160 and 1162 at sites formed at the same time as the batteries were deposited onto the sheet 1300. After placing all the various electronics, the electronic device can be formed merely by dicing two of the wells 1360 and 1362 so that they form a top and bottom of the device enclosure 1380. All sorts of electronic devices could be included, including an LCD or other display device. The LCD may be readable directly

through a sheet if it is transparent or the sheet, or one of the wells 1360 and 1362, may be provided with an opening that would correspond to an opening or face of the display of an LCD or other display device. Thus, the sheet and the deposited battery thereon can ultimately become the exterior surface or the enclosure for the device formed on the sheet. This has a great advantage in that the process steps necessary to form a device are or can be quite easily and efficiently done in a continuous process. This would lead to very efficient manufacturing of electronic devices.

[0243] FIG. 16A is a plan view of a sheet including a plurality of cells 1110 according to this invention. FIG. 16A, 16B and 16C show a way to form a laminated battery cell and possibly laminated battery cell and electronics for a smart card or other invention that includes a battery and electronics within a card. The sheet 1300 shown in FIG. 16A includes cells 1110. The sheet also includes fold lines 1390 and 1392. The sheet 1300 is diced into individual sections, which include fold lines 1390 and 1392, as well as a battery cell site 1110. The battery cell site might also include electronics that are also deposited with the battery or energy source onto the sheet 1300. The diced portion 1400 includes one portion including the cell 1100 and two blank portions 1402 and 1403. The diced portion 1400 is then fan folded, as shown in FIG. 16C. Once a fan fold has been formed, the cell portion 1110 is captured between the two unpopulated sheet portions 1402 and 1403 and will provide an extra protective layer. The excess portions of the sheet 1300 can be trimmed, as shown in FIG. 16D to produce a smart card or card including both a battery 1110 and electronic, as shown in FIG. 16E.

[0244] FIG. 17 is an exploded perspective view of a diced portion of a sheet 1300 which includes one battery cell 1110 rolled around an electrical motor 1500. In this case, the diced portion 1300, which includes a cell 1110, is an elongated strip 1510 from the original sheet 1300. The elongated strip 1510 may include several batteries placed in series or one elongated battery that is laid down as a strip on the sheet 1300. The electrical motor is electrically connected to the anode and cathode of the battery and then rolled on to the electrical motor 1500. In this case, the strip 1510, on which the battery has been deposited, becomes the case for the electrical motor or also can be viewed as being a part of the case of the electrical motor. The electrical motor can be provided with a sprocket 1520 that is used to drive another gear 1530 having a shaft 1532 attached thereto. As shown in FIG. 17, a chuck 1540 is placed upon the shaft 1532 to form a drill or other power tool. Advantageously, the power tool could be light and compact, as well as being capable of being recharged a multiplicity of times. The power tool could be a hand-held drill for homeowner use or a smaller device, such as a Dremel-brand rotary hand tool.

[0245] FIGS. 18A, 18B, 18C and 18D show several other embodiments of an LED light device in which the diced portion of a sheet 1300 becomes the outside case for the penlight or light device.

[0246] FIG. 18A is a planned view of a diced battery cell 1600 which includes a battery or energy device 1110 and a switch 1602 and an LED 1604. The switch 1602, battery 1110 and the LED 1604 form a flashlight or LED lighting device. The sheet, including the diced battery cell and LED, is rolled across its shorter distance starting at the end

including the LED 1604. The LED is merely rolled into the battery and the battery is formed around the first roll to form a spiral, as shown in FIG. 18B.

[0247] FIG. 18B is a perspective view of the diced battery cell and LED after it has been formed into a lighting device in which the sheet 1600 in which the battery is deposited becomes an outer case. The LED can be activated by enabling the switch 1604. By enabling the switch 1604, the LED can be turned on. The sheet 1600 acts as an outer case of the lighting device formed 1620.

[0248] FIGS. 18C and 18D show another embodiment of the invention for a lighting device. In this particular embodiment, again a strip 1600 is provided with a switch 1602 and an LED 1604. In this particular embodiment, the LED is positioned so that it extends beyond the length of the sheet 1600. In this particular embodiment, the sheet 1600 is rolled along its longer dimension around the LED 1604 to form an elongated case having the LED 1604 at one end of the case and a switch 1602 at the other end of the case. This forms a light emitting diode light 1630 in which the dice sheet 1600 is part of the case.

[0249] FIGS. 19A, 19B and 19C, in some instances, is necessary to keep the battery portions 1110 and 1110' of a power source or energy source flat and not curved when it is formed.

[0250] FIG. 19A shows a sheet 1300 which includes a plurality of individual cells such as 1110 and 1110' which are an elongated strips and include fold lines, such as 1710. FIG. 19B is a plan view of a diced strip 1700 including a plurality of battery cells 1110, 1110' and 1110". It should be noted that the battery cells 1110", which are located near one end of the strip 1700, are smaller than the battery cells formed at the other end of the strip 1700. For example, battery cell 1110" has a very thin width while the battery cell 1110 is roughly more rectangularly shaped. The strip 1700 is folded successively along fold lines 1710 to form a box of cells, as shown in FIG. 19C. The smaller cells 1110" are in the inside or inner core of the box while the larger cells 1110 form the outer sides of the box. Each of the cells 1110, 1110' and 1110" and the cells in between those particular cells are placed in series with one another. The end result is a cubically formed battery cell 1720, as shown in FIG. 19C.

[0251] FIG. 20 is a cutaway side view of an enclosure portion that includes a sheet having a plurality of battery cells. It should be noted that we have discussed thus far that a sheet of battery cells, such as the one shown in FIG. 15H, can either be placed on the outside surface of an enclosure or on the inside surface of an enclosure or it can be formed or deposited upon an inside or outside surface of the enclosure. FIG. 20 shows that an enclosure portion 1800 having an interior surface 1801 and an exterior surface 1802 can be injection molded around a battery formed on a sheet. The battery could be a single battery, as is shown in FIG. 15B, or it could be a multi-celled battery, as shown in FIG. 15H. In other words, a sheet 1820 including one or more, or at least one battery cell 1110 formed by the above methods, could be held within a mold and a suitable plastic could be injection molded about or around the battery cell 1820. The mold could also include pins that electrically connect the battery 1820 to the interior surface 1801 of the enclosure portion 1800. The pins are shown by reference numerals 1821 and 1822.

[0252] FIG. 21A is a flow chart that depicts a process for recycling device enclosure portions or for recycling batteries 1110 or battery cells 1110. Because the battery cell 1110 and batteries made from a number of these battery cells 1110 can be recharged many, many times, it is contemplated that any electronics associated with this circuit may become obsolete over time and, therefore, a method of recycling the batteries is also part of this invention.

[0253] The first step, depicted by reference numeral 1900, is to determine if the electronics within a circuit are obsolete. Electronics are typically obsolete due to technology advances in the electronics, which may occur over a number of years. If the electronics are obsolete, then the battery 1110 or series of cells 1110 may be removed from a device cover or enclosure portion, as depicted by reference numeral 1910. The next step is to replace the old electrical components with new electrical components, as depicted by reference numeral 1920. This first process is useful for enclosure portions where the battery or number of cells 1110 cannot be easily removed from the enclosure portion.

[0254] A second process is shown in FIG. 21B. The second process shown in FIG. 19B is useful for devices in which the battery 1110 may be removed easily from the enclosure portion. As before, the first step, depicted by reference numeral 1930, is to determine if the electronics are obsolete. If they are, the battery 1110 is merely removed from the case for the enclosure portion and recycled for use in another enclosure portion having a similar contour, as depicted by reference numeral 1950.

[0255] In some embodiments, multiple cells are stacked in the original device, the manufacture method would include connection tabs that are coupled together to form the appropriate cell capacity and voltage for some particular electronic device. Upon reaching the end of the device's life, such battery stacks could have the tabs clipped or otherwise disconnected from each other so that the battery stack could be disassembled and re-assembled in a different capacity/voltage configuration.

[0256] Design and fabrication of solid-state power sources cofabricated with solid-state integrated circuitry

[0257] FIG. 22A shows a schematic circuit of an embodiment of a device 2200 having an integrated battery 2320 and circuit 2330 sharing a common terminal 2318. In other embodiments, more than one terminal is common between battery 2320 and circuit 2330, for example, when battery 2320 includes a stack having plurality of series-connected cells, and circuit 2330 connects to two or more different taps in the cell stack (e.g., if each cell of a two-cell stack provided an open-circuit potential of 3.6 volts, circuit 2330 could connect to the top of the cell stack for a portion of its circuitry needing 7.2 volts, and also to a center tap of the cell stack for a portion of its circuitry needing 3.6 volts, or a split voltage battery supply could be wired to provide a ground connection at the center tap and plus and minus 3.6 volts at the top and bottom of the stack). Common terminal 2318 connects battery 2320 to circuit 2330, and optionally can be brought out as a connection to other components. In some embodiments, common terminal connects the cathode of battery 2320 to circuit 2330; in other embodiments, terminal 2318 connects the anode of battery 2320 to circuit 2320 as shown in FIG. 22A. In some embodiments, circuit 2330 includes one or more conductors 2317 that are used to

connect to other components and/or to the other connections to battery 2320. In some embodiments, battery 2320 includes one or more conductors 2319 that are used to connect to other components and/or to the other connections to circuit 2330. In other embodiments, terminal 2317 of circuit 2330 is connected directly to terminal 2319 of battery 2320 to form a complete device, and no connection is made to other external devices using terminals 2317, 2318, or 2319. Note that circuit 2330 can include any type of circuitry, for example, as shown in FIGS. 23-26, wiring traces 2332-2337, one or more active or passive devices such as integrated circuit 2340, switches, light sources, LCD displays, photovoltaic cells, etc.

