UNITED STATES PATENT OFFICE.

PAUL WINAND, OF COLOGNE, GERMANY.

EXPLOSIVE MIXTURE FOR COMBUSTION-ENGINES.

No. 914,624

Specification of Letters Patent.

Patented March 9, 1909.

Application filed October 31, 1906. Serial No. 341,510.

To all whom it may concern:

Be it known that I, Paul Winand, engineer, a subject of the King of Belgium, re-·siding at 1 Sudermannstrasse, Cologne-on-5 the-Rhine, Germany, have invented certain new and useful Improvements in Explosive Mixtures for Combustion-Engines; and I do hereby declare the following to be a full, clear, and exact description of the invention, 10 such as will enable others skilled in the art to which it appertains to make and use the same.

For the operation of combustion engines for the propulsion of submarine boats and 15 torpedoes and the like wherein it is impracticable to supply the combustion chamber with air drawn from the external atmosphere, the oxygen necessary for such combustion must be carried on board in storage, 20 either in the form of compressed oxygen or in the form of suitable substances from which the oxygen is obtainable. Where the oxygen is stored in a compressed state under pressure, the weight of the stored 25 oxygen is but 20% of the total weight of itself and its containing receptacle. Where chemicals are stored for the production of the oxygen, as for instance, the perchlorates, the percentage, by weight, of oxygen to the 30 total weight of the chemicals is greater, but it is necessary to employ apparatus of suitable weight for the production of the oxygen from the chemicals and the general operation is not only materially complicated and un-35 certain, but the yield of free oxygen never reaches, in practice, the theoretical value.

In contrast to what has been heretofore proposed, it is a distinguishing characteristic of my invention, as hereinafter de-40 scribed, that the oxygen bearing substance or compound employed by me can be used not only in ordinary temperatures and under low vapor pressure, but also that being already in a liquid state, it can be 45 introduced into the engine without preliminary chemical transformation into said liquid condition. So also, my invention presents the advantage that the containing. 50 or compound, being subjected to no substantial internal vapor pressure, may be of thin wall structure and consequently of but moderate weight, which is not the case with receptacles containing compressed oxygen. 55 Furthermore, in contrast to the difficulties experienced in generating and making avail-

able in the engine oxygen from the perchlorates and the like, my invention offers the great simplification that the oxygenbearer is a fluid, requiring no oxygen gen- so erator for its employment, but capable of being pumped directly into the engine.

In carrying out my invention I employ as the oxygen-bearer such nitrated carbon compounds as contain a large proportion of 65 carbon with respect to the hydrogen present therein and at the same time a smaller proportion of oxygen, as, for instance, the nitrobenzols, C₆H₅NO₂, C₆H₄N₂O₄, and C₆H₃ N₃O₆; or such nitrated carbon compounds as 70 contain a large proportion of nitrogen in comparison to the carbon present, as, for instance, tetranitromethane CN₄O₈. These nitrated carbon compounds, moreover, have the great advantage that they are not explo- 75 sive, which is not the case with those substances which contain substantially enough oxygen to effect the complete combustion of the carbon and hydrogen present in them. In fact it is clear that these latter substances 83 if ignited at any point are immediately consumed throughout their entire mass without the access of air.

The oxygen contained in the NO₂ groups of the nitro compounds is as available for 85 combustion as free oxygen and it can take the place of a like weight of the latter. If the substance (as in the case for the nitrobenzols) contains less oxygen then is necessary for the complete combustion of the 90 carbon and hydrogen contained therein there must be employed, in order to obtain complete combustion, a supplemental quantity of additional oxygen. This supplemental oxygen is, however, less than is necessary for 95 the production of the same heating effect with an ordinary fuel. The nitro-benzols and the like may, therefore, be spoken of as oxygen-bearing fuels. By their employment from 7 to 20% of the otherwise neces- 10 sary oxygen is saved. Tetranitro-methane, CN₄O₈, on the other hand, is a substance which contains an excess of oxygen masmuch as of the eight atoms of oxygen, only receptacles for the oxygen-bearing substance | two are necessary for the complete combus- 10 tion of the one atom of carbon to carbonic dioxid. Three-fourths of the oxygen is therefore in excess and may be employed for the combustion of supplemental fuel introduced into the combustion chamber. Tet-11 ranito-methane may, therefore, be designated as not only an oxidizing agent or oxy-

gen-bearer, but also as a fuel bearer. By its use the employment of supplemental oxygen for the combustion of its own carbon and of additional fuel may be entirely dispensed with. The ideal mixture to be supplied, in accordance with my invention, to the combustion chamber of the combustion engine for ignition therein would consist of a nitrobenzol and tetranitro-methane, both of which are individually non-explosive and which would mutually contribute to produce the necessary explosive mixture in the working cylinder, the one supplying the excess of fuel to combine with the excess of oxygen of the other. The usual diluent for maintaining the required limitation of temperature, would preferably be supplied either by the products of combustion from the engine or from the external sea water, so that it would not be necessary to carry upon the submarine any special diluent or receptacle therefor.

Having thus described my invention, what I claim and desire to secure by Letters Pat-

ent of the United States, is:

1. An explosive mixture for combustion engines consisting of a suitable diluent, an oxidizing agent, and a non-explosive nitrated carbon compound containing hydrogen and wherein the percentage of oxygen is low and the percentage of carbon with respect to the hydrogen present is large; substantially as described.

2. An explosive mixture for combustion engines consisting of a suitable diluent, and an oxidizing agent, together with nitro-ben-

zol; substantially as described.

3. An explosive mixture for combustion

engines consisting of a suitable diluent, and two nitrated carbon compounds, which are 40 separately non-explosive, but which are explosive when mixed; substantially as described.

4. An explosive mixture for combustion engines consisting of a suitable diluent, a 45 non-explosive nitrated carbon compound containing a large percentage of nitrogen in proportion to the carbon present and an excess of oxygen over the quantity necessary to consume its carbon, and a combustible adapt-50 ed to be consumed by said excess of oxygen;

substantially as described.

5. An explosive mixture for combustion engines consisting of a suitable diluent, a non-explosive nitrated carbon compound 55 containing a large percentage of nitrogen in proportion to the carbon present and an excess of oxygen over the quantity necessary to consume its carbon, and a combustible adapted to be consumed by said excess of 60 carbon, said combustible consisting of a non-explosive nitrated carbon compound containing hydrogen and wherein a small percentage of oxygen is present; substantially as described.

6. An explosive mixture for combustion engines consisting of a suitable diluting agent, nitro-benzol, and tetranitro methane;

substantially as described.

In testimony whereof I have affixed my 70 signature, in presence of two witnesses.

PAUL WINAND.

Witnesses:

Bessie F. Dunlap, Louis Vandorn.