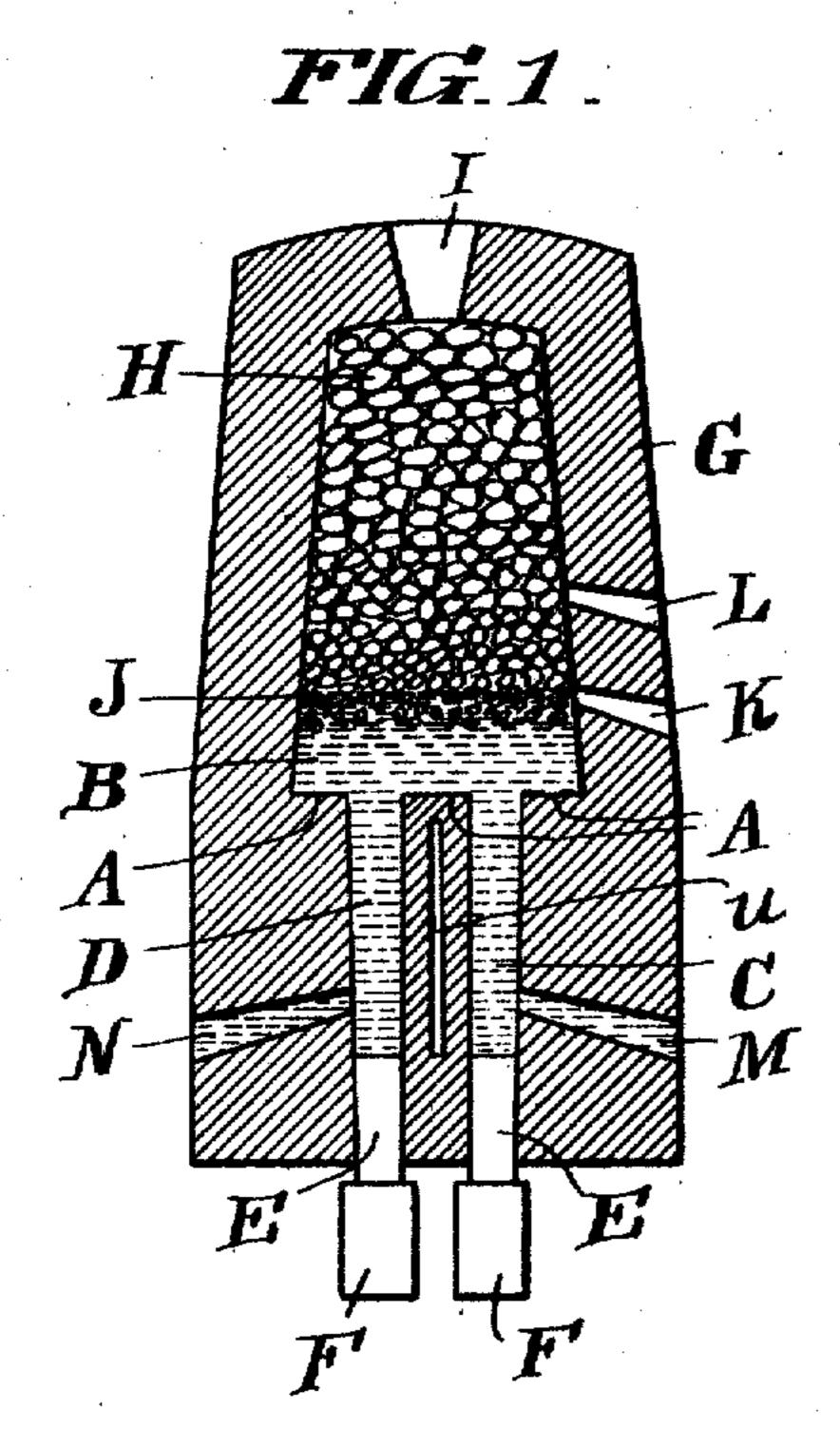
C. HERING. ELECTRIC FURNACE. APPLICATION FILED MAY 2, 1911.

