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(54) **METHODS OF MANUFACTURING A HERMETICALLY SEALED WET ELECTROLYTIC CAPACITOR AND A HERMETICALLY SEALED WET ELECTROLYTIC CAPACITOR**

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See application file for complete search history.

(56) **References Cited**

U.S. PATENT DOCUMENTS

3,763,409 A \* 10/1973 Sheard ..... H01G 4/0085  
106/1.13  
4,780,797 A 10/1988 Libby  
(Continued)

FOREIGN PATENT DOCUMENTS

CN 101271769 A 9/2008  
CN 101271772 A 9/2008  
(Continued)

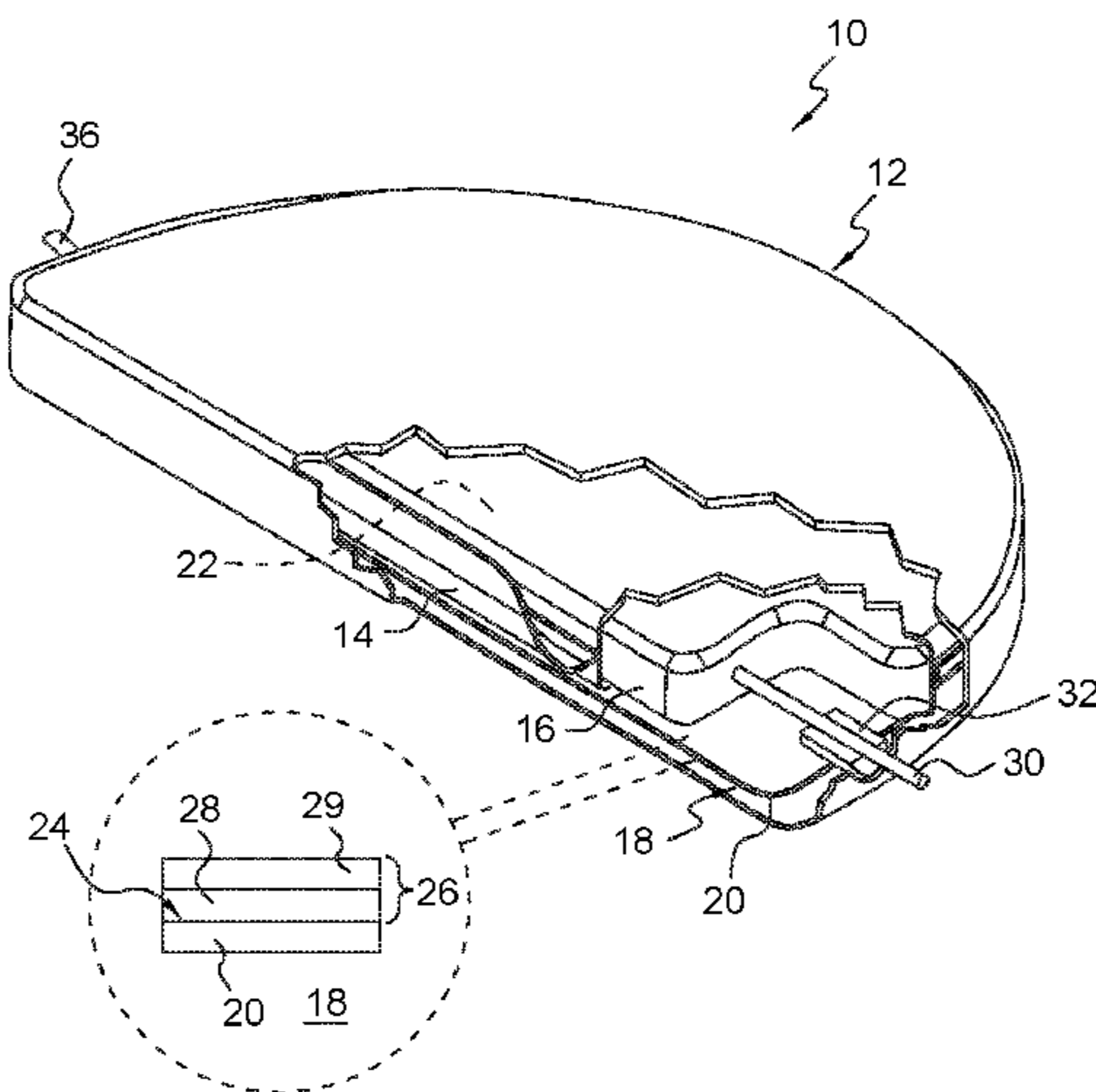
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(57) **ABSTRACT**

