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(54) **PROCESS FOR PREPARING
INTERMEDIATES FOR THE SYNTHESIS OF
OPTICALLY ACTIVE BETA-AMINO
ALCOHOLS BY ENZYMATIC REDUCTION
AND NOVEL SYNTHESIS INTERMEDIATES**

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CPC **C07D 405/06** (2013.01); **C07D 207/408** (2013.01); **C12P 13/02** (2013.01); **C07B 2200/07** (2013.01)

(58) **Field of Classification Search**

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See application file for complete search history.

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(57) **ABSTRACT**

Subject-matter of the present invention is a process for preparing intermediates for the synthesis of optically active beta-amino alcohols by enzymatic reduction of the corre- sponding beta-amino ketones. Subject-matter of the inven- tion are also said novel synthesis intermediates and the use thereof in the preparation of active pharmaceutical ingredi- ents, among which vilanterol and the salts thereof.

5 Claims, No Drawings

Specification includes a Sequence Listing.

1

**PROCESS FOR PREPARING
INTERMEDIATES FOR THE SYNTHESIS OF
OPTICALLY ACTIVE BETA-AMINO
ALCOHOLS BY ENZYMATIC REDUCTION
AND NOVEL SYNTHESIS INTERMEDIATES**

This application is a U.S. national stage of PCT/IB2018/060335 filed on 19 Dec. 2018, which claims priority to and the benefit of Italian Patent Application No. 102017000147611 filed 20 Dec. 2017, the contents of which are incorporated herein by reference in their entireties.

A subject-matter of the present invention is a process for preparing intermediates for the synthesis of optically active beta-amino alcohols by enzymatic reduction of the corresponding beta-amino ketones. Subject-matter of the invention are also said novel synthesis intermediates and the use thereof in the preparation of active pharmaceutical ingredients, among which vilanterol and the salts thereof.

TECHNICAL FIELD

Amino alcohols, in particular chiral phenyl-beta-amino alcohols are very important synthons for the synthesis of active pharmaceutical ingredients; their basic structure is for example present in the hormones adrenalin and nor-adrenalin, in the drugs used for to treat cardiac problems, for example sotalol and arbutamine, in those used for treating asthma or chronic obstructive pulmonary disease (COPD) such as vilanterol, salbutamol, colterol, isoproterenole, and in different natural products such as for example the natural antiviral product (+) tembamide.

Optically active beta-amino alcohols are also of industrial interest as they can be used as chiral ligands or auxiliaries in different types of asymmetric syntheses.

Given the importance of such molecules, a number of synthesis methods have been developed over the years.

Initially the most used synthesis path, as it is promising in terms of optical purity, was the chiral resolution by optically active chemical compounds of the racemic amino alcohol but unfortunately such a synthesis path was not convenient in terms of yield.

Recently different enantioselective synthesis methods have been developed, that are more effective than the resolution, in terms of yield. For example, methods that are based on the chemical reduction of ketones and methods based on the hydrogenation in the presence of stereoselective catalysts and ligands are known.

The enantioselective reductions of chemical type imply the use of potentially hazardous reagents, very expensive ligands and in general do not yield to alcohols with enantiomeric excess higher than 97-98%. Therefore the raw compounds obtained need a subsequent purification.

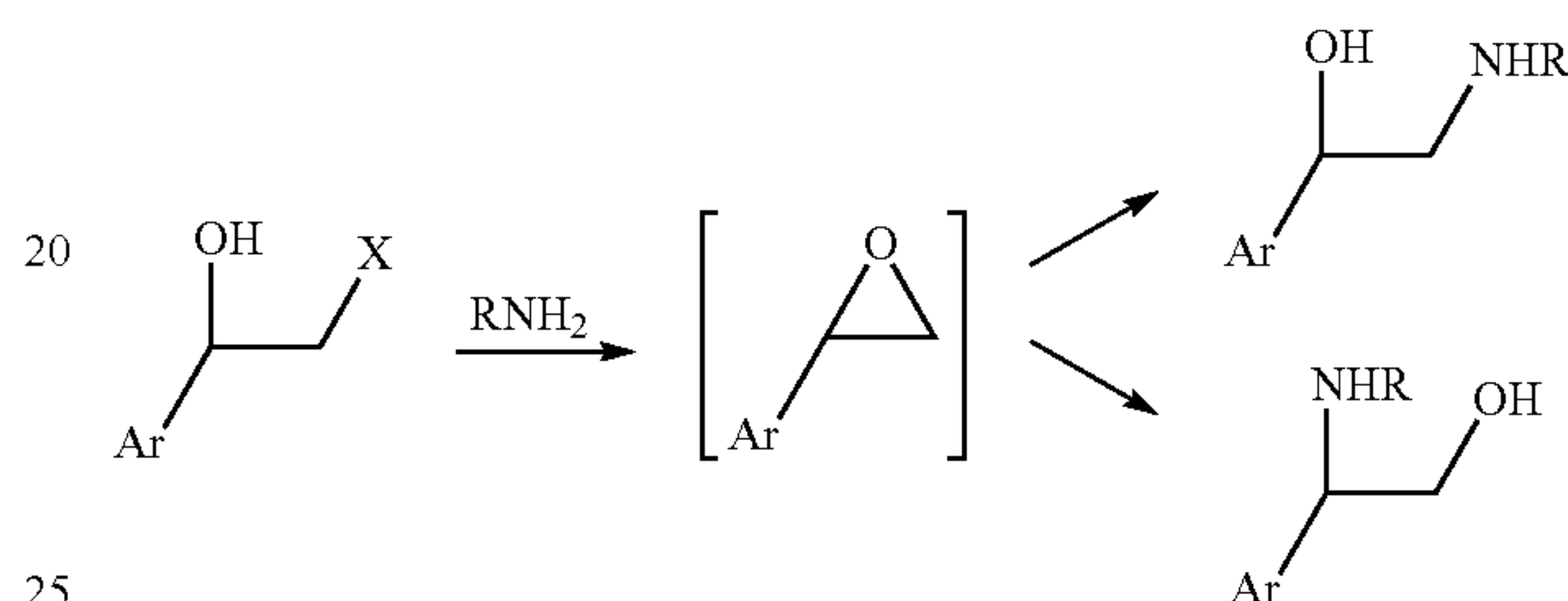
The hydrogenation often involves the use of high pressures, expensive metal catalysts and often yields to impurities due to an excessive reduction ("overreduction") or to secondary reactions on other parts of the molecule.

The use of biochemical agents able to reduce carbonyl groups to hydroxy groups is also known. Among the biochemical methods the most used in this type of reaction is the use of yeasts. However, their use brings to many disadvantages such as the difficult reproducibility of the results over time because of the proliferation of hybrid breeds of yeast which tend to progressively change, because of the need of very high dilutions and the difficult collection of the desired product due to the presence of high amounts of biomasses.

2

Another biochemical method provides for the use of enzymes for the enantioselective reduction of carbonyl groups in particular of alpha halogen-ketones, as described in the Application WO2017/001907. This method provides for the reduction of a halogen ketone to halogen alcohol, followed by an alkylation with an amine, to yield the desired amino alcohol. The disadvantage is that, in the presence of an amine (base), the halogen alcohol tends to give the corresponding epoxide, which, in the presence of the amine itself, is opened in the 1 and 2 positions with low selectivity, bringing to the formation of by-products.

Reaction Scheme:



For this reason, in the procedure used in WO2017/001907, at the end of the enzymatic reduction, the benzyl OH is protected, for being deprotected after the alkylation of the amine.

With the aim of solving the problems above, the Applicant tried the direct enzymatic reduction of beta-amino ketones to beta-amino alcohols, so as to have the amine group already introduced in the desired position. However the use of enzymes on this type of substrates showed considerable difficulties and it was not possible to effectively carry out the reduction.

OBJECTS OF THE INVENTION

It is an object of the invention to provide a process suitable for preparing intermediates for the synthesis of optically active beta-amino alcohols, with good yields and high enantiomeric excesses even on industrial scale.

It is another object of the invention to provide an enzymatic process suitable for preparing intermediates for the synthesis of optically active beta-amino alcohols, that overcomes the disadvantages of the prior art.

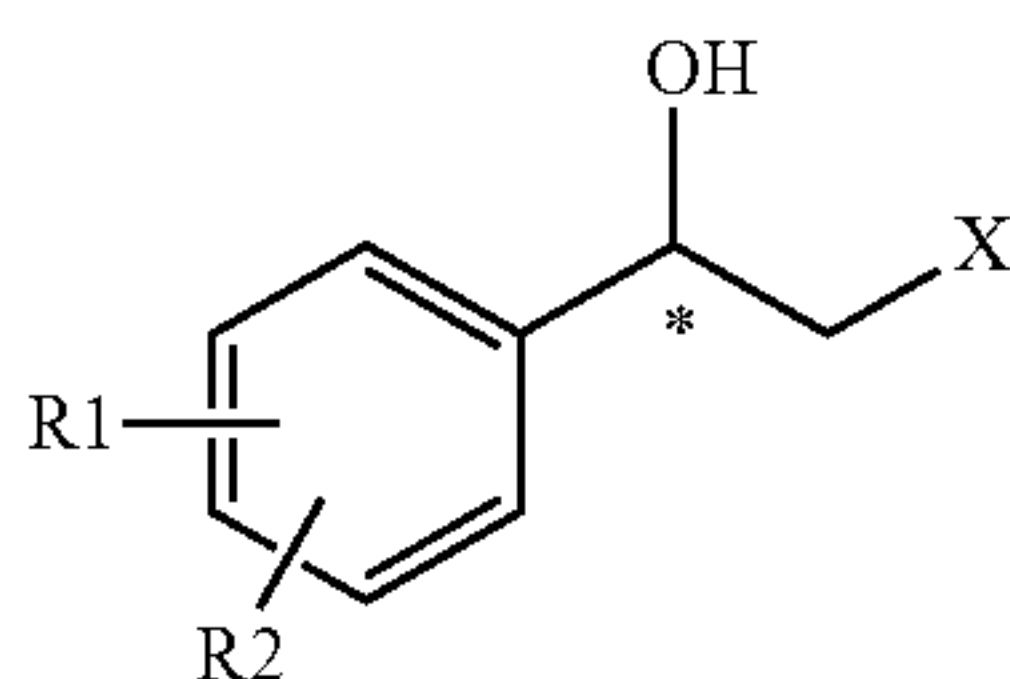
It is a further object of the invention to provide novel intermediates useful in particular, but not only, for preparing vilanterol, adrenalin, noradrenalin, R-salbutamol, R-colterol, R-isoproterenole, (-)-arbutamine and the salts thereof.

DESCRIPTION OF THE INVENTION

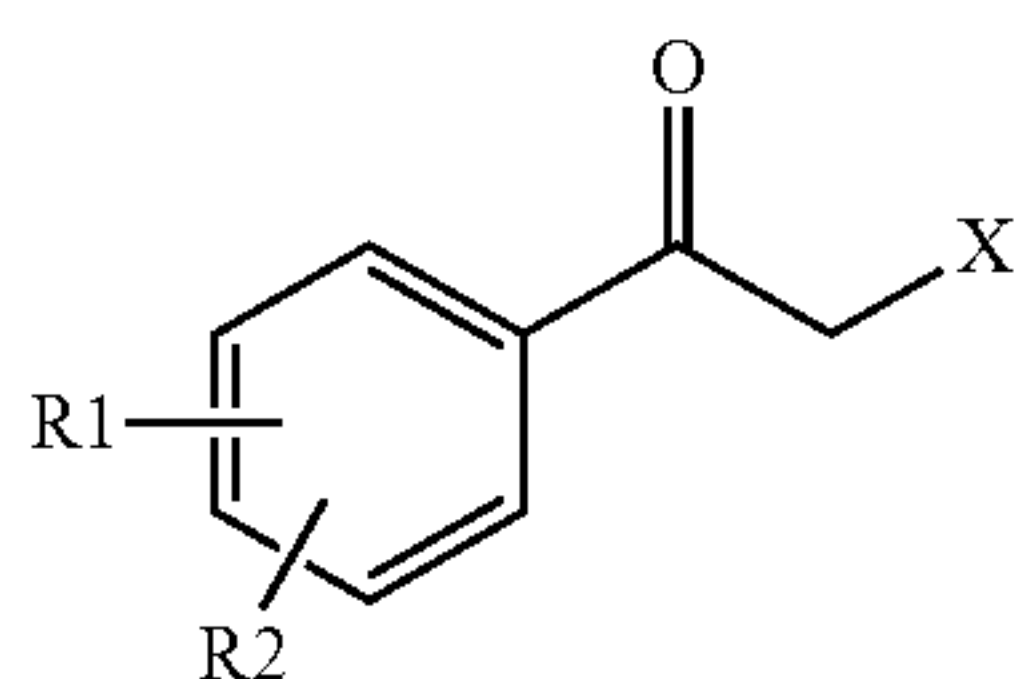
The Applicant has surprisingly found that, by using a beta-amino ketone wherein the amine group has lost its basicity, for example by transforming to an amide, imide or carbamate group, it is possible to carry out its enzymatic reduction and obtain optically active amino-alcohols, with high yields and purities.

3

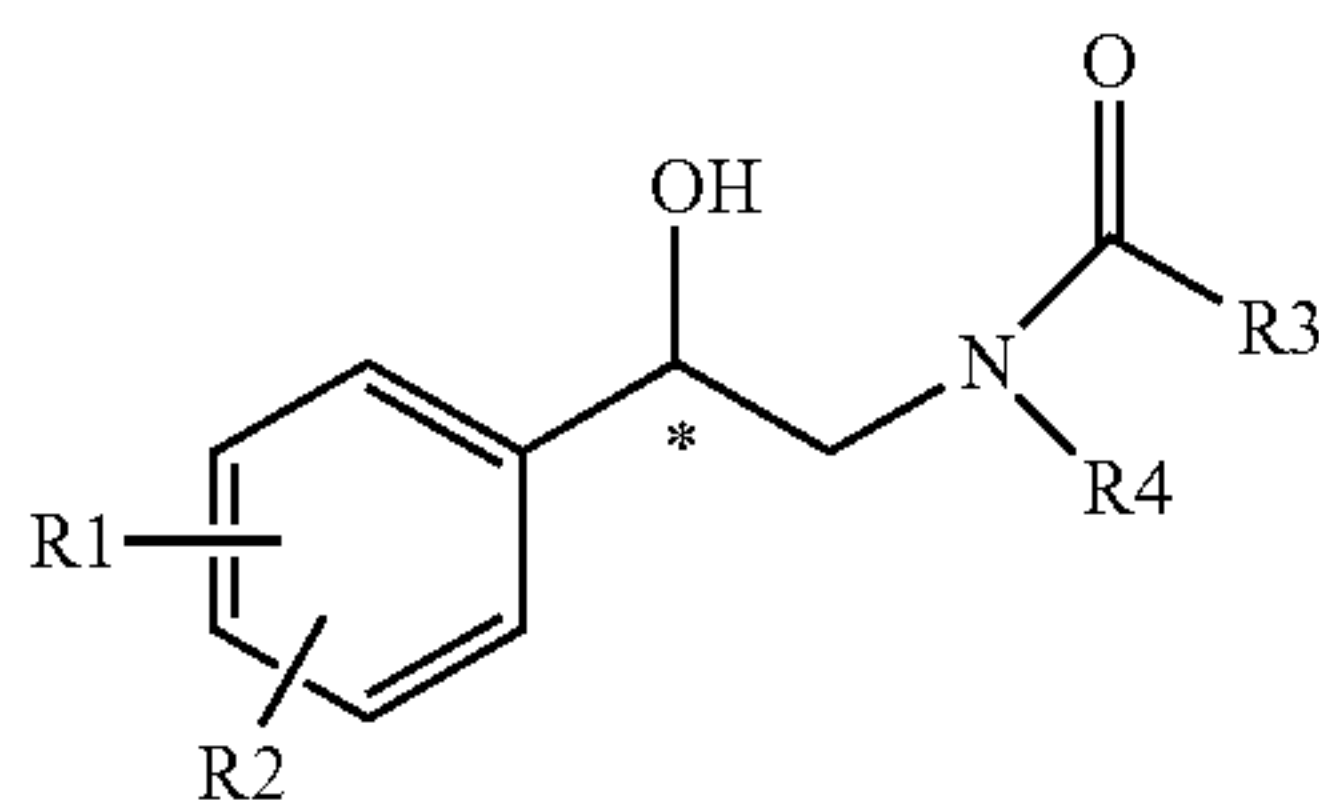
Thus, according to one of its aspects, subject-matter of the invention is a process for preparing an optically active compound of Formula (I)



wherein
the asterisk indicates the chiral carbon is in the (S) form or (R) form;
X represents an amide, imide or carbamate residue bound to the compound of formula (I) through the nitrogen atom;
R₁ and R₂ are independently selected from hydrogen, —OPr, —CH₂OPr, —OH, —CH₂OH and C₁-C₄-alkoxy; wherein Pr is a hydroxy protective group and, when two Pr protective groups are in the compound of formula (I):
said two Pr protective groups can be the same or different from one another; or
said two Pr protective groups, together with the oxygen atoms to which they are bound, may form a cycle fused with the benzene;
said process comprising the enzymatic reduction of the compound of Formula (II)



wherein X, R₁ and R₂ are as defined above.
According to a preferred embodiment, the compound of Formula (I) is a compound of Formula (I')



wherein
the asterisk, R₁ and R₂ are as defined above;
R₃ is selected from an hydrogen, alkyl, alkoxy, aryl, alkylaryl and arylalkyl;
R₄ is selected from hydrogen, alkyl, aryl, alkylaryl and arylalkyl;
or else
R₃ and R₄, together with the carboxamide group (HN—C=O) bonding them, constitute an imide.
The expression “chiral carbon is in the (S) form or in the (R) form” herein means that at least 80%, preferably at least

4

90-95%, more preferably 98-99.9% of the compound of Formula (I) and (I') is in said configuration.

According to a preferred embodiment, the chiral carbon of the compound of Formula (I) and (I') is in the (R) configuration.

The term “alkyl” herein means a saturated, linear or branched, alkyl residue, preferably having 1 to 6 carbon atoms, advantageously 1 to 4 carbon atoms, for example the methyl or ethyl group.

