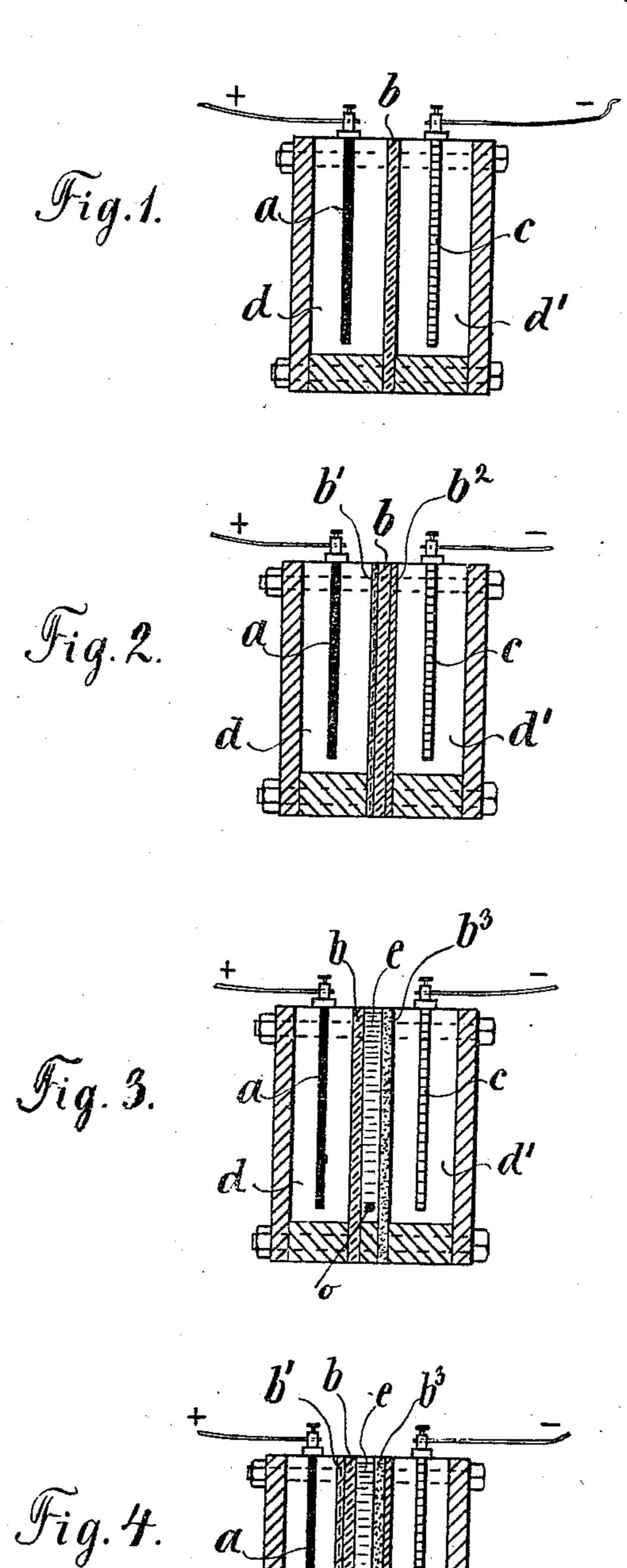
(No Model.)

## C. HOEPFNER. ELECTROLYTICAL APPARATUS.

No. 518,065.

Patented Apr. 10, 1894.



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## United States Patent Office.

CARL HOEPFNER, OF GIESSEN, GERMANY.

## ELECTROLYTICAL APPARATUS.

SPECIFICATION forming part of Letters Patent No. 518,065, dated April 10, 1894.

Application filed February 8, 1892. Serial No. 420,774. (No model.) Patented in Germany February 22, 1889, No. 58,133; in Norway September 1, 1890, No. 1,967; in England September 1, 1890, No. 13,735; in Italy September 30, 1890, XXIV, 28,112, LV, 196; in Spain November 11, 1890, No. 11,205, and in Austria-Hungary December 12, 1890, No. 39,156 and No. 66,246.

To all whom it may concern:

Be it known that I, CARL HOEPFNER, a subject of the King of Prussia, residing at Giessen, Germany, have invented certain new and 5 useful Improvements in and Relating to Electrolytical Apparatus and to the Diaphragms Employed Therewith, (for which Letters Patent have been obtained in Germany, No. 58,133, dated February 22, 1889; in Norwáy, to No. 1,967, dated September 1, 1890; in Austria-Hungary, No. 39,156 and No. 66,246, dated December 12, 1890; in Great Britain, No. 13,735, dated September 1, 1890; in Italy, Vol. XXIV, No. 28,112, and Vol. LV, No. 196, dated 15 September 30, 1890, and in Spain, Libro XI, fol. 516, No. 11,205, dated November 11, 1890;) and I do hereby declare the following to be a full, clear, and exact description of the invention, such as will enable others skilled in the 20 art to which it appertains to make and use the same.

My present invention relates to electrolytical apparatus, and more especially to the dia-

phragms employed therewith.

