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(54) **IODINE COMPOSITIONS**

(71) Applicant: **DERMALIQ THERAPEUTICS, INC.**, Wilmington, DE (US)

(72) Inventors: **Frank Löscher**, Schriesheim (DE);
Ralf Grillenberger, Ilvesheim (DE)

(73) Assignee: **DERMALIQ THERAPEUTICS, INC.**, Wilmington, DE (US)

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(58) **Field of Classification Search**

None

See application file for complete search history.

(56)

References Cited

U.S. PATENT DOCUMENTS

2,616,927 A	11/1952	Kauck et al.
4,452,818 A	6/1984	Haidt
5,077,036 A	12/1991	Long
5,126,127 A	6/1992	Bhagwat et al.
5,254,338 A	10/1993	Sakai et al.
5,326,566 A	7/1994	Parab
5,336,175 A	8/1994	Mames
5,370,313 A	12/1994	Beard
5,518,731 A	5/1996	Meadows
5,667,809 A	9/1997	Trevino et al.
5,849,291 A	12/1998	Kessler
5,851,544 A	12/1998	Penska et al.
5,874,469 A	2/1999	Maniar et al.
5,874,481 A	2/1999	Weers et al.
5,904,933 A	5/1999	Riess et al.
5,980,936 A	11/1999	Krafft et al.
5,981,607 A	11/1999	Ding et al.
6,042,845 A	3/2000	Sun et al.
6,060,085 A	5/2000	Osborne
6,113,919 A	9/2000	Reiss et al.
6,140,374 A	10/2000	May et al.
6,159,977 A	12/2000	Reeves
6,177,477 B1	1/2001	George et al.

(Continued)

FOREIGN PATENT DOCUMENTS

CN	200977281 Y	11/2007
CN	202136470 U	2/2012

(Continued)

OTHER PUBLICATIONS

Narayanan et al ("Dry Eye Disease and Microbial Keratitis: Is There a Connection?"), The Ocular Surface, vol. 11(2) (Apr. 2013), p. 75-92) (Year: 2013).*

Frentz et al., "Repeated Exposure to Benzalkonium Chloride in the Ex Vivo Eye Irritation Test (EVEIT): Observation of the Isolated Corneal Damage and Healing," Altern to Lab Anim, 2008, (36), p. 25-32.

Schrage et al., "The Ex Vivo Eye Irritation Test (EVEIT) in evaluation of artificial tears: Purite—preserved versus unpreserved eye drops," Graefes Arch Clin Exp Ophthalmol, 2012, 250(9): 1333-1340.

Bardin et al., "Long-Range Nanometer-Scale Organization of Semifluorinated Alkane Monolayers at the Air/Water Interface," Langmuir, 2011, 27, 13497-13505.

(Continued)

Primary Examiner — Sin J Lee

(74) *Attorney, Agent, or Firm* — Hoxie & Associates LLC

(57)

ABSTRACT

The present invention relates to compositions comprising molecular iodine and a vehicle comprising a semifluorinated alkane. The compositions of the invention may be used to treat or prevent diseases or conditions caused by, or associated with microorganisms.

25 Claims, No Drawings

(56)

References Cited

U.S. PATENT DOCUMENTS

6,197,323 B1	3/2001	Georgieff	2008/0112895 A1	5/2008	Kottayil et al.
6,224,887 B1	5/2001	Samour et al.	2008/0153909 A1	6/2008	Dana et al.
6,262,105 B1	7/2001	Johnstone	2008/0207537 A1	8/2008	Turner et al.
6,262,126 B1	7/2001	Meinert	2008/0234389 A1	9/2008	Mecozzi et al.
6,294,563 B1	9/2001	Garst	2008/0254106 A1	10/2008	Bell
6,335,335 B2	1/2002	Higashiyama	2008/0260656 A1	10/2008	Mallard
6,372,243 B2	4/2002	Kobuch	2009/0104243 A1	4/2009	Utkhede et al.
6,391,879 B1	5/2002	Reeves	2009/0169601 A1	7/2009	Koch et al.
6,399,087 B1	6/2002	Zhang et al.	2009/0226875 A1	9/2009	Meinert et al.
6,458,376 B1	10/2002	Meadows	2010/0006600 A1	1/2010	Dascanio
6,486,212 B2	11/2002	Meinert	2010/0008996 A1	1/2010	Meinert
6,489,367 B1	12/2002	Meinert	2010/0016814 A1	1/2010	Gokhale et al.
6,730,328 B2	5/2004	Maskiewicz et al.	2010/0137252 A1	6/2010	Matsumura et al.
7,001,607 B1	2/2006	Menz et al.	2010/0189765 A1	7/2010	Erickson et al.
7,026,359 B1	4/2006	Gross et al.	2010/0210720 A1	8/2010	Pilotaz et al.
7,258,869 B1	8/2007	Berry et al.	2010/0226997 A1	9/2010	Bowman et al.
7,335,379 B2	2/2008	Carrara et al.	2010/0274215 A1	10/2010	Wong et al.
7,740,875 B2	6/2010	Dechow	2010/0305081 A1	12/2010	Dechow
8,029,977 B2	10/2011	Meinert et al.	2011/0223208 A1	9/2011	Hill et al.
8,222,292 B2	7/2012	Goskonda et al.	2011/0269704 A1	11/2011	Seigfried
8,470,873 B2	6/2013	Chen	2012/0010280 A1	1/2012	Aleo et al.
8,614,178 B2	12/2013	Theisinger et al.	2012/0095097 A1	4/2012	Tabuchi et al.
8,796,340 B2	8/2014	Theisinger et al.	2012/0100183 A1	4/2012	Schlessinger et al.
8,916,157 B2	12/2014	Krause et al.	2012/0219640 A1	8/2012	Wright
8,916,178 B2	12/2014	Krause	2012/0238639 A1	9/2012	Theisinger et al.
8,986,738 B2	3/2015	Meinert	2013/0011484 A1	1/2013	Bevier
9,241,900 B2	1/2016	Wilson	2013/0046014 A1	2/2013	Theisinger et al.
9,308,262 B2	4/2016	Günther et al.	2013/0084250 A1	4/2013	Hagedorn et al.
9,757,459 B2	9/2017	Günther et al.	2013/0172282 A1 *	7/2013	Parks A61K 31/365 514/450
9,757,460 B2	9/2017	Günther et al.	2013/0177522 A1	7/2013	Liang et al.
9,770,508 B2	9/2017	Günther et al.	2013/0266652 A1	10/2013	Theisinger et al.
9,968,678 B2	5/2018	Theisinger et al.	2013/0303473 A1	11/2013	Wilson
10,045,996 B2	8/2018	Theisinger et al.	2013/0336557 A1	12/2013	Cruzat et al.
10,058,615 B2	8/2018	Günther et al.	2014/0004197 A1	1/2014	Theisinger et al.
10,064,944 B2	9/2018	Wilson	2014/0100180 A1 *	4/2014	Gunther A61K 31/436 514/29
10,130,707 B2	11/2018	Günther et al.	2014/0140942 A1	5/2014	Günther et al.
10,273,298 B2	4/2019	Günther et al.	2014/0274879 A1 *	9/2014	Song A61K 38/164 435/375
10,369,117 B2	8/2019	Günther et al.	2014/0303219 A1	10/2014	Bingaman et al.
10,449,164 B2	10/2019	Günther et al.	2014/0369993 A1	12/2014	Günther et al.
10,507,132 B2	12/2019	Graf et al.	2015/0045282 A1	2/2015	Elschly et al.
10,525,062 B2	1/2020	Theisinger et al.	2015/0099019 A1	4/2015	Johnson
10,576,154 B2	3/2020	Günther et al.	2015/0224064 A1 *	8/2015	Gunther A61K 31/02 514/757
10,682,315 B2	6/2020	Scherer et al.	2015/0238605 A1	8/2015	Günther et al.
10,813,976 B2	10/2020	Löscher et al.	2015/0258040 A1	9/2015	Lynch et al.
10,813,999 B2	10/2020	Günther et al.	2016/0000941 A1	1/2016	Keller et al.
11,154,513 B2	10/2021	Scherer et al.	2016/0101178 A1	4/2016	Wilson
11,160,865 B2	11/2021	Theisinger et al.	2016/0159902 A1	6/2016	Günther et al.
11,273,174 B2 *	3/2022	Löscher A61K 9/0048	2016/0184259 A1	6/2016	Anastassov et al.
11,357,738 B2	6/2022	Scherer et al.	2016/0243189 A1	8/2016	Gu et al.
11,510,855 B2	11/2022	Löscher et al.	2017/0020726 A1	1/2017	Labombarbe et al.
2002/0004063 A1	1/2002	Zhang	2017/0044096 A1	2/2017	Laskin et al.
2002/0006442 A1	1/2002	Mishra et al.	2017/0065647 A1 *	3/2017	Kim A61K 35/747
2002/0064565 A1	5/2002	Karagoezian	2017/0087100 A1	3/2017	Scherer et al.
2002/0128527 A1	9/2002	Meinert	2017/0087101 A1	3/2017	Scherer et al.
2003/0018044 A1	1/2003	Peyman	2017/0143832 A1	5/2017	Günther et al.
2003/0027833 A1	2/2003	Cleary et al.	2017/0182060 A1	6/2017	Wiedersberg et al.
2003/0170194 A1	9/2003	Piotrowiak	2017/0224531 A1	8/2017	Chauhan et al.
2003/0194447 A1	10/2003	Scholz et al.	2018/0021434 A1	1/2018	Günther et al.
2004/0101551 A1	5/2004	Selzer	2018/0360908 A1	12/2018	Beier et al.
2004/0265362 A1	12/2004	Susilo	2019/0274970 A1	9/2019	Günther et al.
2004/0266702 A1	12/2004	Dawson et al.	2019/0328717 A1	10/2019	Günther et al.
2005/0075407 A1	4/2005	Tamarkin et al.	2019/0343793 A1	11/2019	Günther et al.
2005/0079210 A1	4/2005	Gupta	2020/0060987 A1	2/2020	Günther et al.
2005/0084553 A1	4/2005	Moon et al.	2020/0129543 A1	4/2020	Löscher et al.
2005/0175541 A1	8/2005	Lanza et al.	2020/0188318 A1	6/2020	Günther et al.
2005/0274744 A1	12/2005	Spada et al.	2020/0206241 A1	7/2020	Theisinger et al.
2005/0288196 A1	12/2005	Horn	2020/0246463 A1	8/2020	Günther et al.
2006/0013820 A1	1/2006	Bonnet et al.	2020/0268648 A1	8/2020	Günther et al.
2006/0078577 A1	4/2006	Dechow	2020/0268682 A1	8/2020	Günther et al.
2006/0078580 A1	4/2006	Dechow	2020/0338015 A1	10/2020	Scherer et al.
2006/0153905 A1	7/2006	Carrara et al.	2021/0023166 A1	1/2021	Löscher et al.
2007/0249730 A1	10/2007	Daftary et al.	2021/0069014 A1	3/2021	Löscher et al.
2008/0019926 A1	1/2008	Krafft et al.	2021/0100904 A1	4/2021	Günther et al.
2008/0050335 A1	2/2008	Faour et al.	2021/0106558 A1	4/2021	Löscher et al.
2008/0089923 A1	4/2008	Burkstrand et al.	2021/0121471 A1	4/2021	Löscher et al.

(56)

References Cited

U.S. PATENT DOCUMENTS

2021/0228595 A1 7/2021 Löscher et al.
 2021/0236591 A1 8/2021 Leo et al.
 2021/0308024 A1 10/2021 Löscher et al.
 2021/0315832 A1 10/2021 Scherer et al.
 2021/0346313 A1 11/2021 Beier et al.
 2022/0008397 A1 1/2022 Xu et al.
 2022/0031844 A1 2/2022 Mauden et al.
 2022/0079980 A1 3/2022 Löscher et al.
 2022/0152096 A1 5/2022 Löscher et al.
 2022/0218828 A1 7/2022 Günther et al.

FOREIGN PATENT DOCUMENTS

CN 203524843 U 4/2014
 EP 0 670 159 9/1995
 EP 1 152 749 4/2006
 JP S5721312 A 2/1982
 JP S6452722 2/1989
 JP H0764702 B2 7/1995
 JP 2001/158734 6/2001
 JP 2011/006348 1/2011
 JP 2011/024841 A 2/2011
 WO WO 93/00991 1/1993
 WO WO 1998/005301 12/1998
 WO WO 00/54588 9/2000
 WO WO 2010/146536 12/2010
 WO WO 2012/007776 1/2012

OTHER PUBLICATIONS

Bertilla et al., "Semifluorinated Alkanes as Stabilizing Agents of Fluorocarbon Emulsions," Springer, Tokyo, 2005, International Symposia for Life Sciences and Medicine, vol. 12, pp. 237-251.
 Broniatowski, M. et al., "Langmuir Monolayers Characteristic of (Perfluorodecyl)-Alkanes," Journal of Physical Chemistry B, 2004, 108, 13403-13411.
 Costa Gomes et al., "Solubility of dioxygen in seven fluorinated liquids," Journal of Fluorine Chemistry, 2004, 125, 1325-1329.
 Davies, "Biopharmaceutical Considerations in Topical Ocular Drug Delivery," Clin. Exper. Pharmacol. Physiol., 2000, 27:558-562.
 Dembinski, R. et al., "Semi-fluorinated Alkanes as Carriers for Drug Targeting in Acute Respiratory Failure," Experimental Lung Research, 2010, 36, 499, 507.
 Deschamps, J. et al., "Solubility of oxygen, carbon dioxide and water in semifluorinated alkanes and in perfluorooctylbromide by molecular simulation", Journal of Fluorine Chemistry, Elsevier, vol. 125, No. 3, 2004.
 Dias et al., "Solubility of oxygen in liquid perfluorocarbons," Fluid Phase Equilibria, 2004, 222-223:325-330.
 "Does Iodine Dissolve in Ethanol?"—an internet article obtained from the webpage <https://www.enotes.com/homework-help/does-iodine-dissolve-ethanol-557516> (dated Nov. 19, 2015) (Year: 2015).
 Dutescu et al., "Semifluorinated alkanes as a liquid drug carrier system for topical ocular drug delivery," European Journal of Pharmaceutics and Biopharmaceutics, 2014, 88(1):123-128, Abstract Only (2 pages).
 English-language machine translation of EP0670159 (A1) issued in U.S. Appl. No. 14/122,025 on Apr. 1, 2015, 10 pages.
 Freiburger Dokumentenserver (FreiDok), Albert-Ludwigs-Universität Freiburg, retrieved from <http://www.freidok.uni-freiburg.de/volltexte/5682/>, retrieved on Feb. 5, 2014, 2 pages.
 Gehlsen, U., et al., "Cyclosporine A using F4H5 as liquid drug carrier is effective in treating experimental dry-eye disease," Investigative Ophthalmology & Visual Science, 2015, 56(7):319, Abstract Only (2 pages).
 Gehlsen et al., "A semifluorinated alkane (F4H5) as novel carrier for cyclosporine A: a promising therapeutic and prophylactic option for topical treatment of dry eye," Graefes Arch. Clin. Exp. Ophthalmol., (2017) 255(4):767-775.

