

(No Model.)

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METHOD OF AND APPARATUS FOR PRODUCING LEAD CARBONATE.

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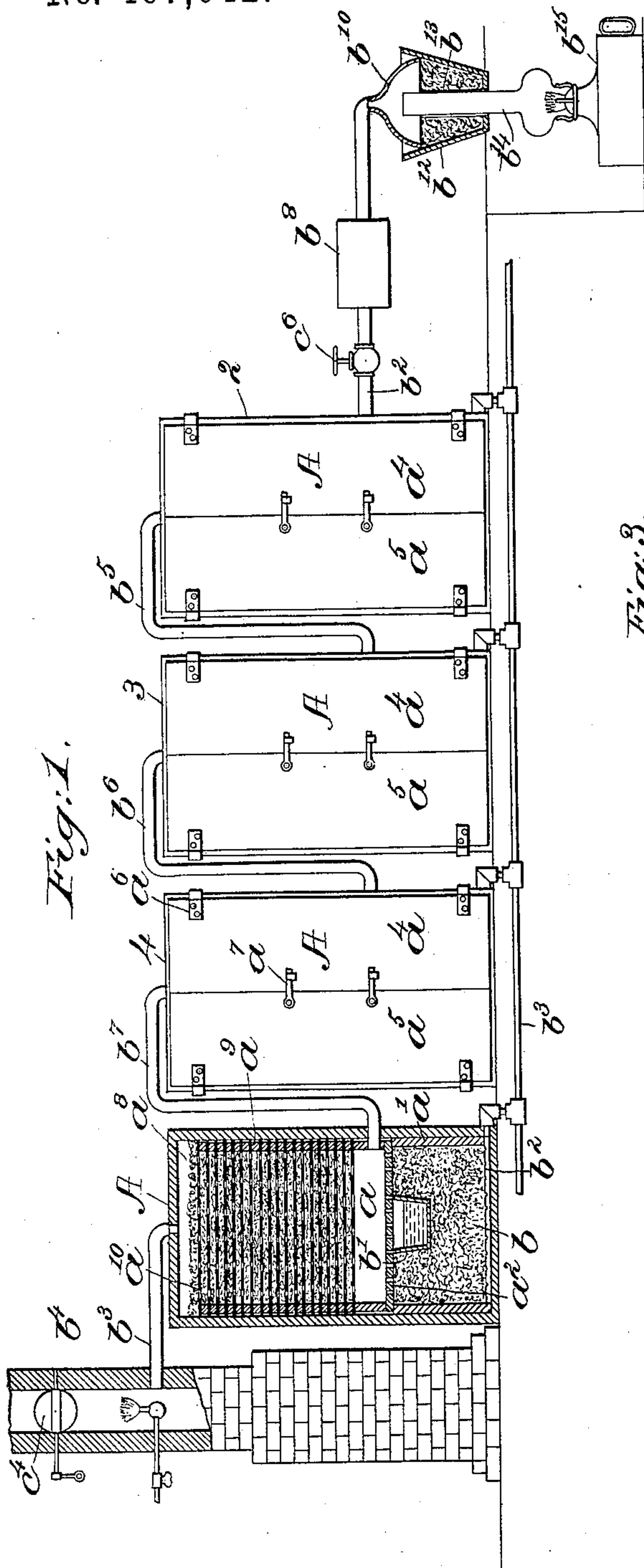
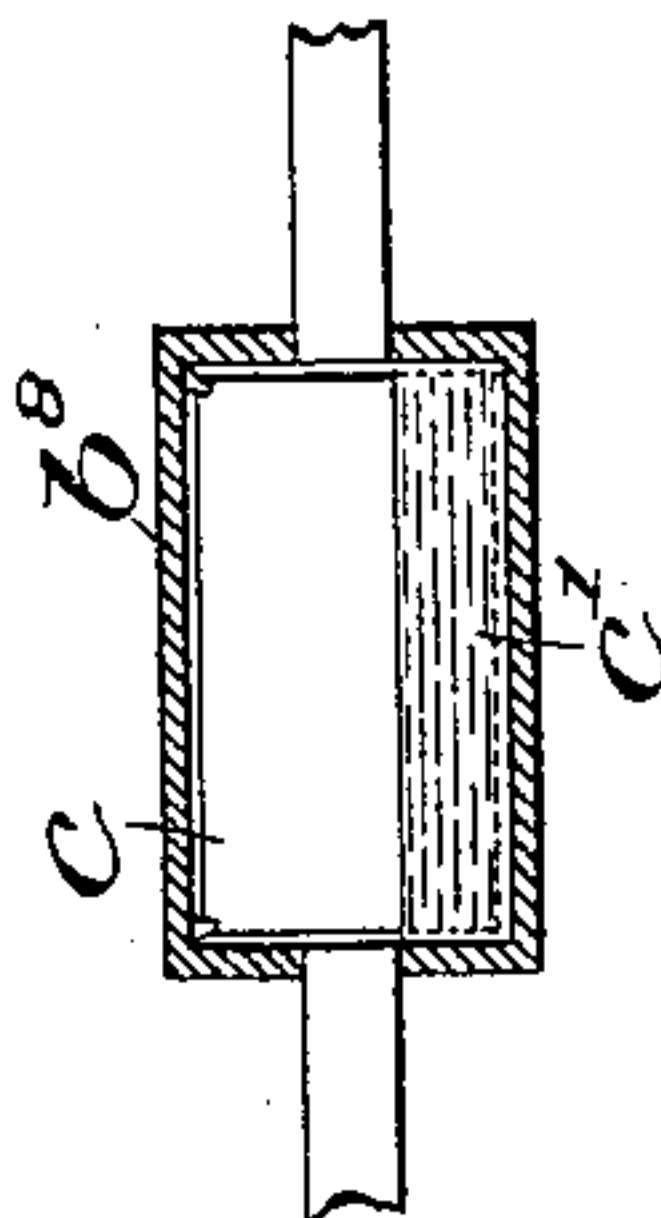
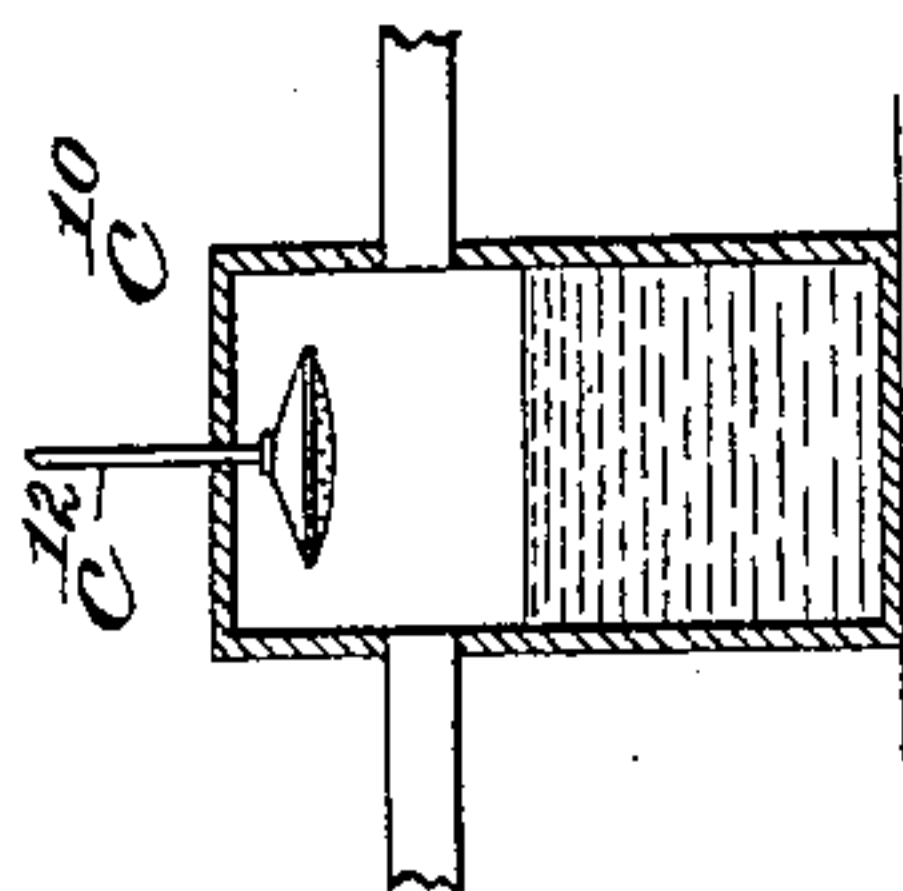