[0258] FIG. 22B shows a block diagram perspective view of an integrated device 2201 implementing circuit 2200 of FIG. 22A having the circuit 2330 built on the battery 2320. According to the present invention, in some embodiments such as shown in FIG. 22B, battery 2320 is deposited or fabricated first (for example, onto a polymer substrate), and later circuit 2330 is deposited or fabricated to a surface of battery 2320. In some embodiments as shown in FIG. 22B, a top surface of the device implementing circuit 2330 includes one or more conductors 2317 that are used to connect to other components and/or to the other connections to battery 2320. In some embodiments, a bottom surface of battery 2320 includes one or more conductors 2319 that are used to connect to other components and/or to the other connections to circuit 2330. In some embodiments, a top surface of battery 2320 (the surface fabricated adjacently to circuit 2330) is partially exposed and includes one or more conductors 2318 that are used to connect to other components and/or to the other connections to circuit 2330. FIG. 23 and FIG. 24A show some examples of devices 2300 and 2400 that are exemplary embodiments of device 2201 of FIG. 22B.

[0259] FIG. 22C shows a block diagram perspective view of an integrated device 2202 implementing circuit 2200 of FIG. 22A having the battery 2320 built on the circuit 2330. According to the present invention, in some embodiments such as shown in FIG. 22C, circuit 2330 is deposited or fabricated first (for example, an integrated circuit chip built onto a silicon substrate), and later battery 2320 is deposited or fabricated to a surface of battery 2320. In some embodiments as shown in FIG. 22B, a top surface of the device implementing circuit 2330 is left partially exposed and includes one or more conductors 2317 that are used to connect to other components and/or to the other connections to battery 2320. In some embodiments, a top surface of battery 2320 includes one or more conductors 2319 that are used to connect to other components and/or to the other connections to circuit 2330. In some embodiments, a top surface of circuit 2330 (the surface fabricated adjacently to circuit 2330) is partially exposed and includes one or more conductors 2318 that are used to connect to other components and/or to the other connections to circuit 2330. FIG. 25A and FIG. 26A show some examples of devices 2500 and 2600 that are exemplary embodiments of device 2202 of FIG. 22C.

[0260] FIG. 22D shows a schematic circuit 2205 of an embodiment of an integrated battery 2320 and circuit 2330 each having separate, electrically isolated terminals. Such embodiments are substantially identical to the embodiments of FIGS. 22A, 22B, and 22C, except that an insulator

between terminal 2318 of the battery 2320 and terminal 2316 of the circuit 2330 keeps these electrically separate.

[0261] FIG. 22E shows a block diagram perspective view of an integrated device 2206 implementing circuit 2205 of FIG. 22D having the circuit built on the battery. Such embodiments are substantially identical to the embodiments of FIG. 22B except that an insulator 2331 is deposited on battery 2320 before the rest of circuit 2330 is deposited or fabricated. In some embodiments, a portion of the top surface of battery 2320 is left partially exposed and includes one or more conductors 2318 that are used to connect to other components and/or to the other connections to circuit 2330. In some embodiments, a portion of the top surface of insulator layer 2331 is coated with a conductor and is left partially exposed and includes one or more conductors 2316 from circuit 2330 that are used to connect to other components and/or to the other connections to battery 2320.

[0262] FIG. 22F shows a block diagram perspective view of an integrated device 2207 implementing circuit 2205 of FIG. 22D having the battery 2320 built on but insulated from the circuit 2330. Such embodiments are substantially identical to the embodiments of FIG. 22C except that an insulator 2331 is deposited on circuit 2330 before the rest of battery 2320 is deposited or fabricated. In some embodiments, a portion of the top surface of circuit 2330 is left partially exposed and includes two or more conductors 2316 and 2317 that are used to connect to other components and/or to the other connections to battery 2320. In some embodiments, a portion of the top surface of insulator layer 2331 is coated with a conductor and is left partially exposed and includes one or more conductors 2318 from battery 2320 that are used to connect to other components and/or to the other connections to circuit 2330.

[0263] FIG. 22G shows a block diagram perspective view of an integrated device 2207 implementing circuit 2200 of FIG. 22A having the battery 2320 and the circuit 2330 built side-by-side on a substrate 2310. In some embodiments, a pattern of conductive areas or traces is deposited on substrate 2310, and the successive layer(s) of battery 2320 and circuit 2330 are then deposited. In some embodiments, circuit 2330 consists only of these conductive traces. In other embodiments, one or more of the process steps or deposited layers of battery 2320 and circuit 2330 are common, and thus performed at substantially the same time for both circuit 2330 and battery 2320, thus increasing the reliability, speed and yield of fabrication and lowering the cost of fabrication. In the embodiment shown, trace 2318 is deposited on substrate 2310 and forms a common bottom electrical connection for both circuit 2330 and battery 2320. Other aspects of FIG. 22G can be understood by reference to FIGS. 22A-22C.

[0264] FIG. 22H shows a block diagram perspective view of an integrated device 2208 implementing circuit 2205 of FIG. 22D having the battery 2320 and the circuit 2330 built side-by-side on a substrate 2310. This embodiment is substantially identical to that of FIG. 22G, except that separate traces are provided for signals 2316 and 2318.

[0265] FIG. 23 shows a perspective view of an embodiment 2300 of the present invention having a battery 2320 overlaid with circuitry. In some embodiments, substrate 2310 is a conductor such as a thin sheet of metal, and is overlaid with an insulator layer 2312, and then the bottom

conductor layer **2322** of battery **2320**. In other embodiments, insulator layer **2312** and bottom conductor layer **2322** are omitted, and a conductive substrate **2310** itself forms the bottom conductive layer for battery **2320**. In some embodiments, battery **2320** is a thin-film battery deposited by a process, and having a structure, as described in **FIGS. 1B to 8** herein. In the embodiment shown, battery **2320** includes a bottom conductive layer/electrical contact **2322** and a top conductive layer/electrical contact **2324**, and is covered by a protective/electrically insulating layer **2331** having one or more openings or vias for electrical connections, for example, a via through which pad/trace **2332** connects to battery **2320**. In some embodiments, the top conductor **2324** of battery **2320** is the anode connection. In the embodiment shown, the connection to the lower conductive layer/electrical contact **2322** from pad/trace **2334** is a conductive trace deposited over the side of battery **2320** to extended contact area **2333**. In some embodiments, additional connection pads/traces **2335**, **2336**, and **2337** are deposited, for example, using a shadow mask that defines where the traces will go, and a metal-evaporation source, PVD source, CVD source, sputter source or other source to supply the conductor being laid down. In other embodiments, a conductive layer for circuit **2330** is deposited over an entire upper surface, and the unneeded portions are removed, for example, using photolithography and etching techniques. In some embodiments, multiple layers are successively deposited, wherein these layers include conductors, insulators, semiconductors (e.g., polysilicon or polymer semiconductors), electrolytes, passivation layers, mechanical-protection layers, sealants, reactants (such as sensor materials that react with, e.g., smoke, carbon dioxide, antibodies, DNA, etc.) and/or decorative pattern, topography, design or color layers.

[0266] Some embodiments further include a separately fabricated circuit **2340** that is bonded (e.g., by adhesive or solder) to the rest of the deposited circuitry **2330**, for example, a flip-chip integrated circuit **2340** having bump, ball or ball-grid array connections **2341** as shown in **FIG. 23**. In other embodiments, packaged chips are used, e.g., J-leaded, gull-wing leaded, in-line-pin, or other plastic- or ceramic-encapsulated chip packages.

[0267] **FIG. 24A** shows a perspective view of an embodiment **2400** of the present invention having a battery **2320** overlaid with an integrated device **2430**. In some embodiments, integrated device **2340** is a so-called supercapacitor relying on either charge accumulation on opposing sides on an insulator (as in a capacitor) or ion transport across an electrolyte (as in a battery), or both charge accumulation and ion transport to store electrical energy. In some embodiments, integrated device **2340** includes a photovoltaic cell of conventional construction deposited directly on battery **2320**.

[0268] Some embodiments further include a separately fabricated circuit device such as an integrated circuit chip **2440** that is wire-lead bonded to device **2430** using wire **2441**, to device-battery common terminal **2324** using wire **2443**, and to bottom battery contact **2322** using wire **2442**. For example, in one embodiment having a supercapacitor device **2430**, integrated circuit **2430** includes a wireless communication circuit that uses the battery for overall power needs and uses supercapacitor device **2430** for quick-burst power needs such as for transmitting short burst of data

to an antenna. Other embodiments include other fabricated circuit devices such as switches, LEDs or other light sources, LCD displays, antennas, sensors, capacitors, resistors, etc., wired to device **2400**.

[0269] In one embodiment, battery **2320** includes a bottom conductor layer of platinum (e.g., 0.5 micrometers thick), a cathode of lithium cobalt oxide covered by a LiPON electrolyte and a carbon anode, and a top electrode of platinum. On top of these depositions, device **2430** includes a layer of ruthenium oxide, an electrolyte of LiPON, another layer of ruthenium oxide and a top layer of platinum. Such a device **2430** would store energy by transporting lithium ions derived from the LiPON electrolyte from one to another of the top and bottom surface of the electrolyte, as well as perhaps moving charge (electrons) to an opposing surface. Such a device exhibits a higher-current discharge rate than a comparable battery, and a higher energy storage than a comparable capacitor. The present invention including ion-assist deposition provides for higher quality cathode films (better crystal orientation) and better electrolyte films (more complete isolation and fewer pinhole defects for any given thickness, thus allowing thinner electrolyte films that increase ion transport rates), and better capacitor dielectric films (more complete isolation and fewer pinhole defects for any given thickness, thus allowing thinner dielectric films that increase dielectric isolation, capacitance, and charge storage). In some embodiments, a capacitor insulator layer is made of a barium strontium titanate.

[0270] In some embodiments, a cathode layer of lithium-cobalt-oxide is covered by a LiPON electrolyte layer and a lithium(0.5)-cobalt-oxide anode layer. This anode layer is non-stoichiometric deposited using a source that has excess cobalt and oxygen relative to lithium as compared to that used for the cathode, and various embodiments use different lithium ratios.