999,720.

Patented Aug. 1, 1911.



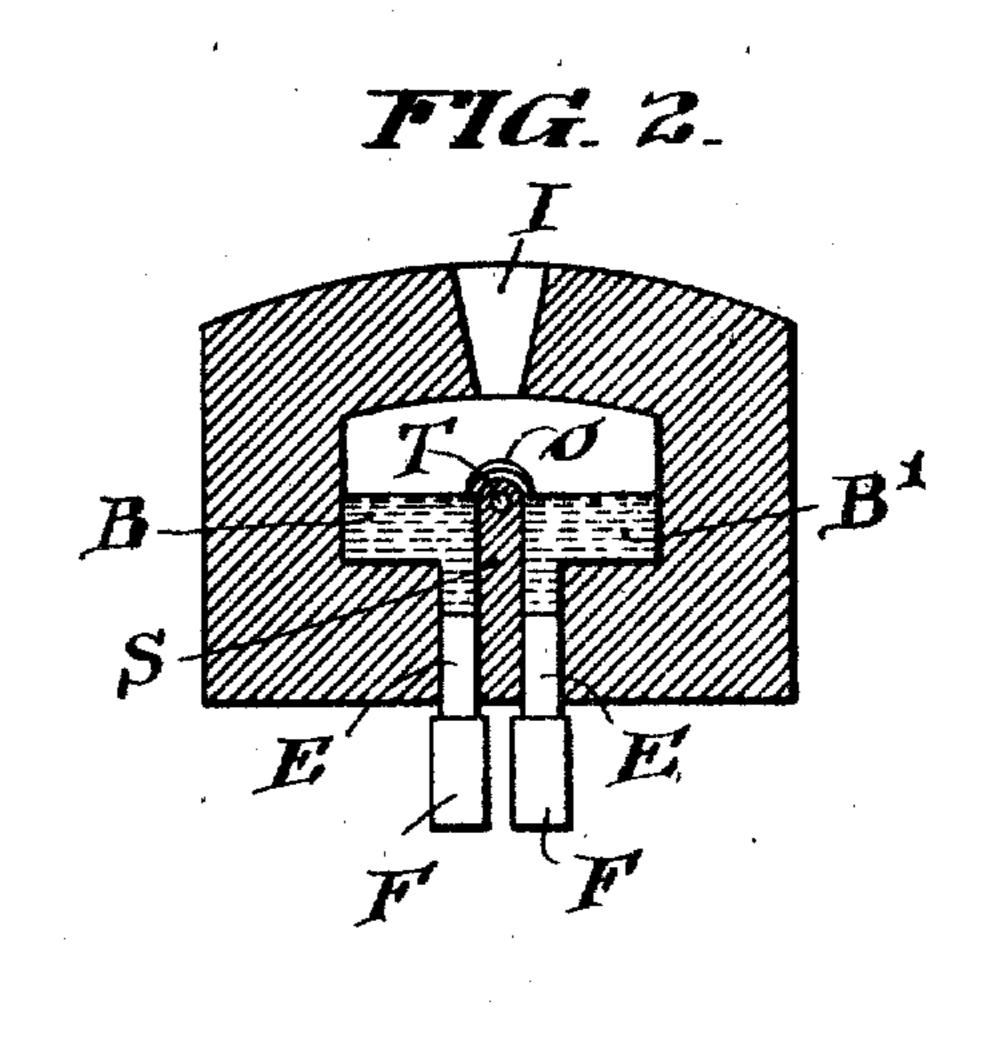


FIG.3

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Sumentor

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UNITED STATES PATENT OFFICE.

CARL HERING, OF PHILADELPHIA, PENNSYLVANIA.

ELECTRIC FURNACE.

999,720.

Specification of Letters Patent.

Patented Aug. 1, 1911.

Original application filed February 17, 1911, Serial No. 609,123. Divided and this application filed May 2, 1911. Serial No. 624,616.