The term “alkoxy” herein means a saturated, linear or branched, alkoxy residue, preferably having 1 to 6 carbon atoms, advantageously 1 to 4 carbon atoms, for example the methyl or ethyl group.

The term “aryl” herein means an aromatic hydrocarbon residue, preferably selected from phenyl and naphthyl.

The term “alkylaryl” herein means an aromatic hydrocarbon residue, preferably selected from phenyl and naphthyl, substituted with one or more alkyls, as defined above.

The term “arylalkyl” herein means a saturated, linear or branched, alkyl residue, preferably having 1 to 6 carbon atoms, advantageously 1 to 4 carbon atoms as defined above, substituted with one or more aryls, as defined above.

The expression “hydroxy protective group” means any protective group able to protect the hydroxy function without interfering with the reduction reaction. Such groups can be for example selected from those mentioned in T. W. Greene, John Wiley & Sons, Ltd, “*Protective Groups in Organic Synthesis*”, 5th edition, 2014. According to a preferred embodiment of the invention, the —OPr groups, as defined above, include esters and acetonides.

Other protective groups according to the present invention include the silanes, for example the silyl group; alternatively, when there are two Pr groups, said silane protective groups, together with the two oxygen atoms to which they are bound, may form a cycle fused with the benzene.

According to another embodiment, when the groups R₁ and R₂ are on adjacent carbon atoms and there are two Pr groups, said two Pr groups represent a carbonyl group (=CO) bound to the two oxygen atoms.

According to a preferred embodiment, R₁ and R₂ are both —OH or OPr, wherein the Pr groups can be the same or different from one another, preferably the same.

According to another preferred embodiment, one of R₁ and R₂ is OH or —OPr and the other is —CH₂OH or —CH₂OPr.

According to another preferred embodiment, R₁ and R₂ are different from hydrogen and are on adjacent positions on the benzene ring, preferably in the 3 and 4 positions of the benzene ring.

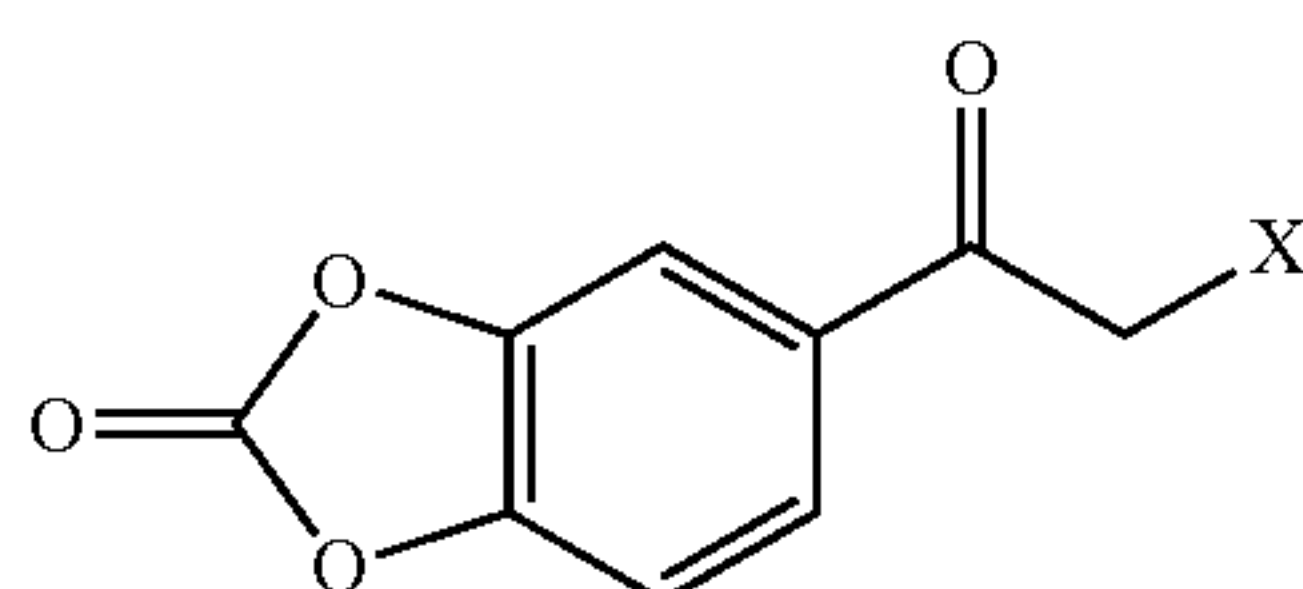
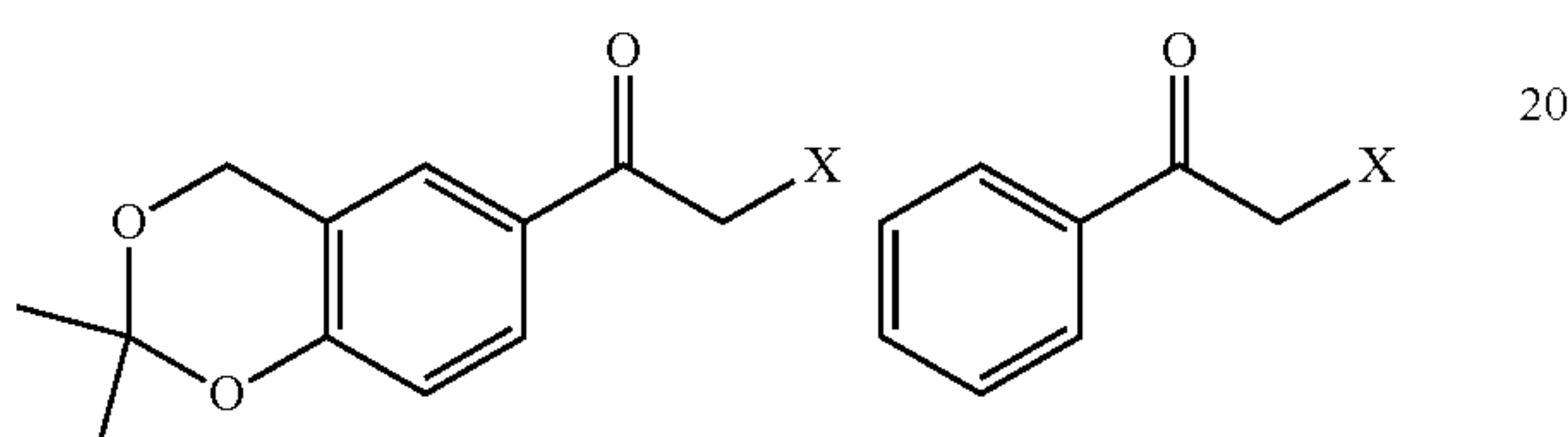
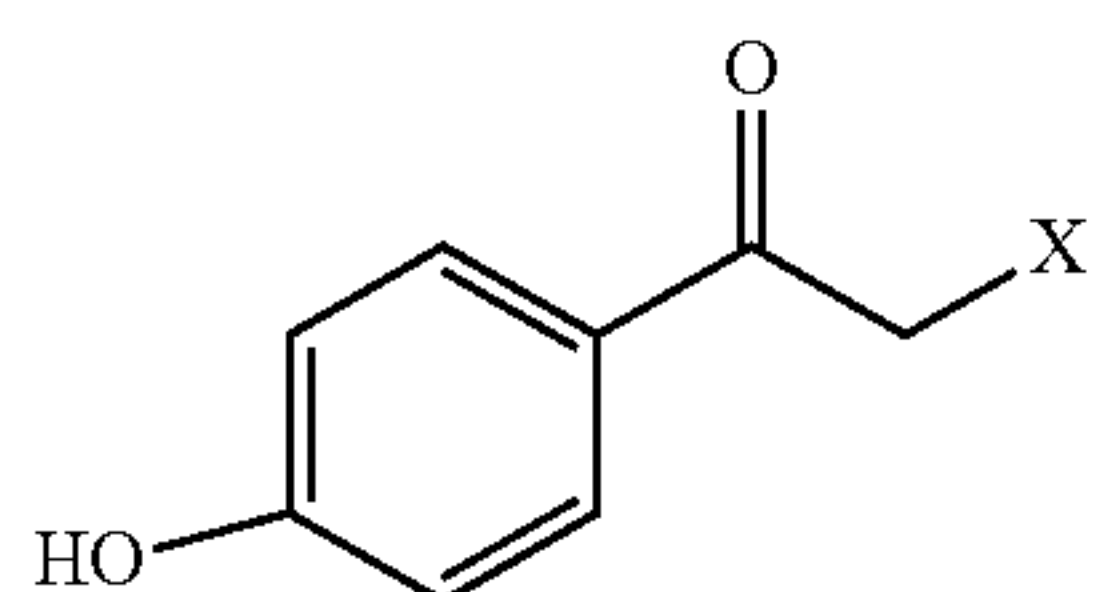
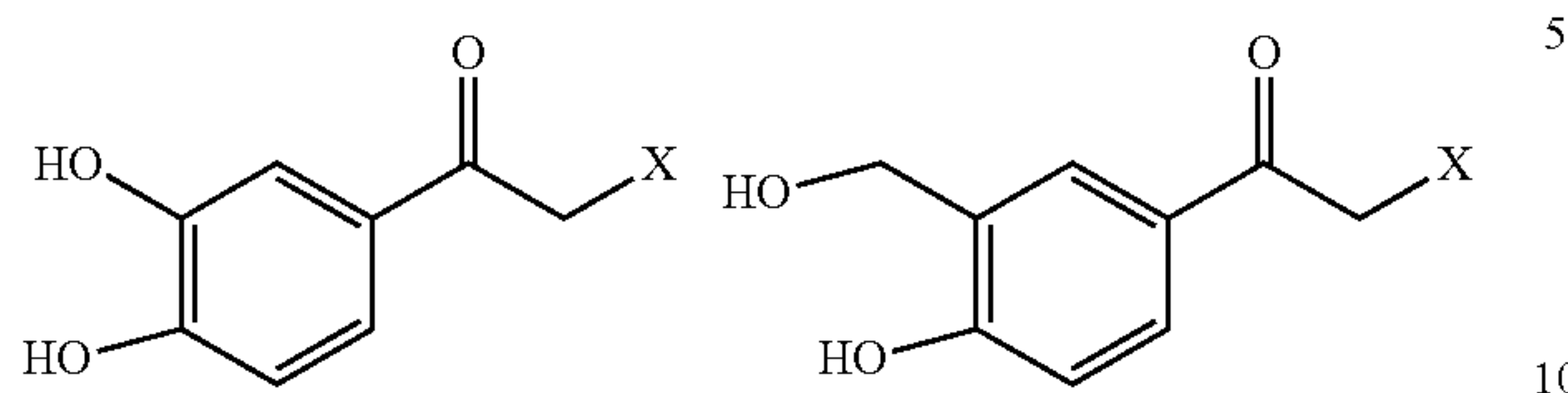
According to another preferred embodiment, when R₁ and R₂ are each independently —OPr or —CH₂OPr, they form a cycle fused with the benzene ring; in this case, according to a more preferred embodiment, they are on adjacent positions of the benzene ring.

According to a preferred embodiment, R₃ is hydrogen, saturated and linear C₁-C₆ alkyl, preferably methyl or ethyl and R₄ is hydrogen or methyl.

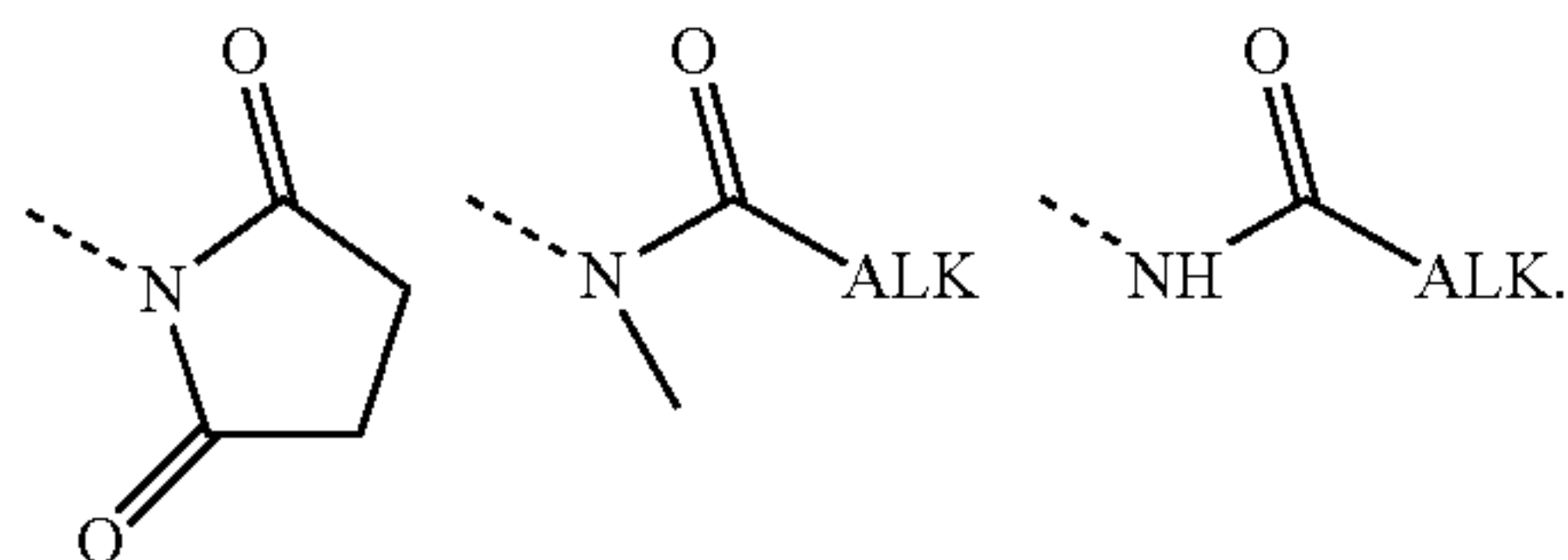
According to another preferred embodiment, R₃ and R₄, together with the carboxamide group bonding them, represent a succinimide or a phthalimide possibly substituted, advantageously the succinimide or substituted succinimide, for example with one or more fluorine atoms.

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Preferred compounds of Formula (II) are selected from the following:

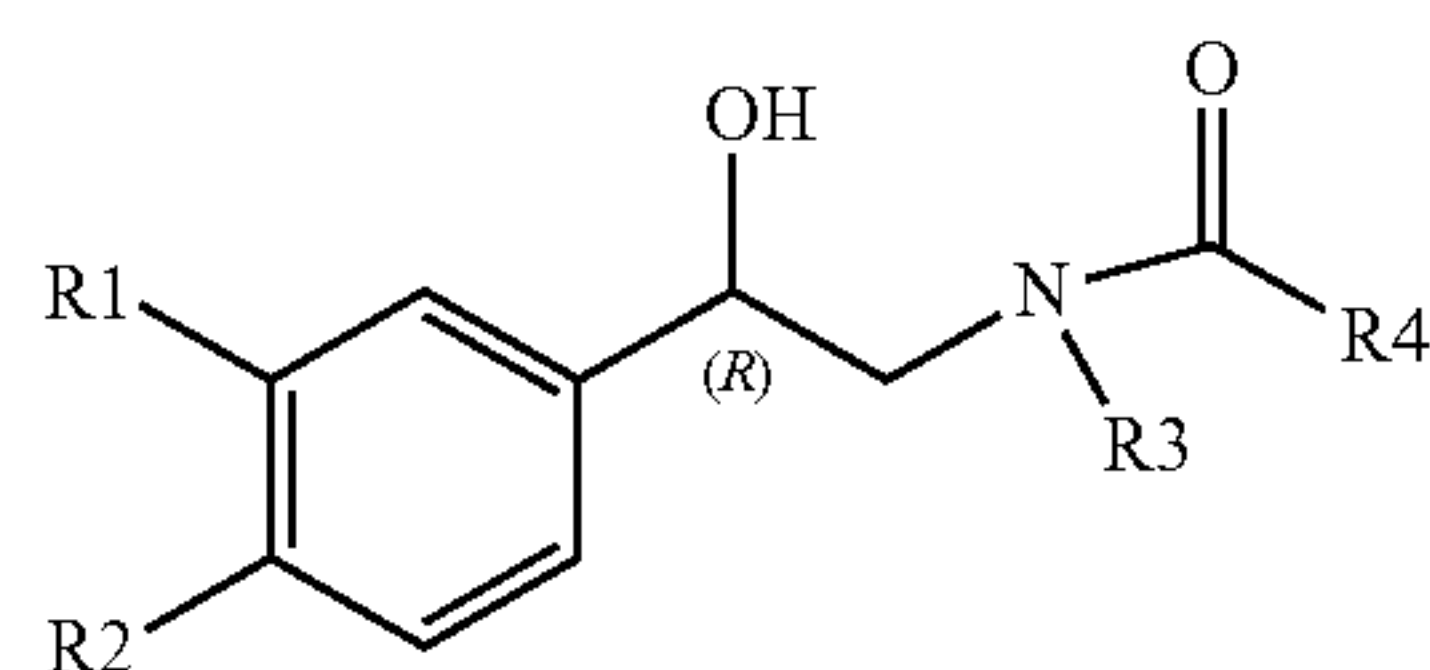


wherein X is a substituent selected from



wherein ALK represents an alkyl as defined above, advantageously methyl or ethyl, said substituent X being bound to the molecule through the nitrogen atom, as depicted by the dashed line.

