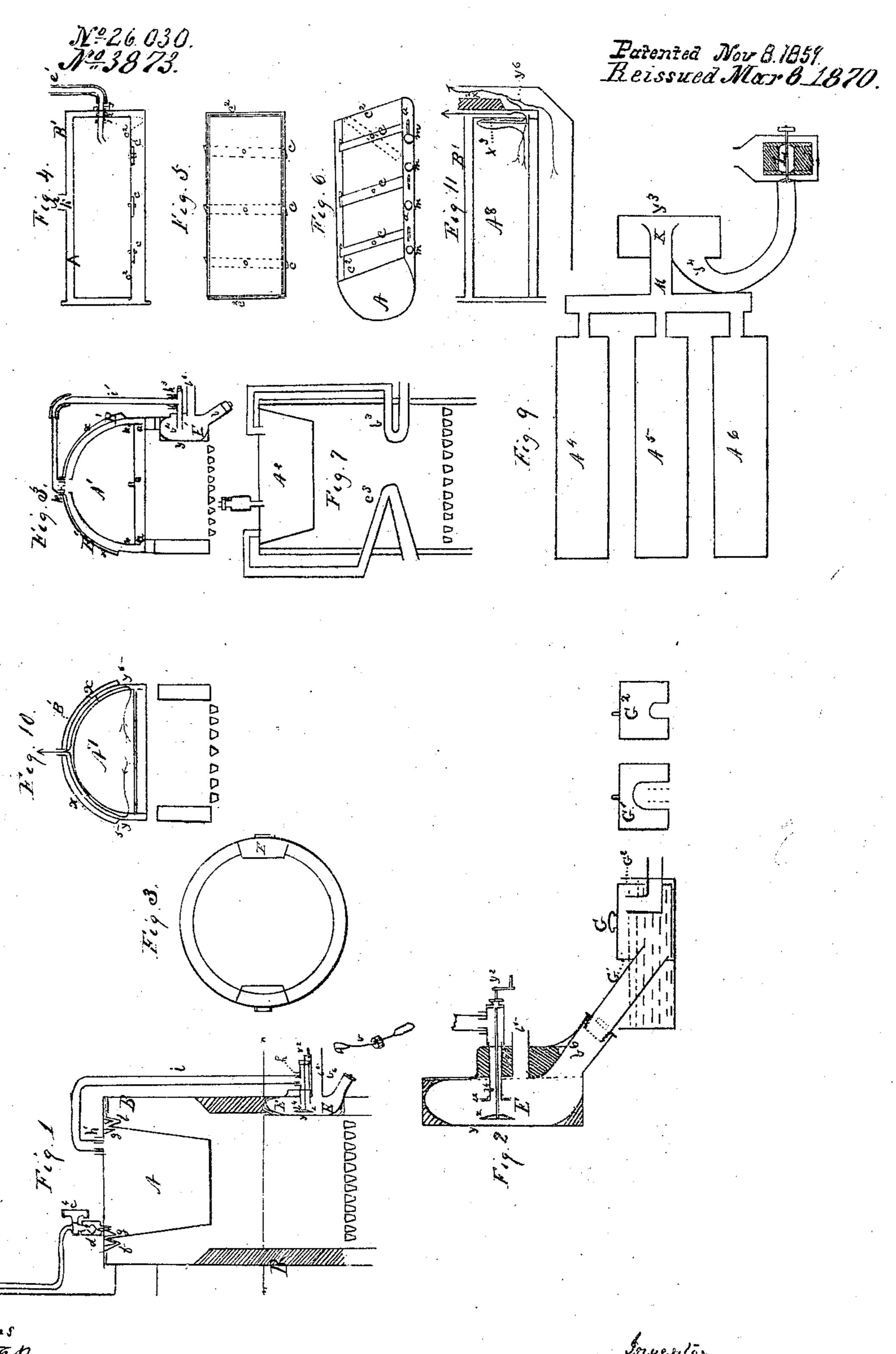
I.I.G. Gale.

Illuminating Gas Process.



Hetnesses Edwi F. Brown IAMenrill

Inventor VIII Lale

## UNITED STATES PATENT OFFICE.

LEONARD D. GALE, OF WASHINGTON, DISTRICT OF COLUMBIA.

## MANUFACTURE OF GAS.

Specification forming part of Letters Patent No. 26,030, dated November 8, 1859; Reissued March 8, 1870, No. 3,873.

