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(54) **COMPOSITE 3D-PRINTED REACTORS FOR GAS ABSORPTION, PURIFICATION, AND REACTION**

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B29K 83/00 (2006.01)
B33Y 10/00 (2015.01)

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(58) **Field of Classification Search**
CPC B01D 53/1475
See application file for complete search history.

(56) **References Cited**

U.S. PATENT DOCUMENTS

8,361,195 B2 1/2013 Aines et al.
8,394,350 B2 3/2013 Aines

(Continued)

FOREIGN PATENT DOCUMENTS

WO 2014-153535 A2 9/2014
WO 2016-154278 A1 9/2016

OTHER PUBLICATIONS

Vericella et al., "Encapsulated Liquid Sorbents for Carbon Dioxide Capture," Nature Comm., 2015, pp. 1-7.

(Continued)

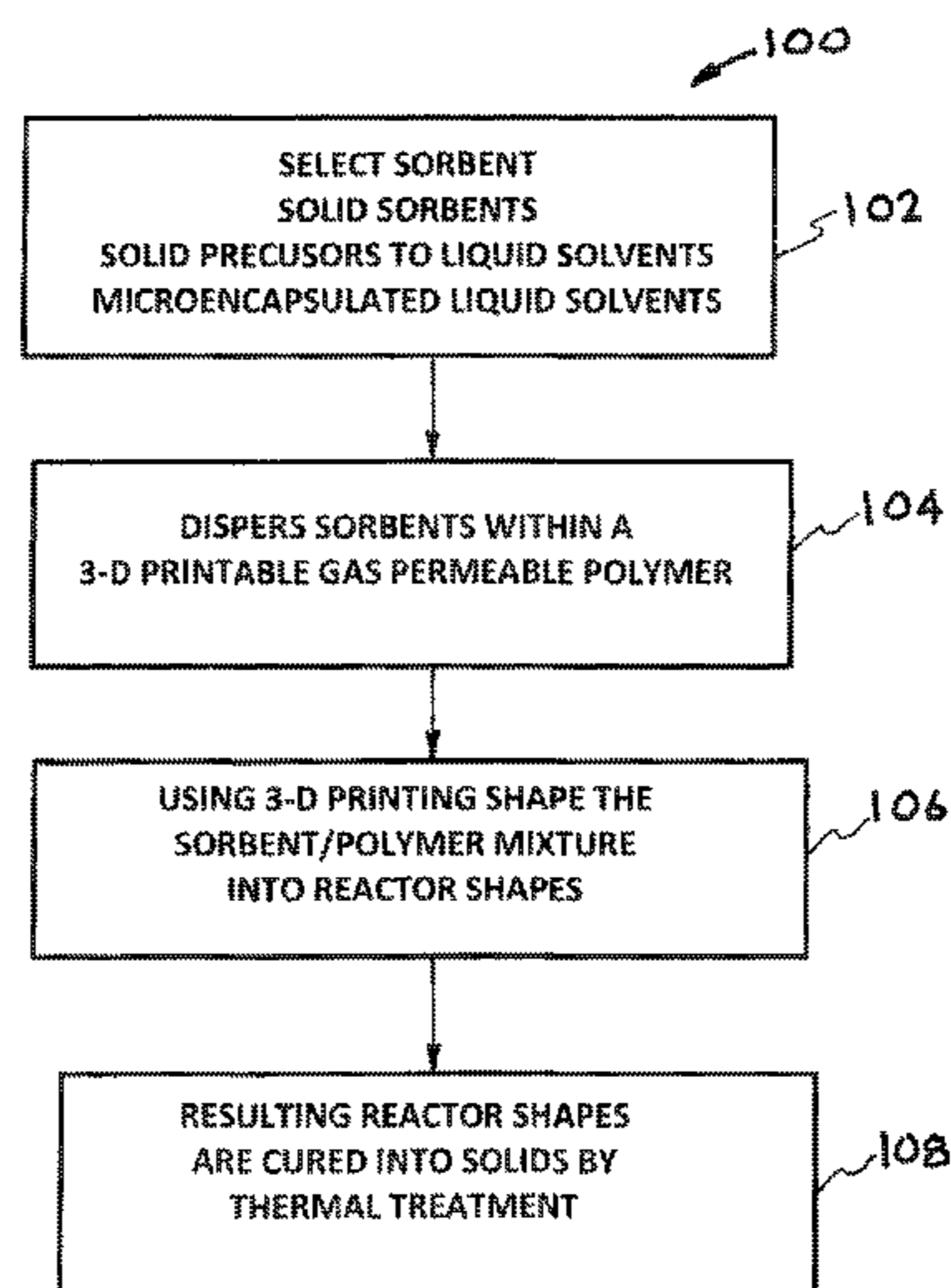
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(57) **ABSTRACT**

A composite material for gas capture, notably CO₂ capture and storage. The composite material includes a mixture of a solid or liquid reactive filler and a gas-permeable polymer such that the reactive filler forms micron-scale domains in the polymer matrix.

9 Claims, 10 Drawing Sheets



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B33Y 80/00 (2015.01)
B29C 64/106 (2017.01)
B29C 64/386 (2017.01)
B29K 509/02 (2006.01)

(56) **References Cited**

U.S. PATENT DOCUMENTS

8,834,605	B2	9/2014	Aines	
9,062,586	B2 *	6/2015	Jiang	F01N 3/2828
2013/0243675	A1 *	9/2013	Halder	B01D 53/02 423/220
2015/0010461	A1	1/2015	Cronin	
2016/0067891	A1	3/2016	Worsley et al.	

OTHER PUBLICATIONS

International Search Report and Written Opinion for PCT/US2018/023889 corresponding to U.S. Appl. No. 15/468,297, 13 pages.

