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## PROCESS OF TREATING VANADIUM ORES.

No. 828,850.

Specification of Letters Patent.

Patented Aug. 14, 1906.

Application filed December 1, 1905. Serial No. 289,737.

To all whom it may concern:

Be it known that we, JUSTIN H. HAYNES and WILBER D. ENGLE, citizens of the United States, residing in the city and county of 5 Denver and State of Colorado, have invented a certain new and useful Process of Treating Vanadium Ores; and we do declare the following to be a full, clear, and exact description of the invention, such as will enable 10 others skilled in the art to which it appertains to make and use the same.

Our invention relates to a process for treat-

ing ore containing vanadium. In carrying out our process the ore is first 15 suitably crushed, preferably so that it will pass through a twelve-mesh screen or finer, and is then boiled with a solution of alkaline carbonate, preferably sodium carbonate or potassium carbonate, until the vanadium zo contained in the ore is brought into solution. The strength of the sodium-carbonate solution and the length of the time necessary to boil is determined by the vanadium in the ore and will probably vary with every district. 25 We would state, however, that one hundred pounds of sodium carbonate for one ton of ore for every per cent. of vanadium will give good results. So far as the time during which the boiling treatment should be con-30 tinued, we may state that ordinarily this step of the process should be continued about one hour. It must be understood, however, that the proportion of the sodium carbonate or other alkaline carbonate employed to a 35 ton of ore may be varied, and we therefore do not intend to limit ourselves to any exact proportions. Neither do we wish it understood that we are limited as to any exact time or period for boiling the ore.

40 After the vanadium is dissolved it is precipitated from the solution, either with or without neutralization, as calcium vanadate by the addition of water-slaked lime.

The usual method for the treatment of this 45 ore has been to roast the ore with dry salt. This converts the vanadium into sodium vanadate soluble in water. The mass is then treated with water until all of the vanadium is dissolved. Our invention consists

in the treating of the ore, either with or with- 50 out a preliminary roasting, with a solution of alkaline carbonate, preferably sodium carbonate, at any temperature that is deemed advisable. This treatment brings the vanadium into solution preferably of the formula 55 Na<sub>2</sub>V<sub>2</sub>O<sub>7</sub>. All the other elements present in the ore are undissolved and are consequently left in the residue or tailings. This method in some cases works better by giving the ore a preliminary roasting. It may also be said 60 that the temperature at which the ore is to be treated varies somewhat with the district from which the ore comes.

Having thus described our invention, what

we claim is-

1. The herein-described process of treating ores containing vanadium, consisting in boiling the crushed ore with a solution of alkaline carbonate until the vanadium contained in the ore is brought into solution.

2. The herein-described process of treating ore containing vanadium, consisting in boiling the crushed ore with a solution of alkaline carbonate until the vanadium contained in the ore is brought into solution; and 75 finally separating the vanadium from the solution.

3. The herein-described process of separating vanadium from ore, consisting in first crushing the ore to be treated; second, boil- 80 ing the ore with a solution of alkaline carbonate until the metal to be extracted is placed in solution; third, drawing off the solution containing the dissolved metal; and finally precipitating the said metal as calcium vana- 85 date by the addition of water-slaked lime.

In testimony whereof we affix our signatures in presence of two witnesses.

JUSTIN H. HAYNES. WILBER D. ENGLE. Witnesses as to the signature of Justin H.

Haynes: N. E. STETMAN,

F. R. HYND. Witnesses as to the signature of Wilber D. Engle:

A. J. O'BRIEN, DENA NELSON.