Methods of manufacturing a hermetically sealed wet electrolytic capacitor and a hermetically sealed wet electrolytic capacitor are described. A method of manufacturing a wet electrolytic capacitor includes forming a cathode of the capacitor by forming a case comprising a metal substrate, the metal substrate having an alloyed surface, depositing a smooth film comprising palladium and copper as a tacking layer on the alloyed surface of the metal substrate, and depositing a rough, high surface area layer on the tacking layer to achieve a high capacitance cathode. A first terminal is electrically connected to the cathode. An anode is formed. A second terminal is electrically connected to the anode. An electrolytic solution is disposed within the case, and the case is hermetically sealed.

**19 Claims, 1 Drawing Sheet**



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(56) **References Cited**

U.S. PATENT DOCUMENTS

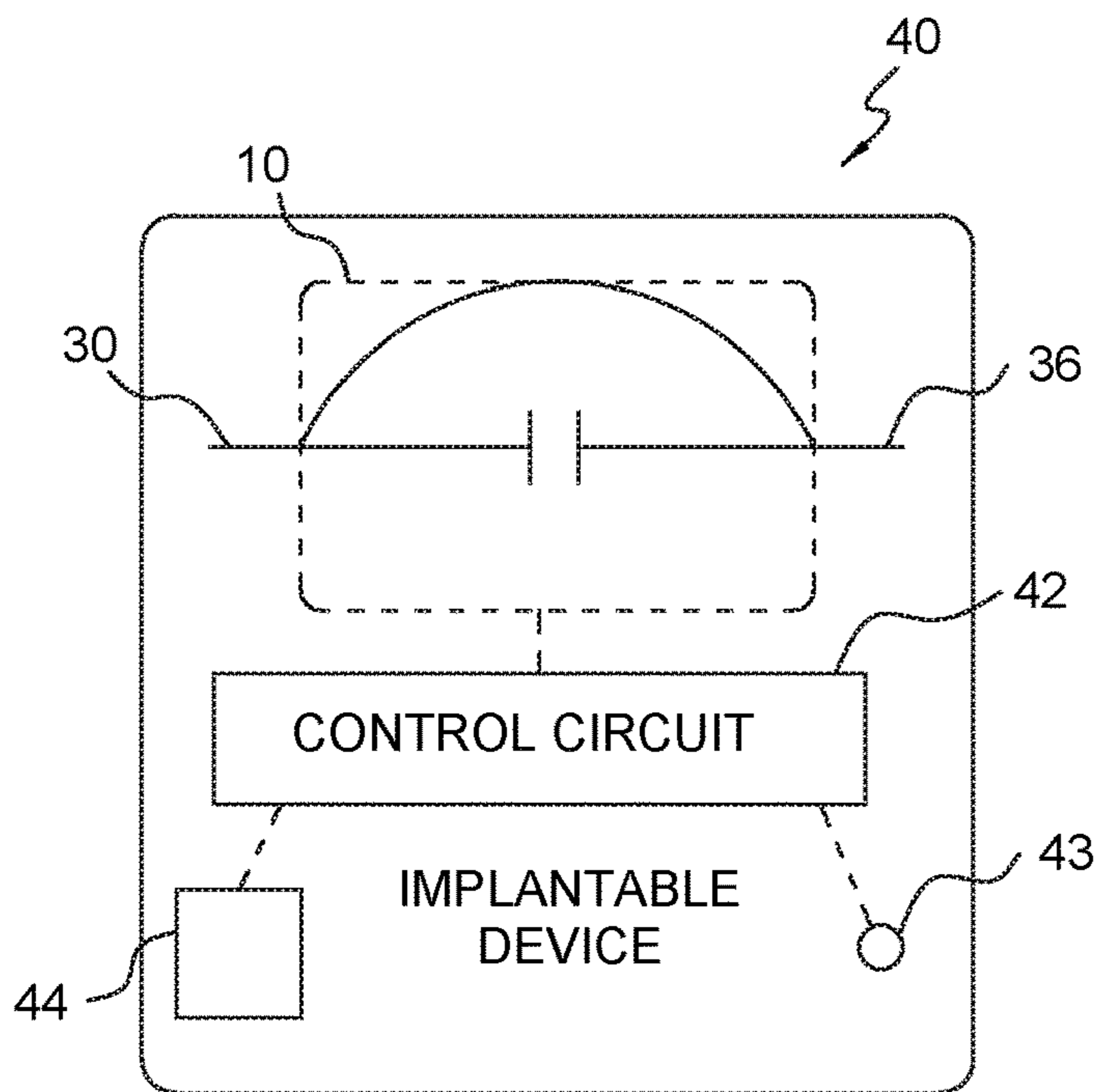
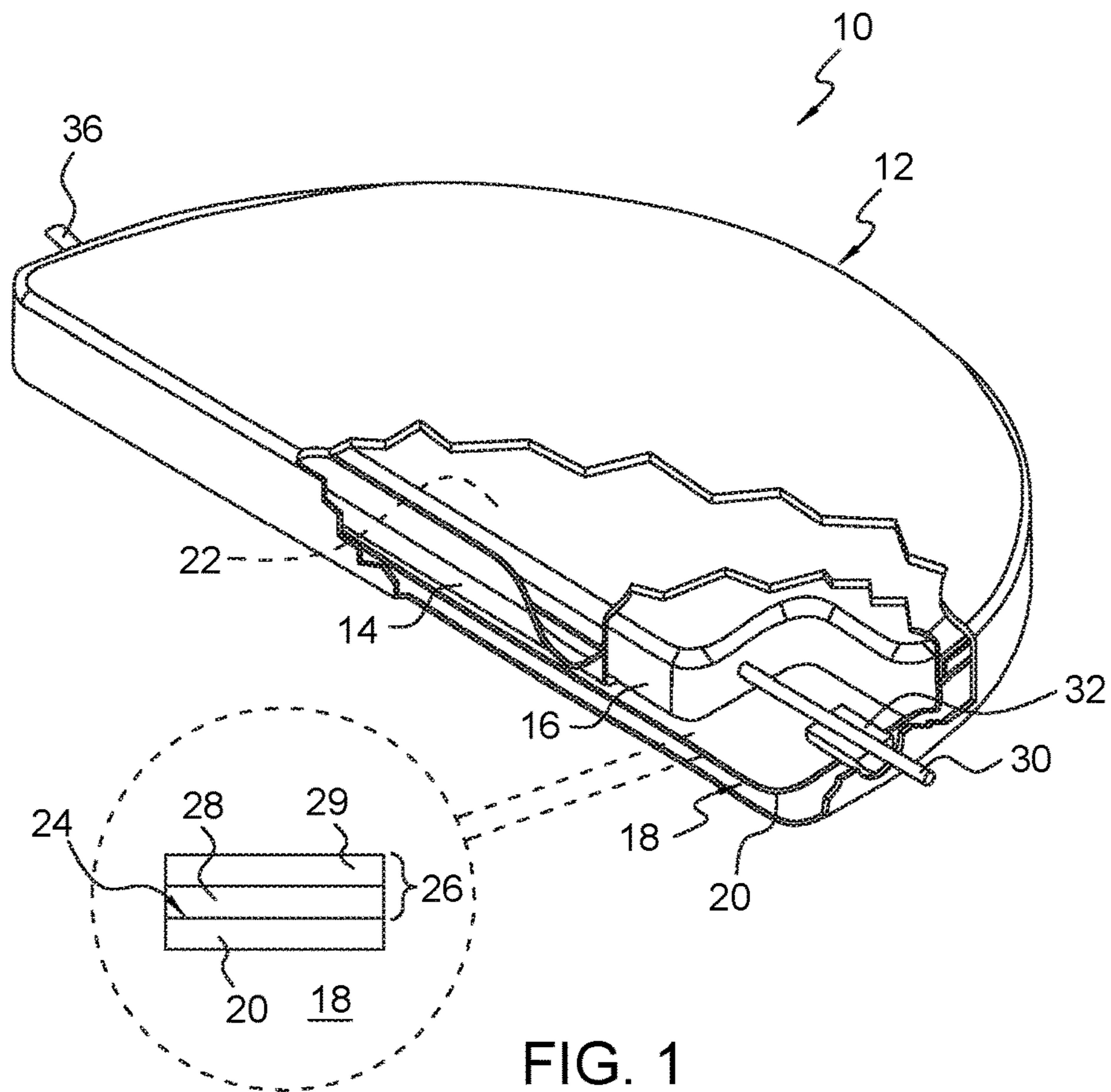
4,942,500	A	7/1990	Libby
5,043,849	A	8/1991	Libby
5,334,219	A	8/1994	Kroll
5,607,454	A	3/1997	Cameron et al.
5,621,608	A	4/1997	Arai et al.
5,926,362	A	7/1999	Muffoletto et al.
6,334,879	B1	1/2002	Muffoletto et al.
6,522,524	B1	2/2003	Feger et al.
6,599,580	B2 *	7/2003	Muffoletto ..... C23C 14/5833 427/250
6,687,117	B2	2/2004	Liu et al.
6,743,370	B1 *	6/2004	Feger ..... H01G 9/035 252/62.2
6,761,728	B2	7/2004	Harguth et al.
6,801,424	B1	10/2004	Nielsen et al.
6,819,544	B1	11/2004	Nielsen et al.
6,965,510	B1	11/2005	Liu et al.