German, E.J., et al., "Reality of drop size from multi-dose eye drop bottles: is it cause for concern?" Eye, 1999, 13:93-100.
 Griffin, W., "Classification of Surface-Active Agnets by 'HLB'," Journal of the Society of Cosmetic Chemists, 1949, 1:311-326.
 Hardung, H., "Semifluorierte und perfluorierte Verbindungen zur topischen und parenteralen Anwendung," 2008, retrieved from http://www.freidok.uni-freiburg.de/volltexte/5682/pdf/Dissertation_Hardung.pdf (retrieved on Oct. 10, 2011).
 Hardung, H., "Semifluorierte und perfluorierte Verbindungen zur topischen und parenteralen Anwendung," 2008, English Language Abstract, 2 pages, retrieved from <https://freidok.uni-freiburg.de/data/5682> (retrieved on Jul. 10, 2017).
 Hoerauf, H. et al., "Combined Use of Partially Fluorinated Alkanes, Perfluorocarbon Liquids and Silicone Oil: An Experimental Study," Graefes Archive for Clinical and Experimental Ophthalmology, 2001, 239 (5), 373-381.
 Holm, R. et al., "A novel excipient, 1-perfluorohexyloctane shows limited utility for the oral delivery of poorly water-soluble drugs," European Journal of Pharmaceutical Sciences, 2011, 42: 416-422.
 JPS5721312A, Green Cross Corp, "Breathable Ointment," Apr. 2, 1982, English language machine translation of abstract, Espacenet, date obtained: Apr. 30, 2021, 1 page <<https://worldwide.espacenet.com/patent/search/family/014132731/publications/JPS5721312A?q=JPS5721312A>>.
 JPH0764702B2, Kanebo LTD, "Cosmetic of Polyphasic Emulsion Type," Jul. 12, 1995, English language machine translation of abstract, Espacenet, date obtained: Apr. 30, 2021, 1 page <<https://worldwide.espacenet.com/patent/search/family/014142733/publication/JPH0764702B2?q=JPH0764702B2>>.
 JPS6452722, English Machine Translation of the Abstract, Description, and Claims, Espacenet, Date Accessed: Feb. 10, 2016, 4 pages.
 Kaercher et al., "NovaTears® as new Therapy in Dry Eye Results from three prospective, multicenter, non-interventional studies in different patient populations," TFOS Conference (Tear Film & Ocular Surface), Sep. 7-10, 2016, Montpellier, France, Poster Session II, Poster No. 60, 1 page.
 Kociok, N., et al, "Influence on Membrane-Mediated Cell Activation by Vesicles of Silicone Oil or Perfluorohexyloctane," Graefes Archive for Clinical and Experimental Ophthalmology, 2005, 243, 345-358.
 Matteucci et al., "Biocompatibility assessment of liquid artificial vitreous replacements: relevance of in vitro studies," Survey of Ophthalmology, 2007, 52(3):289-299, Abstract Only (1 page).
 Meinert, H. et al., "The Use of Semifluorinated Alkanes in Blood-Substitutes," Biomaterials, Artificial Cells, and Immobilization Biotechnology, 1993, 21 (5), 583-595.
 Meinert, H. et al., "Semifluorinated Alkanes—A New Class of Compounds with Outstanding Properties for Use in Ophthalmology," European Journal of Ophthalmology, 2000, 10 (3), 189-197.
 Messmer, et al. "Semifluorierte Alkanes als Therapie bei Meibomdrüsen-Dysfunktion Ergebnisse einer prospektiven, multizentrischen Beobachtungsstudie", Presentation, DOG—Kongress, Sep. 29-Oct. 2, 2016, Berlin DOG (Deutsche Ophthalmologische Gesellschaft), Poster No. PSA03-02, 1 page (German language version).
 Messmer, et al. "Semifluorinated Alkanes as a Therapy for Meibomian Gland Dysfunction Results of a prospective, multi-centered observational study", Presentation, DOG—Kongress, Sep. 29, 2016-Oct. 2, 2016, Berlin DOG (Deutsche Ophthalmologische Gesellschaft), Poster No. PSA03-02, English Translation, 6 pages.
 O'Rourke, M. et al., "Enhancing Delivery of Topical Ocular Drops," Cataract & Refractive Surgery Today Europe, 2016, 2 pages.
 Plassmann, M. et al., "Trace Analytical Methods for Semifluorinated n-Alkanes in Snow, Soil, and Air," Analytical Chemistry, 2010, 82(11), 4551-4557.
 Plassmann, M. et al., "Theoretical and Experimental Simulation of the Fate of Semifluorinated n-Alkanes During Snowmelt," Environmental Science & Technology, 2010, 44 (17), 6692-6697.
 Schmutz et al., "Fluorinated Vesicles Made from Combinations of Phospholipids and Semifluorinated Alkanes. Direct Experimental Evidence of the Location of the Semifluorinated Alkane within the Bilayer", Langmuir, 2003, 19:4889-4894.

(56)

References Cited

OTHER PUBLICATIONS

Steven, P. et al. "Semifluorinated Alkane Eye Drops for Treatment of Dry Eye Disease—A Prospective, Multicenter Noninterventional Study" *Journal of Ocular Pharmacology and Therapeutics* (2015) vol. 31(8):498-503.

Steven, P. et al. "Semifluorinated Alkane Eye Drops for Treatment of Dry Eye Disease—A Prospective, Multicenter Noninterventional Study," *Investigative Ophthalmology & Visual Science*, 2015, 56:4493, Abstract Only (1 page).

Steven et al., "Semifluorinated Alkane Eye Drops for Treatment of Dry Eye Disease Due to Meibomian Gland Disease," *Journal of Ocular Pharmacology and Therapeutics*, 2017, 33(9):1-8.

Wong, D. et al., "Perfluorocarbons and Semifluorinated Alkanes," *Seminars in Ophthalmology*, 2000, 15 (1), 25-35.

Yaoxue Zhuanye Zhishi II (Editor: Jin Xiangqun), Military Medical Science Press, 1st Printing of 2nd Edition, Mar. 2009, p. 158.

Yaoxue Zhuanye Zhishi II (Editor: Jin Xiangqun), Military Medical Science Press, 1st Printing of 2nd Edition, Mar. 2009, p. 158, 3 pages. (English Machine Translation).

Zhang et al., "Surface micelles of semifluorinated alkanes in Langmuir-Blodgett monolayers," *Phys. Chem. Chem. Phys.*, 2004, 6, 1566-1569.

Elkeeb, R., "Transungual Drug Delivery: Current Status," *International Journal of Pharmaceutics*, 2010, 384: 1-8.

Padzik et al., "Effect of Povidone Iodine, Chlorhexidine Digluconate and Toyocamycin on Amphizoic Amoebic Strains, Infectious Agents of *Acanthamoeba keratitis*—a growing threat to human health worldwide," *Annals of Agricultural and Environmental Medicine*, vol. 25, No. 4, p. 725-731, (2018).

PubChem, *Hexamethyldisiloxane* (Chemical and Physical Properties), retrieved from <https://pubchem.ncbi.nlm.nih.gov/compound/24764> (last accessed May 8, 2023).

PubChem, *Octamethyltrisiloxane* (Chemical and Physical Properties), retrieved from <https://pubchem.ncbi.nlm.nih.gov/compound/24705> (last accessed May 8, 2023).

Cooper, R., "Iodine Revisited," *International Wound Journal*, 4(2):124-137, 124 (2007).

Eggers, M., "Infectious Disease Management and Control with Povidone Iodine," *Infect. Dis. Ther.* 8:581-593 (2019).

McDonnell, G., & Russell, A.D., "Antiseptics and Disinfectants: Activity, Action, and Resistance." *Clinical Microbiology Reviews*, 12(1):147-179, 155 (1999).

Lemke, T.L., "Antiseptics and Disinfectants," Chapter 36, Principles of Medicinal Chemistry, 818-819 (Foye, W.O et al., Eds., 1995).

Sherris, J.C. & Plorde, J.J., "Sterilization, Pasteurization, Disinfection, Sanitization, and Asepsis," Chapter 11 in, *Medical Microbiology: An Introduction to Infectious Diseases*, 177 (Sherris, J.C., Ed., Elsevier 1990).

* cited by examiner

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IODINE COMPOSITIONS

CROSS-REFERENCE TO RELATED APPLICATIONS

This application is a continuation of U.S. application Ser. No. 16/606,936, filed on Oct. 21, 2019, which is a U.S. national stage application filed under 35 U.S.C. § 371 of International Application No. PCT/EP2018/060197, filed on Apr. 20, 2018, which claims priority to, and the benefit of, European Application No. 17167561.4, filed Apr. 21, 2017, the contents of each of which are incorporated herein by reference in their entireties.

BACKGROUND OF THE INVENTION

Iodine compositions are useful as topical antiseptics and disinfectants. Tinctures of iodine are compositions based on a hydroalcoholic solution of elemental iodine, iodide salt in ethanol and water which have long been used for such purposes. Such tinctures are however highly irritating and toxic, and are thus not suitable for administration, for example, to sensitive tissues such as the eye. Many iodine compositions in the art are based alternatively on povidone iodine, also known PVP-iodine or polyvinylpyrrolidone-iodine, as a source and reservoir of free molecular iodine, and to improve the solubility of iodine in the aqueous environment.

Examples of ophthalmic compositions comprising povidone-iodine are described for example in EP0526695A1 and WO2011084473A1. Such compositions rely on extensive formulation, requiring the addition of pH buffers or adjusting agents, stabilizing agents, and other excipients to provide stability and adequate shelf-life.

Iodine compositions based on non-aqueous medium, such as organic solvents have also previously been disclosed. For example, US 2012/0219640 discloses iodine compositions for use in treating topical skin conditions such as athlete's foot infection, comprising organic solvents such as ethanol, iso-propyl alcohol and acetone. Such solvents are not universally applicable in terms of therapeutic use due to tissue toxicity or poor biocompatibility. WO 2012/007776 A2 proposes the use of a siloxane vehicle, to prepare disinfectant and antiseptic solutions of iodine. Siloxanes however may tend to be viscous and for topical applications, may feel greasy and have limited spreadability. They also tend to have refractive indexes which are differentiated from that of human tear fluid and thus may not be suitable or patient-friendly for ophthalmic applications.

It is an object of the present invention, to provide iodine compositions which may avoid some of the disadvantages of the prior art, and which may be useful as an antiseptic or a disinfectant, such as in the treatment or prophylaxis of diseases and conditions caused or associated with microorganisms. Further objects of the invention will be clear on the basis of the following description of the invention, examples and claims.

SUMMARY OF THE INVENTION

In a first aspect, the invention relates to a composition comprising molecular iodine and a vehicle comprising a semifluorinated alkane. In some embodiments, the vehicle comprises a semifluorinated alkane is of formula (II): $\text{CF}_3(\text{CF}_2)_n(\text{CH}_2)_m\text{CH}_3$ wherein n and m are each independently selected from an integer from 3 to 10. In another aspect, the composition may be essentially free of water.

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In a further aspect, the invention provides for a composition comprising molecular iodine and a vehicle comprising a semifluorinated alkane, for use as a medicament, preferably for the treatment or prevention of a disease or condition of the skin or mucosal tissue; or the eye or an ophthalmic tissue; preferably wherein the disease or condition is associated with, or caused by a microorganism.

DETAILED DESCRIPTION OF THE INVENTION

The present invention relates to a composition comprising molecular iodine and a vehicle comprising a semifluorinated alkane.

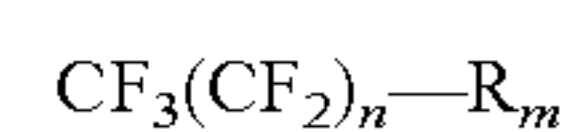
Molecular iodine, which may also be referred to as elemental iodine, has a chemical formula of I_2 , and is metallic grey/dark coloured solid with a molecular weight of 253.81 g/mol. Molecular iodine has been found to be effective in destroying or inhibiting the growth/proliferation of microorganisms such as bacteria, fungi, demodex and viruses.

The vehicle of the composition according to invention serves as a carrier for molecular iodine, and may be in the form of a liquid, semi-solid or solid. Preferably, the molecular iodine is dissolved and/or fully miscible in the vehicle i.e. the composition of the invention does not comprise of any particulate or solid phase iodine, preferably at room temperature conditions i.e. between 15 and 25° C.

In one preferred embodiment, the vehicle is a liquid. Preferably, the molecular iodine is dissolved in the liquid vehicle to form a clear solution. The term "clear solution" or "solution" as such, as understood herein, refers to a liquid solution in which all solutes are fully dissolvable or dissolved under room temperature conditions i.e. between 15 and 25° C. The resulting solution does not comprise of any particulate or solid phase components. In the present invention, the molecular iodine is preferably completely dissolved in the liquid vehicle comprising a semifluorinated alkane, which, depending on degree of dilution, may provide clear, purple to pink coloured solutions.

The composition according to invention comprises a semifluorinated alkane (which may also be referred to in abbreviated form as an SFA), which may be defined as a linear or branched compound composed of at least one perfluorinated segment (F-segment) and at least one non-fluorinated hydrocarbon segment (H-segment). Preferably, the semifluorinated alkane is a linear compound composed of one perfluorinated segment (F-segment) and one non-fluorinated hydrocarbon segment (H-segment). In a branched semifluorinated alkane, the semifluorinated alkane may comprise of a branched non-fluorinated hydrocarbon segment, wherein the hydrocarbon segment comprises as a substituent, one or more of an alkyl group selected from the group consisting of $-\text{CH}_3$, $-\text{C}_2\text{H}_5$, $-\text{C}_3\text{H}_7$ and $-\text{C}_4\text{H}_9$.

In particular, the present invention relates to composition comprising molecular iodine and a vehicle comprising a semifluorinated alkane, wherein the semifluorinated alkane is of formula



Formula (I)

wherein n is an integer selected from 1 to 12, and R is linear or branched alkyl or cycloalkyl with m carbon atoms, wherein m is an integer selected from 2 to 12.

In one preferred embodiment, said vehicle may comprise of a semifluorinated alkane of Formula (I), wherein n is an

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integer selected from 3 to 10, and R is linear or branched alkyl or cycloalkyl with m carbon atoms, wherein m is an integer selected from 4 to 8.

In the above embodiments, where R is preferably a branched alkyl, R may be selected from iso-propyl, any one of the butyl isomers, such as iso-butyl, sec-butyl, or tert-butyl, any one of the pentyl isomers such as neo-pentyl, tert-pentyl, sec-pentyl, iso-pentyl, or 3-pentyl; any one of the hexyl or heptyl isomers, and any one of the octyl isomers, for example such as iso-octyl.