According to a particularly preferred embodiment, subject-matter of the invention is a process for preparing an optically active compound of Formula (I'')



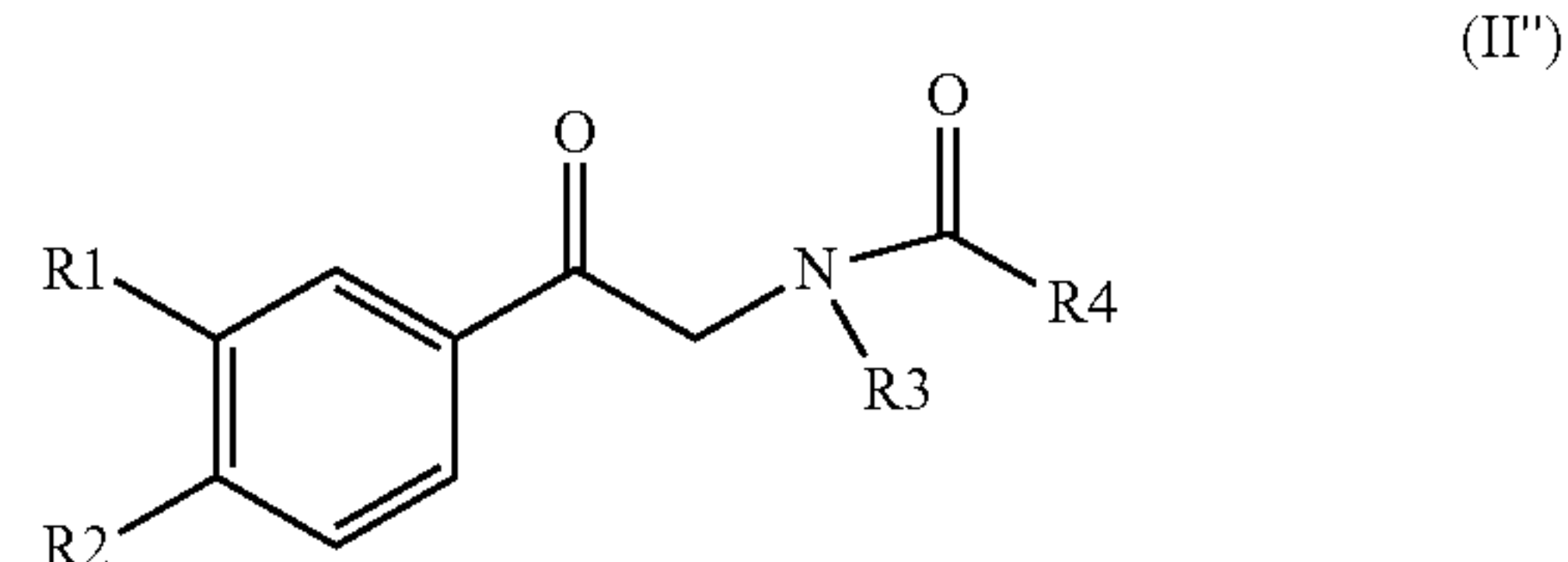
wherein

the asterisk, R₁ and R₂ are as defined above;

R₃ and R₄, together with the carboxamide group (HN—C=O) bonding them, constitute an imide,

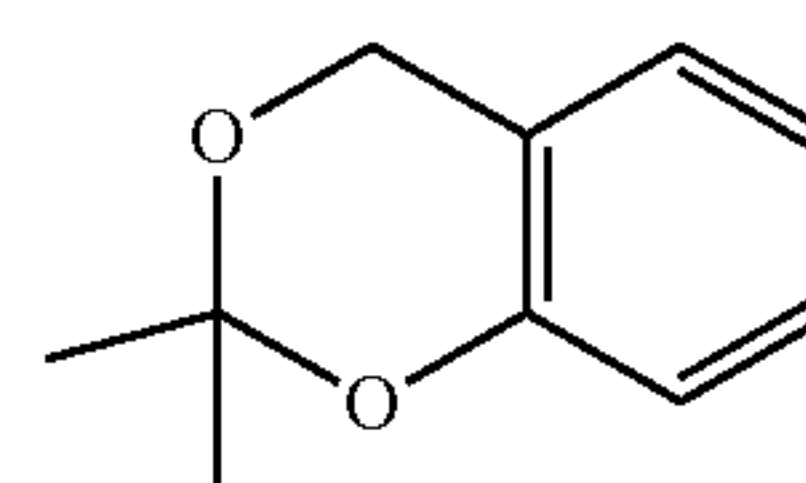
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said process comprising the enzymatic reduction of the compound of Formula (II'')



wherein R₁, R₂, R₃ and R₄ are as defined above.

According to the last embodiment, preferably the R groups form a cycle fused with the benzene, advantageously having the following formula:



Still according to the last embodiment, preferably said imide is succinimide.

The compounds of Formula (II) and (II'') can be easily prepared according to the methods known in the art.

The process of the invention preferably uses at least one oxidoreductase enzyme, which is preferably a polypeptide originating from yeasts or bacteria, advantageously from bacteria.

The enantioselective enzymatic reduction can be carried out by using the oxidoreductase enzyme in suspension in a reaction mixture, or immobilized according to known techniques. The enzyme can be purified, only partially purified, or can be contained in cells. The cells themselves can be in a native state, in a permeabilized state or in a lysed state. Preferably, the enzyme is expressed in *E. coli* and used as a suspension of native cells.

According to a preferred embodiment, said enzymatic reduction is carried out with at least one oxidoreductase enzyme, in the presence of at least one cofactor and at least one co-substrate that regenerates said cofactor.

The process of enzymatic reduction of the compounds of Formula (II) or (II'') can be carried out for example in a reaction mixture comprising said compound of Formula (II), an oxidoreductase enzyme, NADH or NADPH as cofactor, a co-substrate.

According to a preferred embodiment, said at least one oxidoreductase enzyme that is used in the process of the invention, has a sequence selected from the sequences SEQ ID No: 1, herein below also indicated as "OX 56", described in EP1963516 (IEP) and SEQ ID No.: 2 herein below also indicated as "OX 62", described in EP1929001 (BASF).

According to a more preferred embodiment, said at least one oxidoreductase enzyme that is used in the process of the invention has the sequence SEQ ID No: 1.

The sequences are described in the sequence listing herein attached and depicted below:

SEQ ID NO. 1

Met Arg Leu Lys Gly Lys Ala Ala Ile Val Thr Gly

Gly Ala Ser Gly Ile Gly Arg Ala Thr Ala Ile Arg

-continued

Phe Ala Glu Glu Gly Ala Lys Val Ala Val Ser Asp
 Ile Asn Glu Glu Gly Gly Glu Glu Thr Val Arg Leu
 Ile Arg Glu Lys Gly Gly Glu Ala Ile Phe Val Gln
 Thr Asp Val Ala Asp Ser Lys Gln Val Ser Arg Leu
 Val Gln Thr Ala Val Asp Ala Phe Gly Gly Leu His
 Ile Leu Phe Asn Asn Ala Gly Ile Gly His Ser Glu
 Val Arg Ser Thr Asp Leu Ser Glu Glu Glu Trp Asp
 Arg Val Ile Asn Val Asn Leu Lys Gly Val Phe Leu
 Gly Ile Lys Tyr Ala Val Pro Val Met Lys Gln Cys
 Gly Gly Gly Ala Ile Val Asn Thr Ser Ser Leu Leu
 Gly Ile Lys Gly Lys Lys Tyr Glu Ser Ala Tyr Asn
 Ala Ser Lys Ala Gly Val Ile Leu Leu Thr Lys Asn
 Ala Ala Leu Glu Tyr Gly Lys Phe Asn Ile Arg Val
 Asn Ala Ile Ala Pro Gly Val Ile Asp Thr Asn Ile
 Ile Thr Pro Trp Lys Gln Asp Glu Arg Lys Trp Pro
 Ile Ile Ser Lys Ala Asn Ala Leu Gly Arg Ile Gly
 Thr Pro Glu Glu Val Ala Asn Ala Val Leu Phe Leu
 Ala Ser Asp Glu Ala Ser Phe Ile Thr Gly Ala Thr
 Leu Ser Val Asp Gly Gly Gly Leu Thr Phe

SEQ ID No. 2

Met Thr Thr Thr Ser Asn Ala Leu Val Thr Gly Gly
 Ser Arg Gly Ile Gly Ala Ala Ser Ala Ile Lys Leu
 Ala Gln Glu Gly Tyr Asn Val Thr Leu Ala Ser Arg
 Ser Val Asp Lys Leu Asn Glu Val Lys Ala Lys Leu
 Pro Ile Val Gln Asp Gly Gln Lys His Tyr Ile Trp
 Glu Leu Asp Leu Ala Asp Val Glu Ala Ala Ser Ser
 Phe Lys Gly Ala Pro Leu Pro Ala Arg Ser Tyr Asp
 Val Phe Val Ser Asn Ala Gly Val Ala Ala Phe Ser
 Pro Thr Ala Asp His Asp Asp Lys Glu Trp Gln Asn
 Leu Leu Ala Val Asn Leu Ser Ser Pro Ile Ala Leu
 Thr Lys Ala Leu Leu Lys Asp Val Ser Glu Arg Pro
 Val Asp Lys Pro Leu Gln Ile Ile Tyr Ile Ser Ser
 Val Ala Gly Leu His Gly Ala Ala Gln Val Ala Val
 Tyr Ser Ala Ser Lys Ala Gly Leu Asp Gly Phe Met
 Arg Ser Val Ala Arg Glu Val Gly Pro Lys Gly Ile
 His Val Asn Ser Ile Asn Pro Gly Tyr Thr Lys Thr
 Glu Met Thr Ala Gly Ile Glu Ala Leu Pro Asp Leu
 Pro Ile Lys Gly Trp Ile Glu Pro Glu Ala Ile Ala
 Asp Ala Val Leu Phe Leu Ala Lys Ser Lys Asn Ile
 Thr Gly Thr Asn Ile Val Val Asp Asn Gly Leu Ile
 Ala

It has been found that polypeptides that have an amino acid sequence having at least 60%, preferably at least 80%, advantageously at least 90% of the amino acids identical to the amino acid sequence of SEQ ID No: 1 and SEQ ID No: 2 lead to the reduction of the compound of Formula (II) or (II") in the (R) configuration with high yields and high enantiomeric selectivity. In fact, the obtained enantiomeric excess of the compound of Formula (I), (I') and (I'') in the (R) configuration is at least about 90%, preferably at least about 95% and more preferably at least about 99%. According to an advantageous embodiment of the invention, the cofactor is selected from nicotinamide adenine dinucleotide phosphate (NADP) and nicotinamide adenine dinucleotide (NAD).

The cofactor is preferably in the reaction mixture at a concentration from about 0.01 mM to about 5 mM, advantageously from about 0.05 mM to about 0.5 mM. According to an advantageous embodiment of the invention the co-substrate is a secondary alcohol, preferably a secondary alcohol up to 10 carbon atoms, such as 2-propanol, 2-butanol, 2-pentanol, 4-methyl-2-pentanol, 2-heptanol and 2-octanol, preferably 2-propanol or 4-methyl-2-pentanol, more preferably 2-propanol. According to a preferred embodiment, the co-substrate is present in the reaction mixture from about 10% to about 80% (v/v), more preferably from about 10% to about 50%, advantageously at the rate of 15-25%.

Preferably the process of enzymatic reduction of the compounds of Formula (II) or (II") is carried out in a solvent, advantageously in the presence of a suitable buffer. The reaction of enzymatic reduction of the invention can be carried out in a mono-phase system or a bi-phase system of the kind of water/organic solvent, according to known techniques. In the latter case, an organic solvent that is not involved in the regeneration of the cofactor can be added to the reaction mixture. Examples of such solvents include diethyl ether, tert-butyl methyl ether, diisopropyl ether, dibutyl ether, ethyl acetate, butyl acetate, heptane, hexane or cyclohexane. Such a solvent can be present at a rate of about 1% to 50% in volume based on the volume of the reaction mixture.

When the enzyme is used as a suspension of native cells, the reaction mixture preferably contains from about 100 to 2000 g cells per kg of raw product that has been produced by the reduction.

The buffer used can, for example, be selected from potassium phosphate buffer or triethanolamine buffer, and can further comprise ions for stabilizing the enzyme, for example a source of magnesium ions. Other additives that can be present in the buffer for stabilizing the enzymes can include a polyol, such as glycerol, sorbitol and the like, sulfur compounds such as 1,4-DL-dithiothreitol, glutathione, cysteine or the like, amino acids and peptides or detergents, such as DMSO.

A preferred stabilizer for the enzyme is a polyol, in particular glycerol, that can be present at a rate of about 10-80%, preferably about 50% by weight based on the weight of the cell suspension.

The oxidized cofactor that is formed during the reduction of the compound of Formula (II) or (II") is regenerated by oxidation of the co-substrate and the oxidation can also be catalyzed by the oxidoreductase itself. Therefore, a particular practical and economical advantage of the present process is that the oxidoreductase affects both the reduction of the compound of Formula (I), (I') and (I'') and the oxidation of the co-substrate, and thus it is not necessary to use other enzymes for regenerating the cofactor.

The pH of the reaction mixture, after the addition of all the components, will be between 5 and 10, preferably from 7 to 9 and optimally about 7.5. The enzymatic reduction according to the present invention is carried out at a temperature of about 10-45° C., preferably about 20-40° C., preferably about 35-40° C. The process of enantioselective reduction is convenient and eco-friendly, in addition to providing the alcohols of Formula (I), (I') and (I'') with high yields and high enantiomeric selectivity.

The compound of Formula (I), (I') and (I'') in the (R) configuration with high optical purity can be obtained in the presence of the oxidoreductase enzyme in the reaction conditions mentioned above in about 2 to 96 hours, preferably from about 4 to 24 hours. During the incubation, the pH of the mixture is preferably maintained within the ranges indicated above. The efficiency of the enantioselective enzymatic reduction can be expressed by the total turnover number (TTN) that is the moles of the chiral alcohol of Formula (I), (I') and (I'') produced per mole of cofactor used. The TTN of the enantioselective enzymatic reduction is from about 102 to 105, preferably >103.

According to an embodiment, the at least one oxidoreductase enzyme that is used in the process of the invention, as described above, has a sequence selected from the sequences SEQ. ID No: 1 and SEQ. ID No: 2.

In addition to the sequences SEQ ID No: 1 and SEQ ID No: 2, it has been found that polypeptides that have an amino acid sequence having at least 60%, preferably at least 80%, advantageously at least 90%, identical to the amino acid sequence of SEQ ID No: 1 and SEQ ID No: 2 lead to the reduction of the compound of Formula (II) or (II'') in the (R) configuration with high yields and high enantiomeric selectivity. In fact, the enantiomeric excess of the compound of Formula (I) or (I') is at least about 90%, preferably at least about 95% and more preferably at least about 99%. According to a preferred embodiment, subject-matter of the invention is a process of enantioselective reduction of the preferred compounds of Formula (II) and (II''), as reported above, comprising the use of at least one enzyme selected from the sequences SEQ. ID No: 1 and SEQ. ID No: 2 or of polypeptides that have an amino acid sequence having at least 60%, preferably at least 80%, advantageously at least 90%, of the amino acids identical to the amino acid sequence of SEQ. ID No: 1 and SEQ. ID No: 2.

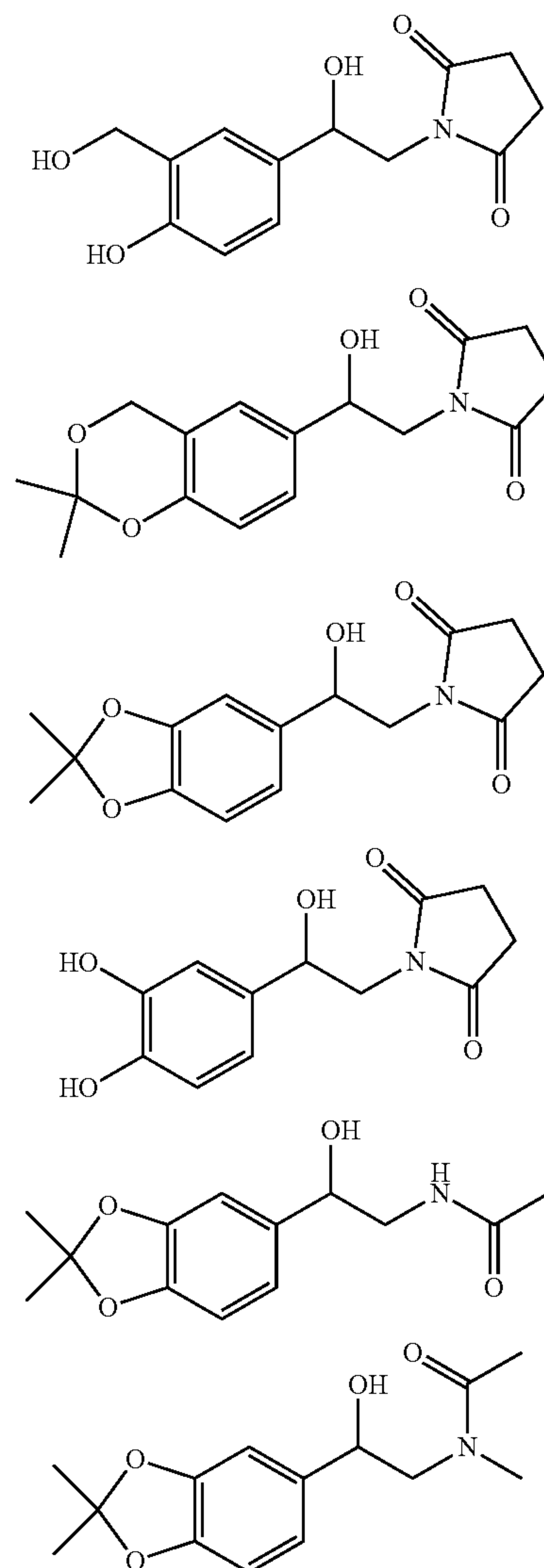
The compound of Formula (I), (I') and (I'') can be easily transformed in the corresponding optically active beta-amino alcohol, for example by treating with bases or acids, according to the techniques well known to the person skilled in the art.

Some detailed examples of the reactions described above are reported in the following Experimental section.

Therefore it is understood that, in a completely unexpected way, it has been found that the enzymatic reduction of beta-amino ketones to give optically active beta-amino alcohols, can be carried out provided that the amine group is deprived of its basicity. This fact was not at all conceivable a priori and is an important technical progress in the field of stereoselective reductions.