7,169,284	B1	1/2007	Jiang et al.	
2002/0067589	A1 *	6/2002	Marshall .....	H01G 9/022 361/503
2004/0211043	A1 *	10/2004	Will .....	H01G 9/00 29/25.03
2004/0240149	A1	12/2004	Lessner et al.	
2005/0077342	A1	4/2005	Chen et al.	
2005/0177193	A1	8/2005	Nielsen et al.	
2005/0180094	A1 *	8/2005	Muffoletto .....	H01G 9/035 361/504
2005/0219787	A1	10/2005	Stevenson	
2006/0198082	A1 *	9/2006	Eberhard .....	H01G 9/04 361/516
2006/0279907	A1	12/2006	Doffing et al.	
2008/0068779	A1 *	3/2008	Restorff .....	H01G 9/02 361/508
2008/0232029	A1	9/2008	Ning	
2008/0232032	A1	9/2008	Jones et al.	
2010/0268292	A1	10/2010	Eidelman et al.	
2012/0179217	A1 *	7/2012	Bates .....	B22F 1/0055 607/5

FOREIGN PATENT DOCUMENTS

EP	1053763	A2	11/2000	
GB	2447726	A	9/2008	
JP	H02-280310	A	11/1990	
JP	H05-114680	A	6/1993	
JP	07254532	A *	10/1995	..... H01L 24/48
JP	H08-509385	A	10/1996	
JP	H10-312936	A	11/1998	
JP	2003-265627	A	9/2003	
JP	2007-526008	A	9/2007	
JP	2008-235895		10/2008	
WO	94/00193	A1	1/1994	
WO	2005-001997	A2	1/2005	

\* cited by examiner



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**METHODS OF MANUFACTURING A  
HERMETICALLY SEALED WET  
ELECTROLYTIC CAPACITOR AND A  
HERMETICALLY SEALED WET  
ELECTROLYTIC CAPACITOR**

This application is a continuation of U.S. patent application Ser. No. 12/759,769, filed Apr. 14, 2010, which claims the benefit of U.S. Provisional Patent Application No. 61/169,764, filed on Apr. 16, 2009.

**FIELD OF THE INVENTION**

The present invention relates to capacitors, and more specifically to a capacitor suitable for use in medical applications such as implantable cardioverter defibrillators.

**BACKGROUND**

Capacitors are used in a wide range of electronic applications. Certain applications require a capacitor which is capable of a rapid electrical charge to a pre-determined voltage and, once charged, is also capable of delivering sizeable pulses of energy. One example of such an application is in implantable devices. In such an application, it is also important that the capacitor be compact in size and highly reliable.