* cited by examiner

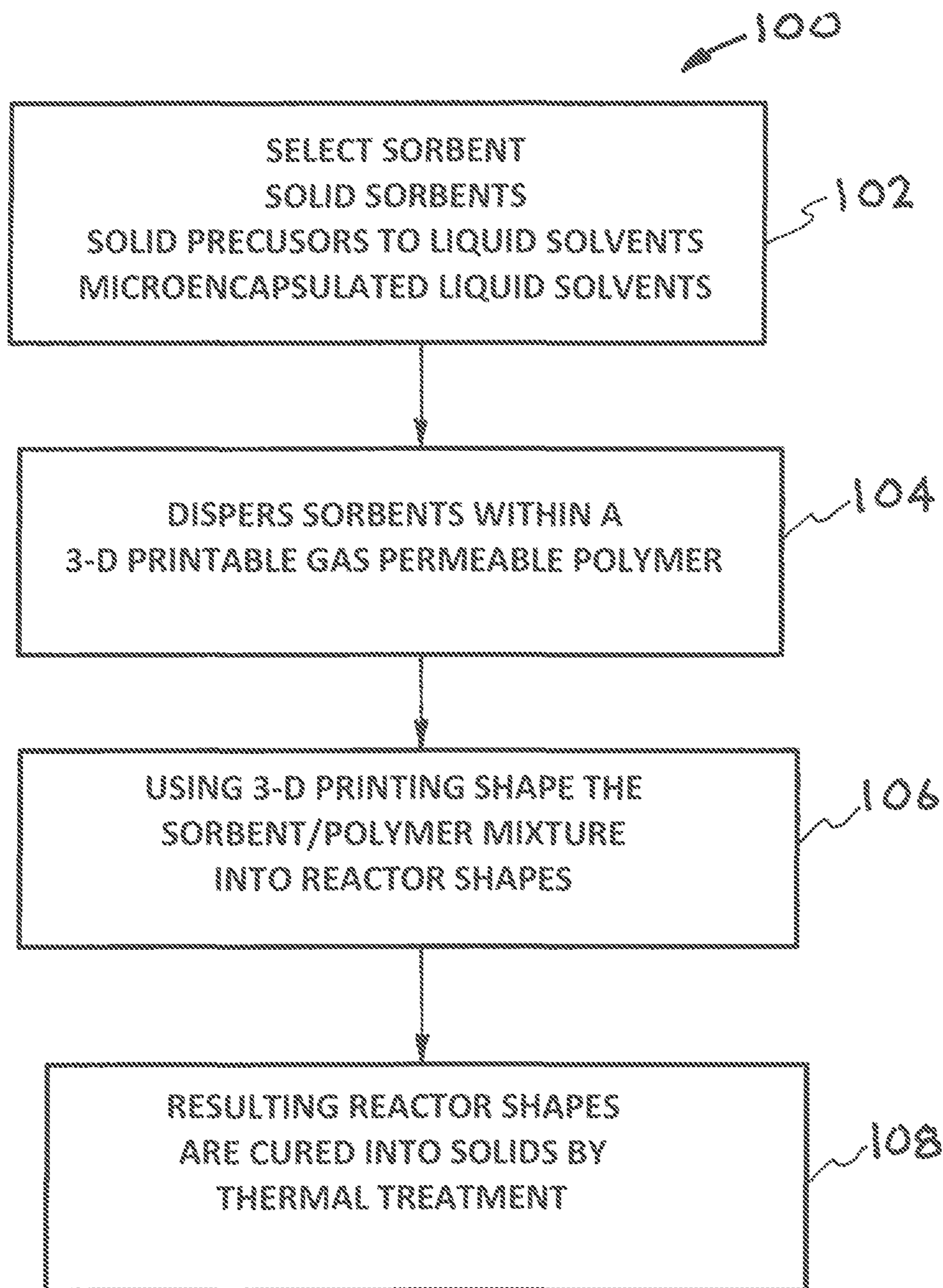


FIG. 1

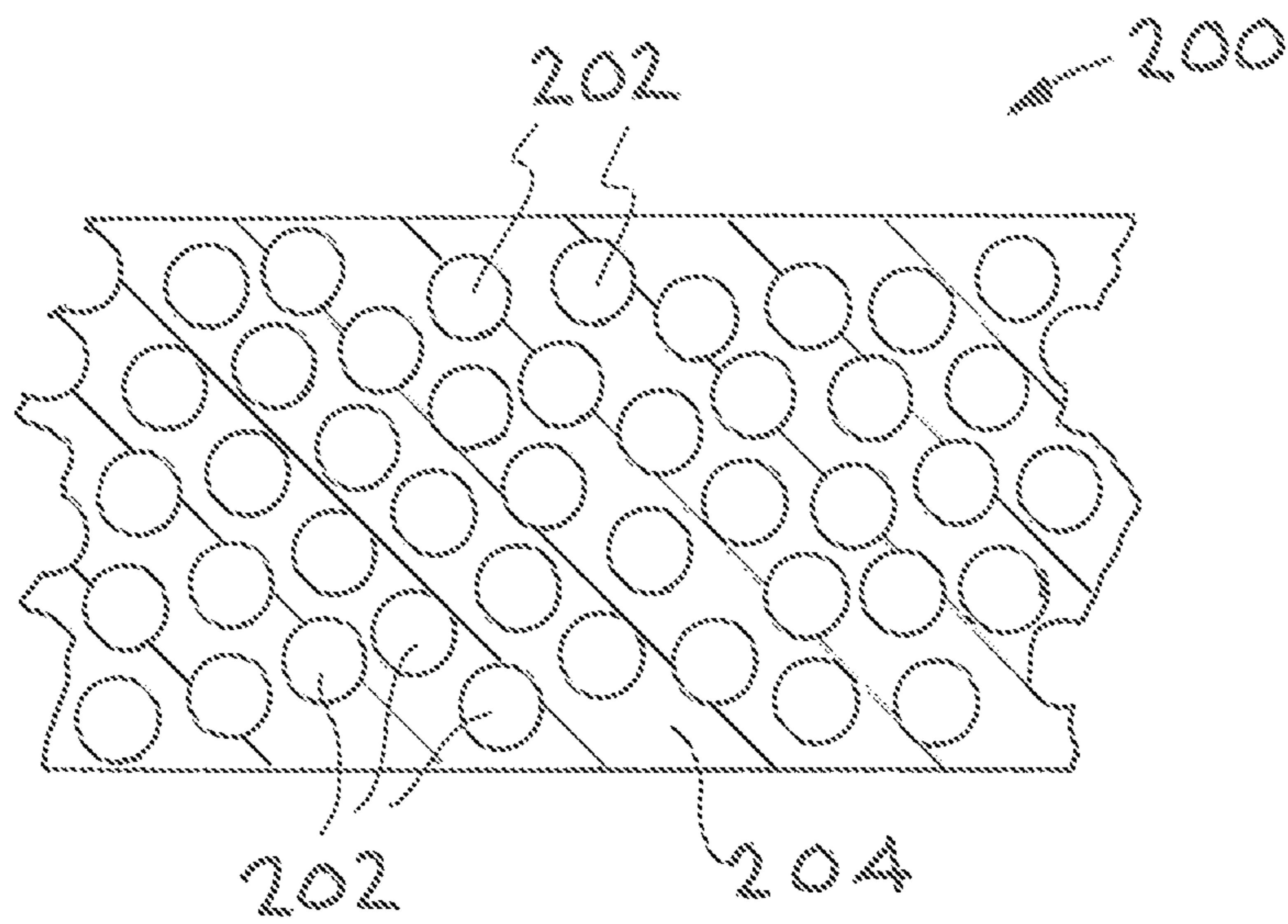


FIG. 2