[0271] Design and fabrication of solid-state power sources fabricated as a laminate on a rigid or flexible direct energy conversion material such as photovoltaic

[0272] Virtually all electronics require energy to operate and perform the designed functions. This energy typically comes from either an AC source such as a home wall electrical outlet or a battery mounted in the packaging of the electronic device. More recently, advances in the conversion of heat and light into energy have fueled research in the area of direct energy conversion (e.g., by photovoltaic cells). This has the potential to supply a large percentage of the world energy needs in a clean and safe manner. One problem with these methods of energy supply has been the cyclical nature of the energy being converted. Whether heat or light, the source usually goes away for a 6- to 12-hour period resulting in zero output from the unit. One way around this problem is to supply a battery with the unit to supply power during periods of low light or heat input. This is however not an ideal solution as today's rechargeable batteries are bulky and failure prone after several charge/discharge cycles. The present invention solves this problem by integrating its solid-state Lithium battery directly on the energy conversion substrate. The present battery has a distinct advantage over current technologies, in that it is not prone to failure or memory problems over tens of thousands of charge/discharge cycles, has very high capacity, is lightweight, can be fabricated on nearly any substrate and is cheap to manufac-

ture. The resultant product is a reliable, portable power source with steady output over extended periods or rain or shine, night or day, warm or cold.

[0273] According to the present invention, solid-state processes are used to cofabricate direct energy conversion materials and energy storage on the same substrate. This is possible by using the low-temperature processes for solid-state batteries described above.

[0274] FIG. 24B shows a block diagram of a battery-layer-deposition system 2460. In some embodiments, system 2460 includes a supply reel 2461, a deposition chamber 2462 that deposits one or more layers of battery 2320 onto a substrate 2410 as described above, and a takeup reel 2463. Typically, deposition chamber 2462 is a vacuum chamber that encloses supply reel 2461 and takeup reel 2463, and successively deposits a plurality of layers, wherein each of one or more of the layers is immediately treated (e.g., by ion assist, laser surface anneal, heat surface anneal, or kinetic treatment), according to the present invention, to impart a high-quality surface structure to that layer or those layers before subsequent layers are deposited, and without substantial heating of the underlying layer(s) or substrate. For layers that need to be thicker, a longer deposition station is provided than the station for thinner layers. In some embodiments, the lower contact layer 2322 is deposited onto a starting substrate film, fabric, or foil 2410, then the cathode, electrolyte, anode, and anode-contact layers are deposited, wherein the cathode layer and/or the electrolyte layer are treated (e.g., by an ion-assist beam) before subsequent layers are deposited.

[0275] FIG. 24C shows the resulting item 2464, which is a continuous sheet of substrate material 2410 having batteries 2320 deposited on it. This partially built item 2464 is then used as the supply reel 2466 of layer-deposition system 2465 of FIG. 24D.

[0276] FIG. 24D shows a block diagram of a energy-conversion-layer-deposition system 2465. In some embodiments, system 2465 deposits layers that form a photovoltaic cell 2430 onto battery 2320 of FIG. 24A. In some embodiments, system 2460 and system 2465 are merged into a single system having a single supply reel 2461 and a single takeup reel 2468, and having layers of the battery 2320 and of the photovoltaic cell 2430 successively deposited. In other embodiments, other types of devices 2430 are deposited such as capacitors, antennae, circuitry, transducers, sensors, magneto-resistors (e.g., of the giant magneto-resistor type), etc.

[0277] FIG. 24E shows a perspective view of a processed sheet 2469 that is the result of processing by system 2460 and system 2465. Sheet 2469 is then cut or diced into individual devices 2400. FIG. 24F shows a perspective view of three diced final devices 2400. In other embodiments, sheet 2469 is cut into any desired number of devices 2400.

[0278] In other embodiments, system 2460 and system 2465 deposit a battery 2320 and a photovoltaic cell 2330 side-by-side on one face of substrate 2310, such as shown in FIG. 22G and FIG. 22H. In some such embodiments, one or more of the layers deposited for battery 2320 are also deposited for photovoltaic cell 2330 simultaneously of the same deposition material, thus saving process steps but making a wider device than if stacked as in FIG. 24A.

[0279] FIG. 25A shows a perspective view of an embodiment 2500 of the present invention having an integrated circuit 2510 overlaid with a battery 2320. In some embodiments, integrated circuit 2510 includes a top insulator layer 2511 having a plurality of vias or openings 2512 to the active surface of the integrated circuit 2510 (the side with devices and connectors). Two of these vias are used as contacts 2514 and 2515 between integrated circuit 2510 and battery 2320. Battery 2320 is deposited as described for FIG. 23. In some embodiments, battery 2320 is deposited on an integrated circuit wafer before integrated circuit 2510 is diced apart from the other integrated circuits. In some embodiments, battery 2320 is deposited onto integrated circuit 2510 after integrated circuit 2510 is diced apart from the other integrated circuits. Some embodiments further include a passivation layer over the top and sides of battery 2320 such as layer 2331 of FIG. 23.

[0280] In other embodiments, a circuit such as circuit 2330 of FIG. 23 is used in place of integrated circuit 2510 of FIG. 25A. Thus, a pattern of vias and/or other devices or circuitry is deposited on a substrate, and battery 2320 is deposited on the top of the predefined circuitry/substrate, as in FIG. 25A. In some embodiments, a photovoltaic cell is used as such a circuit device/substrate, and battery 2320 is deposited directly on the premanufactured photovoltaic cell. In some embodiments, an integrated circuit such as 2440 of FIG. 24A is wired to the battery 2320 and the premanufactured photovoltaic cell to control charging of the battery from the cell and/or to control using power for other devices (such as a light source or hearing aid) from the photocell during periods of high amounts of light and power available from the photovoltaic cell, and using power from the battery during periods of little or no light and power available from the photovoltaic cell.

[0281] Virtually all electronics require energy to operate and perform the designed functions. This energy typically comes from either an AC source such as a home wall electrical outlet or a battery mounted in the packaging of the electronic device. Until the last few years, this approach has proved to be acceptable even though the inefficiencies caused waste of both energy and natural resources in that the device housing had to be made large enough to incorporate the energy package or conversion electronics. As electronic complexity increases, the wasted real estate and energy begin to become an issue as the demands of operator interface begin to compete with the energy source for area on the device. The application of the solid-state battery process of the present invention allows the cofabricating of electronics and the associated power source together on chip.

[0282] Solid-state processes are used to cofabricate electronics and solid-state rechargeable battery on a common substrate such as silicon used for IC processing. This is possible by using the low-temperature processes for solid-state batteries described above.

[0283] Referring to FIG. 25A, in some embodiments, the integrated circuit (IC) 2510 in wafer form is processed normally through final passivation including bond-pad etch. All thermal processing necessary for the electronics is performed conventionally. The IC in wafer form is sent to backend energy processing. In some embodiments, the design of the IC includes electronics for control of recharge

for the solid-state energy source; contact vias for connecting the cathode plate and anode plate to the circuit. Using shadow masks with sufficient overlay accuracy, the necessary components of the energy structure **2320** are deposited using PVD or CVD as described above. A final passivation coating (such as **2331** of **FIG. 23**) is applied to the energy stack. The IC in wafer form with energy source integrated is sent for test, dicing and packaging. This provides integration of electronics and solid-state rechargeable batteries by cofabrication.

[**0284**] Design and fabrication of solid-state power sources fabricated as a laminate on the packaging for the device the energy source will power

[**0285**] Solid-state processes are used to cofabricate electronics and packaging. This is possible by using the low-temperature processes for solid-state batteries described above.

[**0286**] **FIGS. 25B-25E** show a fabrication sequence for cofabrication of solid-state integrated circuits and solid-state energy source such as that described above, but onto a packaged IC **2540**. **FIG. 25B** shows a plan view and **FIG. 25C** shows an elevational view of IC **2540**. In some embodiments, IC **2540** includes a silicon chip **2545** having integrated components such as transistors, resistors, memory, etc., a lower substrate **2540**, and a wiring superstrate **2544** having deposited wires **2540** that extend to bonding vias **2542**. **FIG. 25D** shows a plan view and **FIG. 25E** shows an elevational view of an integrated battery-IC **2501**. Battery-IC **2501** includes a cathode **2326** (e.g., lithium cobalt oxide), electrolyte layer **2327** (e.g., LiPON), and anode layer **2328** (e.g., including copper, carbon, lithium, lithium-magnesium, and/or other suitable anode material). Passivation overcoat layer **2329** suitable to protect the inner components of battery **2320** is then deposited or grown.

[**0287**] In one embodiment, the product package **2540** is formed by conventional means. All machine work and cleaning is accomplished. The package **2540** is sent to energy processing for deposition of battery **2320** or other energy-storage device. The design of the package included a suitable area **2549** for deposition of battery components. Using shadow masks with sufficient overlay accuracy, the necessary components of the energy structure (e.g., a battery and/or photovoltaic cell) are deposited using the methods described above. A final passivation coating **2329** is applied to the energy stack structure. The package with energy structure integrated is sent for assembly.

[**0288**] In one embodiment, further electronics are attached to the package/energy entity **2501** by way of adhesive. The electronics are then hardwired to the package/energy entity. In a second embodiment, the electronics are mounted directly to the package/energy entity by **2501** way of solder bumps. In some embodiments, the entire assembly is optionally potted, then sealed by the package cover. In other embodiments, the battery is formed on a substrate suitable as a packaging material. The substrate is formed into individual package form factors. The package with energy structure integrated is sent for assembly.

[**0289**] Thus, the present invention provides integrated product packaging and solid-state rechargeable batteries by cofabrication where the battery is deposited on the already-formed package. The present invention also provides inte-

grated product packaging and solid-state rechargeable batteries by cofabrication where the battery is deposited on a suitable package material, then formed into the package.

[**0290**] The present invention also provides a method of attaching electronics to a package/energy hybrid wherein the electronics are mounted with adhesive, then hardwired to the energy source. The present invention further provides a method of attaching electronics to a package/energy hybrid wherein the electronics are attached to the energy source via solder bumps.