Thus, what is needed is a capacitor suitable for use in applications, such as implantable cardioverter defibrillators, where reliability and performance are provided in a small size.

**SUMMARY**

Therefore, it is a primary object, feature, or advantage of the present invention to improve over the state of the art.

It is a further object, feature, or advantage of the present invention to provide a capacitor suitable for use in implantable devices.

A still further object, feature, or advantage of the present invention is to provide a capacitor that is capable of a rapid electrical charge to a pre-determined voltage and, once charged, is also capable of delivering sufficient pulses of energy to restore the normal function of a patient's heart when used in implantable cardioverter defibrillators (ICD).

Another object, feature, or advantage of the present invention is to provide a capacitor which is efficiently constructed and shaped to fit into the limited volume available within an ICD.

Yet another object, feature, or advantage of the present invention is to provide a capacitor with high performance and high reliability.

One or more of these and/or other objects, features, or advantages of the present invention will become apparent from the specification and claims that follow.

According to one aspect of the present invention, a hermetically sealed wet electrolytic capacitor is provided. The capacitor has a hermetically sealed case that encloses a cathode, an anode, an electrical insulator between the anode and the cathode and an electrolytic solution. A first terminal is electrically connected to the anode and a second terminal electrically connected to the cathode. The hermetically sealed wet electrolytic capacitor is able to provide a pulse delivery equal to at least 80 percent of the stored energy.

According to another aspect of the present invention, the capacitor's cathode includes a metal substrate having an alloy layer formed with a noble metal and a noble metal/base

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metal electrode element layer electrochemically deposited thereon, and the electrolytic solution has a conductivity between 10 and 60 mS/cm.

According to another aspect of the present invention, a method of manufacturing a capacitor is provided. The method includes hermetically sealing a case containing an electrolytic solution having a conductivity between 10 and 60 mS/cm. The method further includes electrically connecting a first terminal to an anode, the anode being insulated from a cathode. The method further includes electrically connecting a second terminal to the cathode. The cathode is formed from a metal substrate having an alloy layer formed with a noble metal and a noble metal/base metal electrode element layer electrochemically deposited thereon.

**BRIEF DESCRIPTION OF THE DRAWINGS**

FIG. 1 illustrates one embodiment of a hermetically sealed wet electrolytic capacitor.

FIG. 2 illustrates the capacitor of FIG. 1 placed in an implantable cardioverter defibrillator.

**DETAILED DESCRIPTION OF THE  
PREFERRED EMBODIMENT**

The present invention is now described with respect to a particular embodiment. That which is shown is merely for purposes of illustration and example, and one skilled in the art will understand that the present invention contemplates other options, alternatives, or variations.

FIG. 1 illustrates one embodiment of a capacitor **10** of the present invention. Although shown in a semi-circle shape, the capacitor **10** need not have such a shape. This particular shape is merely an example. In FIG. 1, a hermetically sealed wet electrolytic capacitor **10** is shown. The capacitor **10** has a hermetically sealed case **12**. The capacitor **10** has a cathode **18** and an anode **16**. One example design for an anode **16** would comprise sodium reduced capacitor grade tantalum powder pressed to a green density of between 5.0 and 7.0 grams/cc then vacuum sintered between 1450° C. and 1650° C. Powder, press and sinter conditions may be varied to attain the requisite capacitance. Formation of the anode should be in an electrolyte capable of sustaining the voltage necessary for the required oxide thickness.

An insulator **14**, (preferably, but not required, comprising one or more layers of a polymeric material), is positioned between the anode **16** and the cathode **18** to electrically insulate the anode **16** from the cathode **18**. An electrolytic solution **22** is disposed within the hermetically sealed case **12** and surrounds both the cathode **18** and the anode **16**. The electrolytic solution **22** preferably comprises a gel which includes DI water, organic and inorganic acids and an organic solvent. The constituent components of the electrolytic solution **22** may be admixed in a variety of concentrations to provide conductivity within a preferred range between 10 and 60 mS/cm. One example of such an electrolytic solution **22** would be:

65-80% DI water  
0.2-0.6% phosphoric acid  
15-30% ethylene glycol  
3-6% oxalic acid  
2-4% boric acid

The cathode **18** is formed from a metal substrate **20** having an alloy layer **24** formed with a noble metal and a noble metal/base metal electrode element layer **26** electrochemically deposited on the alloyed surface from a solution

of the metal salts. One example design for the cathode **18** may be a mixture of Pd and Cu electrodeposited on a Ti—Pd alloy. To increase adhesion of the cathode **18** to the alloyed substrate, an initial smooth film of Pd—Cu may be electrodeposited as a tacking layer **28**. A rough, high surface area layer **29** can then be deposited on top of the tacking layer to achieve a high capacitance cathode **18**.

The metal substrate **20** of the cathode **18** can be formed of a valve metal. Examples of such valve metals include tantalum, niobium, hafnium, vanadium, zirconium, titanium or any of their alloys. The metal substrate **20** may have any number of shapes or configurations, including a planar or cylindrical shape. The metal substrate **20** may be a liner of any suitable shape and may represent a part of the capacitor case **12**. Such a construction of the cathode **18** results in high cathode capacitance which assists in efficiently delivering energy stored in the capacitor **10** to a load.

A first terminal **30** is shown extending through a spacer **32**. The first terminal **30** is electrically connected to the anode **16**. A second terminal **36** is electrically connected to the cathode **18**.

FIG. 2 illustrates one embodiment of an implantable cardioverter defibrillator (ICD) device **40**. The device **40** includes the capacitor **10** of FIG. 1 (with a first terminal **30** and a second terminal **36**), and a control circuit **42**, which is electrically coupled to the capacitor **10**, a detector **43** and a battery **44**. The capacitor **10** is configured to provide a pulse delivery of at least 80 percent, (but preferably greater than 87 percent), of stored energy between the first and second terminals **30**, **36**. The detector **43** monitors a patient's condition and provides this patient data to the control circuit **42**. The control circuit **42** monitors the information from the detector and upon detection of an anomaly or a critical condition, (which may be defined as one or more predetermined parameters that have exceeded one or more predetermined thresholds).

By way of example, the detector **43** may detect electrical activity in the heart of a patient and forward this data to the control circuit **42**. The control circuit **42** monitors this electrical activity and if it drops below a certain electrical level, or if the electrical activity becomes irregular (as happens with an arrhythmia), initiates delivery of an electrical shock.

The battery **44** may be used to charge the capacitor **10** and to power the ICD device. The charging of the capacitor **10** may be constant (to counter the effects of charge leakage), such that the capacitor **10** is always ready for discharge; may be periodic (i.e. charging at predetermined intervals to keep the charge level of the capacitor **10** above a predetermined threshold); or may be on demand, such that when the onset of an anomaly is detected, the battery **44** is used to charge the capacitor at that time.

In the application of an ICD device **40**, the capacitor **10** performs the function of delivering electrical shock therapy into the heart of a patient when a control circuit **42** of the ICD device **40** detects an anomaly or a critical condition in the patient. The capacitor **10** allows the capacitor to be capable of providing a rapid electrical charge to a predetermined voltage, and thereafter delivering one or more pulses of sufficient energy to restore normal functions of a patient's heart.

The capacitor **10** as shown in FIG. 1 is efficient in nature and highly compact such that the capacitor **10** is constructed and shaped to fit within a limited volume within an ICD device **40**. Preferably, the size of the capacitor **10** is 1.5-3.0 CC, and comprises a half-moon shape as shown in FIG. 1, although this should not be construed to be limiting to the

present invention. The capacitor **10** is able to conform to any size and shape in order to fit the particular configuration demanded by the person within which it is being implanted.

In order to support the application of an ICD device **40**, the capacitor **10** is able to supply a minimum of 9 J, (but preferably 12 J), upon demand. The amount of energy actually delivered is determined by the control circuit **42**.

A hermetically sealed wet electrolytic capacitor has been described. The present invention is not to be limited to the specific embodiment shown or described herein as the present invention contemplates variations in the size and shape of the capacitor, variations in the materials used, and other variations, alternatives, and options as would be apparent to one skilled in the art.