To all whom it may concern:

Be it known that I, Carl Hering, a citizen of the United States, residing in the city of Philadelphia, county of Philadelphia, and State of Pennsylvania, have invented certain new and useful Improvements in Electric Furnaces, of which the following is a specification.

My invention relates to electric furnaces, and more particularly to electric furnaces employing electrodes of certain proportions.

My invention resides in an electric furnace comprising improved electrodes, for securing minimum electrode loss, in connection with electric furnace resistors, preferably liquid or molten; and the employment of my said improved electrodes in combination with an electric furnace in such arrangement that the power factor of the furnace is great, when alternating current is employed.

For an illustration of some forms of application of my invention, reference is to be had to the accompanying drawing, in

which:

Figure 1 is a vertical sectional view of an electric furnace in which the resistor is in the form of a column or columns of molten material in combination with my improved electrodes. Fig. 2 is a vertical sectional view of an electric arc furnace, involving also resistors in combination with my improved electrodes. Fig. 3 is a side elevational view of a metallic or conducting starter.

35. I have found that the energy loss in the electrodes of an electric furnace may be reduced greatly below what has heretofore been common practice. I have discovered the law of the electrode losses and from it I 40 have found that for a minimum amount of loss in the electrodes, such electrodes must be so proportioned that the C2R loss (heat generated by current in the resistance of the electrodes) shall be equal substantially or 45 approximately to twice the heat conduction loss of the electrodes. By heat conduction loss I mean the loss of heat from the interior of the furnace through the electrodes by heat conduction, when no current is flowing. 50 I have found that for any other relation be-

tween these two losses the combined loss be-

comes greater. As a result of proportioning the electrodes so that this relation shall hold, the losses of energy in and through the electrodes become very small as compared with 55 prior practice. The total loss in the electrodes will then be equal to the electrical resistance loss only (C2R loss) as there will then be no heat lost by conduction, because the temperature of the hot end of the elec- 60 trode will then be equal to that of the furnace, and the electrode will therefore be the equivalent of a perfect heat insulator, allowing no heat to pass through it from the furnace, although the electrode remains a very 65 good electrical conductor. No material is known which has these two qualities combined, namely, heat insulation and electrical conductivity; but by proportioning the electrodes according to the laws which I have 70 discovered, the practical equivalent of these two properties can nevertheless be realized. While the equations and proportions herein given are for electrode materials with ideal properties, such as zero co-efficients for both 75 electric and heat conductivities, these co-efficients of actual materials available -vary somewhat and, therefore, the results in practice are not exactly those indicated by the equations and proportions given, but are, for 80 all practical purposes, substantially those indicated by the equations and proportions given. I have found that for each electrode material, this minimum loss is a constant per ampere of current and per degree of 85 furnace temperature. Also that for a given material this minimum electrode loss is dependent upon the furnace temperature and the current, but not on the electrode dimensions, except that the ratio of the length to 90 the cross section of the electrode must be a certain quantity, as hereinafter pointed out. For any given electrode material, temperature, and current, this minimum electrode loss is fixed and cannot be further reduced 95 by any change in the dimensions of the electrode. For different materials I find that this minimum loss is proportional to the square root of the ratio of the thermal to the electrical conductivities of the electrodes. 100 From these laws I find that the minimum loss is generally least for the metals, and

that it is very considerably less than for the usual materials carbon and graphite.