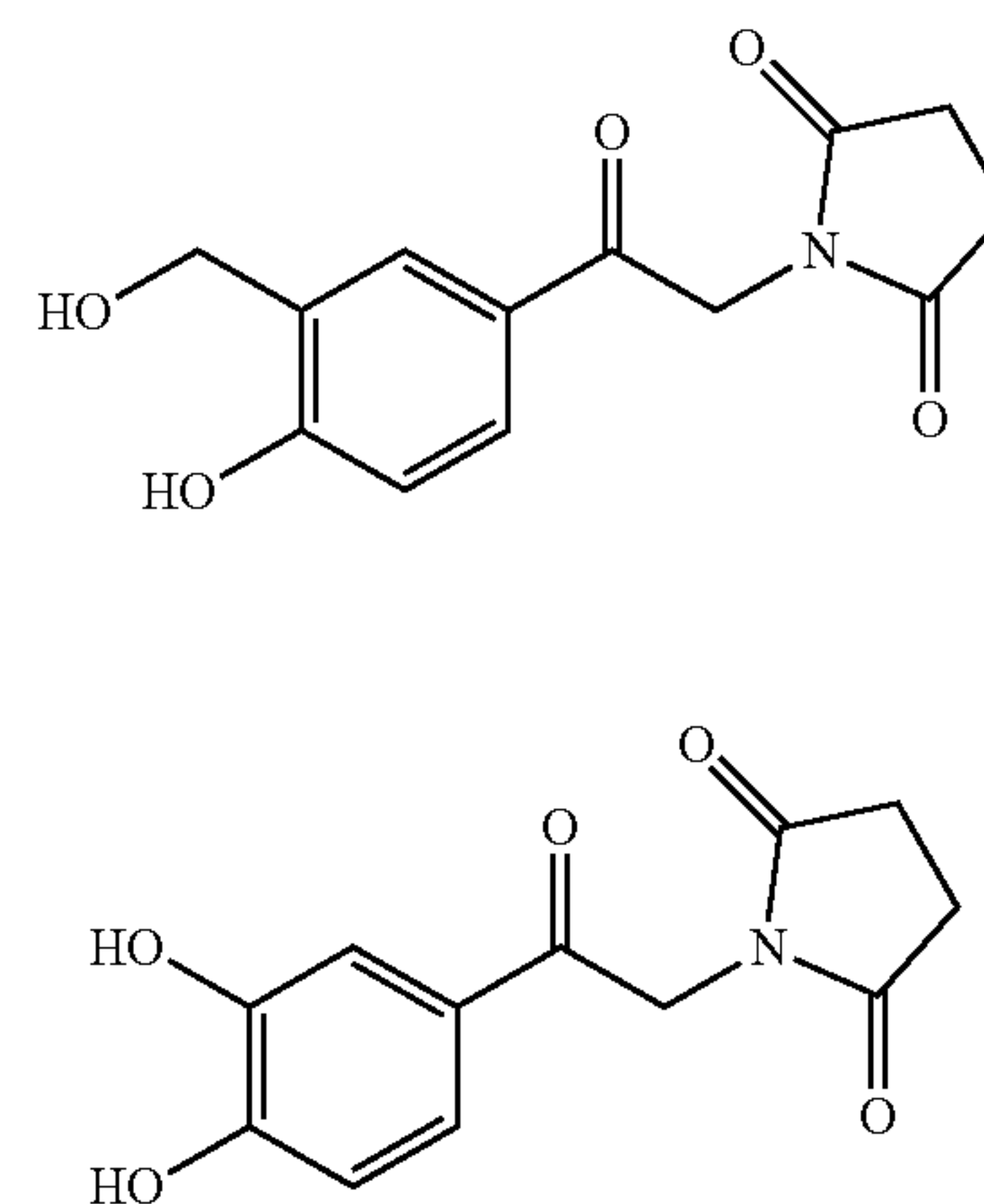
The compounds of Formula (I), (I') and (I'') are versatile synthesis intermediates and can be easily converted to active pharmaceutical ingredients, such as for example adrenalin, noradrenalin, vilanterol, R-salbutamol, R-colterol, R-isoproterenole, (-)-arbutamine and the like.

According to another of its aspects, subject-matter of the invention is a compound of Formula (I) selected from the following compounds:



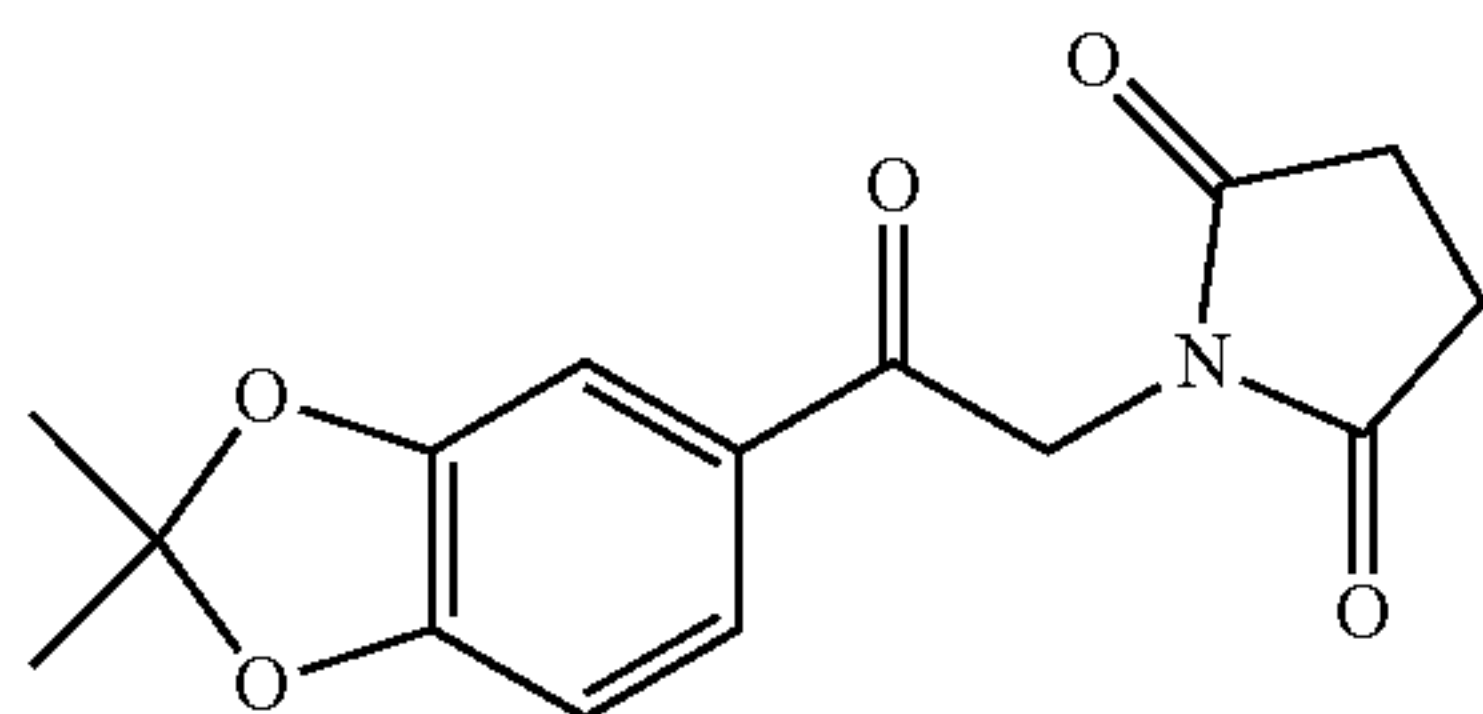
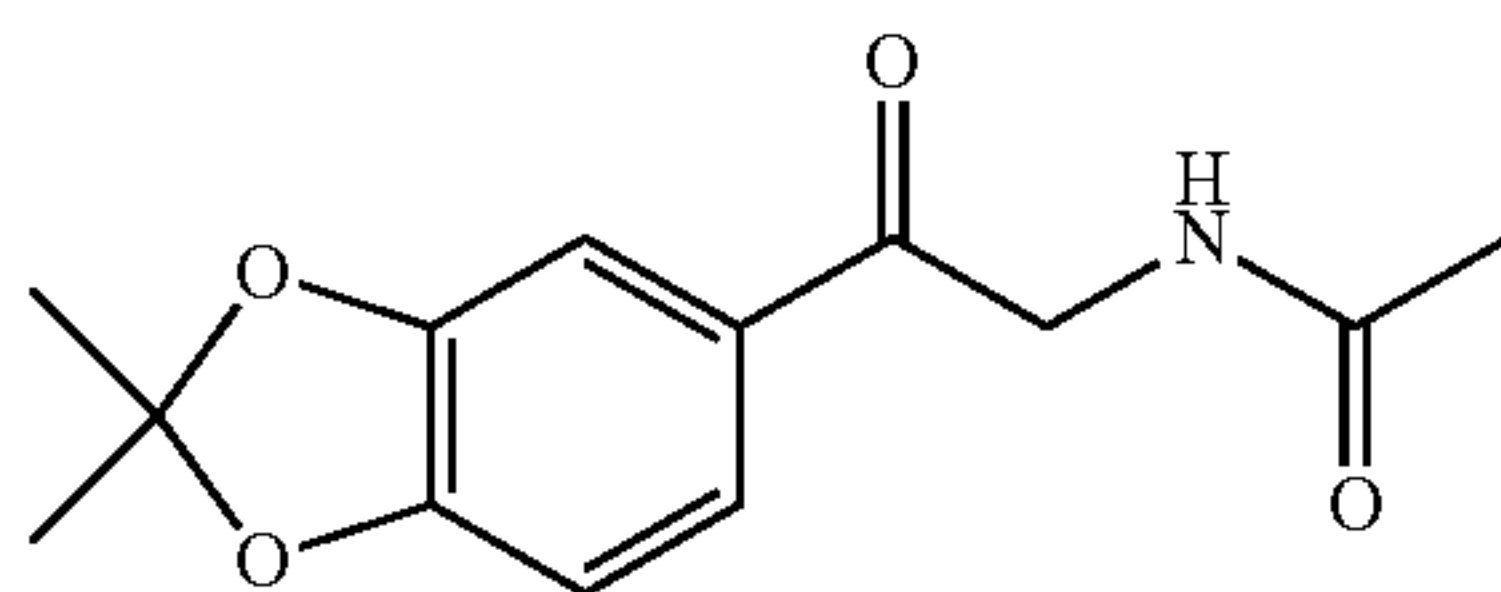
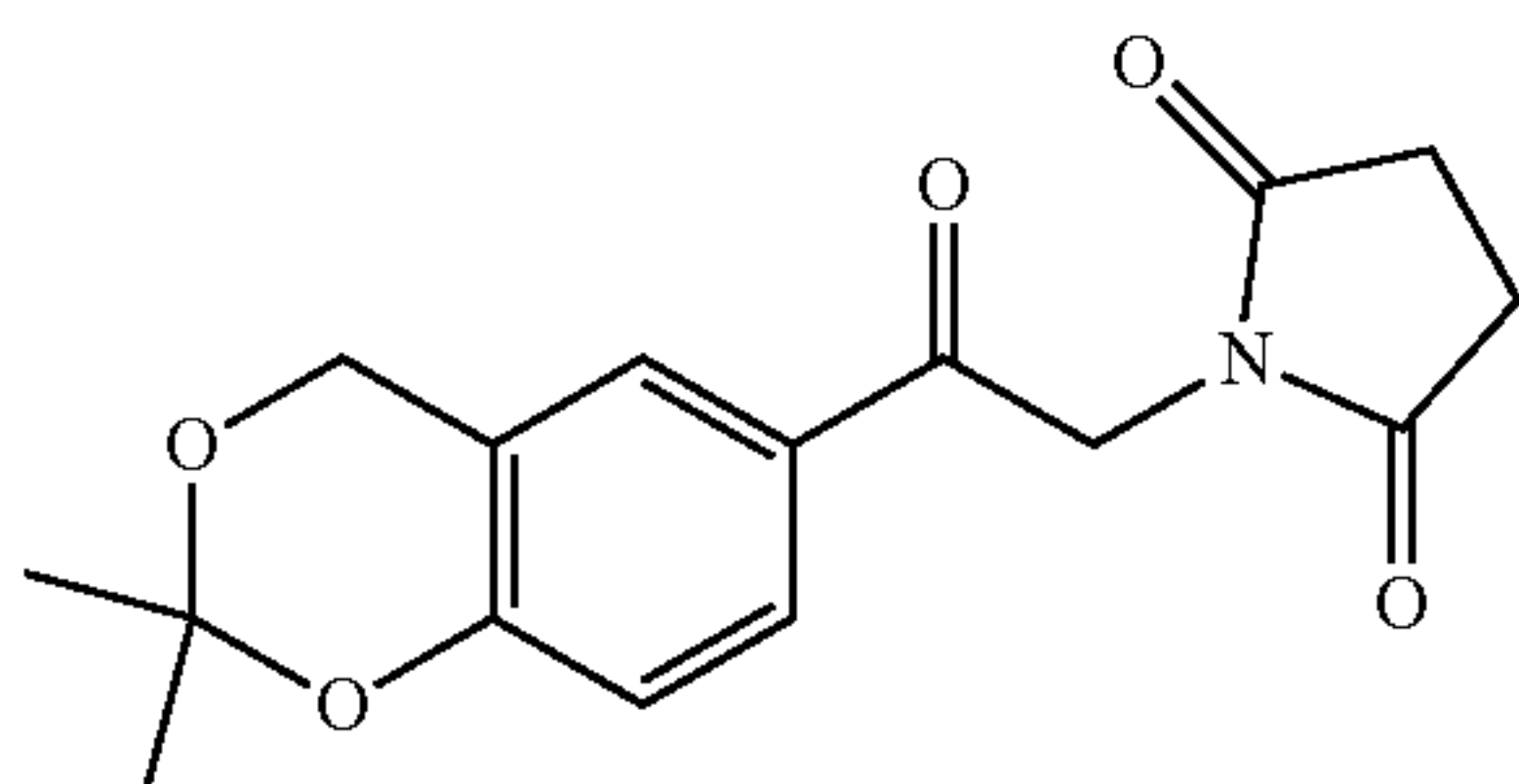
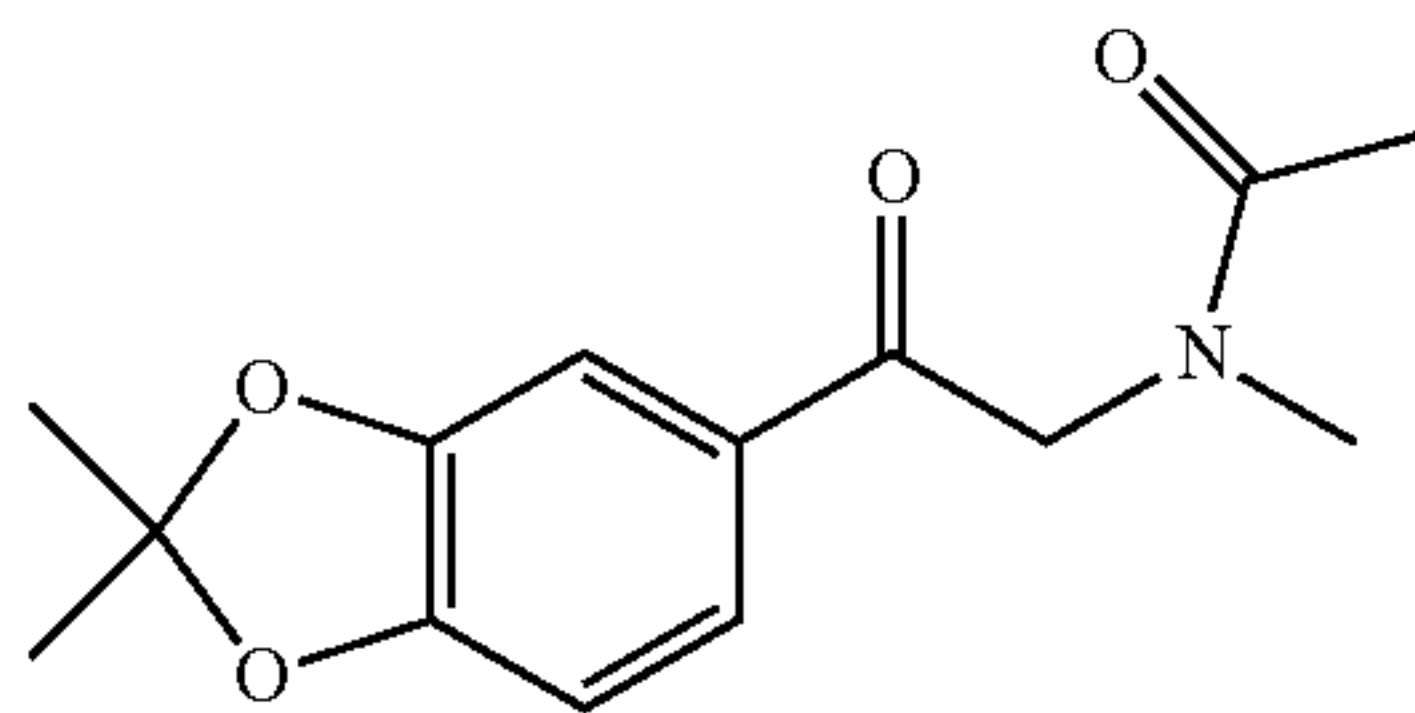
and the salts and/or hydrates and/or solvates thereof.