What is claimed is:

1. A method of manufacturing a hermetically sealed wet electrolytic capacitor, the method comprising:

forming a cathode of the capacitor by:

- forming a case comprising a metal substrate, the metal substrate being formed from a valve metal and having a titanium and palladium alloyed surface, depositing a smooth film comprising palladium and copper as a tacking layer on the alloyed surface of the metal substrate,
- depositing a rough layer on the tacking layer to achieve a high capacitance cathode;
- electrically connecting a first terminal to the cathode;
- forming an anode;
- electrically connecting a second terminal to the anode;
- disposing an electrolytic solution within the case; and
- hermetically sealing the case.

2. The method of claim 1, further comprising: positioning an insulator between the cathode and the anode.

3. The method of claim 1, wherein the metal substrate comprises a titanium alloy.

4. The method of claim 1, further comprising forming the anode from sodium reduced capacitor grade tantalum powder pressed to a green density of between 5.0 and 7.0 grams/cc and then vacuum sintered between 1450 degrees Celsius and 1650 degrees Celsius.

5. The method of claim 1, wherein the electrolytic solution has conductivity between 10 and 60 mS/cm.

6. The method of claim 1, wherein the electrolytic solution comprises water, inorganic acids, an organic acid and an organic solvent.

7. A hermetically sealed wet electrolytic capacitor comprising:

- a hermetically sealed capacitor case comprising a multi-layer cathode, the multi-layer cathode comprising:
  - a metal substrate formed from a valve metal and having a titanium and palladium alloyed surface layer,
  - a smooth film comprising palladium and copper deposited on the alloyed surface layer as a tacking layer, and
  - a rough layer deposited on the tacking layer and configured to achieve high capacitance;
- an anode disposed in the hermetically sealed capacitor case; and
- an electrolytic solution disposed in the hermetically sealed capacitor case.

8. The hermetically sealed wet electrolytic capacitor of claim 7, further comprising:

an insulator disposed between the cathode and the anode.

9. The hermetically sealed wet electrolytic capacitor of claim 7, wherein the metal substrate comprises a titanium alloy.

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10. The hermetically sealed wet electrolytic capacitor of claim 7, wherein the anode comprises sodium reduced capacitor grade tantalum powder pressed to a green density of between 5.0 and 7.0 grams/cc and then vacuum sintered between 1450 degrees Celsius and 1650 degrees Celsius. 5

11. The hermetically sealed wet electrolytic capacitor of claim 7, wherein the electrolytic solution has conductivity between 10 and 60 mS/cm.

12. The hermetically sealed wet electrolytic capacitor of claim 7, wherein the electrolytic solution comprises water, inorganic acids, an organic acid and an organic solvent. 10

13. A method of manufacturing a hermetically sealed wet electrolytic capacitor, the method comprising:

providing a case formed from a valve metal having a titanium and palladium alloyed inner surface; 15

forming a cathode on the alloyed inner surface of the case, the cathode comprising multiple layers, at least one of the cathode layers comprising a smooth film comprising palladium and copper as a tacking layer deposited on the alloyed inner surface of the case; 20

disposing an anode within the case,

electrically connecting a first terminal to an anode, the anode being insulated from the cathode;

electrically connecting a second terminal to the cathode;

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disposing an electrolytic solution within the case; and hermetically sealing the case;

wherein the amount of energy delivered by the capacitor is greater than 87 percent of the stored energy of the capacitor.

14. The method of claim 13, further comprising preparing the electrolytic solution by admixing DI water, phosphoric acid, ethylene glycol and boric acid.

15. The method of claim 13, wherein the step of forming the cathode comprises depositing a layer on the tacking layer, wherein the layer has a rough surface.

16. The method of claim 13, further comprising: positioning an insulator between the cathode and the anode.

17. The method of claim 13, wherein the alloyed inner surface comprises a titanium alloy.

18. The method of claim 13, further comprising forming the anode from sodium reduced capacitor grade tantalum powder pressed to a green density of between 5.0 and 7.0 grams/cc and then vacuum sintered between 1450 degrees Celsius and 1650 degrees Celsius.

19. The method of claim 13, wherein the electrolytic solution has conductivity between 10 and 60 mS/cm.

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