[**0291**] **FIG. 25F** shows a block diagram of a layer-deposition system **2560** much the same as that of **FIG. 24B**, however rather than using a sheet of polymer or other homogenous substrate material **2410**, system **2560** starts with a sheet **2561** having a plurality of processed circuits **2540**.

[**0292**] **FIG. 25G** shows a perspective view of a processed sheet **2569**. Sheet **2569** includes a plurality of preprocessed circuits **2540** each having a battery **2320** deposited on it by system **2560**. Sheet **2569** is then cut or diced into individual devices **2501**.

[**0293**] **FIG. 26A** shows a perspective view of an device **2600** of the present invention having an integrated circuit **2510** overlaid on its back with a battery **2320**. This embodiment is similar to that of **FIG. 25A**, except that the battery **2320** is deposited on the back of IC **2510**, and is wire-lead bonded to contact **2514** using wire **2614** and to contact **2515** using wire **2615**.

[**0294**] In some embodiments, device **2600** further includes device **2650** such as a photovoltaic cell fabricated on a surface of integrated circuit **2510**, for example, on the opposite side as that facing battery **2320**. In some embodiments, such a photovoltaic cell **2650** provides power to IC **2510** for both operation of IC **2510** and for charging of battery **2320** during periods of relatively bright light, and then battery **2320** provides power to IC **2510** for operation during periods of relatively dim or no light. In some embodiments, device **2600** includes one or more devices **2650** such as sound transducers for such applications as a hearing aid having an combined transducer-battery-amplifier device. In some such embodiments, both a photovoltaic cell **2650** and one or more sound transducers **2650** are deposited in order to provide a light-rechargeable hearing aid which could be taken out of the ear at night and placed in a light-emitting recharging stand (e.g., that of **FIG. 27L**), avoiding the need to replace batteries or even to electrically connect to an external recharging circuit. In some embodiments, a photovoltaic cell and/or a sound transducer is/are deposited on one face of device **2600** for recharging and for sound pickup, and a sound transducer is deposited on an opposing face for use as a speaker for applications such as a hearing aid.

[**0295**] In yet other embodiments, **2600** further includes device **2650** such as a magnetoresistive sensor fabricated on a surface of integrated circuit **2510**, for example, on the opposite side as that facing battery **2320**. Such a device **2600** could be used in a compass, for example.

[**0296**] In some embodiments, embodiment **2600** further includes an antenna or electromagnetic radiation receiving loop **2660** fabricated on a surface of integrated circuit **2510**, for example, on the opposite side as that facing battery **2320**. In some such embodiments, device **2600** also includes one

or more devices **2650** such as sound transducers for such applications as a hearing aid having an combined transducer-battery-amplifier device in order to provide a radio frequency-wave-rechargeable hearing aid which could be taken out of the ear at night and placed in a RF-emitting recharging stand (e.g., that of **FIG. 27M**), avoiding the need to replace batteries or even to electrically connect to an external recharging circuit.

[0297] In various embodiments, such an antenna or electromagnetic radiation receiving loop **2660** is fabricated on device **2202**, **2203**, **2203**, **2204**, **2206**, **2207**, **2208**, **2300**, **2400**, or **2500** (or **2700** described below) or other battery devices described herein. In some such embodiments, electromagnetic radiation received wirelessly by antenna **2660** can be such low-frequency radiation as 50- or 60-hertz magnetic radiation from a coil connected to house current (e.g., that of **FIG. 27L**). In other such embodiments, RF radiation such as radio, TV, cellular, etc. having frequencies up to and exceeding 2.4 GHz is received. In some embodiments, multiple antennae are used, e.g., one for transducing communications signals and another for receiving recharging signals.

[0298] **FIG. 26B** shows a block diagram of a layer-deposition system **2660**. System **2660** is much the same as system **2560** of **FIG. 25B**, except that the battery material is deposited on the back of the sheet, i.e., on the side opposite the active parts or connections of circuit **2510**.

[0299] **FIG. 26C** shows a perspective view of a processed sheet **2669**. Sheet **2669** includes a plurality of devices or circuits **2510** each having a battery **2320** on the back. **FIG. 26D** shows a perspective view of diced final devices **2600** after being dices or cut apart. **FIG. 26E** shows a perspective view of wired diced final device **2600** after being wired, e.g., by wires **2615** and **2616** as shown, or by deposited traces (not shown) that extend electrical connections from the top to the bottom of device **2600**.

[0300] In some embodiments, a roll of flexible fabric **2661** suitable for use as a substrate for direct energy conversion has deposited on it the necessary elements and/or layers to form the desired unit (such as a photovoltaic cell) using roll-to-roll concepts. The roll is then taken to the energy deposition tool **2660** that is also configured to operate in a roll-to-roll mode. The battery **2320** is fabricated on the backside (the side opposite the active side of the device, e.g., the side having the light-reception face of a photovoltaic cell) of the roll. Electrical connection is made after fabrication using hardwire techniques, such as shown in **FIG. 26E**.

[0301] In other embodiments such as shown in **FIGS. 24B-24F**, a roll of flexible fabric **2461** suitable for use as a substrate for direct energy conversion (e.g., for a photovoltaic cell) is deposited with materials to form a solid-state lithium battery using roll-to-roll concepts in system **2460**. The resulting roll **2463** is then taken to the direct-energy-conversion materials-deposition tool **2465** that is also configured to operate in a roll-to-roll mode. The direct energy conversion material **2430** is deposited directly on the solid-state battery **2320**. In some embodiments, electrical connection is made through vias formed during battery and device fabrication such as shown in **FIG. 23**.

[0302] In yet other embodiments, roll **2461** above is replaced by a different substrate, such as wafer **2961** of **FIG.**

29A described below, also suitable for use in direct energy conversion. The fabrication tools **2960** and **2965** are also configured to handle the new substrate form factor such as square plates or round wafers.

[0303] In still other embodiments, roll **2661** above is replaced by a different substrate, such as wafer **2971** of **FIG. 29E** below, also suitable for use in direct energy conversion. The fabrication tools **2960** and **2965** are also configured to handle the new substrate form factor such as square plates or round wafers.

[0304] Thus, the present invention provides a method for integrating solid-state lithium batteries with direct energy conversion materials on a flexible fabric. Further, the present invention provides a method for integrating solid-state lithium batteries with direct energy conversion materials on a rigid substrate.

[0305] **FIG. 26F** shows a perspective view of a hearing aid **2690** incorporating a wired diced final device **2600**. In some embodiments, device **2600** includes a photovoltaic cell **2650** for recharging battery **2320** the operates hearing aid **2690**. In some embodiments, sound transducers of conventional materials such as piezo-electric materials are deposited as layers by system **2660** to be used as the microphone and speaker of hearing aid **2690**.

[0306] **FIG. 27A** shows a plan view of a starting substrate **2710** of an embodiment that will have an integrated battery and device sharing a common terminal. **FIG. 27F** shows an elevation view of the starting substrate of **FIG. 27A**. **FIG. 27B** shows a plan view of the substrate **2710** of **FIG. 27A** after deposition of the integrated battery **2320** and device **2430** sharing a common terminal. In some embodiments, integrated battery **2320** and device **2430** are a thin-film battery and supercapacitor having electrical connections **2322**, **2324**, and **2431** such as shown and described in **FIG. 24A** above. **FIG. 27G** shows an elevation view of the partially built device of **FIG. 27B**. **FIG. 27C** shows a plan view of the substrate of **FIG. 27B** after placing and wiring a separately fabricated chip **2440** connected by wires **2441**, **2442**, and **2443** to the integrated battery **2320** and device **2430** sharing common terminal **2324**. **FIG. 27H** shows an elevation view of the partially built device of **FIG. 27C**. **FIG. 27D** shows a plan view of the substrate **2710** of **FIG. 27C** after placing and wiring a loop antenna **2750**. **FIG. 27I** shows an elevation view of the partially built device of **FIG. 27D**. **FIG. 27E** shows a plan view of the final device **2700** having the partially built device of **FIG. 27D** after a top encapsulation layer **2760** has been deposited. **FIG. 27J** shows a cross-section elevation view of the device **2700** of **FIG. 27E**. The elevational views of **FIGS. 27E-27J** are not to scale. In some embodiments, device **2700** is approximately the size and thickness of a common credit card. In some embodiments, a magnetic strip **2770** and raised lettering **2780** are also fabricated on device **2700**.

[0307] **FIG. 27K** shows a perspective view of the device of **FIG. 27E** at a magnetic-recharging station. In the embodiment shown, coil **2790** uses house current to generate a 60 Hz magnetic field, and together with coil **2750**, form a transformer inducing current flow in coil **2750**, which is rectified and used to recharge battery **2320**.

[0308] **FIG. 27L** shows a perspective view of a device **2700** of **FIG. 27E**, but further including a photovoltaic cell

2650, at a light-recharging station that includes lamp. In some embodiments, device **2700** is fabricated in a shape to fit in the ear, includes sound transducers, and functions as a hearing aid that can be recharged an indefinite number of times, eliminating the need to replace its battery.

[0309] FIG. 27M shows a schematic of the device of FIG. 27E at a radio-wave-recharging station **2792**. Radio waves from radio-wave-recharging station **2792** are picked up by antenna **2750**, and the received radio wave's power is scavenged to recharge battery **2320** using a conventional recharging circuit, e.g., implemented in circuit **2440**.

[0310] Solid-state rechargeable batteries such as those described above have the unique ability of being integrated directly with the electronics they will power. Further integration of thin-wire antenna/coil **2660** or **2750** to be used as one of the coils of a two-part transformer such as shown in FIG. 27K and/or RF-scavenging technology such as that used in keyless entry systems allows the recharging of the solid-state thin-film battery **2320** wirelessly (through the air). Using techniques already common in RF I.D. tagging, the communicated energy is converted into a D.C. voltage and used to perform functions on board. In the case where a battery already exists on board, the D.C. voltage is used to power up recharge circuitry to wirelessly recharge the on-board battery.