In another preferred embodiment, said vehicle may comprise of a semifluorinated alkane of Formula (I), wherein n is an integer selected from 3 to 10, and R is linear alkyl with m carbon atoms, wherein m is an integer selected from 4 to 8.

In an alternative preferred embodiment, said vehicle may comprise of a semifluorinated alkane of Formula (I), wherein n is an integer selected from 3 to 10, and R is linear alkyl with m carbon atoms, wherein m is an integer selected from 2 to 8.

As understood herein, the term linear alkyl may refer to unbranched and straight-chain alkyl groups. For example, R of Formula (I) may preferably be selected from ethyl, n-propyl, n-butyl, n-pentyl, n-hexyl, or n-heptyl, and n-octyl.

In another preferred embodiment, said vehicle may comprise of a semifluorinated alkane of Formula (I), wherein n is an integer selected from 3 to 10, and R is cycloalkyl with m carbon atoms, wherein m is an integer selected from 4 to 8.

As understood herein, the term cycloalkyls may refer to cyclic, i.e., ring-forming alkyl groups, such as cyclobutyl, cyclopentyl or cyclohexyl.

In a preferred embodiment, the semifluorinated alkane featured in the compositions according to the invention is a liquid, i.e. a compound that exists in a liquid state at least at one temperature within the temperature range of 4° C. to 50° C. Preferably, the semifluorinated alkane featured in the composition according to the invention is a compound that is a liquid at room temperature (RT), such as between the temperature of 15° C. to 25° C. Alternatively, the semifluorinated alkane, or optionally, a mixture of two or more semifluorinated alkanes, featured in the composition may be semi-solid or solid at room temperature but has a melting point of at least one temperature within the range from 26° C. to 45° C.

In another particularly preferred embodiment, the present invention relates to a composition comprising molecular iodine and a vehicle comprising a semifluorinated alkane of formula (II):



wherein n and m are each independently selected from an integer from 3 to 10.

Preferably, said semifluorinated alkane may be selected from $\text{CF}_3(\text{CF}_2)_3-(\text{CH}_2)_4\text{CH}_3$ (F4H5), $\text{CF}_3(\text{CF}_2)_3-(\text{CH}_2)_5\text{CH}_3$ (F4H6), $\text{CF}_3(\text{CF}_2)_5-(\text{CH}_2)_5\text{CH}_3$ (F6H6), $\text{CF}_3(\text{CF}_2)_5-(\text{CH}_2)_7\text{CH}_3$ (F6H8), $\text{CF}_3(\text{CF}_2)_3-(\text{CH}_2)_7\text{CH}_3$ (F4H8), $\text{CF}_3(\text{CF}_2)_7-(\text{CH}_2)_7\text{CH}_3$ (F8H8) and $\text{CF}_3(\text{CF}_2)_9-(\text{CH}_2)_4\text{CH}_3$ (F10H5). More preferably, said semifluorinated alkane may be selected from $\text{CF}_3(\text{CF}_2)_3-(\text{CH}_2)_4\text{CH}_3$ (F4H5) and $\text{CF}_3(\text{CF}_2)_5-(\text{CH}_2)_7\text{CH}_3$ (F6H8). In another embodiment, the composition may comprise of molecular iodine dissolved in any one of said selected semifluorinated alkanes.

In another alternative embodiment, the present invention relates to a composition comprising molecular iodine and a vehicle comprising a semifluorinated alkane of formula (II):



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wherein n is an integer selected from 2 to 9 and m is an integer from 1 to 7.

Said semifluorinated alkane may be selected from $\text{CF}_3(\text{CF}_2)_3-(\text{CH}_2)_4\text{CH}_3$ (F4H5), $\text{CF}_3(\text{CF}_2)_3-(\text{CH}_2)_5\text{CH}_3$ (F4H6), $\text{CF}_3(\text{CF}_2)_5-(\text{CH}_2)_5\text{CH}_3$ (F6H2), $\text{CF}_3(\text{CF}_2)_5-(\text{CH}_2)_7\text{CH}_3$ (F6H6), $\text{CF}_3(\text{CF}_2)_5-(\text{CH}_2)_7\text{CH}_3$ (F6H8), $\text{CF}_3(\text{CF}_2)_3-(\text{CH}_2)_7\text{CH}_3$ (F4H8), $\text{CF}_3(\text{CF}_2)_7-(\text{CH}_2)_7\text{CH}_3$ (F8H8) and $\text{CF}_3(\text{CF}_2)_9-(\text{CH}_2)_4\text{CH}_3$ (F10H5). More preferably, said semifluorinated alkane may be selected from $\text{CF}_3(\text{CF}_2)_5-(\text{CH}_2)_5\text{CH}_3$ (F6H2), $\text{CF}_3(\text{CF}_2)_3-(\text{CH}_2)_4\text{CH}_3$ (F4H5) and $\text{CF}_3(\text{CF}_2)_5-(\text{CH}_2)_7\text{CH}_3$ (F6H8). In another embodiment, the composition may comprise of molecular iodine dissolved in any one of said selected semifluorinated alkanes.

In another particularly preferred embodiment, the present invention relates to a composition comprising molecular iodine and a vehicle comprising a semifluorinated alkane of formula (II):



wherein n and m are each independently selected from an integer from 3 to 10, and wherein the composition is essentially free of water. In such embodiment, preferably, the semifluorinated alkane may be selected from $\text{CF}_3(\text{CF}_2)_3-(\text{CH}_2)_4\text{CH}_3$ (F4H5), $\text{CF}_3(\text{CF}_2)_3-(\text{CH}_2)_5\text{CH}_3$ (F4H6), $\text{CF}_3(\text{CF}_2)_5-(\text{CH}_2)_5\text{CH}_3$ (F6H6), $\text{CF}_3(\text{CF}_2)_5-(\text{CH}_2)_7\text{CH}_3$ (F6H8), $\text{CF}_3(\text{CF}_2)_3-(\text{CH}_2)_7\text{CH}_3$ (F4H8), $\text{CF}_3(\text{CF}_2)_7-(\text{CH}_2)_7\text{CH}_3$ (F8H8) and $\text{CF}_3(\text{CF}_2)_9-(\text{CH}_2)_4\text{CH}_3$ (F10H5).

In yet a further embodiment, the present invention relates to a composition comprising molecular iodine and a vehicle comprising a semifluorinated alkane of formula (II):



wherein n is an integer selected from 2 to 9 and m is an integer from 1 to 7, and wherein the composition is essentially free of water. In such embodiment, preferably, the semifluorinated alkane may be selected from $\text{CF}_3(\text{CF}_2)_3-(\text{CH}_2)_4\text{CH}_3$ (F4H5), $\text{CF}_3(\text{CF}_2)_3-(\text{CH}_2)_5\text{CH}_3$ (F4H6), $\text{CF}_3(\text{CF}_2)_5-(\text{CH}_2)_5\text{CH}_3$ (F6H2), $\text{CF}_3(\text{CF}_2)_5-(\text{CH}_2)_5\text{CH}_3$ (F6H6), $\text{CF}_3(\text{CF}_2)_5-(\text{CH}_2)_7\text{CH}_3$ (F6H8), $\text{CF}_3(\text{CF}_2)_3-(\text{CH}_2)_7\text{CH}_3$ (F4H8), $\text{CF}_3(\text{CF}_2)_7-(\text{CH}_2)_7\text{CH}_3$ (F8H8) and $\text{CF}_3(\text{CF}_2)_9-(\text{CH}_2)_4\text{CH}_3$ (F10H5).

The alternative nomenclature for the specified semifluorinated alkanes as noted in parentheses above and as may be further used herein, is based on the general formula F_nH_m , wherein F means the linear perfluorinated hydrocarbon segment, H means the linear non-fluorinated hydrocarbon segment and n, m is the number of carbon atoms of the respective segment. For example, F4H5 may be used to denote 1-perfluorobutyl pentane or $\text{CF}_3(\text{CF}_2)_3-(\text{CH}_2)_4\text{CH}_3$ (which may be also, alternatively expressed as formula $\text{F}(\text{CF}_2)_4(\text{CH}_2)_5\text{H}$), which has a linear perfluorinated segment F with four carbons (n=4) and a linear non-fluorinated hydrocarbon segment with 5 carbons (m=5).

Most preferably, the compositions of the invention comprise of molecular iodine and a liquid vehicle comprising, or consisting of a semifluorinated alkane selected from $\text{F}(\text{CF}_2)_6(\text{CH}_2)_2\text{H}$ (F6H2), $\text{F}(\text{CF}_2)_4(\text{CH}_2)_5\text{H}$ (F4H5) and $\text{F}(\text{CF}_2)_6(\text{CH}_2)_8\text{H}$ (F6H8). The particular preferred semifluorinated alkane, F6H2, also known as perfluorohexyl-ethane, and which has the chemical formula $\text{F}(\text{CF}_2)_6(\text{CH}_2)_2\text{H}$ is a chemically and physiologically inert, water-insoluble liquid, with a density of 1.57 g/cm³ at 20° C. and refractive index of 1.295 at 20° C. The particular preferred semifluorinated alkane, F4H5, also known as 1-perfluorobutyl-pentane, and which has the chemical formula $\text{F}(\text{CF}_2)_4(\text{CH}_2)_5\text{H}$ is a chemically and physiologically inert, water-insoluble liquid,

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with a density of 1.284 g/cm³ at 25° C. and refractive index of 1.3204 at 20° C. The particularly preferred semifluorinated alkane, F6H8, comprised by the liquid vehicle of the composition of the present invention also known as 1-perfluorohexyl-octane, is also a chemically and physiologically inert, water-insoluble liquid, and has a density of 1.35 g/cm³ at 25° C. and refractive index of 1.3432 at 20° C. In a preferred embodiment, said compositions may be substantially free of water.

The vehicle of the compositions of the invention comprising of “a” semifluorinated alkane is to be understood herein, as comprising of at least one semifluorinated alkane, preferably of Formula (I) or (II) or any of the other formulas as described above. Optionally, however, the vehicle of the composition may comprise of more than one, for example, a mixture of two or more semifluorinated alkanes of Formula (I) or (II) or any one of the semifluorinated alkane species as described above.

In yet further embodiment, the vehicle of the composition may consist of a semifluorinated alkane of Formula (I) or (II) or any one of the semifluorinated alkanes as specified above. In this context, the term ‘a’ semifluorinated alkane is to be understood as at least one semifluorinated alkane, but may also include the option of more than one, or a plurality of semifluorinated alkane compounds. Accordingly, in one embodiment, the vehicle of the composition may consist of more than one semifluorinated alkane of Formula (I) or (II) or any one of the semifluorinated alkanes as specified above. Preferably, said semifluorinated alkane (or optionally, mixture of two or more semifluorinated alkanes) is present and functionally featured in the composition as a vehicle for the molecular iodine. In an optional embodiment, said compositions may however include a further active ingredient and/or one or more excipients. In an embodiment of the invention, the composition may have a vehicle consisting of a semifluorinated alkane or a mixture of semifluorinated alkanes as defined and optionally, one or more excipients, or a co-solvent.

In a preferred embodiment, the composition of the present invention may consist of molecular iodine, and optionally a further active ingredient, and a vehicle comprising a semifluorinated alkane. In a further preferred embodiment, the composition may consist of molecular iodine, and optionally a further active ingredient, and a vehicle consisting of one or more semifluorinated alkanes and optionally one or more non-polar co-solvents or excipients. Optionally, the composition may consist of molecular iodine and a vehicle consisting of one or more semifluorinated alkanes. Optionally, the composition may consist of molecular iodine and a vehicle consisting of a semifluorinated alkane.

In one embodiment, the invention relates to a composition consisting of a) molecular iodine and optionally, a further active ingredient, and b) a vehicle consisting of a semifluorinated alkane and optionally a non-polar co-solvent or excipient; wherein the semifluorinated alkane is of formula (I) $\text{CF}_3(\text{CF}_2)_n\text{—R}_m$ wherein n is an integer selected from 1 to 12, and R is linear or branched alkyl or cycloalkyl with m carbon atoms, wherein m is an integer selected from 2 to 12.

In another embodiment, the invention relates to a composition consisting of a) molecular iodine and optionally, a further active ingredient, and b) a vehicle consisting of a semifluorinated alkane and optionally a non-polar co-solvent or excipient; wherein the semifluorinated alkane is of formula (I) $\text{CF}_3(\text{CF}_2)_n\text{—R}_m$ wherein n is an integer selected

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from 2 to 12, and R is linear or branched alkyl or cycloalkyl with m carbon atoms, wherein m is an integer selected from 2 to 12.

As used herein, the term “consists” and related terms “consisting” or “consist” is to be understood as meaning that no other features, other than those prefaced by the term are present. In the context of compositions, if any other constituent or component is present in the composition other than those prefaced by such term, then it is present only in trace or residual amounts such as to confer no technical advantage or relevance in respect of the object of the invention, such as may be further understood by the term ‘essentially’ or ‘substantially’ used in conjunction with these terms (e.g. ‘essentially consisting of’). In contrast, the term ‘comprising’ or related terms “comprises” or “comprise” in the context of compositions, is to be understood as meaning that other features, other than those prefaced by the term may be present in the composition.

In an alternative aspect of the invention, the compositions of the invention may comprise of molecular iodine (I₂) and a vehicle comprising a perfluorocarbon. As understood herein, a perfluorocarbon is a hydrocarbon compound wherein all hydrogen atoms are replaced by another atom, namely predominantly or entirely by fluorine atoms. Preferably, the perfluorocarbon or perfluorinated hydrocarbon has 5 to 12 carbon atoms. Preferably, the perfluorocarbon is a liquid at room temperature. In another embodiment, the perfluorocarbon may be compound of formula $\text{F}(\text{CF}_2)_n\text{F}$, wherein n is an integer selected from 5 to 12. Particularly preferred perfluorocarbons are perfluorooctane, and perfluorodecalin. Optionally, said compositions may comprise also a semifluorinated alkane, preferably of Formula (I) or (II) as described above.

Preferably, the amount of iodine incorporated in the composition according to the invention, when used in a therapeutic or prophylactic application, is present in an amount effective to inhibit or prevent growth and/or proliferation of a microorganism. In another aspect, the molecular iodine may preferably be incorporated in the composition in an amount that is useful for producing a desired pharmacological effect, such as a pharmacological effect in the treatment of any one of the conditions or diseases as will be further described in detail below.

In one aspect, a composition of the invention such as described in any of the above embodiments, comprises an amount of molecular iodine of less than 1 mg/mL, or preferably less than 0.9 mg/mL, 0.8 mg/mL, 0.6 mg/mL, 0.5 mg/mL, 0.4 mg/mL, 0.2 mg/mL, 0.1 mg/mL, 0.05 mg/mL, 0.01 mg/mL or less than 0.001 mg/mL.