To all whom it may concern:

Be it known that I, Leonard D. Gale, of the city and county of Washington, in the District of Columbia, have invented a new and improved method of treating rosin, wood, wood-tar, and all resinous, oily, or fatty bodies, using them as liquid or solid feed, in the manner hereinafter specified, or using any of the equivalents of these sev-10 eral substances for converting them into illuminating-gas; and I hereby declare the following to be a full and exact description thereof, reference being had to the accompanying drawings and references marked 15 thereon, which constitute a part of my

specification. The nature of the invention consists in treating any, or all of the above named substances or their equivalents, and, in fact, 20 all gas illuminating materials (except bitumen, bituminous coal and its distillates, which are made the subject of a separate and simultaneous application for letters patent,) by first converting them into vapor and then forcing their vapor into contact with a red hot surface substantially as herein after described. This involves two distinct and separate processes. The first requires only so much heat as will vaporize the materials without converting them into a permanent gas. It is therefore below a cherry red, and may be in the range of 500° to 800° Fahrenheit. The second part of the process transfers the vapor when generated to the decomposing chamber forcing it by the vis a tergo into contact with, or against a cherry red hot surface of iron or other material by which the vapor is decomposed into permanent gas. No permanent gas can be formed in the first part of the process, and in the second, none of the vapor can escape being converted to a per-

In the common methods described for making illuminating gas, the effort has been to convert the gas making material whatever it be, as quickly as possible directly to a permanent gas by heating the crude material at once to a cherry red heat, so that it may be decomposed into vapor and into gas at the same instant, without recognizing the intermediate or vapor state. In my method, the gas making material is first separated, taking from the crude, solid or liquid, matter all that is capable of being back, seen in Figs. 1, 2, and 3, designed to

manent illuminating gas, except by derange-

ment of the functions of the apparatus.

converted to gas, and retaining it in a uniform state of vapor till it is ready to be discharged into the decomposing chamber, where it deposits a little carbon and be- 60 comes a permanent gas. As the vapor all passes through the decomposing chamber, which is sustained at a cherry red heat, no part can escape decomposition into gas, and therefore there can be no appreciable 65 amount of gas tar. This view is substantiated by experiment.

In the ordinary mode of generating coal gas for our cities in the D form retorts, it is only the outer portion of the coal that is 70 heated to redness; the interior parts being generally below that temperature are only vaporized without ever being converted to permanent gas. Consequently when the product is cooled down the vapor is depos- 75 ited as coal tar.

From the above remarks it is quite evident that my process may be illustrated by a variety of different apparatus. It is only necessary to have one vessel to be 80 heated at a little above the temperature of boiling oil or to a temperature less than a red heat controllable within that range, and a second vessel or chamber that shall be kept at a cherry red heat.

In the accompanying drawings, Figure 1 represents a sectional elevation of a gas stove in which A is the vapor generator, and E the gas generating chamber. Fig. 2 is an enlarged view of the gas making chamber. 90 Fig. 3 is a sectional end view of a double 3 retort and gas back. Fig. 4 is a longitudinal sectional elevation of the 2 retort showing the mode of supplying it with liquid feed. Fig. 5 shows the movable bottom of 95 the inner retort. Fig. 6 shows a bottom view of the inner retort and the supporting braces (c, c, c,). It also exhibits vapor holes (m,) &c., which are sometimes used when the gas back is dispensed with.

Besides the above forms of retort there are a variety of modifications that I have used: One of these is seen in Fig. 7 where the U form tube  $c^3$  or  $b^3$  is used as the decomposing chamber. But the objection to this 105 is that the tube is liable to clog with a deposit of charcoal.

Fig. 9 exhibits several retorts A4, A5, A6, in a single bench for converting the gas fuel to vapor, the product being then forced 110 into a decomposing box K, or into a gas

be used where large quantities of gas are manufactured.  $y^3$  is the red hot surface. The reason for using the double a retort instead of a single one is, that the heat of the inner retort is more controllable. It is indispensable that the heat be not so high as a red heat. If it were a single retort some part of it might, and probably would be, heated to redness now and then, which would be fatal to my process. If the single retort could be certainly kept at a heat below redness then it is in some respects preferable to the double one. These retorts for converting vaporizable gas fuel into va-15 por may be made very large so as to contain several tons of material at a time, and discharge their vapor into a very large decomposing chamber, having a suitable capacity to convert the whole material to permanent gas as fast as generated. It is proper to say that the kettle retort A is more especially adapted to liquid feed of all kinds, and the double a retort is adapted to wood and all solid gas fuel, though it can be 25 used for liquid feed.

In all the drawings similar letters and

figures refer to the same devices.