[0311] Certain needs exist within industry that would benefit from the integration of energy, storage communication and electronics on a single platform. One example is control of warehouse inventories where a small "credit card" is attached to an item in the warehouse. On board the "credit card" is an antenna, supercapacitor, solid-state battery and all required electronics. When the controller needs to know something about the package, the warehouse is queried via cellular or other wireless means with the I.D. of the package in question. The query "wakes up" the package and entices it to respond with whatever data is programmed to be released. The supercapacitor discharges into the antennae-driving circuitry bursting the data out to the central computer. At the same time, the electronics on the credit-card form factor device perform a self evaluation to see if any anomalies have or are occurring such as "battery needs charging." If the answer is yes, the central computer sends a signal of appropriate length to recharge the on-board battery using technology described herein.

[0312] Another application seeing significant enhancement from the integration of energy, communication and electronics on a single platform is an implantable device such as a pacemaker. This technology allows a battery having a very large number (if not infinite) charge/discharge cycles to be implanted as part of a pacemaker. When a "battery-low condition" is encountered, the battery is remotely recharged through the body using AC magnetic fields, sound or ultrasound, radio-frequency or other energy sources.

[0313] Solid-state processes are used to integrate electronics, solid-state rechargeable battery, and antenna on a single platform such as a "credit card" form factor. This is possible by using the low-temperature processes for solid-state batteries and supercapacitors described.

[0314] The present invention provides a platform integrating electronics, solid-state rechargeable batteries, and

antenna on a single platform such as a credit card or implantable device allowing remote wireless recharging of the on-board battery.

[0315] FIGS. 27A-27J show a fabrication sequence in some embodiments of an example of a credit-card form factor I.D. tag with remote recharge capability.

[0316] FIG. 31B shows a fabrication sequence for an example of an implantable device such as a pacemaker **3101**. This method starts with a substantially flat sheet deposited with batteries, which is then cut apart and formed into a three-dimensional shape. The method is otherwise similar to that of FIG. 31C.

[0317] FIG. 31C shows one method for making a pacemaker **3102**. The method includes a plurality of steps carrying the reference numbers **3194**, **3195**, **3196** and **3197**. The pacemaker **3102** includes a first half and a second half **3130**. In the initial step, **3194**, the second half **3130** is provided. A battery cell **1110** is formed on an interior surface of the pacemaker **3102**, as shown by step **3195**. The single cell **1110** is deposited on the interior surface, as shown by step **3195** as shown in step **3191**. The electronics **3150** are then placed onto the battery **1110** to form a circuit with the battery **1110**, as depicted by step **3196**. The first half **3131** of the enclosure is placed over the second half **3130** to form the assembled pacemaker **3102**, as depicted by step **3197**.

[0318] Solid-state rechargeable batteries such as those described above have the unique ability of being integrated directly with the electronics they will power. Further integration of thin-wire antenna and an energy burst device such as a supercapacitor would allow the device to communicate over large distances via any possible number of current communication methods including but not limited to cellular.

[0319] This invention relates to solid-state rechargeable batteries and the integration of such with wireless communication (antennae and electronics), supercapacitor and conventional electronics on a single platform.

[0320] Certain needs exist within industry that would benefit from the integration of energy, communication and electronics on a single platform. One example is control of warehouse inventories where a small "credit card" is attached to an item in the warehouse. On board the "credit card" are an antenna, supercapacitor, solid-state battery and all required electronics. This "credit card" allows tracking of location, time at location, description of item in question and/or information on the environment. When the controller needs to know something about the package, the warehouse is queried via cellular or other wireless means with the I.D. of the package in question. The query "wakes up" the package and entices it to respond with whatever data is programmed to be released. The supercapacitor discharges into the circuitry driving the antennae bursting the data out to the central computer. At the same time, the electronics on the "credit card" performs a self evaluation to see if any anomalies have or are occurring such as battery needs charging. If the answer is yes, the central computer could send a signal of appropriate length to allow recharge of on-board battery using technology described above.

[0321] Solid-state processes are used to integrate electronics, solid-state rechargeable battery, supercapacitor and antenna on a single platform such as a "credit card" form

factor. This is possible by using the low-temperature processes for solid-state batteries and supercapacitors described above.

[0322] Thus, the present invention provides for integrating electronics, solid-state rechargeable batteries, supercapacitors and antenna on a single platform such as a credit card or implantable device.

[0323] Method of recycling and re-using solid-state lithium-ion batteries

[0324] FIG. 28A shows an elevation view of a battery 2800 having stacked cells 2801. Each cell includes an anode tab 2802 and a cathode tab 2803, wherein all of the anode tabs 2802 are soldered together, and all of the cathode tabs 2803 are soldered together. Optionally, battery 2800 is encapsulated with a potting material.

[0325] FIG. 28B shows a plan view of a single battery cell 2801 after recycling. In some embodiments, the anode tab 2802 and the cathode tab 2803 are "tinned" (covered with fresh solder) and/or solder bumped to facilitate reassembly soldering operations.

[0326] FIG. 28C shows a process 2810 used for recycling. Process 2810 includes providing batteries 2800 to be recycled into input bin 2820. In some embodiments, the batteries are de-potted at de-pot station 2822, de-soldered at de-solder station 2824, tested at test station 2826, and outputted into sorted output bins 2828 based on the testing results.

[0327] Of the 2 billion rechargeable batteries consumed in the United States in 1998, only about 300 million were actually recycled. That means about 1.7 billion recyclable batteries made it into landfills. Although more and more of these batteries are technically environmentally safe, this still represents a significant load on the landfill situation in the USA. The present invention provides a solution that will have its greatest impact as solid-state lithium-ion batteries begin to dominate the rechargeable battery market. In this invention, solid-state lithium-ion batteries have a date code and/or recycle value associated with them. Because of the very large (over 40,000) number of charge/discharge cycles possible with solid-state lithium batteries, the average expected life of a cell could exceed 100 years. It is therefore very likely that the product in which the cell is placed will lose its usefulness well before the battery cell is depleted. Thus, when the battery reaches the end of its useful life based on the obsolescence of the product it was in, the consumer will be enticed to recycle the battery based on the value returned to the consumer in exchange for recycling. This value could be a function of the date code and application the battery was used in. The recycler 2810 then disassembles the unit 2800, tests the single cells 2801, and then rebuilds the cells in whatever configuration is most in demand at that time. The rebuilt unit 2800' could then be sold at an appropriate cost and warranty on performance.

[0328] This invention relates to recycling of rechargeable batteries, specifically the recycling of batteries that are manufactured in such a way so as to allow the disassembly of the individual battery cells upon recycling.

[0329] For years the automotive industry has recycled certain high-cost components of the automobile. Using this philosophy, the present invention applies those principles to

the recycling of rechargeable batteries. As battery technology advances, the batteries are actually outlasting the products they were designed for. The conventional solution is to depend on the consumer to recycle the no-longer useful battery by taking it to some place that will accept the battery. The data suggests that this is wishful thinking, as fully 80% of Americans do not recycle their rechargeable batteries. Rather, they throw them into the garbage and the battery ends up in a landfill. Although the newer battery chemistries are relatively benign to the environment, the sheer bulk of the disposed batteries can represent an enormous strain on landfills. This invention allows enticement of the consumer to recycle the batteries by offering a cash reward, or other inducement such as reduced cost on new batteries, in exchange for recycling. Since money is involved, this program should be able to be implemented on a wide scale making participation likely.

[0330] In one embodiment, rechargeable battery manufacturers are encouraged to manufacture their products in such a way that upon recycling, the battery can be broken down into individual cells and these cells rebuilt into "new" batteries. In some embodiments of the present invention provide such a recycling program, and provide batteries with features to facilitate recycling, for example, marking one or more of the cells of a battery with a code indicating such information as date of manufacture, voltage, capacity, value, composition physical size, and/or weight. An example is a cell-phone battery having a capacity of 1000 mAh (milli-ampere hours). Some embodiments involve the parallel assembly of approximately 10 individual cells into a battery pack that would have a capacity of 1000 mAh. These individual cells are fabricated on a grid that provides bonding tabs allowing the configuration of the cells in a variety of modes. Upon recycling, the batteries are de-potted, de-soldered and analyzed for robustness. Cells having data codes and test results indicating substantial life remaining would be repackaged according to market needs. In some embodiments, recycling rechargeable batteries involves the breaking down of the battery pack into individual cells that are tested and re-assembled into usable battery packs. Some embodiments include a method of determining the viability of recycled battery cells for use in rebuilt batteries such as measuring the charge-discharge voltage-current curve over one or more cycles. Some embodiments include a method of de-potting batteries such that the individual cells are accessible and not damaged, such as using a plastic potting compound that can later be dissolved using a solvent and/or heat that does not deteriorate the battery. Some embodiments include a method of disconnecting cells from the original battery pack and re-connecting into a new configuration, such as having solder tabs that extend beyond the battery pack so that the solder tabs can be desoldered without substantially heating the battery itself. Some embodiments include a recycling system based loosely on the system used by the automotive industry in rebuilding of starters, alternators etc. and the techniques used by lead acid battery outlets.

[0331] FIG. 29A shows a block diagram of a layer-deposition system 2960. System 2960 has layer deposition sections 2962 much the same as those of FIG. 2460 of FIG. 24B, except that it is set up to deposit layers onto wafers 2961 (or onto diced ICs 2510 rather than onto flexible

substrates. **FIG. 29B** shows a perspective view of a partially processed wafer **2964** having battery material **2320** on wafer **2961** or IC **2410**.

[0332] **FIG. 29C** shows a block diagram of a layer-deposition system **2965**. System **2965** has layer deposition sections **2962** much the same as those of **FIG. 2465** of **FIG. 24D**, except that it is set up to deposit layers onto wafers **2961** (or onto diced ICs **2510** rather than onto flexible substrates. **FIG. 29D** shows a perspective view of a processed sheet **2969** having battery material **2320** on wafer **2961** or IC **2410** and covered by a device **2430** such as a photovoltaic cell.

