

No. 744,223.

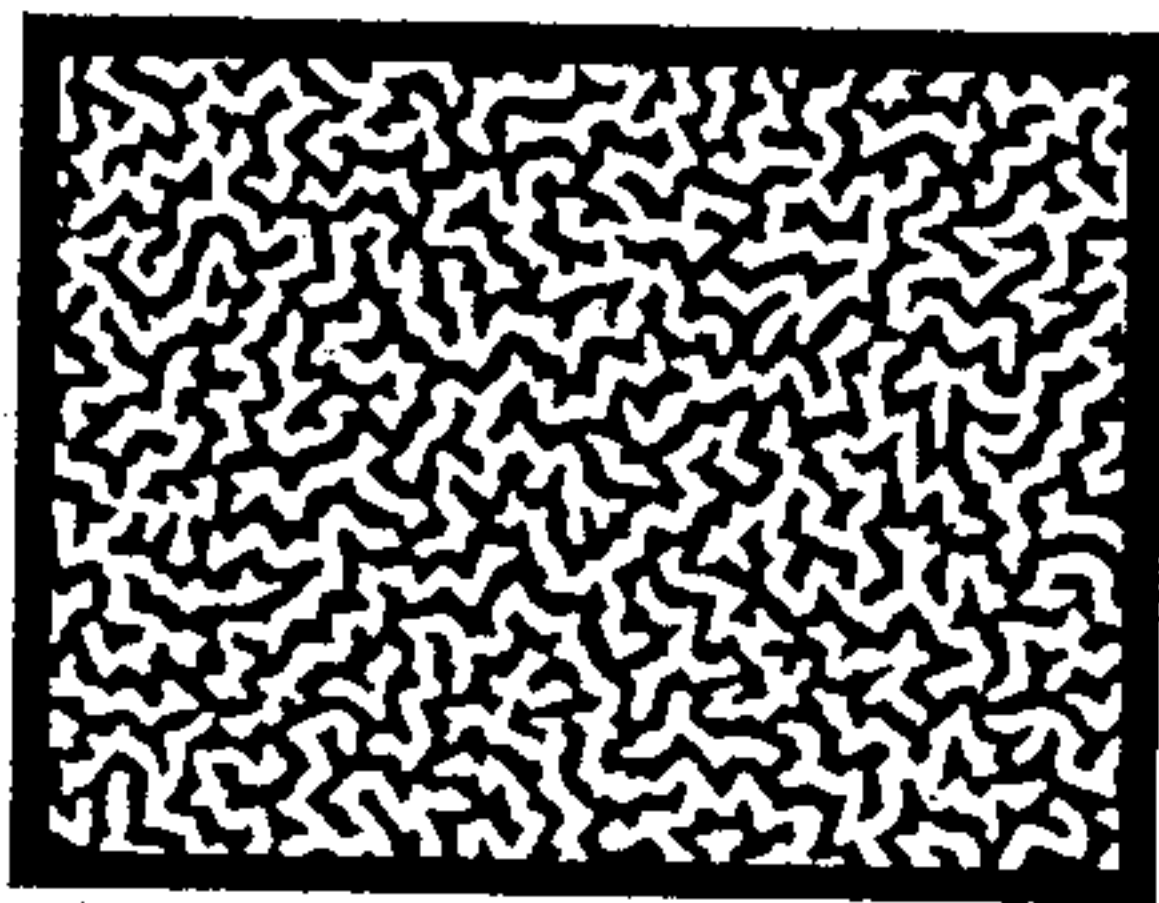
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E. ORTMANN.

PHOTOGRAPHIC REPRODUCTIVE PROCESS.

APPLICATION FILED JUNE 21, 1901.

NO MODEL.



WITNESSES

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PHOTOGRAPHIC REPRODUCTIVE PROCESS.

SPECIFICATION forming part of Letters Patent No. 744,223, dated November 17, 1903.

Application filed June 21, 1901. Serial No. 65,519. (No model.)

To all whom it may concern:

Be it known that I, ERNST ORTMANN, a citizen of the United States, residing at Detroit, county of Wayne, State of Michigan, have invented a certain new and useful Improvement in Photographic Reproductive Processes; and I declare the following to be a full, clear, and exact description of the invention, such as will enable others skilled in the art to which it pertains to make and use the same.