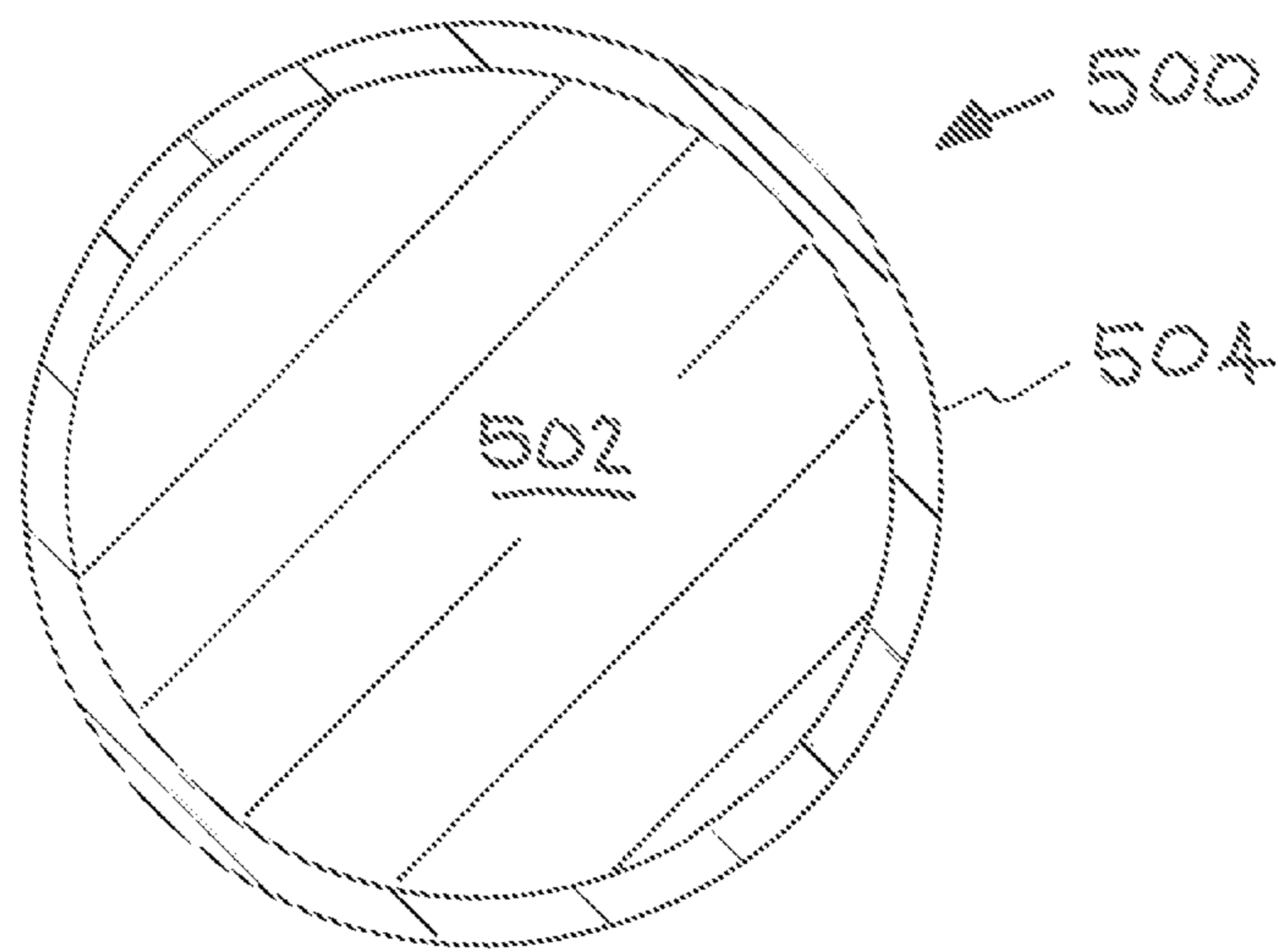


FIG. 5

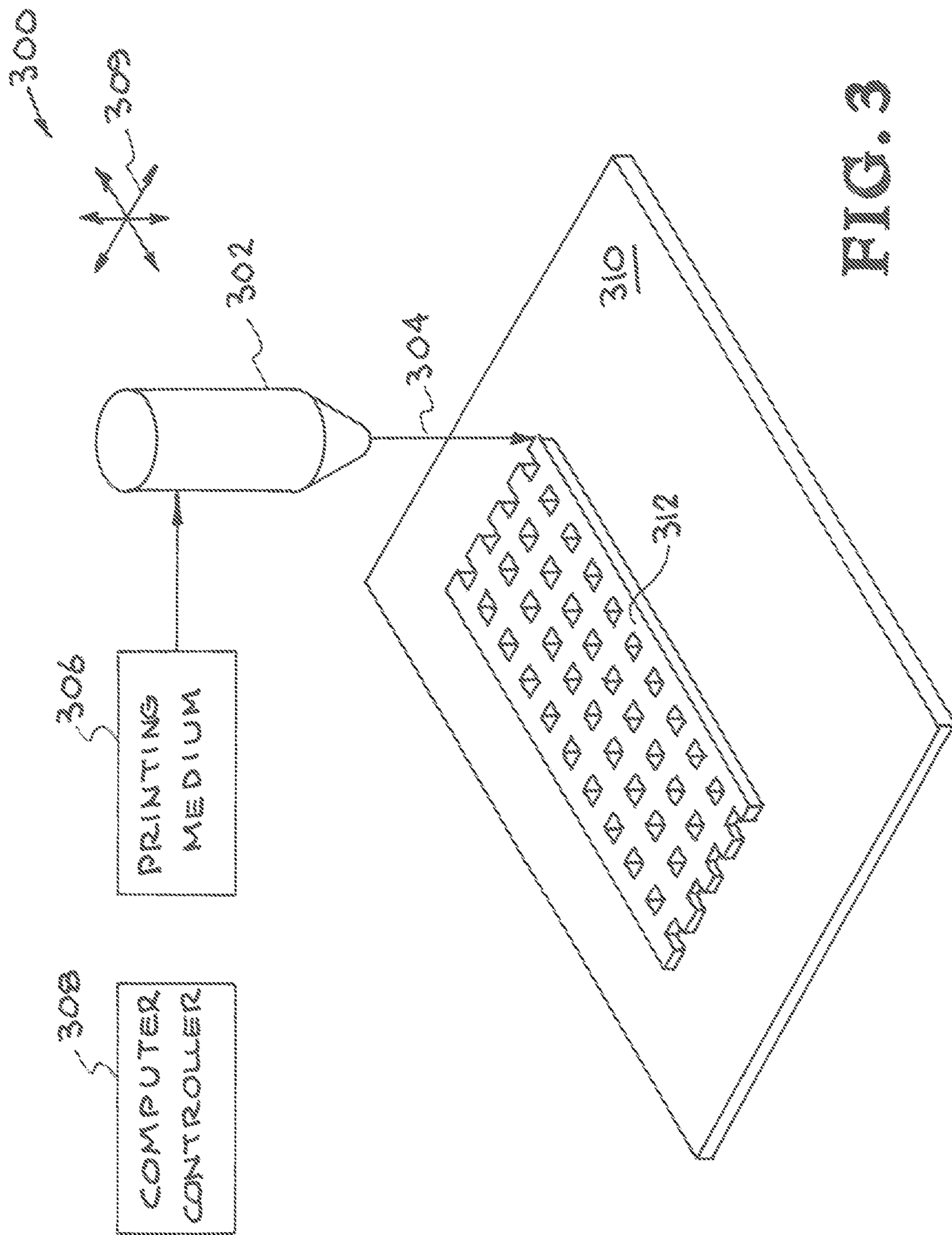
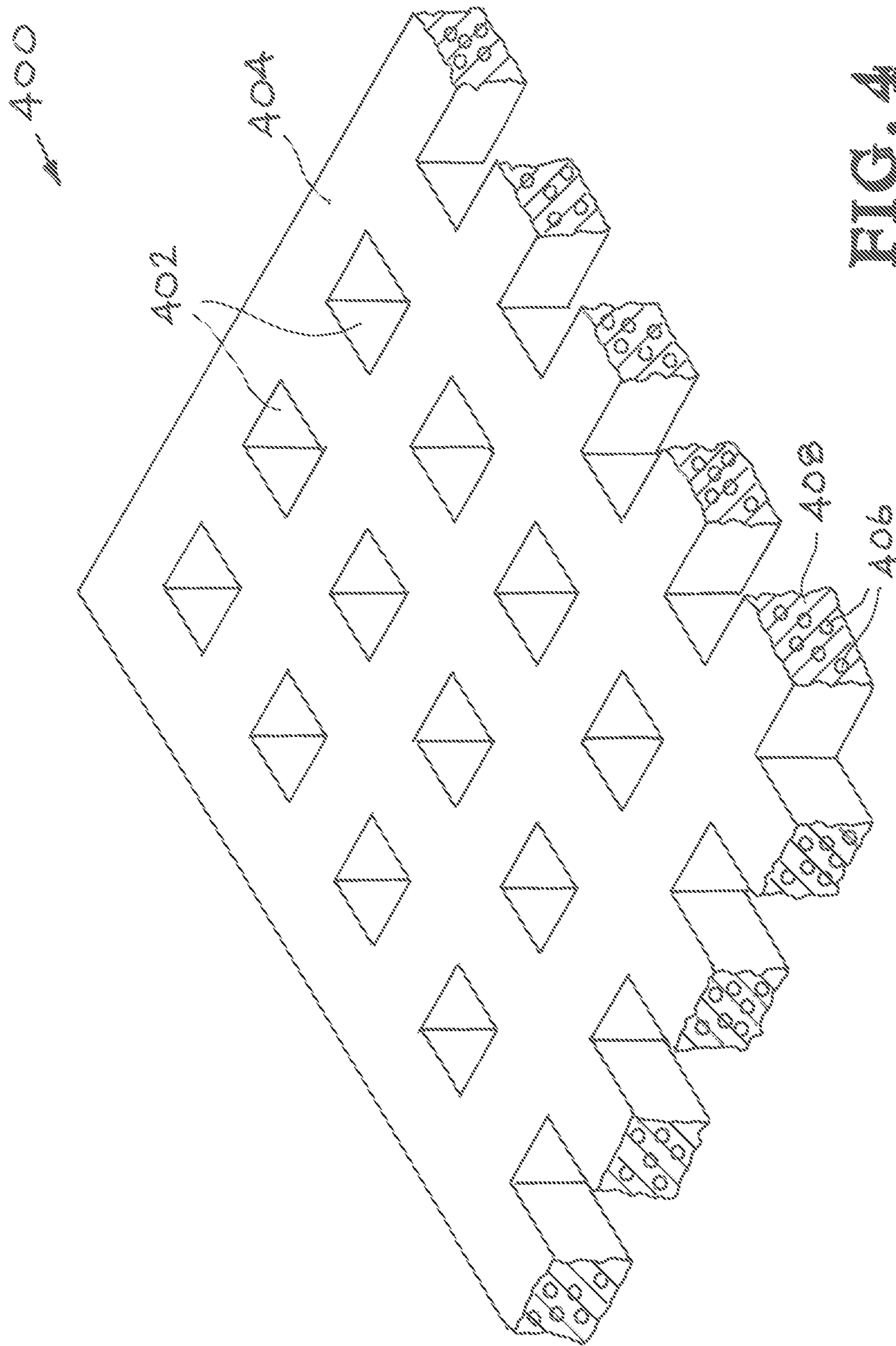