Fig. 1.

Fig. 3.

Fig. 2.



Witnesses.

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METHOD OF AND APPARATUS FOR PRODUCING LEAD CARBONATE.

SPECIFICATION forming part of Letters Patent No. 467,042, dated January 12, 1892.

Application filed March 31, 1891. Serial No. 387,105. (No model.)

To all whom it may concern:

Be it known that we, NORMAN K. MORRIS and JOHN W. BAILEY, both of Denver, county of Arapahoe, State of Colorado, have invented
5 an Improvement in Methods of and Apparatus for Producing Lead Carbonates, of which the following description, in connection with the accompanying drawings, is a specification, like letters on the drawings representing like
10 parts.

This invention relates to a novel method of and apparatus for the manufacture of metallic carbonates, especially lead carbonate or white lead.

15 Prior to our invention we are aware that lead carbonate or white lead has been made by means of carbonic-acid gas acting upon the lead in the presence of acetic-acid vapor.

Lead carbonate manufactured in a stack
20 by what is known as the "Dutch" method, in which the carbonic-acid gas is made from tan-bark, &c., possesses superior qualities; but the objection to this method is the length of time required to convert the metallic lead
25 into the form of the carbonate. Metallic lead has also been converted into the carbonate in a much quicker time than what is required by the Dutch method by means of carbonic-acid gas generated by the combustion of pe-
30 troleum. This method, although much quicker than the Dutch method, does not ordinarily produce a superior quality of lead carbonate owing to the drying out of the metallic lead.

35 This invention has for its object to produce a superior quality of lead carbonate in a substantially short time, whereby the advantages of the old Dutch method and of the petroleum method are combined.

40 In accordance with this invention, the lead to be converted into the carbonate is placed in a corroding-chamber in a finely-divided condition, preferably in the form of thin strips or fibers, and is therein acted upon by
45 a current of cooled hydrous carbonic-acid gas with an oxidizing agent, which for the best results consists of vapors of acetic acid, preferably commingled with the hydrous carbonic-acid gas before being brought into con-
50 tact with the lead, the said gas and vapor before entering the corroding-chamber being cooled to about 95° Fahrenheit, and the tem-

perature of the corroding-chamber is main-
tained at about 95° to 105° Fahrenheit, so that the danger of drying out the metallic lead or
55 stock is prevented, for we have ascertained by a series of long-continued experiments that if the metallic lead is subjected to a too high temperature spots are formed on the sur-
face of the lead which resist the action of the
60 carbonic-acid gas and acetic-acid vapor and cannot be subsequently corroded. These spots give rise to lumps of metallic lead in the carbonate and deteriorate the quality
65 of the carbonate produced. The formation of these spots is not, so far as our experi-
ments show, prevented by the presence of an excess of watery vapor generated by heating water or even acidulated water; but we are
70 led to believe by the results of our experi-
ments that water formed by the combustion of refined petroleum-oil or other unsaturated or saturated hydrocarbon—such, for instance,
as gasoline, kerosene, natural gas, naphtha,
75 alcohol, &c.—has a peculiar property itself or imparts to the carbonic-acid gas produced
by the combustion of the above-mentioned bodies, the property of more energetically act-
ing upon the metallic lead, for the lead car-
80 bonate produced in accordance with this in-
vention is of a beautiful white carbonate of creamy consistency and free from particles of metallic lead, so that subsequent treat-
ments, such as grinding, are entirely dis-
85 pensed with and the carbonate may be taken
from the corroding-chamber and used with-
out any preparatory treatment.

The corroding-chamber referred to will preferably be made in the form of a stack and will preferably contain in its lower por-
90 tion tan-bark, cow-dung, or like material now commonly employed in the Dutch method, and preferably a series of such stacks will be connected together for a circulation of gas and vapor from one to another, as will be de-
95 scribed, the circulation being effected by means of the draft in a chimney, thereby dispensing with a circulating-pump and cheapening the cost of the apparatus, the said draft being controlled by a suitable damper. 100

The bottom or lower chamber of each stack may and preferably will have located in it a steam-pipe, by which the tan-bark or like ma-
terial may be heated.

The particular features of our invention will be pointed out in the claim at the end of this specification.

Figure 1 represents in section and elevation an apparatus embodying this invention and with which the improved method may be carried into effect; Fig. 2, a sectional detail of the gas-cooling apparatus shown in Fig. 1, and Fig. 3 a vertical section of a modified form of cooling apparatus.

Referring to Fig. 1, four stacks or corroding-chambers A are shown and numbered 2 3 4 5, and the said stacks being alike we will specifically describe but one.