My invention relates to an improvement in photographic reproductive processes, particularly in photomechanical printing processes described in the following specification and more particularly pointed out in the claims.

Before entering into a detail description of my invention I will outline briefly the state of the art as heretofore practiced in order that a more perfect understanding may be had of the position occupied by my invention in the reproductive arts.

The so-called "heliotype" process (*Licht-druck*) had or still has its characteristic features in the peculiarity of reproducing by the several steps in its manipulation the unbroken tints or tones of the original picture, which, while only maintaining to a certain extent the color value of the original, made it possible for the printer by the aid of his press to reproduce copies with more or less rapidity. The editions that can be obtained by the employment of this process are very limited, due to the etching solution acting on the thin gelatin layer and the action of the printer's roller and press.

Closely relating to the heliotype process is one whereby ordinary photographic gelatinous dry-plates (gelatin-argentic dry-plates) are immersed in a bath containing a solution of a chromate salt, (bichromate of potash dissolved in water.) Said plates (or films) are rapidly dried, exposed in the usual way, the soluble chromate washed out, dried again, and finally prepared for printing with roller and printer's ink. From my experiments with this process I am convinced of the unreliability of the gelatin-argentic combination for treatment with roller and ink. In fact, it often happens that the very unstable gelatin-argentic film suffers destruction before a copy can be secured in the press. This process was

published (about 1890) by M. Bolagng, Dr. Schnaus, and others.

The necessity for more uniformity in the copies and greater speed in obtaining them than was possible by employing the gelatinous printing-surface produced the invention known as the "Meisenbach" or half-tone process, employing a line screen or grating whereby the tints of the original are broken into dots or dashes, which when printed upon a metallic plate and afterward etched produces a type printing-plate, but lacking the artistic beauty of the heliotype process.

In order to give more brilliancy or color value to the so-called "half-tone" engravings, many expedients have from time to time been resorted to—for instance, reëtching, tool-engraving, the employment of double negatives, and other manipulations—all with more or less success.

I shall now proceed to describe my invention, which is designed to overcome the objections and the limitations of the processes before referred to.

One of the principal objects of my invention is the creation of a uniform and sharply-defined grain or stipple by chemical action alone, which result is brought about as follows:

Any of the various gelatin-argentic dry-plates on the market will answer my purpose where delicate details are to be reproduced; but where a more open stipple is required I prefer to use my own prepared plates, inasmuch as this enables me to deposit on said plates (or flexible films) any desired thickness of the gelatin-argentic emulsion, because a thin coating in connection with the treatment hereinafter described will produce a fine grain or stipple, whereas a thicker coating by the same treatment will produce an open grain or stipple, according to the quantity of emulsion applied to the surface. Any of the various combinations of the gelatin-argentic emulsions employed in the manufacture of dry-plates will give good results. I prefer, however, either one of the following formulas:

1. (A) Three hundred grains bromide of ammonia, (weight,) seven hundred and fifty grains gelatin, (weight,) ten ounces distilled

water, (fluid;) (B) four hundred and fifty grains silver nitrate, (weight,) ten ounces distilled water, (fluid.)

2. (A) Three hundred and sixty grains bro-
 5 mide of potash, (weight,) ten grains iodid of
 potash, (weight,) four hundred grains gelatin,
 (weight,) eight ounces distilled water, (fluid;)
 (B) four hundred and fifty grains silver ni-
 10 trate, (weight,) eight ounces distilled water,
 (fluid.)

"A" is dissolved by dipping the container
 in hot water until contents is melted. "B"
 is dissolved at normal temperature and gradu-
 ally added to "A." The emulsion may be
 15 filtered and is then ready for use, or it may
 be poured into a suitable container, allowed
 to set, cut into slices and kept in a cool place
 until needed. The addition of a small por-
 tion (say from two to four per cent., by weight,
 20 to the dry gelatin used in the formulas above)
 of dextrine, gum-arabic, or albumin is of ad-
 vantage, as clearness and vigor are aided there-
 by. Instead of the bromides, the chlorid of
 ammonium may be used, or the bromide and
 25 chlorid combined in the proportions as given
 above under "A" in formulas 1 and 2. It is
 obvious that the proportion may be varied.
 Before coating the plates or films should be
 leveled in the usual manner.