FIG. 3



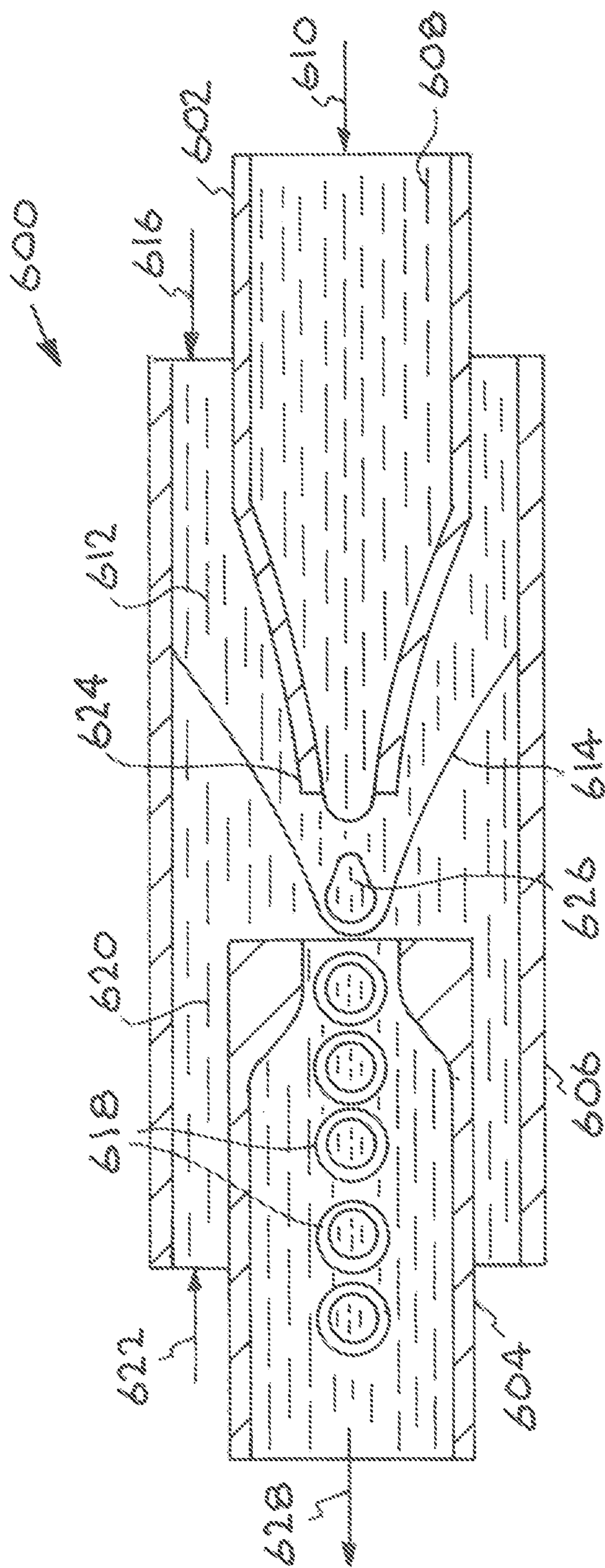


FIG. 6

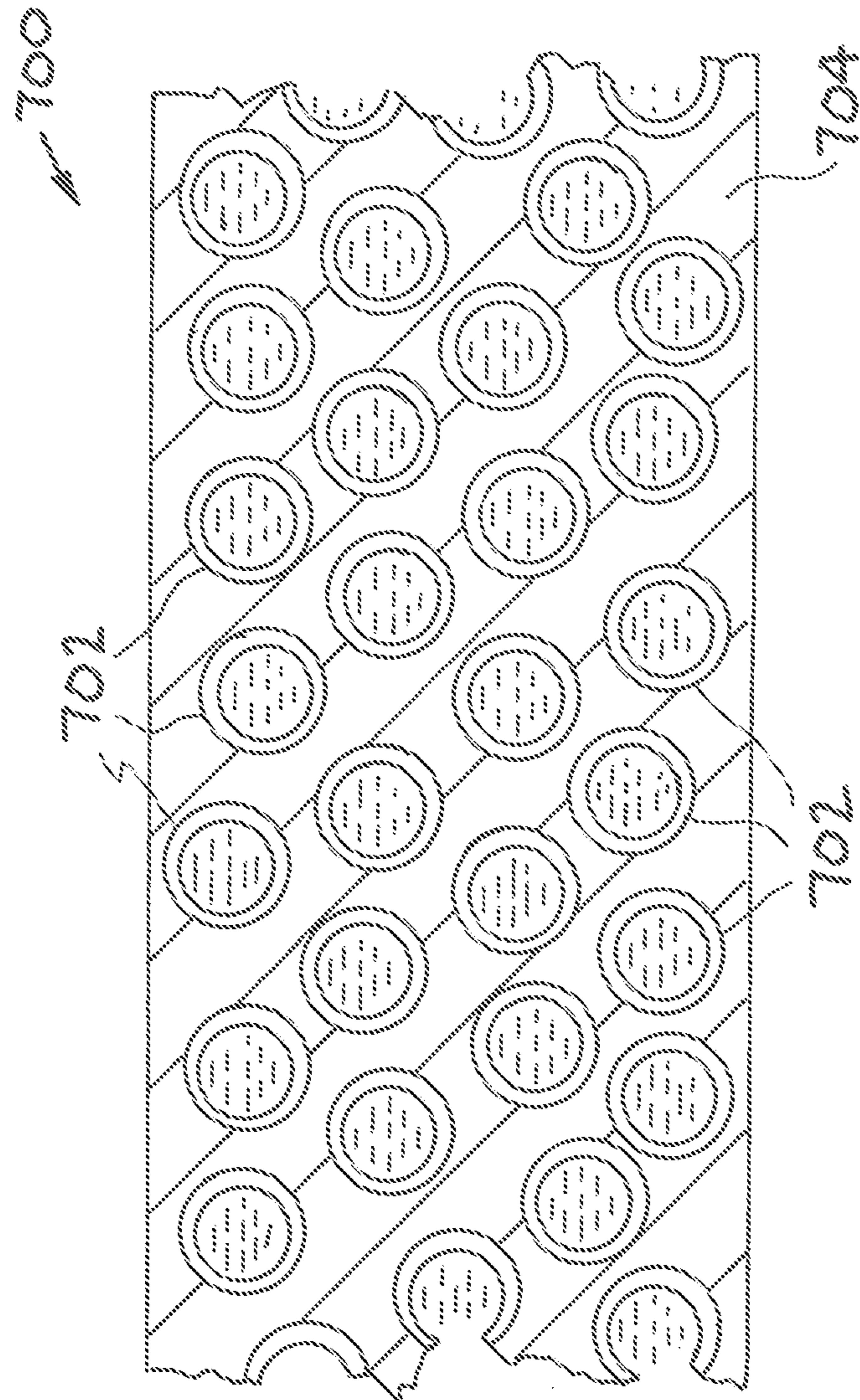


FIG. 7

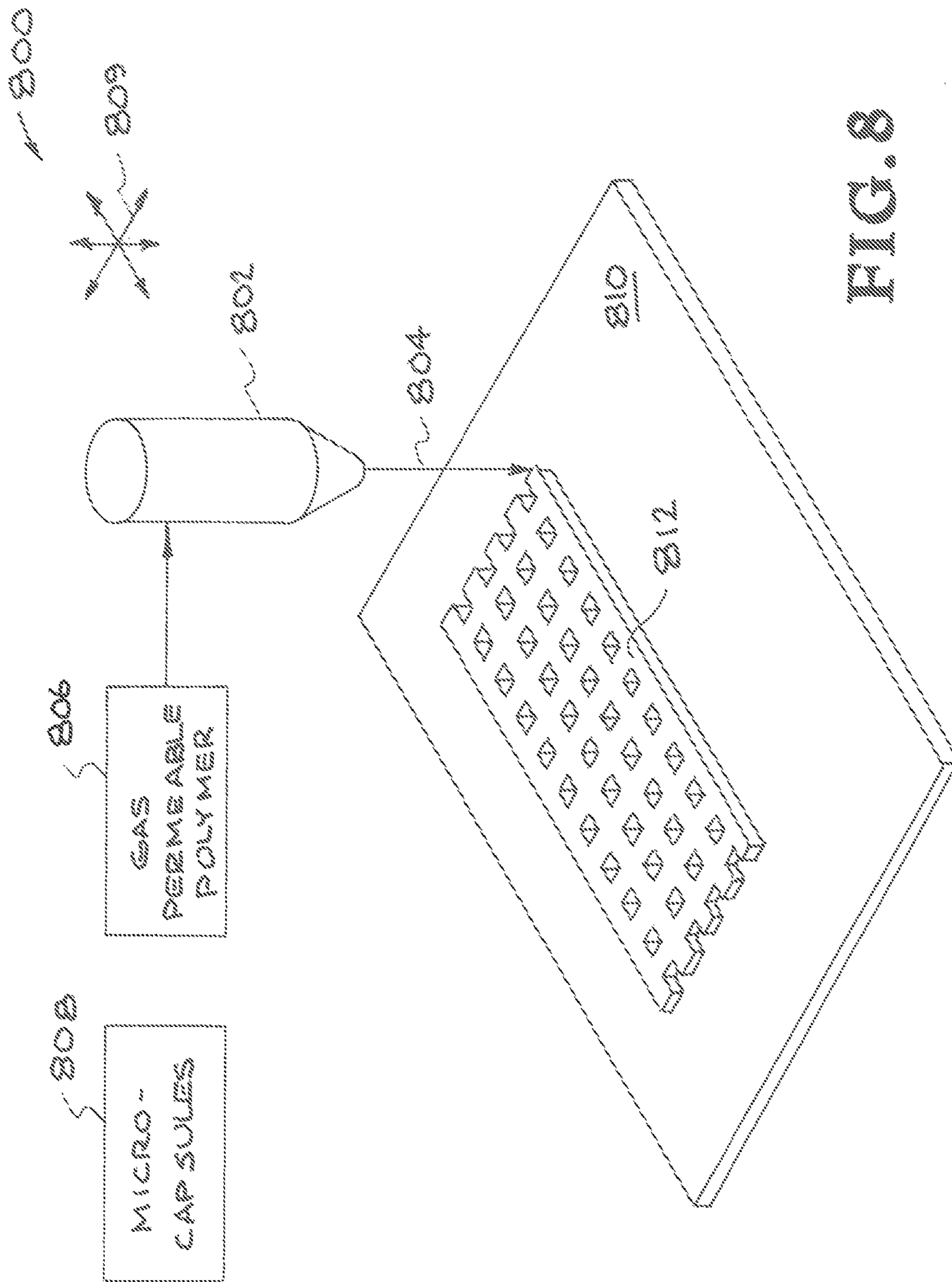


FIG. 8

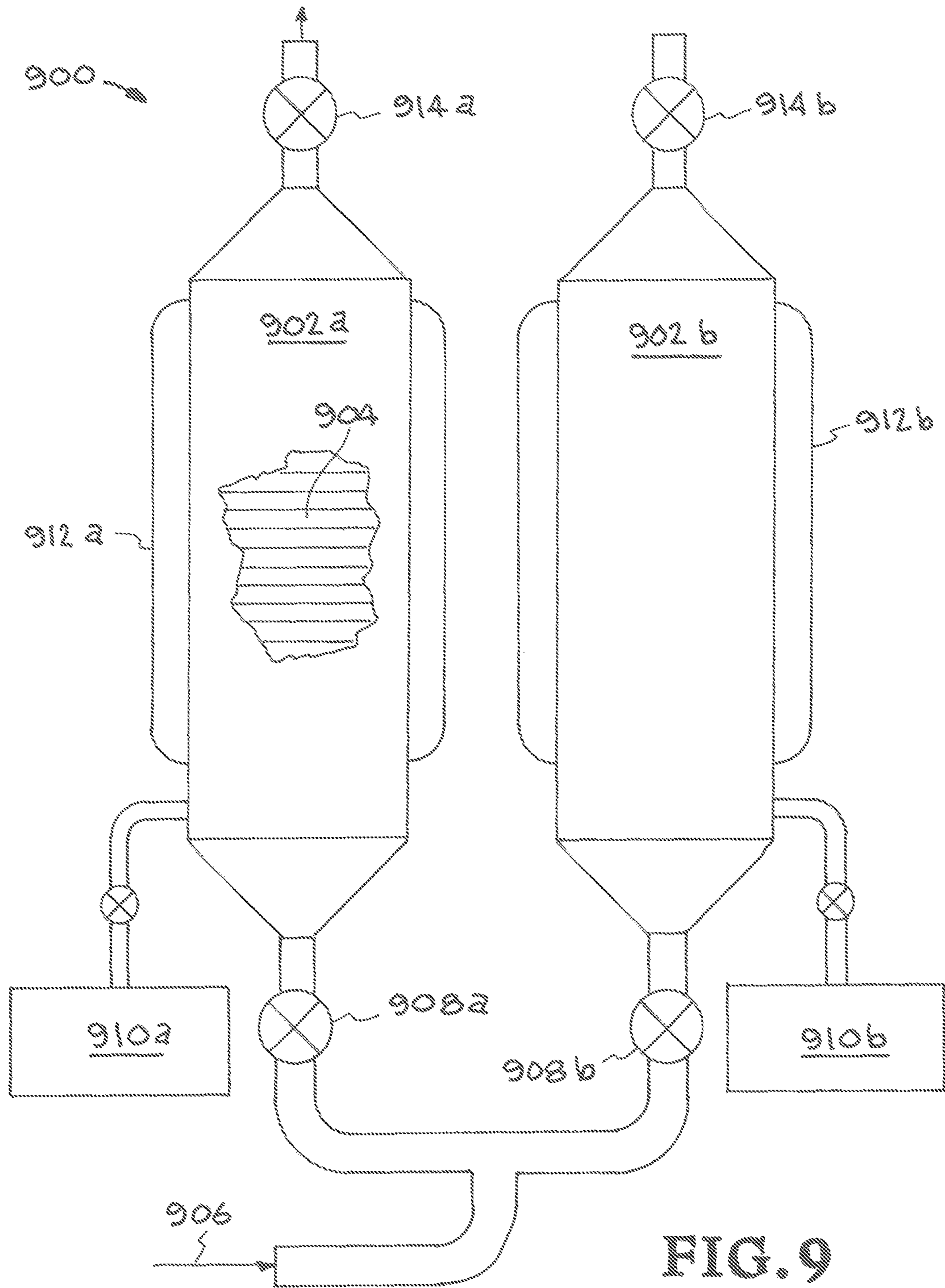


FIG. 9

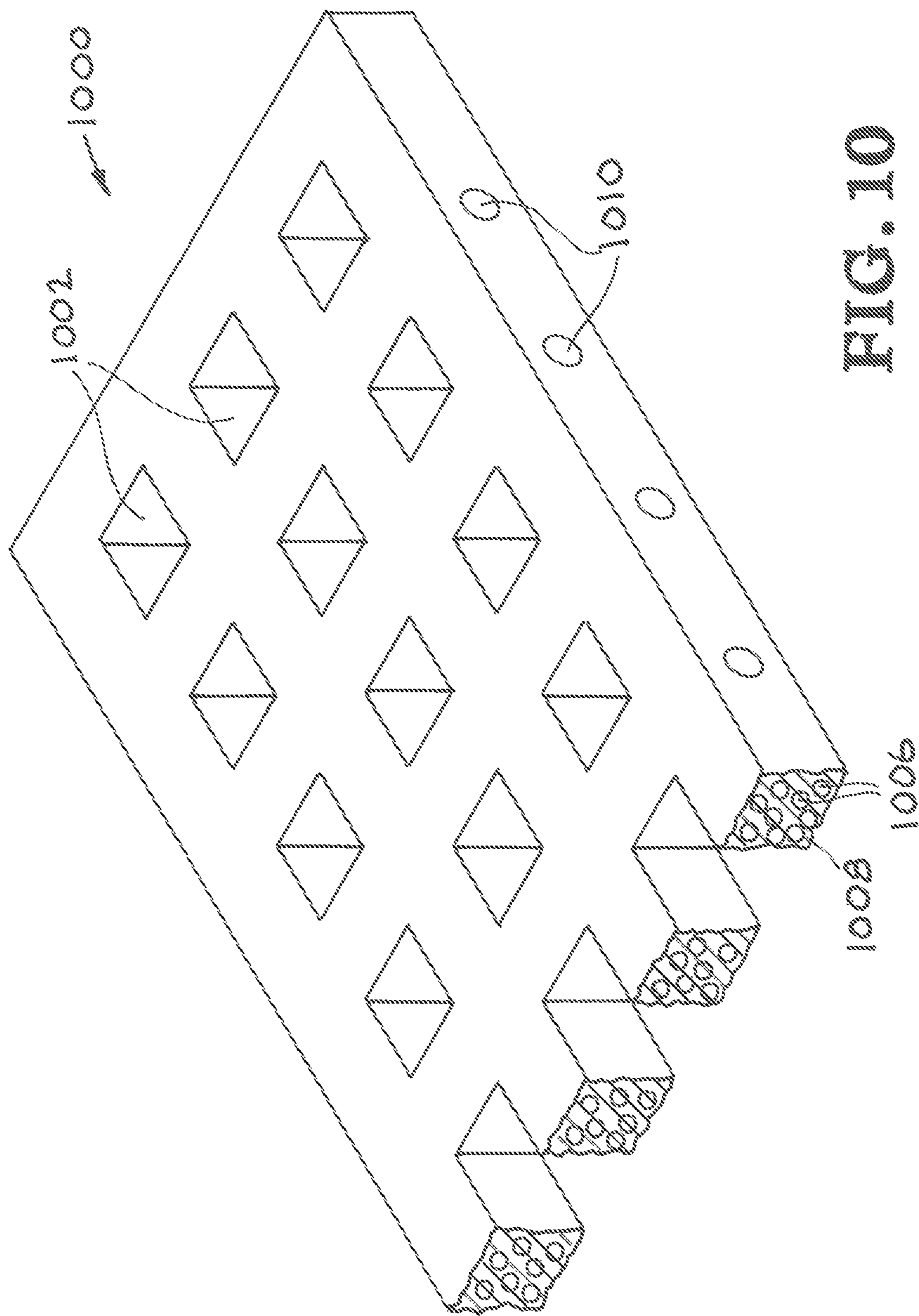


FIG. 10

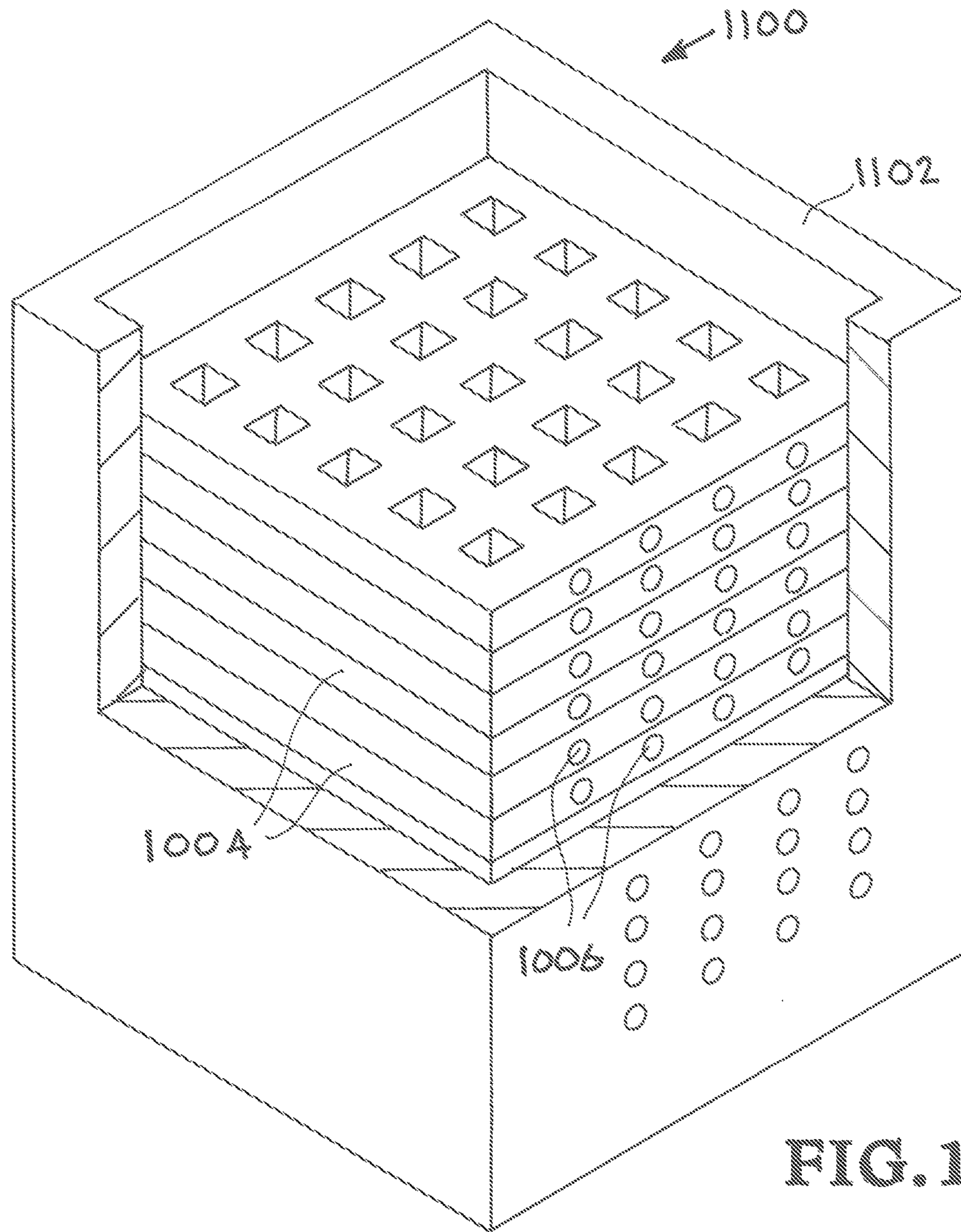


FIG. 11

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COMPOSITE 3D-PRINTED REACTORS FOR GAS ABSORPTION, PURIFICATION, AND REACTION

STATEMENT AS TO RIGHTS TO
APPLICATIONS MADE UNDER FEDERALLY
SPONSORED RESEARCH AND
DEVELOPMENT

The United States Government has rights in this applica-
tion pursuant to Contract No. DE-AC52-07NA27344
between the United States Department of Energy and Law-
rence Livermore National Security, LLC for the operation of
Lawrence Livermore National Laboratory.

BACKGROUND

Field of Endeavor

The present application relates to reactors for gas absorp-
tion, purification, and/or reaction and more particularly to
composite 3-D printed reactors for gas absorption, purifica-
tion, and/or reaction.

State of Technology

This section provides background information related to
the present disclosure which is not necessarily prior art.

The exchange of gas into or out of a liquid is a common
problem in the absorption of gases into a solvent for indus-
trial chemical processes, gas purification, and water purifi-
cation. The potentially largest scale application is for the
absorption of CO₂ for carbon capture and storage from
power plants. Other applications include purification of
natural gas, purification of biogas, and various industrial
gas-to-liquid reactions. The most common method for gas
absorption is the use of a "packed tower" absorption column.
The absorption column is typically a cylindrical reactor
filled with a packing material. Liquid solvent is pumped to
the top of the tower and allowed to flow down over the
packing while gas is blown from the bottom of the tower in
the opposite direction. The liquid solvent forms a film over
the wetted parts of the packing material, resulting in a
gas-liquid interface where the exchange between CO₂ and
solvent takes place.

A major limitation of these tower packings is that the
surface-area to volume ratio of the liquid is limited by the
thickness of the liquid film. This thickness is determined by
the properties of the solvent, but is typically around 1 mm.
Additional area can be put into the tower using finer pack-
ings, but this leads to higher holdup of liquid, and impeded
gas flow.

Solid sorbents are an alternative to liquid solvents in
many applications, including large-scale CO₂ capture. Solid