Each stack A, made of wood or other suitable material, is divided into an upper compartment a and a lower compartment a' , as herein shown, by a perforated wall or partition a^2 , resting, as shown, upon supports a^3 , which may be bars, rods, posts, or of any other suitable construction.

The stack A is provided with, preferably, side doors a^4 a^5 , hinged, as at a^6 , to the main frame of the stack and secured together by one or more latches a^7 or by any other suitable form of locking device. The compartment a of each stack contains within it the metallic lead to be converted into the carbonate, which metallic lead is in a finely-divided form, preferably in strips or fibers represented by a^8 , the said finely divided lead being supported upon a series of trays, preferably composed of a wooden frame a^9 and a body a^{10} of fibrous material, such as cords or strings, but which may be made of any material not effected by the corroding agents, and which will not discolor or otherwise injure the carbonate. The compartment a' of the stack is partially or wholly filled with tan-bark, cow-dung, or like material b , in which is placed a vessel b' , to contain acetic acid or water acidulated with acetic acid, and the tan-bark or like material is heated, as herein represented, by a steam-pipe b^2 , preferably provided with perforations and connected to a supply-pipe b^3 , which communicates with a boiler or other source of supply.

The compartment a of the first stack (marked 2) has an inlet-pipe b^2 , and the compartment a for the last stack has an outlet-pipe b^3 , communicating with a chimney b^4 , and the compartments a of the adjacent stacks are connected together by pipes b^5 b^6 b^7 .

The inlet-pipe b^2 for the stack marked 2 is connected to or communicates with the source of supply for carbonic-acid gas, it being herein shown as connected directly to a cooling apparatus or chamber b^8 , placed in the inlet-pipe b^2 , the latter being provided with a bell-shaped mouth b^{10} , projecting down within a vessel or chamber b^{12} , having a central pipe or tube b^{13} , up into and preferably beyond or above which is projected the chimney b^{14} of a lamp b^{15} . The vessel b^{12} contains water, or it may be dilute acetic acid, which is vaporized by the heat from the products of com-

bustion passing up through the chimney b^{14} , and the vapor thus formed commingles with the products of combustion and passes into the corroding-compartment of the stack A, (marked 2,) the said vapors and products of combustion being cooled on their passage through the cooling apparatus b^8 , the latter, when made, as shown in Figs. 1 and 2, containing suspended therein a number or series of strips or sheets c of flannel or other absorbent material which dip into a bath c' of water, or it may be dilute acetic acid. The lamp or burner b^{15} preferably contains refined petroleum or other hydrocarbon, or it may be alcohol or equivalent body, which is decomposed by combustion into a hydrous carbonic-acid gas—that is, carbonic-acid gas and a plurality of molecules of water, which reaction may be best represented as follows, viz: $2C_6H_6 + (3O)O = 12CO_2 + 6H_2O$.

The hydrous carbonic-acid gas formed as above represented may and preferably will be augmented with watery vapor arising from the vessel b^{12} , and the temperature of the gas and vapors is lowered in the cooling apparatus, so that the said gas and vapor enter the corroding-compartment of the stack marked 2 at a substantially low temperature—that is, at about 95° Fahrenheit—and the lead fiber contained therein is energetically acted upon by the said gas and vapor and converted into the carbonate in a substantially short time without the formation of dry spots on the lead.

The surplus gas and vapor pass from the stack 2 through the pipe b^5 into the stack 3, thence by the pipe b^6 into the stack 4, thence by the pipe b^7 into the stack 5, from which it passes through the pipe b^3 into the chimney b^4 . The chimney b^4 is provided with a damper c^4 , by which the draft of the chimney may be controlled, and the rate of circulation of the gas and vapor through the stacks may be regulated. To assist the natural draft of the chimney, a gas or other burner c^5 is located within the chimney.

The inlet-pipe b^2 for the stack A (marked 2) is provided with a valve c^6 , by which the inlet to the corroding-compartment of the stack 2 may be closed, if desired—as, for instance, at night-time, the lamp or burner b^{15} being at such time extinguished.

During the day-time the lead fiber may be subjected to the corroding action of the hydrous carbonic-acid gas from the lamp or burner b^{15} and to the action of the carbonic-acid gas and vapor arising from the compartment a' of the corroding-stack, as at that time the lamp or burner b^{15} may be watched and controlled by a workman; but at night the lamp may be extinguished and the corroding action still maintained by the carbonic-acid gas and vapor arising from the compartment a' of the stack.