In one embodiment, the composition of the invention may comprise of less than 1 mg/mL of molecular iodine dissolved in the vehicle.

In another embodiment, the composition of the invention may comprise or consist of an amount of molecular iodine of between 0.001 mg/mL and 1 mg/mL, or between 0.001 mg/mL and any one of the values of 0.9 mg/mL, 0.8 mg/mL, 0.6 mg/mL, 0.5 mg/mL, 0.4 mg/mL, 0.2 mg/mL, 0.1 mg/mL, 0.05 mg/mL, 0.01 mg/mL, dissolved in the vehicle, preferably a vehicle consisting of a semifluorinated alkane as defined in any one of the above embodiments, and optionally a compatible e.g. non-polar, co-solvent or an excipient.

In one embodiment, the composition may comprise less than 1 mg/mL, or preferably less than 0.9 mg/mL, 0.8 mg/mL, 0.6 mg/mL, 0.5 mg/mL, 0.4 mg/mL, 0.2 mg/mL, 0.1 mg/mL, 0.05 mg/mL, 0.01 mg/mL or less than 0.001 mg/mL of molecular iodine dissolved in the vehicle. In an

embodiment where the vehicle may consist of only a semifluorinated alkane, the composition may comprise less than 1 mg/mL, or less than 0.9 mg/mL, 0.8 mg/mL, 0.6 mg/mL, 0.5 mg/mL, 0.4 mg/mL, 0.2 mg/mL, 0.1 mg/mL, 0.05 mg/mL, 0.01 mg/mL or less than 0.001 mg/mL of molecular iodine dissolved in said semifluorinated alkane vehicle.

In another embodiment where the vehicle may consist of only a semifluorinated alkane and optionally a non-polar and/or aprotic solvent, or an excipient, the composition may comprise less than 1 mg/mL, or less than 0.9 mg/mL, 0.8 mg/mL, 0.6 mg/mL, 0.5 mg/mL, 0.4 mg/mL, 0.2 mg/mL, 0.1 mg/mL, 0.05 mg/mL, 0.01 mg/mL or less than 0.001 mg/mL of molecular iodine dissolved in said vehicle comprising of a semifluorinated alkane and optional co-solvent(s). Preferably, such compositions are non-staining and/or stable for at least 30, 90 or 180 days at room temperature.

According to an aspect of the invention, the composition may comprise of molecular iodine and a vehicle comprising of a semifluorinated alkane, such as described in any of the above embodiments, wherein the vehicle also further comprises a co-solvent.

As understood herein, a co-solvent is a compound which is suitable for enhancing solubility or solubilizing molecular iodine and/or any other solid or non-miscible component featured in the composition of the invention. Said solid or non-miscible component may be, for example a further active ingredient, or another excipient. Preferably, the co-solvent component of the composition is a liquid that is fully miscible with the semifluorinated alkane, i.e. it mixes with the semifluorinated alkane to form a coherent and single phase, to serve as a vehicle and carrier for molecular iodine and/or any other component, such as a further active ingredient or a further excipient. The co-solvent preferably is a liquid organic compound which is physiologically tolerated and safe for direct topical administration to the eye, eye tissue and/or to skin, dermal, or mucosal tissue.

Preferably, the co-solvent is a non-polar solvent. The non-polar solvent may be non-ionizable, and/or aprotic, preferably such co-solvent does not support the formation of an iodine ion or oxidized iodine species, preferably the co-solvent is a solvent that does not support the formation as of any one (or combination of) the following: iodide (I⁻), triiodide (I₃⁻), iodate (IO₃⁻), or hypoiodate (IO⁻).

Particularly preferred as a co-solvent is a non-polar solvent selected from any one or mixture of a saturated hydrocarbon, preferably a C1-C10 alkane (such as n-pentane, n-hexane, n-heptane, or n-octane); mineral oil; paraffin; siloxane; and a perfluorocarbon. In further embodiment, the non-polar solvent may also be selected from saturated, unbranched or branched hydrocarbons with longer carbon chains such as C10 to C30, or C10 to C40.

In a particular preferred embodiment, the co-solvent is a solvent that does not support the formation of an iodine ion or oxidized iodine species, preferably the co-solvent is a solvent that does not support the formation of any one (or combination of) the following: iodide (I⁻), triiodide (I₃⁻), iodate (IO₃⁻), or hypoiodate (IO⁻).

The co-solvent may be featured in a composition according to the invention in an amount of no more than about 38 wt % based on the total weight of the vehicle. In a further embodiment, the amount of co-solvent may be in an amount of no more than 30% (w/w), 25% (w/w), 15% (w/w) or 10% (w/w), or 5% (w/w) 2% (w/w), or about 1.0% (w/w) in respect to the total amount of the vehicle.

In further embodiments, the co-solvent may be featured in an amount in the range of between 0.001 to 5% (w/w), or

0.001 to 2% (w/w), or 0.01 to 1% (w/w), or 0.1 to 1% (w/w) in respect to the total weight of the vehicle.

In an alternative aspect, the co-solvent may be selected from an alcohol. The co-solvent as an alcohol selected from any one or mixture of ethanol, methanol, propanol, or isopropanol or any alcohol that is physiologically tolerated by the eye or skin. The co-solvent may be an alcohol in absolute/anhydrous form, such as absolute ethanol, methanol, propanol, or isopropanol.

In one embodiment, the co-solvent is ethanol, and is featured in the composition in an amount of no more than 30% (w/w), 25% (w/w), 15% (w/w) or 10% (w/w), or 5% (w/w) 2% (w/w), or about 1.0% (w/w) in respect to the amount of the vehicle, optionally wherein the vehicle consists only of a semifluorinated alkane and ethanol; further optionally wherein the ethanol is absolute. Said composition may comprise an amount of at least 0.25 mg/ml, at least 0.5 mg/ml, at least 1 mg/mL, at least 1.5 mg/mL, at least 2.0 mg/mL, at least 2.5 mg/mL, at least 3.0 mg/mL or at least 5.0 mg/mL of molecular iodine dissolved in the semifluorinated alkane and ethanol vehicle.

Compositions with particularly higher concentrations of iodine may be useful, and advantageous, for example for disinfectant applications, such as for disinfecting inanimate surfaces, such as a surface of a surgical tool, instrument or means.

In one aspect of the invention, a composition according to the invention may comprise of an amount of molecular iodine of up to about 20 mg/ml. In such embodiments, a composition may comprise of an amount of molecular iodine of at least 1 mg/mL, or at least 1.5 mg/mL, or at least 2.0 mg/mL, or at least 2.5 mg/mL, at least 3.0 mg/mL, at least 5.0 mg/mL, at least 10.0 mg/mL or at least 15.0 mg/mL.

In a one embodiment, the composition may comprise an amount of up to 20 mg/mL of molecular iodine dissolved in the vehicle. The composition may comprise of an amount of molecular iodine dissolved in the vehicle of at least 1 mg/mL, or at least 1.5 mg/mL, or at least 2.0 mg/mL, or at least 2.5 mg/mL, at least 3.0 mg/mL, at least 5.0 mg/mL, at least 10.0 mg/mL or at least 15.0 mg/mL. In an embodiment where the vehicle may consist of only a semifluorinated alkane and optionally a co-solvent, the composition may comprise an amount of at least 1 mg/mL, at least 1.5 mg/mL, at least 2.0 mg/mL, at least 2.5 mg/mL, at least 3.0 mg/mL or at least 5.0 mg/mL, or at least 10.0 mg/mL or at least 15.0 mg/mL of molecular iodine dissolved in said vehicle consisting of a semifluorinated alkane and said optional co-solvent.

In yet a further embodiment of the invention relating to the composition comprising molecular iodine, and a vehicle comprising a semifluorinated alkane such as defined in any of the previous embodiments described above, the vehicle preferably comprises a semifluorinated alkane or optionally, a mixture of semifluorinated alkanes in an amount of at least 60% (w/w) with respect to the total weight of the vehicle. In other embodiments, the vehicle may comprise at least 70% (w/w), 75% (w/w), 85% (w/w), 90% (w/w), 95% (w/w), 98% or at least 99% (w/w) of a semifluorinated alkane or a mixture of semifluorinated alkanes, with respect to the total weight of the vehicle.

In one preferred embodiment, the vehicle of the composition according to the invention is 100% (w/w) of a semifluorinated alkane or mixture of semifluorinated alkanes. In another embodiment, the vehicle comprises a semifluorinated alkane or optionally, a mixture of semifluorinated alkanes in an amount of at least 60% (w/w) with respect to

the total weight of the vehicle. In another embodiment, the vehicle may consist of at least 70% (w/w), 75% (w/w), 85% (w/w), 90% (w/w), 95% (w/w), 98% or at least 99% (w/w) of a semifluorinated alkane or a mixture of semifluorinated alkanes, with respect to the total weight of the vehicle, and a co-solvent or excipient, preferably a non-polar co-solvent or excipient.

The term “% (w/w)” as used herein and unless indicated refers to the amount of a component of a composition, (e.g. the composition of a vehicle) as a weight percentage in relation to the total weight of the composition (e.g. total weight of a vehicle), with ‘w’ denoting weight. For instance, the vehicle of a composition according to the present invention may comprise up to about 1.5% (w/w) of a co-solvent such as ethanol, relative to the total weight of the vehicle. The term “% (w/v)” likewise, unless indicated refers to the amount of a component of a composition as a weight percentage in relation to the total volume of the composition (with “w” denoting weight and “v” denoting volume).

In a further embodiment, the composition of the present invention is a rapidly evaporating formulation. Said composition is preferably antimicrobial effective (e.g. against any one or combination of bacteria, fungi, virus). Preferably, the rapidly evaporating formulation evaporates within 5 minutes, preferably within 3 minutes, more preferably within 1 minute from time of topical application to a surface, for example a tissue surface, or surface of any eye, or alternatively to an inanimate surface. Preferably, such rapidly evaporating formulation comprises molecular iodine dissolved in a vehicle comprising, or consisting of a semifluorinated alkane characterized by a boiling point of less than 140° C., even more preferably such rapidly evaporating formulation comprises molecular iodine dissolved in a vehicle comprising a semifluorinated alkane selected from the group consisting of F6H2 or F4H5. In one embodiment, such rapidly evaporating formulation comprises or consists of up to 0.5 mg/ml molecular iodine dissolved in F6H2.

In another embodiment, the compositions according to the invention may comprise further of an active ingredient dissolved in the vehicle. Preferably, the active ingredient is present in a therapeutically effective amount, which as used herein, means the active ingredient is at a dose, concentration or strength which is useful for producing the desired pharmacological effect, more specifically a pharmacological effect in the treatment of any one of the conditions or diseases as further described in detail below.

Preferably, the active ingredient is selected from the group consisting of antibiotics, anti-inflammatory agents, analgesics, anaesthetics, anti-allergic agents, and immunosuppressants. The active ingredient may also be selected from a corticosteroid. Preferably the said active ingredient is suitable for co-formulation with molecular iodine.

Preferred antibiotics include fluoroquinolones (ciprofloxacin, levofloxacin, ofloxacin, moxifloxacin, gatifloxacin, and the like); aminoglycosides (tobramycin, gentamicin, neomycin, and the like); polymyxin B combinations (polymyxin B/trimethoprim, Polysporin polymyxin B/bacitracin, Neosporin polymyxin B/neomycin/gramicidin, and the like) and other antibiotics (azithromycin, ilotycin, erythromycin, bacitracin, and the like).

Preferred anti-inflammatories include corticosteroids (such as dexamethasone, prednisolone, loteprednol, difluprednate, fluoromethalone, rimexolone, medrysone, hydrocortisone) or NSAIDs (such as bromfenac, nepafenac, diclofenac, flurbiprofen, suprofen, celecoxib, naproxen, rofecoxib).

Preferred anesthetics include proparacaine, lidocaine, tetracaine, benzocaine, butamben, dibucaine, oxybuprocaine, pramoxine, proxymetacaine.

Preferred anti-allergic agents include pheniramine maleate, epinastine, emedastine, azelastin, olopatadine, ketotifen, pemirolast, nedocromil, lodoxamide, cromolyn.

Preferred immunosuppressants include ciclosporine A, tacrolimus, sirolimus.

An example of a preferred corticosteroid is dexamethasone.

In an embodiment, the invention provides for a composition consisting of a) molecular iodine and a further active ingredient such as defined above, and b) a vehicle consisting of a semifluorinated alkane and optionally a non-polar co-solvent or excipient; wherein the semifluorinated alkane is of $\text{CF}_3(\text{CF}_2)_n\text{—R}_m$ wherein n is an integer selected from 1 to 12, and R is linear or branched alkyl or cycloalkyl with m carbon atoms, wherein m is an integer selected from 2 to 12.

In another embodiment, the invention provides for a composition consisting of a) molecular iodine and a further active ingredient such as defined above, and b) a vehicle consisting of a semifluorinated alkane and optionally a non-polar co-solvent or excipient; wherein the semifluorinated alkane is of $\text{CF}_3(\text{CF}_2)_n\text{—R}_m$ wherein n is an integer selected from 2 to 12, and R is linear or branched alkyl or cycloalkyl with m carbon atoms, wherein m is an integer selected from 2 to 12.

Optionally, the composition of the present invention may also comprise one or more further excipients as an additional component. The term “excipients” as used herein refers to any pharmaceutically acceptable natural or synthetic substance that may be added to the compositions of the present invention, more specifically to the vehicle of the composition to enhance or otherwise modify its physical or chemical constitution or stability or therapeutic properties.

Compounds or composition components considered as excipients may also fall under the definition of co-solvent as defined above. In one embodiment, the excipient featured in the composition of the invention is an excipient that does not support the formation of an iodine ion or oxidized iodine species, preferably said excipient does not support the formation as of any one (or combination of) the following: iodide (I—), triiodide (I3-), iodate (IO3-), or hypoiodate (IO—).

The present composition may optionally comprise one or more excipients such as, for example, an antioxidant, a preservative, a lipid or oily excipient, a surfactant or a lubricant or a combination of at least 2 excipients thereof.

Lipid or oily excipients for use in the composition of the present invention may include, for example, triglyceride oils (i.e. soybean oil, olive oil, sesame oil, cotton seed oil, castor oil, sweet almond oil), triglycerides, mineral oil (i.e. petrolatum and liquid paraffin), medium chain triglycerides (MCT), oily fatty acids, isopropyl myristate, oily fatty alcohols, esters of sorbitol and fatty acids, oily sucrose esters, or any other oily substance which is physiologically tolerated by the eye or skin. Said lipid or oily excipient may also function as a co-solvent for molecular iodine in accordance with the purpose of a co-solvent, such as defined above.

In one embodiment, the amount of excipient featured in a composition according to the invention may be in an amount of no more than 30% (w/w), 25% (w/w), 15% (w/w) or 10% (w/w), or 5% (w/w) 2% (w/w), or about 1.0% (w/w) in respect to the total weight amount of the vehicle.

