UNITED STATES PATENT OFFICE.

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TREATMENT OF FIBROUS CELLULOSE FOR THE PRODUCTION OF HYDRATED DERIVATIVES.

No Drawing.

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To all whom it may concern:

Cross and Edward J. Bevan, both subjects sorbed alkali. On the other hand, when a of the King of Great Britain, both residing less concentrated solution is used the formab in London, England, have jointly invented tion of soluble derivatives on subsequent or brous Cellulose for the Production of Hy- phide is increasingly imperfect. Thus, when

is a specification.

The action of the alkaline hydrates in strong solution on fibrous celluloses determines structural changes in the fibres, especially marked by shrinkage, and in the case of selected cotton textile fabrics the action, with certain mechanical aids, is ap-20 tic soda of 15-17.5 per cent, (NaOH) at are evidenced in further reaction with 25 (NaOH) the diminution of effect is rapid, The fibrous mass swells considerably, and zation.

35 rivatives (by subsequent action of carbon di- xanthates, a treatment with dilute acid may sulphide) as described in specification of be desirable. British Letters Patent No. 8700 of 1892, it is this fully hydrated condition of the known that the cellulose must be taken to cellulose and the implied corresponding 40 the fibrous cellulose must be treated with a to the novel technical results of the present plete conversion to water soluble derivatives (NaOH), and simultaneously or subsequentin the subsequent reaction, are prepared by ly, with carbon disulphide. treating the air-dry cellulose with caustic soda lye of 17 per cent (NaOH) strength ton fibre may be impregnated with twice its and pressing the product so that the fibrous weight of caustic soda, of 9 per cent strength

mass retains solution amounting to 3 times 50 Be it known that we, Charles Frederick the weight of the cellulose, inclusive of aba certain new and useful Treatment of Fi- simultaneous treatment with carbon disul- 55 drated Derivatives, of which the following cotton or other cellulose is uniformly impregnated with solution of 9 per cent (NaOH) in the proportion of twice the weight of the fibre or fabric, and then ex- 60 posed to carbon disulphide, only a small fraction of the material is converted into water soluble derivatives.

The attendant formation of water-insoluplied to produce the effects known as "mer- ble xanthates is also fractional in amount. 65 cerization". In practice the mercerizing But, as we have discovered, fundamental process consists in applying under the suit- reaction occurs, and the cellulose is proable conditions, an aqueous solution of caus- foundly modified. The new characteristics ordinary temperatures. The effects are di- water; structural changes in the fibre are 70 minished as the concentration of the caustic developed, with large increase of diameter, soda is diminished below 15 per cent; with and thickening of the cell wall, and in the solutions of 12.5 per cent to 10.5 per cent mass there are visible effects of hydration. and it may be said that at strengths below retains a very much increased percentage of 75 12.5 per cent caustic soda solution has no water, after having been pressed or centrisubstantial mercerizing effect. Thus cotton fuged. The fibres, however, though much or cotton fabric treated with an alkaline lye distended remain free, and may be readily of 9 per cent caustic soda (NaOH) is not and quickly washed free from excess of sensibly affected in the direction of merceri- alkali, and the small proportion of soluble 80 cellulose compounds. For subsequent re-For the production of alkali-cellulose in moval of residual alkali and decomposition the synthesis of water soluble cellulose de- of fractional residues of water-insoluble

the stage of full mercerization and therefore plasticity of the substance, which gives rise solution containing the equivalent of up- invention, which consists in modifying the 90 wards of 15.0 per cent of caustic soda cellulose for subsequent full hydration by (NaOH). Alkali-celluloses for industrial treatment of the cellulose with alkali soluproduction of viscose, which implies com- tion of a strength between 6 and 11 per cent

As an example of the treatment, raw cot-

and introduced into a closed chamber together with carbon disulphide amounting to say 15 per cent of the weight of the original fibre. After a period of 6-10 hours, at the 5 ordinary temperature, the mass is transferred to a convenient apparatus, in which the product is now washed with water at ordinary temperature for removal and recovery of alkali and to determine or develop

10 the effects of hydration.

The structural modifications produced in the cotton fibre and the changes accompanying the swelling of the fibre generally resemble those produced by saline solvents of 15 cellulose in the earlier stages of action, that is mainly in the thickening by swelling of the cell wall. A more special effect is the exaggeration of the central canal defined by symmetrical sinuous outlines. In extreme 20 treatment a definite continuous spiral appears in the mass of swollen fibre substance. This development of the ultimate spiral differentations of the external tissues of the cotton is more marked if the preparatory 25 treatment with alkali is carried out at higher temperatures, say 80° to 90° C.