[0333] **FIG. 29E** shows a block diagram of a layer-deposition system **2965**. In some such embodiments, system **2965** deposits layers forming a photovoltaic cell device **2650** onto a wafer **2971** or IC **2510**. **FIG. 29F** shows a perspective view of a partially processed wafer **2974**. **FIG. 29G** shows a block diagram of a layer-deposition system **2960**. In some such embodiments, system **2960** deposits layers of a battery **2320**. **FIG. 29H** shows a perspective view of a processed wafer **2979**. In some embodiments, wafer **2979** represents a single device, and in other embodiments, wafer **2979** is diced or cut into a plurality of individual devices and then wired as necessary to connect the signals on the top of the device to the bottom of the device. **FIG. 29I** shows a perspective view of wired diced final device **2600**.

[0334] Turning now to **FIGS. 30, 31** and **32**, specific examples of devices will now be provided. **FIG. 30** shows an implantable device **3000** used to stimulate specific portions of the brain. One use of such device **3000** is for deep brain neural stimulation, for example, in order to treat Parkinson's Disease. By sending signals to a specific portion of the brain the tremors associated with Parkinson's Disease may be reduced. In the past, a lead or conductor was implanted in the brain so that electrical signals may be sent to the specific area of the brain for reducing tremors. The lead passes under the skull and through the neck to a pocket near the patient's chest in current versions. As shown in **FIG. 30**, after a burr hole has been made in the skull, a port **3010** is placed in the burr hole. The port **3010** includes a cap **3012**, which is used to hold the lead in place during implantation as well as after implantation. In this particular invention, the cap **3012** is made of a suitable biocompatible material. Imbedded within the cap is a battery cell **1110** or a series of battery cells **1110**. The electronics necessary to deliver the signals at a desired rate or programmable rate is also imbedded within the cap **3012**. An RF antenna **3014** is also placed within the cap so that the battery **1110** imbedded within the cap **3012** can be recharged by passing radio frequency into the cap or inductively coupling the required energy into the cap. Another embodiment may use the lead **3020** for an energizing antenna and may include a separate antenna for programming the electronics used to deliver signals to the brain.

[0335] **FIG. 31A** is directed toward a pacemaker **3100**. Rather than include separate batteries within the case of the pacemaker **3100**, the enclosure, or at least one enclosure portion, includes a battery **1110** or a series of cells **1110**. The pacemaker **3100** may include an antenna **3120** which is used to direct radio frequency toward the pacemaker for recharging of the battery **1110** that is positioned within the case or enclosure of the pacemaker **3100**.

[0336] **FIG. 31B** shows the method for making the pacemaker **3100**. The method is comprised of a plurality of steps carrying the reference numbers **3190, 3191, 3192** and **3193**. The pacemaker **3100** includes a first half and a second half **3130**. A plurality of battery cells **1110** are formed on a substrate material **3140**, as shown by step **3190**. The substrate material **3140** is diced or cut resulting in a single cell **1110** on the sheet as diced. The single cell **1110** is adhesively bonded to the second half **3130** of the pacemaker **3100**, as shown in step **3191**. The electronics **3150** are then placed onto the battery **1110** to form a circuit with the battery **1110**, as depicted by step **3192**. The first half **3131** of the enclosure is placed over the second half **3130** to form the assembled pacemaker **3100**.

[0337] **FIG. 32A** is a perspective cutaway view of a watch **3200**. The watch includes a case **3210** and a band **3212** for strapping onto a person's wrist. Within the case **3210** is a solar cell **3220** and an LCD **3222**. The solar cell **3220** is attached to the battery or series of battery cells **1110**. The LCD **3222** is attached to the battery and electronic (not shown). The battery powers the LCD **3222** and is associated to electronics associated with the watch **3200**. The solar cell **3220** recharges the battery **1110** more or less continuously. Both the solar cell **3220** and the LCD **3222** appear at the crystal or glass portion of the watch. Advantageously, this type of watch can be sealed forever so that it can be made absolutely watertight.

[0338] Another embodiment of a watch is shown in **FIG. 32B**. In this particular instance, a circular-shaped solar cell **3240** is positioned atop a circular-shaped battery cell **1110**. The circular-shaped solar cell includes an opening **3241** therein. A set of hands for an analog watch may be inserted through the opening. The crystal or glass face of the watch will then be opened to the solar cell **3240** so that it can continuously charge the battery **1110**, which in turn powers the working portion of the watch.

[0339] Conclusion

[0340] Described above are an improved method and system for fabrication of solid-state energy devices and the solid-state energy devices per se. Solid-state energy devices include storage devices, e.g., batteries and capacitors, and conversion devices, e.g., photovoltaic cells. In some embodiments, the present method focuses the energy for forming films having fewer defects and/or certain crystal properties directly at a certain film. By focusing the energy, collateral damage to other parts of a structure or device are minimized. Accordingly, a wider array of materials, devices and structures are available with which an energy storage device or energy conversion device is fabricated according to the present invention.

[0341] In one embodiment, the method includes depositing a first film on a substrate by depositing a first material to a location on the substrate, and supplying energized ions of a second material different than the first material to the location on the substrate to control growth of a crystalline structure of the first material at the location. Electrolyte and anode layers are subsequently formed.

[0342] In one embodiment, the secondary source is an ion source with the capability of supplying energetic ions having an energy greater than 5 eV. In another embodiment, the energy range is about 5 eV to about 3,000 eV. In one

embodiment, the energy range of is about 5 eV to about 1,000 eV. The energy range in a further embodiment is about 10 eV to about 500 eV. In another embodiment, the energy range is in the range of about 60 eV to 150 eV. In another embodiment, the energy range is about 140 eV.

[0343] In some embodiments, the cathode layer is formed without an anneal step being performed thereon as the present method provides the necessary energy to grow a sufficiently crystalline layer by use of the secondary source.

[0344] In some embodiments, the substrate is formed of a material that will degrade due to thermal effects at a high temperature such as an anneal temperature for conventional cathode thin films. In one embodiment, the substrate will thermally degrade at a temperature of less than 700 degrees Celsius. In one embodiment, the substrate will thermally degrade at a temperature of less than 300 degrees Celsius. In one embodiment, the substrate will thermally degrade at a temperature of less than 250 degrees Celsius. In one embodiment, the substrate will thermally degrade at a temperature of less than 100 degrees Celsius.

[0345] Some embodiments of the method include depositing a seed material on a first film to assist in the low energy deposition of the lithium intercalation material of a subsequent film thereon. In an embodiment, the seed material assists in orienting the crystal structure of the subsequent film.

[0346] Another aspect of the present invention provides structure and methods for controlling the temperature of the substrate and any films formed thereon during subsequent depositing steps. This allows a more temperature sensitive material to be used as the substrate while preventing thermal degradation effects of the substrate.

[0347] In one embodiment of the method, subsequent films, including the electrolyte film and the anode film, are deposited using a secondary source. In one embodiment, the secondary source provides energy to orient the crystalline structure of the electrolyte film and the anode film and influence the stoichiometry of the resulting films.

[0348] In another embodiment of the method, the battery, and/or layers of the battery are subject to a post fabrication cryogenic anneal to correct defects in the crystalline structure.

[0349] Another important aspect of the methods described herein pertains to the quality of the crystalline materials, e.g., size, of the crystallites in the thin layer. Crystallite size relates to the electrical performance of lithium intercalation materials in lithium batteries. Crystallite size in the layers deposited according to the methods described herein is adequately large to achieve commercially viable electrical properties.

[0350] The energy-storage devices, such as thin-film batteries, capacitors, and supercapacitors include an ultra-thinned electrolyte film interposed between two electrode films. Such ultra-thinned electrolyte films improve electrical performance of the energy-storage devices by decreasing the resistance within the electrolyte layer while still maintaining the required insulative property between the two electrodes. By using the methods described herein, such ultra-thin electrolyte films of high structural quality and stoichiometric control are fabricated. In some embodiments, a supercapaci-

tor includes a substrate, a solid-state, electrode first film on the substrate, a solid-state, electrolyte second film on the first electrode film, and a solid-state, electrode second film on the electrolyte. In some embodiments, the supercapacitor has a working voltage is greater than 5 volts. In some embodiments, the supercapacitor has an electrolyte film that is less than about 5,000 Angstroms thick. In some embodiments, the supercapacitor has an electrolyte film that is less than about 1,000 Angstroms thick. In some embodiments, the supercapacitor has an electrolyte film that is less than about 500 Angstroms thick. In some embodiments, the supercapacitor has an electrolyte film in a range of about 600 to 900 Angstroms. In some embodiments, the supercapacitor has an electrolyte film in a range of about 1100 to 1900 Angstroms. In some embodiments, the supercapacitor has an electrolyte film in a range of about 2100 to 2900 Angstroms. In some embodiments, the supercapacitor has an electrolyte film in a range of about 3000 to 4900 Angstroms. In some embodiments, the supercapacitor has an electrolyte film that is less than about 250 Angstroms thick.

[0351] In some embodiments, the supercapacitor has an electrode film that is a metal oxide. In some embodiments, the supercapacitor has an electrode film that is a RuO_2 .

[0352] The electrodes (i.e., plates), in some embodiments, include the same material(s). In some embodiments, the supercapacitor includes both electrode films being formed of a lithium intercalation material. In some embodiments, at least one of the electrode films has a highly ordered crystal structure that provides superior energy charge/discharge rates and electrical energy density. In some embodiments, at least one electrode film includes lattice planes that are essentially perpendicular to a boundary with the electrolyte film.