- 30 One fluid ounce of the filtered emulsion on
 an eight-by-ten-inch surface will suffice for a
 medium fine stipple or grain. After coagu-
 lation of the coating the plates are set aside
 to dry. The dry-plates are then immersed
 35 in a bichromated bath. I prefer to use from
 two to five ounces (weight) of bichromate of
 potash (or bichromate of ammonium) to one
 hundred ounces (fluid) of water, in which
 bath the plates or films are left until the
 40 coating is saturated, which may take from
 five to twenty minutes, according to the thick-
 ness of the layer. More of the bichromate
 will increase contrast; less will produce soft-
 ness. The addition of a hygroscopical salt—
 45 such as chlorid of calcium, chlorid of sodi-
 um or ferrocyanid of potash—will acceler-
 ate the formation of the stipple or grain, as
 will be explained hereinafter. I use from
 one to two per cent., (by weight,) more or less,
 50 according to circumstances, to the volume
 of the chromate-bath. The plates having
 drained sufficiently are then dried, prefer-
 ably in a suitable drying-oven. The plates
 or films are then exposed to white light until
 55 all the details in the resulting positive have
 appeared. The next and most important
 step in the procedure is now the process of
 developing the exposed plates, which may
 take place after the exposure is completed,
 60 or the plates may be immersed in water and
 the soluble bichromate removed before de-
 velopment. The latter course I prefer. Any
 of the developers used by photographers for
 the development of gelatinous dry-plates may
 65 be employed as the action of the developer
 is the very same, and I therefore do not limit

myself to any particular one, but the follow-
 ing formula will be found satisfactory:

(Pyro stock solution:) Water, twelve
 ounces, (fluid;) oxalic acid, twenty grains, 70
 (weight;) bromide potash, thirty grains; pyro,
 one ounce, (weight,) and water to make six-
 teen fluid ounces.

(Soda stock solution:) Water, twelve
 ounces, (fluid;) soda sulfite, crystals, four 75
 ounces; carbonate of soda, two ounces; car-
 bonate of potash, one ounce; dilute two
 ounces of stock No. 2 with seven or eight
 ounces of water for cold weather and ten to
 fourteen ounces of water in summer. 80

To three or four ounces of dilute No. 2 add
 from one and one-half to two and one-half
 drams of No. 1. The more pyro the denser
 the stipple, and vice versa.

During the developing procedure the posi- 85
 tive will gradually turn to a negative. The
 positive parts of the picture caused through
 the action of the light have been changed from
 bichromate to chromic oxid, which will re-
 main indifferent to the action of the developer, 90
 whereas those parts not affected by light will
 absorb the developer, changing the bromide
 or silver (or other silver-salt combination em-
 ployed) into metallic silver, causing those
 parts to become opaque. After the develop- 95
 ment is completed, the extent or duration of
 which must be left to the judgment of the oper-
 ator, as different subjects require more or less
 development to secure satisfactory results, the
 developed plate is rinsed and placed in the 100
 fixing-bath, which is also identical with the
 fixing-bath of dry-plates in general. I prefer,
 however, to use the so-called "sour fixing-
 bath," which is composed of hypo-sulfite of
 soda, sixteen ounces, (weight;) water, forty- 105
 eight ounces, (fluid;) sulfite of soda, two
 ounces; water, six ounces; sulfuric acid, one
 dram, (fluid;) water, two ounces; chrome-
 alum, one ounce; water, eight ounces. Dis-
 solve and add in rotation as given. 110

After fixing the plates or films are washed
 sufficiently to eliminate the hypo. I thus pro-
 duce a duplicate negative in which the con-
 tinuous tones or gradation of the original
 negative (or positive) are discontinued or dis- 115
 solved into an opaque stipple or grain with
 transparent intersections. It is obvious that
 a copy from said stipple-negative will cause
 a stippled positive, which I produce on any
 sensitized surface, preferably on metal or 120
 lithographic stone for making photostipple
 engravings or photostipple lithographs.