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Preferably, the compositions of the invention are substantially free of water. As understood herein, the term 'substantially free', or alternatively 'essentially free' in reference to a composition constituent refers to the presence of said constituent in no more than trace amounts and that if present in trace amounts the constituent does not provide any technical contributions to the composition.

In a preferred embodiment of the invention, the compositions are essentially free of water.

In another further preferred embodiment, the compositions according to the present invention may be substantially free of a preservative, such as an antimicrobial preservative and/or a surfactant. Preferably, the compositions are free of, i.e. do not contain water and/or any one of a preservative, such as an antimicrobial preservative, or a surfactant.

Optionally, the compositions according to the present invention may be substantially free of a polymer or complexing agent which may form complexes or bind with molecular iodine, such as polyvinylpyrrolidone (PVP) or a cadexomer.

In another preferred embodiment, the compositions according to the invention may be substantially free of an iodine ion or oxidized iodine species, preferably as selected from any one (or combination of) the following: iodide (I^-), triiodide (I_3^-), iodate (IO_3^-), or hypoiodate (IO^-). Said species may include any suitable counterion or protic form thereof, for example iodide may have a counterion of potassium or sodium (i.e. NaI or KI).

In a preferred embodiment, the compositions of the present invention, preferably compositions that are substantially free of water and/or compositions that are substantially free of an iodine ion or oxidized iodine species, preferably as selected from any one (or combination of) the following: iodide (I^-), triiodide (I_3^-), iodate (IO_3^-), or hypoiodate (IO^-), are non-staining, preferably when administered or coming into contact to any one of the following surfaces: skin or dermal tissue, eye or eye tissue, and optionally, fabric. As defined herein, the term non-staining refers to the absence of any persisting discolouration or persisting coloured residue on the site or surface to which the composition is topically applied or is coming into contact with. It has been observed that when a composition of the invention comprising molecular iodine and a semifluorinated alkane as the vehicle is initially administered to a site of topical application, such as skin or filter paper surface, some colour may be observed initially on said surface, i.e. immediately post administration, however the colouration does not persist under ambient or skin surface conditions, and no long-term discolouration or residue is observed.

Preferably, any staining or colour which is present on application of the composition of the present invention disappears or dissipates within less than 10 minutes, less than 5 minutes, or preferably less than 1 minute. Accordingly, also no color is transferred to any surface that comes into contact with the site of administration afterwards. Thus, no color is transferred, for example, to a fabric that comes into contact with the site of iodine administration, representing a major advantage over conventional, water-based antiseptic or disinfectant iodine formulations.

Without wishing to be bound by theory, it is assumed by the inventors that the lack of staining or discolouration observed in topical application of compositions according to the present invention are due to the absence of, and lack of formation of species such as iodine ion and/or oxidized iodine species such as triiodide (I_3^-) in the composition. The reddish-brown or yellow colour of many aqueous-based iodine formulations and corresponding discolouration and

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staining of skin and other surfaces are generally attributed to such species, in particular to triiodide. As discolouration to the skin or cross-contamination to clothing may be avoided or reduced, the compositions of the invention may be advantageous in terms of improving patient compliance, where the composition is used as a medicament

In another, and optional aspect, the compositions of the invention may alternatively be formulated so as to provide a staining effect; the composition may comprise at least one component species (i.e. triiodide (I_3^-)), excipient or co-solvent which is coloured or staining, or which results in a composition that is staining. Preferably, said compositions are adapted for use in applications where staining or colouration may be useful as an indicator of the surface area to which the composition has been applied or contacted to, for example, in an application such as surgery. Examples of staining compositions are those comprising a semifluorinated alkane and an alcohol co-solvent.

It has also been found that the compositions of the present invention appear to be highly stable, surprisingly even when formulated at relatively higher as well as at lower concentrations of molecular iodine. State of the art compositions, such as provided for ophthalmic application are typically based on aqueous formulations comprising 5 w/v % povidone-iodine complex (e.g. Minims®, Betadine®), which provide only 0.5% available iodine, with ultimately only around 2.5 ppm (parts per million) molecular iodine as the active species.

As will be described in further detail below in the Examples, the compositions of the invention comprising a vehicle consisting of semifluorinated alkane and/or a perfluorocarbon do not undergo any physical change, such as precipitation of molecular iodine, the formation of solid or particulate species or aggregates out of solution, after prolonged period of storage, and formulations at higher concentrations of the active molecular iodine may be formulated. Compositions based on liquid or solid vehicles of one or a mixture of semifluorinated alkane and co-solvents or excipients, which do not lead to the formation of species which lead to staining, such as hydrocarbon compounds, are also observed to be stable and non-staining, and also permit the formulation of high concentrations of molecular iodine (i.e. more than 100, 250, 500 ppm or up to 1000 ppm molecular iodine).

In the context of therapeutic applications relying on topical administration to a tissue surface of a patient or subject, preferably a human patient, it is important that compositions may be readily and consistently be applied and distributed over said surface. For example, where the tissue surface is the skin, or the surface tissues of the eye, a composition needs also to reach out e.g. to folds/crevices formed by the tissue, which have limited direct accessibility. The molecular iodine compositions formulated with a vehicle based on a semifluorinated alkane according to the invention provide compositions with good spreadability for topical application. Achieving patient-friendly compositions which have positive sensory attributes is also advantageous, for instance in terms of achieving patient compliance. The semifluorinated alkane compositions according to the invention provide a pleasant and 'silky' sensation when applied, for example to the eye or the skin.

In a preferred embodiment, the composition of the present invention, is stable at for at least 30 days, preferably at room temperature. In further embodiments, the composition is preferably stable for at least 90 days, more preferably stable for at least 180 days at room temperature.

In a further preferred embodiment, the composition of the present invention, is stable without protection from light; preferably the composition is stable for at least 30 days; more preferably, stable for at least 90 days; or even more preferably, stable for at least 180 days at room temperature without protection from light.

Composition stability may be determined by comparison of properties to that observed for the composition when first prepared. Properties which may be used to determine stability include the determination of molecular iodine content by analytical methods known in the art for determining molecular iodine content (such as uv-vis measurements), or the presence of other iodine species (e.g. staining tests, uv-vis measurements). Stability may also be assessed by comparing overall visual appearance of the composition for physical changes such as precipitation, formation of particles and changes in colouration.

The iodine compositions of the invention are stable within a broad range of concentrations of molecular iodine, without requiring the inclusion of many additional excipients and composition components. For example, the compositions of the present invention may be prepared without the use of additional stabilizing agents such as buffering or pH-adjusting agents, or complexing agents to enhance solubility of iodine which may be typically required for aqueous based compositions comprising iodine.

In another aspect of the present invention, the composition may be formulated as any one of a solution, semi-solid, cream, lotion, ointment, or as a swab.

The compositions in accordance with the present invention such as described in any one of the above embodiments may also be used in therapy, as a medicine or medicament.

In a particular embodiment, the compositions may be used as an antiseptic and/or as a disinfectant in the treatment or prevention of a disease or condition in a subject in need thereof. Preferably, the compositions of the invention may be used to prevent or inhibit, the growth or proliferation of a microorganism.

As used herein, the term 'antiseptic' refers to a composition which prevents or inhibits the growth of microorganisms and/or may destroy microorganisms, in particular those which may cause infection or sepsis, and may be synonymous with the terms anti-microbial or biocide. The term 'disinfectant' refers to a composition which inhibit or prevent proliferation of a microorganisms and/or destroy microorganisms that is topically applied to a subject, but more preferably to an inanimate surface to which a subject will or may have contact.

As understood herein, the use of a composition of the present invention in the context of prevention of a disease or condition, or in a prophylactic treatment refers to use of said composition so as to prevent occurrence, or exacerbation, or risk of occurrence said disease or condition in a subject. For example, a subject who has a minor cut or surface wound on their skin may use and topically apply the composition to said minor cut or surface wound to the skin to destroy and/or inhibit the growth and proliferation of any microorganism, in order to avoid infection and allow for quicker healing thereof.

The composition according to any of the above embodiments may, in particular, be used for use treatment or prevention of a disease or condition affecting the skin, or mucosal tissue, wherein the composition is topically administered to the skin and/or mucosal tissue. As understood herein, the term 'skin' refers to dermal tissue, including the various layers, associated tissues or cellular layers, i.e. the epidermis (i.e. stratum corneum and vital, or living epider-

mis), the dermis or derma and hypodermis, and in all its forms. The composition may be administered to the skin on any surface of a subject. Preferably, it is administered to the skin which is injured or non-intact, such as to a cut, laceration, abrasion, burn, or a wound.

Mucosal tissue refers to tissue comprising epithelial cells and connective tissue and may preferably be mucosal tissue or membrane of the nose, mouth or buccal cavity.

In a further aspect, the composition according to any of the above embodiments may, also be used for the treatment or prevention of a disease or condition affecting the eye or an ophthalmic tissue, wherein the composition is topically administered to the eye or ophthalmic tissue. Ophthalmic tissue refers to any anatomical aspect of the eye apparatus, for example the cornea, or conjunctiva, or meibomian glands, or the eye lid margins, or eyelashes.

In a preferred embodiment, the compositions may be topically administered to the surface of the eye and to any such region or tissue of the eye that may be accessible to topical administration, such as to the cornea or conjunctiva, or to the eyelid margins or corners.

Preferably, the compositions are used to treat or prevent a disease or condition that is associated with, or caused by a microorganism. Preferably said microorganism is pathogenic, i.e. may cause a disease. The microorganism may be selected from any one or combination of bacteria, demodex, fungi and virus. Further microorganisms for which the present compositions may be optionally therapeutically useful may be amoeba, or yeast.

Bacteria may include gram-positive or gram-negative species. In a preferred embodiment, the compositions of the invention may be used to treat or prevent diseases or conditions associated or caused by bacteria, preferably selected from any one or combination of bacteria from the genus *Streptococcus*, *Staphylococcus*, *Enterococcus*, *Peptostreptococcus*, *Pseudomonas*, *Mycobacteria*, *Haemophilus*, *Clostridium*, *Corynebacterium*, *Escherichia*, and *Klebsiella*.

Viruses may include herpes simplex virus (HSV), adenovirus, herpes zoster virus varicella-zoster virus (VZV), picornavirus, poxvirus, influenza virus and human immunodeficiency virus (HIV).

In one preferred embodiment, the disease or condition is associated or caused by a microorganism selected from any one or combination of the group consisting of *Staphylococcus aureus*, *Pseudomonas aeruginosa*, *Candida albicans*, *Aspergillus Brasiliensis*, *Streptococcus pneumoniae*, *Proteus mirabilis*, and *Propionibacterium acnes*.

Where the composition is for use in the treatment or prevention of a disease or condition of the eye or an ophthalmic tissue, the disease or condition to be treated is preferably selected from the group consisting of blepharitis, bacterial conjunctivitis, ocular dryness, follicular conjunctivitis, giant papillary conjunctivitis, corneal ulcer, keratitis, conjunctival neoplasia, HSV keratitis, eye allergy, neonatal conjunctivitis, ulcerative infectious keratitis, epithelial keratitis, stromal keratitis and keratoconjunctivitis epidemica.

In a particularly preferred embodiment, the compositions for ophthalmic use are formulated wherein the amount of molecular iodine in the composition is preferably less than 0.5% (w/v), less than 0.1% (w/v), less than 0.01% (w/v) or less than 0.001% (w/v).

Where the composition is for use in the treatment or prevention of a disease or condition of the skin or mucosal tissue, it is preferred that the composition may be used to treat non-intact skin or mucosal tissue, preferably selected from a wound, a cut, a surgical wound, burn, abrasion, laceration, ulcer, rash, and sore. In another embodiment, the

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composition may be used to prevent infection or microorganism proliferation in a site where skin or mucosal tissue is not intact, preferably at a site of a wound, a cut, a surgical wound, a burn, an abrasion, a laceration, an ulcer, a rash, or a sore. In further embodiment, the composition of the present invention may be used to treat or prevent a disease or condition, preferably as acne, or an infection, preferably a fungal and/or bacterial infection, for example, athletes foot.

In particular preferred embodiments, the compositions for use in the treatment of a disease or condition of the skin or mucosal tissue are formulated wherein the amount of molecular iodine in the composition is preferably at least 0.01% (w/v), at least 0.1% (w/v), at least 0.5% (w/v).

In a further embodiment of the invention, the compositions for therapeutic or prophylactic use may be topically administered to a tissue or organ, at any one or more instances of prior to, during, or subsequent to a surgical or medical procedure. In a preferred embodiment, the compositions may be topically administered prior to surgery, i.e. pre-operatively to a tissue or organ prior to surgical procedure on said tissue or organ; for example prior to, during, or subsequent to a cataract surgery, LASIK surgery, corneal abrasion or intravitreal injection (to prevent bacteria present on the cornea or conjunctiva from being carried over into the interior of the eye).

The use of a composition as described in any one of the above embodiments, in the manufacture or preparation of a medicament or a medicine for the treatment or prophylactic treatment of a subject in need thereof in relation to any one of diseases or conditions as described herein are also provided for in the context of the present invention.

Further provided for within the context of the present invention, are also methods of treating or prophylactically treating a subject with, or likely to be afflicted with, any one of the diseases or conditions as described herein, wherein the methods may comprise the topical administration of any one of the defined compositions to said subject. For instance, where the composition is administered to the eye, the composition may be directly topically instilled to the eye or an ophthalmic tissue.

Said treatment methods and compositions for therapeutic use are moreover preferably targeted towards human subjects, but also to veterinary subjects, such as mammals, such as canines, felines, horses, or other livestock.

In another aspect, the present invention may relate to the use of a composition according to any of the above embodiments, in a cosmetic preparation.

In yet another aspect, the composition according to any of the above embodiments may also be used as a disinfectant or an antiseptic, wherein the composition is administered to an animate and/or an inanimate surface. In one preferred embodiment, the composition may be used as a disinfectant for an inanimate surface, preferably a surface of a surgical tool, instrument or means.

Moreover, the invention may also provide to a kit comprising a composition according to the invention as described herein, wherein the kit comprises a container adapted for holding and/or storing the composition and a dispenser adapted for topical administration of the composition to a subject or a surface in need thereof.

The present invention may include the following numbered items:

1. A composition comprising molecular iodine and a vehicle comprising a semifluorinated alkane and/or a perfluorocarbon.

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2. A composition according to item 1, wherein the composition is substantially free of water.
3. A composition of any of items 1 or 2, wherein the perfluorocarbon compound is a perfluorinated hydrocarbon with 5 to 12 carbon atoms.
4. A composition of any of the preceding items, wherein the perfluorocarbon is a compound of formula $F(CF_2)_nF$, wherein n is an integer selected from 5 to 12.
5. A composition of any of the preceding items, wherein the perfluorocarbon is selected from perfluorooctane, or perfluorodecalin, and perfluorooctyl bromide.
6. The composition of any one of the preceding items wherein the semifluorinated alkane is of formula (I):



wherein n is an integer selected from 1 to 12, and R is linear or branched alkyl, or a cycloalkyl with m carbon atoms, wherein m is an integer selected from 2 to 12.