B, B, is an ordinary cylinder sheet-iron stove lined with fire brick, except a small 30 space on one side where the fire brick has its place supplied by a hollow casting of iron E. The retort A, is chiefly designed to be supplied with fluid feed from the reservoir (e) controlled by the stopcock (c)35 which discharges its feed drop by drop through the tube (d) which is a metallic tube containing two diamond holes on opposite sides, and having an inner glass tube cemented in it by plaster of paris. By look-40 ing directly through the holes the rapidity of the drop can be seen.

(f) is a circular channel in the retort rim for holding fusible metal made of a mixture of 8 parts by weight of bismuth, 5 of lead 45 and 3 of tin, which melts at the heat of boiling water. But pure tin will answer the purpose. This melts at about 440° and the cover of the retort is generally hot enough

to keep the metal in a liquid state. (g, g) are flanges projecting from the

under side of the cover to prevent the vapor condensed on the same from being drawn up into the metal joints and so getting on the outside. (h') is also a cup metal joint

55 shown in section.

(i) is an iron pipe for conveying the vapor from A to E; (h) a cup joint of the

alloy aforesaid.

(b'') is an iron pipe projecting half way 60 through the chamber E, and discharges the vapor with considerable force against the red hot surface (y) of the gas back E, E.

 $(b^5)$  is the tube for discharging the illuminating gas formed in the chamber E. (z) is a rotary scraping blade designed to

clear off the carbon deposited on plate (y), and  $(y^2)$  is the crank by which it is worked; and when not worked, the shaft, which is surrounded by an air tight collar, is drawn back so that the sharp edge of the blade rests 70 against the end of tube  $(b^4)$ . Pipe  $(b^5)$ should be a third larger than  $(b^4)$ . As a substitute for the rotary shaft and scraping blade, a rod (V) and hand scraper is inserted in the opening  $(b^{\mathfrak{g}})$  and worked through 75 an air tight collar. The scraper should be worked every two hours to insure no incrustation. The charcoal is withdrawn through pipe  $(b^6)$ . A top view of the gas back is seen in Fig. 8 occupying the place 80 of one of the fire bricks, a convenient position to sustain the inner plate at a constant red heat.

The gas back E, E, constitutes a distinguishing feature of my discovery and inven- 85 tion, which is that any suitable hydrocarbon vapor generated at a temperature below a red heat, on being projected against a red hot surface and instantaneously withdrawn therefrom is converted into an illuminating 90 gas of a superior quality; and at the same

time produces little or no gas tar.

The operation is as follows: A fire of hard coal, coke, or hard wood is built on the grate till plate (y) is at a cherry red heat, 95 when kettle A will be hot enough to distil any fatty or resinous body or woody fiber. As the apparatus is designed for liquid feed which is charged in the reservoir, the feed is let on by carefully opening the stop cock 100  $(c^4)$  till the feed drops with a rapid succession of drops and the vapor generated passes through tube (i) and against plate (y) and after becoming permanent gas passes out through pipe  $(b^5)$  and after going through 105 condensing pipes is conducted to the gasometer, unless purification be necessary, in which case the usual processes are used. If wood be used as the gas fuel, then the D retort Fig. 3 is the best adapted for the pur- 110 pose. The mouth and fastening of this retort is constructed in the usual way, and not described here. The inner retort, made of thin cast iron, or of Russia sheet iron where it is only about  $2\frac{1}{2}$  feet long, and 10 to 12 115 inches wide, and 9 to 10 inches high, is laid upon its back, as in Fig. 6, and the braces (c)unlocked by partially rotating them on their centers, and the bottom removed, and the retort charged with wood, the bottom re- 120 placed and keyed down; the retort is then set upright and shoved into the outer retort and the door closed. If the fire be now raised, the gas will come over in a few minutes or as soon as the retort can be 125 heated to the requisite temperature, which is a low red heat.

It is recommended that as soon as the outer retort shows signs of incipient red heat, and the gas back is full cherry red 130

hot, then is the time to put in the inner retort, and close up the door as soon as possible. The joining of this retort is usually made air tight by luting the rim of the door 5 with clay and screwing it fast against the end of the retort. The same gas back used with the kettle retort is used with the D retort; but the size of the gas back must increase with the size of the retort. And it must be borne in mind, that in charging the retort with separate charges put in at intervals, as wood, and coal, and peat, and coal slate, the gas generated comes over with a rush, and then diminishes, so as to make 15 the pressure upon the internal part of the retort and gas back very unequal and, consequently, the jet through pipe (b4) and against plate (y) will also be different at different stages in working off the charge. This irregularity is corrected by making plate (y) and chamber E, sufficiently large to decompose all the vapor supplied when the current is the strongest. And if at any time it be ascertained that vapor still es-25 capes in the glass tube  $(b^5)$  and is deposited as coal tar, the remedy is to screw farther in pipe  $(b^4)$  so that the hollow disk  $(z^2)$  shall be brought nearer to the decomposing plate (y) so as to insure the bringing of all the 30 current of vapor within the sphere of a red heat.