I wish it to be distinctly understood that I
 do not confine myself to the production of
 stipple negatives or positives, but contem- 125
 plate the production of stipple-screens, which
 I use as so-called "mother" plates, which give
 excellent results in a variety of cases where
 stipple reproductions are desired in prefer-
 ence to lines or gratings. 130

The process for making stipple-screens is
 precisely the same as detailed heretofore with

the only variation that in the production of stipple-screens a general and even exposure over the entire surface of the plate or film is required instead of exposing behind a negative or positive. In regard to the formation of the stipple or grain the general theory is as follows: When a bichromated gelatinous film or layer is exposed to white light, the bichromate will be converted into chromic oxid, whereby the gelatin more or less loses its peculiarity of absorbing water relative to the duration of exposure or to the quantity of light received by the sensitive layer and the amount of chromic oxid thus produced. If the light is allowed to penetrate the entire film or layer, the formation of chromic oxid is complete, causing the gelatin to resist the water, whereas a short exposure will form chromic oxid on the surface only relative to the action of light, leaving the bichromated layer underneath in its original state. When said exposed film or plate is submitted to a water-bath, the unaffected gelatin will absorb water and being fixed on a rigid support seeks to force its way through the more or less resisting stratum of the chromic oxid, changing or separating the original smooth surface into minute divisions—in other words, into a stippled or grained surface. The addition of a hygroscopical salt, as pointed out before, will accelerate the absorption of water, increasing the energy of expansion, and thus causing the formation of a larger stipple or grain. It is obvious that the presence of silver salt is necessary for the development of the exposed plates or films.

What I claim is—

1. The within-described method of manufacturing photographic printing-plates consisting in coating the plate with chemicals adapted to produce an opaque stippled surface on the plate by the action of light and developer without the aid of a screen by first, submitting the plate to the action of light, and then to the developer, substantially as described.

2. The within-described method of producing stipple printing-plates, consisting of first preparing the plates with a coating of gelatino-argentic emulsion, then submitting the plates so coated to a bichromate-bath, then exposing the plates, then washing out the soluble bichromate, then developing the plate whereby the washed-out portion takes the developer and becomes opaque to form the stipple, substantially as described.

3. The within-described method of producing stipple printing-plates, consisting of first preparing the plate with a coating of gelatino-argentic emulsion, then submitting the plate so coated to a bichromate-bath, then drying

the plate, then exposing the plate, then washing out the soluble bichromate, then developing the plate whereby the washed-out portion takes the developer and becomes opaque to form the stipple, substantially as described.

4. The within-described method of producing stipple printing-plates, consisting of first preparing the plate with a coating of gelatino-argentic emulsion, then submitting the plate so coated to a bichromate-bath, then drying the plate; then exposing the plate, under a negative, then washing out the soluble bichromate, then developing the plate whereby the washed-out portion takes the developer and becomes opaque to form the stipple, substantially as described.

5. The within-described process of producing stipple printing-plates, consisting of preparing the plate with a coating of sensitized material, then submitting the plate to a bath in chemicals adapted to cause the emulsion to divide into two sets of minute divisions when exposed to light, one set impervious to water, the other set of divisions adapted to take the developer and develop into an opaque stipple; then washing out the soluble chemicals after first exposing the plate to the action of light, then developing the plate whereby the divisions impervious to water refuse the developer and the other set of divisions that take the developer develop into an opaque stipple, then fixing the plate, substantially as described.

6. The within-described method of producing stipple printing-plates, consisting of first preparing the plates with a coating of sensitized emulsion, then submitting the plates so coated to a bichromate-bath, then exposing the plates, then washing out the soluble bichromate, then developing the plates whereby the washed-out portion takes the developer and becomes opaque to form the stipple, substantially as described.

7. The within-described method of producing stipple printing-plates, consisting of first preparing the plates with a coating of sensitized emulsion, then submitting the plates so coated to a bath adapted to form an opaque stipple, then exposing the plates, then washing out the soluble chemicals, then developing the plate whereby the washed-out portion takes the developer and becomes opaque to form the stipple, then fixing the plate, substantially as described.

In testimony whereof I sign this specification in the presence of two witnesses.

ERNST ORTMANN.

Witnesses:

S. E. THOMAS,
C. H. FISK.