It is a part of my invention, therefore, that metal electrodes are employed, prefer-5 ably of the same metal as is melted in the furnace, or of metal or material which does not contaminate the material or metal fused in the furnace, and thereby I can greatly reduce the electrode losses. To obtain this 10 advantage of a lower minimum loss for metal electrodes they must be proportioned | cal to use metal electrodes whenever possible. 75 so that the heat conduction loss, with no current flowing, will be equal to half the C2R loss approximately or substantially. 15 But my improvement is not limited to metal electrodes, for, by observing the novel proportions herein described, carbon and graphite electrodes, giving minimum losses for those materials, may be employed, and the 20 loss will be found to be much smaller than for the carbon and graphite electrodes as heretofore used. I have stated above that this minimum loss is independent of the actual dimensions of the electrodes, that is, 25 they may be large or small and yet have the same minimum loss. To get this minimum loss, however, I find that they must have a certain proportion between their length and their cross section. In order to obtain 30 the best economy of electrode material, therefore, the electrode should be made as short as possible, as the economy increases inversely as the square of the length. The trodes. cross section is then made to correspond 35 with this length in accordance with the laws which I have discovered, in order to obtain the minimum loss. From the laws which I have discovered it follows that this economy of material will be greatest, that is, for any 40 given length the cross section will be least, when the square root of the product of the electrical and the thermal conductivities is greatest. Hence, I find that as far as economy of material is concerned, those ma-45 terials are best in which this product is greatest. In any given material, therefore, the desirable qualities are, to a certain extent, opposed, with respect to economy of power as compared with economy of ma-50 terial. I have discovered that if K and k represent the electrical and thermal conductivities of the material, then the minimum power loss in them, in the form of the heat which leaves them at the outside terminals, 55 is least when K divided by k is greatest. On the other hand, the economy of material is best when the product of k and K is greatest. It is, therefore, the quotient and the product of the electrical and thermal conduc-60 tivities which determine their suitability for electrode materials, and not either of these alone. In comparing different materials quantitatively with each other, it is the square root of this quotient or this product |

which must be compared and not the quo- 65 tients and products directly. From the conductivities of different materials as far as they are known, I find, from the law which I have discovered, that the square root of these quotients and products are as a rule 70 greatest for the metals as distinguished from the usual electrode materials, carbon and graphite. The difference is great. Hence, I have found it much more economi-

When metal electrodes are used and proportioned in accordance with the laws which I have discovered, they will remain solid at the external ends although they will be at the temperature of fusion at the inside or 80 furnace ends. The reason is that when so proportioned there will be no heat conducted by the electrodes from the interior of the furnace, and all the heat generated in them by the current will be led off at the cool or 85 outside end just as fast as it is generated in it; hence their temperature will not increase and they will remain unfused except at their extreme inside ends. If continuously covered with fused metal at their inside or hot 90 ends they will not be consumed and if made of the same metal as that fused in the furnace, or of one which is nonmiscible with the fused material, they will not contaminate the latter. This state, as I have found, 95 is also the state of least total loss in the elec-

From the laws above stated I have deduced the following formula:

$$X = 2.894C \sqrt{kr}T$$

in which X is the total minimum loss in watts in or through the electrodes, 2.894 is a constant involving no physical properties, C is the current in amperes, k is the heat con- 105 ductivity in gram calories per second, cubic inch units, r the electrical resistivity in ohms, cubic inch units, and T the temperature difference between the inside and outside ends of the electrode in centigrade de- 110 grees. And I have deduced the following formula for determining the condition of proper proportions of the electrodes for minimum loss:

$$\frac{S}{L} = 0.3456 \,\mathrm{C} \,\sqrt{\frac{r}{k\mathrm{T}}}$$

in which S is the cross section of the electrodes in square inches, L their length in inches, C the current in amperes, r, k and T 120 being the same as above. The electrodes must have this proportion in order to obtain the minimum loss given in the first formula.

When the thermal conductivity k in the above formulæ is expressed in watts in place 125 of in calories per second, the numerical constants in these formulæ, namely, 2.894 and 0.3456, disappear, and a factor 2 accom-

panies the factor k, so that these formulæ take the form, respectively, as follows:

$$X = C\sqrt{2}krT$$

$$\frac{S}{L} = C \sqrt{\frac{r}{2kT}}.$$

This second formula gives the ratio of the section to the length of the electrodes and 10 therefore leaves a choice of either, but not of both. The length should be made as short as possible; it is usually determined by the general design and thickness of the furnace walls or other considerations. The 15 quantity of electrode material increases as the square of the length. It follows, therefore, that in accordance with my discoveries and invention, I may greatly reduce the size of, and therefore cheapen, the electrodes 20 heretofore used in the art and at the same time secure a minimum loss of energy in the electrodes, thus leaving greater amounts of energy for useful work within the furnace and, in consequence, increasing the efficiency 25 of the furnace.