Instead of the particular form of cooling apparatus, any other desired form of appara-

tus may be employed—such, for instance, as shown in Fig. 3, wherein the gas is subjected to a shower of water from a rose or nozzle c^{10} on the end of a pipe c^{12} , connected to a suitable source of supply.

By means of the natural draft of the chimney a circulation of gas and vapor may be maintained without the use of machinery, such as a pump.

In practice the trays or supporting-frames covered with lead fiber are placed within the stacks and the doors securely closed. The lamp or burner b^{15} is started and the steam is admitted to the compartment a' to start the fermentation or combustion of the tan-bark, cow-dung, or like material, and the corrosion is continued until all the metallic lead is converted into the carbonate, which takes from four to five days, at the end of which time the trays holding the lead carbonate may be removed, and the carbonate, which is then of a creamy consistency, is dried and is ready to be used without subsequent treatment, such as grinding, as more commonly practiced.

We have herein shown a lamp by which the hydrous carbonic-acid gas is generated; but we do not desire to limit ourselves in this respect, as any desired form of burner may be used.

We claim—

1. The herein-described method of producing lead carbonate, which consists in confining finely-divided metallic lead in a closed chamber and subjecting the said finely-divided lead to the action of cooled hydrous carbonic-acid gas and an oxidizing agent, the said gas being generated by the combustion of refined petroleum or its described equivalent and cooled before entering the said chamber, substantially as described.

2. The herein-described method of producing lead carbonate, which consists in confining finely-divided metallic lead in a closed chamber and subjecting the said finely-divided lead to the corroding action of carbonic-acid gas and acetic-acid vapor generated in the said chamber and to the corroding action of hydrous carbonic-acid gas and acetic acid vapor generated outside the chamber and cooled before entering the said chamber, substantially as described.

3. The herein-described method of producing lead carbonate, which consists, first, in producing a purified hydrous carbonic-acid gas, consisting of carbonic-acid gas and chemically-produced water obtained by the combustion of refined petroleum or its herein-described equivalent; second, cooling the hydrous carbonic-acid gas to substantially the temperature herein specified, and, third, acting upon finely-divided metallic lead with the cooled hydrous carbonic acid in the presence

of an oxidizing agent, substantially as described.

4. The herein-described method of producing lead carbonate, which consists in confining finely-divided metallic lead in a closed chamber and subjecting the said finely-divided lead to the corroding action of carbonic-acid gas and acetic-acid vapor generated in the said chamber and to the corroding action of hydrous carbonic-acid gas generated outside the chamber and cooled before entering the said chamber, substantially as described.

5. In an apparatus for the production of lead carbonate, the combination, with a chamber or stack provided with a perforated partition to form compartments, of a gas inlet and outlet for one of said compartments, a series of removable trays in the upper compartment to support the lead to be converted, a fermenting substance in the lower compartment, and means to heat the lower of said compartments, substantially as described.

6. In an apparatus for the production of lead carbonate, the combination, with a stack or chamber to contain the lead, of an inlet-pipe for the said stack or chamber, a vessel to contain fluid, provided with a tube or pipe extended up through the vessel, of a burner to co-operate with and heat said vessel, and a cooling apparatus located in the inlet-pipe, substantially as described.

7. In an apparatus for the production of lead carbonate, the combination, with a stack or chamber to contain the lead, of an inlet-pipe for the said stack or chamber, a vessel to contain fluid, provided with a tube or pipe extended up through the vessel, of a burner to co-operate with and heat said vessel, and a cooling apparatus located in the inlet-pipe, an outlet-pipe for said chamber or stack, and a chimney with which the outlet-pipe communicates, substantially as described.

8. In an apparatus for the production of lead carbonate, the combination, with a stack or chamber to contain the lead, of an inlet-pipe for the said stack or chamber, a vessel to contain fluid, provided with a tube or pipe extended up through the vessel, of a burner to co-operate with and heat said vessel and a cooling apparatus located in the inlet-pipe, an outlet-pipe for said chamber or stack, and a chimney with which the outlet-pipe communicates, and a burner located in said chimney, to operate substantially as described.

In testimony whereof we have signed our names to this specification in the presence of two subscribing witnesses.

NORMAN K. MORRIS.

JOHN W. BAILEY.

Witnesses:

C. T. WARD,

R. T. ROYAL.