sorbents are preferred for air purification for, e.g. small
submarines and personal underwater rebreathers and
removal of volatile organic compounds emitted from certain
industrial processes. Solid sorbents include mineral CO₂
sorbents like soda-lime, designer gas sorbents like metal-
organic frameworks (MOFs), zeolites, and activated car-
bons. Solid sorbents are typically prepared in a powder, and
must be pelletized or formed into monoliths with a binder,
reducing accessible surface area and yielding sub-optimal
gas flow.

SUMMARY

Features and advantages of the disclosed apparatus, sys-
tems, and methods will become apparent from the following

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description. Applicant is providing this description, which
includes drawings and examples of specific embodiments, to
give a broad representation of the apparatus, systems, and
methods. Various changes and modifications within the
spirit and scope of the application will become apparent to
those skilled in the art from this description and by practice
of the apparatus, systems, and methods. The scope of the
apparatus, systems, and methods is not intended to be
limited to the particular forms disclosed and the application
covers all modifications, equivalents, and alternatives falling
within the spirit and scope of the apparatus, systems, and
methods as defined by the claims.

The inventor's apparatus, systems, and methods provide a
composite material for gas capture, notably CO₂ capture and
storage. The composite material includes a mixture of a solid
or liquid reactive filler and a gas-permeable polymer (e.g.
silicone), such that the reactive filler forms micron-scale
domains in the polymer matrix. In contrast to typical absorp-
tion schemes based on liquid solvents or solid sorbent
powders, the composite materials can be fabricated into
arbitrary fixed shapes via additive or conventional manu-
facturing. The gas-permeable polymer matrix acts as a
gas-permeable support while the reactive filler acts as a gas
sorbent or catalyst for chemical reactions. Control over the
material shape allows for the patterning of high surface-
area-to-volume ratio structures for fast reactivity while
minimizing pressure drops typically associated with high
surface area materials and packings. The inventor's appa-
ratus, systems, and methods can be used for the absorption
of gases or catalyzing chemical reactions involving a gas.
This use can be tailored for specific applications such as CO₂
capture from power plants, CO₂ utilization, natural gas
purification, biogas purification, and underwater rebreather