[0353] In some embodiments, the supercapacitor includes a substrate has a thermal degradation temperature of less than 700 degrees Celsius. In some embodiments, the supercapacitor includes a substrate has a thermal degradation temperature of less than about 600 degrees Celsius. In some embodiments, the supercapacitor includes a substrate has a thermal degradation temperature of less than about 500 degrees Celsius. In some embodiments, the supercapacitor includes a substrate has a thermal degradation temperature of less than about 400 degrees Celsius. In some embodiments, the supercapacitor includes a substrate has a thermal degradation temperature of less than about 300 degrees Celsius. In some embodiments, the supercapacitor includes a substrate has a thermal degradation temperature of less than about 200 degrees Celsius. In some embodiments, the supercapacitor includes a substrate has a thermal degradation temperature of less than about 100 degrees Celsius.

[0354] In some embodiments, at least one electrode is non-annealed and has a crystallite size of greater than about 240 Angstroms. In some embodiments, both electrodes are non-annealed and has a crystallite size of greater than about 240 Angstroms. In some embodiments, a non-annealed electrode film has a crystalline orientation with its lattice planes essentially perpendicular to a boundary between the non-annealed electrode film and the electrolyte film. In some embodiments, both electrodes have a crystalline orientation with its lattice planes essentially perpendicular to a boundary between the non-annealed electrode film and the electrolyte film.

[0355] In some embodiments, an electrolyte film of a supercapacitor has an ordered structure that allows higher ion diffusion rates for enhancing the charge/discharge characteristics of the supercapacitor. Another embodiment is directed to a method of fabricating a supercapacitor including forming a solid-state first electrode film on a substrate, forming a solid-state electrolyte film on the first electrode film, and forming another solid-state second electrode film. In some embodiments, forming the electrolyte includes depositing an electrolyte material to a location on the first electrode film, and supplying energized ions of a second material adjacent the location to control growth of a structure of the electrode material at the location. In some embodiments, depositing the primary material includes depositing lithium phosphorus oxynitride as the primary material. In some embodiments, forming the solid-state, first electrode film includes depositing an electrode material to a location on the substrate, and supplying energized ions of a second material adjacent the location to control growth of a crystalline structure of the electrode material at the location. In some embodiments, depositing the electrode material includes depositing a metal oxide.

[0356] In some embodiments, depositing the electrode material includes depositing RuO_2 . In some embodiments, forming both solid-state, electrode films include depositing an electrode material to a location on the substrate, and supplying energized ions of a second material adjacent the location to control growth of a crystalline structure of the electrode material at the location.

[0357] In some embodiments, supplying energized ions includes supplying ions having an energy within the range of about 5 to about 3000 eV. In some embodiments, supplying energized ions includes supplying ions having an energy within the range of about 10 to about 2000 eV. In some embodiments, supplying energized ions includes supplying ions having an energy within the range of about 15 to about 1500 eV. In some embodiments, supplying energized ions includes supplying ions from a source gas including one or more of O_2 and N_2 . In some embodiments, supplying energized ions includes supplying ions from a source gas including one or more of argon, xenon, helium, neon, and krypton.

[0358] In some embodiments, supplying energized ions includes supplying ions having an energy within the range of about 5 to about 200 eV. In some embodiments, supplying energized ions includes supplying ions having an energy within the range of about 10 eV to about 500 eV. In some embodiments, supplying energized ions includes supplying ions having an energy within the range of about 60 eV to about 150 eV. In some embodiments, supplying energized ions includes supplying ions having an energy of about 140 eV.

[0359] In some embodiments, depositing the electrolyte film includes using chemical vapor deposition to direct the primary material to the location on the substrate. In some embodiments, depositing the electrolyte film includes using physical vapor deposition to direct the primary material to the location on the substrate. In some embodiments, supplying energized ions controls the stoichiometry of a growing electrolyte film.

[0360] It is also within the scope of the present invention to fabricate photovoltaics. In some embodiments, the method of fabricating a photovoltaic device includes form-

ing an electrode first film on a substrate, forming a semiconductor second film on the electrode first film, forming a semiconductor third film on the semiconductor second film; and forming an electrode fourth film on the semiconductor third film. In some embodiments, one of forming step includes depositing semiconductor material using a deposition source, and supplying energy to the semiconductor material to deposit the semiconductor material into a desired film structure.

[0361] In some embodiments, supplying energy includes supplying energized particles having energy of greater than about 5 eV. In some embodiments, supplying energy includes supplying energized particles having energy of less than about 5000 eV. In some embodiments, supplying energy includes supplying energized particles having energy of less than about 3000 eV. In some embodiments, supplying energy includes supplying energized particles having energy of less than about 2000 eV. In some embodiments, supplying energy includes supplying energized particles having energy of less than about 1000 eV. In some embodiments, supplying energy includes supplying energized particles having energy of less than about 500 eV. In some embodiments, supplying energy includes supplying energized particles having energy of less than about 300 eV. In some embodiments, supplying energy includes supplying energized particles having energy in the range of about 5 eV to about 500 eV. In some embodiments, supplying energy includes supplying energized particles having energy in the range of about 5 eV to about 250 eV. In some embodiments, supplying energy includes supplying energized particles having energy in the range of about 10 eV to about 200 eV. In some embodiments, supplying energy includes supplying energized particles having energy in the range of about 20 eV to about 40 eV. In some embodiments, the energized particles are focused. In some embodiments, a film includes CdS. In some embodiments, a film includes CdTe.

[0362] In some embodiments, the energized particles include ionized sulfur. In some embodiments, forming a film includes depositing cadmium and reacting the cadmium with an ionized sulfur. In some embodiments, the energy is supplied simultaneously with depositing material from a deposition source. In some embodiments, the energy beam and the material beam are not coincident. In some embodiments, supplying energy includes supplying oxygen ions. In some embodiments, supplying energy includes supplying a noble gas ions. In some embodiments, the substrate is not heated during forming films.

[0363] In some embodiments, forming a semiconductor film on another film includes depositing a high quality first region and then depositing a second highly doped region on the first region.

[0364] In some embodiments, a photovoltaic cell includes an essentially transparent substrate, an essentially transparent electrode first film on the substrate, a semiconductor second film on the electrode first film, a semiconductor third film on the semiconductor second film, and an electrode fourth film on the semiconductor third film. In some embodiments, the third film includes a high quality first region adjacent to the second film and a highly doped second region remote from the second film, and the first region and the second film form a pn junction of the photovoltaic cell.

[0365] In some embodiments, at least one of the semiconductor films has a thickness of about 50 nanometers to about

100 nanometers. In some embodiments, the first region of the third film has a thickness of about 50 nanometers to about 100 nanometers. In some embodiments, the second region of the third film has a thickness of greater than about 1000 nanometers. In some embodiments, the high quality first region of the third film is formed by providing energized ions while material is deposited on the second film to grow the first region such that the first region has a reduced number of defects. In some embodiments, the first region of the third film has large crystal sizes. In some embodiments, the first region includes P-doped CdTe. In some embodiments, the substrate, first film, second film third film, and fourth film provide an efficiency of greater than about 10 percent.

[0366] In some embodiments, at least one of the substrate, the first film and the fourth film have a thermal degradation temperature of less than about 500 degrees Celsius. In some embodiments, the thermal degradation temperature is less than about 450 degrees Celsius. In some embodiments, the thermal degradation temperature is less than about 400 degrees Celsius. In some embodiments, the thermal degradation temperature is less than about 350 degrees Celsius. In some embodiments, the thermal degradation temperature is less than about 300 degrees Celsius. In some embodiments, the thermal degradation temperature is less than about 250 degrees Celsius.

[0367] Another embodiment of the present invention is a system for fabricating photovoltaic cells. In some embodiments, the system includes a chamber in which a substrate on which the cell is to be formed and deposition apparatus connected to the chamber. In some embodiments, the deposition apparatus directs material to be deposited toward the substrate. In some embodiments, a system includes energizing means that provides energy to the material while the material is forming a film on the substrate. In some embodiments, the means for providing energy includes a directed energy source that does not appreciably heat the substrate. In some embodiments, the chamber has a temperature of less than about 600 degrees Celsius. In some embodiments, the chamber has a temperature of less than about 500 degrees Celsius. In some embodiments, the chamber has a temperature of less than about 400 degrees Celsius. In some embodiments, the chamber has a temperature of less than about 300 degrees Celsius. In some embodiments, the chamber has a temperature of less than about 250 degrees Celsius. In some embodiments, the chamber has a temperature of less than about 200 degrees Celsius. In some embodiments, the chamber is adapted to hold the substrate below about 300 degrees Celsius. In some embodiments, the chamber is adapted to hold the substrate below about 200 degrees Celsius.

[0368] Another embodiment includes a system for fabricating photovoltaic cells. The system includes a chamber in which a substrate on which the cell is to be formed, a deposition apparatus connected to the chamber, the deposition apparatus directing material to be deposited toward the substrate, and an energy source directing energy to the material while the material is forming a film on the substrate. In some embodiments, the energy source is an ion beam source. In some embodiments, the ion beam source produces energized ions having energy of about 3 eV to about 3,000 eV. In some embodiments, the ion beam source produces energized ions having energy of about 5 eV to about 5,000 eV. In some embodiments, the ion beam source produces

energized ions having energy of about 10 eV to about 1,000 eV. In some embodiments, the ion beam source produces energized ions having energy of about 15 eV to about 500 eV. In some embodiments, the ion beam source produces energized ions having energy of about 30 eV to about 300 eV. In some embodiments, the ion beam source produces energized ions having energy of about 40 eV to about 400 eV. In some embodiments, the ion beam source produces energized ions having energy of about 40 eV to about 300 eV. In some embodiments, the ion beam source produces energized ions having energy of about 100 eV to about 200 eV.

[0369] It is to be understood that the above description is intended to be illustrative, and not restrictive. Although numerous characteristics and advantages of various embodiments as described herein have been set forth in the foregoing description, together with details of the structure and function of various embodiments, many other embodiments and changes to details will be apparent to those of skill in the art upon reviewing the above description. The scope of the invention should, therefore, be determined with reference to the appended claims, along with the full scope of equivalents to which such claims are entitled.