According to another of its aspects, subject-matter of the invention is a compound of Formula (II) selected from the following compounds:



11

-continued



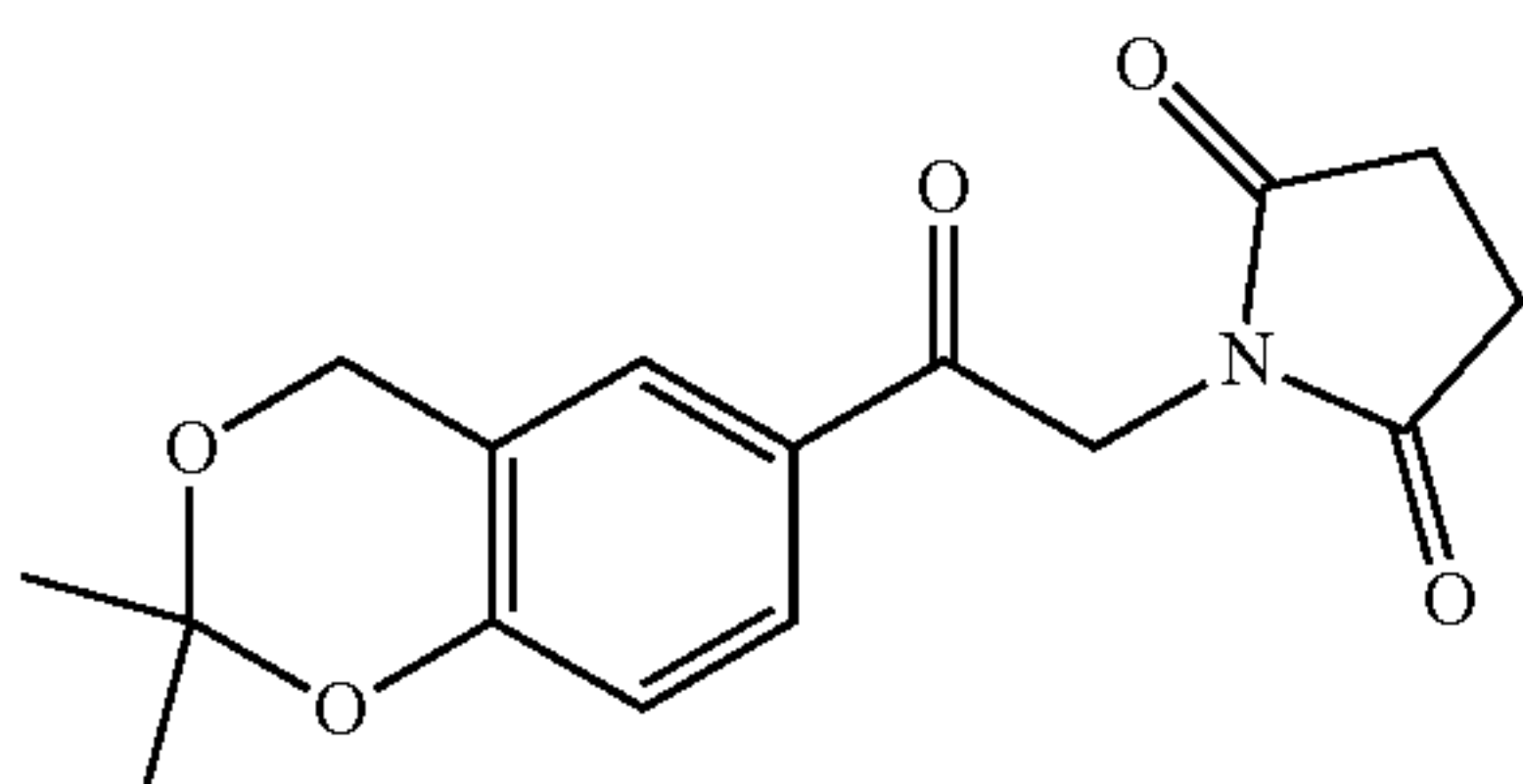
and the salts and/or hydrates and/or solvates thereof.

According to another of its aspects, subject-matter of the invention is the use of at least one compound selected from the compounds of Formula (I), (I'), (I''), (II) or (II'') for preparing a compound selected from adrenalin and nora-

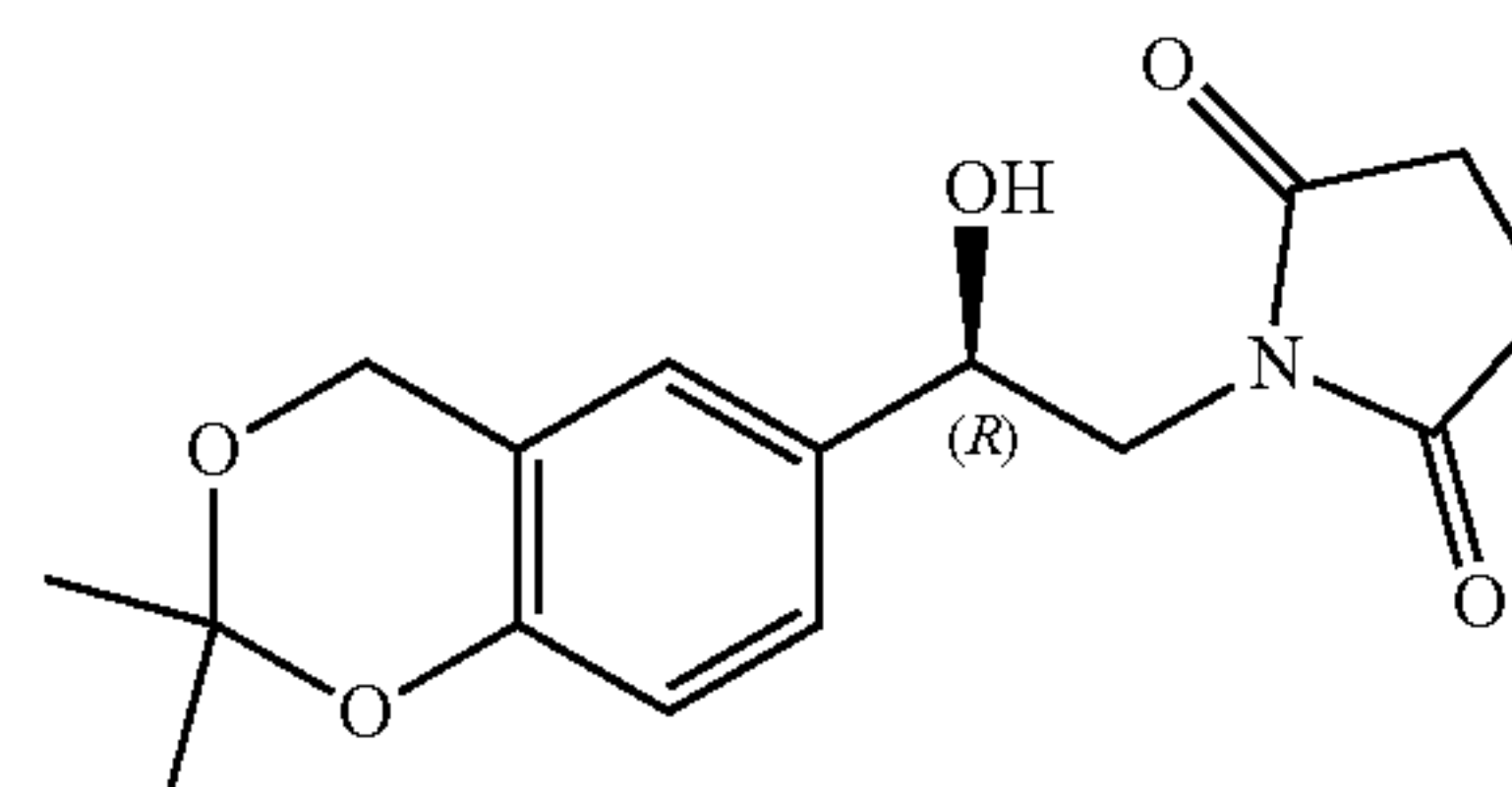
drenalin or one of the salts thereof.

According to another of its aspects, subject-matter of the invention is the use of at least one compound selected from the compounds of Formula (I), (I'), (I''), (II) or (II'') for preparing vilanterol or one of the salts thereof.

According to another of its aspects, subject-matter of the invention is the use of the compound having formula B, C

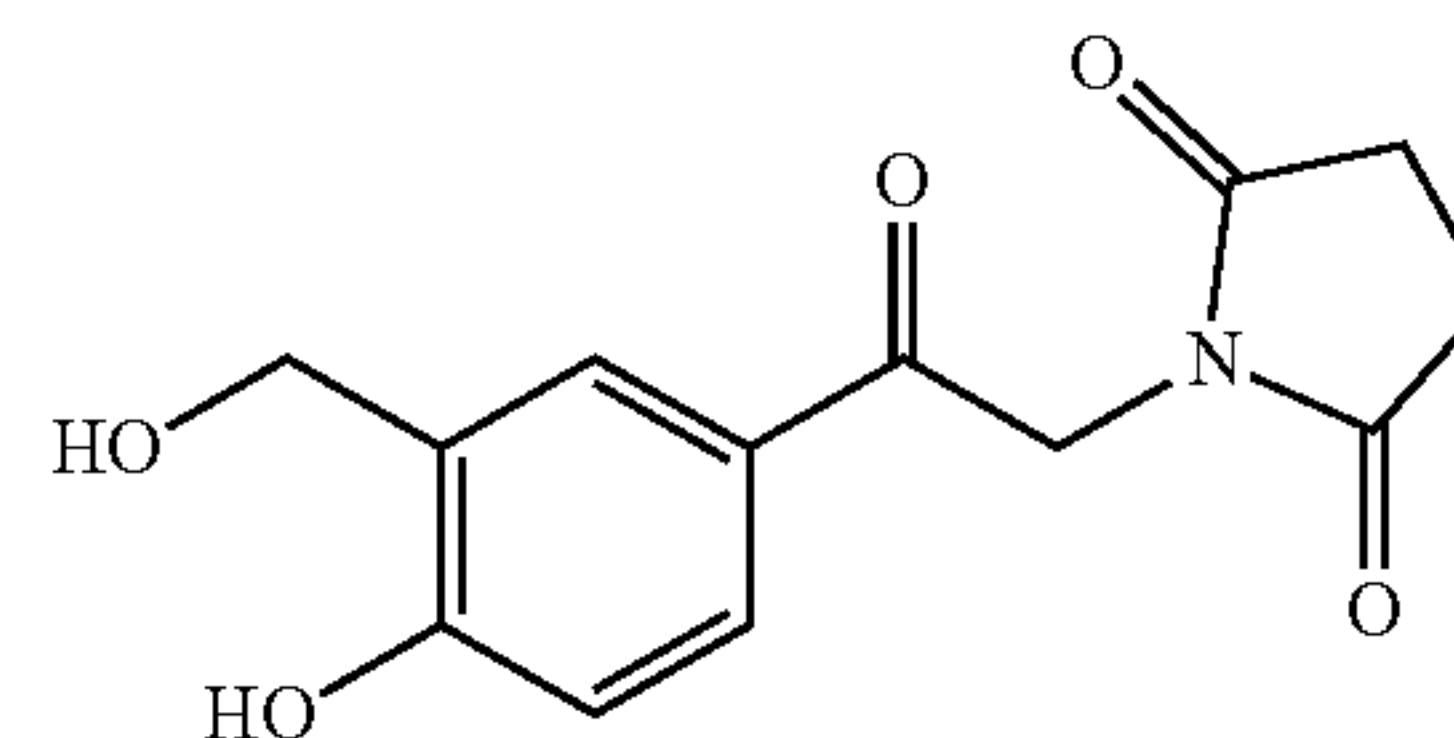
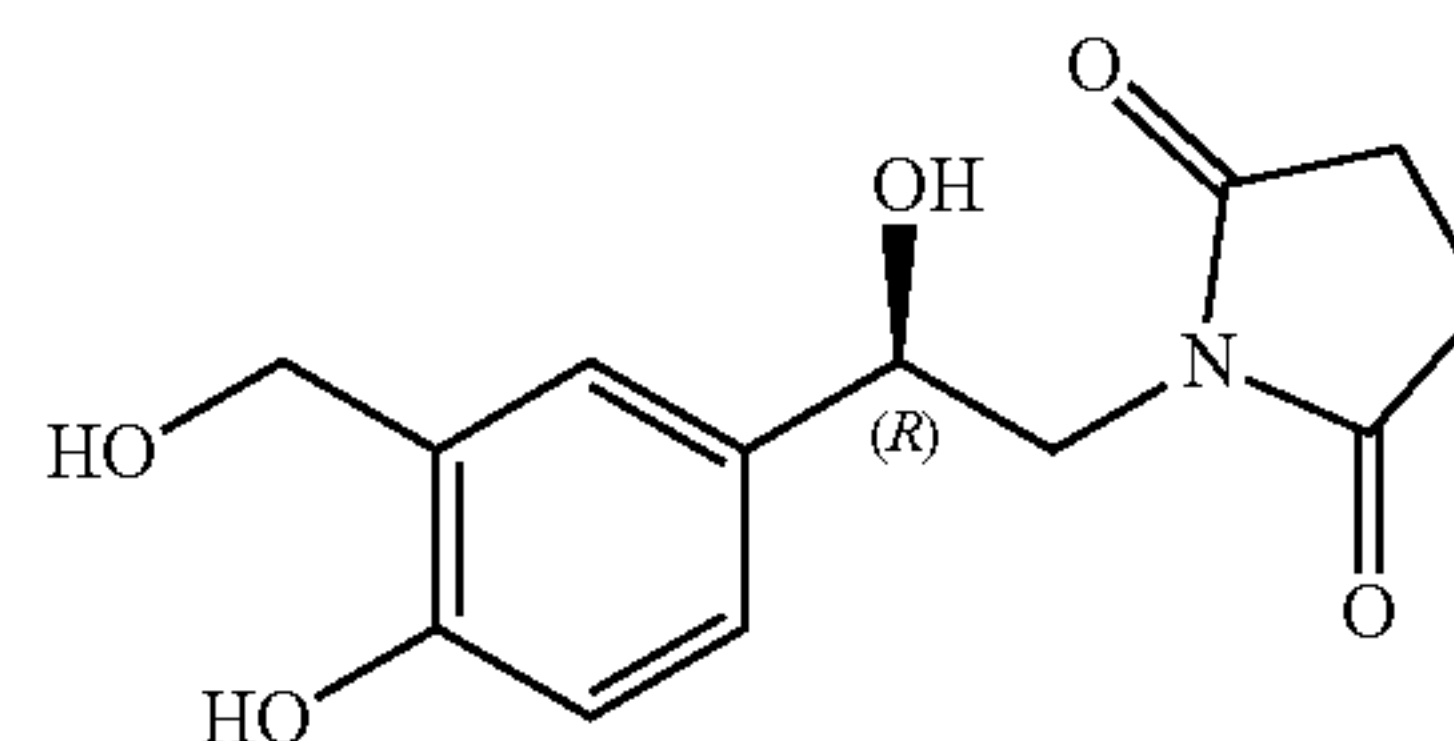
**12**

-continued



and the salts and/or hydrates and/or solvates thereof, for preparing vilanterol.

According to another of its aspects, subject-matter of the invention is the use of the compound having formula Y, Z,



and the salts and/or hydrates and/or solvates thereof, for preparing vilanterol. The examples described in the following Experimental section are provided purely by way of illustration and in no way as limiting.

EXPERIMENTAL SECTION

Abbreviations

DMF=dimethylformamide

DCM=dichloromethane

MTBE=methyl tert-butyl ether

IPA=2-propanol

MP=4-methyl-2-pentanol

THF=tetrahydrofuran

DMS=dimethyl sulfide

Analytic Methods

UPLC purity method (Waters Acquity UPLC™):

Stationary phase: BEH SHIELD RP18 1.7 μm 2.1×50 mm Column, Mobile phase: H₂O+0.05% TFA, CH₃CN+0.05% TFA, Gradient: 5% to 100% CH₃CN.

Chiral HPLC method (HPLC Agilent 1200):

Stationary phase: CHIRALPAK AD-H 250×4.6 mm-5*μm, Mobile phase: heptane/MeOH/iPrOH 90:5:5+0.1% ethanolamine.

13

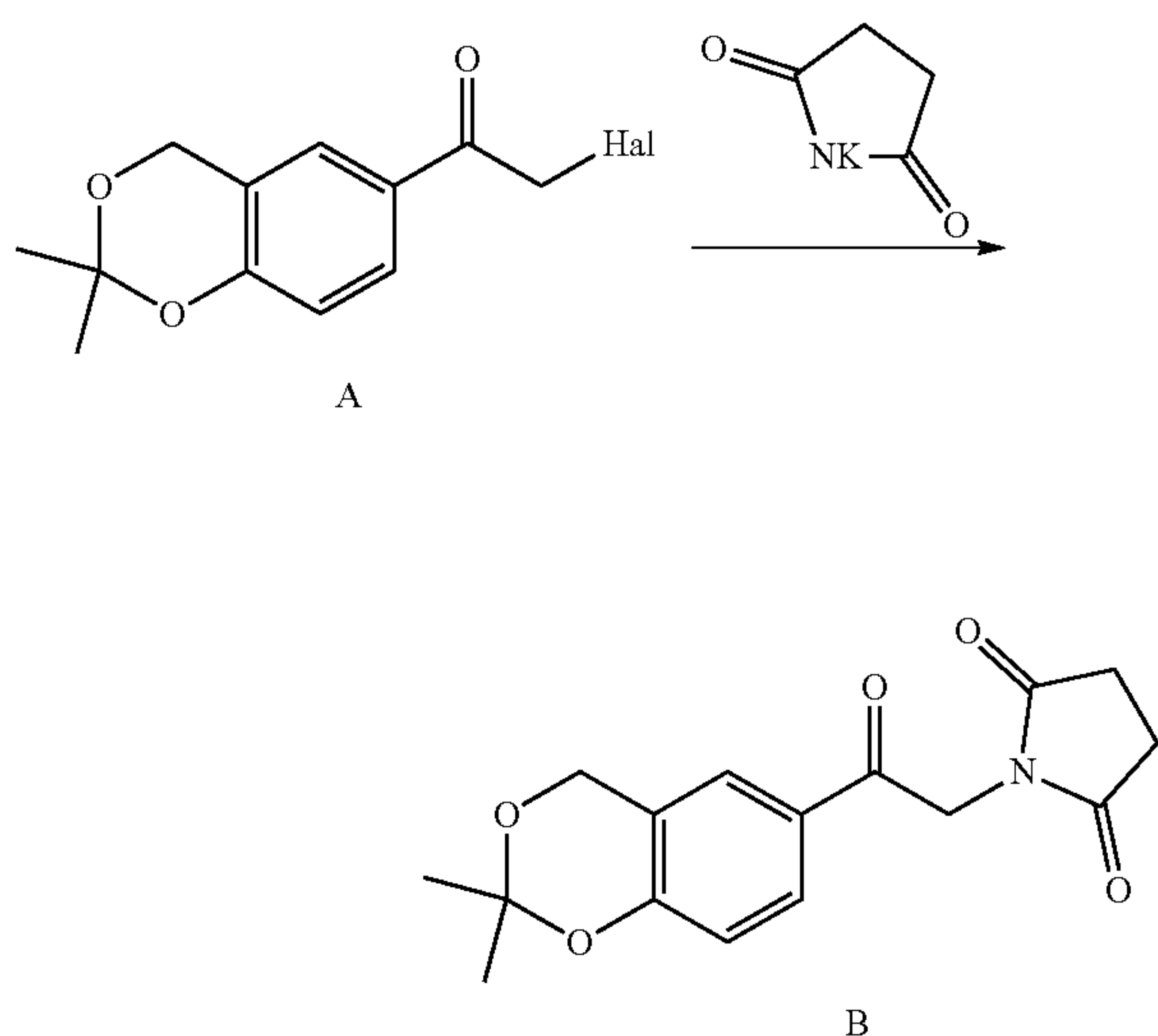
Preparation 1

Preparation of the Enzymatic Solution

Competent *Escherichia coli* cells Starb121 (DE3) (Invitrogen) have been transformed with the construct pET21a-MIX encoding for the oxidoreductases. The colonies of cells transformed with the constructs have been cultured in 200 ml LB medium (1% tryptone, 0.5% yeast extract and 1% NaCl) with 50 µg/ml ampicillin or 40 µg/ml kanamycin, respectively, until an optical density of 0.5 measured at 550 nm is achieved. The expression of the recombinant protein has been induced by adding isopropyl-thiogalactoside (IPTG) with a concentration of 0.1 mM. After 16 hours of induction at 25° C. and 220 revolutions/minute, the cells have been collected and frozen at -20° C. For the preparation of the enzymatic solutions, 30 g cells have been re-suspended in 150 ml triethanolamine buffer (100 mM, pH 7, 2 mM MgCl₂, 10% glycerol) and homogenized by using a high pressure homogenizer. Subsequently, the enzymatic solution has been mixed with 150 ml glycerol and stored at -20° C.