applications. Specific gases and reactions can be targeted
using different reactive fillers.

The apparatus, systems, and methods are susceptible to
modifications and alternative forms. Specific embodiments
are shown by way of example. It is to be understood that the
apparatus, systems, and methods are not limited to the
particular forms disclosed. The apparatus, systems, and
methods cover all modifications, equivalents, and alterna-
tives falling within the spirit and scope of the application as
defined by the claims.

BRIEF DESCRIPTION OF THE DRAWINGS

The accompanying drawings, which are incorporated into
and constitute a part of the specification, illustrate specific
embodiments of the apparatus, systems, and methods and,
together with the general description given above, and the
detailed description of the specific embodiments, serve to
explain the principles of the apparatus, systems, and meth-
ods.

FIG. 1 is a flow chart illustrating one embodiment of the
inventor's apparatus, systems, and methods.

FIG. 2 illustrates one embodiment of the inventors' ink
that can be used to create a monolith reactor.

FIG. 3 an embodiment of the inventor's 3D printing and
otherwise manufacturing of a monolithic reactor is illus-
trated.

FIG. 4 illustrates an embodiment of the inventor's mono-
lith reactor.

FIG. 5 illustrates spheres containing liquid stripping sol-
vent for carbon dioxide removal.

FIG. 6 is an illustration of a system for making the
spheres.

FIG. 7 illustrates another embodiment of the inventors' ink that can be used to create a monolith reactor.

FIG. 8 is an illustration of an additive manufacturing system for making an embodiment of the monolithic reactor.

FIG. 9 illustrates an embodiment of a system utilizing the inventor's reactor for removing the target gas from a fluid or mixture and regenerating the reactor.

FIG. 10 illustrates another embodiment of the reactor.

FIG. 11 illustrates an embodiment of a reactor stack.

DETAILED DESCRIPTION OF SPECIFIC EMBODIMENTS

Referring to the drawings, to the following detailed description, and to incorporated materials, detailed information about the apparatus, systems, and methods is provided including the description of specific embodiments. The detailed description serves to explain the principles of the apparatus, systems, and methods. The apparatus, systems, and methods are susceptible to modifications and alternative forms. The application is not limited to the forms disclosed. The application covers all modifications, equivalents, and alternatives falling within the spirit and scope of the apparatus, systems, and methods as defined by the claims.