The invention claimed is:

1. A supercapacitor, comprising:
 - a substrate;
 - a solid-state, electrode first film on the substrate;
 - a solid-state, electrolyte second film on the first electrode film; and
 - a solid-state, electrode second film on the electrolyte.
2. The supercapacitor of claim 1, wherein the working voltage is greater than 5 volts.
3. The supercapacitor of claim 1, wherein the electrolyte second film is less than about 5,000 Angstroms thick.
4. The supercapacitor of claim 1, wherein the electrolyte second film is less than about 1,000 Angstroms thick.
5. The supercapacitor of claim 1, wherein the electrolyte second film is less than about 500 Angstroms thick.
6. The supercapacitor of claim 1, wherein the electrolyte second film is less than about 250 Angstroms thick.
7. The supercapacitor of claim 1, wherein the electrode first film is a metal oxide.
8. The supercapacitor of claim 1, wherein the electrode first film is a RuO₂.
9. The supercapacitor of claim 1, wherein the electrode third film is a metal oxide.
10. The supercapacitor of claim 1, wherein the electrode third film is a RuO₂.
11. The supercapacitor of claim 1, wherein both the first film and the third film are formed of lithium intercalation material.
12. The supercapacitor of claim 1, wherein the first film and the third film each have a highly ordered crystal structure that provides superior energy charge/discharge rates and electrical energy density.
13. The supercapacitor of claim 1, wherein at least one of the first film and the second film include lattice planes that are essentially perpendicular to a boundary with the electrolyte second film.

14. The supercapacitor of claim 1, wherein the substrate has a thermal degradation temperature of less than 400 degrees Celsius.

15. The supercapacitor of claim 1, wherein the substrate has a thermal degradation temperature of less than 200 degrees Celsius.

16. The supercapacitor of claim 1, wherein the substrate has a thermal degradation temperature of less than 100 degrees Celsius.

17. The supercapacitor of claim 1, wherein the first film is a non-annealed electrode having a crystallite size of greater than about 240 Angstroms.

18. The supercapacitor of claim 17, wherein the first film is a non-annealed electrode film having a crystalline orientation with the lattice planes essentially perpendicular to a boundary between the non-annealed electrode film and the electrolyte film.

19. The supercapacitor of claim 1, wherein the third film has a crystallite size of greater than about 240 Å.

20. The supercapacitor of claim 19, wherein the third film is an electrode film having a crystalline orientation with the lattice planes essentially perpendicular to a boundary between the electrode film and the electrolyte second film.

21. The supercapacitor of claim 1, wherein the electrolyte second film has an ordered structure that allows higher ion diffusion rates for enhancing the charge/discharge characteristics of the supercapacitor.

22. A method of fabricating a supercapacitor, comprising:

providing a substrate;

forming a solid-state first electrode film on the substrate;

forming a solid-state electrolyte second film on the first electrode film, including:

(a) depositing an electrolyte material to a location on the first electrode film, and

(b) supplying energized ions of a second material adjacent the location to control growth of a structure of the electrode material at the location; and

forming a solid-state second electrode third film.

23. The method of claim 22, wherein depositing the primary material includes depositing lithium phosphorus oxynitride as the primary material.

24. The method of 22, wherein forming the solid-state, first electrode film includes:

depositing an electrode material to a location on the substrate, and supplying energized ions of a second material adjacent the location to control growth of a crystalline structure of the electrode material at the location.

25. The method of 24, wherein depositing the electrode material includes depositing a metal oxide.

26. The method of 24, wherein depositing the electrode material includes depositing RuO₂.

27. The method of 22, wherein forming the solid-state, electrode second film includes:

depositing an electrode material to a location on the substrate, and supplying energized ions of a second material adjacent the location to control growth of a crystalline structure of the electrode material at the location.

28. The method of 27, wherein depositing the electrode material includes depositing a metal oxide.

29. The method of 27, wherein depositing the electrode material includes depositing RuO₂.

30. The method of claim 22, wherein supplying energized ions includes supplying ions having an energy within the range of about 5 to about 3000 eV.

31. The method of claim 30, wherein supplying energized ions includes supplying ions from a source gas including one or more of O₂ and N₂.

32. The method of claim 30, wherein supplying energized ions includes supplying ions from a source gas including one or more of argon, xenon, helium, neon, and krypton.

33. The method of claim 22, wherein supplying energized ions includes supplying ions having an energy within the range of about 5 to about 200 eV.

34. The method of claim 22, wherein supplying energized ions includes supplying ions having an energy within the range of about 10 eV to about 500 eV.

35. The method of claim 22, wherein supplying energized ions includes supplying ions having an energy within the range of about 60 eV to about 150 eV.

36. The method of claim 22, wherein supplying energized ions includes supplying ions having an energy of about 140 eV.

37. The method of claim 22, wherein depositing the electrolyte second film includes using chemical vapor deposition to direct the primary material to the location on the substrate.

38. The method of claim 22, wherein supplying energized ions includes controlling stoichiometry of a growing electrolyte film.

39. The method of claim 22, wherein depositing the electrolyte film includes using physical vapor deposition to direct the electrolyte material to the location on the substrate.

40. A method of fabricating a photovoltaic device, comprising:

providing a substrate;

forming an electrode first film on the substrate;

forming a semiconductor second film on the electrode first film;

forming a semiconductor third film on the semiconductor second film; and

forming an electrode fourth film on the semiconductor third, film wherein one of forming the second film and forming the third film includes:

depositing semiconductor material using a deposition source; and

supplying energy to the semiconductor material to deposit the semiconductor material into a desired film structure.

41. The method of claim 40, wherein supplying energy includes supplying energized particles having energy of greater than about 5 eV.

42. The method of claim 40, wherein supplying energy includes supplying energized particles having energy of less than about 3000 eV.

43. The method of claim 40, wherein supplying energy includes supplying energized particles having energy in the range of about 5 eV to about 500 eV.

44. The method of claim 40, wherein supplying energy includes supplying energized particles having energy in the range of about 5 eV to about 250 eV.

45. The method of claim 40, wherein supplying energy includes supplying energized particles having energy in the range of about 10 eV to about 200 eV.

46. The method of claim 40, wherein supplying energy includes supplying energized particles having energy in the range of about 20 eV to about 40 eV.

47. The method of claim 40, wherein forming the second film includes depositing CdS.

48. The method of claim 40, wherein forming the third film includes depositing CdTe.

49. The method of claim 40, wherein forming the second film includes the supplying energy, and wherein the supplying energy includes supplying ionized sulfur.

50. The method of claim 49, wherein forming the second film includes depositing the cadmium and reacting the cadmium with the ionized sulfur.

51. The method of claim 40, wherein forming the third film includes the supplying energy, and wherein the supplying energy includes supplying energized ions.

52. The method of claim 51, wherein forming the third film includes depositing the cadmium.

53. The method of claim 40, wherein supplying energy includes supplying ions simultaneously with depositing material from the deposition source.

54. The method of claim 40, wherein supplying energy includes supplying oxygen ions.

55. The method of claim 40, wherein supplying energy includes supplying a noble gas ions.

56. The method of claim 40, wherein the substrate is not heated during forming the second film or the third film.

57. The method of claim 40, wherein forming the semiconductor third film on the semiconductor second film includes depositing a high quality first region and then depositing a second highly doped region on the first region.

58. The method of claim 40, wherein the one of forming the second film and forming the third film includes providing energy to the semiconductor material being deposited by only means sending the semiconductor material toward the cell and by the means supplying energy.

59. A photovoltaic cell, comprising:

an essentially transparent substrate;

an essentially transparent electrode first film on the substrate

a semiconductor second film on the electrode first film;

a semiconductor third film on the semiconductor second film; and

an electrode fourth film on the semiconductor third film, wherein the third film includes a high quality first region adjacent to the second film and a highly doped second region remote from the second film, and the first region and the second film form a pn junction of the photovoltaic cell.

60. The device of claim 59, wherein the second film has a thickness of about 50 nanometers to about 100 nanometers.

61. The device of claim 59, wherein the first region of the third film has a thickness of about 50 nanometers to about 100 nanometers.

62. The device of claim 61, wherein the second region of the third film has a thickness of greater than about 1000 nanometers.

63. The device of claim 59, wherein the high quality first region of the third film is formed by providing energized ions while material is deposited on the second film to grow the first region such that the first region has a reduced number of defects.

64. The device of claim 63, wherein the first region of the third film has large crystal sizes.

65. The device of claim 64, wherein the first region includes P-doped CdTe.

66. The device of claim 59, wherein the substrate, first film, second film third film, and fourth film provide an efficiency of greater than about 10 percent.

67. The device of claim 59, wherein at least one of the substrate, the first film and the fourth film have a thermal degradation temperature of less than about 500 degrees Celsius.

68. A system for fabricating photovoltaic cells, comprising:

chamber in which a substrate on which the cell is to be formed,

deposition apparatus connected to the chamber, the deposition apparatus directing material to be deposited toward the substrate, and

means for providing energy to the material while the material is forming a film on the substrate.

69. The system of claim 68, wherein the means for providing energy includes a directed energy source that does not appreciable heat the substrate.

70. The system of claim 68, wherein the chamber has a temperature of less than about 300 degrees Celsius.

71. The system of claim 68, wherein the chamber has a temperature of less than about 250 degrees Celsius.

72. The system of claim 68, wherein the chamber is adapted to hold the substrate below about 500 degrees Celsius.

73. The system of claim 68, wherein the chamber is adapted to hold the substrate below about 300 degrees Celsius.

74. The system of claim 68, wherein the chamber is adapted to hold the substrate below about 200 degrees Celsius.

75. A system for fabricating photovoltaic cells, comprising:

a chamber in which a substrate on which the cell is to be formed,

a deposition apparatus connected to the chamber, the deposition apparatus directing material to be deposited toward the substrate, and

an energy source directing energy to the material while the material is forming a film on the substrate.

76. The system of claim 75, wherein the energy source is an ion beam source.

77. The system of claim 76, wherein the ion beam source produces energized ions having energy of about 3 eV to about 3,000 eV.