UNITED STATES PATENT OFFICE.

MARTIN MUGDAN AND WILLY HERRMANN, OF NUREMBERG, GERMANY, ASSIGNORS TO THE FIRM OF CONSORTIUM FÜR ELEKTROCHEMISCHE INDUSTRIE, GESELL-SCHAFT MIT BESCHRÄNKTER HAFTUNG, OF NUREMBERG, GERMANY.

PROCESS OF MAKING INDOXYL, &c.

960,671.

Specification of Letters Patent.

Patented June 7, 1910.

No Drawing.

Application filed January 25, 1910. Serial No. 540,011.

To all whom it may concern:

doctor of philosophy, chemist, and WILLY HERRMANN, doctor of philosophy, chemist, 5 subjects of the King of Prussia, German Emperor, residing at 54 Gugelstrasse, Nuremberg, Kingdom of Bavaria, German Empire, have jointly invented new and useful Improvements in Processes of Manufac-10 turing Indoxyl and Derivatives Thereof, of which the following is a specification.

This invention relates to a process for the production of indoxyl and derivatives thereof from aryl-glycin bodies. It is al-15 ready known that phenylglycin and its derivatives and homologues, on being heated with alkali hydroxids can be converted into indoxyl, or its derivatives, or homologues, and that indigo may be obtained from the 20 indoxyl thus obtained by oxidation for instance by blowing air into the aqueous solution of the melt. The yield of indigo in this case, however, is very bad. We have found that indoxyl and derivatives thereof 25 can be obtained from aryl-glycin bodies by heating these compounds with alloys of silicon and metals of the alkaline earths in the presence of anhydrous caustic alkalies. The process yields good results and is easily 30 carried out, as the alloys of silicon and metals of the alkaline earths are readily and cheaply obtained.

The following are examples showing how the invention is performed but the invention

35 is not limited to these.

Example I: Into an anhydrous melt of 280 parts by weight of caustic potash and of 200 parts by weight of caustic soda, are slowly introduced at about 250° centigrade, · 40 50 parts by weight of pulverized calciumsilicon (containing about 30 per cent. calcium). The calcium-silicon alloy dissolves while a brisk evolution of hydrogen takes place. While the melt is kept well stirred, 45 50 parts by weight of finely pulverized phenylglycin potassium are introduced at a temperature of 220° centigrade. The tem-

water the amount of the dye formed no Be it known that we, Martin Mugdan, | longer increases. The melt is dissolved in 10000 parts by weight of water at 100° centigrade, and the solution thus obtained which contains the indoxyl, is filtered off 55 from any insoluble substances produced. If it be desired to obtain the indoxyl in the solid state, the solution is evaporated to dryness in the presence of an inert gas such as hydrogen. As the indoxyl is generally 60 prepared for the purpose of obtaining indigo, its isolation may be omitted and the indigo prepared directly by introducing air into the solution. The indigo dye thus obtained is of great purity. 65

In carrying out the process the quantitative proportions and temperatures may be varied. It has been found advantageous to introduce the alloy of silicon and alkaline earth metal intimately mixed with the aryl- 70 glycin body into the melt of caustic alkalies.

Example II: Into an anhydrous melt of 300 parts by weight of caustic soda and 500 parts by weight of caustic potash is slowly introduced while stirring, at a temperature 75 of about 220° centigrade, an intimate mixture of 200 parts by weight of phenylglycinpotassium and 130 parts of calcium-silicon (containing about 30 per cent. of calcium). The temperature is then raised to 250° cen- 80 tigrade and the process carried out as described in Example I.

In the above examples the calcium-silicon can be replaced by alloys of silicon with other metals of the alkaline earths, for in- 85 stance barium-silicon may be made.

Instead of phenylglycin-potassium other compounds of phenylglycin can be used, for instance phenylglycin-sodium, phenylglycinamid or phenylglycin-anilid.

If it be desired to produce homologues of indoxyl homologues of phenylglyein are used. From tolyl-glycin-potassium

CH₃.C₆H₄.NH.CH₂COOK

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methyl indoxyl is obtained.

We claim:

perature is then raised to about 250° centi- | 1. A process for producing indoxyl and grade and the reaction is interrupted when | derivatives thereof by heating an aryl-glycin blowing air through a sample dissolved in | body with an alloy of silicon and a metal of 100

the alkaline earths in the presence of anhydrous caustic alkalies.

2. A process for producing indoxyl and derivatives thereof by introducing a mixture of an alloy of silicon and a metal of the alkaline earths and of an aryl-glycin body into melted anhydrous caustic alkalies.

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In testimony whereof we affix our signatures in presence of two witnesses.

MARTIN MUGDAN.

WILLY HERRMANN.

Witnesses:
ERICK Floefer,
ADAM KISCHEA.