The hydration effects are enhanced by increasing the proportion of alkali to fibre in the first stage of treatment, for instance 30 by using an amount of caustic soda solution equal to 2.5 to 3.0 times the weight of the fibre. The effects may also be controlled by varying the temperature of the alkali solution, the action of the latter in 35 determining hydration effects being greater the lower the temperature; on the other hand, raising the temperature alters the structural characteristics of the hydrated product, and particularly in the sense that when formed into a mass, or in sheets and subjected to pressure it yields a parchmentlike material.

The finally reverted air dry product generally retains 10 to 11 per cent of moisture as compared with the 6 to 7 per cent normal to the raw cotton fibre, which is correlative with a generally increased dyeing capacity.

Instead of washing with water the material which has been attacked by carbon disulphide, it may be washed with a strong solution of a neutral alkali salt, for instance brine, to remove excess of alkali; the hydranection with the recovery of the alkali, which is thereby obtained in more concentrated 2. A process of treating fibrous cellulose state, as it is in furnishing a product which can be subsequently hydrated by washing away the brine.

term "fibrous cellulose" includes the well- the ordinary temperature. known industrial forms of cellulose: cotton, 3. A process of treating fibrous cellulose

proportions of alkali-soluble constituents; otherwise, as in the case of flax, it is advantageous to operate on the alkali-scoured material; also papermakers' half stuff, or pulps in the bleached or the unbleached state, in 70 which they retain residues of non-cellulose components.

The invention may also be applied to compound celluloses of the type of jute fibre, in treating which the upper limit of concen- 75 tration of the alkali used in the preliminary

treatment is required.

In regard to the technical uses of the product, its use in paper-making has been noted.

It is to be noted that the hydration effects give the necessary papermaking quality, and are independent therefore of any hydration effects such as produced in the ordinary Hollander or beater. Therefore, 85 also fibrous cellulose of any length of fibre may be brought into the state necessary for wet manipulation with conservation of length. Hence, fine felted sheets of say 15 grammes per sq. meter of area can be produced, showing considerable adhesion in the final reverted state.

The hydrated fibre in admixture with ordinary paper-making cellulose, that is, as a mixed fibre, produces its proportionate 95 effect, and it is characteristic of the reversion that in the final form the fibre retains

elasticity and bulk.

In treating textile materials the process may be applied to roving, to be afterwards 100 spun and drawn by the wet process to yarn. Yarns and cloth may also be treated with or without the tension necessary to oppose shrinkage of the material which occurs in 105 the hydration process.

In this way a range of finishes and effects are obtained with some of the characteristics of mercerization, but differentiated by reason of the relative elasticity and 110 bulk of the fibre or material in the finished

state.

Having thus described the nature of the same invention and the best means we know of carrying the same into practical effect, 115 we claim:—

1. A process of treating fibrous cellulose which consists in subjecting the cellulose tion effect is thus suspended and can be sub- to the action of alkaline solution of strength sequently produced by treatment with wa- between 6 and 11 per cent (NaOH), and ter. This brine treatment is useful in con- of carbon bisulphide at the ordinary temof carbon bisulphide at the ordinary temperature.

which consists in subjecting the cellulose to the action of alkaline solution of strength 125 between 6 and 11 per cent (NaOH) and For the purposes of this invention the then to the action of carbon bisulphide at

flax, hemp, esparto, wood cellulose, etc. which consists in subjecting the cellulose 130 Also, raw cottons which contain only small to the action of alkaline solution of strength

of carbon bisulphide at the ordinary tem- alkali salt. salt.

4. A process of treating fibrous cellulose metrical sinuous outlines. which consists in subjecting the cellulose to the action of alkaline solution of strength between 6 and 11 per cent (NaOH) and then 10 to the action of carbon bisulphide at the ordinary temperature, and finally washing the

between 6 and 11 per cent (NaOH) and product with a strong solution of a neutral

perature, and finally washing the product 5. As a new product, hydrated fibrous celwith a strong solution of a neutral alkali lulose having a thickened cell wall and an 15 exaggerated central canal defined by sym-

In testimony whereof we have signed our

names to this specification.

EDWARD J. BEVAN.