Example 1

Preparation of a Compound of Formula (II) and (II')

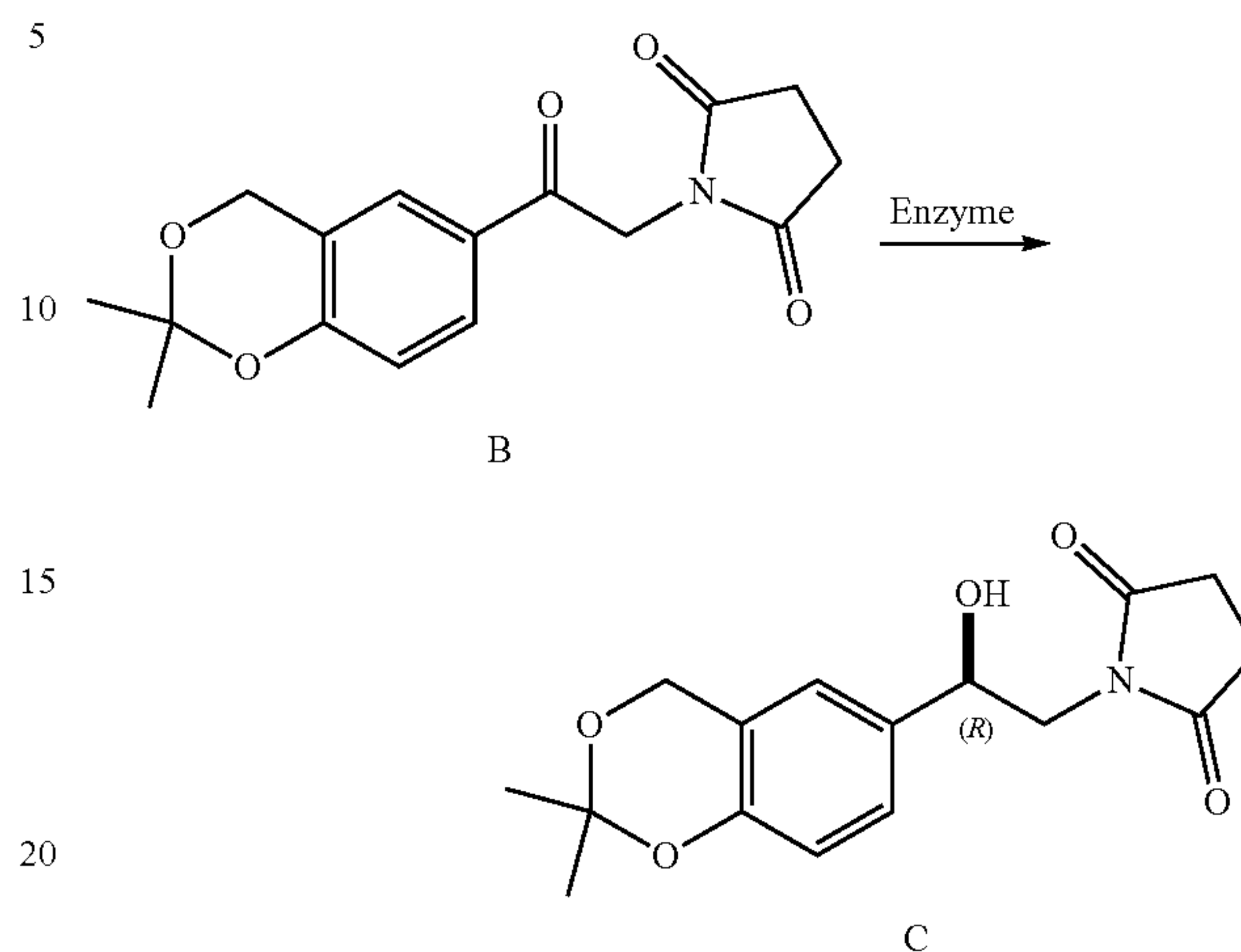


Hal=halogen, preferably bromine or chlorine

5 g compound A (wherein Hal=Cl) is dissolved in 50 ml DMF. 3.5 g succinimide potassium salt has been added to the solution. Upon complete reaction, 50 ml water and 100 ml DCM have been added. The organic phase has been dried over sodium sulphate, filtered and concentrated to a residue. 60 ml heptane and 10 ml DCM have been added, then the bulk was filtered, collecting 4.7 g of compound B that, in the dried form, provided 4.5 g of compound B, with HPLC purity higher than 99%.

14

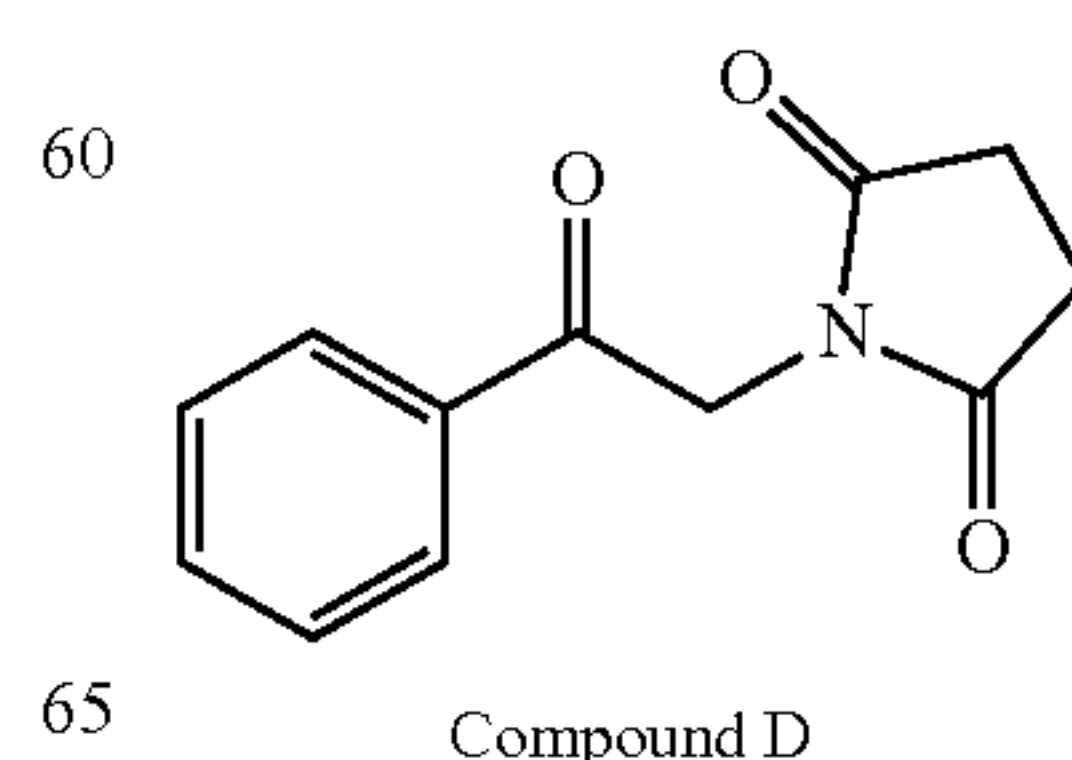
Example 2



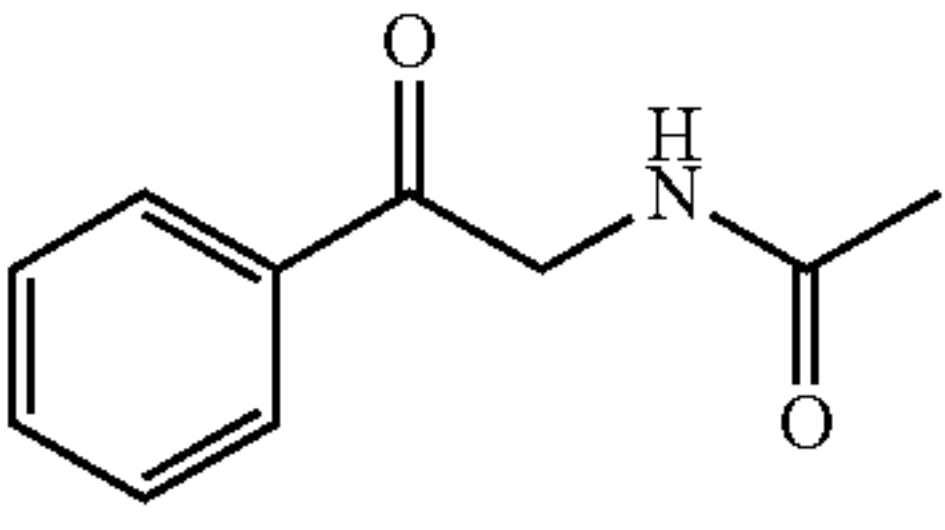
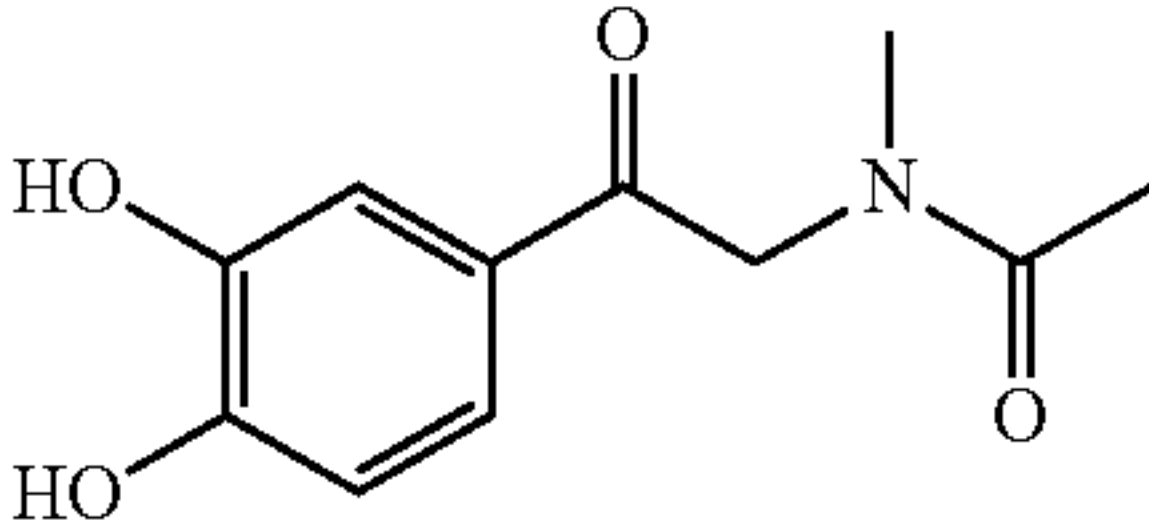
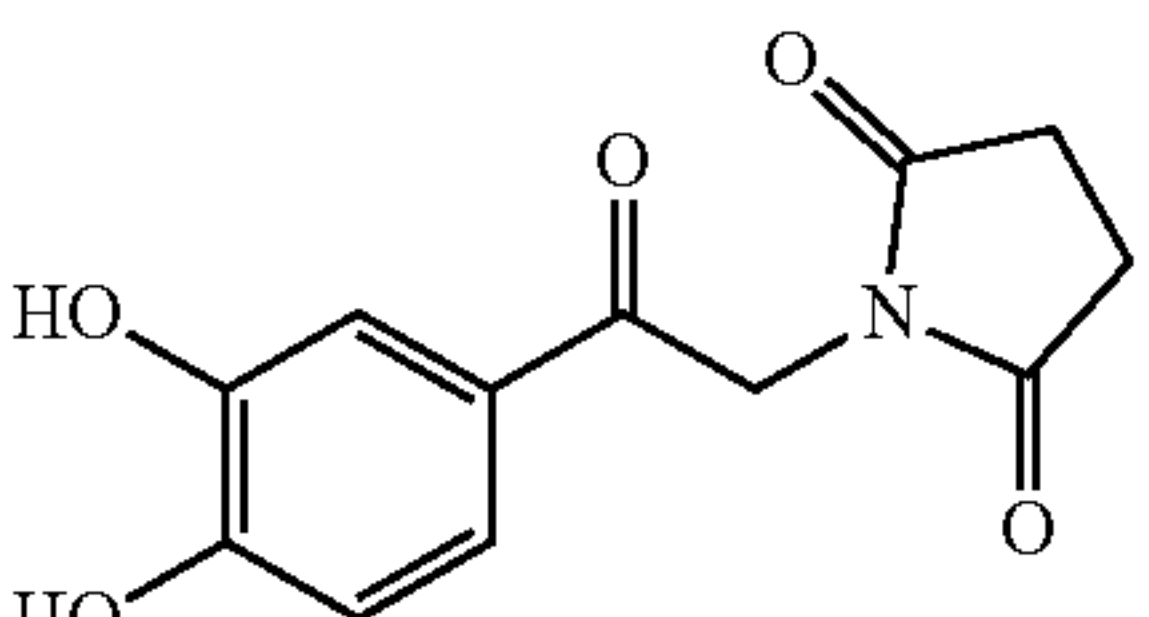
1 g of compound B has been dissolved in 3 ml 2-propanol and 4 ml MTBE. 13 ml aqueous potassium buffer at pH 8, 1 mg NAD and 5 ml enzyme SEQ ID No: 1 (OX 56) have been added. The bulk has been stirred at about 35° C. until complete conversion of ketone to alcohol. The separated aqueous phase has been extracted twice with DCM, then the organic phases have been combined and concentrated to a solid. Yield 900 mg, HPLC purity>96%, S-alcohol not detectable (chiral HPLC analysis).

By operating as described in Example 2, the reactions on compound B and the following compounds have been carried out in the conditions and with the results reported in Table (I):

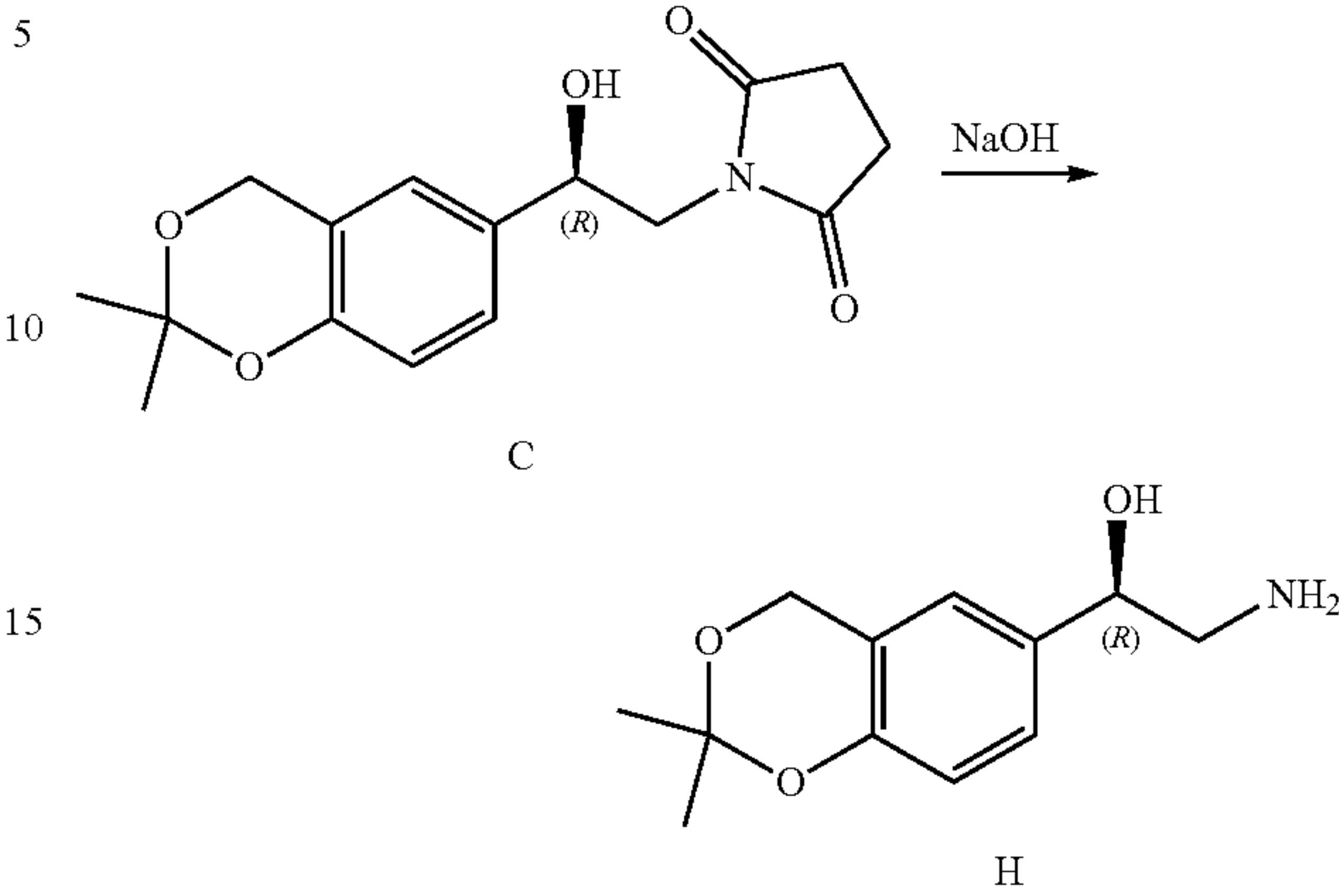
Compound	Enzyme	Cofactor	Solvent	ee (S)	ee (R)
B	SEQ ID No.: 1 (ox56)	NAD	MP	nd	>99.9%
D	SEQ ID No.: 1 (ox56)	NAD	IPA	nd	100%
E	SEQ ID No.: 1 (ox56)	NAD	IPA	0.6%	99.4%
E	SEQ ID No.: 1 (ox56)	NAD	MP	0.7%	99.3%
F	SEQ ID No.: 2 (ox62)	NAD	IPA	2.4%	97.6%
F	SEQ ID No.: 2 (ox62)	NAD	MP	2.0%	98.0%
G	SEQ ID No.: 1 (ox56)	NAD	MP	nd	100%
G	SEQ ID No.: 1 (ox56)	NAD	IPA	nd	100%



15
-continued

Compound	Enzyme	Cofactor	Solvent	ee (S)	ee (R)
					
Compound E					
					
Compound F					
					
Compound G					

16
Example 3



350 mg of compound C has been dissolved in 11 ml ethanol and about 1 g of 20% caustic soda (20 g/100 ml). The bulk has been heated to reflux for about 1 hour. The bulk has been extracted at room temperature with DCM that has been then dried over sodium sulphate, filtered and concentrated to a residue, thus obtaining 210 mg of compound H.