The inventor's apparatus, systems, and methods provide a high-surface-area, hierarchically-structured, reactive composite material for energy-efficient gas purification and the techniques to fabricate this material. The hierarchy consists of, on the one hand, micron-scale domains of solvent or sorbent particles embedded within a gas-permeable polymer to form the composite material and, on the other hand, the submillimeter-scale structures that the composite materials form to create a monolith reactor. The reactor has channels for optimal gas flow and features sized for fast reaction with the gas. The structure optionally consists of hollow tubes where a second solvent, for heat exchange or additional chemical reaction, is flowed through the tube cores.

The reactor is fabricated by dispersing solid sorbents, solid precursors to liquid solvents, or microencapsulated liquid solvents within a 3D-printable, gas-permeable polymer, and then printing or otherwise shaping the mixture into reactor structures. The composite material may contain other reactive components, such as a pH-indicator dye to give visual indication of CO₂ absorption. Various structures can be printed that have desired properties for gas flow, fit into desired reactor housings, or provide other benefits. Features at the centimeter or larger scale can be incorporated into the printed structure, such as a hierarchy of gas channel sizes. The inventors have shown that the resulting structures absorb CO₂ much faster than comparable films of the liquid solvent, indicating much smaller and more efficient reactors are possible using the printed composite concept than with conventional technology. The CO₂ absorption rates are higher for smaller filament sizes (which have higher surface area-to-volume ratios) and they approach or exceed the absorption rates for the microencapsulated solvents we have previously developed. Though the inventors' research focus is on CO₂ capture and early demonstrations use sodium carbonate. This invention is widely applicable to other solvents, other reactions, and other applications involving gas purification or heterogeneous reactions involving a gas.

Several key innovations underpin the inventor's apparatus, systems, and methods:

1. The method of preparing a gas-absorbing composite material, i.e. by mixing solid precursors to liquid solvents with a gas-permeable polymer precursor, cross-linking the polymer, and then hydrating the solid particles to form

micro-scale solvent domains. This innovation is not specific to 3-D printing (it can be achieved by extrusion or other conventional manufacturing techniques), but it is specific to solvents that can be dispersed as solid powders and then re-hydrated, solid sorbents, or microencapsulated solvents.

2. The concept of preparing and 3-D printing a gas-absorptive ink into a monolith reactor with beneficial gas-flow features. This innovation is not specific to liquid solvents and can be applied to absorptive or reactive powders that remain solid, such as metal-organic frameworks.

3. The formation of a composite material as in concepts 1 or 2 into hollow-tube structures, where a second liquid is flowed through the center of the tubes for the purpose of heat exchange. The tubes can be printed with a Direct Ink Write 3-D printing system equipped with core-shell nozzles for printing hollow or liquid-filled tubes.

Referring now to the drawings, and in particular to FIG. 1, an embodiment of the inventor's apparatus, systems, and methods is shown. This embodiment is designated generally by the reference numeral 100. The inventor's apparatus, systems, and methods 100 provide a high-surface-area, hierarchically-structured, reactive composite material for energy-efficient gas purification and the techniques to fabricate this material. The hierarchy consists of, on the one hand, micron-scale domains of solvent or sorbent particles embedded within a gas-permeable polymer to form the composite material and, on the other hand, the submillimeter-scale structures that the composite materials form to create a monolith reactor. The reactor has channels for optimal gas flow and features sized for fast reaction with the gas. The structure optionally consists of hollow tubes where a second solvent, for heat exchange or additional chemical reaction, is flowed through the tube cores.

A flow chart illustrates the inventor's apparatus, systems, and methods 100. The flow chart illustrates a number of individual steps that encompass the inventor's apparatus, systems, and methods 100.

Step 102 comprises selecting the sorbents. For example, the sorbents can be mixing solid precursors to liquid solvents with a gas-permeable polymer precursor, cross-linking the polymer, and then hydrating the solid particles to form micro-scale solvent domains.

Step 104 comprises dispersing the sorbent material within a 3D printable gas permeable polymer. The inventor's apparatus, systems, and methods are not specific to liquid solvents and can be applied to absorptive or reactive powders that remain solid, such as metal-organic frameworks.

Step 106 comprises 3-D printing and shaping the sorbent material within a 3D printable gas permeable polymer into a monolith reactor with beneficial gas-flow features. The inventor's apparatus, systems, and methods are limited to specific to 3-D printing (it can be achieved by extrusion or other conventional manufacturing techniques), but it is specific to solvents that can be dispersed as solid powders and then re-hydrated, solid sorbents, or microencapsulated solvents.

Step 108 comprises curing the monolith reactor by thermal treatment.

Referring now to FIG. 2, one embodiment of the inventors' ink that can be used to create a monolith reactor is shown. This embodiment of the inventors' ink is designated by the reference numeral 200. The ink 200 includes sodium carbonate particles 202 in uncured silicone oil 204. The sodium carbonate particles 202 were produced by milling solid sodium carbonate to a fine powder of ~10-micron average diameter. The sodium carbonate particles 202 were then mixed with an uncured silicone oil 204, for example

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(e.g. Dow Corning® SE1700) via planetary mixing. The resulting ink **200** was supplied to a Direct Ink Write device for 3-D printing into desired structures. The completed structure was then cured. The sodium carbonate particles **202** can be considered solid particles for some operations and the sodium carbonate particles **202** can become liquid units when the sodium carbonate is exposed to moisture.

Referring now to FIG. 3, an embodiment of the inventor's 3D printing and otherwise manufacturing of a monolithic reactor is illustrated. As illustrated in FIG. 3, extruded material **304** is deposited on a surface **310** to be printed by print head **302**. For example, the extruded material **304** can be composed of the ink illustrated in FIG. 2.

The print head **302** extrudes the material **304** onto the surface **310**. Movement of the print head **302** is controlled by computer controller **308** which provides freedom of movement along all axes as indicated by the arrows **309**. The instructions for creating the reactor **312** is fed to the computer controller **308**. The computer controller **308** uses the instructions to move the print head **302** through a series of moments along the surface **310** forming the reactor **312**.

Referring now to FIG. 4 an embodiment of the inventor's monolith reactor is illustrated. This embodiment is designated generally by the reference numeral **400**. The reactor **400** has channels **402** for optimal gas flow and features sized for fast reaction with the gas. The structure can also be optionally constructed to consist of hollow tubes where a second solvent, for heat exchange or additional chemical reaction, is flowed through the tube cores.

As illustrated in FIG. 4, individual channels **402** are made by the shape of the structure **404**. The structure is made of micron-scale domains of solvent and/or sorbent particles embedded within a gas-permeable polymer to form the monolith reactor. The structure **404** includes gas-permeable particles **406** and a solvent matrix **408**. This an embodiment of the inventor's monolith reactor is a lattice like structure. The physical dimensions can be varied by length, width, and height. A smaller filament size yields a higher surface to volume ratio and faster CO₂ absorption rates.

The inventors' ink that is used to create a monolith reactor includes solid or liquid reactive filler and a gas-permeable polymer, such that the reactive filler forms micron-scale domains in the polymer matrix. The solid particle reactive filler is illustrated in FIGS. 2 and 4.

Referring now to FIG. 5, one embodiment of the inventors' liquid reactive filler is illustrated. The inventors' liquid reactive filler utilizes individual spheres that are capsules with liquid stripping solvents for absorbing CO₂ encapsulated within the capsules. One of the capsules **500** is shown in FIG. 5. The capsule **500** has a polymer surface layer **504** that is permeable to carbon dioxide. A liquid stripping solvent **502** for absorbing CO₂ is encapsulated within the polymer surface layer **504** of the spheres **500**.

Referring now to FIG. 6, a system for making the spheres is illustrated. The system for making the spheres is designated generally by the reference numeral **600**. The individual spheres have a polymer surface layer that is permeable to carbon dioxide and have liquid stripping solvents for absorbing carbon dioxide encapsulated within the polymer surface layer and inside the spheres.

One specific example of a system for making the spheres is shown in FIG. 6. The schematically illustrated method **600** is composed of the following items. An injection tube **602** with a ID (um) and OD 1000 (um), a collection tube **604** with an ID of 500 (um) and OD 1000 (um) and an outer tube **606** of square cross section with ID of 1000 (um) and ID of 1200 (um).

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In operation, the inner fluid **610** (Monoethanolamine/H₂O) with a viscosity of 10-50 (cP) and a flow rate of 200-800 (uLh⁻¹) flows in the injection tube **602** in the direction indicated by the inner fluid arrow. As this fluid proceeds it passes thru a droplet forming nozzle **624**. The formed droplet **626** is released from the nozzle **602** and becomes encased in the middle fluid **612** (Norland Optical Adhesive Pre-polymer) with a viscosity of 10-50 (cP) and flow rate of 200-800 (uLh⁻¹), the middle fluid **612** is flowing in the direction indicated by arrow **616**. The inner fluid droplet **626** becomes encased in the middle fluid **612** forming encapsulated microcapsules **618** that have a CO₂ capturing solvent core with a thin CO₂ permeable outer shell. The outer fluid (Polyvinyl Alcohol Stabilizer) with a viscosity of 10-50 (cP) and a flow rate of 200-800 (uLh⁻¹) flowing in the outer tube **606** in the direction indicated by arrow **622**. This outer fluid **620** carries the fabricated microcapsules **618** into the collection tube **604**. There is a boundary layer **614** that prevents the middle fluid **612** and outer fluid **620** from mixing as they have a large difference in both their viscosity and flow rates. The above described method will produce Microcapsules of a controlled size with an inner fluid (solvent/catalyst) enclosed in a CO₂ permeable polymer shell. The fabricated microcapsules **618** move out of the system as indicated by arrow **628**.