In the manufacture of wood gas in domestic apparatus where there is difficulty in equalizing the pressure in the gas back, I 35 have dispensed with the gas back altogether, | using in place of it, the space on each side between the inner and the outer retort; this is done by making a sufficient number of holes (m) along (see Fig. 6) the sides of 40 the inner retort and near its bottom, through which the vapor generated within is discharged against the red hot sides of the outer retort, thence upward between the two retorts, and out by the aperture at  $(k^2)$ . In 45 this case where the gas back is dispensed with, the aperture in the dome of the inner retort is closed and holes (m) substituted therefor, and the upper part of the outer retort is prevented from being heated red hot 50 by the dome shield piece (x, x,) which may be of brick work or of iron. It should be remarked in this connection, that the modification last described requires a greater amount of heat than that where the gas 55 back is used, inasmuch as the bottom and a part of the sides must be maintained at a full red heat. All the remarks applicable | to this modification, apply equally to coal gas and gas from any solid substances. 30 This modification is also seen in Fig. 10, where the vapor generated in A<sup>7</sup> passes out in the direction of the arrows first in con-

tact with red hot plate  $y^5$ , which converts

it into gas, and thence out through the top

of the dome, which dome is protected from \

being heated to redness by shield x; the part,  $y^5$ , being fully exposed to the flue while all above is not so exposed.

Fig. 11 shows another modification of the apparatus last named where the vapor gen- 70 erated in  $A^8$  passes out over plate  $x^3$  through a horizontal slot in end plate of the inner retort and strikes against red hot plate  $y^{\mathfrak{o}}$ and escapes through the top in the direction of the arrow. The other arrow pursues the 75 course of the fire flue. In case of using liquids as feed which are liable to froth up, such as crude coal distillates or other bituminous liquids which may perchance overflow into the gas back and clog pipe,  $b^5$ , 80 I dispense with the use of  $b^5$ , as a gas discharge and allow the gas as well as the drip to pass out together through  $b^{c}$ , and the condenser to the gasometer. The advantages of this use of  $b^{\overline{6}}$  are a mere ca- 85 pacious discharge opening, for the gas and the solid and liquid refuse all in one vessel. That is to say, the gas and the vapor and other refuse are passed out through the drip and washing apparatus, G, (Fig. 2) and 90 thence to the gasometer.

Of the various forms of retort for working my process described above that shown in Figs. 10, or 11, I regard as best for efficiency and cheapness. That in Fig. 1, does 95 its work well, is useful for working very small quantities. The gas it makes is of

superior quality. I am aware that James Hansor obtained an English patent in 1856 in which it pro- 100 posed to convert the materials to vapor in one retort and then pass the vapors so generated into a second retort to convert them to permanent gas. But there was no provision to force the vapor against a red hot 105 surface and to remove the same instantaneously from the red hot chamber to prevent further decomposition which inevitably occurs in the absence of such provisions. am aware that Wm. P. McConnill obtained 110 an American patent for what was called reheating the gas, &c., but as no provision was made to avoid generating permanent gas in the first heating, nor for removing the gas as fast as generated out of the reach 115 of the decomposing heat it does not interfere with my invention. I am also aware that many inventions have been made proposing to improve illuminating gas by passing such gas after being made, through heated flues 120 pipes, or chambers heated to redness; not being aware of the fact that gas once made cannot be heated to redness without suffering decomposition and losing its illuminating power. All these devices I disclaim.

Having fully described my process for manufacturing illuminating gas, and several of the ways in which it may be carried into operation, as applied to woods, resins, fats, oils, tars, and other gas making substances 130 except bitumen, bituminous coal and their distillates,

What I claim in the present application as my invention, and desire to secure by Let5 ters Patent, is—

The treatment of all woody, resinous, and fatty bodies as well as all tarry matter except bitumen, bituminous coal and their distillates, by first converting the volatile portions to a vapor at a temperature below a

cherry red heat, and afterward forcing the vapor so generated into contact with a red hot surface, in such manner that the gas generated thereby may be instantly removed from said heated surface and thus be pre- 15 vented from further decomposition.

L. D. GALE.

Witnesses:

Edw. F. Brown, J. H. Merrill.