If by the electrode efficiency is meant the ratio of the energy set free in the interior of the furnace, that is, between the hot ends of the two electrodes, divided by the total 80 energy between the two cold ends, then for a given minimum loss in the electrodes, this efficiency will evidently be higher the greater the drop of voltage between the hot ends, as compared with the drop of voltage in one 95 of the electrodes. By my invention the latter may be made very small, much smaller than heretofore, hence for a given current and voltage of a furnace there will be more useful heat generated in the furnace. But 40 to increase this efficiency still more the drop of voltage between the two hot ends should be made as great as possible. To do this with a liquid resistor may require this resistor to be made long and small in section, 45 hence I may in those cases prefer to use the arc as this has a relatively high drop of potential in a small space. Or still better I may use several arcs in series.

In Fig. 1 is shown an electric furnace in vertical section, having a hearth A; this hearth may take any suitable or desired form, as the heat producing resistor is practically independent of the proportions of this hearth or the amount of molten ma-55 terial in it. Upon the hearth A is a mass B, of molten iron or other material under treatment, the molten material extending also molten material in these columns making ther from the dividing member S. so electrical end-on contact with the furnace electrodes E, E, proportioned and constructed as hereinbefore described, which extend through the bottom or wall of the furnace, and may terminate outside in conduct-

ing enlargements, F, F, which may be 65 cooled, if desired, by a water jacket. The furnace extends upwardly in the form of a dome G, preferably enlarging toward the bottom, such dome being preferably filled with the charging material H, as iron ore 70 or other material, which may be introduced through the opening I at the top, which is thereby preheated.

The furnace may be started by a charge of molten material or by a casting of pref- 75 erably the same material as that to be treated, extending downwardly in the columns C and D into contact with the electrodes E, E, such casting being continuous and bridging the columns C and D at the top. When the 80 current is turned on, it flows from one electrode through one of the columns and out through the other column and other electrode, the casting, when such is employed for starting, becoming hotter and hotter un- 85 til finally melted.

M is a tap hole communicating with the column C for drawing off the finished material, and a similar tap hole N may be provided for communicating with the other 90 column D for the same purpose, if desired. Or the tap hole may communicate with the bottom of the hearth, if desired, or tap holes may be placed at both places.

In operation, a minimum amount of en- 95 ergy is lost in the electrodes E, E, and the heat is produced by the current in the columns C and D, the molten masses in these columns constituting the resistor.

In Fig. 2, I have shown an arc furnace 100 having the electrodes E, E; proportioned as herein described, which may be metallic, communicating with the separated baths B and B' of molten material, a dividing wall or member S being provided. The 105 arc may be started by a bridge piece m, such as shown in Fig. 3, made of the same metal as that in the baths B and B', by placing the same over the dividing member S; the member then melts and an arc 110 o is formed between the two baths B and B'. Or the arc may be started by granular conducting material extending over the member S into contact with the two baths, or the baths may be agitated to come momen- 115 tarily into contact with each other above the member S, or any other means may be employed. The dividing member S may be kept from fusing by a circulation of water or other cooling material through the open- 120 ing or tube T. Or the magnetic blow-out downwardly into the columns C and D, the | principle may be used to keep the arcs far-

In both the forms of furnaces herein shown, it will be noticed that the two ter- 125 minals or electrodes may be brought out. close together, because of their small size, due to my invention, thereby facilitating the

connections to the transformer and thus increasing the power factor, when alternating current is used, since the area inclosed by the conducting loop formed within the fur-

5 nace is greatly reduced.

The electrodes are not consumed and therefore do not contaminate the fused product and do not have to be advanced into the furnace. In consequence, the construction of to the furnace is greatly simplified and cheapened. Unless proportioned as I have shown, the losses through metal electrodes may become very large, due to their high heat conductivity.