7. The composition of item 6, wherein the semifluorinated alkane is of formula (II):



wherein n and m are each independently selected from an integer from 3 to 10.

8. The composition according item 7, wherein the semifluorinated alkane is selected from F4H5, F4H6, F4H8, F6H6, F6H8, F8H8 and F10H5, preferably selected from F4H5, and F6H8.
9. The composition according to any one of the preceding items, wherein the molecular iodine is dissolved in the vehicle.
10. The composition according to any one of the preceding items, wherein the amount of molecular iodine dissolved in the vehicle is less than 1 mg/mL, 0.8 mg/mL, 0.6 mg/mL, 0.5 mg/mL, 0.2 mg/mL, 0.1 mg/mL, 0.05 mg/mL, 0.02 mg/mL, 0.01 mg/mL, or less than 0.001 mg/mL.
11. The composition according to any one of the preceding items, wherein the composition is free of an iodine species from the group consisting of any one or combination of an iodide (I^-), triiodide (I_3^-), iodate (IO_3^-), or hypoiodate (IO^-).
12. The composition according to any one of the preceding items, wherein the composition further comprises an active ingredient dissolved in the vehicle, optionally selected from the group consisting of antibiotics, anti-inflammatory agents, analgesics, anaesthetics, immunosuppressants, and anti-allergic agents.
13. The composition according to any one of the preceding items, wherein the vehicle further comprises a co-solvent, preferably selected from an alcohol, or non-polar solvent or from a mixture thereof.
14. The composition according to any one of the preceding items, wherein the co-solvent is a solvent that does not support the formation of an iodine ion or oxidized iodine species, preferably wherein the co-solvent is a solvent that does not support the formation of any one or combination of iodide (I^-), triiodide (I_3^-), iodate (IO_3^-), or hypoiodate (IO^-).
15. The composition according to item 13 or 14, wherein the co-solvent is a non-polar solvent selected from any one or mixture of a saturated hydrocarbon or alkane, such as pentane, hexane, heptane, or octane; mineral oil, paraffin, and a siloxane.
16. The composition according to item 13 or 14, wherein the co-solvent is an alcohol selected from any one or

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mixture of ethanol, methanol or isopropanol, preferably absolute ethanol, methanol or isopropanol.

17. The composition according to any one of the items 13 to 16, wherein the vehicle comprises said co-solvent in an amount of up to about 38 wt % relative to the total weight of the vehicle. 5
18. The composition according to any one of the preceding items, wherein the vehicle consists of a semifluorinated alkane and/or a perfluorocarbon.
19. The composition according to any one of the preceding items, wherein the amount of molecular iodine dissolved in the composition or vehicle is at least 1 mg/mL, or up to 20 mg/mL. 10
20. The composition according to any one of the preceding items, wherein the composition comprises molecular iodine and a vehicle consisting of at least one semifluorinated alkane and/or a perfluorocarbon. 15
21. The composition according to any one of the preceding items, wherein the composition consists of molecular iodine and the vehicle, wherein the vehicle consists of a semifluorinated alkane and/or a perfluorocarbon, and optionally, an active ingredient, and further optionally, a co-solvent, such as ethanol, or an excipient. 20
22. The composition according to any one of the preceding items, wherein the composition consists of molecular iodine and a vehicle consisting of a semifluorinated alkane and/or a perfluorocarbon. 25
23. The composition according to any one of the preceding items, wherein the composition is non-staining when applied to a surface or tissue; preferably any one of skin, mucosal tissue, eye or eye tissue, or fabric. 30
24. The composition according to any one of the preceding items, wherein the composition is stable for at least 30 day, preferably stable for at least 90 days, more preferably stable for at least 180 days, optionally the composition is stable without protection from light. 35
25. The composition according to item 24, wherein the composition is stable at room temperature (15-25° C.).
26. The composition according to any one of the preceding claims, wherein the composition is formulated as a liquid, semi-solid, cream, lotion, ointment, or a swab. 40
27. The composition according to any one of the preceding items, for use as a medicine.
28. The composition according to any one of the preceding items, for use as an antiseptic and/or a disinfectant. 45
29. The composition according to item 27 to 28 for use in the treatment or prevention of a disease or condition affecting the skin or mucosal tissue, wherein the composition is topically administered to said organ or tissue. 50
30. The composition according to any one of items 27 to 29 for use in the treatment or prevention of a disease or condition affecting the eye or an ophthalmic tissue, wherein the composition is topically administered to the eye or ophthalmic tissue. 55
31. The composition for use according to any one of items 27 to 30, wherein the disease or condition is associated with, or caused by a microorganism.
32. The composition for use according to item 31, wherein the microorganism selected from any one or combination of bacteria, demodex, fungi, yeast or virus. 60
33. The composition for use according to any one of items 27 to 32, wherein disease or condition is associated or caused by a microorganism is selected from any one or combination of the group consisting of *Staphylococcus aureus*, *Pseudomonas aeruginosa*, *Candida albicans*, 65

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Aspergillus brasiliensis, *Streptococcus pneumoniae*, *Proteus mirabilis*, *Propionibacterium acnes*.

34. The composition for use according to any one of items 27 to 33, wherein the composition is for use in the treatment or prevention of a disease or condition of the skin or mucosal tissue, preferably selected from a wound, a cut, an abrasion, a burn, a surgical site wound, an ulcer, a sore, acne, and an infection.
35. The composition for use according to items 27 to 33, wherein the composition is for use in the treatment or prevention of a disease or condition of the eye or an ophthalmic tissue, preferably selected from the group consisting of blepharitis, bacterial conjunctivitis, ocular dryness, follicular conjunctivitis, giant papillary conjunctivitis, corneal ulcer, keratitis, conjunctival neoplasia, HSV keratitis, eye allergy, ulcerative infectious keratitis, epithelial keratitis, stromal keratitis, keratoconjunctivitis epidemica, and neonatal conjunctivitis.
36. A composition for use according to any one of items 27 to 35, wherein the therapeutic use or preventative use comprises topical administration of the composition to a tissue, at any one instance or combination of prior to, during, or subsequent to a surgical or medical procedure.
37. A composition for use according to items 36, wherein the surgical or medical procedure is an ophthalmic procedure, preferably cataract surgery, LASIK, corneal abrasion, or intravitreal injection.
38. A method of treating or preventing a disease or condition associated or caused by a microorganism, preferably selected from any one or combination of bacteria, demodex, fungi, or virus, comprising topically administering a composition of any one of items 1 to 26 to a subject, preferably to the skin or muscosal tissue, or an eye or ophthalmic tissue of a subject in need thereof; preferably wherein infection or microorganism growth is inhibited or prevented, and optionally wherein the composition is administered at any one instance or combination of a) prior to, b) during, or c) subsequent to a surgical or medical procedure.
39. Use of a composition according to any one of item 1 to 26, in the manufacture of a medicament for use in the treatment of a disease or condition associated or caused by a microorganism, preferably selected from any one or combination of bacteria, demodex, fungi, or virus; preferably wherein the medicament is topically administered to a subject and/or adapted for topical administration to dermal tissue, mucosal tissue, to an eye or ophthalmic tissue of a subject.
40. Use of a composition according to any one of items 1 to 26 as a disinfectant or an antiseptic, preferably wherein the composition is administered to an animate and/or an inanimate surface.
41. The use of a composition according to any one of items 1 to 26 as a disinfectant for an inanimate surface, preferably wherein the inanimate surface is a surface of a surgical tool or means.
42. Use of a composition according to any one of items 1 to 26 in a cosmetic preparation.
43. A kit comprising a composition according to any of items 1 to 26, wherein the kit comprises a container adapted for holding and/or storing the composition and a dispenser adapted for topical administration of the composition to a subject or a surface in need thereof.
44. The composition according to any of the items 6 to 24, wherein the composition consists of up to about 1.0 mg/mL molecular iodine dissolved in a vehicle con-

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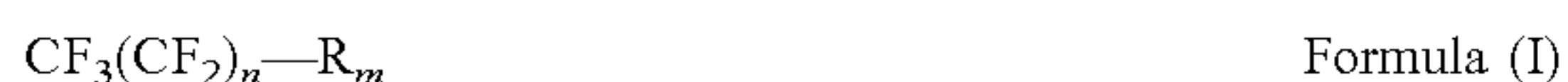
sisting of 1-perfluorohexyl-octane or 1-perfluorbutyl-pentane, optionally comprising a non-polar solvent.

45. The composition according to any of the items 6 to 24, wherein the composition consists of up to about 0.5 mg/mL molecular iodine dissolved in a vehicle comprising or consisting of perfluorohexyl-ethane, optionally comprising a non-polar solvent.

The present invention may also include the following items:

- 1.1 A composition comprising:

- a) molecular iodine, and
b) a vehicle comprising a semifluorinated alkane; wherein the semifluorinated alkane is of formula (I):



wherein n is an integer selected from 1 to 12, and R is linear or branched alkyl or cycloalkyl with m carbon atoms, wherein m is an integer selected from 2 to 12.

- 1.2 A composition consisting of:

- a) molecular iodine and optionally, a further active ingredient, and
b) a vehicle consisting of a semifluorinated alkane and optionally a non-polar co-solvent or excipient; wherein the semifluorinated alkane is of formula (I):



wherein n is an integer selected from 1 to 12, and R is linear or branched alkyl or cycloalkyl with m carbon atoms, wherein m is an integer selected from 2 to 12.

- 1.3 The composition according to any one of the preceding items, wherein the molecular iodine is dissolved in the vehicle in an amount of up to about 20 mg/mL.

- 1.4 The composition according to any one of the preceding items, wherein the molecular iodine is dissolved in the vehicle in an amount of up to about 1 mg/mL, optionally in the range from about 0.001 mg/mL to 1 mg/mL.

- 1.5 The composition according to any one of the preceding items, wherein the composition comprises a further active ingredient dissolved in the vehicle, optionally selected from the group consisting of antibiotics, anti-inflammatory agents, analgesics, anaesthetics, immunosuppressants and anti-allergic agents.

- 1.6 The composition according to any one of the preceding items, wherein the vehicle further comprises a co-solvent, preferably an alcohol or a non-polar solvent and optionally in an amount of no more than 38% (w/w) in respect to the amount of the vehicle.

- 1.7 The composition according to any one of the preceding items, wherein the vehicle comprises a non-polar solvent or excipient.

- 1.8 The composition according to item 1.7, wherein the non-polar solvent or excipient is in an amount of no more than 38% (w/w) in respect to the amount of the vehicle, or in an amount of no more than 30% (w/w), 25% (w/w), 15% (w/w) or 10% (w/w), or 5% (w/w) 2% (w/w), or about 1.0% (w/w) in respect to the total weight amount of the vehicle.

- 1.9 The composition according to claim item 1.7 or 1.8, wherein the non-polar co-solvent is selected from any one or mixture of a saturated hydrocarbon; mineral oil; paraffin; siloxane; and a perfluorocarbon.

- 1.10 The composition according to any one of the preceding items, wherein the composition is essentially free of water.

- 1.11 The composition according to any one of the preceding items, wherein the composition is free of an

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iodine species from the group consisting of any one or combination of iodide (I^-), triiodide (I_3^-), iodate (IO_3^-), or hypiodate (IO^-).

- 1.12 The composition according any one of the preceding items, wherein the composition is non-staining, optionally wherein the composition is stable at room temperature for at least 30 days, preferably stable for at least 90 days, or more preferably, stable for at least 180 days.

- 1.13 The composition according to any one of the preceding items, wherein the composition consists of molecular iodine and a vehicle consisting of a semifluorinated alkane.

- 1.14 The composition of any one of the preceding items, wherein the semifluorinated alkane is of formula (II):



wherein n and m are each independently selected from an integer from 3 to 10, optionally wherein the semifluorinated alkane is selected from $\text{CF}_3(\text{CF}_2)_3\text{—}(\text{CH}_2)_4\text{CH}_3$ (F4H5), $\text{CF}_3(\text{CF}_2)\text{—}(\text{CH}_2)_5\text{CH}_3$ (F4H6), $\text{CF}_3(\text{CF}_2)_5\text{—}(\text{CH}_2)_5\text{CH}_3$ (F6H6), $\text{CF}_3(\text{CF}_2)_5\text{—}(\text{CH}_2)_7\text{CH}_3$ (F6H8), $\text{CF}_3(\text{CF}_2)_3\text{—}(\text{CH}_2)_7\text{CH}_3$ (F4H8), $\text{CF}_3(\text{CF}_2)_7\text{—}(\text{CH}_2)_7\text{CH}_3$ (F8H8) and $\text{CF}_3(\text{CF}_2)_9\text{—}(\text{CH}_2)_4\text{CH}_3$ (F10H5), or more preferably selected from $\text{CF}_3(\text{CF}_2)_3\text{—}(\text{CH}_2)_4\text{CH}_3$ (F4H5) and $\text{CF}_3(\text{CF}_2)_5\text{—}(\text{CH}_2)_7\text{CH}_3$ (F6H8).

- 1.15 The composition of any one of the preceding items, wherein the semifluorinated alkane is of formula (II):



wherein n is an integer selected from 2 to 9 and m is an integer from 1 to 7, optionally wherein the semifluorinated alkane is selected from $\text{CF}_3(\text{CF}_2)_3\text{—}(\text{CH}_2)_4\text{CH}_3$ (F4H5), $\text{CF}_3(\text{CF}_2)_3\text{—}(\text{CH}_2)_5\text{CH}_3$ (F4H6), $\text{CF}_3(\text{CF}_2)_5\text{—CH}_2\text{CH}_3$ (F6H2), $\text{CF}_3(\text{CF}_2)_5\text{—}(\text{CH}_2)_5\text{CH}_3$ (F6H6), $\text{CF}_3(\text{CF}_2)_5\text{—}(\text{CH}_2)_7\text{CH}_3$ (F6H8), $\text{CF}_3(\text{CF}_2)_3\text{—}(\text{CH}_2)_7\text{CH}_3$ (F4H8), $\text{CF}_3(\text{CF}_2)_7\text{—}(\text{CH}_2)_7\text{CH}_3$ (F8H8) and $\text{CF}_3(\text{CF}_2)_9\text{—}(\text{CH}_2)_4\text{CH}_3$ (F10H5).

- 1.16 The composition according to any one of the preceding items, wherein the composition consists of up to about 1.0 mg/mL molecular iodine dissolved in a vehicle consisting of 1-perfluorohexyl-octane or 1-perfluorbutyl-pentane, and optionally non-polar solvent.