SEQUENCE LISTING

<160> NUMBER OF SEQ ID NOS: 2

<210> SEQ ID NO 1

<211> LENGTH: 250

<212> TYPE: PRT

<213> ORGANISM: Geobacillus thermodenitrificans

<400> SEQUENCE: 1

Met Arg Leu Lys Gly Lys Ala Ala Ile Val Thr Gly Gly Ala Ser Gly
1 5 10 15

Ile Gly Arg Ala Thr Ala Ile Arg Phe Ala Glu Glu Gly Ala Lys Val
20 25 30

Ala Val Ser Asp Ile Asn Glu Glu Gly Gly Glu Glu Thr Val Arg Leu
35 40 45

Ile Arg Glu Lys Gly Gly Glu Ala Ile Phe Val Gln Thr Asp Val Ala
50 55 60

Asp Ser Lys Gln Val Ser Arg Leu Val Gln Thr Ala Val Asp Ala Phe
65 70 75 80

Gly Gly Leu His Ile Leu Phe Asn Asn Ala Gly Ile Gly His Ser Glu
85 90 95

Val Arg Ser Thr Asp Leu Ser Glu Glu Glu Trp Asp Arg Val Ile Asn
100 105 110

Val Asn Leu Lys Gly Val Phe Leu Gly Ile Lys Tyr Ala Val Pro Val
115 120 125

Met Lys Gln Cys Gly Gly Gly Ala Ile Val Asn Thr Ser Ser Leu Leu
130 135 140

Gly Ile Lys Gly Lys Lys Tyr Glu Ser Ala Tyr Asn Ala Ser Lys Ala
145 150 155 160

Gly Val Ile Leu Leu Thr Lys Asn Ala Ala Leu Glu Tyr Gly Lys Phe
165 170 175

Asn Ile Arg Val Asn Ala Ile Ala Pro Gly Val Ile Asp Thr Asn Ile
180 185 190

-continued

Ile Thr Pro Trp Lys Gln Asp Glu Arg Lys Trp Pro Ile Ile Ser Lys
195 200 205

Ala Asn Ala Leu Gly Arg Ile Gly Thr Pro Glu Glu Val Ala Asn Ala
210 215 220

Val Leu Phe Leu Ala Ser Asp Glu Ala Ser Phe Ile Thr Gly Ala Thr
225 230 235 240

Leu Ser Val Asp Gly Gly Gly Leu Thr Phe
245 250

<210> SEQ ID NO 2
<211> LENGTH: 241
<212> TYPE: PRT
<213> ORGANISM: Candida magnoliae

<400> SEQUENCE: 2

Met Thr Thr Thr Ser Asn Ala Leu Val Thr Gly Gly Ser Arg Gly Ile
1 5 10 15

Gly Ala Ala Ser Ala Ile Lys Leu Ala Gln Glu Gly Tyr Asn Val Thr
20 25 30

Leu Ala Ser Arg Ser Val Asp Lys Leu Asn Glu Val Lys Ala Lys Leu
35 40 45

Pro Ile Val Gln Asp Gly Gln Lys His Tyr Ile Trp Glu Leu Asp Leu
50 55 60

Ala Asp Val Glu Ala Ala Ser Ser Phe Lys Gly Ala Pro Leu Pro Ala
65 70 75 80

Arg Ser Tyr Asp Val Phe Val Ser Asn Ala Gly Val Ala Ala Phe Ser
85 90 95

Pro Thr Ala Asp His Asp Asp Lys Glu Trp Gln Asn Leu Leu Ala Val
100 105 110

Asn Leu Ser Ser Pro Ile Ala Leu Thr Lys Ala Leu Leu Lys Asp Val
115 120 125

Ser Glu Arg Pro Val Asp Lys Pro Leu Gln Ile Ile Tyr Ile Ser Ser
130 135 140

Val Ala Gly Leu His Gly Ala Ala Gln Val Ala Val Tyr Ser Ala Ser
145 150 155 160

Lys Ala Gly Leu Asp Gly Phe Met Arg Ser Val Ala Arg Glu Val Gly
165 170 175

Pro Lys Gly Ile His Val Asn Ser Ile Asn Pro Gly Tyr Thr Lys Thr
180 185 190

Glu Met Thr Ala Gly Ile Glu Ala Leu Pro Asp Leu Pro Ile Lys Gly
195 200 205

Trp Ile Glu Pro Glu Ala Ile Ala Asp Ala Val Leu Phe Leu Ala Lys
210 215 220

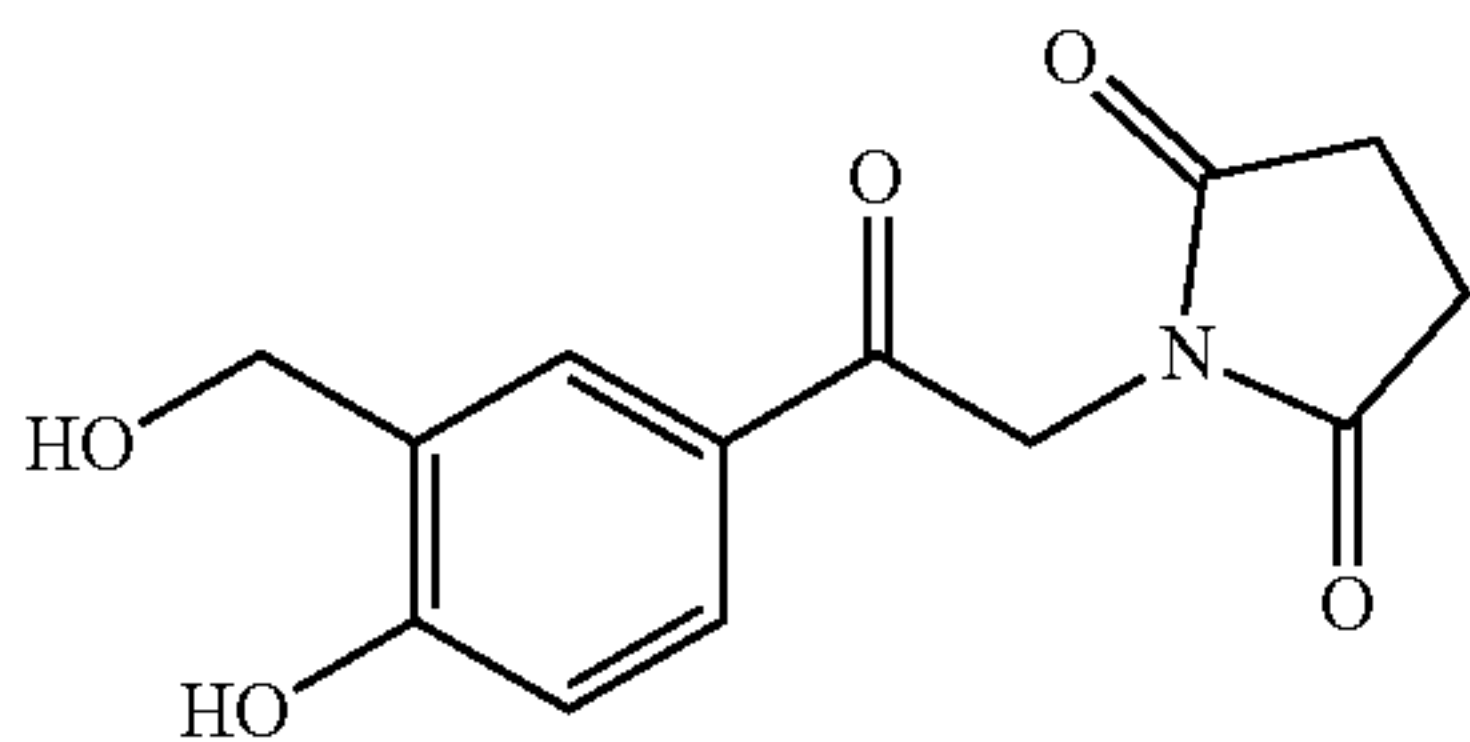
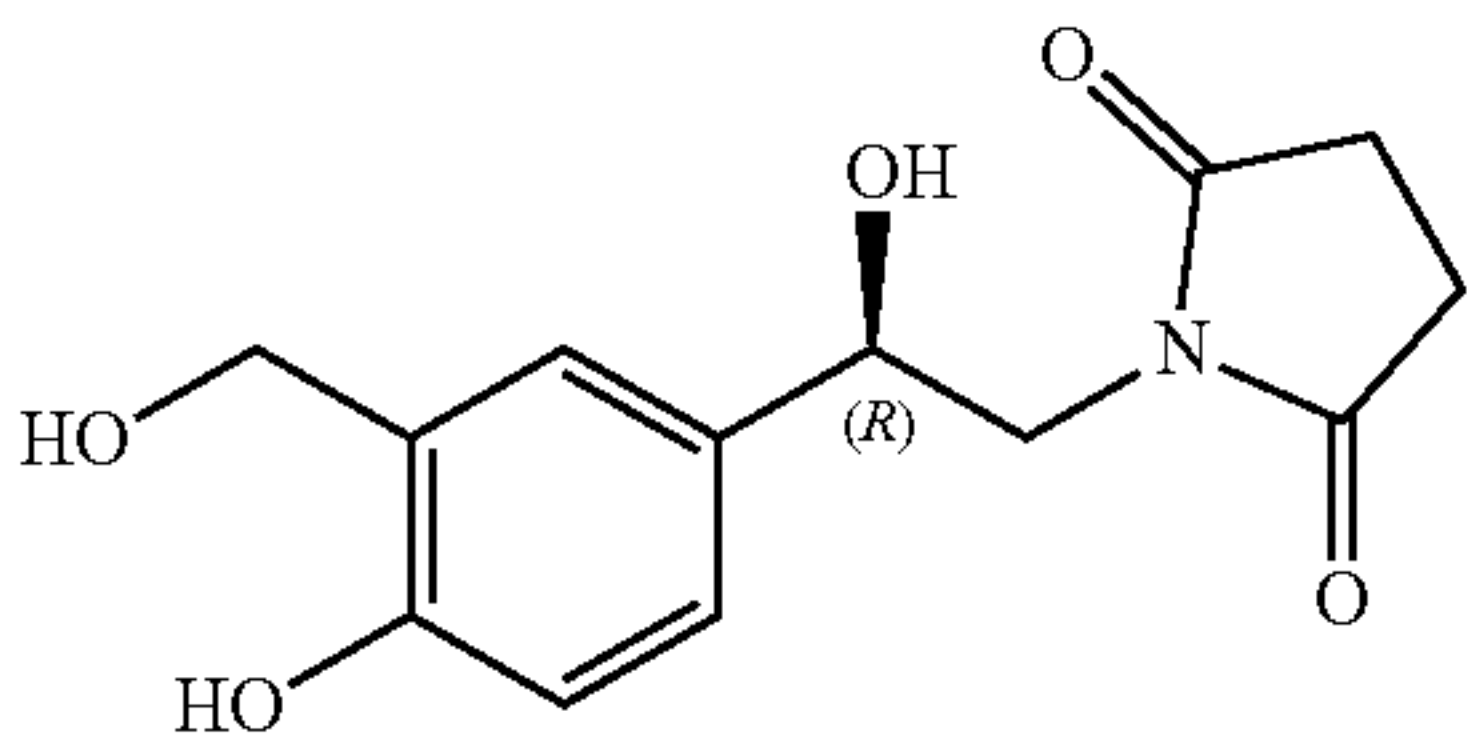
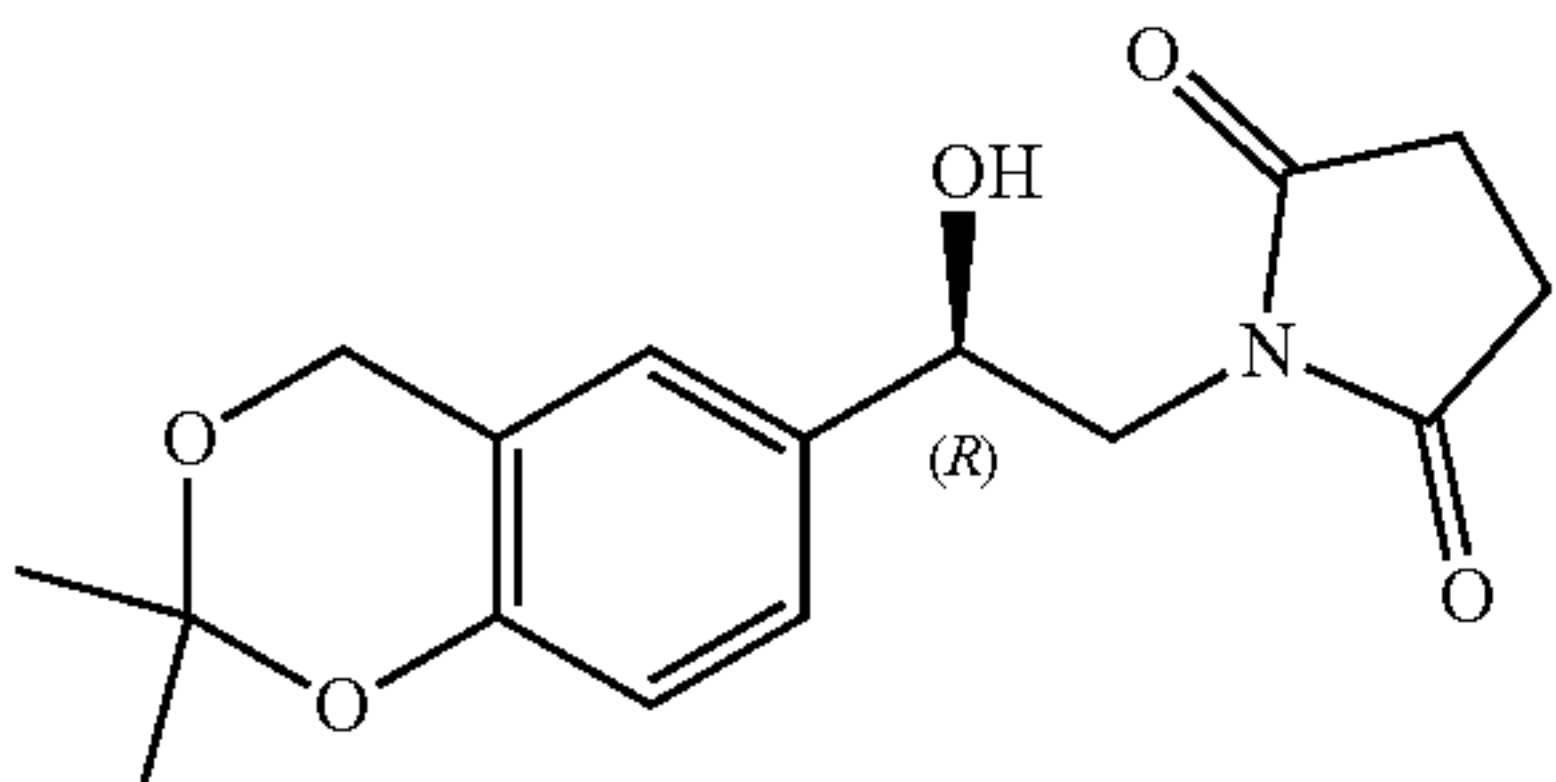
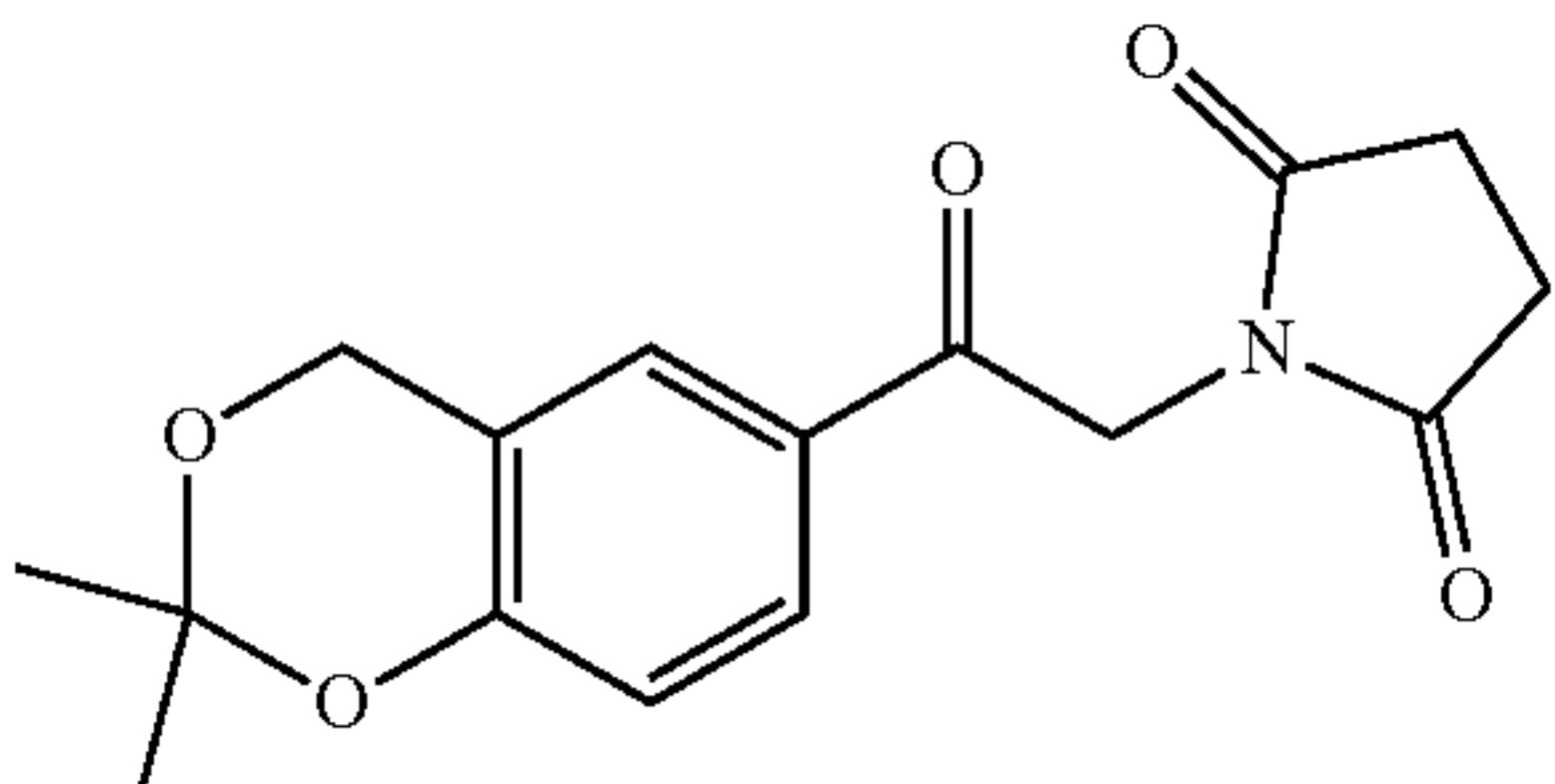
Ser Lys Asn Ile Thr Gly Thr Asn Ile Val Val Asp Asn Gly Leu Ile
225 230 235 240