Referring now to FIG. 7, another embodiment of the inventors' ink that can be used to create a monolith reactor is shown. This embodiment of the inventors' ink is designated by the reference numeral **700**. The ink **700** includes spherical capsules **702** in a gas-permeable polymer **704**. The spherical capsules **700** have a polymer surface layer that is permeable to carbon dioxide and a liquid stripping solvent for absorbing CO₂ encapsulated within the polymer surface layer. The resulting ink **700** can be supplied to a Direct Ink Write device for 3-D printing into desired structures.

Referring now to FIG. 8, another embodiment of the inventors' 3-D printing for manufacturing a monolithic reactor is illustrated. As illustrated in FIG. 8, extruded material **804** is deposited on a surface **810** to be printed by print head **802**. For example, the extruded material **804** can be composed of the ink illustrated in FIG. 7.

The print head **802** extrudes the material **804** onto the surface **810**. Movement of the print head **802** is controlled by computer controller **808** which provides freedom of movement along all axes as indicated by the arrows **809**. The instructions for creating the reactor **812** is fed to the computer controller **808**. The computer controller **808** uses the instructions to move the print head **802** through a series of moments along the surface **810** forming the reactor **812**.

Referring now to FIG. 9, an embodiment of a system utilizing the inventor's reactor for (1) removing the target gas from a fluid or mixture and (2) for regenerating the reactor to remove the captured target gas from the reactor and getting it ready for another cycle of removing target gas from a fluid or mixture. This embodiment is designated generally by the reference numeral **900**.

The system **900** uses a stack **904** of reactors in cylindrical reactor vessels **902a** and **902b** for removing the target gas from a fluid or mixture. An example of using the reactor for removing a target gas from a fluid or mixture include use in the capture of carbon dioxide from gas mixtures containing carbon dioxide (examples: fossil fuel plants, natural gas streams, air). Also, the system **900** can be used for removing and/or capturing other gases including nitrous oxides (NO_x), sulphates (SO_x), hydrogen sulfide, or other trace gases.

The target gas in a fluid or mixture **906** is directed to one of the reactor vessels **902a** or **902b**. This is accomplished by

selectively using the valves **908a** and **908b** to channel the target gas in a fluid or mixture to the selected reactor vessel. The target gas in a fluid or mixture passes through the selected reactor stack and the reactive material in the reactor removes the gas. The fluid or mixture exits the system through valve **914a** or **904b**.

The reactors in the stack **904** of reactors needs to be purged of the trapped target gas. This can be accomplished using the temperature control system **912a** and/or **912b** and the purge system **910a** and/or **910b**. Once the reactors in the stack **904** of reactors have been purged of the trapped target gas they are ready for reuse. This is accomplished by selectively using the valves **908a** and **908b** to channel another target gas in a fluid or mixture to the reactor vessel with the purged reactors.

As illustrated in FIG. **10**, another embodiment of the reactor is illustrated. This embodiment is designated generally by the reference numeral **1000**. The reactor **1000** includes individual channels **1002** formed by the shape of the structural members **1004**. The structure is made of micron-scale domains of solid or liquid sorbent units **1006** embedded within a gas-permeable polymer **1008** to form the monolith reactor **1000**. The physical dimensions can be varied by length, width, and height. A smaller filament size yields higher surfaced to volume ration and faster absorption rates. Temperature control tubes **1010** are in the structural members **1004**. The coolant tubes can be used for heat exchange and for recycling the reactors.

Referring now to FIG. **11**, an embodiment of a reactor stack is illustrated. This embodiment is designated generally by the reference numeral **1100**. A reactor vessel **1102** contains reactors in a reactor stack **1104**. Coolant tubes **1106** are in the individual reactors. The target gas in a fluid or mixture is directed through the reactor stack **1004** wherein the stripping solvent in the individual reactors removes the target gas.

Although the description above contains many details and specifics, these should not be construed as limiting the scope of the application but as merely providing illustrations of some of the presently preferred embodiments of the apparatus, systems, and methods. Other implementations, enhancements and variations can be made based on what is described and illustrated in this patent document. The features of the embodiments described herein may be combined in all possible combinations of methods, apparatus, modules, systems, and computer program products. Certain features that are described in this patent document in the context of separate embodiments can also be implemented in combination in a single embodiment. Conversely, various features that are described in the context of a single embodiment can also be implemented in multiple embodiments separately or in any suitable subcombination. Moreover, although features may be described above as acting in certain combinations and even initially claimed as such, one or more features from a claimed combination can in some cases be excised from the combination, and the claimed combination may be directed to a subcombination or variation of a subcombination. Similarly, while operations are depicted in the drawings in a particular order, this should not be understood as requiring that such operations be performed in the particular order shown or in sequential order, or that all illustrated operations be performed, to achieve desirable results. Moreover, the separation of various system components in the embodiments described above should not be understood as requiring such separation in all embodiments.

Therefore, it will be appreciated that the scope of the present application fully encompasses other embodiments which may become obvious to those skilled in the art. In the claims, reference to an element in the singular is not intended to mean "one and only one" unless explicitly so stated, but rather "one or more." All structural and functional equivalents to the elements of the above-described preferred embodiment that are known to those of ordinary skill in the art are expressly incorporated herein by reference and are intended to be encompassed by the present claims. Moreover, it is not necessary for a device to address each and every problem sought to be solved by the present apparatus, systems, and methods, for it to be encompassed by the present claims. Furthermore, no element or component in the present disclosure is intended to be dedicated to the public regardless of whether the element or component is explicitly recited in the claims. No claim element herein is to be construed under the provisions of 35 U.S.C. 112, sixth paragraph, unless the element is expressly recited using the phrase "means for."

While the apparatus, systems, and methods may be susceptible to various modifications and alternative forms, specific embodiments have been shown by way of example in the drawings and have been described in detail herein. However, it should be understood that the application is not intended to be limited to the particular forms disclosed. Rather, the application is to cover all modifications, equivalents, and alternatives falling within the spirit and scope of the application as defined by the following appended claims.

The invention claimed is:

1. A method of making a reactor for removing carbon dioxide from a gas containing the carbon dioxide, comprising the steps of:

- providing a sorbent material;
- providing a gas-permeable polymer;
- combining said sorbent material and said gas-permeable polymer to produce a 3-D ink;
- using a 3-D print head of a 3-D printer to extrude said 3-D ink onto a surface;
- controlling said 3-D print head to extrude said 3-D ink onto said surface to form a green form of the reactor;
- and
- curing said green form of the reactor to produce the reactor.

2. The method of making a reactor for removing carbon dioxide from a gas containing the carbon dioxide of claim **1** wherein said step of providing a sorbent material comprises providing particles of carbon dioxide stripping material.

3. The method of making a reactor for removing carbon dioxide from a gas containing the carbon dioxide of claim **1** wherein said step of providing a sorbent material comprises providing particles of carbon dioxide stripping material encapsulated in capsules.

4. The method of making a reactor for removing carbon dioxide from a gas containing the carbon dioxide of claim **1** wherein said step of providing a sorbent material comprises providing particles of carbon dioxide stripping material encapsulated in spherical capsules that have a gas-permeable polymer surface layer encapsulating said particles of carbon dioxide stripping material.

5. The method of making a reactor for removing carbon dioxide from a gas containing the carbon dioxide of claim **1** wherein said step of providing a sorbent material comprises providing liquid units of carbon dioxide stripping material.

6. The method of making a reactor for removing carbon dioxide from a gas containing the carbon dioxide of claim **5** wherein said step of providing liquid units of carbon dioxide

stripping material comprises providing solid units of carbon dioxide stripping material and hydrating said solid units of carbon dioxide stripping material to provide said liquid units of carbon dioxide stripping material.

7. The method of making a reactor for removing carbon dioxide from a gas containing the carbon dioxide of claim 1 wherein said step of providing a sorbent material comprises providing liquid droplets of carbon dioxide stripping material encapsulated in capsules. 5

8. The method of making a reactor for removing carbon dioxide from a gas containing the carbon dioxide of claim 1 wherein said step of providing a sorbent material comprises providing liquid droplets of carbon dioxide stripping material encapsulated in spherical capsules that have a gas-permeable polymer surface layer encapsulating said liquid droplets of carbon dioxide stripping material. 10 15

9. The method of making a reactor for removing carbon dioxide from a gas containing the carbon dioxide of claim 1 wherein said step of providing a gas-permeable polymer comprises providing uncured silicone. 20

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