- 1.17 The composition according to any one of the preceding items, wherein the composition consists of up to about 0.5 mg/mL molecular iodine dissolved in a vehicle consisting of perfluorohexyl-ethane, and optionally a non-polar solvent.

- 1.18 The composition according to any one of the preceding items, wherein the composition is formulated as a liquid, semi-solid, cream, lotion, ointment, or a swab.

- 1.19 The composition according to any one of the preceding items for use as a medicine.

- 1.20 The composition according to item 1.19, for use as an antiseptic and/or as a disinfectant in the treatment or prevention of a disease or condition in a subject in need thereof.

- 1.21 The composition according to any one of items 1.19 or 1.20, for use in the treatment or prevention of a disease or condition affecting:

- a) the skin or mucosal tissue; or
b) the eye or an ophthalmic tissue; wherein the composition is topically administered to said organ or tissue.

- 1.22 The composition for use according to any one of items 1.20 or 1.21, wherein the disease or condition is associated with, or caused by a microorganism, pref-

erably selected from any one or combination of bacteria, demodex, fungi, or virus.

1.23 The composition for use according to any of the items 1.19 to 1.22, wherein the composition is for use in the treatment or prevention of a disease or condition of:

a) the eye or an ophthalmic tissue selected from the group consisting of blepharitis, bacterial conjunctivitis, ocular dryness, follicular conjunctivitis, giant papillary conjunctivitis, corneal ulcer, keratitis, conjunctival neoplasia, HSV keratitis, eye allergy, ulcerative infectious keratitis, epithelial keratitis, stromal keratitis, neonatal conjunctivitis and keratoconjunctivitis epidemica.

b) the skin, epithelial or mucosal tissue selected from a wound, a cut, a surgical wound, burn, abrasion, laceration, ulcer, rash, sore, acne, and infection.

1.24 A composition for use according to any one of items 1.19 to 1.23, wherein the treatment or preventative use comprises topical administration of the composition to a tissue, at any one or more instances of prior to, during, or subsequent to a surgical or medical procedure.

1.25 A method of treating or preventing a disease or condition associated or caused by a microorganism, preferably selected from any one or combination of bacteria, demodex, fungi, or virus, comprising topically administering a composition of any one of items 1.1 to 1.18 to a subject, preferably to the skin or mucosal tissue, or an eye or ophthalmic tissue of a subject in need thereof; preferably wherein infection or microorganism growth is inhibited or prevented, and optionally wherein the composition is administered at any one instance or combination of a) prior to, b) during, or c) subsequent to a surgical or medical procedure.

1.26 Use of a composition according to any one of items 1.1 to 1.18, in the manufacture of a medicament for use in the treatment of a disease or condition associated or caused by a microorganism, preferably selected from any one or combination of bacteria, demodex, fungi, or virus; preferably wherein the medicament is topically administered to a subject and/or adapted for topical administration to dermal tissue, mucosal tissue, to an eye or ophthalmic tissue of a subject.

1.27 Use of a composition according to any one of items 1.1 to 1.18 as a disinfectant or an antiseptic, preferably wherein the composition is administered to an animate and/or an inanimate surface.

1.28 The use of a composition according to any one of items 1.1 to 1.18 as a disinfectant for an inanimate surface, preferably wherein the inanimate surface is a surface of a surgical tool or means.

1.29 Use of a composition according to any one of items 1.1 to 1.18 in a cosmetic preparation.

1.30 A kit comprising a composition according to any of items 1.1 to 1.18, wherein the kit comprises a container adapted for holding and/or storing the composition and a dispenser adapted for topical administration of the composition to a subject or a surface in need thereof.

The following examples serve to illustrate the invention, however should not to be understood as restricting the scope of the invention.

EXAMPLES

Example 1—Compositions

(A) I₂ in F4H5

Molecular iodine (I₂) solutions in the semifluorinated alkane F4H5 at target concentrations of 0.01 mg/ml, 0.05

mg/ml, 0.1 mg/mL, 0.5 mg/mL, 0.55 mg/ml and 0.94 mg/mL were prepared. Molecular iodine was weighed into clear glass vials, followed by addition of the calculated volume of semifluorinated alkane F4H5. The compositions were stirred for 2 hours at room temperatures. The compositions provided clear pink solutions, ranging from strong pink/violet color at high concentrations (0.55 mg/ml) to very dilute pinkish colour at lower concentrations (0.01 mg/ml).

The 0.94 mg/ml solution was prepared by adding additional F4H5 to a composition comprising 1.0 mg/mL molecular iodine until all remaining molecular iodine particles were dissolved, providing a clear solution of deep pink/violet colour.

(B) I₂ in F6H8

Molecular iodine (I₂) solutions in the semifluorinated alkane F6H8 at target concentrations of 0.1 mg/mL, 0.5 mg/mL and 1.0 mg/mL were prepared. Molecular iodine was weighed into clear glass vials, followed by addition of the calculated volume of semifluorinated alkane F6H8. The compositions were stirred for 2 hours at room temperature.

The compositions at 0.1 mg/mL and 0.5 mg/mL concentrations provided clear pink solutions, ranging from strong pink/violet color at high concentrations (0.50 mg/ml) to dilute pink colour at lower concentration (0.1 mg/ml).

Additional F6H8 was added to the target 1.0 mg/mL composition until all remaining molecular iodine particles were dissolved, providing a clear solution, with a concentration of about 0.88 mg/mL. The solution is a deep pink/violet in colour.

(C) I₂ in Perfluorodecalin (Octadecafluorodecalin, C₁₀F₁₈)

A 0.2 mg/ml molecular iodine (I₂) solution in perfluorodecalin was prepared by weighing molecular iodine into a clear glass vial, followed by addition of the calculated volume of perfluorodecalin. The composition was stirred overnight at room temperature to provide a clear pink solution.

(D) I₂ in F8H8 or F10H5

A composition of 0.2 mg/ml molecular iodine (I₂) and F8H8 and a composition of 0.1 mg/ml of molecular iodine (I₂) and F10H5 were prepared, respectively, by weighing molecular iodine into a clear glass vial, followed by addition of the calculated amount of semifluorinated alkane to the vial. The molecular iodine was dissolved in the respective semifluorinated alkanes by stirring with waterbath (45° C.) heating to result in clear pink solutions. Upon cooling to room temperature, the composition prepared with F8H8 provided a pink coloured semi-solid, while the composition with F10H5 resulted in a pinkish coloured solid. On application to the skin, both compositions were observed to change to a liquid state upon contact with the skin.

(E) I₂ in F4H5+EtOH

Solutions with higher I₂ concentrations may be prepared with the addition of ethanol (non-absolute or alternatively absolute). F4H5/EtOH-Concentrations of about 20 mg/mL of I₂ in F4H5 are achieved when ethanol is added to an amount of 38% (w/w), relative to the weight of the vehicle. The resulting composition is a red-brown coloured solution.

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F4H5/EtOH-Solutions comprising 3 mg/mL iodine were prepared by adding 1.5% (w/w) ethanol (or alternatively 1.0% (w/w)) relative to the weight of the F4H5/EtOH liquid vehicle. The compositions are brown-pink coloured clear solutions.

(F) I₂ in F6H8+EtOH

F6H8/EtOH-Solutions with higher I₂ concentrations may be prepared with the addition of ethanol (non-absolute or alternatively absolute). Concentrations of about 20 mg/mL of I₂ in F6H8 are achieved when ethanol is added to an amount of 38% (w/w), relative to the weight of the vehicle.

F6H8/EtOH-comprising 3 mg/ml iodine were prepared by adding 1.5% (w/w) ethanol (or alternatively 1.0% (w/w)) relative to the weight of the F6H8/EtOH liquid vehicle. The compositions are reddish-brown coloured clear solutions.

Example 2—Composition Stability

Saturated iodine solutions as prepared in Example 1 above of I₂ in F4H5 (0.94 mg/mL) and F6H8 (0.88 mg/mL) were stored standing under ambient atmosphere and temperature conditions under a capped clear glass vial for 52 weeks without protection from light. No visible changes in terms of colour, or appearance of particles forming either in solution, or on the glass walls were observed.

In contrast, the formation of particles in solution and on the glass walls were observed for the Iodine solutions as prepared in Example 1 above of I₂ in the F4H5/Ethanol or F6H8/Ethanol liquid vehicles.

The stability of compositions of iodine (I₂) dissolved the semifluorinated alkanes F4H5 and F6H8 at further concentrations of 0.01 mg/mL, 0.05 mg/mL, 0.1 mg/mL, 0.5 mg/mL and 0.9 mg/mL, stored in closed clear-glass crimp cap vials (minimum composition volume of 10-mL) are tested. Comparative compositions, such as with alternative vehicles such as aqueous-based and other organic compounds are also formulated and tested for comparison.

Samples of the compositions are stored at ambient conditions (room temperature), while some samples are aged or stress-condition testing, for example under UV-stress conditions (to evaluate light sensitivity) or stored at non-ambient temperatures.

The compositions are assessed at intervals over time, for example via: visual inspection of the vials to observe and record any visible changes to the compositions such as precipitation, formation of deposits, etc, as well as staining tests; measuring the content of molecular iodine (and other iodine species) e.g. by uv-vis measurements, and other analytical methods in the art for analysing iodine content.

Example 3—Antimicrobial Effectiveness

(A) Ph. Eur. 5.1.3

Solutions of I₂ in F4H5 (0.85 mg/mL) and I₂ in F6H8 (0.85 mg/mL) were tested for their efficacy of antimicrobial preservation according to current edition Ph. Eur. 5.1.3. under the acceptance criterium for parenteral preparations, eye preparations, intrauterine preparations and intramammary preparations. Both solutions were found to be fully compliant within the guidelines. The tests were conducted for both formulations with the following microorganisms: *Pseudomonas aeruginosa*, *Staphylococcus aureus*, *Candida albicans*, and *Aspergillus brasiliensis*.

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(B) Hemmhof Test/Agar Diffusion Test

Hemmhof tests, or Agar diffusion tests were performed on the following compositions:

I₂ (3 mg/mL) solution in F6H8 and EtOH 1.5% (w/w) as prepared in Example 1

I₂(3 mg/mL) in F6H8 and EtOH 1.5% (w/w) as prepared in Example 1

Organisms tested included *Staphylococcus aureus* ATCC 6538, *Pseudomonas aeruginosa* ATCC 9027, *Streptococcus pneumoniae* ATCC 33400, *Proteus mirabilis* ATCC 14153, *Candida albicans* ATCC 10231, *Propionibacterium acnes* ATCC 11828 (Anaerobier).

Both formulations exhibited inhibition zones, demonstrating effectiveness in inhibition of growth of the organisms tested.

Example 4—Staining

Solutions of I₂ in F4H5 (0.85 mg/mL) and I₂ in F6H8 (0.85 mg/mL) were tested in respect of staining on skin. A drop of the respective compositions were administered to skin on a subject. On initial contact with the skin, the pink colour which is characteristic of the compositions was still observed, however vanished within a few seconds. No staining of the skin was observed. No color was transferred to a piece of fabric in contact with the skin afterwards.

No staining of the skin was also observed for solutions of I₂ at lower concentrations in F4H5 or F6H8, as well as with the compositions of molecular iodine in F8H8, F10H5 and perfluorodecalin when these were topically applied to the skin.

Further staining tests are conducted on cellulose filter paper (Whatman, CAT No. 1005-070). In the tests, a drop of the respective formulations (ca. 50 µL) was added to the filter paper. The staining behaviour of the compositions were recorded afterwards.

No staining of the Whatman filter was observed for solutions of I₂ in F4H5, F6H8, F8H8, F10H5 and perfluorodecalin.

When the same tests were conducted for the compositions of I₂ prepared in the F4H5/ethanol and F6H8/ethanol liquid vehicles of Example 1, staining of the Whatman filter paper was observed. Yellow-brown spots on the site of administration were left behind on the filter paper.

Example 5—Further Compositions—Stability and Staining

Further compositions of molecular iodine, in a vehicle consisting of a semifluorinated alkane and optionally a co-solvent as summarized in Table 1 below were prepared according to the methods described in Example 1, and then tested similarly to the protocols described in Examples 2 and 4 as to their stability and for staining.

Stability—Vials containing the compositions were visually inspected to determine stability. Visual inspection was performed directly after fabrication, after 1 day and after 1 week. A “Yes” in Table 1 indicates the recording of no visual changes, such as discoloration of the initial pink color of the composition, and no formation of precipitates or deposits.

Staining—Staining was assessed as detailed in Example 4 on cellulose filter paper. A “No” in Table 1 indicates that no staining effects were observed upon application of a sample of the composition to the cellulose filter.

TABLE 1

Vehicle 1 (V1)	Vehicle 2 (V2)	Ratio V1:V2 (w/w)	Ratio V1:V2 (v/v)	I ₂ (mg/mL)	I ₂ (ppm)	Staining	Stability
F6H2	—	—	—	0.01	10	yes	no
F6H2	—	—	—	0.05	50	yes	no
F6H2	—	—	—	0.10	100	yes	no
F6H2	—	—	—	0.25	250	yes	no
F6H2	—	—	—	0.50	500	yes	no
F4H5	—	—	—	0.01	10	yes	no
F4H5	—	—	—	0.05	50	yes	no
F4H5	—	—	—	0.10	100	yes	no
F4H5	—	—	—	0.25	250	yes	no
F4H5	—	—	—	0.50	500	yes	no
F4H5	—	—	—	1.00	1000	yes	no
F6H8	—	—	—	0.01	10	yes	no
F6H8	—	—	—	0.05	50	yes	no
F6H8	—	—	—	0.10	100	yes	no
F6H8	—	—	—	0.25	250	yes	no
F6H8	—	—	—	0.50	500	yes	no
F6H8	—	—	—	1.00	1000	yes	no
F8H8	—	—	—	0.10	100	yes	no
Perfluoro- octane	—	—	—	0.10	100	yes	no
F6H8	paraffin wax	4:1	4:1	0.20	200	yes	no
F6H2	F4H5	1.17:1	1:1	0.25	250	yes	no
F6H2	F6H8	1.12:1	1:1	0.125	125	yes	no
F4H5	F6H8	0.96:1	1:1	0.125	125	yes	no
F6H8	Decamethylcyclo- pentasiloxan	1.41:1	4:1	0.20	200	yes	no
2-perfluoro- hexyl-octane	—	—	—	0.50	500	yes	no
F6H8	Squalane	8.33:1	5:1	0.208	208	yes	no
F4H5:F6H8 (1:1)	Squalane	8.15:1	5:1	0.227	227	yes	no
F6H8	paraffinum perliquidum	6.67:1	4:1	0.80	800	yes	no
F4H5	paraffinum perliquidum	6.37:1	4:1	0.80	800	yes	no
F6H8	mineral oil	6.67:1	4:1	0.80	800	yes	no
F4H5	mineral oil	6.37:1	4:1	0.80	800	yes	no
F6H8	octane	7.71:1	4:1	0.80	800	yes	no
F4H5	octane	7.37:1	4:1	0.80	800	yes	no

The composition of molecular iodine in a vehicle based on 4:1 ratio of F6H8 and paraffin wax was a solid formulation.