In electrolysis, and more especially in the electrolytical treatment of chlorides, haloid salts, haloid acids and the like, it is necessary that the cell diaphragms should be constructed of a material capable of resisting the action 30 of the electrolyte, or a component or components thereof, or that of the products of decomposition. Not only are the materials heretofore employed comparatively costly, but the diaphragms are more or less defective in that 35 they do not possess the necessary power of resistance to pressure in apparatus in which said diaphragms are subjected to the pressure of the electrolyte, while the diaphragms for apparatus designed for the electrolytical treat-40 ment of substances such as described, had to be constructed of materials capable of resisting the deleterious action of the electrolyte or its components or the products of decomposition. My invention has for its object the provis-

ion of means whereby diaphragms of sufficient strength to resist any pressure to which they may be subjected are obtained, and whereby said diaphragms are made proof against the 50 deleterious action of the electrolyte or its com-

ponents or the products of decomposition, and this I attain by the use of simple or compound diaphragms rendered proof against the deleterious influences referred to by nitrating the same or by nitrating the side thereof that faces 55 the anode cell.

The invention consists therefore in the construction and nitration of the diaphragms, as will now be fully described, reference being had to the accompanying drawings in which 60 a simple form of electrolytical apparatus is

illustrated by cross-sectional views.

Figure 1 shows a diaphragm composed of a material containing a nitrated organic substance. Fig. 2 shows a diaphragm that is me- 65 chanically braced or supported on opposite sides by a textile material, that side facing the anode being nitrated or containing nitrated organic substances. Fig. 3 shows a diaphragm in which the nitrated membrane or 70 that part that faces the anode is so arranged relatively to the diaphragm proper as to form a cell between the two for the reception of a liquid or semi-liquid that will more or less hinder diffusion between the anode and cath- 75 ode cells, and Fig. 4 shows a similar diaphragm in which the membranes are reinforced by additional membranes.

In carrying out my invention I construct the diaphragm, b, of an organic substance, 80 animal or vegetable, as for instance, paper, or parchment paper, felt, leather, textiles, veneer, or the like, which is nitrated by saturating the material in a solution of nitric acid, or equivalent solution, or by subjecting the 85 material to the action of a nitrating gas or gases. On the other hand, the nitration may be effected by combining an organic or inorganic material, such as wood or paper pulp or asbestus, with nitro-cellulose, and forming the 90 diaphragm from the compound by molding or pressure, as shown in Fig. 1; the same result may be attained by coating an organic or inorganic material with a nitro-cellulose or equivalent material to protect the same 95 against the action of the electrolyte.

In Fig. 2, the diaphragm, b, made of any of the materials hereinabove referred to, is reinforced on both sides by a textile material, b', and  $b^2$ , respectively, one of which, namely the

reinforcing sheet, b', facing the anode, a, only, is nitrated as described. The object of reinforcing the diaphragm as described, is to enable it to withstand much higher pressures than would be the case otherwise. On the other hand, to prevent diffusion as much as possible, I construct the diaphragm in the form of a cell for the reception of a semi-fluid or a solid substance, two or a greater number of diaphragms being employed in the construction and arranged to form a cell for the reception of the interposed material. The individual diaphragms may be constructed of a material and treated as above set forth.

In Fig. 3 the cell, e, is formed between two diaphragms, b, b³, the one, b, facing the anode, a, or forming one of the walls of the anode cell, d, is nitrated as set forth. In order to impart to these cell diaphragms the necessary strength to resist pressure when they are made of paper or paper parchment, I reinforce both by means of a strong textile fabric or several layers thereof, or with equivalent materials, as leather, gelatine, veneer, asbestus, felt, or the like applied to the outer face of the diaphragms, b, and b³, as shown at b', b², Fig. 4, such reinforcing material or at least the material, b', facing the anode, a, being nitrated as described.

In the electrolytical treatment of the chlorides of the alkalies, carbonic acid, for instance, may be introduced into the cell, e, between the diaphragms by means of a suitable pipe connected with said cell near the bottom thereof as shown at o, Figs. 3 and 4, or by means of a pipe projecting into the cell to

near the bottom thereof, as may be found most convenient for the purpose of preventing any deleterious action of the electrolyte in the cell, e, upon the diaphragms by the formation 40 of carbonates or bi-carbonates.

Of course it will be understood that the diaphragms herein referred to are supported in the electrolytical apparatus in suitable frames between which they are clamped, as is well 45 known.

Having thus described my invention, what I claim as new therein, and desire to secure by Letters Patent, is—

1. In an electrolytic apparatus having ad- 50 jacent cells of unlike name, a partition between said cells comprising a diaphragm constructed of a nitrated textile material, substantially as and for the purpose set forth.

2. In an electrolytic apparatus having ad- 55 jacent cells of unlike name, a partition between said cells comprising a diaphragm constructed of a nitrated textile material and one or more auxiliary diaphragms, said nitrated diaphragm facing the insoluble electrode, substantially as and for the purpose set forth.

3. An electrolytic apparatus having adjacent cells of unlike name, and a chambered partition dividing said cells having its wall facing the insoluble electrode constructed of 65 a nitrated organic substance as a nitrated textile material, substantially as and for the purpose set forth.

CARL HOEPFNER.

Witnesses:

GEORGE LOUBIER, RICHARD SCHMIDT.