Ala

19

The invention claimed is:

1. A compound selected from the compounds of formula B, C, Y, Z

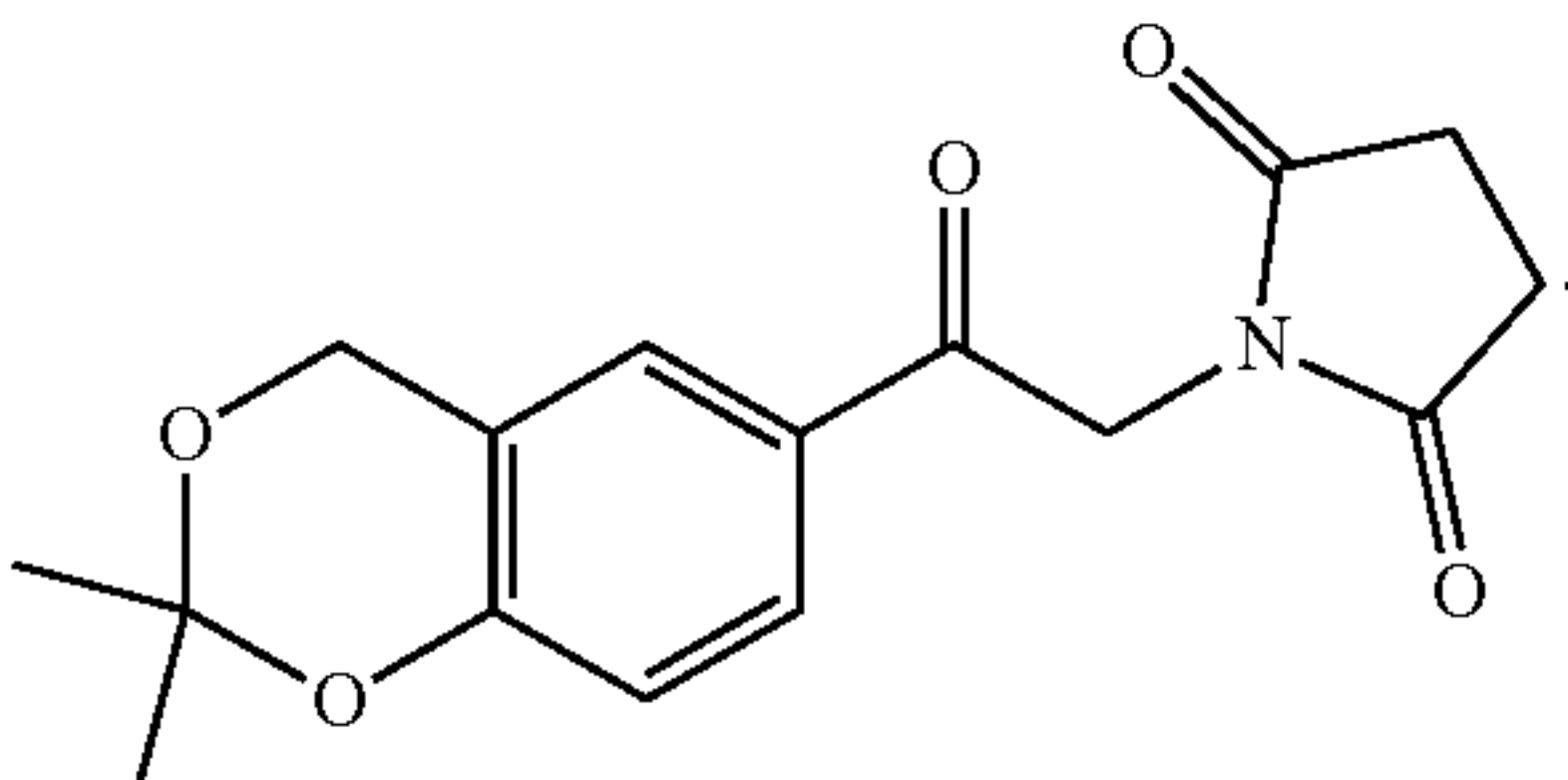


and the salts thereof.

20

2. A compound of formula B:

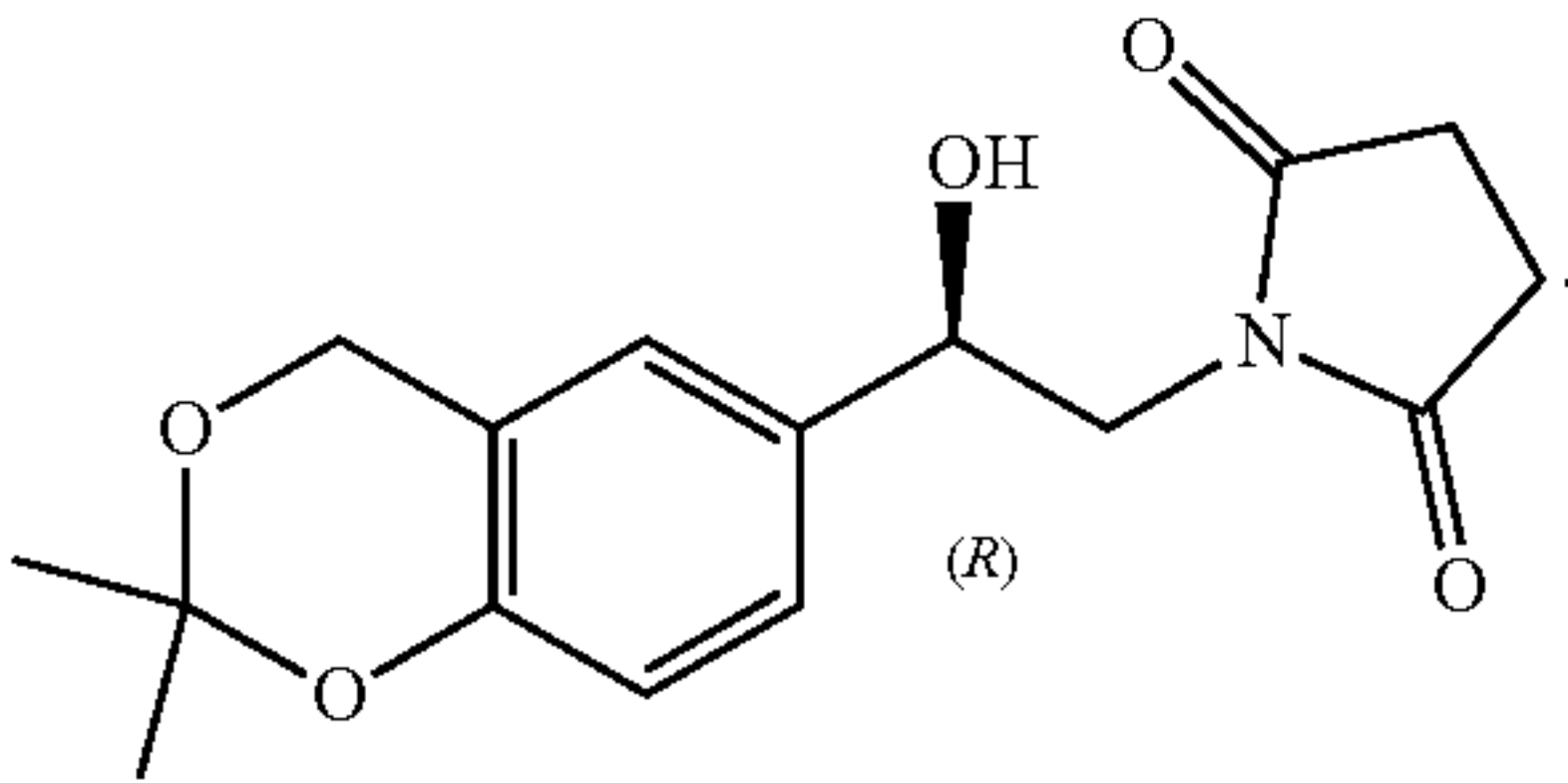
5
B



B

3. A compound of formula C:

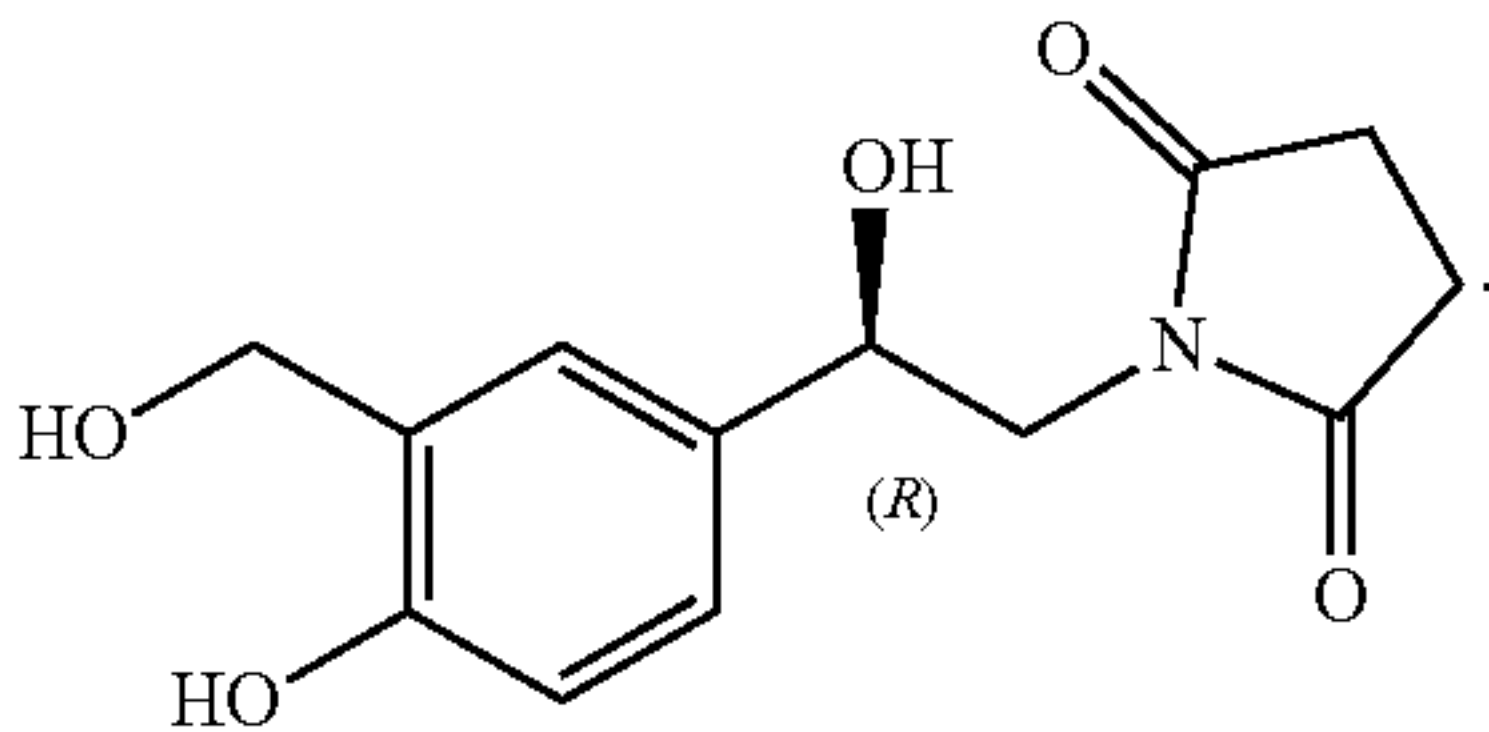
15
C



C

4. A compound of formula Y:

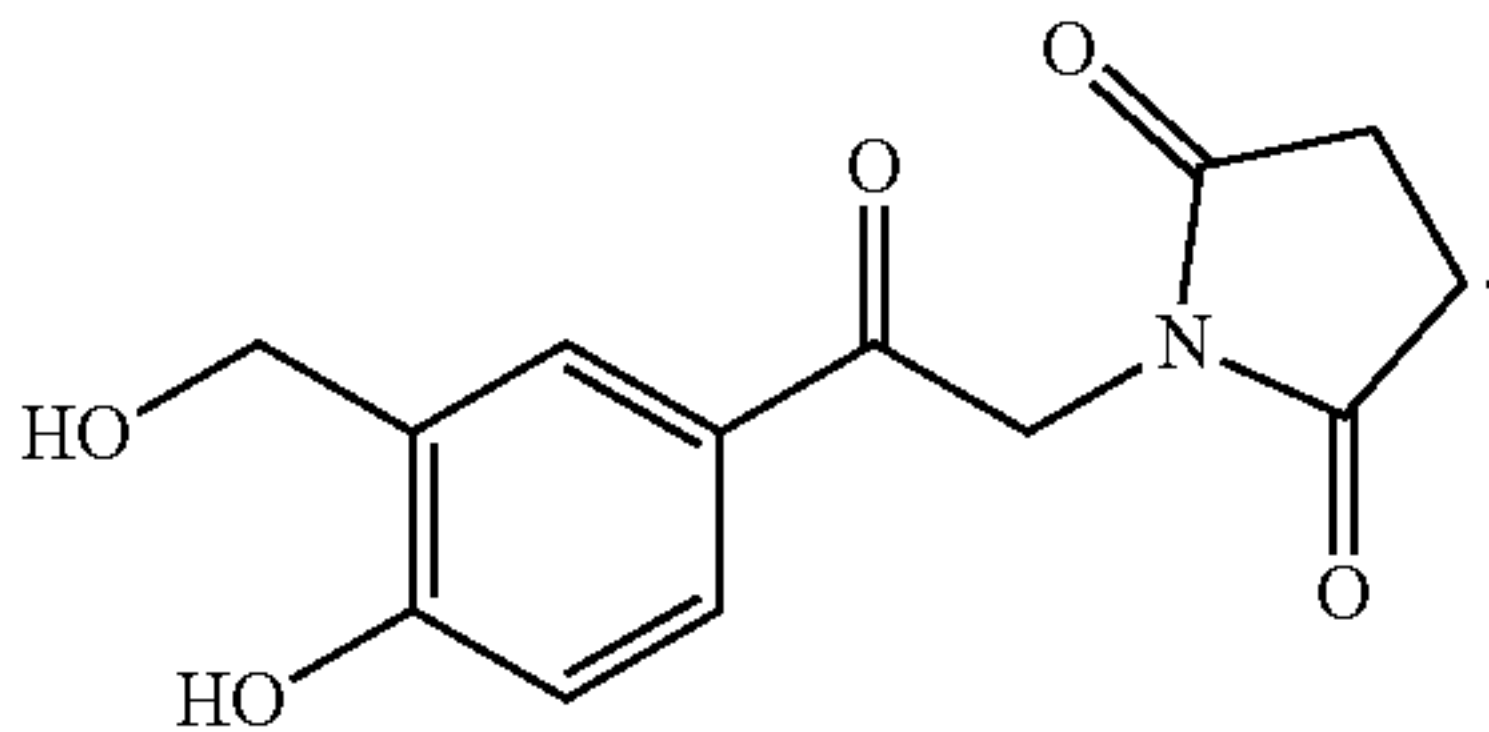
25
Y
30
35



Y

5. A compound of formula Z:

40
Z
45



Z

* * * * *