A composition comprising a semifluorinated alkane, molecular iodine and an exemplary active ingredient, cyclosporine, was also prepared. The same tests as conducted for compositions of Table 1 was conducted. The composition was also found to be stable and non-staining.

TABLE 2

Vehicle 1 (V1)	Vehicle 2 (V2)	Ratio V1:V2 (w/w)	Ratio V1:V2 (v/v)	API (mg/mL)	I ₂ (mg/mL)	I ₂ (ppm)	Stability	Staining
F6H8	F4H5	10.5:1	10:1	cyclosporine (0.091)	0.27	227	yes	no

Example 6—EpiOcular Eye Irritation Test

The following compositions were evaluated in terms of risk for eye irritation. A model of ocular irritation, the EpiOcular Eye Irritation Test (OCL-200-EIT; MaTek Corporation), was used. The EpiOcular protocol is approved as a new OECD Test Guideline No. 492 (Reconstructed human Cornea-like Epithelium (RhCE) test method for identifying chemicals not requiring classification and labeling for eye irritation or serious eye damage).

The following compositions were evaluated:

- (a) 1.0 mg/ml iodine (1000 ppm) in 1-perfluorobutyl-pentane (F4H5)
- (b) 1.0 mg/ml iodine (1000 ppm) in 1-perfluorohexyl-octane (F6H8)
- (c) 0.5 mg/ml iodine (500 ppm) in 1-perfluorohexyl-ethane (F6H2).

A test material-treated tissue viability of higher than 60% relative to the negative control-treated tissue is classified as a non-irritant; a test material-treated tissue viability below 60% is considered as an irritant.

These compositions, which have relatively high concentrations of iodine, were unexpectedly found to be non-irritating. Iodine ophthalmic products in the art based on povidone-iodine complex aqueous formulations typically provide significantly lower concentrations of molecular iodine, for example about 2.5 ppm. All of the tested compositions of molecular iodine in semifluorinated alkane were

classified based on this model as non-irritating, with results of (a) 95.4%, (b) 98.4% and (c) 107.8% tissue viability respectively.

Example 7—Ex Vivo Eye Irritation Test (EVEIT)

A comparison in respect of corneal healing process was conducted for a composition comprising molecular iodine (0.25 mg/ml; 250 ppm) in 1-perfluorohexyl-octane (Composition B) with vehicle 1-perfluorohexyl-octane (Composition A) and hyaluronic acid (HYLO-COMOD®) as a reference and 0.01% BAC (benzalkonium chloride) as a control using an Ex Vivo Eye Irritation Test (EVEIT), similar to as described in M. Frentz et al, *Altern. to Lab. Anim.*, 2008 (36) p 25-32; and N. Schrage et al, *Graefes Arch Clin Exp Ophthalmol* 2012 (250), 1330-1340).

Method. Rabbit corneas were obtained and placed in an artificial anterior ocular chamber which was gently filled with serum-free minimal essential medium (Eagle's MEM) containing Earle's salts and HEPES buffer for nutrition. The medium was constantly replenished by a micropump to imitate the physiological condition of the eye. The culture chambers were held at 32° C. under normal air without supplementary CO₂ and >95% relative humidity. Five corneas per test substance (n=5) were used except for the positive control with which two corneas (n=2) were tested.

After 12 h of stabilization in the culture chamber, the corneas were evaluated by microscopy and corneas with intact epithelium and without opacities were selected. Four small abrasions (2.3-4.3 mm²) were applied to the surface of the selected corneas with a cornea drill. All defects were monitored by fluorescein sodium staining (0.17% aq. solution) and microscopy.

The test substances were administered one hour after induction of the corneal erosion and were applied six times daily onto the apex of the corneas (30-50 µL every four hours). A soft-tipped cannula, with continuous suction was placed on the lowest part of the corneoscleral region within the culturing chamber to remove any excess fluid. Experiments were terminated after 3 days of application. Biomicroscopic images of the corneas were taken daily to document the corneal healing process using a phase-contrast microscope integrated camera (KY-F1030U, JVC, (Bad Vilbel, DE) mounted on a Z16 APO Microscope (Wetzlar, DE)). All defects were monitored by fluorescein sodium stains (0.17% aq. solution) with yellow green fluorescence indicating the areas of epithelial defects. Erosion sizes were determined using a software tool of the microscope (DIS-KUS). At the end of the 3 days, the experiment was terminated and all corneas were fixed in 3.7% formaldehyde and stained with a hematoxylin-eosin dye for microscopic evaluation. To monitor the metabolic activity of the cornea, glucose and lactate concentrations were photometrically quantified in the outflow medium from the artificial anterior chambers.

Results. The composition comprising molecular iodine in F6H8 (Composition B) did not harm the corneal healing process after the induction of corneal erosion. Instead, a similar positive outcome in respect of corneal healing was observed with Composition A (F6H8 alone) or with the standard reference hyaluronic acid composition (HYLO-COMOD®).

TABLE 3

Corneal Erosion Size Measurements/Mean mm ² (SD)				
Composition	Day 0	Day 1	Day 2	Day 3
A (n = 5); F6H8	11.29 (1.02)	3.51 (0.69)	0 (0)	0 (0)
B (n = 5); molecular iodine in F6H8	12.09 (0.86)	3.37 (1.17)	0 (0)	0 (0)
0.01% BAC	12.91 (0.32)	7.42 (0.03)	26.40 (2.18)	42.08 (0.43)
HYLO	12.12 (0.51)	3.59 (1.07)	0 (0)	0 (0)
COMOD®				

No significant differences in terms of an effect of positive corneal healing was noted between composition A and composition B (comprising 250 ppm molecular iodine). With both compositions, as with the reference composition (HYLO COMOD®), the mechanically induced epithelial erosions were found to be significantly reduced and essentially absent after day 2 of treatment. Furthermore, no corneal toxicity, based on the metabolic activity as indicated by the glucose/lactate measurements was observed for these compositions. In significant contrast, the control comprising 0.01% of the preservative BAC, a progressive increase of the induced epithelial lesions was observed over the course of the three days of the experiment.

Example 7—Antimicrobial Effectiveness Test (F6H2)

Formulations of molecular iodine dissolved in the semi-fluorinated alkane F6H2 (0.01, 0.05, 0.1, 0.25 and 0.5 mg/ml) were tested for antimicrobial effectiveness.

Staphylococcus aureus (10⁷ cells) were plated onto an inert plate. After a short drying time, 100 µl of the formulations of molecular iodine dissolved in perfluorohexyl-ethane (F6H2) were applied to the germs on the plate. After a specified contact time, the remaining *Staphylococcus aureus* cells were collected by shaking out with medium and transferred and plated to an agar plate. After incubation at 37° C. overnight, the *Staphylococcus aureus* cell counts were determined.

It was observed, within a contact time of 30 seconds of the *Staphylococcus aureus* cells on the plate, that all of the tested iodine formulations (0.01, 0.05, 0.1, 0.25 and 0.5 mg/ml) in F6H2 lead to a complete killing of all germs. No viable cell counts were observed after the incubation step overnight.

The invention claimed is:

1. A method for treatment of a disease or condition affecting the skin, mucosal tissue, eye, or ophthalmic tissue in a subject in need thereof, wherein the disease or condition is associated with or caused by a microorganism selected from bacteria, demodex, fungi, virus, or a combination thereof, comprising topical application of a composition to the skin, mucosal tissue, eye, or ophthalmic tissue of the subject, wherein the composition consists of:

- molecular iodine, and
 - a vehicle;
- wherein the vehicle consists of (i) a semifluorinated alkane and optionally (ii) a non-polar cosolvent, wherein the cosolvent is not a siloxane; and wherein the semifluorinated alkane is of Formula (I) shown below:



Formula (I)

wherein n is an integer selected from 1 to 12, and R_m is a linear or branched alkyl or cycloalkyl with m carbon atoms, wherein m is an integer selected from 2 to 12.

2. The method according to claim 1, wherein the composition is non-staining when administered or coming into contact with a surface selected from the group consisting of skin, dermal tissue, eye, eye tissue and fabric.

3. The method according to claim 1, wherein the disease or condition is a disease or condition:

a) selected from the group consisting of blepharitis, bacterial conjunctivitis, ocular dryness, follicular conjunctivitis, giant papillary conjunctivitis, corneal ulcer, keratitis, conjunctival neoplasia, HSV keratitis, eye allergy, ulcerative infectious keratitis, epithelial keratitis, stromal keratitis, neonatal conjunctivitis and keratoconjunctivitis epidemica; or

b) selected from the group consisting of a wound, a cut, a surgical wound, burn, abrasion, laceration, ulcer, rash, sore, acne, and infection.

4. The method according to claim 1, wherein the semifluorinated alkane is selected from the group consisting of $CF_3(CF_2)_3-(CH_2)_4CH_3$ (F4H5), $CF_3(CF_2)_3-(CH_2)_5CH_3$ (F4H6), $CF_3(CF_2)_5-CH_2CH_3$ (F6H2), $CF_3(CF_2)_5-(CH_2)_5CH_3$ (F6H6), $CF_3(CF_2)_5-(CH_2)_7CH_3$ (F6H8), $CF_3(CF_2)_3-(CH_2)_7CH_3$ (F4H8), $CF_3(CF_2)_7-(CH_2)_7CH_3$ (F8H8) and $CF_3(CF_2)_9-(CH_2)_4CH_3$ (F10H5).

5. The method according to claim 1, wherein the composition consists of molecular iodine and a vehicle, wherein the vehicle consists of the semifluorinated alkane.

6. The method according to claim 1, wherein the composition consists of molecular iodine dissolved in F4H5, F6H2, F6H8, or a mixture thereof.

7. The method according to claim 1, wherein the composition consists of up to about 1.0 mg/mL of molecular iodine dissolved in a vehicle consisting of 1-perfluorohexyl-octane (F6H8) or 1-perfluorobutyl-pentane (F4H5).

8. The method according to claim 1, wherein the composition consists of up to about 0.5 mg/ml of molecular iodine dissolved in 1-perfluorohexyl-ethane (F6H2).

9. The method according to claim 1, wherein the composition consists of molecular iodine and a vehicle, wherein the vehicle consists of the semifluorinated alkane and the non-polar cosolvent.

10. The method according to claim 9, wherein the non-polar cosolvent is selected from the group consisting of a saturated hydrocarbon, a perfluorocarbon, and a combination thereof.

11. The method according to claim 1, wherein the composition consists of up to about 1.0 mg/mL of molecular iodine dissolved in a vehicle consisting of 1-perfluorohexyl-octane (F6H8) or 1-perfluorobutyl-pentane (F4H5) and optionally a non-polar cosolvent, wherein the cosolvent is not a siloxane.

12. The method according to claim 1, wherein the composition is topically administered to an ophthalmic tissue selected from the group consisting of the cornea, conjunctiva, meibomian glands, eye lid margins, and eyelashes.

13. The method of claim 1, wherein the non-polar cosolvent is present and does not support the formation of any one of, or combination of, the following: iodide ion (I^-), triiodide ion (I_3^-), iodate ion (IO_3^-), or hypiodate ion (IO^-).

14. The method of claim 1, wherein the non-polar cosolvent is present and is selected from saturated, unbranched or branched, hydrocarbons.

15. The method according to claim 14, wherein the non-polar cosolvent is present and selected from C10 to C30, or from C10 to C40 saturated, unbranched or branched hydrocarbons.

16. The method according to claim 1, wherein the disease or condition affects the eye or ophthalmic tissue, and the composition is applied to the eye or ophthalmic tissue of the subject.

17. The method according to claim 1, wherein the disease or condition affects the skin or mucosal tissue, and the composition is applied to the skin or mucosal tissue of the subject.

18. The method of claim 17, wherein the semifluorinated alkane is selected from the group consisting of $CF_3(CF_2)_3-(CH_2)_4CH_3$ (F4H5), $CF_3(CF_2)_3-(CH_2)_5CH_3$ (F4H6), $CF_3(CF_2)_5-CH_2CH_3$ (F6H2), $CF_3(CF_2)_5-(CH_2)_5CH_3$ (F6H6), $CF_3(CF_2)_5-(CH_2)_7CH_3$ (F6H8), $CF_3(CF_2)_3-(CH_2)_7CH_3$ (F4H8), $CF_3(CF_2)_7-(CH_2)_7CH_3$ (F8H8) and $CF_3(CF_2)_9-(CH_2)_4CH_3$ (F10H5).

19. The method according to claim 17, wherein the composition consists of molecular iodine and a vehicle, wherein the vehicle consists of the semifluorinated alkane.

20. The method according to claim 17, wherein the composition consists of molecular iodine dissolved in F4H5, F6H2, F6H8, or a mixture thereof.

21. The method according to claim 17, wherein the composition consists of up to about 1.0 mg/mL of molecular iodine dissolved in a vehicle consisting of 1-perfluorohexyl-octane (F6H8) or 1-perfluorobutyl-pentane (F4H5), and optionally a non-polar cosolvent, wherein the cosolvent is not a siloxane.

22. The method of claim 17, wherein the non-polar cosolvent is present and selected from saturated, unbranched or branched, hydrocarbons.

23. A method for providing an antiseptic or disinfectant action in a subject in need thereof, comprising application of a composition to the skin, mucosal tissue, eye or ophthalmic tissue of the subject, wherein the composition consists of:

- a) molecular iodine, and
- b) a vehicle;

wherein the vehicle consists of (i) a semifluorinated alkane and optionally (ii) a non-polar cosolvent, wherein the cosolvent is not a siloxane; and wherein the semifluorinated alkane is of Formula (I) shown below:



wherein n is an integer selected from 1 to 12, and R_m is a linear or branched alkyl or cycloalkyl with m carbon atoms, wherein m is an integer selected from 2 to 12.

24. The method of claim 23, wherein the non-polar cosolvent is present and is selected from saturated, unbranched or branched, hydrocarbons.

25. The method according to claim 24, wherein the non-polar cosolvent is present and selected from C10 to C30, or from C10 to C40 saturated, unbranched or branched hydrocarbons.

* * * * *

UNITED STATES PATENT AND TRADEMARK OFFICE
CERTIFICATE OF CORRECTION

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INVENTOR(S) : Löscher et al.

Page 1 of 1

It is certified that error appears in the above-identified patent and that said Letters Patent is hereby corrected as shown below:

In the Specification

Column 25, Line 4, in the heading of Table 1, “Staining” should be changed to “Stability” and
“Stability” should be changed to “Staining”

Signed and Sealed this
Seventeenth Day of June, 2025



Coke Morgan Stewart
Acting Director of the United States Patent and Trademark Office