By constructing a furnace electrode as hereinbefore described, the electrode section is not so large that heat conduction can occur therethrough from the furnace; nor is the section of the electrode so small that the 20 electrode is raised within the furnace wall to a temperature higher than the furnace

. This application is a division from my application Serial No. 309,123, which is a 25 division from my application Serial No. | 505,963, upon which latter was issued Let-

ters Patent of the United States No. 988,936.

I do not herein claim the electrodes themselves, proportioned, as herein described, for 30 minimum loss, but claim said electrodes herein only in combination with other features; the electrodes themselves, proportioned for minimum loss, are claimed in the above - mentioned application Serial No. 35 609,123.

What I claim is:

temperature.

1. In an electric furnace, molten conduct: 7. In an electric furnace, a container for to tion with said resistor for communicating and cross section of said electrode being so related that the resistor heating current passed through said electrode raises said electrode 15 to a temperature preventing heat conduction through said electrode from said molten material.

2. In an electric furnace, a resistor, a plurality of electrodes electrically communicatio ing with said resistor, and an insulating dividing wall between said electrodes, said electrodes proportioned for minimum electrode loss substantially as described, whereby the distance between said electrodes is 5 small and the area inclosed by the furnace circuit is small and the power factor great when alternating current is passed through said electrodes.

3. In an electric furnace, a resistor, and o an electrode for communicating heating current to said resistor, said electrode being so proportioned that the C'R heat developed in said electrode by the resistor heating current passed through said electrode raises

said electrode to a temperature preventir; 65 heat conduction through said electrode from said resistor.

4. In an electric furnace, a container for a mass of molten metal, a metal electrode extending through the furnace wall and dis- 70 posed in end on contact with said mass of molten metal, said metal electrode having a resistance such that the C2R heat developed therein by current transmitted therethrough to said mass shall be substantially equal to 75 twice the heat conduction loss when no cur-

rent flows.

5. In an electric furnace, an insulating . dividing wall, columns of conducting material on opposite sides of said wall, and elec- 80 trodes on opposite sides of said wall electrically communicating with said conducting columns, said electrodes proportioned for minimum electrode losses substantially as described, whereby the distance between 85 said electrodes is small and the area of the furnace circuit is small and the power factor great when alternating current is passed through said electrodes.

6. In an electric furnace, a dividing wall 90 within said furnace, containers for molten material on opposite sides of said dividing wall within the furnace wall, electrodes proportioned for minimum electrode losses substantially as herein described extending 95 through the furnace wall and disposed in end on contact with said separate masses on either side of said dividing wall, said masses being adapted to be connected by an arc extending over said dividing wall.

ing mater al serving as a furnace resistor, a mass of molten metal, a metal electrode and an electrode in electrical communica- extending through the furnace wall in electrical communication with said mass, the: heating current to said resistor, the length cross section of said electrode so related to 105 the length of said electrode that the C2R heat developed in said electrode by current transmitted therethrough to said mass raises said electrode to a temperature preventing heat conduction through said electrode from 110 said mass.

8. In an electric furnace, a molten resistor, an electrode in end-on contact therewith for communicating heating current to said resistor, said electrode having such resistance 115 that the C2R heat developed in said electrode by said resistor heating current raises the end of said electrode in contact with said resistor to a temperature substantially equal to the temperature of said resistor.

9. In an electric furnace, the combination with a container for material to be treated within the furnace wall, of an electrode extending through said wall and disposed in electrical communication with said material, 125 the cross section of said electrode being so small that the C2R heat developed in said electrode by current passed therethough to

said material raises said electrode to a temperature preventing heat conduction through said electrode from said material within said furnace.

10. In an electric furnace, the combination with a container for material to be treated within the furnace wall, of an electrode extending through said wall and disposed in electrical communication with said material, the cross section of said electrode being uniform and so small that the C2R heat developed in said electrode by current

passed therethrough to said material raises said electrode to a temperature preventing heat conduction outwardly through said 15 electrode.

In testimony whereof I have hereunto affixed my signature in the presence of the two subscribing witnesses.

CARL HERING.

Witnesses:

ELEANOR T. McCall, Anna E. Steinbock.